# Divalent Ytterbium Hydrido Complex Supported by $\beta$-Diketiminato Based Tetradentate Ligand: Synthesis, Structure and Reactivity 

Qingqing Wen, ${ }^{\dagger}$ Bin Feng, ${ }^{\dagger}$ Li Xiang, ${ }^{\dagger}{ }^{\dagger *}$ Xuebing Leng, ${ }^{\dagger}$ and Yaofeng Chen ${ }^{\dagger}{ }^{*}$
${ }^{\dagger}$ State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, University of Chinese Academy of Sciences, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, P. R. China.
${ }^{\ddagger}$ Chongqing Key Laboratory of Green Synthesis and Applications, College of Chemistry, Chongqing Normal University, Chongqing 401331, P. R. China.

## Supporting Information

## Contents

1. General ..... S1
2. Synthesis and characterization data of new complexes ..... S2
3. X-ray crystallography ..... S9
4. NMR spectra of new complexes ..... S13
5. References ..... S41

General. All manipulations were performed under an atmosphere of nitrogen using Schlenk techniques or in a nitrogen-filled glovebox. THF, toluene, benzene, hexane, $\mathrm{C}_{6} \mathrm{D}_{6}, d_{8}$-toluene, $d_{8}$-THF and mesitylene were dried over $\mathrm{Na} / \mathrm{K}$ alloy, transferred under vacuum, and stored in the glovebox. Cyclohexane, $\mathrm{PhSiH}_{3}$ and pyridine were dried over $\mathrm{CaH}_{2}$, transferred under vacuum, degassed by three freeze-pump-thaw cycles, and stored in the glovebox. $\mathrm{YbI}_{2}(\mathrm{THF})_{2}{ }^{1}, \mathrm{KCH}_{2} \mathrm{SiMe}_{3}{ }^{2}$ and $\mathrm{PhSiD}_{3}{ }^{3}$ were synthesized following the literature procedures. The highly pure $\mathrm{H}_{2}(99.999 \%)$ was further dired by passing through activated $4 \AA$ molecular sieve. 4-(Dimethylamino)pyridine (DMAP), bipyridine (bpy), triphenylphosphine oxide, triphenylphosphine sulfide and diphenyl disulfide were purified by sublimation before use. ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ and ${ }^{31} \mathrm{P}\left\{{ }^{1} \mathrm{H}\right\}$ spectra were recorded on a Varian 400 MHz , an Agilent 400 MHz , a Bruker 400 MHz or an Agilent 600 MHz spectrometer. ${ }^{2} \mathrm{H}$ spectra were recorded on a Bruker 400 MHz spectrometer. Chemical shifts $\delta$ were reported in ppm with references to the residual resonance of the deuterated solvents for proton and carbon spectroscopies, to internal $\mathrm{C}_{6} \mathrm{D}_{6}$ for ${ }^{2} \mathrm{H}$ chemical shifts, and to external $\mathrm{H}_{3} \mathrm{PO}_{4}(85 \%)$ for phosphorus chemical shifts. The assignment of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ resonances was assisted with gCOSY, gHSQC and gHMBC spectra. Elemental analyses were performed by the Analytical Laboratory of Shanghai Institute of Organic Chemistry.
$\mathbf{L K}\left(\mathrm{L}=\left[{ }^{t} \mathrm{BuC}(\mathrm{NDipp}) \mathrm{CHC}\left({ }^{t} \mathrm{Bu}\right) \mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{~N}(\mathrm{Me}) \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NMe}_{2}\right]^{-}\right.$, Dipp $\left.=2,6-\left({ }^{( }{ }^{\mathrm{Pr}}\right)_{2} \mathrm{C}_{6} \mathrm{H}_{3}\right)$ : LK was synthesized similar to that of $\mathrm{LLi}^{4}{ }^{4}$ A THF solution $\left(\begin{array}{lll}20 & \mathrm{~mL}\end{array}\right)$ of $\mathrm{Me}\left({ }^{( } \mathrm{Bu}\right) \mathrm{CNCH}_{2} \mathrm{CH}_{2} \mathrm{~N}(\mathrm{Me}) \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NMe}_{2}(5.00 \mathrm{~g}, 22.0 \mathrm{mmol})$ was precooled to $-78{ }^{\circ} \mathrm{C}$, and then ${ }^{n} \mathrm{BuLi}(22.0 \mathrm{mmol}, 9.2 \mathrm{~mL}$ of a 2.40 M solution in hexane) was added dropwise. The mixture was allowed to warm to room temperature and stirred for 2 h , resulting in an orange solution. This mixture was precooled to $0^{\circ} \mathrm{C}$, and $\mathrm{ClC}\left({ }^{( } \mathrm{Bu}\right)=\mathrm{N}(\mathrm{Dipp})(6.16 \mathrm{~g}, 22.0 \mathrm{mmol})$ was added dropwise. After stirring at room temperature for 6 h , the volatiles of the reaction mixture were removed under vacuum. After that, water $(8 \mathrm{~mL})$ and hexane $(50 \mathrm{~mL})$ were added to the residue. The organic phase was collected, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and filtered. The solvent of the filtration was removed under vacuum. The crude product LH was obtained as a yellow oil $(9.80 \mathrm{~g}, 95 \%$ yield $)$. The crude product could not be purified
by distillation due to its high boiling point, and it is also difficult to be purified by column chromatography due to a hydrolysis of the imine unit in LH. However, the obtained crude product can be used for the synthesis of LK without purification. KH ( $1.32 \mathrm{~g}, 33.0 \mathrm{mmol}$ ) was added to a THF solution ( 20 mL ) of $\mathrm{LH}(9.80 \mathrm{~g}, 20.8 \mathrm{mmol})$ at room temperature, the precipitate was separated by centrifugation after stirring for 12 h . The solvent of the solution was removed under vacuum, the residue was stirred in 10 mL of hexane, and the volatiles were removed again under vacuum. This operation was repeated for three times to remove THF completely. The residue was washed with hexane $(15 \mathrm{~mL} \times 4)$ and dried under vacuum to give LK as a yellow solid ( $6.70 \mathrm{~g}, 63 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, d_{8}-\mathrm{THF}, 25{ }^{\circ} \mathrm{C}$ ): $\delta 6.71\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=7.2 \mathrm{~Hz}, 2 \mathrm{H}, m-\mathrm{ArH}\right.$ of Dipp), $6.29\left(\mathrm{t},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=7.2 \mathrm{~Hz}\right.$, $1 \mathrm{H}, p-\mathrm{Ar} H$ of Dipp), $4.28\left(\mathrm{~s}, 1 \mathrm{H},{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 3.67$ (sept, $\left.{ }^{3} \mathrm{~J}_{\mathrm{H}-\mathrm{H}}=6.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{ArCH} \mathrm{Me}_{2}\right), 2.96(\mathrm{~m}$, $\left.\left.2 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.35\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.26\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.13(\mathrm{~s}, 9 \mathrm{H}, \mathrm{N} M e \text { and } \mathrm{NMe})_{2}\right), 2.10(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{NCH}_{2}\right), 1.26\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CMe} e_{3}\right), 1.18\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.2 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{ArCHMe} 2\right), 1.08\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CMe} e_{3}\right), 1.01\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}\right.$ $\left.=6.2 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{ArCH} M e_{2}\right) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \operatorname{NMR}\left(100 \mathrm{MHz}, d_{8}\right.$-THF, $25^{\circ} \mathrm{C}$ ): $\delta 180.6,165.3$ (imine $C$ ), 152.6 ( $i-\operatorname{ArC}$ of Dipp), 138.2 ( $o-\operatorname{ArC}$ of Dipp), 120.4 ( $m-\operatorname{ArC}$ of Dipp), 113.6 ( $p-\operatorname{ArC}$ of Dipp), 78.8 $\left({ }^{( } \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 63.0,58.3,57.6,53.2\left(\mathrm{NCH}_{2}\right), 45.8(\mathrm{NMe} 2), 43.1(\mathrm{NMe}), 41.9,39.8\left(\mathrm{CMe}_{3}\right), 32.0,31.3$ ( $\mathrm{CMe}_{3}$ ), 28.6 ( $\mathrm{ArCHMe}_{2}$ ), 24.4, 22.2 ( $\mathrm{ArCHMe}_{2}$ ).
$[\mathbf{L Y b}(\boldsymbol{\mu}-\mathbf{I})]_{2}(\mathbf{1}):$ To a THF solution $(3 \mathrm{~mL})$ of $\mathrm{YbI}_{2}(\mathrm{THF})_{2}(571 \mathrm{mg}, 1.0 \mathrm{mmol})$ was added a THF solution ( 3 mL ) of LK ( $509 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) at room temperature. The reaction was stirred for 2 h , resulting in a dark brown solution with gray precipitates. After filtration, the solvent of the filtration was removed under vacuum. The residue was stirred in 5 mL of hexane, and the volatiles were removed again under vacuum. This operation was repeated for three times to remove THF completely, and then the residue was extracted with toluene ( 15 mL ). The solvent of the extraction was removed under vacuum, and the residue was washed with hexane ( $5 \mathrm{~mL} \times 3$ ) and dried under vacuum to give complex 1 as a dark red solid ( $400 \mathrm{mg}, 52 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta 6.98(\mathrm{~m}, 1 \mathrm{H}, \mathrm{p}-\mathrm{ArH}$ of Dipp), 6.91 (m, 2H, m-ArH of Dipp), 5.35 (s, 1H, $\left.{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 4.12\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 3.62(\mathrm{br}, 1 \mathrm{H}$, $\mathrm{ArCH} \mathrm{Me}_{2}$ ), 3.15 (br, $2 \mathrm{H}, \mathrm{NCH}_{2}$ and $\mathrm{ArCH} \mathrm{Me}_{2}$ ), 2.67 (m, $1 \mathrm{H}, \mathrm{NCH}$ ), 2.25 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{NMe}$ ), $2.10-1.90$
$\left(\mathrm{m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.72\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.53\left(\mathrm{br}, 10 \mathrm{H}, \mathrm{NMe}_{2}, \mathrm{NCH}_{2}\right.$ and ArCHMe 2$), 1.46(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CMe} 3)$, $1.33\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{ArCH} M e_{2}\right), 1.21$ (br, $12 \mathrm{H}, \mathrm{CMe}_{3}$ and $\mathrm{ArCH} M e_{2}$ ), 1.09 (br, $3 \mathrm{H}, \mathrm{ArCH} M e_{2}$ ). ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta$ 172.1, 171.2 (imine $C$ ), 147.5 ( $i-\mathrm{ArC}$ of Dipp), 140.3, 139.1 ( $o-A r C$ of Dipp), 124.0, 123.2 ( $m-A r C$ of Dipp), 123.0 ( $p-\mathrm{ArC}$ of Dipp), 89.8 ( $\left.{ }^{t} \mathrm{BuC}(\mathrm{N}) C \mathrm{H}\right), 57.9,57.2$, 55.3, $47.7\left(\mathrm{NCH}_{2}\right), 44.7(\mathrm{NMe}), 44.4\left(\mathrm{CMe}_{3}\right), 43.5(\mathrm{NMe}), 40.8\left(\mathrm{CMe}_{3}\right), 32.4$, $31.0\left(\mathrm{CMe}_{3}\right), 29.3$ ( $\mathrm{ArCHMe} e_{2}$ ), 27.8 ( $\mathrm{ArCHMe}_{2}$ ), 26.1, 23.8, 22.8 ( $\mathrm{ArCHMe} e_{2}$ ). Anal. Calcd for $\mathrm{C}_{60} \mathrm{H}_{106} \mathrm{I}_{2} \mathrm{~N}_{8} \mathrm{Yb}_{2}: \mathrm{C} 46.81$; H 6.94; N 7.28. Found: C 46.97; H 6.99; N 7.35.
$\mathbf{L Y b C H} 2 \mathbf{S i M e}_{3}$ (2): Complex $1(154 \mathrm{mg}, 0.10 \mathrm{mmol})$ and $\mathrm{KCH}_{2} \mathrm{SiMe}_{3}(27.0 \mathrm{mg}, 0.20 \mathrm{mmol})$ were mixed in 4 mL of benzene at room temperature, resulting in a dark brown solution with gray precipitates. The reaction was kept at room temperature for 30 minutes, and then the solvent was removed under vacuum. The residue was extracted with hexane $(8 \mathrm{~mL})$. The solvent of the extraction was removed under vacuum to afford complex $\mathbf{2}$ as a dark red solid ( $132 \mathrm{mg}, 90 \%$ yield). ${ }^{1} \mathrm{H}$ NMR (400 MHz, C ${ }_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta 7.00$ (m, 1H, $p-\mathrm{ArH}$ of Dipp), 6.92 (m, 2H, $m-\mathrm{ArH}$ of Dipp), 5.34 (s, 1 H , $\left.{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{C} H\right), 4.16\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 3.78\left(\mathrm{sept},{ }^{3} \mathrm{~J}_{\mathrm{H}-\mathrm{H}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArCHMe}_{2}\right), 3.15\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right)$, 3.12 (sept, $\left.{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArCHMe} 2\right), 2.65\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.06(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N} M e), 1.95(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{NCH}_{2}\right), 1.80(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N} M e), 1.58\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.52\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \operatorname{ArCH} M e_{2}\right), 1.49(\mathrm{~s}, 9 \mathrm{H}$, $\left.\left.\mathrm{CMe}_{3}\right), 1.41\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{ArCH} M e_{2}\right), 1.35(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH} 2), 1.26(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CMe})^{3}\right), 1.20\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}\right.$ $=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{ArCHMe}), 1.06(\mathrm{~m}, 6 \mathrm{H}, \mathrm{ArCHMe} 2$ and NMe$), 0.47\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CH}_{2} \mathrm{SiMe}_{3}\right),-1.19(\mathrm{~d}, 2 \mathrm{H}$, $\left.\mathrm{YbCH}_{2}\right) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}$ ): $\delta 171.2,170.6$ (imine $C$ ), 148.2 ( $i-\mathrm{ArC}$ of Dipp), 140.1, 139.5 ( $o-\operatorname{Ar} C$ of Dipp), 124.0 ( $p-\operatorname{Ar} C$ of Dipp), 123.1, 122.3 ( $m-\operatorname{Ar} C$ of Dipp), 91.9 $\left({ }^{t} \mathrm{BuC}(\mathrm{N}) C \mathrm{H}\right), 57.3,56.9,55.4,47.4\left(\mathrm{NCH}_{2}\right), 44.6\left(\mathrm{CMe}_{3}\right), 44.5,43.8,41.9(\mathrm{~N} M e), 41.2\left(C \mathrm{Me}_{3}\right), 32.5$, 31.2 ( $\mathrm{CMe}_{3}$ ), 27.9 ( ArCHMe 2 ), 27.7, 27.5 ( $\mathrm{ArCHMe}_{2}$ ), 25.9, 24.1, 22.8 ( $\mathrm{ArCHMe} e_{2}$ ), $17.95\left(\mathrm{YbCH}_{2}\right)$, 5.97 (SiMe 3 ). Anal. Calcd for $\mathrm{C}_{34} \mathrm{H}_{64} \mathrm{~N}_{4} \mathrm{SiYb}$ : C 55.94; H 8.84; N 7.67. Found: C 56.04; H 9.04; N 7.77.
$[\mathbf{L Y b}(\mu-\mathbf{H})]_{2}(\mathbf{3}):$ Method $A$. A hexane solution $(3 \mathrm{~mL})$ of $\mathbf{2}(100 \mathrm{mg}, 0.14 \mathrm{mmol})$ was placed in tube
with a Teflon stopcock. The tube was taken out of the glovebox and connected to a Schlenk line. The solution of $\mathbf{2}$ was degassed at low temperature, and then exposed to 1.0 atm of $\mathrm{H}_{2}$ at room temperature. After standing at room temperature for 3 h , the resulting solution was concentrated to about 0.5 mL and stood at $-35^{\circ} \mathrm{C}$ for 12 h . Complex $\mathbf{3}$ was isolated as dark red crystals ( $43 \mathrm{mg}, 49 \%$ yield). Method B. In the glove box, to a hexane solution ( 3 mL ) of $\mathbf{2}(150 \mathrm{mg}, 0.21 \mathrm{mmol})$ was added a hexane solution $(131 \mathrm{mg})$ of $\mathrm{PhSiH}_{3}(22 \mathrm{mg}, 0.21 \mathrm{mmol})$ at room temperature, resulting in a dark red solution. After standing at room temperature for 10 minutes, the solution was concentrated to about 2 mL and stood at $-35^{\circ} \mathrm{C}$ for 12 h . The resulting dark red crystalline solids were collected, washed with hexane $(1 \mathrm{~mL} \times 3)$, and dried under vacuum to afford $\mathbf{3}(116 \mathrm{mg}, 88 \%$ yield). There are two isomers in about 4:1 ratio. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}$ ): isomer $\mathbf{I}, \delta 8.95\left(\mathrm{t},{ }^{1} J_{\mathrm{Yb}-\mathrm{H}}=358 \mathrm{~Hz}, 1.6 \mathrm{H}, \mathrm{Yb} H\right)$. isomer II, $\delta 9.04\left(\mathrm{t},{ }^{1} J_{\mathrm{Yb}-\mathrm{H}}=352 \mathrm{~Hz}, 0.4 \mathrm{H}, \mathrm{Yb} H\right)$. Most other signals of two isomers are overlapped. $7.12(\mathrm{~m}$, $2 \mathrm{H}, p-\mathrm{ArH}$ of Dipp), $7.05\left(\mathrm{~m}, 4 \mathrm{H}, m-\mathrm{ArH}\right.$ of Dipp), $4.95\left(\mathrm{~s}, 2 \mathrm{H},{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 4.16\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right)$, 3.61 (m, 2H, ArCHMe 2 ), $3.20\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArCHMe}\right.$ ), 2.83 (m, 4H, NCH $\mathrm{N}_{2}$, 2.41 ( $\mathrm{s}, 6 \mathrm{H}, \mathrm{NMe}$ ), 2.33 (m, $\left.4 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.18\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.90\left(\mathrm{br}, 12 \mathrm{H}, \mathrm{N} M e_{2}\right), 1.48-1.44\left(\mathrm{~m}, 30 \mathrm{H}, \mathrm{ArCH} \mathrm{Me}_{2}\right.$ and CMe 3 ), 1.32-1.28 (m, 30H, $\mathrm{ArCH} \mathrm{Me}_{2}$ and CMe 3 ). As some signals in the ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{3}$ are overlapped and 3 easily decomposes into another complex in solution, we were failed to determine the specific structure of the isomers by ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NOESY spectroscopy. The poor solubility and instability of $\mathbf{3}$ also caused the difficulty in recording the ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ and ${ }^{171} \mathrm{Yb}$ NMR spectra of this complex. Anal. Calcd for $\mathrm{C}_{60} \mathrm{H}_{108} \mathrm{~N}_{8} \mathrm{Yb}_{2}$ : C 55.97; H 8.45; N 8.70. Found: C 55.76; H 8.24; N 8.74 .
$[\mathbf{L Y b}(\boldsymbol{\mu}-\mathbf{D})]_{2}(\mathbf{3}-\mathbf{D})$ : To a hexane solution ( 3 mL ) of $\mathbf{2}(100 \mathrm{mg}, 0.14 \mathrm{mmol})$ was added a hexane solution ( 86 mg ) of $\mathrm{PhSiD}_{3}(16 \mathrm{mg}, 0.14 \mathrm{mmol})$ at room temperature, resulting in a dark red solution. After standing at room temperature for 10 minutes, the solution was concentrated to about 2 mL and stood at $-35^{\circ} \mathrm{C}$ for 12 h . The resulting dark red crystalline solids were collected, washed with hexane $(1 \mathrm{~mL}$ $\times 3$ ), and dried under vacuum to afford 3-D ( $78 \mathrm{mg}, 88 \%$ yield). ${ }^{2} \mathrm{H}$ NMR ( $60 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{H}_{6}, 25{ }^{\circ} \mathrm{C}$ ): $\delta$ $9.03(\mathrm{br}, \mathrm{Yb} D)$.
$\mathbf{L Y b}\left(\mathbf{N C}_{5} \mathbf{H}_{6}\right) \mathbf{( 5 )}$ : To a toluene solution ( 1 mL ) of $\mathbf{3}(80 \mathrm{mg}, 0.06 \mathrm{mmol})$ was added a toluene solution ( 155 mg ) of pyridine $(10 \mathrm{mg}, 0.12 \mathrm{mmol})$ at room temperature, resulting in a dark brown solution. After 10 minutes, the volatiles of the reaction solution were removed under vacuum. The residue was washed with cold hexane ( 1 mL ) and dried under vacuum to afford complex 5 as a dark brown solid ( 65 mg , $72 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta 7.02\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{H}-\mathrm{H}}=5.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{NCH}=\mathrm{CH}\right), 6.97(\mathrm{~m}, 1 \mathrm{H}$, $p-\mathrm{ArH}$ of Dipp), $6.89\left(\mathrm{~m}, 2 \mathrm{H}, m-\mathrm{Ar} H\right.$ of Dipp), $6.55\left(\mathrm{dd},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=8.4 \mathrm{~Hz},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=5.6 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $\left.\mathrm{CH}_{2} \mathrm{CHCH}\right), 5.39\left(\mathrm{~m}, 2 \mathrm{H},{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right.$ and $\left.\mathrm{NCH}=\mathrm{CH}\right)$, $4.73\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2} \mathrm{CH}\right), 4.12\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{NCH}_{2}\right.$ and $\mathrm{NCH}_{2} \mathrm{CH}$ ), $3.66\left(\mathrm{sept},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArCHMe} 2\right), 3.16\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArCHMe} 2\right.$ and $\left.\mathrm{NCH}_{2}\right), 2.57$ $\left(\mathrm{m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.03(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N} M e), 1.85\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.79-1.61\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{N} M e\right.$ and $\left.\mathrm{NCH}_{2}\right), 1.46$ ( $\mathrm{s}, 10 \mathrm{H}, \mathrm{CMe}_{3}$ and $\mathrm{NCH}_{2}$ ), $1.41\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{ArCH} M e_{2}\right), 1.33\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.5 \mathrm{~Hz}, 3 \mathrm{H}\right.$, ArCHMe $)$, 1.24 (s, 9H, CMe3), $1.18\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{ArCHMe} e_{2}\right), 1.07\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.5 \mathrm{~Hz}, 3 \mathrm{H}\right.$, ArCHMe 2 ), 1.02 (br, $3 \mathrm{H}, \mathrm{NMe}$ ). ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}$ ): $\delta 172.8,170.5$ (imine $C$ ), $148.8(\mathrm{NCH}), 147.2$ ( $i-\mathrm{ArC}$ of Dipp), 140.4, 139.4 ( $o-\mathrm{ArC}$ of Dipp), $128.5\left(\mathrm{CH}_{2} \mathrm{CHCH}\right), 124.2(p-\mathrm{ArC}$ of Dipp), 123.1, 122.5 ( $m-\operatorname{ArC}$ of Dipp), $92.9\left(\mathrm{CH}_{2} \mathrm{CH}\right), 92.57,92.52\left(\mathrm{NCHCH}\right.$ and $\left.{ }^{t} \mathrm{BuC}(\mathrm{N}) C \mathrm{H}\right), 57.8$, 56.9, $55.1\left(\mathrm{NCH}_{2}\right), 48.6\left(\mathrm{NCH}_{2} \mathrm{CH}\right), 47.7\left(\mathrm{NCH}_{2}\right), 44.4\left(\mathrm{CMe}_{3}\right), 44.2,41.3(\mathrm{NMe}), 41.1\left(\mathrm{CMe}_{3}\right), 32.3$, $31.2\left(\mathrm{CMe}_{3}\right), 27.90,27.85\left(\mathrm{ArCHMe}_{2}\right), 27.6,25.8,24.0,22.4\left(\mathrm{ArCH} \mathrm{Ce}_{2}\right)$. Anal. Calcd for $\mathrm{C}_{35} \mathrm{H}_{59} \mathrm{~N}_{5} \mathrm{Yb}$ : C, 58.15; H, 8.23; N, 9.69. Found: C, 58.43; H, 8.31; N, 9.21.
$\mathbf{L}^{\prime} \mathbf{Y b}(\mathbf{D M A P})$ (6): To a toluene solution ( 1 mL ) of $\mathbf{3}(80 \mathrm{mg}, 0.06 \mathrm{mmol})$ was added a toluene solution $(1 \mathrm{~mL})$ of 4-dimethylamino pyridine $(15 \mathrm{mg}, 0.12 \mathrm{mmol})$ at room temperature, resulting in a dark red solution. After standing at room temperature for 10 minutes, the resulting solution was concentrated to about 0.5 mL and stood at $-35^{\circ} \mathrm{C}$. Layering hexane $(2 \mathrm{~mL})$ on the toluene solution afforded complex 6 as dark red crystals ( $85 \mathrm{mg}, 89 \%$ yield). Anal. Calcd for $\mathrm{C}_{37} \mathrm{H}_{64} \mathrm{~N}_{6} \mathrm{Yb}$ : C, $58.02 ; \mathrm{H}, 8.42 ; \mathrm{N}, 10.97$. Found: C, 57.32; H, 8.56; N, 10.78. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta 4.79\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.0 \mathrm{~Hz}, 1 \mathrm{H}\right.$, $\left.{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 4.29\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right.$ and $\left.\mathrm{ArCH} \mathrm{Me}_{2}\right), 3.91\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{H}-\mathrm{H}}=6.0 \mathrm{~Hz}, 1 \mathrm{H},{ }^{t} \mathrm{BuCH}\right)$. As complex 6 is unstable in solution and the signals of $\mathbf{6}$ are overlapped with those of the decomposed complex, the assignment of other signals of 6 is unsuccessful. The ${ }^{1} \mathrm{H}$ NMR spectral monitoring of $\mathbf{6}$ in $\mathrm{C}_{6} \mathrm{D}_{6}$
indicated that $\mathbf{6}$ converted into a new complex $\mathbf{6}^{\prime}$ via a ligand redistribution reaction, accompanied by a releasing of 4-dimethylamino pyridine. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$ of $\mathbf{6}^{\prime}: \delta 7.13(\mathrm{~m}, 2 \mathrm{H}, m-$ ArH of Dipp), $6.96\left(\mathrm{~m}, 1 \mathrm{H}, p-\mathrm{Ar} H\right.$ of Dipp), $4.23\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=7.2 \mathrm{~Hz}, 1 \mathrm{H},{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{C} H\right), 3.84\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}\right.$ $\left.=7.2 \mathrm{~Hz}, 1 \mathrm{H},{ }^{t} \mathrm{BuCH}\right), 3.57\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{ArCH} \mathrm{Me}_{2}\right.$ and $\left.\mathrm{NCH}_{2}\right), 3.36(\mathrm{~m}, 1 \mathrm{H}, \mathrm{ArCHMe}), 3.11(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{NCH}_{2}\right), 2.92\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.65\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.22(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N} M e), 2.13\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.01(\mathrm{~m}$, $\left.1 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.90(\mathrm{~s}, 6 \mathrm{H}, \mathrm{NMe} 2), 1.45-1.19\left(\mathrm{~m}, 30 \mathrm{H}, \mathrm{ArCH} M e_{2}\right.$ and $\left.\mathrm{CMe} e_{3}\right)$. Attempts to grow the single crystals of 6' did not succeed, complex 6' might be a dimer or oligomer as [L'Yb] $]_{\mathrm{n}}$ ( L ' = $\left.\left[{ }^{t} \mathrm{BuC}(\mathrm{NDipp}) \mathrm{CHCH}\left({ }^{t} \mathrm{Bu}\right) \mathrm{NCH}_{2} \mathrm{CH}_{2} \mathrm{~N}(\mathrm{Me}) \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NMe}_{2}\right]^{2-}\right)$.
$\mathbf{L}^{\prime} \mathbf{Y b}\left(\mathbf{b p y} \boldsymbol{0}^{-}\right) \mathbf{( 7 )}$ : To a toluene solution ( 1 mL ) of $\mathbf{3}(50 \mathrm{mg}, 0.04 \mathrm{mmol})$ was added a toluene solution $(1 \mathrm{~mL})$ of bipyridine ( $12 \mathrm{mg}, 0.08 \mathrm{mmol}$ ) at room temperature, resulting in an orange solution. After standing for 10 minutes, the volatiles of the reaction solution were removed under vacuum. The residue was washed with cold hexane ( 1 mL ) and dried under vacuum to afford complex 7 as an orange solid ( $50 \mathrm{mg}, 80 \%$ yield). This complex is paramagnetic. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta 253.60,154.10$, $142.25,105.26,99.37,77.30,53.83,16.34,7.16,2.11,1.22,0.70,0.60,-2.81,-4.90,-5.60,-16.76,-$ $24.72,-33.43,-35.86,-38.78,-40.22,-46.77,-49.37,-79.36,-84.48,-112.08,-136.42,-165.19,-193.82$, -203.19, -241.20, -264.07. Anal. Calcd for $\mathrm{C}_{40} \mathrm{H}_{62} \mathrm{~N}_{6} \mathrm{Yb}: \mathrm{C}, 60.05 ; \mathrm{H}, 7.81 ; \mathrm{N}, 10.50$. Found: C, 59.49; H, 7.84; N, 10.55.
$\mathbf{L Y b}(\mathbf{O P P h} 2) \mathbf{( 8 )}$ : To a toluene solution ( 1 mL ) of $\mathbf{3}(80 \mathrm{mg}, 0.06 \mathrm{mmol})$ was added a toluene solution $(1 \mathrm{~mL})$ of triphenylphosphine oxide ( $35 \mathrm{mg}, 0.12 \mathrm{mmol}$ ) at room temperature, resulting in a red solution. After standing for 10 minutes, the volatiles of the reaction solution were removed under vacuum. The residue was washed with hexane $(1 \mathrm{~mL} \times 3)$ and dried under vacuum to afford complex 8 as a red solid ( $77 \mathrm{mg}, 73 \%$ yield). ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right): \delta 7.94\left(\mathrm{~m}, 4 \mathrm{H}, o-\mathrm{Ph} H\right.$ of $\left.\mathrm{OPPh}_{2}\right)$, $7.32\left(\mathrm{~m}, 4 \mathrm{H}, m-\mathrm{Ph} H\right.$ of $\left.\mathrm{OPPh}_{2}\right), 7.11\left(\mathrm{~m}, 2 \mathrm{H}, p-\mathrm{Ph} H\right.$ of $\left.\mathrm{OPPh}_{2}\right), 6.91(\mathrm{~m}, 3 \mathrm{H}, m-$ and $p-\mathrm{Ar} H$ of $\operatorname{Dipp})$, $5.39\left(\mathrm{~s}, 1 \mathrm{H},{ }^{t} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 4.17\left(\mathrm{br}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 3.57\left(\mathrm{br}, 1 \mathrm{H}, \mathrm{ArCHMe}_{2}\right), 3.19\left(\mathrm{br}, 2 \mathrm{H}, \mathrm{ArCHMe}_{2}\right.$ and $\mathrm{NCH}_{2}$ ), $2.61\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 2.04(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N} M e), 1.82(\mathrm{~m}, 3 \mathrm{H}, \mathrm{NCH}), 1.70\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.48(\mathrm{~s}$,

9H, CMes), 1.31 (m, 15H, CMe,$~ \mathrm{ArCH} M e_{2}$ and NMe ), 1.20 (m, 6H, ArCHMe ${ }_{2}$ and NMe ), 1.09 (m, $\left.6 \mathrm{H}, \mathrm{ArCH} M e_{2}\right) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (100 MHz, $\mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}$ ): $\delta 172.6,170.5$ (imine $C$ ), $155.66\left(\mathrm{~d},{ }^{1} J_{\mathrm{P}-\mathrm{C}}=\right.$ $36.5 \mathrm{~Hz}, i-\mathrm{Ph} C$ of $\mathrm{OPPh}_{2}$ ), $155.62\left(\mathrm{~d},{ }^{1} J_{\mathrm{P}-\mathrm{C}}=36.9 \mathrm{~Hz}, i-\mathrm{Ph} C\right.$ of $\left.\mathrm{OPPh}_{2}\right), 147.2(i-\mathrm{Ar} C$ of Dipp), 140.4 , 139.2 ( o-ArC of Dipp), $129.04\left(\mathrm{~d},{ }^{2} J_{\mathrm{P}-\mathrm{C}}=20.5 \mathrm{~Hz}, o-\mathrm{PhC}\right.$ of $\mathrm{OPPh}_{2}$ ), $128.99\left(\mathrm{~d},{ }^{2} J_{\mathrm{P}-\mathrm{C}}=20.8 \mathrm{~Hz}, o-\right.$ $\mathrm{Ph} C$ of $\left.\mathrm{OPPh}_{2}\right), 126.6\left(\mathrm{~d},{ }^{4} J_{\mathrm{P}-\mathrm{C}}=6.9 \mathrm{~Hz}, p-\mathrm{Ph} C\right.$ of $\left.\mathrm{OPPh}_{2}\right), 124.1,123.1(m-\mathrm{Ar} C$ of Dipp $), 122.5(p-$ ArC of Dipp), $91.9\left({ }^{\mathrm{t}} \mathrm{BuC}(\mathrm{N}) \mathrm{CH}\right), 57.9,56.8,55.2,47.9\left(\mathrm{NCH}_{2}\right), 44.57,44.53\left(\mathrm{CMe}_{3}\right.$ and NMe 2$), 41.8$ (NMe), 41.1 (CMe $)$, 32.4, 31.3 ( $\mathrm{CMe}_{3}$ ), 27.8, 27.6 ( $\mathrm{ArCHMe}_{2}$ ), 27.3, 26.0, 23.9, 22.5 ( $\mathrm{ArCHMe}_{2}$ ), the signals of $m-\mathrm{Ph} C$ of $\mathrm{OPPh}_{2}$ were overlapped with those of $\mathrm{C}_{6} \mathrm{D}_{6} .{ }^{31} \mathrm{P}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(162 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25\right.$ ${ }^{\circ} \mathrm{C}$ ): $\delta$ 84.05. Anal. Calcd for $\mathrm{C}_{42} \mathrm{H}_{63} \mathrm{~N}_{4} \mathrm{OPYb}: \mathrm{C}, 59.77 ; \mathrm{H}, 7.52$; $\mathrm{N}, 6.64$. Found: C, 59.38; H, 7.27; N , 6.36 .
$[\mathbf{L Y b}(\boldsymbol{\mu}-\mathbf{S P h})]_{2} \mathbf{( 9 )}$ : To a toluene solution $(1 \mathrm{~mL})$ of $\mathbf{3}(80 \mathrm{mg}, 0.06 \mathrm{mmol})$ was added a toluene solution $(1 \mathrm{~mL})$ of diphenyl disulfide $(13.5 \mathrm{mg}, 0.06 \mathrm{mmol})$ at room temperature, resulting in a dark brown solution. After standing at room temperature for 10 minutes, the resulting solution was concentrated to about 0.5 mL and stood at $-35^{\circ} \mathrm{C}$. Layering hexane $(2 \mathrm{~mL})$ on the toluene solution afforded complex 9 as brown crystals ( $60 \mathrm{mg}, 64 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}$ ) $\delta 7.83\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{H}-\mathrm{H}}=7.6 \mathrm{~Hz}\right.$, $2 \mathrm{H}, o-\mathrm{Ph} H$ of SPh $), 7.14\left(\mathrm{t},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=7.6 \mathrm{~Hz}, 2 \mathrm{H}, m-\mathrm{Ph} H\right.$ of SPh$), 6.92(\mathrm{~m}, 4 \mathrm{H}, p-\mathrm{Ph} H$ of SPh , and $m-$ and $p-\mathrm{Ar} H$ of Dipp), $5.37\left(\mathrm{~s}, 1 \mathrm{H},{ }^{\mathrm{t}} \mathrm{BuC}(\mathrm{N}) \mathrm{C} H\right), 4.13\left(\mathrm{br}, 1 \mathrm{H}, \mathrm{NCH}_{2}\right), 3.41\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{ArCH} \mathrm{Me}_{2}\right), 3.15(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{NCH}_{2}$ and ArCHMe 2 ), 2.67 (br, $1 \mathrm{H}, \mathrm{NCH}_{2}$ ), 2.17 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{NMe}$ ), 2.08-1.87 (m, 3H, $\mathrm{NCH}_{2}$ ), 1.66 $\left(\mathrm{m}, 2 \mathrm{H}, \mathrm{NCH}_{2}\right), 1.50(\mathrm{~m}, 6 \mathrm{H}, \mathrm{NMe} 2), 1.43\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CMe} e_{3}\right), 1.24\left(\mathrm{~m}, 15 \mathrm{H}, \mathrm{CMe} e_{3}\right.$ and ArCHMe 2$), 1.18(\mathrm{~d}$, $\left.\left.{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \operatorname{ArCH} M e_{2}\right), 1.07\left(\mathrm{~d},{ }^{3} J_{\mathrm{H}-\mathrm{H}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \operatorname{ArCHMe}\right)^{2}\right) .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , $\mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}$ ): $\delta 172.0,171.5$ (imine $C$ ), 150.2 ( $i-\mathrm{Ph} C$ of SPh), 147.4 ( $i-\mathrm{ArC}$ of Dipp), 140.5, 139.3 (oArC of Dipp), 134.2 ( $o-\mathrm{PhC}$ of SPh), 128.1 ( $m-\mathrm{PhC}$ of SPh), 124.0 ( $p-\mathrm{Ar} C$ of Dipp), 123.1, 122.9 ( $m-$ ArC of Dipp), 120.6 ( $p-\mathrm{Ph} H$ of SPh ), 91.4 ( $\left.{ }^{( } \mathrm{BuC}(\mathrm{N}) C \mathrm{H}\right), 57.7,57.2,55.4,47.7\left(\mathrm{NCH}_{2}\right), 44.6(\mathrm{NMe})$, $44.4\left(\mathrm{CMe}_{3}\right), 42.3$ (NMe), $40.9\left(\mathrm{CMe}_{3}\right), 32.4,31.0\left(\mathrm{CMe}_{3}\right), 28.2,27.7$ ( $\mathrm{ArCHMe}_{2}$ ), 27.6, 26.1, 23.8, 22.7 (ArCHMe2). Anal. Calcd for $\mathrm{C}_{72} \mathrm{H}_{116} \mathrm{~N}_{8} \mathrm{~S}_{2} \mathrm{Yb}_{2}$ : C, 57.50; H, 7.77; N, 7.45. Found: C, 57.10; H, 7.83; N, 7.37.
$\mathbf{L Y b}(\mathbf{S P h})_{2} \mathbf{( 1 0 )}$ : To a toluene solution $(1 \mathrm{~mL})$ of $\mathbf{3}(71 \mathrm{mg}, 0.055 \mathrm{mmol})$ was added a toluene solution $(1 \mathrm{~mL})$ of diphenyl disulfide ( $24 \mathrm{mg}, 0.11 \mathrm{mmol}$ ) at room temperature, resulting in an orange solution. After standing for 10 minutes, the volatiles of the reaction solution were removed under vacuum. The residue was washed with hexane $(1 \mathrm{~mL} \times 3)$ and dried under vacuum to afford complex $\mathbf{1 0}$ as a yellow solid ( $72 \mathrm{mg}, 76 \%$ yield). This complex is paramagnetic. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}$ ) $\delta 96.53$, $38.88,23.38,18.51,17.00,15.16,11.04,7.35,5.93,3.68,1.21,-0.02,-6.23,-11.11,-14.27,-24.84,-$ 81.06, -86.90, -90.17. Anal. Calcd for $\mathrm{C}_{42} \mathrm{H}_{63} \mathrm{~N}_{4} \mathrm{~S}_{2} \mathrm{Yb}$ : C, $58.58 ; \mathrm{H}, 7.37$; N, 6.51. Found: C, 59.19; H, 7.50; $\mathrm{N}, 6.44$. In addition, a ${ }^{1} \mathrm{H}$ NMR investigation in $\mathrm{C}_{6} \mathrm{D}_{6}$ also showed that complex 9 reacts with 1.0 equiv. of PhSSPh to give complex 10 (see Figure S28).

## X-ray Crystallography:

Single crystals of 1 suitable for single-crystal X-ray diffraction were grown from a THF/hexane mixture, those of $\mathbf{2}$ were from a hexane solution, those of $\mathbf{3}$ were from a toluene solution, those of $\mathbf{5}$, $\mathbf{6 , 8}$ and $\mathbf{9}$ were from a toluene/hexane mixture, those of $\mathbf{7}$ were from a mesitylene solution, and those of $\mathbf{1 0}$ were from a toluene/cyclohexane mixture. The crystals were mounted under a nitrogen atmosphere on a glass fiber at low temperature. Data collection of $\mathbf{1 , 3}, 5,6,7,9$ and $\mathbf{1 0}$ was performed on a Bruker D8 Venture with $\mathrm{Ga} \mathrm{K} \alpha$ radiation $(\lambda=1.34139 \AA)$, that of $\mathbf{2}$ and $\mathbf{8}$ was performed on a Bruker APEX-II CCD with graphite-monochromated Mo $\mathrm{K} \alpha$ radiation ( $\lambda=0.71073 \AA$ ). The SMART program package was used to determine the unit cell parameters. The absorption correction was applied using SADABS program. ${ }^{5}$ All structures were solved by direct methods and refined on $F^{2}$ by fullmatrix least-squares techniques with anisotropic thermal parameters for non-hydrogen atoms. Hydrogen atoms were placed at calculated positions and were included in the structure calculation, except for the hydrogen atoms of $\mathrm{Yb}-\mathrm{H}$, which were located in the Fourier different map. Calculations were carried out using the SHELXL-97, SHELXL-2014 or Olex2 program. ${ }^{6}$ Crystallographic data and refinement parameters are listed in Table S1.

Table S1. Crystallographic data and refinement parameters

|  | 1 | 2 | 0.5(3) |
| :---: | :---: | :---: | :---: |
| formula | $\mathrm{C}_{60} \mathrm{H}_{106} \mathrm{I}_{2} \mathrm{~N}_{8} \mathrm{Yb}_{2}$ | $\mathrm{C}_{34} \mathrm{H}_{64} \mathrm{~N}_{4} \mathrm{SiYb}$ | $\mathrm{C}_{30} \mathrm{H}_{54} \mathrm{~N}_{4} \mathrm{Yb}$ |
| fw | 1539.40 | 730.02 | 643.81 |
| color | red | red | red |
| crystal system | monoclinic | monoclinic | monoclinic |
| space group | $P 2{ }_{1} / c$ | $P 2{ }_{1} / n$ | $P 2{ }_{1} / n$ |
| $a, \AA$ | 10.240(1) | 11.376(1) | 11.917(1) |
| $b, \AA$ | 28.616(1) | 23.594(1) | 20.310(1) |
| $c, \AA$ | 11.527(1) | 14.219(1) | 13.183(1) |
| $\alpha, \operatorname{deg}$ | 90 | 90 | 90 |
| $\beta$, deg | 109.493(3) | 102.100(1) | 92.514(2) |
| $\gamma, \operatorname{deg}$ | 90 | 90 | 90 |
| $V, \AA^{3}$ | 3184.1(3) | 3731.6(1) | 3187.9(2) |
| Z | 2 | 4 | 4 |
| $D_{\text {calcd }}, \mathrm{mg} / \mathrm{m}^{3}$ | 1.606 | 1.299 | 1.341 |
| absorption coefficient, $\mathrm{mm}^{-1}$ | 15.071 | 8.616 | 9.814 |
| $F(000)$ | 1536 | 1520 | 1328 |
| $T(\mathrm{~K})$ | 170(2) | 170(2) | 173(2) |
| $\theta$ range, deg | 3.786, 54.995 | 3.210, 54.926 | 3.787, 54.933 |
| no. of reflns collected | 21040 | 33590 | 32617 |
| no. of unique reflns | 6035 | 7067 | 6042 |
| no. of obsd reflns ( $I>2 \sigma(I)$ ) | 4051 | 6417 | 5156 |
| no. of params | 338 | 381 | 333 |
| final $R, w R(I>2 \sigma(I))$ | 0.067, 0.168 | 0.024, 0.056 | 0.034, 0.085 |
| goodness of fit on $F^{2}$ | 1.108 | 1.049 | 1.026 |
| $\Delta \rho_{\text {max,min }}, \mathrm{e} \AA^{-3}$ | 2.002, -1.594 | 0.436, -1.037 | 1.258, -1.129 |


|  | 5 | 6 | 7•mesitylene |
| :---: | :---: | :---: | :---: |
| formula | $\mathrm{C}_{35} \mathrm{H}_{59} \mathrm{~N}_{5} \mathrm{Yb}$ | $\mathrm{C}_{37} \mathrm{H}_{64} \mathrm{~N}_{6} \mathrm{Yb}$ | $\mathrm{C}_{49} \mathrm{H}_{74} \mathrm{~N}_{6} \mathrm{Yb}$ |
| fw | 722.91 | 765.98 | 920.18 |
| color | red | red | orange |
| crystal system | triclinic | monoclinic | monoclinic |
| space group | $P \overline{1}$ | C2/c | $P 2{ }_{1} / \mathrm{c}$ |
| $a, \AA$ | 9.777(1) | 31.728(1) | 21.092(1) |
| $b, \AA$ | 10.429(1) | 12.136(1) | 13.185(1) |
| $c, \AA$ | 19.793(1) | 21.846(1) | 16.504(1) |
| $\alpha$, deg | 78.372(2) | 90 | 90 |
| $\beta$, deg | 77.701(2) | 101.593(2) | 94.546(2) |
| $\gamma, \operatorname{deg}$ | 65.313(2) | 90 | 90 |
| $V, \AA^{3}$ | 1777.2(1) | 8239.9(4) | 4575.6(3) |
| Z | 2 | 8 | 4 |
| $D_{\text {calce }}, \mathrm{mg} / \mathrm{m}^{3}$ | 1.351 | 1.235 | 1.336 |
| absorption coefficient, $\mathrm{mm}^{-1}$ | 8.848 | 8.307 | 6.954 |
| $F(000)$ | 748 | 3184 | 1920 |
| $T(\mathrm{~K})$ | 170(2) | 173(2) | 173(2) |
| $\theta$ range, deg | 4.092, 55.173 | 3.594, 54.884 | 3.738, 54.944 |
| no. of reflns collected | 20165 | 38830 | 48464 |
| no. of unique reflns | 6737 | 7815 | 8668 |
| no. of obsd reflns ( $I>2 \sigma(I)$ ) | 6239 | 6589 | 7415 |
| no. of params | 383 | 412 | 536 |
| final $R, w R(I>2 \sigma(I))$ | 0.038, 0.099 | 0.034, 0.076 | 0.028, 0.065 |
| goodness of fit on $F^{2}$ | 1.042 | 1.053 | 1.028 |
| $\Delta \rho_{\text {max,min }}, \mathrm{e} \AA^{-3}$ | 1.247, -1.791 | 0.658, -0.902 | 0.431, -0.833 |


|  | 8 | 9 | 10 |
| :---: | :---: | :---: | :---: |
| formula | $\mathrm{C}_{42} \mathrm{H}_{63} \mathrm{~N}_{4} \mathrm{OPYb}$ | $\mathrm{C}_{72} \mathrm{H}_{116} \mathrm{~N}_{8} \mathrm{~S}_{2} \mathrm{Yb}_{2}$ | $\mathrm{C}_{42} \mathrm{H}_{63} \mathrm{~N}_{4} \mathrm{~S}_{2} \mathrm{Yb}$ |
| fw | 843.97 | 1503.92 | 861.12 |
| color | red | brown | yellow |
| crystal system | monoclinic | monoclinic | monoclinic |
| space group | $P 2{ }_{1} / n$ | $P 2{ }_{1} / c$ | $P 2{ }_{1} / c$ |
| $a, ~ \AA \begin{aligned} & \text { a } \\ & \\ & \text { a }\end{aligned}$ | 19.463(1) | 15.231(1) | 19.433(1) |
| $b, \AA$ | 11.075(1) | 34.584(1) | 13.508(1) |
| $c, \AA$ | 20.645(1) | 21.662(1) | 17.331(1) |
| $\alpha$, deg | 90 | 90 | 90 |
| $\beta$, deg | 113.247(3) | 95.572(2) | 100.930(2) |
| $\gamma, \operatorname{deg}$ | 90 | 90 | 90 |
| $V, \AA^{3}$ | 4088.6(4) | 11356.5(5) | 4466.6(3) |
| Z | 4 | 6 | 4 |
| $D_{\text {calcd, }}\left(\mathrm{mg} / \mathrm{m}^{3}\right)$ | 1.371 | 1.319 | 1.281 |
| absorption coefficient, $\mathrm{mm}^{-1}$ | 7.935 | 8.644 | 7.648 |
| $F(000)$ | 1744 | 4656 | 1780 |
| $T(\mathrm{~K})$ | 173(2) | 185(2) | 192(1) |
| $\theta$ range, deg | 4.021, 55.010 | 2.769, 55.001 | 3.488, 54.987 |
| no. of refns collected | 34631 | 124671 | 35229 |
| no. of unique refns | 7654 | 21642 | 8460 |
| no. of obsd refns ( $I>2 \sigma(I)$ ) | 6347 | 14445 | 6217 |
| no. of params | 455 | 1174 | 455 |
| final $R, w R(\mathrm{I}>2 \sigma(\mathrm{I})$ ) | 0.066, 0.180 | 0.053, 0.129 | 0.067, 0.174 |
| goodness of fit on $F^{2}$ | 1.109 | 1.043 | 1.043 |
| $\Delta \rho_{\text {max }, \text { min }}, \mathrm{e} \AA^{-3}$ | 1.806, -2.377 | 2.276, -1.469 | 1.807, -2.742 |



Figure S1. ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{L K}\left(400 \mathrm{MHz}, d_{8}\right.$-THF, $\left.25^{\circ} \mathrm{C}\right)$.


Figure S2. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of $\mathbf{L K}\left(100 \mathrm{MHz}, d_{8}-\mathrm{THF}, 25{ }^{\circ} \mathrm{C}\right)$.


Figure S3. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $\mathbf{1}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S4. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of complex $\mathbf{1}\left(100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S5. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $2\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S6. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of complex $2\left(100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S7. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $\mathbf{3}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$. A small amount of complex $\mathbf{3}$ have decomposed into complex 4 .


Figure S8. ${ }^{1} \mathrm{H}$ NMR spectrum of complex 3-D $\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


[^0]Figure S9. ${ }^{2} \mathrm{H}$ NMR spectrum of complex 3-D $\left(60 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{H}_{6}, 25{ }^{\circ} \mathrm{C}\right) . \mathrm{C}_{6} \mathrm{D}_{6}$ as the reference.


Figure S10. ${ }^{1} \mathrm{H}$ NMR spectral monitoring on transformation of $\mathbf{3}$ to $\mathbf{4}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}\right)$.


Figure S11. ${ }^{1} \mathrm{H}$ NMR spectral monitoring on transformation of 3-D to 4-D $\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}\right)$.


Figure S12. ${ }^{2} \mathrm{H}$ NMR spectral monitoring on transformation of 3-D to 4-D $\left(60 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{H}_{6}, 25^{\circ} \mathrm{C}\right) . \mathrm{C}_{6} \mathrm{D}_{6}$ as the reference.


Figure S13. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectra of $\mathbf{4}$ and 4-D, which indicate the presence of ${ }^{t} \mathrm{BuCH}$ in $\mathbf{4}$ and ${ }^{t} \mathrm{BuCD}$ in 4-D $\left(100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S14. gCOSY spectra of the mixture of $\mathbf{3}$ and $\mathbf{4}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S15. gHSQC spectra of $4\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S16. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $5\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S17. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of complex $5\left(100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S18. ${ }^{1} \mathrm{H}$ NMR spectrum showing of the transformation of complex 6 into DMAP and $\mathbf{6}^{\prime}\left(600 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right)$

## 



Figure S19. ${ }^{1} \mathrm{H}$ NMR spectrum of complex 6' $\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S20. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $7\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S21. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $\mathbf{8}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}\right)$.


Figure S22. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of complex $\mathbf{8}\left(100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S23. ${ }^{31} \mathrm{P}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of complex $\mathbf{8}\left(162 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25{ }^{\circ} \mathrm{C}\right)$.


```
9.0
```

Figure S24. ${ }^{1} \mathrm{H}$ NMR spectrum of the reaction of complex $\mathbf{3}$ with $\mathrm{Ph}_{3} \mathrm{P}=\mathrm{O}$ in $d_{8}$-toluene $\left(400 \mathrm{MHz}, 25^{\circ} \mathrm{C}\right)$, showing the formation of $\mathrm{C}_{6} \mathrm{H}_{6}$.


Figure S25. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $9\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S26. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR spectrum of complex $9\left(100 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$.


Figure S27. ${ }^{1} \mathrm{H}$ NMR spectrum of complex $\mathbf{1 0}\left(400 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}, 25^{\circ} \mathrm{C}\right)$


Figure S28. ${ }^{1} \mathrm{H}$ NMR spectrum of the reaction of complex 9 with 1.0 euiv. PhSSPh in $\mathrm{C}_{6} \mathrm{D}_{6}$ in $20 \mathrm{~min}\left(400 \mathrm{MHz}, 25^{\circ} \mathrm{C}\right)$, showing the formation of complex 10.

## References

(1) Watson, P. L.; Tulip, T. H.; Williams, I. Defluorination of Perfluoroolefins by Divalent Lanthanoid Reagents: Activating Carbon-Fluorine Bonds. Organometallics 1990, 9, 1999-2009.
(2) Hitchcock, P. B.; Khvostov, A. V.; Lappert, M. F. Synthesis and Structures of Crystalline Bis(trimethylsilyl)methanido Complexes of Potassium, Calcium and Ytterbium. J. Organomet. Chem. 2002, 663, 263-268.
(3) Zhou, W.; Marquard, S. L.; Bezpalko, M. W.; Foxman, B. M.; Thomas, C. M. Catalytic Hydrosilylation of Ketones Using a $\mathrm{Co} / \mathrm{Zr}$ Heterobimetallic Complex: Evidence for an Unusual Mechanism Involving Ketyl Radicals. Organometallics 2013, 32, 1766-1772.
(4) Kurogi, T.; Chu, J. X.; Chen, Y. F.; Mindiola, D. J. Neutral and Anionic Monomeric Zirconium Imides Prepared via Selective $\mathrm{C}=\mathrm{N}$ Bond Cleavage of a Multidentate and Sterically Demanding $\beta$ Diketiminato Ligand. Chem. Asian J. 2019, 14, 2629-2638.
(5) Sheldrick, G. M. SADABS: An Empirical Absorption Correction Program for Area Detector Data; University of Göttingen: Göttingen, Germany, 1996.
(6) (a) Sheldrick, G. M.; SHELXS-97 and SHELXL-97; University of Göttingen: Göttingen, Germany, 1997 and 2008. (b) Sheldrick, G. M. SHELXS-2014; University of Göttingen: Göttingen, Germany, 2014. (c) Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. OLEX2: A Complete Structure Solution, Refinement and Analysis Program. J. Appl. Cryst. 2009, 42, 339-341. (d) SMART, Version 5.628; Bruker AXS Inc.: Madison, WI, 2002. (e) SAINT+, Vdrsion 6.22a; Bruker AXS Inc.: Madison, WI, 2002. (f) SAINT+, Vdrsion v7.68A; Bruker AXS Inc.: Madison, WI, 2009. (g) SHELXTL NT/2000, Version 6.1; Bruker AXS Inc.: Madison, WI, 2002.


[^0]:    $\begin{array}{llllllllllllllllllllllllllllllllllll}14 & 13 & 12 & 11 & 10 & 9 & 8 & 7 & 6 & 5 & 4 & 3 & 2 & 1 & 0 & \mathbf{- 1} & \mathbf{- 2} & \mathbf{- 3} & \mathbf{- 4} & \mathbf{- 5} & \mathbf{- 6} & -7 & \mathbf{- 8} & \mathbf{- 9} & \mathbf{- 1 0} & \mathbf{- 1 1} & \mathbf{- 1 2} & \mathbf{- 1 3} & \mathbf{- 1 4} & \mathbf{- 1}\end{array}$ chemical shift(ppm)

