

Electronic Supplementary Information

How End-Capped Acceptors Regulate the Photovoltaic Performance of the Organic Solar Cells: A Detailed Density Functional Exploration of Their Impact on the A-D- π -D-A Type Small Molecular Electron Donors

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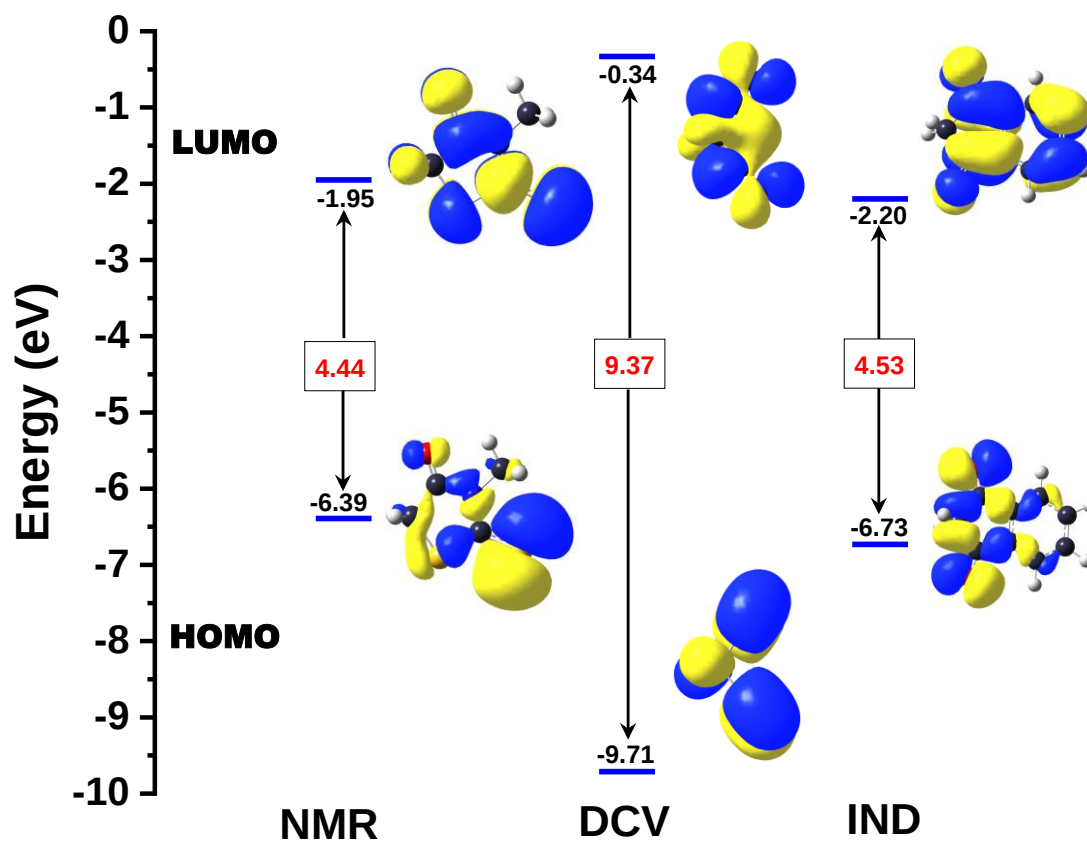


Figure S1. DFT computed HOMO, LUMO, HOMO-LUMO gap and isodensity plots of the bare acceptors obtained at the DFT/B3LYP/6-311G(d, p) level of theory.

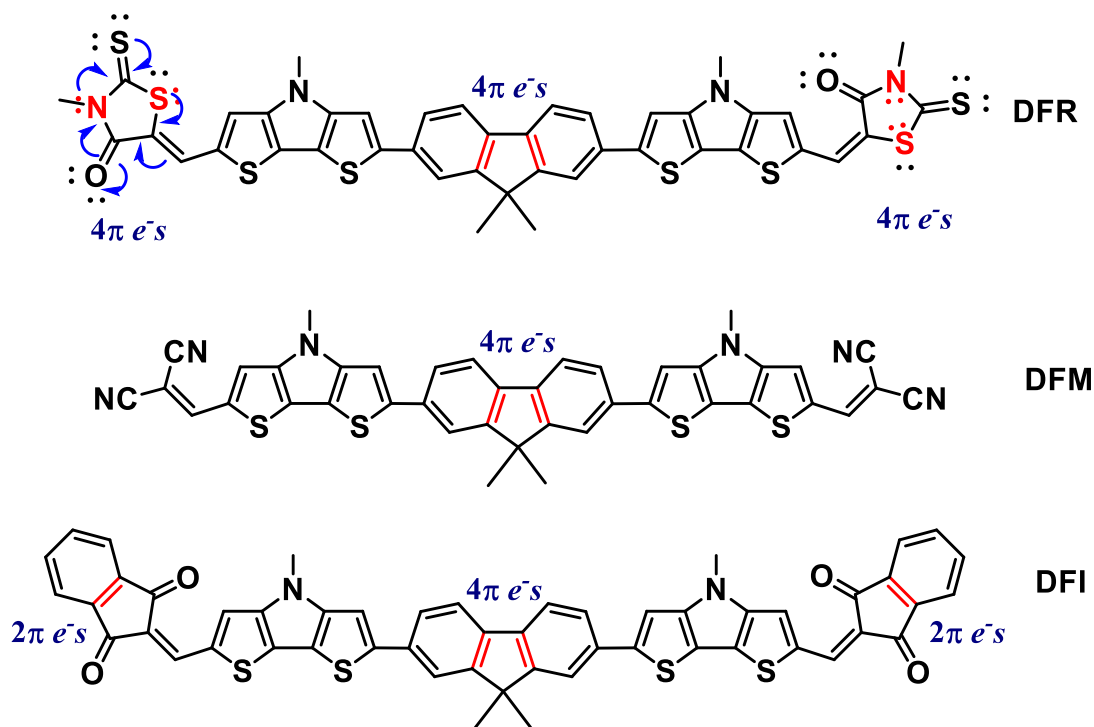


Figure S2. Existence of anti-aromatic units in the SMEDs. (Indicated red in color)

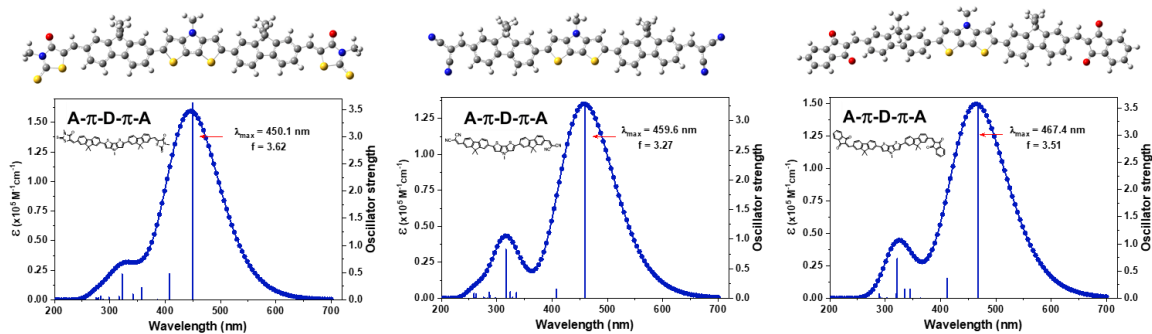


Figure S3. Optimized geometries, simulated absorptions and oscillator strength of the A- π -D- π -A framework using DFT/B3LYP/6-311G(d, p)/CPCM(CH₂Cl₂)/TDDFT/M062X/6-311G(d, p)/CPCM(CH₂Cl₂) level of theory.

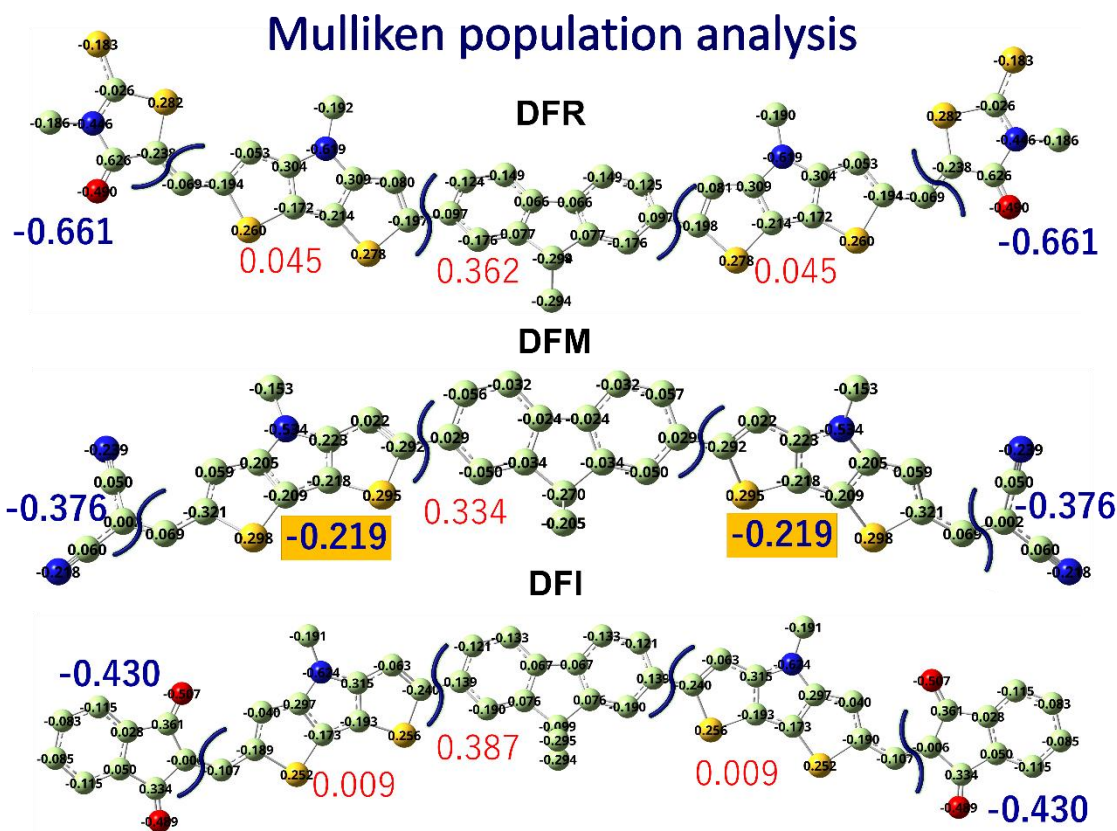


Figure S4. Mulliken population analysis on the grouped segments (D, π and A) of the SMEDs obtained from DFT/B3LYP/6-311++G (d, p) level of theory.

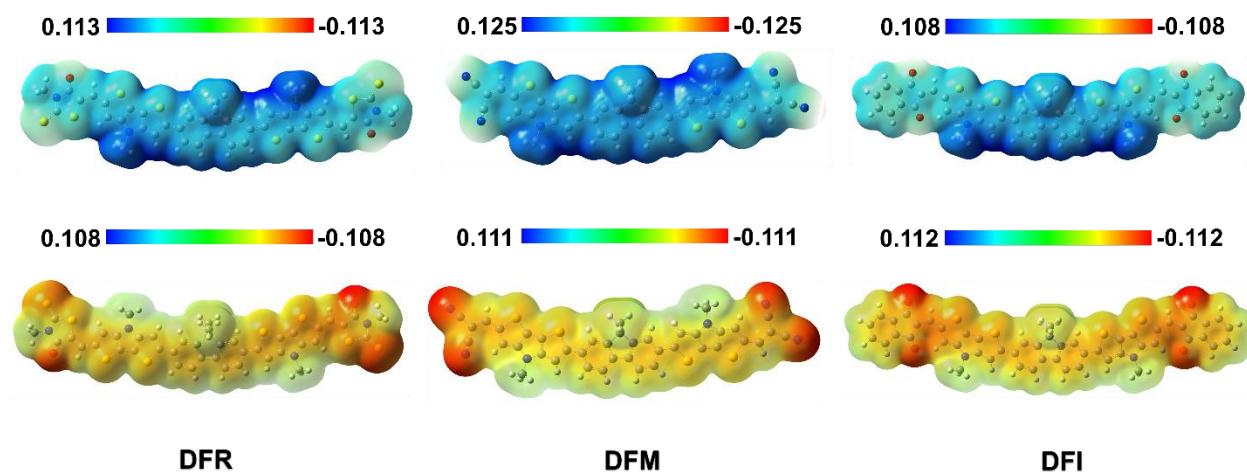


Figure S5. Variation of electrostatic environment revealed from MESP plots of the SMEDs at the oxidized ($-1e^-$) and reduced ($+1e^-$) states.

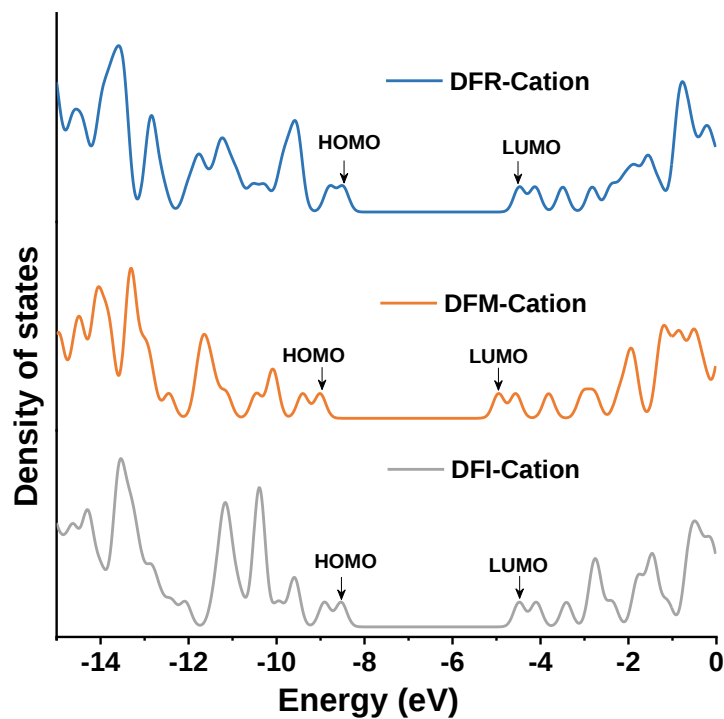


Figure S6 Variation of energy levels in terms of DOS population corresponding to the HOMO and LUMO of the SMEDs at the oxidized ($-1e^-$) state.

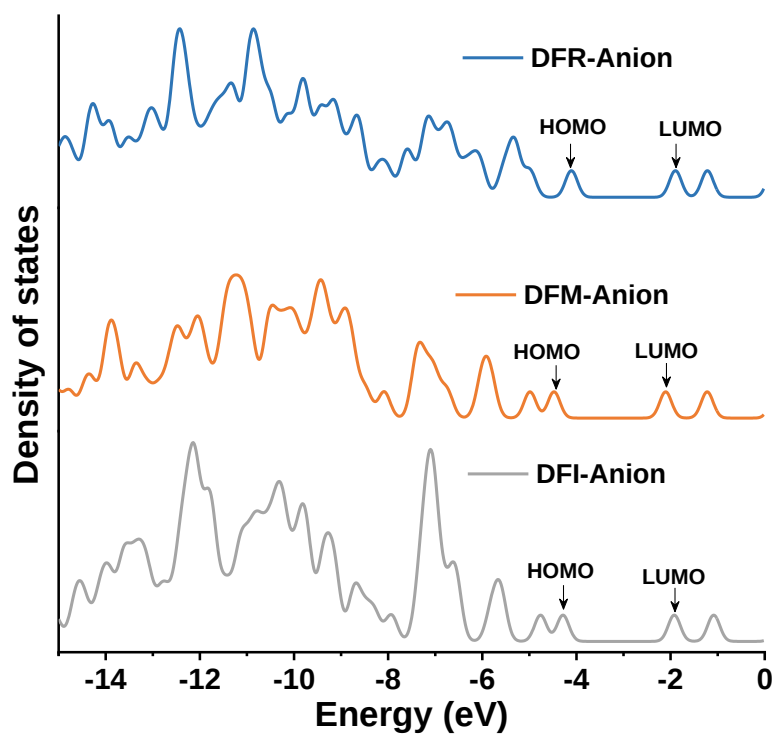


Figure S7. Variation of energy levels in terms of DOS population corresponding to the HOMO and LUMO of the SMEDs at the reduced ($+1e^-$) state.

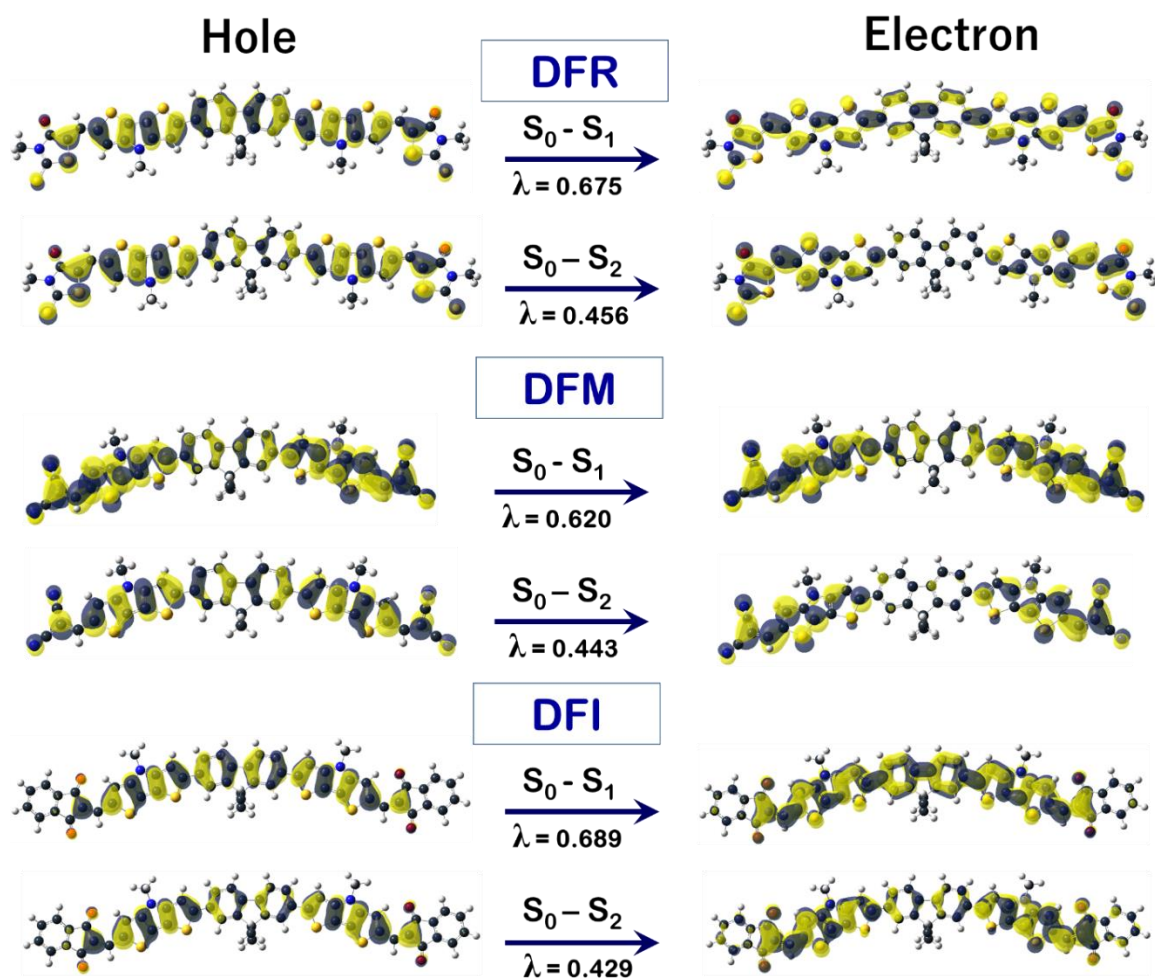


Figure S8. Natural transition orbital (NTOs) analysis corresponding to S_0-S_1 and S_0-S_2 transitions.

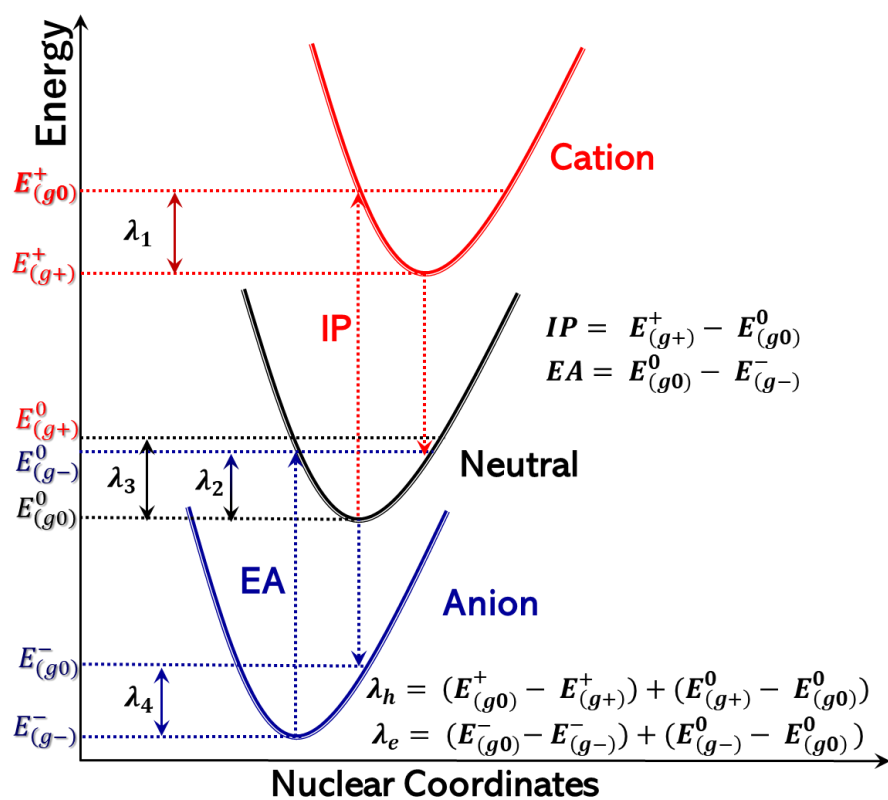


Figure S9. Schematic diagram represents the computational methodology for reorganization energy calculations.

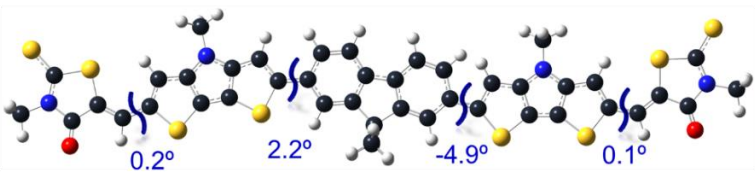
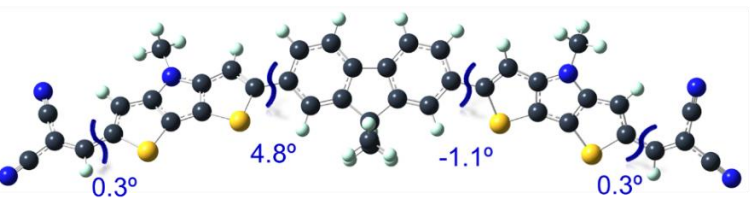
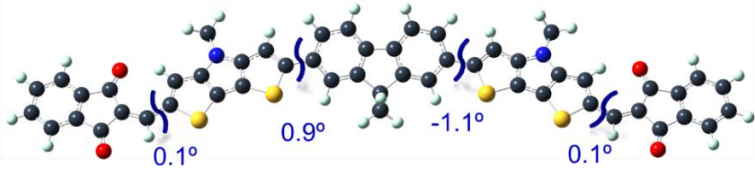
	Cation-Optimized Geometry
DFR	
DFM	
DFI	

Figure S10. Optimized geometry of SMEDs with selected dihedral angles in their first oxidized (cationic) state.

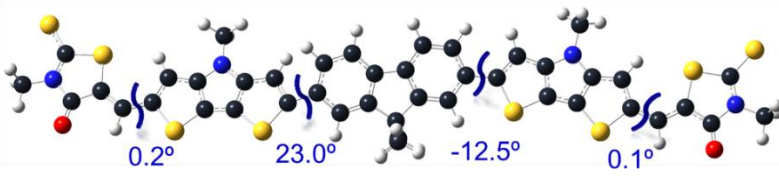
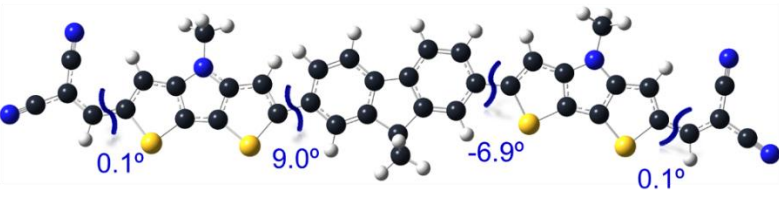
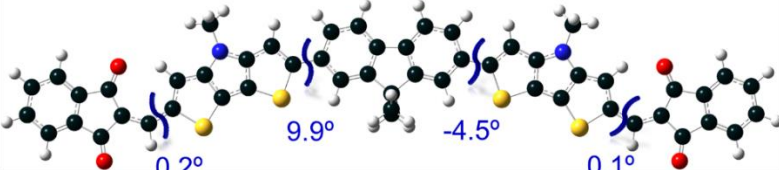
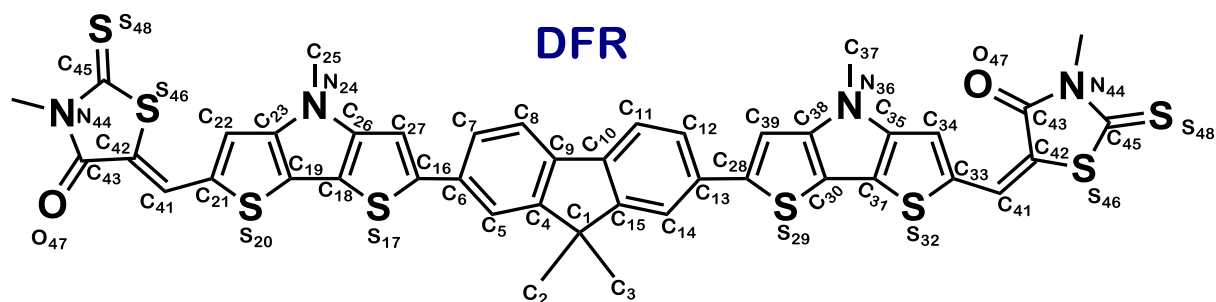
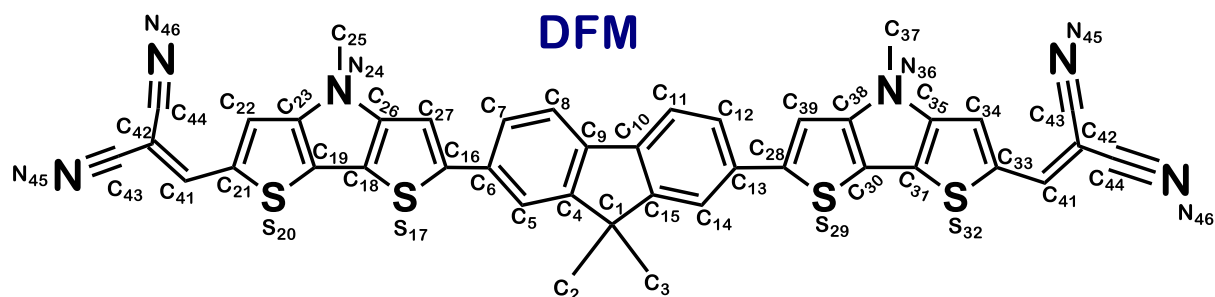
	Anion-Optimized Geometry
DFR	
DFM	
DFI	

Figure S11. Optimized geometry of SMEDs with selected dihedral angles in their first reduced (anionic) state.



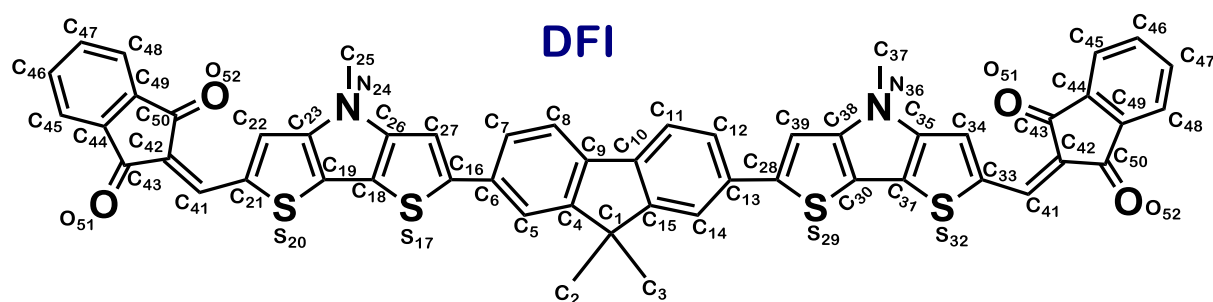
DFR	Neutral	Cation	Anion	DFR	Neutral	Cation	Anion
C ₁ -C ₂	1.544	1.541	1.545	C ₂₆ -C ₂₇	1.414	1.397	1.404
C ₁ -C ₃	1.544	1.541	1.545	C ₂₇ -C ₁₆	1.379	1.394	1.388
C ₁ -C ₄	1.529	1.527	1.529	C ₁₃ -C ₂₈	1.463	1.440	1.459
C ₄ -C ₅	1.383	1.376	1.382	C ₂₈ -S ₂₉	1.774	1.784	1.781
C ₅ -C ₆	1.411	1.418	1.416	S ₂₉ -C ₃₀	1.729	1.733	1.740
C ₆ -C ₇	1.408	1.422	1.417	C ₃₀ -C ₃₁	1.406	1.392	1.401
C ₇ -C ₈	1.389	1.379	1.386	C ₃₁ -S ₃₂	1.729	1.731	1.730
C ₈ -C ₉	1.395	1.405	1.400	S ₃₂ -C ₃₃	1.782	1.781	1.788
C ₄ -C ₉	1.408	1.417	1.409	C ₃₃ -C ₃₄	1.391	1.395	1.396
C ₉ -C ₁₀	1.461	1.441	1.456	C ₃₄ -C ₃₅	1.402	1.397	1.398
C ₁₀ -C ₁₁	1.395	1.404	1.398	C ₃₅ -N ₃₆	1.385	1.380	1.383
C ₁₁ -C ₁₂	1.388	1.380	1.389	N ₃₆ -C ₃₇	1.451	1.459	1.455
C ₁₂ -C ₁₃	1.408	1.418	1.410	N ₃₆ -C ₃₈	1.382	1.378	1.393
C ₁₃ -C ₁₄	1.411	1.420	1.414	C ₃₈ -C ₃₉	1.414	1.397	1.412
C ₁₄ -C ₁₅	1.383	1.375	1.381	C ₃₉ -C ₂₈	1.462	1.394	1.381
C ₁₅ -C ₁₀	1.408	1.417	1.412	C ₂₁ -C ₄₁	1.423	1.422	1.415
C ₁₅ -C ₁	1.529	1.528	1.529	C ₄₁ -C ₄₂	1.359	1.361	1.367
C ₆ -C ₁₆	1.463	1.442	1.451	C ₃₃ -C ₄₁	1.421	1.422	1.419
C ₁₆ -S ₁₇	1.774	1.783	1.789	C ₄₂ -C ₄₃	1.415	1.420	1.420
S ₁₇ -C ₁₈	1.737	1.734	1.741	C ₄₃ -N ₄₄	1.473	1.462	1.462
C ₁₈ -C ₁₉	1.406	1.392	1.348	N ₄₄ -C ₄₅	1.368	1.359	1.359
C ₁₉ -S ₂₀	1.729	1.731	1.743	C ₄₅ -S ₄₆	1.777	1.756	1.766
S ₂₀ -C ₂₁	1.782	1.781	1.806	C ₄₂ -S ₄₆	1.648	1.653	1.663
C ₂₁ -C ₂₂	1.391	1.395	1.400	C ₄₃ -O ₄₇	1.213	1.213	1.219
C ₂₂ -C ₂₃	1.402	1.398	1.408				
C ₂₃ -N ₂₄	1.385	1.379	1.384				
N ₂₄ -C ₂₅	1.451	1.459	1.453				
N ₂₄ -C ₂₆	1.382	1.379	1.386				

Figure S12. Optimized geometrical coordinates of DFR at the neutral, cation and anionic states.



DFM	Neutral	Cation	Anion	DFM	Neutral	Cation	Anion
C ₁ -C ₂	1.544	1.545	1.544	N ₂₄ -C ₂₅	1.453	1.459	1.455
C ₁ -C ₃	1.544	1.545	1.544	N ₂₄ -C ₂₆	1.380	1.377	1.381
C ₁ -C ₄	1.528	1.527	1.530	C ₂₆ -C ₂₇	1.413	1.397	1.404
C ₄ -C ₅	1.383	1.380	1.378	C ₂₇ -C ₁₆	1.379	1.395	1.388
C ₅ -C ₆	1.409	1.421	1.419	C ₁₃ -C ₂₈	1.463	1.440	1.448
C ₆ -C ₇	1.409	1.423	1.415	C ₂₈ -S ₂₉	1.774	1.783	1.788
C ₇ -C ₈	1.388	1.378	1.386	S ₂₉ -C ₃₀	1.726	1.732	1.741
C ₈ -C ₉	1.396	1.406	1.400	C ₃₀ -C ₃₁	1.403	1.392	1.393
C ₄ -C ₉	1.407	1.418	1.414	C ₃₁ -S ₃₂	1.726	1.726	1.735
C ₉ -C ₁₀	1.460	1.438	1.450	S ₃₂ -C ₃₃	1.786	1.782	1.804
C ₁₀ -C ₁₁	1.396	1.405	1.402	C ₃₃ -C ₃₄	1.395	1.398	1.411
C ₁₁ -C ₁₂	1.388	1.378	1.384	C ₃₄ -C ₃₅	1.396	1.393	1.390
C ₁₂ -C ₁₃	1.409	1.421	1.418	C ₃₅ -N ₃₆	1.385	1.380	1.385
C ₁₃ -C ₁₄	1.409	1.421	1.417	N ₃₆ -C ₃₇	1.453	1.459	1.454
C ₁₄ -C ₁₅	1.383	1.374	1.380	N ₃₆ -C ₃₈	1.380	1.376	1.382
C ₁₅ -C ₁₀	1.407	1.419	1.412	C ₃₈ -C ₃₉	1.413	1.397	1.403
C ₁₅ -C ₁	1.528	1.527	1.530	C ₃₈ -C ₂₈	1.463	1.394	1.390
C ₆ -C ₁₆	1.463	1.441	1.448	C ₂₁ -C ₄₁	1.418	1.419	1.404
C ₁₆ -S ₁₇	1.774	1.782	1.787	C ₄₁ -C ₄₂	1.379	1.376	1.395
S ₁₇ -C ₁₈	1.738	1.732	1.741	C ₃₃ -C ₄₁	1.422	1.419	1.398
C ₁₈ -C ₁₉	1.402	1.392	1.394	C ₄₂ -C ₄₃	1.425	1.425	1.412
C ₁₉ -S ₂₀	1.726	1.726	1.733	C ₄₂ -C ₄₄	1.422	1.422	1.415
S ₂₀ -C ₂₁	1.786	1.782	1.801	C ₄₃ -N ₄₅	1.186	1.156	1.161
C ₂₁ -C ₂₂	1.395	1.398	1.409	C ₄₄ -N ₄₆	1.187	1.157	1.161
C ₂₂ -C ₂₃	1.396	1.394	1.390				
C ₂₃ -N ₂₄	1.385	1.380	1.384				

Figure S13. Optimized geometrical coordinates of DFM at the neutral, cation and anionic states.



DFI	Neutral	Cation	Anion	DFI	Neutral	Cation	Anion
C ₁ -C ₂	1.544	1.545	1.541	C ₃₀ -C ₃₁	1.404	1.391	1.392
C ₁ -C ₃	1.544	1.545	1.541	C ₃₁ -S ₃₂	1.726	1.729	1.735
C ₁ -C ₄	1.528	1.527	1.526	S ₃₂ -C ₃₃	1.790	1.786	1.801
C ₄ -C ₅	1.383	1.375	1.379	C ₃₃ -C ₃₄	1.394	1.396	1.410
C ₅ -C ₆	1.409	1.419	1.418	C ₃₄ -C ₃₅	1.396	1.395	1.391
C ₆ -C ₇	1.409	1.422	1.417	C ₃₅ -N ₃₆	1.387	1.381	1.387
C ₇ -C ₈	1.388	1.378	1.387	N ₃₆ -C ₃₇	1.452	1.459	1.454
C ₈ -C ₉	1.396	1.405	1.401	N ₃₆ -C ₃₈	1.379	1.377	1.381
C ₄ -C ₉	1.407	1.417	1.412	C ₃₈ -C ₃₉	1.414	1.397	1.405
C ₉ -C ₁₀	1.461	1.440	1.450	C ₃₈ -C ₂₈	1.461	1.395	1.391
C ₁₀ -C ₁₁	1.396	1.405	1.401	C ₂₁ -C ₄₁	1.416	1.421	1.419
C ₁₁ -C ₁₂	1.388	1.378	1.387	C ₄₁ -C ₄₂	1.369	1.366	1.396
C ₁₂ -C ₁₃	1.409	1.422	1.418	C ₃₃ -C ₄₁	1.420	1.421	1.411
C ₁₃ -C ₁₄	1.409	1.419	1.418	C ₄₂ -C ₄₃	1.476	1.482	1.456
C ₁₄ -C ₁₅	1.383	1.375	1.378	C ₄₃ -C ₄₄	1.395	1.493	1.508
C ₁₅ -C ₁₀	1.406	1.417	1.379	C ₄₄ -C ₄₅	1.386	1.390	1.384
C ₁₅ -C ₁	1.528	1.527	1.526	C ₄₅ -C ₄₆	1.395	1.395	1.403
C ₆ -C ₁₆	1.462	1.442	1.450	C ₄₆ -C ₄₇	1.401	1.402	1.394
C ₁₆ -S ₁₇	1.774	1.783	1.789	C ₄₇ -C ₄₈	1.395	1.395	1.403
S ₁₇ -C ₁₈	1.738	1.733	1.741	C ₄₈ -C ₄₉	1.395	1.390	1.385
C ₁₈ -C ₁₉	1.404	1.391	1.393	C ₄₉ -C ₅₀	1.398	1.489	1.499
C ₁₉ -S ₂₀	1.726	1.729	1.734	C ₄₂ -C ₅₀	1.480	1.493	1.472
S ₂₀ -C ₂₁	1.790	1.786	1.805	C ₄₃ -O ₅₁	1.223	1.218	1.230
C ₂₂ -C ₂₃	1.396	1.396	1.397	C ₄₉ -O ₅₂	1.223	1.222	1.235
C ₂₃ -N ₂₄	1.387	1.395	1.388				
N ₂₄ -C ₂₅	1.452	1.459	1.455				
N ₂₄ -C ₂₆	1.379	1.377	1.380				
C ₂₆ -C ₂₇	1.414	1.422	1.405				
C ₂₇ -C ₁₆	1.379	1.397	1.390				
C ₁₃ -C ₂₈	1.462	1.442	1.450				
C ₂₈ -S ₂₉	1.774	1.783	1.789				
S ₂₉ -C ₃₀	1.726	1.733	1.741				

Figure S14. Optimized geometrical coordinates of DFI at the neutral, cation and anionic states.

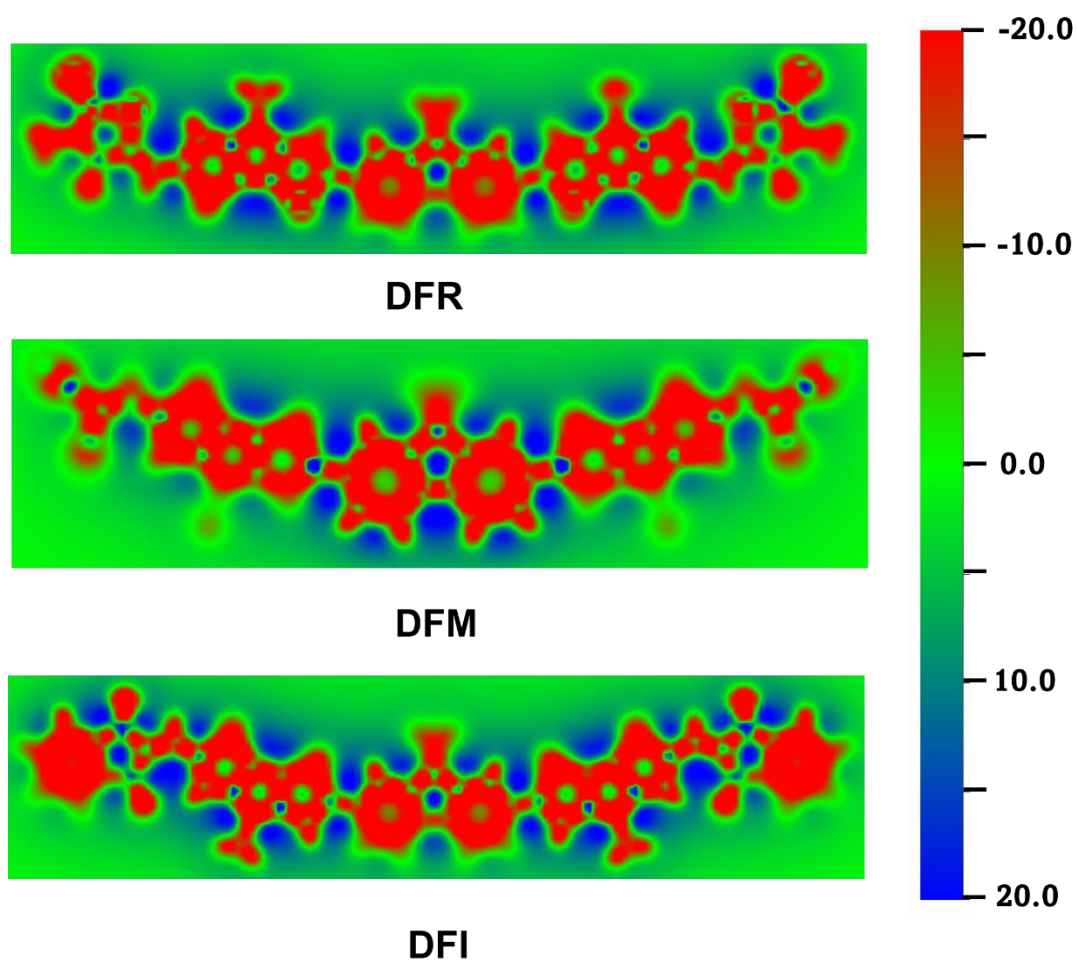


Figure S15. 2D-ICSS Map of the SMEDs showing aromatic and anti-aromatic characteristics.

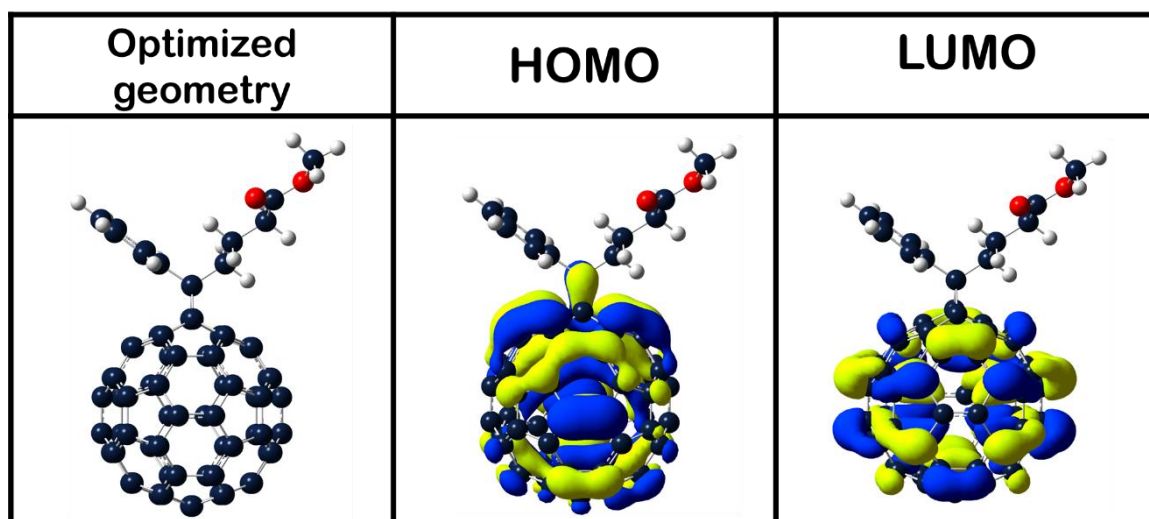


Figure S16. Optimized geometry and computed isosurface of frontier molecular orbitals of PC₆₁BM electron acceptor.

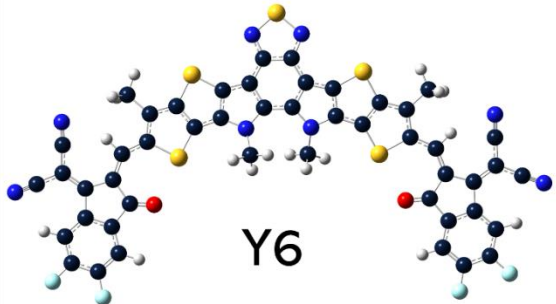
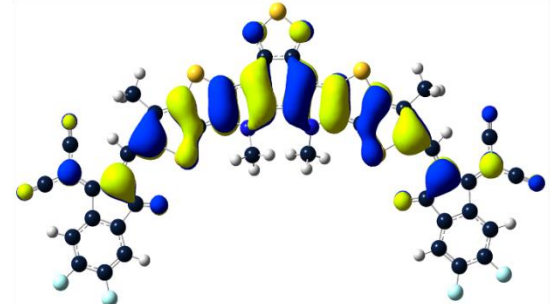
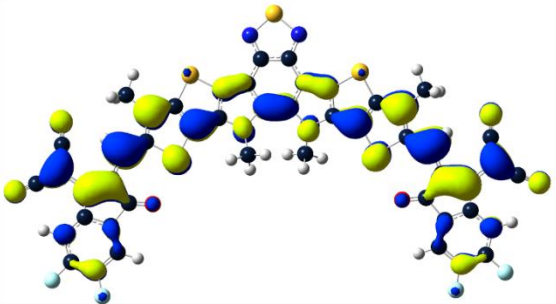
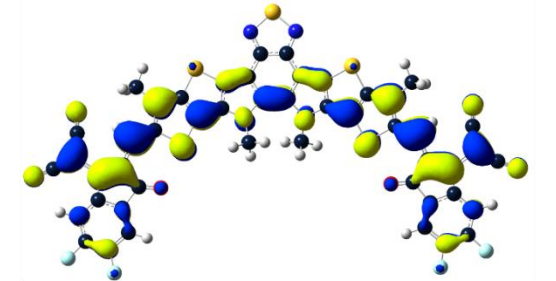
Optimized geometry	 Y6
HOMO	
LUMO	
LUMO+1	

Figure S17. Optimized geometry and computed isosurface plots of the frontier molecular orbitals of Y6 electron acceptor.

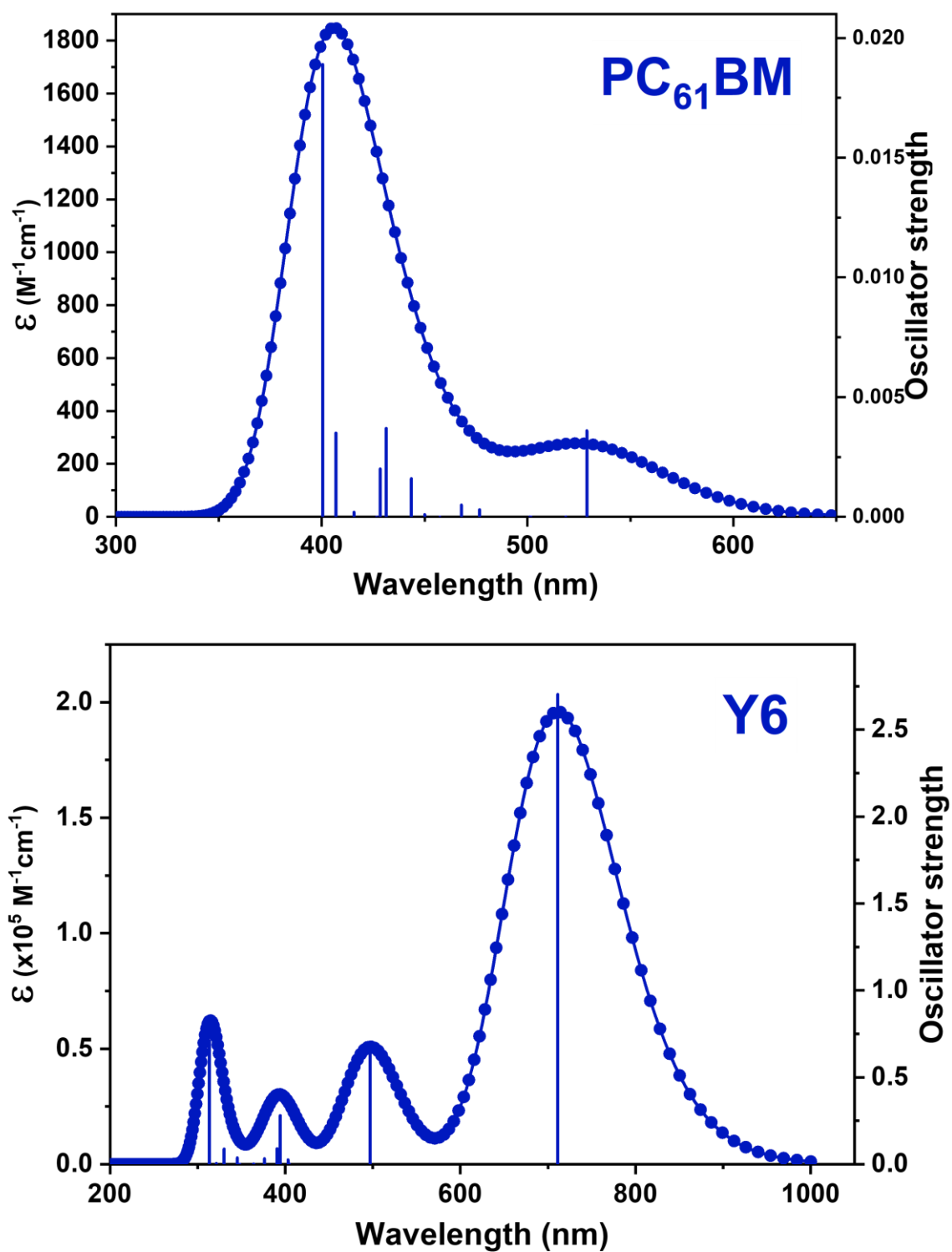


Figure S18. Simulated absorption profiles of PC₆₁BM and Y6 acceptors obtained at the TDDFT/M06-2X/6-311G (d, p)/CPCM(CHCl₃) level of theory.

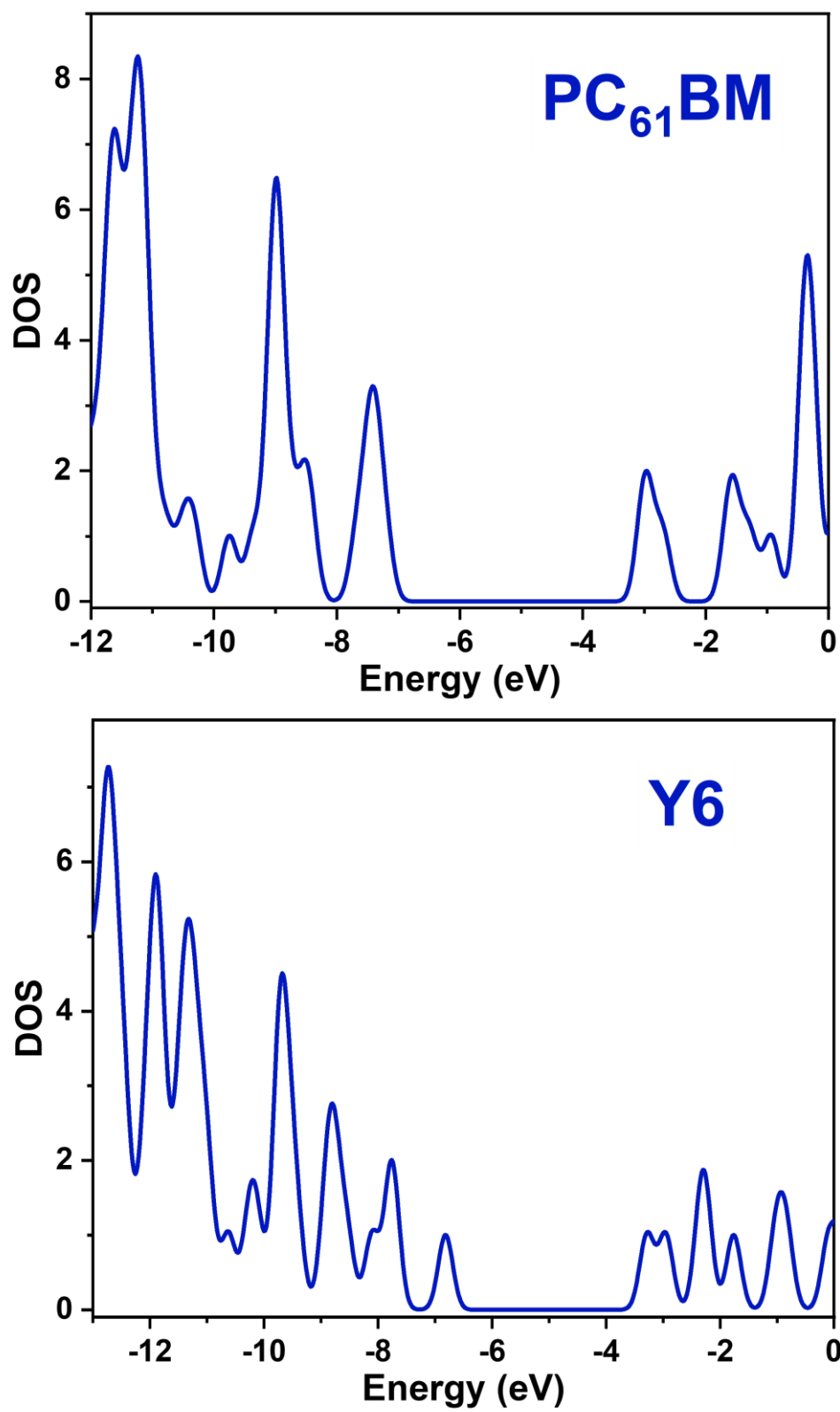


Figure S19. DOS population corresponding to the frontier energy levels of PC₆₁BM and Y6 electron acceptors.

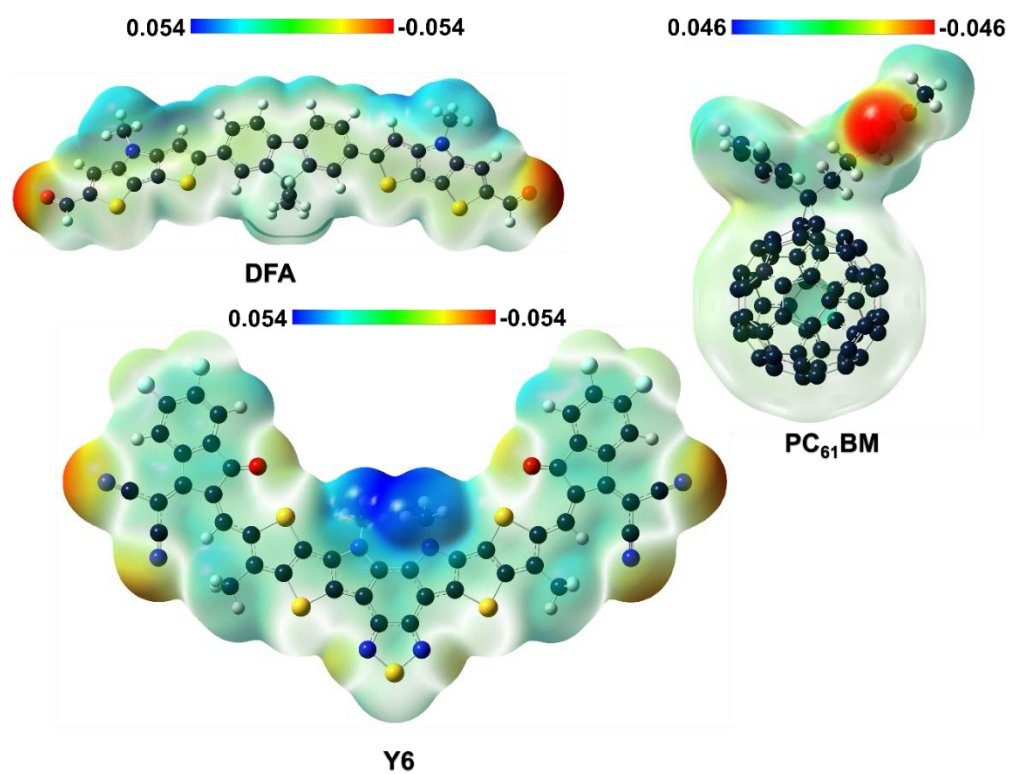


Figure S20. ESP plots of the SMEDs obtained from DFT/B3LYP/6-311G++ (d, p) level of theory.

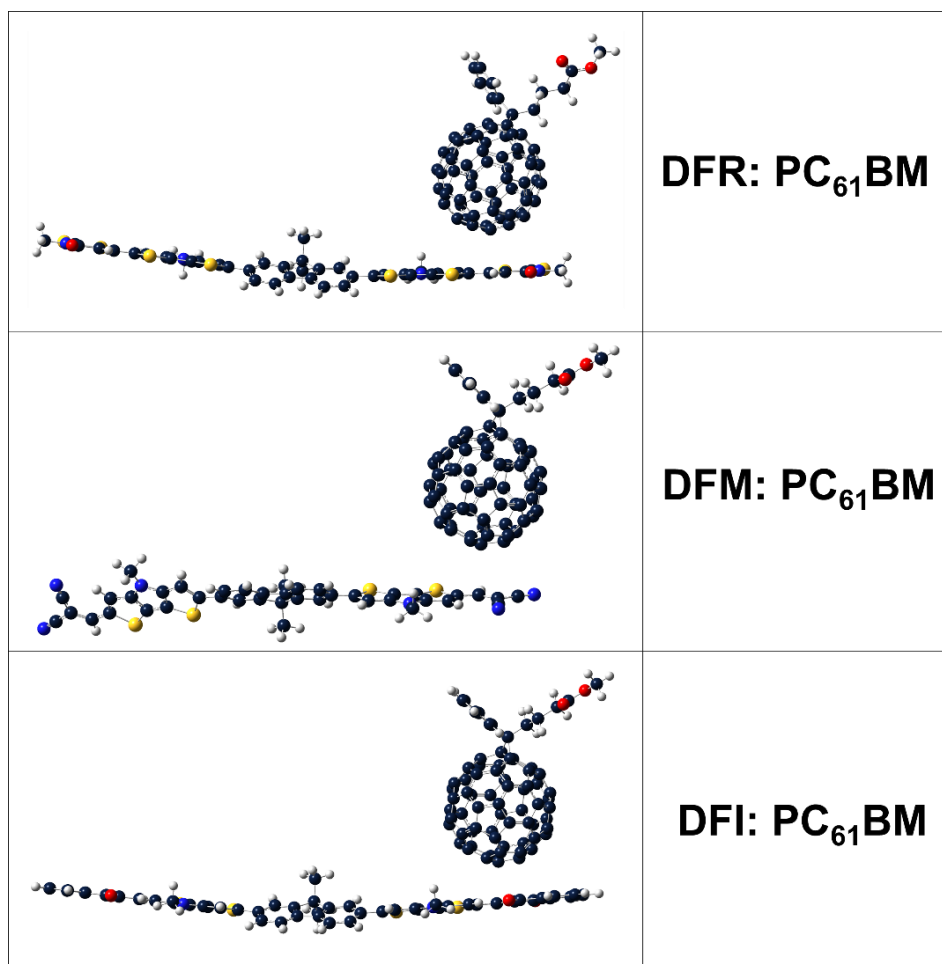


Figure S21. Optimized geometries of SMEDs-PC₆₁BM combination.

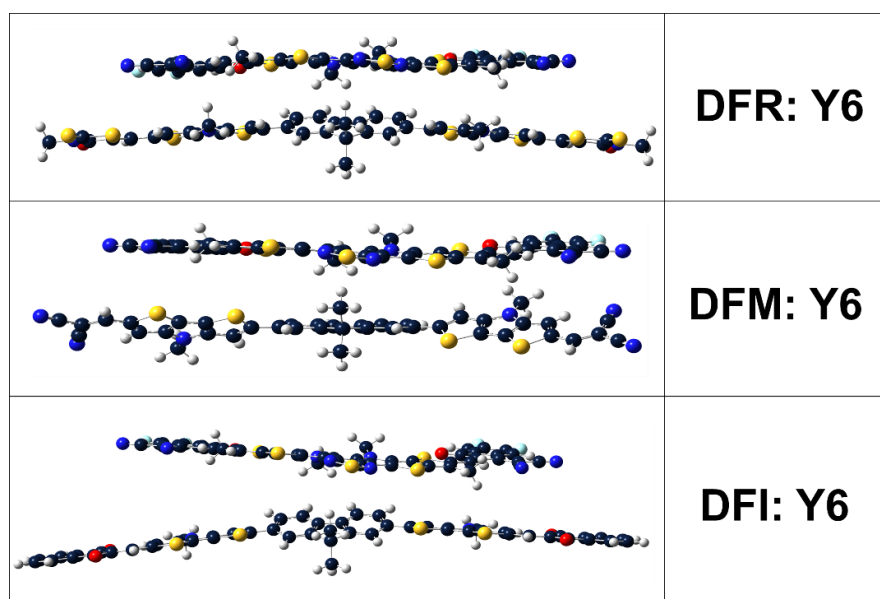


Figure S22. Optimized geometries of SMEDs-Y6 combination.

Tables

Table S1. Computed S_0 - S_1 excitation energies with electron volt in parenthesis, oscillator strength (f), major transitions and transient dipole moment of the dyes obtained from B3LYP/6-311G(d, p)/C-PCM(CH_2Cl_2) level of theory.

B3LYP		λ_{max} (nm)	f	Major transitions	μ_e (D)
DFR	S_0 - S_1	489.3 (2.534 eV)	4.27	HOMO->LUMO (57%), H-1->L+1 (33%)	3.33
	S_0 - S_2	463.8 (2.673 eV)	0.16	H-1->LUMO (44%), HOMO->L+1 (46%)	
DFM	S_0 - S_1	476.6 (2.601 eV)	3.82	H-1->L+1 (30%), HOMO->LUMO (63%), H-4->L+2 (2%)	8.44
	S_0 - S_2	446.1 (2.779 eV)	0.18	H-1->LUMO (44%), HOMO->L+1 (49%), H-4->L+1 (4%)	
DFI	S_0 - S_1	494.3 (2.508 eV)	4.41	H-1->L+1 (32%), HOMO->LUMO (60%), H-4->L+4 (2%)	12.12
	S_0 - S_2	464.8 (2.667 eV)	0.16	H-1->LUMO (44%), HOMO->L+1 (47%), H-4->L+1 (4%)	

Table S2. Computed S_0 - S_1 excitation energies with electron volt in parenthesis, oscillator strength (f), major transitions and transient dipole moment of the dyes obtained from CAM-B3LYP/6-311G(d, p)/C-PCM(CH_2Cl_2) level of theory.

CAM-B3LYP		λ_{max} (nm)	f	Major transitions	μ_e (D)
DFR	S_0 - S_1	538.4 (2.303 eV)	4.37	H-1->L+1 (29%), HOMO->LUMO (64%)	3.69
	S_0 - S_2	465.9 (2.661 eV)	0.16	H-1->LUMO (43%), HOMO->L+1 (50%), H-4->L+1 (4%)	
DFM	S_0 - S_1	511.6 (2.423 eV)	3.81	H-1->L+1 (27%), HOMO->LUMO (67%), H-4->L+2 (2%)	8.25
	S_0 - S_2	445.1 (2.786 eV)	0.18	H-1->LUMO (42%), HOMO->L+1 (52%), H-4->L+1 (3%)	
DFI	S_0 - S_1	556.2 (2.229 eV)	4.25	H-1->L+1 (31%), HOMO->LUMO (62%)	11.96
	S_0 - S_2	465.1 (2.666 eV)	0.16	H-1->LUMO (44%), HOMO->L+1 (49%), H-4->L+1 (3%)	

Table S3. Computed S_0 - S_1 excitation energies with electron volt in parenthesis, oscillator strength (f), major transitions and transient dipole moment of the dyes obtained from M06-2X/6-311G(d, p)/C-PCM(CH_2Cl_2) level of theory.

M06-2X		λ_{max} (nm)	f	Major transitions	μ_e (D)
DFR	S_0 - S_1	555.5 (2.232 eV)	3.26	HOMO->LUMO (95%), H-1->L+1 (3%)	3.13
	S_0 - S_4	491.2 (2.524 eV)	0.71	H-1->L+1 (95%), HOMO->LUMO (4%)	
DFM	S_0 - S_1	537.0 (2.309 eV)	3.34	HOMO->LUMO (97%)	8.86
	S_0 - S_4	446.4 (2.777 eV)	0.66	H-3->L+1 (25%), H-2->LUMO (49%), H-1->L+1 (25%)	
DFI	S_0 - S_1	579.6 (2.534 eV)	3.47	HOMO->LUMO (96%), H-1->L+1 (3%)	12.5
	S_0 - S_4	475.8 (2.534 eV)	0.71	H-1->L+1 (93%), HOMO->LUMO (3%)	

Table S4. Computed S_0 - S_1 excitation energies with electron volt in parenthesis, oscillator strength (f), major transitions and transient dipole moment of SMEDs-PC₆₁BM acceptor combination obtained from M06-2X/6-311G(d, p)/C-PCM(CH_2Cl_2) level of theory.

M06-2X		λ_{max} (nm)	f	Major transitions	μ_e (D)
DFR-PC ₆₁ BM	S_0 - S_2	509.4 (2.434 eV)	0.14	H-2->LUMO (91%), H-7->L+1 (3%)	7.14
	S_0 - S_4	506.1 (2.450 eV)	1.22	H-3->LUMO (16%), HOMO->LUMO (11%), HOMO->L+1 (13%), HOMO->L+3 (17%), H-2->L+1 (8%), H-1->LUMO (7%), H-1->L+1 (7%), H-1->L+4 (8%)	
DFM-PC ₆₁ BM	S_0 - S_5	480.8 (2.579 eV)	3.63	H-2->LUMO (92%), H-5->L+1 (3%)	10.33
	S_0 - S_8	449.9 (2.756 eV)	0.19	H-1->L+4 (26%), HOMO->L+3 (62%), H-9->L+8 (2%), HOMO->L+2 (5%)	
DFI-PC ₆₁ BM	S_0 - S_1	525.6 (2.359 eV)	0.58	H-2->LUMO (55%), H-3->LUMO (3%), H-1->LUMO (5%), HOMO->LUMO (5%), HOMO->L+1 (5%), HOMO->L+3 (6%)	12.3
	S_0 - S_4	517.2 (2.397 eV)	1.63	H-2->LUMO (30%), H-2->L+1 (10%), HOMO->L+3 (23%) H-1->L+1 (5%), H-1->L+4 (8%), HOMO->L+1 (9%)	

Table S5. Computed S0- S1 excitation energies with electron volt in parenthesis, oscillator strength (f), major transitions and transient dipole moment of SMEDs-Y6 acceptor combination obtained from M06-2X/6-311G(d, p)/C-PCM(CH₂Cl₂) level of theory.

M06-2X		λ_{max} (nm)	f	Major transitions	μ_e (D)
DFR-Y6	S ₀ -S ₁	608.8 (2.037 eV)	2.38	H-1->LUMO (83%), H-6->L+1 (7%), H-5->L+1 (2%), H-1->L+4 (2%)	7.95
	S ₀ -S ₅	486.9 (2.546 eV)	4.15	H-2->L+3 (29%), HOMO->L+2 (61%), H-7->L+7 (2%)	
DFM-Y6	S ₀ -S ₁	606.2 (2.045 eV)	2.31	H-1->LUMO (84%), H-6->L+1 (7%)	8.87
	S ₀ -S ₅	493.8 (2.511 eV)	4.48	H-2->L+3 (29%), HOMO->L+2 (64%) H-7->L+9 (2%)	
DFI-Y6	S ₀ -S ₁	611.8 (2.027 eV)	2.26	H-1->LUMO (71%), HOMO->LUMO (14%), H-6->L+1 (4%), H-3->L+1 (5%)	12.98
	S ₀ -S ₅	473.9 (2.616 eV)	3.70	H-2->L+3 (26%), HOMO->L+2 (57%), H-1->L+2 (7%)	

Table S6. Computed bond length alternation and energy levels of the SMEDs at the cation and anionic states

	BLA _{neu} ^a (Å)	BLA _{cat} ^b (Å)	BLA _{ani} ^c (Å)	BLA _{N→C} ^d (Å)	BLA _{N→A} ^e (Å)	HOMO _{cat} ^f (eV)	LUMO _{cat} ^f (eV)	H- L _{cat} ^f (eV)	HOMO _{ani} ^g (eV)	LUMO _{ani} ^g (eV)	H- L _{ani} ^g (eV)
DFR	-0.161	-0.061	-0.082	-0.154	-0.03	-8.52	-4.48	4.04	-4.08	-1.90	2.18
DFM	-0.357	-0.037	-0.005	-0.173	-0.072	-9.02	-4.94	4.08	-4.48	-2.10	2.38
DFI	-0.126	-0.042	-0.004	-0.202	-0.141	-8.54	-4.48	4.06	-4.28	-1.92	2.36

^{a,b,c}-variation of bond length alternation of the SMEDs at the neutral, oxidized and reduced state respectively; ^d-variation of BLA transformation from neutral to oxidized/reduced charged states respectively; ^{f,g}- frontier energy levels at the oxidized and reduced states respectively.

Table S7. Interaction energy of the SMEDs upon combination with the electron acceptors PC₆₁BM and Y6.

D-A	Interaction energy (kcal/mol)	D-A	Interaction energy (kcal/mol)
DFR-PC ₆₁ BM	-169.70	DFR-Y6	-166.29
DFM-PC ₆₁ BM	-156.37	DFM-Y6	-142.76
DFI-PC ₆₁ BM	-169.92	DFI-Y6	-153.20