

## SUPPORTING INFORMATION

### Synthesis of Sulfonamide-Based Ynamides and Ynamines in Water

Lei Zhao,<sup>a</sup> Hongyi Yang,<sup>a</sup> Ruikun Li,<sup>a</sup> Ye Tao,<sup>a</sup> Xiao-Feng Guo,<sup>b</sup> Edward A. Anderson,<sup>c</sup> Andrew Whiting<sup>d</sup> and Na Wu \*<sup>e</sup>

[a] State Key Laboratory for the Chemistry and Molecular Engineering of Medicinal Resources, School of Chemistry and Pharmaceutical Science, Guangxi Normal University, Guilin, Guangxi, 541004, China.

[b] College of Biomedical Engineering, Taiyuan University of Technology, Taiyuan, Shanxi, 030024, China.

[c] Chemistry Research Laboratory, Chemistry Department, Oxford University, 12 Mansfield Road, OX1 3TA, Oxford, United Kingdom

[d] Chemistry Department, Science Site, Chemistry Department, Durham University, South Road, DH1 3LE, Durham, United Kingdom

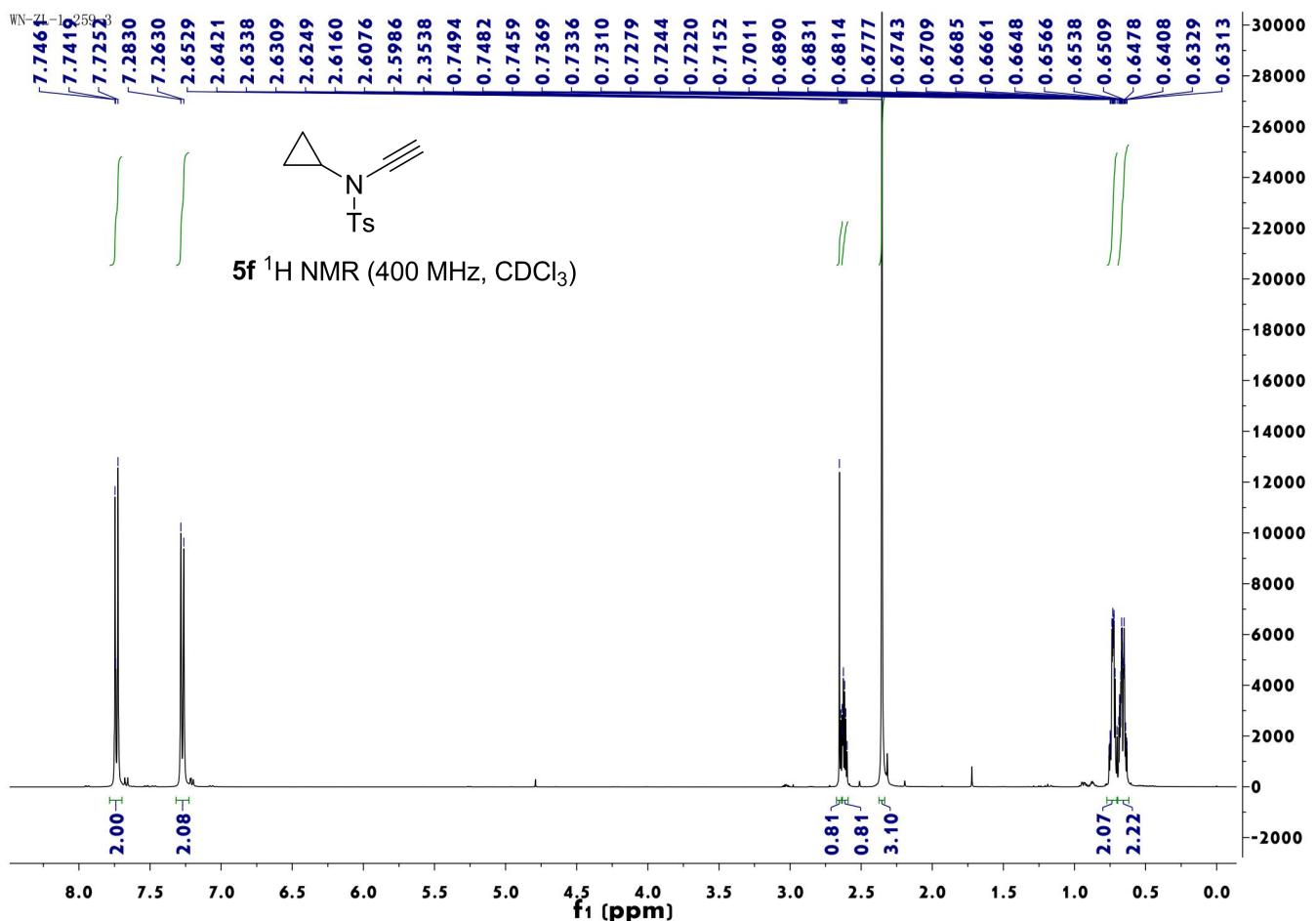
[e] School of Chemistry and Bioscience, University of Bradford, Richmond Road, Bradford, BD7 1DP, United Kingdom

Corresponding author: n.wu1@brad.ac.uk

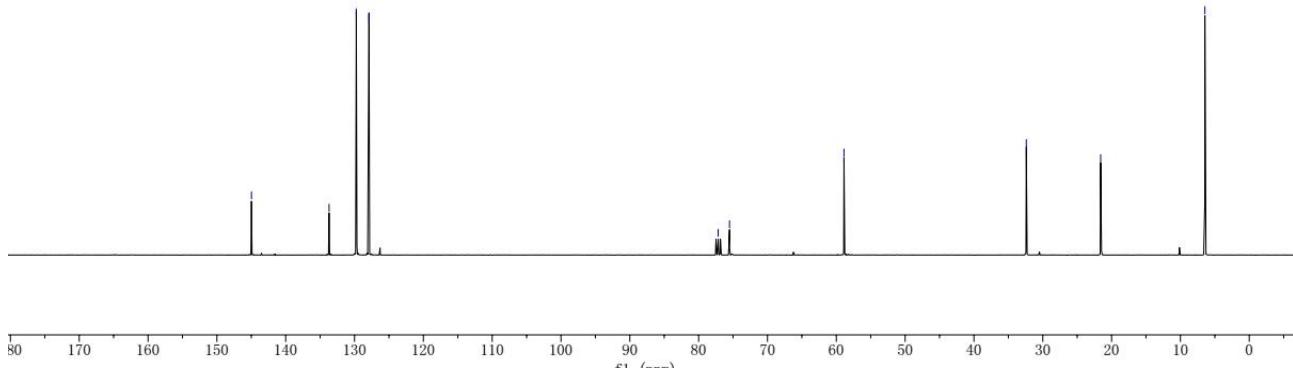
### Contents

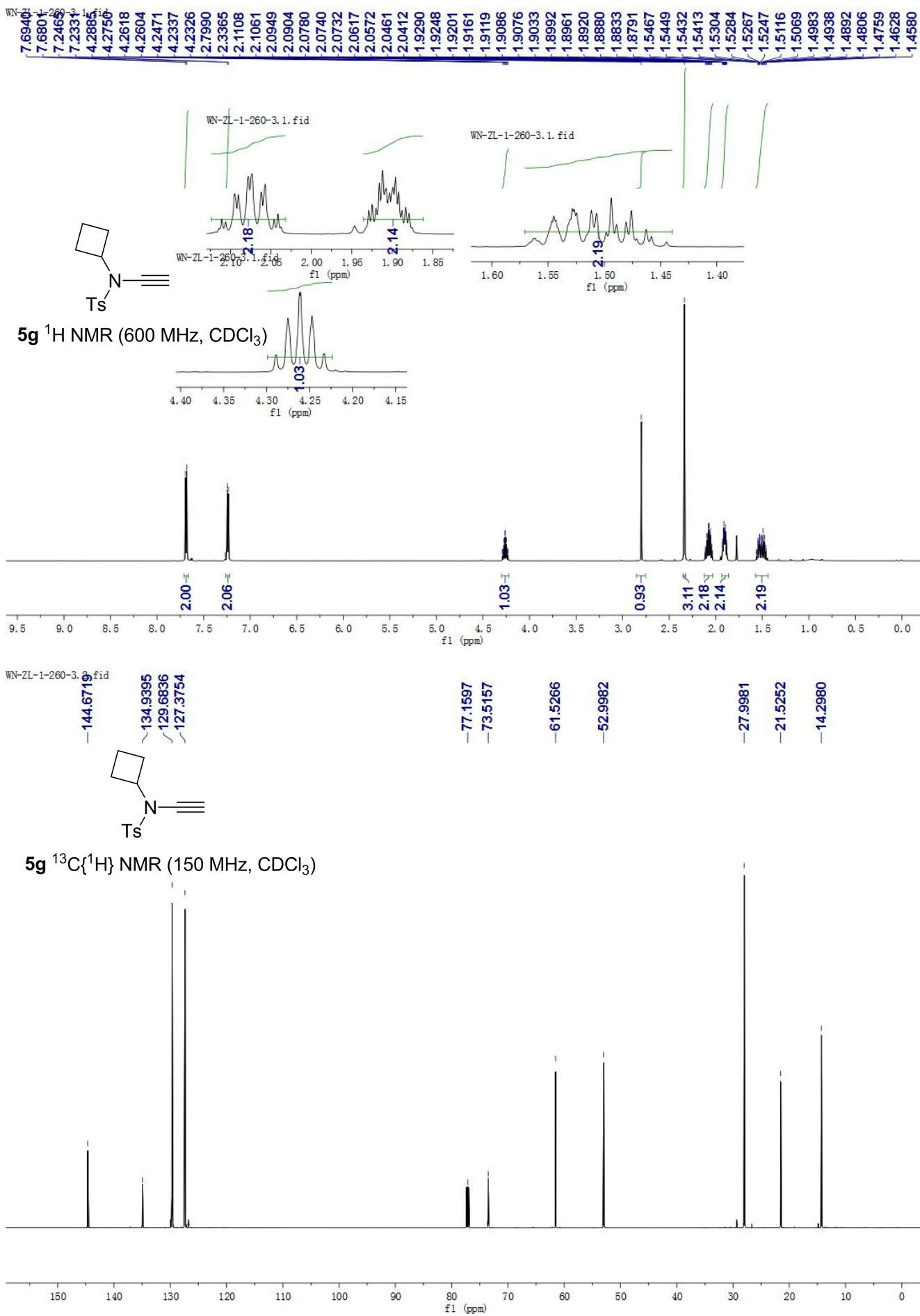
|  |     |
|--|-----|
| 1. <sup>1</sup> H and <sup>13</sup> C NMR spectra..... | S2  |
| 2. X-ray crystal structure of 8aj.....                 | S43 |
| 3. Reference.....                                      | S45 |

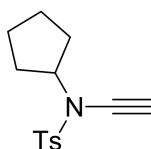
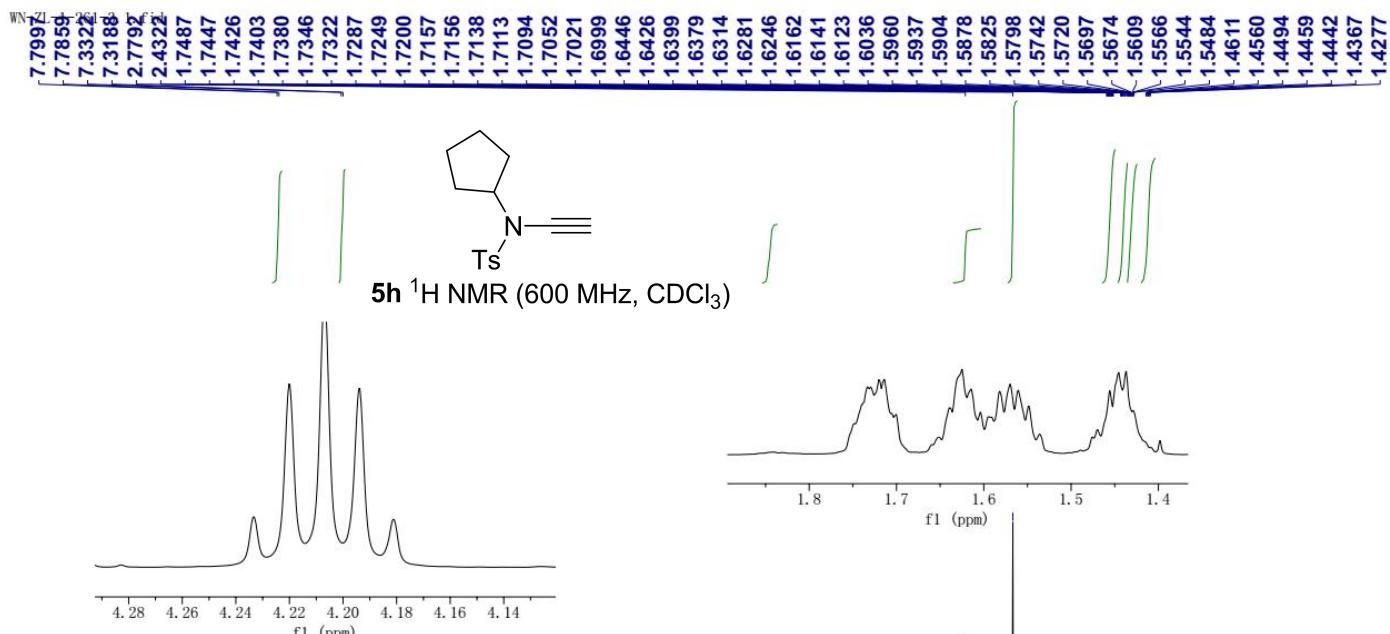
# 1. $^1\text{H}$ and $^{13}\text{C}$ NMR spectra



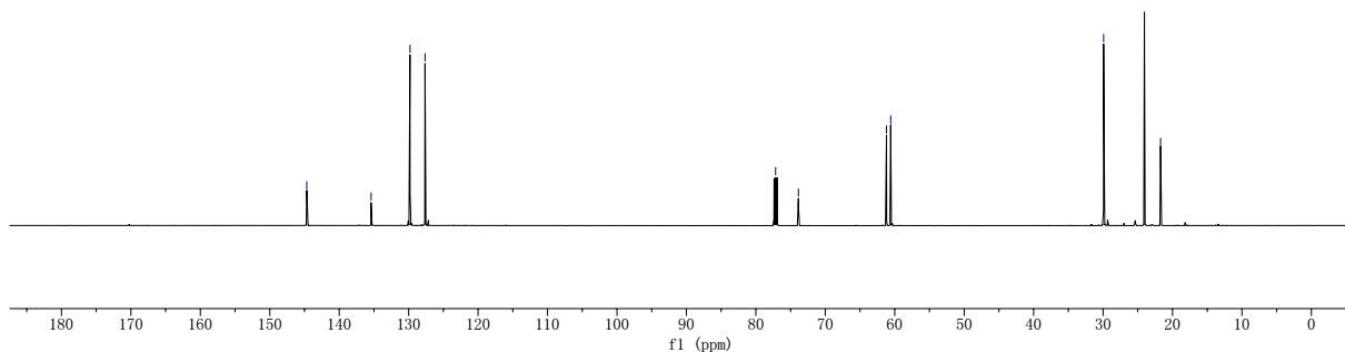
ZL-1259-3. 2. fid



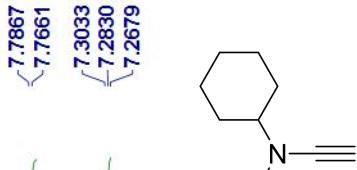




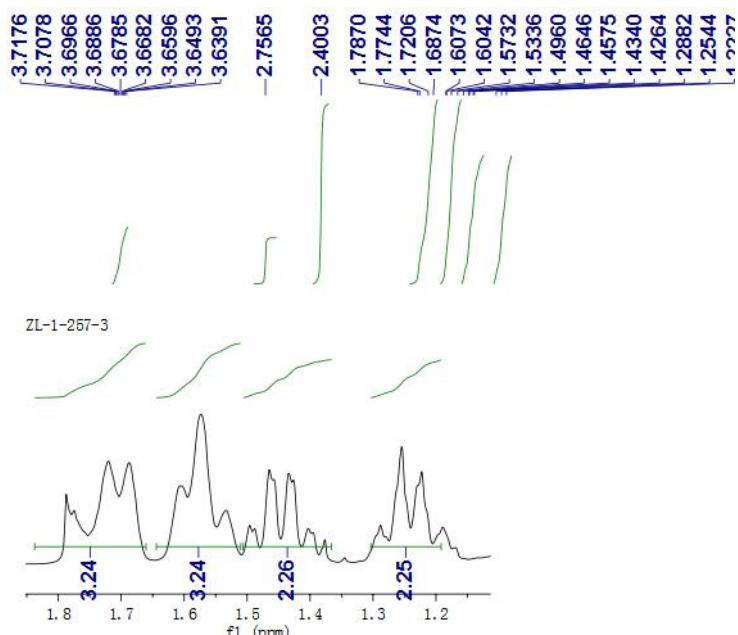
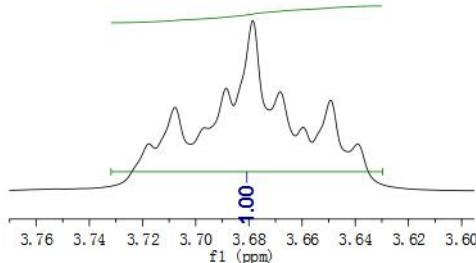
**5h**  $^{13}\text{C}\{\text{H}\}$  NMR (150 MHz,  $\text{CDCl}_3$ )



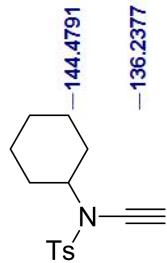
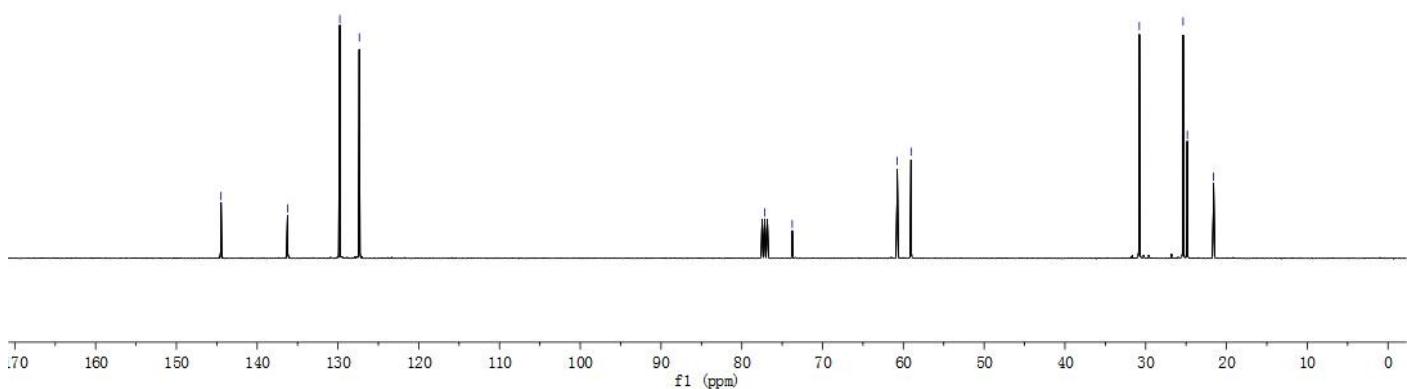
ZL-1-257-3

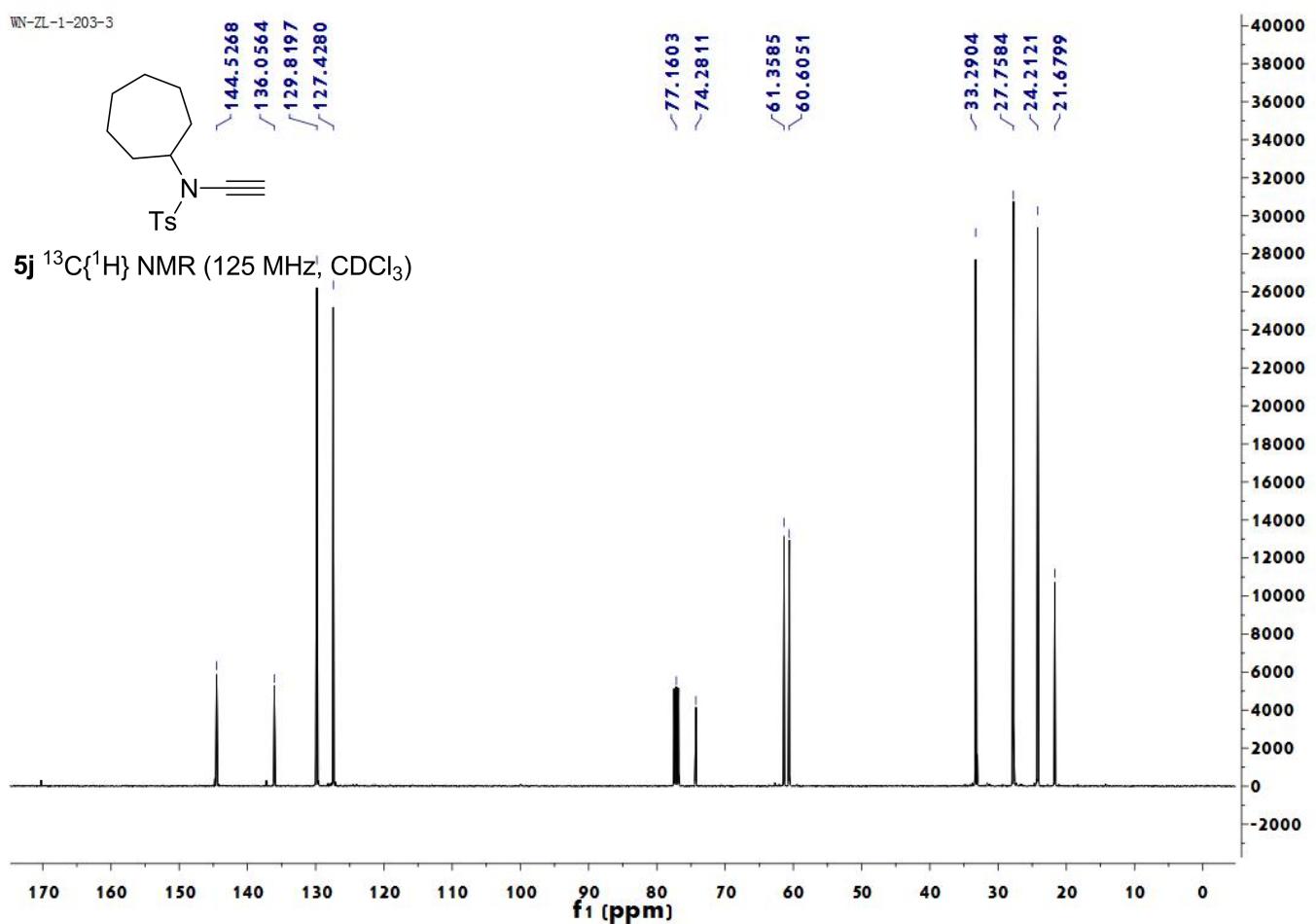
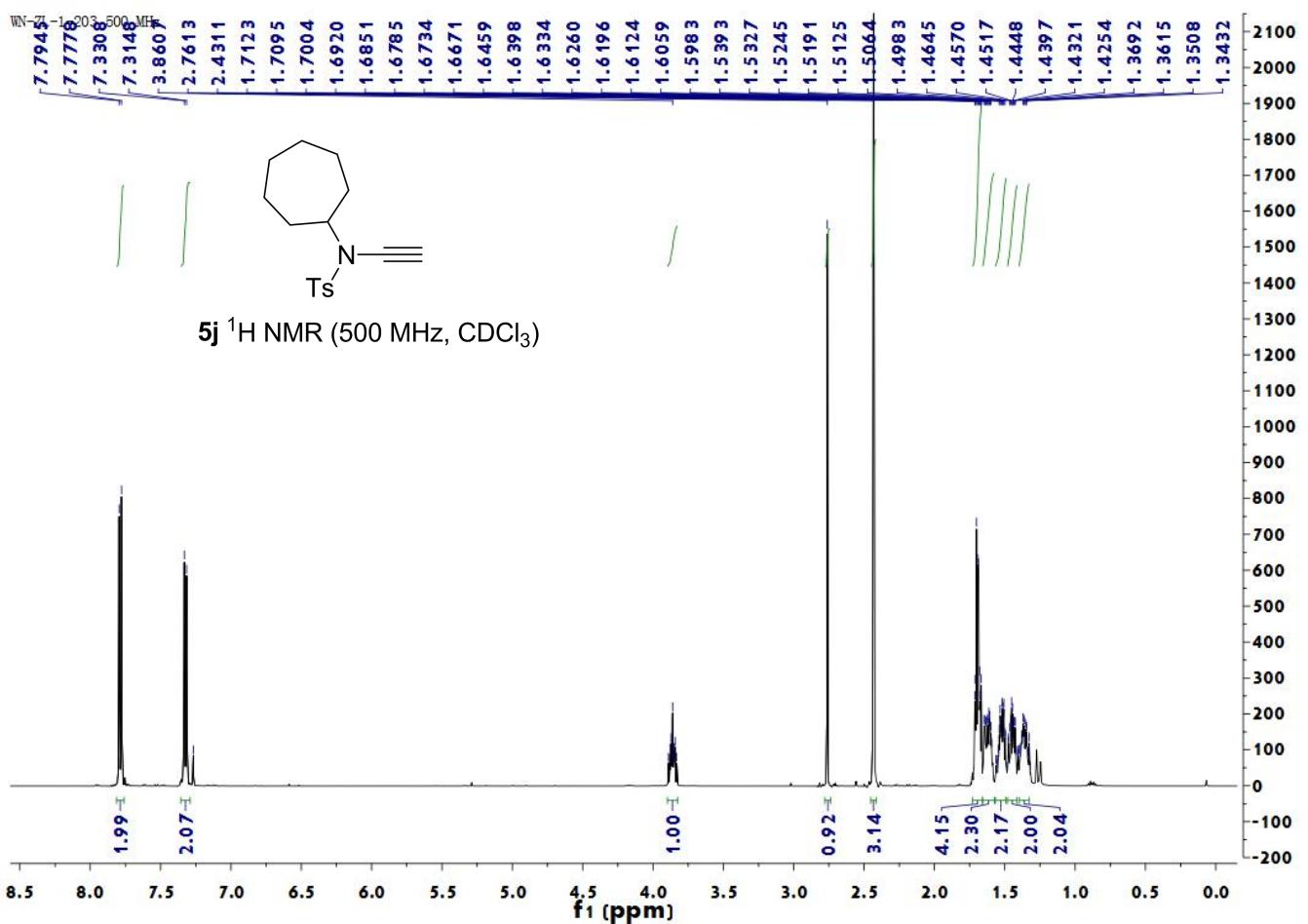
**5i**  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )

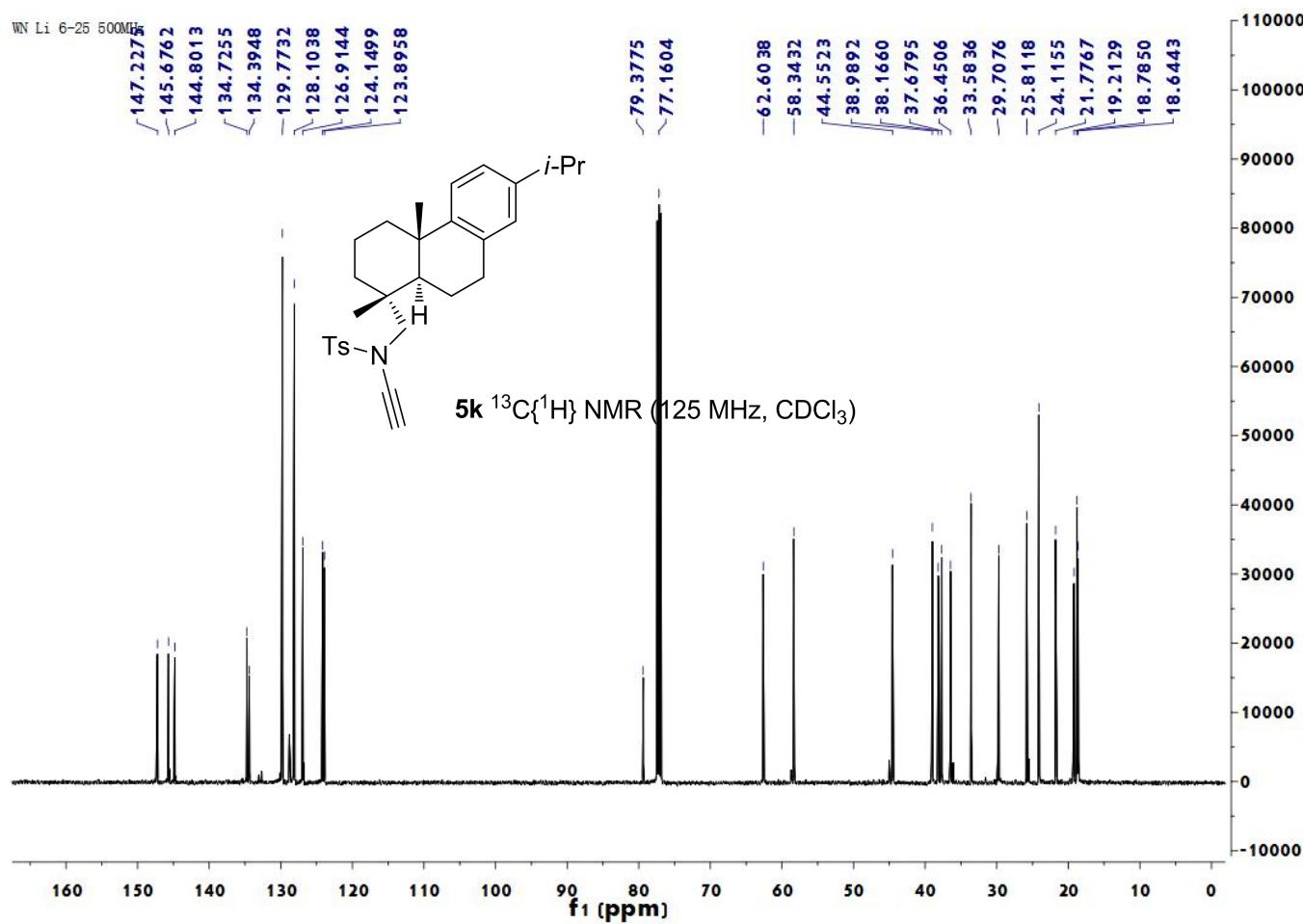
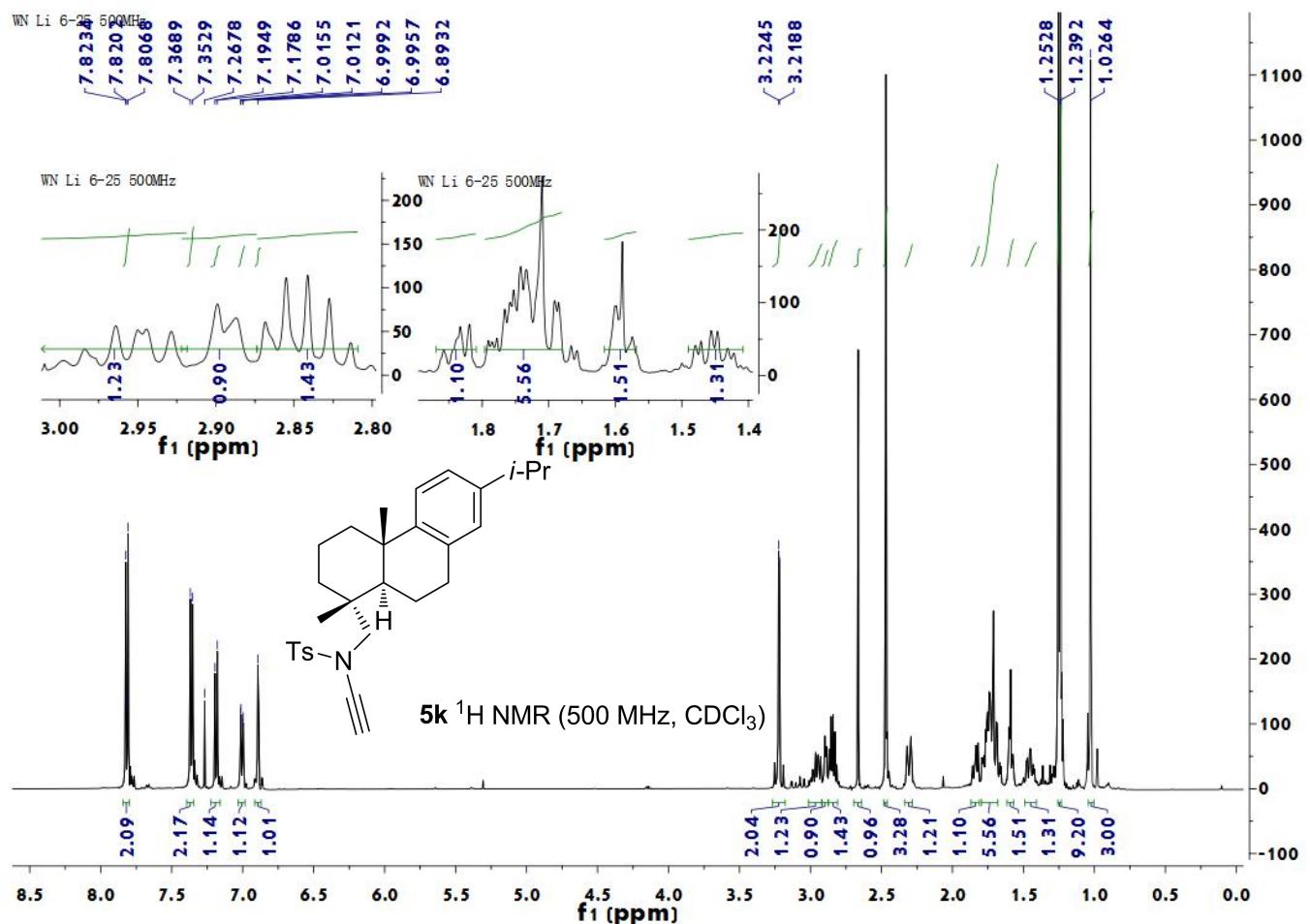
ZL-1-257-3



ZL-1-257-3

**5i**  $^{13}\text{C}\{\text{H}\}$  NMR (100 MHz,  $\text{CDCl}_3$ )

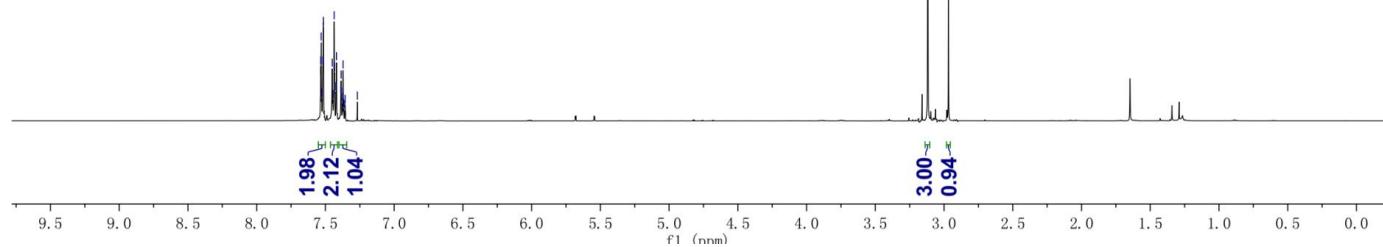




WN Li 6-21 500MHz



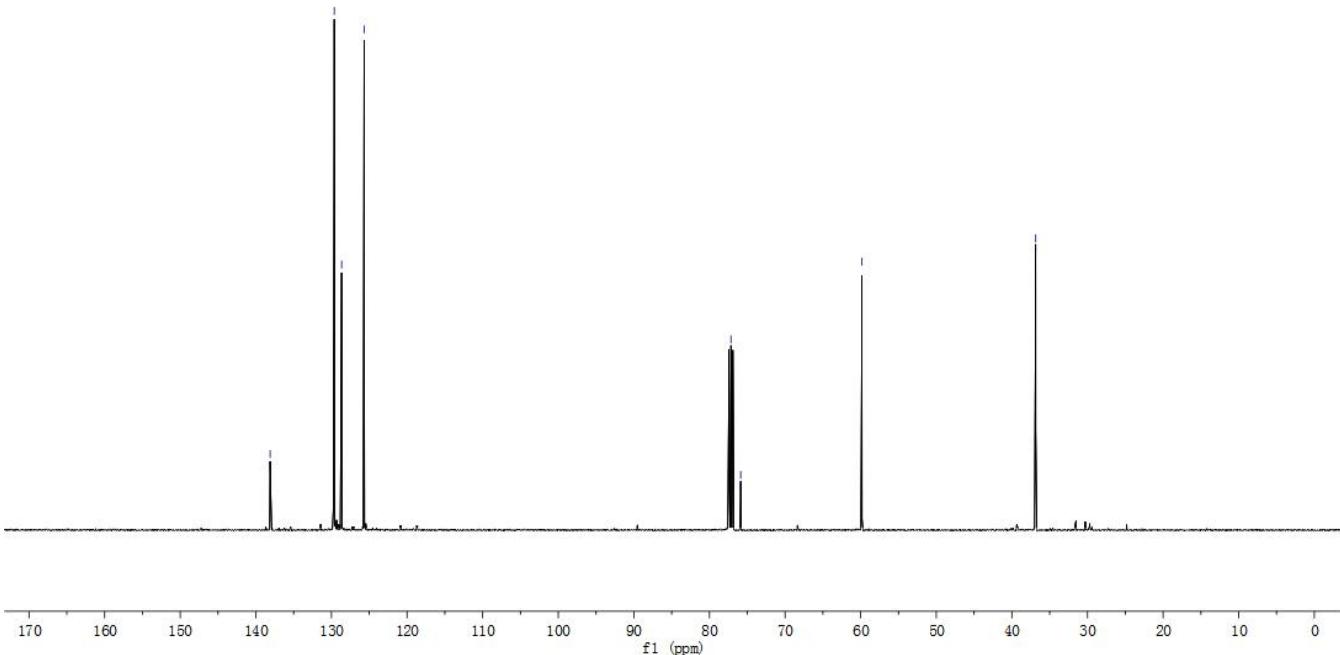
**5I**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )



WN Li 6-24 500MHz

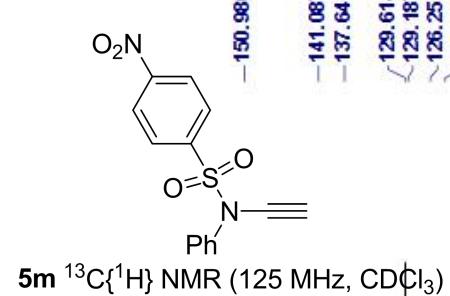
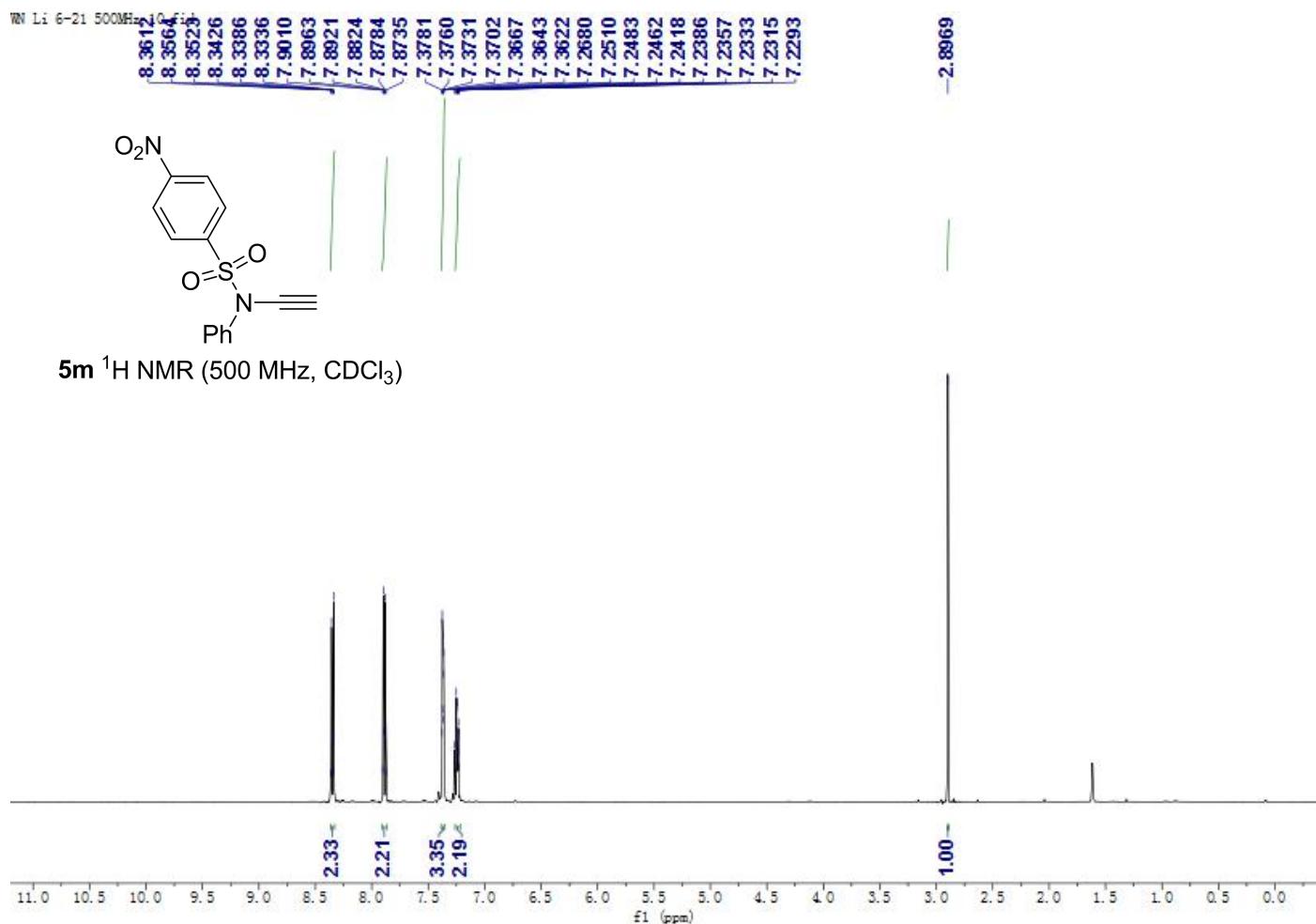


**5I**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

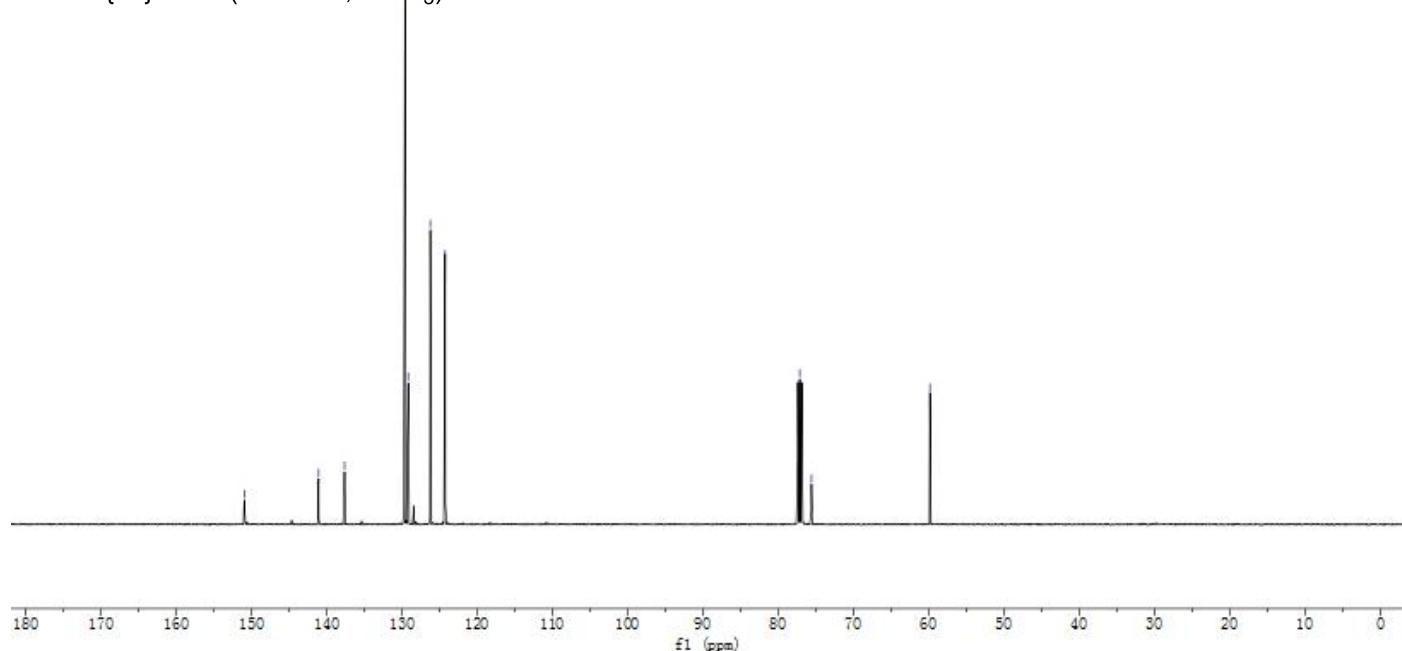


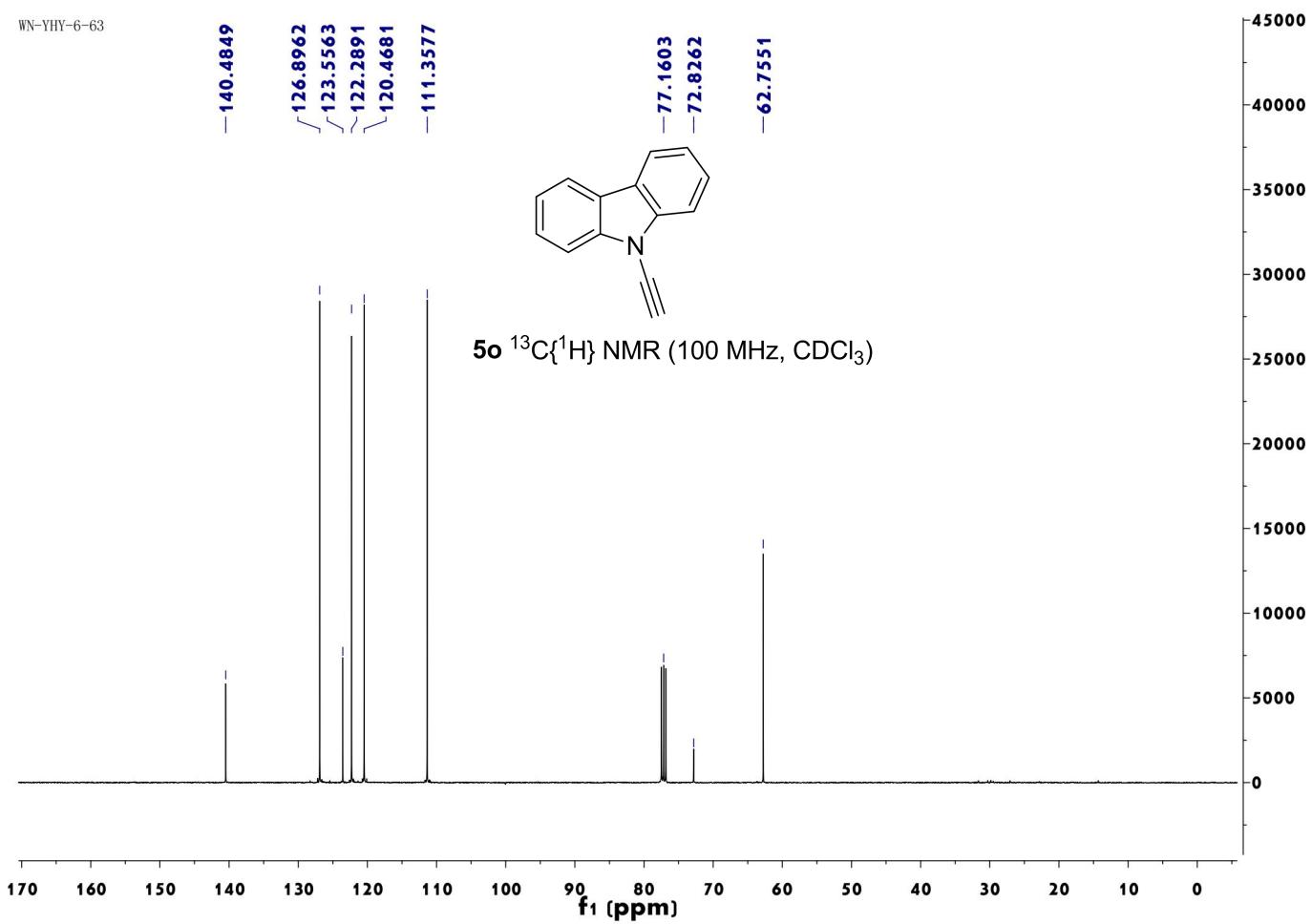
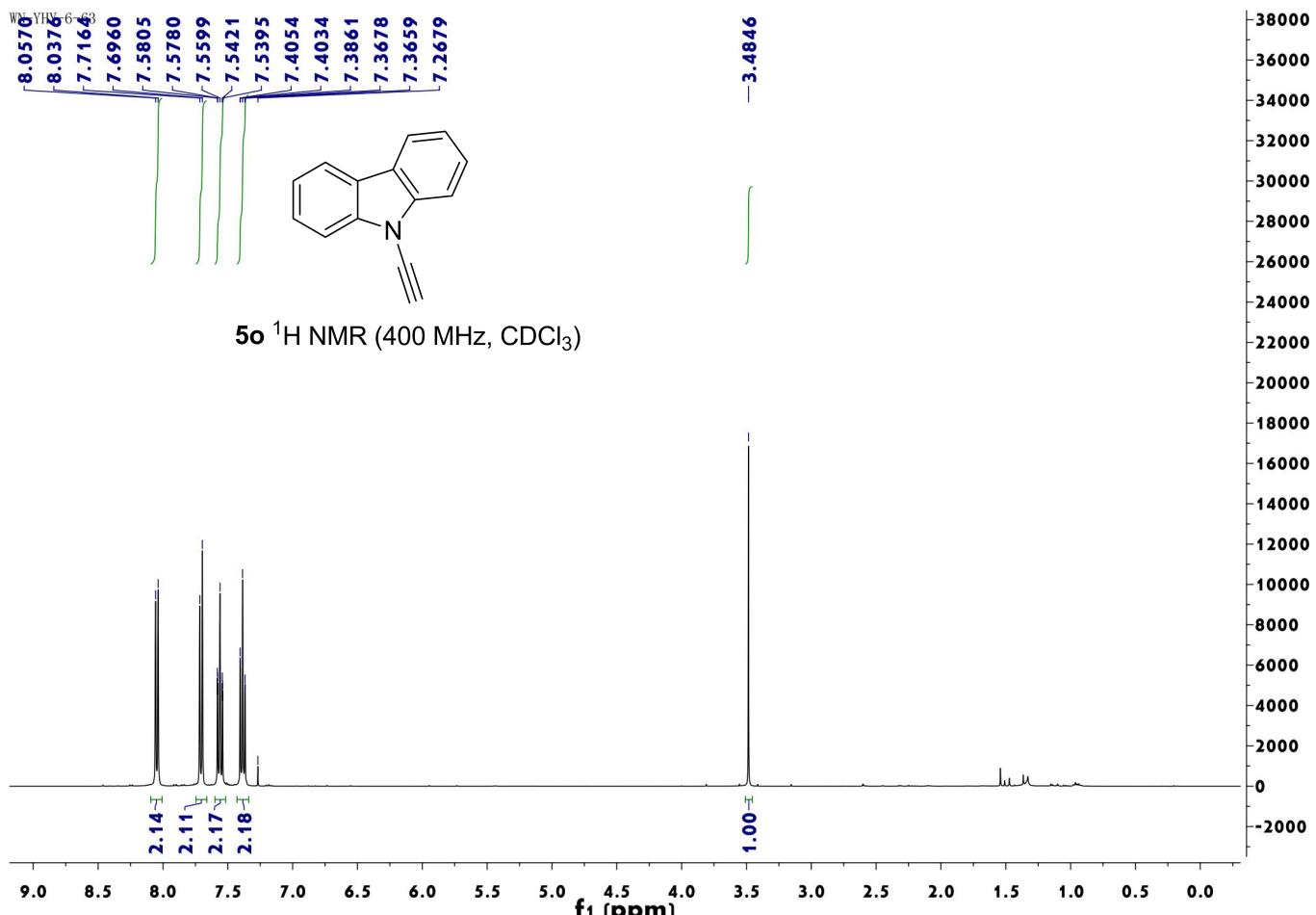


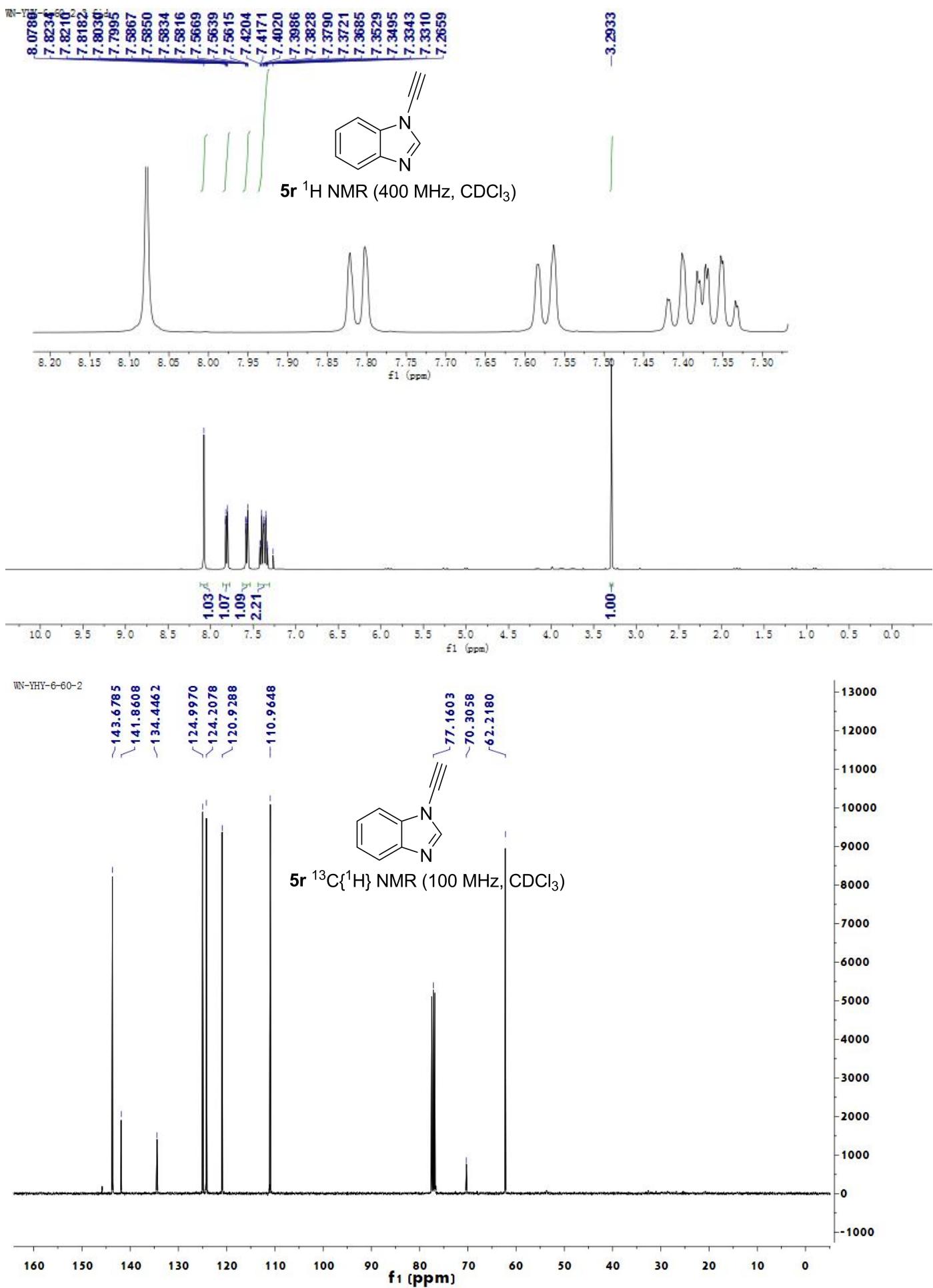
**5m**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )

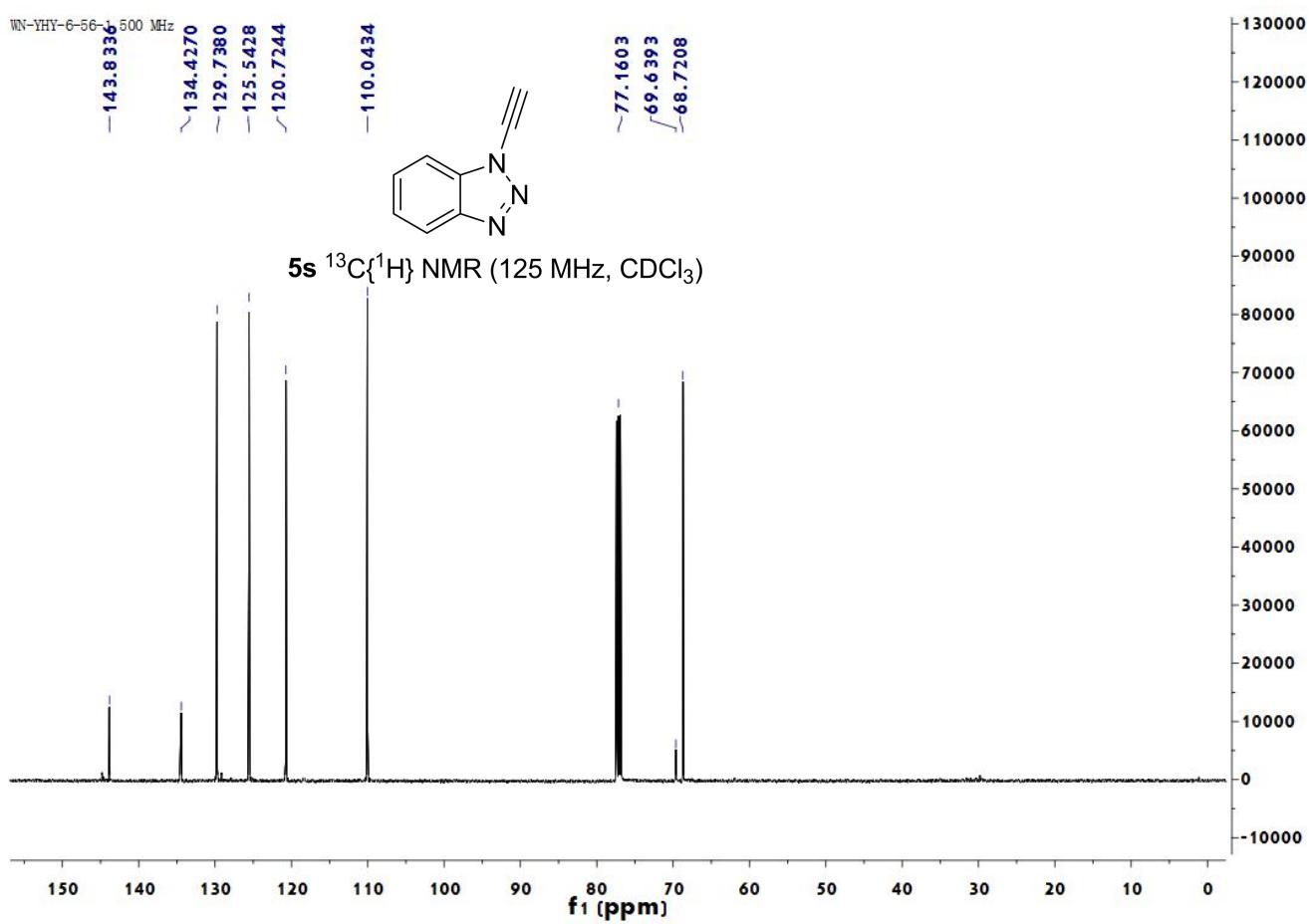
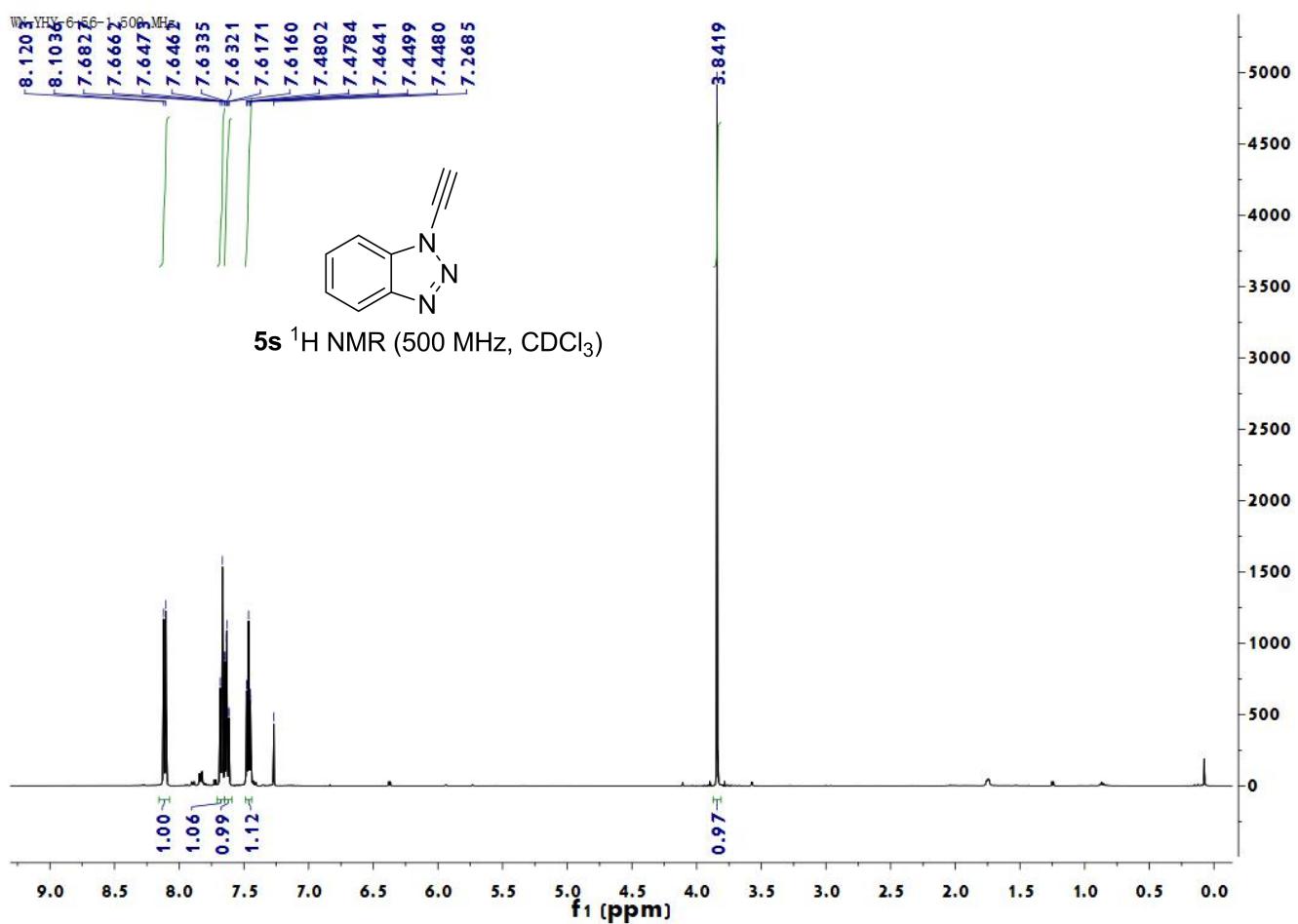


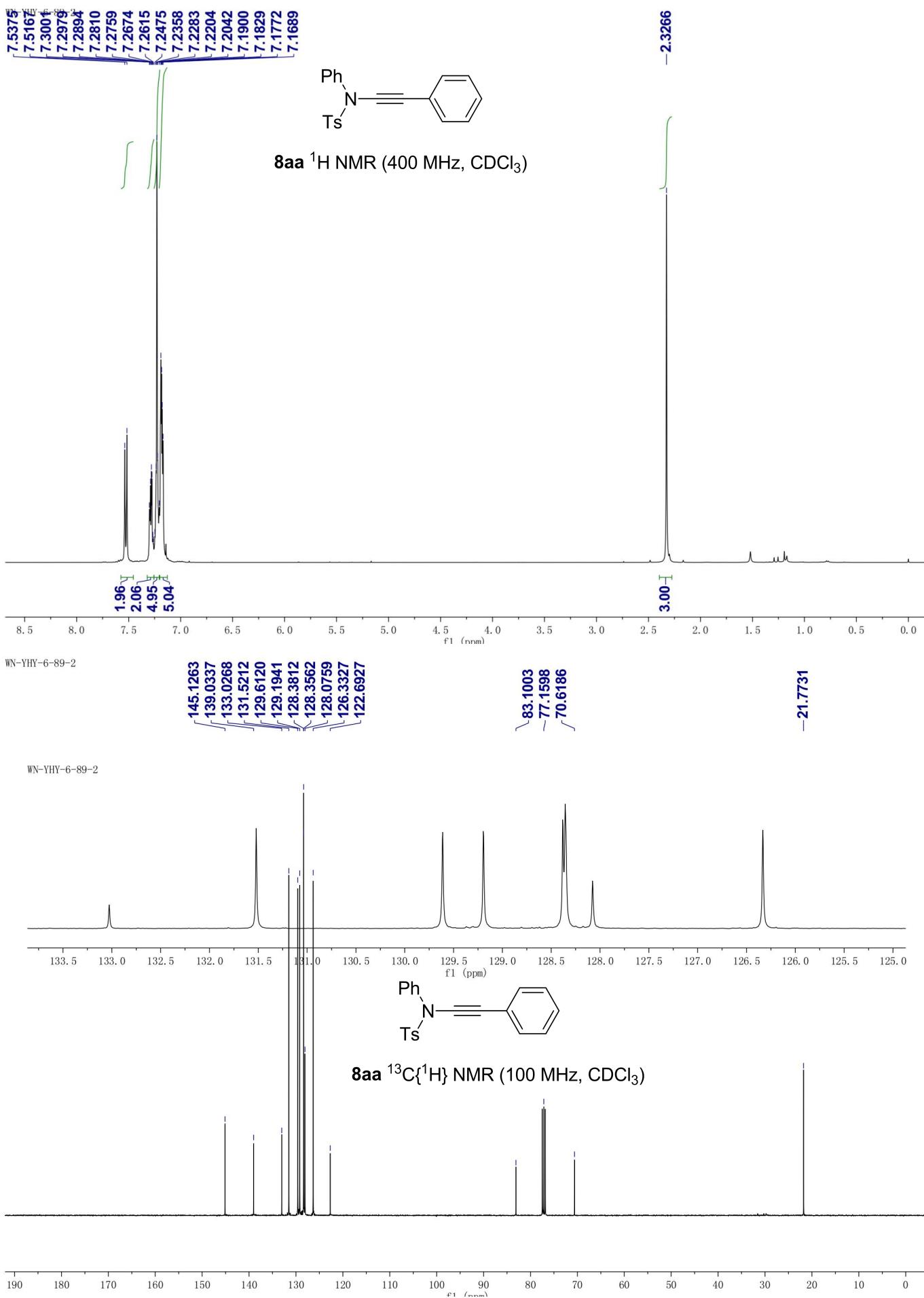
**5m**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

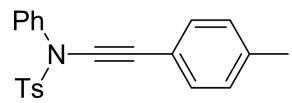
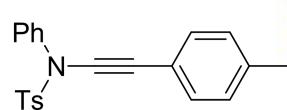
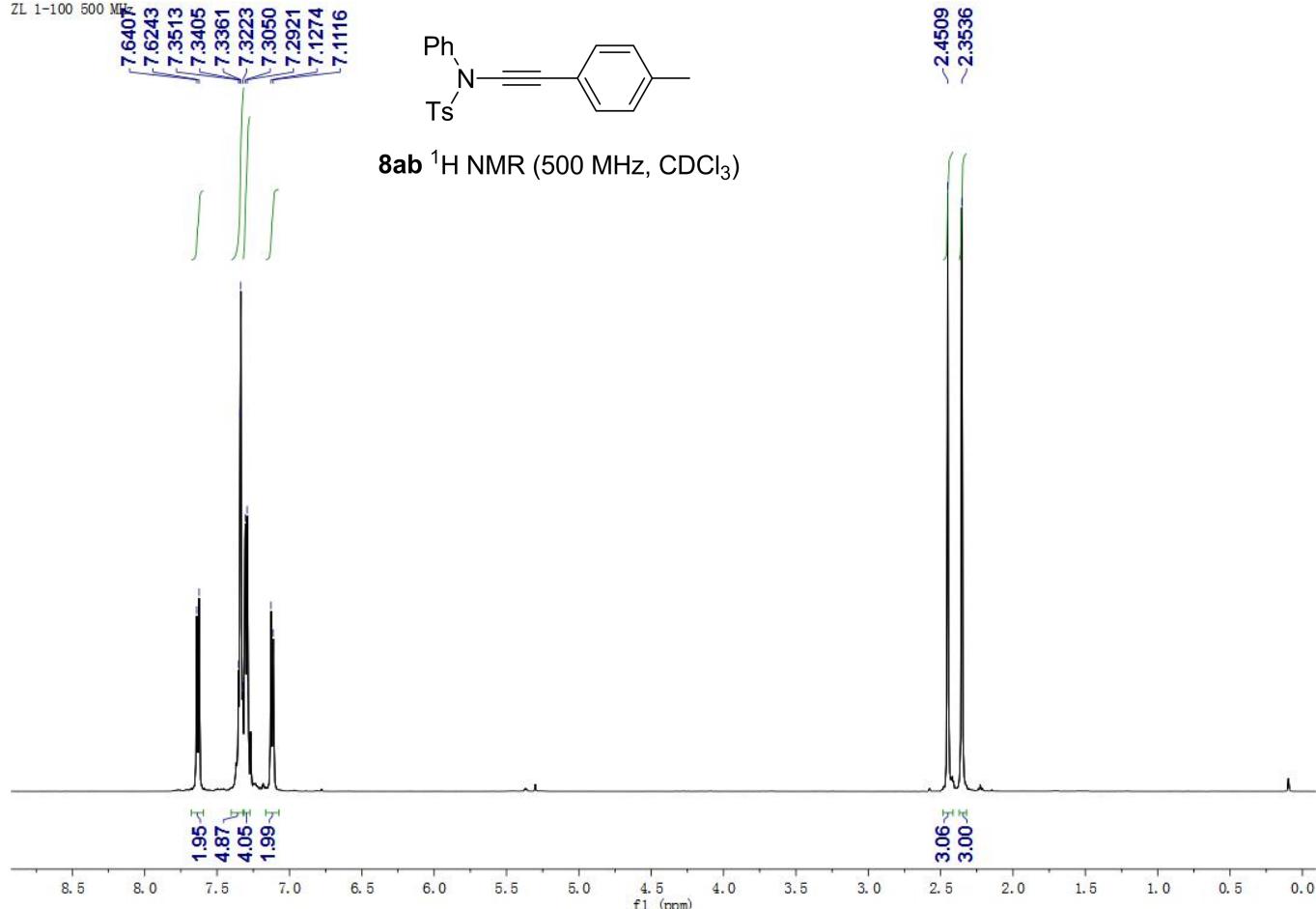
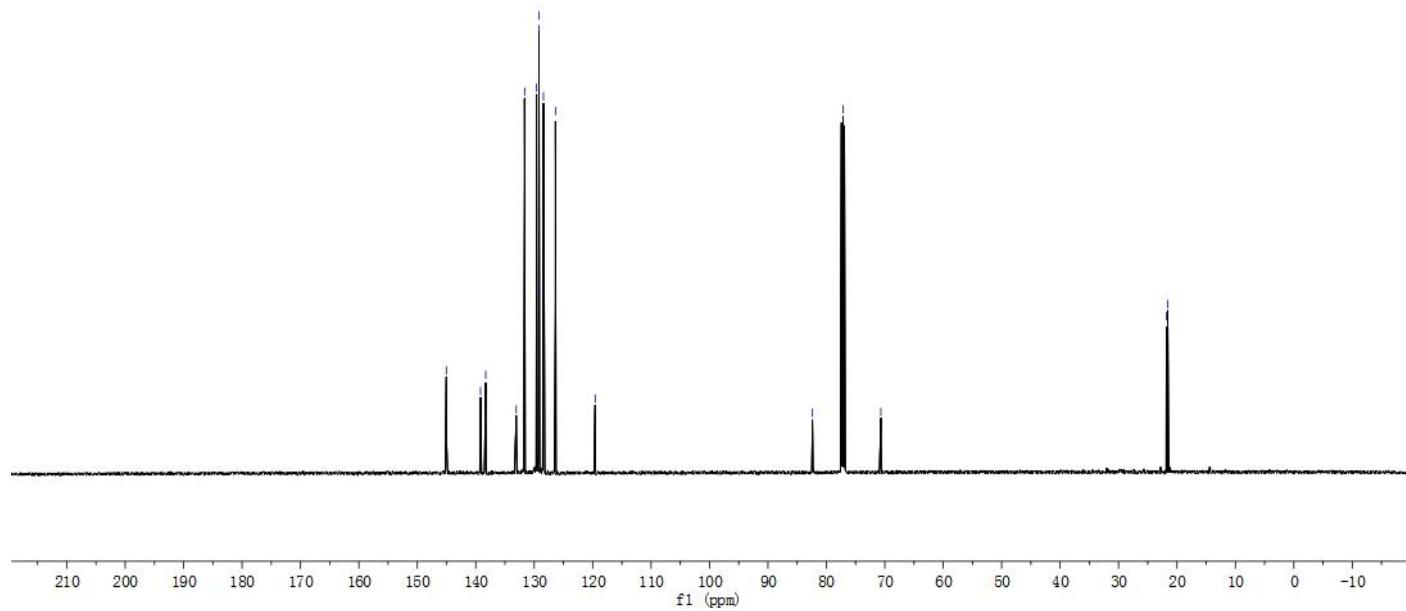


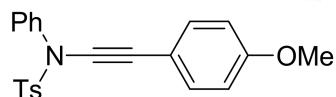
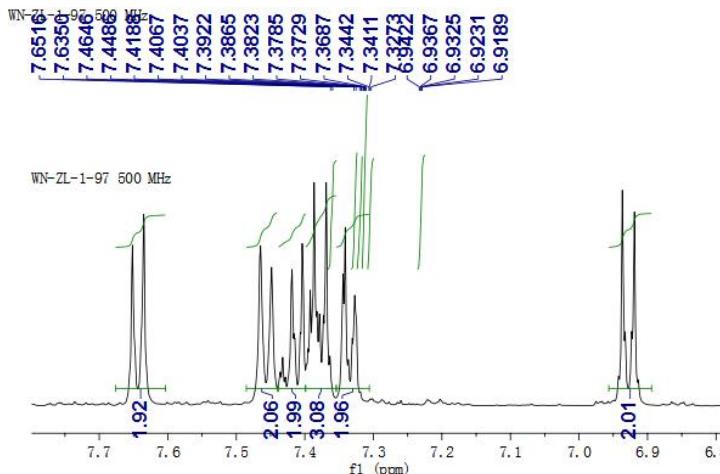




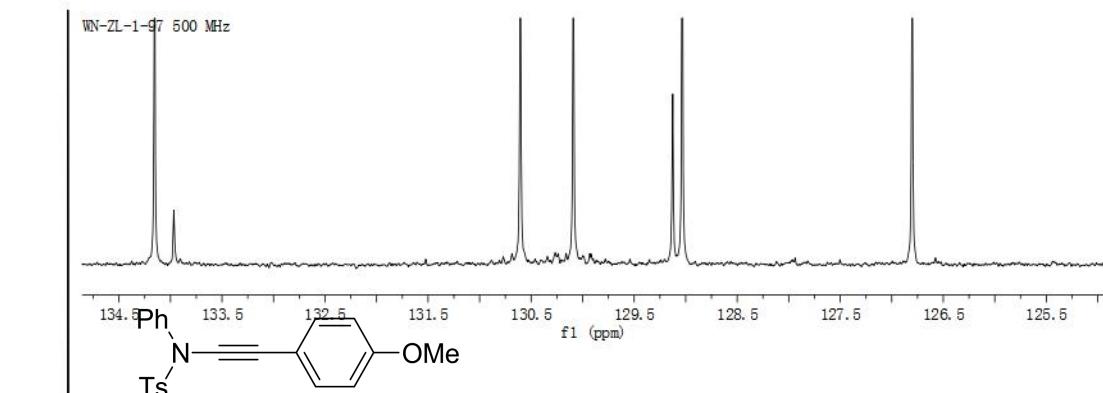
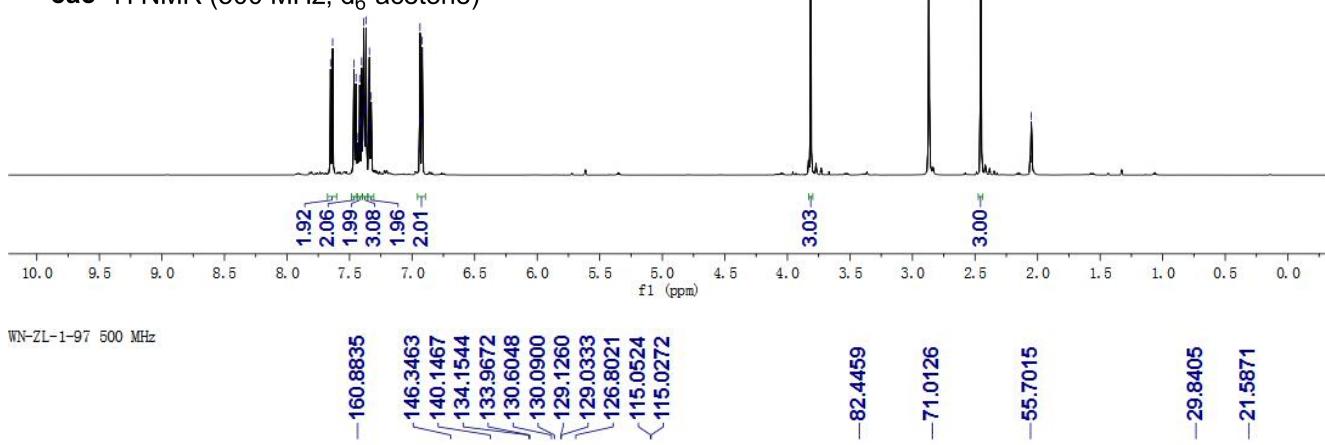




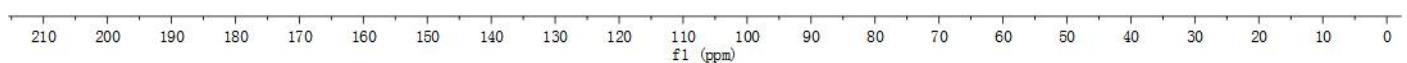
**8ab**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )**8ab**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

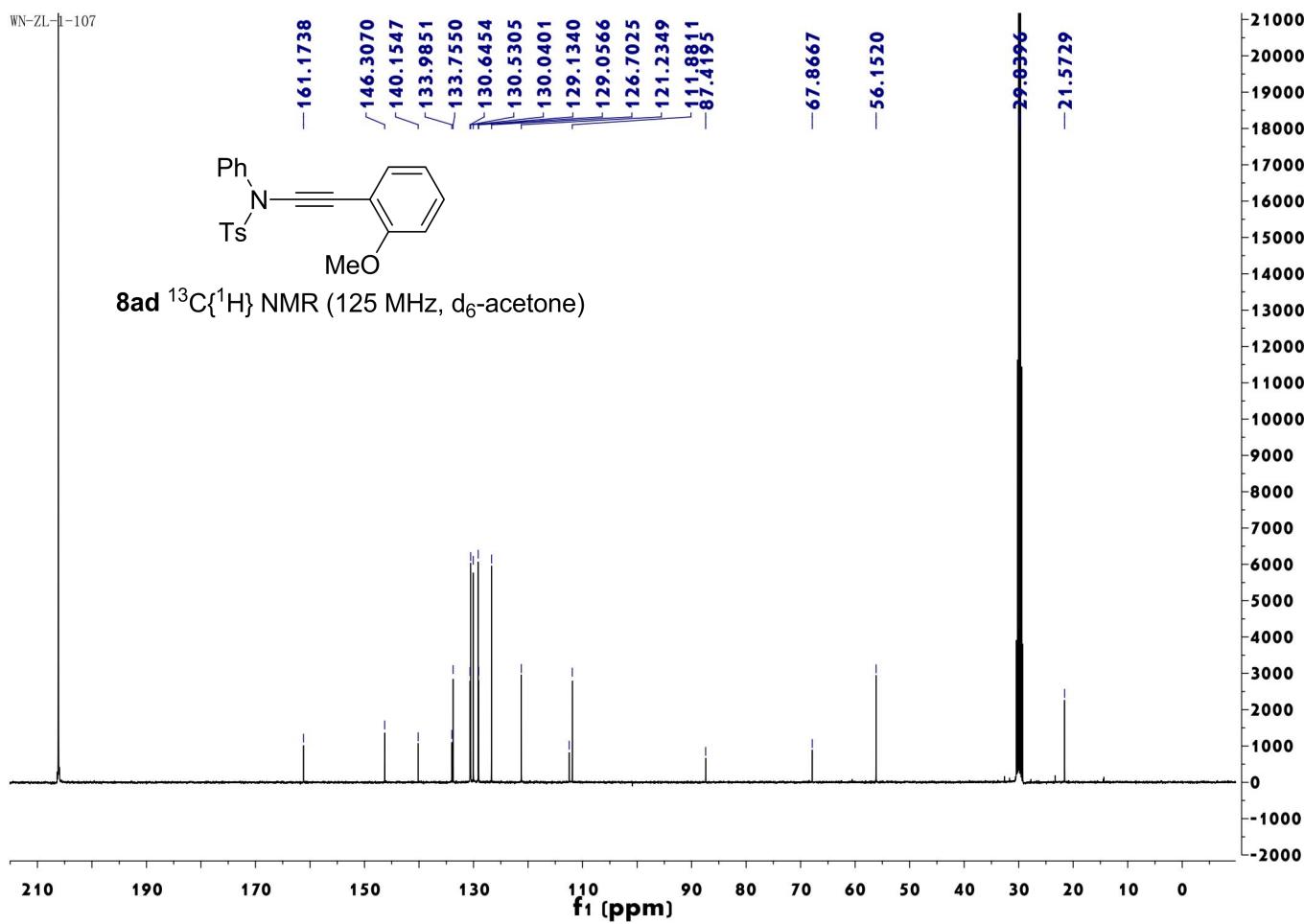
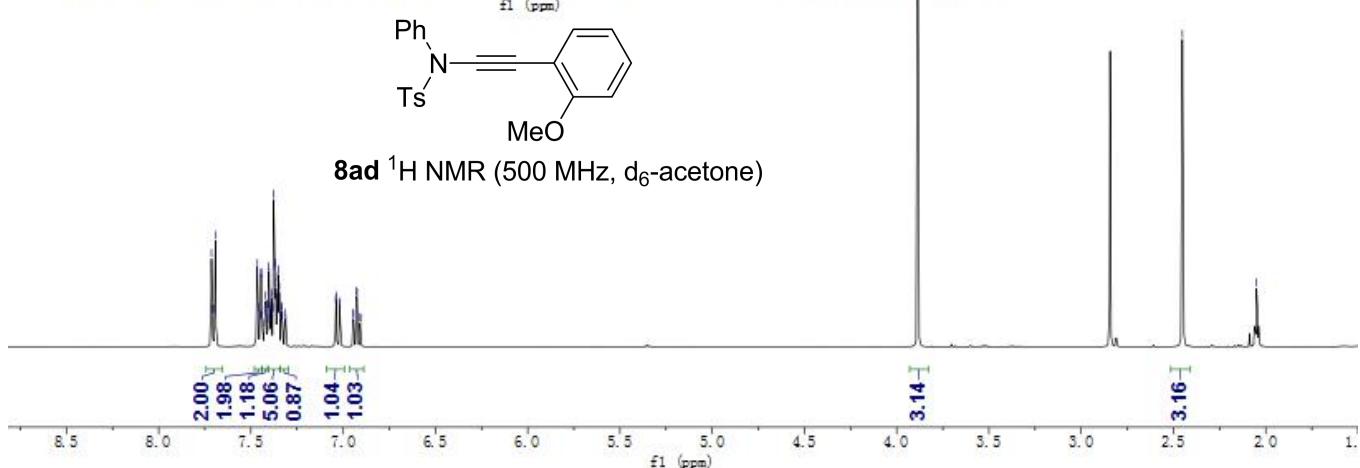
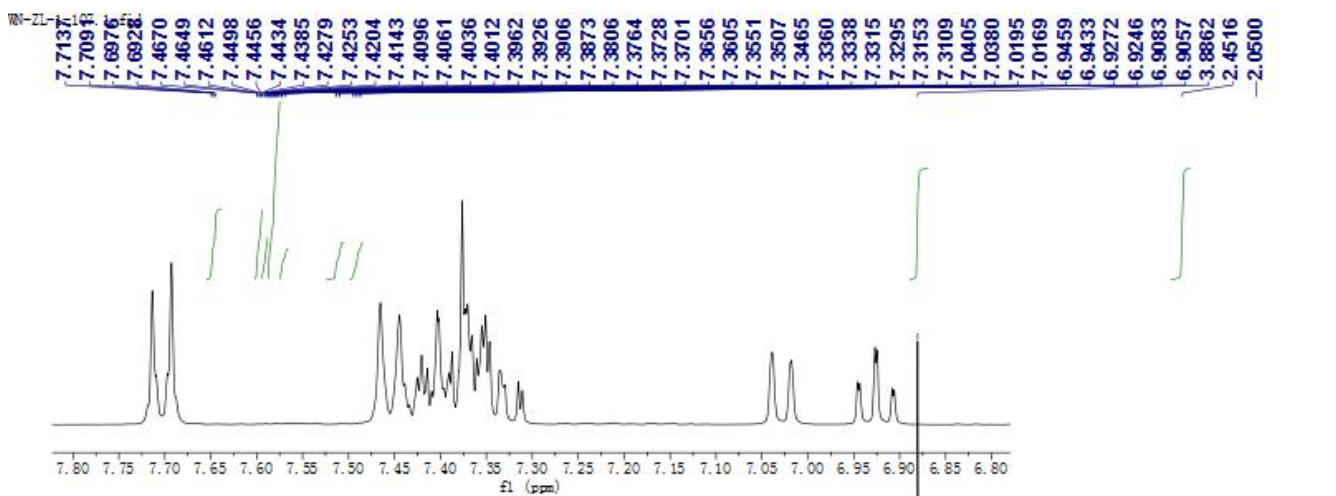


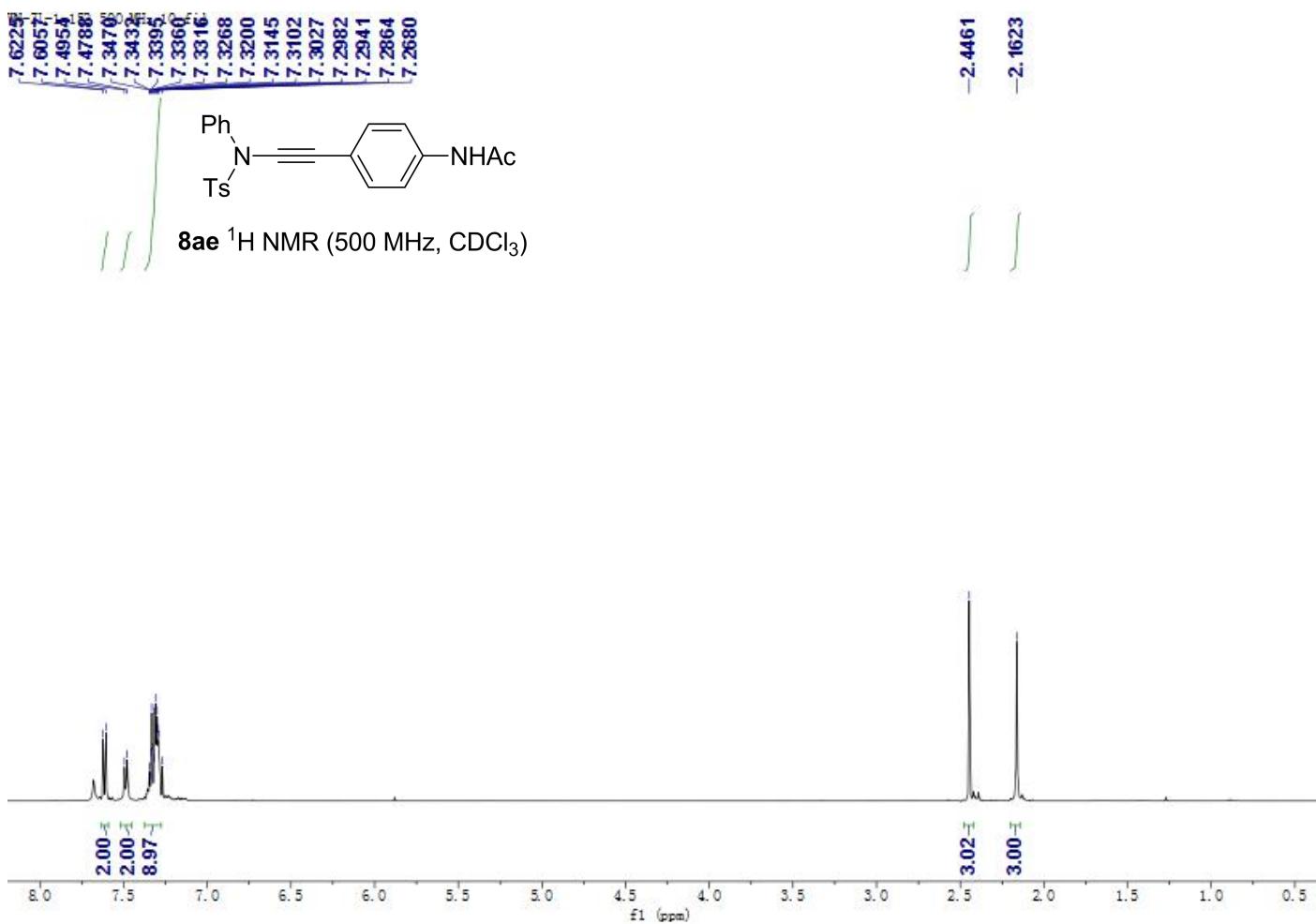
**8ac**  $^1\text{H}$  NMR (500 MHz,  $\text{d}_6$ -acetone)



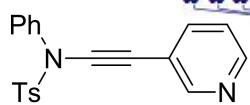
**8ac**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{d}_6$ -acetone)



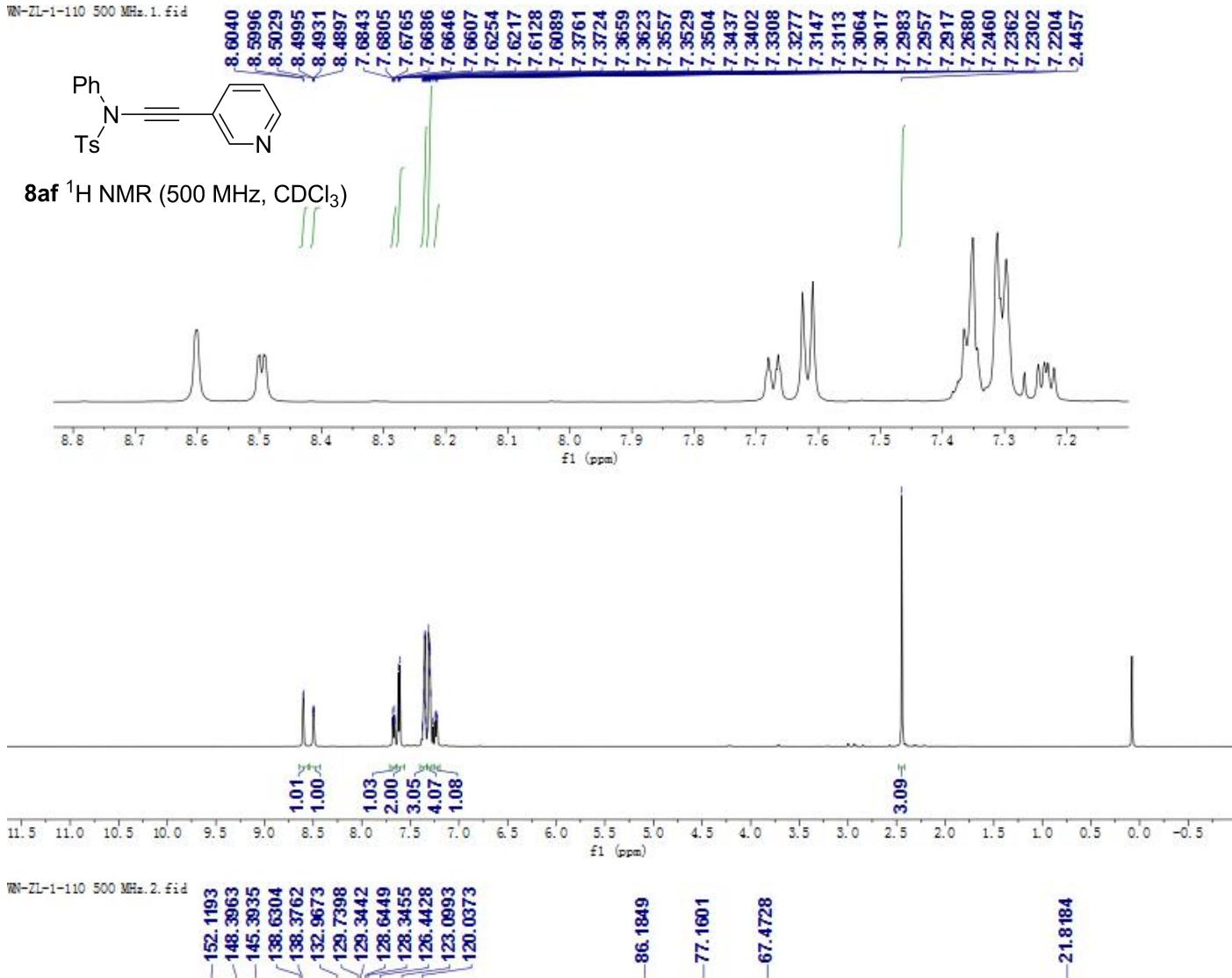




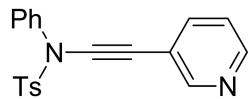
WZL-1-110 500 MHz.1.fid



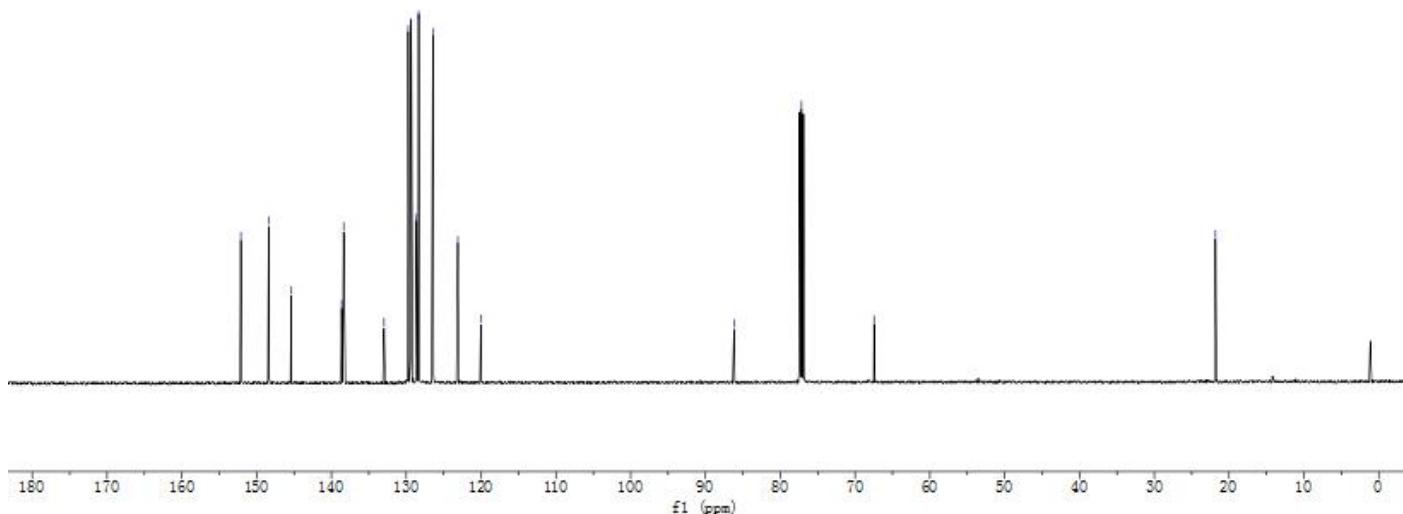
**8af**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )



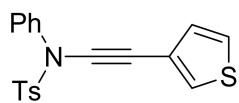
WZL-1-110 500 MHz.2.fid



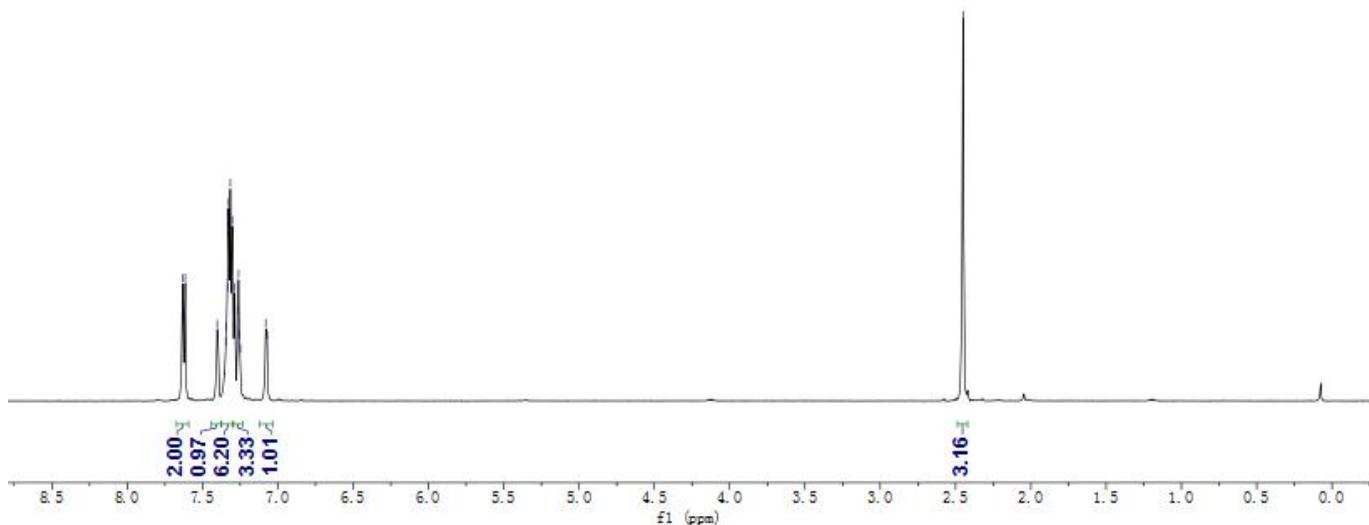
**8af**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )



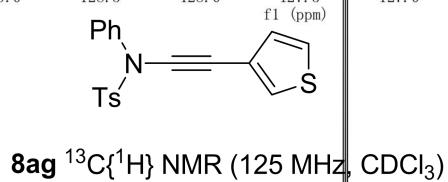
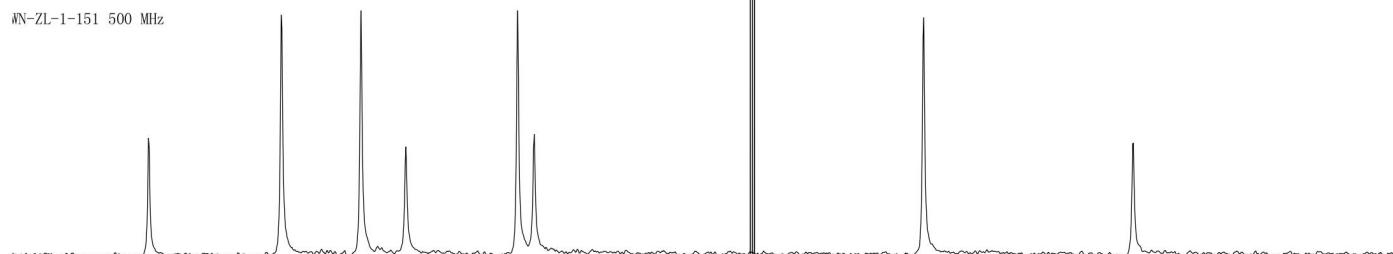
7.6319  
7.6199  
7.6152  
7.4032  
7.3425  
7.3346  
7.3282  
7.3230



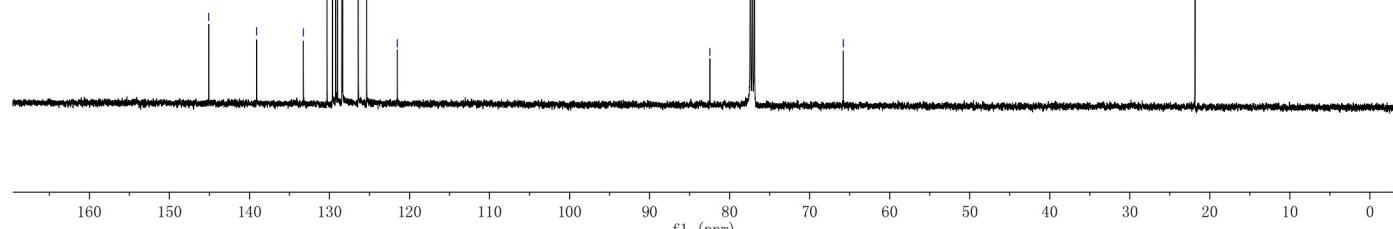
8ag  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )

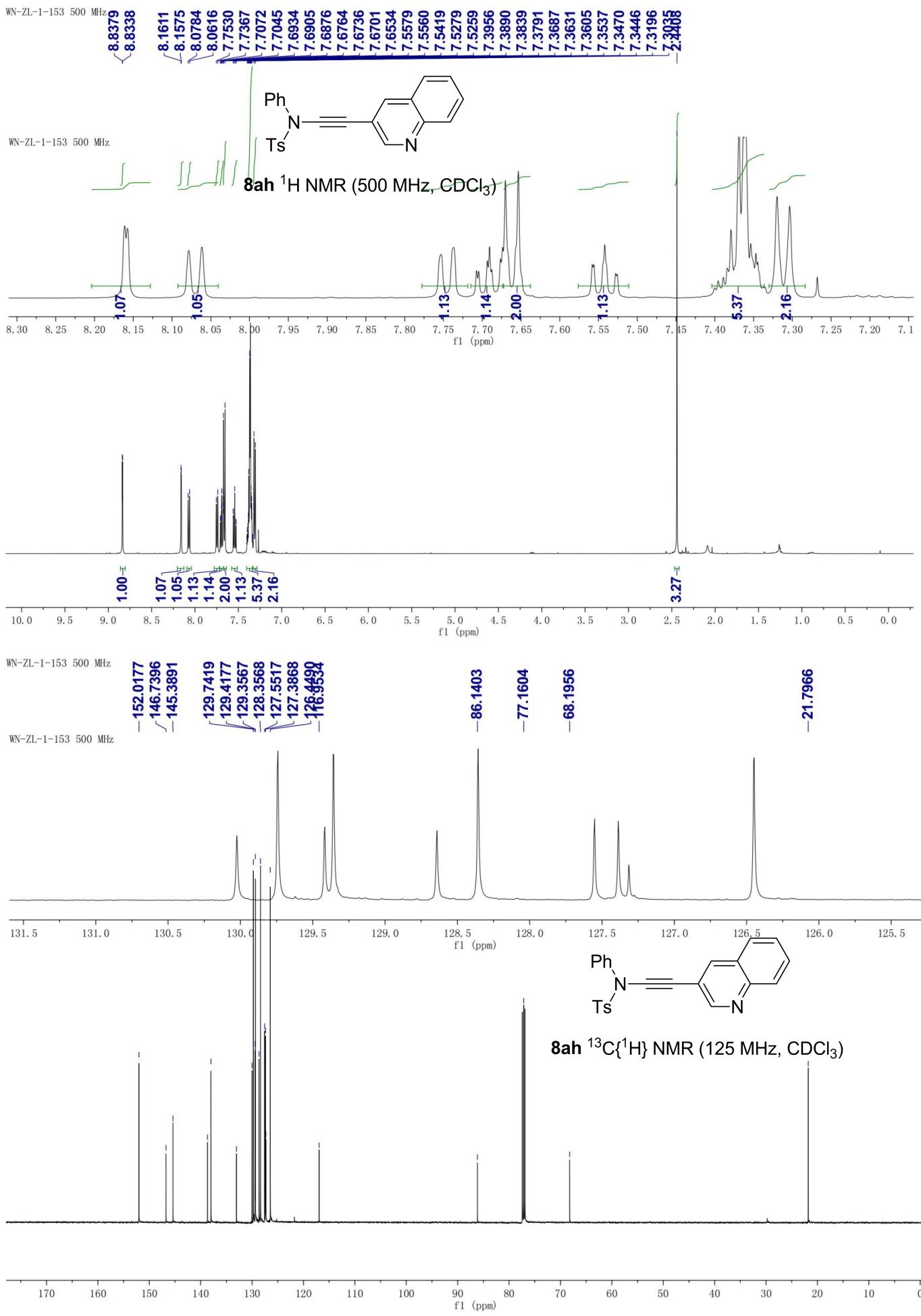


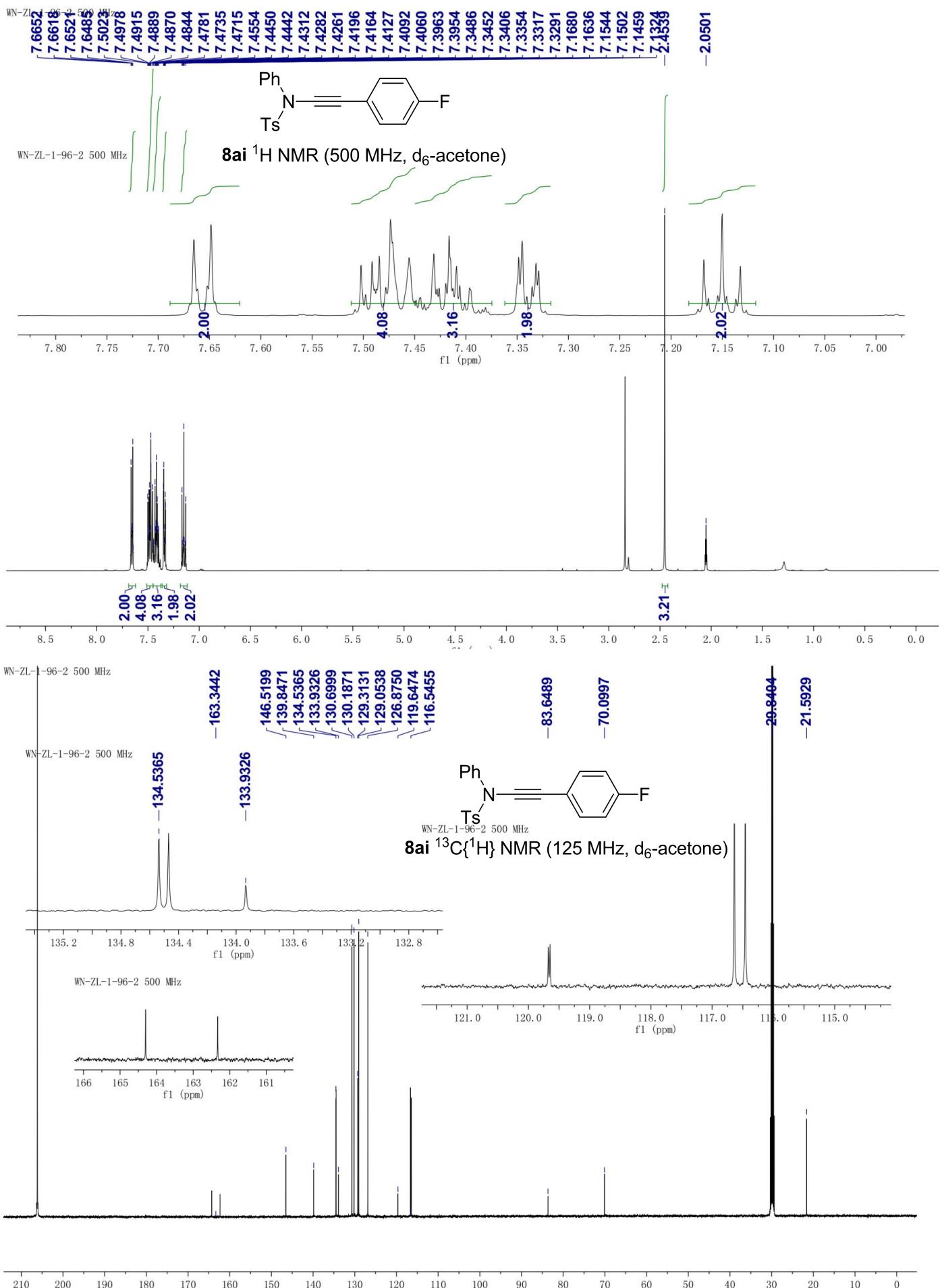
WN-ZL-1-151 500 MHz



8ag  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

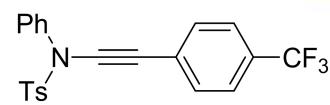




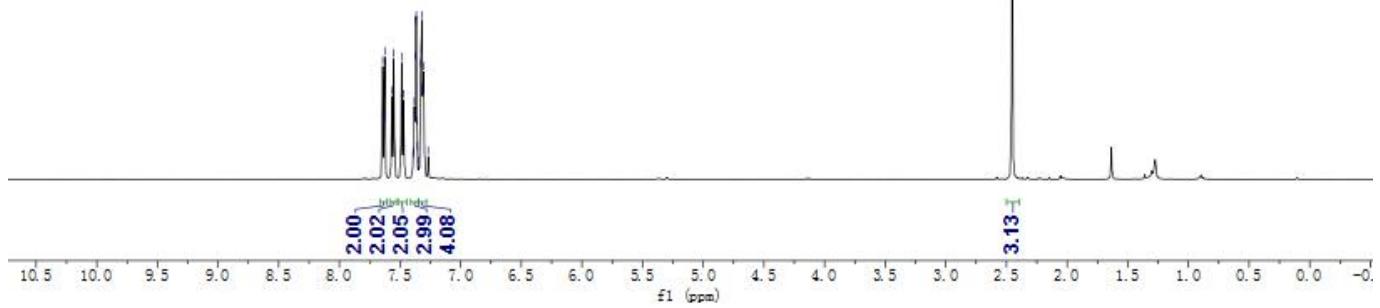
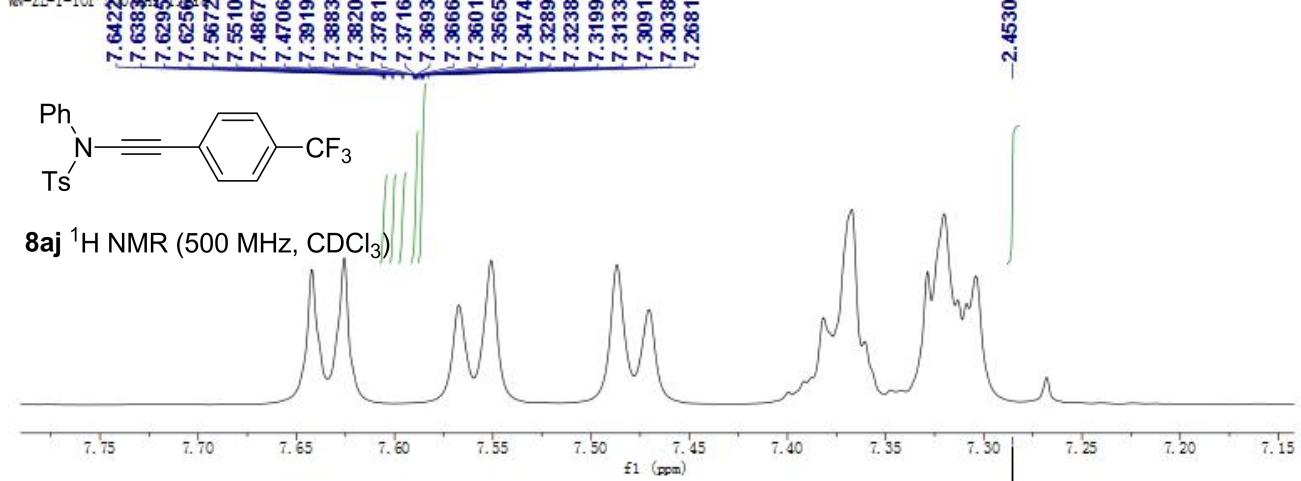


WN-ZL-1-101 500 MHz

7.6422 7.6384 7.6295 7.6256 7.5672 7.5510 7.4867 7.4706 7.3919 7.3883 7.3820 7.3781 7.3716 7.3693 7.3666 7.3601 7.3565 7.3474 7.3289 7.3238 7.3199 7.3133 7.3091 7.3038 7.2681



**8aj**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )



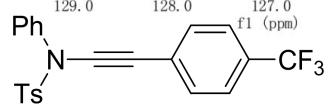
WN-ZL-1-101 500 MHz

WN-ZL-1-101 500 MHz

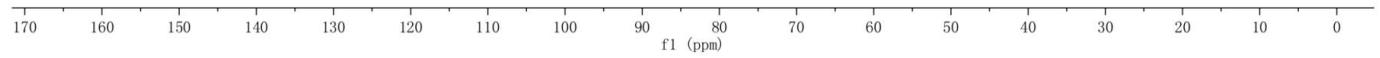
-145.4084  
-138.6512  
-133.0527  
-131.2565  
-129.7609  
-129.5425  
-129.3663  
-128.7589  
-128.6670  
-128.3509  
-126.8107  
-126.4625  
-125.3487  
-124.1649

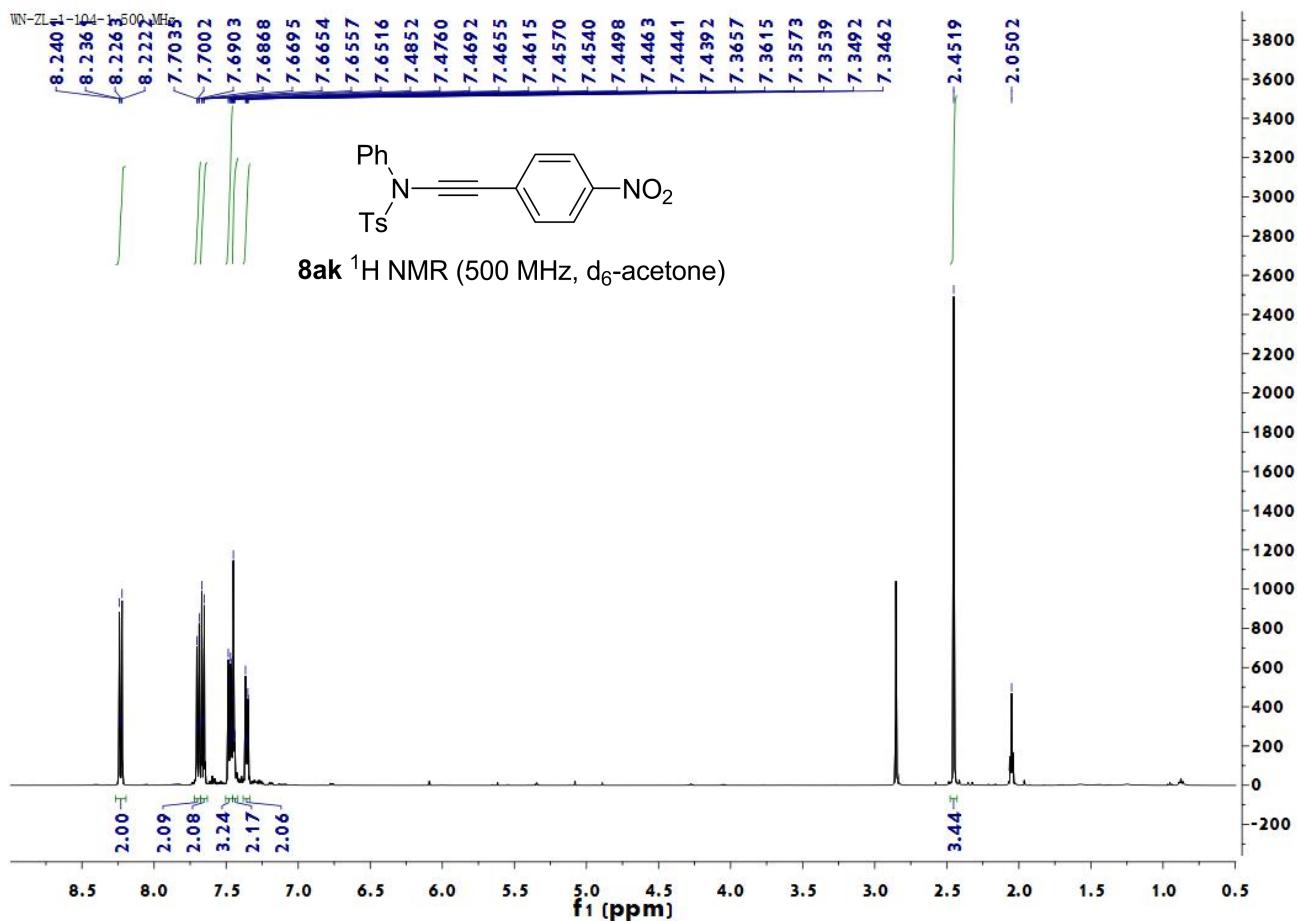
-85.6940  
-77.1601  
-69.7718

-21.8225

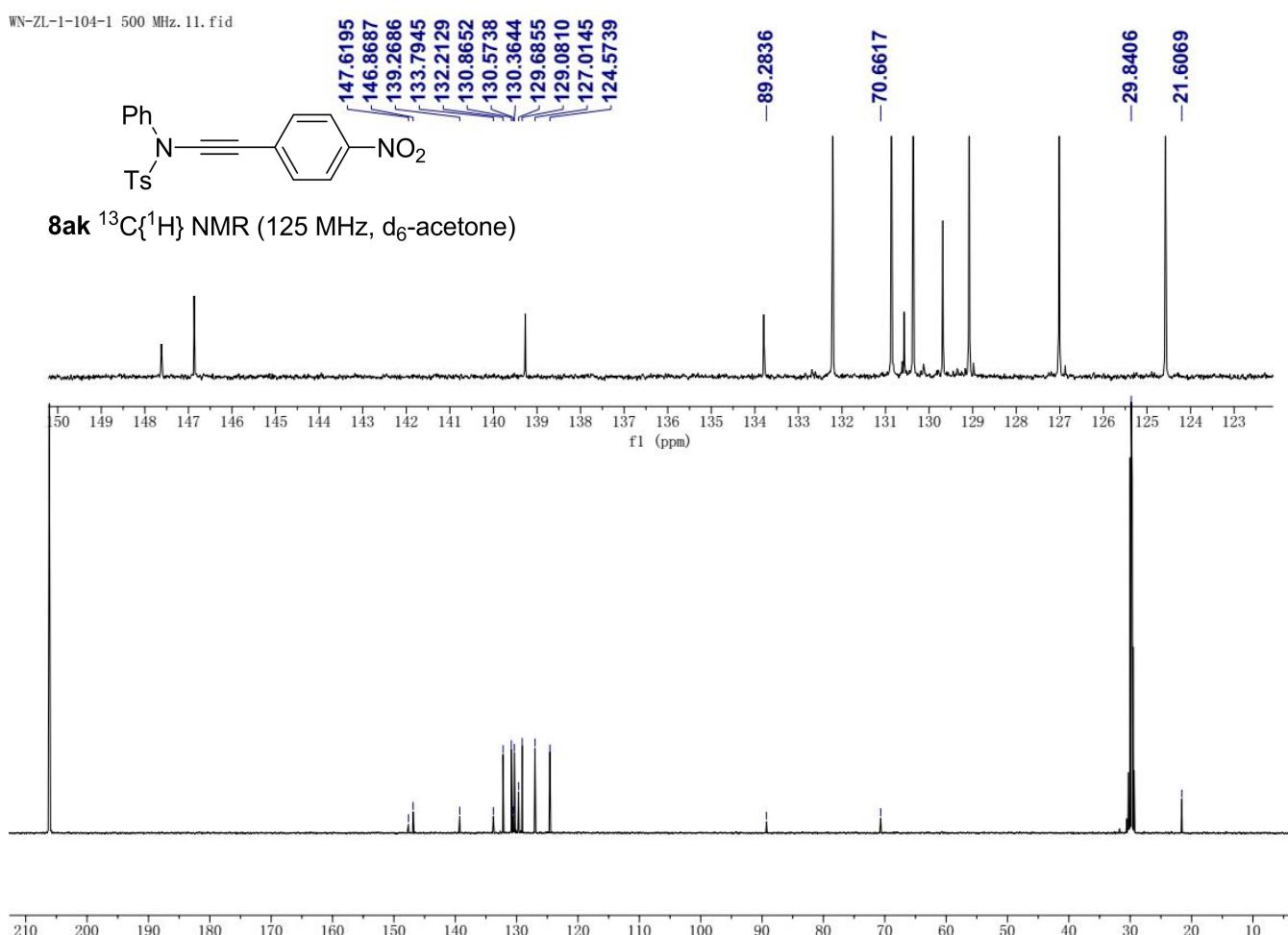


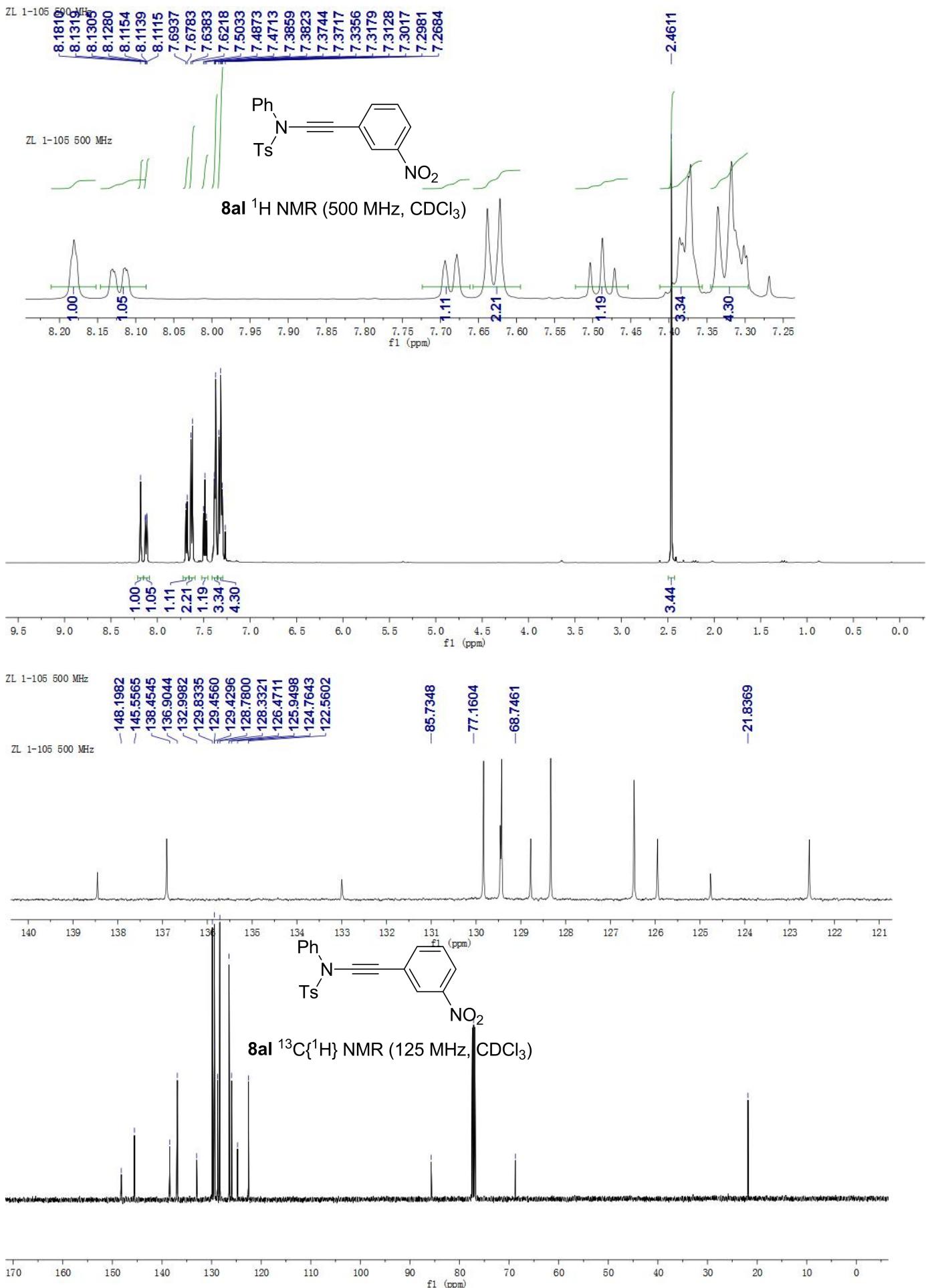
**8aj**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

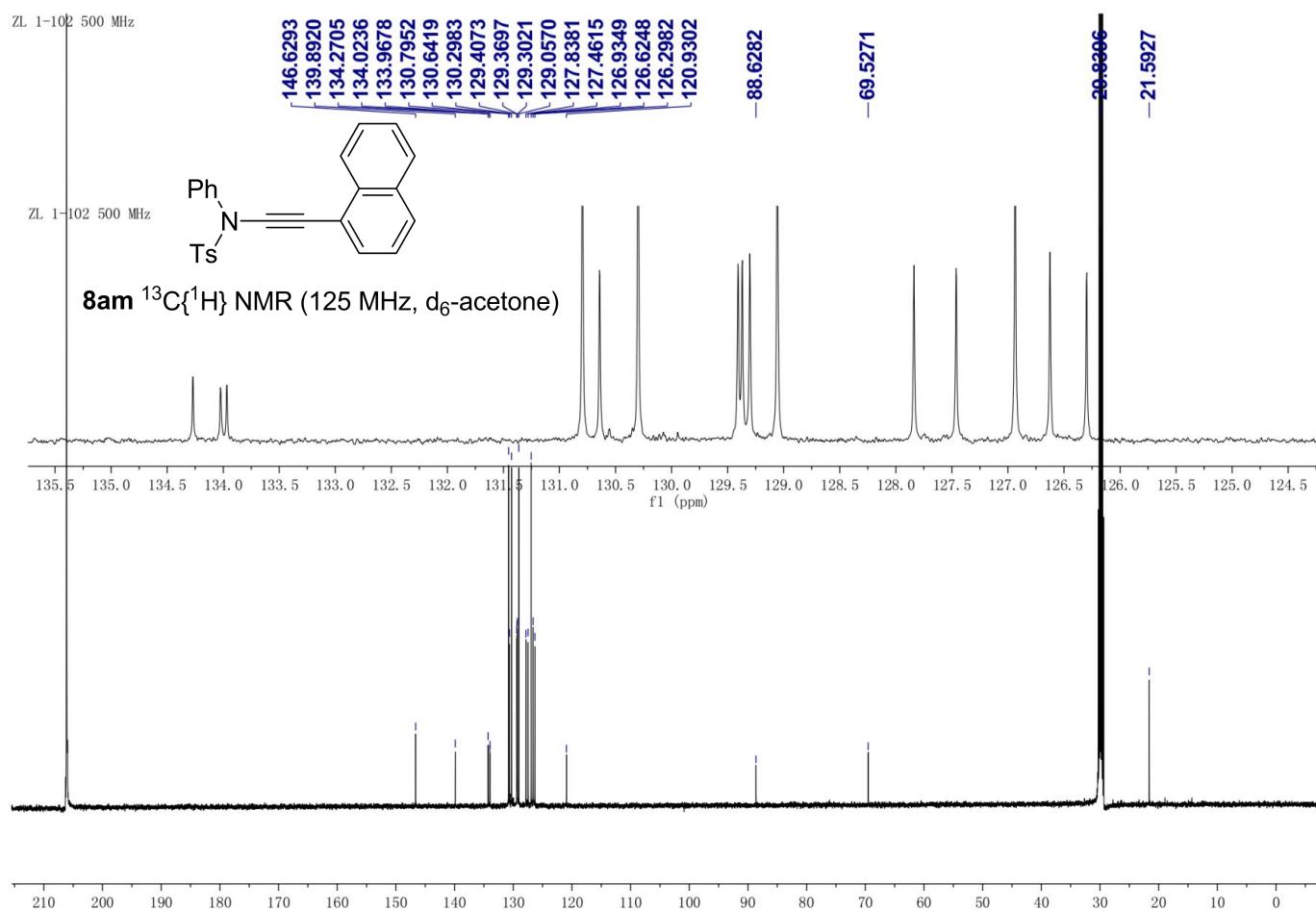
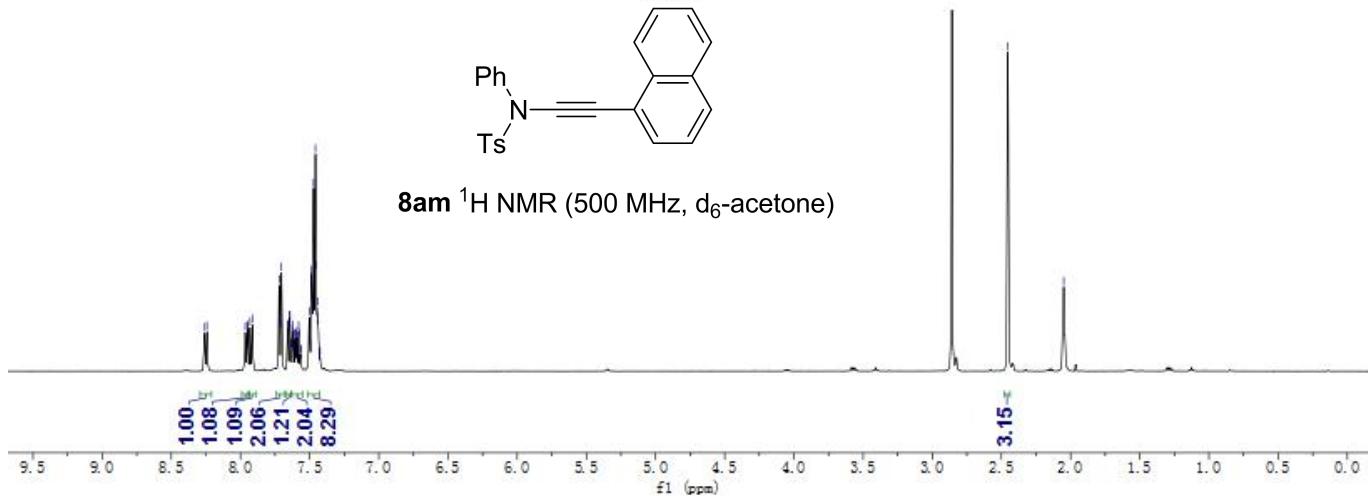
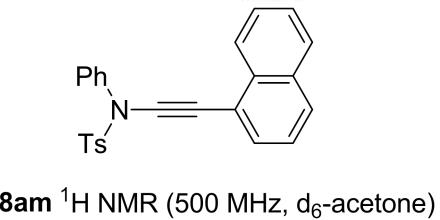
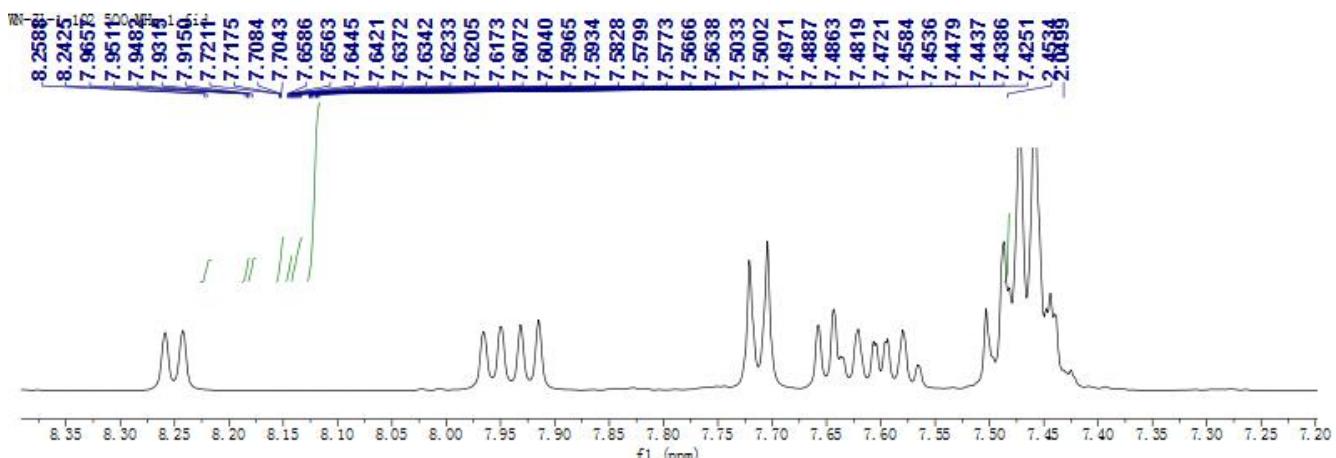


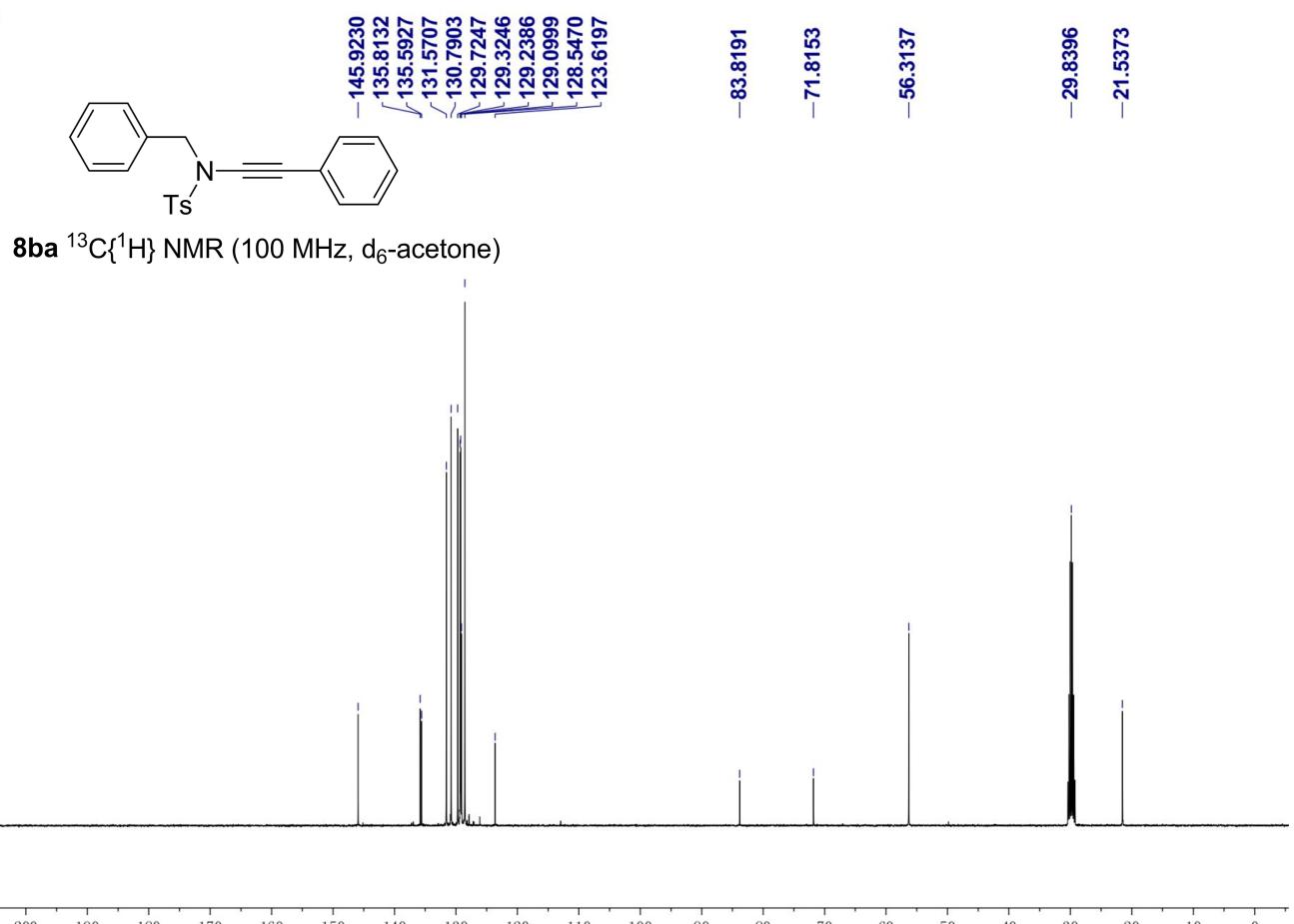
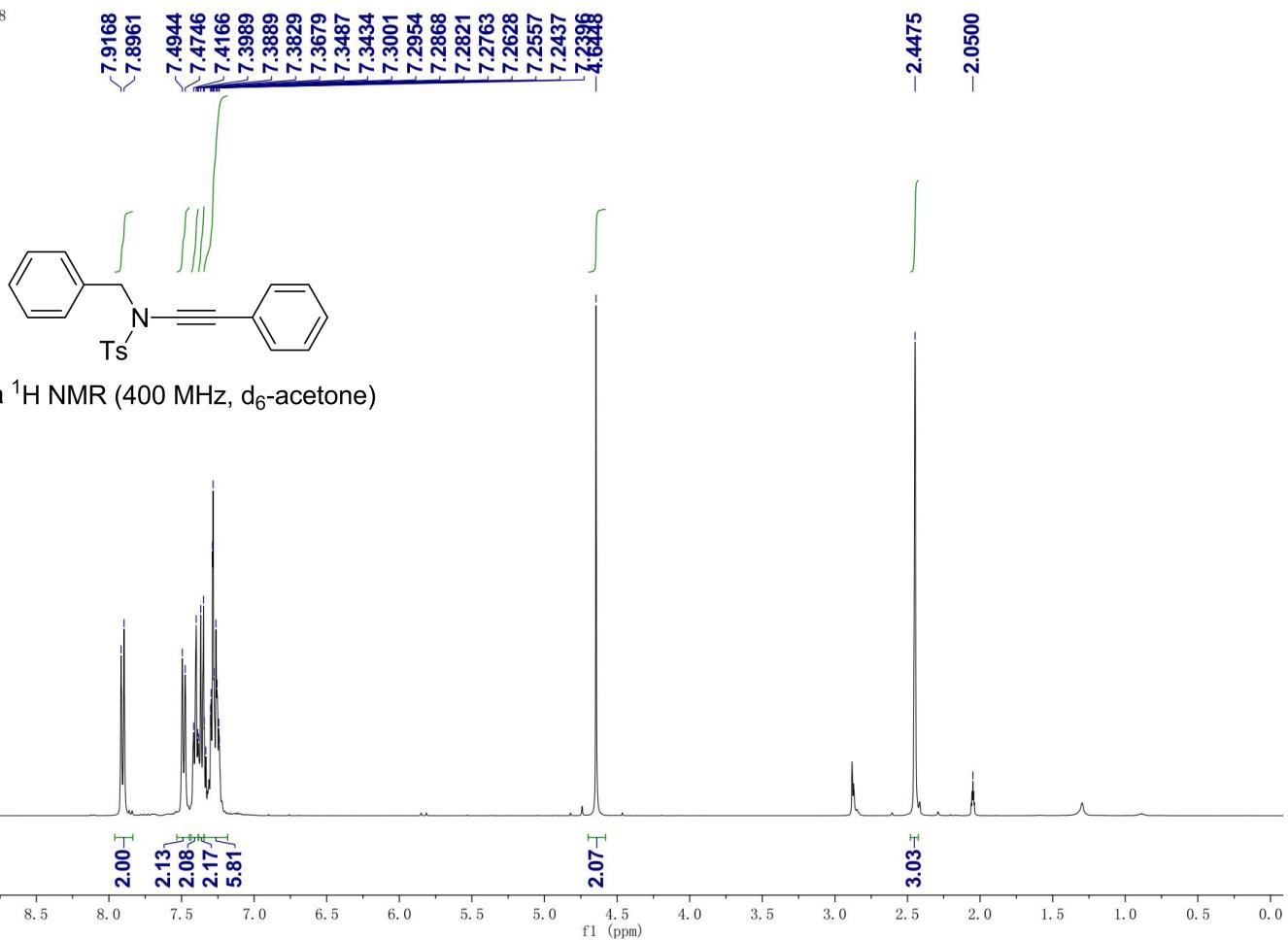


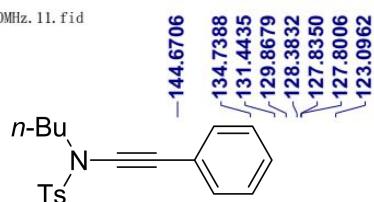
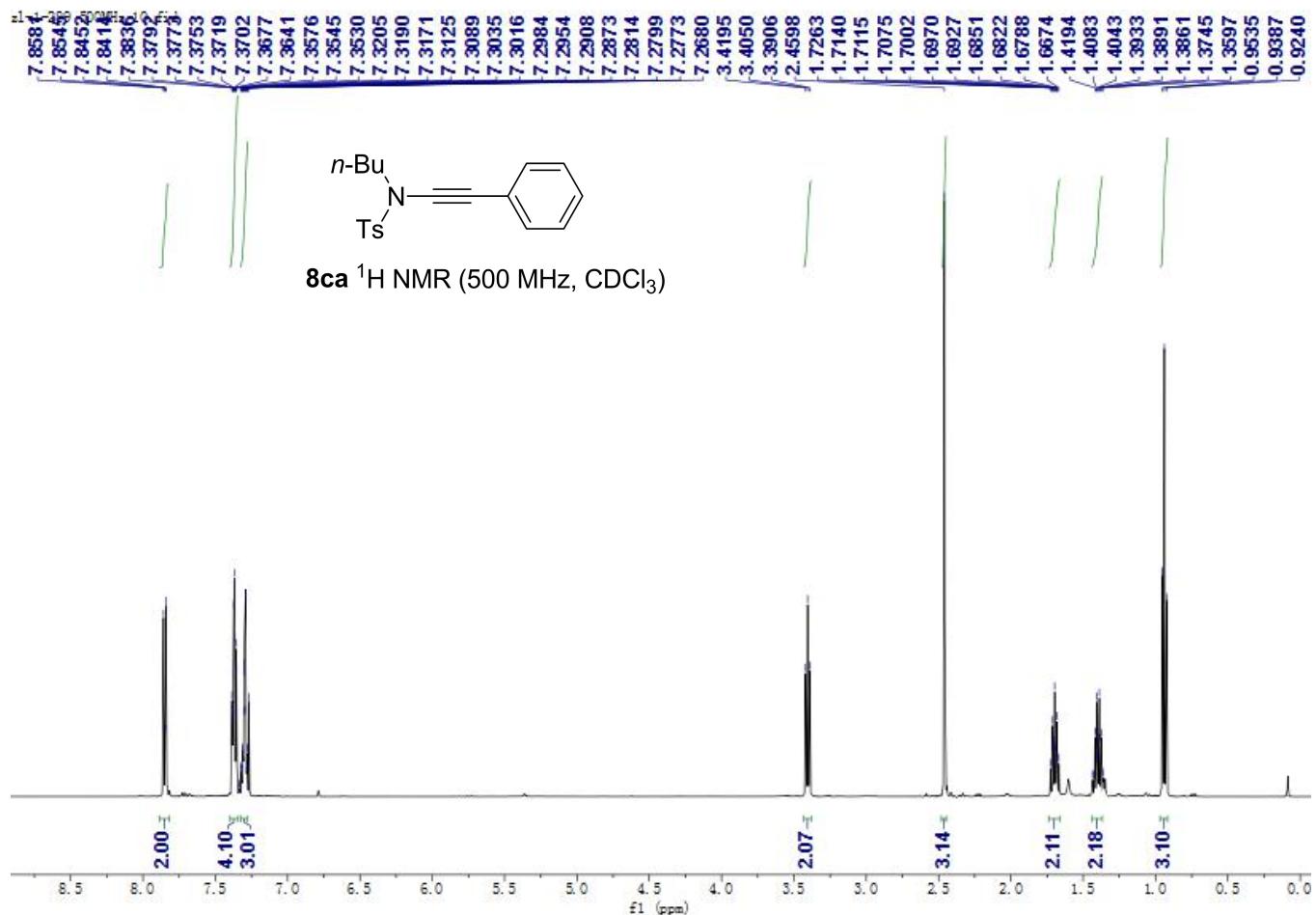
WN-ZL-1-104-1 500 MHz, 11.fid



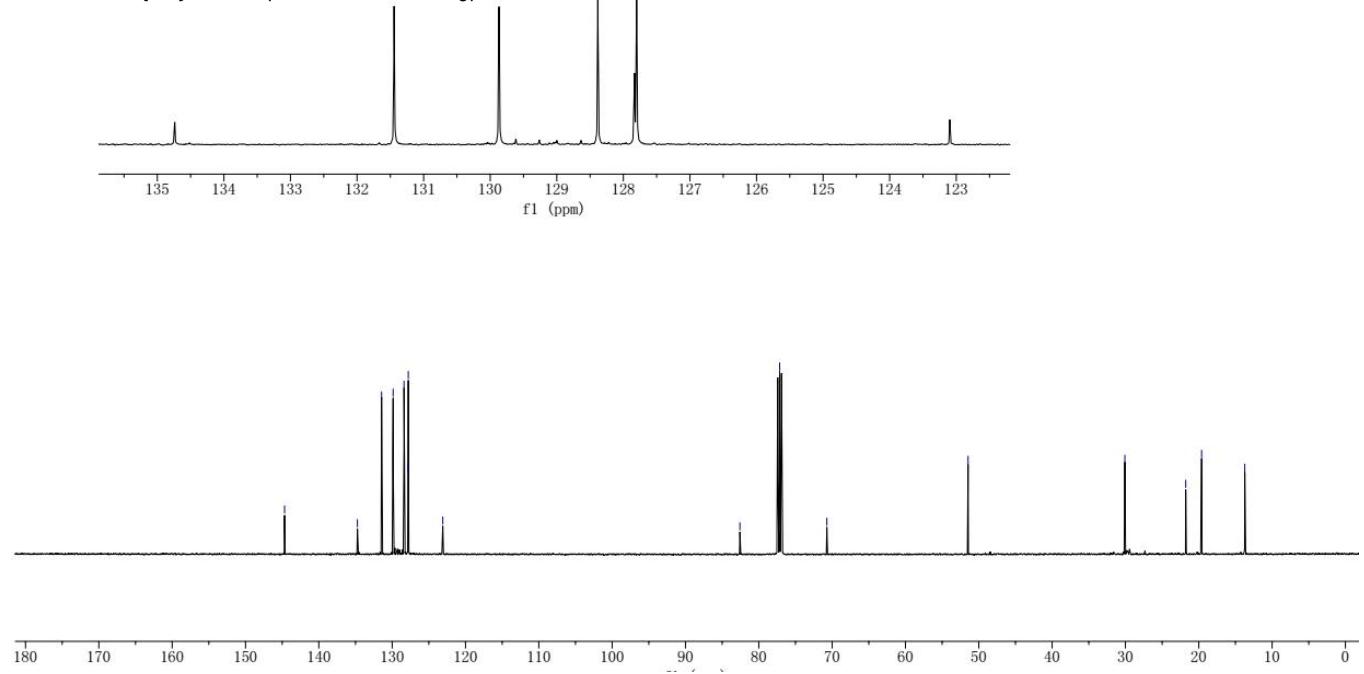


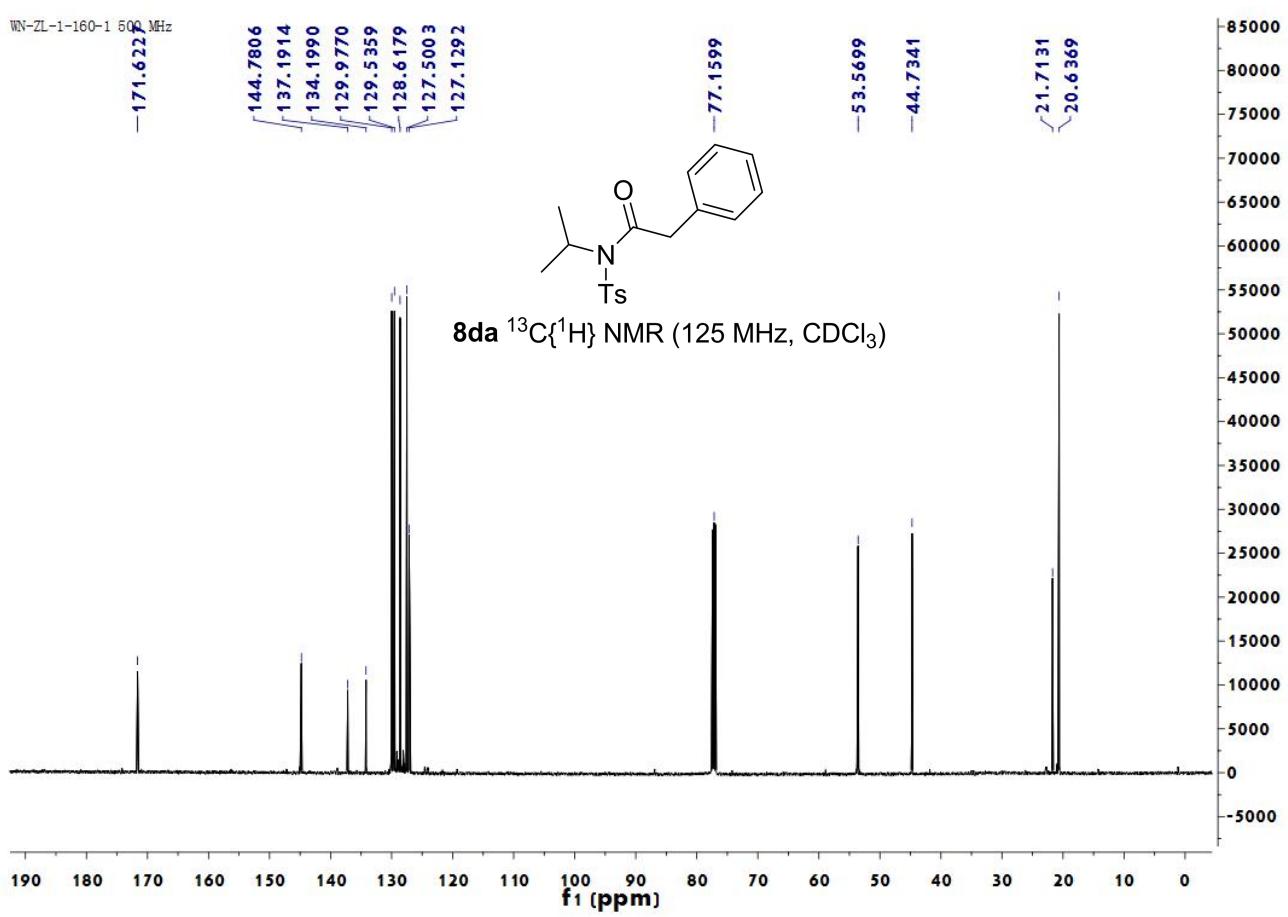
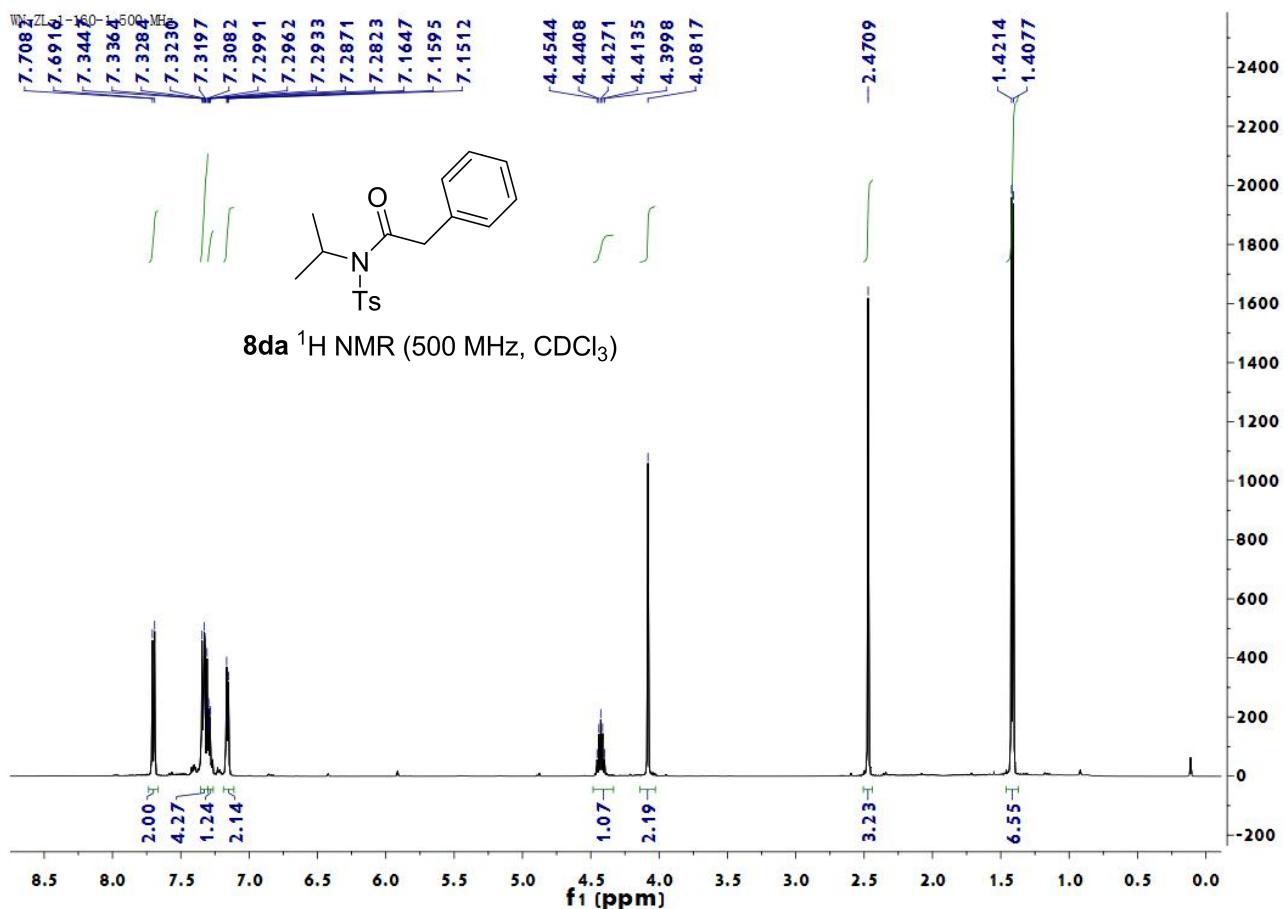


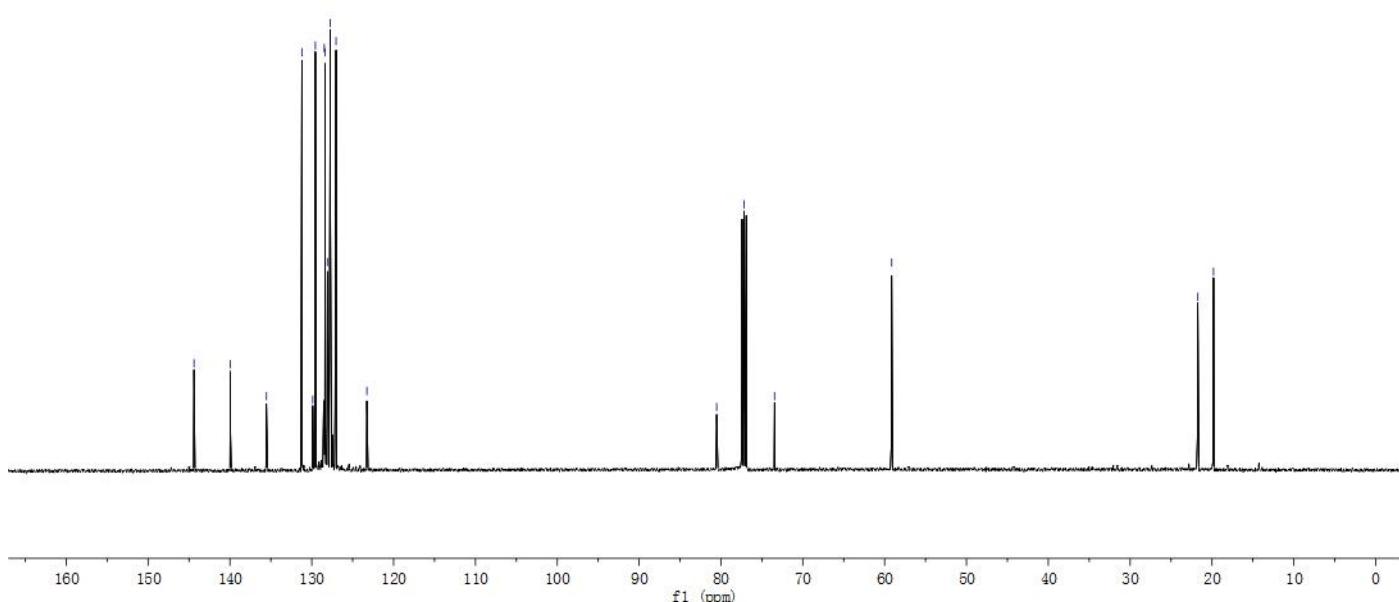
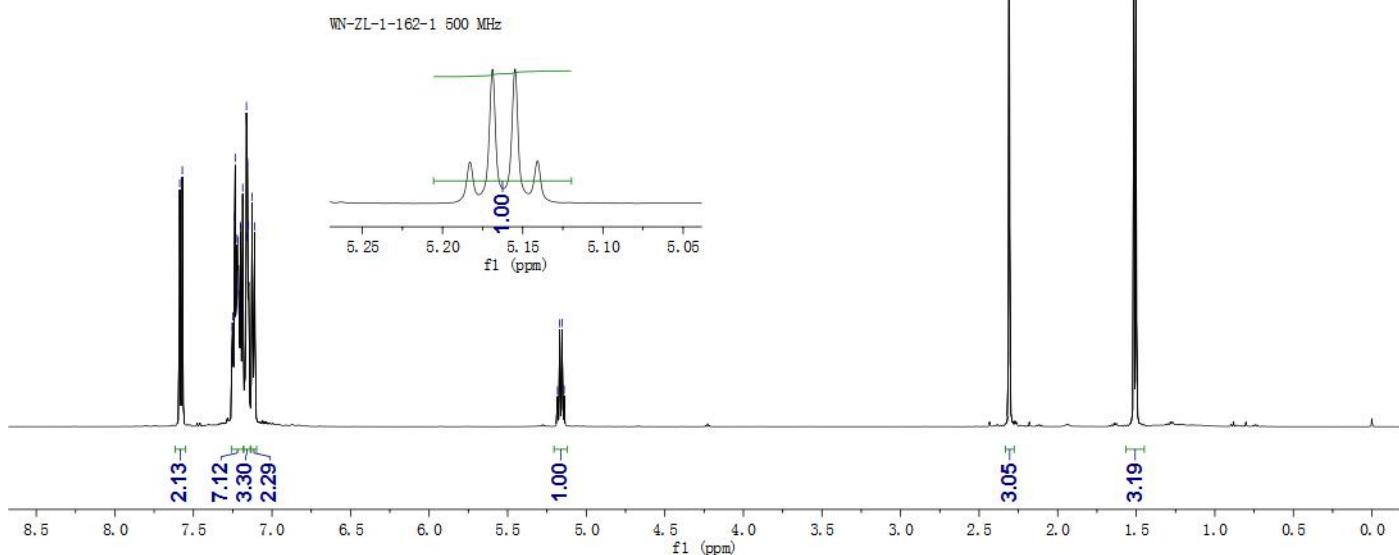
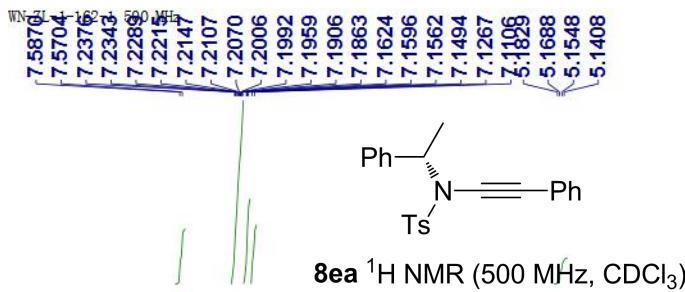


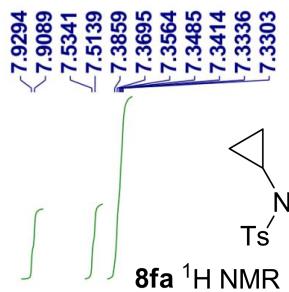
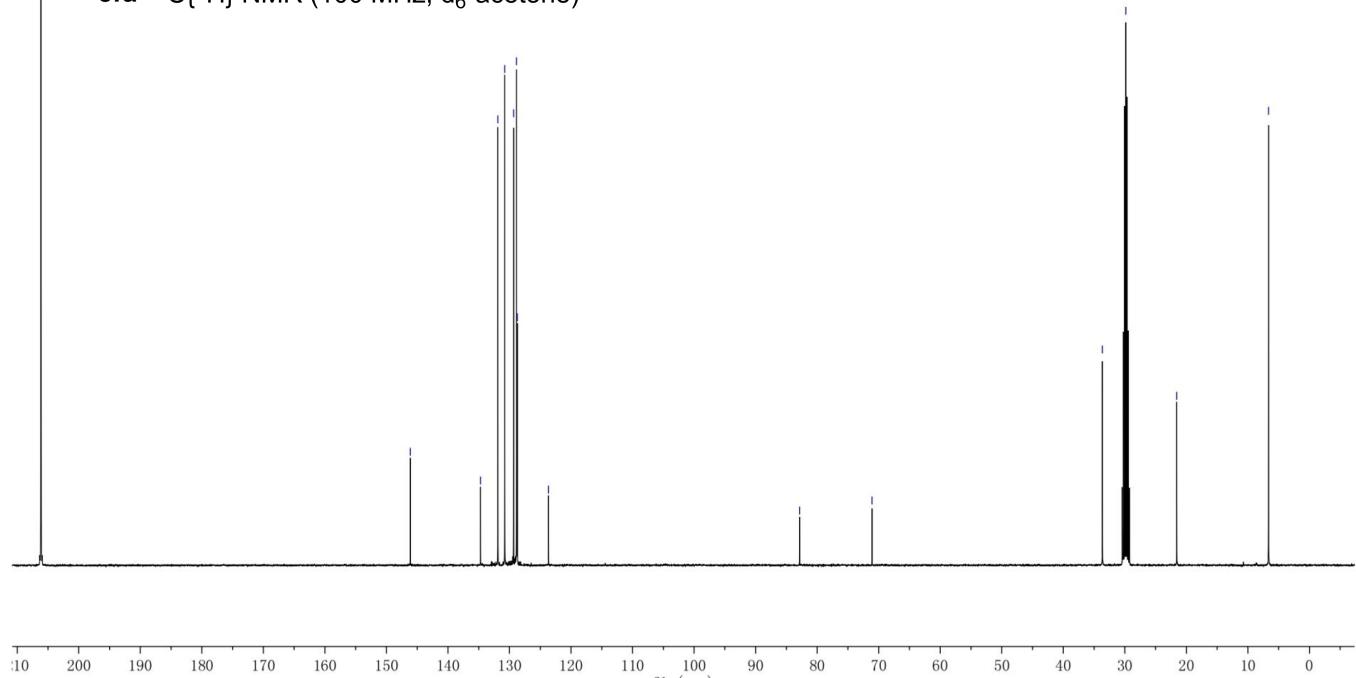
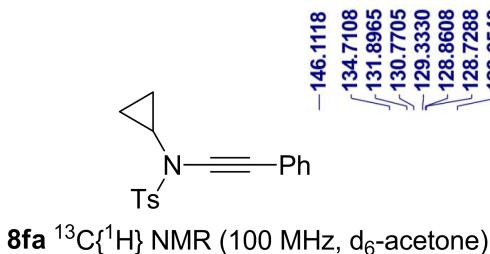
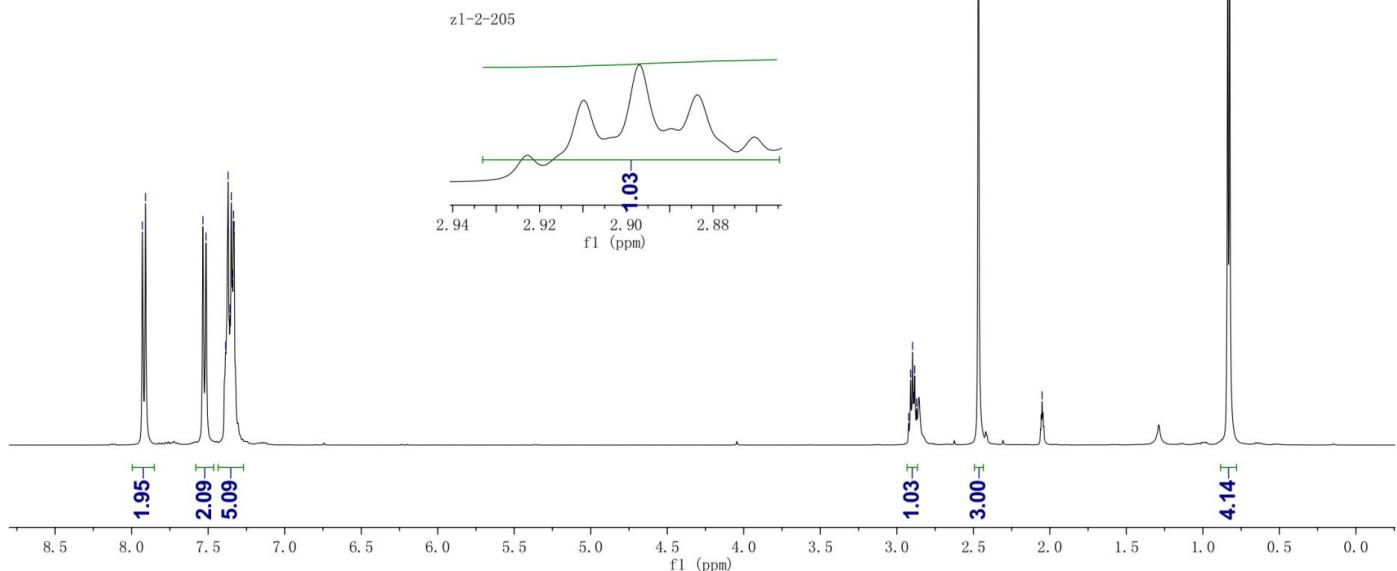


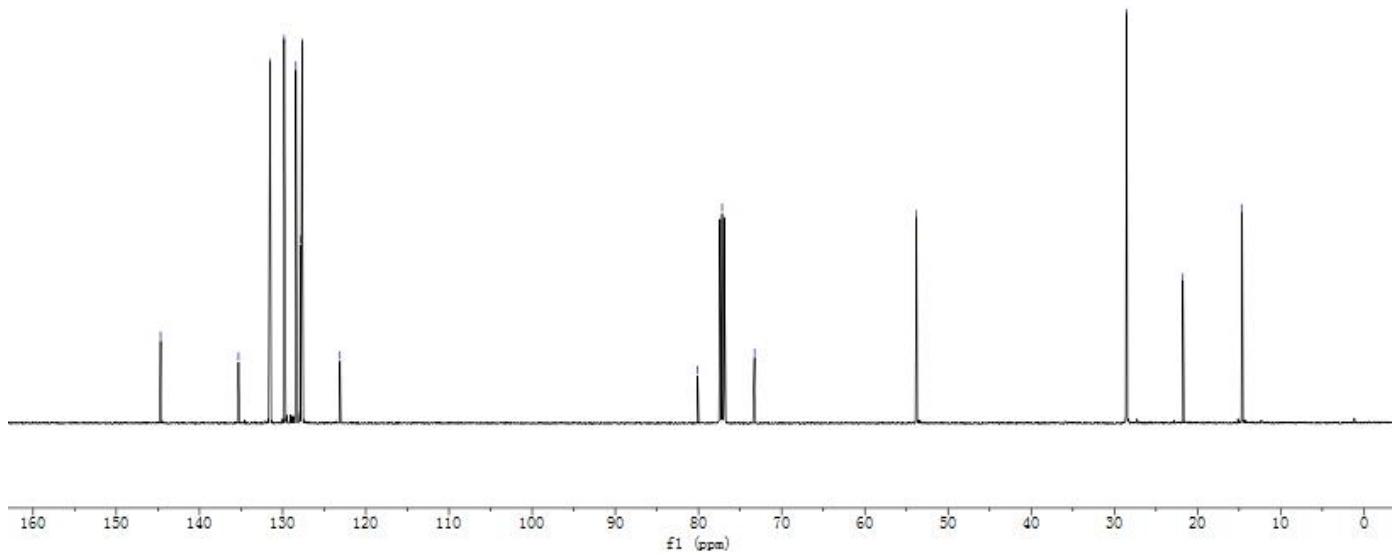
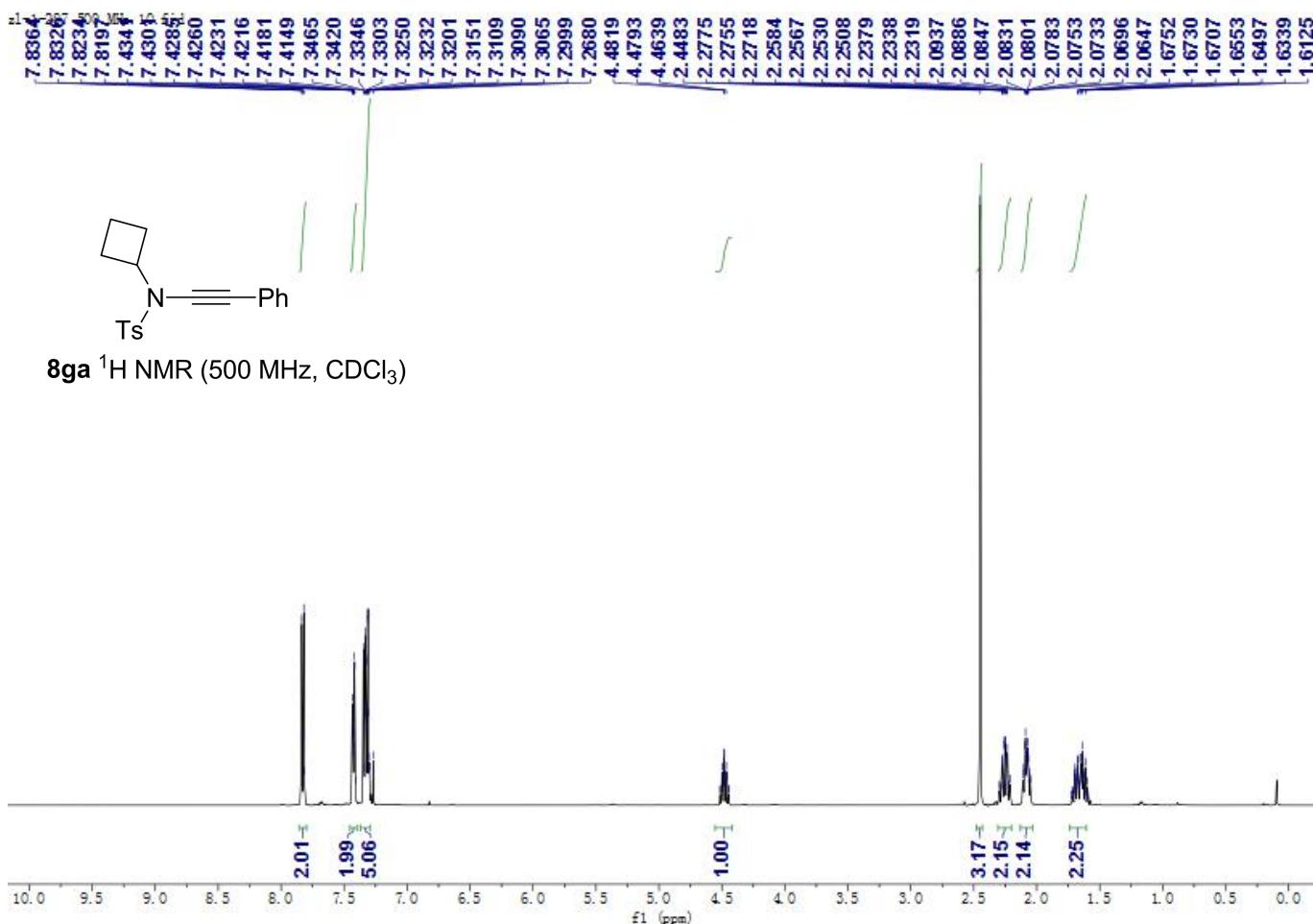
**8ca**  $^{13}\text{C}\{\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

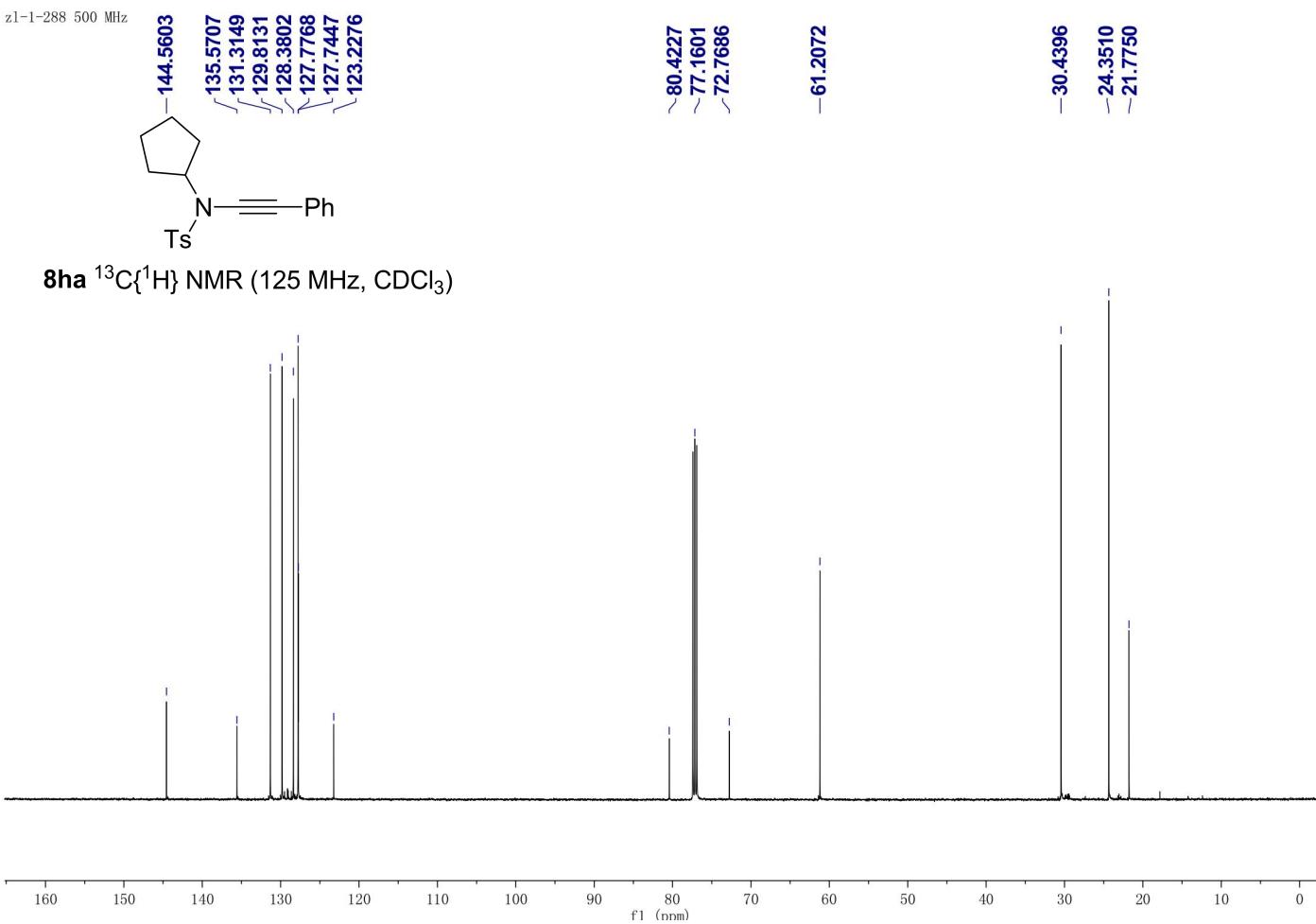
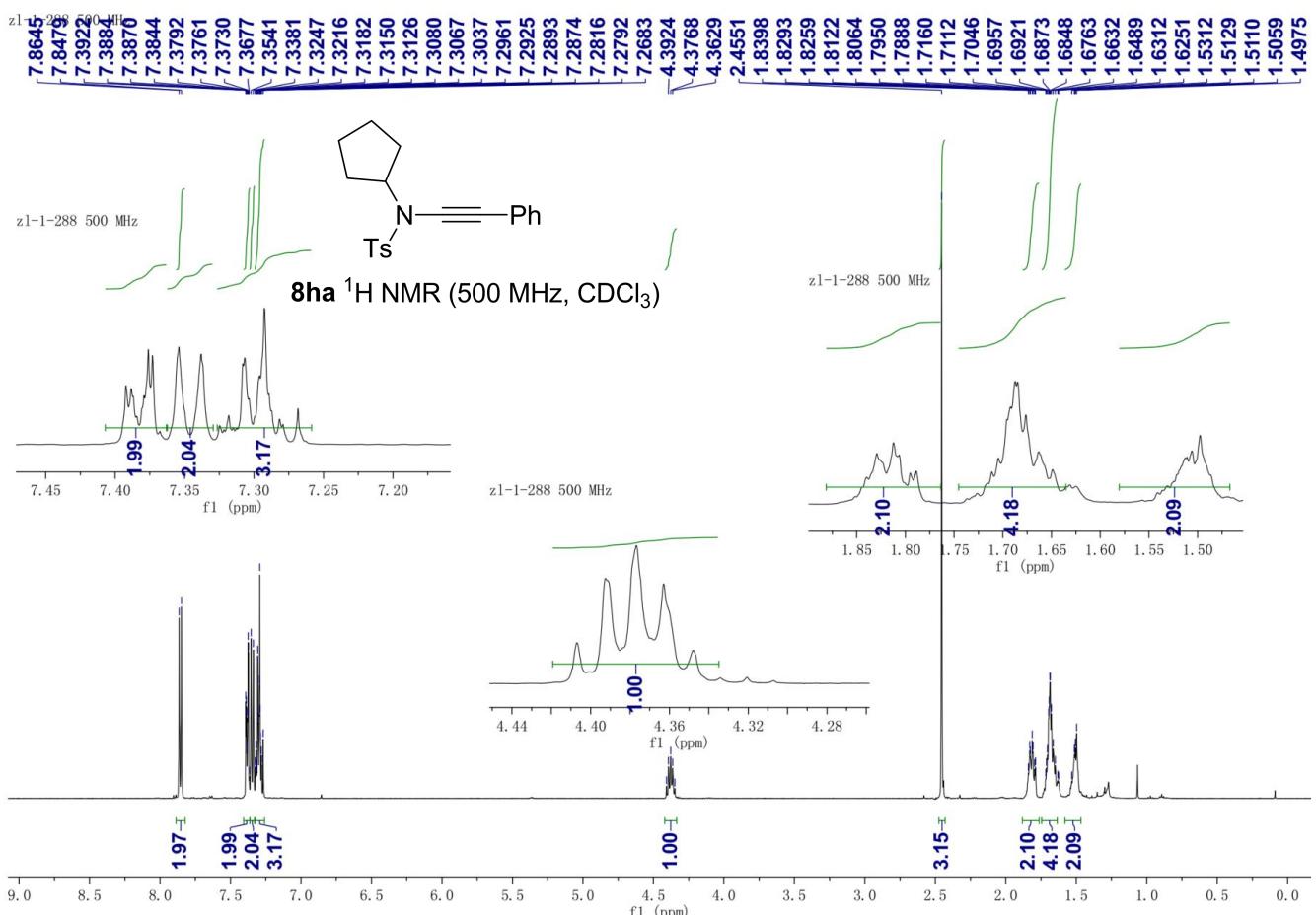




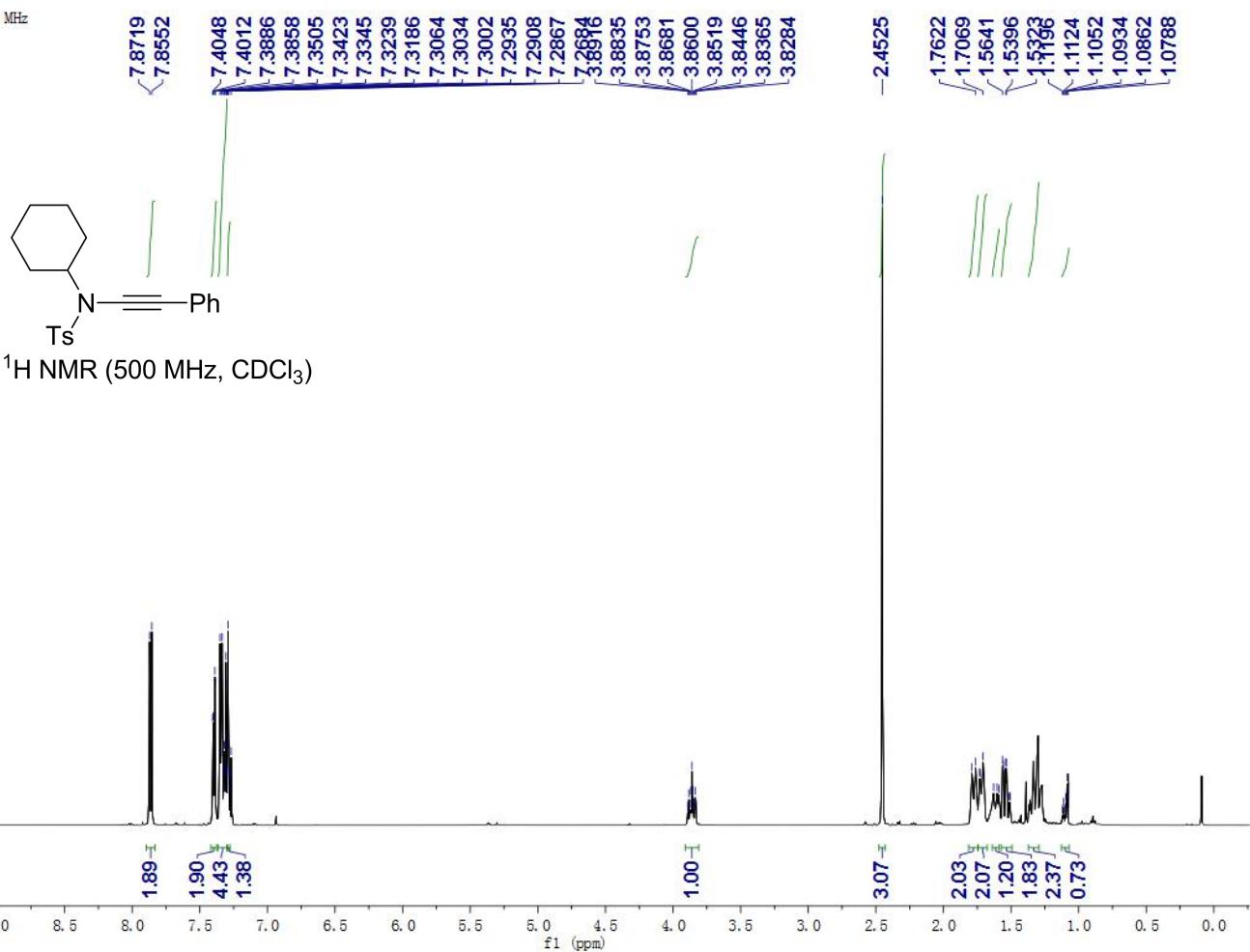


**8fa**  $^1\text{H}$  NMR (400 MHz,  $\text{d}_6$ -acetone)

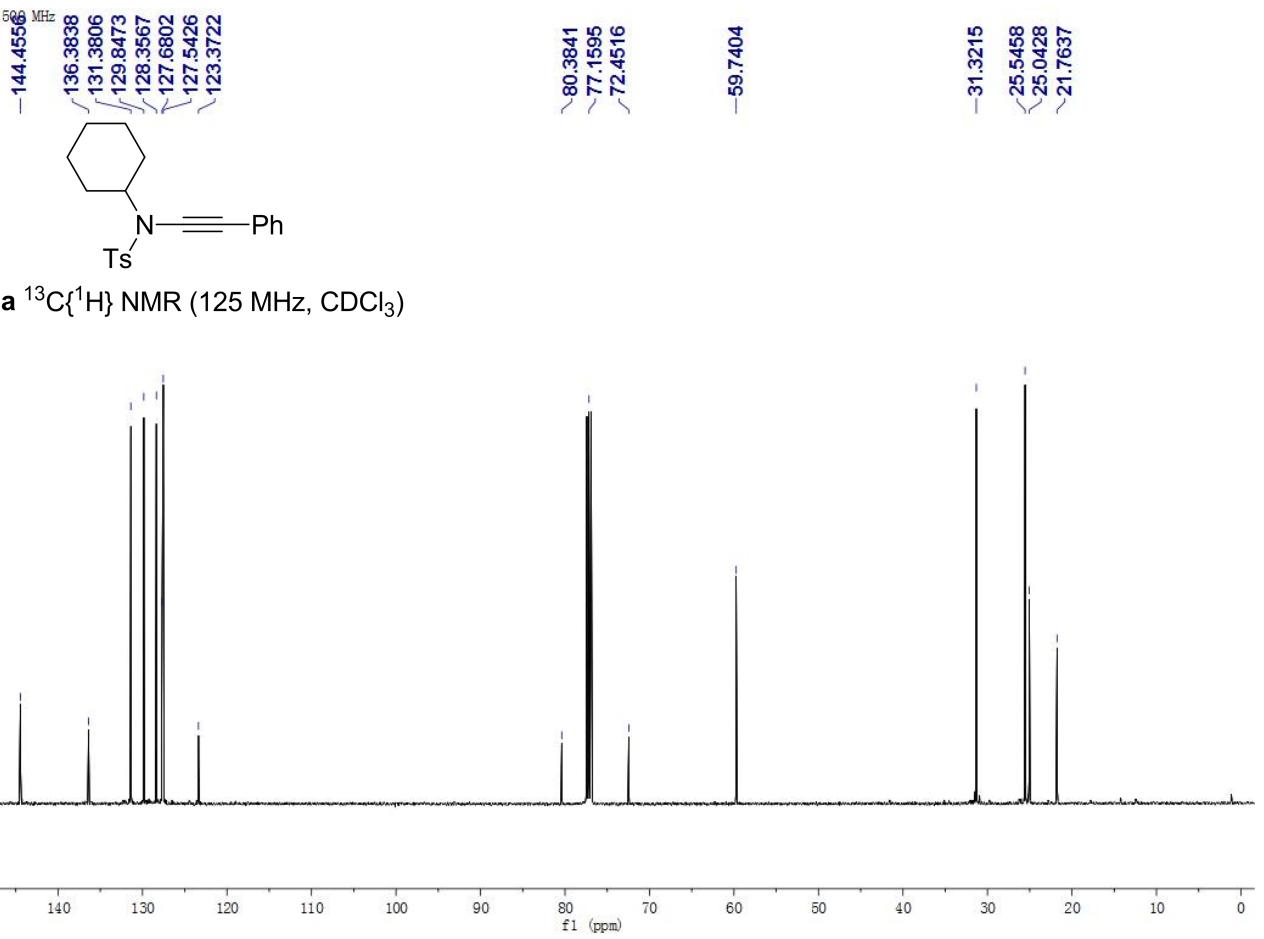




z1-1-284 500 MHz

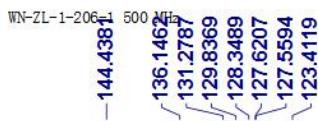
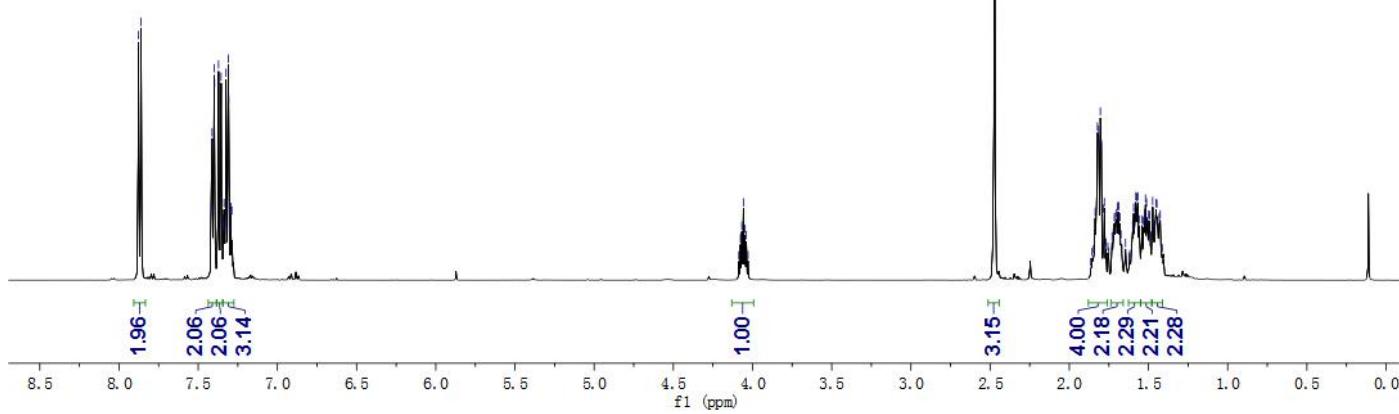


z1-1-284 500 MHz

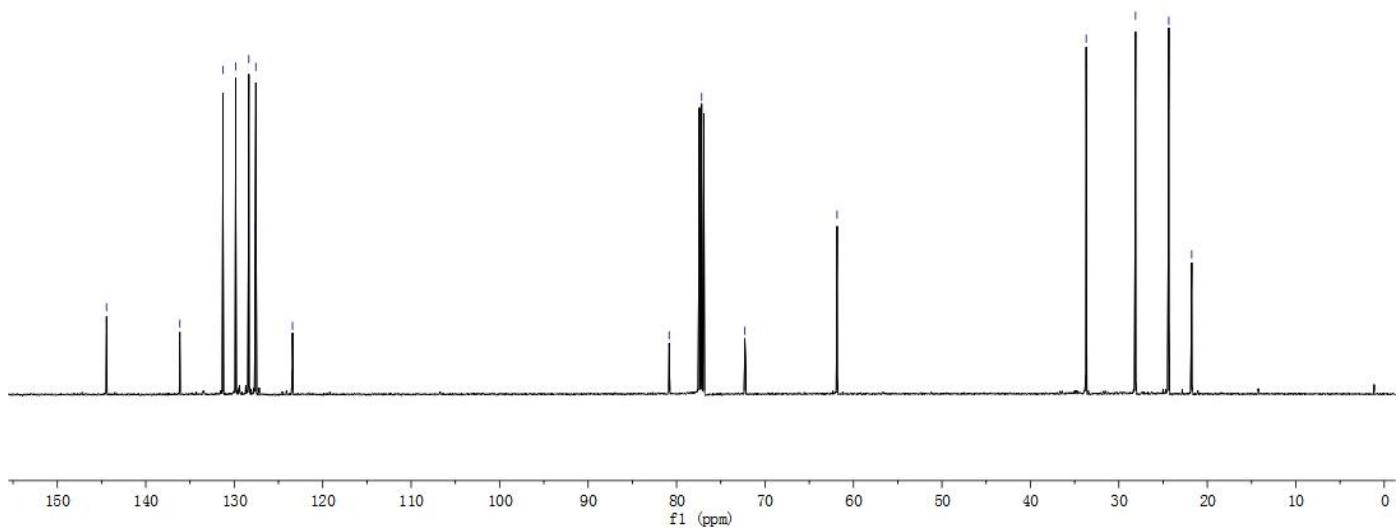


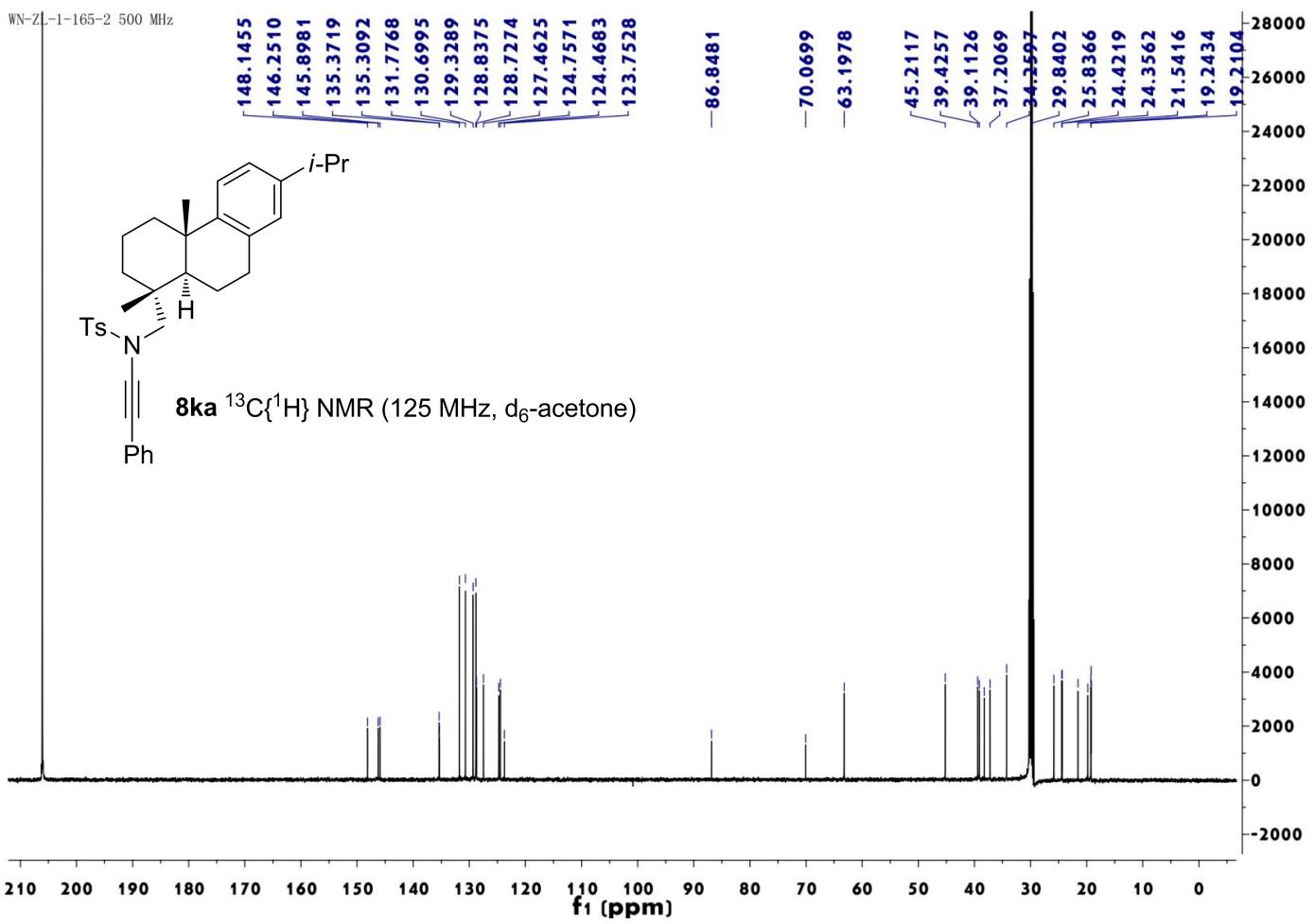
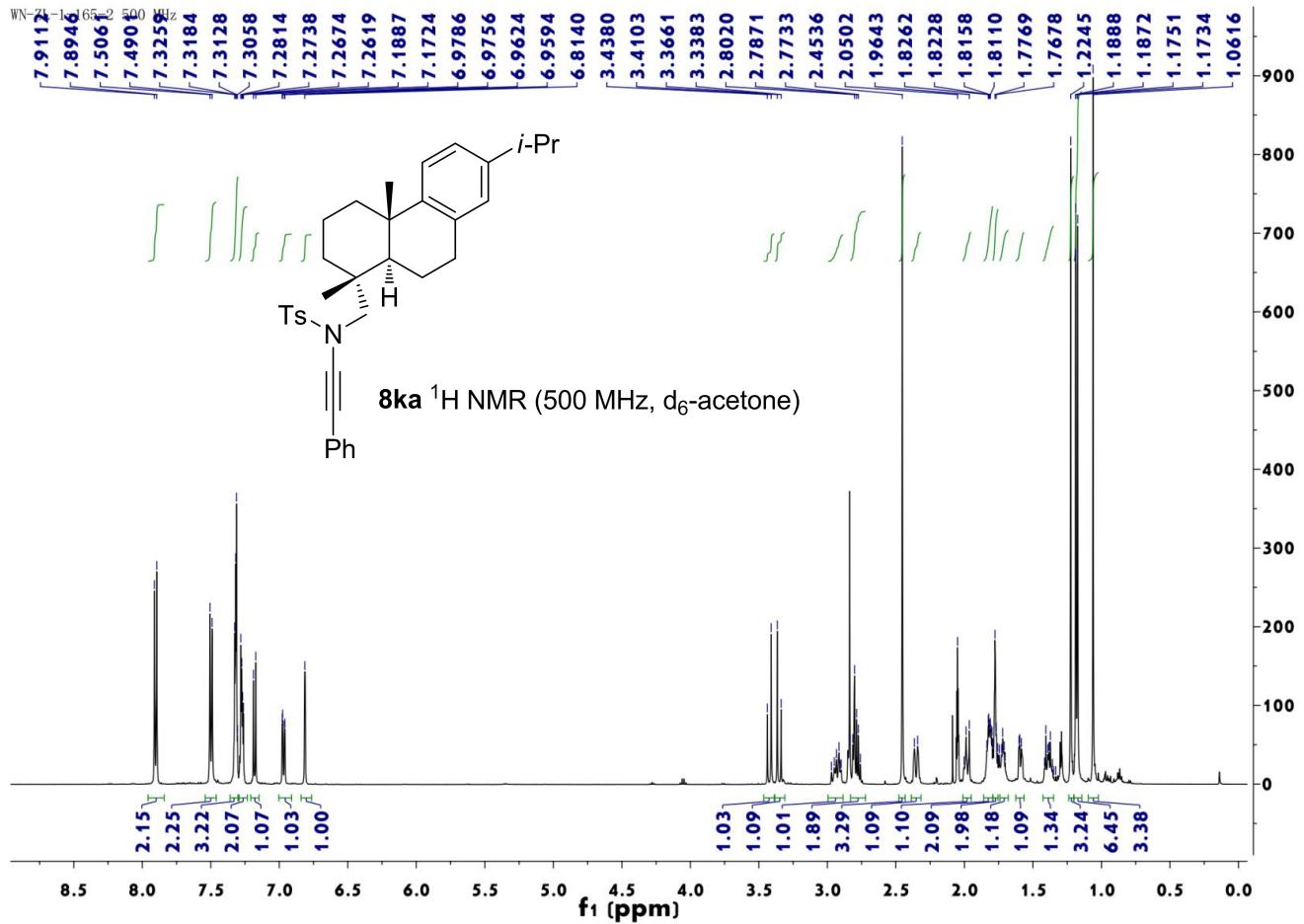


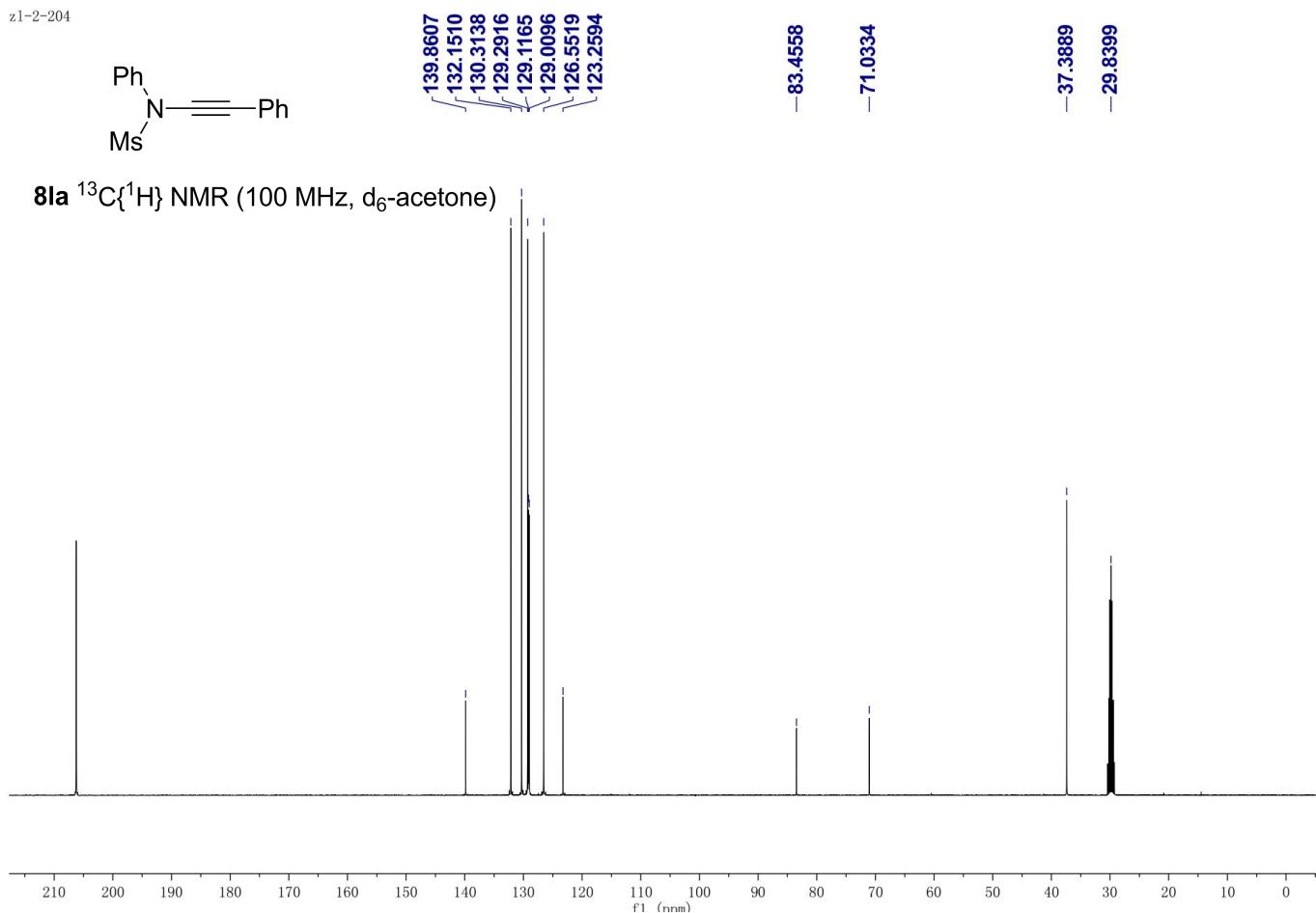
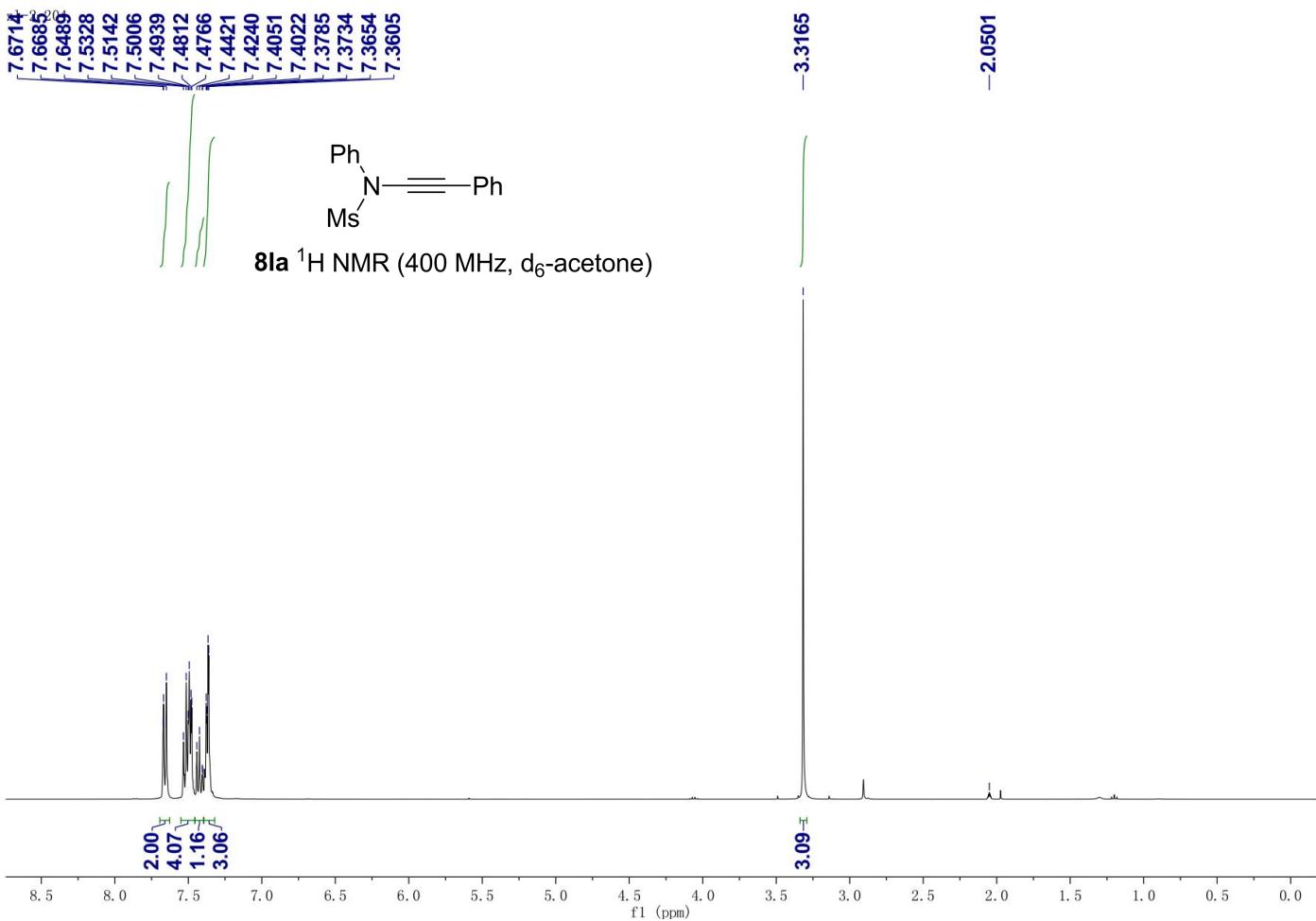
**8ja**  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )

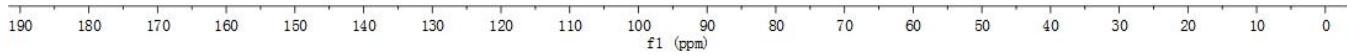
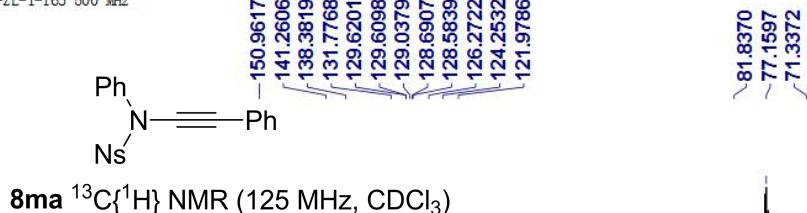
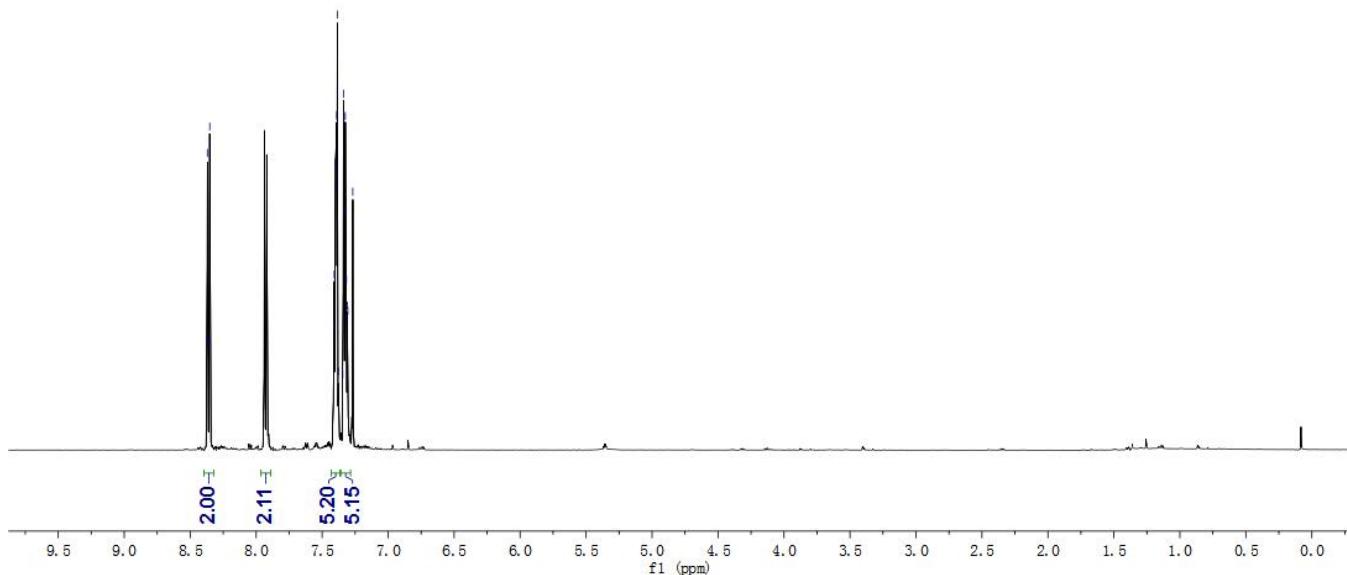
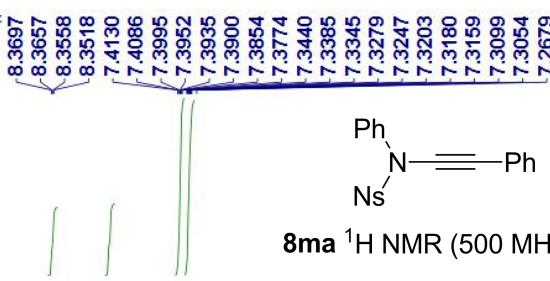


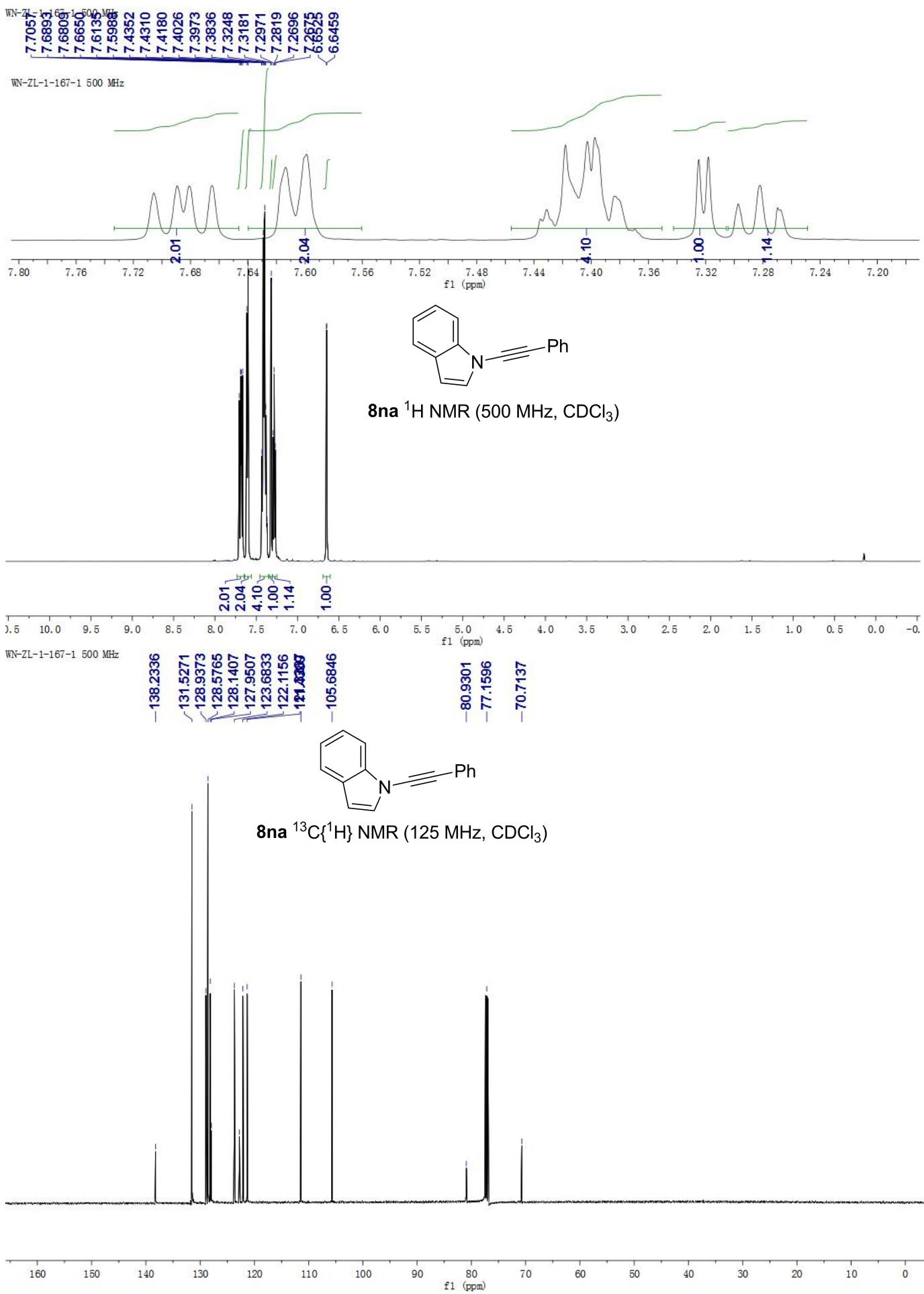
**8ja**  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz,  $\text{CDCl}_3$ )

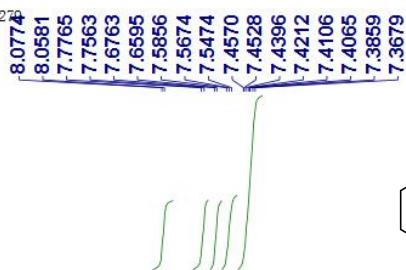
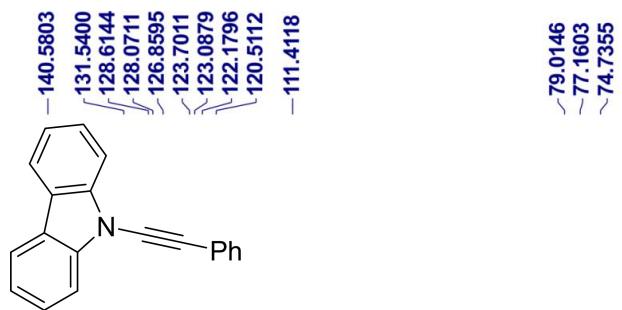
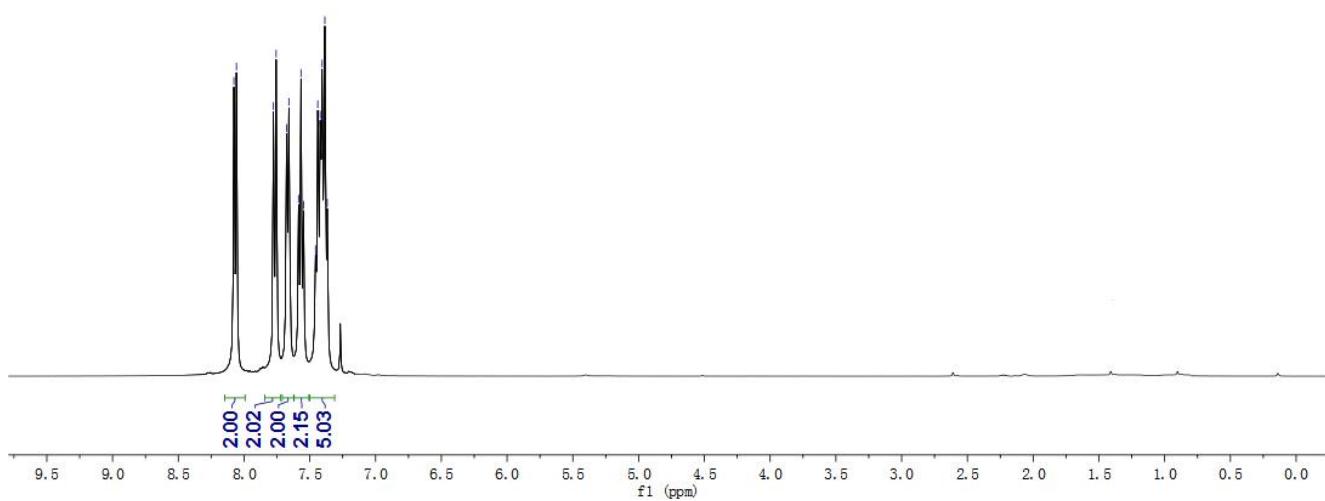
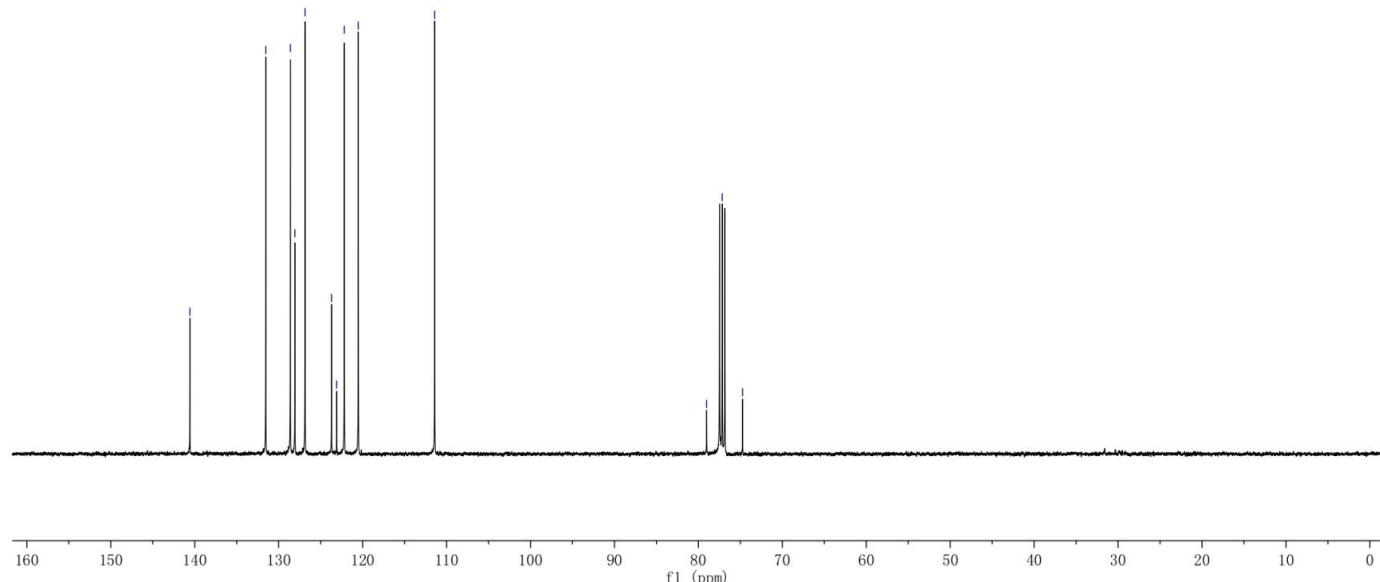


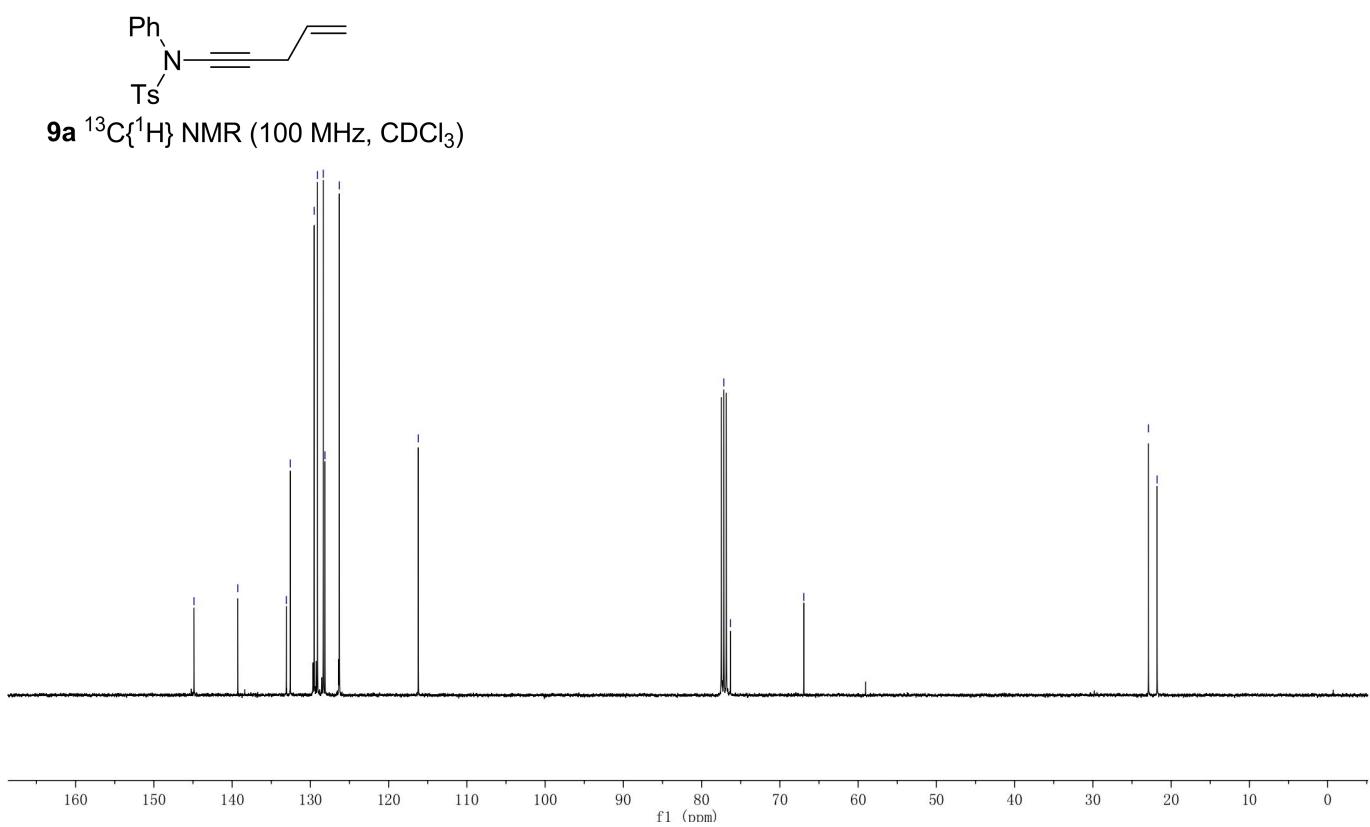
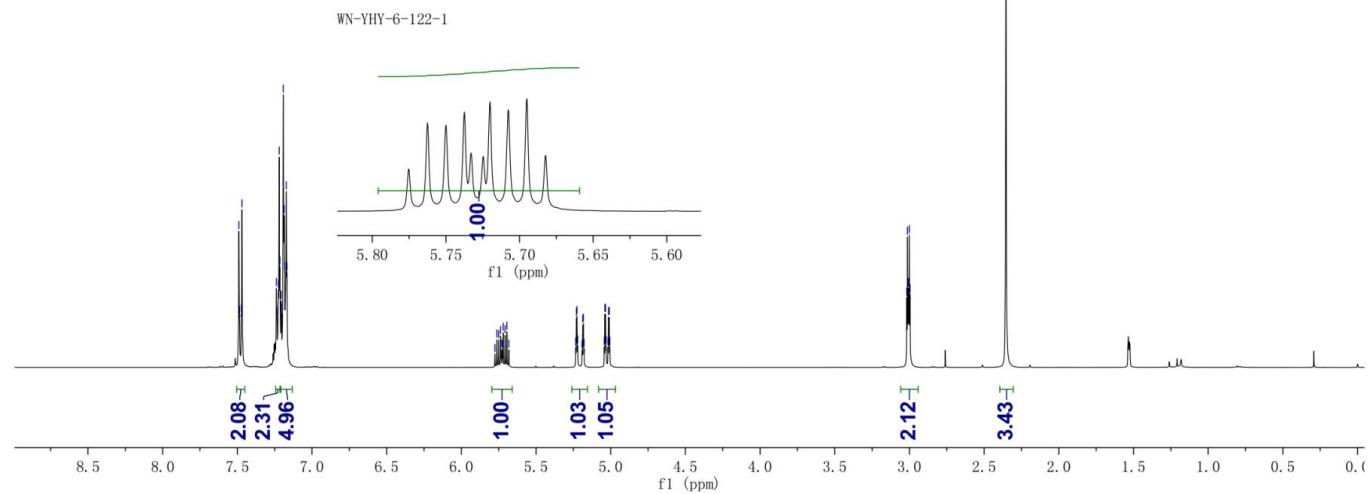
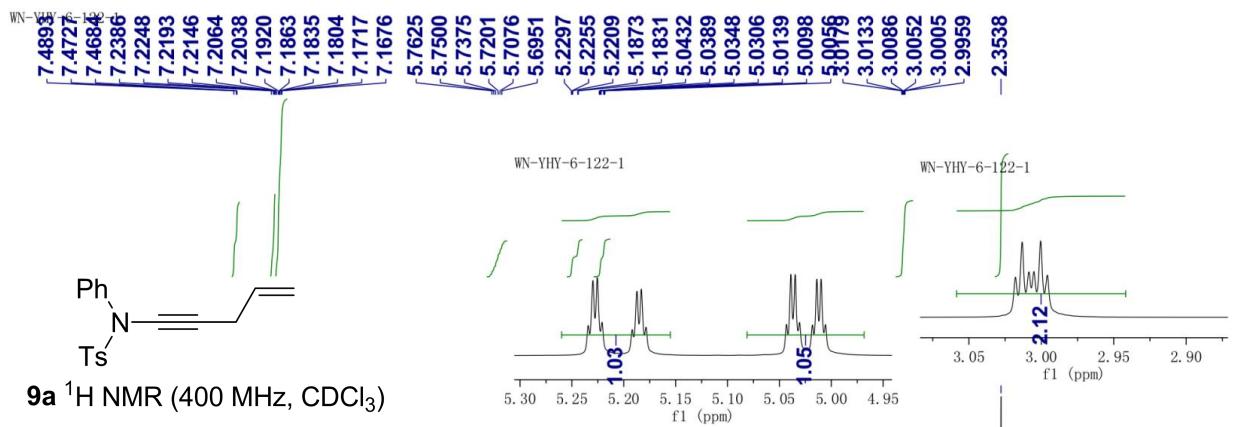


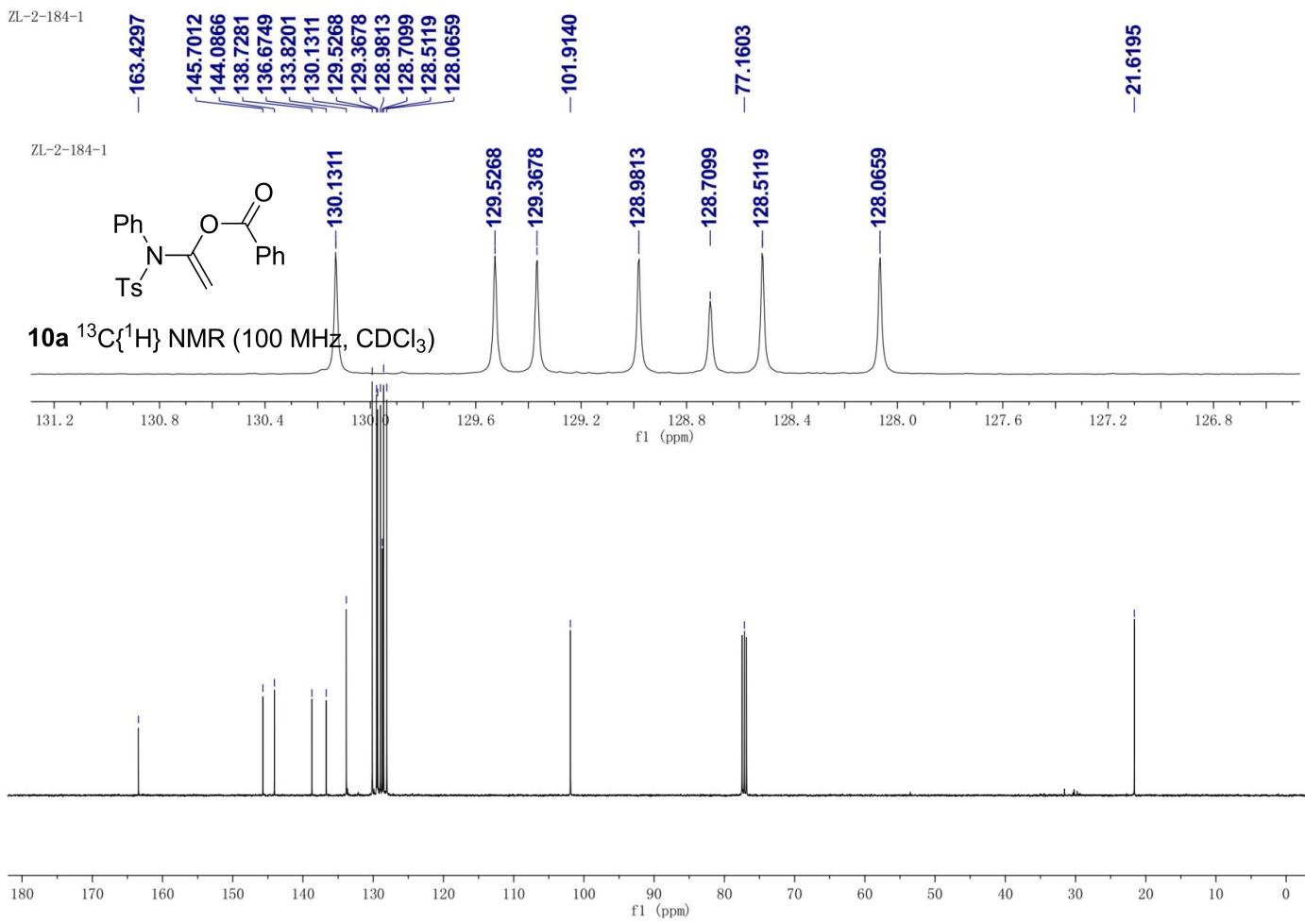
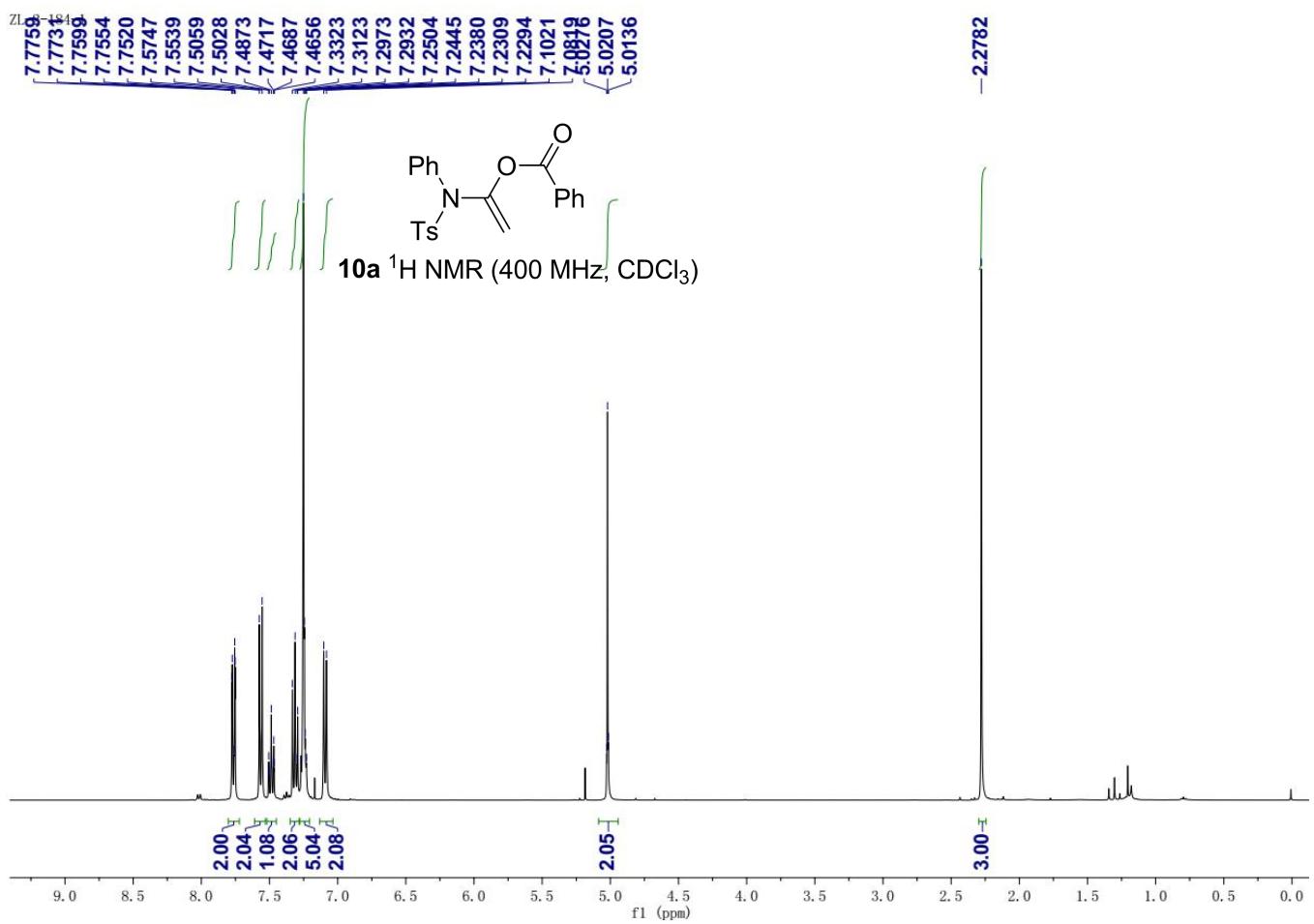


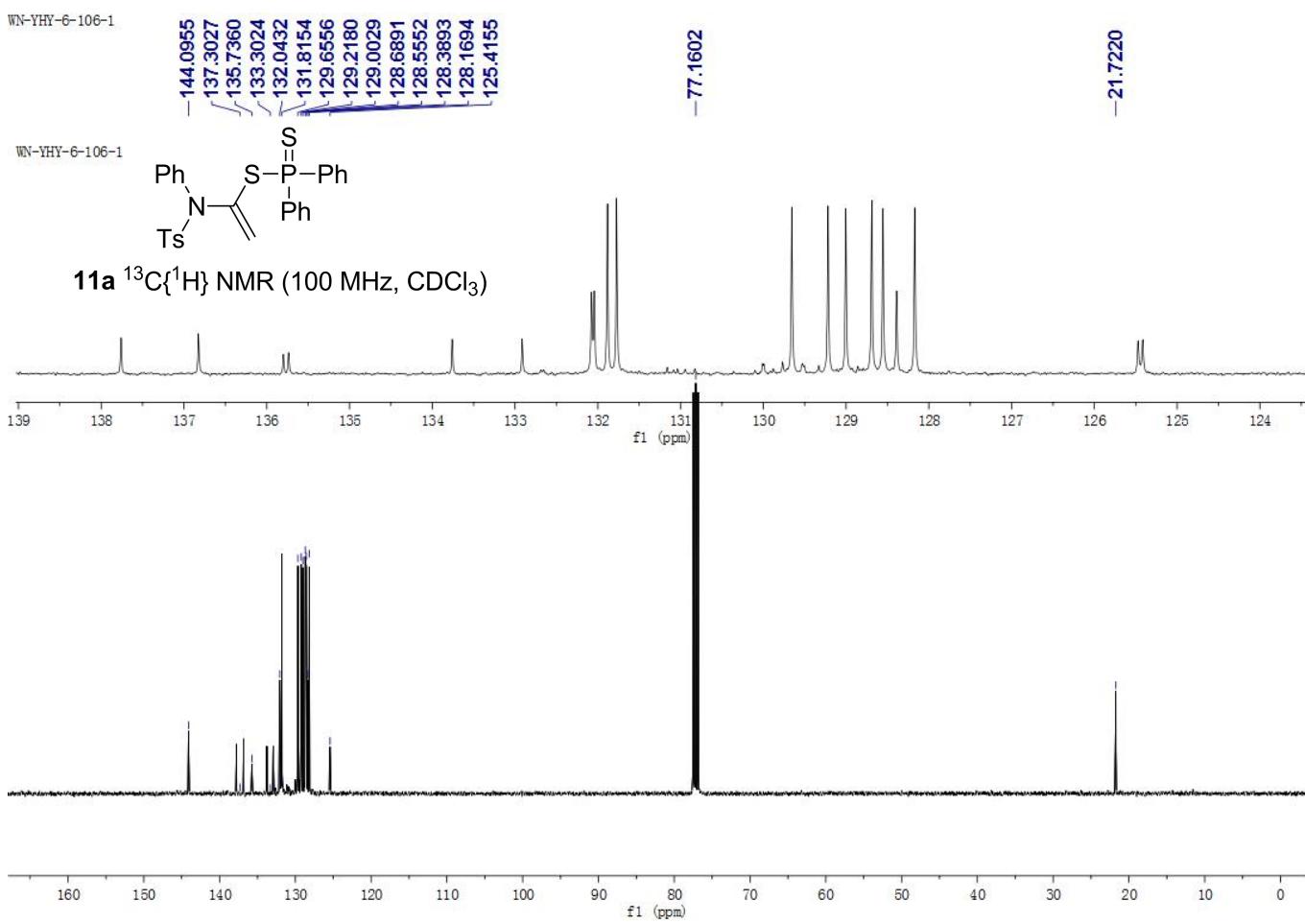
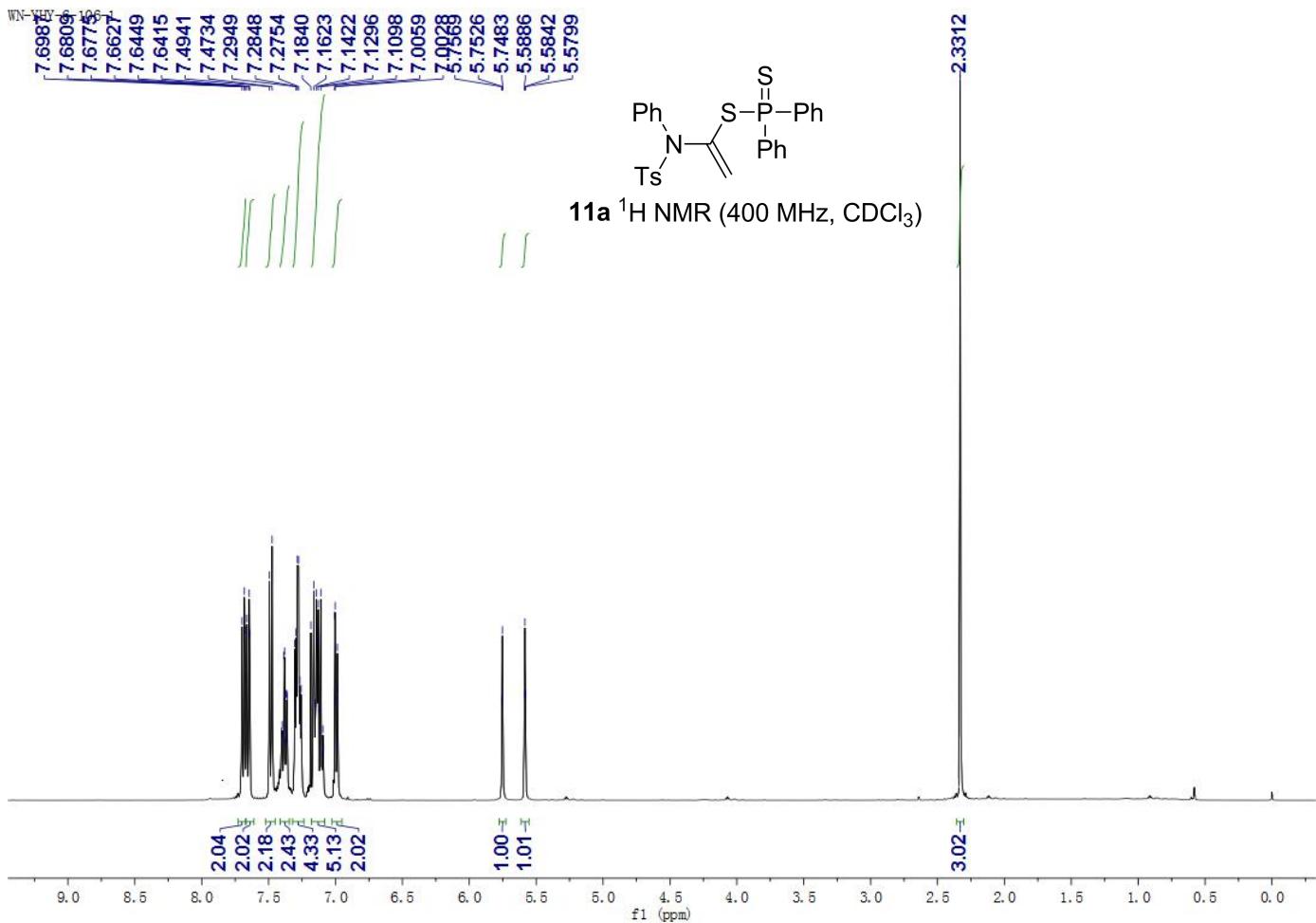




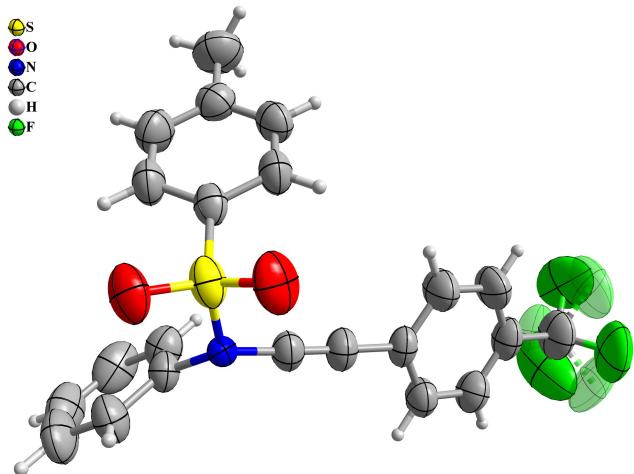
**8oa** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)**8oa** <sup>13</sup>C{<sup>1</sup>H} NMR (100 MHz, CDCl<sub>3</sub>)







## 2. X-ray crystal structure of **8aj**



ORTEP drawing of **8aj** showing thermal ellipsoids at the 50% probability level.

Crystal evaluation and data collection were performed at room temperature on a Bruker APEX2 CCD area detector diffractometer using graphite monochromated Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ , sealed X-ray tube). Using Olex2,<sup>1</sup> the structure was solved with the ShelXS<sup>2</sup> structure solution program using charge flipping and refined with the ShelXL<sup>3</sup> refinement package using least-squares minimization. All nonhydrogen atoms were refined with anisotropic displacement parameters. The hydrogen atoms on the structure were placed in idealized positions and refined using a riding model. The detailed crystallographic data and structure refinement parameters were summarized in Table S1. Selected bond lengths for compound **8aj** were collected in Table S2. Crystallographic data for the structural analyses has been deposited at the Cambridge Crystallographic Data Centre (CCDC reference numbers: 1890688).

White prisms; m.p. 151.8 - 152.6 °C; recrystallised in ethyl acetate and petroleum ether.

**Table S1.** Crystallographic data and structure refinement parameters of **8aj**

| Identification code | <b>8aj</b>   |
|---------------------|--|
| Empirical formula   | C <sub>22</sub> H <sub>16</sub> F <sub>3</sub> N <sub>3</sub> O <sub>2</sub> S |
| Formula weight (M)  | 415.42   |
| Crystal system      | monoclinic   |
| Space group         | <i>P</i> 2 <sub>1</sub> / <i>n</i>   |
| a (Å)               | 13.923(4)  |
| b (Å)               | 8.463(2)   |
| c (Å)               | 18.232(5)  |
| <i>α</i> (°)        | 90   |
| <i>β</i> (°)        | 109.888(4)   |
| <i>γ</i> (°)        | 90   |
| V/(Å <sup>3</sup> ) | 2020.0(9)  |

|                       |                 |
|-----------------------|-----------------|
| Z                     | 4               |
| Dc( Mg m-3)           | 1.366           |
| F(000)                | 856             |
| Reflections collected | 23039 / 4137    |
| unique                | R(int) = 0.0459 |
| Goodness-of-fit on F2 | 1.037           |
| Final R indices       | R1 = 0.0432     |
| I > 2σ(I)             | ωR2 = 0.1187    |
| R indices             | R1 = 0.0667     |
| (all data)            | ωR2 = 0.1358    |

**Table S2** Selected bond length for **8aj**

1.

### Atomic Distances

[Å]

|        |             |         |            |
|--------|-------------|---------|------------|
| S1—O1  | 1.4279 (15) | C8—F3'  | 1.261 (12) |
| S1—O2  | 1.4216 (15) | C8—F2'  | 1.308 (13) |
| S1—N1  | 1.6909 (18) | C8—F1'  | 1.29 (2)   |
| S1—C5  | 1.742 (2)   | C9—C10  | 1.379 (3)  |
| N1—C16 | 1.370 (2)   | C9—C14  | 1.372 (3)  |
| N1—C17 | 1.454 (2)   | C10—C11 | 1.373 (3)  |
| C1—C2  | 1.501 (3)   | C11—C12 | 1.390 (3)  |
| C2—C3  | 1.384 (3)   | C12—C13 | 1.375 (3)  |
| C2—C7  | 1.379 (3)   | C12—C15 | 1.442 (3)  |
| C3—C4  | 1.373 (3)   | C13—C14 | 1.379 (3)  |
| C4—C5  | 1.386 (3)   | C15—C16 | 1.180 (3)  |
| C5—C6  | 1.392 (3)   | C17—C18 | 1.380 (3)  |
| C6—C7  | 1.373 (3)   | C17—C22 | 1.372 (3)  |
| C8—C9  | 1.491 (3)   | C18—C19 | 1.379 (3)  |
| C8—F1  | 1.295 (8)   | C19—C20 | 1.354 (4)  |
| C8—F2  | 1.342 (6)   | C20—C21 | 1.371 (4)  |

---

|       |           |         |           |
|-------|-----------|---------|-----------|
| C8—F3 | 1.323 (6) | C21—C22 | 1.384 (4) |
|-------|-----------|---------|-----------|

---

### 3. Reference:

- (1) Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. OLEX<sup>2</sup>: a complete structure solution, refinement and analysis program. *J. Appl. Crystallogr.* **2009**, *42*, 339-341.
- (2) Sheldrick, G. M. A short history of SHELX. *Acta. Crystallogr. A*. **2008**, *A64*, 112-122.
- (3) Sheldrick, G. M. Crystal structure refinement with SHELXL. *Acta. Crystallogr. C*. **2015**, *C71*, 3-8.