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**Supporting Information for:****Boosting Highly Efficient Hydrocarbon Solvent Processed All-polymer based Organic Solar Cells by Modulating Thin Film Morphology**

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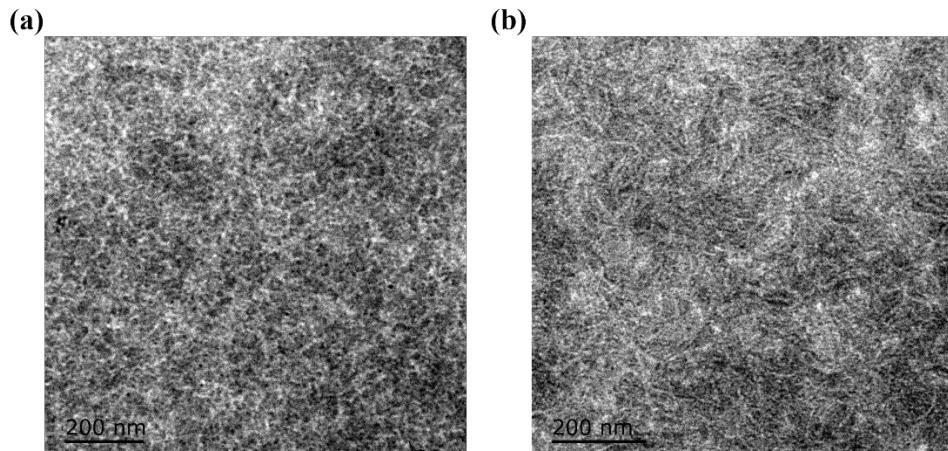
### Solar cell fabrication and characterization

Solar cells were fabricated in a conventional device configuration of ITO/PEDOT:PSS/active layers/PNDIT-F3N/Ag. The ITO substrates were first scrubbed by detergent and then sonicated with deionized water, acetone and isopropanol subsequently, and dried overnight in an oven. The glass substrates were treated by UV-Ozone for 30 min before use. PEDOT:PSS (Heraeus Clevios P VP AI 4083) was spin-cast onto the ITO substrates at 4000 rpm for 30 s, and then dried at 150 °C for 15 min in air. The PM6:PY-IT blends (1:1 weight ratio), were dissolved in chloroform/toluene (the concentration of donor was 7 mg mL<sup>-1</sup> for all blends), with 1-chloronaphthalene (1% vol) as additive, and stirred overnight in a nitrogen-filled glove box. The chloroform blend solution was spin-cast at 2000 rpm for 30 s onto PEDOT:PSS films followed by a temperature annealing of 95°C for 5 min. The 95°C toluene blend solution was spin-cast at 2000 rpm for 30 s onto 95°C preheated PEDOT:PSS films (to imitate the film formation kinetics of chloroform) followed by a temperature annealing of 95°C for 5 min. A thin PNDIT-F3N layer (0.5mg/mL in methanol, 2000 rpm) was coated on the active layer, followed by the deposition of Ag (evaporated under 3×10<sup>-4</sup> Pa through a shadow mask). The optimal active layer thickness measured by a Bruker Dektak XT stylus profilometer was about 100 nm. The current density-voltage (*J-V*) curves of all encapsulated devices (by Epoxy) were measured using a Keithley 2400 Source Meter in the air under AM 1.5G (100 mW cm<sup>-2</sup>) using a Newport solar simulator. The light intensity was calibrated using a standard Si diode (with KG5 filter, purchased from PV Measurement to bring spectral mismatch to unity). An optical microscope (Olympus BX51) was used to define the device area (5.9 mm<sup>2</sup>). EQEs were measured using an Enlitech QE-S EQE system equipped with a standard Si diode. Monochromatic light was generated from a Newport 300W lamp source.

### SCLC Measurements

The electron and hole mobilities were measured by using the method of space-charge

limited current (SCLC) for electron-only devices with the structure of ITO/ZnO/active layer/PNDIT-F3N/Ag and hole-only devices with the structure of ITO/PEDOT:PSS/active layers/MoO<sub>x</sub>/Ag. The charge carrier mobility was determined by fitting the dark current to the model of a single carrier SCLC according to the equation:  $J = 9\epsilon_0\epsilon_r\mu V^2/8d^3$ , where  $J$  is the current density,  $d$  is the film thickness of the active layer,  $\mu$  is the charge carrier mobility,  $\epsilon_r$  is the relative dielectric constant of the transport medium, and  $\epsilon_0$  is the permittivity of free space.  $V = V_{\text{app}} - V_{\text{bi}}$ , where  $V_{\text{app}}$  is the applied voltage,  $V_{\text{bi}}$  is the built-in voltage. The carrier mobility is calculated from the slope of the  $J^{1/2} \sim V$  curves.



**Figure S1.** TEM images of PM6:PY-IT films cast from (a) CF and (b) TMB.

**Table S1.** Morphology parameters of (100) peaks fitted from GIWAXS data.

PM6:PY-IT	Position ( $\text{\AA}^{-1}$ )	d-spacing ( $\text{\AA}$ )	CCL ( $\text{\AA}$ )
Chloroform	0.30	20.9	42.1
Toluene	0.30	20.9	54.3

**Table S2.** Morphology parameters of (010) peaks fitted from GIWAXS data.

PM6:PY-IT	Position ( $\text{\AA}^{-1}$ )	d-spacing ( $\text{\AA}$ )	CCL ( $\text{\AA}$ )
Chloroform	1.67	3.76	24.2
Toluene	1.67	3.76	21.1

**Table S3.** Photovoltaic performances of all-polymer based OSCs summary

Ref	Solvents	$V_{\text{oc}}$ (V)	$J_{\text{SC}}$ ( $\text{mA cm}^{-2}$ )	FF (%)	PCE (%)
1	chloroform	0.96	20.6	72.1	14.3
2	chloroform	0.891	23.03	73.98	15.17

<b>3</b>	chloroform	0.9	22.3	70	14.4
<b>4</b>	chloroform	0.87	23.96	72.67	15.11
<b>5</b>	chloroform	0.955	22.65	74.3	16.09
<b>6</b>	chloroform	0.896	24.7	71.3	15.8
<b>7</b>	chlorobenzene	0.946	20.65	74	14.45
<b>8</b>	chloroform	0.933	22.3	72.3	15.05
<b>9</b>	chloroform	0.9	22.7	75.3	15.4
<b>10</b>	chloroform	0.936	22.77	73.2	15.62
<b>11</b>	chloroform	0.93	20.71	66.33	13.44
<b>12</b>	chloroform	0.901	23.3	72.4	15.2
<b>13</b>	chloroform	0.91	22	71.53	14.32
<b>14</b>	chloroform	0.899	21.33	65.3	12.53
<b>15</b>	chloroform	0.99	16.48	66.1	10.77
<b>16</b>	chloroform	0.943	15.75	68.2	10.13
<b>17</b>	chloroform	0.947	22.6	74.9	16.04
<b>18</b>	chloroform	0.938	21.5	66.66	13.43
<b>19</b>	chlorobenzene	1	15.2	69	10.5
<b>20</b>	chloroform	0.91	23.2	75	15.8
<b>21</b>	chloroform	0.96	15.27	68	10.3
<b>22</b>	chloroform	0.97	17.96	67	12.06
<b>23</b>	chloroform	0.89	23.41	67.73	14.1
<b>24</b>	chlorobenzene	0.86	18.55	64	10.1
<b>25</b>	chlorobenzene	1.17	13.39	64	10.07
<b>26</b>	chloroform	0.86	22.28	69	13.22
<b>27</b>	2-MeTHF	0.92	22.47	66.7	13.8
<b>28</b>	2-MeTHF	0.88	17.62	75.78	11.76
<b>29</b>	Toluene	0.96	17.1	68.2	11.2
<b>30</b>	CPME	0.85	16.5	77.9	11
<b>31</b>	CPME	0.85	17.2	77.9	11.5
<b>32</b>	o-xylene	0.91	21.22	73.12	14.28
<b>This work</b>	Toluene	<b>0.933</b>	<b>22.52</b>	<b>73.8</b>	<b>15.51</b>

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