Supporting Information:

Role of Intermolecular Charge Fluxes in the Hydrogen-Bond-Induced Frequency Shifts of the OH Stretching Mode of Water

Hajime Torii^{1,2}* and Ryota Ukawa¹

Supplementary figure

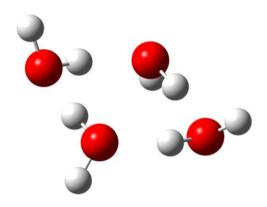


Figure S1. Structure of the water cyclic tetramer with the S₄ symmetry.

¹ Department of Applied Chemistry and Biochemical Engineering, Faculty of Engineering, Shizuoka University, 3-5-1 Johoku, Naka-ku, Hamamatsu 432-8561, Japan.

² Department of Optoelectronics and Nanostructure Science, Graduate School of Science and Technology, Shizuoka University, 3-5-1 Johoku, Naka-ku, Hamamatsu 432-8561, Japan.

Supplementary table

Table S1. Coefficients of the polynomials of the fitting to some of the correlations among the structural, vibrational, and electrostatic properties

x	у	species	equation
field on H / $10^{-2} E_h e^{-1} a_0^{-1}$	dipole derivative / D Å-1 amu-1/2	all non-H-bonded	y = 0.8355 x + 0.6313 y = 0.2683 x + 1.0753
field on H / 10^{-2} $E_{\rm h}$ e^{-1} a_0^{-1} hydrogen-bond distance / Å hydrogen-bond distance / Å	frequency (scaled) / cm ⁻¹ dipole derivative / D Å ⁻¹ amu ^{-1/2} frequency (scaled) / cm ⁻¹	all H-bonded H-bonded	$y = -20.362 x^{2} - 8.534 x + 3715.03$ $y = 4.6991 (x - 2)^{2} - 5.2002 (x - 2) + 2.9443$ $y = 2612.1 (x - 2)^{3} - 1563.7 (x - 2)^{2} + 568.7 (x - 2) + 3520.3$