

Mechanisms of fluorine-induced separation of mass interference during TOF-SIMS analysis

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Table S1. PVD deposition parameters of Zr, Mo and Ag thin films.

Layer	Power mode	Plasma power [W]	Plasma current [mA]	Deposition time [s]	Base pressure [mbar]	Process pressure [mbar]	Argon flow [sccm]
Zr	DC	73	200	1320	4.7×10^{-7}	5.2×10^{-3}	35
Mo	DC	72	200	1760	9.0×10^{-7}	5.0×10^{-3}	20
Ag	DC	111	200	582	5.3×10^{-7}	5.0×10^{-3}	20

Table S2. PVD deposition parameters of AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr multilayer sample.

Layer		Plasma power [W]	Plasma current [mA]	Deposition time [s]	Base pressure [mbar]	Process pressure [mbar]	Argon flow [sccm]
AgMo	Ag	35	80	701	1.6×10^{-8}	5.0×10^{-3}	35
	Mo	86	250				
ZrMo	Zr	93	250	655	2.7×10^{-7}	5.0×10^{-3}	35
	Mo	74	220				
ZrAg	Zr	94	250	618	2.3×10^{-7}	5.0×10^{-3}	35
	Ag	33	80				
Zr		72	200	1322	2.4×10^{-7}	5.1×10^{-3}	35

Table S3: ALD deposition parameters of AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr multilayer sample. 42 cycles were conducted to obtain 5 nm Al₂O₃ thin films.

Precursor				Substrate temperature [°C]	Argon flow [sccm]
H ₂ O		Al(CH ₃) ₃			
Precursor temperature [°C]	Pulse time [ms]	Precursor temperature [°C]	Pulse time [ms]		
45	0.05	72	0.05	120	50

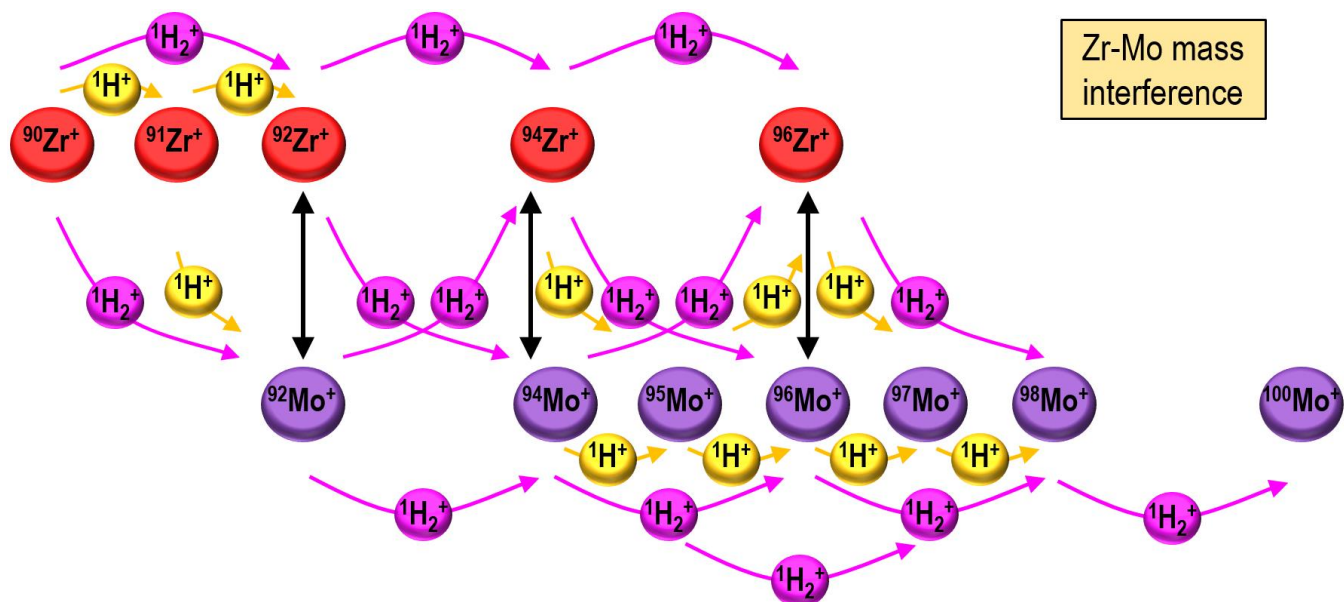


Figure S1. Graphical representation of potential mass interference variants between different metal isotopes (isobaric interference, 1st order mass interference), their hydrides (2nd order mass interference) and dihydrides (3rd order mass interference) in the case of a system containing Zr (given in red) and Mo (given in violet). ¹H⁺ ions, which contribute to ZrH and MoH ions formation, are shown in yellow. ¹H₂⁺ ions, which contribute to ZrH₂ and MoH₂ ions formation, are given in pink. The arrows indicate the direction of ions' influence on the signal measured at a given m/q . For example, the signal measured at $m/q=92$, corresponding to ⁹²Zr⁺ ions, is a sum of signals coming from ⁹²Zr⁺, ⁹²Mo⁺, ⁹¹Zr¹H⁺, ⁹⁰Zr¹H₂⁺ secondary ions.

Table S4. Natural isotope abundance of Zr, Mo and Ag. Source: TOF-SIMS Explorer version 1.4.4.0 software from TOFWERK.

m/q	90	91	92	94	95	96	97	98	100	107	109
Zr	51.5%	11.2%	17.2%	17.4%		2.8%					
Mo			14.8%	9.2%	15.9%	16.7%	9.6%	24.2%	9.7%		
Ag										51.8%	48.2%

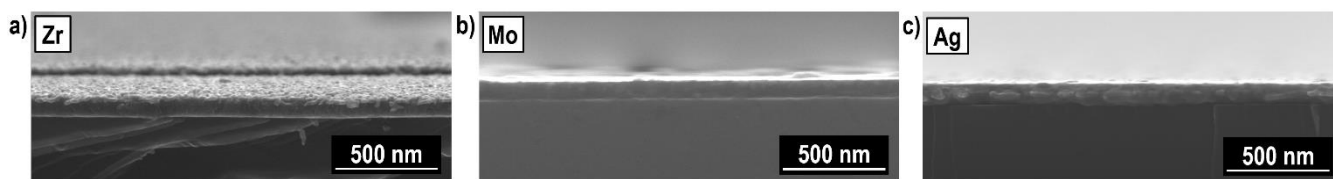


Figure S2. SEM images of mechanically obtained sample cross-sections. The Si substrate is at the bottom. The thin film thickness of $h_{Zr}=81\pm3$ nm, $h_{Mo}=88\pm4$ nm, $h_{Ag}=100\pm2$ nm were measured in the case of Zr (a), Mo (b) and Ag (c), respectively.

Table S5. Summary of the thin film thicknesses, h , measured on the samples' cross-sections using SEM (Figures S2) and calculated sputtering rates.

Sample	h [nm]	Sputtering rate [nm/frame]	
		Without gas	With fluorine gas
Zr	81 ± 3	1.8 ± 0.2	2.8 ± 0.2
Mo	88 ± 4	3.3 ± 0.6	5.9 ± 0.7
Ag	100 ± 2	4.3 ± 0.7	14 ± 3

Table S6: Summary of the results conducted on metal thin films: $t_{interface}$ denotes the time of reaching the interface between the metal thin film and the Si substrate, S_{total} is the total secondary ion signal of a given isotope, the indexes *reference* and *fluorine* refer to the measurements conducted without and with fluorine gas. The given values are the averages of three measurements.

Isotope	Ga beam current [pA]		Moment of reaching the interface [frames]		Total integrated current [counts/pixel/pA]	
	$I_{reference}$	$I_{fluorine}$	$t_{interface_reference}$	$t_{interface_fluorine}$	$S_{total_reference}$	$S_{total_fluorine}$
$^{90}\text{Zr}^+$	117.3	119	46 ± 2	29 ± 1	$(4.0 \pm 0.2) \times 10^{-4}$	$(1.50 \pm 0.02) \times 10^{-3}$
$^{98}\text{Mo}^+$	118	119.3	27 ± 3	15 ± 1	$(3.6 \pm 0.2) \times 10^{-5}$	$(1.61 \pm 0.06) \times 10^{-3}$
$^{107}\text{Ag}^+$	118	119.3	23 ± 3	7 ± 1	$(7 \pm 2) \times 10^{-6}$	$(3.3 \pm 0.3) \times 10^{-3}$

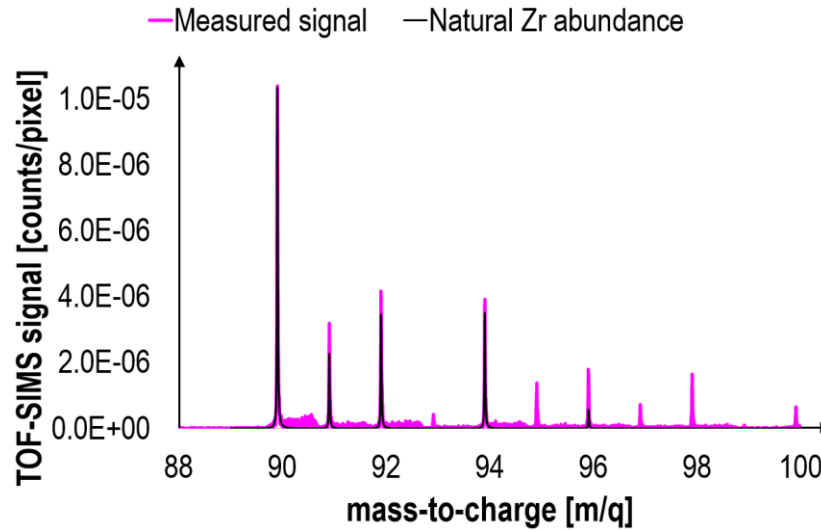


Figure S3. Recognition of mass interference. The measured Zr signal (pink line) in AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr sample is fitted to the Zr natural isotope abundance (black line). The difference between the two lines suggests the contribution of other ions (Mo in this case) to the total measured signal. Data acquired under standard vacuum conditions.

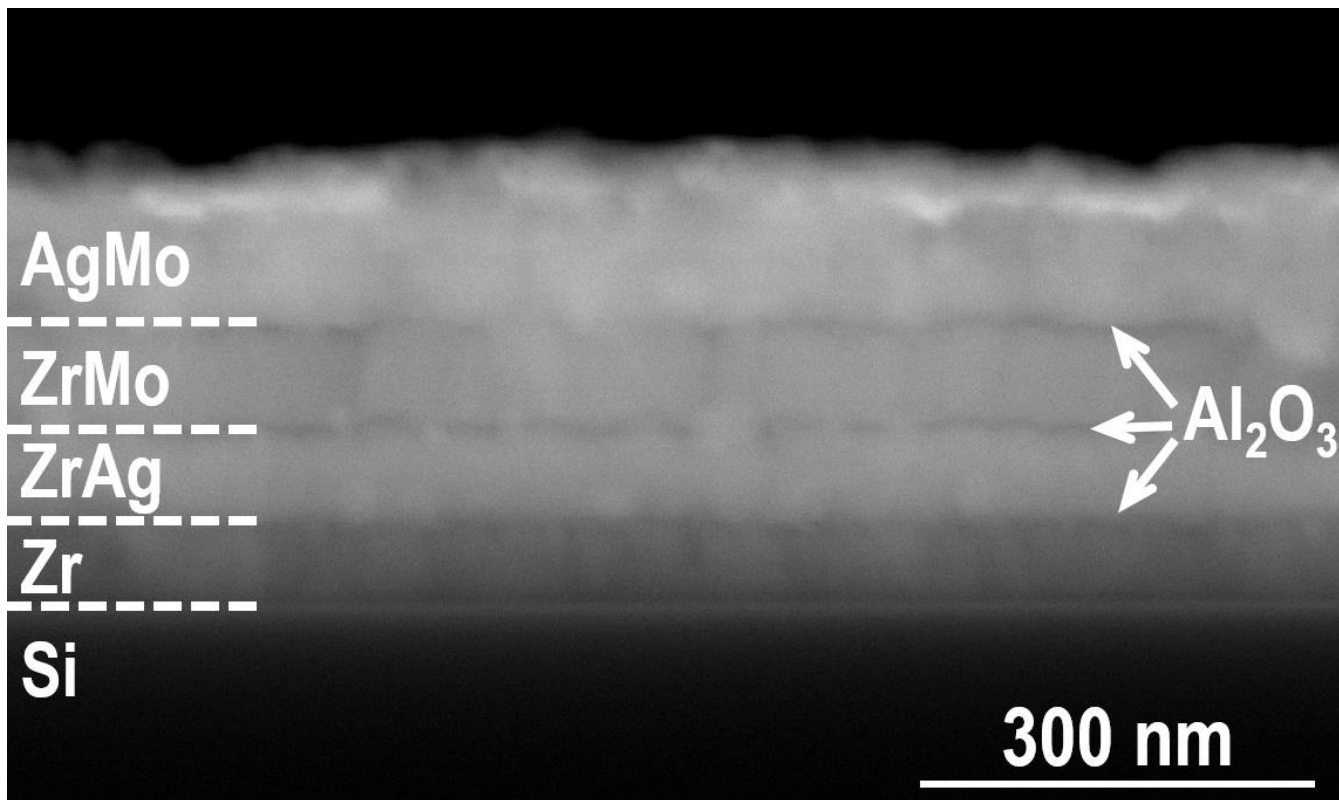


Figure S4. The SEM image of mechanically obtained AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr cross-section. The roughness of the PVD-deposited metallic layers was sufficiently low to prevent potential misinterpretation during further analysis of TOF-SIMS depth profiles. All metallic thin films were well-separated and the layer thicknesses in the order of 100 nm were achieved (i.e. $h_{\text{AgMo}}=133\pm9$ nm, $h_{\text{ZrMo}}=77\pm3$ nm, $h_{\text{ZrAg}}=69\pm4$ nm and $h_{\text{Zr}}=61\pm2$ nm). The ceramic layers were visible as well but they were too shallow to measure their size.

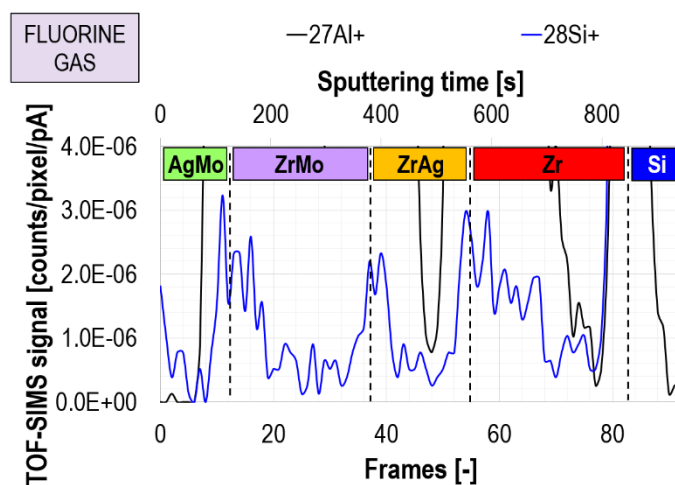


Figure S5. The TOF-SIMS signals depth profiles measured at $m/q=27$ and $m/q=28$, corresponding to $^{27}\text{Al}^+$ and $^{28}\text{Si}^+$. Data obtained during the fluorine gas-assisted measurement.

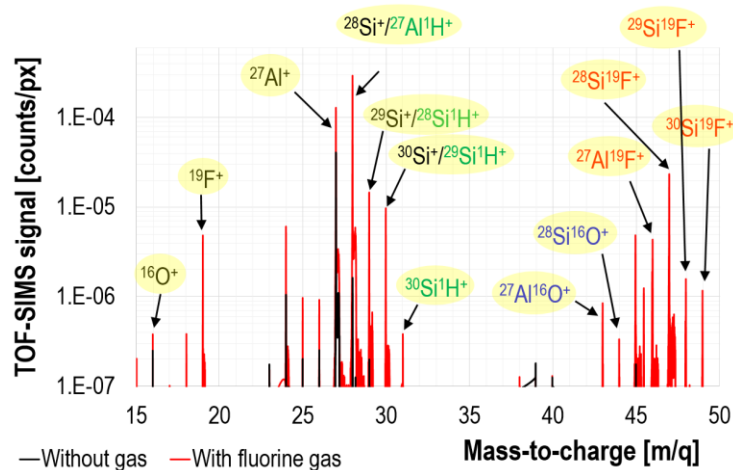


Figure S6. Part of AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr mass spectrum corresponding to Al- and Si-containing ions. Contribution of ²⁹Si¹⁶O⁺ and ³⁰Si¹⁶O⁺ ions to the signals measured at $m/q=45$ and $m/q=46$ are negligible due to the low contribution of ²⁹Si and ³⁰Si isotopes to the Si natural isotope abundance (i.e. 92.2% ²⁸Si, 4.7% ²⁹Si, 3.1% ³⁰Si) and relatively low ²⁸Si¹⁶O⁺ signal. Note logarithmic scale.

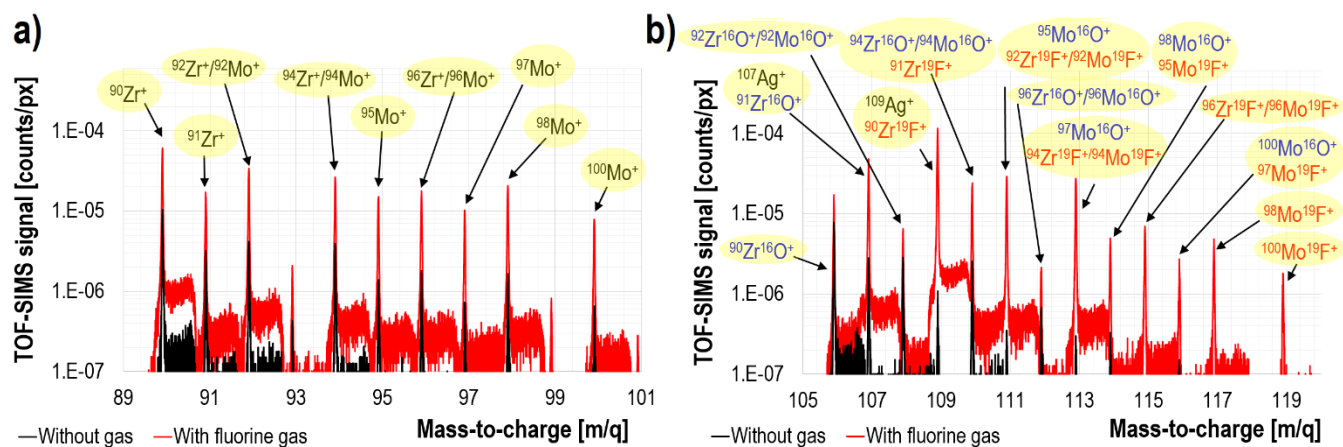


Figure S7. Comparison of mass spectra obtained for AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr under standard vacuum conditions (black lines) and in the presence of fluorine gas (red line). A) a mass spectrum range corresponding to Zr and Mo ions, b) a mass spectrum range corresponding to Ag ions as well as Zr- and Mo- oxides and fluorides.

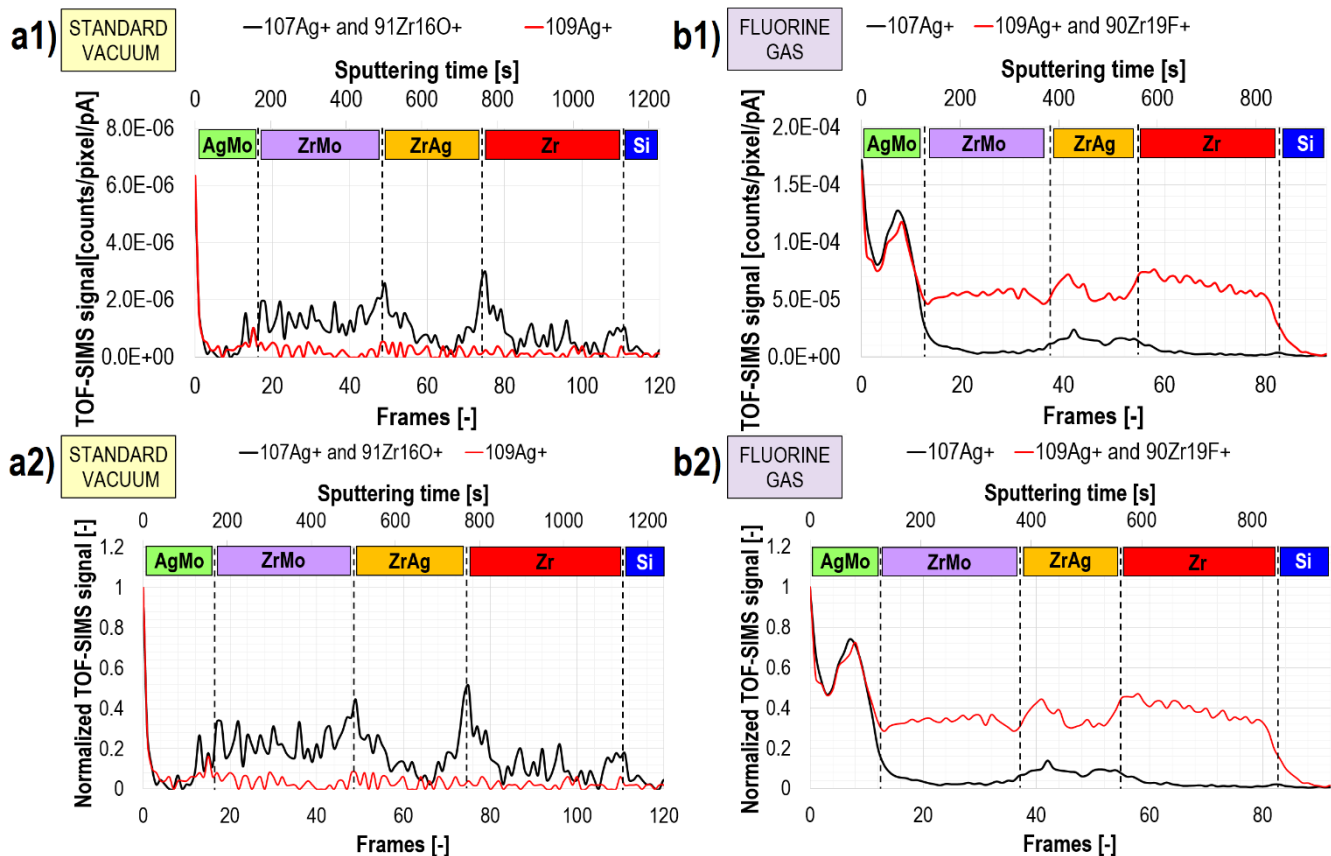


Figure S8. TOF-SIMS depth profiles of signals at $m/q=107$ and 109 , corresponding to $^{107}\text{Ag}^+$ and $^{109}\text{Ag}^+$, respectively, measured without (a1 and a2) and with (b1 and b2) fluorine gas. Figures a1 and b2 are given in absolute values, whilst Figures a2 and b2 show values normalized to 1.

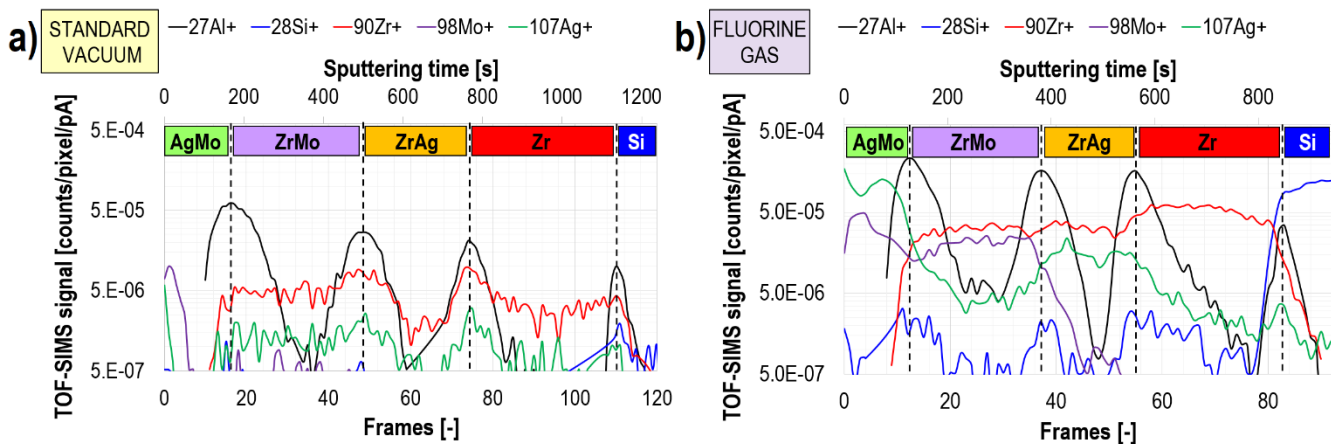


Figure S9. Depth profiles (in logarithmic scale) of AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr multilayer's most prominent isotope signals acquired without (a) and with (b) fluorine gas. Data integration over $5\mu\text{m}\times 5\mu\text{m}$ in the x-y plane.

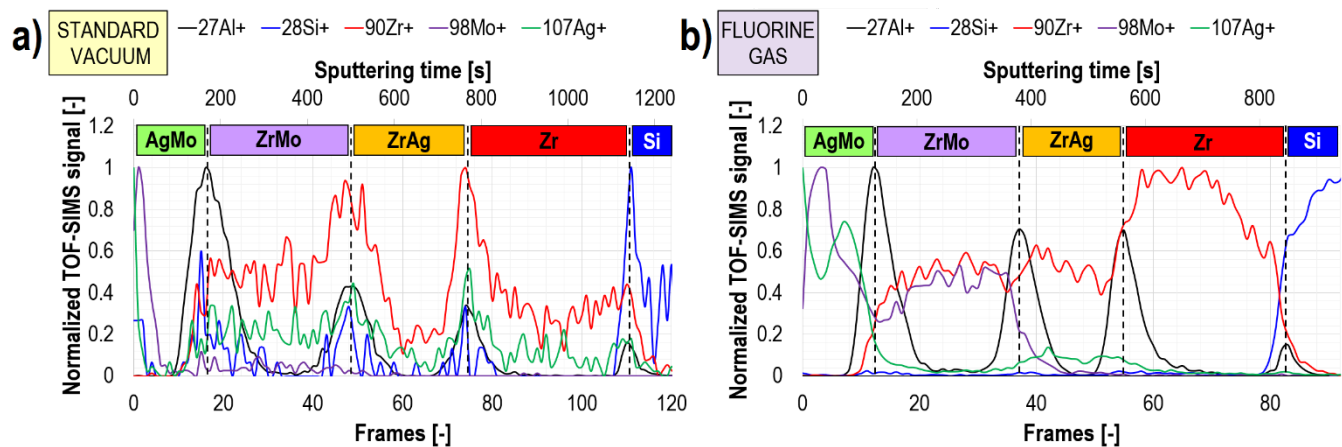


Figure S10. Normalized to 1 depth profiles of AgMo/Al₂O₃/ZrMo/Al₂O₃/ZrAg/Al₂O₃/Zr multilayer's most prominent isotope signals acquired without (a) and with (b) fluorine gas. Data integration over 5 μ m \times 5 μ m in the *x-y* plane. Depth profiles with absolute values are given in Figure S9.