## Designing Molecular Electrides from Defective Unit Cells of Cubic Alkaline Earth Oxides. Supporting Information

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Figure S1. HOMOs of studied structures. Contour value is 0.043.



Figure S2. LOL isosurfaces of studied structures. Isovalue = 0.5. (PBE0-DKH/aug-cc-pVTZ-DK).

Table S1. GM of  $Be_4O_3$ 4 2.141356000 0.844043000 0.979691000

| 4 -0.496011000<br>4 1.700912000<br>4 -0.110812000<br>8 1.105869000<br>8 0.369641000<br>8 2.726936000           | 2.611644000<br>3.102590000<br>0.636626000<br>-0.099888000<br>3.675017000<br>2.099801000                   | 0.860034000<br>0.459846000<br>1.314900000<br>1.350026000<br>0.480716000<br>0.586576000                   |
|--|---|--|
| $\begin{array}{r llllllllllllllllllllllllllllllllllll$   | 0.837367000<br>0.837367000<br>-1.674734000<br>0.000000000<br>1.642065000<br>-0.821032000<br>-0.821032000  | -0.660272000<br>-0.660272000<br>-0.660272000<br>1.372196000<br>0.304310000<br>0.304310000<br>0.304310000 |
| $\begin{array}{llllllllllllllllllllllllllllllllllll$   | -1.389034000<br>1.898855000<br>-0.509797000<br>-0.000111000<br>1.197235000<br>-1.636871000<br>0.439851000 | -0.713268000<br>-0.713243000<br>-0.713090000<br>1.711767000<br>0.356601000<br>0.356695000                |
| $\begin{array}{r llllllllllllllllllllllllllllllllllll$   | 1.063593000<br>1.063593000<br>-2.127186000<br>0.000000000<br>1.747558000<br>-0.873779000<br>-0.873779000  | -0.721511000<br>-0.721511000<br>-0.721511000<br>1.908382000<br>0.405573000<br>0.405573000<br>0.405573000 |
| Table S5. GM Ba4O356-0.00097300056-2.280799000560.884701000561.39705700081.7574430008-1.0758960008-0.681448000 | -0.002222000<br>-0.294909000<br>2.123714000<br>-1.826465000<br>0.227964000<br>1.407499000<br>-1.636288000 | 2.081775000<br>-0.753247000<br>-0.750538000<br>-0.753246000<br>0.410061000<br>0.409809000<br>0.406917000 |

Table S6. At PBE0-DKH/aug-cc-pVTZ-DK level of theory.

| System                         | <α>, a.u. | β <sub>tot</sub> , a.u. | Excitation energy, eV |
|--------------------------------|-----------|-------------------------|-----------------------|
| Be <sub>4</sub> O <sub>3</sub> | 66.79     | 524.61                  | 5.25                  |
| Mg <sub>4</sub> O <sub>3</sub> | 243.86    | 5689.96                 | 2.06                  |
| Ca4O3                          | 508.11    | 37703.84                | 1.42                  |
| Sr4O3                          | 634.175   | 46218.29                | 1.31                  |
| Ba4O3                          | 837.99    | 65747.73                | 1.18                  |

Table S7. Dissociation energy of  $Ca_4O_3$ . (PBE0/def2-TZVP + ZPE).

| Fragments   | Energy, kcal/mol |
|---|------------------|
| Ca <sub>2</sub> O <sub>2</sub> +Ca <sub>2</sub> O | 378088.9         |

| 3xCaO+Ca                            | 298.5    |
|-------------------------------------|----------|
| Ca <sub>3</sub> O <sub>3</sub> +Ca  | 81.9     |
| Ca <sub>2</sub> O+2xCaO             | 378206.7 |
| Ca <sub>3</sub> O <sub>2</sub> +CaO | 123.8    |
| 4Ca+3O                              | 789.3    |