

Supporting Information

Nitrogen-Doped MoS₂/Ti₃C₂T_x Heterostructures as Ultra-Efficient Alkaline HER Electrocatalysts

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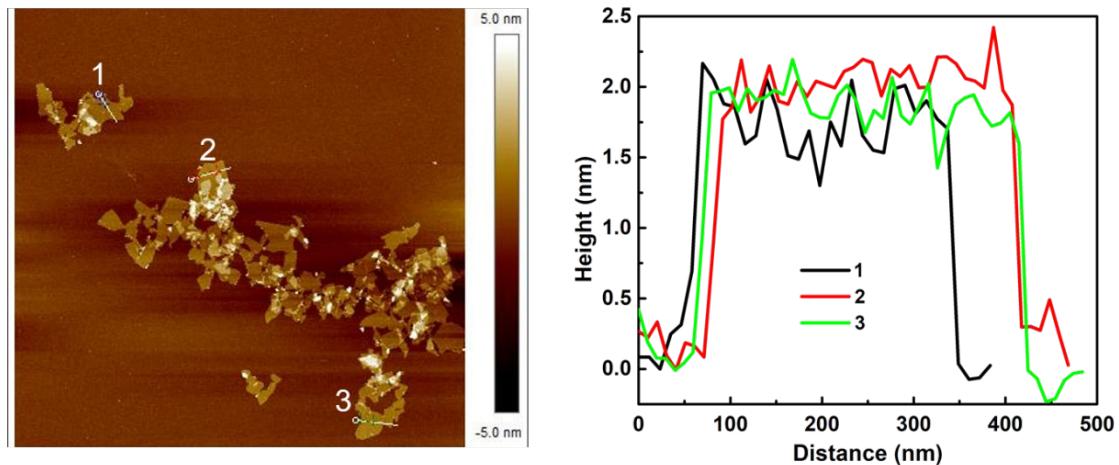


Figure S1. The AFM image and the corresponding height profile of single-layer $\text{Ti}_3\text{C}_2\text{T}_x$

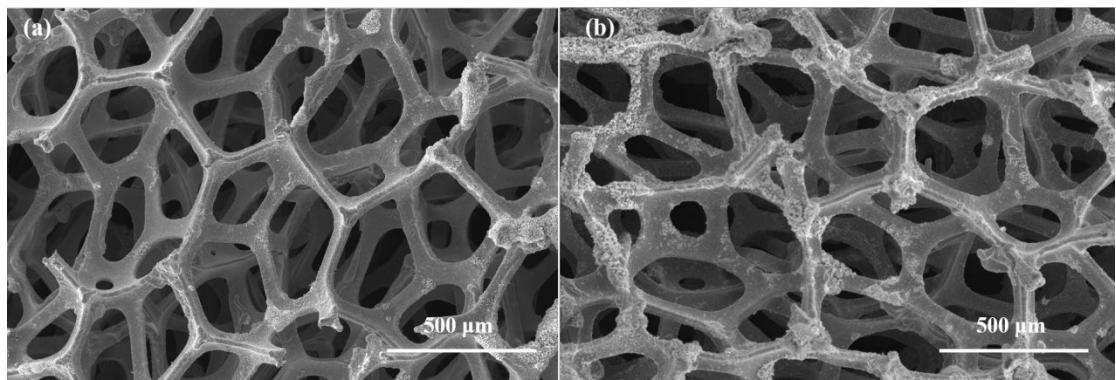


Figure S2. The SEM images of (a) $\text{MoS}_2/\text{Ti}_3\text{C}_2\text{T}_\text{X}$ and (b) N-doped $\text{MoS}_2/\text{Ti}_3\text{C}_2\text{T}_\text{X}$ heterostructures on Nickel foam.

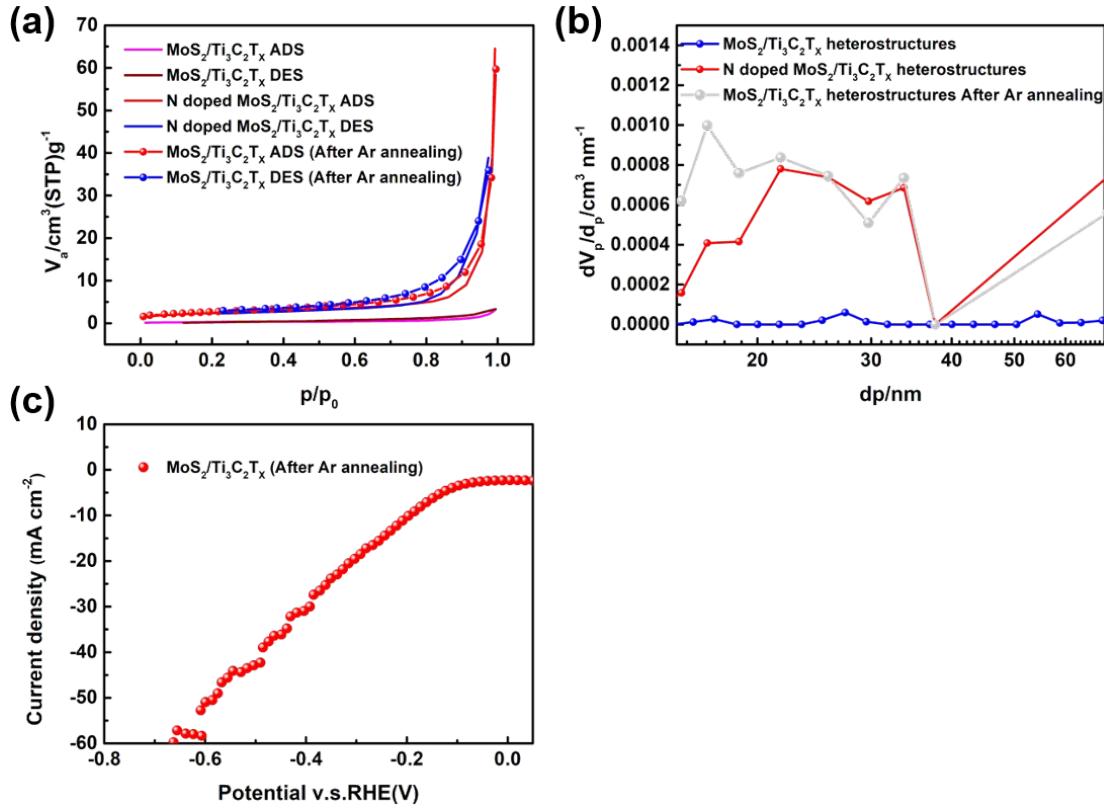


Figure S3. (a) N₂ adsorption-desorption isotherms and (b) the pore size distribution of N-doped MoS₂/Ti₃C₂T_x heterostructures and MoS₂/Ti₃C₂T_x heterostructures on Ni foam. (c) LSV curves of MoS₂/Ti₃C₂T_x heterostructures after Ar annealing in 1.0 M KOH

Then we annealed MoS₂/Ti₃C₂T_x heterostructures in Ar. We found that the pore size distributions of MoS₂/Ti₃C₂T_x heterostructures and N-doped MoS₂/Ti₃C₂T_x heterostructures are not much different, but the overpotential of N-doped MoS₂/Ti₃C₂T_x heterostructures (80 mV at 10 mA cm⁻²) is much smaller than that of MoS₂/Ti₃C₂T_x heterostructures (195 mV at 10 mA cm⁻²), thus indicating that the main source of activity in N-doped MoS₂/Ti₃C₂T_x heterostructures is not microstructure difference but N-doped.

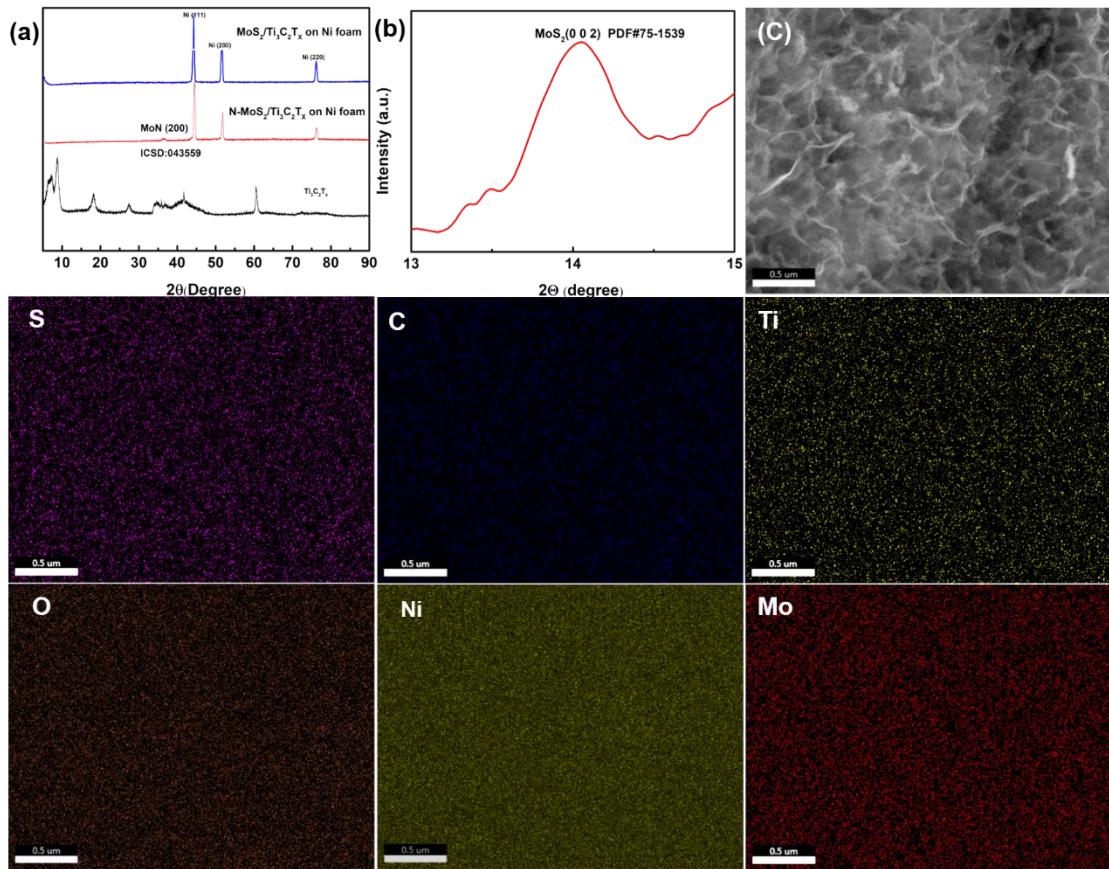


Figure S4 (a) The XRD patterns of MoS₂/Ti₃C₂T_x Heterostructures on Ni foam, and N-doped MoS₂/Ti₃C₂T_x Heterostructures on Ni foam. and (b) The low-angle XRD patterns of MoS₂/Ti₃C₂T_x Heterostructures (c) MoS₂/Ti₃C₂T_x heterostructures on Ni foam SEM and the corresponding EDS mapping images.

As shown in Figure S4a, the patterns of samples coating on Ni foam show obvious Ni peak (PDF#04-0850) and week diffraction peaks of MoS₂ that is due to low mass loading of MoS₂/Ti₃C₂T_x compared to the crystal Ni foam as the frameworks. According to previous studies, the dried restacked Ti₃C₂T_x nanosheets show obvious XRD diffraction, in line with our XRD pattern of the Ti₃C₂T_x precursor (Figure S4a). However, no obvious diffraction can be observed in the final MoS₂/Ti₃C₂T_x heterostructures on Ni foam. This result may be explained by the relatively low content of the Ti₃C₂T_x in the hybrid, as well as the inhibition of the restacking of Ti₃C₂T_x nanosheets by the MoS₂ crystallites. The low-angle XRD pattern of MoS₂/Ti₃C₂T_x heterostructures (Figure S4b) shows the MoS₂ (002) plane. After

nitridation process, the MoS₂ (002) peak disappeared, which is because N doping causes the crystallinity of MoS₂ to deteriorate.[1] The presence of MoN (200) peak indicates that the Nitrogen atom successfully entered MoS₂. The SEM image and EDS mapping images of MoS₂/Ti₃C₂T_X Heterostructures (Figure S4c) shows the conformal assembly of 2D MoS₂/Ti₃C₂T_X heterostructures on the 3D Ni foam. Above all, Figure S4 demonstrated the successful self-assembly of MoS₂/Ti₃C₂T_X heterostructures on Ni foam. Additionally, in this work all the catalysts that were mentioned are on the Ni foam if there is no exceptional explanation.

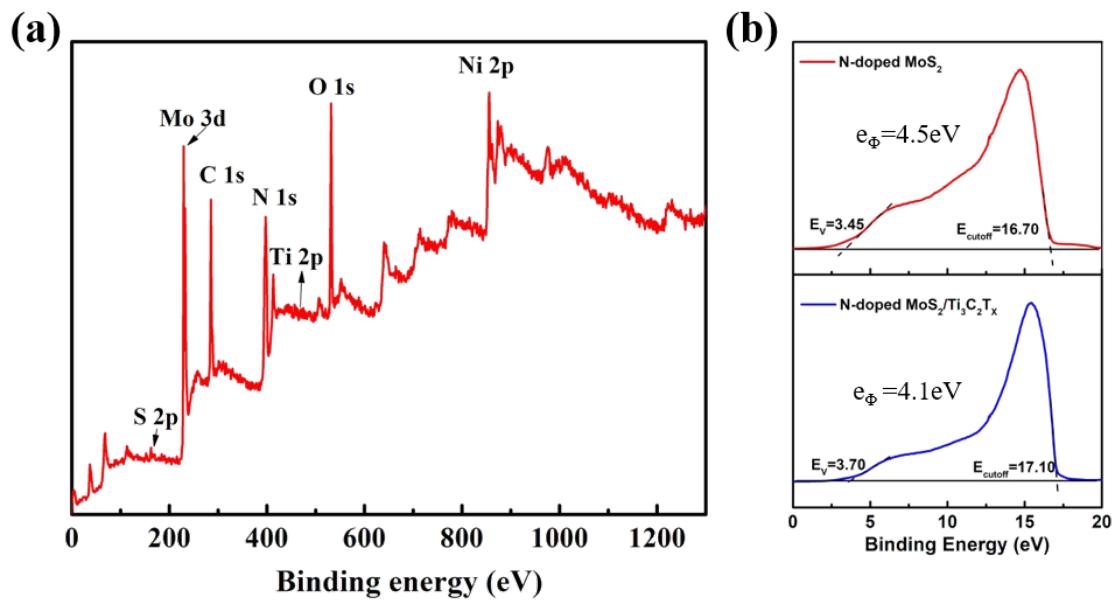


Figure S5. (a) The XPS spectrum of N-doped $\text{MoS}_2/\text{Ti}_3\text{C}_2\text{T}_x$ (b) UPS spectra of N-doped MoS_2 and N-doped $\text{MoS}_2/\text{Ti}_3\text{C}_2\text{T}_x$.

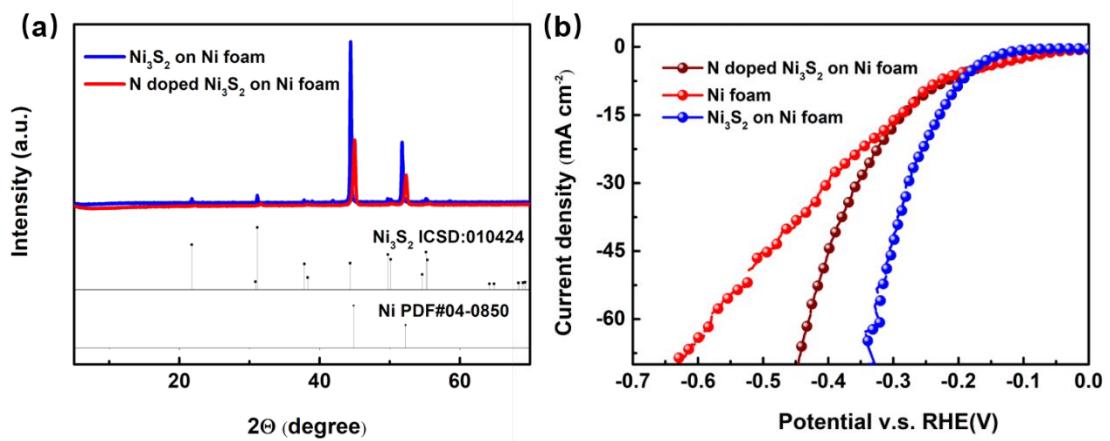


Figure S6. (a) XRD pattern. (b) LSV curves of Ni foam before and after the sulfidizing and nitriding treatment in 1.0 M KOH

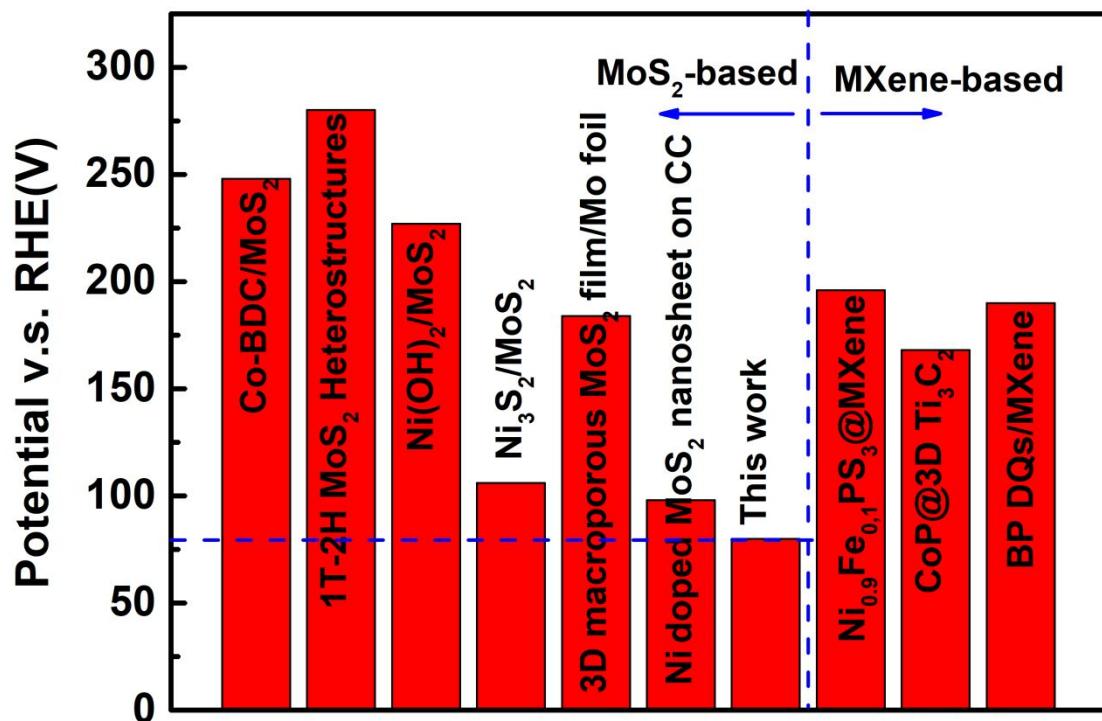


Figure S7. The comparison of the HER activity of this work with the reported MoS₂-/MXene-based HER catalysts.

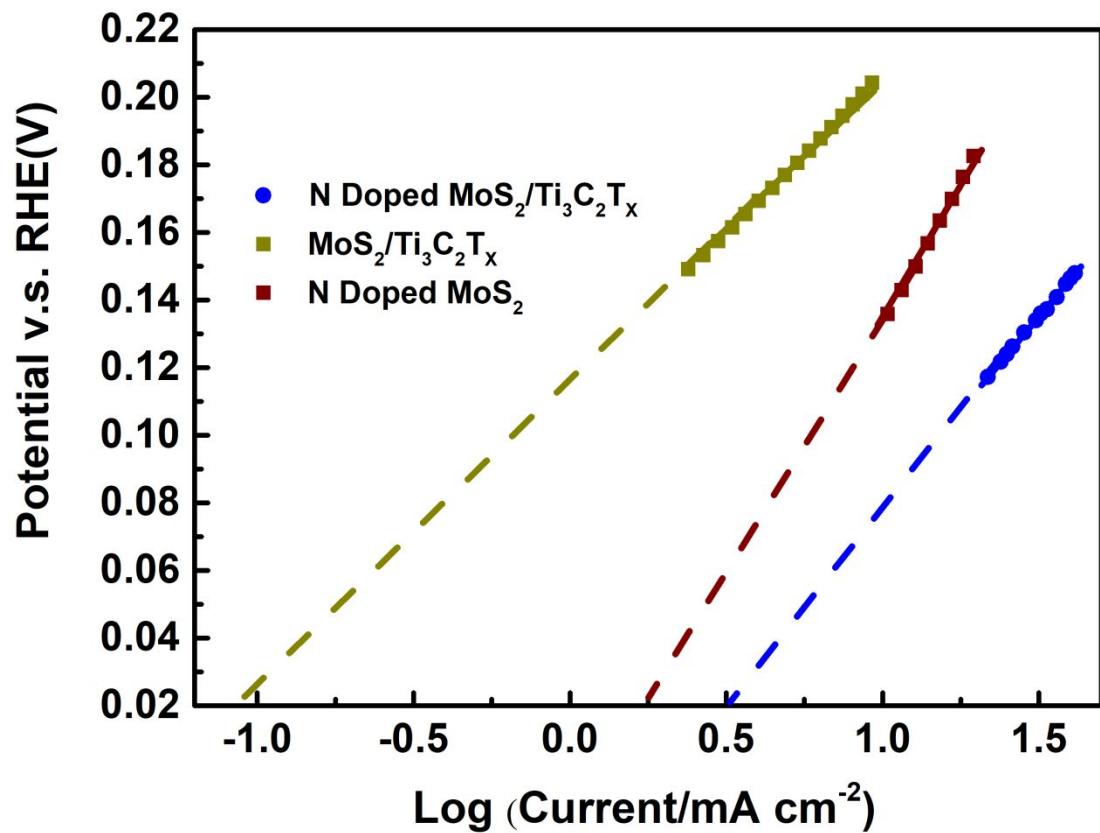


Figure S8. Tafel plot in the region of current densities of N-doped MoS₂, N-doped MoS₂/Ti₃C₂T_x and MoS₂ in 1M KOH.

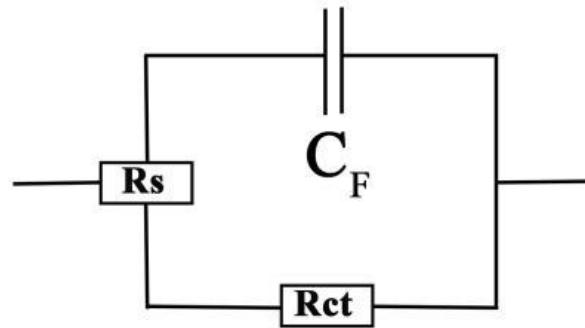


Figure S9. Equivalent circuit model fitted from the EIS data.

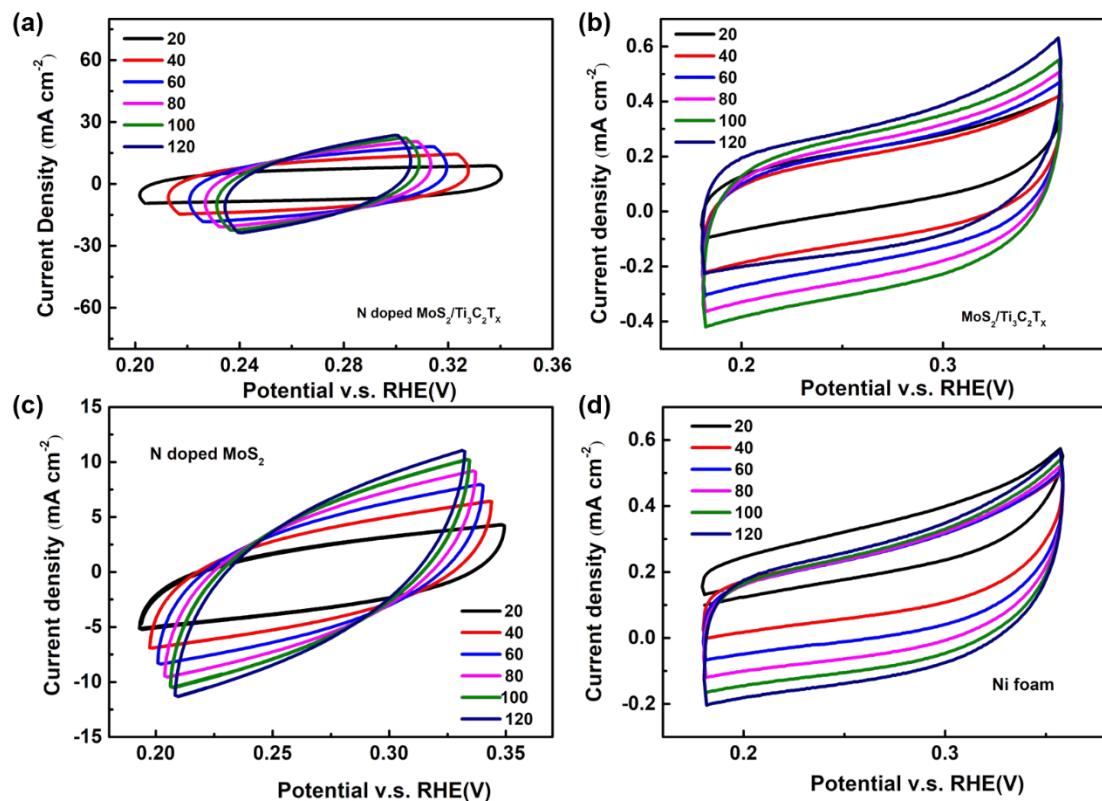


Figure S10 CV curves at different scan rate in the range from 20 mV s^{-1} to 120 mV s^{-1} , of (a) N-doped $\text{MoS}_2/\text{Ti}_3\text{C}_2\text{T}_x$, (b) $\text{MoS}_2/\text{Ti}_3\text{C}_2\text{T}_x$, (c) N-doped MoS_2 and (d) Ni foam.

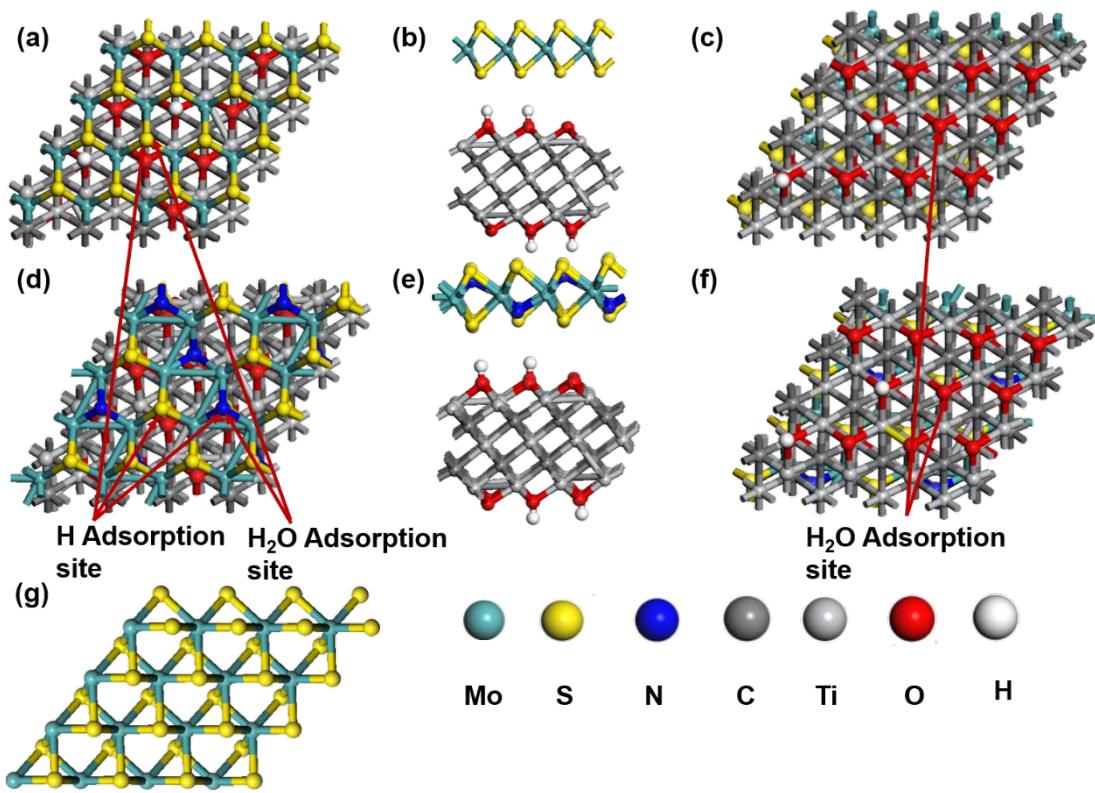


Figure S11. (a) The top, (b) side and (c) bottom view of MoS₂/Ti₃C₂T_X model; (d) The top, (e) side and (f) bottom view of model of N-doped MoS₂/Ti₃C₂T_X. (g) The model of MoS₂.

Table S1. Comparison of HER electrocatalytic performances between N-doped MoS₂/Ti₃C₂T_x in this work with reported MXene-/MoS₂-based HER catalysts in alkaline solution

Sample	Electrolyte	η_{10} (mV vs RHE)	Tafel slope (mV dec ⁻¹)	Ref
N doped MoS₂/ Ti₃C₂T_x	1M KOH	80	100	This work
PtO_aPdO_b	1M KOH	57	95	[2]
Ni_{0.9}Fe_{0.1}PS₃@MXene	1M KOH	196	114	[3]
CoP@3D Ti₃C₂	1M KOH	168	58	[4]
BP QDs/MXene	1M KOH	190	83	[5]
Co-BDC/MoS₂	1M KOH	248		[6]
1T-2H MoS₂	1.0M KOH	280		[7]
Heterostructures				
Ni(OH)₂/MoS₂	1.0M KOH	227		[8]
MoS₂/Ni₃S₂	1.0M KOH	110		[9]
D macroporous MoS₂ film/Mo foil	1.0M KOH	184		[10]
Ni doped MoS₂ nanosheets on CC	1.0M KOH	98		[11]

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