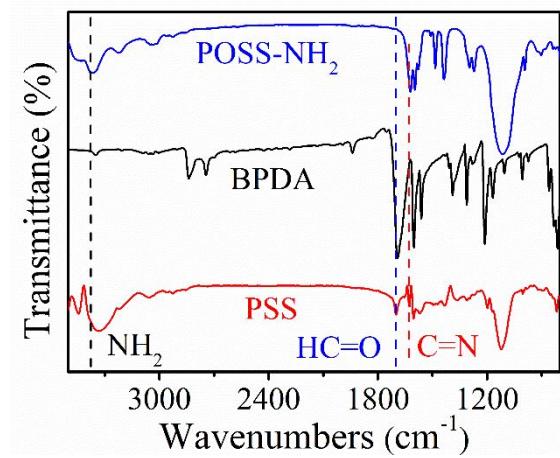


# **Supporting Information**

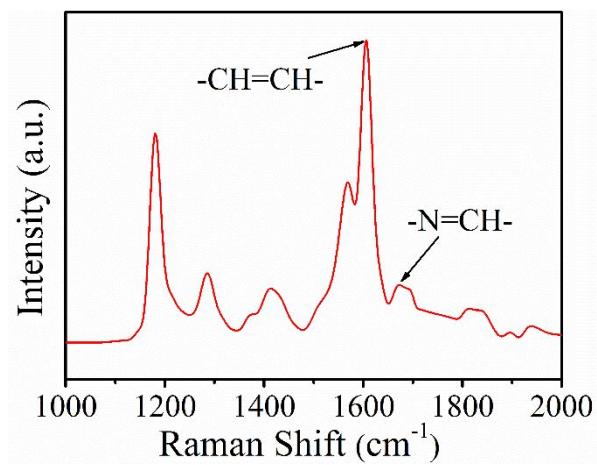
## **Molecular Level Design of Nitrogen–Doped Well–Defined Microporous Carbon Spheres for Selective Adsorption and Electrocatalysis**

Zirun Chen, Shaohong Liu, Junlong Huang, Wen Huang, Luyi Chen, Yin Cui, Yang Du, and Ruowen Fu\*

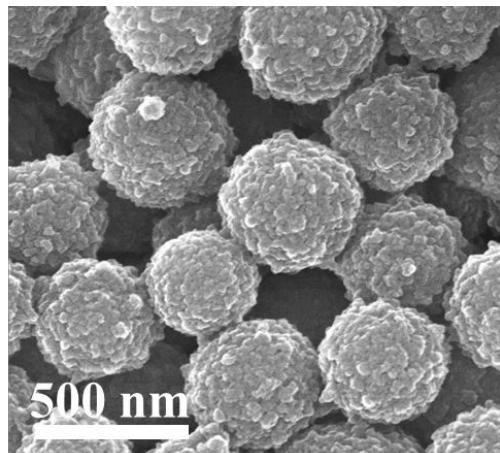
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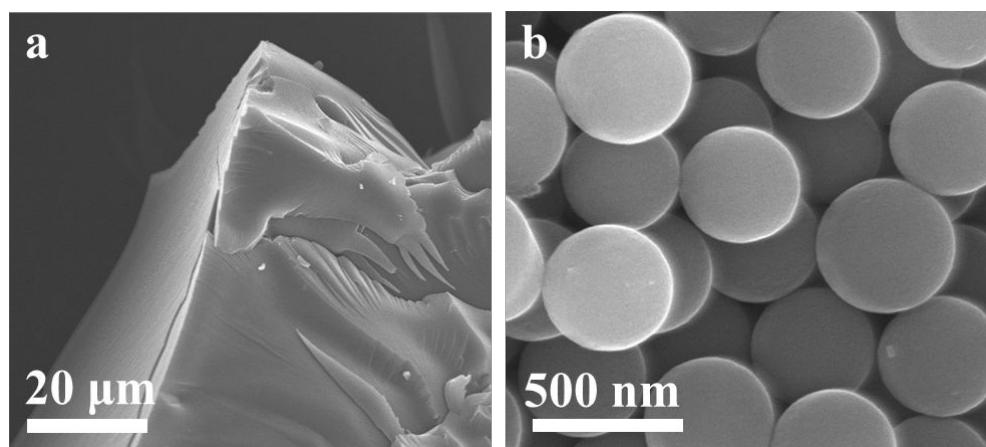
**Figure S1.** FT-IR spectra of PSS.



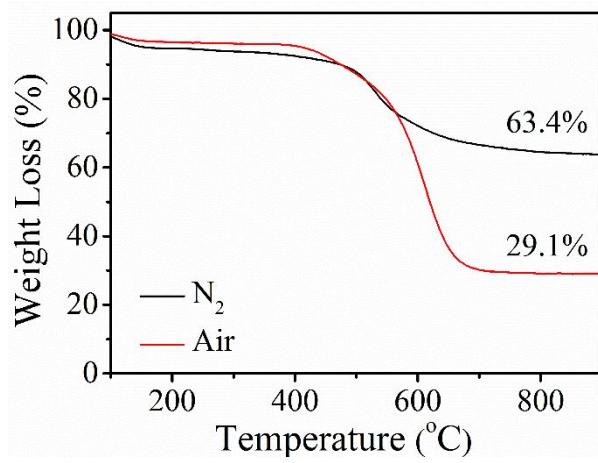
**Figure S2.** Raman spectra of PSS.



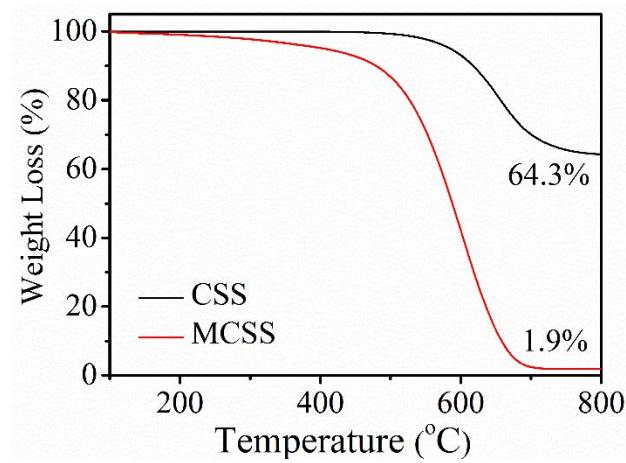
**Figure S3.** SEM image of PSS.



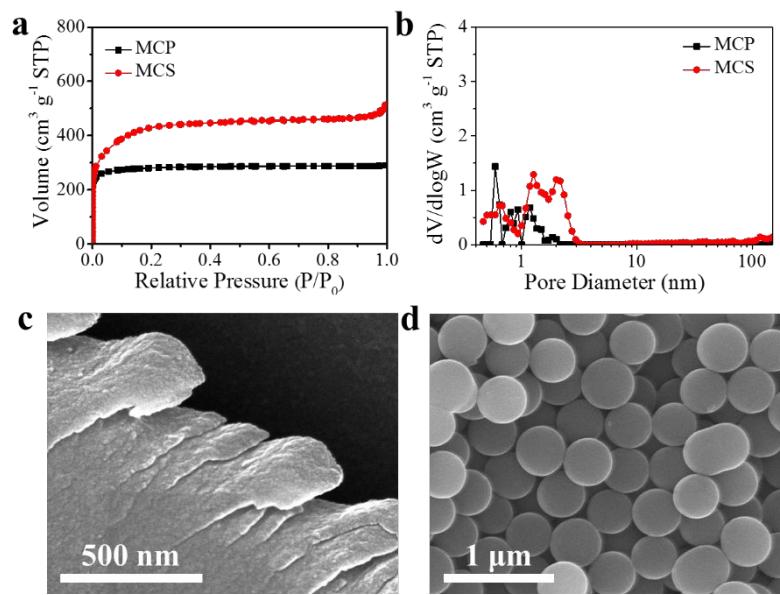
**Figure S4.** SEM images of (a) polymer particles and (b) polymer spheres.



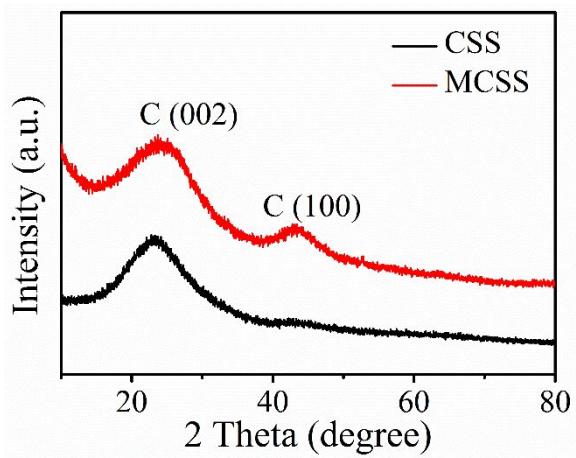
**Figure S5.** TGA curves of PSS.



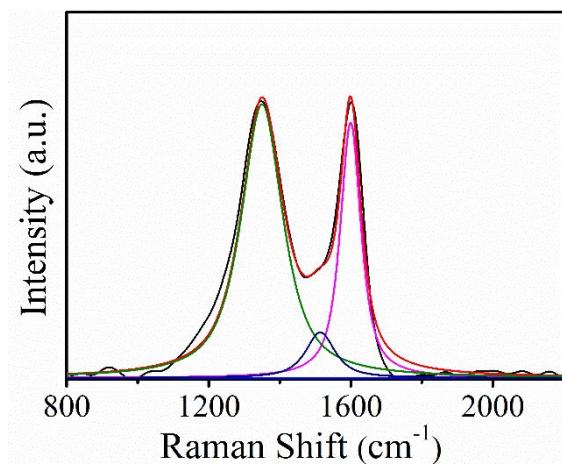
**Figure S6.** TGA curves of CSS and MCSS.



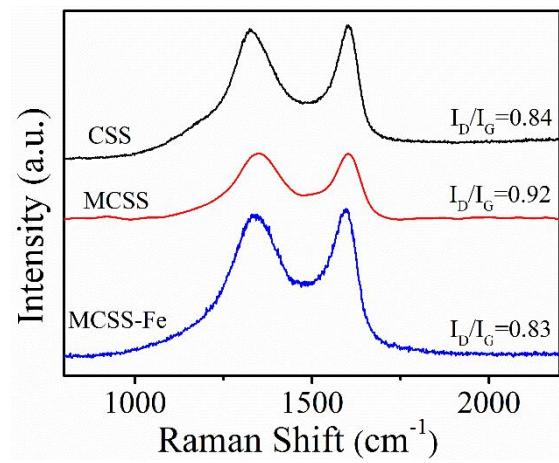
**Figure S7.** (a, b)  $\text{N}_2$  adsorption-desorption isotherms and Pore size distribution of MCP and MCS. SEM images of (c) MCP and (d) MCS.



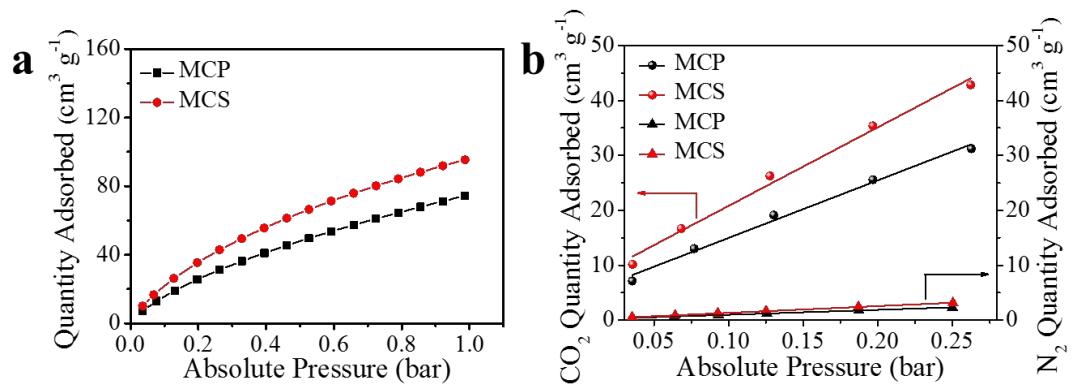
**Figure S8.** XRD patterns of CSS and MCSS.



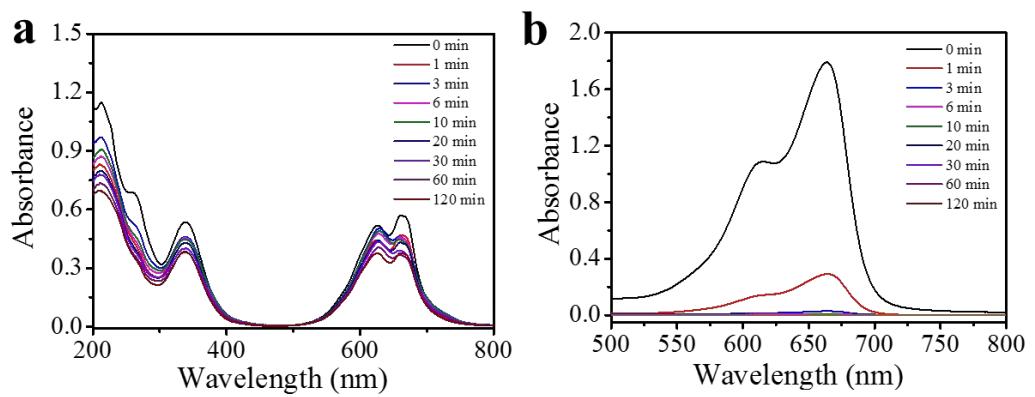
**Figure S9.** Raman spectra of MCSS.



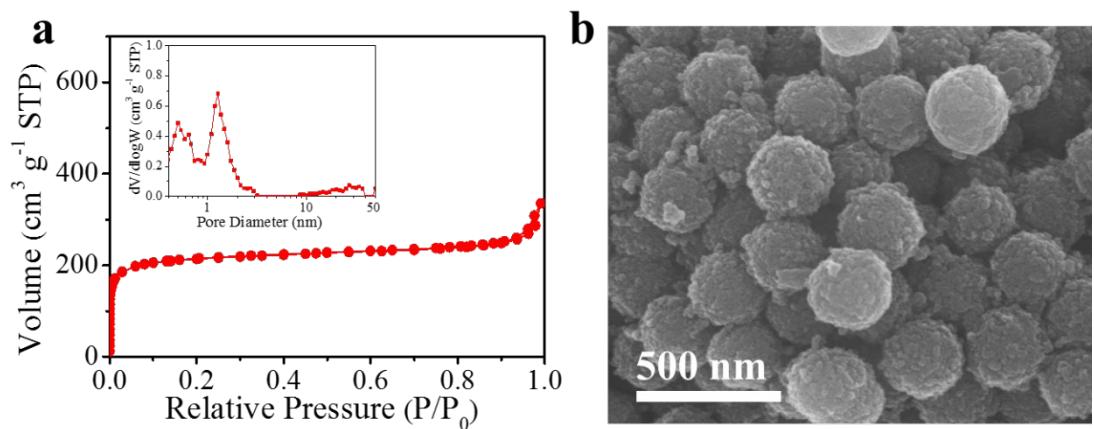
**Figure S10.** Raman spectra of MCSS.



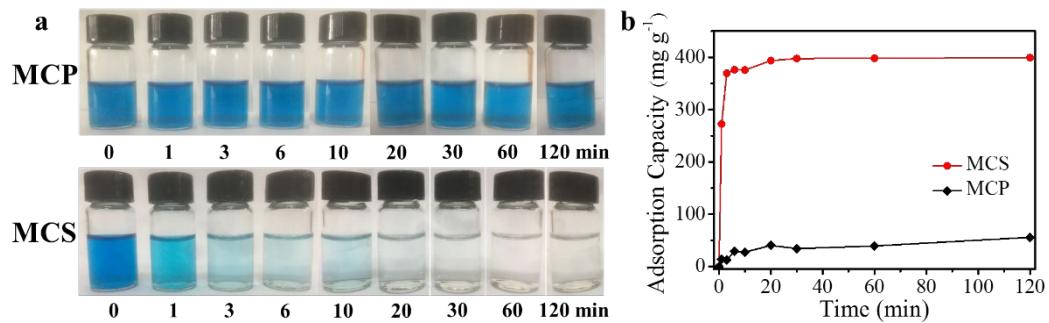
**Figure S11.** (a) CO<sub>2</sub> adsorption isotherms of MCP and MCS at 273 K, and (b) Initial N<sub>2</sub> and CO<sub>2</sub> uptake slopes of MCP and MCS used for selectivity calculations (CO<sub>2</sub> over N<sub>2</sub>) in a pressure window of 0-0.26 bar.



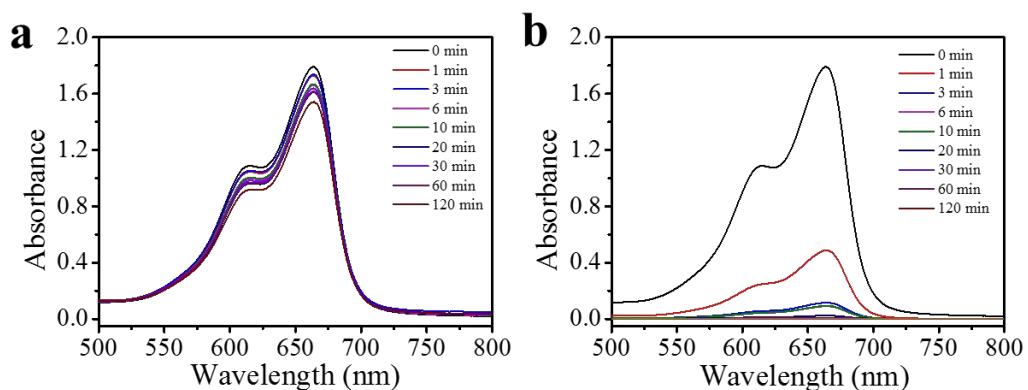
**Figure S12.** UV-Vis spectra of (a) RB21 and (b) MB for MCSS.



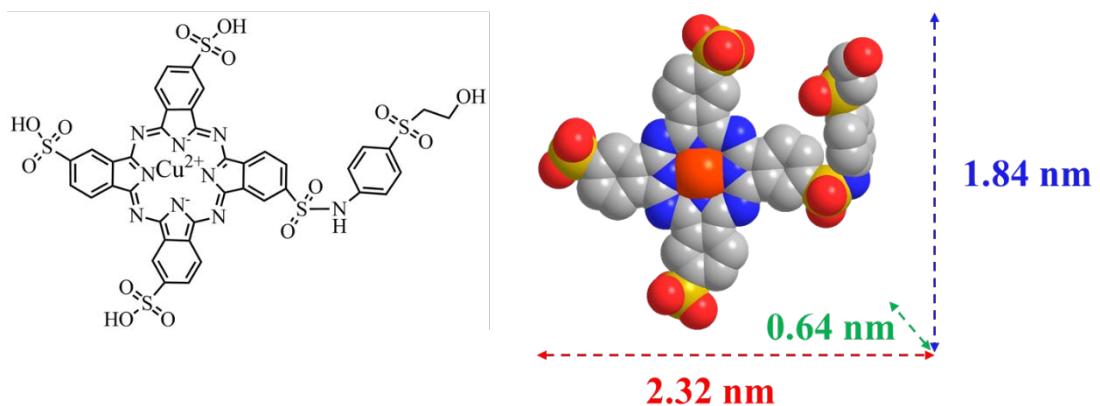
**Figure S13.** (a)  $N_2$  adsorption-desorption isotherms and corresponding Pore size distribution of MCSS after adsorption of MB. (b) SEM image of MCSS after adsorption of MB.



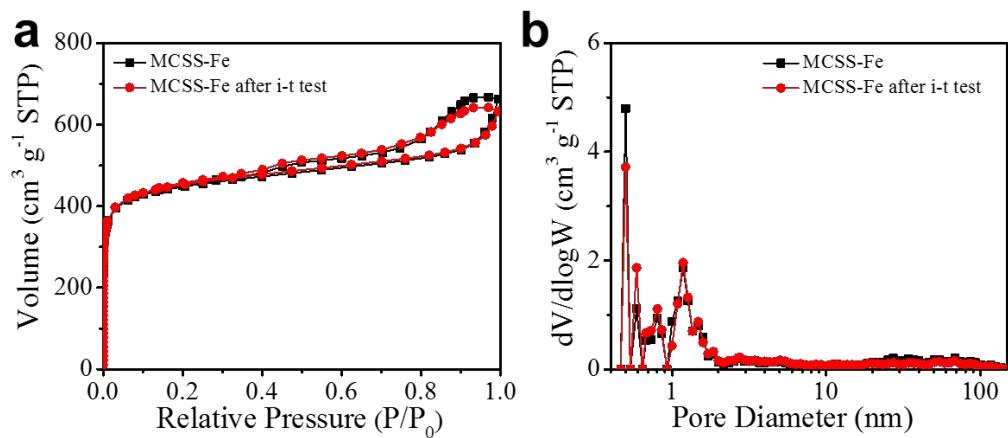
**Figure S14.** (a) Photograph of MB solutions with various times. (b) Time-dependent adsorption of MB by MCP and MCS.



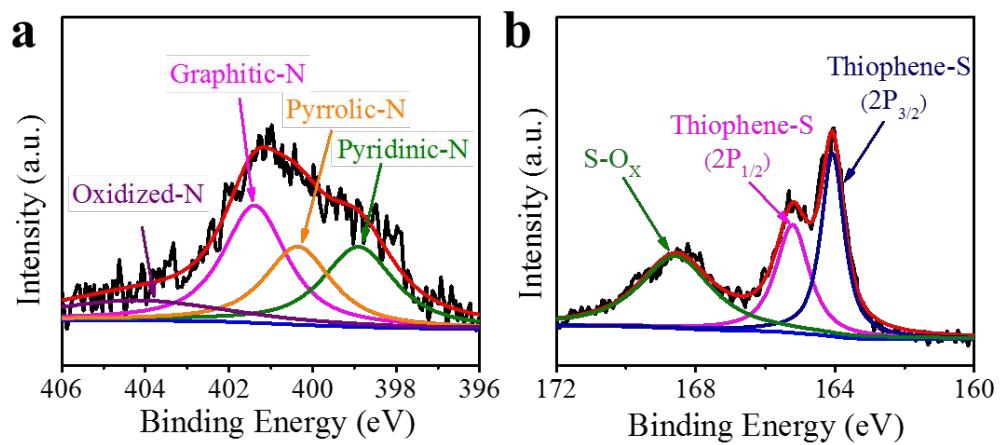
**Figure S15.** UV-Vis spectra of MB for (a) MCP and (b) MCS.



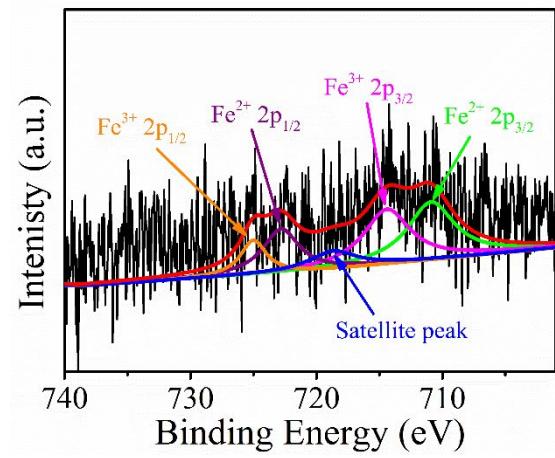
**Figure S16.** Molecule structure and 3D model of RB21. For 3D model, the H atoms are hidden for clarity.



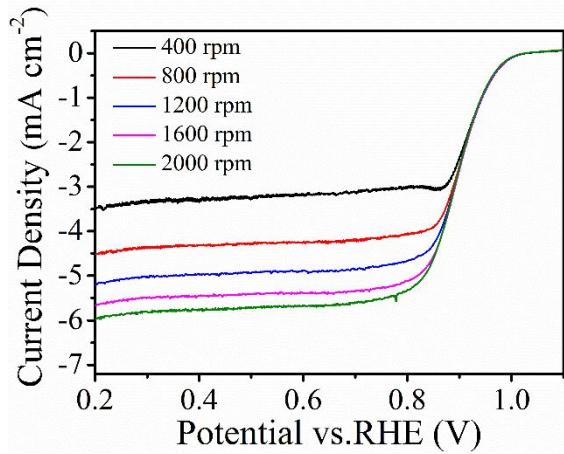
**Figure S17.** (a) N<sub>2</sub> adsorption-desorption isotherms of MCSS-Fe and MCSS-Fe after i-t test. (b) Pore size distribution of MCSS-Fe and MCSS-Fe after i-t test.



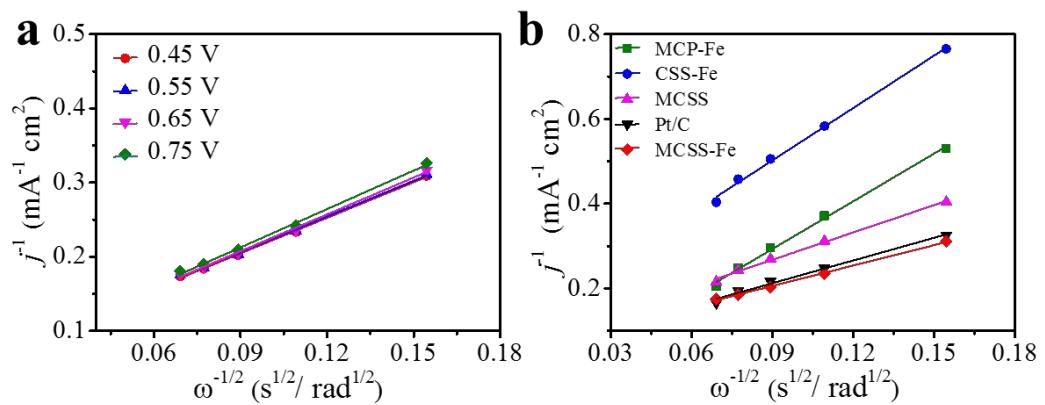
**Figure S18.** High resolution XPS spectra (a) N 1s and (b) S 2p of MCSS-Fe.

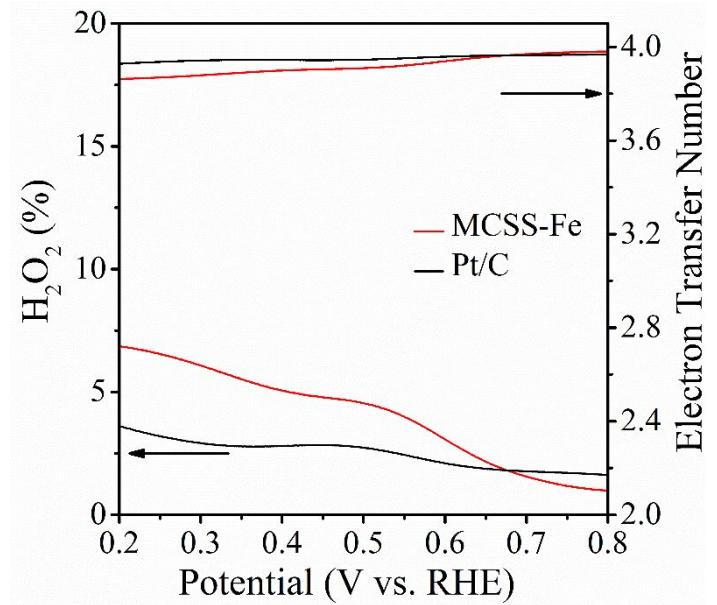


**Figure S19.** High resolution XPS spectra Fe 2p of MCSS-Fe.

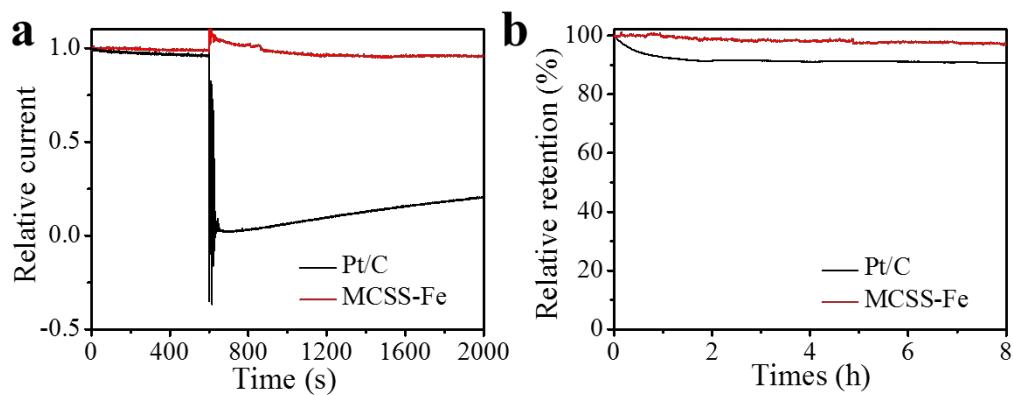


**Figure S20.** LSV curves of MCSS-Fe at different rotating speeds.

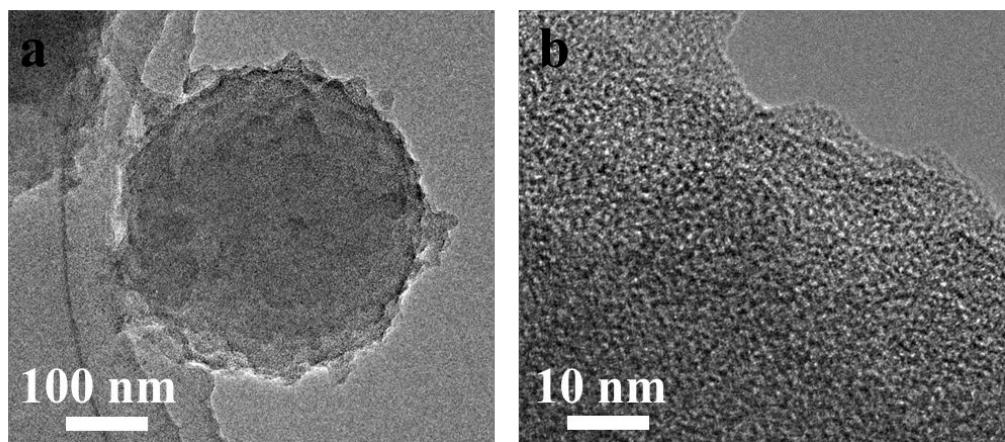




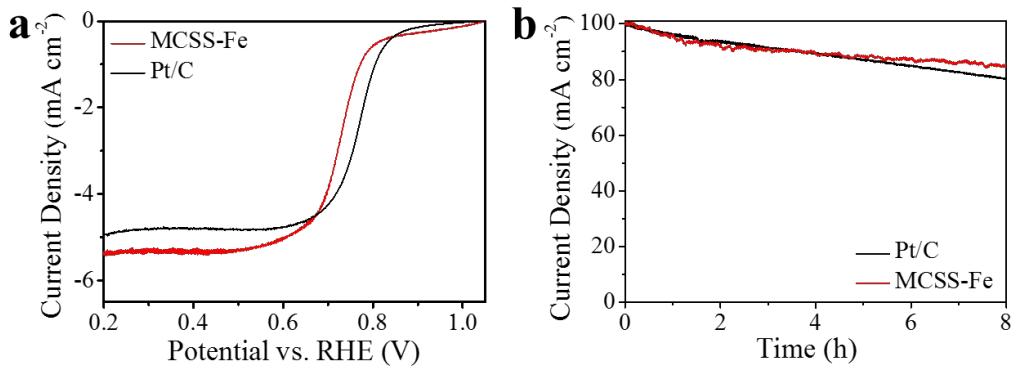
**Figure S22.** Peroxide yield and electron-transfer number of MCSS-Fe and Pt/C catalyst at various potentials based on the RRDE data.



**Figure S23.** (a) Chronoamperometric curves of MCSS-Fe and Pt/C at 0.565 V followed by introducing methanol and (b) Chronoamperometric curves of MCSS-Fe and Pt/C at 0.57 V in O<sub>2</sub>-saturated 0.1 M KOH at 600 rpm.



**Figure S24.** (a,b) HR-TEM images of MCSS-Fe after i-t test.



**Figure S25.** (a) ORR polarization curves of Pt/C, and MCSS-Fe samples in 0.5 M  $\text{H}_2\text{SO}_4$  and (b) Chronoamperometric curves of MCSS-Fe and Pt/C at 0.4 V in  $\text{O}_2$ -saturated 0.5 M  $\text{H}_2\text{SO}_4$  at 600 rpm.

**Table S1.** Element content of samples.

Samples	C content (at%)	N content (at%)	Si content (at%)	Fe content (at%)	S content (at%)	O content (at%)
CSS	36.63	1.11	18.86	0	0	43.40
MCSS	85.87	2.35	1.86	0	0	9.93
MCSS-Fe	85.07	2.84	2.04	0.37	2.04	8.38
MCP-Fe	91.30	2.91	0.72	0.2	0.71	4.17

**Table S2.** Comparison in the BET surface area and pore structure of MCSS and carbon spheres obtained through the silica templates.

Samples	$S_{\text{BET}}$ ( $\text{m}^2 \text{ g}^{-1}$ )	Pore size (nm)	Silica template	Reference
<b>MCSS</b>	<b>2036</b>	<b>1.2</b>	<b>Nanosilica domain</b>	<b>This work</b>
O-MCS	1186	5.9	$\text{SiO}_2$	1
MCNs	1117	7	$\text{SiO}_2$	2
HMCNs	646	3.8	$\text{SiO}_2$	3
MC-MS-1	526	8.1	Silica inverse opal	4
N-HPCS	1137	2.5	$\text{SiO}_2$	5
HMCSSs	771	2.1	$\text{SiO}_2$	6
egg-CMS	970	4	$\text{SiO}_2$	7
MCS	648	3	$\text{SiO}_2$	8
NHC	1538	3.2	$\text{SiO}_2$	9
HC-C18	1620	4.3	Organosilica	10
PHCSSs	1520	1-25	$\text{SiO}_2$	11
P-N-CS	95	>7	$\text{SiO}_2$	12

**Table S3.** Comparison of ORR catalytic activity between MCSS-Fe and other Fe-based carbonous electrocatalysts in 0.1 M KOH solution.

Samples	Half-wave Potential (V vs.RHE)	References
<b>MCSS-Fe</b>	<b>0.89</b>	<b>This work</b>
FeP/Fe <sub>2</sub> O <sub>3</sub> @NPCA	0.81	13
Fe <sub>20</sub> @N/HCSs	0.85	14
Ni-N <sub>4</sub> /GHSs/Fe-N <sub>4</sub>	0.83	15
CAN-Pc(Fe/Co)	0.85	16
Fe-N-C HNSs	0.87	17
Fe2-Z8-C	0.87	18
PFA-Fe10-900-ALP	0.86	19
S,N-Fe/N/C-CNT	0.85	20
Fe/SNC	0.86	21
Fe <sub>1</sub> -HNC-500-850	0.84	22
Fe/NS/C-g-C <sub>3</sub> N <sub>4</sub> /TPTZ-1000	0.87	23
Fe-CNT/PC	0.88	24
Fe-SNC@900	0.83	25
p-Fe-N-CNFs	0.82	26

FeS/Fe <sub>3</sub> C@N-S-C-800	0.87	27
OM-NCNF-FeN <sub>x</sub>	0.84	28

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