# Incorporation of Basic $\alpha$-Hydroxy Acid Residues 

# into Primitive Polyester Microdroplets for RNA 

## Segregation

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## Supplementary Methods

Matrix-assisted laser desorption ionization time-of-flight mass spectrometry (MALDI-TOF-
MS) peak identification. Monoisotopic peaks were identified by isolating the highest intensity peak in an isotope envelope that corresponded to a polymer product using a peak list generated from mMass ${ }^{1-3}$ (Open Source Software, Prague, Czech Republic) after smoothing (SavitskyGolay, window size $=0.3 \mathrm{~m} / \mathrm{z}, 2$ cycles), baselining ( 100 precision and 0 relative offset), and thresholding above an intensity of 5000, in that order. Major peaks were isolated by applying the following parameters: $\mathrm{S} / \mathrm{N}>10$, absolute peak intensity $>5000$, picking height at $100 \%$, while applying baselining, smoothing, and deisotoping (maximum charge $=1$, isotope mass tolerance $=$ $0.1 \mathrm{~m} / \mathrm{z}$, isotope intensity tolerance $=50 \%$, isotope mass shift $=0.0$, remove isotopes, remove unknown, label envelope tool $=1$ st selected, envelope intensity $=$ envelope maximum). Analytical settings, matrix, and thresholding parameters were identical for each experiment. Although other adducts were also observed, we list only the following adducts: $\mathrm{M}+\mathrm{H}, \mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic), $\mathrm{M}+\mathrm{Na}-$ $\mathrm{H}_{2} \mathrm{O}$ (cyclic), and $\mathrm{M}+\mathrm{Na}+$ ethanolamine (" $\mathrm{M}+\mathrm{H} / \mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species), which were the major adducts observed, and also ones that most clearly showed mass ladders. Mass accuracy in ppm was calculated by comparing the observed mass with the calculated mass for major peaks corresponding to polymerization products. Detailed peak lists are presented in Tables S1-S12.

Electrospray Ionization (ESI)-MS. Polymerized 4a2h homopolymer and 1:1 4a2h: $\alpha$ HA ( $\alpha$ hydroxy acid) samples (of GA, LA, PA, SA, and MA) were first rehydrated in $500 \mu \mathrm{~L}$ water. Then, a 1 mL sample of each polyester diluted 1000 -fold (in HPLC-grade water) was subjected to positive-ion mode ESI-MS acquisition, on a Xevo TQ-S quadrupole time-of-flight mass spectrometer (Waters, Milford, Massachusetts, USA) with direct infusion. The following acquisition parameters: sample infusion flow rate of $10 \mu \mathrm{~L} \mathrm{~min}^{-1}$, an ion source temperature of $100^{\circ} \mathrm{C}$, a desolvation gas temperature of $250^{\circ} \mathrm{C}$, a desolvation gas flow rate of $500 \mathrm{Lh}^{-1}$, a cone voltage of 10 V , a cone gas flow rate of $100 \mathrm{~L} \mathrm{~h}^{-1}$, and a capillary voltage of 3.0 kV . A water blank was injected between all samples. Peaks were identified using the information from the MALDI-TOF-MS peaklists (Tables S1-S12). Analytical settings, matrix, and thresholding parameters were identical among different runs of experiments. Although we also observed other adducts,
including some of those listed in Tables S1-S12, we list only the $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic or dehydrated) adducts for ease of visualization.

Fluorescence Microscopy Analysis. For Figs. 8, S16-S25, and S28, fluorescence acquisition (excitation illumination strength and exposure time) and image processing parameters (brightness/contrast) were constant down each column. This was important to ensure the ability to directly compare fluorescence intensity values between the relevant images within each figure. Kymographs were acquired with Fiji across the same line, which was consistent down each row for each relevant figure. For Fig. S41, the powdery poly(4a2h) synthesis product was deposited directly onto a glass slide and observed by microscope.

Hammerhead Ribozyme Reaction. The hammerhead ribozyme reaction was performed by preparing a $5 \quad \mu \mathrm{~L}$ sample containing $4 \quad \mu \mathrm{M} \quad \mathrm{HH} 1 \quad$ (5'-FAM-CGCGCCGAAACACCGUGUCUCGAGC-3'; FAM = 6-carboxyfluorescein), $6 \mu \mathrm{M} \mathrm{HH2} \mathrm{(5'-}$ GGCUCGACUGAUGAGGCGCG-3'), 200 mM MES pH 5.7 , and $100 \mathrm{mM} \mathrm{MgCl}_{2}$ in the presence or absence of polyester components. This was done by first mixing $3 \mu \mathrm{~L}$ of the appropriate stock polyester sample (after rehydration with $500 \mu \mathrm{~L} 4: 1$ water:acetonitrile; in this particular study, all samples were dissolved in $4: 1$ water:acetonitrile even if they did not require acetonitrile for rehydration, resulting in a $12 \%$ aqueous acetonitrile solvent) with $1.5 \mu \mathrm{~L}$ of the other components not including magnesium ( $0.2 \mu \mathrm{~L}$ HH1 ( $100 \mu \mathrm{M}$ stock), $0.3 \mu \mathrm{~L} \mathrm{HH} 2$ ( $100 \mu \mathrm{M}$ stock), and $1 \mu \mathrm{~L}$ MES pH 5.7 (1 M stock)). Otherwise in the absence of polyester samples, the reaction occurred in pure water. Then, $0.5 \mu \mathrm{~L}$ of a $1 \mathrm{M} \mathrm{MgCl}_{2}$ stock solution (final $\mathrm{MgCl}_{2}$ concentration of 100 mM ) was added to initiate the reaction. All buffers used in this section were either made with RNAsefree water and subsequently filtered, or were purchased as RNAse free (except the sample loading buffer purchased from Thermo-Fisher). The reactions were stopped after 24 hours by adding 200 $\mu \mathrm{L}$ of $4: 1$ ethanol:RNAse-free water, $50 \mu \mathrm{~g} / \mathrm{mL}$ glycogen, and 125 mM ammonium acetate solution, followed by 15 minutes incubation at $-20^{\circ} \mathrm{C}$ for precipitation. After incubation, samples were subjected to centrifugation at $21,500 \mathrm{~g}$ for 15 minutes $\left(4^{\circ} \mathrm{C}\right)$ in a himac CT 15 RE benchtop centrifuge (Hitachi, Chiyoda-ku, Tokyo, Japan), followed by supernatant removal by pipet. The remaining pellet was allowed to air dry for several minutes, followed by dissolution in $5 \mu \mathrm{~L}$ of 2 X Novex TBE-Urea Sample Buffer (Thermo-Fisher Scientific) by pipetting. $2.5 \mu \mathrm{~L}$ of each sample
was loaded onto a pre-cast 12-well $6 \%$ Novex TBE-Urea gel, followed by electrophoresis ( 150 V , 35 minutes) in a Novex gel running box (Thermo-Fisher Scientific). The gel was then imaged (Blue channel ( 460 nm ) on automatic mode) on an Amersham Imager 600 (GE Life Sciences, Shinjuku-ku, Tokyo, Japan) after separation from the plastic casing. Finally, the total gel band intensities were measured by FIJI, and the fraction of accumulated product was calculated by dividing the total intensity of the product fraction by the total intensity of both the product and remaining reactant fractions. Means and standard errors to the mean were calculated accordingly ( $n=4$ ).

## Figure S1



Positive ion-mode MALDI-TOF mass spectrum of the polymerization product ( $\Delta 101.11 \mathrm{Da}$ ) of complete drying of a 4 a 2 h solution at $80^{\circ} \mathrm{C}$ for one week. Indicated peaks are proton adducts. MALDI-TOF-MS was chosen rather than ESI-MS due to the required dissolution (and possible hydrolysis) of the sample upon dilution in the aqueous sample running solvent. The peak list is shown in Tables S1 and S2.

## Figure S2



Photograph of white, powdery product resulting from drying 4 a 2 h solution at $80^{\circ} \mathrm{C}$ for one week.
Figure S3


Positive ion mode MALDI-TOF mass spectrum of the SDHB matrix (9:1 (w/w) mixture of 2,5Dihydroxybenzoic (DHB) and 2-hydroxy-5-methoxybenzoic acid).

## Figure S4



1:1 4a2h:GA


1:1 4a2h:PA


1:1 4a2h:SA

1:1 4a2h:LA



1:1 4a2h:MA

Photographs of the polymerization products of drying of 1:14a2h:uncharged $\alpha \mathrm{HA}$ solutions and a solution of an equimolar combination of all six $\alpha \mathrm{HAs}$ at $80^{\circ} \mathrm{C}$ for one week (Fig. 1). All samples contained 500 mM total monomer before polymerization. The brown color could be the result of products formed through degradation of the starting materials followed by melanoidin formation akin to Maillard-type reaction products or intermediates ${ }^{4}$.

Figure S5


Photograph of the polymerization product of drying a solution of an equimolar combination of all six $\alpha$ HAs ( 500 mM total monomer before polymerization) at $80^{\circ} \mathrm{C}$ for one week.

## Figure S6



Zoomed-in positive ion mode MALDI-TOF mass spectra ( $400-600 \mathrm{~m} / \mathrm{z}$ ) of polymerization products resulting from drying of 1:1 (a) 4a2h:GA and (b) 4a2h:LA solutions ( 500 mM total monomer concentration) (Fig. 2). For ease of visualization, indicated peaks are only those which may be cyclic or dehydrated protonated species (red and black) and sodiated ethanolamine (Fig. S10) conjugated adducts (blue and black). Red double-sided arrows show mass difference indicative of addition of $4 \mathrm{a} 2 \mathrm{~h}(\Delta 101.11 \mathrm{Da})$; black arrows show mass difference indicative of addition of GA ( 458.01 Da ) or LA ( $\Delta 72.02 \mathrm{Da}$ ). Peak lists: Tables S3-S4 $(4 \mathrm{a} 2 \mathrm{~h} / \mathrm{GA})$ and Tables S5-S6 (4a2h/LA). For ease of interpretation, we have included a set of tables sorted by mass (Tables S3 and S5), as well as a set of tables sorted first by adduct and then by mass (Tables S4 and S6). The only homopolymers observed were pure 4a2h homopolymers; no non-4a2hcontaining homopolymers were observed. This could be potentially due to a preference of incorporation of 4 a 2 h into the polymer products, or the increased stability of poly(4a2h) products. While it could be theoretically possible that the 4 a 2 h oligomers are not polyesters, but rather linear or branched polymers assembled from the formation of an amide bond between the 4 a 2 h side chain amino group and the $\alpha$-hydroxy group, other similar studies did not observe such a phenomenon with 4a2h (although they observed such amide bond formation from similar polymerization of 6-amino-4-hydroxybutyric acid monomers) ${ }^{5}$. Thus, we believe that the homopolymer structures formed are likely to be polyesters, although it cannot be ruled out that poly(4a2h) peptides, depsipeptides, or even branched/hyperbranched structures were also present due to formation of amide bonds; a future study utilizing FTIR and/or NMR is planned.

## Figure S7



Zoomed-in positive ion mode MALDI-TOF mass spectra ( $400-600 \mathrm{~m} / \mathrm{z}$ ) of the polymerization products of complete drying of 1:1 (a) 4a2h:PA, (b) $4 \mathrm{a} 2 \mathrm{~h}: \mathrm{SA}$, and (c) $4 \mathrm{a} 2 \mathrm{~h}: \mathrm{MA}$ solutions at $80^{\circ} \mathrm{C}$ for one week ( 500 mM total monomer concentration) (Fig. 3). For ease of visualization, indicated peaks are only those which may be cyclic or dehydrated protonated species (red and black) and sodiated ethanolamine (Fig. S10) conjugated adducts (blue and black). Red double-sided arrows indicate a mass difference indicative of addition of $4 \mathrm{a} 2 \mathrm{~h}(\Delta 101.11 \mathrm{Da})$, while black arrows indicate a mass difference indicative of addition of PA $(\Delta 148.05 \mathrm{Da})$, SA $(\Delta 132.02 \mathrm{Da})$, or MA $(\Delta 114.07$ Da). Peak lists: Tables S7-S8 (4a2h/PA), S9-S10 (4a2h/SA), and S11-S12 (4a2h/MA).

## Figure S8



Positive ion of ESI mass spectra of polymerization products resulting from drying of (a) 1:1 4a2h:GA and (b) 4a2h:LA solutions at 500 mM total monomer concentration. Only product masses consistent with cyclic or dehydrated protonated adducts are annotated here for ease of visualization.

Figure S9


Positive ion of ESI mass spectra of polymerization products resulting from drying of (a) $1: 1$ 4a2h:PA, (b) 4a2h:SA, and (c) 4a2h:MA solutions at 500 mM total monomer concentration. For ease of visualization, indicated peaks are only those which may be cyclic or dehydrated protonated species.

## Figure S10



b)


Combined mass: 286.13787
One of the peak series detected in all of the mixed samples initially reduced to monomeric additions (of 4 a 2 h and the appropriate $\alpha \mathrm{HA}$ ) to a $\sim 286 \mathrm{Da}$ species. We speculate this compound to be a sodiated 4a2h oligoester dimer containing an ethanolamine conjugated to the free carboxyl group. a) Degradation of a cyclized 4a2h monomer lactam (3-hydroxypyrrolidin-2-one) ( $\sim 101 \mathrm{Da}$ detected previously in similar conditions ${ }^{5}$ ) into ethanolamine (and other various degradation products); this product could also have contributed to the brown coloration of the mixed sample products). b) The ethanolamine then could form an ester with the carboxyl group of a linear 4a2h ester dimer, resulting in a mass of $\sim 286 \mathrm{Da}$ after sodiation that matches that detected in all of the mixed sample product MALDI-TOF spectra. Co-existence of a peptide, via nucleophilic addition of the amine moiety of ethanolamine to the C-terminus of 4a2h dimer, with the identical mass is possible. All exact masses in the figure are represented in Da .

Figure S11


100\% 4a2h


1:1 4a2h:GA


1:1 4a2h:SA


1:1 4a2h:MA


All six aHA

Rehydration of the dried polyester products (total of 500 mM monomer before polymerization, depicted in Figs. S2, S3, and $\mathbf{S 5}$ ) in $500 \mu \mathrm{~L}$ water resulted in a range of phase separation properties, including full solubility for some samples $(100 \% 4 a 2 h, 1: 14 a 2 h: G A, 1: 14 a 2 h: L A, ~ a n d ~ 1: 1$ $4 \mathrm{a} 2 \mathrm{~h}: \mathrm{SA}$ ), and formation of microdroplets for other samples ( $1: 14 \mathrm{a} 2 \mathrm{~h}: \mathrm{PA}, 1: 14 \mathrm{a} 2 \mathrm{~h}: \mathrm{MA}$, and All six monomers).

## Figure S12



Representative optical microscopy image of a poly(4a2h) sample synthesized by drying of 500 mM 4 a 2 h followed by rehydration. No droplets are observed (the small dots are likely impurities on the sample slide glass or objective). Scale bar is $100 \mu \mathrm{~m}$.

Figure S13


1:9 4a2h:LA


1:4 4a2h:LA


1:4 4a2h:SA

Representative optical microscopy images of samples containing 1:94a2h:LA, 1:4 4a2h:LA, and 1:4 4a2h:SA ( 500 mM total monomer concentration), followed by drying. Rehydration (in $4: 1$ water:acetonitrile) of 1:9 4a2h:LA and 1:4 4a2h:SA samples resulted in droplet assembly, suggesting that decreasing the ratio of 4 a 2 h monomers in the starting mixture, followed by synthesis, can decrease the solubility of the polyester products, resulting in droplet assembly. Acetonitrile was also required to be added to the rehydration solvent, further suggesting the insolubility of the gel phase in pure water. Scale bars are $100 \mu \mathrm{~m}$.

Figure S14


Rehydration of dried polyester products composed of varying ratios of 4 a 2 h to PA and MA total of 500 mM monomer before polymerization) in $500 \mu \mathrm{~L}$ water (except for $2: 34 \mathrm{a} 2 \mathrm{~h}: \mathrm{PA}$ and 2:3 4a2h:MA samples, which were rehydrated in $4: 1$ water:acetonitrile) resulted in non-turbid solutions. This suggests that higher amounts of 4 a 2 h incorporation, even to originally fairly apolar and insoluble polyesters, result in greater solubility of polyester products (although a some smaller droplets still form at higher 4 a 2 h ratios). Turbid solutions formed at lower 4 a 2 h proportions (and droplets as observed through microscopy in Fig. 5) suggest that droplet formation occurs readily at lower 4a2h ratios due to a decrease in solubility.

Figure S15


A photograph depicting rehydrated samples of polyPA and 1:4 4a2h:PA ( 500 mM total starting monomer concentration) after 24 hours. Droplets which contain 4 a 2 h appear to coalesce much more quickly than those which do not contain 4 a 2 h , as indicated by the observation of macroscopic phase separation. The brown color could be the result of products or intermediates formed through degradation of the starting materials followed by melanoidin formation akin to Maillard-type reaction products or intermediates ${ }^{4}$.

Figure S16


Representative brightfield (left) and fluorescence (middle) microscopy images of polyLA droplets ( 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}-$ GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). PolyLA is not intrinsically fluorescent in the red channel, while the fluorescent RNA appears to segregate somewhat weakly into the polyLA droplets based on kymograph analysis (in our previous study, a fluoresceinlabeled RNA $25-$ mer was not observed to segregate into polyLA droplets ${ }^{6}$; however, the different chemical structure of the Cy 3 fluorescent tag as well as the shorter length may assist in weak segregation of fluorescent RNA into polyLA droplets in this study). Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S17


#### Abstract

Red Brightfield Fluorescence    

Representative brightfield (left) and fluorescence (middle) microscopy images of polyPA droplets ( 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}-$ GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). PolyPA is not intrinsically fluorescent in the red channel, while kymograph analysis shows that that fluorescent RNA segregates into the polyPA droplets ${ }^{6}$. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.


Figure S18


Representative brightfield (left) and fluorescence (middle) microscopy images of polyMA droplets ( 500 mM total monomer concentration before synthesis) with or without fluorescent RNA (5'-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). PolyMA is not intrinsically fluorescent in the red channel, while kymograph analysis also shows that the fluorescent RNA does not appear to segregate into the polyMA droplets ${ }^{6}$. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S19


Representative brightfield (left) and fluorescence (middle) microscopy images of polySA droplets ( 500 mM total monomer concentration before synthesis) with or without fluorescent RNA (5'-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). PolySA appears to be fluorescent in the red channel, while the images and kymograph indicate that fluorescent RNA likely segregates into the polySA droplets based on the high fluorescence intensities localized to the droplets (in our previous study, a fluorescein-labeled RNA 25-mer was not observed to segregate into polyLA droplets ${ }^{6}$; however, the different chemical structure of the Cy3 fluorescent tag as well as the shorter length may assist in weak segregation of fluorescent RNA into polySA droplets in this study). However, the intrinsic fluorescent character of polySA may contribute to some of the increase in localized droplet fluorescence intensity even in the presence of fluorescent RNA; further quantitative analyses using confocal microscopy are planned. Scale bars are $100 \mu \mathrm{~m}$ (615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S20


Representative brightfield (left) and fluorescence (middle) microscopy images of poly(4a2h-LA) droplets (1:9 4a2h:LA; 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}$-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). poly(4a2h-LA) droplets appear slightly fluorescent in the red channel. Although the fluorescence intensities localized to some droplets in the presence of RNA are higher than some droplets in the absence of RNA (which may show the ability of these droplets to segregate RNA), the current images and analyses are inconclusive on whether fluorescent RNA can definitively segregate into the droplets. The fluorescent character of poly $(4 a 2 h-L A)$ may contribute to some of the increase in localized droplet fluorescence intensity even in the presence of fluorescent RNA; further quantitative analyses using confocal microscopy are required to confirm whether these droplets can segregate RNA. Due to the fact that polyLA droplets showed the ability to segregate RNA, and no appearance of significant amounts of background fluorescence (due to significant amounts of RNA in solution and not segregated within droplets), it is likely that poly(4a2h-LA) droplets can segregate fluorescent RNA. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S21


Representative brightfield (left) and fluorescence (middle) microscopy images of poly(4a2h-PA) droplets (1:4 4a2h:PA; 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}$-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). poly(4a2h-PA) appears slightly fluorescent in the red channel, however the significant increase in localized fluorescence intensity to the droplets in the presence of RNA suggests that fluorescent RNA segregates strongly into the poly(4a2h-PA) droplets based on kymograph analysis. The fluorescent character of poly $(4 \mathrm{a} 2 \mathrm{~h}-\mathrm{PA})$ may contribute to some of the increase in localized droplet fluorescence intensity even in the presence of fluorescent RNA; further quantitative analyses using confocal microscopy are planned. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S22


Representative brightfield (left) and fluorescence (middle) microscopy images of poly(4a2h-MA) droplets (1:4 4a2h:MA; 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}$-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). poly(4a2h-MA) appears intrinsically slightly fluorescent in the red channel. However, kymograph analysis suggests that fluorescent RNA does indeed segregate within the poly(4a2h-MA) droplets. The fluorescent intensity localized to droplets (and the fluorescence intensity increase compared to the background) is greater in the presence of RNA. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S23


Representative brightfield (left) and fluorescence (middle) microscopy images of poly(4a2h-SA) droplets (1:4 4a2h:SA; 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}$-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). poly(4a2h-SA) may be intrinsically slightly fluorescent, but kymograph analysis suggests that fluorescent RNA clearly segregates to the poly(4a2h-SA) droplets. Scale bars are $100 \mu \mathrm{~m}(615.38$ pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S24


Representative brightfield (left) and fluorescence (middle) microscopy images of poly(4a2h-PA) droplets (1:1 4a2h:PA; 500 mM total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}$-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). poly( $4 \mathrm{a} 2 \mathrm{~h}-\mathrm{PA}$ ) droplets appear to be intrinsically slightly fluorescent in the red channel, and although there is some localized fluorescence intensity to the droplets in the presence of fluorescent, RNA, kymograph analysis is inconclusive as to whether these droplets retain the ability to segregate fluorescent RNA upon increasing the 4 a 2 h :PA ratio to 1:1. In particular, the fluorescence intensity localized to the droplets is less compared to poly( $4 \mathrm{a} 2 \mathrm{~h}-\mathrm{PA}$ ) droplets prepared with a $4 \mathrm{a} 2 \mathrm{~h}:$ PA ratio of 1:4. However, more quantitative studies, such as with confocal microscopy, are required to confirm whether these droplets can segregate RNA or not. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.

Figure S25


Representative brightfield (left) and fluorescence (middle) microscopy images of poly(4a2h-MA) droplets ( $1: 14 \mathrm{a} 2 \mathrm{~h}: \mathrm{MA} ; 500 \mathrm{mM}$ total monomer concentration before synthesis) with or without fluorescent RNA ( $5^{\prime}$-GCGUAGACUGACUGG-Cy3-3') in the red emission channel ( 585 nm emission). Kymographs depict the same region of analysis and direction for each row (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). poly(4a2h-MA) droplets do not appear to be intrinsically fluorescent in the red channel. However, the fluorescence intensity in the presence of RNA is not localized to any droplets, suggesting that the ability to segregate fluorescent RNA may have been lost upon increasing the 4 a 2 h :MA ratio to 1:1. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant between the rows.


An overnight self-cleaving hammerhead ribozyme reaction was performed under a variety of conditions ( $4 \mu \mathrm{mHH}, 6 \mu \mathrm{~m} \mathrm{HH} 2,200 \mathrm{mM}$ MES $\mathrm{pH} 5.7,100 \mathrm{mM} \mathrm{MgCl} 2$ ) including in the various ratios of $4 \mathrm{a} 2 \mathrm{~h}: \mathrm{PA}$-containing polyester systems and polymerized pure poly $(4 \mathrm{a} 2 \mathrm{~h})$ and polyPA (the identical ribozyme reaction was observed to occur in the presence of polyPA previously ${ }^{6}$ ). The fraction of product accumulated for each of the hammerhead ribozyme reactions is presented above each respective lane. The representative gel-shift assay indicates that the reaction (fluorescent starting material band above, and fluorescent product band below) occurred under all conditions tested, suggesting that the presence of polyester samples (poly( 4 a 2 h ), polyPA, and poly(4a2h-PA) with different 4 a 2 h :PA ratios) does not completely inhibit the reaction from occurring.

Figure S27

|  | No Polyesters <br> (Control) | $\mathbf{1}: 4$ <br> polymerized <br> 4a2h:PA | $\mathbf{1 0 0 \%}$ polymerized PA |
| :--- | :---: | :---: | :---: |
| Fraction of Product <br> Accumulated | $0.68 \pm 0.04$ | $0.61 \pm 0.10$ | $0.54 \pm 0.12$ |

The mean and the standard error to the mean $(n=4)$ of the fraction of product accumulated for hammerhead ribozyme reactions without polyesters and in the presence of poly(4a2h-PA) (1:4 $4 \mathrm{a} 2 \mathrm{~h}: \mathrm{PA}$ ratio) and polyPA. The gel-shift assay indicates that the amount of product that accumulated in the presence of poly( $4 \mathrm{a} 2 \mathrm{~h}-\mathrm{PA}$ ) ( $1: 44 \mathrm{a} 2 \mathrm{~h}: \mathrm{PA}$ ratio) was similar to that in the absence of any polyesters as well as that of polyPA (within error). This suggests that 4a2hcontaining polyesters do not result in complete inhibition of the self-cleavage function of the hammerhead ribozyme, although we cannot make any statements about the potential inhibition of the reaction rate by polyesters (this study was an endpoint measurement and not a kinetic study). However, as this experiment was conducted on only one specific, simple ribozyme system, further studies regarding activity of other functional RNA (such as ribozyme-catalyzed nucleic acid ligation or nonenzymatic polymerization) within polyester microdroplet systems are still warranted to determine global patterns of RNA function in polyester microdroplets.

Figure S28


Representative brightfield and fluorescence microscopy images of poly(4a2h-LA) droplets (1:9 4a2h:LA; 500 mM total monomer concentration before synthesis) in blue ( 470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. poly(4a2h-LA) droplets may show intrinsic fluorescence in all channels. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down each column.

Figure S29


Representative brightfield and fluorescence microscopy images of poly(4a2h-PA) droplets (1:4 4a2h:PA; 500 mM total monomer concentration before synthesis) in blue ( 470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. poly(4a2h-PA) droplets may show intrinsic fluorescence in all channels, although the intensities may be variable. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down each column.

Figure S30


Representative brightfield and fluorescence microscopy images of poly(4a2h-MA) droplets (1:4 4a2h:MA; 500 mM total monomer concentration before synthesis) in blue ( 470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. poly(4a2h-MA) droplets may show intrinsic fluorescence in all channels, although the intensities may be variable. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down each column.

## Figure S31



Representative brightfield and fluorescence microscopy images of poly(4a2h-SA) droplets (1:4 4a2h:SA; 500 mM total monomer concentration before synthesis) in blue ( 470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. poly(4a2h-SA) droplets may show intrinsic fluorescence in all channels, although the intensities may be variable. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down each column.

Figure S32


Additional fluorescence microscopy images of poly(4a2h-LA) droplets (1:9 4a2h:LA; 500 mM total monomer concentration before synthesis) in all emission channels. Each column represents a single sample. We observed some variability in the intrinsic fluorescence, and further detailed investigations are planned. Scale bars are $100 \mu \mathrm{~m}$. Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant down each column.

Figure S33


Additional fluorescence microscopy images of poly(4a2h-PA) droplets (1:4 4a2h:PA; 500 mM total monomer concentration before synthesis) in all emission channels. Each column represents a single sample. We observed some variability in the intrinsic fluorescence, and further detailed investigations are planned. Scale bars are $100 \mu \mathrm{~m}$. Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant down each column.

Figure S34


Additional fluorescence microscopy images of poly(4a2h-MA) droplets (1:4 4a2h:MA; 500 mM total monomer concentration before synthesis) in all emission channels. Each column represents a single sample. We observed some variability in the intrinsic fluorescence, and further detailed investigations are planned. Scale bars are $100 \mu \mathrm{~m}$. Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant down each column.

Figure S35


Additional fluorescence microscopy images of poly(4a2h-SA) droplets (1:4 4a2h:SA; 500 mM total monomer concentration before synthesis) in all emission channels. Each column represents a single sample. We observed some variability in the intrinsic fluorescence, and further detailed investigations are planned. Scale bars are $100 \mu \mathrm{~m}$. Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant down each column.

Figure S36


Representative brightfield and fluorescence microscopy images of homopolyester droplets in blue ( 470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels. Each column represents a single sample. While polyMA and polySA show slight intrinsic fluorescence in one or more channels (Figs. S36-S38), polyLA and polyPA show little to no fluorescence (Figs. S39S40). This suggests that incorporation of 4 a 2 h into polyLA or polyPA systems may convey the emergence of fluorescent properties within 4a2h-containing heteropolyester droplets observed in Figure 8. Incorporation of 4 a 2 h into the polyMA system in particular may expand the type of fluorescence exhibited, as potentially stronger blue fluorescence was observed (in addition to the green fluorescence originally observed in polyMA) in the poly(4a2h-MA) system (Figs. 8 and S34). Scale bars are $100 \mu \mathrm{~m}$. Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (brightness, contrast) parameters were constant down each column.

Figure S37


Representative brightfield and fluorescence microscopy images of polyMA droplets in blue (470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. polyMA apparently shows some intrinsic fluorescence (especially green fluorescence) based on kymograph analysis. Scale bars are $100 \mu \mathrm{~m}(615.38$ pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down the column.

Figure S38


Representative brightfield and fluorescence microscopy images of polySA droplets in blue (470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. polySA shows some intrinsic fluorescence in all channels as evidenced by the kymograph analysis. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down the column.

Figure S39


Representative brightfield and fluorescence microscopy images of polyLA droplets in blue (470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. polyLA shows little to no intrinsic fluorescence. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down the column.

Figure S40


Representative brightfield and fluorescence microscopy images of polyPA droplets in blue (470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels with accompanying kymographs. The region and direction of analysis for each image were identical (in black for brightfield image and in white for fluorescence images; arrows indicate direction of analysis). The column represents a single sample. polyPA shows no clearly observable intrinsic fluorescence based on kymograph analysis. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down the column.

Figure S41


Representative brightfield and fluorescence microscopy images of the white powdery product from drying of 4a2h monomer solution (Fig. S2) in blue ( 470 nm emission), green ( 527 nm emission), and red ( 585 nm emission) channels. Each column represents a single sample. This product may show some apparent fluorescence. Scale bars are $100 \mu \mathrm{~m}$ ( 615.38 pixels). Fluorescence acquisition (excitation illumination strength and exposure time) and image processing (contrast, brightness) parameters were constant down the column; this acquisition and image analysis was performed once.

Table S1. Peak list sorted by mass for poly(4a2h) positive mode MALDI-TOF-MS (Fig. S1); protonated peaks. "M $+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 120.0731659 | 510367.5448 | 120.065519 | 4a2h | $\mathbf{M}+\mathbf{H}$ | 63.68934282 |
| 203.1046448 | 15413.79948 | 203.102637 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 9.885519113 |
| 221.1161804 | 69864.22229 | 221.113202 | (4a2h) ${ }_{2}$ | $\mathbf{M}+\mathbf{H}$ | 13.47011383 |
| 304.1588745 | 10586.59585 | 304.15032 | (4a2h)3 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 28.12593457 |
| 322.1725769 | 91069.28237 | 322.160885 | (4a2h) ${ }_{3}$ | $\mathbf{M}+\mathbf{H}$ | 36.29212777 |
| 423.2255859 | 58388.26055 | 423.208568 | (4a2h) ${ }_{4}$ | $\mathbf{M}+\mathbf{H}$ | 40.21170479 |
| 524.2819214 | 26539.99417 | 524.256251 | (4a2h) ${ }_{5}$ | $\mathbf{M}+\mathbf{H}$ | 48.96534271 |
| 625.3421021 | 8320.135667 | 625.303934 | (4a2h) ${ }_{6}$ | $\mathbf{M}+\mathbf{H}$ | 61.03919858 |

Table S2. Peak list sorted by adduct and then mass and also colorcoded (for ease of visualization) for poly(4a2h) positive mode MALDI-TOF-MS (Fig. S1); protonated peaks. "M+H- $\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 120.0731659 | 510367.5448 | 120.065519 | 4a2h | M+H | 63.68934282 |
| 221.1161804 | 69864.22229 | 221.113202 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathbf{M}+\mathrm{H}$ | 13.47011383 |
| 322.1725769 | 91069.28237 | 322.160885 | $(4 a 2 h)_{3}$ | $\mathbf{M}+\mathbf{H}$ | 36.29212777 |
| 423.2255859 | 58388.26055 | 423.208568 | $(4 a 2 h)_{4}$ | $\mathrm{M}+\mathrm{H}$ | 40.21170479 |
| 524.2819214 | 26539.99417 | 524.256251 | (4a2h)5 | $\mathbf{M}+\mathrm{H}$ | 48.96534271 |
| 625.3421021 | 8320.135667 | 625.303934 | (4a2h) ${ }_{6}$ | $\mathrm{M}+\mathrm{H}$ | 61.03919858 |
| 203.1046448 | 15413.79948 | 203.102637 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 9.885519113 |
| 304.1588745 | 10586.59585 | 304.15032 | (4a2h)3 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 28.12593457 |

Table S3. Peak list sorted by mass for poly(4a2h-GA) positive mode MALDI-TOF-MS (Figs. 2 and S6); sodiated and protonated peaks. " $\mathrm{M}+\mathrm{H} / \mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 124.0400009 | 40223.07344 | 124.036896 | 4a2h | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 25.03219687 |
| 182.0437775 | 14999.38361 | 182.042376 | (4a2h)GA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 7.69857014 |
| 200.0528717 | 8975.386506 | 200.052941 | (4a2h)GA | $\mathbf{M}+\mathbf{N a}$ | 0.346388309 |
| 203.1046448 | 335548.2198 | 203.102637 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 9.885519113 |
| 221.1161804 | 11685.60477 | 221.113202 | $(4 a 2 h)_{2}$ | $\mathbf{M}+\mathbf{H}$ | 13.47011383 |
| 225.0892334 | 14974.01782 | 225.084579 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 20.67844017 |
| 261.1126099 | 135610.0097 | 261.108117 | (4a2h) $\mathbf{2}^{\text {GA }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 17.20690667 |
| 283.0907593 | 82496.61156 | 283.090059 | (4a2h) $\mathbf{2}^{\text {GA }}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 2.47368983 |
| 286.1491699 | 441457.0766 | 286.13789 | $(4 a 2 h)_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 39.42128042 |
| 301.1065063 | 19510.88378 | 301.100624 | (4a2h) $\mathbf{2}^{\text {GA }}$ | $\mathbf{M}+\mathbf{N a}$ | 19.53615347 |
| 304.1588745 | 195246.9344 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 28.12593457 |
| 319.1213379 | 17488.47027 | 319.113597 | (4a2h) $\mathbf{2}^{\mathbf{G A}}{ }_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 24.25747782 |
| 326.1412048 | 20728.48335 | 326.132262 | (4a2h)3 | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 27.42088116 |
| 341.1045532 | 36320.30733 | 341.095539 | (4a2h) $\mathbf{2}^{\mathbf{G A}}{ }_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 26.42726735 |
| 344.1555176 | 108391.6045 | 344.143365 | (4a2h) $\mathbf{2}^{\text {GA }}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 35.31254482 |
| 359.1139221 | 20225.99359 | 359.106104 | (4a2h) $\mathbf{2}^{\mathbf{G A}}{ }_{\mathbf{2}}$ | $\mathbf{M}+\mathbf{N a}$ | 21.77105572 |
| 362.1670837 | 191480.3936 | 362.1558 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 31.15714286 |
| 384.1511536 | 94434.368 | 384.137742 | (4a2h)3 ${ }_{3} \mathbf{G A}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 34.91342436 |
| 387.204071 | 185011.0484 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 47.78857098 |
| 399.1156616 | $\mathbf{9 4 7 9 . 5 7 7 4 6 6}$ | 399.101019 | $(4 \mathrm{a} 2 \mathrm{~h}) \mathbf{2 G A}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 36.689009331 |
| 402.1612854 | 31872.25501 | 402.148845 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 30.9348147 |
| 402.1612854 | 31872.25501 | 402.148307 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}$ | $\mathbf{M}+\mathrm{Na}$ | 32.27267099 |
| 405.2134399 | 74933.90811 | 405.198003 | $(4 a 2 h)_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 38.09727809 |


| 417.1277466 | 10683.35167 | 417.111584 | (4a2h) $\mathbf{2 F A}_{3}$ | $\mathbf{M}+\mathbf{N a}$ | 38.74882075 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 420.1789246 | 61115.46674 | 420.16128 | $(4 a 2 h)_{3} \mathbf{G A ~}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 41.99473355 |
| 427.1964722 | 17597.17866 | 427.179945 | (4a2h) ${ }_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 38.689000728 |
| 442.1593018 | 72077.419 | 442.143222 | (4a2h)3 $\mathbf{H A}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 36.36775868 |
| 445.2150574 | 158052.4843 | 445.191048 | (4a2h) ${ }_{3} \mathrm{GA}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 53.93049368 |
| 460.1722412 | 31517.05646 | 460.154325 | (4a2h)2 $\mathbf{H A}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 38.93522244 |
| 460.1722412 | 31517.05646 | 460.153787 | (4a2h)3 $\mathbf{H A A}_{2}$ | $\mathbf{M}+\mathbf{N a}$ | 40.1044423 |
| 463.2295227 | 133427.5362 | 463.203483 | (4a2h)4GA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 56.21655699 |
| 478.1901855 | 12458.87168 | 478.16676 | (4a2h) $\mathbf{3}^{\text {GA }}$ 3 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 48.99032923 |
| 485.2112732 | 65778.20395 | 485.185425 | (4a2h)4 ${ }_{4}$ GA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 53.27487527 |
| 488.2608948 | 59412.83895 | 488.233251 | $(4 a 2 h)_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 56.62001706 |
| 500.1743164 | 28151.63465 | 500.148702 | $(4 a 2 h) 3 \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 51.21358088 |
| 503.2250366 | 59062.58119 | 503.196528 | $(4 a 2 h) 3 \mathrm{GA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 56.65504314 |
| 503.2250366 | 59062.58119 | 503.19599 | (4a2h)4 ${ }_{4}$ GA | $\mathbf{M}+\mathbf{N a}$ | 57.72426962 |
| 506.2736511 | 24984.10188 | 506.245686 | (4a2h) ${ }_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.24021986 |
| 518.1901245 | 20255.6906 | 518.159805 | $(4 a 2 h) 2 \mathbf{G A}_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 58.51382471 |
| 518.1901245 | 20255.6906 | 518.159267 | (4a2h)3 $\mathbf{H A}_{3}$ | $\mathbf{M}+\mathbf{N a}$ | 59.55217626 |
| 521.2373657 | 78040.06133 | 521.208963 | (4a2h) $\mathbf{4}^{\mathbf{G A}}{ }_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 54.49392665 |
| 528.2578735 | 9805.778115 | 528.227628 | (4a2h) ${ }_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 57.2585253 |
| 543.2216797 | 74038.11748 | 543.190905 | (4a2h) $\mathbf{4}^{\mathbf{G A}}{ }_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 56.65538159 |
| 546.2705078 | 95953.86078 | 546.238731 | (4a2h)4GA | $\mathrm{M}+$ ethanolamine +Na | 58.17385366 |
| 558.190979 | 7890.231182 | 558.154182 | $(4 a 2 h) 3 \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.92623541 |
| 561.2334595 | 35523.90839 | 561.202008 | $(4 a 2 h) 3 \mathrm{GA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 56.04305144 |
| 561.2334595 | 35523.90839 | 561.20147 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}$ | 57.00176266 |
| 564.2841797 | 66350.08315 | 564.251166 | (4a2h) ${ }_{5} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 58.50885207 |
| 576.2038574 | 8625.756098 | 576.165285 | $(4 \mathrm{ah})_{2} \mathrm{GA}_{5}$ | $\mathrm{M}+$ ethanolamine +Na | 66.94679982 |


| 576.2038574 | 8625.756098 | 576.164747 | $(4 a 2 h) 3 \mathrm{GA}_{4}$ | $\mathbf{M}+\mathbf{N a}$ | 67.88062304 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 579.2522583 | 25827.92332 | 579.214443 | (4a2h) ${ }_{4} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.2872204 |
| 586.2652588 | 35965.79904 | 586.233108 | (4a2h) ${ }_{5} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 54.84301136 |
| 589.3154907 | 18672.64635 | 589.280934 | (4a2h)5 | M+ethanolamine + Na | 58.64218746 |
| 601.234314 | 42399.11653 | 601.196385 | (4a2h) $\mathbf{4 G A}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.08914349 |
| 604.2858276 | 59753.9657 | 604.244211 | (4a2h) $\mathbf{4 G A}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 68.87386961 |
| 604.2858276 | 59753.9657 | 604.243673 | (4a2h) ${ }_{5} \mathbf{G A}$ | $\mathbf{M}+\mathbf{N a}$ | 69.7643002 |
| 619.2484741 | 26191.3836 | 619.207488 | (4a2h)3 $\mathbf{G A}_{4}$ | $\mathrm{M}+$ ethanolamine $+\mathbf{N a}$ | 66.19125543 |
| 619.2484741 | 26191.3836 | 619.20695 | (4a2h) $\mathbf{4 G A}_{3}$ | $\mathbf{M}+\mathbf{N a}$ | 67.06016623 |
| 622.3010254 | 60049.53752 | 622.256646 | (4a2h) ${ }_{5} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.32007554 |
| 644.2799072 | 51772.70637 | 644.238588 | (4a2h) $)^{\text {GAA }}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 64.13652918 |
| 647.3355103 | 42949.31003 | 647.286414 | (4a2h) ${ }_{5} \mathrm{GA}$ | $\mathrm{M}+$ ethanolamine +Na | 75.84935036 |
| 659.2451172 | 16151.97222 | 659.201865 | $(4 a 2 h) 4 \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.61296364 |
| 662.296814 | 35461.14964 | 662.249691 | (4a2h) $\mathbf{4}^{\text {GA }} 3$ | $\mathrm{M}+$ ethanolamine +Na | 71.15588824 |
| 662.296814 | 35461.14964 | 662.249153 | (4a2h) $\mathbf{5 G A}^{\text {G }}$ | $\mathbf{M}+\mathbf{N a}$ | 71.96832912 |
| 665.3424072 | 27933.75054 | 665.298849 | (4a2h)6GA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.47167046 |
| 677.2575073 | 13511.1714 | 677.212968 | (4a2h)3 ${ }^{\text {GA5 }}$ | $\mathrm{M}+$ ethanolamine + Na | 65.76856337 |
| 677.2575073 | 13511.1714 | 677.21243 | (4a2h)4 $\mathbf{G A}_{4}$ | $\mathbf{M}+\mathbf{N a}$ | 66.56304876 |
| 680.3108521 | 30417.34061 | 680.262126 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.6283461 |
| 687.3244019 | 16780.62446 | 687.280791 | (4a2h) ${ }_{6} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.45420325 |
| 702.2949829 | 39848.25195 | 702.244068 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.50315427 |
| 705.3435669 | 40697.91891 | 705.291894 | $(\mathbf{4 a 2 h})_{5} \mathrm{GA}_{\mathbf{2}}$ | $\mathrm{M}+$ ethanolamine +Na | 73.26455251 |
| 705.3435669 | 40697.91891 | 705.291356 | (4a2h) ${ }_{6} \mathrm{GA}$ | $\mathbf{M}+\mathbf{N a}$ | 74.02741371 |
| 717.263855 | 5412.926919 | 717.207345 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.79169168 |
| 720.303894 | 24036.04273 | 720.255171 | $(4 a 2 h) 4 \mathrm{GA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 67.64691871 |
| 720.303894 | 24036.04273 | 720.254633 | (4a2h) $)^{\text {GA }} 3$ | $\mathrm{M}+\mathrm{Na}$ | 68.39392729 |


| 723.3572388 | 35069.74033 | 723.304329 | (4a2h) $\mathbf{6}^{\mathbf{G A}}{ }_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.15008065 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 735.2769165 | 5993.886221 | 735.218448 | $(4 a 2 h) 3 \mathrm{GA}_{6}$ | M+ethanolamine + Na | 79.52534945 |
| 735.2769165 | 5993.886221 | 735.21791 | (4a2h)4 $\mathbf{G A F}_{5}$ | $\mathrm{M}+\mathrm{Na}$ | 80.25716348 |
| 738.3204346 | 11056.77839 | 738.267606 | (4a2h) $\mathbf{5 G A}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.55748074 |
| 745.3424072 | 28520.67782 | 745.286271 | (4a2h) $\mathbf{6 G A}_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.32169743 |
| 748.3928223 | 17103.13495 | 748.334097 | (4a2h)6GA | $\mathbf{M + e t h a n o l a m i n e + N a ~}$ | 78.4746629 |
| 760.3064575 | 20013.44797 | 760.249548 | $(4 \mathrm{ah}))_{5} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 74.85636808 |
| 763.3522339 | 28262.33855 | 763.297374 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{3}$ | $\mathbf{M}+$ ethanolamine +Na | 71.87223338 |
| 763.3522339 | 28262.33855 | 763.296836 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{GA}_{2}$ | $\mathbf{M}+\mathbf{N a}$ | 72.57712123 |
| 766.4041748 | 11138.89027 | 766.346532 | (4a2h) ${ }_{7} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.21767581 |
| 778.3182983 | 15354.55148 | 778.260651 | (4a2h)4 $\mathbf{G A}_{5}$ | M+ethanolamine+Na | 74.07202192 |
| 778.3182983 | 15354.55148 | 778.260113 | $(4 a 2 h) 5 \mathrm{GA}_{4}$ | $\mathbf{M}+\mathbf{N a}$ | 74.76335871 |
| 781.3716431 | 24798.47332 | 781.309809 | (4a2h) $\mathbf{6}^{\text {GA }} 3$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.14154576 |
| 788.3884888 | 7519.70457 | 788.328474 | (4a2h) ${ }_{7} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 76.12914157 |
| 803.352478 | 27697.001 | 803.291751 | (4a2h) $)^{\text {GA }}{ }_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.59772265 |
| 806.4040527 | 21790.57475 | 806.339577 | (4a2h) $\mathbf{6 G A}_{\mathbf{2}}$ | M+ethanolamine+Na | 79.96101871 |
| 806.4040527 | 21790.57475 | 806.339039 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}$ | 80.62828519 |
| 818.3193359 | 7880.62899 | 818.255028 | (4a2h) ${ }_{5} \mathrm{GA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.59155862 |
| 821.362854 | 19850.48399 | 821.302854 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 73.05466395 |
| 821.362854 | 19850.48399 | 821.302316 | (4a2h) $\mathbf{6 G A}_{3}$ | $\mathbf{M}+\mathbf{N a}$ | 73.70976901 |
| 824.4120483 | 17648.81544 | 824.352012 | $(4 a 2 h)_{7} \mathbf{G A}_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.82852365 |
| 836.3344727 | 7486.395318 | 836.266131 | $(4 \mathrm{ah}))_{4} \mathrm{GA}_{6}$ | M+ethanolamine+Na | 81.72237696 |
| 836.3344727 | 7486.395318 | 836.265593 | (4a2h) $)^{\text {GA }} 5$ | $\mathbf{M}+\mathbf{N a}$ | 82.36576582 |
| 839.381897 | 12227.83666 | 839.315289 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.35989475 |
| 846.395874 | 14387.72797 | 846.333954 | (4a2h) ${ }_{7} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.16263599 |
| 861.3621826 | 17605.83479 | 861.297231 | (4a2h)6GA4 | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.4113849 |


| 864.4101563 | 18499.07725 | 864.345057 | (4a2h) $\mathbf{6 G A}_{3}$ | M+ethanolamine+Na | 75.31627499 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 864.4101563 | 18499.07725 | 864.344519 | (4a2h) ${ }_{7} \mathbf{G A}_{2}$ | $\mathbf{M}+\mathbf{N a}$ | 75.93875886 |
| 879.3763428 | 13311.35027 | 879.308334 | (4a2h) ${ }_{5} \mathrm{GA}_{5}$ | M+ethanolamine +Na | 77.34348734 |
| 879.3763428 | 13311.35027 | 879.307796 | (4a2h) $\mathbf{6 G A}_{4}$ | $\mathbf{M}+\mathbf{N a}$ | 77.95537957 |
| $\mathbf{8 8 2 . 4 2 5 8 4 2 3}$ | 16284.32368 | 882.357492 | (4a2h) $\mathbf{7 G A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.46325681 |
| 904.4099731 | 16316.53843 | 904.339434 | (4a2h)7GA3 | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.00073993 |
| 907.456665 | 9863.059684 | 907.38726 | (4a2h) $\mathbf{7}_{\mathbf{G A}}^{2}$ | $\mathrm{M}+$ ethanolamine+Na | 76.488888414 |
| 907.456665 | 9863.059684 | 907.386722 | (4a2h) ${ }_{8} \mathrm{GA}$ | $\mathbf{M}+\mathbf{N a}$ | 77.08184097 |
| 919.3787231 | 8939.006052 | 919.302711 | (4a2h) $\mathbf{6 H A}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 82.68456526 |
| 922.4274292 | 13749.00442 | 922.350537 | (4a2h) ${ }_{6} \mathrm{GA}_{4}$ | M+ethanolamine +Na | 83.36548407 |
| 922.4274292 | 13749.00442 | 922.349999 | (4a2h) $\mathbf{7 G A}_{3}$ | $\mathrm{M}+\mathrm{Na}$ | 83.94882537 |
| 937.3886719 | 7765.893903 | 937.313814 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{6}$ | $\mathrm{M}+$ ethanolamine +Na | 79.86426091 |
| 937.38866719 | 7765.893903 | 937.313276 | (4a2h) $\mathbf{6}^{\text {GA }}{ }_{5}$ | $\mathrm{M}+\mathrm{Na}$ | 80.43828774 |
| 940.4359741 | 9876.554511 | 940.362972 | (4a2h) $\mathbf{7 G A}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.63185405 |
| $\mathbf{9 6 2 . 4 2 6 6 3 5 7}$ | 13010.21676 | 962.344914 | (4a2h) $\mathbf{7 G A}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 84.91938889 |
| 965.467041 | 10383.62124 | 965.39274 | (4a2h) $\mathbf{7 G A}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 76.96454813 |
| 965.467041 | 10383.62124 | 965.392202 | (4a2h) $\mathbf{8 G A}^{\text {G }}$ | $\mathbf{M}+\mathbf{N a}$ | 77.52187748 |
| $\mathbf{9 8 0 . 4 3 8 4 7 6 6}$ | 10019.87347 | 980.356017 | (4a2h) $\mathbf{6 G A}_{5}$ | $\mathrm{M}+$ ethanolamine +Na | 84.11185383 |
| 980.4384766 | 10019.87347 | 980.355479 | (4a2h) $\mathbf{7}^{\text {GA }} 4$ | $\mathbf{M}+\mathbf{N a}$ | 84.660688052 |
| 1020.439514 | 7355.481173 | 1020.350394 | (4a2h) $\mathbf{7}^{\text {GA }}$ 5 | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.34270161 |
| 1023.480896 | 8763.829756 | 1023.39822 | (4a2h) $\mathbf{7}^{\text {GA }} 4$ | $\mathbf{M}+$ ethanolamine +Na | 80.78575708 |
| 1023.480896 | 8763.829756 | 1023.397682 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{GA}_{3}$ | $\mathbf{M}+\mathrm{Na}$ | 81.31149939 |
| 1038.444336 | 6218.391766 | 1038.361497 | $(4 a 2 h) 6 \mathrm{GA}_{6}$ | $\mathrm{M}+$ ethanolamine +Na | 79.77851667 |
| 1038.444336 | 6218.391766 | 1038.360959 | (4a2h) ${ }_{7} \mathrm{GA}_{5}$ | $\mathbf{M}+\mathbf{N a}$ | 80.29668226 |
| 1081.49353 | 6526.155217 | 1081.4037 | (4a2h) ${ }_{7} \mathrm{GA}_{5}$ | M+ethanolamine+Na | 83.06821033 |
| 1081.49353 | 6526.155217 | 1081.403162 | (4a2h) $\mathbf{8 G A}_{4}$ | $\mathbf{M}+\mathbf{N a}$ | 83.56575344 |

Table S4. Peak list sorted by adduct and then mass and also colorcoded (for ease of visualization) for poly(4a2h-GA) positive mode MALDI-TOF-MS (Figs. 2 and S6); sodiated and protonated peaks. "M $\mathrm{H}+\mathrm{H} / \mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 286.1491699 | 441457.0766 | 286.13789 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | M+ethanolamine+Na | 39.42128042 |
| 344.1555176 | 108391.6045 | 344.143365 | (4a2h) ${ }_{2} \mathrm{GA}$ | M+ethanolamine+Na | 35.31254482 |
| 387.204071 | 185011.0484 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 47.78857098 |
| 402.1612854 | 31872.25501 | 402.148845 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 30.9348147 |
| 445.2150574 | 158052.4843 | 445.191048 | (4a2h)3 ${ }_{3}$ GA | $\mathrm{M}+$ ethanolamine+Na | 53.93049368 |
| 460.1722412 | 31517.05646 | 460.154325 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 38.93522244 |
| 488.2608948 | 59412.83895 | 488.233251 | $(4 a 2 h)_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 56.62001706 |
| 503.2250366 | 59062.58119 | 503.196528 | (4a2h)3GA2 | $\mathbf{M}+$ ethanolamine +Na | 56.65504314 |
| 518.1901245 | 20255.6906 | 518.159805 | (4a2h) ${ }_{2} \mathrm{GA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 58.51382471 |
| 546.2705078 | 95953.86078 | 546.238731 | (4a2h) ${ }_{4} \mathrm{GA}$ | $\mathrm{M}+$ ethanolamine+Na | 58.17385366 |
| 561.2334595 | 35523.90839 | 561.202008 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 56.04305144 |
| 576.2038574 | 8625.756098 | 576.165285 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{5}$ | $\mathrm{M}+$ ethanolamine+Na | 66.94679982 |
| 589.3154907 | 18672.64635 | 589.280934 | $(4 \mathrm{a} 2 \mathrm{~h})_{5}$ | $\mathrm{M}+$ ethanolamine+Na | 58.64218746 |
| 604.2858276 | 59753.9657 | 604.244211 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 68.87386961 |
| 619.2484741 | 26191.3836 | 619.207488 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 66.19125543 |
| 647.3355103 | 42949.31003 | 647.286414 | (4a2h) $)_{5} \mathrm{GA}$ | $\mathrm{M}+$ ethanolamine+Na | 75.84935036 |
| 662.296814 | 35461.14964 | 662.249691 | $(4 a 2 h) 4 \mathrm{GA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 71.15588824 |
| 677.2575073 | 13511.1714 | 677.212968 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GAA}_{5}$ | $\mathrm{M}+$ ethanolamine+Na | 65.76856337 |
| 705.3435669 | 40697.91891 | 705.291894 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 73.26455251 |
| 720.303894 | 24036.04273 | 720.255171 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 67.64691871 |
| 735.2769165 | 5993.886221 | 735.218448 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{6}$ | $\mathrm{M}+$ ethanolamine+Na | 79.52534945 |
| 748.3928223 | 17103.13495 | 748.334097 | (4a2h)6GA | $\mathbf{M}+$ ethanolamine+Na | 78.4746629 |
| 763.3522339 | 28262.33855 | 763.297374 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 71.87223338 |


| 778.3182983 | 15354.55148 | 778.260651 | (4a2h)4 ${ }^{\text {GA }}{ }_{5}$ | $\mathrm{M}+$ ethanolamine+Na | 74.07202192 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 806.4040527 | 21790.57475 | 806.339577 | (4a2h) ${ }_{6} \mathrm{GA}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 79.96101871 |
| 821.362854 | 19850.48399 | 821.302854 | (4a2h) ${ }_{5} \mathrm{GA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 73.05466395 |
| 836.3344727 | 7486.395318 | 836.266131 | (4a2h)4GA6 | M+ethanolamine+Na | 81.72237696 |
| 864.4101563 | 18499.07725 | 864.345057 | (4a2h)6GA3 | $\mathbf{M}+$ ethanolamine+Na | 75.31627499 |
| 879.3763428 | 13311.35027 | 879.308334 | (4a2h) $)^{\text {GA }}{ }_{5}$ | $\mathrm{M}+$ ethanolamine +Na | 77.34348734 |
| 907.456665 | 9863.059684 | 907.38726 | (4a2h) $\mathbf{7 G A}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 76.48888414 |
| 922.4274292 | 13749.00442 | 922.350537 | (4a2h)6GA ${ }_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 83.36548407 |
| 937.3886719 | 7765.893903 | 937.313814 | (4a2h) $)^{\text {GA }}{ }_{6}$ | $\mathbf{M}+$ ethanolamine +Na | 79.86426091 |
| 965.467041 | 10383.62124 | 965.39274 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{3}$ | M+ethanolamine+Na | 76.96454813 |
| 980.4384766 | 10019.87347 | 980.356017 | (4a2h) $)_{6 \mathrm{GA}_{5}}$ | $\mathrm{M}+$ ethanolamine+ Na | 84.11185383 |
| 1023.480896 | 8763.829756 | 1023.39822 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 80.78575708 |
| 1038.444336 | 6218.391766 | 1038.361497 | (4a2h) $)^{\text {GA }}{ }_{6}$ | $\mathrm{M}+$ ethanolamine+ Na | 79.77851667 |
| 1081.49353 | 6526.155217 | 1081.4037 | (4a2h) $)^{\text {GA }}{ }_{5}$ | $\mathrm{M}+$ ethanolamine +Na | 83.06821033 |
| 221.1161804 | 11685.60477 | 221.113202 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathbf{M}+\mathbf{H}$ | 13.47011383 |
| 203.1046448 | 335548.2198 | 203.102637 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 9.885519113 |
| 261.1126099 | 135610.0097 | 261.108117 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 17.20690667 |
| 304.1588745 | 195246.9344 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 28.12593457 |
| 319.1213379 | 17488.47027 | 319.113597 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 24.25747782 |
| 362.1670837 | 191480.3936 | 362.1558 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 31.15714286 |
| 405.2134399 | 74933.90811 | 405.198003 | $(4 a 2 h)_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 38.09727809 |
| 420.1789246 | 61115.46674 | 420.16128 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 41.99473355 |
| 463.2295227 | 133427.5362 | 463.203483 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 56.21655699 |
| 478.1901855 | 12458.87168 | 478.16676 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 48.99032923 |
| 506.2736511 | 24984.10188 | 506.245686 | (4a2h)5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.24021986 |
| 521.2373657 | 78040.06133 | 521.208963 | (4a2h)4 $\mathbf{G A ~}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 54.49392665 |


| 564.2841797 | 66350.08315 | 564.251166 | (4a2h) ${ }_{5} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 58.50885207 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 579.2522583 | 25827.92332 | 579.214443 | $(4 a 2 h) 4 \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.2872204 |
| 622.3010254 | 60049.53752 | 622.256646 | (4a2h) $)^{\text {GAA }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.32007554 |
| 665.3424072 | 27933.75054 | 665.298849 | (4a2h) ${ }_{6} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.47167046 |
| 680.3108521 | 30417.34061 | 680.262126 | (4a2h) $)^{\text {GA }} 3$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.6283461 |
| 723.3572388 | 35069.74033 | 723.304329 | (4a2h) $\mathbf{6 G A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.15008065 |
| 738.3204346 | 11056.77839 | 738.267606 | (4a2h) $)^{\text {GA }}{ }_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.55748074 |
| 766.4041748 | 11138.89027 | 766.346532 | (4a2h) ${ }_{7} \mathrm{GA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.21767581 |
| 781.3716431 | 24798.47332 | 781.309809 | (4a2h) ${ }_{6} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.14154576 |
| 824.4120483 | 17648.81544 | 824.352012 | (4a2h) $\mathbf{7 G A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.82852365 |
| 839.381897 | 12227.83666 | 839.315289 | (4a2h) $\mathbf{6 G A}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.35989475 |
| 882.4258423 | 16284.32368 | 882.357492 | (4a2h) $)^{\text {GA }}{ }_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.46325681 |
| 940.4359741 | 9876.554511 | 940.362972 | (4a2h) $)_{7} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.63185405 |
| 200.0528717 | 8975.386506 | 200.052941 | (4a2h)GA | $\mathbf{M}+\mathrm{Na}$ | 0.346388309 |
| 301.1065063 | 19510.88378 | 301.100624 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}$ | $\mathbf{M}+\mathbf{N a}$ | 19.53615347 |
| 359.1139221 | 20225.99359 | 359.106104 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{2}$ | $\mathbf{M}+\mathrm{Na}$ | 21.77105572 |
| 402.1612854 | 31872.25501 | 402.148307 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}$ | $\mathbf{M}+\mathrm{Na}$ | 32.27267099 |
| 417.1277466 | 10683.35167 | 417.111584 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}$ | 38.74882075 |
| 460.1722412 | 31517.05646 | 460.153787 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{2}$ | $\mathbf{M}+\mathrm{Na}$ | 40.1044423 |
| 503.2250366 | 59062.58119 | 503.19599 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}$ | 57.72426962 |
| 518.1901245 | 20255.6906 | 518.159267 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}$ | 59.55217626 |
| 561.2334595 | 35523.90839 | 561.20147 | (4a2h) $\mathbf{4 G A}_{2}$ | $\mathbf{M}+\mathbf{N a}$ | 57.00176266 |
| 576.2038574 | 8625.756098 | 576.164747 | $(4 a 2 h) 3 \mathrm{GA}_{4}$ | $\mathbf{M}+\mathbf{N a}$ | 67.88062304 |
| 604.2858276 | 59753.9657 | 604.243673 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}$ | $\mathbf{M}+\mathbf{N a}$ | 69.7643002 |
| 619.2484741 | 26191.3836 | 619.20695 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{3}$ | $\mathbf{M}+\mathbf{N a}$ | 67.06016623 |
| 662.296814 | 35461.14964 | 662.249153 | (4a2h) $\mathbf{S G A}_{2}$ | $\mathbf{M}+\mathbf{N a}$ | 71.96832912 |


| 677.2575073 | 13511.1714 | 677.21243 | (4a2h)4GA4 | $\mathrm{M}+\mathrm{Na}$ | 66.56304876 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 705.3435669 | 40697.91891 | 705.291356 | (4a2h) ${ }_{6} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}$ | 74.02741371 |
| 720.303894 | 24036.04273 | 720.254633 | (4a2h) $\mathbf{5}^{\text {GA }} 3$ | $\mathrm{M}+\mathrm{Na}$ | 68.39392729 |
| 735.2769165 | 5993.886221 | 735.21791 | (4a2h)4GA5 | $\mathbf{M}+\mathrm{Na}$ | 80.25716348 |
| 763.3522339 | 28262.33855 | 763.296836 | (4a2h) ${ }_{6} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}$ | 72.57712123 |
| 778.3182983 | 15354.55148 | 778.260113 | (4a2h) ${ }_{5} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}$ | 74.76335871 |
| 806.4040527 | 21790.57475 | 806.339039 | (4a2h) $)^{\text {GA }}$ ( | $\mathbf{M}+\mathrm{Na}$ | 80.62828519 |
| 821.362854 | 19850.48399 | 821.302316 | $(4 a 2 h) 6 \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}$ | 73.70976901 |
| 836.3344727 | 7486.395318 | 836.265593 | (4a2h) $\mathbf{S G A}_{5}$ | $\mathrm{M}+\mathrm{Na}$ | 82.36576582 |
| 864.4101563 | 18499.07725 | 864.344519 | (4a2h) ${ }_{7} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}$ | 75.93875886 |
| 879.3763428 | 13311.35027 | 879.307796 | $(4 \mathrm{a} 2)_{6} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}$ | 77.95537957 |
| 907.456665 | 9863.059684 | 907.386722 | (4a2h) ${ }_{8} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}$ | 77.08184097 |
| 922.4274292 | 13749.00442 | 922.349999 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}$ | 83.94882537 |
| 937.3886719 | 7765.893903 | 937.313276 | (4a2h) ${ }_{6} \mathrm{GA}_{5}$ | $\mathrm{M}+\mathrm{Na}$ | 80.43828774 |
| 965.467041 | 10383.62124 | 965.392202 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{GA}_{2}$ | $\mathbf{M}+\mathrm{Na}$ | 77.52187748 |
| 980.4384766 | 10019.87347 | 980.355479 | (4a2h) $\mathbf{7 G A}_{4}$ | $\mathbf{M}+\mathrm{Na}$ | 84.66068052 |
| 1023.480896 | 8763.829756 | 1023.397682 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{GA}_{3}$ | $\mathbf{M}+\mathbf{N a}$ | 81.31149939 |
| 1038.444336 | 6218.391766 | 1038.360959 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{5}$ | $\mathbf{M}+\mathrm{Na}$ | 80.29668226 |
| 1081.49353 | 6526.155217 | 1081.403162 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{GA}_{4}$ | $\mathbf{M}+\mathrm{Na}$ | 83.56575344 |
| 124.0400009 | 40223.07344 | 124.036896 | 4a2h | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 25.03219687 |
| 182.0437775 | 14999.38361 | 182.042376 | (4a2h)GA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 7.69857014 |
| 225.0892334 | 14974.01782 | 225.084579 | $(4 a 2 h)_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 20.67844017 |
| 283.0907593 | 82496.61156 | 283.090059 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 2.47368983 |
| 326.1412048 | 20728.48335 | 326.132262 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 27.42088116 |
| 341.1045532 | 36320.30733 | 341.095539 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 26.42726735 |
| 384.1511536 | 94434.368 | 384.137742 | (4a2h)3GA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 34.91342436 |


| 399.1156616 | 9479.577466 | 399.101019 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 36.68900931 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 427.1964722 | 17597.17866 | 427.179945 | $(4 a 2 h)_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 38.68900728 |
| 442.1593018 | 72077.419 | 442.143222 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 36.36775868 |
| 485.2112732 | 65778.20395 | 485.185425 | (4a2h)4GA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 53.27487527 |
| 500.1743164 | 28151.63465 | 500.148702 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 51.21358088 |
| 528.2578735 | 9805.778115 | 528.227628 | (4a2h) ${ }^{\text {a }}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 57.2585253 |
| 543.2216797 | 74038.11748 | 543.190905 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 56.65538159 |
| 558.190979 | 7890.231182 | 558.154182 | $(4 \mathrm{ah})_{3} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.92623541 |
| 586.2652588 | 35965.79904 | 586.233108 | (4a2h) ${ }_{5} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 54.84301136 |
| 601.234314 | 42399.11653 | 601.196385 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.08914349 |
| 644.2799072 | 51772.70637 | 644.238588 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 64.13652918 |
| 659.2451172 | 16151.97222 | 659.201865 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 65.61296364 |
| 687.3244019 | 16780.62446 | 687.280791 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.45420325 |
| 702.2949829 | 39848.25195 | 702.244068 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.50315427 |
| 717.263855 | 5412.926919 | 717.207345 | (4a2h) ${ }_{4} \mathrm{GA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.79169168 |
| 745.3424072 | 28520.67782 | 745.286271 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.32169743 |
| 760.3064575 | 20013.44797 | 760.249548 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 74.85636808 |
| 788.3884888 | 7519.70457 | 788.328474 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 76.12914157 |
| 803.352478 | 27697.001 | 803.291751 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.59772265 |
| 818.3193359 | 7880.62899 | 818.255028 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{GA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.59155862 |
| 846.395874 | 14387.72797 | 846.333954 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.16263599 |
| 861.3621826 | 17605.83479 | 861.297231 | $(4 a 2 h) 6 \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 75.4113849 |
| 904.4099731 | 16316.53843 | 904.339434 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.00073993 |
| 919.3787231 | 8939.006052 | 919.302711 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{GA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 82.68456526 |
| 962.4266357 | 13010.21676 | 962.344914 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{GA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 84.91938889 |
| 1020.439514 | 7355.481173 | 1020.350394 | (4a2h) $)^{\text {GA }}{ }_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.34270161 |

Table S5. Peak list sorted by mass for poly(4a2h-LA) positive mode MALDI-TOF-MS (Figs. 2 and S6); sodiated and protonated peaks. "M+H-H2O (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 203.1046448 | 310156.1954 | 203.102637 | $(4 a 2 h)_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 9.885519113 |
| 275.1292725 | 121712.8269 | 275.123747 | (4a2h) ${ }_{2} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 20.08354808 |
| 286.1534729 | 399893.0661 | 286.13789 | (4a2h) ${ }_{2}$ | M+ethanolamine+Na | 54.45940767 |
| 304.1632996 | 219114.1385 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 42.67482277 |
| 347.1585693 | 28440.59692 | 347.144857 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 39.50032882 |
| 358.1759644 | 130060.0404 | 358.158995 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}$ | $\mathbf{M}+$ ethanolamine +Na | 47.37939082 |
| 376.1924133 | 222088.7496 | 376.17143 | $(4 a 2 h){ }_{3} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.78129631 |
| 387.2090759 | 215042.5396 | 387.185568 | $(4 a 2 h) 3$ | $\mathbf{M}+$ ethanolamine + Na | 60.714886998 |
| 405.2236633 | 106882.5888 | 405.198003 | (4a2h)4 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.32787874 |
| 419.190094 | 8245.149344 | 419.165967 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 57.5595251 |
| 430.2052612 | 31175.58651 | 430.180105 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 58.47836687 |
| 448.222168 | 98427.2027 | 448.19254 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 66.10544879 |
| 459.2406921 | 204851.9006 | 459.206678 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}$ | $\mathrm{M}+$ ethanolamine +Na | 74.07152515 |
| 477.2516174 | 178425.8422 | 477.219113 | (4a2h) ${ }_{4} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 68.11217555 |
| 488.2664795 | 95265.03239 | 488.233251 | $(4 a 2 h)_{4}$ | M+ethanolamine+Na | 68.05864191 |
| 502.2393799 | 10106.00743 | 502.201215 | $(4 a 2 h) 2 \mathrm{LA}_{3}$ | $\mathbf{M}+$ ethanolamine +Na | 75.99520244 |
| 506.2850952 | 43273.87977 | 506.245686 | (4a2h) 5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.84602633 |
| 520.2515259 | 35344.78352 | 520.21365 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.80831443 |
| 531.2701416 | 92994.80409 | 531.227788 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 79.72776078 |
| 549.2802124 | 129416.6699 | 549.240223 | (4a2h) $\mathbf{4 L A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.80858234 |
| 560.3005981 | 147956.14 | 560.254361 | $(4 a 2 h) 4 \mathrm{LA}$ | M+ethanolamine+Na | 82.52884443 |
| 578.3167114 | 106401.573 | 578.266796 | (4a2h)sLA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.31902496 |
| 589.3216553 | 39550.91519 | 589.280934 | (4a2h) 5 | M+ethanolamine + Na | 69.10332687 |


| 592.2870483 | 13760.19135 | 592.23476 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.28988694 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 603.2990112 | 34314.20048 | 603.248898 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{3}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 83.07222801 |
| 607.3388672 | 17711.5739 | $\mathbf{6 0 7 . 2 9 3 3 6 9}$ | (4a2h)6 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 74.91961929 |
| 621.3122559 | 64947.32759 | 621.261333 | (4a2h) $\mathbf{4 L A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 81.96688945 |
| 632.3282471 | 103996.2997 | 632.275471 | (4a2h) $\mathbf{4}^{\text {LA }}{ }_{2}$ | M+ethanolamine +Na | 83.47005763 |
| 650.3464355 | 106994.1225 | 650.287906 | (4a2h) $\mathbf{5 L A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.00559054 |
| 661.3550415 | 81982.80034 | 661.302044 | (4a2h) ${ }_{5} \mathrm{LA}$ | $\mathrm{M}+$ ethanolamine +Na | 80.14114652 |
| 664.326416 | 6966.899115 | 664.25587 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 106.203075 |
| 675.3336792 | 12023.88379 | 675.270008 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 94.28998511 |
| 679.3762207 | 52742.31982 | 679.314479 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.88824824 |
| 690.3737183 | 17061.40467 | 690.328617 | $(4 a 2 h)_{6}$ | $\mathrm{M}+$ ethanolamine +Na | 65.33303254 |
| 693.3427734 | 27939.0476 | 693.282443 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.02144214 |
| 704.3581543 | 52496.08738 | 704.296581 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{LA}_{3}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 87.42523911 |
| 708.3920898 | 7487.961148 | 708.341052 | (4a2h $)_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.05264167 |
| 722.3729248 | 71394.94506 | 722.309016 | $(4 a 2 h) 5 \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.47848163 |
| 733.3890991 | 77791.51538 | 733.323154 | $(4 a 2 h) 5 \mathrm{LA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 89.92641326 |
| 736.3734741 | 5038.285162 | 736.27698 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LAA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 131.0568218 |
| 747.3637695 | 5890.81099 | 747.291118 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LAA}_{5}$ | $\mathrm{M}+$ ethanolamine +Na | 97.21985081 |
| 751.4077148 | 68382.78517 | 751.335589 | $(4 a 2 h) 6 \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.99684223 |
| 762.4111938 | 40423.64349 | 762.349727 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 80.62814982 |
| 765.3768311 | 12201.02662 | 765.303553 | (4a2h)4LA5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.75031334 |
| 776.3899536 | 21872.28955 | 776.317691 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{LA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 93.08381586 |
| 780.4337158 | 23835.15482 | 780.362162 | $(4 \mathrm{ah})_{7} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.69309262 |
| 791.4186401 | 8773.753124 | 791.3763 | $(4 \mathbf{a} 2 \mathrm{~h})_{7}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 53.50190169 |
| 794.40448 | 37760.8861 | 794.330126 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.60589202 |
| 805.4151001 | 51204.31114 | 805.344264 | (4a2h) $)^{\text {LA }} 3$ | $\mathrm{M}+$ ethanolamine +Na | 87.95753713 |


| 823.43396 | 57880.14138 | 823.356699 | (4a2h) $\mathbf{6}^{\text {LA }}{ }_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.83656087 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 834.4454956 | 48137.42963 | 834.370837 | (4a2h) $\mathbf{6}^{\text {L }}{ }^{\text {a }}$ | $\mathrm{M}+$ ethanolamine +Na | 89.47892435 |
| 837.4158936 | 6953.435255 | 837.324663 | (4a2h) $\mathbf{4}^{\text {L }}{ }^{\text {d }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 108.9548165 |
| 848.4168701 | 9679.751505 | 848.338801 | (4a2h)4LA5 | $\mathrm{M}+$ ethanolamine +Na | 92.02587092 |
| 852.4661255 | 37074.11195 | 852.383272 | (4a2h) $\mathbf{7}^{\mathbf{L A}}{ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.20215157 |
| 863.4682617 | 19473.37411 | 863.39741 | (4a2h) ${ }_{7} \mathrm{LA}$ | M+ethanolamine+Na | 82.06153757 |
| 866.4376221 | 18277.77089 | 866.351236 | (4a2h) $)^{\text {L }} \mathrm{LA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.71252583 |
| 877.4467163 | 25930.19433 | 877.365374 | $(4 a 2 h) 5 \mathrm{LA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 92.71201191 |
| 881.4817505 | 10476.86646 | 881.409845 | (4a2h) ${ }_{8} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 81.58008265 |
| 892.4644165 | 5588.566316 | 892.423983 | (4a2h)8 | $\mathrm{M}+$ ethanolamine +Na | 45.30750492 |
| 895.4604492 | 36897.1879 | 895.377809 | $(4 a 2 h) 6 \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.29647884 |
| 906.4762573 | 37945.76358 | 906.391947 | $(4 a 2 h) 6 \mathrm{LA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 93.01751221 |
| 924.4910889 | 38167.21718 | 924.404382 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.79755082 |
| 935.505127 | 26021.91744 | 935.41852 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 92.5863142 |
| 938.463623 | 9315.307001 | 938.372346 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.27167194 |
| 949.4743652 | 12739.08974 | 949.386484 | (4a2h)sLA5 | M+ethanolamine+Na | 92.56634203 |
| $\mathbf{9 5 3 . 5 2 2 8 2 7 1}$ | 17961.87844 | 953.430955 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.35951877 |
| 964.5188599 | 9479.26976 | 964.445093 | (4a2h) ${ }_{8} \mathrm{LA}$ | $\mathbf{M}+$ ethanolamine +Na | 76.48632725 |
| 967.491333 | 20212.78028 | 967.398919 | $(4 a 2 h) 6 \mathrm{LA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.52833499 |
| 978.5040283 | 23606.18319 | 978.413057 | $(4 a 2 h) 6 \mathrm{LA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 92.97844029 |
| 996.520874 | 28319.96781 | 996.425492 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.72418988 |
| 1007.528015 | 24642.33925 | 1007.43963 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 87.73244308 |
| 1009.531433 | 6000.015279 | 1010.393456 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 853.1556542 |
| 1021.504334 | 6606.929504 | 1021.407594 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{6}$ | M+ethanolamine +Na | 94.71194513 |
| 1025.549072 | 21264.62702 | 1025.452065 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 94.59951695 |
| 1036.551514 | 13597.46102 | 1036.466203 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 82.3091672 |


| 1039.526733 | 10551.73249 | 1039.420029 | (4a2h) ${ }_{6} \mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 102.6576331 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1050.529785 | 12881.88462 | 1050.434167 | (4a2h) ${ }_{6} \mathrm{LAA}_{5}$ | M+ethanolamine+Na | 91.0272752 |
| 1065.556641 | 5067.830871 | 1065.492776 | (4a2h)9LA | $\mathbf{M}+$ ethanolamine + Na | 59.93904552 |
| 1068.548218 | 17510.4293 | 1068.446602 | (4a2h) ${ }_{7} \mathrm{LAA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.10608187 |
| 1079.56189 | 16979.74775 | 1079.46074 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{4}$ | M+ethanolamine+Na | 93.70387106 |
| 1081.560547 | 5295.226308 | 1082.414566 | (4a2h) $\mathbf{5 L A}_{8}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 788.9944822 |
| 1097.582886 | 18080.18737 | 1097.473175 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.96667117 |
| 1108.584229 | 13885.34982 | 1108.487313 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 87.43042781 |
| 1111.551025 | 5835.771056 | 1111.441139 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.86838461 |
| 1122.5625 | 6667.675158 | 1122.455277 | $(4 a 2 h) 6 \mathbf{L A}_{6}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 95.52540952 |
| 1137.598022 | 7258.682181 | 1137.513886 | $(4 a 2 h) 9 \mathbf{L A}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 73.96521575 |
| 1140.577637 | 9875.486724 | 1140.467712 | (4a2h) $\mathbf{7 L A}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.38564849 |
| 1151.593872 | 10620.58957 | 1151.48185 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{5}$ | M+ethanolamine +Na | 97.28513741 |
| 1169.605347 | 13049.81845 | 1169.494285 | $(4 a 2 h))_{8}{ }^{\text {LA }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 94.96556026 |
| 1180.612427 | 11601.27408 | 1180.508423 | $(4 \mathrm{a} 2 \mathrm{~h})_{8 L A}{ }_{4}$ | M+ethanolamine +Na | 88.10081993 |
| 1209.63916 | 7651.434319 | 1209.534996 | (4a2h) $)^{\text {LA }} 3$ | M+ethanolamine +Na | 86.11917832 |
| 1212.605591 | 5658.185677 | 1212.488822 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.30506928 |
| 1223.617432 | 6097.388143 | 1223.50296 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{6}$ | M+ethanolamine + Na | 93.56057463 |
| 1252.644043 | 8275.685694 | 1252.529533 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{5}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 91.42297006 |
| 1281.6658894 | 7024.973073 | 1281.556106 | $(4 \mathrm{a} 2 \mathrm{~h})_{9} \mathrm{LA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 85.66737694 |
| 1324.674805 | 5106.571028 | 1324.550643 | $(4 a 2 h) 8{ }_{8}{ }^{\text {L }}{ }_{6}$ | M+ethanolamine + Na | 93.73872615 |

Table S6. Peak list sorted by adduct and then mass and also colorcoded (for ease of visualization) for poly(4a2h-LA) positive mode MALDI-TOF-MS (Figs. 2 and S6); sodiated and protonated peaks. "M+H- $\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 286.1534729 | 399893.0661 | 286.13789 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 54.45940767 |
| 358.1759644 | 130060.0404 | 358.158995 | (4a2h) $)_{2} \mathrm{LA}$ | $\mathrm{M}+$ ethanolamine+Na | 47.37939082 |
| 387.2090759 | 215042.5396 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 60.71488698 |
| 430.2052612 | 31175.58651 | 430.180105 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine+Na | 58.47836687 |
| 459.2406921 | 204851.9006 | 459.206678 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}$ | M+ethanolamine +Na | 74.07152515 |
| 488.2664795 | 95265.03239 | 488.233251 | $(4 a 2 h)_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 68.05864191 |
| 502.2393799 | 10106.00743 | 502.201215 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}_{3}$ | M+ethanolamine+Na | 75.99520244 |
| 531.2701416 | 92994.80409 | 531.227788 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 79.72776078 |
| 560.3005981 | 147956.14 | 560.254361 | (4a2h) ${ }_{4} \mathrm{LA}$ | $\mathbf{M}+$ ethanolamine+Na | 82.52884443 |
| 589.3216553 | 39550.91519 | 589.280934 | (4a2h)5 | M+ethanolamine+Na | 69.10332687 |
| 603.2990112 | 34314.20048 | 603.248898 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{3}$ | $\mathbf{M}+$ ethanolamine+ Na | 83.07222801 |
| 632.3282471 | 103996.2997 | 632.275471 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{LA}_{2}$ | M+ethanolamine+Na | 83.47005763 |
| 661.3550415 | 81982.80034 | 661.302044 | (4a2h)sLA | $\mathbf{M}+$ ethanolamine+ Na | 80.14114652 |
| 675.3336792 | 12023.88379 | 675.270008 | $(4 a 2 h) 3 \mathrm{LA}_{4}$ | M+ethanolamine+Na | 94.28998511 |
| 690.3737183 | 17061.40467 | 690.328617 | $(4 a 2 h)_{6}$ | M+ethanolamine+Na | 65.33303254 |
| 704.3581543 | 52496.08738 | 704.296581 | (4a2h)4 $\mathrm{LA}_{3}$ | $\mathbf{M}+$ ethanolamine+Na | 87.42523911 |
| 733.3890991 | 77791.51538 | 733.323154 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 89.92641326 |
| 747.3637695 | 5890.81099 | 747.291118 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{5}$ | $\mathbf{M}+$ ethanolamine +Na | 97.21985081 |
| 762.4111938 | 40423.64349 | 762.349727 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}$ | $\mathbf{M}+$ ethanolamine +Na | 80.62814982 |
| 776.3899536 | 21872.28955 | 776.317691 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{LA}_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 93.08381586 |
| 791.4186401 | 8773.753124 | 791.3763 | (4a2h) ${ }_{7}$ | $\mathbf{M}+$ ethanolamine +Na | 53.50190169 |
| 805.4151001 | 51204.31114 | 805.344264 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{3}$ | $\mathbf{M}+$ ethanolamine+Na | 87.95753713 |
| 834.4454956 | 48137.42963 | 834.370837 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}_{2}$ | M+ethanolamine+Na | 89.47892435 |


| 848.4168701 | 9679.751505 | 848.338801 | (4a2h) ${ }_{4} \mathrm{LA}_{5}$ | $\mathbf{M}+$ ethanolamine +Na | 92.02587092 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 863.4682617 | 19473.37411 | 863.39741 | (4a2h) ${ }_{7} \mathrm{LA}$ | M+ethanolamine+Na | 82.06153757 |
| 877.4467163 | 25930.19433 | 877.365374 | (4a2h) $\mathbf{L S A}_{4}$ | $\mathbf{M}+$ ethanolamine+Na | 92.71201191 |
| 892.4644165 | 5588.566316 | 892.423983 | (4a2h)8 | $\mathrm{M}+$ ethanolamine+ Na | 45.30750492 |
| 906.4762573 | 37945.76358 | 906.391947 | (4a2h)6LA3 | $\mathbf{M}+$ ethanolamine+ Na | 93.01751221 |
| 935.505127 | 26021.91744 | 935.41852 | (4a2h) $\mathbf{L L A}_{2}$ | M+ethanolamine+Na | 92.5863142 |
| 949.4743652 | 12739.08974 | 949.386484 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{5}$ | M+ethanolamine+Na | 92.56634203 |
| 964.5188599 | 9479.26976 | 964.445093 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}$ | $\mathbf{M}+$ ethanolamine +Na | 76.48632725 |
| 978.5040283 | 23606.18319 | 978.413057 | $(4 \mathrm{a} 2 \mathrm{~h})_{6 L A}{ }_{4}$ | $\mathbf{M}+$ ethanolamine +Na | 92.97844029 |
| 1007.528015 | 24642.33925 | 1007.43963 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{3}$ | $\mathrm{M}+$ ethanolamine+ Na | 87.73244308 |
| 1021.504334 | 6606.929504 | 1021.407594 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{6}$ | $\mathrm{M}+$ ethanolamine +Na | 94.71194513 |
| 1036.551514 | 13597.46102 | 1036.466203 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 82.3091672 |
| 1050.529785 | 12881.88462 | 1050.434167 | (4a2h) ${ }_{6} \mathrm{LA}_{5}$ | $\mathrm{M}+$ ethanolamine+ Na | 91.0272752 |
| 1065.556641 | 5067.830871 | 1065.492776 | (4a2h) $)_{\text {L }}$ L | $\mathbf{M}+$ ethanolamine +Na | 59.93904552 |
| 1079.56189 | 16979.74775 | 1079.46074 | (4a2h) ${ }_{7} \mathrm{LA}_{4}$ | $\mathbf{M}+$ ethanolamine+Na | 93.70387106 |
| 1108.584229 | 13885.34982 | 1108.487313 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{3}$ | M+ethanolamine+Na | 87.43042781 |
| 1122.5625 | 6667.675158 | 1122.455277 | (4a2h)6LA6 | $\mathbf{M}+$ ethanolamine+ Na | 95.52540952 |
| 1137.598022 | 7258.682181 | 1137.513886 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{9} \mathrm{LA}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 73.96521575 |
| 1151.593872 | 10620.58957 | 1151.48185 | (4a2h) $\mathbf{L L A}_{5}$ | $\mathbf{M}+$ ethanolamine +Na | 97.28513741 |
| 1180.612427 | 11601.27408 | 1180.508423 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{4}$ | M+ethanolamine+Na | 88.10081993 |
| 1209.63916 | 7651.434319 | 1209.534996 | $(4 a 2 h){ }_{9} \mathrm{LA}_{3}$ | $\mathrm{M}+$ ethanolamine+ Na | 86.11917832 |
| 1223.617432 | 6097.388143 | 1223.50296 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{6}$ | M+ethanolamine+Na | 93.56057463 |
| 1252.644043 | 8275.685694 | 1252.529533 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{5}$ | $\mathbf{M}+$ ethanolamine +Na | 91.42297006 |
| 1281.665894 | 7024.973073 | 1281.556106 | $(4 \mathrm{a} 2 \mathrm{~h})_{9} \mathrm{LA}_{4}$ | $\mathbf{M}+$ ethanolamine+Na | 85.66737694 |
| 1324.674805 | 5106.571028 | 1324.550643 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{6}$ | $\mathrm{M}+$ ethanolamine +Na | 93.73872615 |
| 203.1046448 | 310156.1954 | 203.102637 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 9.885519113 |


| 275.1292725 | 121712.8269 | 275.123747 | (4a2h) ${ }_{2} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 20.08354808 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 304.1632996 | 219114.1385 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 42.67482277 |
| 347.1585693 | 28440.59692 | 347.144857 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 39.50032882 |
| 376.1924133 | 222088.7496 | 376.17143 | (4a2h) ${ }_{3} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.78129631 |
| 405.2236633 | 106882.5888 | 405.198003 | (4a2h)4 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.32787874 |
| 419.190094 | 8245.149344 | 419.165967 | (4a2h) $\mathbf{L}^{2} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 57.5595251 |
| 448.222168 | 98427.2027 | 448.19254 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 66.10544879 |
| 477.2516174 | 178425.8422 | 477.219113 | (4a2h) ${ }_{4} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 68.11217555 |
| 506.2850952 | 43273.87977 | 506.245686 | (4a2h) ${ }_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.84602633 |
| 520.2515259 | 35344.78352 | 520.21365 | $(4 a 2 h) 3 \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.80831443 |
| 549.2802124 | 129416.6699 | 549.240223 | $(4 a 2 h) 4 \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.80858234 |
| 578.3167114 | 106401.573 | 578.266796 | (4a2h) ${ }_{5} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.31902496 |
| 592.2870483 | 13760.19135 | 592.23476 | $(4 a 2 h) 3 \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.28988694 |
| 607.3388672 | 17711.5739 | 607.293369 | $(4 a 2 h)_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 74.91961929 |
| 621.3122559 | 64947.32759 | 621.261333 | $(4 a 2 h) 4 \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 81.96688945 |
| 650.3464355 | 106994.1225 | 650.287906 | $(4 a 2 h) 5 L_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.00559054 |
| 664.326416 | 6966.899115 | 664.25587 | (4a2h)3LA5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 106.203075 |
| 679.3762207 | 52742.31982 | 679.314479 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.88824824 |
| 693.3427734 | 27939.0476 | 693.282443 | $(4 a 2 h) 4 \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.02144214 |
| 708.3920898 | 7487.961148 | 708.341052 | $(4 \mathrm{a} 2 \mathrm{~h})_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 72.05264167 |
| 722.3729248 | 71394.94506 | 722.309016 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.47848163 |
| 736.3734741 | 5038.285162 | 736.27698 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 131.0568218 |
| 751.4077148 | 68382.78517 | 751.335589 | $(4 \mathrm{a} 2 \mathrm{~h})_{6 L A}{ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.99684223 |
| 765.3768311 | 12201.02662 | 765.303553 | $(4 \mathrm{a} 2 \mathrm{~h})_{4 \mathrm{LA}}^{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.75031334 |
| 780.4337158 | 23835.15482 | 780.362162 | (4a2h) ${ }_{7} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.69309262 |
| 794.40448 | 37760.8861 | 794.330126 | $(4 \mathrm{ah})_{5} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.60589202 |


| 823.43396 | 57880.14138 | 823.356699 | (4a2h) ${ }_{6} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.83656087 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 837.4158936 | 6953.435255 | 837.324663 | (4a2h)4 $\mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 108.9548165 |
| 852.4661255 | 37074.11195 | 852.383272 | (4a2h) ${ }_{7} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.20215157 |
| 866.4376221 | 18277.77089 | 866.351236 | (4a2h) SAA5 $^{\text {L }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.71252583 |
| 881.4817505 | 10476.86646 | 881.409845 | (4a2h) ${ }_{8} \mathrm{LA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 81.58008265 |
| 895.4604492 | 36897.1879 | 895.377809 | $(4 a 2 h) 6 \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.29647884 |
| 924.4910889 | 38167.21718 | 924.404382 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.79755082 |
| 938.463623 | 9315.307001 | 938.372346 | $(4 a 2 h) 5 \mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.27167194 |
| 953.5228271 | 17961.87844 | 953.430955 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.35951877 |
| 967.491333 | 20212.78028 | 967.398919 | $(4 a 2 h) 6 \mathrm{LA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.52833499 |
| 996.520874 | 28319.96781 | 996.425492 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.72418988 |
| 1009.531433 | 6000.015279 | 1010.393456 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 853.1556542 |
| 1025.549072 | 21264.62702 | 1025.452065 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{LA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 94.59951695 |
| 1039.526733 | 10551.73249 | 1039.420029 | $(4 a 2 h){ }_{6} \mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 102.6576331 |
| 1068.548218 | 17510.4293 | 1068.446602 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LAA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.10608187 |
| 1081.560547 | 5295.226308 | 1082.414566 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{LA}_{8}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 788.9944822 |
| 1097.582886 | 18080.18737 | 1097.473175 | $(4 a 2 h) 8 \mathrm{LA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.96667117 |
| 1111.551025 | 5835.771056 | 1111.441139 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{LA}_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.86838461 |
| 1140.577637 | 9875.486724 | 1140.467712 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{6}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.38564849 |
| 1169.605347 | 13049.81845 | 1169.494285 | $(4 a 2 h)_{8} \mathrm{LA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 94.96556026 |
| 1212.605591 | 5658.185677 | 1212.488822 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{LA}_{7}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.30506928 |

Table S7. Peak list sorted by mass for poly(4a2h-PA) positive mode MALDI-TOF-MS (Figs. 3 and S7); sodiated and protonated peaks. " $\mathrm{M}+\mathrm{H} / \mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 203.1010284 | 137059.9526 | 203.102637 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 7.919926712 |
| 286.1448669 | 187418.8148 | 286.13789 | (4a2h) ${ }_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 24.38314968 |
| 304.1588745 | 75173.2404 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 28.12593457 |
| 351.1687317 | 135816.358 | 351.155067 | (4a2h) $\mathbf{2}^{\text {PA }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 38.91354642 |
| 373.1539917 | 12961.63055 | 373.137009 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 45.51330635 |
| 387.204071 | 48415.42214 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 47.78857098 |
| 405.2185669 | 28441.91319 | 405.198003 | (4a2h) ${ }_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 50.75023778 |
| 434.2140503 | 100207.6384 | 434.190315 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine +Na | 54.66564357 |
| 452.2330322 | 140878.5462 | 452.20275 | (4a2h)3PA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 66.96603902 |
| 474.2145996 | 15305.32683 | 474.184692 | (4a2h)3PA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.07164593 |
| 488.2664795 | 18613.46149 | 488.233251 | (4a2h)4 | $\mathrm{M}+$ ethanolamine +Na | 68.05864191 |
| 499.2427368 | 84734.30597 | 499.207497 | $(4 a 2 h){ }_{2} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 70.59151998 |
| 506.2736511 | 10785.32174 | 506.245686 | (4a2h) ${ }_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.24021986 |
| 521.225769 | 17451.59261 | 521.189439 | $(4 a 2 h){ }_{2} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 69.70602296 |
| 535.2783813 | 78575.30583 | 535.237998 | (4a2h) ${ }_{3} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine +Na | 75.44932937 |
| 553.2901611 | 86865.58201 | 553.250433 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.80858908 |
| 568.239502 | 9981.198548 | 568.194186 | (4a2h) $\mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.75434124 |
| 575.2769165 | 11680.47309 | 575.232375 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{4} \mathrm{PA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.43219251 |
| 582.2903442 | 63748.79773 | 582.242745 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 81.75153475 |
| 589.3216553 | 7628.842563 | 589.280934 | (4a2h)5 | $\mathrm{M}+$ ethanolamine +Na | 69.10332687 |
| 600.3060913 | 119488.2189 | 600.25518 | $(4 a 2 h) 3 \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 84.81610938 |
| 622.2883301 | 20752.57418 | 622.237122 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 82.2967261 |
| 636.3359985 | 41665.89492 | 636.285681 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{4} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine+Na | 79.08009956 |


| 647.3161011 | 56645.12028 | 647.259927 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{2} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.78750477 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 654.352478 | 42251.91545 | 654.298116 | (4a2h) ${ }_{5} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 83.08449264 |
| 669.2957764 | 19558.7082 | 669.241869 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{2} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 80.54990206 |
| 676.331604 | 7467.884307 | 676.280058 | (4a2h) ${ }_{5} \mathrm{PA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 76.21990829 |
| 683.3511963 | 62198.33413 | 683.290428 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 88.93478748 |
| 701.3654785 | 94700.75957 | 701.302863 | $(4 \mathrm{ah}) 4 \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 89.28455779 |
| 723.3504028 | 17013.52254 | 723.284805 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.69433167 |
| 730.3626099 | 30695.39719 | 730.295175 | $(4 \mathrm{a} 2 \mathrm{~h}) \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 92.33918737 |
| 737.3949585 | 19879.16011 | 737.333364 | $(4 \mathrm{ah})_{5} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine +Na | 83.53683559 |
| 748.3719482 | 97161.30323 | 748.30761 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 85.9783345 |
| 755.4136353 | 18682.30459 | 755.345799 | $(4 \mathrm{ah}){ }_{6} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 89.80820982 |
| 770.3582764 | 23621.76205 | 770.289552 | $(4 a 2 h) 3 \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 89.21887467 |
| 784.4096069 | 41680.91185 | 784.338111 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 91.15448172 |
| 795.3866577 | 12735.68778 | 795.312357 | $(4 \mathrm{ah})_{2} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.42331267 |
| 802.4229736 | 56445.73916 | 802.350546 | (4a2h)5 $\mathbf{P A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.26931353 |
| 817.3812256 | 7418.371807 | 817.294299 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 106.3589776 |
| 824.4047241 | 11305.92803 | 824.332488 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.62983632 |
| 831.4217529 | 47071.89637 | 831.342858 | $(4 a 2 h) 3 \mathrm{PA}_{3}$ | $\mathbf{M}+$ ethanolamine +Na | 94.90059275 |
| 838.4538574 | 9516.653737 | 838.381047 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine +Na | 86.84645515 |
| 849.4319458 | 89008.77399 | 849.355293 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.24821724 |
| 856.4654541 | 8301.332901 | 856.393482 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathbf{P A}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 84.04092688 |
| 871.4228516 | 21206.89675 | 871.337235 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.2588125 |
| 878.4338379 | 6997.332038 | 878.347605 | $(4 a 2 h) 2 \mathrm{PA}_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 98.17626929 |
| 885.4654541 | 23825.25791 | 885.385794 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 89.97219352 |
| 896.4425049 | 37989.79361 | 896.36004 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.99973149 |
| 903.4847412 | 29744.27566 | 903.398229 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.76309563 |


| 918.4304199 | 12478.51659 | 918.341982 | $(4 a 2 h) 3 \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.30173044 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 925.4657593 | 7209.437315 | 925.380171 | $(4 a 2 h) 6 \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.4898541 |
| 932.4813843 | 40030.55625 | 932.390541 | $(4 \mathrm{ah}))_{4} \mathrm{PA}_{3}$ | M+ethanolamine+Na | 97.43050042 |
| 939.4924316 | 5238.321615 | 939.42873 | $(4 \mathbf{2 h})_{7} \mathbf{P A}$ | M+ethanolamine+Na | 67.80891298 |
| 943.4725952 | 5709.115833 | 943.364787 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 114.2805164 |
| 950.4935303 | 59568.77598 | 950.402976 | $(4 \mathrm{ah}))_{5} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.27987105 |
| 972.4821167 | 14865.20188 | 972.384918 | $(4 \mathrm{ah}))_{5} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.95907711 |
| 979.4909058 | 14764.80323 | 979.395288 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{4}$ | M+ethanolamine +Na | 97.62938741 |
| 986.5249023 | 12427.25465 | 986.433477 | $(4 \mathrm{ah}))_{6} \mathrm{PA}_{2}$ | M+ethanolamine +Na | 92.68272634 |
| 997.5007324 | 52151.51073 | 997.407723 | $(4 \mathrm{ah}){ }_{4} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.25115482 |
| 1004.534729 | 14309.10144 | 1004.445912 | $(\mathbf{4 a 2 h})_{7} \mathbf{P A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.42387523 |
| 1019.489014 | 14606.61892 | 1019.389665 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.45897316 |
| 1033.539795 | 25627.29835 | 1033.438224 | $(4 a 2 h) 5 \mathrm{PA}_{3}$ | M+ethanolamine+Na | 98.28446214 |
| 1044.519409 | 8049.175896 | 1044.41247 | $(4 a 2 h) 3 \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 102.3917112 |
| 1051.55249 | 33206.36071 | 1051.450659 | (4a2h)6 $\mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.84832011 |
| 1073.535278 | 8977.046764 | 1073.432601 | $(4 \mathrm{a2h}) 6 \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.65325285 |
| 1080.548462 | 19051.87764 | 1080.442971 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{4}$ | M+ethanolamine +Na | 97.636722 |
| 1087.57605 | 6355.580857 | 1087.48116 | $(\mathbf{4 a 2 h})_{7} \mathbf{P A}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 87.25650015 |
| 1098.560791 | 45769.06684 | 1098.455406 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.9392793 |
| 1105.587646 | 6610.105902 | 1105.493595 | $(4 a 2 h) 8 \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 85.07645854 |
| 1120.551636 | 12461.83988 | 1120.437348 | $(4 \mathrm{ah}))_{5} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 102.002794 |
| 1134.596558 | 13748.86947 | 1134.485907 | $(4 a 2 h) 6 \mathrm{PA}_{3}$ | M+ethanolamine +Na | 97.53371048 |
| 1145.575317 | 12678.41333 | 1145.460153 | $(4 \mathrm{ah}))_{4} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 100.5398396 |
| 1152.604248 | 15819.89438 | 1152.498342 | $\left(4 \mathbf{a}^{2 h}\right)_{7} \mathbf{P A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.89258339 |
| 1174.587891 | 5035.794002 | 1174.480284 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathbf{P A}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.62062698 |
| 1181.609253 | 15856.70201 | 1181.490654 | $(4 \mathrm{ah}))_{5} \mathrm{PA}_{4}$ | M+ethanolamine + Na | 100.3807602 |


| 1199.615845 | 29918.04387 | 1199.503089 | $(4 a 2 h) 6 \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 94.00203387 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1221.606689 | 8716.703415 | 1221.485031 | $(4 a 2 h) 6 \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.59880548 |
| 1235.64502 | 6982.360499 | 1235.53359 | (4a2h) $\mathbf{7}^{\text {PA }} 3$ | $\mathrm{M}+$ ethanolamine +Na | 90.18737402 |
| 1246.626343 | 15159.01792 | 1246.507836 | (4a2h) $\mathbf{5 P A}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.07101887 |
| 1253.661987 | 7539.637977 | 1253.546025 | $(4 a 2 h) 8 \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.50741312 |
| 1268.61731 | 5055.243486 | 1268.489778 | $(4 \mathrm{ah}))_{5} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 100.5381141 |
| 1282.659058 | 10846.77469 | 1282.538337 | $(4 \mathrm{ah}))_{6} \mathrm{PA}_{4}$ | M+ethanolamine +Na | 94.12632474 |
| 1300.675781 | 16468.14796 | 1300.550772 | $(\mathbf{4 a 2 h})_{7} \mathbf{P A}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.1202382 |
| 1322.665771 | 5318.028955 | 1322.532714 | $(\mathbf{4 a 2 h})_{7} \mathbf{P A}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 100.6080822 |
| 1347.688599 | 13565.55857 | 1347.555519 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.75632441 |
| 1383.717651 | 6322.22457 | 1383.58602 | $\left(4 \mathbf{a 2 h}_{7} \mathbf{P A}_{4}\right.$ | $\mathrm{M}+$ ethanolamine + Na | 95.13782887 |
| 1401.734741 | 8172.217933 | 1401.598455 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.236273 |
| 1448.746338 | 9724.808944 | 1448.603202 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.80959106 |
| 1549.799072 | 5826.728473 | 1549.650885 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.62622874 |

Table S8. Peak list sorted by adduct and then mass and also colorcoded (for ease of visualization) for poly( $4 \mathrm{a} 2 \mathrm{~h}-\mathrm{PA}$ ) positive mode MALDI-TOF-MS (Figs. 3 and S7); sodiated and protonated peaks. "M $\mathrm{H}+\mathrm{H} / \mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 286.1448669 | 187418.8148 | 286.13789 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 24.38314968 |
| 387.204071 | 48415.42214 | 387.185568 | $(4 \mathrm{a} 2 \mathrm{~h})_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 47.78857098 |
| 434.2140503 | 100207.6384 | 434.190315 | (4a2h) ${ }_{2} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine+Na | 54.66564357 |
| 488.2664795 | 18613.46149 | 488.233251 | (4a2h) ${ }_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 68.05864191 |
| 535.2783813 | 78575.30583 | 535.237998 | (4a2h) ${ }_{3} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine +Na | 75.44932937 |
| 582.2903442 | 63748.79773 | 582.242745 | $(4 \mathrm{ah})_{2} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 81.75153475 |
| 589.3216553 | 7628.842563 | 589.280934 | (4a2h)5 | $\mathrm{M}+$ ethanolamine+Na | 69.10332687 |
| 636.3359985 | 41665.89492 | 636.285681 | (4a2h) ${ }_{4} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine +Na | 79.08009956 |
| 683.3511963 | 62198.33413 | 683.290428 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 88.93478748 |
| 730.3626099 | 30695.39719 | 730.295175 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 92.33918737 |
| 737.3949585 | 19879.16011 | 737.333364 | (4a2h)sPA | $\mathrm{M}+$ ethanolamine+Na | 83.53683559 |
| 784.4096069 | 41680.91185 | 784.338111 | $(4 \mathrm{ah}){ }_{4} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 91.15448172 |
| 831.4217529 | 47071.89637 | 831.342858 | $(4 a 2 h) 3 \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 94.90059275 |
| 838.4538574 | 9516.653737 | 838.381047 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{6} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine+Na | 86.84645515 |
| 878.4338379 | 6997.332038 | 878.347605 | $(4 \mathrm{ah}){ }_{2} \mathrm{PA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 98.17626929 |
| 885.4654541 | 23825.25791 | 885.385794 | $(4 \mathrm{ah})_{5} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 89.97219352 |
| 932.4813843 | 40030.55625 | 932.390541 | $(4 \mathrm{a} 2 \mathrm{~h}){ }_{4} \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 97.43050042 |
| 939.4924316 | 5238.321615 | 939.42873 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{PA}$ | $\mathrm{M}+$ ethanolamine+Na | 67.80891298 |
| 979.4909058 | 14764.80323 | 979.395288 | $(4 \mathrm{ah}))_{3} \mathrm{PA}_{4}$ | $\mathrm{M}+$ ethanolamine +Na | 97.62938741 |
| 986.5249023 | 12427.25465 | 986.433477 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 92.68272634 |
| 1033.539795 | 25627.29835 | 1033.438224 | $(4 \mathrm{ah})_{5} \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 98.28446214 |
| 1080.548462 | 19051.87764 | 1080.442971 | $(4 \mathrm{ah}){ }_{4} \mathrm{PA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 97.636722 |
| 1087.57605 | 6355.580857 | 1087.48116 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{PA}_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 87.25650015 |


| 1134.596558 | 13748.86947 | 1134.485907 | $(4 a 2 h) 6 \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 97.53371048 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1181.609253 | 15856.70201 | 1181.490654 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{4}$ | $\mathbf{M}+$ ethanolamine+Na | 100.3807602 |
| 1235.64502 | 6982.360499 | 1235.53359 | (4a2h) ${ }_{7} \mathrm{PA}_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 90.18737402 |
| 1282.659058 | 10846.77469 | 1282.538337 | $(4 \mathrm{ah}){ }_{6} \mathrm{PA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 94.12632474 |
| 1383.717651 | 6322.22457 | 1383.58602 | $(4 \mathrm{ah})_{7} \mathrm{PA}_{4}$ | $\mathrm{M}+$ ethanolamine+Na | 95.13782887 |
| 203.1010284 | 137059.9526 | 203.102637 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 7.919926712 |
| 304.1588745 | 75173.2404 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 28.12593457 |
| 351.1687317 | 135816.358 | 351.155067 | (4a2h) ${ }_{2} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 38.91354642 |
| 405.2185669 | 28441.91319 | 405.198003 | $(4 \mathrm{a} 2 \mathrm{~h})_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 50.75023778 |
| 452.2330322 | 140878.5462 | 452.20275 | (4a2h) ${ }_{3} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 66.96603902 |
| 499.2427368 | 84734.30597 | 499.207497 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 70.59151998 |
| 506.2736511 | 10785.32174 | 506.245686 | $(4 \mathrm{a} 2 \mathrm{~h})_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.24021986 |
| 553.2901611 | 86865.58201 | 553.250433 | $(4 \mathrm{ah}){ }_{4} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 71.80858908 |
| 600.3060913 | 119488.2189 | 600.25518 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 84.81610938 |
| 647.3161011 | 56645.12028 | 647.259927 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.78750477 |
| 654.352478 | 42251.91545 | 654.298116 | (4a2h) ${ }_{\text {P }} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 83.08449264 |
| 701.3654785 | 94700.75957 | 701.302863 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 89.28455779 |
| 748.3719482 | 97161.30323 | 748.30761 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 85.9783345 |
| 755.4136353 | 18682.30459 | 755.345799 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 89.80820982 |
| 795.3866577 | 12735.68778 | 795.312357 | $(4 \mathrm{ah}){ }_{2} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.42331267 |
| 802.4229736 | 56445.73916 | 802.350546 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.26931353 |
| 849.4319458 | 89008.77399 | 849.355293 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.24821724 |
| 856.4654541 | 8301.332901 | 856.393482 | $(4 \mathrm{ah})_{7} \mathrm{PA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 84.04092688 |
| 896.4425049 | 37989.79361 | 896.36004 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.99973149 |
| 903.4847412 | 29744.27566 | 903.398229 | $(4 \mathrm{ah}){ }_{6} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.76309563 |
| 943.4725952 | 5709.115833 | 943.364787 | (4a2h)2 $\mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 114.2805164 |


| 950.4935303 | 59568.77598 | 950.402976 | $(4 \mathrm{ah}))_{5} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.27987105 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 997.5007324 | 52151.51073 | 997.407723 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.25115482 |
| 1004.534729 | 14309.10144 | 1004.445912 | (4a2h) ${ }_{7} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.42387523 |
| 1044.519409 | 8049.175896 | 1044.41247 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 102.3917112 |
| 1051.55249 | 33206.36071 | 1051.450659 | $(4 a 2 h) 6 \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.84832011 |
| 1098.560791 | 45769.06684 | 1098.455406 | (4a2h) ${ }_{5} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.9392793 |
| 1105.587646 | 6610.105902 | 1105.493595 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 85.07645854 |
| 1145.575317 | 12678.41333 | 1145.460153 | $(4 \mathrm{ah})_{4} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 100.5398396 |
| 1152.604248 | 15819.89438 | 1152.498342 | $(4 a 2 h)_{7} \mathbf{P A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.89258339 |
| 1199.615845 | 29918.04387 | 1199.503089 | $(4 a 2 h) 6 \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 94.00203387 |
| 1246.626343 | 15159.01792 | 1246.507836 | (4a2h) $)^{\text {PA }}{ }_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.07101887 |
| 1253.661987 | 7539.637977 | 1253.546025 | $(4 \mathrm{ah}))_{8} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.50741312 |
| 1300.675781 | 16468.14796 | 1300.550772 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.1202382 |
| 1347.688599 | 13565.55857 | 1347.555519 | $(4 a 2 h) 6 \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.75632441 |
| 1401.734741 | 8172.217933 | 1401.598455 | $(4 \mathrm{a} 2 \mathrm{~h})_{8} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.236273 |
| 1448.746338 | 9724.808944 | 1448.603202 | $(4 \mathrm{a} 2)_{7} \mathrm{PAA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.80959106 |
| 1549.799072 | 5826.728473 | 1549.650885 | (4a2h) ${ }_{8} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.62622874 |
| 373.1539917 | 12961.63055 | 373.137009 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 45.51330635 |
| 474.2145996 | 15305.32683 | 474.184692 | (4a2h)3PA | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.07164593 |
| 521.225769 | 17451.59261 | 521.189439 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 69.70602296 |
| 568.239502 | 9981.198548 | 568.194186 | (4a2h)PA3 | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.75434124 |
| 575.2769165 | 11680.47309 | 575.232375 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.43219251 |
| 622.2883301 | 20752.57418 | 622.237122 | $(4 \mathrm{ah})_{3} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 82.2967261 |
| 669.2957764 | 19558.7082 | 669.241869 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 80.54990206 |
| 676.331604 | 7467.884307 | 676.280058 | (4a2h) ${ }_{5} \mathrm{PA}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 76.21990829 |
| 723.3504028 | 17013.52254 | 723.284805 | $(4 \mathrm{ah}){ }_{4} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.69433167 |


| 770.3582764 | 23621.76205 | 770.289552 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 89.21887467 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 817.3812256 | 7418.371807 | 817.294299 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 106.3589776 |
| 824.4047241 | 11305.92803 | 824.332488 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.62983632 |
| 871.4228516 | 21206.89675 | 871.337235 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.2588125 |
| 918.4304199 | 12478.51659 | 918.341982 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 96.30173044 |
| 925.4657593 | 7209.437315 | 925.380171 | (4a2h) ${ }_{6} \mathrm{PA}_{2}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.4898541 |
| 972.4821167 | 14865.20188 | 972.384918 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.95907711 |
| 1019.489014 | 14606.61892 | 1019.389665 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 97.45897316 |
| 1073.535278 | 8977.046764 | 1073.432601 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 95.65325285 |
| 1120.551636 | 12461.83988 | 1120.437348 | $(4 \mathrm{ah})_{5} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 102.002794 |
| 1174.587891 | 5035.794002 | 1174.480284 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{PA}_{3}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.62062698 |
| 1221.606689 | 8716.703415 | 1221.485031 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 99.59880548 |
| 1268.61731 | 5055.243486 | 1268.489778 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{PA}_{5}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 100.5381141 |
| 1322.665771 | 5318.028955 | 1322.532714 | $(4 \mathrm{a} 2 \mathrm{~h})_{7} \mathrm{PA}_{4}$ | $\mathrm{M}+\mathrm{Na}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 100.6080822 |

Table S9. Peak list sorted by mass for poly(4a2h-SA) positive mode MALDI-TOF-MS (Figs. 3 and S7); sodiated and protonated peaks. " $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 203.1082611 | 119627.7243 | 203.102637 | $(4 a 2 h)_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 27.69096494 |
| 286.1534729 | 138296.5019 | 286.13789 | (4a2h) ${ }_{2}$ | $\mathrm{M}+$ ethanolamine+Na | 54.45940767 |
| 304.1632996 | 79547.49402 | 304.15032 | (4a2h) ${ }_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 42.67482277 |
| 335.145752 | 45208.77016 | 335.127147 | (4a2h) ${ }_{2} \mathrm{SA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.51610237 |
| 387.2140808 | 64667.31716 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 73.64120297 |
| 405.2236633 | 31962.4951 | 405.198003 | $(4 a 2 h)_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.32787874 |
| 418.1920471 | 34312.11079 | 418.162395 | (4a2h) ${ }_{2} \mathrm{SA}$ | $\mathrm{M}+$ ethanolamine +Na | 70.91053465 |
| 436.2015381 | 61431.72595 | 436.17483 | (4a2h)3SA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 61.23252458 |
| 467.1861877 | 10903.82403 | 467.151657 | (4a2h) $\mathbf{S A}^{\text {S }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.91763142 |
| 488.2720947 | 24187.12329 | 488.233251 | (4a2h)4 | $\mathbf{M}+$ ethanolamine +Na | 79.55977378 |
| 506.2850952 | 10026.07337 | 506.245686 | (4a2h)5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.84602633 |
| 519.2550659 | 37057.09256 | 519.210078 | (4a2h) ${ }_{3} \mathrm{SA}$ | $\mathbf{M}+$ ethanolamine +Na | 86.64685049 |
| 537.2675781 | 36690.33756 | 537.222513 | (4a2h) ${ }_{4} \mathrm{SA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 83.88539927 |
| 550.2336426 | 7235.814997 | 550.186905 | (4a2h) $\mathbf{S A}^{\text {S }}$ | M+ethanolamine +Na | 84.94854671 |
| 568.2455444 | 21107.43784 | 568.19934 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{SA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 81.31729614 |
| 589.3278198 | 7174.857536 | 589.280934 | $(4 a 2 h)_{5}$ | $\mathrm{M}+$ ethanolamine +Na | 79.56446797 |
| 620.3115845 | 20729.73856 | 620.2578 | (4a2h) ${ }_{4} \mathrm{SA}$ | $\mathrm{M}+$ ethanolamine +Na | 86.77597667 |
| 638.3253784 | 16077.44795 | 638.270196 | (4a2h) ${ }_{5} \mathrm{SA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.45620357 |
| 651.2932739 | 12060.34898 | 651.234588 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{SA}_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 90.11487885 |
| 669.3023071 | 17325.23547 | 669.247023 | (4a2h)4SA2 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 82.60646234 |
| 700.2886353 | 5774.033225 | 700.22385 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{SA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.52077603 |
| 721.3756104 | 8676.655065 | 721.305444 | (4a2h) ${ }_{5}$ SA | $\mathrm{M}+$ ethanolamine +Na | 97.27689231 |
| 739.3847046 | 6555.611082 | 739.317879 | (4a2h)6SA1 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.38816982 |


| 752.355896 | $\mathbf{8 9 5 8 . 8 5 0 7 9 1}$ | 752.2823 | $(4 a 2 h)_{4}$ SA $_{2}$ | M+ethanolamine+Na | 97.8688437 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 770.3652954 | 10377.94951 | 770.294706 | $(4 a 2 h) 5 S A 2$ | M+H-H2O (cyclic) | 91.6394848 |
| 801.3500977 | $\mathbf{6 0 8 4 . 1 5 5 7 5 8}$ | $\mathbf{8 0 1 . 2 7 1 5 3 3}$ | $(4 a 2 h)_{4}$ SA $_{3}$ | M+H-H2O (cyclic) | $\mathbf{9 8 . 0 4 9 9 7 7 7 7}$ |

Table S10. Peak list sorted by adduct and then mass and also colorcoded (for ease of visualization) for poly(4a2h-SA) positive mode MALDI-TOF-MS (Figs. 3 and S7); sodiated and protonated peaks. "M+H- $\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 286.1534729 | 138296.5019 | 286.13789 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathbf{M}+$ ethanolamine +Na | 54.45940767 |
| 387.2140808 | 64667.31716 | 387.185568 | $(4 a 2 h)_{3}$ | M+ethanolamine+Na | 73.64120297 |
| 418.1920471 | 34312.11079 | 418.162395 | (4a2h) ${ }_{2} \mathrm{SA}$ | $\mathrm{M}+$ ethanolamine +Na | 70.91053465 |
| 488.2720947 | 24187.12329 | 488.233251 | (4a2h)4 | $\mathrm{M}+$ ethanolamine +Na | 79.55977378 |
| 519.2550659 | 37057.09256 | 519.210078 | (4a2h) ${ }_{3} \mathrm{SA}$ | $\mathrm{M}+$ ethanolamine+Na | 86.64685049 |
| 550.2336426 | 7235.814997 | 550.186905 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{SA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 84.94854671 |
| 589.3278198 | 7174.857536 | 589.280934 | (4a2h)5 | $\mathrm{M}+$ ethanolamine +Na | 79.56446797 |
| 620.3115845 | 20729.73856 | 620.2578 | (4a2h)4SA | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 86.77597667 |
| 651.2932739 | 12060.34898 | 651.234588 | $\left(4 \mathrm{a} 2 \mathrm{~h} 3_{3} \mathrm{SA}_{2}\right.$ | $\mathbf{M}+$ ethanolamine +Na | 90.11487885 |
| 721.3756104 | 8676.655065 | 721.305444 | (4a2h) ${ }_{5} \mathrm{SA}$ | $\mathbf{M}+$ ethanolamine +Na | 97.27689231 |
| 752.355896 | 8958.850791 | 752.2823 | (4a2h)4SA ${ }_{2}$ | $\mathbf{M}+$ ethanolamine $+\mathbf{N a}$ | 97.8688437 |
| 203.1082611 | 119627.7243 | 203.102637 | $(4 a 2 h)_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 27.69096494 |
| 304.1632996 | 79547.49402 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 42.67482277 |
| 335.145752 | 45208.77016 | 335.127147 | (4a2h) ${ }_{2} \mathrm{SA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 55.51610237 |
| 405.2236633 | 31962.4951 | 405.198003 | (4a2h)4 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.32787874 |
| 436.2015381 | 61431.72595 | 436.17483 | (4a2h) ${ }_{3} \mathrm{SA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 61.23252458 |
| 467.1861877 | 10903.82403 | 467.151657 | $\left(4 \mathrm{a} 2 \mathrm{~h} 2_{2} \mathrm{SA}_{2}\right.$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.91763142 |
| 506.2850952 | 10026.07337 | 506.245686 | (4a2h)5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.84602633 |
| 537.2675781 | 36690.33756 | 537.222513 | (4a2h)4SA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 83.88539927 |
| 568.2455444 | 21107.43784 | 568.19934 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{SA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 81.31729614 |
| 638.3253784 | 16077.44795 | 638.270196 | (4a2h) ${ }_{5} \mathrm{SA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.45620357 |
| 669.3023071 | 17325.23547 | 669.247023 | (4a2h)4SA ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 82.60646234 |
| 700.2886353 | 5774.033225 | 700.22385 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{SA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.52077603 |


| 739.3847046 | 6555.611082 | 739.317879 | $(4 \mathrm{a} 2 \mathrm{~h}) 6 \mathrm{SA} 1$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 90.38816982 |
| :--- | :--- | :--- | :--- | :--- | :--- |
| 770.3652954 | 10377.94951 | 770.294706 | $(4 \mathrm{a} 2 \mathrm{~h}) 5 \mathrm{SA} 2$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.6394848 |
| 801.3500977 | 6084.155758 | 801.271533 | $(4 \mathrm{a} 2 \mathrm{~h}) 4 \mathrm{SA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 98.04997777 |

Table S11. Peak list sorted by mass for poly(4a2h-MA) positive mode MALDI-TOF-MS (Figs. 3 and S7)); sodiated and protonated peaks. "M+H-H2O (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 203.1118774 | 61217.23984 | 203.102637 | (4a2h) ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 45.49641076 |
| 286.1534729 | 59098.51242 | 286.13789 | (4a2h) ${ }_{2}$ | M+ethanolamine + Na | 54.45940767 |
| 304.1677246 | 38826.89896 | 304.15032 | $(4 a 2 h)_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 57.22370767 |
| 317.1872253 | 46808.71956 | 317.170717 | (4a2h) $\mathbf{2}^{\text {MA }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 52.04875833 |
| 387.2140808 | 21416.12425 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 73.64120297 |
| 400.2329102 | 22744.67772 | 400.205965 | (4a2h) $\mathbf{2}^{\text {MA }}$ | $\mathrm{M}+$ ethanolamine +Na | 67.32822186 |
| 405.2236633 | 15895.80825 | 405.198003 | (4a2h)4 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.32787874 |
| 418.2492065 | 51464.6359 | 418.2184 | (4a2h)3MA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.66137645 |
| 431.2703247 | 25806.98307 | 431.238797 | $(4 a 2 h){ }_{2} \mathrm{MA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.10962562 |
| 488.2720947 | 7823.612667 | 488.233251 | (4a2h)4 | $\mathrm{M}+$ ethanolamine +Na | 79.55977378 |
| 501.2944946 | 19271.03419 | 501.253648 | (4a2h)3MA | $\mathrm{M}+$ ethanolamine + Na | 81.4889411 |
| 506.2850952 | 5339.720985 | 506.245686 | (4a2h) ${ }^{\text {a }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.84602633 |
| 514.3158569 | 10465.10194 | 514.274045 | $(4 a 2 h){ }_{2} \mathrm{MA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 81.30282756 |
| 519.3071899 | 30128.49217 | 519.266083 | (4a2h)4MA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.16353936 |
| 532.3309326 | 33528.91296 | 532.28648 | (4a2h) $\mathbf{3}^{\mathbf{M A}}{ }_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 83.51257954 |
| 545.3502197 | 12750.27915 | 545.306877 | $(4 a 2 h){ }_{2} \mathrm{MA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.48318429 |
| 602.3504639 | 10335.776 | 602.301331 | (4a2h)4MA | M+ethanolamine + Na | 81.57522567 |
| 615.3770752 | 10687.08136 | 615.321728 | (4a2h) $\mathbf{3}^{\mathbf{M A}}{ }_{\mathbf{2}}$ | $\mathrm{M}+$ ethanolamine +Na | 89.948383888 |
| 620.3622437 | 14124.68433 | 620.313766 | (4a2h) ${ }_{5} \mathrm{MA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.15021149 |
| 633.3897705 | 25799.72225 | 633.334163 | (4a2h) $\mathbf{4}^{\text {MA }}{ }_{\mathbf{2}}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.8012134 |


| 646.4109497 | 19576.36245 | 646.35456 | (4a2h)3 $\mathbf{M A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.24268457 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 716.4333496 | 7321.153815 | 716.369411 | (4a2h) $\mathbf{4}^{(1)}{ }^{\text {a }}$ | M+ethanolamine +Na | 89.25368395 |
| 721.4234009 | 5715.255404 | 721.361449 | (4a2h)6MA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 85.88188222 |
| 734.4498291 | 14634.74789 | 734.381846 | (4a2h) $\mathbf{5 M A}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.57187166 |
| 747.4680176 | 16847.41138 | 747.402243 | (4a2h) $\mathbf{M M A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.00425556 |
| 760.4887085 | 7543.129313 | 760.42264 | (4a2h)3MA4 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.88391498 |
| 835.5036621 | 7313.521706 | 835.429529 | (4a2h) $\mathbf{6}^{(1)}{ }^{\text {M }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.73651987 |
| 848.5279541 | 11279.42892 | 848.449926 | (4a2h) $\mathbf{5 M A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.96547682 |
| 949.5841064 | 6558.255092 | 949.497609 | (4a2h) $\mathbf{6}^{(1)}{ }^{\text {M }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.09811776 |
| 962.6081543 | 6296.545552 | 962.518006 | (4a2h) $\mathbf{S M A}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.65881619 |

Table S12. Peak list sorted by adduct and then mass and also colorcoded (for ease of visualization) for poly(4a2h-MA) positive mode MALDI-TOF-MS (Figs. 3 and S7)); sodiated and protonated peaks. "M $\mathrm{H}-\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic)": possibility of both dehydrated or cyclic species.

| Observed Mass (m/z) | Intensity | Calculated Mass | Identity | Adduct | PPM |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 286.1534729 | 59098.51242 | 286.13789 | (4a2h) ${ }_{2}$ | M+ethanolamine+Na | 54.45940767 |
| 387.2140808 | 21416.12425 | 387.185568 | $(4 a 2 h)_{3}$ | $\mathrm{M}+$ ethanolamine+Na | 73.64120297 |
| 400.2329102 | 22744.67772 | 400.205965 | (4a2h) ${ }_{2} \mathrm{MA}$ | M+ethanolamine +Na | 67.32822186 |
| 488.2720947 | 7823.612667 | 488.233251 | (4a2h) ${ }_{4}$ | $\mathrm{M}+$ ethanolamine+ Na | 79.55977378 |
| 501.2944946 | 19271.03419 | 501.253648 | (4a2h)3MA | M+ethanolamine+Na | 81.4889411 |
| 514.3158569 | 10465.10194 | 514.274045 | $(4 \mathrm{a} 2 \mathrm{~h})_{2} \mathrm{MA}_{2}$ | $\mathrm{M}+$ ethanolamine +Na | 81.30282756 |
| 602.3504639 | 10335.776 | 602.301331 | (4a2h)4MA | $\mathrm{M}+$ ethanolamine+Na | 81.57522567 |
| 615.3770752 | 10687.08136 | 615.321728 | (4a2h)3MA ${ }_{3}$ | $\mathrm{M}+$ ethanolamine +Na | 89.94838388 |
| 716.4333496 | 7321.153815 | 716.369411 | (4a2h) $\mathbf{4}^{(1)}{ }^{\text {a }}$ | $\mathrm{M}+$ ethanolamine+Na | 89.25368395 |
| 203.1118774 | 61217.23984 | 203.102637 | $(4 \mathrm{a} 2 \mathrm{~h})_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 45.49641076 |
| 304.1677246 | 38826.89896 | 304.15032 | $(4 \mathrm{a} 2 \mathrm{~h})_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 57.22370767 |
| 317.1872253 | 46808.71956 | 317.170717 | (4a2h)2MA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 52.04875833 |
| 405.2236633 | 15895.80825 | 405.198003 | $(4 a 2 h)_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 63.32787874 |


| 418.2492065 | 51464.6359 | 418.2184 | (4a2h)3MA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.66137645 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 431.2703247 | 25806.98307 | 431.238797 | (4a2h) $\mathbf{2}^{\mathbf{M A}}{ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 73.10962562 |
| 506.2850952 | 5339.720985 | 506.245686 | (4a2h)5 | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 77.84602633 |
| 519.3071899 | 30128.49217 | 519.266083 | (4a2h)4MA | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.16353936 |
| 532.3309326 | 33528.91296 | 532.28648 | (4a2h)3MA ${ }_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 83.51257954 |
| 545.3502197 | 12750.27915 | 545.306877 | (4a2h) $2^{(M A}{ }_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 79.48318429 |
| 620.3622437 | 14124.68433 | 620.313766 | (4a2h) ${ }_{5} \mathrm{MA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 78.15021149 |
| 633.3897705 | 25799.72225 | 633.334163 | (4a2h) $\mathbf{4}^{(1)}{ }^{\text {M }}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.8012134 |
| 646.4109497 | 19576.36245 | 646.35456 | (4a2h) $\mathbf{3}^{(1)}{ }_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 87.24268457 |
| 721.4234009 | 5715.255404 | 721.361449 | (4a2h) ${ }_{6} \mathrm{MA}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 85.88188222 |
| 734.4498291 | 14634.74789 | 734.381846 | $(4 \mathrm{a} 2 \mathrm{~h})_{5} \mathrm{MA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 92.57187166 |
| 747.4680176 | 16847.41138 | 747.402243 | $(4 \mathrm{a} 2 \mathrm{~h})_{4} \mathrm{MA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.00425556 |
| 760.4887085 | 7543.129313 | 760.42264 | $(4 \mathrm{a} 2 \mathrm{~h})_{3} \mathrm{MA}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 86.88391498 |
| 835.5036621 | 7313.521706 | 835.429529 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{MA}_{2}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 88.73651987 |
| 848.5279541 | 11279.42892 | 848.449926 | (4a2h) $\mathbf{S M A}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.96547682 |
| 949.5841064 | 6558.255092 | 949.497609 | $(4 \mathrm{a} 2 \mathrm{~h})_{6} \mathrm{MA}_{3}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 91.09811776 |
| 962.6081543 | 6296.545552 | 962.518006 | (4a2h) $\mathbf{5 M A}_{4}$ | $\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}$ (cyclic) | 93.65881619 |

Table S13. Compositional ratios of polyesters containing 4a2h and one other $\alpha \mathrm{HA}$ which assemble into droplets. Cells with " $X$ " indicate that the conditions were not tested in this study, as these conditions were generally not near the "Droplet"/"No Droplet" transition. Relevant microscopy images can be found in Figs. 4, 5, S11, S12, and S23 and in reference ${ }^{6}$; the associated phase diagram is depicted as Fig. 6.

|  | 4a2h:GA | 4a2h:LA | 4a2h:PA | 4a2h:SA | 4a2h:MA |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{0 \%} \mathbf{4 a 2 h}$ | No Droplets | Droplets | Droplets | Droplets | Droplets |
| $\mathbf{1 0 \%} \mathbf{4 a 2 h}$ | X | Droplets | X | X | X |
| $\mathbf{2 0 \%} \mathbf{4 a 2 h}$ | X | No Droplets | Droplets | Droplets | Droplets |
| $\mathbf{4 0 \%} \mathbf{4 a 2 h}$ | X | X | Droplets | X | Droplets |
| $\mathbf{5 0 \%} \mathbf{4 a 2 h}$ | No Droplets | No Droplets | Droplets | No Droplets | Droplets |
| $\mathbf{6 0 \%} \mathbf{4 a 2 h}$ | X | X | Droplets | X | No Droplets |
| $\mathbf{8 0 \%} \mathbf{4 a 2 h}$ | X | X | No Droplets | X | No Droplets |

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