

Supporting Information for

Mixed Solvothermal Synthesis of T_n Cluster-Based Indium and Gallium Sulfides Using Versatile Ammonia or Amine Structure-Directing Agents

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1. Tables

Table S1. Bond valence sum (BVS) values for the In and S atoms in **1**. The BVS calculations were performed on the KDist software (Version 3.75) distributed by K. Knížek.

Atom	BVS value	Atom	BVS value
In1	3.23	S4	-0.79
In2	3.22	S5	-1.66
In3	3.20	S6	-1.62
In4	3.25	S7	-1.62
S1	-0.76	S8	-1.61
S2	-0.77	S9	-1.65
S3	-0.77	S10	-1.66

Table S2. Summary of alkaline metal/ammonia/ammine templated Ga-Q compounds.

Compounds	SG	D	Reactants	Solvent	Yield	Ref.
K ₈ Ga ₄ S ₁₀	<i>Pnma</i>	0D	Ga ₂ S ₃ ; 160 °C, 6 d	K ₂ S	NA	[1]
[C ₁₀ N ₄ H ₂₆] _{0.5} [GaS ₂]	<i>P2₁/n</i>	1D	Ga ₂ S ₃ + S; 160 °C, 6 d	BAPP	NA	[2]
[C ₆ H ₈ N] ₂ [Ga ₁₀ S ₁₆ (NC ₆ H ₇) ₂ (N ₂ C ₁₂ H ₁₂)]	<i>Pcc</i> a	1D	Ga + S + DPE; 170 °C, 5 d	4-methylpyridine	NA	[3]
[C ₂ H ₈ N] ₂ [Ga ₁₀ S ₁₆ (N ₂ C ₁₂ H ₁₂)(NC ₂ H ₇) ₂]	<i>P3₂1</i>	1D	Ga + S + DPE + TPP; 200 °C, 20 d	Py	NA	[4]
[Mn(atep)][Ga ₂ S ₄]	<i>P2₁/n</i>	1D	Ga ₂ S ₃ + MnS + S; 180 °C, 21 d	tепа	15%	[5]
[Mn(en) ₃] _{0.5} [GaS ₂]	<i>Cmcm</i>	1D	Ga ₂ S ₃ + MnCl ₂ 4H ₂ O + S; 170 °C, 12 d	en	80%	[6]
[Co(en) ₃] _{0.5} [GaS ₂]	<i>Cmcm</i>	1D	Ga ₂ S ₃ + CoCl ₂ 6H ₂ O + S; 160 °C, 20 d	en	NA	
[Ni(en) ₃] _{0.5} [GaS ₂]	<i>Cmcm</i>	1D	Ga ₂ S ₃ + NiS + S; 180 °C, 6 d	en	NA	
[enH ₂][Ga ₄ S ₇ (en) ₂]	<i>P2₁/c</i>	2D	Ga ₂ S ₃ + S; 170 °C, 5 d	en	80%	
CsGaS ₂	<i>C2/c</i>	2D	Ga ₂ S ₃ + Se + CsN ₃ ; 500°C/700°C	NA	NA	[7]
NaGaS ₂	<i>C2/c</i>	2D	Ga + S + Na ₂ S; 650 °C, 5 h	NA	NA	[8]
[C ₇ H ₁₀ N] ₂ [Ga ₁₀ S ₁₆ (NC ₇ H ₉)(N ₂ C ₁₂ H ₁₀) _{3/2}]	<i>P\bar{1}</i>	2D	Ga + S + DPE; 200 °C, 20 d	3,5-methylpyridine	NA	[3]
[C ₅ H ₆ N] ₃ [Ga ₁₀ S ₁₆ (OH)(N ₂ C ₁₃ H ₁₄)]	<i>Pbcm</i>	2D	Ga + S + TMP + TPP; 200 °C, 20 d	Py	NA	[4]
UCR-18GaS-AEP	<i>C2/c</i>	3D	Ga + S; 190 °C, 6 d	AEP	86%	[9]
UCR-7GaS-TETA	<i>I4₁/acd</i>	3D	NA	NA	NA	
UCR-7GaS-TAEA	<i>I4₁/acd</i>	3D	NA	NA	NA	
UCR-7GaS-DBA	<i>P4₁2₁2</i>	3D	NA	NA	NA	
[C ₄ NH ₁₂] ₆ [Ga ₁₀ S ₁₈]	<i>P4₃2₁2</i>	3D	Ga + S; 170 °C, 8 d	DEA	NA	[10]
[C ₄ NH ₁₂] ₁₂ [Ga ₂₀ S _{35.5} (S ₃) _{0.5} O]	<i>C2/c</i>	3D	Ga + S; 190 °C, 10 d	DEA	NA	
[Dy ₂ (en) ₆ (μ ₂ -OH) ₂]Ga ₄ S ₈	<i>P\bar{1}</i>	1D	Ga ₂ S ₃ + DyCl ₃ + S; 170 °C, 10 d	en	NA	[11]
[Mn(dap) ₃] _{0.5} [GaSe ₂]	<i>Cmcm</i>	1D	Ga + Mn + Se; 180 °C, 16 d	dap + H ₂ O	13%	[5]
(C ₈ H ₁₁ N ₂)GaSe ₂	<i>P2₁/c</i>	1D	Ga + Se; 200 °C, 8 d	piperazine + H ₂ O	92%	[12]
[1,3-pdaH ₂][Ga ₂ Se ₂ (Se ₂)Se ₃]	<i>P2₁</i>	1D	Ga + Se; 160 °C, 6 d	1,3-pda + CH ₃ OH	22.4%	[13]
[1,4-bdaH ₂][Ga ₂ Se ₃ (Se ₂)]	<i>C2/c</i>	1D	Ga + Se; 160 °C, 6 d	1,4-bda + CH ₃ OH	9.9%	
[Me ₂ NH ₂] ₂ [Ga ₂ Se ₂ (Se ₂) ₂]	<i>P2₁/c</i>	1D	Ga ₂ O ₃ + Se; 150 °C, 5 d	DMF + N ₂ H ₄ ·H ₂ O	18.4%	
α-[AEPH] ₂ [Ga ₂ Se ₂ (Se ₂) ₂]	<i>Pn</i>	1D	Ga ₂ O ₃ + Se; 120 °C, 5 d	AEP	NA	
β-[AEPH] ₂ [Ga ₂ Se ₂ (Se ₂) ₂]	<i>P2₁/c</i>	1D	Ga ₂ O ₃ + Se; 170 °C, 5 d	AEP	11.9%	
[C ₆ H ₁₀ N ₂][Ga ₂ Se ₃ (Se ₂)]	<i>C2/c</i>	1D	Ga + Se; 170 °C, 10 d	trans-1,4-diaminocyclohexane + H ₂ O	NA	[14]
[bappH ₂][Ga ₂ Se ₄]	<i>P\bar{1}</i>	1D	Ga + Se + bapp; 170 °C, 6 d	H ₂ O	46%	[15]

[Mn(en) ₃][Ga ₂ Se ₅]	<i>Pbcn</i>	1D	Ga + Se + Mn(CH ₃ COO) ₂ 4H ₂ O + en; 170 °C, 6 d	H ₂ O	53%	
[H ₂ dap] ₂ Ga ₄ Se ₈	<i>C2/c</i>	2D	Ga + Se; 170 °C, 5 d	dap + H ₂ O	43%	[5]
(Ph ₄ P)[Ga(Se ₆) ₂]	<i>P4</i>	2D	Ga + (Ph ₄ P) ₂ Se ₅ + Se; 200 °C, 2 d	NA	60%	[15]
RbGaSe ₂	<i>C2/c</i>	2D	GaSe + Se + RbN ₃ ; 850°C	NA	NA	[16]
Ga ₂ Se ₇ (en) ₂ ·(enH) ₂	<i>P2₁/c</i>	2D	Ga + Se; 140 °C, 5-7 d	en	90%	[17]
(C ₁₃ H ₂₆ N ₂ H ₂) ₃ (Ga ₁₀ Se ₁₈)	<i>I42d</i>	3D	Ga ₂ O ₃ + Se; 200 °C, 28 d	TMDP + H ₂ O	77%	[9]
(C ₁₁ H ₂₂ N ₂ H ₂) ₃ (Ga ₁₀ Se ₁₈)	<i>I4m2</i>	3D	Ga ₂ O ₃ + Se; 200 °C, 19 d	DPM + H ₂ O	84%	

Abbreviations: en = ethylenediamine; BAPP = 1,4-bis(3-aminopropyl)piperazine; DPE = 1,2-di(4-pyridyl)ethylene; TPP = tetraphenylphosphonium bromide; Py = pyridine; TMP = 4,4'-Trimethylenedipyridine; TAEA = Tris(2-aminoethyl)amine; TETA = triethylenetetramine; DBA = di-n-butylamine; AEP = 1-(2-aminethyl)piperazine; DEA = diethylamine; dap = diaminopropane; 1,3-pda = 1,3-diaminopropane; 1,4-pda = 1,4-diaminopropane; TMDP = 4,4-trimethylenedipiperidine; DPM = dipiperidinomethane.

Table S3. Summary of ammonia/ammine templated In-S compounds.

Compounds	SG	D	Reactants	Solvent	Yield	Ref.
(Ph ₄ P) ₄ In ₂ S ₂₇	<i>P1</i>	0D	InCl ₃ + K ₂ S ₅ ; 110 °C, 10 d	Ph ₄ PCl + CH ₃ OH	58%	[18]
(Ph ₄ P) ₂ [In ₂ S ₁₄] _{0.5} [In ₂ S ₁₆] _{0.5}	<i>P1</i>	0D	InCl ₃ + K ₂ S ₅ ; RT, 2d	Ph ₄ PCl + DMF	48%	
[In(en) ₂ S] ₂ ·2Cl	<i>P2₁/n</i>	0D	In + S; 190 °C, 5 d	en + CH ₂ Cl ₂	40%	[19]
[Mg(en) ₃][In ₂ S ₄]	<i>Cmcm</i>	1D	In + S + Mg; 190 °C, 10 d	en	80%	
[C ₁₀ N ₄ H ₂₆] _{0.5} [InS ₂]	<i>P2₁/n</i>	1D	In + S; 145 °C, 5 d	BAPP	NA	[2]
[Ni(dien) ₂] _{0.5} [InS ₂]	<i>C2/c</i>	1D	In ₂ S ₃ + Ni + S; 157 °C, 6 d	dien + H ₂ O	31%	[20]
[Ni(dap) ₃] _{0.5} [InS ₂]	<i>Cmcm</i>	1D	In ₂ S ₃ + Ni + S; 157 °C, 6 d	dap + H ₂ O	47%	
[Ni(tepa)] ₂ [In ₄ S ₇ (SH) ₂]·H ₂ O	<i>P2₁/c</i>	1D	In ₂ S ₃ + Ni + S 157 °C, 6 d	tepa + H ₂ O	25%	
{[Ni(dach) ₃][In ₂ S ₄]3H ₂ O} _n	<i>C2/c</i>	1D	In + Ni + L-cysteine + (BMIm)Br; 160, °C, 9 d	dach + H ₂ O	35%	[21]
[Fe(phen) ₃] ₄ [H ₄ In ₂₀ S ₃₈]Hphen 3·HDMA·8H ₂ O	<i>P1</i>	1D	In + Fe + Na ₂ S + phen; 140, °C, 18 d	DMF + H ₂ O	30%	[22]
[Ni(phen) ₃] ₄ [H ₄ In ₂₀ S ₃₈]·2Hphen·2HDMA·3H ₂ O	<i>C2/c</i>	1D	In + Ni + Na ₂ S + phen; 140, °C, 32 d	DMF + H ₂ O	21%	
[Ni(phen) ₃] ₄ [In ₂₀ S ₃₇]·6Hphen·4H ₂ O	<i>C2/c</i>	2D	In + Ni + Na ₂ S + phen; 140, °C, 24 d	DMF + H ₂ O	32%	
[Fe(phen) ₃] ₄ [In ₂₀ S ₃₇]·6Hphen·4H ₂ O	<i>C2/c</i>	2D	In + Fe + Na ₂ S + phen; 140, °C, 24 d	DMF + H ₂ O	41%	
[In ₈ S ₁₃ (S ₃) _{1/2} (SH)][In ₄ S ₆ (S ₃) _{1/2} (SH)] (TMDPH ₂) ₅ (HCF-1)	<i>P3₁c</i>	2D	In + S; 150 °C, 5 d	TETA + TMDP + H ₂ O	65%	[23]
[(C ₃ H ₇) ₂ NH ₂] ₃ In ₆ S ₁₁ H	<i>P2₁/c</i>	2D	In + S; 180 °C, 5 d	DPA	NA	[24]
[CH ₃ CH ₂ NH ₃] ₆ In ₆ S ₁₂	<i>P2₁/n</i>	2D	InCl ₃ + TTCA; 160 °C, 4 d	EA + EtOH	46.1%	[25]
[CH ₃ CH ₂ NH ₃] ₆ In ₈ S ₁₅	<i>P2₁/n</i>	2D	In + S; 140 °C, 4 d	EA + EtOH	54%	[26]
[(CH ₃ CH ₂) ₂ NH ₂] ₆ In ₁₀ S ₁₈	<i>I4₁/amd</i>	2D	In + S; 180 °C, 7 d	DEA + H ₂ O	NA	[27]
[C ₆ H ₁₆ N] ₄ In ₄ S ₁₀ H ₄	<i>C2/c</i>	0D	In + S; 180 °C, 5 d	DPA	NA	
[C ₁₃ H ₁₄ N ₂] ₄ In ₉ S ₁₇	<i>P2₁/c</i>	2D	In + S; 180 °C, 5 d	TMDP	NA	
[(CH ₃ CH ₂) ₂ NH ₂] ₆ In ₁₀ S ₁₈	<i>P4₃2₁2</i>	3D	In + S; 180 °C, 7 d	DEA	NA	
UCR-7InS-AEP ([In ₁₀ S ₁₈] ⁶⁻)	<i>P4₁/acd</i>	3D	NA	NA	NA	[9]
[(CH ₃) ₂ NH ₂] ₆ In ₁₀ S ₁₈	<i>Pbca</i>	3D	In + S; 180 °C, 7 d	DMA	NA	[28]
In ₁₀ S ₁₈ (HPP) ₆ (H ₂ O) ₁₅	<i>I43m</i>	3D	In + S; 135 °C	HPP	10%	[29]
In ₁₀ S ₁₈ (DPM) ₃ (H ₂ O) ₇	<i>I4m2</i>	3D	In + S; 135 °C	DPM	50%	
In ₁₀ S ₁₈ (C ₆ H ₁₂ NH ₂) ₆ (C ₆ H ₁₂ NH)(H ₂ O) ₅	<i>I42d</i>	3D	In + S; 150 °C, 4 d	C ₆ H ₁₂ NH + H ₂ O	43%	[30]
[In ₃₃ S ₅₆](C ₁₃ H ₂₆ N ₂ H ₂) _{6,5}	<i>R32</i>	3D	NA	NA	NA	[31]
(In ₃₄ S ₅₄)(In ₁₀ S ₁₈)(C ₁₁ H ₂₄ N ₂) ₆	<i>I4₁/a</i>	3D	In + S + CdCl ₂ ·2.5H ₂ O; 190 °C, 6 d	DPM	50%	[32]
[In ₁₀ S ₁₈](H ₂ O) ₄ (C ₆ H ₁₈ N ₄ H ₂) _{1,5} (C ₆ H ₁₈ N ₄ H ₃)	<i>P4₁2₁2</i>	3D	In + S; 170 °C, 5 d	TETA + Py + H ₂ O	91%	[33]
(H ₂ DAH) ₃ In ₁₀ S ₁₈ ·6H ₂ O	<i>I4m2</i>	3D	In + S; 170 °C, 4 d	DAH + H ₂ O	41%	[34]

(H ₂ DAH) ₃ In ₁₀ S ₁₈	<i>I</i> 42 <i>d</i>	3D	In + S; 170 °C, 8 d	DAH + H ₂ O	58%	
SCIF-1 ([In ₁₀ S ₁₆ (IM) ₂] ⁴⁻)	<i>Pbca</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + imidazole + 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	[35]
SCIF-2 ([In ₁₀ S ₁₆ (2-MIM) ₂] ⁴⁻)	<i>C2/c</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + 2-methylimidazole + 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	
SCIF-3 ([In ₁₀ S ₁₆ (2-EIM) ₂] ⁴⁻)	<i>I</i> 42 <i>d</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + 2-ethylimidazole + 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	
SCIF-4 ([In ₁₀ S ₁₆ (4-MIM) ₂] ⁴⁻)	<i>Pbca</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + 4-methylimidazole + 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	
SCIF-5 ([In ₁₀ S ₁₆ (BIM) ₂] ⁴⁻)	<i>P2₁2₁2₁</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + Benzimidazole+ 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	
SCIF-6 ([In ₁₀ S ₁₆ (MBIM) ₂] ⁴⁻)	<i>Pbca</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + 5-methylbenzimidazole + 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	
SCIF-7 ([In ₁₀ S ₁₆ (DMBIM) ₂] ⁴⁻)	<i>Pbca</i>	3D	In(NO ₃) ₃ • xH ₂ O + Li ₂ S + 5,6-dimethylbenzimidazole + 2-amino-1-butanol + DBU; 150 °C, 12 d	NA	NA	
OCF-31 ([In ₈ S ₁₅] ⁶⁻)	<i>P2₁2₁2₁</i>	3D	In + S + BIM + hexamethyleneimine	H ₂ O	NA	

Abbreviations: en = ethylenediamine; dien = diethylenetriamine; dap = diaminopropane; tepa = tetraethylenepentamine; Ph₄PCl = tetraphenylphosphonium chloride; CH₃OH = methanol; EtOH = ethanol; DMF = Dimethylformamide; phen = 1,10-phenanthroline; dach = 1,2-diaminocyclohexane; TETA = triethylenetetramine; TMAP = 4,4'-trimethylenedipiperidine; DEA = diethylamine; TTCA = trithiocyanuric acid; EA = ethylamine; TMDP = 4,4-trimethylenedipiperidine; DPA = dipropylamine; DMA = dimethylamine; BPP = 1,3,4,6,7,8-hexahydro-2H-pyrimido[1,2-a]pyrimidine; DPM = dipiperidinomethane; C₆H₁₂NH = hexamethyleneimine; TMDP = 4,4-trimethylenedipiperidine; BAPP = 1,4-bis(3-aminopropyl)piperazine; Py = pyridine; DBU = 1,8-diazabicyclo[5.4.0]-7-undecene; BIM = benzimidazole.

Table S4. Summary of ammonia/ammine templated In-Se compounds.

Compounds	SG	D	Reactants	Solvent	Yield	Ref.
[Ph ₄ P] ₄ In ₂ Se ₂₁	<i>P</i> 1	0D	InCl ₃ + Na ₂ Se ₅ ; RT, 2 d	Ph ₄ PCl + DMF	75%	[36]
(Pr ₄ N) ₄ [In ₂ (Se ₄) ₄ (Se ₅)]	<i>P2/c</i>	0D	InCl ₃ + Na ₂ Se ₅ ; RT, 2 d	Pr ₄ NBr + DMF	68%	[37]
(Et ₄ N) ₄ [In ₂ (Se ₄) ₄ (Se ₅)]	<i>P</i> 1	0D	InCl ₃ + Na ₂ Se ₅ ; RT	Et ₄ NBr + DMF	72%	
(Pr ₄ N) ₂ [In ₂ Se ₂ (Se ₄) ₂]	<i>P</i> 1	0D	InCl ₃ + Na ₂ Se ₅ ; 110 °C, 3 d	Pr ₄ NBr + H ₂ O	80%	
[(Ph ₃ P) ₂ N] ₂ [In ₂ Se ₂ (Se ₄) ₂]	<i>P</i> 1	0D	InCl ₃ + Na ₂ Se ₅ ; 110 °C, 3 d	[(Ph ₃ P) ₂ N]Cl + H ₂ O	60%	
(Et ₄ N) ₃ [In ₃ Se ₃ (Se ₄) ₃]	<i>P2₁c</i>	0D	InCl ₃ + Na ₂ Se ₅ ; RT, 4 d InCl ₃ + Na ₂ Se ₅ ; 110, 3 d	Et ₄ NBr + CH ₃ CN/ Et ₄ NBr + H ₂ O	76%/85%	
[Mn(en) ₃][In ₂ Se ₅]	<i>Pbcn</i>	1D	In ₂ Se ₃ + Se + MnCl ₂ ; 170 °C, 6 d	en + py/thiophenol	21%	[38]
[C ₆ H ₁₆ N ₂][In ₂ Se ₃ (Se ₂)]	<i>C2/c</i>	1D	In + Se; 170 °C, 10 d	trans-1,4-diaminocycl- ohexane + H ₂ O	N _A	[39]
(bdaH)InSe ₂	<i>Fdd2</i>	1D	In + Se; 160 °C, 6 d	bda	31.8%	[40]
{[Ni(phen) ₃]In ₂ Se ₅ H ₂ O} _n	<i>P2₁2₁2₁</i>	1D	In + Ni + Se; 140 °C, 4 d	phen + din-butylamine + H ₂ O	31%	[41]
{[Fe(phen) ₃]In ₂ Se ₅ H ₂ O} _n	<i>P2₁2₁2₁</i>	1D	In + Fe + Se; 140 °C, 4 d	phen + di-n-butylamine + H ₂ O	35%	
[C ₉ H ₁₇ N ₂] ₃ [In ₅ Se _{8.07} (Se ₂) _{0.93}]	<i>P2₁c</i>	1D	In + Se + imidazole; 200 °C, 10 d	DBU + acetonitrile	NA	[42]

[C ₉ H ₁₇ N ₂] ₃ [In ₅ Se _{8.74} (Se ₂) _{0.26}]	<i>P</i> 1̄	1D	InCl ₃ + Se; 200 °C, 10 d	DBU	NA	
[C ₆ H ₁₂ N ₂] ₄ [C ₆ H ₁₄ N ₂] ₃ [In ₁₀ Se ₁₅ (Se ₂) ₃]	<i>P</i> 2 ₁ /c	1D	In + Se; 140 °C, 10 d	DABDO + DMF	NA	
(Ph ₄ P)[In(Se ₆) ₂]	<i>P</i> 4̄	2D	In + (Ph ₄ P) ₂ Se ₅ + Se; 200 °C, 2 d	NA	60%	[15]
[C ₇ H ₁₀ N][In ₃ Se ₅]	<i>P</i> 2 ₁ /c	2D	In + Se; 185 °C, 10 d	3,5-dimethylpyridine	NA	[43]
[C ₇ H ₁₀ N][In ₃ Se ₅]	<i>P</i> 2 ₂ 1 ₂ 1	2D	In + Se; 200 °C, 10 d	2,6-dimethylpyridine	NA	
[C ₆ H ₁₄ N ₂]·[(C ₆ H ₁₂ N ₂) ₂ NaIn ₅ Se ₉]	<i>C</i> 2/c	3D	In + Se; 140 °C, 10 d	DABDO + NaOH + acetonitrile	NA	[42]
[enH ₂][NH ₄][In ₇ Se ₁₂]	<i>P</i> 2 ₁	3D	In + Se; 200 °C, 14 d	DABDO + H ₂ O	NA	
[μ ₃ -Se ₄] _{3.27} ·[In _{49.88} Se _{95.92}]·(C ₅ H ₁₂ N) _{26.0} ·C ₂ H ₈ N) _{42.4}	<i>R</i> 3c	3D	In + Se; 170 °C, 5 d	PR + DMF	NA	[44]
[In ₄ Se ₁₀]·(C ₇ H ₁₆ N) _{1.8} ·(C ₂ H ₈ N) _{2.2}	<i>P</i> nma	3D	In + Se; 170 °C, 7 d	1,4-dioxane + 3,5-DMP + DMF	NA	
[In ₂₀ Se ₃₉]·(C ₆ H ₁₄ N) ₁₂	<i>P</i> 2 ₁ /c	3D	In + Se; 180 °C, 6 d	H ₂ O + 1-MPR	NA	
[In ₄₀ Se ₇₉] ₃₈ ·38H ⁺ ·3,5-DMP	<i>I</i> 43d	3D	In + Se; 170 °C, 5 d	3,5-DMP + ATMP	NA	[45]
[In ₃₆ Se ₆₉] ₃₀ ·30H ⁺ ·3,5-DMP	<i>P</i> 43n	3D	In + Se; 180 °C, 7 d	3,5-dimethylpyridine + H ₂ O	NA	
[C ₇ H ₁₀ N][In ₉ Se ₁₄]	<i>P</i> 6 ₃ /m	3D	In + Se; 200 °C, 16 d	3,5-dimethylpyridine	NA	[46]
UCR-7InSe-TETA ([In ₁₀ Se ₁₈] ⁶⁻)	<i>I</i> 4 ₁ /acd	3D	In + Se; 190 °C, 6 d	TETA + H ₂ O	33%	[47]
UCR-7InSe-PPZ ([In ₁₀ Se ₁₈] ⁶⁻)	<i>I</i> 4 ₁ /acd	3D	In + Se; 190 °C, 6 d	PPZ + H ₂ O	NA	
UCR-7InSe-DABCO ([In ₁₀ Se ₁₈] ⁶⁻)	<i>I</i> 4 ₁ /acd	3D	In + Se; 190 °C, 6 d	DABCO + H ₂ O	NA	
UCR-7InSe-AEAE ([In ₁₀ Se ₁₈] ⁶⁻)	<i>P</i> 4 ₁ 2 ₁ 2	3D	In + Se; 190 °C, 6 d	AEAE + H ₂ O	NA	
UCR-7InSe-DMAPA ([In ₁₀ Se ₁₈] ⁶⁻)	<i>P</i> 4 ₁ 2 ₁ 2	3D	In + Se; 190 °C, 6 d	DMAPA + H ₂ O	NA	
[In ₂₈ Se ₅₄ (H ₂ O) ₄] ²⁴⁻ ·24(H ⁺ -PR)·nH ₂ O	<i>F</i> 4 ₁ 3c	3D	In + Se; 170 °C, 7 d	PR + H ₂ O	NA	[48]
(H ⁺ -DMP) _{58.26} ·[In ₅₆ Se _{113.13}]	<i>P</i> 4/ncc	3D	In + Se; 170 °C, 5 d	3,5-DMP + ATMP	NA	[49]
(H ⁺ -MPR) _{57.94} ·[In ₆₀ Se _{118.97}]	<i>C</i> 2/c	3D	In + Se; 170 °C, 7 d	MPR + H ₂ O	NA	
(NH ₄) ₄ In ₁₂ Se ₂₀	<i>P</i> ca2 ₁	3D	In + Se; 250 °C, 2 d	NH ₄ OH	NA	[50]
(NH ₄) ₂ In ₁₂ Se ₁₉	<i>R</i> 3̄	3D	In + Se; 250 °C, 1 d	NH ₄ OH + H ₂ O	NA	
UCR-2InSe-DIPA ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	DIPA + H ₂ O	NA	[31]
UCR-2InSe-TETA ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	TETA + H ₂ O	NA	
UCR-2InSe-TAA ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	TAA + H ₂ O	NA	
UCR-2InSe-AEAE ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	AEAE + H ₂ O	NA	
UCR-2InSe-DPA ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	DPA + H ₂ O	NA	
UCR-2InSe-TMHD ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	TMHD + H ₂ O	NA	
UCR-2InSe-BAPP ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	BAPP + H ₂ O	NA	
UCR-2InSe-TOTD ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	TOTD + H ₂ O	NA	
UCR-2InSe-HMI ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	HMI + H ₂ O	NA	
UCR-2InSe-DAO ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	DAO + H ₂ O	NA	
UCR-2InSe-DMMP ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	DMMP + H ₂ O	NA	
UCR-2InSe-APM ([In ₃₃ Se ₅₆] ¹³⁻)	<i>R</i> 32	3D	In + Se; 200 °C, 10 d	APM + H ₂ O	NA	

Abbreviations: Ph₄PCl = tetraphenylphosphonium chloride; DMF = Dimethylformamide; en = ethylenediamine; Py = pyridine; bda = 1,4-butanediamine; phen = 1,10-Phenanthroline; DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene; DABDO = 1,4-diazabicyclo[2.2.2]octane; PR = piperidine; TETA = triethylenetetramine; PPZ = piperazine; AEAE = 2-(2-aminoethylaminoethanol); DMAPA = 3-dimethylaminopropylamine; 3,5-DMP/3,5-DMP = 3,5-dimethylpiperidine; 1-MPR = 1-methylpiperidine; H⁺-PR = protonated piperidine; ATMP = amino trimethylene phosphonic acid; MPR = 3-methylpiperidine.

2. Figures

2.1 Synthesis and Structures

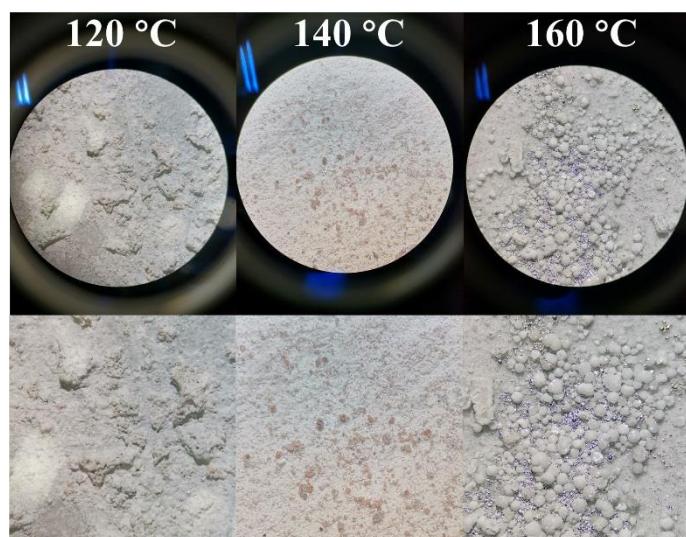


Figure S1. Photographs of the products obtained from the optimizing reactions for **2** ($\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$:urea = 0.65:1) performed at different temperatures. Top line: products washed by distilled water; bottom line: magnified imaging of the products.

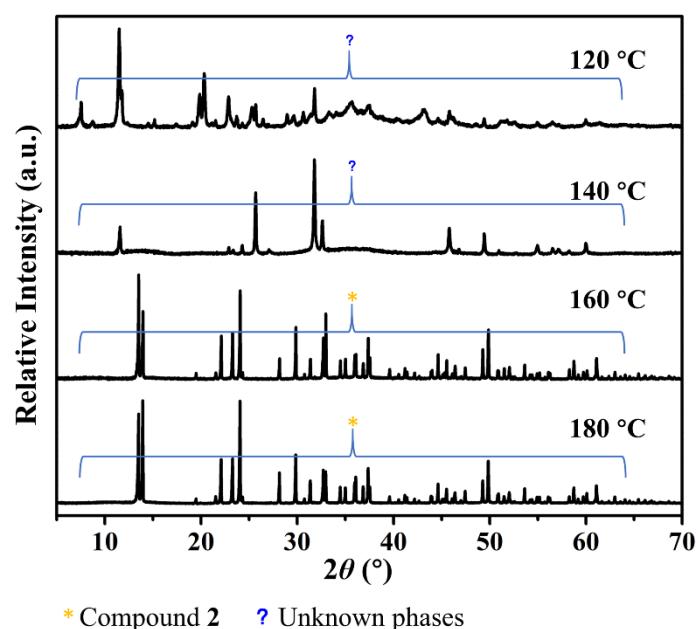


Figure S2. Powder XRD patterns of the products obtained from the optimizing reactions for **2** performed at different temperatures.

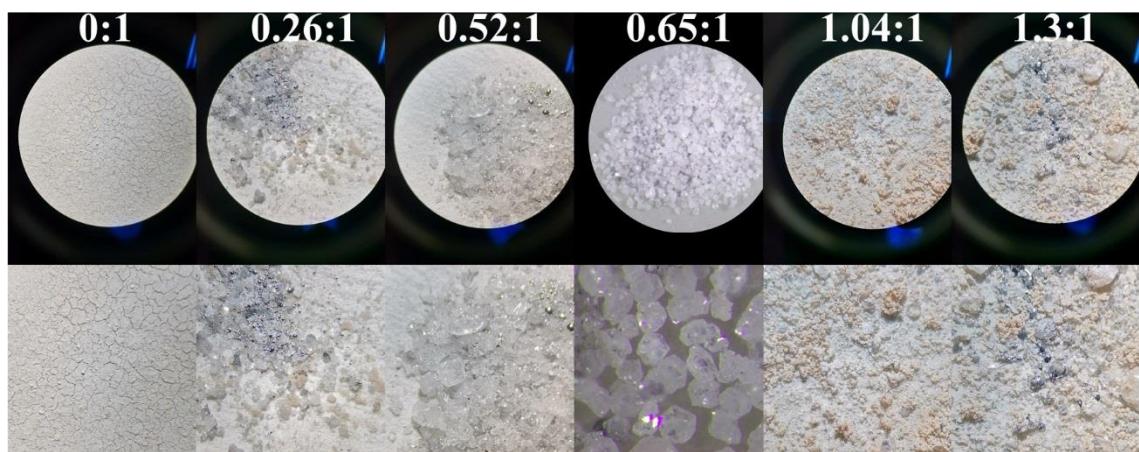


Figure S3. Photographs of the products obtained from the optimizing reactions for **2** with different $\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$:urea molar ratios at 180°C . Top line: products washed by distilled water; bottom line: magnified imaging of the products.

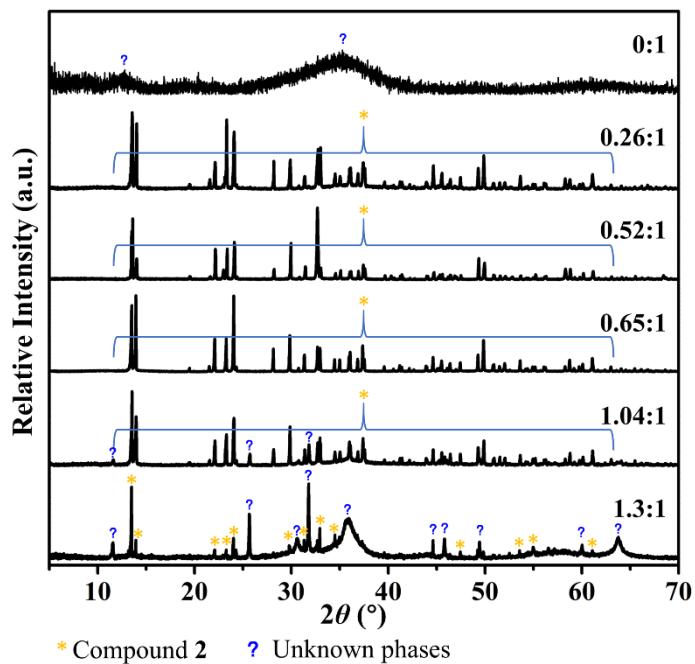


Figure S4. Powder XRD patterns of the products from optimizing reactions for **2** with different $\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$:urea ratio at 180°C .

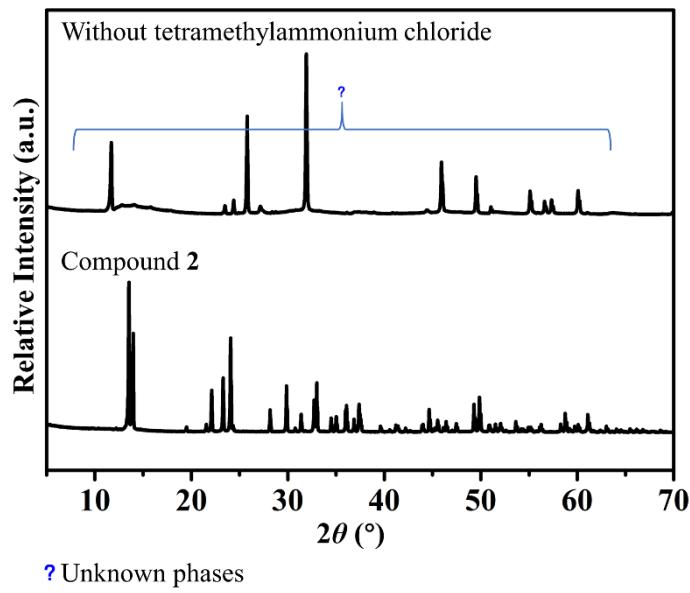


Figure S5. Powder XRD patterns of compound 2 and the product from the reaction without adding tetramethylammonium chloride.

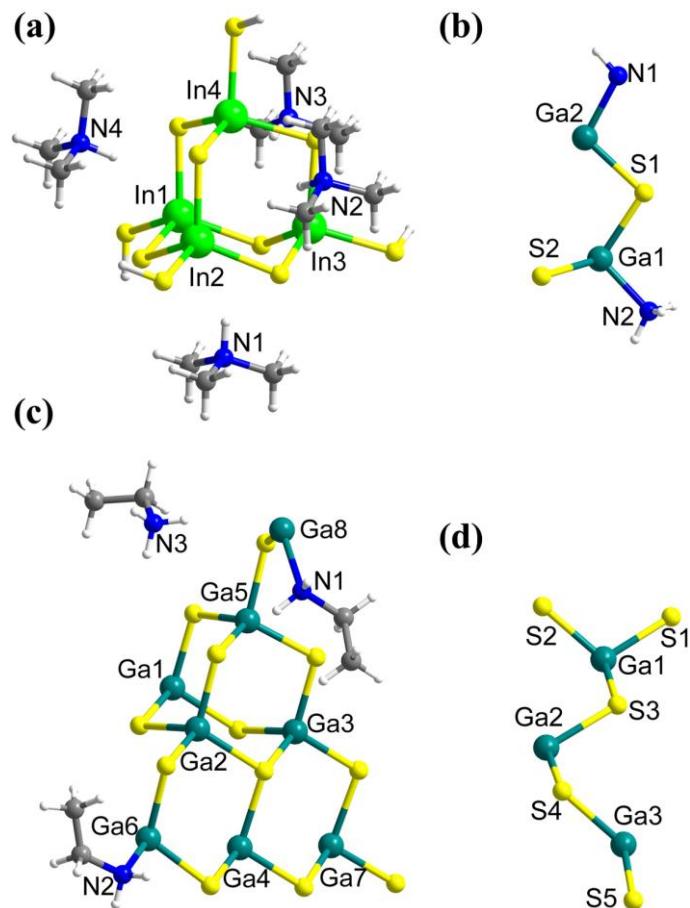


Figure S6. Asymmetric units of (a) 1, (b) 2, (c) 3, and (d) 4. Color code: In (bright green), Ga (teal), S (yellow), N (blue), C (gray), H (white). Note: some of the atoms shown in the figure are not fully occupied due to their location at special positions.

2.2 Characterizations

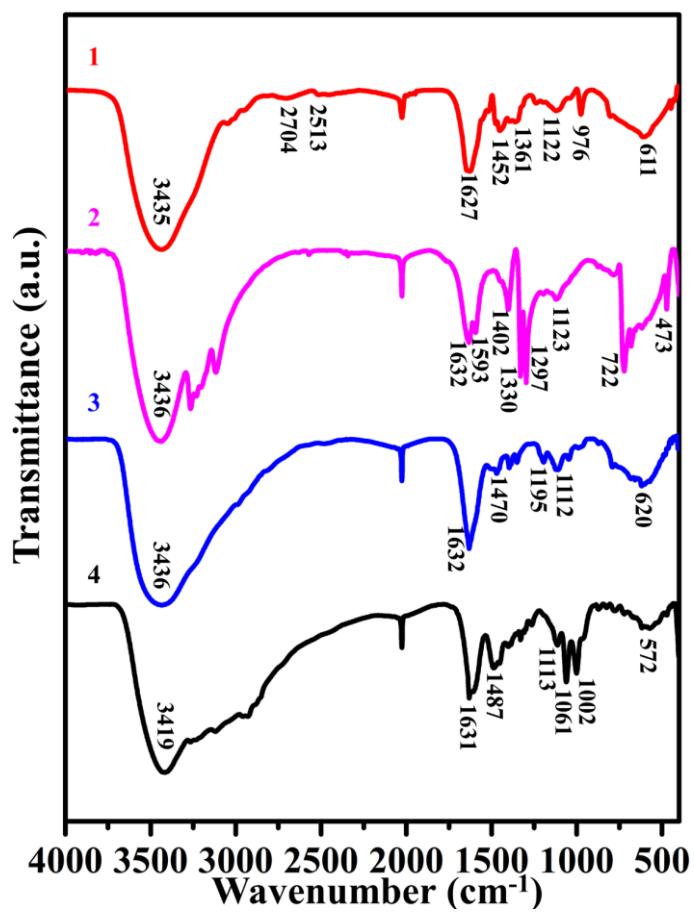


Figure S7. FTIR spectra of 1–4.

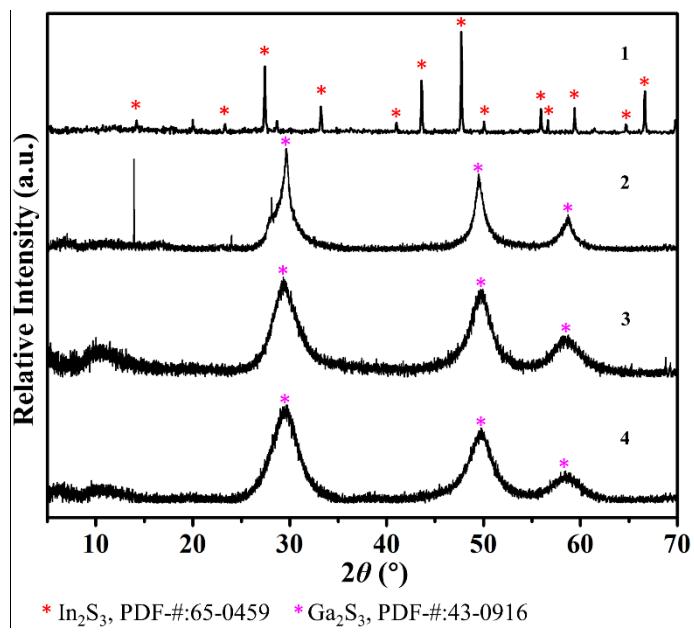


Figure S8. Powder XRD patterns for the TG residues of 1–4.

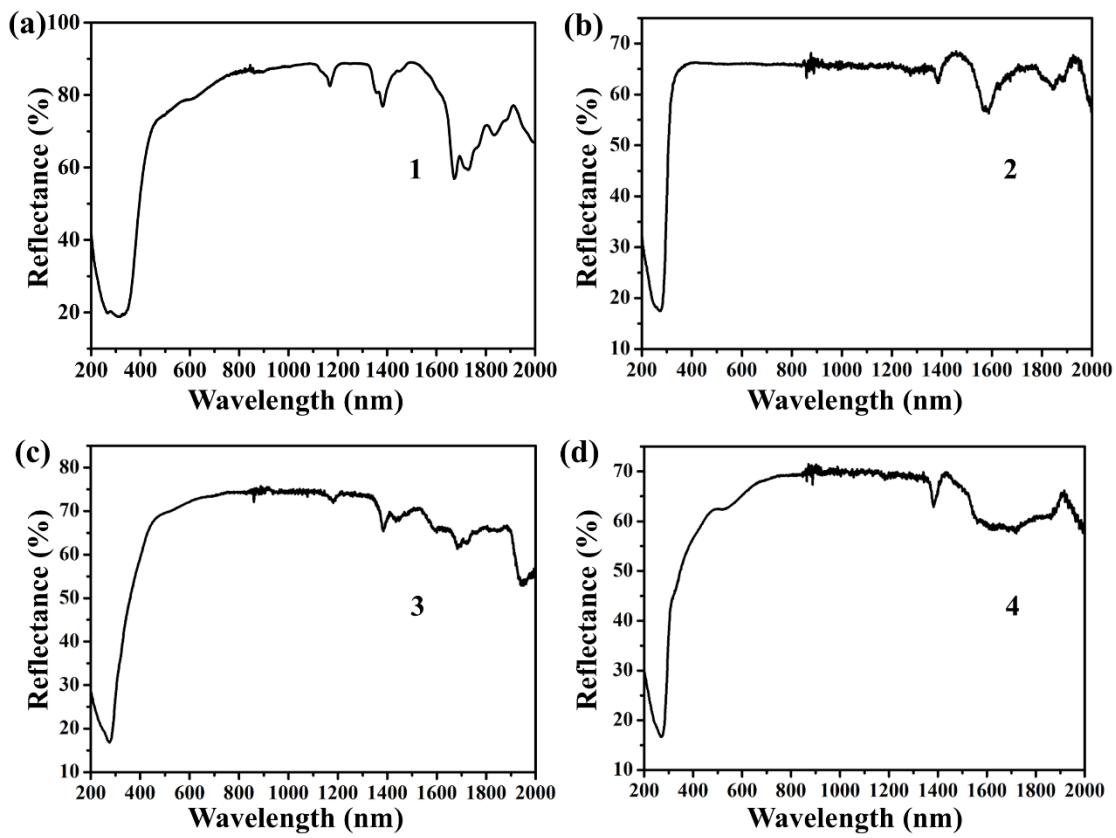


Figure S9. UV-vis reflectance spectra of compounds 1-4.

2.3 Organic template identification for 1-4 by ^1H and ^{13}C NMR.

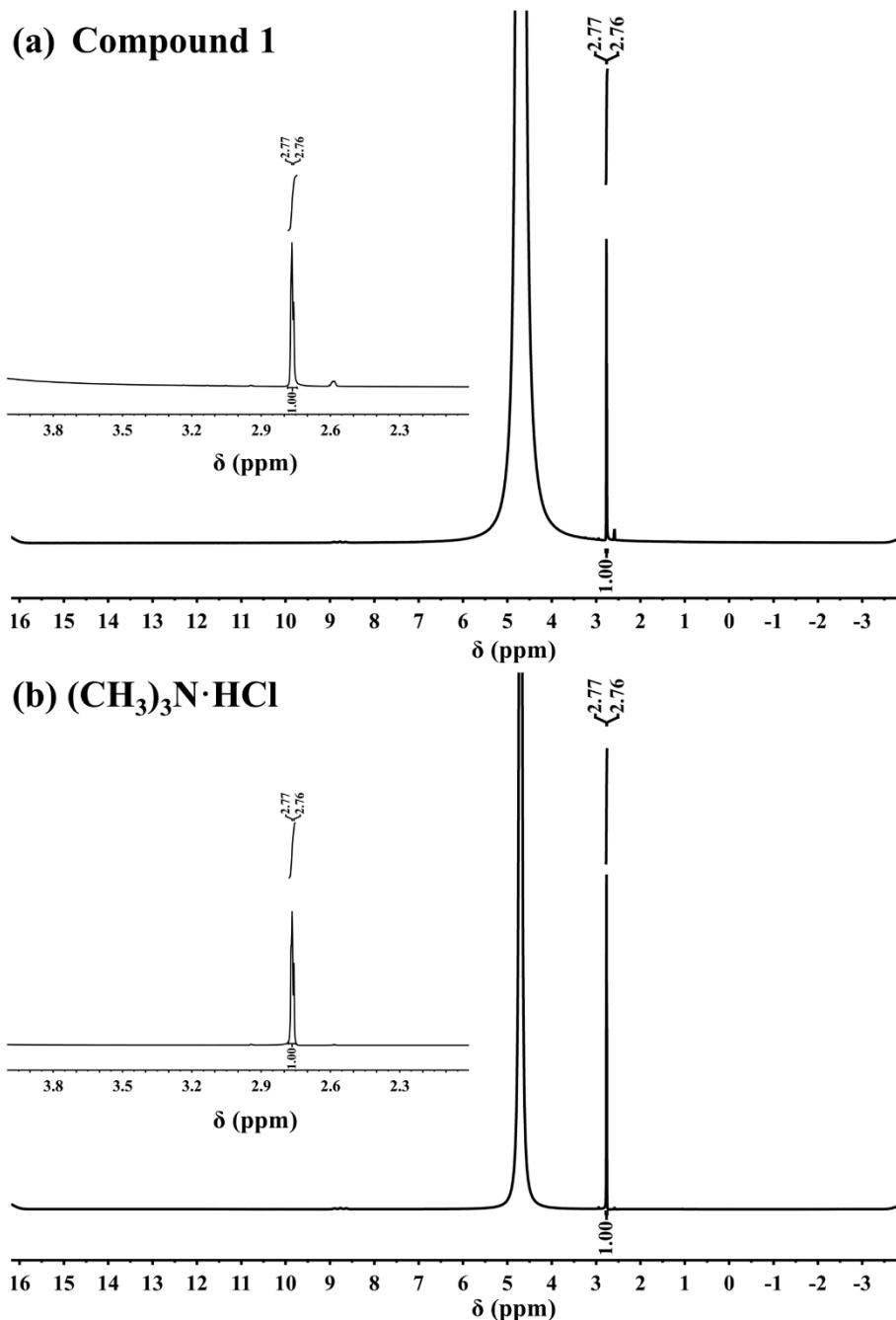
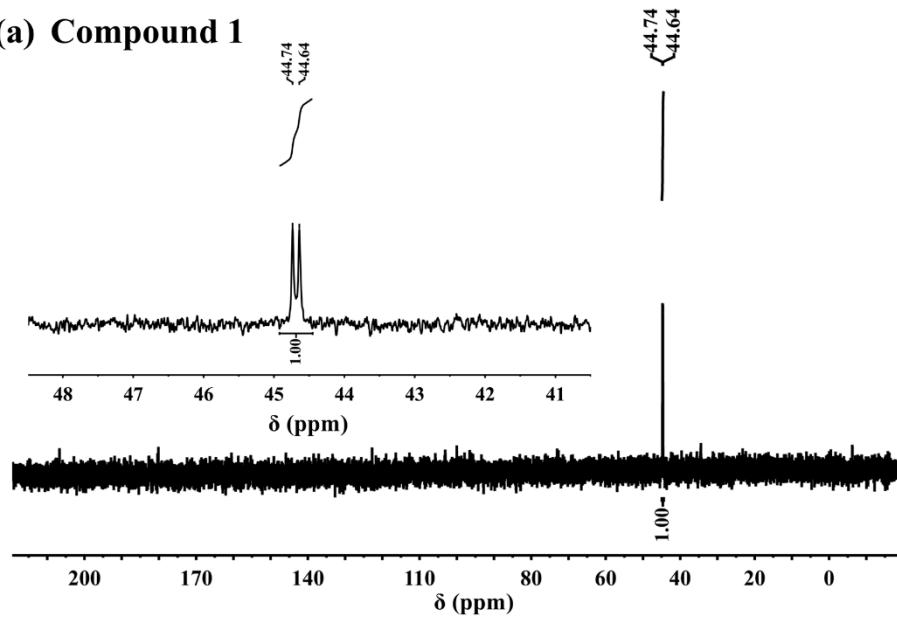


Figure S10. ^1H NMR spectra of **1** and $(\text{CH}_3)_3\text{N}\cdot\text{HCl}$ dissolved in mixed HCl (0.1 M)/D₂O recorded at room temperature.

(a) Compound 1



(b) (CH₃)₃N·HCl

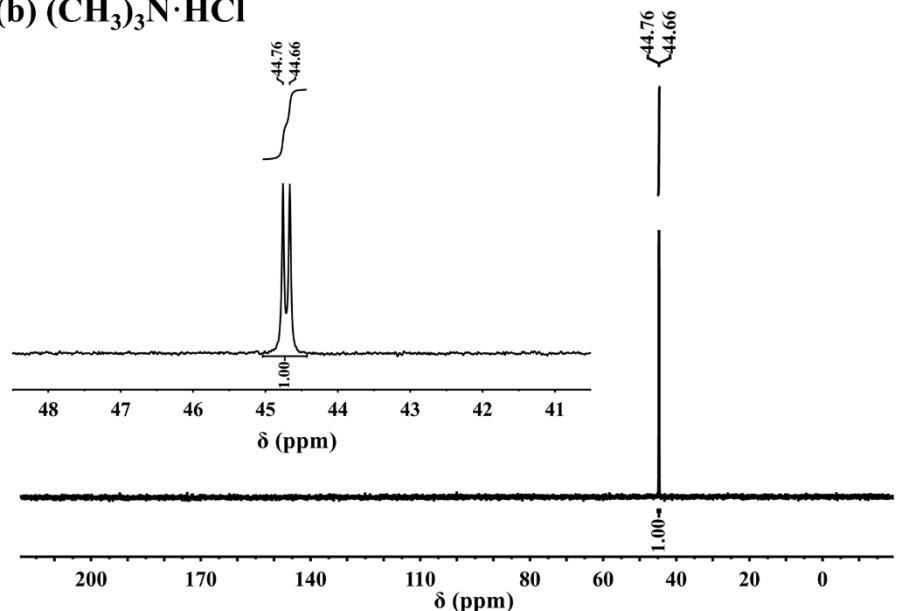
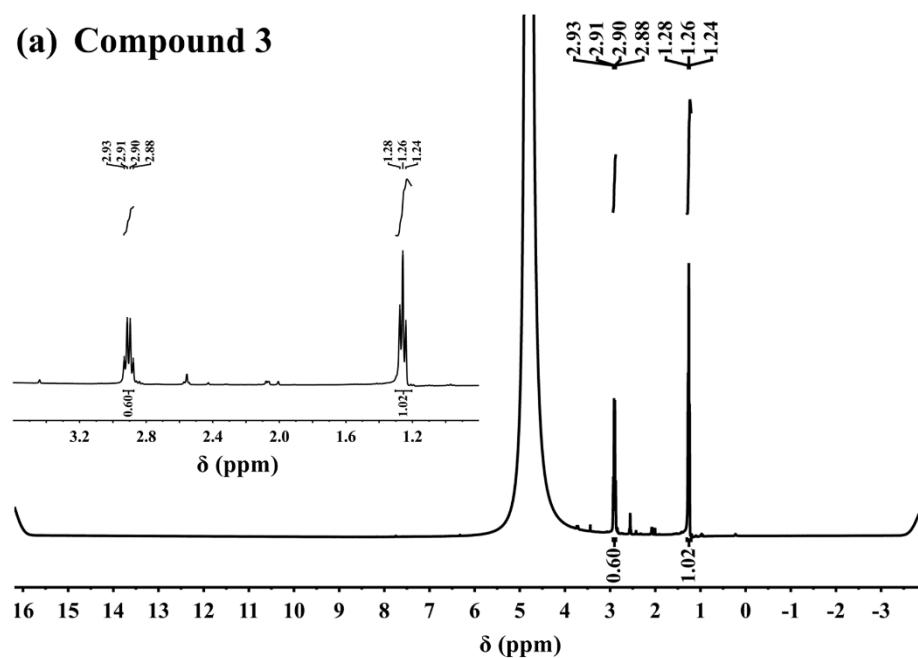


Figure S11. ¹³C NMR spectra of **1** and (CH₃)₃N·HCl dissolved in mixed HCl (0.1 M)/D₂O recorded at room temperature.

(a) Compound 3



(b) CH₃CH₂NH₂·HCl

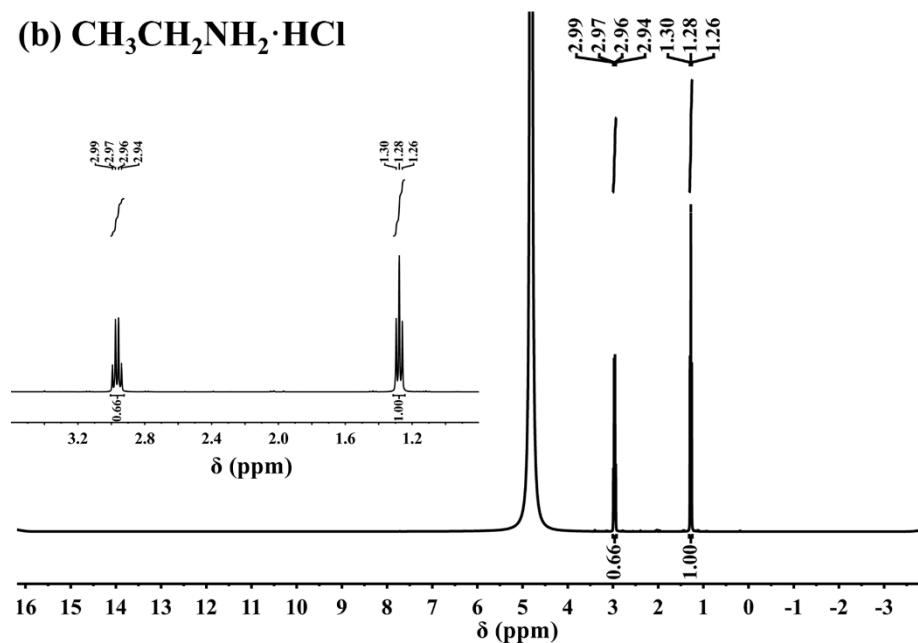
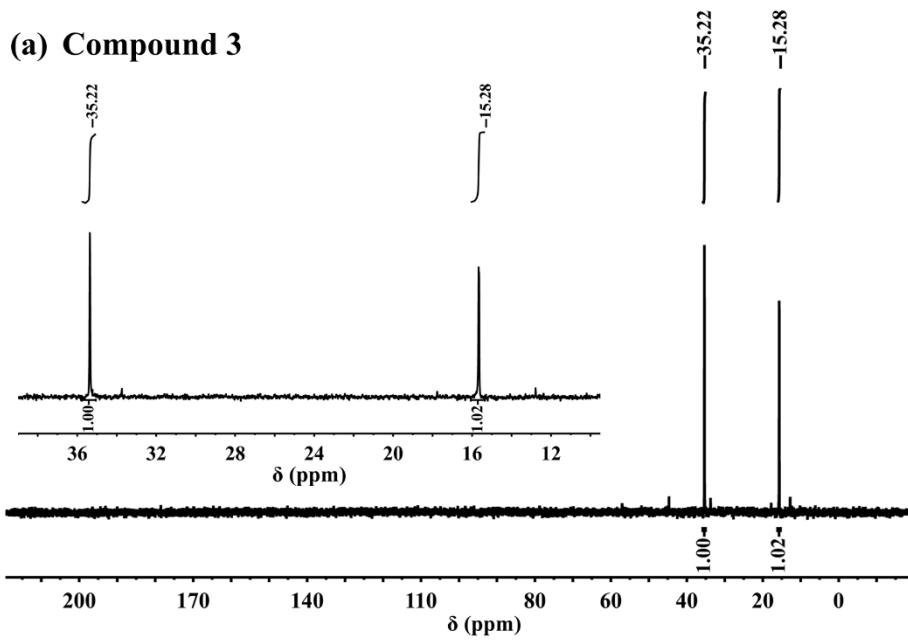


Figure S12. ¹H NMR spectra of **3** and CH₃CH₂NH₂·HCl dissolved in N₂H₄·H₂O (98%)/D₂O recorded at room temperature.

(a) Compound 3



(b) CH₃CH₂NH₂·HCl

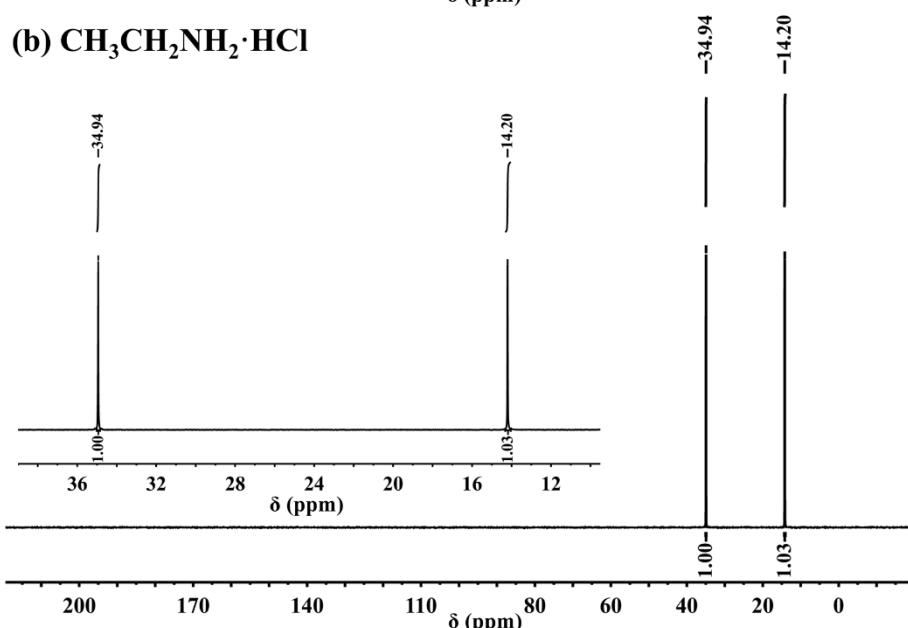
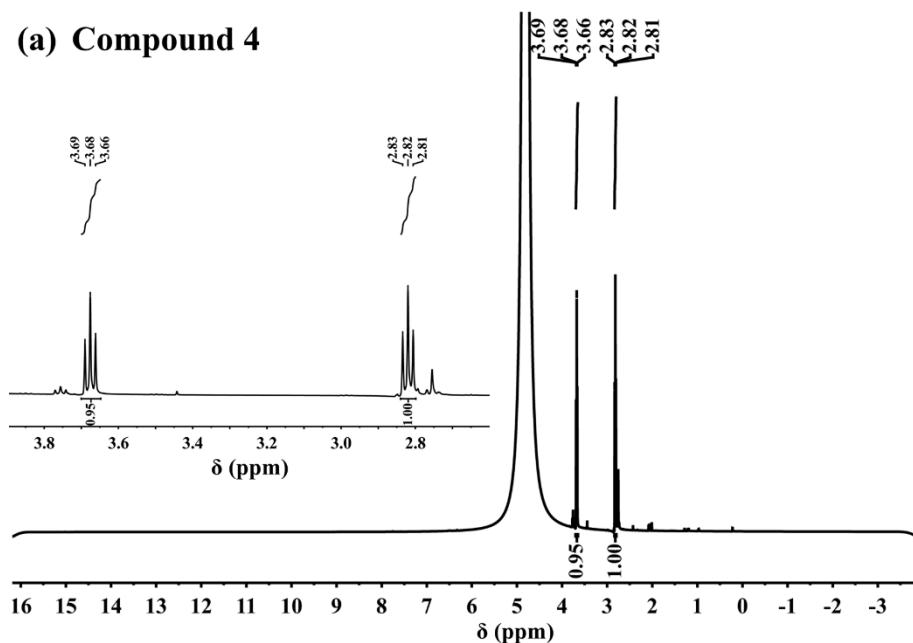


Figure S13. ¹³C NMR spectra of **3** and CH₃CH₂NH₂·HCl dissolved in N₂H₄·H₂O (98%)/D₂O recorded at room temperature.

(a) Compound 4



(b) HOCH₂CH₂NH₂

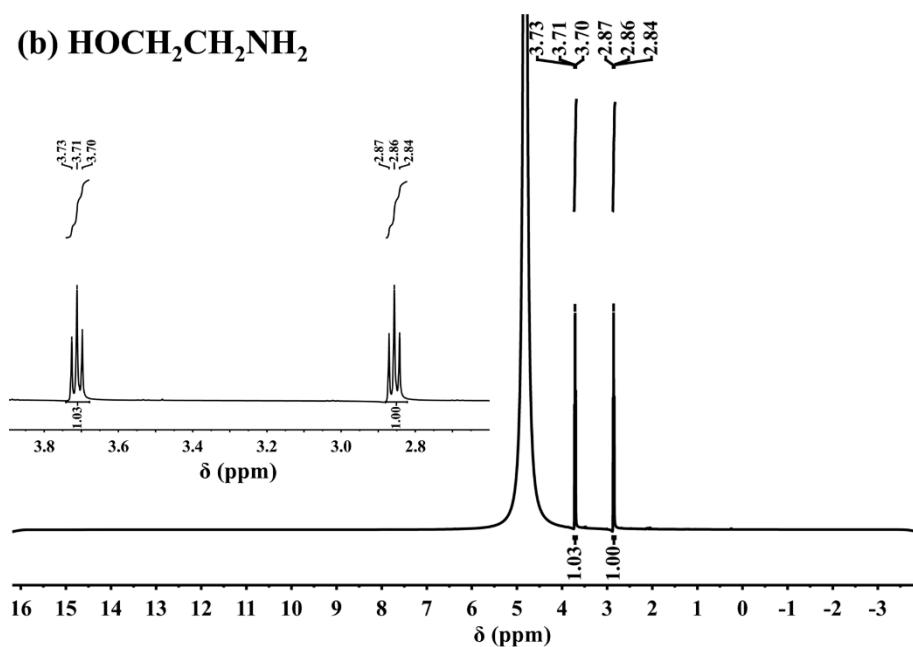


Figure S14. ¹H NMR spectra of **4** and HOCH₂CH₂NH₂ dissolved in N₂H₄·H₂O (98%)/D₂O recorded at room temperature.

(a) Compound 4

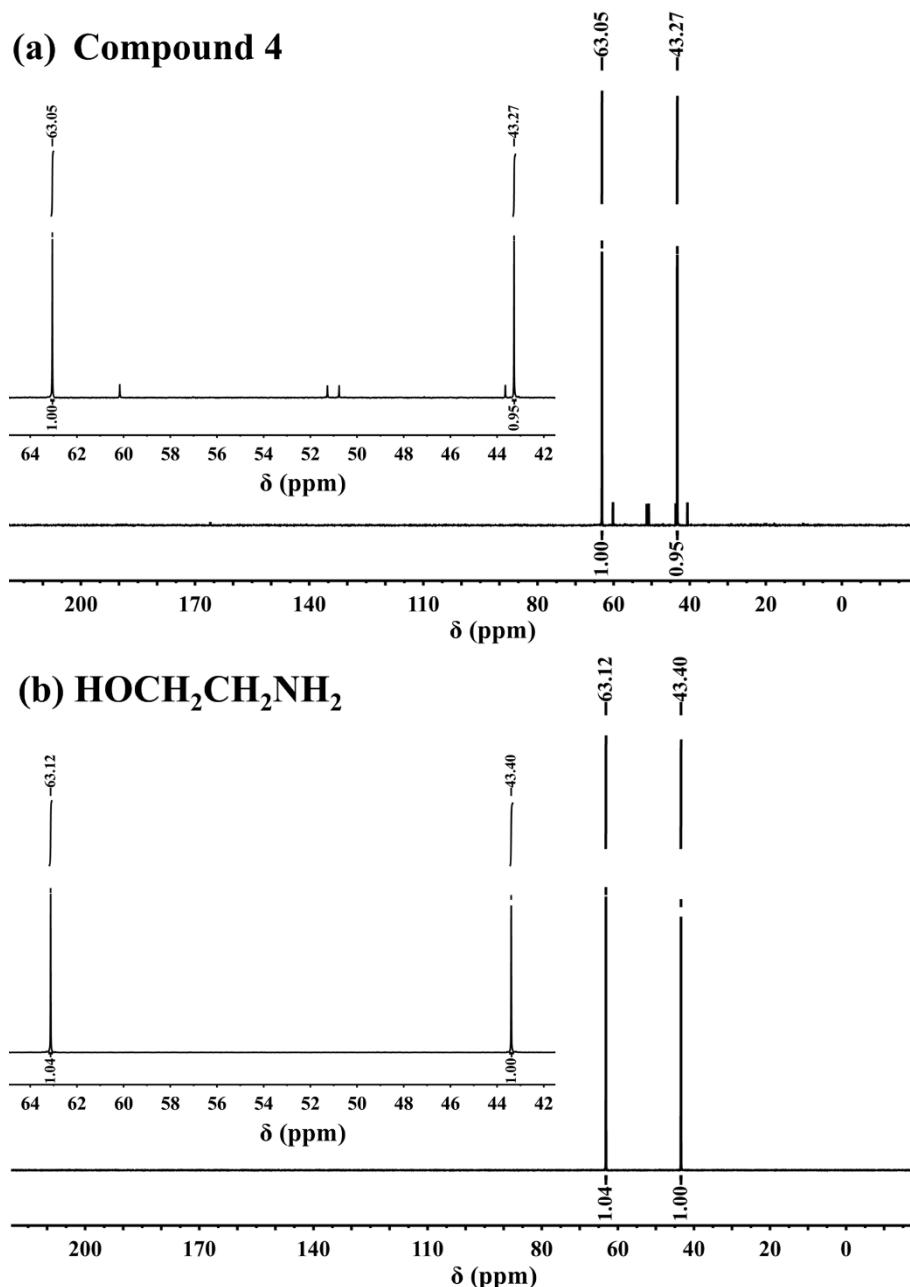


Figure S15. ^{13}C NMR spectra of **4** and $\text{HOCH}_2\text{CH}_2\text{NH}_2$ dissolved in $\text{N}_2\text{H}_4 \cdot \text{H}_2\text{O}$ (98%)/ D_2O recorded at room temperature.

References:

- (1) Krebs, B.; Voelker, D.; Stiller, K.-O. Novel adamantane-like thio- and selenoanions from aqueous solution: $\text{Ga}_4\text{S}_{10}^{8-}$, $\text{In}_4\text{S}_{10}^{8-}$, $\text{In}_4\text{Se}_{10}^{8-}$. *Inorganica Chimica Acta* **1982**, 65, L101-L102.
- (2) Vaqueiro, P. Solvothermal synthesis and characterisation of new one-dimensional indium and gallium sulphides: $[\text{C}_{10}\text{N}_4\text{H}_{26}]_{0.5}[\text{InS}_2]$ and $[\text{C}_{10}\text{N}_4\text{H}_{26}]_{0.5}[\text{GaS}_2]$. *J. Solid State Chem.* **2006**, 179, 302-307.
- (3) Vaqueiro, P.; Romero, M. L. Gallium-sulfide supertetrahedral clusters as building blocks of covalent organic-inorganic networks. *J. Am. Chem. Soc.* **2008**, 130, 9630-9631.
- (4) Vaqueiro, P.; Romero, M. L.; Rowan, B. C.; Richards, B. S. Arrays of chiral nanotubes and a layered coordination polymer containing gallium-sulfide supertetrahedral clusters. *Chem. Eur. J.* **2010**, 16, 4462-4465.
- (5) Zhou, J.; Zhang, Y.; Bian, G.-Q.; Li, C.-Y.; Chen, X.-X.; Dai, J. Structural study of organic-inorganic hybrid thiogallates and selenidogallates in view of effects of the chelate amines. *Cryst. Growth. Des.* **2008**, 8, 2235-2240.
- (6) Vaqueiro, P. From one-dimensional chains to three-dimensional networks: solvothermal synthesis of thiogallates in ethylenediamine. *Inorg. Chem.* **2006**, 45, 4150-4156.
- (7) Friedrich, D.; Schlosser, M.; Weihrich, R.; Pfitzner, A. Polymorphism of CsGaS_2 - structural characterization of a new two-dimensional polymorph and study of the phase-transition kinetic. *Inorg. Chem. Front.* **2017**, 4, 393-400.
- (8) Vladislav, V.; Klepov, A. A.; Berseneva, K. A.; Pace, V. K.; Sun, M.-Q.; Qiu, P.; Wang, H.; Chen, F.-L.; Theodore, M. B.; Hans-Conrad, Z. L. NaGaS_2 : An elusive layered compound with dynamic water. Absorption and wide-ranging ion-exchange properties. *Angew. Chem. Int. Ed.* **2020**, 59, 10836-10841.
- (9) Zheng, N.-F.; Bu, X.-H.; Feng, P.-Y. Nonaqueous synthesis and selective crystallization of gallium sulfide clusters into three-dimensional photoluminescent superlattices. *J. Am. Chem. Soc.* **2003**, 125, 1138-1139.
- (10) Vaqueiro, P.; Romero, M. L. Three-dimensional gallium sulphide open frameworks. *J Phys Chem Solids.* **2007**, 68, 1239-1243.
- (11) Li, C.-Y.; Zhou, J.; Bian, G.-Q.; Zhang, M.-H.; Dai, J. The first polymeric thiogallate with lanthanide-containing counter cation, $[\text{Dy}_2(\text{en})_6(\mu_2\text{-OH})_2]\text{Ga}_4\text{S}_8$. *Inorg. Chem. Commun.* **2008**, 11, 1327-1329.
- (12) Bu, X.-H.; Zheng, N.-F.; Wang, X.; Wang, B.; Feng, P.-Y. Three dimensional frameworks of gallium selenide supertetrahedral clusters. *Angew. Chem.* **2004**, 116, 1528-1531.
- (13) Xiong, W.-W.; Li, J.-R.; Feng, M.-L.; Huang, X.-Y. Solvothermal syntheses, crystal structures, and characterizations of a series of one-dimensional organic-containing gallium polyselenides. *Crystengcomm.* **2011**, 13, 6206-6211.
- (14) Ewing, S. J.; Romero, M. L.; Hutchinson, J.; Powell, A. V.; Vaqueiro, P. Solvothermal synthesis of one-dimensional chalcogenides containing group 13 elements. *Z. Anorg. Allg. Chem.* **2012**, 638, 2526-2531.
- (15) Dhingra, S.; Kanatzidis, M. G. Open framework structures based on Se_x^{2-} fragments: synthesis of $(\text{Ph}_4\text{P})[\text{M}(\text{Se}_6)_2]$ ($\text{M} = \text{Ga, In, TI}$) in Molten $(\text{Ph}_4\text{P})_2\text{Se}_x$. *Science*. **1992**, 258, 1769-1772.
- (16) Friedrich, D.; Schlosser , M.; Pfitzner, A. Synthesis and structural characterization of the layered selenogallate RbGaSe_2 . *Z. Anorg. Allg. Chem.* **2017**, 643, 1589-1592.
- (17) Dong, Y.; Peng, Q.; Wang, R.; Li, Y. Synthesis and characterization of an open framework

- gallium selenide: $\text{Ga}_4\text{Se}_7(\text{en})_2 \cdot (\text{enH})_2$. *Inorg. Chem.* **2003**, *42*, 1794-1796.
- (18) Dhingra, S. S.; Kanatzidis, M. G. Syntheses and characterization of the new homoleptic indium-polysulfide complexes $[\text{In}_2\text{S}_{27}]^{4-}$, $[\text{In}_2\text{S}_{14}]^{2-}$, and $[\text{In}_2\text{S}_{16}]^{2-}$. *Inorg. Chem.* **1993**, *32*, 3300-3305.
- (19) Quiroga-González, E.; Kienle, L.; Näther, C.; Chakravadhanula, V. S. K.; Lühmann, H.; Bensch, W. Zero- and one-dimensional thioindates synthesized under solvothermal conditions yielding α - In_2S_3 , β - In_2S_3 or MgIn_2S_4 as thermal decomposition products. *J. Solid State Chem.* **2010**, *183*, 2805-2812
- (20) Zhou, J.; Bian, G.-Q.; Zhang, Y.; Zhu, Q.-Y.; Li, C.-Y.; Dai, J. One-dimensional indium sulfides with transition metal complexes of polyamines. *Inorg. Chem.* **2007**, *46*, 16, 6347-6352.
- (21) Wang, Y.-H.; Jiang, J.-B.; Wang, P.; Sun, X.-Lu.; Zhu, Q.-Y.; Dai, J. Polymeric supertetrahedral InS clusters assembled by new linkages. *CrystEngComm* **2013**, *15*, 6040-6045.
- (22) Zhang, X.; Luo, W.; Zhang, Y.-P.; Jiang, J.-B.; Zhu, Q.-Y.; Dai, J. Indium-sulfur supertetrahedral polymers integrated with $[\text{M}(\text{phen})_3]^{2+}$ cations ($\text{M} = \text{Ni}$ and Fe). *Inorg. Chem.* **2011**, *50*, 6972-6978.
- (23) Zhang, Q.-C.; Bu, X.-H.; Han, L.; Feng, P.-Y. Two-dimensional indium sulfide framework constructed from pentasupertetrahedral P_1 and supertetrahedral T_2 clusters. *Inorg. Chem.* **2006**, *45*, 6684-6687.
- (24) Cahill, C. L.; Gugliottaa, B.; Parisea, J. B. A novel layered indium sulfide material consisting of corner and edge shared InS_4 tetrahedra: synthesis and structural characterization of DPA-InS-SB3. *Chem. Commun.* **1998**, 1715-1716.
- (25) Wang, K.-Y.; Sun, M.; Ding, D.; Liu, H.-W.; Cheng, L.; Wang, C. Di-lacunary $[\text{In}_6\text{S}_{15}]^{12-}$ cluster: building block for a highly negative charged material with good Sr^{2+} ion exchange property. *Chem. Commun.* **2020**, *56*, 3409-3412.
- (26) Sun, M.; Wang, K.-Y.; Ding, D.; Zhu, J.-Y.; Zhao, Y.-M.; Cheng, L.; Wang, C. Removal of Sr^{2+} ions by a high capacity indium sulfide exchanger containing permeable layers with large pores. *2020*, *59*, 13822-13826.
- (27) Cahill, C. L.; Parise, J. B. On the formation of framework indium sulfides. *J. Chem. Soc., Dalton Trans.* **2000**, 1475-1482.
- (28) Cahill, C. L.; Younghhee, Ko.; Parise, J. B. A novel 3-dimensional open framework sulfide based upon the $[\text{In}_{10}\text{S}_{20}]^{10-}$ Supertetrahedron: DMA-InS-SB1. *Chem. Mater.* **1998**, *10*, 19-21.
- (29) Li, H.; Laine, A.; O'Keeffe, M.; Yaghi, O. M. Supertetrahedral Sulfide Crystals with Giant Cavities and Channels. *Science* **1999**, *283*, 1145–1147.
- (30) Li, H.; Eddaoudi, M.; Laine, A.; O'Keeffe, M.; Yaghi, O. M. Noninterpenetrating indium sulfide supertetrahedral cristobalite framework. *J. Am. Chem. Soc.* **1999**, *121*, 6096-6097.
- (31) Wang, C.; Bu, X.-H.; Zheng, N.-F.; Feng, P.-Y. A 3D open-framework indium telluride and its selenide and sulfide analogues. *Angew. Chem. Int. Ed.* **2002**, *41*, 1959-1961.
- (32) Wang, C.; Bu, X.-H.; Zheng, N.-F.; Feng, P.-Y. Nanocluster with one missing core atom: a three-dimensional hybrid superlattice built from dual-sized supertetrahedral clusters. *J. Am. Chem. Soc.* **2002**, *124*, 10268-10269.
- (33) Su, Z.; Li, X.-L.; Lan, Y.; Wen, J.-T.; Jin, G.-M.; Xie, J.-L.; Zheng, C.; Jin, J.; Li, S.-J. A new chalcogenide ion-exchanging material: porous indium sulfide built from condensed $[\text{In}_{10}\text{S}_{18}]^{6-}$ T_3 supertetrahedral clusters. *Mater. Lett.* **2008**, *62*, 2802-2805.
- (34) Zhang, M.-H.; Zhu, Q.-Y.; Bian, G.-Q.; Lei, Z.-X.; Jiang, J.-B.; Dai, J. Two related porous

- thioindates with T₃ clusters, (H₂DAH)₃In₁₀S₁₈·6H₂O and (H₂DAH)₃In₁₀S₁₈. *Z. Anorg. Allg. Chem.* **2010**, 636, 1137-1141.
- (35) Wu, T.; Khazhakyan, R.; Wang, L.; Bu, X.-H.; Zheng, S.-T.; Chau, V.; Feng, P.-Y. Three-dimensional covalent Co-assembly between inorganic supertetrahedral clusters and imidazolates. *Angew. Chem. Int. Ed.* **2011**, 50, 2536-2539.
- (36) Kanatzidis, M. G.; Dhingra, S. Soluble poly chalcogenide chemistry of indium: synthesis and characterization of [In₂Se₂₁]⁴⁻, the first indium polyselenide. *Inorg. Chem.* **1989**, 28, 2024-2026.
- (37) Dhingra, S. S.; Kanatzidis, M. G. Polyselenide chemistry of indium and thallium in dimethylformamide, acetonitrile, and water. syntheses, structures, and properties of the new complexes [In₂(Se₄)₂(Se₅)]⁴⁻, [In₂Se₂(Se₄)₂]²⁻, and [In₃Se₃(Se₄)₃]³⁻. *Inorg. Chem.* **1993**, 32, 1350-1362.
- (38) Yao, H.-G.; Ji, M.; S.-H. Ji; An, Y.-L. Solvothermal synthesis and characterization of one-dimensional indium polyselenides with transition metal complexes. *Z. Anorg. Allg. Chem.* **2012**, 638, 638-687.
- (39) Sarah, J. E.; Anthony, V. P.; Vaqueiro, P. Hydrothermal synthesis of [C₆H₁₆N₂][In₂Se₃(Se₂)]: A new one-dimensional indium selenide. *J. Solid State Chem.* **2011**, 184, 1800-1804.
- (40) Du, K.-Z.; Hu, W.-B.; Hu, B.; Guan, X.-F.; Huang, X.-Y. Synthesis, characterization, and anomalous dielectric and conductivity performance of one-dimensional (bdaH)InSe₂ (bha = 1,4-butanediamine). *Mater. Res. Bull.* **2011**, 46, 1969-1974.
- (41) Zhang, X.; Lei, Z.-X.; Luo, W.; Mu, W.-Q.; Zhang, X.; Zhu, Q.-Y.; Dai, J. 1-D Selenidoindates {[In₂Se₅]}_∞ directed by chiral metal complex cations of 1,10-Phenanthroline. *Inorg. Chem.* **2011**, 50, 10872-10877.
- (42) Ewinga, S. J.; Vaqueiro, P. Structural complexity in indium selenides prepared using bicyclic amine as structure-directing agents. *Dalton. Trans.* **2015**, 44, 1592-1600.
- (43) Ewinga, S. J.; Anthony, V. P.; Vaqueiro, P. [C₇H₁₀N][In₃Se₅]: A layered selenide with two indium coordination environments. *Inorg. Chem.* **2012**, 51, 7404-7409.
- (44) Xue, C.-H.; Lin, J.; Yang, H.-J.; Wang, W.; Wang, X.; Hu, D.-D.; Wu, T. Supertetrahedral clusters-based In-Se open frameworks with unique polyselenide ion as linker. *Cryst. Growth. Des.* **2018**, 18, 2690-2693.
- (45) Xue, C.-H.; Hu, D.-D.; Zhang, Y.-Y.; Yang, H.-J.; Wang, X.; Wang, W.; Wu, T. Two unique crystalline semiconductor zeolite analogues based on indium selenide clusters. *Inorg. Chem.* **2017**, 56, 14763-14766.
- (46) Vaqueiro, P. A three-dimensional open-framework indium selenide: [C₇H₁₀N][In₉Se₁₄]. *Inorg. Chem.* **2008**, 47, 20-22.
- (47) Wang, C.; Bu, X.-H.; Zheng, N.-F.; Feng, P.-Y. Indium selenide superlattices from (In₁₀Se₁₈)⁶⁻ supertetrahedral clusters. *Chem. Commun.* **2002**, 1344-1345.
- (48) Lin, J.; Dong, Y.-Z.; Zhang, Q.; Hu, D.-D.; Li, N.; Wang, L.; Liu, Y.; Wu, T. Interrupted chalcogenide-based zeolite-analogue semiconductor: atomically precise doping for tunable electro-/photoelectrochemical properties. *Angew. Chem. Int. Ed.* **2015**, 54, 5103-5107.
- (49) Xue, C.-Z.; Zhang, L.; Wang, X.-L.; Wang, X.; Zhang, J.-X.; Wu, T. Highly open chalcogenide frameworks built from unusual defective supertetrahedral clusters. *Dalton. Trans.* **2019**, 48, 10799-10803.
- (50) Manos, M. J.; Malliakas, C. D.; Kanatzidis, M. G. Heavy-metal-ion capture, ion exchange, and, exceptional acid stability of the open-framework chalcogenide (NH₄)₄In₁₂Se₂₀. *Chem. Eur. J.* **2007**,

13, 51-58.