

Supporting Information for: Relationship between the Relaxation of Ionic Liquid Structural Motifs and That of the Shear Viscosity

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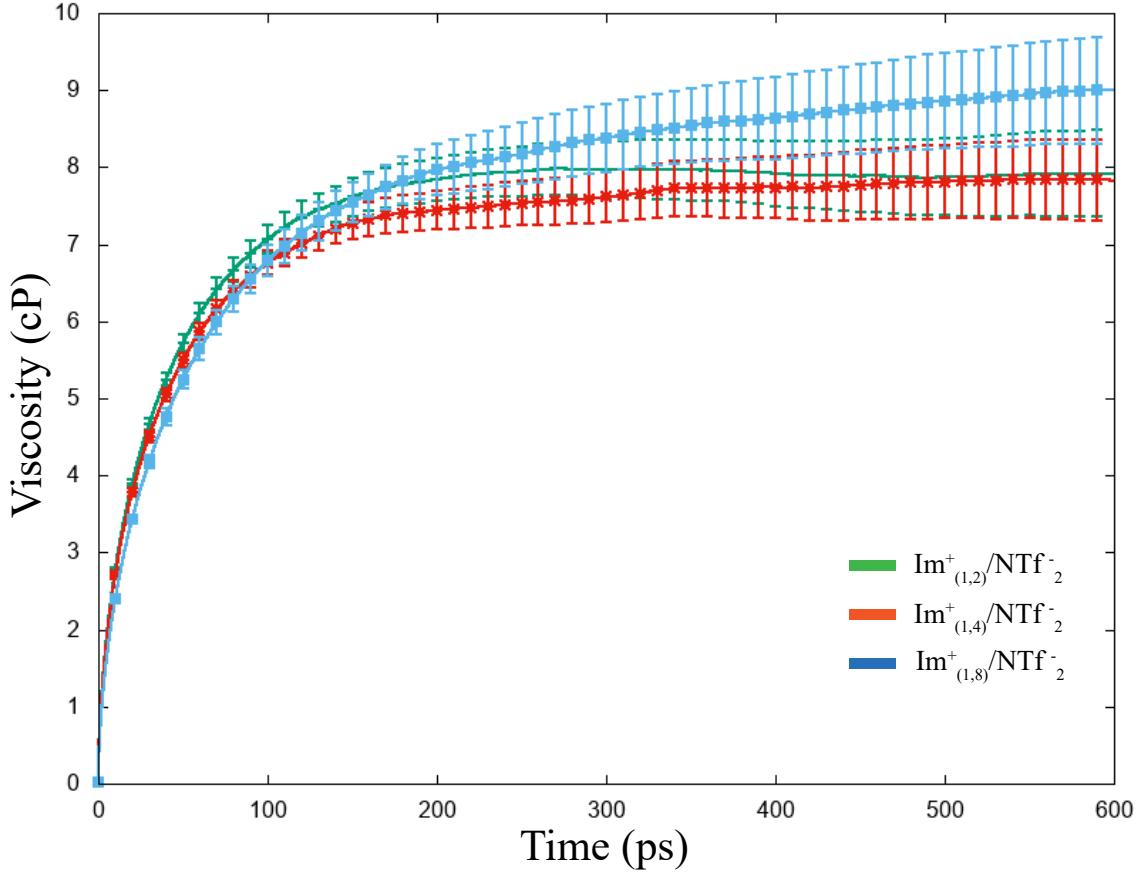


Figure S.1: The Green-Kubo expression $\frac{1}{k_B T V} \int_0^t \langle \sigma^{xx}(0) \sigma^{xx}(t') \rangle dt'$ and error bars at the 95 percent confidence level for $\text{Im}^+_{(1,2)}/\text{NTf}_2^-$, $\text{Im}^+_{(1,4)}/\text{NTf}_2^-$, and $\text{Im}^+_{(1,8)}/\text{NTf}_2^-$. The overlap in the error bars becomes even larger close to the corresponding t_∞ (data not shown for clarity).

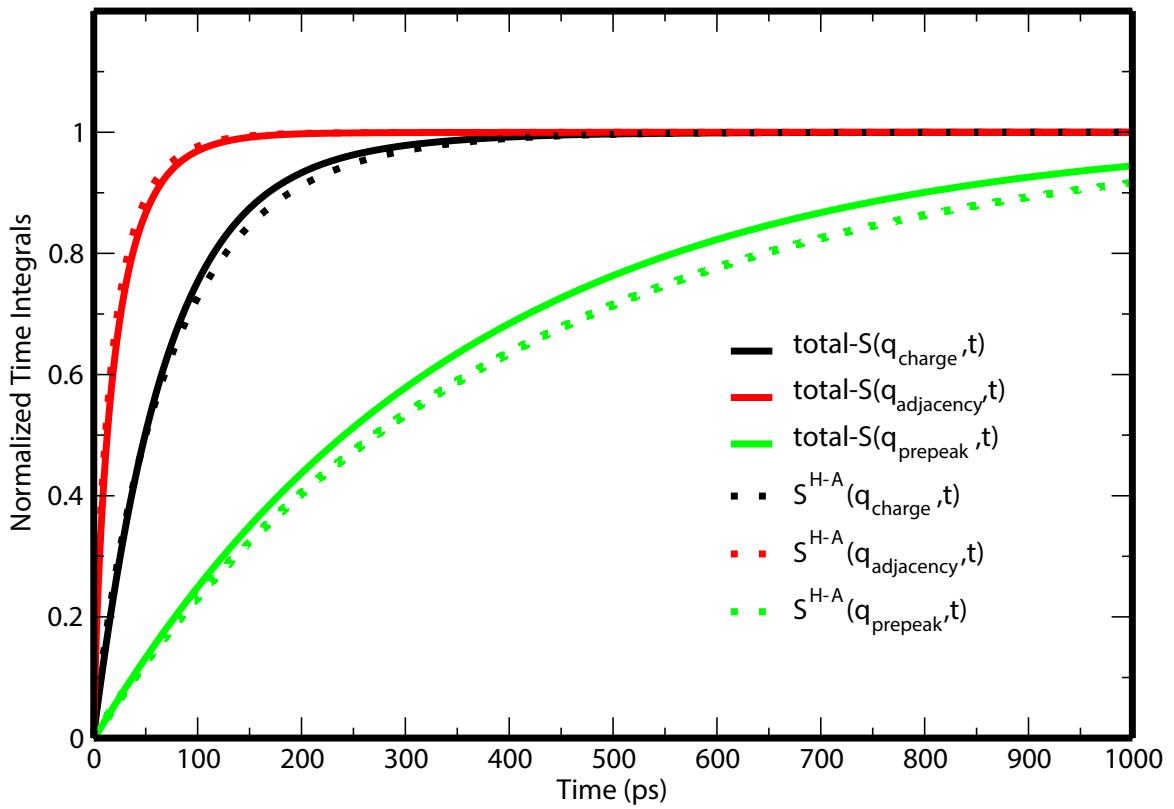


Figure S.2: For $\text{Im}_{1,8}^+/\text{NTf}_2^-$, a comparison between $\int_0^t S(q,t')^2 dt' / \int_0^{t_\infty} S(q,t)^2 dt$ and $\alpha(q,t)$, at the q -regions associated with the structural motifs.

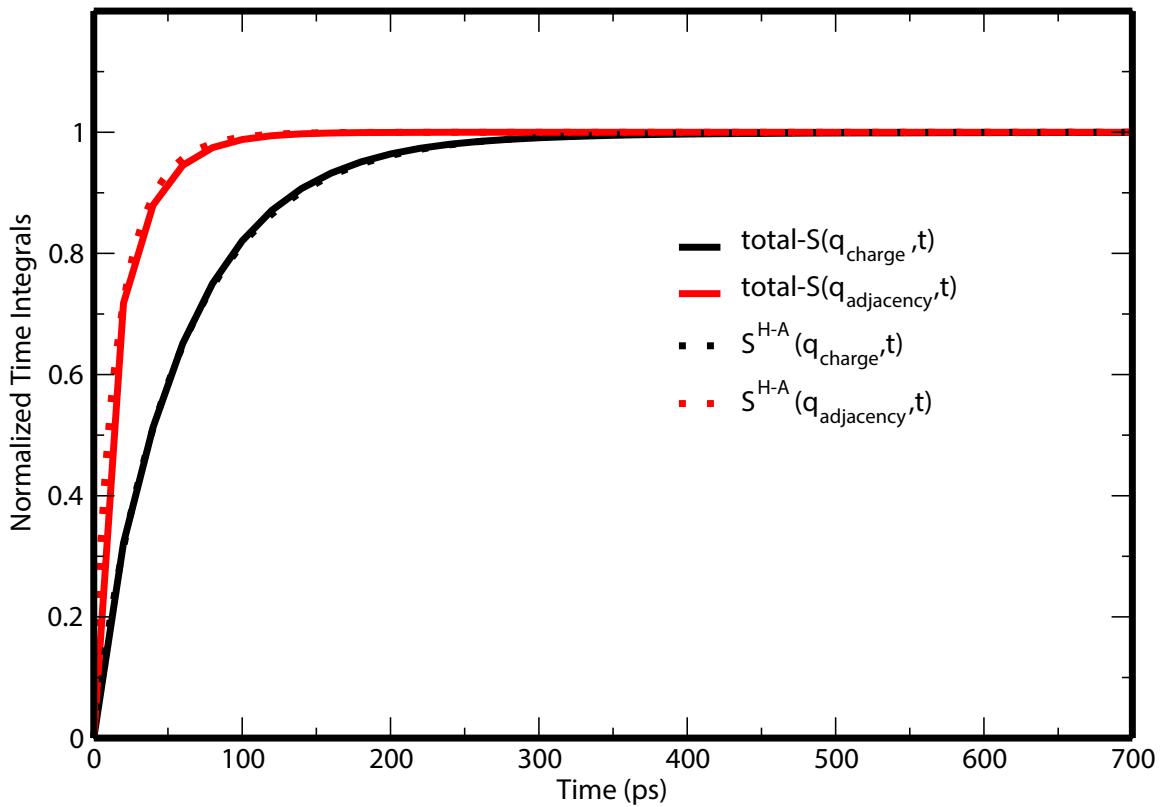


Figure S.3: For $\text{Im}_{1,4}^+/\text{NTf}_2^-$, a comparison between $\int_0^t S(q, t')^2 dt' / \int_0^{t_\infty} S(q, t)^2 dt$ and $\alpha(q, t)$, at the q-regions associated with the structural motifs.

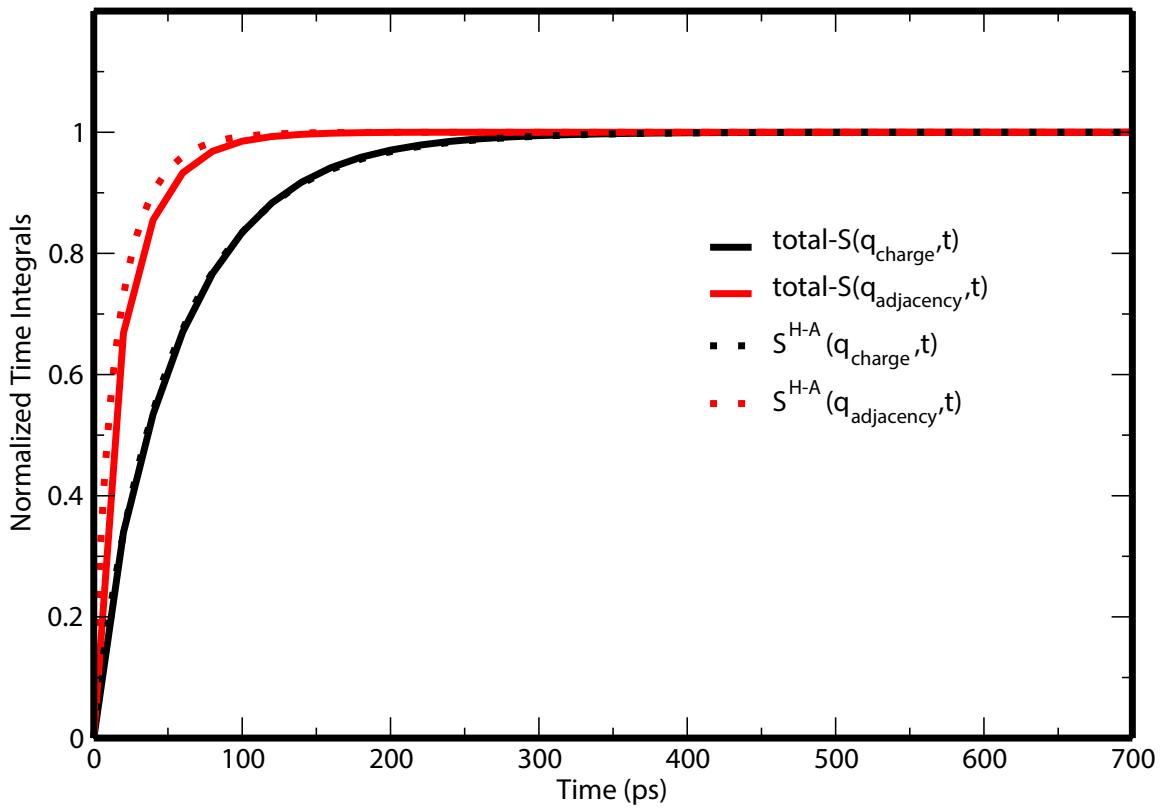


Figure S.4: For $\text{Im}_{1,2}^+/\text{NTf}_2^-$, a comparison between $\int_0^t S(q, t')^2 dt' / \int_0^{t_\infty} S(q, t)^2 dt$ and $\alpha(q, t)$, at the q-regions associated with the structural motifs.

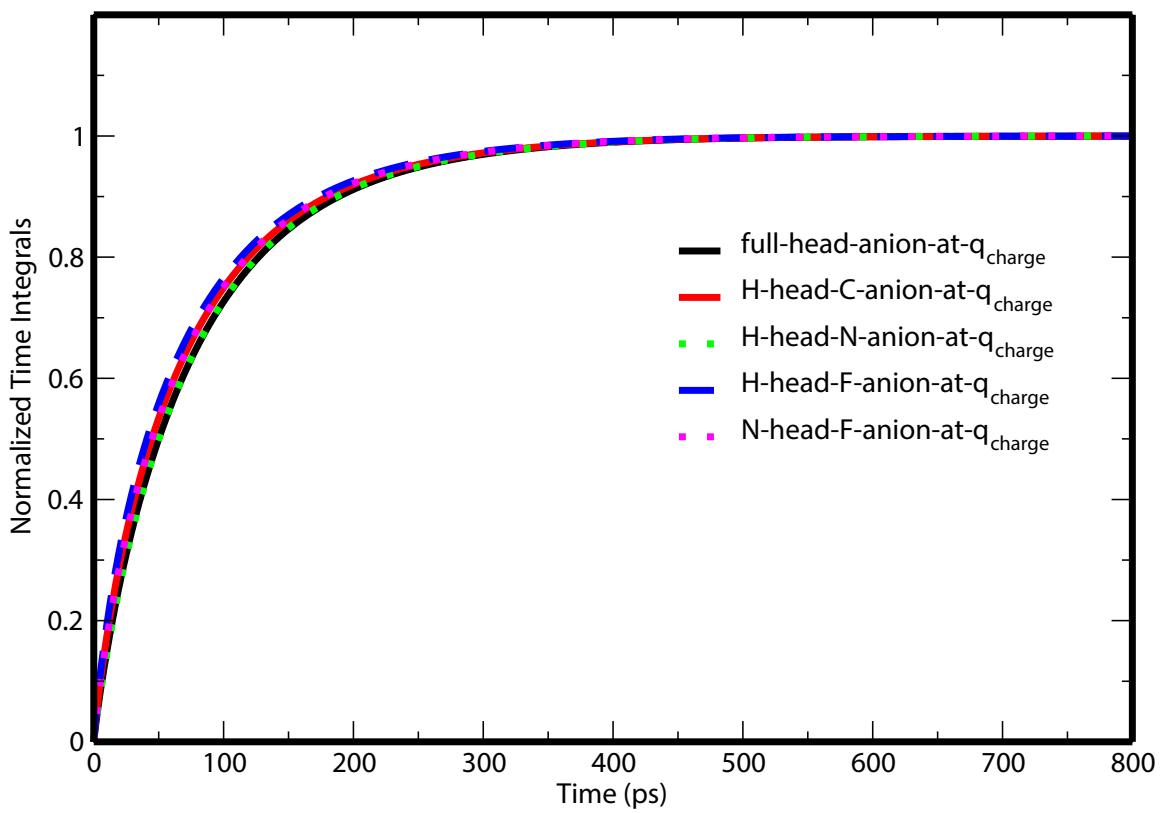


Figure S.5: For $\text{Im}_{1.8}^+/\text{NTf}_2^-$, $\alpha(q,t)$ compared with some of its partial atomic subcomponents.

Table S.1: Biexponential time constants (τ_1 and τ_2), as well as parameters A and α used to fit the relaxation of $\zeta(t)$ as defined in reference 1; computed viscosities for our three ILs. $\zeta(t) = A\alpha\tau_1(1 - e^{-t/\tau_1}) + A(1 - \alpha)\tau_2(1 - e^{-t/\tau_2})$.

IL	Viscosity (cP)	A	α	τ_1 (ps)	τ_2 (ps)
Im _{1,8} ⁺ /NTf ₂ ⁻	8.8 (at 377.4 K)	0.028491	0.135804	135.514	19.5938
Im _{1,4} ⁺ /NTf ₂ ⁻	7.7 (at 363.3 K)	0.055910	0.175726	64.8021	7.86063
Im _{1,2} ⁺ /NTf ₂ ⁻	7.9 (at 354.4 K)	0.081545	0.196336	47.9468	3.44041

References

- (1) Zhang, Y.; Otani, A.; Maginn, E. J. Reliable Viscosity Calculation from Equilibrium Molecular Dynamics Simulations: A Time Decomposition Method. *Journal of Chemical Theory and Computation* **2015**, *11*, 3537–3546.