# **Supporting Information**

# Complex Phase Behavior in Particle-Forming AB/AB' Diblock Copolymer Blends with Variable Core Block Lengths

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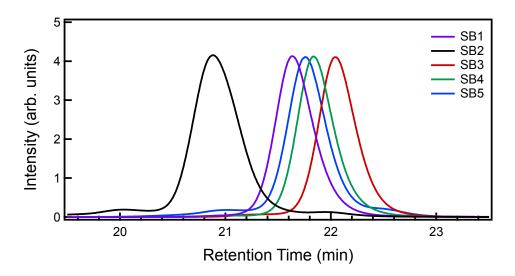
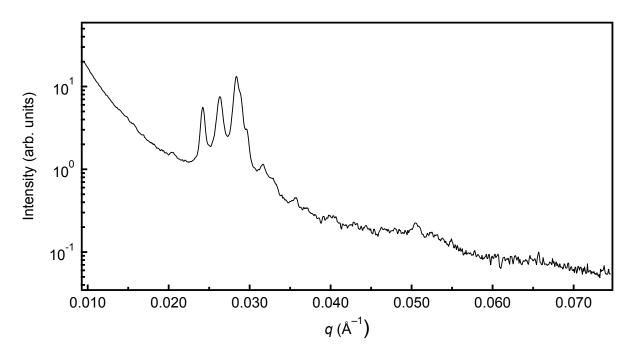


Figure S1. Size exclusion chromatography light scattering traces collected in tetrahydrofuran.



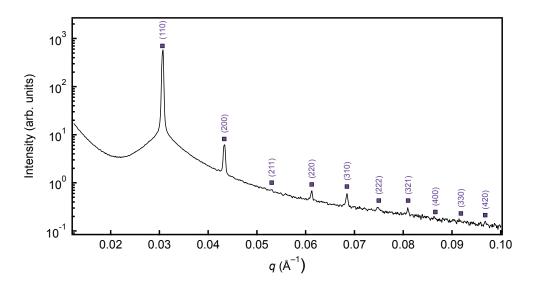
**Figure S2.** Extended 1D SAXS trace for the C14 Laves phase observed in SB1/SB2 blends with  $\phi_2 = 0.075$  at 150 °C following the thermal processing outlined in Figure 2 and the main text. Indexing and residuals can be found in Table S1.

Miller Indices ( <i>hkl</i> )	$q_{ m obs}$ (1/Å)	$q_{ m calc}$ $(1/{ m \AA})$	% Residual $(\Delta q/q_{calc} \times 100)$
(100)	0.013948	0.013952	0.03
(002)	0.014846	0.014827	-0.13
(101)	0.015744	0.015800	0.35
(102)	0.020325	0.020359	0.17
(110)	0.024188	0.024166	-0.09
(103)	0.026292	0.026254	-0.14
(200)	_	0.027905	_
(112)	0.028319	0.028352	0.12
(201)	0.028858	0.028873	0.05
(004)	0.029577	0.029653	0.26
(202)	0.031643	0.031599	-0.14
(104)	0.032900	0.032772	-0.39
(203)	0.035685	0.035683	-0.01
(210)	0.036942	0.036914	-0.07

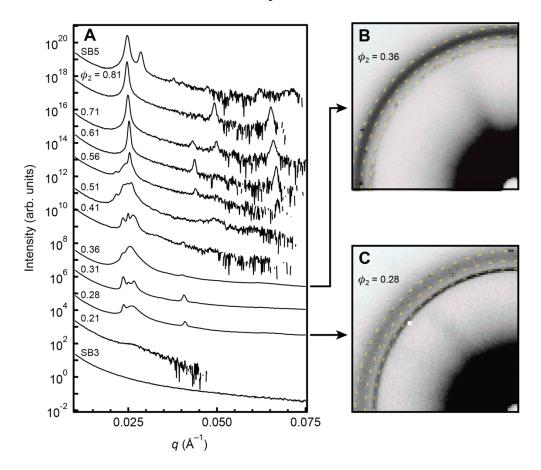
**Table S1.** Observed and calculated peak positions for the C14 Laves phase in Figures 2 and S2; data were collected from SB1/SB2 blends with  $\phi_2 = 0.075$  at 150 °C. Peak positions were calculated as  $q_{\rm hkl} = 2\pi \left[ (4/3) (h^2 + hk + k^2)/a^2 + l^2/c^2 \right]^{1/2}$  based on  $P6_3/mmc$  space group symmetry with lattice parameters a = 520.0 Å and c = 847.5 Å

Miller Indices	$q_{ m obs}$	$q_{ m calc}$	% Residual
(hkl)	(1/Å)	(1/Å)	$(\Delta q/q_{ m calc}  imes 100)$
(110)	_	0.009682	_
(200)	_	0.013692	_
(101)	_	0.014654	_
(210)	_	0.015308	_
(111)	_	0.016174	_
(220)	_	0.019363	_
(211)	0.020048	0.020055	0.04
(310)	0.021753	0.021649	-0.48
(221)	0.023369	0.023298	-0.30
(301)	0.024267	0.024283	0.07
(320)	0.024626	0.024684	0.23
(311)	0.025255	0.025230	-0.10
(002)	0.025917	0.025913	-0.02
(400)	0.027320	0.027384	0.23
(112)	_	0.027663	_
(321)	0.027858	0.027877	0.07
(410)	0.028217	0.028227	0.03
(330)	0.029025	0.029045	0.07
(202)	0.029295	0.029308	0.04
(212)	_	0.030097	_
(420)	_	0.030616	_
(411)	0.03109	0.031058	-0.10
(331)	0.031809	0.031804	-0.02
(222)	0.032347	0.032348	0.00
(421)	0.033155	0.033245	0.27
(312)	0.033694	0.033766	0.21
(430)	_	0.034230	_
(510)	_	0.034908	_
(322)	0.035849	0.035788	-0.17
(501)	0.036657	0.036600	-0.16
(520)	_	0.036867	_
(511)	0.037195	0.037235	0.11

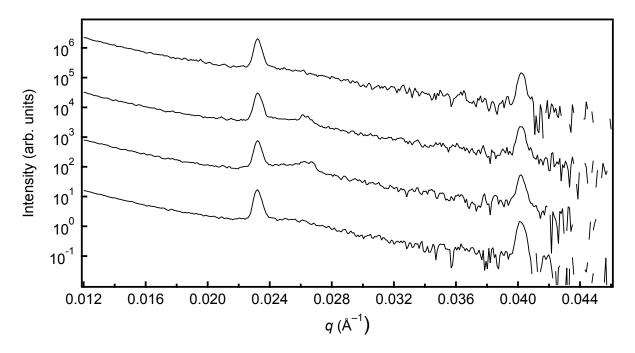
**Table S2.** Observed and calculated peak positions for the  $\sigma$  phase coexisting with HEX<sub>C</sub> in Figures 3 and 4. Data were collected from an SB3/SB4 blend with  $\phi_2 = 0.80$  at 150 °C. Peak positions were calculated as  $q_{hkl} = 2\pi [(h^2 + hk + k^2)/a^2 + l^2/c^2]^{1/2}$  based on  $P4_2/mnm$  space group symmetry with lattice parameters a = 917.8 Å and c = 484.9 Å.



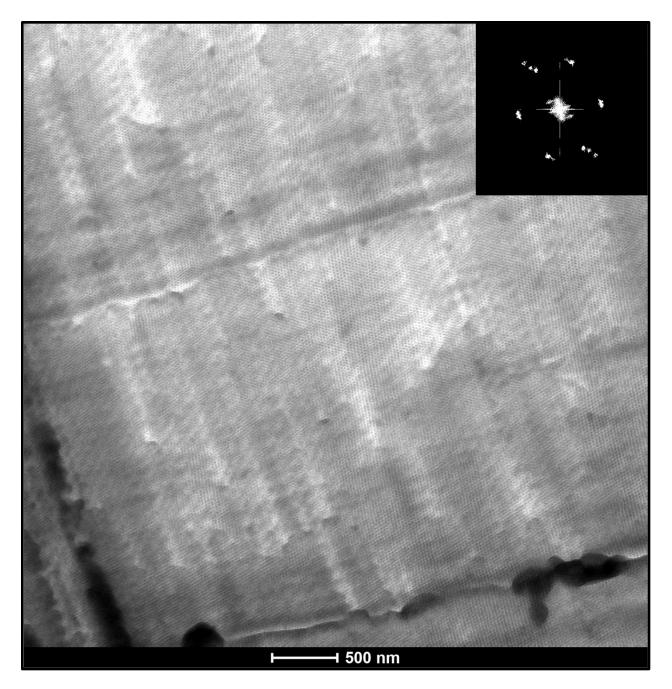
**Figure S3.** 1D SAXS trace collected from a SB3/SB4 blend with  $\phi_2 = 0.80$  following a 169 h anneal at 180 °C. The trace is indexed to a BCC phase.



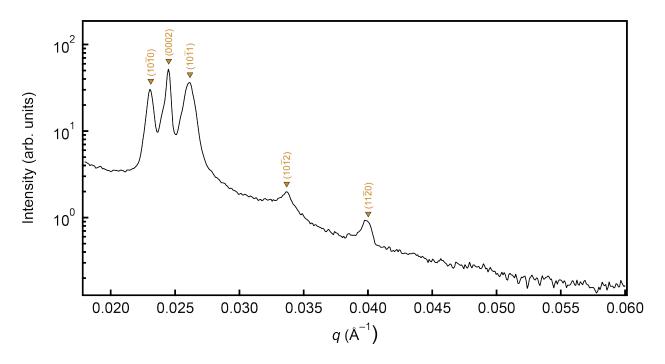
**Figure S4**. (A) 1D and (B,C) 2D SAXS data collected from SB3/SB5 blends following extended annealing (66-92 h) at 120 °C. 2D SAXS data in B and C was collected following 92 h of annealing at 120 °C. 2D data is indexed to the HCP phase (yellow dashed lines).



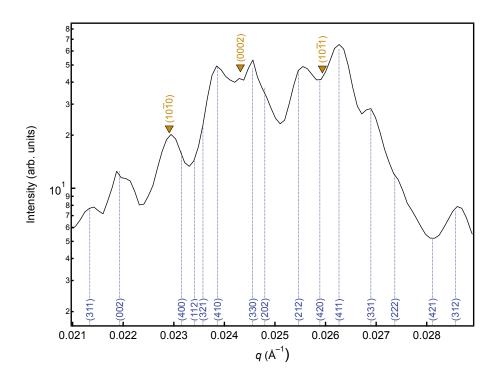
**Figure S5.** 1D SAXS traces collected from different locations on a SB3/SB5 blend with  $\phi_2 = 0.31$  following 66 h of annealing at 150 °C.



**Figure S6.** Transmission electron microscopy (TEM) micrograph collected from a thin (~70 nm) microtomed section of a SB3/SB5 blend with  $\phi_2 = 0.31$ . Prior to microtoming, the sample was annealed for 66 h at 150 °C, after which the SAXS patterns in Figures 5A,C and S5 were collected and the sample was vitrified in liquid nitrogen. The inset in the upper right corner is a Fourier transform of the image displaying the 6-fold rotation symmetry present over large areas.



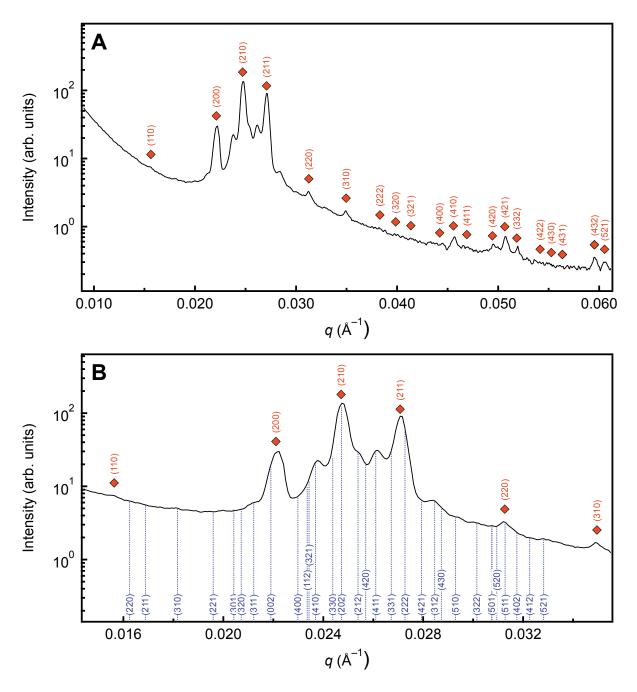
**Figure S7.** 1D SAXS trace collected from a SB3/SB5 blend with  $\phi_2 = 0.41$  following 92 h of annealing a 180 °C. The trace is indexed to a HCP phase. Additional patterns can be found in Figures 5 and S9. Indexing and residuals can be found in Table S4.



**Figure S8.** 1D SAXS trace collected from a SB3/SB5 blend with  $\phi_2 = 0.46$  after annealing at 180 °C for 66 h. Lines and inverted triangles denote peaks associated with the  $\sigma$  and HCP phases, respectively. Indexing and residuals can be found in Table S3.

**Table S3.** Observed and calculated peak positions for the  $\sigma/\text{HCP}$  phase coexistence evidenced in Figures 5 and S10; data were collected from SB3/SB5 blends with  $\phi_2 = 0.46$  at 180 °C. Peak positions for the HCP phase were calculated as  $q_{hkl} = 2\pi [(4/3) (h^2 + hk + k^2)/a^2 + l^2/c^2]^{1/2}$  based on  $P6_3/mmc$  space group symmetry with lattice parameters a = 316.7 Å and c = 517.1 Å. Peak positions for the  $\sigma$  phase were calculated as  $q_{hkl} = 2\pi [(h^2 + k^2)/a^2 + l^2/c^2]^{1/2}$  based on  $P4_2/mnm$  space group symmetry with lattice parameters a = 1085.5 Å and c = 573.0 Å.

Miller Indices ( <i>hkil</i> ) or ( <i>hkl</i> )	$q_{ m obs}$ $(1/ m \AA)$	$q_{ m calc} \ (1/{ m \AA})$	% Residual $(\Delta q/q_{calc} \times 100)$
		CP	
(1010)	0.022946	0.022910	-0.16
(0002)	0.024293	0.024300	0.03
(1011)	_	0.025933	-0.26
(1012)	0.033539	0.033397	-0.43
(1120)	0.039643	0.039682	0.10
(1013)	_	0.043052	_
(2020)	0.046016	0.045820	-0.43
(1122)	0.046555	0.046531	-0.05
(2021)	0.047363	0.047404	0.09
(0004)	0.048799	0.048599	-0.41
(2022)	0.051761	0.051865	0.20
		5	
(211)	0.016932	0.016964	0.19
(310)	0.018279	0.018304	0.14
(221)	0.019625	0.019705	0.40
(301)	0.020613	0.020537	-0.37
(320)	_	0.020870	_
(311)	0.021331	0.021337	0.03
(002)	0.021959	0.021931	-0.13
(400)	_	0.023153	_
(112)	_	0.023409	_
(321)	_	0.023575	_
(410)	0.023844	0.023866	0.09
(330)	0.024562	0.024558	-0.02
(202)	0.024832	0.024799	-0.13
(212)	0.02555	0.025465	-0.33
(420)	_	0.025886	_
(411)	0.026268	0.026264	-0.01
(331)	0.026896	0.026895	-0.01



**Figure S9.** 1D SAXS trace collected from a SB3/SB5 blend with  $\phi_2 = 0.51$  after annealing at 180 °C for 66 h. Lines and diamonds denote peaks associated with  $\sigma$  and A15 phases, respectively. Indexing and residuals can be found in Table S6.

**Table S4.** Observed and calculated peak positions for the A15/ $\sigma$  phase coexistence evidenced in Figures 5 and S11; data were collected from SB3/SB5 blends with  $\phi_2 = 0.51$  at 180 °C. Peak positions for the A15 phase were calculated as  $q_{hkl} = 2\pi [(h^2 + k^2 + l^2)/a^2]^{1/2}$  based on  $Pm\bar{3}n$  space group symmetry with a lattice parameter of a = 568.5 Å. Peak positions for the  $\sigma$  phase were calculated as  $q_{hkl} = 2\pi [(h^2 + k^2)/a^2 + l^2/c^2]^{1/2}$  based on  $P4_2/mnm$  space group symmetry with lattice parameters a = 1093.3 Å and c = 573.7 Å.

Miller Indices ( <i>hkl</i> )	$q_{ m obs} \ (1/{ m \AA})$	$q_{ m calc} \ (1/{ m \AA})$	% Residual $(\Delta q/q_{calc} \times 100)$
(11.11)		15	$(\Delta q/q_{calc} \times 100)$
(110)	0.015586	0.015630	0.28
(200)	0.022228	0.022104	-0.56
(210)	0.024742	0.024713	-0.12
(211)	0.027165	0.027072	-0.34
(220)	0.031205	0.031260	0.18
(310)	0.034885	0.034950	0.19
(222)	0.038207	0.038285	0.20
(320)	0.039733	0.039849	0.29
(321)	0.041348	0.041353	0.01
(400)	0.0444	0.044208	-0.43
(410)	0.045657	0.045569	-0.19
(411)	0.047004	0.046890	-0.24
(420)	0.049517	0.049426	-0.18
(421)	0.050684	0.050647	-0.07
(332)	0.051941	0.051839	-0.20
		5	
(410)	0.023754	0.023695	-0.25
(330)	_	0.024382	_
(202)	_	0.024735	_
(212)	0.02546	0.025394	-0.26
(420)	_	0.025701	_
(411)	0.026178	0.026103	-0.29
(331)	_	0.026728	_
(222)	_	0.027275	_
(421)	_	0.027937	_
(312)	0.028422	0.028460	0.13
(430)	_	0.028734	_
(510)	0.02941	0.029303	-0.36
(322)	0.030038	0.030150	0.37

**Table S5.** Observed and calculated peak positions for the QC evidenced in Figures 5 and S12; data were collected from SB3/SB5 blends with  $\phi_2 = 0.51$  at 150 °C. Peak positions were calculated as described by Iwami and Ishimasa [1] based on a  $P12_6/mmc$  space group symmetry, a tiling edge length a = 572.9 Å, and a periodicity of c = 575.6 Å.

Miller Indices $(a_1 a_2 a_3 a_4 a_5)$	$q_{ m obs} \ (1/{ m \AA})$	$q_{ m calc} \ (1/{ m \AA})$	% Residual $(\Delta q/q_{calc} \times 100)$
(11000)	_	0.012233	_
(00002)	0.021780	0.021832	0.24
(12100)	0.023844	0.023633	-0.89
(01102)	0.024921	0.025026	0.42
(22011)	0.025999	0.026032	0.13
(11102)	0.028063	0.027856	-0.74
(12202)	_	0.036531	_
(00004)	0.043664	0.043664	0.00
(01104)	0.045298	0.045345	0.10

**Table S6.** Observed and calculated peak positions for the A15 phase coexisting with HEX<sub>C</sub> evidenced in Figures 5 and S13; data was collected from SB3/SB5 blends with  $\phi_2 = 0.56$  at 180 °C. Peak positions for the A15 phase were calculated as  $q_{\rm hkl} = 2\pi \left[ (h^2 + k^2 + l^2) / a^2 \right]^{1/2}$  based on  $Pm\bar{3}n$  space group symmetry with a lattice parameter of a = 577.0 Å.

Miller Indices ( <i>hkl</i> )	$q_{ m obs}$ (1/Å)	$q_{ m calc} \ (1/{ m \AA})$	% Residual $(\Delta q/q_{calc} \times 100)$
(110)	_	0.015400	_
(200)	0.021753	0.021779	0.12
(210)	0.024357	0.024349	-0.03
(211)	0.026601	0.026673	0.27
(220)	_	0.030800	_

### Calculation of mean particle radii:

For a periodic particle packing, the mean particle radius  $\langle R \rangle$  can be calculated as:

$$\langle R \rangle = \left(\frac{3V_{\rm UC}}{4\pi\rho_{\rm P,UC}}\right)^{1/3} \tag{S1}$$

where  $V_{UC}$  is the unit cell volume, which can be readily determined *via* SAXS, and  $\rho_{P,UC}$  is the number of particles per unit cell set by the packing. In the most general form, unit cell volume can be calculated as:

$$V_{\rm UC} = a \ b \ c \left(1 - \cos^2 \alpha - \cos^2 \beta - \cos^2 \gamma + 2 \cos \alpha \cos \beta \cos \gamma\right)^{1/2}$$
(S2)

where *a*, *b*, and *c* are lattice constants and the angles  $\alpha$ ,  $\beta$ , and  $\gamma$  are lattice parameters. These values are determined from SAXS from the relations:

$$q_{\rm hkl,cubic} = 2\pi \left( \frac{\left(h^2 + k^2 + l^2\right)}{a^2} \right)^{1/2}$$
 (S3)

$$q_{\rm hkl,hexagonal} = 2\pi \left( \frac{4(h^2 + hk + k^2)}{3a^2} + \frac{l^2}{c^2} \right)^{1/2}$$
(S4)

$$q_{\rm hkl,tetragonal} = 2\pi \left( \frac{(h^2 + k^2)}{a^2} + \frac{l^2}{c^2} \right)^{1/2}$$
 (S5)

for cubic, hexagonal, or tetragonal lattices, where  $q_{hkl}$  is the scattering wavevector for plane (*hkl*),  $\alpha = \beta = \gamma = 90^{\circ}$  for a cubic or tetragonal lattice, and  $\alpha = \beta = 90^{\circ}$  and  $\gamma = 120^{\circ}$  for a hexagonal lattice.

From these relations,  $\langle R \rangle$  can be calculated for the phases observed in this work as:

$$\langle R \rangle_{\rm BCC} = R_{\rm BCC} = \frac{3^{1/3} 2^{1/2} \pi^{2/3}}{q_{110}}$$
 (S6)

$$\langle R \rangle_{\rm FCC} = R_{\rm FCC} = \frac{3^{5/6} \pi^{2/3}}{2^{1/3} q_{111}}$$
 (S7)

S14

$$\langle R \rangle_{\rm HCP} = R_{\rm HCP} = \frac{2^{2/3} \pi^{2/3} (c/a)^{1/3}}{q_{10\bar{1}0}}$$
 (S8)

$$\langle R \rangle_{\sigma} = \frac{2\pi^{2/3}}{5^{1/3} (c/a)^{2/3} q_{002}}$$
(S9)

$$\langle R \rangle_{\rm A15} = \frac{3^{1/3} \pi^{2/3}}{2^{1/6} q_{110}}$$
 (S10)

$$\langle R \rangle_{C14} = \frac{2^{1/3} \pi^{2/3} (c/a)^{1/3}}{3^{1/3} q_{10\bar{1}0}}$$
 (S11)

Owing to a lack of translational symmetry,  $\langle R \rangle$  for a dodecagonal quasicrystal (QC) or a liquidlike packing (LLP) can only be estimated. For a QC, this is best done by taking the (00002) reflection as the  $\sigma q_{002}$  peak owing to the close structural relationship between the two phases and the invariance of this reflection on transition to the  $\sigma$  phase.<sup>2,3</sup> For LLP the principal reflection can be taken as the  $q_{110}$  peak of the BCC phase by a similar argument. On calculating  $\langle R \rangle$ , the core radius can be calculated as:

$$\langle R_{\rm core} \rangle = f_{\rm core}^{-1/3} \langle R \rangle$$
 (S12)

where  $f_{core}$  is volume fraction of the core domain assuming complete segregation of both blocks. This core radius can then be used to calculate the underlying spherical form factor.

A similar strategy can be used to calculate the radius of cylinders in the hexagonally-packed cylinder (HEX<sub>C</sub>) phase. However, the functional form changes slightly owing to periodicity in only two dimensions.  $\langle R \rangle$  is instead calculated as:

$$R_{\rm cyl} = \left(\frac{A_{\rm UC}}{\pi\rho_{\rm C,UC}}\right)^{1/2}$$
(S13)

where  $A_{UC}$  is the unit cell area, which can be readily determined *via* SAXS, and  $\rho_{C,UC} = 1$  is the number of cylinders per unit cell. The unit cell area can be calculated as:

$$A_{\rm UC} = \frac{3^{1/2} a^2}{2} \tag{S14}$$

The lattice parameter *a* is determined from SAXS according to the relation:

$$q_{\rm hk,hexagonal2D} = 2\pi \left(\frac{4(h^2 + hk + k^2)}{3a^2}\right)^{1/2}$$
 (S15)

From these equations,  $\langle R \rangle$  can be calculated for the HEX<sub>C</sub> phase as:

$$R_{\rm cyl} = \frac{2^{3/2} \pi^{1/2}}{3^{1/4} q_{10}} \tag{S16}$$

On calculating  $R_{cyl}$ , the cylinder core radius can then be calculated as:

$$R_{\rm core,cyl} = f_{\rm core}^{1/2} R_{\rm cyl} \tag{S17}$$

where  $f_{core}$  is again the volume fraction of the core domain assuming complete segregation of both blocks. This core radius can then be used to calculate the underlying cylindrical form factor.

#### **Dispersity:**

Core block dispersity  $D \approx 1.12$  was estimated for SB3/SB4 blends with  $\phi_2 = 0.80$  approximating the constituent diblock copolymers as monodisperse via the relation:

$$D = \frac{M_{\rm w}}{M_{\rm n}} \tag{S18}$$

where  $M_w$  and  $M_n$  are the weight- and number-average block molecular weights calculated as:

$$M_{\rm n} = x_1 M_{\rm n,1} + x_2 M_{\rm n,2} \tag{S19}$$

$$M_{\rm w} = \frac{x_1 M_{\rm n,1}^2 + x_2 M_{\rm n,2}^2}{x_1 M_{\rm n,1} + x_2 M_{\rm n,2}}$$
(S20)

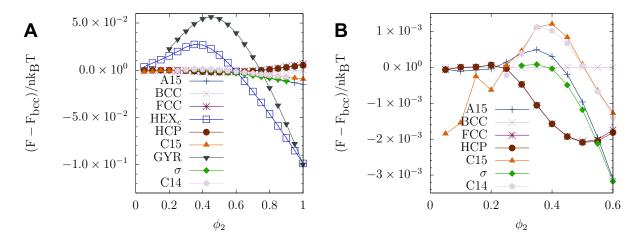
where values of  $M_n$  for each block and polymer can be found in Table S1 and  $x_i$  is the mole fraction of copolymer *i* in the core domain. A similar approach can be used to estimate the dispersity of the corona as  $D \approx 1.001$ . Notably, this approach only provides an estimate and neglects the dispersity of each polymer, each determined via size exclusion chromatography (SEC) to be  $D \approx 1.01$ .

To compare this value with previously investigated 1,4-polyisoprene-*block*-poly( $\pm$ -lactide) (IL) diblock copolymers, we assume a I precursor dispersity of 1.06, a reasonable estimate for low molecular weight I synthesized *via* anionic polymerization,<sup>4</sup> and independent block dispersities. Using the rule for the sum of variances of statistical distributions (*i.e.*,  $D_{PIPLA} = w_{PI}^2(D_{PI} - 1) + w_{PLA}^2(D_{PLA} - 1) + 1$ , where  $w_i$  is the mass fraction of block i), it can be found that many of the  $\sigma$ -forming IL diblock copolymers reported in the literature likely had core (L) block dispersities greater than the  $D_{core} = 1.2$  we found in this work to be sufficient to drive  $\sigma$ -formation.<sup>5-12</sup> It should be noted that the accuracy of such calculations is limited by that of the assumed I precursor dispersity and the resolution of the instruments used for dispersity for the second block. First, ring-opening polymerization of lactides generally results in higher dispersity (~1.1–2.0) than anionic polymerization of isoprene (< 1.1).<sup>5,14,15</sup> Second, it is anticipated that some fraction of PI precursor will be present in the final diblock copolymer due to a combination of early termination,

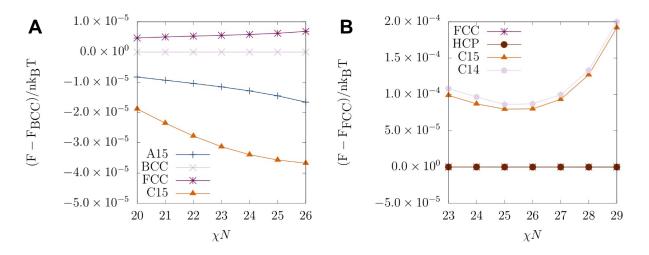
incomplete reinitiation, and challenges inherent to purification.<sup>13</sup> Third, when considering the low molecular weight of these polymers, generally characterized by an average block length of 45 isoprene and 6 lactide monomer units, it is clear that the loss or gain of only a few lactide monomer units over the course of the polymerization would have a dramatic effect on the block dispersity.

#### Self-consistent mean-field theory (SCFT):

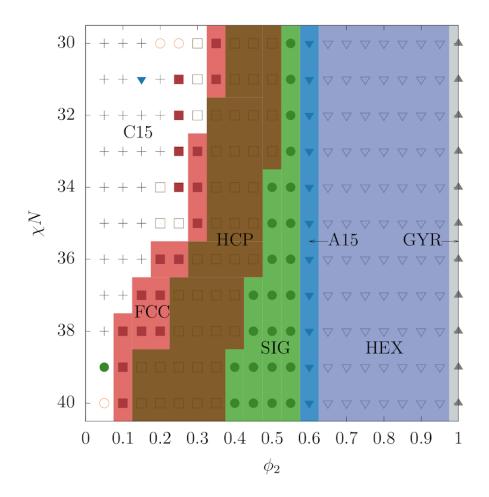
Calculations based on self-consistent mean-field theory were performed using the open-source Polymer Self-Consistent Field software package (PSCF).<sup>16</sup> We performed binary blend canonical ensemble calculations at  $N_2/N_1 = 1.4$ ,  $N_{B,1} = N_{B,2}$ ,  $f_{A,1} = 0.12$ ,  $f_{A,2} = 0.38$  and a conformational asymmetry of  $\varepsilon = b_{\rm B}/b_{\rm S}$  ( $v_{\rm S}/v_{\rm B}$ ) = 1.7. Note this differs modestly from experiments, where  $f_{\rm A,2}$  = 0.388, but is within experimental error and, thus, has a negligible impact on the applicability of the calculations to the experimental results. We scanned a range of  $\chi N$  (20–30) and  $\phi_2$  (0–1), mimicking the parameter space explored experimentally. The candidate phases tested in the canonical ensemble calculations include the body-centered cubic (BCC), face-centered cubic (FCC), hexagonally close-packed (HCP), hexagonally-packed cylinder (HEX<sub>C</sub>), double gyroid (GYR), and disordered (DIS) phases as well as the Frank-Kasper (FK) σ, A15, C14, and C15 phases. SCFT calculations were performed using a grid size of 64×64×64 for the three-dimensional structures except for the  $\sigma$  and two-dimensional HEX<sub>C</sub> phases, which were performed at a grid size of 96×96×48 and 64×64, respectively. Calculations were performed with a contour length step size of  $\Delta s = 0.01$  and a convergence criterion of  $10^{-5}$  as defined by Arora *et al.*<sup>17</sup> As shown in Figure S10, the free energies were almost degenerate at  $\phi_2 = 0.15$  and 0.25. To better resolve the phase behavior at those compositions, we repeated the calculations using a more stringent convergence criterion of 10<sup>-6</sup>. Figure S11 shows the results under this stricter convergence criterion, revealing that C15 and HCP phases offer the lowest free energy at low  $\phi_2$ . Then, we performed grand canonical ensemble calculations between neighboring phases to resolve phase coexistence. Further calculation details can be found elsewhere.<sup>18</sup>



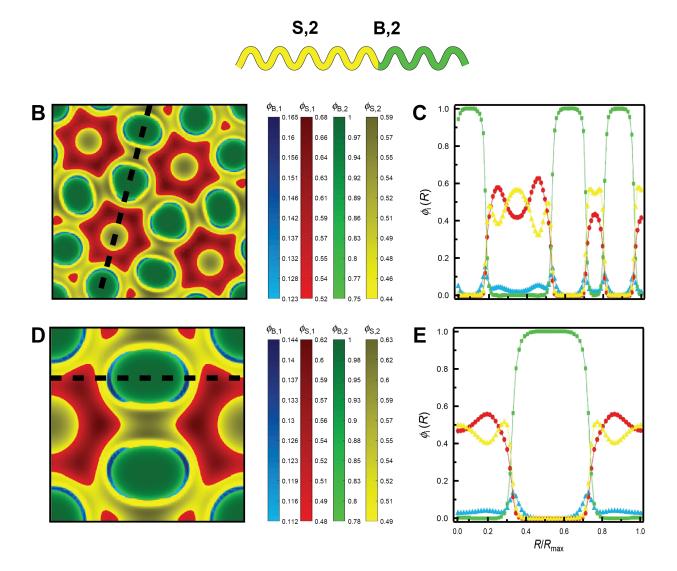
**Figure S10.** Normalized free energy relative to BCC *versus*  $\phi_2$  at  $\chi N = 28$  for (A) all studied phases and (B) only particle phases.



**Figure S11**. Normalized free energy relative to (A) BCC and (B) FCC *versus*  $\chi N$  at (A)  $\phi_2 = 0.15$  and (B)  $\phi_2 = 0.25$  for select phases under the more stringent convergence criterion of 10<sup>-6</sup>.



**Figure S12**. Binary blend phase diagram generated from canonical ensemble SCFT calculations over an extended range of  $\chi N (N_2/N_1 = 1.4, N_{S,1} = N_{S,2}, f_{B,1} = 0.12, \text{ and } f_{B,2} = 0.38)$ . Symbols correspond to the double gyroid (gray  $\blacktriangle$ ), hexagonally-packed cylinder (purple  $\bigtriangledown$ ), A15 (blue  $\checkmark$ ),  $\sigma$  (green  $\textcircled{\bullet}$ ), hexagonal close-packed  $\Box$ ), face-centered cubic (red  $\blacksquare$ ), C14 (orange  $\circ$ ), and C15 phases (black +).



Α

S,1

**B**,1

**Figure S13**. (B,D) Composition maps and (C,E) 1D composition profiles for the (B,C)  $\sigma$  and (D,E) A15 phases in the (001) planes. Data were calculated *via* SCFT for SB3/SB5 blends with (B,C)  $\phi_2 = 0.55$  and (D,E) 0.60 at  $\chi \langle N \rangle = 30$ .  $\phi_i$  is the volume fraction of block i at each position. *R*/*R*<sub>max</sub> is the nondimensional distance along the dashed lines in (B,D), where 0 corresponds to the (C) bottom or (E) left edge of the composition map. The schematic in (A) shows the relative copolymer block lengths. Blue and red shading correspond to the B and S blocks of SB3, whereas green and yellow shading correspond to the B and S blocks of SB5, respectively.

## References

- Iwami, S.; Ishimasa, T. Dodecagonal Quasicrystal in Mn-Based Quaternary Alloys Containing Cr, Ni and Si. *Philos. Mag. Lett.* 2015, 95, 229–236. https://doi.org/10.1080/09500839.2015.1038332.
- Lindsay, A. P.; Jayaraman, A.; Peterson, A. J.; Mueller, A. J.; Weigand, S.; Mahanthappa, M. K.; Lodge, T. P.; Bates, F. S. Reevaluation of Poly(Ethylene- Alt -Propylene)- Block Polydimethylsiloxane Phase Behavior Uncovers Topological Close-Packing and Epitaxial Quasicrystal Growth. *ACS Nano* 2021, *ASAP Article*. https://doi.org/10.1021/acsnano.1c02420.
- Lindsay, A. P.; Lewis, R. M.; Lee, B.; Peterson, A. J.; Lodge, T. P.; Bates, F. S. A15, σ, and a Quasicrystal: Access to Complex Particle Packings via Bidisperse Diblock
   Copolymer Blends. ACS Macro Lett. 2020, 9, 197–203.
   https://doi.org/10.1021/acsmacrolett.9b01026.
- Schmidt, S. C.; Hillmyer, M. A. Morphological Behavior of Model Poly(Ethylene-Alt-Propylene)-b-Polylactide Diblock Copolymers. *J. Polym. Sci. Part B Polym. Phys.* 2002, 40, 2364–2376. https://doi.org/10.1002/polb.10291.
- Lynd, N. A.; Hillmyer, M. A. Influence of Polydispersity on the Self-Assembly of Diblock Copolymers. *Macromolecules* 2005, *38*, 8803–8810. https://doi.org/10.1021/ma051025r.
- Cooke, D. M.; Shi, A.-C. Effects of Polydispersity on Phase Behavior of Diblock
   Copolymers. *Macromolecules* 2006, *39*, 6661–6671. https://doi.org/10.1021/ma060717s.
- Lee, S.; Bluemle, M. J.; Bates, F. S. Discovery of a Frank-Kasper σ Phase in Sphere-Forming Block Copolymer Melts. *Science* 2010, *330*, 349–353. https://doi.org/10.1126/science.1195552.
- (8) Gillard, T. M.; Lee, S.; Bates, F. S. Dodecagonal Quasicrystalline Order in a Diblock

Copolymer Melt. *Proc. Natl. Acad. Sci. U. S. A.* **2016**, *113*, 5167–5172. https://doi.org/10.1073/pnas.1601692113.

- Lee, S.; Leighton, C.; Bates, F. S. Sphericity and Symmetry Breaking in the Formation of Frank-Kasper Phases from One Component Materials. *Proc. Natl. Acad. Sci. U. S. A.*2014, *111*, 17723–17731. https://doi.org/10.1073/pnas.1408678111.
- Schulze, M. W.; Lewis, R. M.; Lettow, J. H.; Hickey, R. J.; Gillard, T. M.; Hillmyer, M. A.; Bates, F. S. Conformational Asymmetry and Quasicrystal Approximants in Linear Diblock Copolymers. *Phys. Rev. Lett.* 2017, *118*, 207801. https://doi.org/10.1103/PhysRevLett.118.207801.
- Kim, K.; Arora, A.; Lewis, R. M.; Liu, M.; Li, W.; Shi, A. C.; Dorfman, K. D.; Bates, F. S. Origins of Low-Symmetry Phases in Asymmetric Diblock Copolymer Melts. *Proc. Natl. Acad. Sci. U. S. A.* 2018, *115*, 847–854. https://doi.org/10.1073/pnas.1717850115.
- (12) Kim, K.; Schulze, M. W.; Arora, A.; Lewis, R. M.; Hillmyer, M. A.; Dorfman, K. D.; Bates, F. S. Thermal Processing of Diblock Copolymer Melts Mimics Metallurgy. *Science* 2017, *356*, 520–523. https://doi.org/10.1126/science.aam7212.
- (13) Hiemenz, P. C.; Lodge, T. P. *Polymer Chemistry*, 2nd Editio.; CRC Press: Boca Raton, 2007.
- (14) Dechy-Cabaret, O.; Martin-Vaca, B.; Bourissou, D. Controlled Ring-Opening Polymerization of Lactide and Glycolide. *Chem. Rev.* 2004, *104*, 6147–6176. https://doi.org/10.1021/CR040002S.
- Schmidt, S. C.; Hillmyer, M. A. Synthesis and Characterization of Model
   Polyisoprene–Polylactide Diblock Copolymers. *Macromolecules* 1999, *32*, 4794–4801.
   https://doi.org/10.1021/MA9900277.

- Arora, A.; Qin, J.; Morse, D. C.; Delaney, K. T.; Fredrickson, G. H.; Bates, F. S.;
   Dorfman, K. D. Broadly Accessible Self-Consistent Field Theory for Block Polymer
   Materials Discovery. *Macromolecules* 2016, 49, 4675–4690.
   https://doi.org/10.1021/acs.macromol.6b00107.
- (17) Arora, A.; Morse, D. C.; Bates, F. S.; Dorfman, K. D. Accelerating Self-Consistent Field Theory of Block Polymers in a Variable Unit Cell. J. Chem. Phys. 2017, 146, 244902. https://doi.org/10.1063/1.4986643.
- Liu, M.; Qiang, Y.; Li, W.; Qiu, F.; Shi, A. C. Stabilizing the Frank-Kasper Phases via Binary Blends of AB Diblock Copolymers. *ACS Macro Lett.* 2016, *5*, 1167–1171. https://doi.org/10.1021/acsmacrolett.6b00685.