# Supporting Information 

# Regioselective C-H Alkenylation and Unsymmetrical BisOlefination of Heteroarene Carboxylic Acids with Ruthenium Catalysis in Water 

Anup Mandal, Ratnadeep Bera, and Mahiuddin Baidya*
Department of Chemistry, Indian Institute of Technology Madras,
Chennai-600036, Tamil Nadu, India
E-mail: mbaidya@iitm.ac.in

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## Mechanistic Studies:

## 1) H-D Exchange Study



5-Methyl-2-thiophenecarboxylic acid $\mathbf{1 i}(0.2 \mathrm{mmol})$, phenyl vinyl sulfone 2a ( 1.1 equiv), $[\mathrm{Ru}(p$ cymene $\left.) \mathrm{Cl}_{2}\right]_{2}(5 \mathrm{~mol} \%)$, and $\mathrm{Cu}(\mathrm{OAc})_{2} \cdot \mathrm{H}_{2} \mathrm{O}$ (1 equiv) were added in an oven-dried screw cap reaction tube with a magnetic stir bar under open-air. Then, $\mathrm{D}_{2} \mathrm{O}(1 \mathrm{~mL})$ was added with a syringe. The reaction tube was capped and the resulting mixture was heated at $100^{\circ} \mathrm{C}$ (in oil-bath) for 35 minutes. After that, it was allowed to cool at room temperature and then quenched with AcOH and diluted with $\mathrm{NH}_{4} \mathrm{Cl}$ solution. The mixture was extracted with ethyl acetate ( 10 mL , two times). The organic layer was dried over anhydrous sodium sulfate and concentrated under reduced pressure. The resulting deuterium incorporated 5-methyl-2-thiophenecarboxylic acid and the olefinated product were purified by silica gel column chromatography. The H/D exchange result was determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy.


Figure S1. ${ }^{1}$ H NMR spectrum of deuterium-incorporated 5-methyl-2-thiophenecarboxylic acid $\mathbf{1 i}$ '.

## 2) Radical Scavenger Experiment



2-Thiophenecarboxylic acid $\mathbf{1 a}(0.2 \mathrm{mmol})$ and phenyl vinyl sulfone $\mathbf{2 a}$ (1.1 equiv), $\left[\mathrm{Ru}(p \text {-cymene }) \mathrm{Cl}_{2}\right]_{2}$ ( $5 \mathrm{~mol} \%$ ), $\mathrm{Cu}(\mathrm{OAc})_{2} \cdot \mathrm{H}_{2} \mathrm{O}$ (1 equiv), and corresponding radical scavenger (3 equiv) were added in an oven-dried screw cap reaction tube with a magnetic stir bar under open-air. Then, $\mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL})$ was added with a syringe. The reaction tube was capped and the resulting mixture was heated at $100^{\circ} \mathrm{C}$ (in oil-bath) for 24 h . After that, it was allowed to cool at room temperature and then quenched with AcOH and diluted with $\mathrm{NH}_{4} \mathrm{Cl}$ solution. The mixture was extracted with ethyl acetate ( 10 mL , two times). The organic layer was dried over anhydrous sodium sulfate and concentrated under reduced pressure. The crude product was treated with $\mathrm{K}_{2} \mathrm{CO}_{3}$ (2 equiv), MeI ( 3 equiv) in MeCN ( 1 mL ) at room temperature for 4 h and then concentrated under reduced pressure. The resulting residue was purified by silica gel column chromatography.

## 3) Kinetic Isotope Effect Study via Parallel Experiment




$$
K I E=k_{\mathrm{H}} / k_{\mathrm{D}}=1.68
$$

5-Methyl-2-thiophenecarboxylic acid $\mathbf{1 i}$ and deuterated-5-methyl-2-thiophenecarboxylic acid $\mathbf{1 i \prime}$ were independently reacted with 2a for five different time intervals (10-40 minutes) under the standard reaction conditions [ $\mathbf{1 i}(0.2 \mathrm{mmol})$ or $\mathbf{1 i}^{\prime \prime}(0.2 \mathrm{mmol})$, phenyl vinyl sulfone $\mathbf{2 a}$ ( 1.1 equiv), $\left[\mathrm{Ru}(p \text {-cymene }) \mathrm{Cl}_{2}\right]_{2}(5$ $\mathrm{mol} \%$ ), and $\mathrm{Cu}(\mathrm{OAc})_{2} \cdot \mathrm{H}_{2} \mathrm{O}$ (1 equiv), $\left.\mathrm{H}_{2} \mathrm{O}(1 \mathrm{~mL})\right]$. The product distributions were analyzed from the worked-up crude reaction mixture by ${ }^{1} \mathrm{H}$ NMR spectroscopy using dibromomethane as an internal standard.

| time (min) | 10 min | 15 min | 20 min | 25 min | 30 min | 35 min | 40 min |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1 i} \rightarrow$ Product (\%) | 27.1 | 33.2 | 36.3 | 41.2 | 43.2 | - | - |
| $\mathbf{1 1}^{\prime \prime} \rightarrow$ Product (\%) | - | 17.4 | 20.9 | - | 24.2 | 26.7 | 30.3 |



Figure S2. Time vs yield plot for KIE determination.

## Crystallographic Experimental Section:

Single Crystal X-ray Crystallography: X-ray data of the crystals were collected and integrated using a Bruker Axs (Kappa Apex 2) CCD diffractometer equipped with graphite monochromatic Mo ( $\mathrm{K} \alpha$ ) radiation. Crystals were mounted over fine nylon loop which was attached to the copper mounting pin held on by a magnetic base. The APEX 3 and APEX 3-SAINT/Bruker SAINT programs were used for the data collection and unit-cell determination, respectively. The crystal structures were solved by direct methods using SHELXL-2014/4 or SHELXS-97 and refined by full-matrix least-squares on $\mathrm{F}^{2}$ method using program SHELXL-2014/7 or SHELXL-2018/3.

Method of Crystallization: All the single crystals were grown in a small glass vial by slow evaporation technique from DCM/Hexane solvent system at room temperature over a period of 1-2 weeks.

Crystal Structure of Compound 3c: ORTEP representation (40\% ellipsoids probability) of 3c (CCDC 1975675).


Table S1. Crystal data and structure refinement for 3c (CCDC 1975675).

| Identification code | 532 |
| :--- | :--- |
| Empirical formula | C 14 H 12 O 5 S |
| Formula weight | 292.30 |
| Temperature | $296(2) \mathrm{K}$ |
| Wavelength | $0.71073 \AA$ |
| Crystal system, space group | Orthorhombic, Pbca |
| Unit cell dimensions | $\mathrm{a}=7.6046(2) \AA \alpha=90^{\circ}$ |
|  | $\mathrm{b}=15.5125(5) \AA \beta=90^{\circ}$ |
|  | $\mathrm{c}=23.2195(7) \AA \gamma=90^{\circ}$ |
| Volume | $2739.12(14) \mathrm{A}^{\wedge} 3$ |
| Z, Calculated density | $8,1.418 \mathrm{Mg} / \mathrm{m}^{\wedge} 3$ |
| Absorption coefficient | $0.252 \mathrm{~mm} \wedge-1$ |
| F(000) | 1216 |
| Crystal size | $0.250 \times 0.220 \times 0.100 \mathrm{~mm}$ |
| Theta range for data collection | 2.769 to 24.999 deg. |
| Limiting indices | $-7<=\mathrm{h}<=9,-18<=\mathrm{k}<=15,-27<=1<=27$ |
| Reflections collected / unique | $15544 / 2406[\mathrm{R}(\mathrm{int})=0.0244]$ |
| Completeness to theta $=25.000$ | $99.9 \%$ |
| Absorption correction | None |
| Refinement method | Full-matrix least-squares on $\mathrm{F} \wedge 2$ |
| Data / restraints / parameters | $2406 / 0 / 182$ |
| Goodness-of-fit on $\mathrm{F} \wedge 2$ | 1.054 |

Final R indices $[\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})]$
$\mathrm{R} 1=0.0352, \mathrm{wR} 2=0.0837$
R indices (all data)
Extinction coefficient
$\mathrm{R} 1=0.0445, \mathrm{wR} 2=0.0916$

Largest diff. peak and hole
n/a
0.184 and -0.305 e. $\mathrm{A}^{\wedge}-3$

Crystal Structure of Compound 3f: ORTEP representation ( $40 \%$ ellipsoids probability) of $\mathbf{3 f}$ (CCDC 1975672).


Table S2. Crystal data and structure refinement for $\mathbf{3 f}$ (CCDC 1975672).

| Identification code | 787 |
| :--- | :--- |
| Empirical formula | C 18 H 14 O 5 S |
| Formula weight | 342.35 |
| Temperature | $296(2) \mathrm{K}$ |
| Wavelength | $0.71073 \AA$ |
| Crystal system, space group | Orthorhombic, Pbca |
| Unit cell dimensions | $\mathrm{a}=17.7265(7) \AA \alpha=90^{\circ}$ |
|  | $\mathrm{b}=11.8090(4) \AA \beta=90^{\circ}$ |
|  | $\mathrm{c}=15.2926(4) \AA \gamma=90^{\circ}$ |
| Volume | $3201.23(19) \mathrm{A}^{\wedge} 3$ |
| Z, Calculated density | $8,1.421 \mathrm{Mg} / \mathrm{m}^{\wedge} 3$ |
| Absorption coefficient | $0.227 \mathrm{~mm} \wedge-1$ |
| F(000) | 1424 |
| Crystal size | 0.250 x 0.220 x 0.100 mm |
| Theta range for data collection | 2.072 to 24.998 deg. |

Limiting indices
Reflections collected / unique
Completeness to theta $=25.000$
Absorption correction
Refinement method
Data / restraints / parameters
Goodness-of-fit on $\mathrm{F}^{\wedge} 2$
Final R indices $[\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})]$
R indices (all data)
Extinction coefficient
Largest diff. peak and hole
$-13<=\mathrm{h}<=21,-13<=\mathrm{k}<=14,-18<=1<=12$
$10818 / 2815[\mathrm{R}($ int $)=0.0224]$
99.9 \%

None
Full-matrix least-squares on $\mathrm{F}^{\wedge} 2$
2815/0/218
1.053
$\mathrm{R} 1=0.0403, \mathrm{wR} 2=0.0959$
$R 1=0.0531, w R 2=0.1040$
n/a
0.277 and -0.301 e. $\mathrm{A}^{\wedge}-3$

Crystal Structure of Compound $\mathbf{3} \mathbf{h}$ : ORTEP representation ( $40 \%$ ellipsoids probability) of $\mathbf{3 h}$ (CCDC 1975673).


Table S3. Crystal data and structure refinement for $\mathbf{3 h}$ (CCDC 1975673).

Identification code
Empirical formula
Formula weight
Temperature
Wavelength
Crystal system, space group
Unit cell dimensions

28
C16 H14 O6 S2
366.39

296(2) K
0.71073 Å

Orthorhombic, Pbca
$\mathrm{a}=8.0160(2) \AA \alpha=90^{\circ}$

$$
\begin{aligned}
& \mathrm{b}=19.6309(8) \AA \beta=90^{\circ} \\
& \mathrm{c}=21.6558(8) \AA \gamma=90^{\circ}
\end{aligned}
$$

Volume
Z, Calculated density
Absorption coefficient
F(000)
Crystal size
Theta range for data collection
Limiting indices
Reflections collected / unique
Completeness to theta $=25.000$
Absorption correction
Refinement method
Data / restraints / parameters
Goodness-of-fit on $\mathrm{F}^{\wedge} 2$
Final R indices $[\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})]$
R indices (all data)
Extinction coefficient
Largest diff. peak and hole
3407.8(2) A^3
$8,1.428 \mathrm{Mg} / \mathrm{m}^{\wedge} 3$
$0.341 \mathrm{~mm}^{\wedge}-1$
1520
$0.250 \times 0.220 \times 0.100 \mathrm{~mm}$
1.881 to 24.996 deg.
$-9<=\mathrm{h}<=9,-21<=\mathrm{k}<=23,-25<=1<=19$
$7967 / 2995[R($ int $)=0.0402]$
99.9 \%

None
Full-matrix least-squares on $\mathrm{F}^{\wedge} 2$
2995 / 0 / 219
1.009
$\mathrm{R} 1=0.0486, \mathrm{wR} 2=0.0978$
$\mathrm{R} 1=0.0957, \mathrm{wR} 2=0.1207$
n/a
0.243 and -0.300 e. $\mathrm{A}^{\wedge}-3$

Crystal Structure of Compound 4g: ORTEP representation (40\% ellipsoids probability) of $\mathbf{4 g}$ (CCDC 1975676).


Table S4. Crystal data and structure refinement for $\mathbf{4 g}$ (CCDC 1975676).

| Identification code | 891 |
| :---: | :---: |
| Empirical formula | C17 H22 N O5 P |
| Formula weight | 351.32 |
| Temperature | 296(2) K |
| Wavelength | 0.71073 A |
| Crystal system, space group | Triclinic, P-1 |
| Unit cell dimensions | $\begin{aligned} & a=7.4902(15) \AA \alpha=88.249(6)^{\circ} \\ & b=10.012(2) \AA \beta=82.343(6)^{\circ} \\ & c=25.476(5) \AA \gamma=71.214(6)^{\circ} \end{aligned}$ |
| Volume | 1792.4(6) $\AA^{\wedge} \wedge 3$ |
| Z, Calculated density | $4,1.302 \mathrm{Mg} / \mathrm{m}^{\wedge} 3$ |
| Absorption coefficient | $0.179 \mathrm{~mm}^{\wedge}-1$ |
| F(000) | 744 |
| Crystal size | $0.200 \times 0.150 \times 0.150 \mathrm{~mm} \wedge 3$ |
| Theta range for data collection | 0.807 to 24.997 deg . |
| Limiting indices | $-8<=\mathrm{h}<=8,-11<=\mathrm{k}<=11,-30<=1<=30$ |
| Reflections collected / unique | $32648 / 32648$ [R(int) $=$ ?] |
| Completeness to theta $=25.000$ | 99.9\% |
| Absorption correction | Semi-empirical from equivalents |
| Max. and min. transmission | 0.7452 and 0.5243 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{\wedge} 2$ |
| Data / restraints / parameters | 32648 / 39 / 454 |
| Goodness-of-fit on $\mathrm{F}^{\wedge} 2$ | 1.107 |
| Final R indices [ $\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})$ ] | $\mathrm{R} 1=0.1044, \mathrm{wR} 2=0.2823$ |
| R indices (all data) | $\mathrm{R} 1=0.1436, \mathrm{wR} 2=0.3062$ |
| Extinction coefficient | 0.017(3) |
| Largest diff. peak and hole | 0.470 and -0.495 e. $\AA^{\wedge}$ - 3 |

Crystal Structure of Compound $\mathbf{5 g}$ : ORTEP representation ( $40 \%$ ellipsoids probability) of $\mathbf{5 g}$ (CCDC 1975674).


Table S5. Crystal data and structure refinement for $\mathbf{5 g}$ (CCDC 1975674).

| Identification code | 21 |
| :--- | :--- |
| Empirical formula | C 15 H 15 N O 4 |
| Formula weight | 546.56 |
| Temperature | $296(2) \mathrm{K}$ |
| Wavelength | $0.71073 \AA$ |
| Crystal system, space group | Triclinic, $\mathrm{P}-1$ |
| Unit cell dimensions | $\mathrm{a}=10.9033(4) \AA \alpha=100.1998(15)^{\circ}$ |
|  | $\mathrm{b}=11.1118(4) \AA \beta=92.2428(17)^{\circ}$ |
|  | $\mathrm{c}=13.1220(4) \AA \gamma=118.6290(14)^{\circ}$ |
| Volume | $1359.16(8) \mathrm{A}^{\wedge} 3$ |
| Z, Calculated density | $2,1.336 \mathrm{Mg} / \mathrm{m} \wedge 3$ |
| Absorption coefficient | $0.098 \mathrm{~mm} \wedge-1$ |
| F(000) | 576 |
| Crystal size | $0.250 \times 0.220 \times 0.100 \mathrm{~mm}$ |
| Theta range for data collection | 1.593 to 24.999 deg. |
| Limiting indices | $-12<=\mathrm{h}<=12,-13<=\mathrm{k}<=12,-15<=1<=15$ |
| Reflections collected $/$ unique | $20423 / 4779[\mathrm{R}(\mathrm{int})=0.0272]$ |

Completeness to theta $=25.000$
Absorption correction
Refinement method
Data / restraints / parameters
Goodness-of-fit on $\mathrm{F}^{\wedge} 2$
Final R indices $[\mathrm{I}>2 \operatorname{sigma}(\mathrm{I})]$
R indices (all data)
Extinction coefficient
Largest diff. peak and hole
99.6 \%

None
Full-matrix least-squares on $\mathrm{F}^{\wedge} 2$
4779 / 0 / 368
1.026
$\mathrm{R} 1=0.0453, \mathrm{wR} 2=0.1172$
$\mathrm{R} 1=0.0625, \mathrm{wR} 2=0.1333$
0.014(2)
0.305 and -0.264 e. $\mathrm{A}^{\wedge}-3$

Crystal Structure of Compound 6b: ORTEP representation (40\% ellipsoids probability) of $\mathbf{6 b}$ (CCDC 2017084).


Table S6. Crystal data and structure refinement for $\mathbf{6 b}$ (CCDC 2017084).

Identification code
Empirical formula
Formula weight
Temperature
Wavelength
Crystal system
Space group
Unit cell dimensions

999
C9 H7 N O2 S
193.22

296(2) K
$0.71073 \AA$
Orthorhombic
Fdd 2

$$
\begin{array}{ll}
\mathrm{a}=21.3018(14) \AA & \alpha=90^{\circ} . \\
\mathrm{b}=43.351(3) \AA & \beta=90^{\circ} . \\
\mathrm{c}=3.9547(3) \AA & \gamma=90^{\circ} .
\end{array}
$$

| Volume | $3652.0(4) \AA^{3}$ |
| :--- | :--- |
| Z | 16 |
| Density (calculated) | $1.406 \mathrm{Mg} / \mathrm{m}^{3}$ |
| Absorption coefficient | $0.317 \mathrm{~mm}^{-1}$ |
| $\mathrm{~F}(000)$ | 1600 |
| Crystal size | $0.300 \times 0.250 \times 0.200 \mathrm{~mm}^{3}$ |
| Theta range for data collection | 3.407 to $28.310^{\circ}$. |
| Index ranges | $-28<=\mathrm{h}<=28,-56<=\mathrm{k}<=56,-5<=1<=5$ |
| Reflections collected | 11871 |
| Independent reflections | $2267[\mathrm{R}($ int $)=0.0603]$ |
| Completeness to theta $=25.242^{\circ}$ | $99.4 \%$ |
| Absorption correction | $\mathrm{Semi}-\mathrm{empirical}$ from equivalents |
| Max. and min. transmission | 0.7457 and 0.5387 |
| Refinement method | $\mathrm{Full-matrix} \mathrm{least-squares} \mathrm{on} \mathrm{F}^{2}$ |
| Data / restraints / parameters | $2267 / 1 / 119$ |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.119 |
| Final R indices [I>2sigma(I)] | $\mathrm{R} 1=0.0433, \mathrm{wR} 2=0.0983$ |
| R indices (all data) | $\mathrm{R} 1=0.0560, \mathrm{wR} 2=0.1076$ |
| Absolute structure parameter | $0.04(4)$ |
| Extinction coefficient | $\mathrm{n} / \mathrm{a}$ |
| Largest diff. peak and hole | 0.219 and $-0.277 \mathrm{e} . \AA^{-3}$ |

Crystal Structure of Compound $\mathbf{8 p}$ : ORTEP representation ( $40 \%$ ellipsoids probability) of $\mathbf{8 p}$ (CCDC 2017085).


Table S7. Crystal data and structure refinement for $\mathbf{8 p}$ (CCDC 2017085).

| Identification code | 1013 |
| :---: | :---: |
| Empirical formula | C15 H16 O7 |
| Formula weight | 308.28 |
| Temperature | 293(2) K |
| Wavelength | 0.71073 Å |
| Crystal system | Triclinic |
| Space group | P-1 |
| Unit cell dimensions | $\mathrm{a}=4.03230(10) \AA \quad \alpha=94.501(2)^{\circ}$. |
|  | $\mathrm{b}=12.1406(2) \AA \quad \beta=91.3350(10)^{\circ}$. |
|  | $\mathrm{c}=15.7776(2) \AA \quad \gamma=93.6930(10)^{\circ}$. |
| Volume | $768.09(2) \AA^{3}$ |
| Z | 2 |
| Density (calculated) | $1.333 \mathrm{Mg} / \mathrm{m}^{3}$ |
| Absorption coefficient | $0.107 \mathrm{~mm}^{-1}$ |
| F(000) | 324 |
| Crystal size | $0.300 \times 0.250 \times 0.200 \mathrm{~mm}^{3}$ |
| Theta range for data collection | 2.976 to $24.997^{\circ}$. |
| Index ranges | $-4<=\mathrm{h}<=4,-14<=\mathrm{k}<=14,-18<=1<=18$ |
| Reflections collected | 23407 |
| Independent reflections | $2695[\mathrm{R}(\mathrm{int})=0.0939]$ |
| Completeness to theta $=24.997^{\circ}$ | 99.8 \% |
| Absorption correction | Semi-empirical from equivalents |
| Max. and min. transmission | 0.7454 and 0.4080 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Data / restraints / parameters | 2695 / 0 / 202 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.047 |
| Final R indices [ $\mathrm{I}>2 \operatorname{sigma}(\mathrm{I}$ ] $]$ | $\mathrm{R} 1=0.0599, \mathrm{wR} 2=0.1440$ |
| R indices (all data) | $\mathrm{R} 1=0.0911, \mathrm{wR} 2=0.1702$ |
| Extinction coefficient | n/a |
| Largest diff. peak and hole | 0.192 and -0.207e..$^{-}{ }^{-3}$ |

## Analytical Data:

Compounds 5a, ${ }^{\text {S1 }}(\mathbf{5 b}, \mathbf{g}),{ }^{\mathrm{S} 2} \mathbf{6 a},{ }^{\mathrm{S}}(\mathbf{7 b}$, and $\mathbf{8 a}, \mathbf{n}){ }^{54}$ are known in literature and thus only ${ }^{1} \mathrm{H}$ NMR data of these compounds are provided.

## References:

(S1) Oger, N.; Grognec, E. L.; Felpin, F.-X. J. Org. Chem. 2014, 79, 8255.
(S2) Padala, K.; Pimparkar, S.; Madasamy, P.; Jeganmohan, M. Chem. Commun. 2012, 48, 7140.
(S3) Ueyama, T.; Mochida, S.; Fukutani, T.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2011, 13, 706.
(S4) Mandal, A.; Mehta, G.; Dana, S.; Baidya, M. Org. Lett. 2019, 21, 5879.

## NMR Spectra of Synthesized Compounds




$\operatorname{ln\mathrm {CDCl}_{3}:400\mathrm {MHz}}$



| 00 | 190 | 180 | 170 | 160 |  | 140 |  | 120 | 110 |  |  |  |  |  |  |  |  |  |  |  |
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| 00 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 |  |  | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |


In $\mathrm{CDCl}_{3}: 400 \mathrm{MHz}$
In $\mathrm{CDCl}_{3}: 100 \mathrm{MHz}$







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In $\mathrm{CDCl}_{3}: 400 \mathrm{MHz}$




In $\mathrm{CDCl}_{3}: 100 \mathrm{MHz}$







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| 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |

(ln CDCl $3: 400 \mathrm{MHz}$

$\frac{\ln \mathrm{CDCl}_{3}: 100 \mathrm{MHz}}{\text { 4a }}$





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In $\mathrm{CDCl}_{3}: 400 \mathrm{MHz}$







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In $\mathrm{CDCl}_{3}: 400 \mathrm{MHz}$



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In $\mathrm{CDCl}_{3}: 100 \mathrm{MHz}$


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| 00 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | ${ }^{100}$ |  | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |



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In $\mathrm{CDCl}_{3}: 400 \mathrm{MHz}$


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| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 00 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |






[^0]




|  |  |  | $\mathrm{AR}$ |  | 范 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ｜｜ | ｜Vにて｜ | $11 \backslash$ | $\checkmark$ | V＇ | ｜ | V／ |



##  

In CDCl $: 400 \mathrm{MHz}$




|  |  |  |  | 160 | 15 | 14 | 13 |  | 1 |  |  | 1 | 7 | 1 |  | 1 | 1 | 1 | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| :00 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | ${ }^{100}$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |







## 






[^1]






[^2]

```
氃郱
##
```







[^0]:    

[^1]:    

[^2]:    

