Electronic Supplementary Information (ESI)

Functionalization Methodology for Synthesis of Silane-End-Functionalized Linear and Star Poly(aryl isocyanide)s by Combination of Cationic Polymerization and Hydrosilylation Reaction

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References

EXPERIMENTAL SECTION

Materials

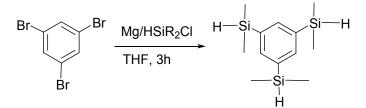
All manipulations of air and moisture-sensitive compounds were performed under a dry and oxygen free nitrogen atmosphere by using schlenk techniques or under nitrogen atmosphere in an Mbraunglovebox. Nitrogen (Beijing AP beifen gases Industrial Co.Ltd.) was purified through a dry clean column (4A molecular sieves, Dalian Replete Science and Technology Co.Ltd.) and a gas clean column (Dalian Replete Science and Technology Co.Ltd.). The nitrogen in glovebox was constantly circulated through copper/molecular sieves catalyst unit. The oxygen and moisture concentrations in the glovebox atmosphere were monitored by an O_2/H_2O Combi-Analyzer (Mbraun) to ensure both were always below 0.1ppm. THF, Hexane, Toluene, Chlorobenzene, Dichlorobenzene and 1,1,2,2-tetrachloroethane were purified by a solvent purification system (SPS-800, Mbraun), and dried over fresh Na Chips in the glovebox. [Ph₃C][B(C₆F₅)₄], [PhMe₂NH][B(C₆F₅)₄], and B(C₆F₅)₃ were purchased from Tosoh Finechem Corporation and used without purification. All Silane monomers were also purchased from Tosoh Finechem Corporation and dried with CaH₂ and Distilled.

Isocyanide monomers were synthesized according to literatures.^[1] The 1,3,5-tris(dimethylsilyl)benzene also synthesized according to literature.^[2] The deuterated solvents benzene-d6 (99.6 atom% D) and Chloroforum-d1 (99.8 atom% D) were obtained from Cambridge Isotope.

General Methods

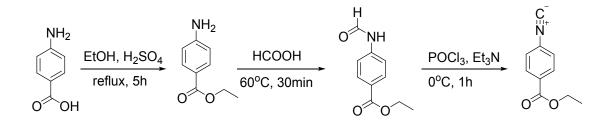
¹H and ¹³C NMR spectra were recorded on a BrukerAvance (III-HD 400 MHz) Spectrometer. The molecular Weights and molecular weight distributions of EPI Polymers were determined against polystyrene standard at 25°C by GPC on waters HPLC-515 apparatus, CHCl₃ was employed as the eluent at a flow rate of 1 ml/min. The molecular Weights and molecular weight distributions of the D-(or L-) IMCI and ITPA Polymers were determined against polystyrene standard at 25 °C by GPC on waters HPLC-8320GPC apparatus, THF was employed as the eluent at a flow rate of 1 mL/min.

FT-IR spectra were recorded on a thermo IS5 FT-IR system using KBr Pellets at room temperature. The UV-Vis spectra were recorded on a HITACHI F-7000 Fluorescence spectrometer. Quartz cells with 10.0 mm length were used in UV-Vis and fluorescence measurement, and the slit widths were set at 5.0 nm for both excitation and emission during the fluorescence measurement. Circular dichroism spectra were collected on a Jasco J-810 and quartz cell length was 1.0 nm. Optical rotations were measured on a Kruss P8000-T polarimeter using a 0.5 cm cell with a Na 589 nm filter. High resolution mass spectra were collected on an Agilent 6520 Accurate-mass Q-TOF LC/MS.



Scheme S1. Synthesis of 1,3,5-Tris(dimethylsilyl) Benzene

Synthesis of 1,3,5-tris(dimethylsilyl) benzene: The 1,3,5-tris(dimethylsilyl) benzene was synthesized according to the literature procedures.^[3]To a mixture of dimethylchlorosilane (4.73 g, 50.0 mmol) and magnesium (1.22 g, 50.0 mmol) in dry THF (15 mL) was added a solution of 1,3,5-tribromobenzene (3.15 g, 100 mmol) in THF (10 mL). The rate of the addition was adjusted to maintain a reflux. When the addition was finished the reaction was heated under reflux for 3 h. The volatiles were removed in vacuo and the product extracted from the solid residue of hexane (3 × 20 mL). The solvent was removed in vacuo. The product was purified by distillation under reduced pressure, Yield 75%. ¹H-NMR (CDCl₃): 0.21 (18H, d, – CH₃, J ¹/₄ 3 Hz), 4.29 (3H, septet, Si–H, J ¹/₄ 3 Hz), 7.58 (3H, s, H–Ar).



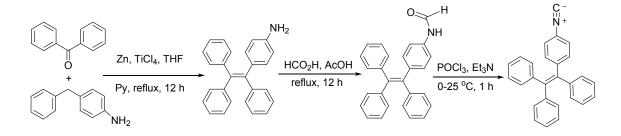
Scheme S2. Synthesis of 4-Ethoxycarbonyl Phenyl Isocyanides (EPI)

Synthesis of ethyl-4-aminobenzoate: To a solution of 4-aminobenzoic acid (5.26 g, 30.0 mmol) in 120 mL of EtOH was slowly added 16.3 mL of aqueous con.H₂SO₄ (300 mmol) at room temperature. The mixture was refluxed for 7 h, cooled to room temperature, neutralized with a saturated K₂CO₃ aqueous solution and extracted with ethyl acetate (3×60 mL), the combined organic phases were washed with brine (2×40 mL), dried over anhydrous Na₂SO₄ and concentrated in vacuum, the residue was purified by column chromatography (silica gel, 4:1-2:1 hexane to ethyl acetate, v/v) to afford compoundas a white solid (4.35 g, 93% yield). ¹H NMR (400 MHz, CDCl₃): 1.36 (t, *J*= 7.0 Hz, 3H), 4.06 (br, 2H), 4.31 (q, *J*= 7.2 Hz, 2H), 6.63 (d, *J*= 8.8 Hz, 2H), 7.85 (d, *J*= 8.4 Hz, 2H).

Synthesis of ethyl-4-formamidobenzoate: Compoundethyl 4-aminobenzoate (4.35 g, 26.3 mmol) was dissolved in a formic acid (1.5 mL, 39.5 mmol) the resulting mixture was heated at 60°C for 30min. After the reaction mixture was cooled to room temperature, the excess formic acid was removed under reduced pressure, the residue was washed with saturated aqueous Na₂CO₃ (25 mL) and filtered, the filter cake was washed twice with water and dried in vacuum to afford crude compoundas a white solid (4.71 g, crude), this compound was used directly for the next step without purification.

Synthesis of 4-ethoxycarbonyl phenyl isocyanide: Compound ethyl-4-formamidobenzoate (4.71 g, 28.5 mmol) and triethylamine (27.60 mL, 191.0 mmol) were dissolved in dry THF (45 mL) under an atmosphere of nitrogen, after the mixture was cooled to 0°C, POCl₃ (4.50 mL, 48.4 mmol) was added drop wise to the mixture, the resulting mixture was slowly warmed to room temperature and stirred for 1h, then the reaction mixture was slowly poured into 40 mL saturated aqueous Na₂CO₃ and stirred at room temperature for 1h, the mixture was extracted with CH_2Cl_2 (3 ×60 mL), the combined organic layers were washed with brine, dried over anhydrous Na₂SO₄ and concentrated under reduced pressure, the residue was purified by column chromatography (neutral Al₂O₃, 10:1 hexane to ethyl acetate, v/v) to afford the desired compound as brown solid (3.71 g, 87% yield). ¹H NMR (400

MHz,CDCl₃): 1.40 (t, *J*= 7.2 Hz, 3H), 4.39 (q, *J*= 7.2 Hz, 2H), 7.43 (d, *J*= 8.4 Hz, 2H), 7.43 (d, *J*= 8.4 Hz, 2H), 8.08 (dt, *J*= 2.0, 8.8 Hz, 2H); ¹³C NMR (100 MHz,CDCl₃):14.33, 61.66, 126.48, 129.94 , 130.88, 131.38, 165.06, 167.10.



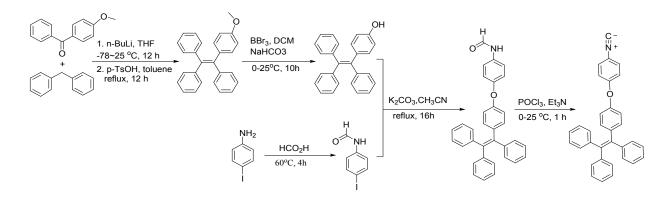
Scheme S3. Synthesis of 4-Isocyano-4-(1,1,2-Triphenylvinyl)-1-Phenyl (ITPP)

Synthesis of (2-(4-aminophenyl)ethene-1,1,2-triyl)tribenzene: The (2-(4-aminophenyl)ethene-1,1,2-triyl)tribenzene was synthesized by the typical McMurry reaction according to the literature procedures.^[4] Into a two-necked round-bottom flask (100 mL) with reflux condenser were added of Zinc powder (Zn, 1.49 g, 22.8 mmol) and THF (60 mL). The flask was evacuated and flushed with dry nitrogen three times. After cooling to -15° C, TiCl₄ (4.33 g, 22.8 mmol) was slowly added. The mixture was stirred for 2.5 h at room temperature. After cooling to -5° C, again, Pyridine (1.53 mL, 19.0 mmol) was added and the mixture was stirred for 10 min. Then the solution of benzophenone (1.50 g, 7.6 mmol) and 4-benzylaniline (1.66 g, 9.1 mmol) in THF (40 mL) was added, the mixture was refluxed overnight. Afterwards, K₂CO₃ solution was added to quench the reaction. After cooling to room temperature, THF was removed by a rotary evaporator. The solution was poured in to water and extracted with DCM. The collected organic layer was dried over anhydrous Na₂SO₄. After extraction and evaporation of solvent, the crude product was recrystallized from Methanol. A yellow solid was obtained in 56%. ¹H NMR (400 MHz, CDCl₃): 7.13–6.99 (m, 17H); 6.85–6.82 (d, 2H); 6.52–6.49 (d, 2H).¹³C NMR (100MHz, CDCl₃): 114.69, 144.57, 144.52, 144.44, 141.25, 139.87, 135.04, 132.94, 131.89.

Synthesis of (2-(4-formamidophenyl) ethene-1,1,2-triyl) tribenzene: Into a 100 mL two-necked flask was added compound (2-(4-aminophenyl)ethene-1,1,2-triyl)tribenzene(1.50 g, 4.3 mmol) and THF (40 mL). After cooling to 0°C, acetic formal anhydride was tardily added by a syringe. The mixture was stirred at room temperature for 2 h. Then the reaction was quenched by the saturated solution of NaHCO₃. The mixture was extracted with ethyl acetate (EA), and then the collected organic layer was dried over anhydrous Na₂SO₄

and concentrated in Vacuum. The crude product was recrystallized from Methanol. A crude yellow solid was obtained in 82% and used directly for next step.

Synthesis of 4-Isocyano-4-(1,1,2-Triphenylvinyl)-1-Phenyl(ITPP): Compound (2-(4-formamidophenyl) ethene-1,1,2-triyl) tribenzene(1.39 g, 3.7 mmol) and triethylamine (3.50 mL, 24.8 mmol) were dissolved in dry THF (20 mL) under an atmosphere of nitrogen, after the mixture was cooled to 0°C, POCl₃ (0.60 mL, 6.3 mmol) was added drop wise to the mixture, the resulting mixture was slowly warmed to room temperature and stirred for 1h, then the reaction mixture was slowly poured into 15 mL saturated aqueous Na₂CO₃ and stirred at room temperature for 1 h, the mixture was extracted with CH₂Cl₂ (3 × 20 mL), the combined organic layers were washed with brine, dried over anhydrous Na₂SO₄ and concentrated under reduced pressure, the residue was purified by column chromatography (neutral Al₂O₃, 30:1 hexane to ethyl acetate, v/v) to afford the desired compound as light yellow solid (0.84g, 63% yield). ¹H NMR (400 MHz, CDCl₃): 7.17–7.10 (m, 11H), 7.21–6.93 (m, 8H); ¹³C NMR (101 MHz, CDCl₃): 167.01, 148.40, 145.13, 145.08, 144.94, 144.78, 141.27, 133.38, 133.15, 132.35, 132.29, 129.09, 129.05, 128.86, 128.12, 126.99, 127.96, 126.88, 125.55.



Scheme S4. Synthesis of 4-Isocyano-4-(1,1,2-Triphenylvinyl)-1-Phenyl-1-Anisol(ITPPA)

Synthesis of 4-methoxytetraphenylethene: Under nitrogen atmosphere, compounddiphenylmethane(8.60 g, 51.0 mmol) was dissolved in 50 mL of dry THF, after the solution was cooled to -78° C, n-BuLi (22.30 mL, 2.5 M in hexane) was added drop wise and the resulting mixture was stirred at -10° C for 2 h, then compound (4-methoxyphenyl)(phenyl)methanone (9.0 g, 42.4 mmol) in 15 mL THF was added drop wise and the mixture was allowed to warmed to room temperature and stirred for 10 h. Then the reaction mixture was quenched with an aqueous solution of ammonium chloride, extracted with CH₂Cl₂ (3 × 100 mL), the combined organic layers were dried over anhydrous Na₂SO₄ and evaporated to remove the solvent. The

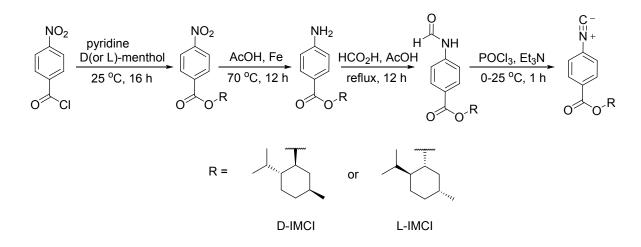
residue was dissolved in toluene (100 mL), p-toluene sulfonic acid (1.06 g, 6.2 mmol) as catalyst was added, the resulting mixture was refluxed for 10 h. after the reaction mixture was cooled to room temperature, the solvent was removed under reduced pressure, the residue was recrystallized from methanol to afford the desired compound 4-methoxytetraphenyletheneas a white solid (14.8 g, 86% yield). ¹H NMR (400 MHz, CDCl₃): 7.21–7.01 (m, 15H, Ar–*H*), 6.97 (d, J = 8.4 Hz, 2H, Ar–*H*), 6.67 (d, J = 8.6 Hz, 2H, Ar–*H*), 3.77 (s, 3H, CH₃).

Synthesis of 4-hydroxytetraphenylethene: Two-necked flask containing compound 4methoxytetraphenylethene (8.25 g, 22.7 mmol) and dried DCM (120 mL) was cooled to -20°C, and BBr₃ (17.10 g, 68.2 mmol) was slowly added. The mixture was warmed to room temperature and allows standing for 8 h, and then poured into saturated aqueous solution of NaHCO₃. The organic phase was dried over Na₂SO₄, and the crude product was recrystallized from petroleum ether to obtained white solid (7.65 g, 97% yield).¹H NMR (400 MHz, CDCl₃): 7.01–7.13 (m, 15H, Ar–*H*). 6.90 (d, J = 8.0 Hz, 2H, Ar–*H*). 6.56 (d, J = 8.0 Hz, 2H, Ar–*H*),4.60 (s, 1H, O–*H*). ¹³C NMR (100 MHz, CDCl₃): 154.08, 144.12, 144.01, 140.54, 140.34, 136.54, 132.87, 131.49, 131.46, 127.84,127.74, 126.52, 126.4, 114.72.

Synthesis of N-(4-iodophenyl) formamide: The synthetic procedure was the same with that of compound (2-(4-formamidophenyl) ethene-1, 1, 2-triyl) tribenzene, and the crude product was directly used for the next step without further purification (11.5g, 92% yield).

Synthesis of N-(4'-(1,2,2-triphenylvinyl)-[1-phenyl-1-anisol]-4-yl) formamide: Compound N-(4iodophenyl) formamide (4.40 g, 17.8 mmol), 4-hydroxytetraphenylethene (5.64 g, 16.2 mmol) and K₂CO₃ (3.26 g, 24.3 mmol) in 100 mL CH₃CN were reflux overnight (16 h). After the reaction completed, water was added and the resulting mixture was extracted with ethyl acetate. The organic layer was collected, washed with water and brine, dried over anhydrous MgSO₄ and evaporated. The crude product was washed with Petroleum ether to obtain a yellow solid (8.5 g, 84.7% yield) and directly used for next the step without further purification.

Synthesis of 4-isocyano-4'-(1,2,2-triphenylvinyl)-1-phenyl-1-anisol (ITPPA): The synthetic procedure was the same with that of compound 4-Isocyano-4-(1, 1, 2-Triphenylvinyl)-1-Phenyl (ITPP). (Weight solid, 83% yield). ¹H NMR (400 MHz, CDCl₃): 7.03–7.15 (m, 17H), 7.32 (d, *J*= 8.0 Hz, 2H), 7.40 (d, *J*= 8.4 Hz, 2H), 7.56 (dt, *J*= 1.6, 8.4 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃): 125.50, 126.36, 126.69, 126.73, 126.76, 126.83, 127.81, 127.84, 127.90, 127.95, 131.43, 131.46, 131.50, 132.15, 137.06, 140.28, 141.75, 141.99, 143.64, 143.69, 144.02, 164.69.



Scheme S5. Synthesis of (*1S*,*2R*,*5S*)-2-Isopropyl-5-Metheylcyclohexyl-4-Isocyanobezoate (**D-IMCI**) and (*1R*,*2S*,*5R*)-2-Isopropyl-5-Metheylcyclohexyl-4-Isocyanobezoate (**L-IMCI**)

Synthesis of (*1S*,*2R*,*5S*)-2-isopropyl-5-metheylcyclohexyl-4-nitrobezoate: Under nitrogen atmosphere, compound 4-nitrobenzoyl chloride (1.80 g, 9.7 mmol) was dissolved in dry pyridine (20 mL), then D-menthol (1.50 g, 9.7 mmol) was added in one portion and the resulting mixture was stirred at room temperature for 16 h, after removal of pyridine under reduced pressure, the residue was dissolved indichloromethane(30 mL) and washed with 1N HCl, saturated NaHCO₃ aqueous solution and brine, the separated organic layer was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure, the residue was purified by column chromatography (silica gel, 10:1 hexane to ethyl acetate, v/v) to afford the desired compound (*1S*,*2R*,*5S*)-2-isopropyl-5-metheylcyclohexyl-4-nitrobezoate as a yellow solid (2.40 g, 81% yield) ¹H NMR (400 MHz, CDCl₃): 0.79 (d, *J* = 7.2 Hz, 3H), 0.93 (t, *J* = 6.4 Hz, 6H), 0.88-0.98 (m, 1H), 1.08–1.17 (m, 2H), 1.54–1.62 (m, 2H), 1.74 (d, *J* = 12.4 Hz, 2H), 1.88–1.95 (m, 1H), 2.12 (d, *J* = 11.6 Hz, 1H), 4.97 (dt, *J* = 4.4, 11.2 Hz, 1H), 8.20 (d, *J* = 8.8 Hz, 2H), 8.28 (d, *J* = 8.4 Hz, 2H).

Synthesis of (*1S*,2*R*,5*S*)-2-isopropyl-5-metheylcyclohexyl-4-aminobezoate: Under nitrogen atmosphere, compound (*1S*,2*R*,5*S*)-2-isopropyl-5-metheylcyclohexyl-4-nitrobezoate (2.40 g, 7.9 mmol) was dissolved in 30 mL of acetic acid, then iron powder (4.4 g, 78.6 mmol) was added in one portion, the resulting mixture was stirred at 70°C overnight. Then the mixture was filtered and the filter cake was washed with ethyl acetate (20 mL), the filtrate was concentrated under reduced pressure, the residue was purified by column chromatography (silica gel, 4:1 hexane to ethyl acetate, v/v) to afford the desired compound (*1S*,2*R*,5*S*)-2-isopropyl-5-metheylcyclohexyl-4-aminobezoate as yellow solid (1.55 g, 72% yield) ¹H NMR (400 MHz, CDCl₃): 0.78 (d, J = 7.2 Hz, 3H), 0.90 (d, J = 6.8 Hz, 3H), 0.91 (d, J = 6.4 Hz, 3H), 0.85–0.96 (m, 1H),

1.02–1.14 (m, 2H), 1.48–1.54 (m, 2H), 1.69–1.72 (m, 2H), 1.94–1.98 (m, 1H), 2.09–2.12 (m, 1H), 4.04 (s, 2H), 4.87 (dt, *J* = 4.4, 10.8 Hz, 1H), 6.63 (d, *J* = 8.4 Hz, 2H), 7.85 (d, *J* = 8.8 Hz, 2H).

Synthesis of (1S, 2R, 5S)-2-isopropyl-5-metheylcyclohexyl 4-formamidobezoate: Compound (1S, 2R, 5S)-2-isopropyl-5-metheylcyclohexyl-4-aminobezoate (1.55 g, 5.6 mmol) was dissolved in a mixture of formic acid (16 mL) and acetic acid (3 mL), the resulting mixture was refluxed overnight. After the reaction mixture was cooled to room temperature, the solvents were removed under reduced pressure, the residue was washed with saturated aqueous Na₂CO₃ (10 mL) and filtered, the filter cake was washed twice with water and dried in vacuum to afford crude compound (1S, 2R, 5S)-2-isopropyl-5-metheylcyclohexyl 4-formamidobezoateas a white solid (1.70 g, crude), this compound was used directly for the next step without purification.

Synthesis of (*1S*, *2R*, *5S*)-2-isopropyl-5-metheylcyclohexyl 4-isocyanobezoate: Compound (*1S*, *2R*, *5S*)-2-isopropyl-5-metheylcyclohexyl 4-formamidobezoate (1.70g, crude) and triethylamine (5.2 mL, 37.5 mmol) were dissolved in dry THF (15 mL) under an atmosphere of nitrogen, after the mixture was cooled to 0°C, POCl₃ (0.90 mL, 9.5 mmol) was added drop wise to the mixture, the resulting mixture was slowly warmed to room temperature and stirred for 1.5 h, then the reaction mixture was slowly poured into 20 mL saturated aqueous Na₂CO₃ and stirred at room temperature for 1 h, the mixture was extracted with DCM (3×20 mL), the combined organic layers were washed with brine, dried over anhydrous Na₂SO₄ and concentrated under reduced pressure, the residue was purified by column chromatography (neutral Al₂O₃, 10:1 hexane to ethyl acetate, v/v) to afford the desired compound (*1S*, *2R*, *5S*)-2-isopropyl-5-metheylcyclohexyl 4-isocyanobezoateas a black syrup (1.35g, 81% yield for two steps) ¹H NMR (400 MHz, CDCl₃): 0.77 (d, *J*= 6.8 Hz, 3H), 0.91 (dd, *J*= 5.6, 6.8 Hz, 6H), 0.86–0.96 (m, 1H), 1.05–1.17 (m, 2H), 1.50–1.59 (m,2H), 1.70–1.74 (m, 2H), 1.86–1.94 (m, 1H), 2.07–2.12 (m, 1H), 4.93 (dt, *J*= 4.4, 10.8 Hz, 1H), 7.43 (d, *J*= 8.4 Hz, 2H), 8.07 (dt, *J*= 2.0, 8.8 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃): 16.60, 20.82, 22.10, 23.71, 26.66, 31.53, 34.32, 40.98, 47.30, 75.75, 126.48, 129.88, 130.90, 131.74, 164.58, 167.05.

Synthesis of (*1R*,*2S*,*5R*)-2-isopropyl-5-metheylcyclohexyl 4-nitrobezoate: was the same with that of (*1S*,*2R*,*5S*)-2-isopropyl-5-methylcyclohexyl 4-isocyanobenzoate.

Synthesis of (*1R*,*2S*,*5R*)-2-isopropyl-5-metheylcyclohexyl 4-aminobezoate: (yellow solid, 83% yield). ¹H NMR (400 MHz, CDCl₃): 0.79 (d, *J*= 6.8 Hz, 3H), 0.93 (t, *J*= 6.2 Hz, 6H), 0.88–0.98 (m, 1H), 1.08–1.19 (m, 2H), 1.54–1.61 (m, 2H), 1.71–1.78 (m, 2H), 1.85–1.97 (m, 1H), 2.13 (d, *J*= 12.0 Hz, 1H), 4.97 (dt, *J*= 4.4, 10.8 Hz, 1H), 8.20 (d, *J*= 9.2 Hz, 2H), 8.28 (d, *J*= 8.8 Hz, 2H).

Synthesis of (*1R*,*2S*,*5R*)-2-isopropyl-5-metheylcyclohexyl 4-formamidobezoate: (yellow oil, 72% yield). ¹H NMR (400 MHz, *d6*-DMSO): 0.73 (d, *J* = 6.8 Hz, 3H), 0.87 (t, *J* = 7.4 Hz, 6H), 0.82–0.92 (m, 1H), 0.97–1.09 (m, 2H), 1.43–1.49 (m, 2H), 1.62–1.66 (m, 2H), 1.82–1.89 (m, 1H), 1.92–1.95 (m, 1H), 4.72 (dt, *J* = 4.4, 10.8 Hz, 1H), 5.92 (s, 2H), 6.56 (d, *J* = 8.8 Hz, 2H), 7.62 (d, *J* = 8.8 Hz, 2H).

Synthesis of (*1R*,*2S*,*5R*)-2-isopropyl-5-metheylcyclohexyl 4-isocyanobezoate: (black syrup, 80% yield). ¹H NMR (400 MHz, CDCl₃): 0.78 (d, *J* = 6.8 Hz, 3H), 0.92 (t, *J* = 6.4 Hz, 6H), 0.87–0.90 (m, 1H), 1.04–1.18 (m, 2H), 1.51–1.60 (m, 2H), 1.70–1.76 (m, 2H), 1.86–1.94 (m, 1H), 2.08–2.13 (m, 1H), 4.93 (dt, *J* = 4.4, 11.2 Hz, 1H), 7.43 (d, *J* = 8.4 Hz, 2H), 8.07 (dt, *J* = 2.0, 8.8 Hz,2H); ¹³C NMR (100 MHz, CDCl₃): 16.63, 20.86, 22.13, 23.74, 26.69, 31.57, 34.35, 41.01, 47.35, 75.81, 126.52, 129.92, 130.94, 131.78, 164.63, 167.03.

A typical procedure for the linear polymerization of 4-ethoxycarbonyl phenyl isocyanide (EPI) with Et_3SiHas end-functionalized agent (Table 1, entry 4). In the glove box, a 50 mL round bottom flask was charged with a solution of $[Ph_3C][B(C_6F_5)_4]$ (9.22 mg, 10.0 µmol) in chlorobenzene (2 mL), then 11.5 mg (10 equiv.) of triethylsilane was added. After stirring for 5 min, a solution of 4-ethoxycarbonyl phenyl isocyanide (875 mg, 5 mmol) in chlorobenzene (3 mL) was added in one portion. The reaction mixture was stirred at 25°C for 1min, then the flask was taken out of the glove box and the reaction mixture was poured into methanol (100 mL) to precipitate the polymer product, the yellow polymer solid was collected by filtration, and dried in vacuum at 40°C to a constant weight (868 mg, 99% yield), The product obtained is soluble thoroughly in CHCl₃ at 25°C.

A typical procedure for the three star random and homocopolymerization of (1S,2R,5S)-2-isopropyl-5methylcyclohexyl-4-isocyanobenzoate (D-IMCI) with 4-isocyano-4'-(1,2,2-triphenylvinyl)-1-phenyl-1-anisol (ITPPA) with 1,3,5-tris(dimethylsilyl)benzene as core-first functional agent (Table 2, entry 5). In the glove box, a 50 mL round bottom flask was charged with a solution of [Ph₃C][B(C₆F₅)₄] (7.5 mg, 8.1 µmol) in chlorobenzene (2 mL). Then 1,3,5-tris(dimethylsilyl)benzene (0.68 mg, 2.7 µmol) was added to the bottom, after stirring for 5 min, 10.8 µmol (4 equiv.) of 1,3,5-tris(dimethylsilyl)benzene was added and subsequently,a solution of (1S,2R,5S)-2-isopropyl-5-methylcyclohexyl 4-isocyanobenzoate (77 mg, 0.3 mmol) and 4-isocyano-4'-(1,2,2-triphenylvinyl)-1-biphenyl-1-anisol (121 mg, 0.3 mmol) in chlorobenzene (3 mL) was added, the reaction mixture was stirred at 25°C for 2 min, then the flask was taken out of the glove box and the reaction mixture was poured into methanol (100 mL) to precipitate the copolymer product, the orange copolymer solid was collected by filtration, and dried in vacuum at 40°C to a constant weight (149 mg, 91% yield). The product obtained is soluble thoroughly in CHCl₃ and THF at 25°C.

Calculation for the activity of catalyst

A=
$$m_{polymer} / (n_{cat.} \cdot t)$$

A: the activity of (co)polymerization (g of polymer/($mol_{cat.}\cdot h$)), $m_{polymer}$: the mass of (co)polymer (g), t: the reaction time of (co)polymerization (h), $n_{cat.}$: molar amount of catalyst (mol).

$$n_{cat.} = m_{catalyst} / M_{catalyst}$$

m_{catalyst}: the mass of catalysts (g), and M_{catalysts}: the relative molecular weight of catalyst.

Calculation of the IMCI contents of the copolymers

The IMCI contents of the copolymers were calculated from the ¹H NMR spectra according to the following formula

$$\omega(\text{mol}\%)_{\text{IMCI}} = \{ [23(I_{\text{H3}}+I_{\text{H4}})] / [19(I_{\text{H1}}+I_{\text{H2}}+I_{\text{H3}}+I_{\text{H4}})] \} \times 100$$

In which I_{H1} is the integration of the peak at 7.05 ppm which assigned to the aryl protons of ITPPA units and the β -H of the aryl ring of IMCI units. I_{H2} is the integration of the peak at 5.76 ppm which assigned to the α -H of the aryl ring of IMCI units. I_{H3} is the integration of the peak at 4.86 ppm ascribed to the proton of the cyclohexyl carbon connected with the oxygen. I_{H4} is the integration of the peaks between 0.3 to 2.5 ppm which assigned to the rest protons of the cyclohexyl group as well as the substituted methyl and the isopropyl.

The IMCI contents of poly(D-IMCI-*co*-ITPA)s and poly(L-IMCI-*co*-ITPA)s were calculated from the ¹H NMR spectra according to the following formula:

$$\omega(\text{mol}) \text{IMCI} = \{ [5 \times I_{\text{H4}} - 6(I_{\text{H1}} + I_{\text{H2}} + I_{\text{H3}})] / [12(I_{\text{H1}} + I_{\text{H2}} + I_{\text{H3}})] \} \times 100$$

In which I_{H1} is the integration of the peak at 7.05 ppm which assigned to the aryl protons of ITPPA units and the H of the aryl ring of IMCI units. I_{H2} is the integration of the peak at 5.76 ppm which assigned to the H of the aryl ring of IMCI and ITPA units. I_{H3} is the integration of the peak at 4.86 ppm ascribed to the proton of the cyclohexyl carbon connected with the oxygen and the proton of the isopropyl carbon connected with the

oxygen. I_{H4} is the integration of the peaks between 0.3 to 2.5 ppm which assigned to the rest protons of the cyclohexyl group as well as the substituted methyl and the isopropyl and the rest protons of the isopropyl.

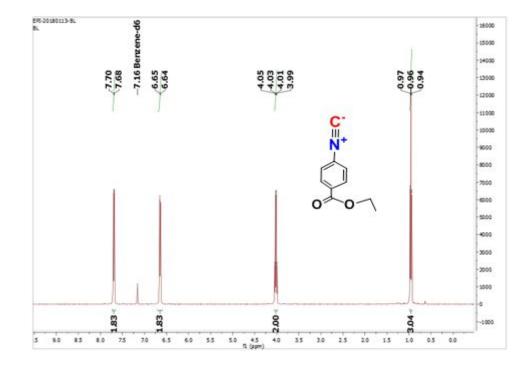


Figure S1. ¹H NMR spectrum of 4-ethoxycarbonyl phenyl isocyanide (a).

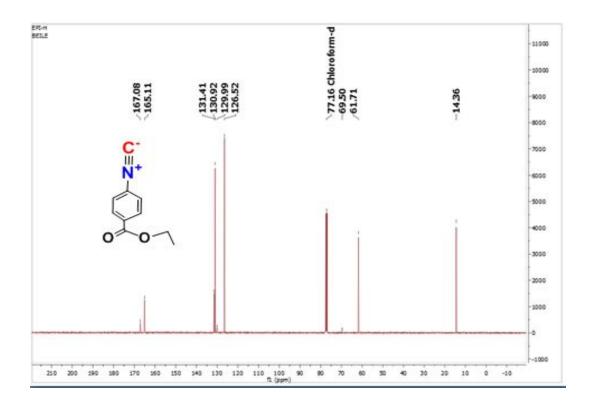


Figure S2. ¹³C NMR spectrum of 4-ethoxycarbonyl phenyl isocyanide (a).

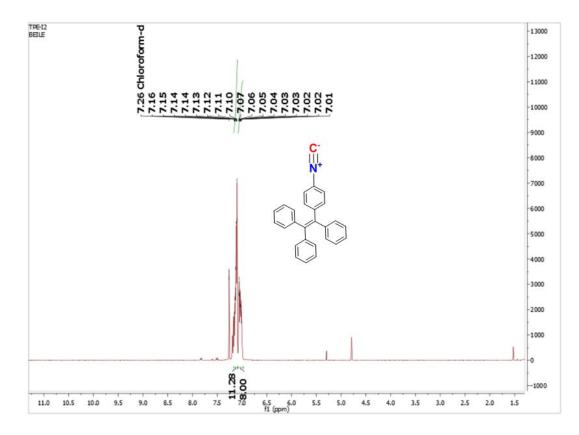


Figure S3. ¹H NMR spectrum of 4-isocysno-4-(1,1,2-triphenylvinyl)-1-phenyl (ITPP) (b).

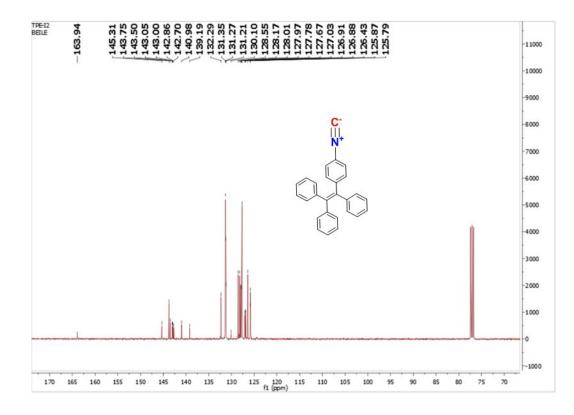


Figure S4. ¹³C NMR spectrum of 4-isocysno-4-(1,1,2-triphenylvinyl)-1-phenyl (ITPP) (b).

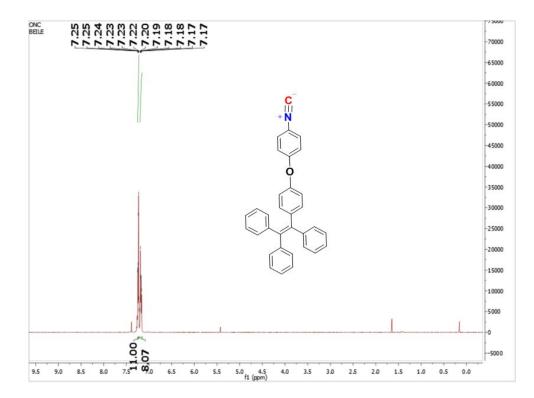


Figure S5. ¹H NMR spectrum of 4-isocysno-4-(1,1,2-triphenylvinyl)-1-phenoxy-1-anisol (ITPPA) (c).

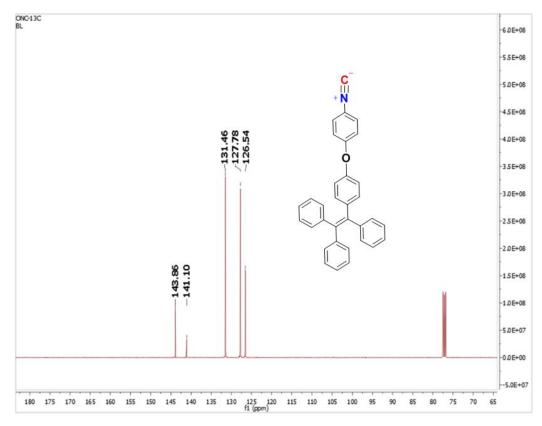


Figure S6. ¹³C NMR spectrum of 4-isocysno-4-(1,1,2-triphenylvinyl)-1-phenoxy-1-anisol (ITPPA) (c).

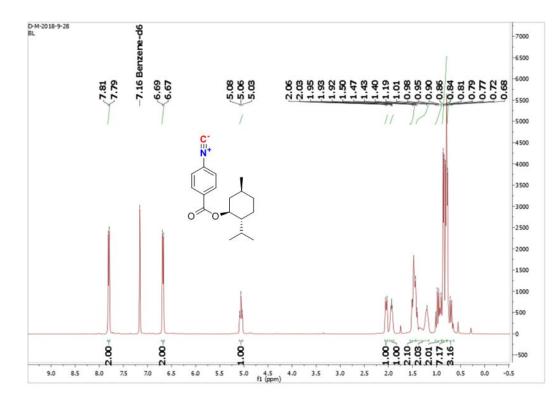


Figure S7. ¹H NMR spectrum of (*1S*, *2R*, *5S*)-2-isopropyl-5-metheylcyclohexyl-4-isocyanobezoate (**d**).

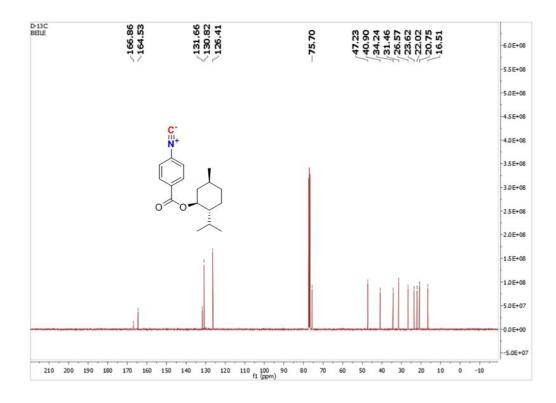


Figure S8. ¹³C NMR spectrum of (*1S*, 2*R*, 5*S*)-2-isopropyl-5-metheylcyclohexyl-4-isocyanobezoate (d).

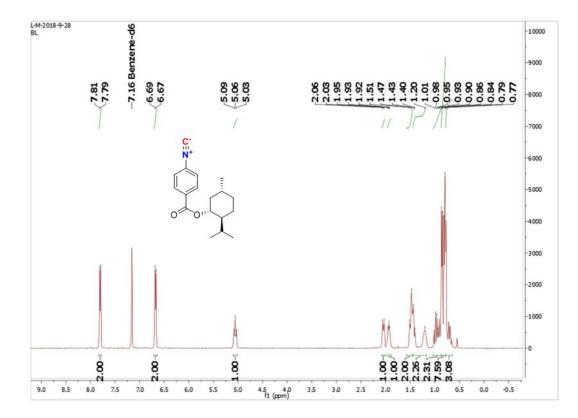


Figure S9. ¹H NMR spectrum of (*1R*, *2S*, *5R*)-2-isopropyl-5-metheylcyclohexyl-4-isocyanobezoate (e).

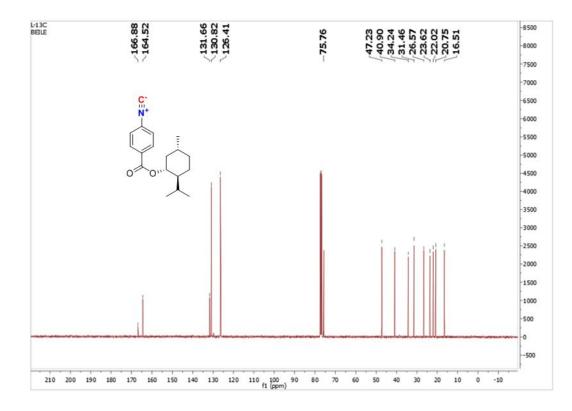


Figure S10. ¹³C NMR spectrum of (*1R*, *2S*, *5R*)-2-isopropyl-5-metheylcyclohexyl-4-isocyanobezoate (e).

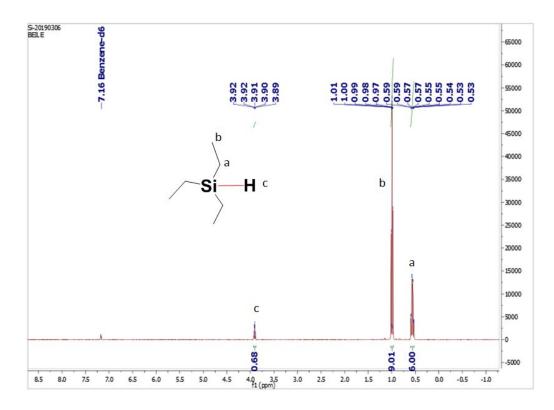


Figure S11. ¹H NMR spectra of Et₃SiH.

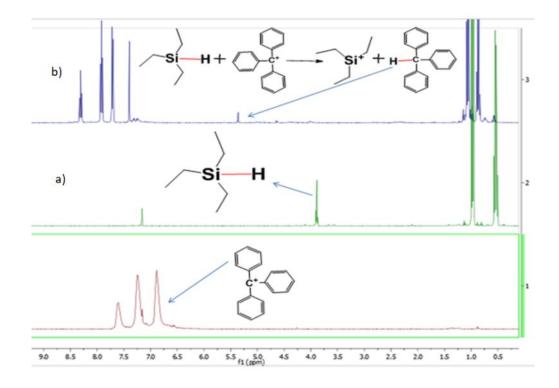


Figure S12. (a) ¹H NMR spectra of Et_3SiH and (b) corresponding cationic species *in situ* generated by reaction of Et_3SiH with catalyst $[Ph_3C][B(C_6F_5)_4]$.

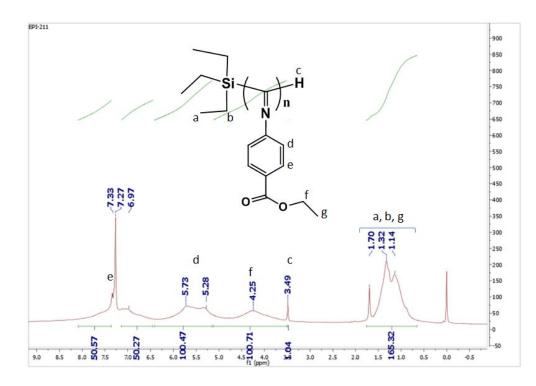


Figure S13. ¹H NMR of poly(EPI) with Et₃SiH as end group (Table 1, entry 4).

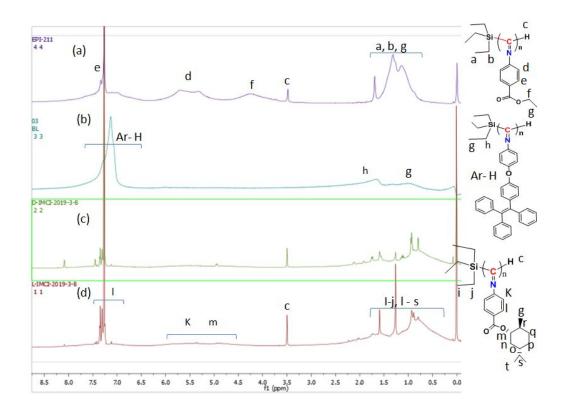


Figure S14. (a) ¹H NMR of poly(EPI) (Table 2, entry 4), (b) Poly(ITPPA) (Table 2, entry 8), (c) Poly(D-IMCI) (Table 2, entry 9) and (d) Poly(L-IMCI) (Table 2, entry 10).

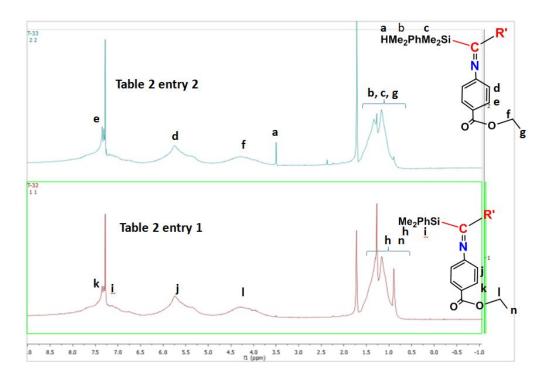


Figure S15. ¹H NMR of poly(EPI) with different silane source as end group (Table 2, entries 1–2).

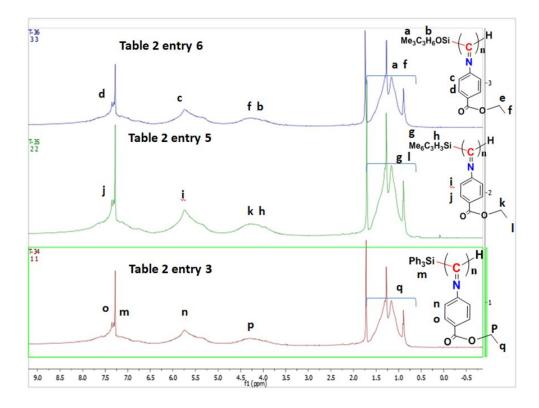


Figure S16. ¹H NMR of poly(EPI) with different silane source as end group (Table 2, entries 3,5–6).

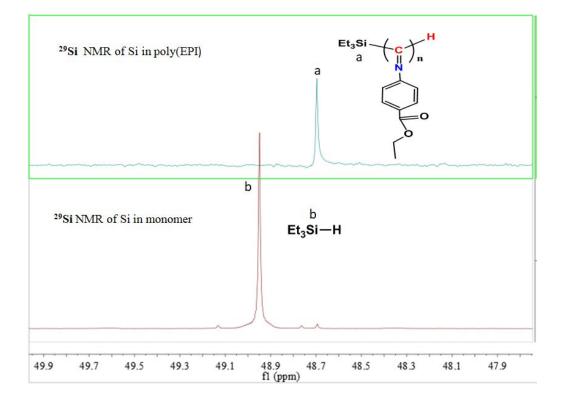


Figure S17. ²⁹SiNMR spectrum of Et₃SiH and Et₃Si-end-functionalized poly(EPI) (Table 2, entry 4).

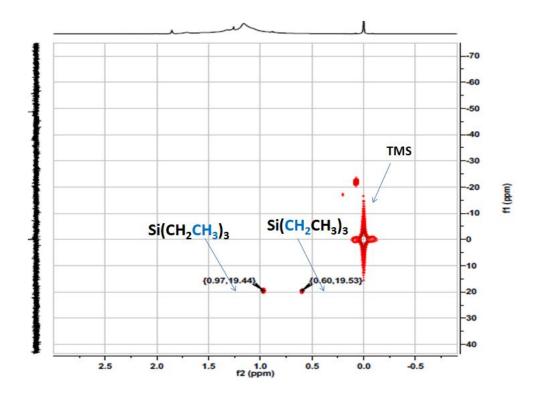


Figure S18. ($^{1}H + ^{29}Si$) gHMBC NMR spectrum of Et₃Si-end-functionalized poly(EPI) (Table 2, entry 4).

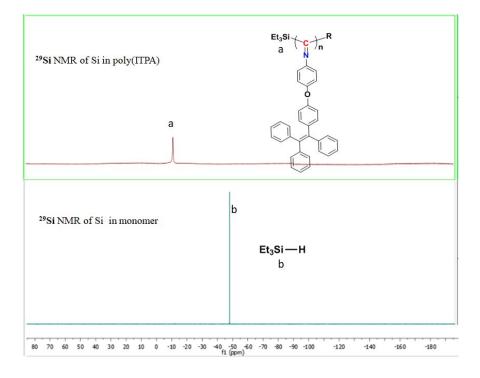


Figure S19. ²⁹Si-NMR spectrum of Et₃SiH and Et₃Si-end-functionalized poly(ITPPA) (Table 2, entry 8).

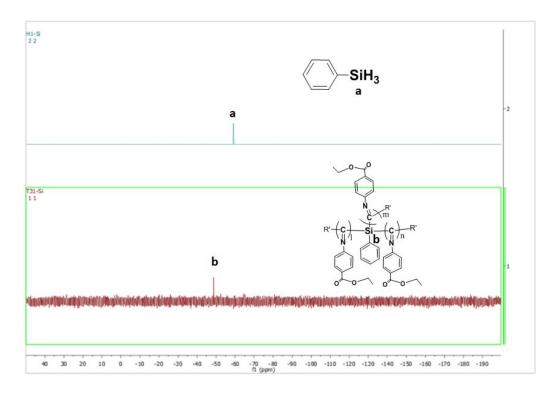


Figure S20. ²⁹Si-NMR spectrum of PhSiH₃ and PhSi-end-functionalized star poly(EPI) (Table 3, entry1).

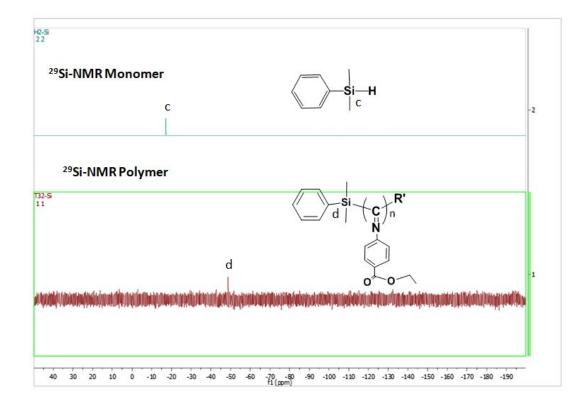


Figure S21. ²⁹Si-NMR spectrum of PhMe₂Si-end-functionalized poly(EPI) (Table 2, entry 1).

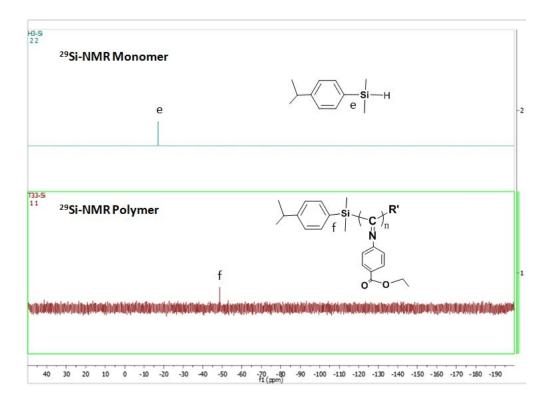


Figure S22. ²⁹Si-NMR spectrum of (4-^{*i*}PrC₆H₄)Me₂Si-end-functionalized poly(EPI) (Table 2, entry 2).

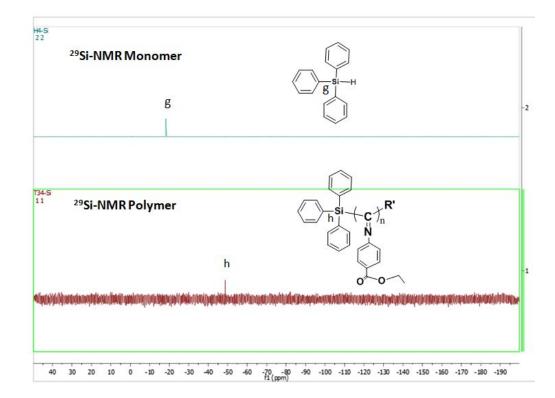


Figure S23. ²⁹Si-NMR spectrum of Ph₃Si-end-functionalized poly(EPI) (Table 2, entry 3).

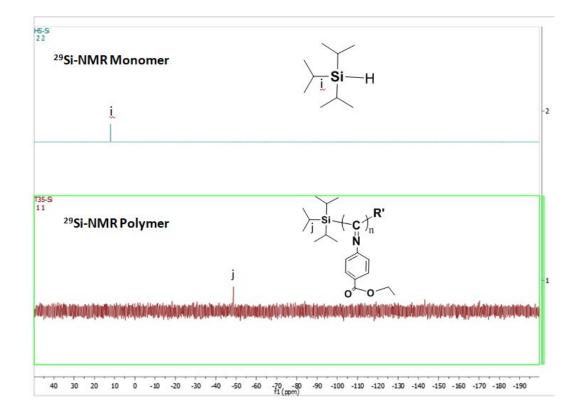


Figure S24. ²⁹Si-NMR spectrum of ^{*i*}Pr₃Si-end-functionalized poly(EPI) (Table 2, entry 5).

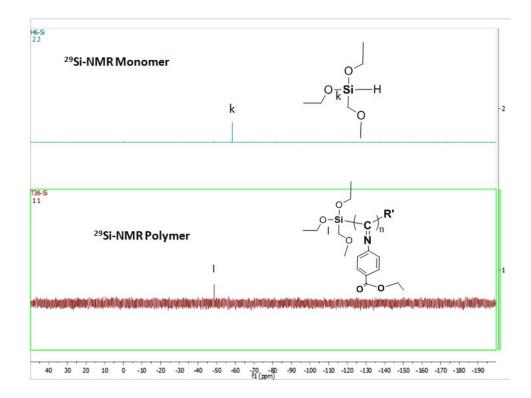


Figure S25. ²⁹Si-NMR spectrum of (OEt)₃Si-end-functionalized poly(EPI) (Table 2, entry 6).

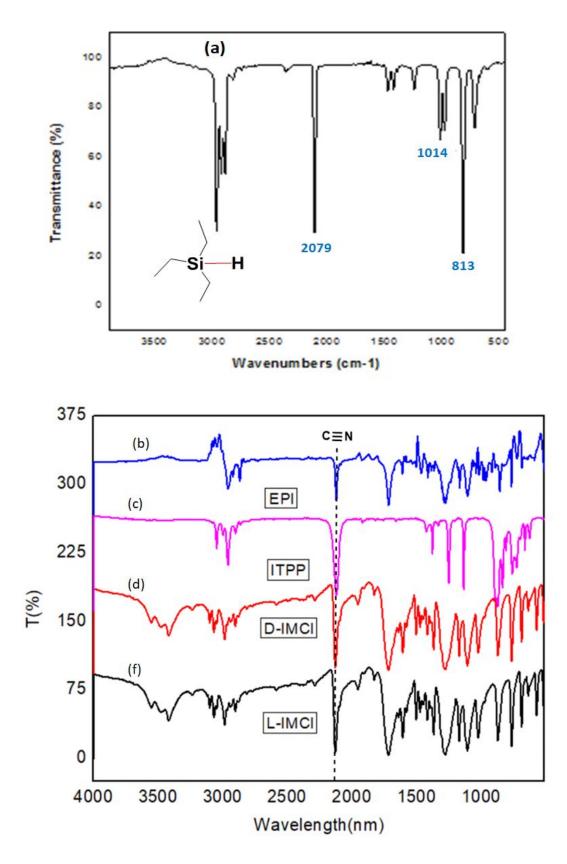


Figure S26. FT-IR spectra of monomers (a) Et₃SiH, (b) EPI, (c) ITPPA, (d) D-IMCI and (e) L-IMCI.

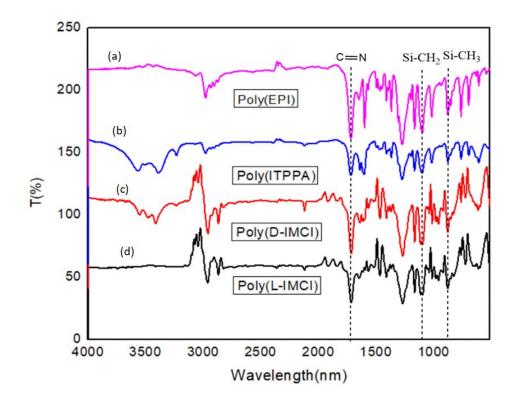


Figure S27. FT-IR spectra of (a) Poly(EPI) (Table 2, entry 4), (b) Poly(ITPPA) (Table 2, entry 8), (c) Poly(D-IMCI) (Table 2, entry 9), (d) Poly(L-IMCI) (Table 2, entry 10).

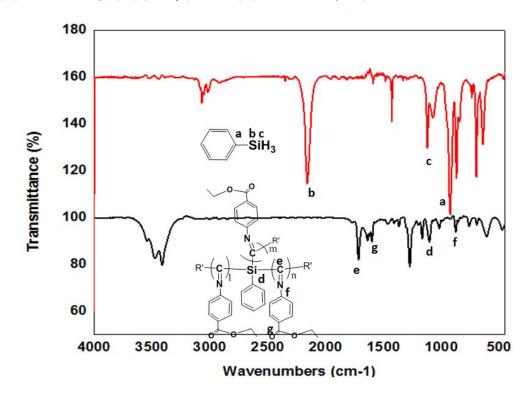


Figure S28. FT-IR spectra of PhSiH₃ and PhSi-end-functionalized star poly(EPI) (Table 3, entry1).

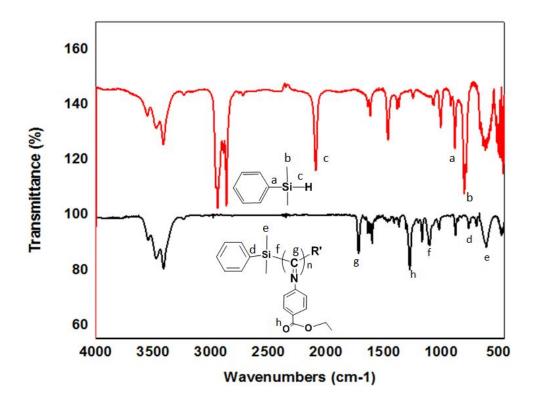


Figure S29. FT-IR spectra of PhMe₂SiH and PhMe₂Si-end-functionalized poly(EPI) (Table 2, entry 1).

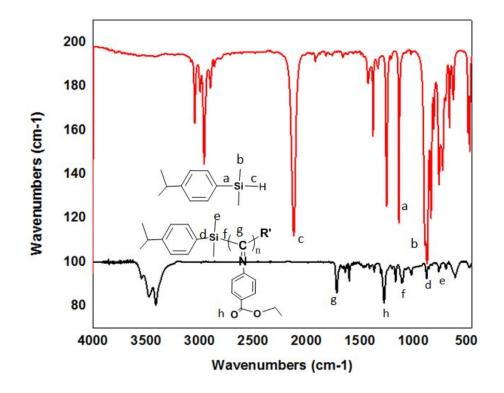


Figure S30. FT-IR spectra of (4-*i*PrC₆H₄)Me₂SiH and (4-*i*PrC₆H₄)Me₂Si-end-functionalized poly(EPI) (Table 2, entry 2).

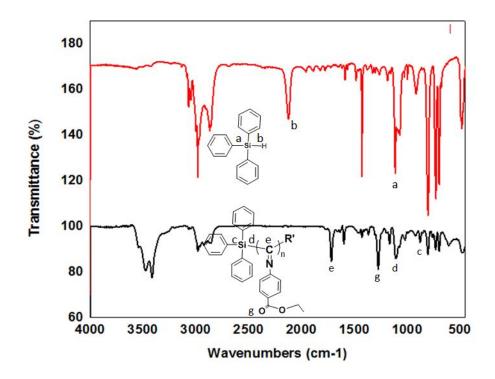


Figure S31. FT-IR spectra of Ph₃SiH and Ph₃Si-end-functionalized poly(EPI) (Table 2, entry 3).

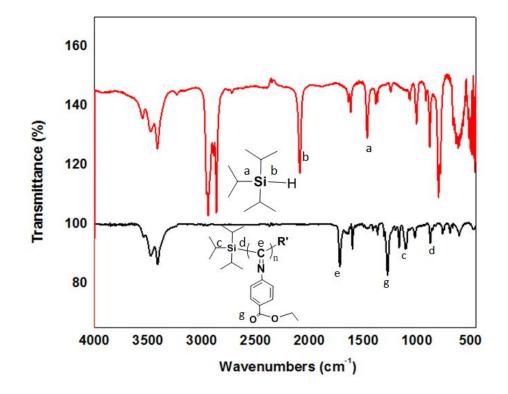


Figure S32. FT-IR spectra of ^{*i*}Pr₃SiH and ^{*i*}Pr₃Si-end-functionalized poly(EPI) (Table 2, entry 5).

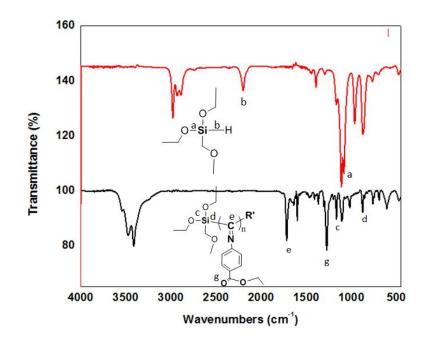


Figure S33. FT-IR spectra of (OEt)₃SiH and (OEt)₃Si-end-functionalized poly(EPI) (Table 2, entry 6).

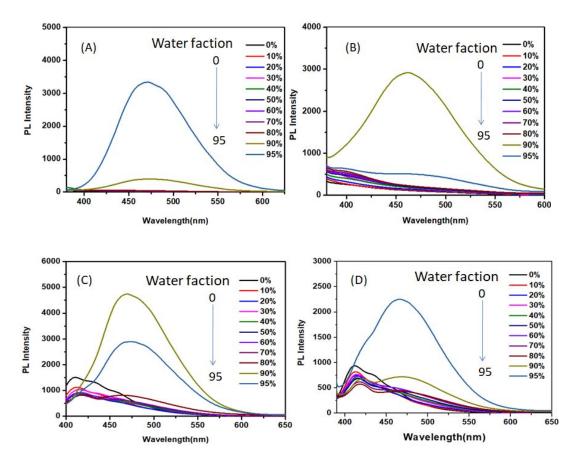


Figure S34. Plots of fluorescence intensity vs water fraction in THF/water mixture (0.01 mg/mL) (A) ITPP,

(B) Poly(ITPP) (Table 2, entry 7), (C) ITPPA, (D) Poly(ITPPA) (Table 3, entry 8) (conditions: EX wavelength: 290 nm, EX slit: 5 nm, EM slit: 5 nm, 700 V).

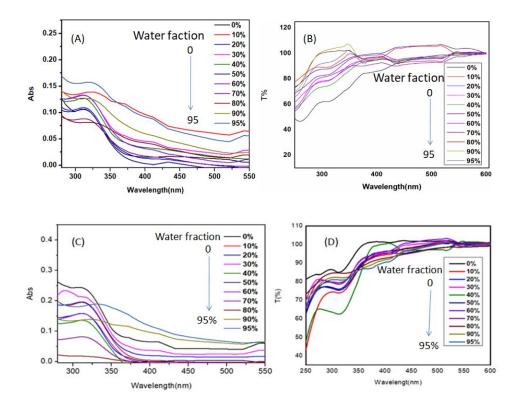


Figure S35. UV absorption and transmittance spectra of (A) ITPP, (B) Poly(ITPP) (Table 2, entry 7), (C) ITPPA and (D) Poly(ITPPA) (Table 2, entry 8) with the water fraction in THF/water mixture ranging from 0 to 95%.

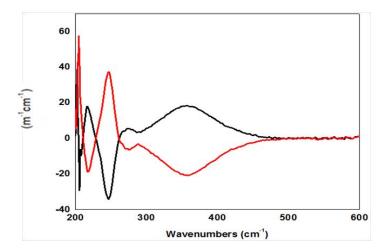


Figure S36. CD Spectra ofEt₃Si-end-functionalized poly(D-IMCI) and Et₃Si-end-functionalizedpoly(L-IMCI) (Table 2, entries 9–10).

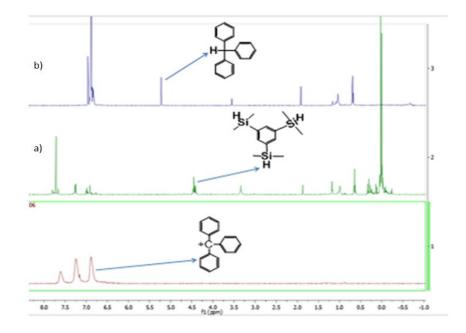


Figure S37. (a) ¹H NMR spectra of 1,3,5-(SiMe₂H)₃-C₆H₃ and (b) corresponding cationic species *in situ* generated by reaction of Poly-Si-H with catalyst [Ph₃C][B(C₆F₅)₄].

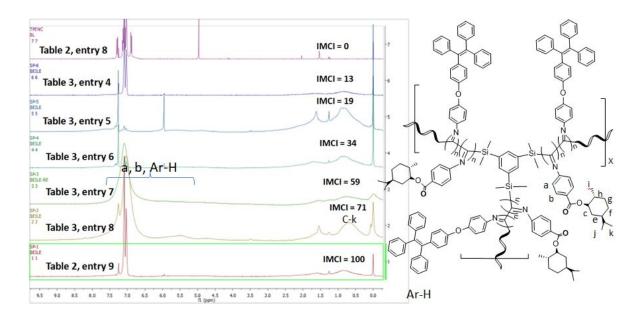


Figure S38. ¹H NMR spectra of Poly(ITPPA) (Table 2, entry 8), Poly(D-IMCI-*co*-ITPPA)s (Table 3, entries 4–8) and Poly(D-IMCI) (Table 3, entry 9).

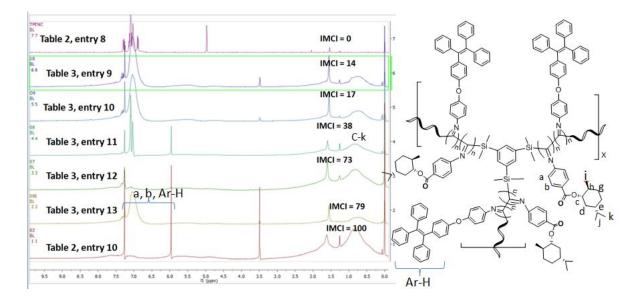


Figure S39. ¹H NMR spectra of Poly(ITPPA) (Table 2, entry 8), Poly(L-IMCI-*co*-ITPPA)s (Table 4, entries 9–13) and Poly(L-IMCI) (Table 3, entry 10).

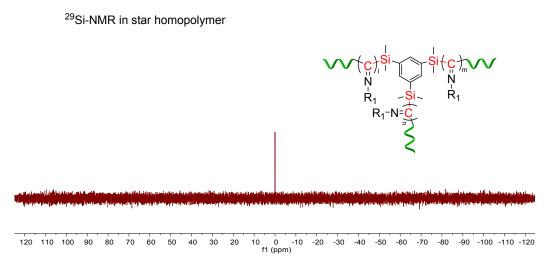


Figure S40. ²⁹Si-NMR spectrum of poly(D-IMCI) (Table 3, entry 2).

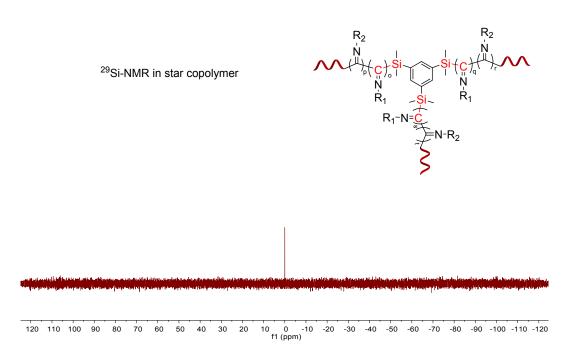
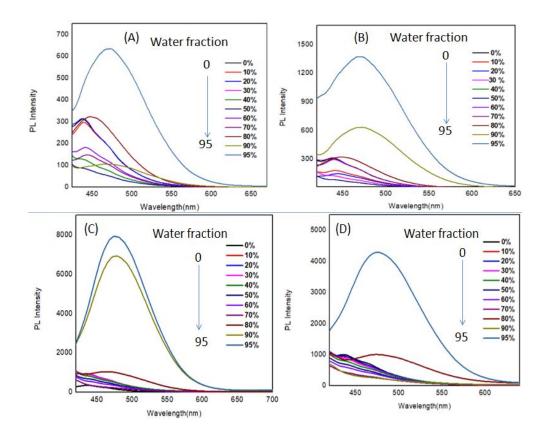


Figure S41. ²⁹Si-NMR spectrum of poly(D-IMCI-co-ITPPA) (Table 3, entry 6).



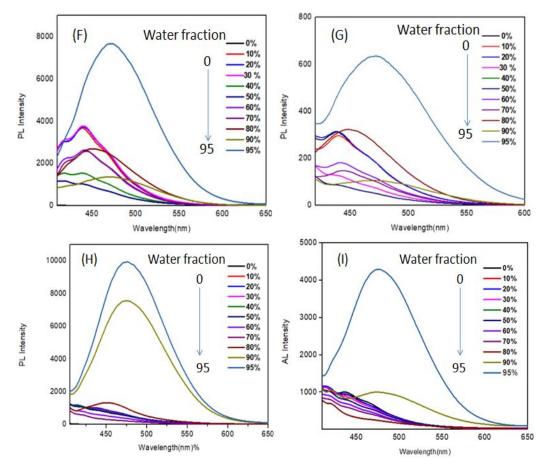


Figure S42. Plots of fluorescence intensity vs water fraction in THF/water mixture (0.01 mg/mL) (A) Poly(D-IMCI-*co*-ITPPA) (Table 3, entry 4), (B) Poly(D-IMCI-*co*-ITPPA) (Table 3, entry 5), (C) Poly(D-IMCI-*co*-ITPPA) (Table 3, entry 6), (D) Poly(D-IMCI-*co*-ITPPA) (Table 3, entry 8), (F) Poly(L-IMCI-*co*-ITPPA) (Table 3, entry 9), (G) Poly(L-IMCI-*co*-ITPPA) (Table 3, entry 10), (H) Poly(L-IMCI-*co*-ITPPA) (Table 3, entry 11) and (I) Poly(L-IMCI-*co*-ITPPA) (Table 3, entry 13) (conditions: EX wavelength: 290 nm, EX slit: 5 nm, EM slit: 5 nm, 700 V).

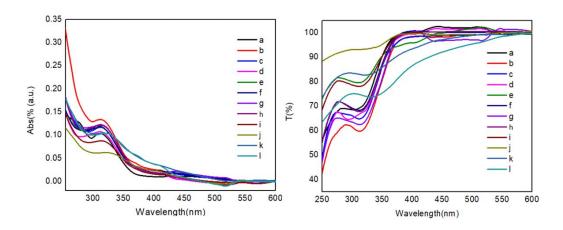


Figure S43. UV absorption and transmittance spectra of (a) baseline, (b) ITPPA, (c) Poly(ITPPA) (Table 2, entry 8), (d-g) (Poly(D-IMCI-*co*-ITPPA)s (Table 3, entries 4–8) and (h-l) Poly(L-IMCI-*co*-ITPPA)s (Table 3, entries 9–13) in THF.

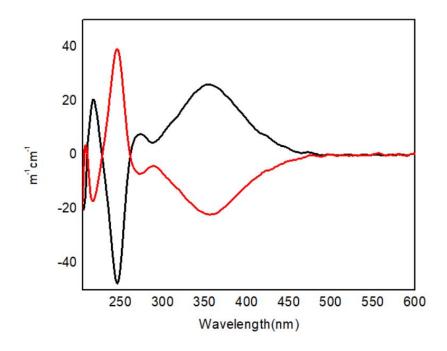


Figure S44. CD spectra of poly(D-IMCI) (Table 3, entry 2) and poly(L-IMCI) (Table 3, entry 3).

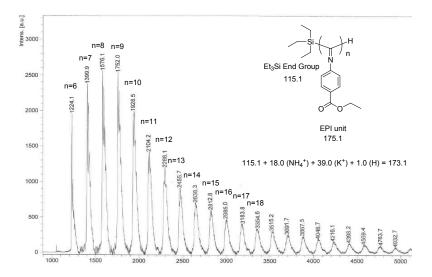
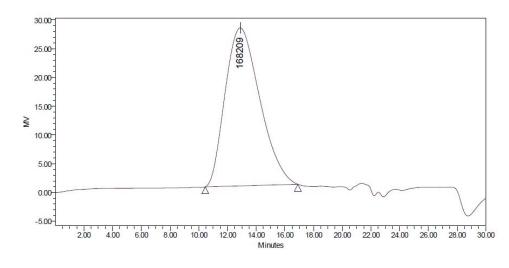


Figure S45. MALDI-TOF mass spectrum of EPI oligomer obtained by the binary $[Ph_3C][B(C_6F_5)_4]/Et_3SiH$ system.

GPC Curve Synthesis of Silane-End-Functionalized Linear Poly(aryl isocyanide)s through Cationic Polymerization of Aryl Isocyanides Promoted by Cationic Initiator Borates or Borane in the Presence of Hydrosilane.



Broad Unknown Modified Universal Peak Table

		Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
Γ	1						59102	229311	168209	<mark>54</mark> 9354	912038	3.879944	2.395673

Figure S46. GPC curve of Poly(EPI)by cationic catalyst[Ph_3C][$B(C_6F_5)_4$ in Table 1, entry 1.

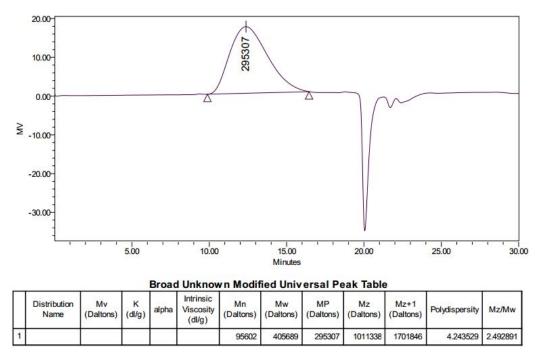
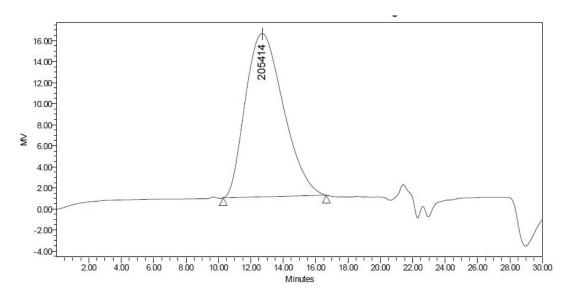


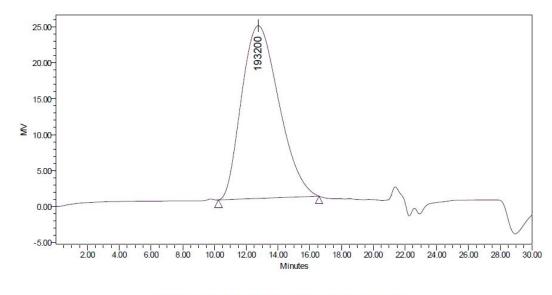
Figure S47. GPC curve of Poly(EPI) by cationic catalyst [(Et₃Si)₂H][B(C₆F₅)₄] in Table 1, entry 2.



Broad Unknown Modified Universal Peak Table

2-12 2	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						77439	281900	205414	657661	1081943	3.640275	2.332958

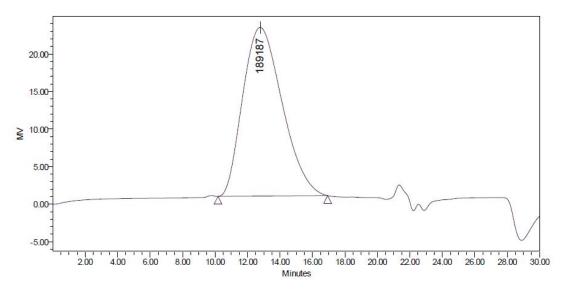
Figure S49. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 3.



Broad Unknown Modified Universal Peak Table

8	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						75636	272626	193200	648352	1087787	3.604437	2.378169

Figure S49. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 4.



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						67507	268870	189187	665567	1132481	3.982856	2.475423

Figure S49. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 5.

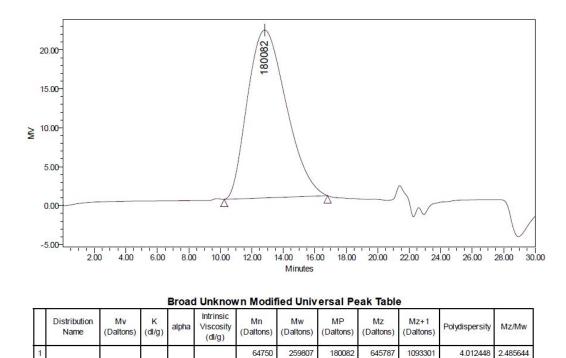
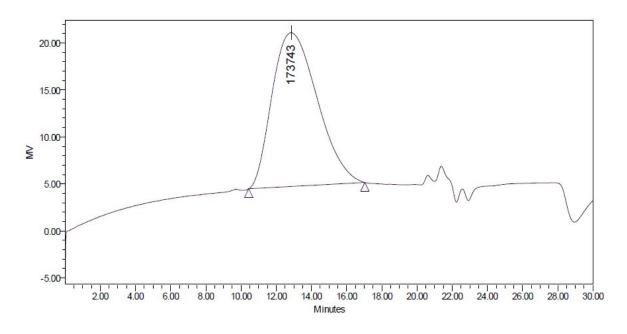


Figure S50. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 6.



	Broad Unknown Modified Universal Peak Table													
	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw		
1				3 		55478	237557	173743	589276	975952	4.282008	2.480563		

Figure S51. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 7.

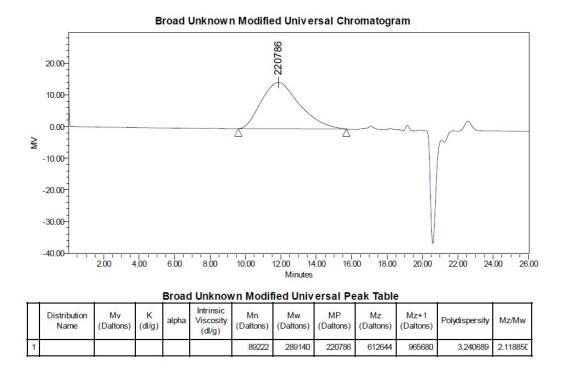
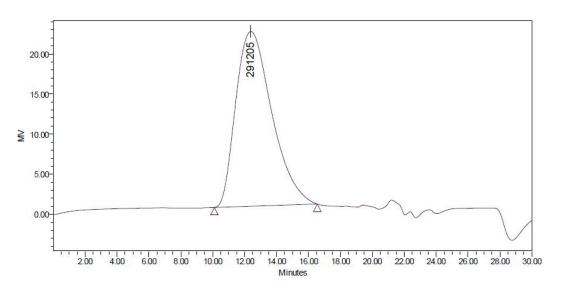


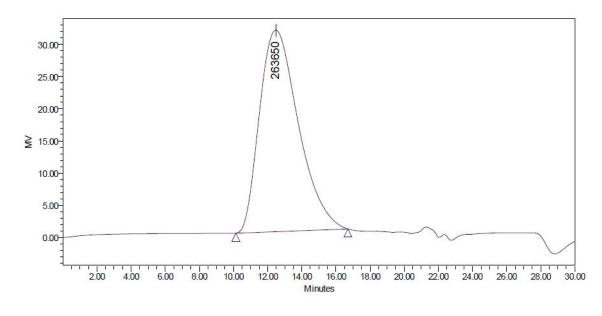
Figure S52. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 8.



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						95716	355640	291205	783629	1253092	3.715588	2.203431

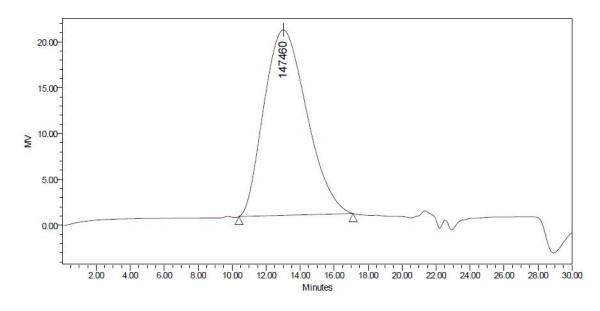
Figure S53. GPC curve of Et₃Si-End-Functionalized Poly(EPI) in Table 1, entry 9.



Broad Unknown Modified Universal Peak Table

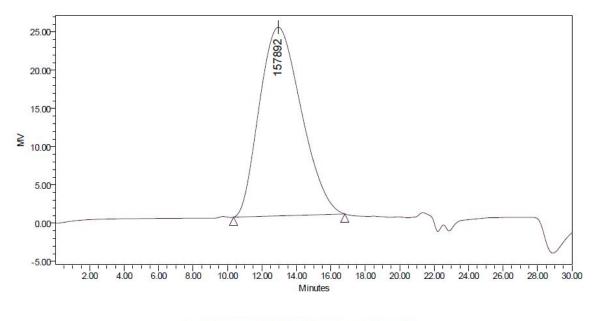
2		Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
	1						86536	326841	263650	733599	<mark>1189990</mark>	3.776956	2.244511

Figure S54. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 10.



	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						53165	224438	147460	579055	988682	4.221503	2.580021

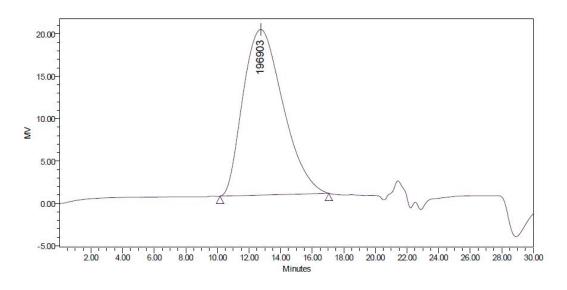
Figure S55. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 11.



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						<mark>61</mark> 134	231339	157892	576059	984709	3.784110	2.490105

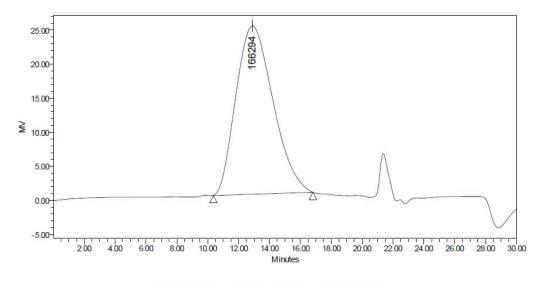
Figure S56. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 12.



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						63580	286366	196903	726279	1225562	4.504001	2.536188

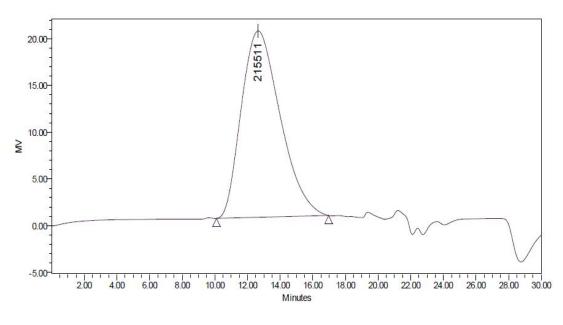
Figure S57. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 13.



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		Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
Ι	1						62297	239798	166294	590950	1002743	3.849237	2.464370

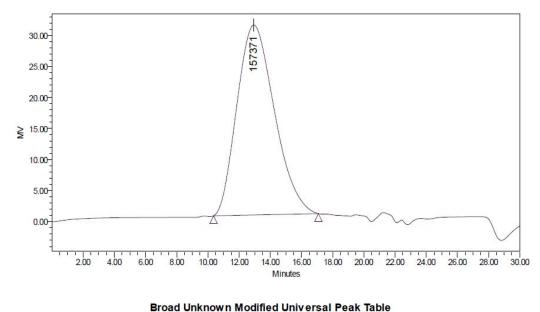
Figure S58. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 14.



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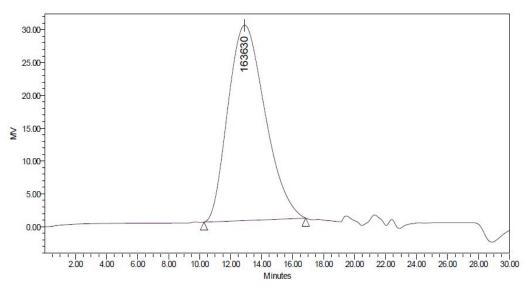
0 1	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						72213	295664	2155 <mark>11</mark>	721397	1226832	4.094348	2.439918

Figure S59. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 15.



	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						58365	232014	157371	585538	1012901	3.975224	2.523718

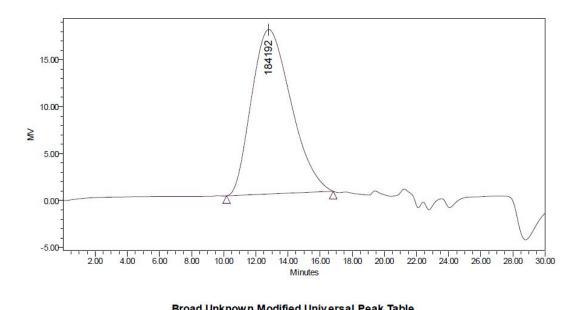
Figure S60. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 16.



Broad Unknown Modified Universal Peak Table

		Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
I	1						61905	238908	163630	596029	1033177	3.859246	2.494803

Figure S61. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 17.



			Dioac	Olivilor	an moun	ieu oniv	ersuir e				
Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw

262249

184192

672196

1168290

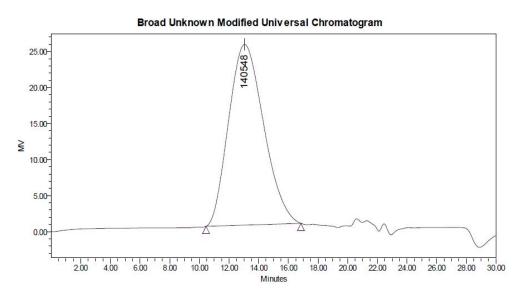
4.147401

2.563199

63232

Figure S62. GPC curve of Et₃Si-end-functionalized poly(EPI) in Table 1, entry 18.

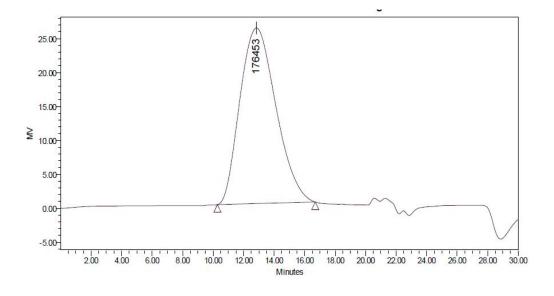
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Broad Unknown Modified Unive	ersal Peak Ta	ble
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	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						57855	209883	140548	521933	902777	3.62773 <mark>4</mark>	2.486782

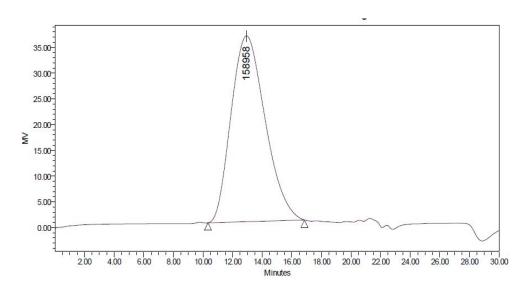
Figure S63. GPC curve of PhMe₂Si-end-functionalized poly(EPI) in Table 2, entry 1.



Broad Unknown Modified Universal Pea	ak ladie
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	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						71822	260754	176453	629250	1064576	3.630539	2.413193

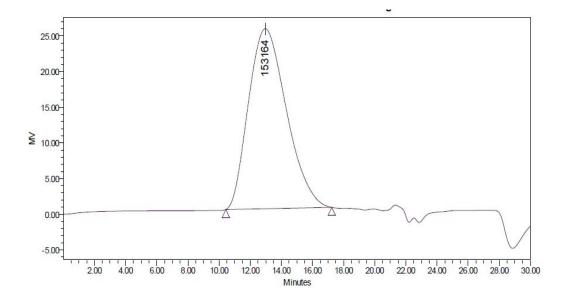
Figure S64. GPC curve of (4-^{*i*}PrC₆H₄)Me₂Si-end-functionalized poly(EPI) in Table 2, entry 2.



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						64122	226677	158958	540900	928216	3.535077	2.386221

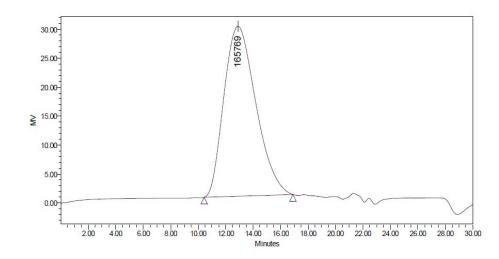
Figure S65. GPC curve of Ph₃Si-end-functionalized poly(EPI) in Table 2, entry 3.



Broad Unknown Modified Universal Peak Table

	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						54327	220242	153164	546000	925151	4.054013	2.479095

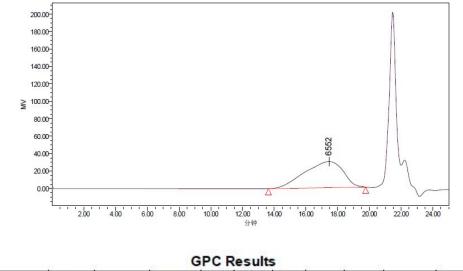
Figure S66. GPC curve of ^{*i*}Pr₃Si -end-functionalized poly(EPI) in Table 2, entry 5.



Broad Unknown Modified Universal Peak Table

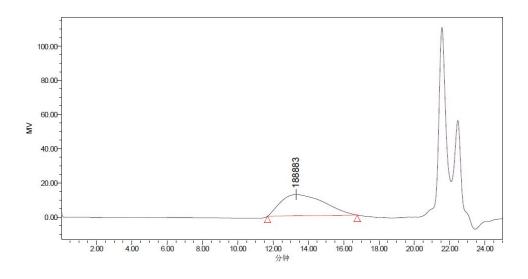
	Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1						62394	225135	165769	522785	869546	3.608279	2.322095

Figure S67. GPC curve of (OEt)₃Si-end-functionalized poly(EPI) in Table 2, entry 6.



	Dist Name	100000000000000000000000000000000000000	Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw	Mz+1/Mw
1		17.457	17.457	17.457	6920	11488	6552	20113	32894	1.750815	2.863346

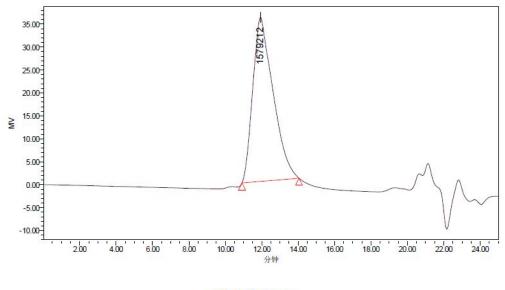
Figure S68. GPC curve of (OEt)₃Si-end-functionalized poly(EPI) in Table 2, entry 7.



GPC Results

	Dist Name	1. S. S. S. L. S.	Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw
1		13.307	13.307	13.307	68717	232328	188883	626163	1032547	2.695171

Figure S69. GPC curve of Et₃Si-end-functionalized poly(ITPP) in Table 2, entry 8.



GPC Results

	Dist Name		Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw
1		11.930	11.930	11.930	711137	1643957	1579212	3131660	4798522	1.904953

Figure S70. GPC curve of Et₃Si-end-functionalized poly(ITPPA) in Table 2, entry 9.

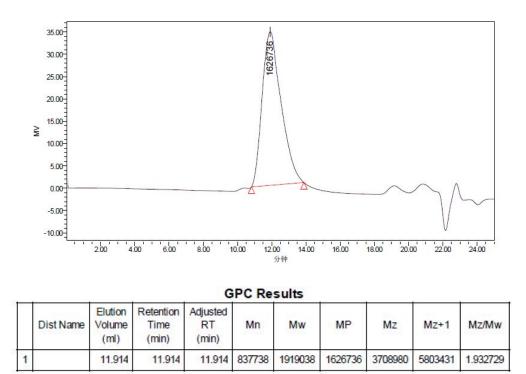
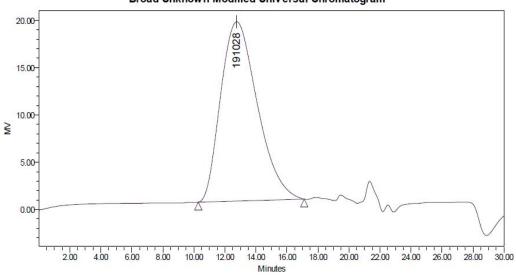


Figure S71. GPC curve of Et₃Si-end-functionalized poly(D-IMCI) in Table 2, entry 10



Broad Unknown Modified Universal Chromatogram

Broad Unknown Modified Universal Peak Table

		Distribution Name	Mv (Daltons)	K (dl/g)	alpha	Intrinsic Viscosity (dl/g)	Mn (Daltons)	Mw (Daltons)	MP (Daltons)	Mz (Daltons)	Mz+1 (Daltons)	Polydispersity	Mz/Mw
1	1						65341	262042	191028	623403	1043958	4.010387	2.379017

Figure S72. GPC curve of PhSi-end-functionalized star poly(EPI) in Table 3, entry 1.

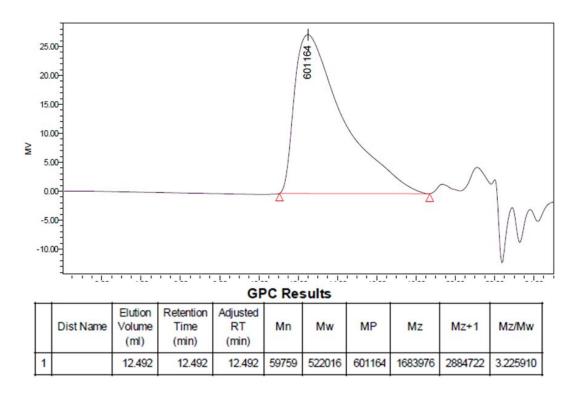


Figure S73. GPC curve of C₆H₃-1,3,5-(Me₂Si)₃-end-functionalized poly(D-IMCI) in Table 3, entry 2.

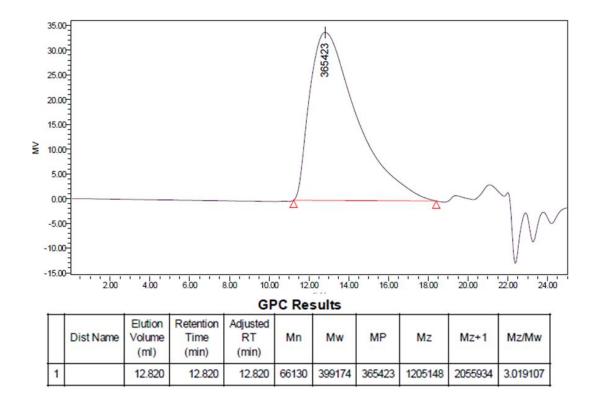


Figure S74. GPC curve of C₆H₃-1,3,5-(Me₂Si)₃-end-functionalized poly(L-IMCI) in Table 3, entry 3.

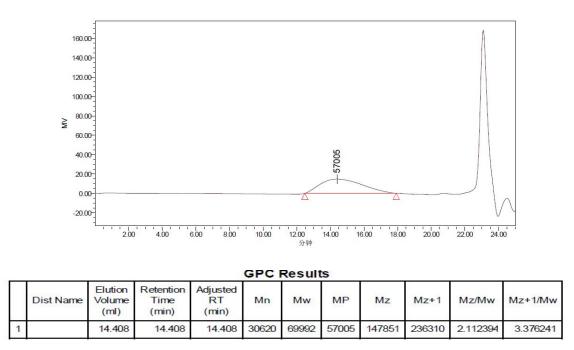
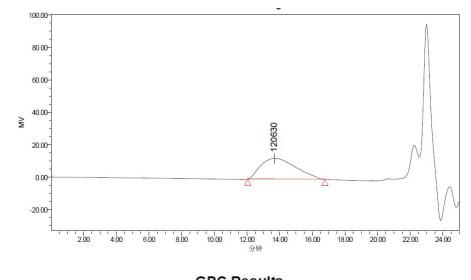


Figure S75. GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(D-IMCI-*co*-ITPPA) in Table 3, entry 4.



_	GPC Results												
	Dist Name	Elution Volume (ml)	Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw			
1		13.681	13.681	13.681	61084	143429	120630	300062	469015	2.092061			

Figure S76. GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(D-IMCI-*co*-ITPPA) in Table 3, entry 5.

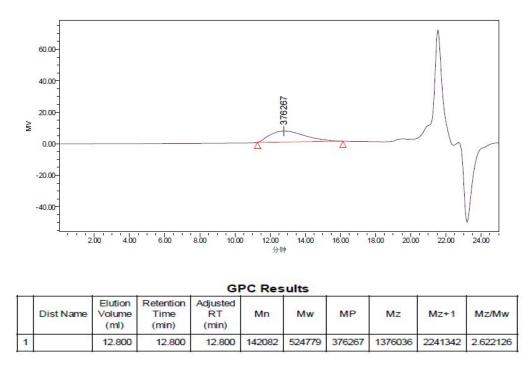
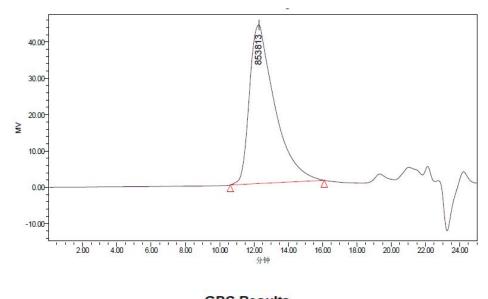


Figure S77. GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(D-IMCI-*co*-ITPPA) in Table 3, entry 6.



	GPC Results												
	Dist Name	Elution Volume (ml)	Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw			
1		12.278	12.278	12.278	251163	925273	853813	2518422	6204822	2.721816			

Figure S78. GPC curve of C₆H₃-1,3,5-(Me₂Si)₃-end-functionalized poly(D-IMCI-*co*-ITPPA) in Table 3, entry 7.

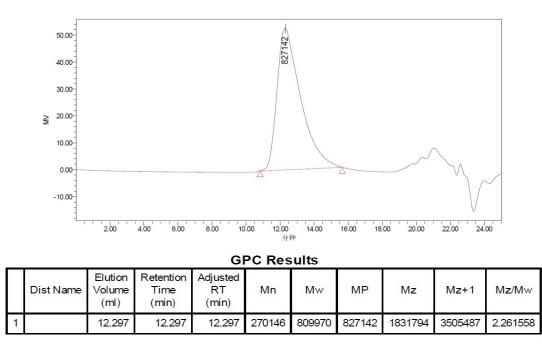
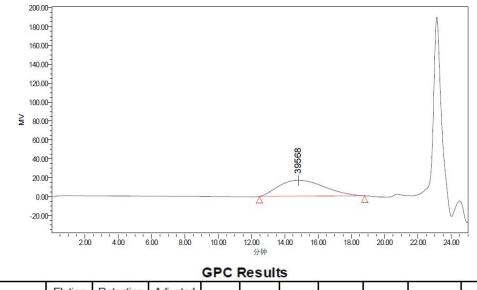


Figure S79. GPC curve of C₆H₃-1,3,5-(Me₂Si)₃-end-functionalized poly(D-IMCI-*co*-ITPPA) in Table 3, entry 8.



	Dist Name		Retention Time (min)	Adjusted RT (min)	Mn	Mw	MP	Mz	Mz+1	Mz/Mw	Mz+1/Mw
1		14.821	<mark>14.821</mark>	<mark>14</mark> .821	22567	55195	39568	<mark>126988</mark>	216504	2.300719	3.922520

Figure S80. GPC curve of C₆H₃-1,3,5-(Me₂Si)₃-end-functionalized poly(L-IMCI-*co*-ITPPA) in Table 3, entry 9.

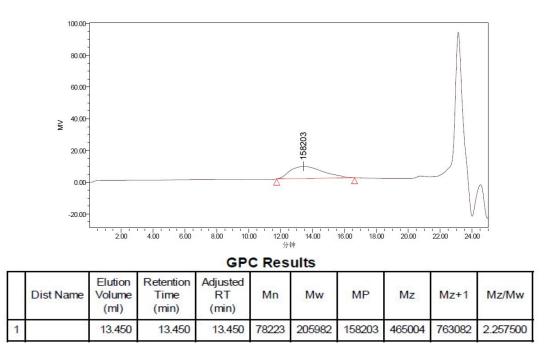


Figure S81. GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(L-IMCI-*co*-ITPPA) in Table 3, entry 10.

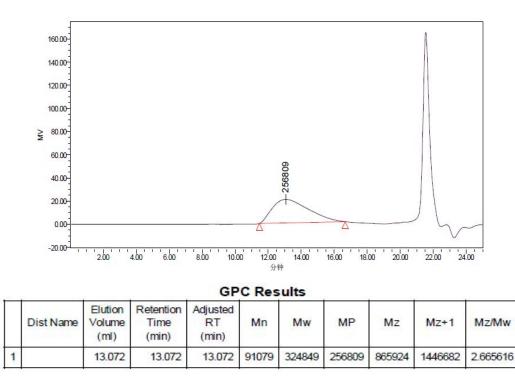


Figure S82. GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(L-IMCI-*co*-ITPPA) in Table 3, entry 11.

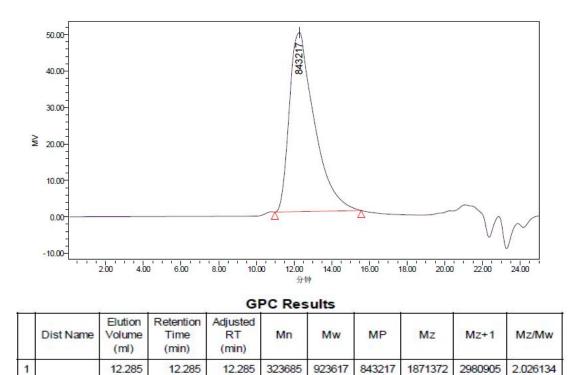


Figure S83. GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(L-IMCI-*co*-ITPPA) in Table 3, entry 12.

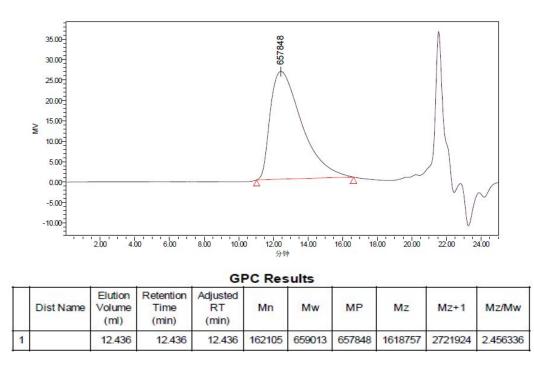


Figure S84.GPC curve of C_6H_3 -1,3,5-(Me₂Si)₃-end-functionalized poly(L-IMCI-*co*-ITPPA) in Table 3, entry 13.

COMPUTATIONALMETHODS

All calculations presented in this paper were performed using density functional theory (DFT) with the hybrid functional B3LYP⁵⁻⁷ as implemented in Gaussian 09 package.⁸ Geometry optimizations were carried out with the 6-31G(d, p) basis set. On the basis of the optimized geometries, more accurate energies were obtained by performing single-point calculations with a larger 6-311+G (2d, 2p) basis set. Using an empirical formula by Grimme et al,⁹⁻¹²dispersion effects were taken into account throughout geometry optimizations and single-point calculations. Solvation effects were also considered throughout geometry optimizations and single-point calculations, using a conductor-like polarizable continuum model (CPCM)¹³⁻¹⁶method with chlorobenzene as the solvent. Frequency calculations were performed at the same level of theory as in the optimizations to further confirm the nature of stationary points and to obtain zero-point energies (ZPE) and entropy effects. The energies reported in this paper are the free energies which have been corrected for dispersion, solvation, ZPE, and entropy effects.

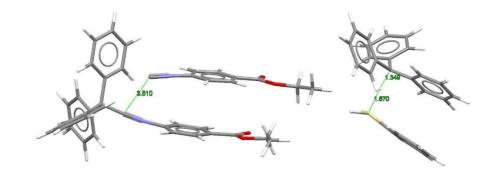
Density functional calculations

Initiation

We first considered which species is the one essentially initiating the polymerization. The adduct of Ph_3C^+ and a substrate **a** (see Ph_3C -(a)₁⁺ in Figure S85) was optimized, having an energy 1.6 kcal/mol higher than the reactant state of Ph_3C^+ + **a** (Figure S86). This indicates that Ph_3C^+ is an effective species to initiate the polymerization, especially considering that no transition state exists during the addition between Ph_3C^+ and **a** because the nature of this step is a simple one bond formation.

The species of A^+ (Figure S87) was also taken into account as an initiating species. Its formation should be activated by Ph_3C^+ via a hydride transfer transition state (see TS_A in Figure S85), which shows a barrier of 16.4 kcal/mol (Figure S87). Although such an activation is energetically accessible, the subsequent addition of the substrate **a** is unreachable with an accumulated barrier of 28.6 kcal/mol (see **TS1**' in Figure

S88 for structures and Figure S87 for energies). This means that A^+ is unable to work as a species to initiate the polymerization, at least in the first polymer chain.



a TS_A

Figure S85. Optimized structures of the adduct between Ph_3C^+ and a substrate **a** (Ph_3C -(**a**)₁⁺) and the transition state (**TS**_A) for the A⁺ formation activated by Ph_3C^+ .

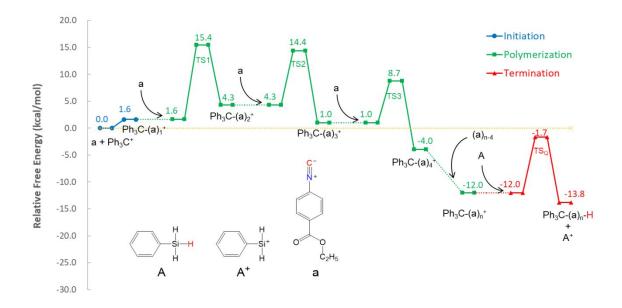


Figure S86. Free energy profile for the polymerization initiated by Ph₃C⁺.

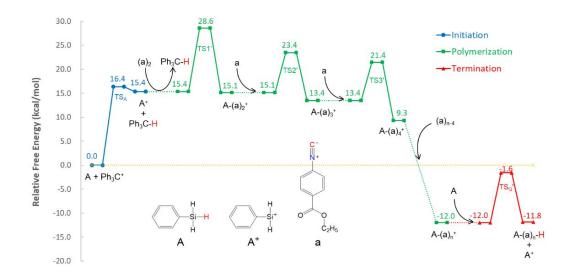
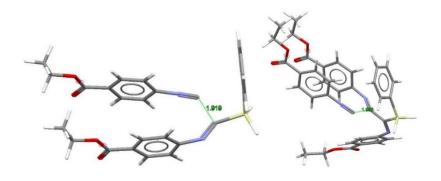
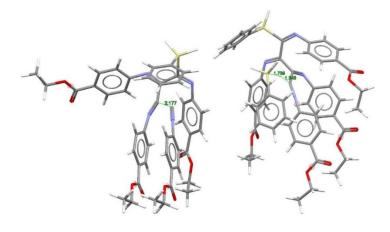


Figure S87. Free energy profile for the polymerization initiated by A⁺.



TS1'

TS2'



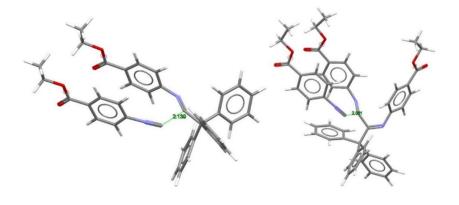


TSQ'

Figure S88. Optimized structures of stationary points in the polymerization initiated by A⁺.

Polymerization and termination

Once the adduct of $Ph_3C-(a)_1^+$ is formed, it is able to accept the addition of the second substrate **a** via a C-C bond formation transition state (see **TS1** in Figure S89 for structures and Figure S86 for energies). The barrier for this step is quite feasible, with a value of 15.4 kcal/mol accumulated from the initial state of Ph_3C^+ + **a** (Figure S86). The subsequent additions of the third and fourth substrates (via **TS2** and **TS3** in Figure S89, respectively) are both energetically reachable with a little exothermicity (see Figure S86 for energies). It is expected that the following polymerization steps should have the same nature with reasonable barrier and exothermicity, making the overall polymerization reaction energetically feasible. With an adequate number (estimated as n in Figure S86) of substrates polymerized, the accumulated exothermicity should be sufficient for the final chain termination through a hydride transfer transition state from an A to the cyano carbon of the late substrate in the chain (see **TS**_Q in Figure S89 for structures and Figure S86 for energies), which has a barrier of 10.3 kcal/mol. In summary, Ph_3C^+ is an effective species to initiate the polymerization, which is able to be terminated by A.



TS1

TS2

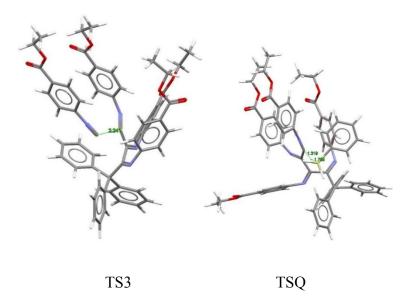


Figure 89. Optimized structures of stationary points in the polymerization initiated by Ph₃C⁺.

Interestingly, the termination step in the Ph_3C^+ -initiated polymerization is slightly exothermic by 1.8 kcal/mol (see red curve in Figure S86), making the A⁺ formation quite feasible, unlike the activation by Ph_3C^+ described above (see the blue curve in Figure S87). The A⁺ obtained from the termination of the Ph_3C^+ -initiated polymerization is able to initiate the addition of substrate via **TS1'** with a reasonable barrier of 13.2 kcal/mol. The following addition of more substrates are also shown to have feasible barriers and slight exothermicities (see **TS2'** and **TS3'** in Figure xx4 for structures and the green curve in Figure S87 for energies), making the polymerization sustainable. Such an A⁺-initiated polymerization chain can also be terminated by a hydride transfer from an A to the cyano carbon of the late substrate in the chain (see **TS_Q'** in Figure xx4 for structures and Figure S87 for energies), which has a barrier of 10.4 kcal/mol.

With these computational results, it can be concluded that the first polymerization chain should be initiated by Ph_3C^+ , while the following chains can be initiated by Ph_3C^+ or A^+ . The A^+ species is most likely formed in the termination steps of polymerization chains through a hydride transfer from A to the cyano carbon of the late substrate in the chain, instead in the activation by Ph_3C^+ .

a

1 C1 3.9307 -0.0336 0.1056 2 C2 4.6797 0.1784 1.4417 3 C3 5 5291 1 2764 1 6016 4 C4 4 5618 -0 7639 2 4716 5 C5 6.2534 1.4292 2.7853 6 H6 5.6288 2.0078 0.8084 7 C7 5 2781 -0 6016 3 6564 8 H8 3.9230 -1.6345 2.3505 9 C9 6.1266 0.4964 3.8156 10 H10 6.9143 2.2824 2.8996 11 H11 5.1751 -1.3340 4.4505 12 H12 6.6858 0.6228 4.7371 13 C13 3.6910 1.2990 -0.6536 14 C14 4.3919 1.5935 -1.8243 15 C15 2 7899 2 2357 -0 1288 16 C16 4.1828 2.8143 -2.4705 17 H17 5.0965 0.8798 -2.2332 18 C18 2.5739 3.4460 -0.7836 19 H19 2.2603 2.0295 0.7972 20 C20 3.2695 3.7369 -1.9600 21 H21 4.7342 3.0373 -3.3783 22 H22 1.8675 4.1603 -0.3730 23 H23 3.1024 4.6790 -2.4722 24 C24 4.5640 -1.1361 -0.7826 25 C25 3.8552 -1.6008 -1.9008 26 C26 5.8345 -1.6436 -0.5031

27 C27 4 4156 -2 5741 -2 7254 28 H28 2.8827 -1.1810 -2.1450 29 C29 6.3946 -2.6116 -1.3405 30 H30 6 3902 -1 2875 0 3557 31 C31 5.6872 -3.0826 -2.4466 32 H32 3.8614 -2.9291 -3.5884 33 H33 7.3853 -2.9967 -1.1214 34 H34 6.1243 -3.8388 -3.0909 35 C35 0.1666 -1.1386 0.8732 36 C36 -0.4801 -2.0151 -0.0060 37 C37 -1.8533 -2.1767 0.1235 38 C38 -2.5550 -1.4765 1.1153 39 C39 -1.8755 -0.6438 2.0126 40 C40 -0.5034 -0.4649 1.9012 41 H41 0.0787 -2.5168 -0.7865 42 H42 -2 3889 -2 8202 -0 5624 43 H43 -2.4411 -0.1146 2.7700 44 H44 0.0372 0.2024 2.5617 45 N45 1.5066 -0.8582 0.6657 46 C46 2.5873 -0.5242 0.4396 47 C47 -4.0483 -1.5280 1.2121 48 048 -4.6882 -0.9152 2.0482 49 049 -4.5908 -2.2878 0.2542 50 C50 -6.0457 -2.3412 0.2196 51 H51 -6.3959 -2.8070 1.1452 52 H52 -6.4166 -1.3153 0.1859 53 C53 -6.4339 -3.1326 -1.0110

54 H54 -7 5239 -3 1979 -1 0740 55 H55 -6.0644 -2.6448 -1.9175 56 H56 -6.0290 -4.1475 -0.9690 57 C57 -1.5289 0.9736 -1.4904 58 C58 -2.6767 0.3569 -2.0058 59 C59 -3.9097 0.6228 -1.4212 60 C60 - 3.9969 1.4940 - 0.3263 61 C61 -2.8435 2.1217 0.1620 62 C62 -1.6063 1.8699 -0.4166 63 H63 -2.5884 -0.3323 -2.8373 64 H64 -4.8029 0.1423 -1.8001 65 H65 -2.9309 2.7921 1.0090 66 H66 -0.7022 2.3348 -0.0408 67 N67 -0.2883 0.6590 -2.0209 68 C68 0.7730 0.3564 -2.4328 69 C69 -5 2869 1 7644 0 3770 70 070 -5.3857 2.4796 1.3570 71 071 -6.3258 1.1298 -0.1942 72 C72 -7.6187 1.3074 0.4472 73 H73 -7.5745 0.8370 1.4344 74 H74 -7.7952 2.3770 0.5875 75 C75 -8.6591 0.6699 -0.4506 76 H76 -9.6501 0.7914 -0.0042 77 H77 -8.6629 1.1443 -1.4358 78 H78 -8.4718 -0.3996 -0.5810

1 Si1 0 6752 0 0534 -2 3839 2 H2 0.1528 -1.1786 -2.9890 3 H3 0.2982 1.3137 -3.0363 4 C39 2.3404 -0.0071 -1.6463 5 C40 2.8190 -1.2009 -1.0621 6 C41 3.1109 1.1717 -1.5302 7 C42 4.0302 -1.2120 -0.3771 8 H43 2.2447 -2.1198 -1.1408 9 C44 4.3298 1.1505 -0.8578 10 H45 2.7527 2.1027 -1.9610 11 C46 4.7826 -0.0370 -0.2734 12 H47 4.3880 -2.1300 0.0776 13 H48 4.9211 2.0570 -0.7777 14 H49 5.7260 -0.0473 0.2640 15 H4 -0.2797 0.0829 -1.0137 16 C5 -0 9437 0 0228 0 1592 17 C6 -1.6886 1.3299 0.1134 18 C7 -1.7460 2.1682 1.2375 19 C8 -2.4091 1.6859 -1.0404 20 C9 -2.5044 3.3401 1.2021 21 H10 -1.2185 1.8964 2.1442 22 C11 -3.1512 2.8613 -1.0786 23 H12 -2.3881 1.0339 -1.9096 24 C13 -3.2003 3.6935 0.0461 25 H14 -2.5477 3.9747 2.0814 26 H15 -3.6918 3.1291 -1.9808 27 H16 -3.7805 4.6105 0.0182

TSA

28 C17 0 1941 -0 0718 1 1253 29 C18 0.3604 -1.1789 1.9770 30 C19 1.1355 0.9764 1.1865 31 C20 1 4383 -1 2331 2 8593 32 H21 -0 3637 -1 9838 1 9691 33 C22 2.2157 0.9132 2.0577 34 H23 1.0205 1.8365 0.5360 35 C24 2 3728 -0 1964 2 8948 36 H25 1.5461 -2.0878 3.5193 37 H26 2.9403 1.7199 2.0757 38 H27 3.2180 -0.2488 3.5740 39 C28 -1.7809 -1.2088 -0.0506 40 C29 -1.1784 -2.4229 -0.4291 41 C30 -3.1686 -1.1788 0.1617 42 C31 -1.9404 -3.5738 -0.5972 43 H32 -0 1055 -2 4640 -0 5883 44 C33 -3.9297 -2.3386 0.0074 45 H34 -3.6498 -0.2571 0.4658 46 C35 -3.3222 -3.5340 -0.3778 47 H36 -1.4603 -4.4993 -0.8983 48 H37 -4.9996 -2.3028 0.1862 49 H38 -3.9188 -4.4315 -0.5079

TS1'

1 Si1 -2.3979 0.8078 1.5743 2 H2 -1.3459 0.1622 2.3728 3 H3 -2.3584 2.2775 1.5948 4 C4 -4 0895 0 1085 1 8129 5 C5 -4.2576 -1.2399 2.1849 6 C6 -5.2317 0.8968 1.5682 7 C7 -5 5356 -1 7845 2 3085 8 H8 -3 3911 -1 8661 2 3827 9 C9 -6.5087 0.3486 1.6927 10 H10 -5.1257 1.9405 1.2849 11 C11 -6.6599 -0.9909 2.0608 12 H12 -5.6556 -2.8235 2.5995 13 H13 -7.3827 0.9649 1.5061 14 H14 -7.6542 -1.4161 2.1592 15 C15 -0.9688 -0.5153 -2.2838 16 C16 -1.7085 -1.5772 -2.8223 17 C17 -1.6470 -1.8080 -4.1920 18 C18 -0.8464 -0.9974 -5.0075 19 C19 -0 0594 0 0136 -4 4376 20 C20 -0.1066 0.2594 -3.0714 21 H21 -2.3481 -2.1730 -2.1813 22 H22 -2.2452 -2.5943 -4.6347 23 H23 0.5625 0.6205 -5.0856 24 H24 0.4702 1.0591 -2.6215 25 N25 -1.1566 -0.1791 -0.9406 26 C26 -1.9471 0.4025 -0.2382 27 C27 -0.8329 -1.1391 -6.4957 28 028 -0.1297 -0.4710 -7.2322 29 029 -1.7041 -2.0646 -6.9228 30 C30 -1.7783 -2.2762 -8.3601

31 H31 -2.0242 -1.3232 -8.8362 32 H32 -0.7940 -2.5940 -8.7161 33 C33 -2.8447 -3.3288 -8.5878 34 H34 -2.9629 -3.5095 -9.6602 35 H35 -2.5713 -4.2719 -8.1067 36 H36 -3.8035 -2.9947 -8.1826 37 C37 -3.8849 0.7728 -3.7974 38 C38 -4.8131 -0.2043 -4.1853 39 C39 -4.9958 -0.4490 -5.5403 40 C40 -4.2589 0.2717 -6.4920 41 C41 -3.3485 1.2554 -6.0849 42 C42 -3.1544 1.5165 -4.7345 43 H43 -5.3563 -0.7645 -3.4333 44 H44 -5.6998 -1.2053 -5.8642 45 H45 -2.7853 1.7952 -6.8371 46 H46 -2.4392 2.2582 -4.4004 47 N47 -3.6273 0.9586 -2.4509 48 C48 -3.3782 1.0922 -1.3154 49 C49 -4.4002 0.0234 -7.9601 50 050 -3.7299 0.5917 -8.8049 51 051 -5.3331 -0.8973 -8.2389 52 C52 -5.5572 -1.1828 -9.6488 53 H53 -5.8940 -0.2624 -10.1346 54 H54 -4.6047 -1.4747 -10.0994 55 C55 -6.5903 -2.2874 -9.7216 56 H56 -6.7982 -2.5238 -10.7691 57 H57 -6.2269 -3.1938 -9.2289

58 H58 -7.5245 -1.9795 -9.2439

TS2'

25 N25 -0 9696 1 5294 -0 6385 26 C26 -2.2266 1.2864 -0.6427 27 C27 2.2422 2.0547 -5.3035 28 O28 3.0422 2.9657 -5.4288 29 029 2.0240 1.1175 -6.2400 30 C30 2.8084 1.2293 -7.4573 31 H31 2.6095 2.2049 -7.9110 32 H32 3.8694 1.1916 -7.1924 33 C33 2.3998 0.0819 -8.3587 34 H34 2.9683 0.1276 -9.2920 35 H35 2.6003 -0.8801 -7.8784 36 H36 1.3344 0.1373 -8.5998 37 C37 -4.1721 1.9485 -3.9338 38 C38 -5.5602 2.0943 -3.8147 39 C39 -6.3473 1.8932 -4.9427 40 C40 - 5 7514 1 5680 - 6 1688 41 C41 -4.3545 1.4963 -6.2782 42 C42 -3.5498 1.6933 -5.1644 43 H43 -6.0031 2.3019 -2.8474 44 H44 -7.4255 1.9543 -4.8664 45 H45 -3.9162 1.2541 -7.2395 46 H46 -2.4705 1.6073 -5.2199 47 N47 -3.3934 1.9746 -2.7795 48 C48 -3.0046 1.2546 -1.9023 49 C49 -6.5634 1.2421 -7.3817 50 050 -6.0783 0.9577 -8.4620 51 051 -7.8799 1.2744 -7.1327 52 C52 -8 7615 0 9689 -8 2492 53 H53 -8.5677 1.6868 -9.0513 54 H54 -8.5163 -0.0312 -8.6164 55 C55 -10 1785 1 0598 -7 7194 56 H56 -10.8868 0.8044 -8.5128 57 H57 -10.3205 0.3649 -6.8876 58 H58 -10.4007 2.0720 -7.3703 59 C59 -5.5950 -1.2601 -3.7978 60 C60 -6.9461 -1.1884 -3.4283 61 C61 -7.9131 -1.4945 -4.3770 62 C62 -7.5335 -1.8614 -5.6771 63 C63 -6.1783 -1.9356 -6.0241 64 C64 -5.1963 -1.6371 -5.0875 65 H65 -7.2136 -0.8761 -2.4257 66 H66 -8.9630 -1.4398 -4.1176 67 H67 -5.9089 -2.2126 -7.0365 68 H68 -4.1442 -1.6718 -5.3426 69 N69 -4.6324 -0.8840 -2.8785 70 C70 -3.8435 -0.5443 -2.0819 71 C71 -8.5387 -2.1826 -6.7369 72 072 -8.2395 -2.4527 -7.8874 73 073 -9.7987 -2.1322 -6.2831 74 C74 -10.8440 -2.4735 -7.2372 75 H75 -10.7382 -1.8302 -8.1146 76 H76 -10.6907 -3.5088 -7.5556 77 C77 -12.1707 -2.2749 -6.5347 78 H78 -12.9864 -2.5360 -7.2150 79 H79 -12.2415 -2.9114 -5.6484 80 H80 -12.2975 -1.2325 -6.2284

TS3'

1 Si1 -6.2580 0.6503 -1.8228 2 H2 -7.2254 1.7631 -1.7523 3 H3 -6.3540 -0.0442 -3.1187 4 C4 -6.5257 -0.5169 -0.3852 5 C5 -6.9098 -0.0097 0.8707 6 C6 -6.3245 -1.9048 -0.5102 7 C7 -7.0929 -0.8608 1.9651 8 H8 -7.0765 1.0575 0.9989 9 C9 -6.5123 -2.7583 0.5792 10 H10 -6.0147 -2.3209 -1.4646 11 C11 -6.8979 -2.2369 1.8192 12 H12 -7 3967 -0 4519 2 9243 13 H13 -6.3626 -3.8276 0.4613 14 H14 -7.0488 -2.9008 2.6654 15 C15 -3.3868 3.3704 -0.7219 16 C16 -2.8505 3.0122 0.5288 17 C17 -1.7142 3.6525 1.0190 18 C18 -1.1076 4.6666 0.2712 19 C19 -1.7067 5.0870 -0.9308 20 C20 -2.8436 4.4617 -1.4214 21 H21 -3.3470 2.2500 1.1147 22 H22 -1.2822 3.3421 1.9636 23 H23 -1.2379 5.8886 -1.4914

24 H24 -3 2891 4 7706 -2 3605 25 N25 -4.5167 2.7062 -1.2128 26 C26 -4.5493 1.4760 -1.5746 27 C27 0 2161 5 2438 0 6328 28 O28 0.8667 5.9465 -0.1241 29 029 0.6261 4.8696 1.8595 30 C30 1.9228 5.3525 2.3115 31 H31 2.1605 6.2752 1.7801 32 H32 1.7802 5.5710 3.3719 33 C33 2.9897 4.2913 2.1008 34 H34 3.9497 4.6573 2.4783 35 H35 2.7351 3.3811 2.6511 36 H36 3.1142 4.0487 1.0416 37 C37 -2.7044 -1.6709 -2.1013 38 C38 -2.6229 -2.6358 -1.0846 39 C39 -1 5360 -3 5030 -1 0469 40 C40 -0.5717 -3.4698 -2.0633 41 C41 -0.7517 -2.6144 -3.1587 42 C42 -1.8021 -1.7026 -3.1759 43 H43 -3.3897 -2.6664 -0.3201 44 H44 -1.4233 -4.1999 -0.2253 45 H45 -0.0248 -2.6323 -3.9639 46 H46 -1.9025 -0.9861 -3.9833 47 N47 -3.7025 -0.6735 -1.9993 48 C48 - 3.3839 0.5348 - 1.7004 49 C49 0.6838 -4.2680 -1.9895 50 050 1.4237 -4.4802 -2.9324

51 051 0.9389 -4.6726 -0.7284 52 C52 2.1768 -5.4032 -0.5199 53 H53 2.1186 -6.3514 -1.0627 54 H54 3 0000 -4 8216 -0 9457 55 C55 2.3291 -5.6067 0.9744 56 H56 3.2512 -6.1580 1.1793 57 H57 2.3795 -4.6466 1.4960 58 H58 1.4880 -6.1782 1.3775 59 C59 0.1769 1.6016 -2.2739 60 C60 1.0440 0.5259 -2.5202 61 C61 2.4104 0.6860 -2.3188 62 C62 2.9239 1.9128 -1.8772 63 C63 2.0610 3.0070 -1.7326 64 C64 0.6975 2.8627 -1.9539 65 H65 0.6406 -0.4382 -2.7997 66 H66 3 0736 -0 1609 -2 4465 67 H67 2.4566 3.9651 -1.4142 68 H68 0.0292 3.7007 -1.8347 69 N69 -1 2227 1 4244 -2 3757 70 C70 -1.9319 0.8866 -1.4711 71 C71 4.3403 2.0686 -1.4484 72 072 4,7829 3,0708 -0.9100 73 073 5.0678 0.9631 -1.6762 74 C74 6.4441 0.9807 -1.2136 75 H75 6.9795 1.7792 -1.7359 76 H76 6.4487 1.2041 -0.1435 77 C77 7.0194 -0.3900 -1.5113

78 H78 8 0473 -0 4479 -1 1415 79 H79 6.4283 -1.1689 -1.0215 80 H80 7.0277 -0.5866 -2.5870 81 C81 1.0319 -0.0998 0.6144 82 C82 1.5480 -1.3927 0.4228 83 C83 2.8822 -1.6283 0.7254 84 C84 3.6882 -0.6014 1.2350 85 C85 3.1431 0.6726 1.4517 86 C86 1.8184 0.9386 1.1420 87 H87 0.9156 -2.1794 0.0318 88 H88 3.3036 -2.6126 0.5637 89 H89 3.7843 1.4547 1.8406 90 H90 1.3937 1.9268 1.2698 91 N91 -0.2443 0.1654 0.2143 92 C92 -1.3566 0.3599 -0.1715 93 C93 5 1231 -0 8156 1 5750 94 094 5.8469 0.0475 2.0437 95 095 5.5332 -2.0645 1.3003 96 C96 6.9090 -2.3863 1.6408 97 H97 7.5677 -1.6577 1.1607 98 H98 7.0277 -2.2877 2.7242 99 C99 7.1644 -3.7988 1.1572 100 H100 8.1859 -4.0959 1.4115 101 H101 6.4714 -4.5022 1.6277 102 H102 7.0453 -3.8644 0.0718 103 Si103 -3.1758 -0.6000 2.0462 104 H104 -3.6572 0.6420 2.6635

105 H105 -4.1861 -1.5119 1.5110 106 C107 -1.6640 -1.3229 2.7683 107 C108 -0.7481 -0.4966 3.4570 108 C109 -1.3106 -2.6651 2.5040 109 C110 0.4867 -0.9979 3.8624 110 H111 -0.9955 0.5422 3.6602 111 C112 -0.0766 -3.1617 2.9117 112 H113 -2.0031 -3.3184 1.9818 113 C114 0.8254 -2.3247 3.5798 114 H115 1.1908 -0.3542 4.3790 115 H116 0.1911 -4.1913 2.6989 116 H117 1.7963 -2.7080 3.8780 117 H106 -2.3823 0.0430 0.6384

TSQ'

1 Si1 -6.2580 0.6503 -1.8228 2 H2 -7.2254 1.7631 -1.7523 3 H3 -6.3540 -0.0442 -3.1187 4 C4 -6.5257 -0.5169 -0.3852 5 C5 -6.9098 -0.0097 0.8707 6 C6 -6.3245 -1.9048 -0.5102 7 C7 -7.0929 -0.8608 1.9651 8 H8 -7.0765 1.0575 0.9989 9 C9 -6.5123 -2.7583 0.5792 10 H10 -6.0147 -2.3209 -1.4646 11 C11 -6.8979 -2.2369 1.8192 12 H12 -7.3967 -0.4519 2.9243

13 H13 -6.3626 -3.8276 0.4613 14 H14 -7.0488 -2.9008 2.6654 15 C15 -3.3868 3.3704 -0.7219 16 C16 -2 8505 3 0122 0 5288 17 C17 -1 7142 3 6525 1 0190 18 C18 -1.1076 4.6666 0.2712 19 C19 -1.7067 5.0870 -0.9308 20 C20 -2 8436 4 4617 -1 4214 21 H21 -3.3470 2.2500 1.1147 22 H22 -1.2822 3.3421 1.9636 23 H23 -1.2379 5.8886 -1.4914 24 H24 -3.2891 4.7706 -2.3605 25 N25 -4.5167 2.7062 -1.2128 26 C26 -4.5493 1.4760 -1.5746 27 C27 0.2161 5.2438 0.6328 28 028 0 8667 5 9465 -0 1241 29 029 0.6261 4.8696 1.8595 30 C30 1.9228 5.3525 2.3115 31 H31 2.1605 6.2752 1.7801 32 H32 1.7802 5.5710 3.3719 33 C33 2.9897 4.2913 2.1008 34 H34 3.9497 4.6573 2.4783 35 H35 2.7351 3.3811 2.6511 36 H36 3.1142 4.0487 1.0416 37 C37 -2.7044 -1.6709 -2.1013 38 C38 -2.6229 -2.6358 -1.0846 39 C39 -1.5360 -3.5030 -1.0469 40 C40 -0 5717 -3 4698 -2 0633 41 C41 -0.7517 -2.6144 -3.1587 42 C42 -1.8021 -1.7026 -3.1759 43 H43 -3.3897 -2.6664 -0.3201 44 H44 -1.4233 -4.1999 -0.2253 45 H45 -0.0248 -2.6323 -3.9639 46 H46 -1.9025 -0.9861 -3.9833 47 N47 -3.7025 -0.6735 -1.9993 48 C48 -3.3839 0.5348 -1.7004 49 C49 0.6838 -4.2680 -1.9895 50 050 1.4237 -4.4802 -2.9324 51 051 0.9389 -4.6726 -0.7284 52 C52 2.1768 -5.4032 -0.5199 53 H53 2.1186 -6.3514 -1.0627 54 H54 3.0000 -4.8216 -0.9457 55 C55 2 3291 -5 6067 0 9744 56 H56 3.2512 -6.1580 1.1793 57 H57 2.3795 -4.6466 1.4960 58 H58 1.4880 -6.1782 1.3775 59 C59 0.1769 1.6016 -2.2739 60 C60 1.0440 0.5259 -2.5202 61 C61 2.4104 0.6860 -2.3188 62 C62 2.9239 1.9128 -1.8772 63 C63 2.0610 3.0070 -1.7326 64 C64 0.6975 2.8627 -1.9539 65 H65 0.6406 -0.4382 -2.7997 66 H66 3.0736 -0.1609 -2.4465

67 H67 2 4566 3 9651 -1 4142 68 H68 0.0292 3.7007 -1.8347 69 N69 -1.2227 1.4244 -2.3757 70 C70 -1.9319 0.8866 -1.4711 71 C71 4.3403 2.0686 -1.4484 72 072 4.7829 3.0708 -0.9100 73 073 5.0678 0.9631 -1.6762 74 C74 6.4441 0.9807 -1.2136 75 H75 6.9795 1.7792 -1.7359 76 H76 6.4487 1.2041 -0.1435 77 C77 7.0194 -0.3900 -1.5113 78 H78 8.0473 -0.4479 -1.1415 79 H79 6.4283 -1.1689 -1.0215 80 H80 7.0277 -0.5866 -2.5870 81 C81 1.0319 -0.0998 0.6144 82 C82 1 5480 -1 3927 0 4228 83 C83 2.8822 -1.6283 0.7254 84 C84 3.6882 -0.6014 1.2350 85 C85 3.1431 0.6726 1.4517 86 C86 1.8184 0.9386 1.1420 87 H87 0.9156 -2.1794 0.0318 88 H88 3.3036 -2.6126 0.5637 89 H89 3.7843 1.4547 1.8406 90 H90 1.3937 1.9268 1.2698 91 N91 -0.2443 0.1654 0.2143 92 C92 -1.3566 0.3599 -0.1715 93 C93 5.1231 -0.8156 1.5750

94 094 5 8469 0 0475 2 0437 95 095 5.5332 -2.0645 1.3003 96 C96 6.9090 -2.3863 1.6408 97 H97 7.5677 -1.6577 1.1607 98 H98 7 0277 -2 2877 2 7242 99 C99 7.1644 -3.7988 1.1572 100 H100 8.1859 -4.0959 1.4115 101 H101 6.4714 -4.5022 1.6277 102 H102 7.0453 -3.8644 0.0718 103 Si103 -3.1758 -0.6000 2.0462 104 H104 -3.6572 0.6420 2.6635 105 H105 -4.1861 -1.5119 1.5110 106 C107 -1.6640 -1.3229 2.7683 107 C108 -0.7481 -0.4966 3.4570 108 C109 -1.3106 -2.6651 2.5040 109 C110 0 4867 -0 9979 3 8624 110 H111 -0.9955 0.5422 3.6602 111 C112 -0.0766 -3.1617 2.9117 112 H113 -2.0031 -3.3184 1.9818 113 C114 0.8254 -2.3247 3.5798 114 H115 1.1908 -0.3542 4.3790 115 H116 0.1911 -4.1913 2.6989 116 H117 1.7963 -2.7080 3.8780 117 H106 -2.3823 0.0430 0.6384

TS1

1 C1 3.7941 -0.0178 0.0729

2 C2 4.6931 1.1121 -0.5044 3 C3 5.8245 0.7910 -1.2626 4 C4 4.4597 2.4478 -0.1548 5 C5 6 6980 1 7972 -1 6801 6 H6 6.0299 -0.2384 -1.5304 7 C7 5.3287 3.4506 -0.5810 8 H8 3.6075 2.7149 0.4614 9 C9 6.4512 3.1290 -1.3475 10 H10 7.5707 1.5328 -2.2688 11 H11 5.1300 4.4821 -0.3078 12 H12 7.1286 3.9098 -1.6781 13 C13 3.7717 -1.3055 -0.7927 14 C14 3.8021 -2.5813 -0.2181 15 C15 3.5854 -1.1840 -2.1793 16 C16 3.6921 -3.7167 -1.0234 17 H17 3.9109 -2.6936 0.8530 18 C18 3.4707 -2.3188 -2.9792 19 H19 3.5376 -0.2001 -2.6346 20 C20 3.5287 -3.5903 -2.4029 21 H21 3.7284 -4.7000 -0.5654 22 H22 3.3350 -2.2084 -4.0503 23 H23 3.4416 -4.4751 -3.0254 24 C24 4.2132 -0.2131 1.5386 25 C25 3.4603 0.2957 2.6014 26 C26 5.4489 -0.8178 1.8053 27 C27 3.9224 0.1794 3.9142

28 H28 2.5100 0.7888 2.4211

29 C29 5 9065 -0 9374 3 1155 30 H30 6.0540 -1.1915 0.9861 31 C31 5.1432 -0.4413 4.1759 32 H32 3 3242 0 5765 4 7281 33 H33 6 8628 -1 4139 3 3067 34 H34 5.5015 -0.5335 5.1963 35 C35 0.2339 1.4107 -0.7715 36 C36 -0.4815 2.2598 0.0809 37 C37 -1.8599 2.3421 -0.0724 38 C38 -2.5056 1.5955 -1.0690 39 C39 -1.7597 0.8091 -1.9559 40 C40 -0.3813 0.7134 -1.8195 41 H41 0.0318 2.7972 0.8696 42 H42 -2.4431 2.9569 0.6014 43 H43 -2.2781 0.2453 -2.7220 44 H44 0 2085 0 0750 -2 4668 45 N45 1.5794 1.1670 -0.5157 46 C46 2.3715 0.4244 -0.0385 47 C47 -3.9955 1.5505 -1.1839 48 048 -4.5882 0.8543 -1.9897 49 049 -4.5984 2.3311 -0.2780 50 C50 -6.0531 2.3271 -0.2873 51 H51 -6.3927 2.7465 -1.2388 52 H52 -6.3899 1.2909 -0.2275 53 C53 -6.5081 3.1476 0.9010 54 H54 -7.6010 3.1862 0.9228 55 H55 -6.1587 2.7001 1.8359

56 H56 -6 1286 4 1712 0 8388 57 C57 -1.3331 -1.4553 0.6325 58 C58 -2.1944 -0.7407 1.4738 59 C59 -3 5665 -0 8298 1 2626 60 C60 -4.0677 -1.6193 0.2195 61 C61 -3.1903 -2.3418 -0.6009 62 C62 -1.8193 -2.2689 -0.4001 63 H63 -1 7868 -0 1142 2 2578 64 H64 -4.2479 -0.2728 1.8932 65 H65 -3.5996 -2.9475 -1.4008 66 H66 -1.1258 -2.8046 -1.0376 67 N67 0.0364 -1.2962 0.7744 68 C68 1.1935 -1.1256 0.8478 69 C69 -5.5313 -1.7221 -0.0692 70 070 -5.9967 -2.4026 -0.9639 71 071 -6 2734 -0 9877 0 7773 72 C72 -7.7159 -1.0689 0.6035 73 H73 -7.9686 -0.6602 -0.3797 74 H74 -8.0072 -2.1226 0.6157 75 C75 -8.3473 -0.2845 1.7346 76 H76 -9.4363 -0.3197 1.6404 77 H77 -8.0693 -0.7094 2.7029 78 H78 -8.0339 0.7626 1.7114

TS2

1 C1 3.8815 -1.3763 0.0199 2 C2 5.3567 -1.2667 -0.4491 3 C3 5 8964 -2 1072 -1 4297 4 C4 6.2202 -0.3773 0.2136 5 C5 7.2505 -2.0322 -1.7703 6 H6 5 2745 -2 8369 -1 9323 7 C7 7 5663 -0 2958 -0 1329 8 H8 5.8400 0.2449 1.0138 9 C9 8.0887 -1.1202 -1.1331 10 H10 7 6433 -2 6960 -2 5345 11 H11 8.2106 0.4053 0.3889 12 H12 9.1394 -1.0602 -1.4001 13 C13 3.0249 -2.3179 -0.8657 14 C14 2.2447 -3.3590 -0.3454 15 C15 2.9293 -2.0388 -2.2421 16 C16 1.4038 -4.1051 -1.1758 17 H17 2.2803 -3.5877 0.7121 18 C18 2 0936 -2 7825 -3 0716 19 H19 3.5105 -1.2241 -2.6635 20 C20 1.3249 -3.8226 -2.5394 21 H21 0.8133 -4.9108 -0.7503 22 H22 2.0402 -2.5480 -4.1299 23 H23 0.6722 -4.4044 -3.1825 24 C24 3.9434 -1.8255 1.5008 25 C25 3.5142 -1.0269 2.5643 26 C26 4.5280 -3.0690 1.7891 27 C27 3.6413 -1.4676 3.8854 28 H28 3.0896 -0.0468 2.3832 29 C29 4.6484 -3.5128 3.1032

30 H30 4 8940 - 3 6873 0 9756 31 C31 4.2025 -2.7125 4.1599 32 H32 3.3011 -0.8294 4.6950 33 H33 5 0985 -4 4804 3 3027 34 H34 4 3008 - 3 0555 5 1850 35 C35 3.0498 2.3559 -0.0133 36 C36 1.9679 2.6131 0.8502 37 C37 1 3503 3 8582 0 8337 38 C38 1.7952 4.8521 -0.0490 39 C39 2.8889 4.5982 -0.8898 40 C40 3.5320 3.3686 -0.8583 41 H41 1.6332 1.8497 1.5451 42 H42 0.5230 4.0626 1.5026 43 H43 3.2286 5.3826 -1.5565 44 H44 4.3856 3.1646 -1.4949 45 N45 3 7188 1 1264 -0 0166 46 C46 3.1613 -0.0122 -0.0944 47 C47 1.1498 6.1961 -0.1223 48 048 1.5223 7.0884 -0.8639 49 049 0.1077 6.3094 0.7179 50 C50 -0.5849 7.5864 0.7158 51 H51 0.1327 8.3719 0.9707 52 H52 -0.9518 7.7817 -0.2963 53 C53 -1.7094 7.4859 1.7256 54 H54 -2.2598 8.4304 1.7578 55 H55 -2.4067 6.6886 1.4531 56 H56 -1.3166 7.2804 2.7254

57 C57 -0 4624 0 3335 -1 4576 58 C58 -1.2315 1.4146 -1.0056 59 C59 -2.6137 1.3319 -1.1155 60 C60 -3.2108 0.1968 -1.6849 61 C61 -2.4153 -0.8325 -2.2049 62 C62 -1.0325 -0.7723 -2.1025 63 H63 -0.7491 2.2650 -0.5379 64 H64 -3.2354 2.1303 -0.7302 65 H65 -2.8972 -1.6951 -2.6491 66 H66 -0.4005 -1.5805 -2.4515 67 N67 0.8982 0.3125 -1.1769 68 C68 1.7023 -0.0765 -0.3847 69 C69 -4.6927 0.0030 -1.6997 70 070 -5.2372 -0.9923 -2.1455 71 071 -5.3452 1.0252 -1.1308 72 C72 -6.7940 0.9100 -1.0636 73 H73 -7.1891 0.9471 -2.0831 74 H74 -7.0381 -0.0633 -0.6346 75 C75 -7.2931 2.0553 -0.2086 76 H76 -8.3847 2.0196 -0.1490 77 H77 -6.8885 1.9850 0.8052 78 H78 -7.0037 3.0194 -0.6358 79 C79 -1.7683 -1.7120 1.0935 80 C80 -2.7404 -0.8317 1.5838 81 C81 -4.0822 -1.1688 1.4406 82 C82 -4.4427 -2.3664 0.8095 83 C83 -3.4535 -3.2437 0.3426

84 C84 -2 1103 -2 9268 0 4826 85 H85 -2.4414 0.1022 2.0440 86 H86 -4.8500 -0.4946 1.7987 87 H87 -3.7563 -4.1637 -0.1431 88 H88 -1.3306 -3.5759 0.1014 89 N89 -0.4350 -1.3445 1.1432 90 C90 0.6993 -1.0507 1.1298 91 C91 -5.8715 -2.7525 0.5936 92 092 -6.2165 -3.7767 0.0349 93 093 -6.7267 -1.8455 1.0941 94 C94 -8.1430 -2.1504 0.9581 95 H95 -8.3893 -2.1710 -0.1080 96 H96 -8.3235 -3.1484 1.3664 97 C97 -8.9043 -1.0762 1.7064 98 H98 -9.9782 -1.2678 1.6288 99 H99 -8 6286 -1 0719 2 7644 100 H100 -8.7005 -0.0859 1.2902

TS3

1 C1 3.7991 -2.4783 -0.0211 2 C2 5.1854 -1.8911 -0.3821 3 C3 5.7748 -2.1038 -1.6335 4 C4 5.9215 -1.2092 0.5975 5 C5 7.0552 -1.6183 -1.9097 6 H6 5.2400 -2.6551 -2.3981 7 C7 7.1975 -0.7210 0.3231 8 H8 5.4990 -1.0652 1.5863 9 C9 7 7682 -0 9178 -0 9365 10 H10 7.4933 -1.7934 -2.8876 11 H11 7.7454 -0.1884 1.0944 12 H12 8 7608 -0 5355 -1 1538 13 C13 2 9021 -2 6872 -1 2849 14 C14 2.1423 -3.8485 -1.4804 15 C15 2.7706 -1.6623 -2.2437 16 C16 1 2462 - 3 9586 - 2 5477 17 H17 2.2338 -4.6814 -0.7972 18 C18 1.8660 -1.7624 -3.3004 19 H19 3.4006 -0.7853 -2.1890 20 C20 1.0841 -2.9092 -3.4509 21 H21 0.6711 -4.8721 -2.6635 22 H22 1.7896 -0.9477 -4.0141 23 H23 0.3824 -2.9934 -4.2752 24 C24 3 9743 -3 7928 0 7751 25 C25 2.8943 -4.3639 1.4731 26 C26 5.1940 -4.4794 0.7784 27 C27 3.0349 -5.5705 2.1545 28 H28 1.9318 -3.8671 1.4742 29 C29 5.3370 - 5.6911 1.4607 30 H30 6.0464 -4.0735 0.2483 31 C31 4.2613 -6.2403 2.1552 32 H32 2.1841 -5.9871 2.6854 33 H33 6.2957 -6.2006 1.4456 34 H34 4.3729 -7.1793 2.6889 35 C35 2.0833 -0.5732 2.7627

36 C36 0 7557 -0 8670 3 1185 37 C37 -0.0613 0.1344 3.6300 38 C38 0.4521 1.4207 3.8528 39 C39 1 8091 1 6751 3 6087 40 C40 2.6221 0.6911 3.0573 41 H41 0.3662 -1.8619 2.9376 42 H42 -1.1017 -0.0747 3.8498 43 H43 2.2041 2.6633 3.8175 44 H44 3.6580 0.9017 2.8152 45 N45 2.8298 -1.5705 2.0897 46 C46 3.0693 -1.4197 0.8469 47 C47 -0.4033 2.5454 4.3197 48 048 0.0239 3.5943 4.7674 49 049 -1.7163 2.2849 4.1498 50 C50 -2.6380 3.3168 4.5948 51 H51 -2 5388 3 4212 5 6796 52 H52 -2.3483 4.2664 4.1367 53 C53 -4.0313 2.8847 4.1878 54 H54 -4.7624 3.6112 4.5535 55 H55 -4.1200 2.8318 3.0993 56 H56 -4.2758 1.9056 4.6091 57 C57 3.2144 2.0317 -0.6864 58 C58 2.3949 2.9622 -0.0218 59 C59 2.2384 4.2374 -0.5513 60 C60 2.8744 4.5881 -1.7503 61 C61 3.7060 3.6614 -2.3964 62 C62 3.9044 2.3995 -1.8552

63 H63 1.9220 2.6974 0.9167 64 H64 1.6192 4.9629 -0.0381 65 H65 4.2050 3.9554 -3.3126 66 H66 4.5704 1.6841 -2.3252 67 N67 3.4652 0.7535 -0.1829 68 C68 2.6252 -0.1295 0.1840 69 C69 2.7147 5.9371 -2.3702 70 070 3.2519 6.2728 -3.4115 71 071 1.9099 6.7376 -1.6530 72 C72 1.6932 8.0738 -2.1807 73 H73 2.6626 8.5723 -2.2741 74 H74 1.2618 7.9864 -3.1824 75 C75 0.7691 8.7895 -1.2173 76 H76 0.5791 9.8050 -1.5765 77 H77 -0.1881 8.2668 -1.1357 78 H78 1.2187 8.8530 -0.2225 79 C79 -1.0651 1.0366 -0.4524 80 C80 -1.9025 1.3635 0.6235 81 C81 -3.2730 1.3643 0.4024 82 C82 -3.7885 1.0605 -0.8662 83 C83 -2.9260 0.8097 -1.9405 84 C84 -1.5519 0.7989 -1.7450 85 H85 -1.4935 1.5586 1.6079 86 H86 -3.9507 1.5613 1.2228 87 H87 -3.3478 0.5814 -2.9118 88 H88 -0.8652 0.5536 -2.5465 89 N89 0.2806 0.8135 -0.2125

90 C90 1.1726 0.0862 0.0410
91 C91 -5.2590 0.9258 -1.1072
92 092 -5.7400 0.6353 -2.1882
93 093 -5.9715 1.1214 0.0088
94 C94 -7.4148 0.9793 -0.1089
95 H95 -7.7877 1.7837 -0.7497
96 H96 -7.6226 0.0253 -0.5959
97 C97 -7.9829 1.0462 1.2927
98 H98 -9.0733 0.9700 1.2509
99 H99 -7.6016 0.2231 1.9040
100 H100 -7.7225 1.9919 1.7760
101 C101 -2.2497 -2.3175 -0.5059
102 C102 -3.2651 -2.1047 0.4343
103 C103 -4.5910 -2.2171 0.0290
104 C104 -4.8949 -2.5288 -1.3029
105 C105 -3.8638 -2.7527 -2.2263
106 C106 -2.5357 -2.6542 -1.8360
107 H107 -3.0101 -1.8372 1.4528
108 H108 -5.3901 -2.0430 0.7386
109 H109 -4.1216 -2.9930 -3.2510
110 H110 -1.7227 -2.8074 -2.5363
111 N111 -0.9303 -2.1273 -0.1317
112 C112 0.1897 -1.9247 0.1499
113 C113 -6.3027 -2.6285 -1.7965
114 0114 -6.5975 -2.8693 -2.9522
115 0115 -7.2039 -2.4351 -0.8180
116 C116 -8.6030 -2.5574 -1.1982

117 H117 -8.8398 -1.7539 -1.9025 118 H118 -8.7401 -3.5108 -1.7154 119 C119 -9.4206 -2.4712 0.0738 120 H120 -10.4835 -2.5598 -0.1676 121 H121 -9.1528 -3.2786 0.7608 122 H122 -9.2620 -1.5161 0.5817

TSQ

1 C1 4.5515 -1.3926 0.1447 2 C2 5.1279 -0.7964 1.4639 3 C3 5.8389 0.4096 1.5106 4 C4 4.8295 -1.4373 2.6784 5 C5 6.2641 0.9415 2.7291 6 H6 6.0467 0.9513 0.5975 7 C7 5.2526 -0.9070 3.8981 8 H8 4 2623 -2 3622 2 6686 9 C9 5.9740 0.2876 3.9284 10 H10 6.8176 1.8759 2.7372 11 H11 5.0128 -1.4274 4.8204 12 H12 6.3019 0.7067 4.8749 13 C13 5.0884 -0.7305 -1.1403 14 C14 6.4764 -0.6528 -1.3341 15 C15 4.2608 -0.3390 -2.1983 16 C16 7.0120 -0.1609 -2.5237 17 H17 7.1434 -0.9912 -0.5480 18 C18 4.7891 0.1466 -3.3963 19 H19 3.1853 -0.4024 -2.0988 20 C20 6.1704 0.2458 -3.5624 21 H21 8.0901 -0.1043 -2.6412 22 H22 4.1162 0.4506 -4.1930 23 H23 6 5865 0 6280 -4 4894 24 C24 4.9088 -2.8895 -0.0387 25 C25 4.2669 -3.6156 -1.0541 26 C26 5.9382 -3.5164 0.6689 27 C27 4.6108 -4.9350 -1.3290 28 H28 3.4989 -3.1287 -1.6433 29 C29 6.2923 -4.8424 0.3927 30 H30 6.4758 -2.9783 1.4403 31 C31 5.6276 - 5.5604 - 0.5989 32 H32 4.0973 -5.4758 -2.1194 33 H33 7.0948 -5.3070 0.9581 34 H34 5.9018 -6.5893 -0.8111 35 C35 1 1735 -2 2061 1 5096 36 C36 0.1125 -3.0499 1.1278 37 C37 -1.1079 -2.9673 1.7853 38 C38 -1.2734 -2.0786 2.8608 39 C39 -0.1749 -1.3359 3.3188 40 C40 1.0416 -1.3980 2.6584 41 H41 0.2476 -3.7266 0.2938 42 H42 -1.9407 -3.5768 1.4549 43 H43 -0.2986 -0.6932 4.1835 44 H44 1.8923 -0.8189 3.0024 45 N45 2.3828 -2.2554 0.8205 46 C46 3.0256 -1.2238 0.4105

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74 H74 1.1731 8.9744 -0.4553
75 C75 2.7630 9.6165 0.8898
76 H76 2.5848 10.6855 0.7405
77 H77 2.2502 9.3048 1.8043
78 H78 3.8374 9.4600 1.0219
79 C79 -0.8629 1.8084 0.9146
80 C80 -1.7638 1.4274 1.9222
81 C81 -3.1193 1.6875 1.7567
82 C82 -3.5790 2.3664 0.6188
83 C83 -2.6513 2.8648 -0.3076
84 C84 -1.2960 2.5981 -0.1656
85 H85 -1.4219 0.8641 2.7793
86 H86 -3.8213 1.3266 2.4990
87 H87 -3.0148 3.4165 -1.1672
88 H88 -0.5760 2.9367 -0.9025
89 N89 0.4721 1.3780 0.9642
90 C90 0.9403 0.4061 0.2881
91 C91 -5.0251 2.5087 0.2984
92 092 -5.4497 3.0874 -0.6885
93 093 -5.8192 1.8884 1.1904
94 C94 -7.2454 1.9691 0.9282
95 H95 -7.5405 3.0226 0.9354
96 H96 -7.4368 1.5677 -0.0700
97 C97 -7.9502 1.1687 2.0034
98 H98 -9.0332 1.2624 1.8791
99 H99 -7.6850 0.1109 1.9357
100 H100 -7.6854 1.5332 2.9999

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- 120 H120 -10.6887 -0.9588 -1.0814
- 121 H121 -9.3718 -2.0741 -0.6668
- 122 H122 -9.4691 -0.4825 0.1148
- 123 Si123 0.6667 -1.9468 -3.0707
- 124 H124 -0.1770 -1.0597 -3.8792
- 125 H125 2.0890 -1.9897 -3.4246
- 126 C127 -0.0649 -3.5105 -2.4770
- 127 C128 -1.4467 -3.6112 -2.2056
- 128 C129 0.7767 -4.6009 -2.1641
- 129 C130 -1.9671 -4.7653 -1.6247
- 130 H131 -2.1155 -2.7886 -2.4388
- 131 C132 0.2489 -5.7576 -1.5942
- 132 H133 1.8421 -4.5455 -2.3593
- 133 C134 -1.1199 -5.8353 -1.3151
- 134 H135 -3.0292 -4.8303 -1.4113
- 135 H136 0.9035 -6.5900 -1.3574
- 136 H137 -1.5270 -6.7308 -0.8561
- 137 H126 -0.6344 -1.0179 -1.582

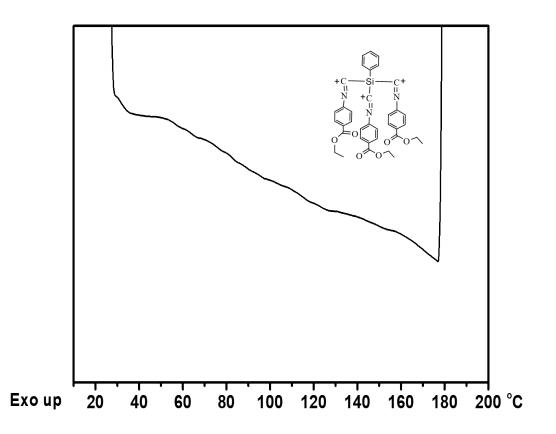


Figure S90. DSC curve of PhSi-end-functionalized star poly(EPI) in Table 3, entry 1.

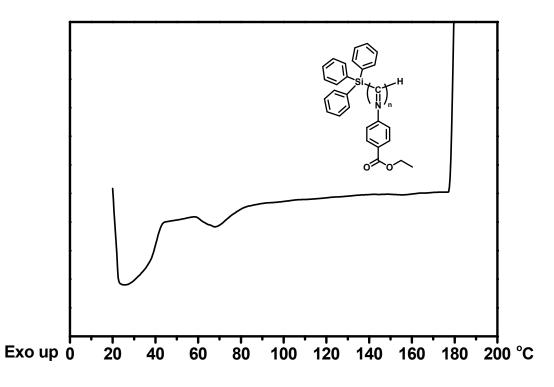


Figure S91. DSC curve of Ph₃Si-end-functionalized poly(EPI) in Table 2, entry 3.

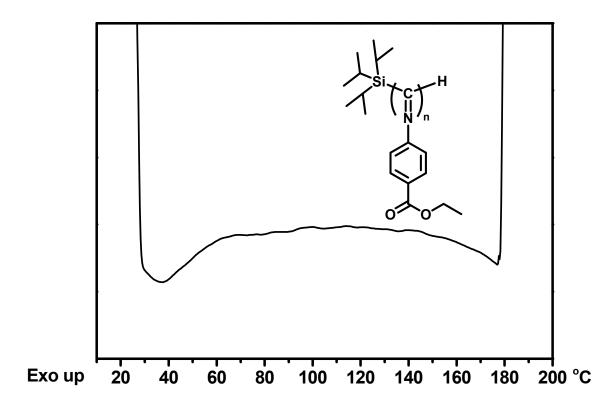


Figure S92. DSC curve of ^{*i*}Pr₃Si-end-functionalized poly(EPI) in Table 2, entry 5.

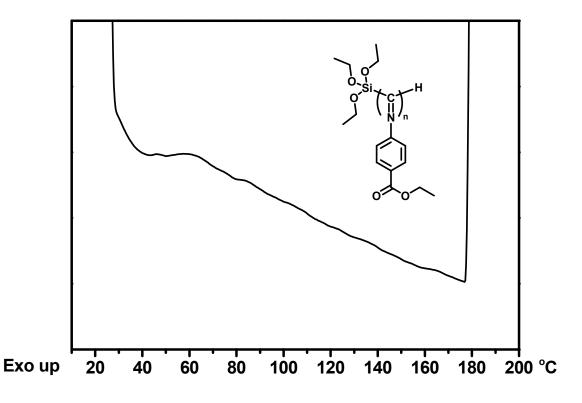


Figure S93. DSC curve of (OEt)₃Si-end-functionalized poly(EPI) in Table 2, entry 6.

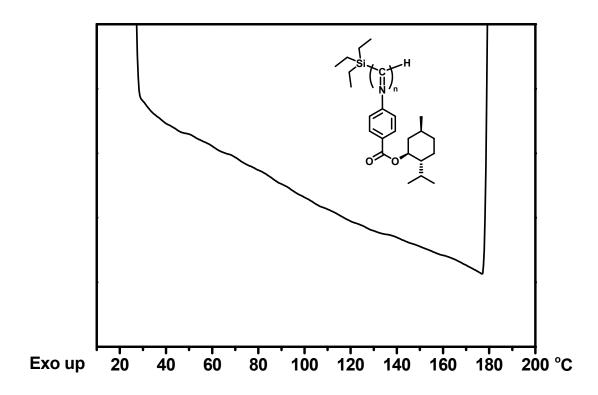


Figure S94. DSC curve of triethylsilane-end-capped Poly(D-IMCI) Table 2, entry 9.

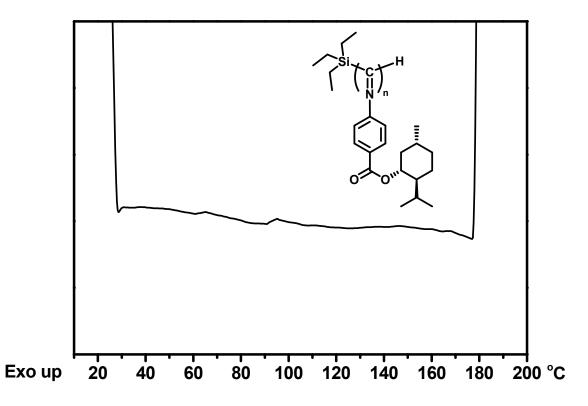


Figure S95. DSC curve of triethylsilane-end-capped Poly(L-IMCI) in Table 2, entry 10.

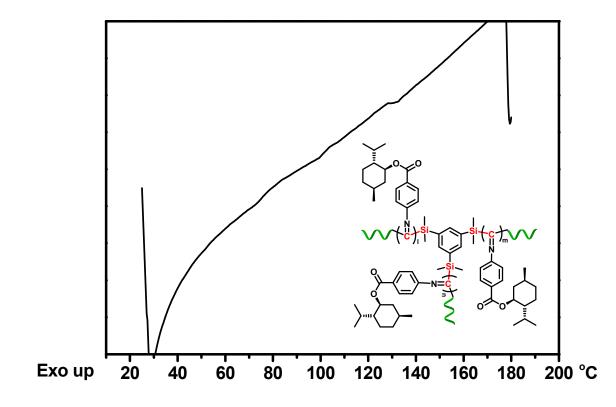


Figure S96. DSC curve of silane-end-capped star Poly(D-IMCI) in Table 3, entry 2.

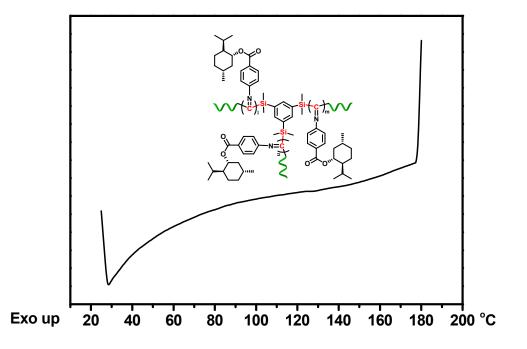


Figure S97. DSC curve of silane-end-capped star Poly(L-IMCI) in Table 3, entry 3.

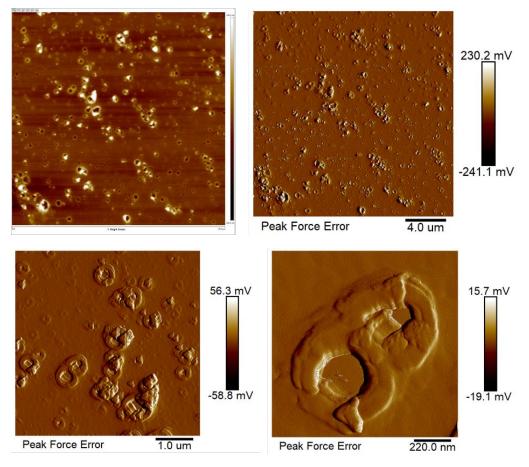


Figure S98. AFM image of silane-end-capped star Poly(L-IMCI) in Table 3, entry 3.

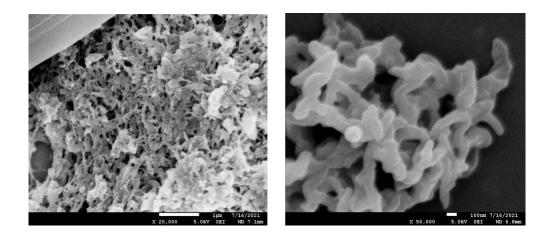


Figure S99. SEM image of silane-end-capped star Poly(L-IMCI) in Table 3, entry 3.

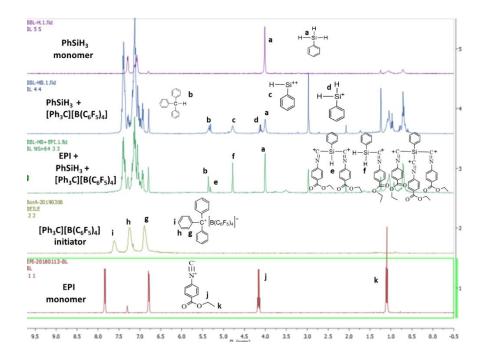


Figure S100. The *in situ* ¹H NMR spectra of the polymerization of EPI by the $[Ph_3C][B(C_6F_5)_4]/PhSiH_3$ binary system under the molar ratio of $[EPI]/{[Ph_3C][B(C_6F_5)_4]}/[PhSiH_3]$ as 10:1:10 at room temperature in d^8 -THF.

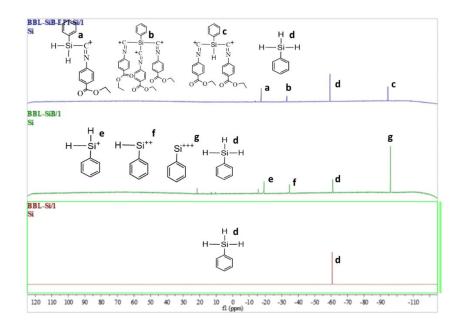


Figure S101. The *in situ* ²⁹Si NMR spectra of the polymerization of EPI by the $[Ph_3C][B(C_6F_5)_4]/PhSiH_3$ binary system under the molar ratio of $[EPI]/{[Ph_3C][B(C_6F_5)_4]}/[PhSiH_3]$ as 10:1:10 at room temperature in d^8 -THF.

 $[Ph_{3}C]^{+}[B(C_{6}F_{5})_{4}]^{-} + PhSiH_{3} \longrightarrow Ph_{3}CH + [PhSiH_{2}]^{+}[B(C_{6}F_{5})_{4}]^{-}$ $2[Ph_{3}C]^{+}[B(C_{6}F_{5})_{4}]^{-} + PhSiH_{3} \longrightarrow 2Ph_{3}CH + [PhSiH]^{2+}[B(C_{6}F_{5})_{4}]^{-}_{2}$ $3[Ph_{3}C]^{+}[B(C_{6}F_{5})_{4}]^{-} + PhSiH_{3} \longrightarrow 3Ph_{3}CH + [PhSi]^{3+}[B(C_{6}F_{5})_{4}]^{-}_{3}$

Figure S102. The reaction of $[Ph_3C][B(C_6F_5)_4]$ and $PhSiH_3$ under the molar ratio of $[EPI]/\{[Ph_3C][B(C_6F_5)_4]\}/[PhSiH_3]$ as 10:1:10 at room temperature in 1h.

References

(1) X. W. Yan, S. W. Zhang, D. Q. Peng, J.G. Zhi, P. F. Zhang, X. L. Wu, L. Wang, Y. P. Dong, and X. F. Li*, *Polym. Chem.***2018**, *9*, 984–993.

(2) (a) M. A-Cortes, B. Heinrich, B. Donnio, K. E. Evans, C. W. Smith and D. W. Bruce*, *J. Mater. Chem.*2011, *21*, 8427–8435. (b) A. Naka, Y. Matsumoto, T. Itano, K. Hasegawa, T. Shimamura, J. Osshita, A. Kunai, T. Takae, and M. Ishikawa*, *J. Organomet. Chem.*2009, *694*, 345–352.

(3) (a), M. Amela-Cort, B. Heinrich, B. Donnio, K. E. Evans, C. W. Smitha, D. W. Bruce, J. Mater. Chem., 2011, 21, 8427–8435. (b), H. Pleino, *J. Organomet. Chem.*1992, 435, 21.

(4) Z.J. Zhao, J.W.Y. Lam, B.Z. Tang, J. Mater. Chem. 2012, 22, 23723–23740.

(5) A. D. Becke, J. Chem. Phys., 1993, 98, 1372-1377.

(6) A. D. Becke, J. Chem. Phys., 1993, 98, 5648-5652.

(7) C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B.*, 1988, **37**, 785-789.

(8) M. J. Frisch, G. W.Trucks, H. B.Schlegel, G. E.Scuseria, M. A.Robb, J. R.Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N.Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K.Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C.Pomelli, J. W.Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A.

Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, D. J. Fox, *Gaussian 09*, Revision D.01; Gaussian: Wallingford, CT, USA, 2013.

- (9) S. Grimme, J. Comput. Chem., 2004, 25, 1463.
- (10) S. Grimme, J. Comput. Chem., 2006, 27, 1787.
- (11) S. Grimme, J. Antony, S. Ehrlich, H. Krieg, J. Chem. Phys., 2010, 132, 154104.
- (12) L. Goerigk, S. Grimme, Phys. Chem. Chem. Phys., 2011, 13, 6670-6688.
- (13) A. Klamt, G. Schüürmann, J. Chem. Soc. Perkin. Trans., 1993, 2, 799.
- (14) R. Cammi, B. Mennucci, J. Tomasi, J. Phys. Chem. A, 1999, 103, 9100.
- (15) V. Barone, M. Cossi, J. Phys. Chem. A, 1998, 102, 1995.
- (16) J. Tomasi, B. Mennucci, R. Cammi, Chem. Rev., 2005, 105, 2999.