

Supporting Information

Theoretical Insights into the Hydrogen Evolution Reaction on VGe₂N₄ and NbGe₂N₄ Monolayers

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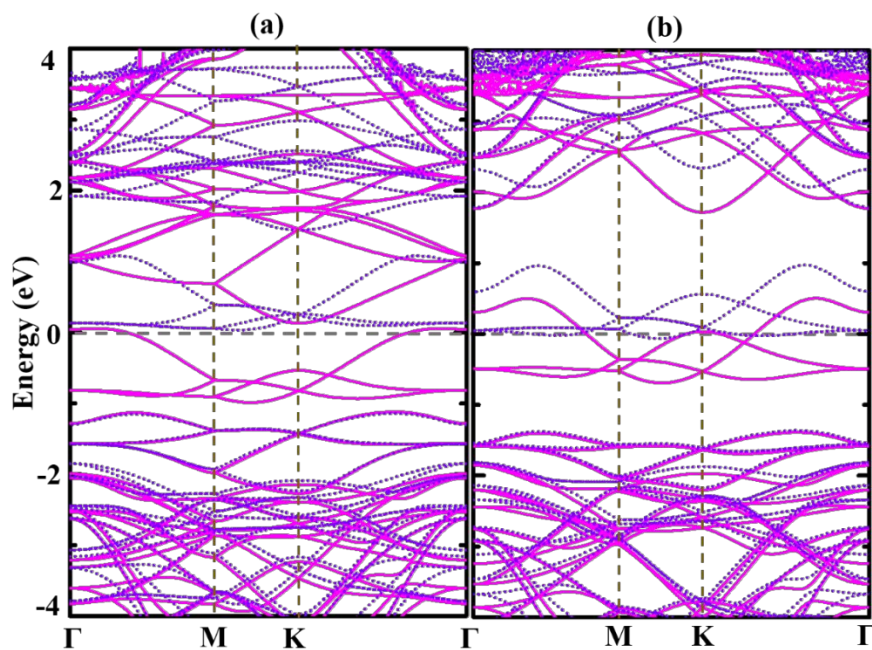


Figure S1 The band structure of (a) VGe₂N₄ and (b) NbGe₂N₄ monolayers within PBE functional. Spin-up and spin-down bands are represented by solid magenta and dotted violet lines respectively.

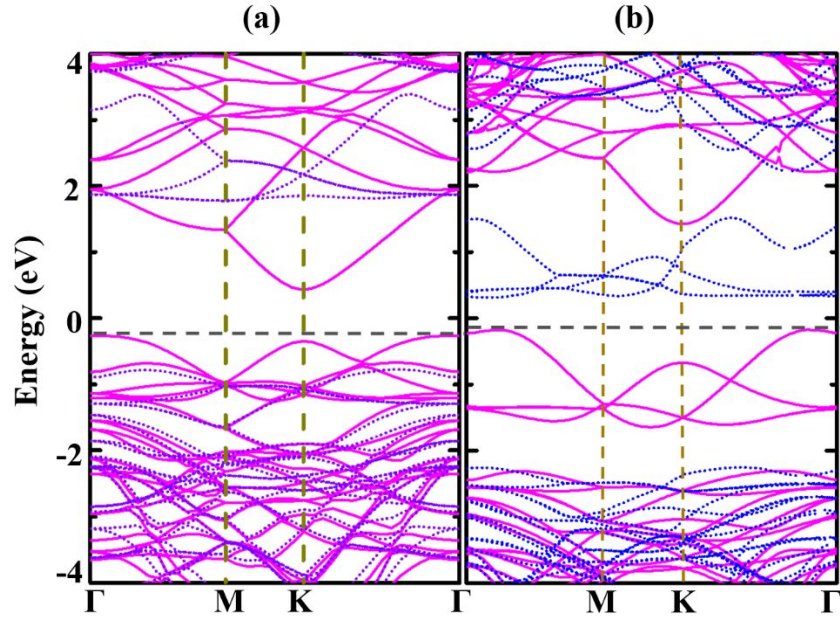


Figure S2 HSE band structure of (a) VGe_2N_4 and (b) NbGe_2N_4 monolayers. Spin-up and spin-down bands are represented by solid magenta and dotted violet lines respectively.

Table S1 The values of band gaps for VGe_2N_4 and NbGe_2N_4 monolayers for different U and HSE functional. Up, down, and total indicate the value for spin-up, spin-down, and total band gaps. For VGe_2N_4 and NbGe_2N_4 , the HSE total band gap matches with $U=4.5$ eV and $U=4$ eV respectively.

U (eV)	Band Gap (eV)					
	VGe_2N_4			NbGe_2N_4		
	Up	Down	Total	Up	Down	Total
2	0.01	1.64	0.01	0.01	1.64	0.01
3	0.02	1.82	0.02	1.15	1.77	0.24
3.5	0.26	1.88	0.26	1.14	1.82	0.33
4	0.49	1.95	0.49	1.12	1.87	0.44
4.5	0.67	2.10	0.67	1.07	1.90	0.55
5	0.63	2.03	0.63	0.89	1.95	0.62
HSE	0.69	2.55	0.69	1.56	2.57	0.48

Table S2 The adsorption Hydrogen Gibbs free energy ΔG_{H^*} for N-vacancy VGe_2N_4 surface doped by different atoms.

Doped atom (X)	ΔG_{H^*} (eV) at different Position of H		
	H above N	H above Ge	H above X
B	-0.95	-1.60	-1.60
C	-1.04	-2.31	-2.31
O	0.72	2.46	1.74
P	-0.38	1.24	1.29
S	0.71	1.30	1.44

Structural, electronic and catalytic properties of P-doped $NbGe_2N_4$

Optimized structure of P atom doped at the N-vacancy of the monolayer is shown in the Figure S3 (a) and (b). The P atom is situated slightly above the outermost layer formed by N atoms. The optimized N-P bond length is found to be 2.27 Å. To check the HER activity of the doped surface in the basal plane, we have considered three different sites for hydrogen adsorption: (i) above N (H_N), (ii) above P (H_P) and (iii) above Ge (H_Ge) atom. Among the three different sites, for H_Ge case, a large positive value for binding energy obtained implying there is no interaction between H and the doped surface. Through Bader, we have found that there is almost no charge transfer between H and surface. Hence we have not proceeded further for details calculations of Ge site. When H is placed above N atom, the binding energy is found to be -0.13 eV whereas for P site, the value is equal to 0.42 eV. Thus ΔG_{H^*} values are 0.11 eV and 0.66 eV for N and P sites respectively. This indicates that N site can be considered an active site for HER. The equivalent H-N and H-P bond length was found to be 1.02 Å and 1.42 Å respectively. To get more insight, we have calculated the amount of charge transfer between H atom and the doped-surface through Bader charge analysis. For case of N-site, a considerable amount of charge transfer (0.49e) from H whereas a small charge (0.28e) is transferred towards H for P-site. The charge density difference diagrams for both the sites are shown in Figure S3 (c) and (d) to understand the charge

redistribution after H adsorption. In case N-site (Figure S3 (c)), a large charge depletion region is observed around H atom and some of the charges are accumulated near the Nb atoms just below it. But the scenario is different in case of P-site as a small amount of charge get accumulated near H indicating very weak interaction (Figure S3 (d)). Further, we have plotted projected density of states (PDOS) of doped monolayer and H-adsorbed above the doped-monolayer (Figure S4) to reveal the electronic structure of H adsorption at different sites. The changes in the DOS near the Fermi level are clearly observed due to H adsorption (Figure S4 (b) and (c)).

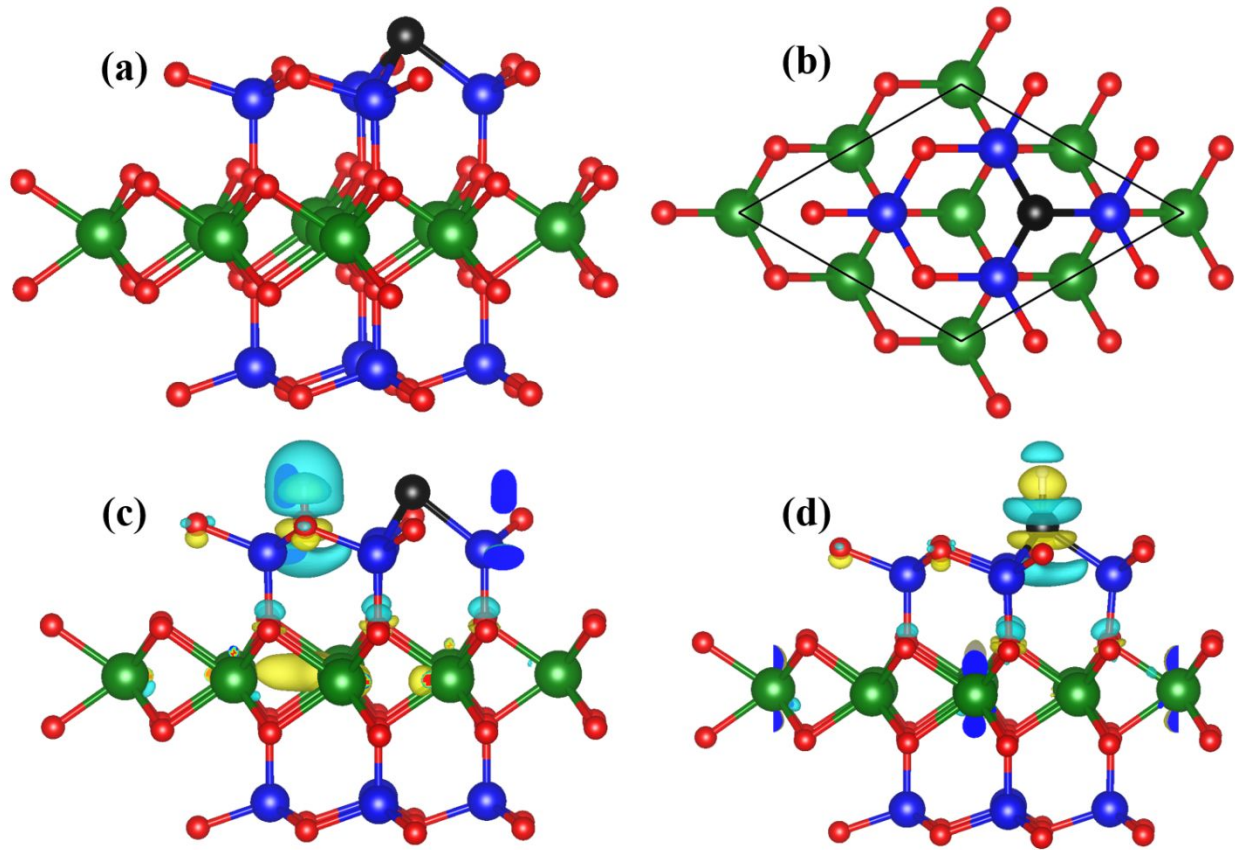


Figure S3 Optimized structure (a) side (b) top view of P-doped NbGe₂N₄ monolayer. The charge density difference for (c) H adsorbed above N atom (d) H adsorbed above P atom of the doped monolayer (iso-surface = 0.005 e/Å³). Nb, Ge, N, C and H atoms are represented by green, blue, red, black and white spheres, respectively. Here P atom is doped in place of N-vacancy of the monolayer. The charge accumulation and depletion regions are represented by yellow and cyan colors, respectively.

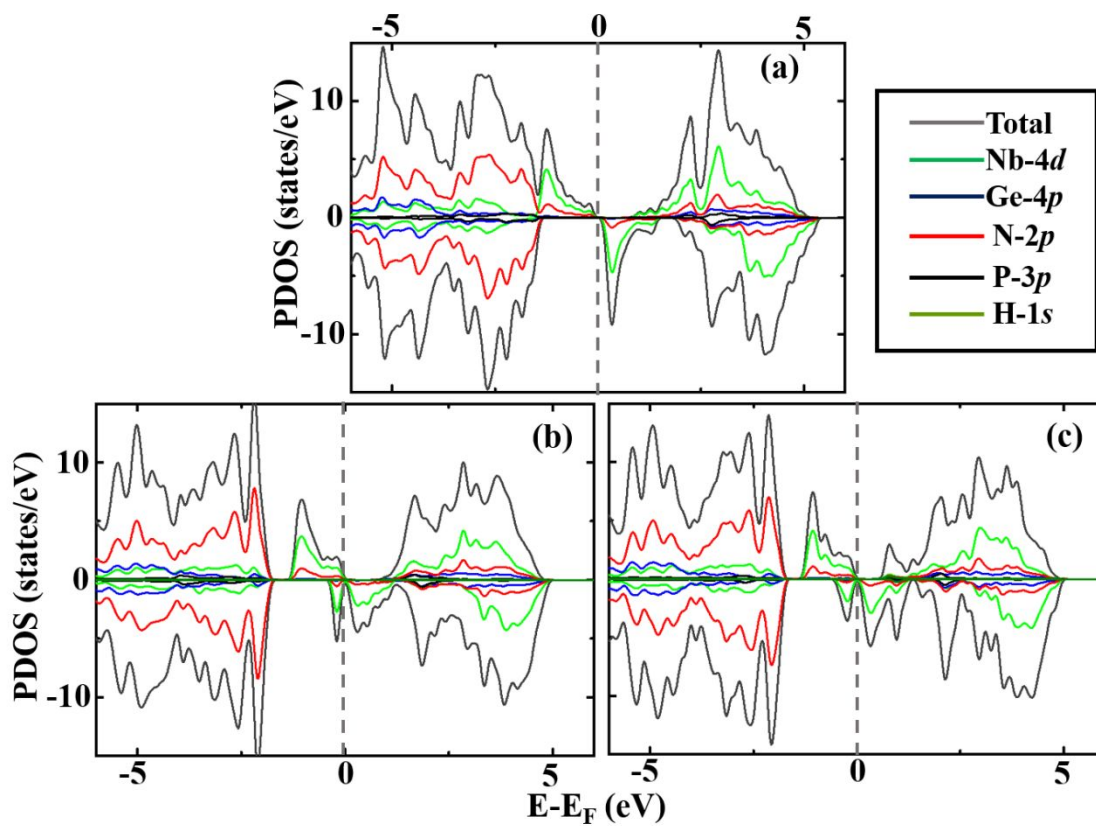


Figure S4 Projected density of states (PDOS) of (a) P-doped at N-vacancy of NbGe₂N₄ monolayer (b) H-adsorbed at N-site (c) H-adsorbed at P-site. Fermi level is set at 0 eV.

Electronic structures under biaxial strains:

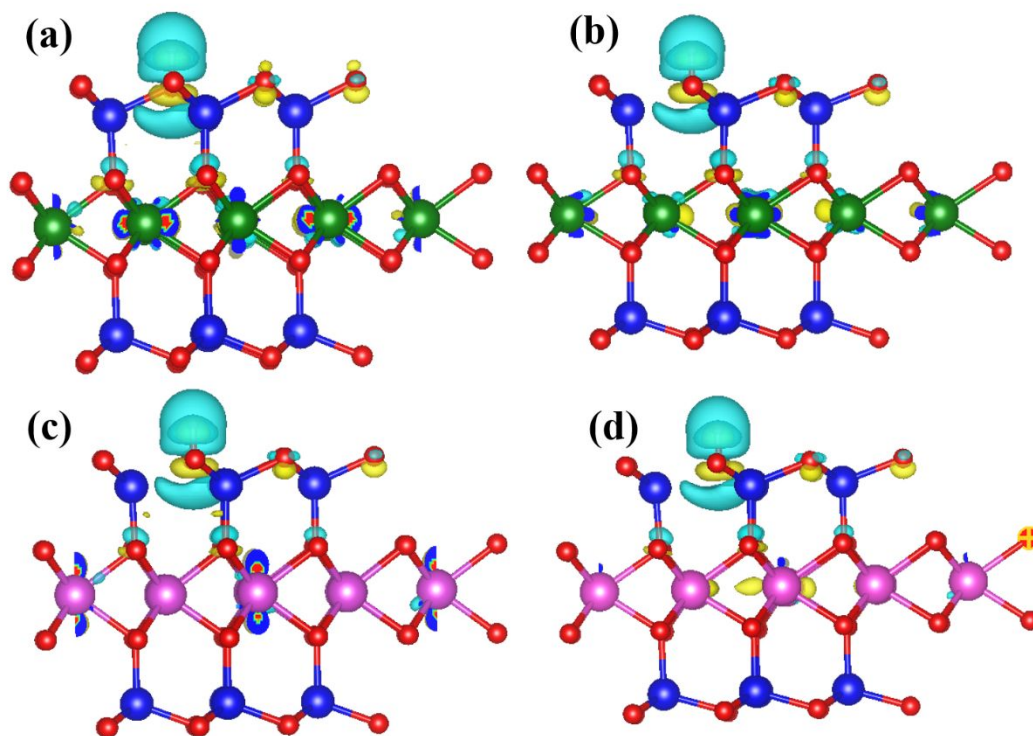


Figure S5 Three-dimensional charge density difference plots for VGe₂N₄ at (a) -3 % (b) +3% strains (upper panel), for NbGe₂N₄ monolayers at (c) -3% and (d) +3% strain (lower panel). V, Nb, Ge, and N atoms are represented by green, magenta, blue, and red spheres respectively. (Iso-surface is 0.005 e/Å³). Yellow and cyan color represent electron accumulation and depletion regions respectively.

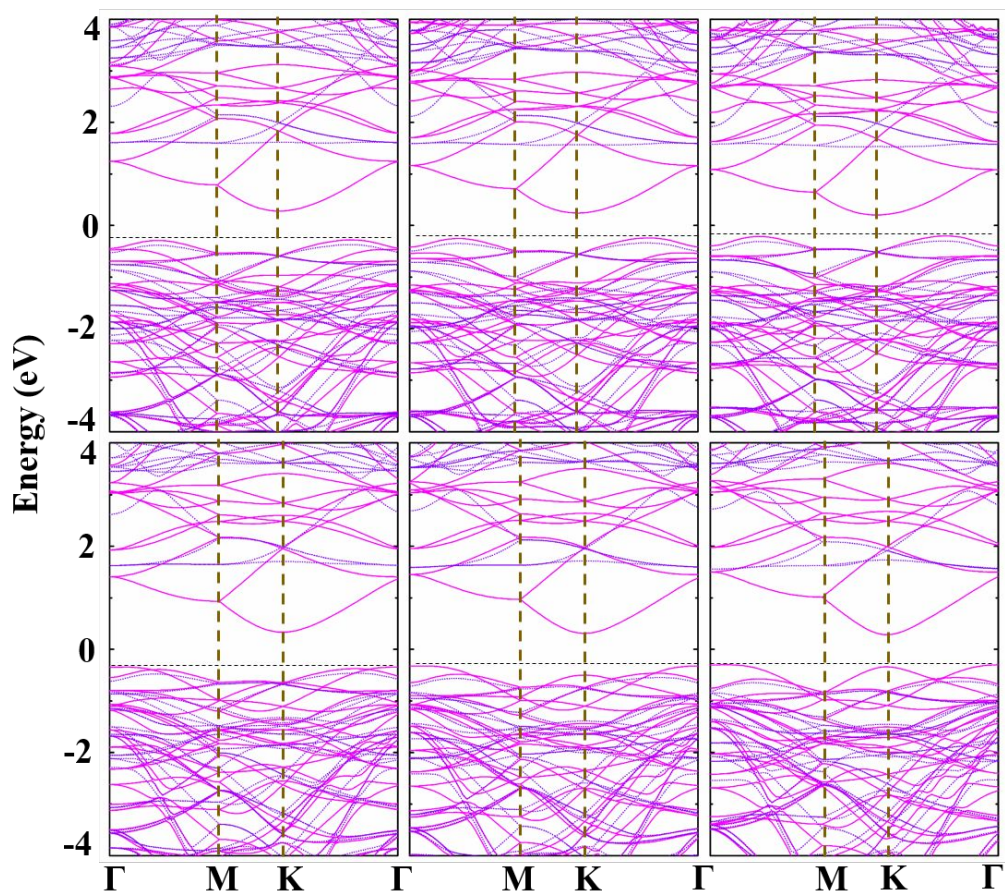


Figure S6 Electronic band structures of VGe₂N₄ monolayer under biaxial strains. Upper panel represents the bands under tensile strains from 1% to 3% while going from left to right. Lower panel represents the bands under compressive strains from -1% to -3% while going from left to right.

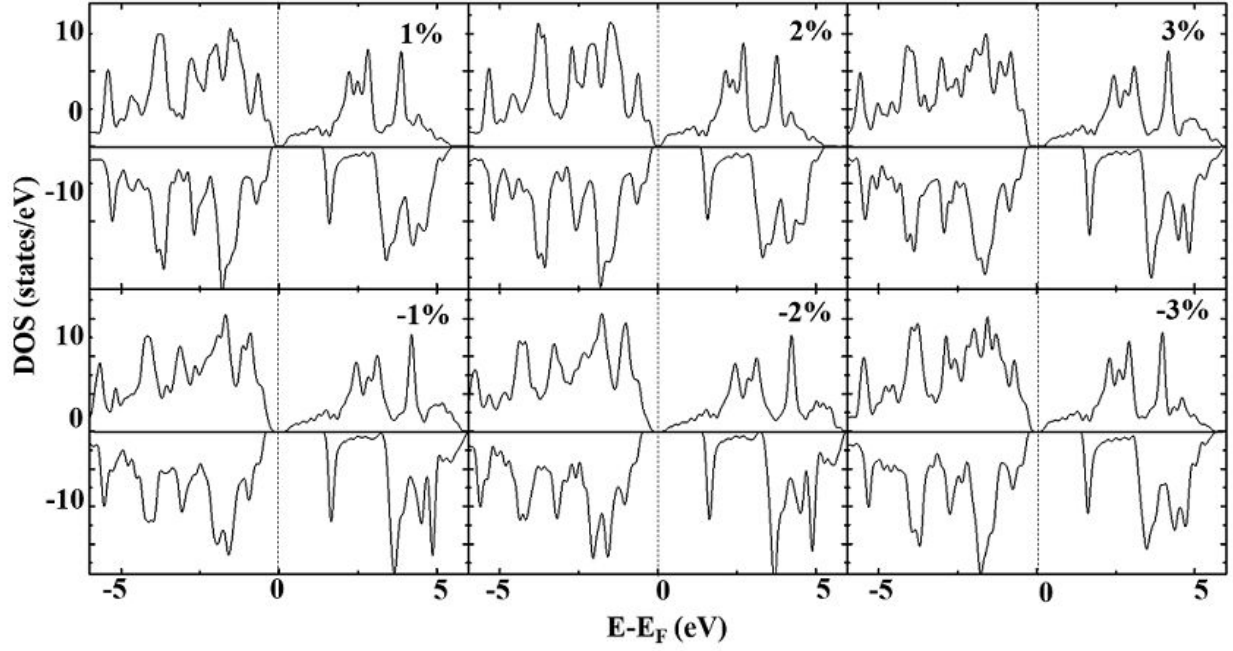


Figure S7 Total density of states (DOS) of VGe_2N_4 monolayer under biaxial strains. Upper panel (tensile strains) and lower panel (compressive strains).

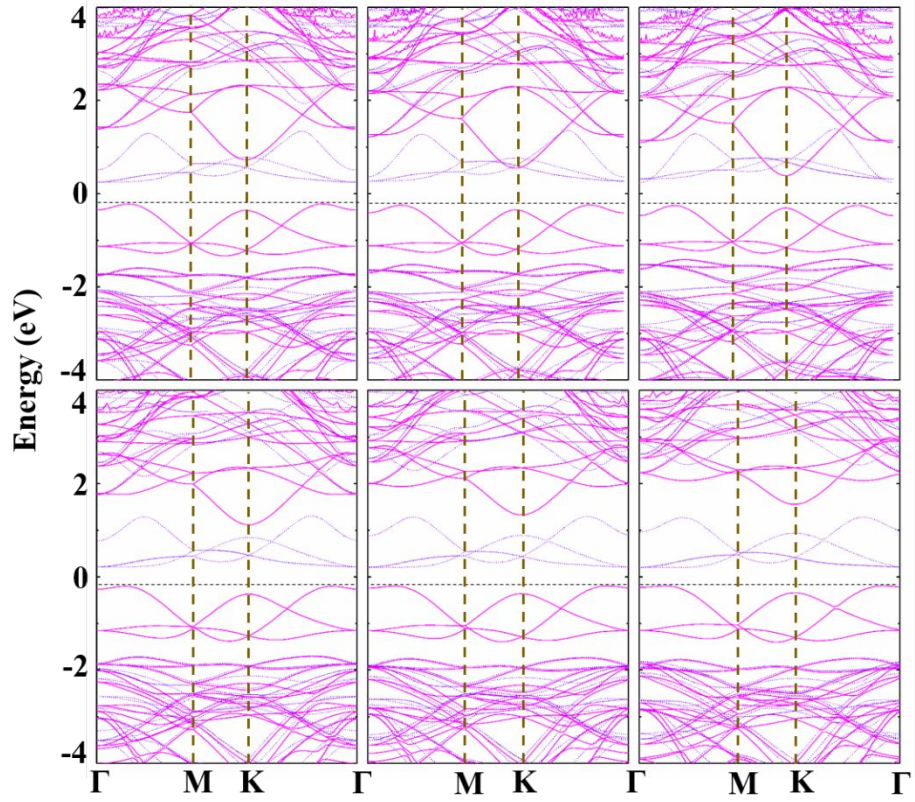


Figure S8 Electronic band structures of NbGe_2N_4 monolayer under biaxial strains. Upper panel represents the bands under tensile strains from 1% to 3% while going from left to right. Lower

panel represents the bands under compressive strains from -1% to -3% while going from left to right.

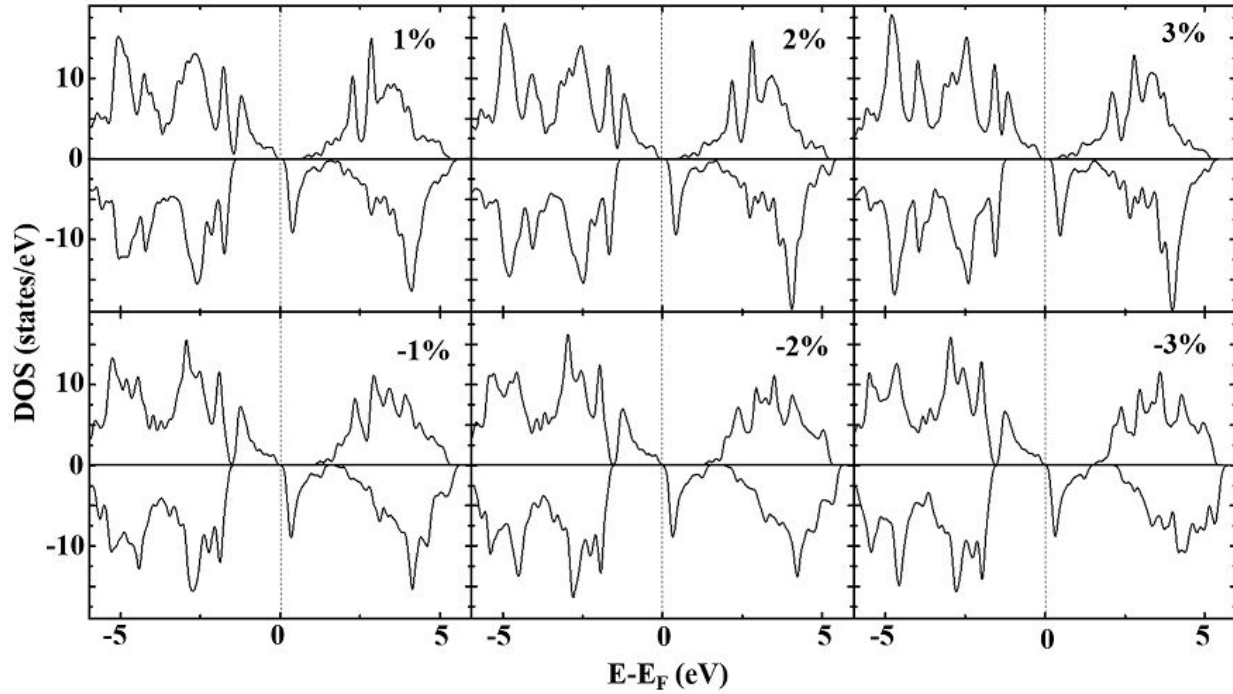


Figure S9 Total density of states (DOS) of NbGe₂N₄ monolayer under biaxial strains. Upper panel (tensile strains) and lower panel (compressive strains)

Table S3 The values of band gaps for VGe₂N₄ and NbGe₂N₄ monolayers for different biaxial strains. Up, down, and total indicate the value for spin-up, spin-down, and total band gaps.

Strain (%)	Band Gap (eV)					
	VGe ₂ N ₄			NbGe ₂ N ₄		
	Up	Down	Total	Up	Down	Total
1	0.56	1.96	0.56	0.95	1.82	0.47
2	0.48	1.89	0.48	0.79	1.77	0.51
3	0.40	1.82	0.40	0.63	1.63	0.49
-1	0.67	2.08	0.67	1.31	1.90	0.41
-2	0.63	2.13	0.63	1.52	1.92	0.40
-3	0.58	2.17	0.58	1.74	1.93	0.39

