Supporting Information

Regioselective C-H Alkenylation and Unsymmetrical Bis-Olefination of Heteroarene Carboxylic Acids with Ruthenium Catalysis in Water

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Mechanistic Studies:

1) H-D Exchange Study

5-Methyl-2-thiophenecarboxylic acid **1i** (0.2 mmol), phenyl vinyl sulfone **2a** (1.1 equiv), [Ru(*p*-cymene)Cl₂]₂ (5 mol %), and Cu(OAc)₂·H₂O (1 equiv) were added in an oven-dried screw cap reaction tube with a magnetic stir bar under open-air. Then, D₂O (1 mL) was added with a syringe. The reaction tube was capped and the resulting mixture was heated at 100 °C (in oil-bath) for 35 minutes. After that, it was allowed to cool at room temperature and then quenched with AcOH and diluted with NH₄Cl solution. The mixture was extracted with ethyl acetate (10 mL, two times). The organic layer was dried over anhydrous sodium sulfate and concentrated under reduced pressure. The resulting deuterium incorporated 5-methyl-2-thiophenecarboxylic acid and the olefinated product were purified by silica gel column chromatography. The H/D exchange result was determined by ¹H NMR spectroscopy.

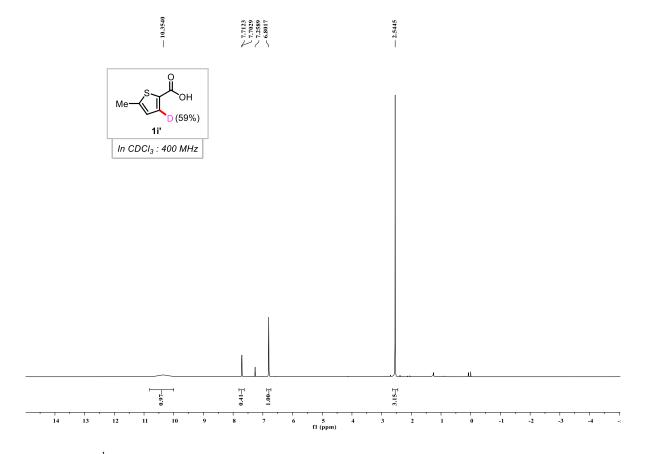


Figure S1. ¹H NMR spectrum of deuterium-incorporated 5-methyl-2-thiophenecarboxylic acid 1i'.

2) Radical Scavenger Experiment

2-Thiophenecarboxylic acid **1a** (0.2 mmol) and phenyl vinyl sulfone **2a** (1.1 equiv), [Ru(p-cymene)Cl₂]₂ (5 mol %), Cu(OAc)₂·H₂O (1 equiv), and corresponding radical scavenger (3 equiv) were added in an oven-dried screw cap reaction tube with a magnetic stir bar under open-air. Then, H₂O (1 mL) was added with a syringe. The reaction tube was capped and the resulting mixture was heated at 100 °C (in oil-bath) for 24 h. After that, it was allowed to cool at room temperature and then quenched with AcOH and diluted with NH₄Cl solution. The mixture was extracted with ethyl acetate (10 mL, two times). The organic layer was dried over anhydrous sodium sulfate and concentrated under reduced pressure. The crude product was treated with K₂CO₃ (2 equiv), MeI (3 equiv) in MeCN (1 mL) at room temperature for 4 h and then concentrated under reduced pressure. The resulting residue was purified by silica gel column chromatography.

3) Kinetic Isotope Effect Study via Parallel Experiment

$$\begin{array}{c} \text{Me} & \begin{array}{c} \text{CO}_2\text{H} \\ \text{1i} \end{array} \begin{array}{c} \text{+} \begin{array}{c} \text{SO}_2\text{Ph} \\ \text{2a} \end{array} \end{array} \begin{array}{c} \begin{array}{c} \text{[Ru}(p\text{-cymene)Cl}_2]_2 \text{ (5 mol \%)} \\ \text{Cu}(\text{OAc)}_2 \cdot \text{H}_2\text{O} \text{ (1 equiv)} \end{array} \end{array} \\ \begin{array}{c} \text{H}_2\text{O}, \ 100 \, ^{\circ}\text{C}, \ 10\text{-}30 \, \text{min}} \\ \text{under aerial condition} \end{array} \\ \begin{array}{c} \text{[Ru}(p\text{-cymene)Cl}_2]_2 \text{ (5 mol \%)} \\ \text{Cu}(\text{OAc)}_2 \cdot \text{H}_2\text{O} \text{ (1 equiv)} \end{array} \\ \begin{array}{c} \text{Me} \\ \text{(27\%) D} \end{array} \begin{array}{c} \text{CO}_2\text{H} \\ \text{Me} \\ \text{(27\%) D} \end{array} \end{array} \\ \begin{array}{c} \text{[Ru}(p\text{-cymene)Cl}_2]_2 \text{ (5 mol \%)} \\ \text{Cu}(\text{OAc)}_2 \cdot \text{H}_2\text{O} \text{ (1 equiv)} \end{array} \\ \begin{array}{c} \text{H}_2\text{O}, \ 100 \, ^{\circ}\text{C}, \ 15\text{-}40 \, \text{min}} \\ \text{under aerial condition} \end{array} \\ \begin{array}{c} \text{KIE} = k_H/k_D = 1.68 \end{array}$$

5-Methyl-2-thiophenecarboxylic acid **1i** and deuterated-5-methyl-2-thiophenecarboxylic acid **1i"** were independently reacted with **2a** for five different time intervals (10-40 minutes) under the standard reaction conditions [**1i** (0.2 mmol) or **1i"** (0.2 mmol), phenyl vinyl sulfone **2a** (1.1 equiv), [Ru(*p*-cymene)Cl₂]₂ (5 mol %), and Cu(OAc)₂·H₂O (1 equiv), H₂O (1 mL)]. The product distributions were analyzed from the worked-up crude reaction mixture by ¹H NMR spectroscopy using dibromomethane as an internal standard.

time (min)	10 min	15 min	20 min	25 min	30 min	35 min	40 min
$1i \rightarrow Product (\%)$	27.1	33.2	36.3	41.2	43.2	-	-
1i" → Product (%)	-	17.4	20.9	-	24.2	26.7	30.3

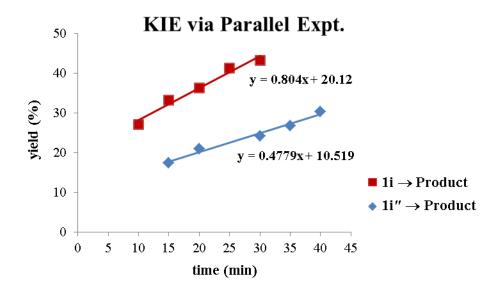


Figure S2. Time vs yield plot for KIE determination.

Crystallographic Experimental Section:

Single Crystal X-ray Crystallography: X-ray data of the crystals were collected and integrated using a Bruker Axs (Kappa Apex 2) CCD diffractometer equipped with graphite monochromatic Mo ($K\alpha$) radiation. Crystals were mounted over fine nylon loop which was attached to the copper mounting pin held on by a magnetic base. The APEX 3 and APEX 3-SAINT/Bruker SAINT programs were used for the data collection and unit-cell determination, respectively. The crystal structures were solved by direct methods using SHELXL-2014/4 or SHELXS-97 and refined by full-matrix least-squares on F^2 method using program SHELXL-2014/7 or SHELXL-2018/3.

Method of Crystallization: All the single crystals were grown in a small glass vial by slow evaporation technique from DCM/Hexane solvent system at room temperature over a period of 1-2 weeks.

Crystal Structure of Compound 3c: ORTEP representation (40% ellipsoids probability) of **3c** (CCDC 1975675).

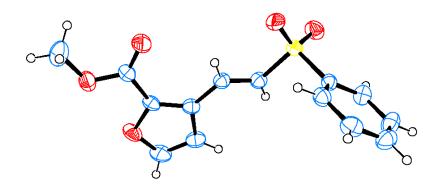


Table S1. Crystal data and structure refinement for **3c** (CCDC 1975675).

Identification co	ode	532

Empirical formula C14 H12 O5 S

Formula weight 292.30Temperature 296(2) K Wavelength 0.71073 Å

Crystal system, space group Orthorhombic, Pbca

Unit cell dimensions $a = 7.6046(2) \text{ Å } \alpha = 90^{\circ}$

 $b = 15.5125(5) \text{ Å } \beta = 90^{\circ}$ $c = 23.2195(7) \text{ Å } \gamma = 90^{\circ}$

Volume 2739.12(14) A^3

Z, Calculated density 8, 1.418 Mg/m³

Absorption coefficient 0.252 mm^-1

F(000) 1216

Crystal size 0.250 x 0.220 x 0.100 mm

Theta range for data collection 2.769 to 24.999 deg.

Limiting indices -7<=h<=9, -18<=k<=15, -27<=l<=27

Reflections collected / unique 15544 / 2406 [R(int) = 0.0244]

Completeness to theta = 25.000 99.9 %
Absorption correction None

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 2406 / 0 / 182

Goodness-of-fit on F² 1.054

Final R indices [I>2sigma(I)] R1 = 0.0352, wR2 = 0.0837 R indices (all data) R1 = 0.0445, wR2 = 0.0916

Extinction coefficient n/a

Largest diff. peak and hole 0.184 and -0.305 e.A^-3

Crystal Structure of Compound 3f: ORTEP representation (40% ellipsoids probability) of **3f** (CCDC 1975672).

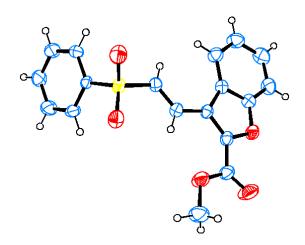


Table S2. Crystal data and structure refinement for **3f** (CCDC 1975672).

Identification code 787

Empirical formula C18 H14 O5 S

Formula weight 342.35
Temperature 296(2) K
Wavelength 0.71073 Å

Crystal system, space group Orthorhombic, Pbca

Unit cell dimensions $a = 17.7265(7) \text{ Å } \alpha = 90^{\circ}$

 $b = 11.8090(4) \text{ Å } \beta = 90^{\circ}$ $c = 15.2926(4) \text{ Å } \gamma = 90^{\circ}$

Volume 3201.23(19) A^3 Z, Calculated density 8, 1.421 Mg/m^3

Absorption coefficient 0.227 mm^-1

F(000) 1424

Crystal size 0.250 x 0.220 x 0.100 mm

Theta range for data collection 2.072 to 24.998 deg.

Limiting indices -13 <= h <= 21, -13 <= k <= 14, -18 <= l <= 12

Reflections collected / unique 10818 / 2815 [R(int) = 0.0224]

Completeness to theta = 25.000 99.9 %
Absorption correction None

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 2815 / 0 / 218

Goodness-of-fit on F² 1.053

Final R indices [I>2sigma(I)] R1 = 0.0403, wR2 = 0.0959 R indices (all data) R1 = 0.0531, wR2 = 0.1040

Extinction coefficient n/a

Largest diff. peak and hole 0.277 and -0.301 e.A^-3

Crystal Structure of Compound 3h: ORTEP representation (40% ellipsoids probability) of **3h** (CCDC 1975673).

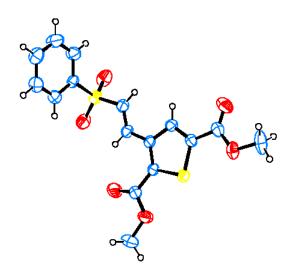


Table S3. Crystal data and structure refinement for **3h** (CCDC 1975673).

Identification code 28

Empirical formula C16 H14 O6 S2

Formula weight 366.39Temperature 296(2) K Wavelength 0.71073 Å

Crystal system, space group Orthorhombic, Pbca Unit cell dimensions $a = 8.0160(2) \text{ Å } \alpha = 90^{\circ}$

 $b = 19.6309(8) \text{ Å } \beta = 90^{\circ}$ $c = 21.6558(8) \text{ Å } \gamma = 90^{\circ}$

Volume 3407.8(2) A^3
Z, Calculated density 8, 1.428 Mg/m^3
Absorption coefficient 0.341 mm^-1

F(000) 1520

Crystal size 0.250 x 0.220 x 0.100 mm

Theta range for data collection 1.881 to 24.996 deg.

Limiting indices -9<=h<=9, -21<=k<=23, -25<=l<=19

Reflections collected / unique 7967 / 2995 [R(int) = 0.0402]

Completeness to theta = 25.000 99.9 % Absorption correction None

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 2995 / 0 / 219

Goodness-of-fit on F² 1.009

Final R indices [I>2sigma(I)] R1 = 0.0486, wR2 = 0.0978 R indices (all data) R1 = 0.0957, wR2 = 0.1207

Extinction coefficient n/a

Largest diff. peak and hole 0.243 and -0.300 e.A^-3

Crystal Structure of Compound 4g: ORTEP representation (40% ellipsoids probability) of **4g** (CCDC 1975676).

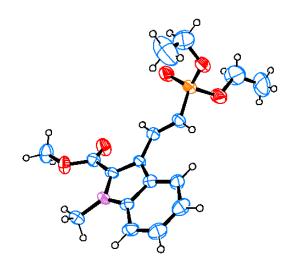


Table S4. Crystal data and structure refinement for **4g** (CCDC 1975676).

Identification code 891

Empirical formula C17 H22 N O5 P

Formula weight 351.32

Temperature 296(2) K

Wavelength 0.71073 Å

Crystal system, space group Triclinic, P-1

Unit cell dimensions $a = 7.4902(15) \text{ Å} \quad \alpha = 88.249(6)^{\circ}$

 $b = 10.012(2) \; \text{Å} \; \; \beta = 82.343(6)^\circ \\ c = 25.476(5) \; \text{Å} \; \; \gamma = 71.214(6)^\circ \\$

Volume 1792.4(6) Å^3
Z, Calculated density 4, 1.302 Mg/m^3
Absorption coefficient 0.179 mm^-1

F(000) 744

Crystal size 0.200 x 0.150 x 0.150 mm³

Theta range for data collection 0.807 to 24.997 deg.

Limiting indices -8 <= h <= 8, -11 <= k <= 11, -30 <= l <= 30

Reflections collected / unique 32648 / 32648 [R(int) = ?]

Completeness to theta = 25.000 99.9 %

Absorption correction Semi-empirical from equivalents

Max. and min. transmission 0.7452 and 0.5243

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 32648 / 39 / 454

Goodness-of-fit on F² 1.107

Final R indices [I>2sigma(I)] R1 = 0.1044, wR2 = 0.2823 R indices (all data) R1 = 0.1436, wR2 = 0.3062

Extinction coefficient 0.017(3)

Largest diff. peak and hole 0.470 and -0.495 e.Å^-3

Crystal Structure of Compound 5g: ORTEP representation (40% ellipsoids probability) of **5g** (CCDC 1975674).

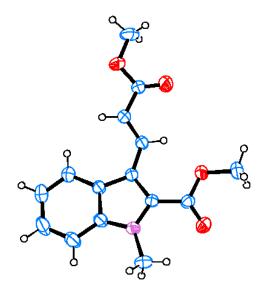


Table S5. Crystal data and structure refinement for **5g** (CCDC 1975674).

Identification code 21

Empirical formula C15 H15 N O4

Formula weight 546.56

Temperature 296(2) K

Wavelength 0.71073 Å

Crystal system, space group Triclinic, P-1

Unit cell dimensions $a = 10.9033(4) \text{ Å } \alpha = 100.1998(15)^{\circ}$

 $b=11.1118(4)~\textrm{Å}~\beta=92.2428(17)^\circ$

 $c = 13.1220(4) \text{ Å } \gamma = 118.6290(14)^{\circ}$

Volume 1359.16(8) A^3

Z, Calculated density 2, 1.336 Mg/m³

Absorption coefficient 0.098 mm^-1

F(000) 576

Crystal size 0.250 x 0.220 x 0.100 mm

Theta range for data collection 1.593 to 24.999 deg.

Limiting indices -12 <= h <= 12, -13 <= k <= 12, -15 <= l <= 15

Reflections collected / unique 20423 / 4779 [R(int) = 0.0272]

Completeness to theta = 25.000 99.6 %
Absorption correction None

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 4779 / 0 / 368

Goodness-of-fit on F² 1.026

Final R indices [I>2sigma(I)] R1 = 0.0453, wR2 = 0.1172 R indices (all data) R1 = 0.0625, wR2 = 0.1333

Extinction coefficient 0.014(2)

Largest diff. peak and hole 0.305 and -0.264 e.A^-3

Crystal Structure of Compound 6b: ORTEP representation (40% ellipsoids probability) of **6b** (CCDC 2017084).

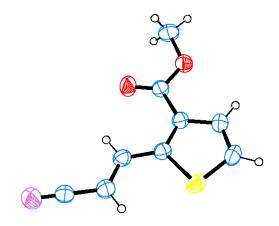


Table S6. Crystal data and structure refinement for **6b** (CCDC 2017084).

Identification code 999

Empirical formula C9 H7 N O2 S

Formula weight 193.22
Temperature 296(2) K
Wavelength 0.71073 Å

Crystal system Orthorhombic

Space group F d d 2

Unit cell dimensions a = 21.3018(14) Å $\alpha = 90^{\circ}$.

b = 43.351(3) Å $\beta = 90^{\circ}$.

c = 3.9547(3) Å $\gamma = 90^{\circ}$.

Volume $3652.0(4) \text{ Å}^3$

Z 16

Density (calculated) 1.406 Mg/m³
Absorption coefficient 0.317 mm⁻¹

F(000) 1600

Crystal size $0.300 \times 0.250 \times 0.200 \text{ mm}^3$

Theta range for data collection 3.407 to 28.310°.

Index ranges -28<=h<=28, -56<=k<=56, -5<=l<=5

Reflections collected 11871

Independent reflections 2267 [R(int) = 0.0603]

Completeness to theta = 25.242° 99.4 %

Absorption correction Semi-empirical from equivalents

Max. and min. transmission 0.7457 and 0.5387

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 2267 / 1 / 119

Goodness-of-fit on F^2 1.119

Final R indices [I>2sigma(I)] R1 = 0.0433, wR2 = 0.0983 R indices (all data) R1 = 0.0560, wR2 = 0.1076

Absolute structure parameter 0.04(4)
Extinction coefficient n/a

Largest diff. peak and hole 0.219 and -0.277 e.Å-3

Crystal Structure of Compound 8p: ORTEP representation (40% ellipsoids probability) of **8p** (CCDC 2017085).

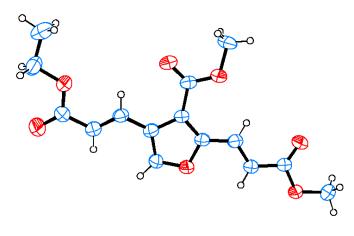


Table S7. Crystal data and structure refinement for **8p** (CCDC 2017085).

Identification code 1013

Empirical formula C15 H16 O7

Formula weight 308.28

Temperature 293(2) K

Wavelength 0.71073 Å

Crystal system Triclinic

Space group P-1

Unit cell dimensions a = 4.03230(10) Å $\alpha = 94.501(2)^{\circ}$.

b = 12.1406(2) Å $\beta = 91.3350(10)^{\circ}.$

c = 15.7776(2) Å $\gamma = 93.6930(10)^{\circ}.$

Volume 768.09(2) Å³

Z 2

Density (calculated) 1.333 Mg/m³

Absorption coefficient 0.107 mm⁻¹

F(000) 324

Crystal size $0.300 \times 0.250 \times 0.200 \text{ mm}^3$

Theta range for data collection 2.976 to 24.997°.

Index ranges -4<=h<=4, -14<=k<=14, -18<=l<=18

Reflections collected 23407

Independent reflections 2695 [R(int) = 0.0939]

Completeness to theta = 24.997° 99.8 %

Absorption correction Semi-empirical from equivalents

Max. and min. transmission 0.7454 and 0.4080

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 2695 / 0 / 202

Goodness-of-fit on F^2 1.047

Final R indices [I>2sigma(I)] R1 = 0.0599, wR2 = 0.1440 R indices (all data) R1 = 0.0911, wR2 = 0.1702

Extinction coefficient n/a

Largest diff. peak and hole 0.192 and -0.207 e.Å-3

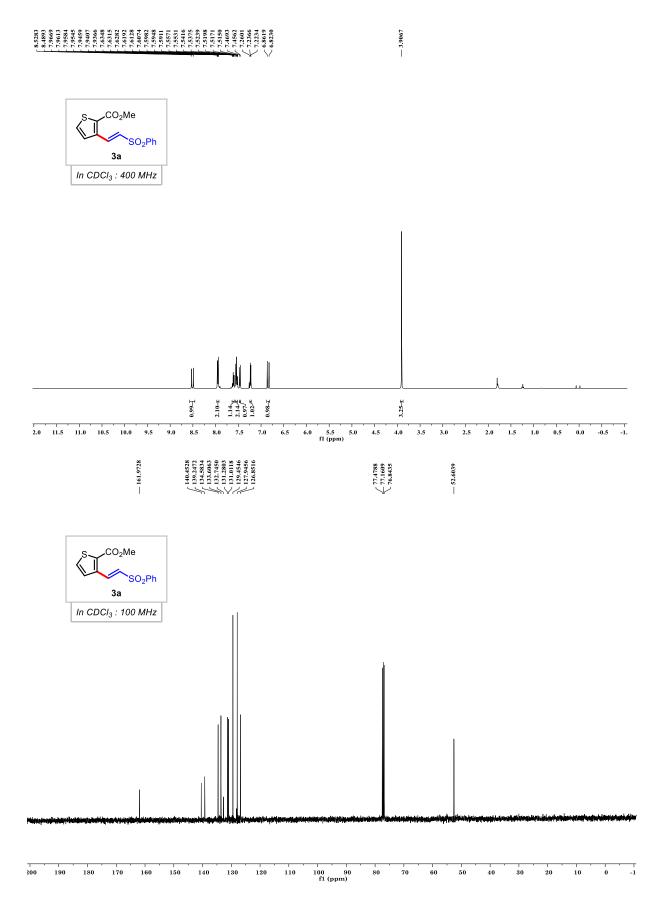
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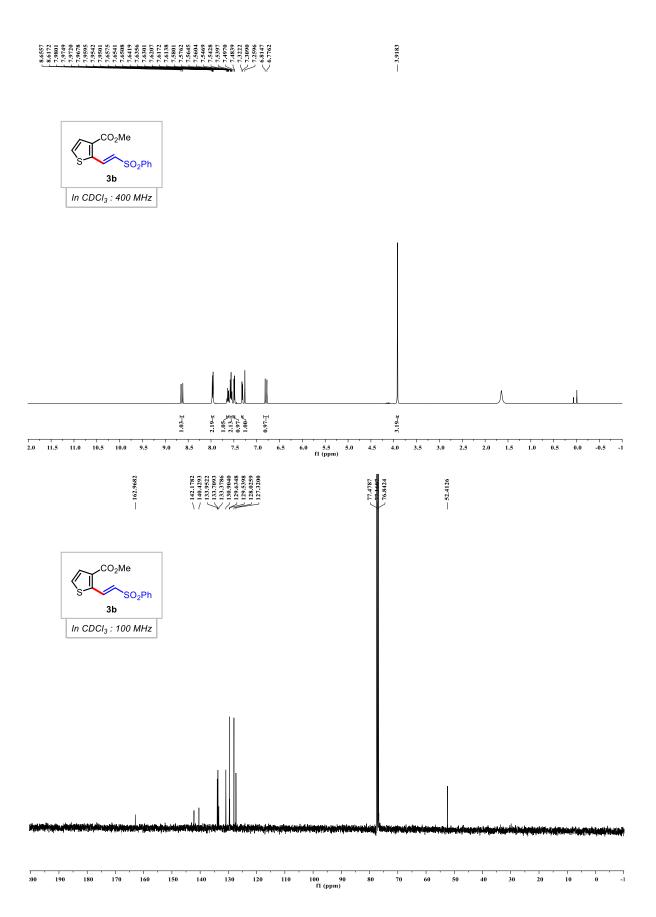
Compounds **5a**, S1 (**5b**,**g**), S2 **6a**, S3 (**7b**, and **8a**,**n**) S4 are known in literature and thus only H NMR data of these compounds are provided.

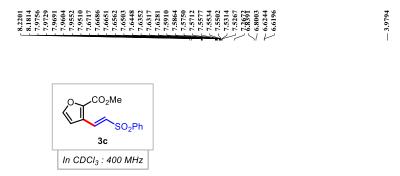
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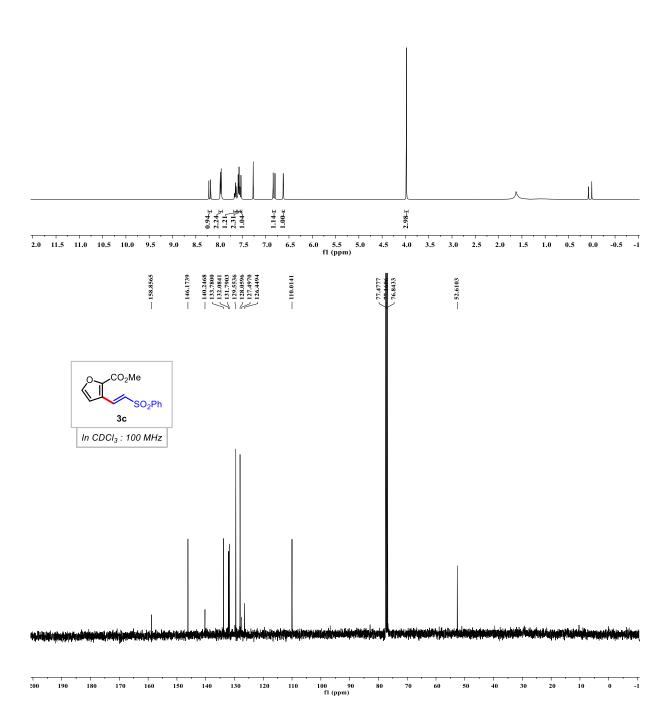
- (S1) Oger, N.; Grognec, E. L.; Felpin, F.-X. J. Org. Chem. 2014, 79, 8255.
- (S2) Padala, K.; Pimparkar, S.; Madasamy, P.; Jeganmohan, M. Chem. Commun. 2012, 48, 7140.
- (S3) Ueyama, T.; Mochida, S.; Fukutani, T.; Hirano, K.; Satoh, T.; Miura, M. Org. Lett. 2011, 13, 706.
- (S4) Mandal, A.; Mehta, G.; Dana, S.; Baidya, M. Org. Lett. 2019, 21, 5879.

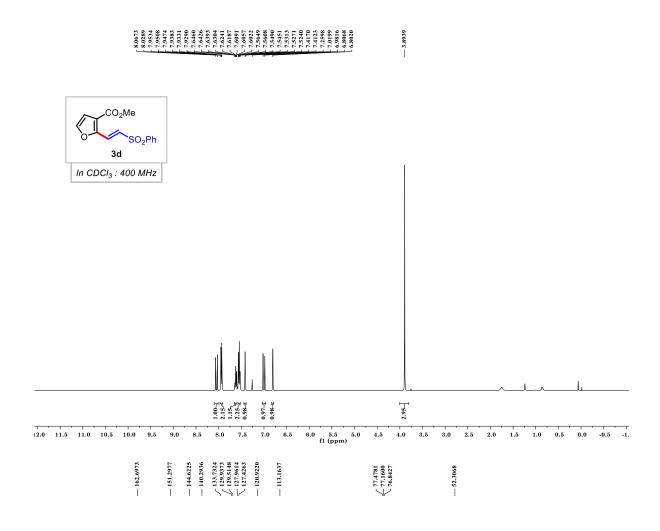
NMR Spectra of Synthesized Compounds



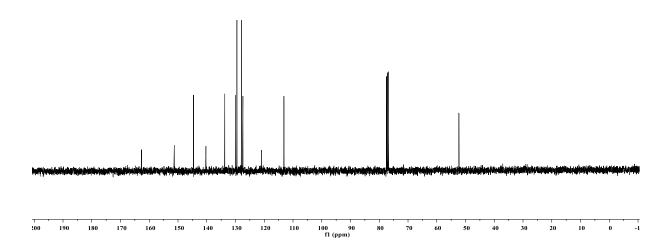


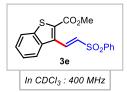


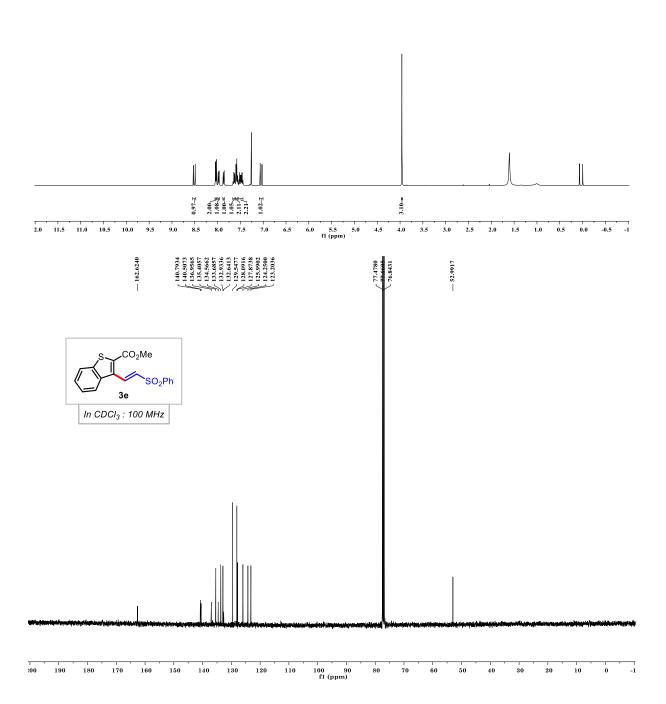




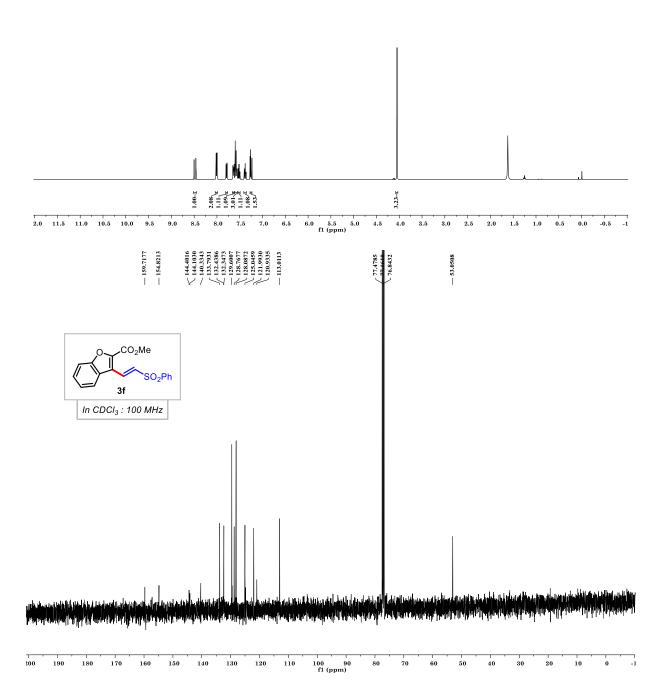




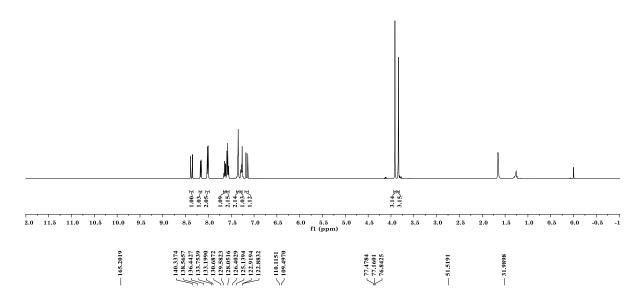


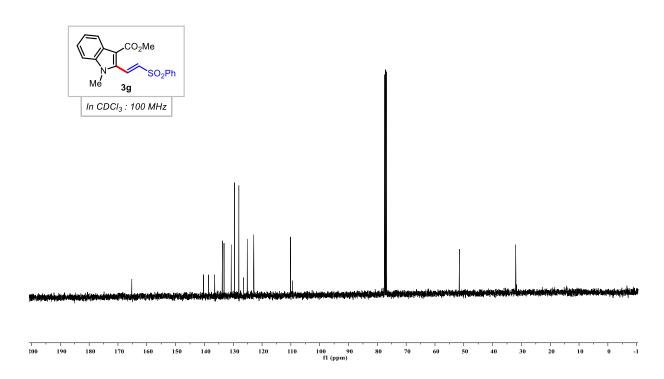




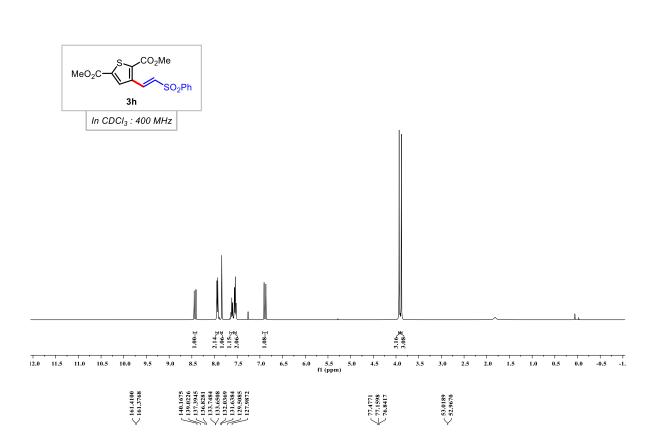


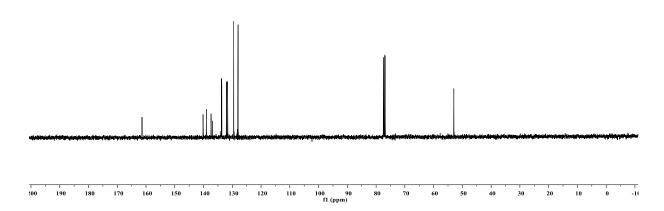






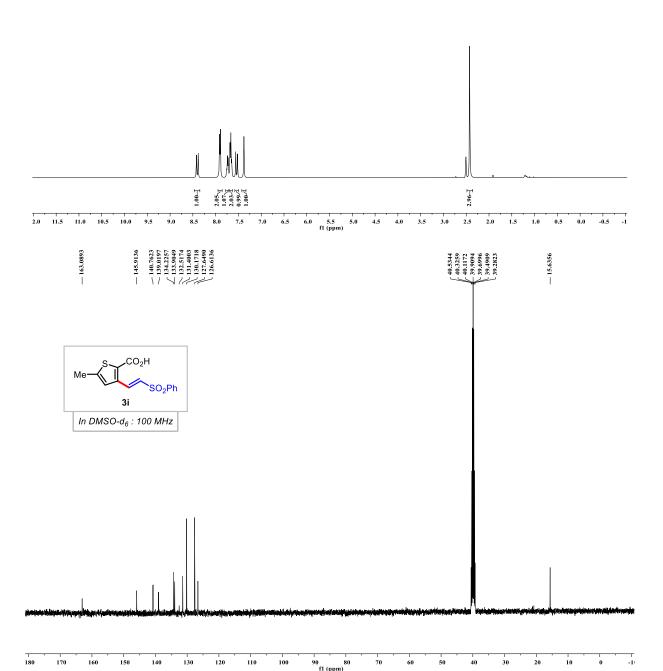




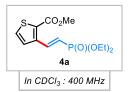


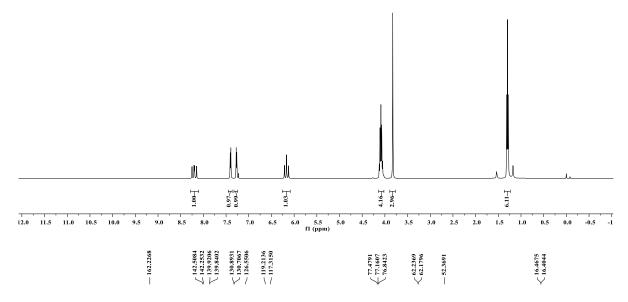




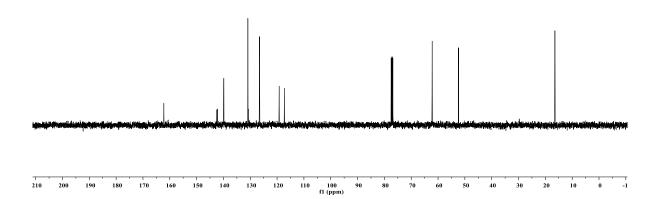




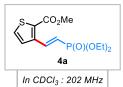


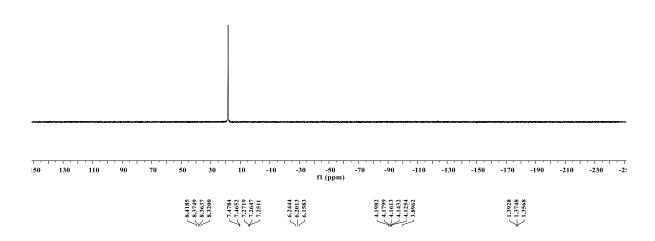




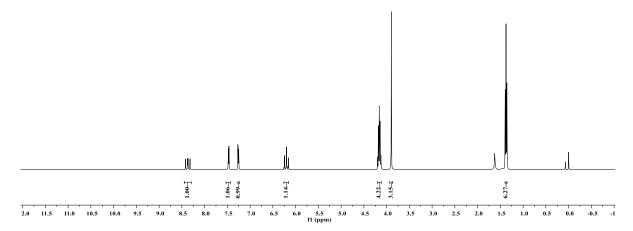


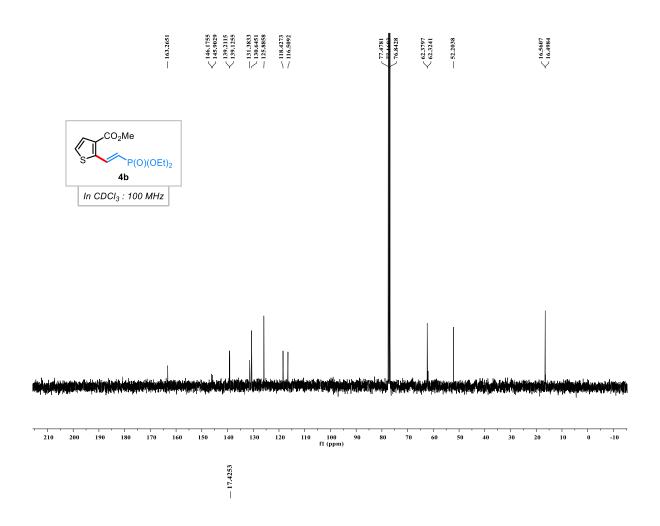


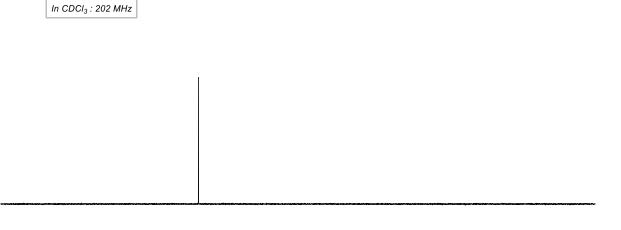












P(O)(OEt)₂

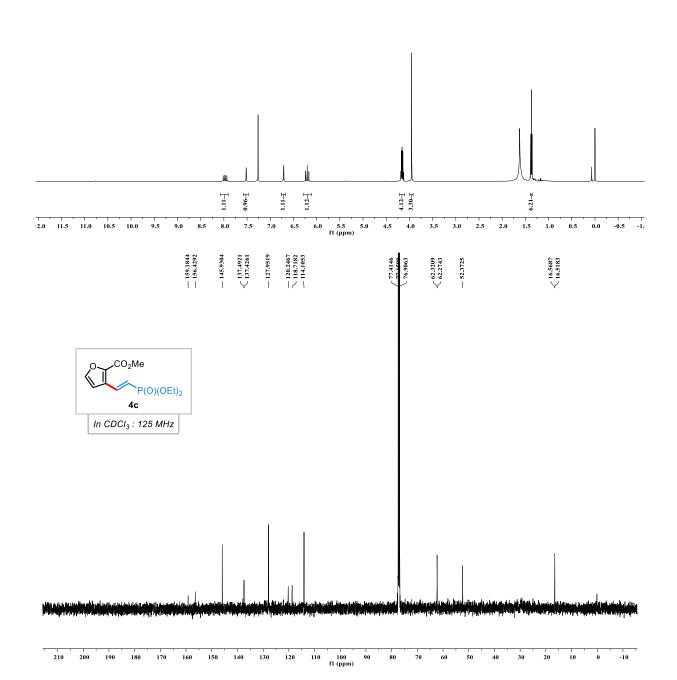
4b

10

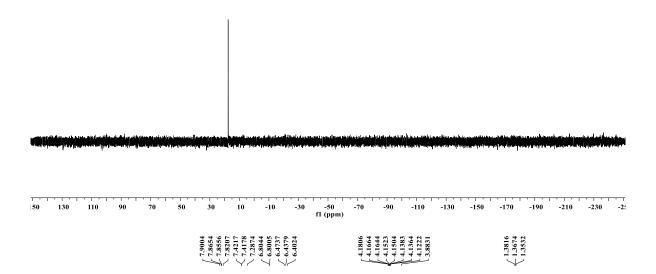
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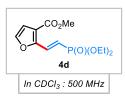


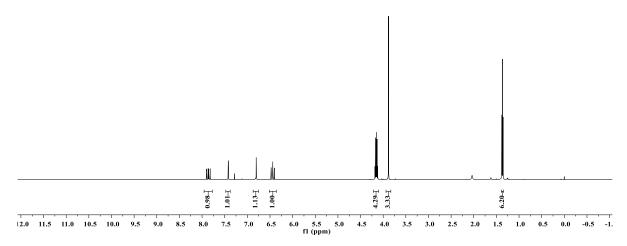




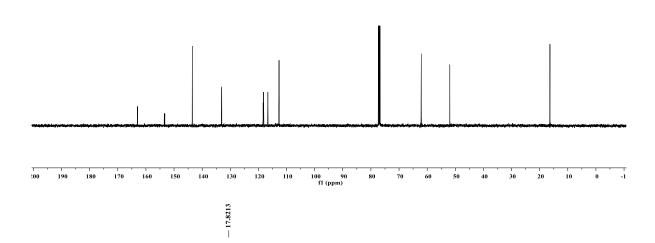


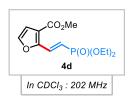


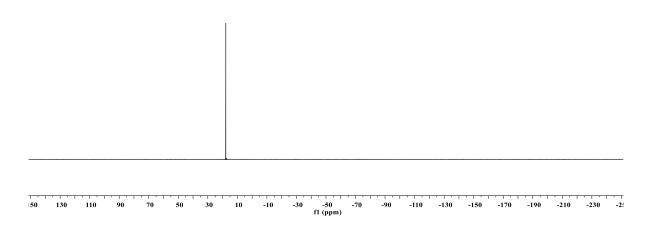




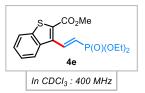


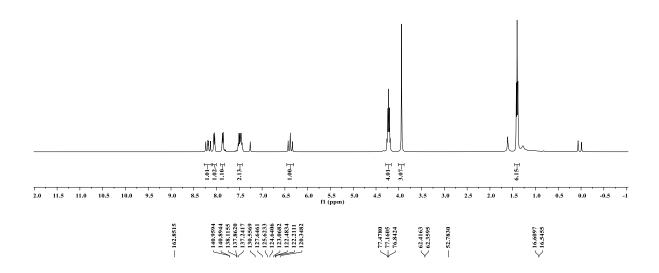


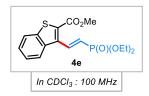


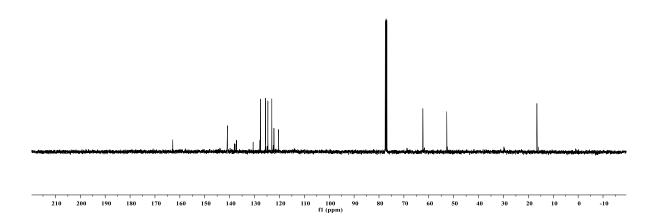










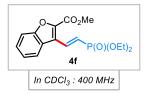


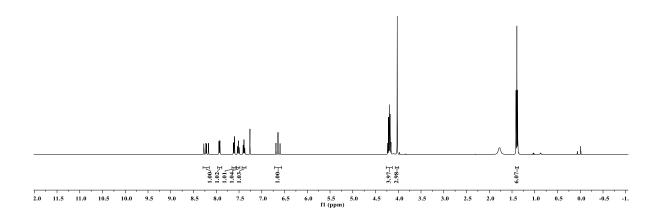


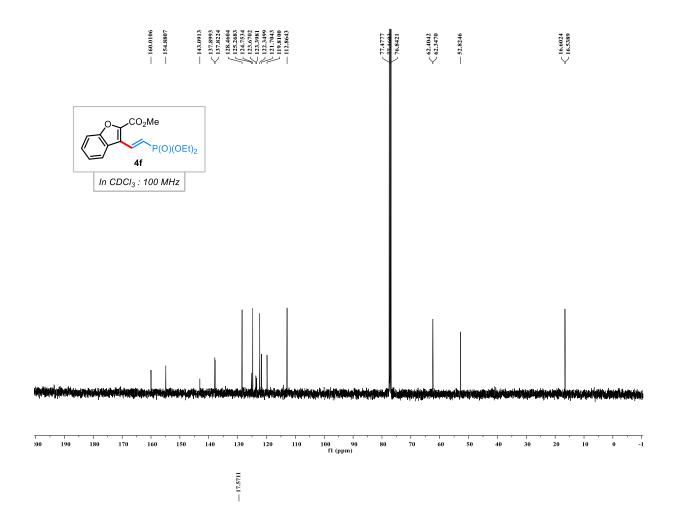
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8.2284 8.2284 8.2162 8.2162 8.2162 7.9399 7.9399 7.5316 7.534 7.5316 7.5316 7.5318 7.4918 7.4918 7.4918 7.4918 7.4918 7.4918 7.4918 7.4918 7.4918 7.4918 7.4918

4.2333 4.2150 4.1970 4.1782 4.1604 $\left\langle \frac{1.4112}{1.3935} \right\rangle$

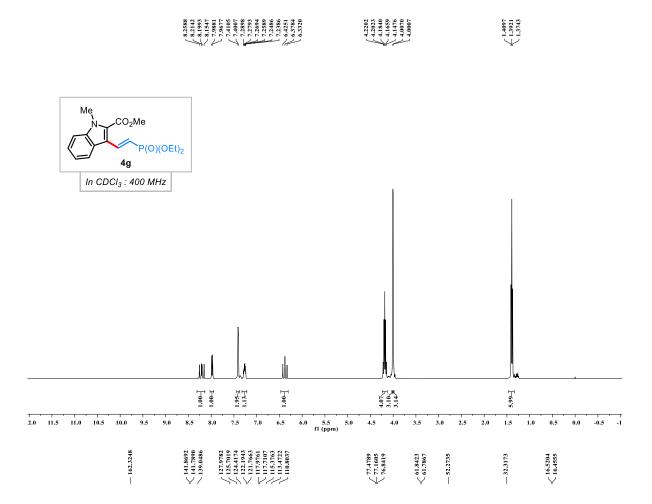


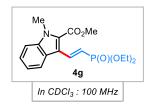


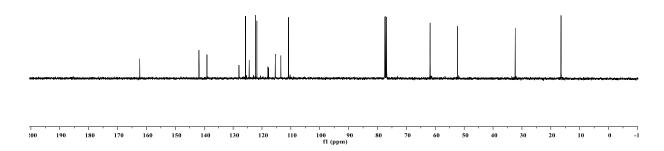




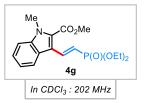
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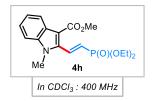


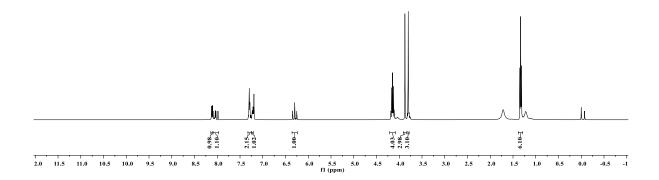


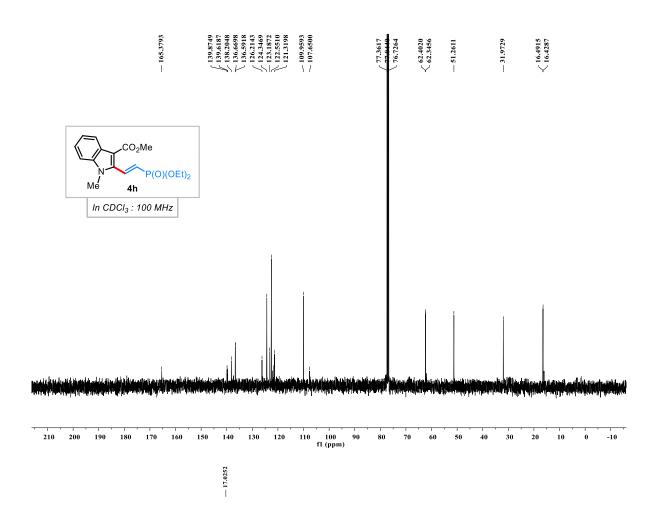


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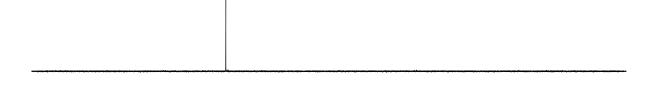






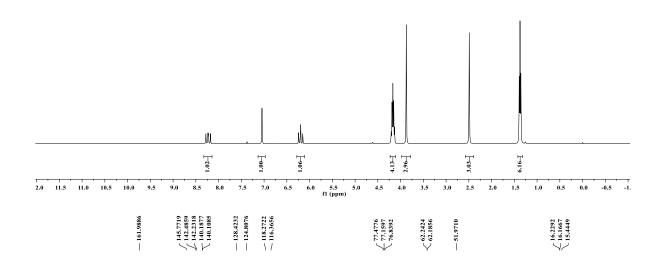


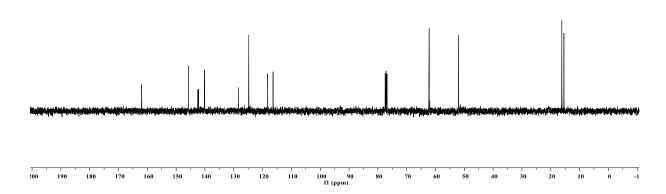




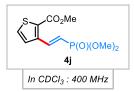
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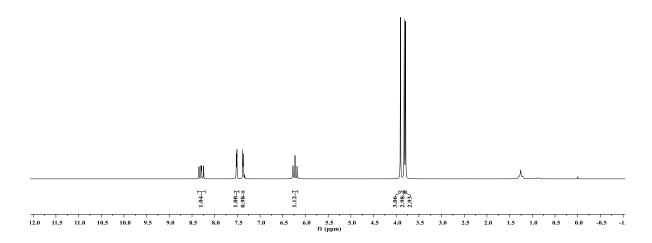




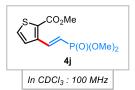


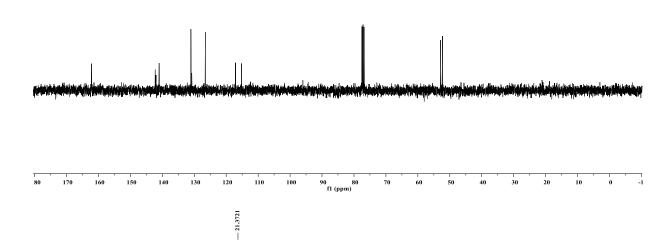




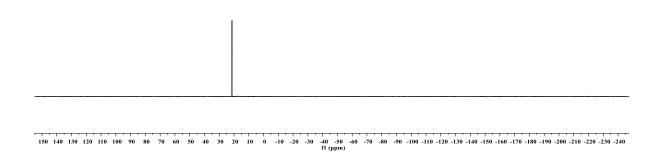






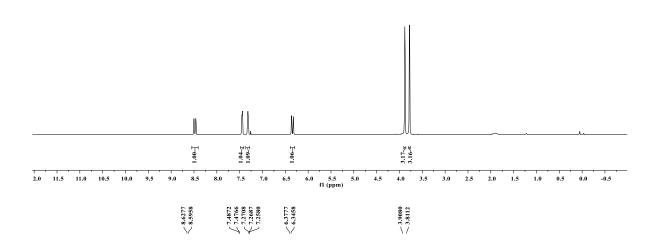




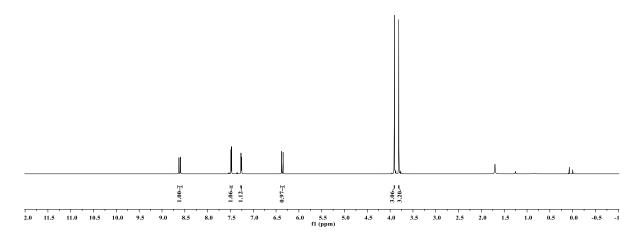






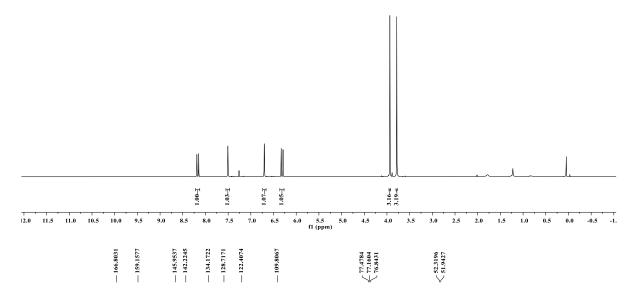




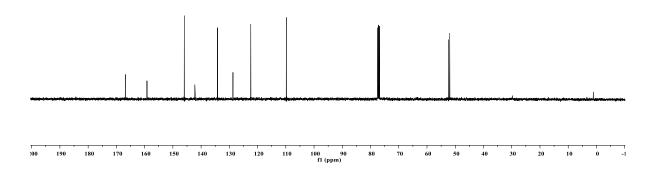




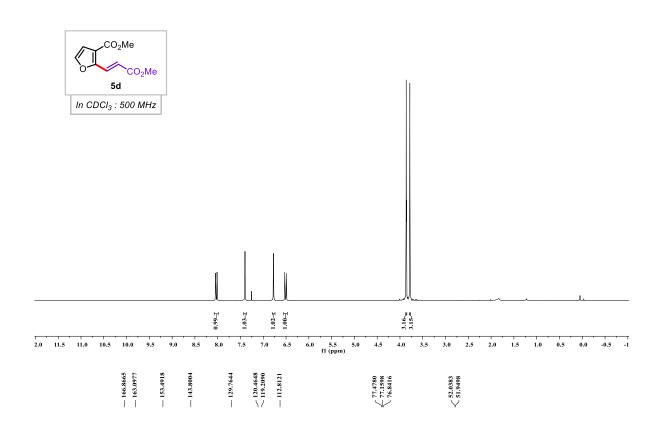




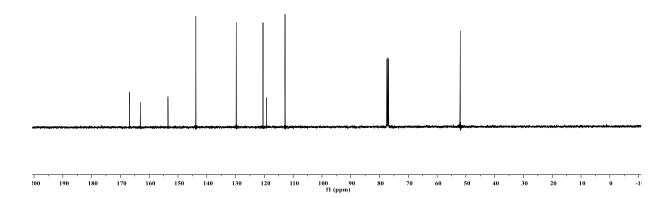




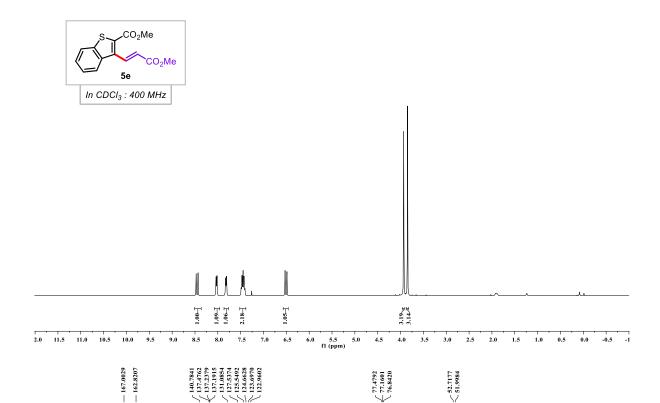


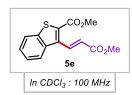


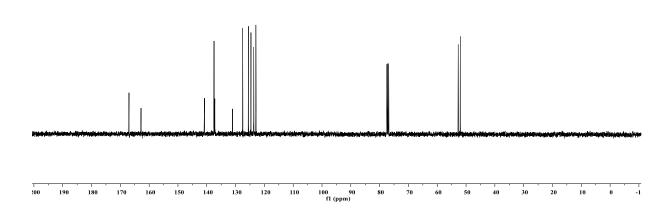




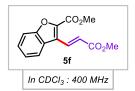


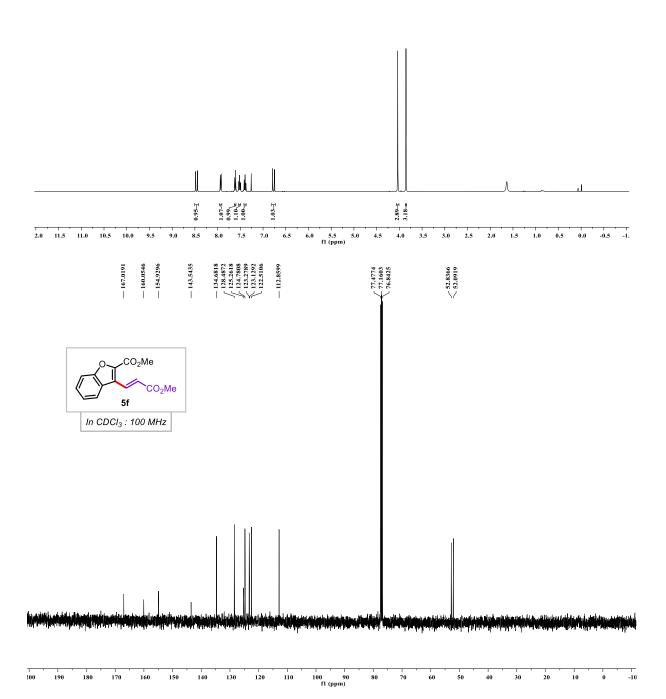


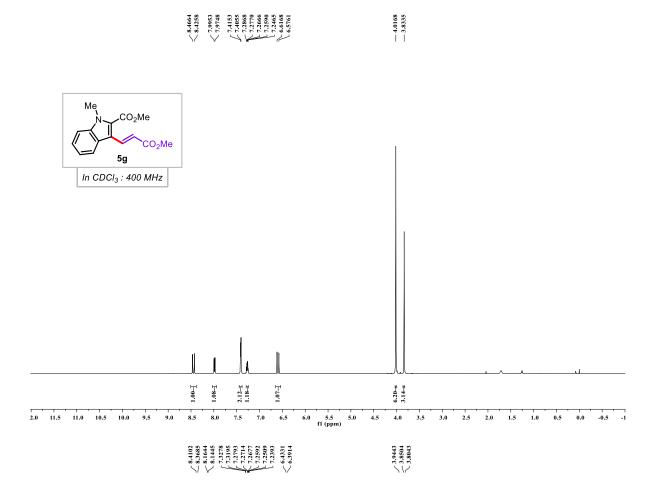




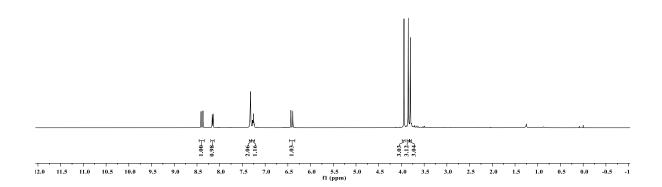




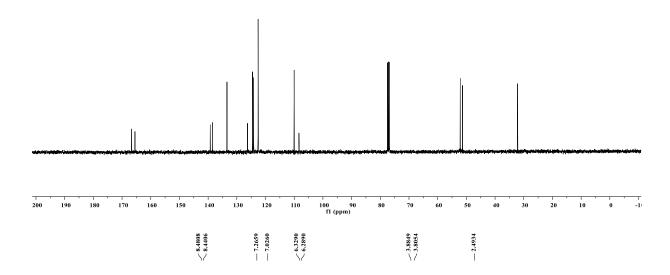




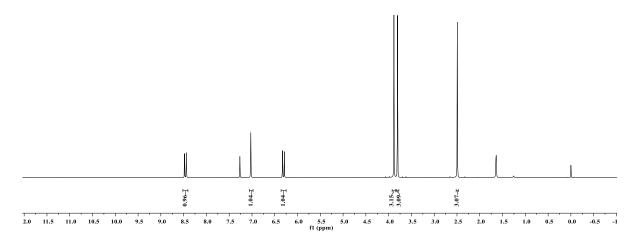


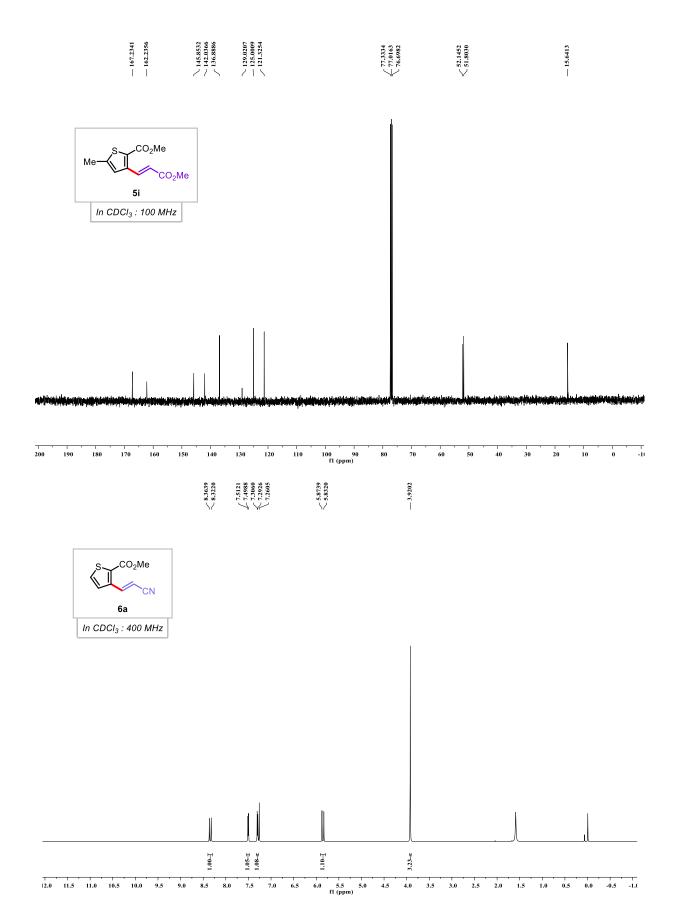






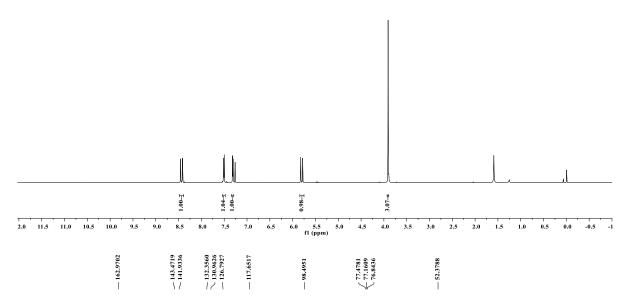


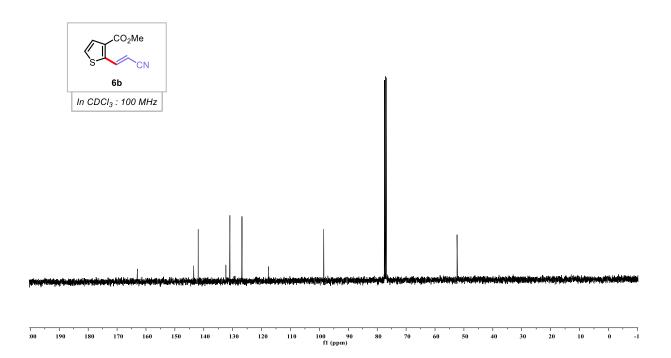






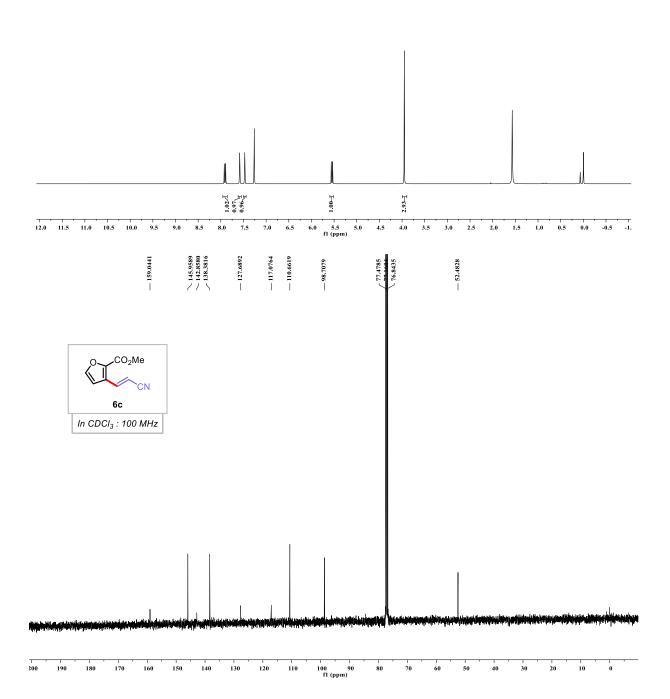




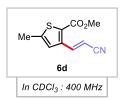


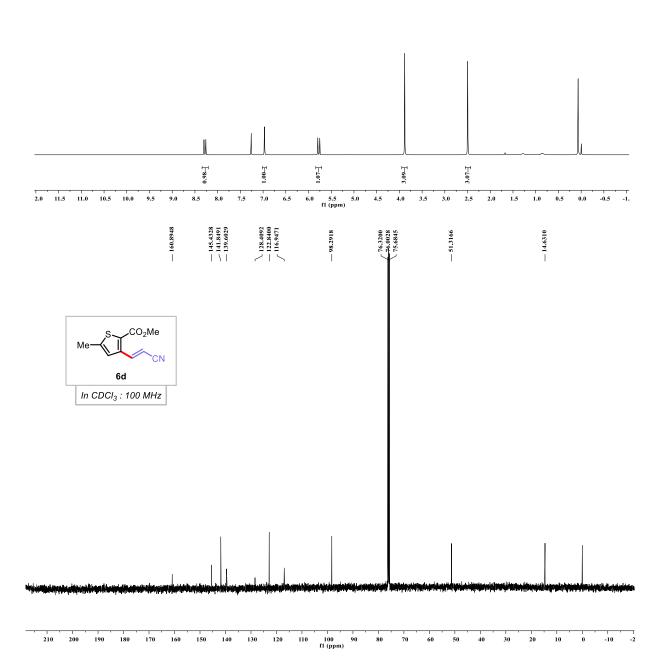






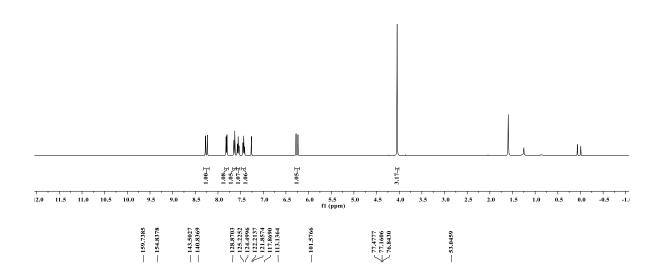


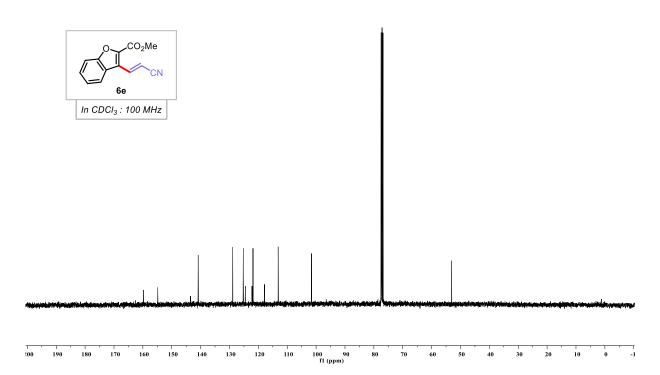


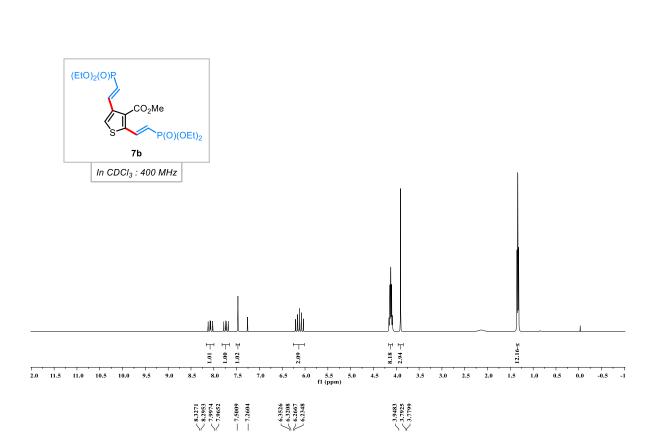




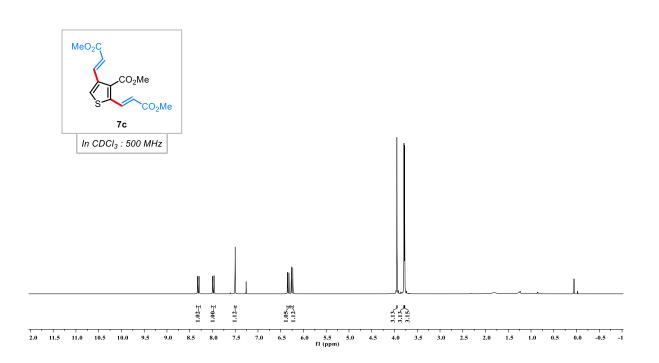


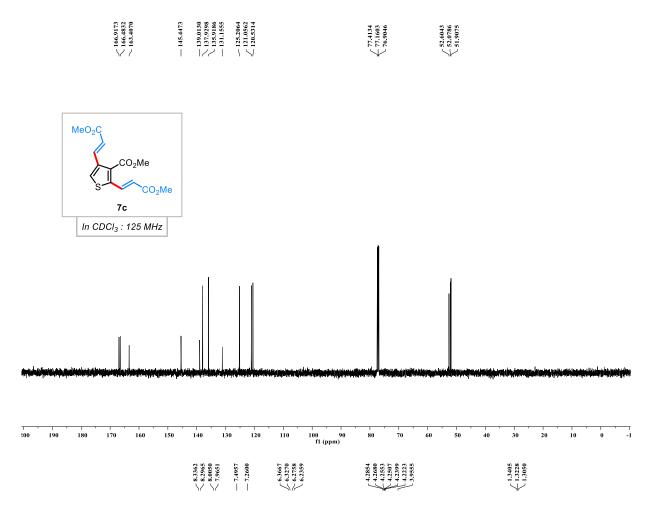




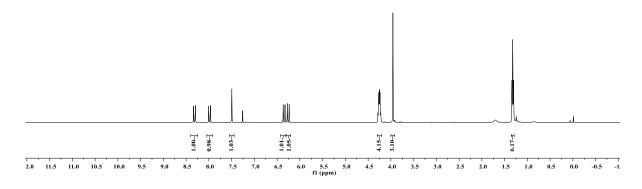


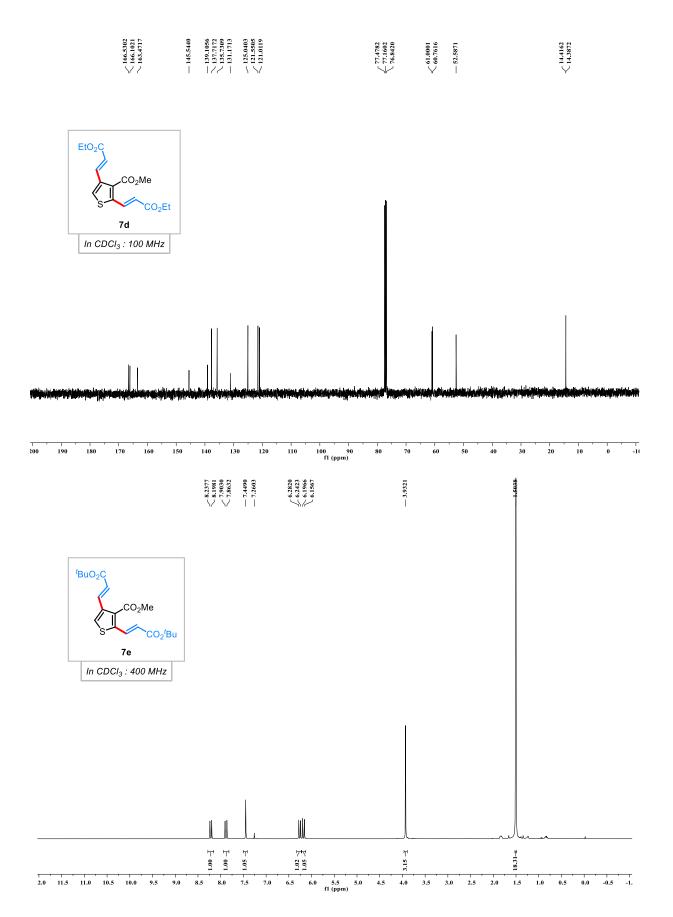
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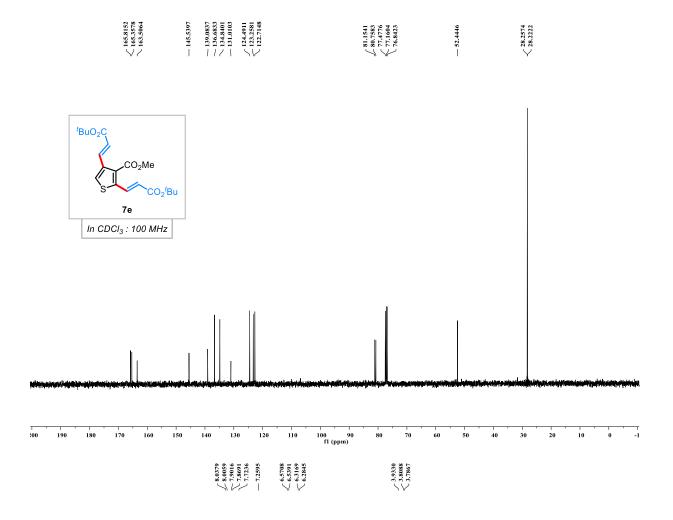


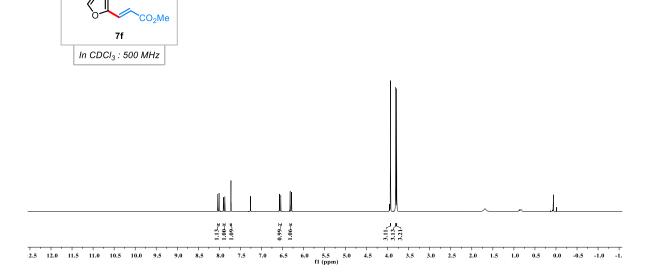


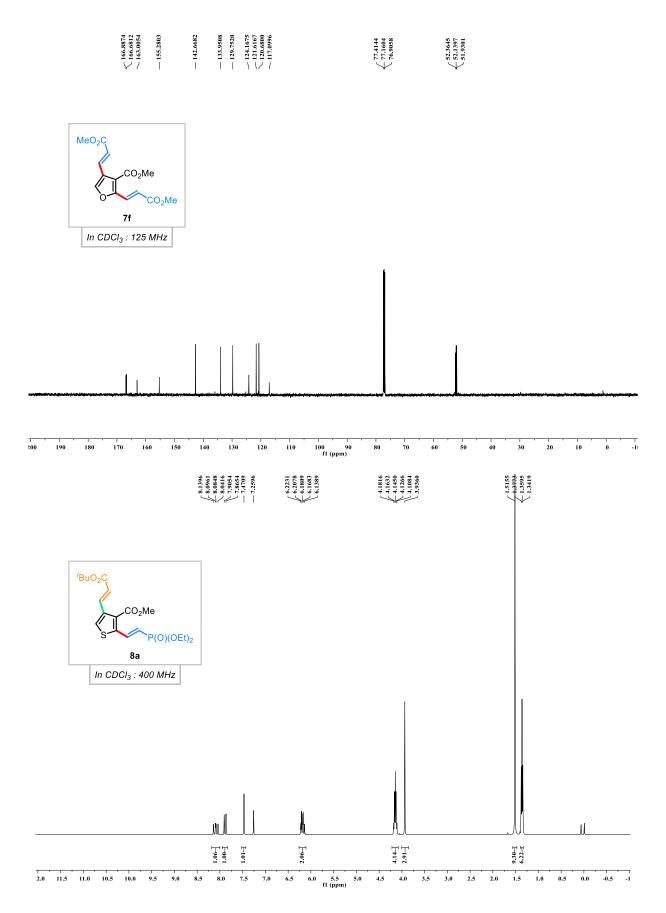






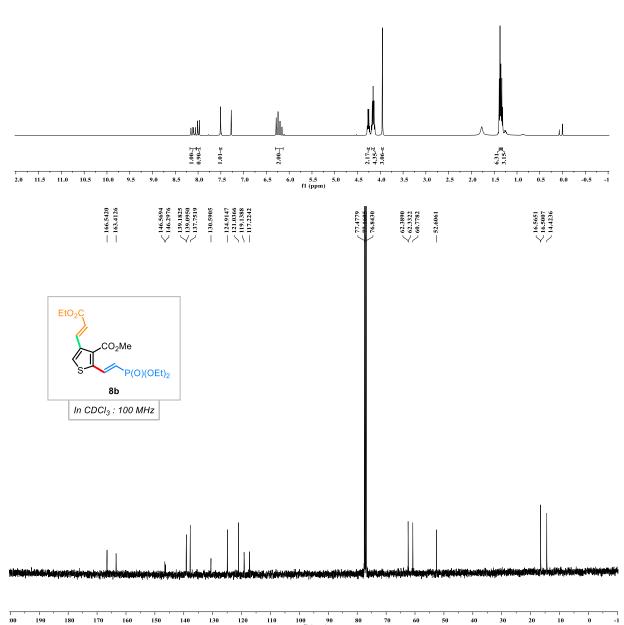










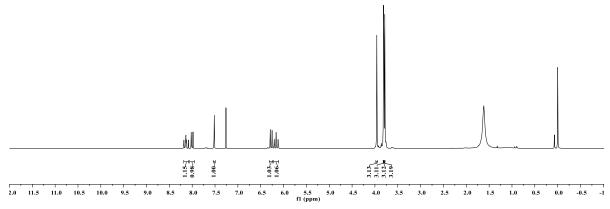


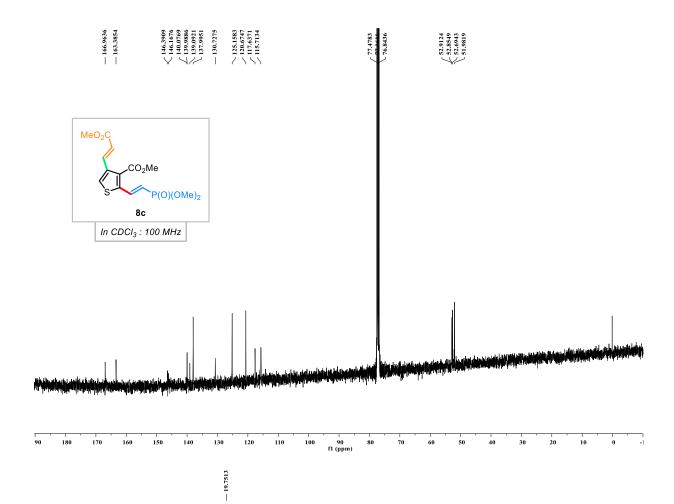


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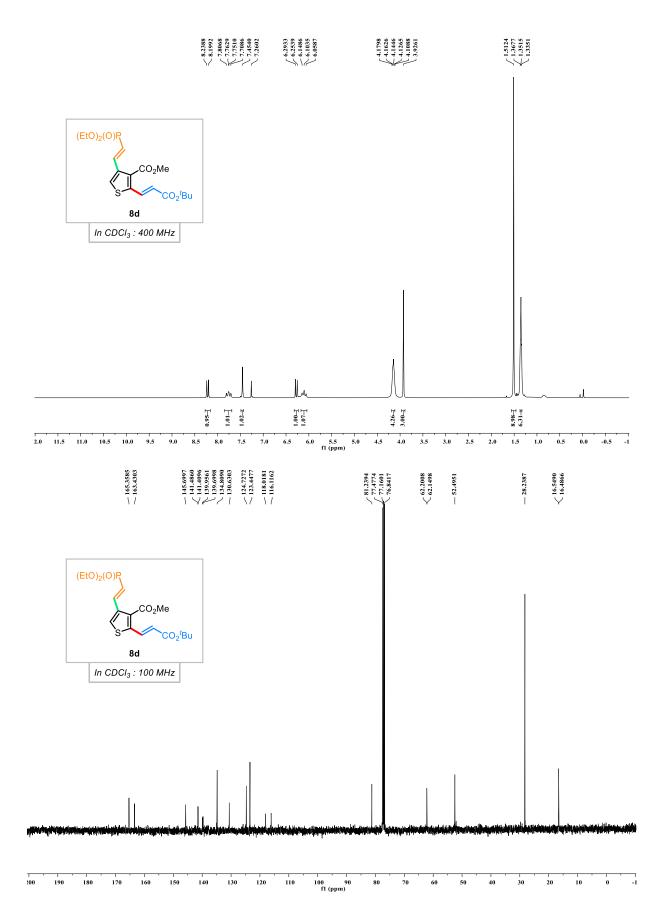
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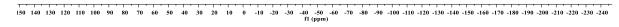




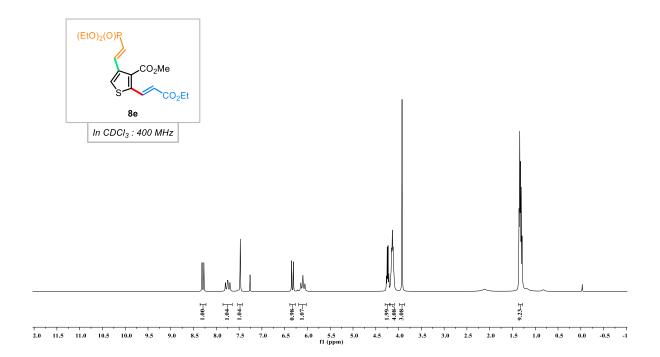


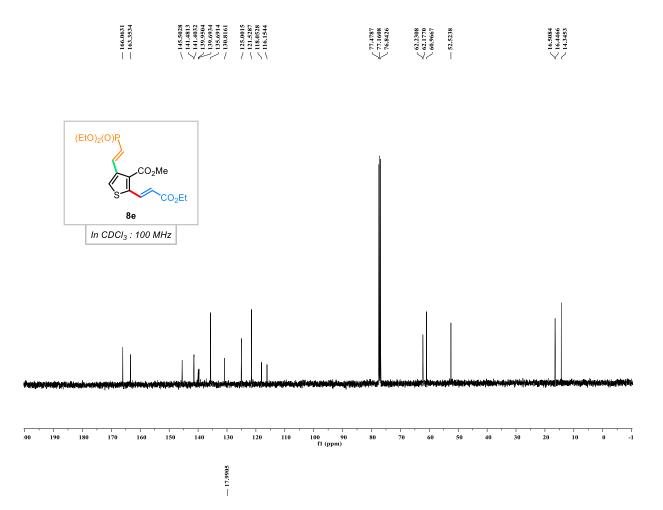


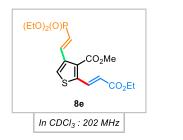


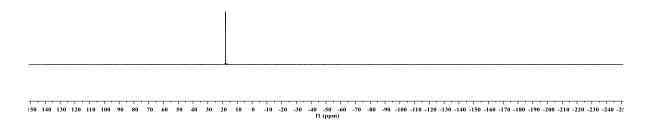






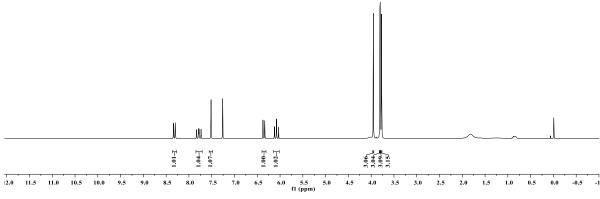


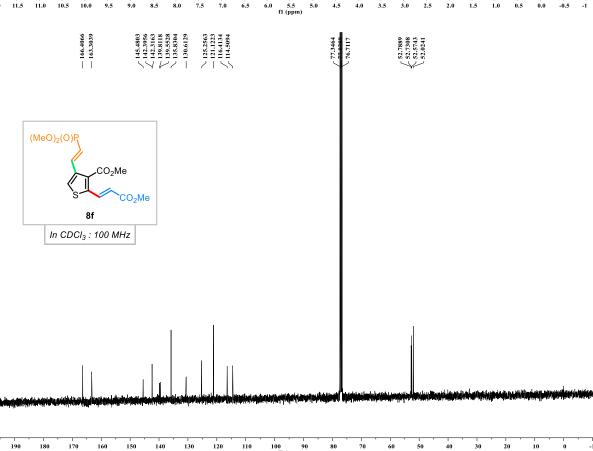




\$3992 \$2993 \$2293 \$7,7829 \$7,7769 \$7,7313 \$7,7

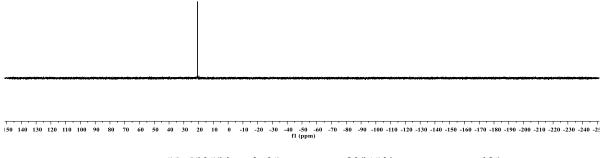












\$ 8.2934 7.7933 7.7419 7.7419 7.7419 6.35619 6.35619 6.4136 6.



