Osmolytes can Destabilize Proteins in Cells by Modulating Electrostatics and Quinary Interactions

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**Supporting Information** 

 $^{\perp}$ These two authors contributed equally to the work.

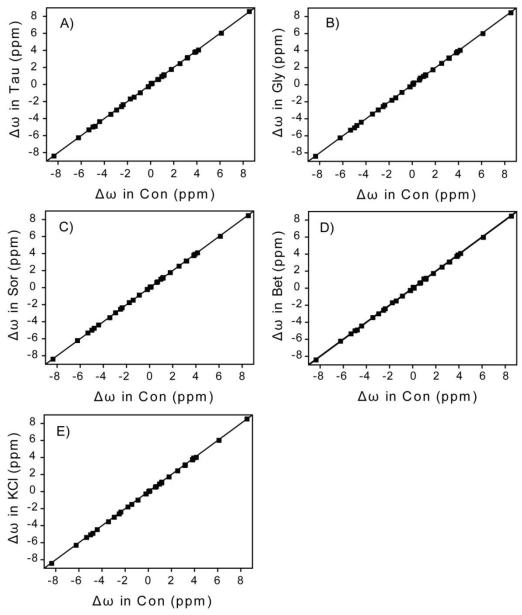


Figure S1. Correlation between  $\Delta \omega$ , the chemical shift difference between unfold state and fold state *in vitro* in the absence (Con) or presence of different small molecules.

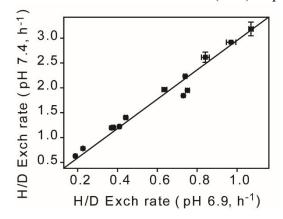
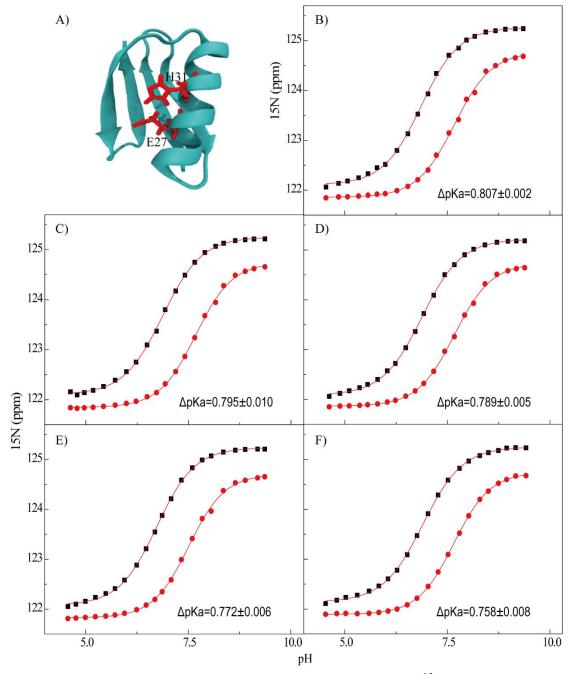


Figure S2. Correlation between the H/D exchange rates of 13 slowest exchanged



amides of MutX in two pH conditions. The best fitted line is y = 2.94x.

Figure S3. (A) Salt bridge H31-E27 in the structure of GB3. <sup>15</sup>N chemical shift of H31 of H31-E27 (red) and H31-Q27 (black) at 298 K and different pHs without osmolyte (B), or in the presence of 500 mM sorbitol (C), glycerol (D), betaine (E) and taurine (F). The chemical shift was fitted to the Henderson-Hasselbalch equation to yield corresponding p*K*a values.

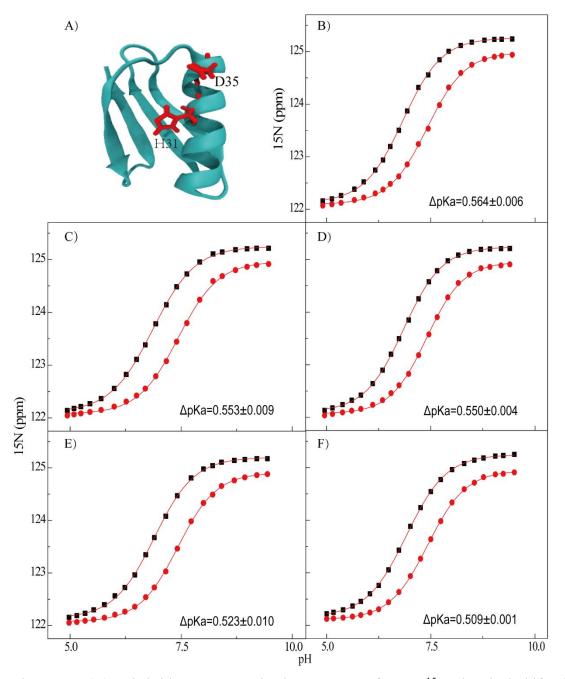


Figure S4. (A) Salt bridge H31-D35 in the structure of GB3. <sup>15</sup>N chemical shift of H31 of H31-D35 (red) and H31-N35 (black) at 298 K and different pHs without osmolyte (B), or in the presence of 500 mM sorbitol (C), glycerol (D), betaine (E) and taurine (F). The chemical shift was fitted to the Henderson-Hasselbalch equation to yield corresponding  $pK_a$  values.

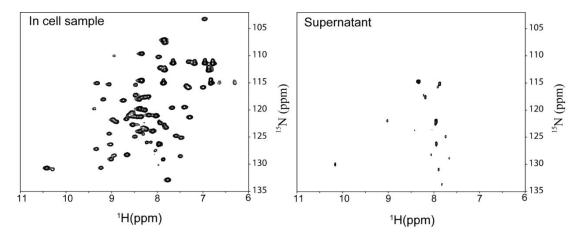


Figure S5. <sup>1</sup>H-<sup>15</sup>N HSQC spectra of the GB3 mutant MutX in *E. coli* cells and in supernatant after 3.5 h of NMR CPMG measurement. The in-cell sample was dissolved in 40 mM bis–tris propane, and 40 mM HEPES, pH 7.8.

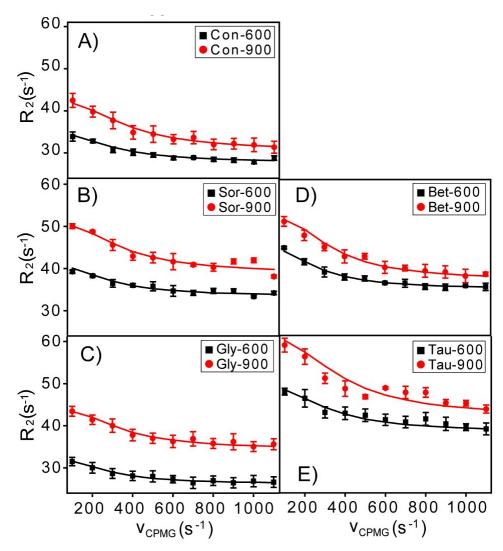


Figure S6. Global fitting of <sup>15</sup>N CPMG profiles for A48 at 600 and 900 MHz magnetic fields in cells with or without osmolytes.

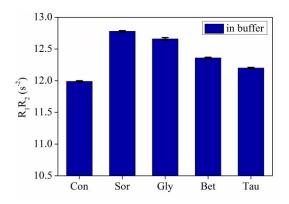


Figure S7. Product of average  $R_1R_2$  of MutX in *in vitro* in the absence or presence of 400 mM osmolytes.