Supporting Information

Phosphorescent Tetradentate Platinum(II) Complexes Containing Fused 6/5/5 or 6/5/6 Metallocycles

Guijie Li,^{*,1} Gang Shen,¹ Xiaoli Fang,¹ Yun-Fang Yang,^{*,1} Feng Zhan,¹ Jianbing Zheng,¹ Weiwei Lou,¹ Qisheng Zhang² and Yuanbin She^{*,1}

¹College of Chemical Engineering, Zhejiang University of Technology, Hangzhou, Zhejiang 310014, P. R. China

²MOE Key Laboratory of Macromolecular Synthesis and Functionalization, Department of Polymer Science and Engineering, Zhejiang University, Hangzhou 310027, P. R. China

E-mail: guijieli@zjut.edu.cn; yangyf@zjut.edu.cn; sheyb@zjut.edu.cn

Table of Contents

1	General Information	S-3-S-4
2	Table S1. DFT calculations for Pt(II) complexes	S-5
3	Table S2. Selected bond lengths, bond angles and dihedral angles of the Pt(II) complexes based on the DFT calculation	S-6
4	Table S3. Crystal data and structure refinement for Pt(bp-6)	S-7
5	Table S4. Crystal data and structure refinement for Pt(bp-7)	S-8
6	Figure S1. Density functional theory calculations of frontier orbitals and spin densities of T_1 states for Pt(II) complexes based on optimized S_0 and T_1 geometries.	S-9
7	Figure S2: Cyclic voltammograms of the Pt(II) complexes	S-10
8	Figure S3: Absorption spectra of the Pt(II) complexes and their ligands	S-11
9	Tables S5-S8: DFT calculations of excited state energy level, wavelength andfrontier orbitals for the Pt(II) complexes	S-11-S-12
10	Figure S4. Natural transition orbitals (NTOs) of the S_1 and T_1 states for Pt(II) complexes based on optimized S_0 geometry	S-13
11	Figure S5. Luminescence spectra of the Pt(II) complexes at 77K in 2-MeTHF, at RT in DCM solution and at RT in PMMA film	S-14
12	Figure S6. Luminescence spectra comparison measured by different spectrometers.	S-14
13	Detailed synthetic procedures of the Pt(II) complexes	S-15-S-25
14	¹ H, ¹³ C NMR spectra and HRMS of the Pt(II) complexes	S-26-S-36
15	Cartesian Coordinates of the Structures	S-37-S-51
16	References	S-52

General Information.

Synthesis and Characterization. Unless noted, all commercial reagents were purchased and used as received without further purification. ¹H NMR spectra were recorded at 400 or 500 MHz, and ¹³C NMR spectra were recorded at 100 or 150 MHz NMR instruments in CDCl₃ or DMSO-*d*₆ solutions and chemical shifts were referenced to tetramethylsilane (TMS) or residual protiated solvent. If CDCl₃ was used as solvent, ¹H and ¹³C NMR spectra were recorded with TMS ($\delta = 0.00$ ppm) and CDCl₃ ($\delta = 77.00$ ppm) as internal references, respectively. If DMSO-*d*₆ was used as solvent, ¹H and ¹³C NMR spectra were recorded with TMS ($\delta = 0.00$ ppm) as internal references, respectively. If DMSO-*d*₆ ($\delta = 39.52$ ppm) as internal references, respectively. The following abbreviations (or combinations thereof) were used to explain ¹H NMR ultiplicities: s = singlet, d = doublet, t = triplet, q = quartet, p = quintet, m = multiplet, br = broad. All of the new compounds were analyzed for HRMS on a mass spectrometer using electrospray ionization in positive ion mode on ESI-QTOF mass spectrometer from Applied Biosystems.

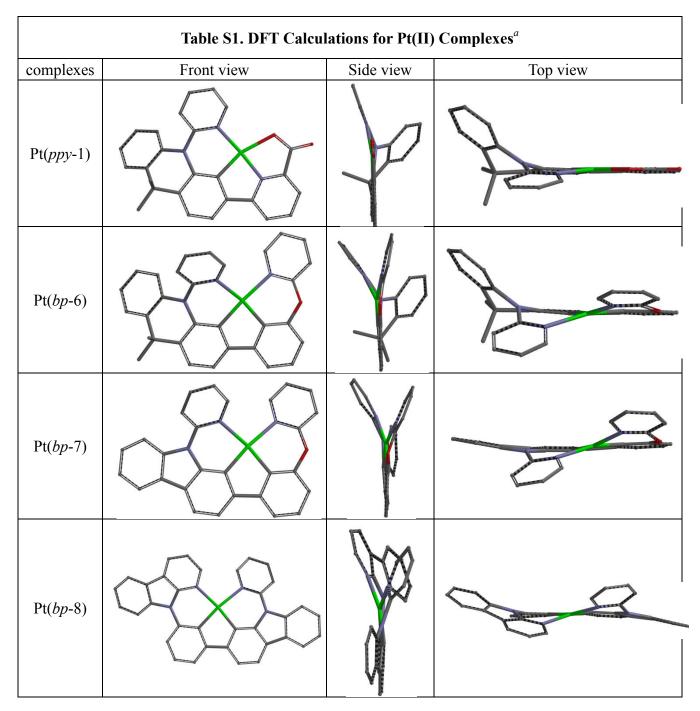
Electrochemistry. Cyclic voltammetry and different pulsed voltammetry were performed using a CH1760E electrochemical analyzeraccording previous report.¹ 0.1 M tetra-*n*-butylammonium hexafluorophosphate was used as the supporting electrolyte, anhydrous *N*,*N*-dimethylformamide, was used as the solvents for the E_{ox} and E_{red} measurements, and the solutions were bubbled with nitrogen for 15 min prior to the test. Silver wire, platinum wire and glassy carbon were used as pseudoreference electrode, counter electrode, and working electrode respectively. Scan rate was 300 mV/s. The redox potentials are based on the values measured from different pulsed voltammetry and are reported relative to an internal reference ferrocenium/ferrocene (Cp₂Fe/Cp₂Fe⁺).² The reversibility of reduction or oxidation was determined using CV.³ As defined, if the magnitudes of the peak anodic and the peak cathodic current have an equal magnitude as scan speeds of 100 mV/s or slower, then the process is considered reversible; if the magnitudes of the peak anodic and the process is considered reversible; is considered irreversible.^{2,3}

DFT Calculations. The theoretical calculations of the Pt(II) complexes were performed using Gaussian 09. The molecular geometries of ground states (S₀) were optimized with the density functional theory (DFT) method. The DFT calculations were performed using a B3LYP function with

a basis set of 6-31G(d) for C, H, O and N atoms and a LANL2DZ basis set for Pt atom.⁴ The energies of the singlet and triplet excited states of the complexes were calculated using TD-B3LYP method with a basis set of SV for H, SVP for C, O and N atoms, and a def2-TZVP basis set for Pt atom base on the optimized S₀ geometry.⁴

Photophysical Measurements. The absorption spectra were measured on an Agilent 8453 UV–VS Spectrometer. Steady state emission experiments and lifetime measurements were performed on a Horiba Jobin Yvon FluoroLog-3 spectrometer. Low temperature (77 K) emission spectra and lifetimes were measured in 2-MeTHF cooled with liquid nitrogen.

Device Fabrication and Characterization. All devices were fabricated by vacuum thermal evaporation, and were tested outside glove box after encapsulation. Prior to deposition, the prepatterned ITO coated glass substrates were cleaned by subsequent sonication in deionized water, acetone, and isopropanol. Organic layers were deposited at rates of 0.5 to 2.0 Å/s, monitored by crystal oscillator, in a custom-made vacuum thermal evaporation chamber built by LN Inc (LN-1082FS). The Al cathode was deposited through a shadow mask without breaking vacuum, defining device areas of 0.09 cm². The current-voltage-luminance characteristics were measured using a Keithley 2400 SourceMeter in conjunction with a PMTH-S1-CR131A Photodiode. Electroluminescent spectra were measured with an Ocean Optics USB2000 spectrometer.



^{*a*}Optimized S_0 were calculated using a B3LYP method with a basic set of 6-31G(d) for C, H, O and N atoms and a LANL2DZ basic set for Pt atom.

	Pt(ppy-1)		$Pt C^{1} C^{2} = $ II) complexes	
Pt(<i>ppy</i> -1)	Pt-N ¹	Pt-C	Pt-N ²	Pt-O
$Pt(ppy-1) S_0$	2.047	1.952	1.973	2.186
$Pt(ppy-1)_T_1$	2.055	1.921	1.948	2.132
complexes	Pt-N ¹	Pt-C ¹	Pt-C ²	Pt-N ²
Pt(<i>bp</i> -6)_X-ray	2.115(2)	1.981(3)	1.968(3)	2.121(3)
$Pt(bp-6)_S_0$	2.187	1.984	1.980	2.195
$Pt(bp-6)_T_1$	2.176	1.956	1.954	2.166
Pt(<i>bp</i> -7)_X-ray	2.1460(19)	1.953(2)	2.126(2)	1.973(2)
$Pt(bp-7)_S_0$	2.232	1.970	1.981	2.196
$Pt(bp-7)_T_1$	2.209	1.942	1.961	2.195
$Pt(bp-8)_S_0$	2.237	1.970	1.993	2.184
$Pt(bp-8)_T_1$	2.214	1.942	1.972	2.174

Table S2. Selected Bond Lengths (Å), Bond Angles (°) and Dihedral Angles (°) for
Tetradentate Pt(II) Complexes Based on the DFT Calculations and X-ray analysis.

Pt(<i>ppy</i> -1)	N ¹ -Pt-C	C-Pt-N ²	N ² -Pt-O	O-Pt-N ¹	N^1 -Pt- N^2	C-Pt-O	dihedral angle ^a
$Pt(ppy-1)_S_0$	93.14	82.43	79.33	104.99	173.27	161.75	16.6
$Pt(ppy-1)_T_1$	93.12	83.65	79.84	103.36	174.76	163.49	16.5
complexes	N^1 -Pt- C^1	C^1 -Pt- C^2	C^2 -Pt- N^2	N^2 -Pt- N^1	N^1 -Pt- C^2	C^1 -Pt- N^2	dihedral angle ^a
Pt(<i>bp</i> -6)_ X-ray	90.24(10)	82.17(12)	89.65(12)	99.86(9)	164.93(10)	167.23(10)	55.2
$Pt(bp-6)_S_0$	89.04	82.25	89.69	100.63	164.85	168.12	48.9
$Pt(bp-6)_T_1$	89.22	82.93	89.61	99.80	165.86	168.50	46.9
Pt(<i>bp</i> -7)_X-ray	88.56(9)	81.43(10)	89.96(9)	101.46(7)	165.37(9)	167.62(9)	51.2
$Pt(bp-7)_S_0$	88.67	81.44	89.48	102.09	165.12	166.27	52.4
$Pt(bp-7)_T_1$	88.79	82.15	89.40	101.28	166.29	166.47	50.8
$Pt(bp-8)_S_0$	88.65	81.95	90.15	100.95	165.65	166.89	52.6
$Pt(bp-8)_T_1$	88.61	82.60	90.22	100.21	166.41	167.37	50.0

^aDihedral angle between terminal pyridine and carboxyl planes for Pt(*ppy*-1), between two terminal pyridine planes for Pt(bp-6) and Pt(bp-7), between terminal pyridine and aza carbazole planes for Pt(bp-8). Optimized S₀ were calculated using a B3LYP method with a basic set of 6-31G(d) for C, H, O and N atoms and a LANL2DZ basic set for Pt atom.

Tuble 50. Ci ystai uutu ulu sti ut	
Identification code	201013_BP_6_0m_sq
Empirical formula	$C_{31}H_{23}N_3OPt$
Formula weight	648.61
Temperature/K	170.0
Crystal system	monoclinic
Space group	$P2_1/n$
a/Å	15.274(6)
b/Å	11.865(3)
c/Å	16.488(5)
α/°	90
β/°	117.245(14)
γ/°	90
Volume/Å ³	2656.5(15)
Z	4
$\rho_{calc}g/cm^3$	1.622
μ/mm^{-1}	5.310
F(000)	1264.0
Crystal size/mm ³	$0.35\times0.16\times0.03$
Radiation	MoKa ($\lambda = 0.71073$)
2Θ range for data collection/°	4.416 to 58.182
Index ranges	$-18 \le h \le 20, -16 \le k \le 16, -22 \le l \le 22$
Reflections collected	35564
Independent reflections	7107 [$R_{int} = 0.0397$, $R_{sigma} = 0.0336$]
Data/restraints/parameters	7107/0/327
Goodness-of-fit on F ²	1.038
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0238, wR_2 = 0.0534$
Final R indexes [all data]	$R_1 = 0.0388, wR_2 = 0.0598$
Largest diff. peak/hole / e Å ⁻³	0.64/-0.86

Table S3. Crystal data and structure refinement for Pt(*bp*-6) (CCDC 2036928).

Table 54. Crystal uata anu struct	ure rennement for r (<i>op-7</i>) (CCDC 2030)
Identification code	200930_0929_1_0m
Empirical formula	$C_{29}H_{19}Cl_2N_3OPt$
Formula weight	691.46
Temperature/K	170.0
Crystal system	monoclinic
Space group	$P2_1/n$
a/Å	8.0827(3)
b/Å	11.0133(4)
c/Å	25.8648(10)
α/°	90
β/°	98.4000(10)
$\gamma/^{\circ}$	90
Volume/Å ³	2277.71(15)
Z	4
$\rho_{calc}g/cm^3$	2.016
μ/mm^{-1}	6.426
F(000)	1336.0
Crystal size/mm ³	$0.26\times0.16\times0.09$
Radiation	MoKa ($\lambda = 0.71073$)
2Θ range for data collection/°	4.88 to 61.034
Index ranges	$-7 \le h \le 11, -15 \le k \le 15, -36 \le l \le 36$
Reflections collected	26825
Independent reflections	6942 [$R_{int} = 0.0374$, $R_{sigma} = 0.0331$]
Data/restraints/parameters	6942/0/325
Goodness-of-fit on F ²	1.041
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0219, wR_2 = 0.0505$
Final R indexes [all data]	$R_1 = 0.0245, wR_2 = 0.0519$
Largest diff. peak/hole / e Å ⁻³	0.76/-1.54

Table S4. Crystal data and structure refinement for Pt(*bp-7*) (CCDC 2036927).

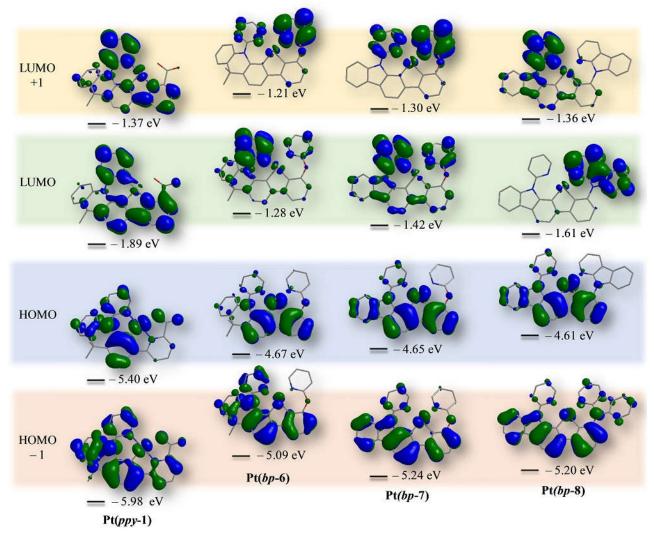


Figure S1. Density functional theory calculations of frontier orbitals and spin densities of T_1 states for Pt(II) complexes based on optimized S_0 and T_1 geometries. The H atoms were omitted for clarity.

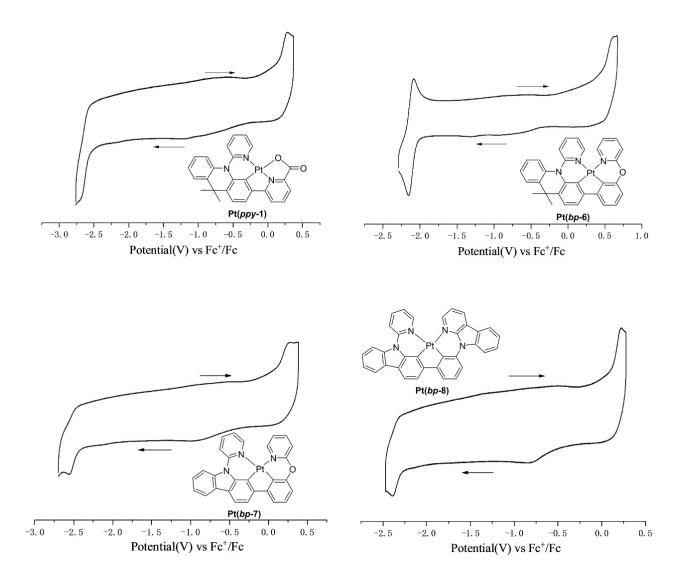


Figure S2. Cyclic voltammograms of Pt(II) complexes in *N*,*N*-dimethylformamide.

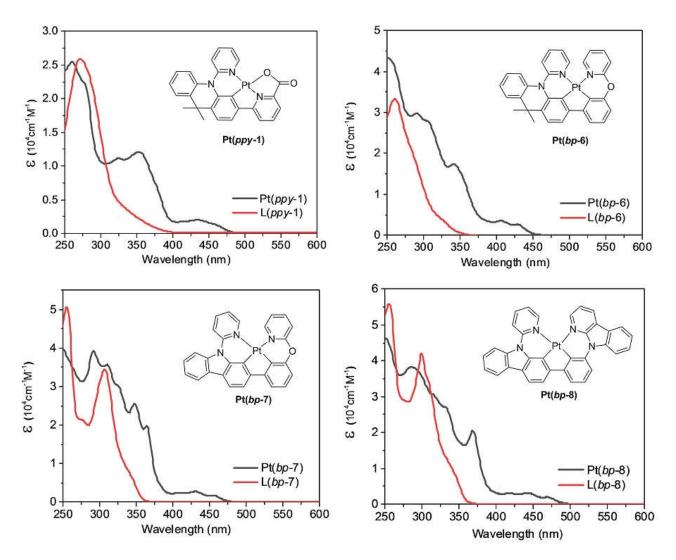


Figure S3. Absorption spectra of Pt(II) complexes and their ligands in dichloromethane solution.

excited state	energy [eV]	wavelength [nm]	f	major contributions
T ₁	2.472	502	0.0000	HOMO \rightarrow LUMO (87%)
\mathbf{S}_1	2.716	457	0.0196	HOMO \rightarrow LUMO (96%)
T ₂	2.759	449	0.0000	HOMO-1 \rightarrow LUMO (39%) HOMO-1 \rightarrow LUMO+2 (4%) HOMO \rightarrow LUMO (8%) HOMO \rightarrow LUMO+1 (25%)
S_2	3.222	385	0.0158	HOMO \rightarrow LUMO+1 (88%) HOMO \rightarrow LUMO+2 (7%)

Table S5. TD-B3LYP/SV/SVP/def2-TZVP results of Pt(ppy-1) at optimized S₀ geometry.

excited state	energy [eV]	wavelength [nm]	f	major contributions
T_1	2.472	502	0.0000	HOMO \rightarrow LUMO (80%) HOMO \rightarrow LUMO+4 (6%)
T ₂	2.587	479	0.0000	HOMO-1 \rightarrow LUMO (11%) HOMO \rightarrow LUMO (4%) HOMO \rightarrow LUMO+1 (70%) HOMO \rightarrow LUMO+4 (5%)
S_1	2.669	465	0.0212	$HOMO \rightarrow LUMO (97\%)$
S_2	2.764	449	0.0075	$HOMO \rightarrow LUMO+1 (96\%)$

Table S6. TD-B3LYP/SV/SVP/def2-TZVP results of Pt(bp-6) at optimized S₀ geometry.

Table S7. TD-B3LYP/SV/SVP/def2-TZVP results of Pt(*bp*-7) at optimized S₀ geometry.

excited state	energy [eV]	wavelength [nm]	f	major contributions
T ₁	2.249	551	0.0000	HOMO \rightarrow LUMO (72%) HOMO \rightarrow LUMO+2 (7%) HOMO \rightarrow LUMO+3 (9%)
T ₂	2.519	492	0.0000	HOMO-1 \rightarrow LUMO (6%) HOMO \rightarrow LUMO (8%) HOMO \rightarrow LUMO+1 (73%) HOMO \rightarrow LUMO+3 (5%)
\mathbf{S}_1	2.519	492	0.0073	HOMO \rightarrow LUMO (97%)
S_2	2.687	461	0.0081	$HOMO \rightarrow LUMO+1 (96\%)$

Table S8. TD-B3LYP/SV/SVP/def2-TZVP results of **Pt(bp-8)** at optimized S0 geometry.

excited state	energy [eV]	wavelength [nm]	f	major contributions
T ₁	2.196	565	0.0000	HOMO \rightarrow LUMO (57%) HOMO \rightarrow LUMO+1 (21%) HOMO \rightarrow LUMO+2 (8%) HOMO \rightarrow LUMO+3 (5%)
T ₂	2.344	529	0.0000	HOMO \rightarrow LUMO (32%) HOMO \rightarrow LUMO+1 (55%)
\mathbf{S}_1	2.407	515	0.0338	HOMO \rightarrow LUMO (94%)
S_2	2.533	490	0.0002	$HOMO \rightarrow LUMO+1 (95\%)$

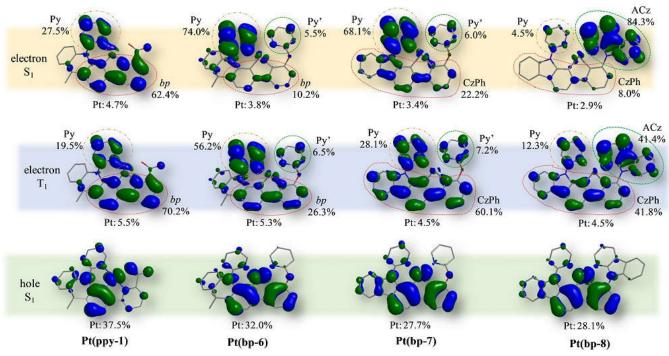


Figure S4. Natural transition orbitals (NTOs) of the S_1 and T_1 states for Pt(II) complexes based on optimized S_0 geometry. All the H atoms are omitted for clarity.

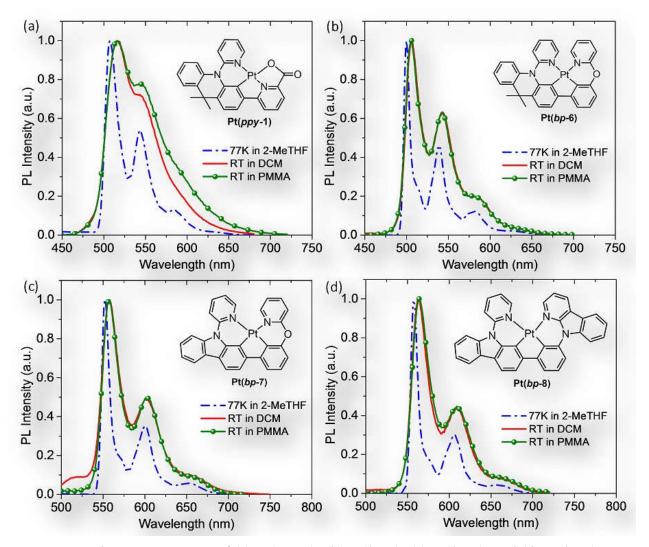


Figure S5. Luminescence spectra of (a) Pt(*ppy*-1), (b) Pt(*bp*-6), (c) Pt(*bp*-7), and (d) Pt(*bp*-8) at 77K in 2-MeTHF (dash-dotted lines), at RT in DCM solution (solid lines) and at RT in PMMA film (solid-ball lines). The chemical structure of each Pt(II) complex is shown in the inset.

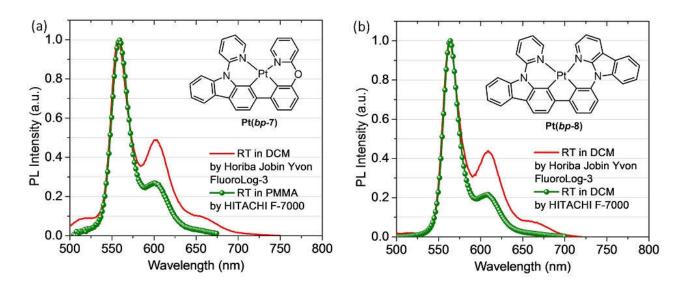
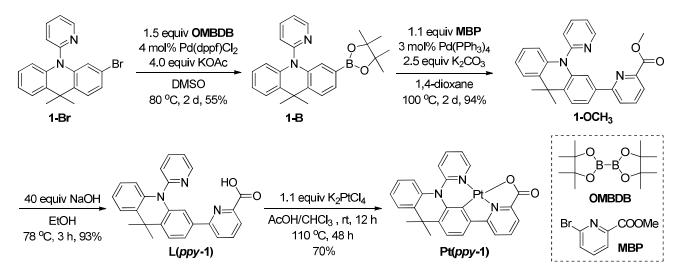


Figure S6. Luminescence spectra comparison measured by different spectrometers.

Experimental Procedures

Synthesis of Pt(ppy-1):



Synthesis of **1-B**: 3-Bromo-9,9-dimethyl-10-(pyridin-2-yl)-9,10-dihydroacridine **1-Br**⁵ (1.50 g, 4.11 mmol, 1.0 equiv), OMBDB (1.56 g, 6.15 mmol, 1.5 equiv), Pd(dppf)Cl₂ (121 mg, 0.16 mmol, 4 mol%), KOAc (1.61 g, 16.50 mmol, 4.0 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then DMSO (25 mL) were added under nitrogen atmosphere, and then the flask was placed in an oil bath and heated at 80 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 2 days, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was washed with water, and then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 10:1-5:1 as eluent to afford the desired product as a white solid 896 mg in 55% yield. ¹H NMR (500 MHz, CDCl₃): δ 1.28 (s, 12H), 1.62 (s, 6H), 7.01–7.05 (m, 2H), 7.10 (td, *J* = 8.0, 1.5 Hz, 1H), 7.17 (ddd, *J* = 7.0, 4.5, 0.5 Hz, 1H), 7.28 (d, *J* = 8.5 Hz, 1H), 7.39 (s, 1H), 7.43 (dd, J = 8.0, 1.5 Hz, 1H), 7.46 (d, J = 8.0 Hz, 1H), 7.50 (dd, J = 7.5, 1.0 Hz, 1H), 7.76 (td, J = 8.0, 2.0 Hz, 1H), 8.62 (dd, J = 5.0, 1.5 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) : δ 24.76, 29.07, 36.97, 83.57, 118.93, 119.11, 120.07, 122.56, 123.90, 124.29, 124.74, 126.08, 129.26, 135.32, 138.75, 138.80, 139.94, 140.43, 149.68, 155.58. HRMS (ESI): calcd for $C_{26}H_{30}N_2O_2B [M+H]^+$ 413.2395, found 413.2415.

Synthesis of 1-OCH₃: 1-B (0.68 g, 1.65 mmol, 1.0 equiv), 6-bromopyridine-2-carboxylic acid

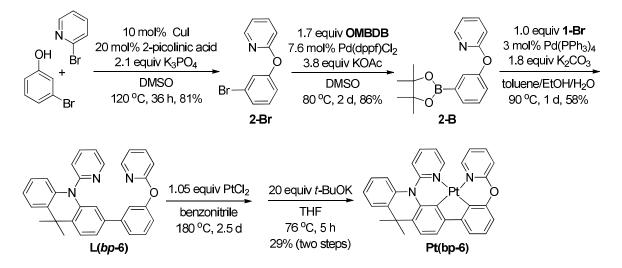
methyl ester (0.39 g, 1.81 mmol, 1.1 equiv), Pd(PPh₃)₄ (57 mg, 0.05 mmol, 3 mol%), K₂CO₃ (0.57 g, 4.12 mmol, 2.5 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then 1,4-dioxane (30 mL) were added under nitrogen atmosphere, and then the flask was placed in an oil bath and heated at 100 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 2 days, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The filtrate was concentrated under reduced pressure and the residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 10:1-3:1 as eluent to afford the desired product as a yellow oily solid 640 mg in 92% yield. ¹H NMR (500 MHz, DMSO-d₆): δ 1.69 (s, 6H), 4.00 (S, 3H), 6.82 (dd, J = 8.0, 1.0, 1H), 7.05 (td, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1.5 Hz, 1H), 7.12 (m, 1H), 7.28-7.31(m, 1H), 7.38 (dt, J = 7.5, 1H), 7.28 (dt, J = 7.8.0, 1.0, 1H), 7.48–7.50 (m, 2H), 7.58 (d, J = 8.5 Hz, 1H), 7.31–7.31(m, 2H), 7.83 (t, J = 2.5 Hz, 1H), 7.87–7.88 (m,1H), 8.00 (dd, J = 7.5, 1.0 Hz, 1H), 8.72–8.74 (m, 1H). ¹³C NMR (125 MHz, CDCl₃) : 8 29.80, 36.60, 52.61, 116.14, 117.59, 121.12, 121.15, 122.33, 123.02, 123.52, 124.67, 125.27, 126.26, 133.84, 135.44, 136.67, 137.44, 139.13, 140.28, 140.71, 147.77, 150.11, 155.48, 157.60, 165.97. HRMS (ESI): calcd for $C_{27}H_{24}N_3O_2$ [M+H]⁺ 422.1877, found 422.1873.

Synthesis of **L**(*ppy*-1): 1-OCH₃ (0.64 g, 1.55 mmol, 1.0 equiv) and NaOH (2.48g, 62.00 mmol, 40 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. Then EtOH (30 mL) was added and the flask was placed in an oil bath and heated at 78 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 3 hours, the resulting mixture was cooled down to room temperature, filtered and add hydrochloric acid to adjust pH to 6, washed with CH₂Cl₂. The organic layer was separated and dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 3:1 and then ethyl acetate/CH₃OH = 10:1 as eluent to afford the desired product as a maroon solid 587 mg in 93% yield. ¹H NMR (500 MHz, DMSO-*d*₆): δ 1.63 (s, 6H), 6.64 (dd, *J* = 8.0, 1.5 Hz, 1H), 7.03 (td, *J* = 7.5, 1.5 Hz, 1H), 7.09 (td, *J* = 8.0, 2.0 Hz, 1H), 7.40 (d, *J* = 2.0 Hz, 1H), 7.43-7.46 (m, 2H), 7.53 (dd, *J* = 7.5, 1.5 Hz, 1H), 7.65 (d, *J* = 8.5 Hz, 1H), 7.77 (dd, *J* = 8.5, 2.0 Hz, 1H), 7.90-7.93 (m, 2H), 7.99 (t, *J* = 7.5 Hz, 1H), 8.03 (td, *J* = 7.5, 2.0 Hz, 1H), 8.67-8.69 (m, 1H), 12.73 (br, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 30.18, 36.81, 115.71, 117.18, 120.52, 121.72, 121.96, 121.99, 122.64, 124.69, 125.05,

125.69, 126.67, 133.35, 135.15, 135.90, 139.29, 139.58, 140.28, 141.18, 145.79, 150.55, 155.45, 156.39, 164.23. HRMS (ESI): calcd for $C_{26}H_{21}N_3O_2Na [M+Na]^+ 430.1526$, found 430.1520.

Synthesis of Pt(ppy-1): A mixture of L(ppy-1) (300 mg, 0.74 mmol, 1.00 equiv) and K₂PtCl₄ (336 mg, 0.81 mmol, 1.10 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then AcOH (40 mL) and CHCl₃ (4mL) were added into the flask under nitrogen atmosphere. The reaction mixture was bubbled with nitrogen for 15 minutes and then stirred at room temperature for 12 h. Then the flask was placed in an oil bath and heated at 110 ^oC with stirring. After 48 hours, the resulting mixture was cooled down to room temperature and concentrated in vacuum and the residue was diluted with dichloromethane. The mixture was washed with water, the organic layer was separated and dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/dichloromethane= 2:1, and then dichloromethane/ethyl acetate/CH₃OH =1:1:0.1 as eluent to afford the desired product as a yellow solid 309 mg in 70% yield. m.p.: >320 °C. ¹H NMR (500 MHz, DMSO- d_6): δ 1.34 (s, 3H), 1.88 (s, 3H), 7.20–7.24 (m, 2H), 7.25-7.28 (m, 3H), 7.57 (dd, J = 7.5, 1.5 Hz, 1H), 7.60-7.63 (m, 2H), 7.66 (dd, J = 7.0, 1.5 Hz, 1H), 7.60-7.63 (m, 2H), 7.66 (m, 2H), 1H), 8.03–8.07 (m, 1H), 8.08–8.13 (m, 2H), 8.85 (dd, J = 6.0, 1.5 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃): 8 22.96, 31.32, 37.26, 118.29, 118.94, 119.87, 120.92, 121.29, 121.34, 122.37, 122.79, 124.31, 125.51, 126.73, 135.59, 137.12, 138.36, 139.32, 139.53, 141.41, 142.46, 150.93, 151.24, 151.79, 162.62, 172.86. HRMS (ESI): calcd for C₂₆H₁₉N₃O₂Na¹⁹⁵Pt [M+Na]⁺ 623.1017, found 623.1008. Anal. for C₂₆H₁₉N₃O₂Pt, Calcd.: C, 52.00, H, 3.19, N, 7.00; Found: C, 51.65, H, 3.20, N, 6.95.

Synthesis of Pt(bp-6):



Synthesis of 2-Br: CuI (440 mg, 2.31 mmol, 10 mol%), 2-picolinic acid (568 mg, 4.62 mmol, 20 mol%) and K₃PO₄ (10.31 g, 48.55 mmol, 2.1 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuate ion and backfill procedure was repeated twice. Then *m*-bromophenol (4.00 g, 23.12 mmol, 1.00 equiv), 2-bromopyridine (5.48 g, 34.68 mmol, 1.50 equiv) and DMSO (30 mL) were added under nitrogen atmosphere, and then the flask was placed in an oil bath and heated at 120 ^oC with stirring. The reaction was monitored by TLC until the reaction was completed. After 36 hours, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was washed with water, and then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/dichloromethane = 40:1-20:1 as eluent to afford the desired product as a white solid 4.67 g in 81% yield. ¹H NMR (500 MHz, CDCl₃): δ 7.08 (d, J= 8.5 Hz, 1H), 7.14-7.18 (m, 2H), 7.36-7.43 (m, 3H), 7.86-7.90 (m, 1H), 7.08 (ddd, J= 4.5, 2.0, 0.5 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) : δ 111.81, 118.96, 119.85, 122.58, 124.45, 127.66, 130.62, 139.61, 147.67, 154.84, 163.06. HRMS (ESI): calcd for $C_{11}H_9NOBr [M+H]^+$ 249.9862, found 249.9862.

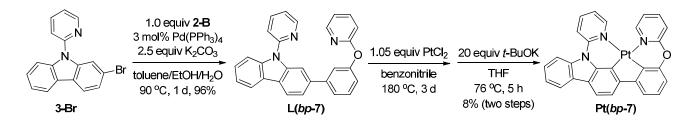
Synthesis of **2-B**: 2-(3-Bromophenoxy)pyridine **2-Br** (4.00 g, 16.00 mmol, 1.00 equiv), **OMBDB** (6.92 g, 27.25 mmol, 1.70 equiv), $Pd(dppf)Cl_2$ (886 mg, 1.21 mmol, 7.6 mol%), KOAc (5.94 g, 60.53 mmol, 3.8 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill

procedure was repeated twice. Then DMSO (45 mL) was added under nitrogen atmosphere, and then the flask was placed in an oil bath and heated at 80 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 2 days, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was washed with water, and then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 20:1-14:1 as eluent to afford the desired product as a white solid 4.07 g in 86% yield. ¹H NMR (500 MHz, CDCl₃): δ 1.33 (s, 12 H), 6.87–6.89 (m, 1H), 6.95–6.99 (m, 1H), 7.24 (ddd, J = 4.0, 3.0, 1.5 Hz, 1H), 7.41 (t, J = 7.5 Hz, 1H), 7.56 (dd, J = 2.5, 1.0 Hz, 1H), 7.64–7.66 (m, 1H), 7.67–7.68 (m, 1H), 8.19 (ddd, J = 3.0, 2.0, 1.0 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) : δ 24.82, 83.88, 111.36, 118.22, 124.35, 127.16, 129.15, 131.15, 139.33, 147.67, 153.62, 163.88. HRMS (ESI): calcd for C₁₇H₂₁NO₃B [M+H]⁺ 298.1609, found 298.1623.

Synthesis of L(bp-6): 3-Bromo-9,9-dimethyl-10-(pyridin-2-yl)-9,10-dihydroa cridine 1-Br (1.30 g, 3.57 mmol, 1.00 equiv), 2-(3-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2yl)phenoxy) pyridine **2-B** (1.16 g, 3.93 mmol, 1.10 equiv), Pd(PPh₃)₄ (124 mg , 0.11 mmol, 3 mol%), K₂CO₃ (888 mg, 6.43 mmol, 1.8 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. Then toluene (25 mL), EtOH (5 mL) and H₂O (5 mL) were added, and then the flask was placed in an oil bath and heated at 90 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 1 day, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was separated and washed with water, then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 10:1-5:1 as eluent to afford the desired product as a white solid 950 mg in 58% yield. ¹H NMR (500 MHz, CDCl₃): δ 1.68 (s, 6H), 6.78 (dd, J = 8.0, 1.5 Hz, 1H), 6.89 (dt, J = 8.5, 1.5 Hz, 1H), 6.97–6.98 (m, 1H), 6.99-7.01 (m, 1H), 7.00-7.09 (m, 3H), 7.20-7.25 (m, 3H), 7.27 (dt, J = 6.5, 1.0 Hz, 1H), 7.33 (dt, J = 8.0, 2.0 Hz, 1H), 7.37 (t, J = 8.0 Hz, 1H), 7.47 (dd, J = 8.0, 2.0 Hz, 1H), 7.48 (d, J = 9.5Hz, 1H), 7.66–7.69 (m, 1H), 7.82 (td, J = 8.0, 2.0 Hz, 1H), 8.18–8.20 (m, 1H), 8.68 (dd, J= 9.5, 1.5 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 29.49, 35.86, 110.96, 115.53, 116.79, 117.90, 118.99, 119.16, 120.46, 120.69, 120.72, 121.64, 122.80, 124.16, 124.62, 125.66, 129.13, 132.75, 133.19, 138.05, 138.60, 138.80, 139.71, 140.06, 142.45, 147.24, 149.69, 153.86, 154.85, 163.11. HRMS (ESI): calcd for $C_{31}H_{26}N_3O [M+H]^+$ 456.2070, found 456.2068.

Synthesis of Pt(bp-6): A mixture of L(bp-6) (250 mg, 0.55 mmol, 1.00 equiv) and PtCl₂ (153 mg, 0.58 mmol, 1.05 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then benzonitrile (30 mL) was added into the flask, and the flask was placed in an oil bath and heated at 180 °C with stirring. After 2.5 days, the mixture was cooled to room temperature. THF (12 mL) and t-BuOK (1.23 g, 10.98 mmol, 20.00 equiv) were added to the mixture under nitrogen atmosphere and stirred in an oil bath held at 76 °C for 5 h. The resulting mixture was cooled down to room temperature and concentrated in vacuum. The residue was diluted with dichloromethane, and washed with water. The organic layer was separated, and the aqueous phase was extracted with dichloromethane twice. The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The crude solid was purified through column chromatography on silica gel using petroleum ether/dichloromethane = 3:1-1:1 as eluent to afford the desired product as a yellow solid 104 mg in 29% yield. m.p.: 314.5-315.5 °C. ¹H NMR (500 MHz, DMSO-d₆): δ 1.27 (s, 3H), 1.84 (s, 3H), 6.75 (dd, J = 8.0, 1.0 Hz, 1H), 6.91-6.98 (m, 2H), 7.10-7.14 (m, 2H), 7.19 (td, J = 8.5, 1.5 Hz, 1H),7.21–7.24 (m, 2H), 7.28 (dd, J = 8.0, 1.5 Hz, 1H), 7.33–7.35 (m, 1H), 7.44 (d, J = 8.5 Hz, 1H), 7.48 (dt, J = 8.5, 1.0 Hz, 1H), 7.53 (dd, J = 8.0, 2.0 Hz, 1H), 7.93-7.98 (m, 1H), 8.16-8.21 (m, 1H), 8.60(dd, J= 6.0, 2.0 Hz, 1H), 8.66 (dd, J= 6.0, 2.0 Hz, 1H). ¹³C NMR (125 MHz, DMSO- d_6): δ 22.72, 32.33, 36.29, 113.26, 116.06, 116.18, 118.12, 119.75, 119.88, 120.53, 120.90, 123.80, 124.26, 122.80, 124.54, 126.48, 128.43, 132.33, 135.28, 137.09, 138.21, 139.27, 140.19, 141.48, 148.89, 150.39, 151.74, 154.07, 156.96, 157.51. HRMS (ESI): calcd for C₃₁H₂₃N₃ONa¹⁹⁵Pt [M+Na]⁺ 671.1381, found 671.1384. Anal. for C₃₁H₂₃N₃OPt•0.5H₂O, Calcd.: C, 56.62, H, 3.68, N, 6.39; Found: C, 56.65, H, 3.70, N, 6.26.

Synthesis of Pt(bp-7):

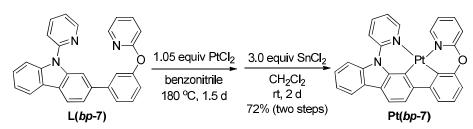


Synthesis of L(bp-7): 2-(3-(4,4,5,5-Tetramethyl-1,3,2-dioxaborolan-2yl)phenoxy)pyridine 2-B (1.00 g, 3.4 mmol, 1.0 equiv), 2-bromo-9-(pyridin-2-yl)-9H-carbazole **3-Br**⁶ (1.20 g, 3.7 mmol, 1.1 equiv), Pd(PPh₃)₄ (116 mg, 0.10 mmol, 3 mol%), K₂CO₃ (1.16 g, 8.4 mmol, 2.5 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then toluene (20 mL), EtOH(4 mL) and H₂O (4 mL) were added, and then the flask was placed in an oil bath and heated at 90 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 24 hours, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was washed with water, and then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 10:1-5:1 as eluent to afford the desired product as a white solid 1.32 g in 96% yield. ¹H NMR (500 MHz, CDCl₃): δ 6.94 (dd, J = 8.5 Hz, 1H), 7.00 (ddd, J = 6.0, 5.0, 1.0 Hz, 1H), 7.10-7.13 (m, 1H), 7.30-7.34 (m, 2H),7.42–7.45 (m, 2H), 7.47 (d, J = 8.0, 1.0 Hz, 1H), 7.51 (dt, J = 7.5, 1.0 Hz, 1H), 7.55 (dd, J = 8.0, 1.5Hz, 1H), 7.64–7.66 (m, 1H), 7.67–7.70 (m, 1H), 7.82 (dd, J= 7.5, 1.0 Hz, 1H), 7.94 (td, J= 8.0 Hz, 1H), 8.02 (m, 1H), 8.12 (d, J = 7.5, 1.0 Hz, 1H), 8.15 (d, J = 8.0 Hz, 1H), 8.20–8.23 (m, 1H), 8.72-8.75 (m, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 109.95, 111.26, 111.66, 118.65, 119.36, 119.89, 120.40, 120.48, 120.55, 120.73, 121.20, 121.51, 123.88, 124.11, 124.14, 126.44, 129.99, 138.71, 139.08, 139.57, 140.21, 140.27, 144.04, 147.98, 149.89, 151.86, 154.67, 163.93. HRMS (ESI): calcd for $C_{28}H_{20}N_{3}O[M+H]^{+}$ 414.1601, found 414.1596.

Synthesis of Pt(bp-7): A mixture of L(bp-7) (350 mg, 0.85 mmol, 1.00 equiv) and $PtCl_2$ (236 mg, 0.89 mmol, 1.05 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then benzonitrile (40 mL) was added into the flask ,and the flask was

placed in an oil bath and heated at 180 °C with stirring. After 3 days, the mixture was cooled to room temperature. THF (15 mL) and t-BuOK (1.90 g, 16.93 mmol, 20.00 equiv) were added to the mixture under nitrogen atmosphere and stirred in an oil bath at 76 °C for 5 h. The resulting mixture was cooled down to room temperature and concentrated in vacuum. The residue was diluted with dichloromethane, and washed with water. The organic layer was separated, and the aqueous phase was extracted with dichloromethane twice. The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The crude solid was purified through column chromatography on silica gel using petroleum ether/dichloromethane = 3:1-1:1 as eluent to afford the desired product as a yellow solid 40 mg in 8% yield. m.p.: 317.6–318.9 °C. ¹H NMR (500 MHz, DMSO- d_6): δ 6.75 (dd, J = 8.0, 1.0 Hz, 1H), 6.94 (t, J = 7.5 Hz, 1H), 7.18 (dd, J = 7.5, 1.0 Hz, 1H), 7.31–7.34 (m, 2H), 7.35–7.41 (m, 2H), 7.45–7.49 (m, 1H), 7.52 (dd, J = 7.5, 1.0 Hz, 1H), 7.60 (d, J = 7.5 Hz, 1H), 8.03 (d, J = 8.0 Hz, 1H), 8.10 (dd, J = 7.5, 1.5 Hz, 1H), 8.03 (d, J = 8.0 Hz, 1H), 8.10 (dd, J = 7.5, 1.5 Hz, 1H), 8.03 (d, J = 8.0 Hz, 1H), 8.10 (dd, J = 7.5, 1.5 Hz, 1H), 8.10 (dd, J = 7.5, 1H), 8.10 (dd, J = 7.5Hz,1H), 8.15–8.22 (m, 2H), 8.28 (d, J = 8.5 Hz, 1H), 8.70 (dd, J = 6.0, 1.0 Hz, 1H), 8.78 (dd, J = 6.0, 1.5 Hz, 1H). ¹³C NMR (150 MHz, DMSO-*d*₆): δ 113.91, 114.23, 114.91, 116.44, 116.50, 116.62, 120.16, 120.62, 120.95, 122.76, 124.94, 125.91, 127.67, 128.15, 129.11, 138.50, 139.54, 140.04, 140.64, 148.98, 152.17, 152.63, 152.68, 156.98, 158.37. HRMS (DART positive ion mode): calcd for C₂₈H₁₈ON₃Pt [M+H]⁺ 607.1092, found 607.1092. Anal. for C₂₈H₁₇ON₃Pt•0.5CH₂Cl₂, Calcd.: C, 52.74, H, 2.80, N, 6.47; Found: C, 53.05, H, 2.95, N, 6.44.

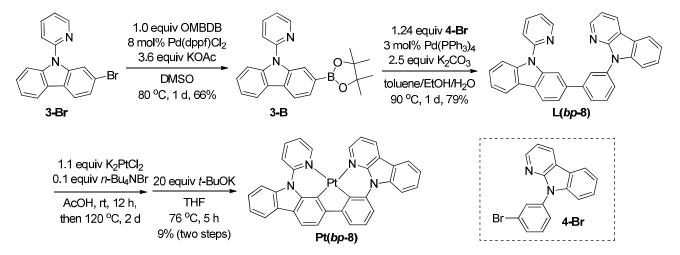
Improved synthesis of **Pt(bp-7)**:



A mixture of L(*bp*-7) (200 mg, 0.48 mmol, 1.00 equiv) and PtCl₂ (135 mg, 0.51 mmol, 1.05 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and back-filled with nitrogen, this evacuation and backfill procedure was repeated twice. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then benzonitrile (30 mL) was added into the flask , and the flask was placed in an oil bath and heated at 180 °C with stirring. After 36 hours, the mixture was cooled to

room temperature and concentrated in vacuum. The residue was diluted with dichloromethane, and washed with water. The organic layer was separated, and the aqueous phase was extracted with dichloromethane twice. The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The crude product and anhydrous SnCl₂ (274 mg, 1.45 mmol, 3.00 equiv) were added to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then dichloroethane (30 mL) was added under nitrogen atmosphere and stirred at room temperature for two days. Then the mixture was washed with water, separated and the aqueous phase was extracted with dichloromethane. The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The crude solid was purified through column chromatography on silica gel using petroleum ether/dichloromethane = 3:1-2:1 as eluent to afford the desired product as a yellow solid 213 mg in 72% yield.¹H NMR (500 MHz, DMSO- d_6) :¹H NMR (500 MHz, DMSO- d_6) : δ 6.75 (dd, J = 8.0, 1.0 Hz, 1H), 6.93 (t, J = 7.5 Hz, 1H), 7.17 (dd, J= 7.0, 1.0 Hz, 1H), 7.30–7.33 (m, 2H), 7.34–7.40 (m, 2H), 7.47 (td, J= 8.0, 1.0 Hz, 1H), 7.52 (dd, J= 8.5, 1.0 Hz, 1H), 7.60 (d, J= 7.5 Hz, 1H), 8.02 (d, J= 8.5 Hz, 1H), 8.08-8.11 (m, 1H), 8.16–8.22 (m, 2H), 8.27 (d, J= 8.5 Hz, 1H), 8.69 (dd, J= 6.0, 2.0 Hz, 1H), 8.78 (dd, J= 6.0, 2.0 Hz, 1H). The ¹H NMR is agreement with the data obtained above.

Synthesis of Pt(bp-8):

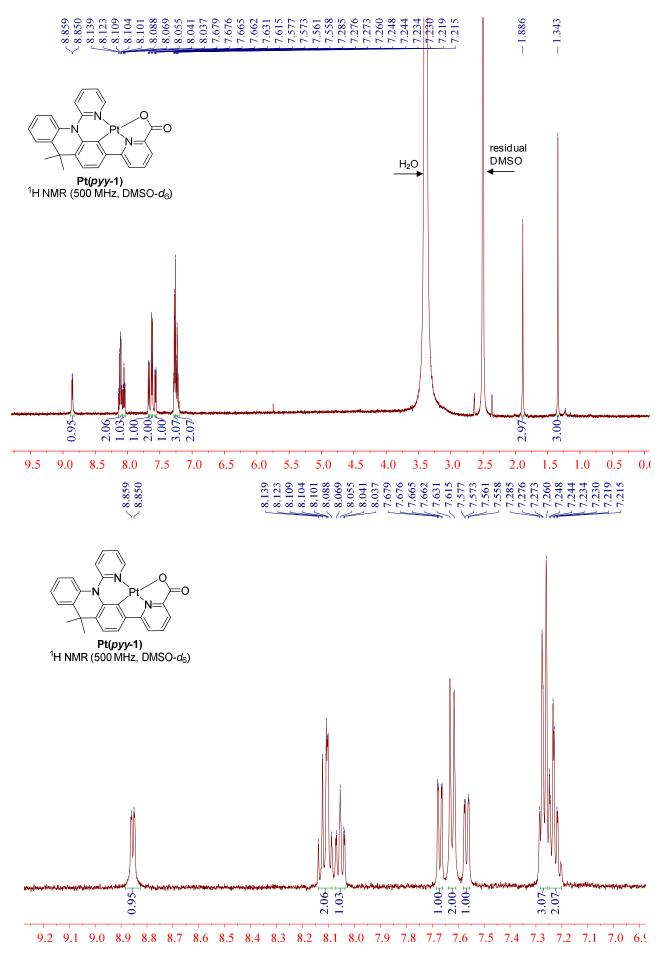


Synthesis of **3-B**: 2-Bromo-9-(pyridin-2-yl)-9*H*-carbazole **3-Br**⁶ (1.50 g, 4.64 mmol, 1.02 equiv), **OMBDB** (1.15 g, 4.53 mmol, 1.00 equiv), Pd(dppf)Cl₂ (263 mg, 0.36 mmol, 8 mol%), KOAc (1.62 g, 16.48 mmol, 3.6 equiv) were added sequentially to a dry three-necked flask equipped with a

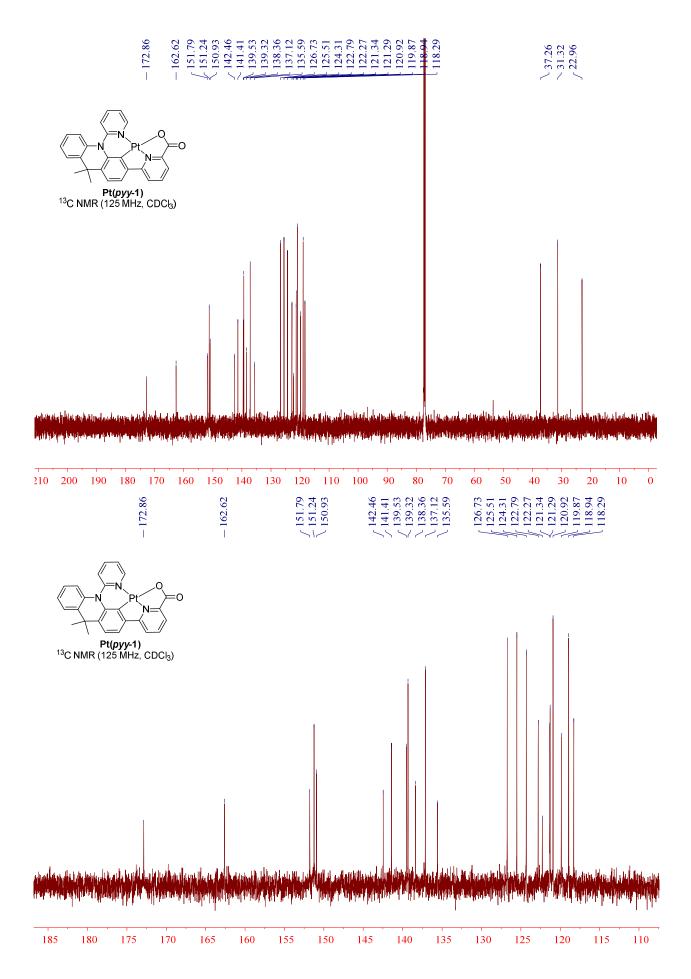
magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then DMSO (20 mL) were added under nitrogen atmosphere, and then the flask was placed in an oil bath and heated at 80 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 1day, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was washed with water, and then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 20:1–10:1 as eluent to afford the desired product as a white solid 1.10 g in 72% yield. ¹H NMR (500 MHz, CDCl₃): δ 1.34 (s, 12H), 7.22 (dd, *J* = 7.5, 5.0 Hz, 1H), 7.30–7.34 (m, 1H), 7.42 (dt, *J* = 8.0, 1.0Hz, 1H), 7.45–7.48 (m, 1H), 7.63 (t, *J* = 8.0 Hz, 1H), 7.72 (ddd, *J* = 7.5, 2.0, 1.0 Hz, 1H), 7.92 (dt, *J* = 7.5, 1.5 Hz, 1H), 8.03 (dd, *J* = 2.5, 1.0 Hz, 1H), 8.12 (dt, *J* = 7.5, 1.0 Hz, 1H), 8.38 (dd, *J* = 7.5, 1.5 Hz, 1H), 8.48 (dd, *J* = 4.5, 1.5 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃) : δ 24.87, 83.76, 111.19, 117.04, 119.47, 119.49, 120.54, 120.82, 121.30, 123.94, 126.77, 127.07, 138.61, 139.12, 140.13, 149.67, 151.64.HRMS (ESI): calcd for C₂₃H₂₄An₂O₂B [M+H]⁺ 371.1925, found 371.1945.

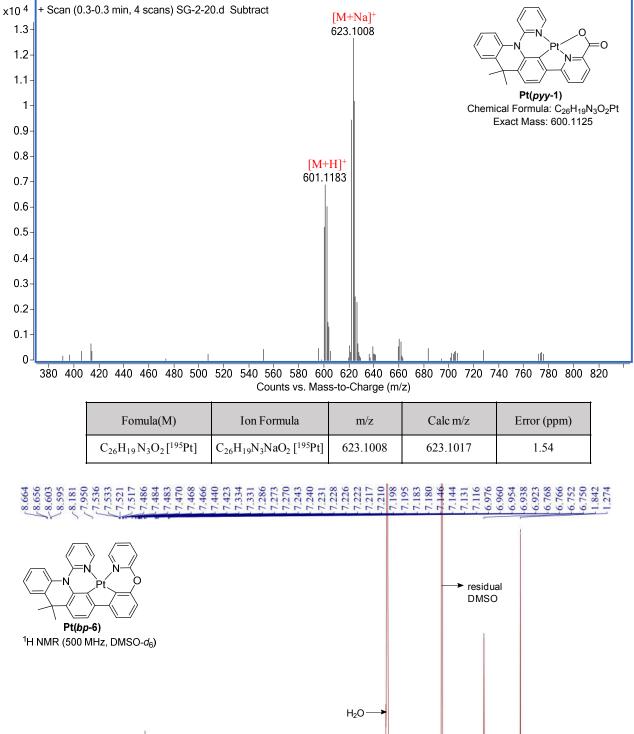
Synthesis of L(bp-8): 3-B (1.00 g, 2.70 mmol, 1.00 equiv), 4-Br⁵ (1.08 g, 3.34 mmol, 1.24 equiv), Pd(PPh₃)₄ (94 mg, 0.08 mmol, 3 mol%), K₂CO₃ (933 mg, 6.75 mmol, 2.50 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. Then toluene (20 mL), EtOH (4 mL) and H₂O (4 mL) were added, then the flask was placed in an oil bath and heated at 90 °C with stirring. The reaction was monitored by TLC until the reaction was completed. After 24 hours, the resulting mixture was cooled down to room temperature, filtered and washed with ethyl acetate. The organic layer was washed with water, and then dried over Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue was purified through column chromatography on silica gel using petroleum ether/ethyl acetate = 10:1-3:1 as eluent to afford the desired product as a white solid 1.04 g in 79% yield. ¹H NMR (500 MHz, CDCl₃): δ 7.28–7.34 (m, 2H), 7.33–7.37 (m, 2H), 7.43–7.46 (m, 1H), 7.46–7.49 (m, 1H), 7.56 (d, J = 8.0 Hz, 1H), 7.63 (dd, J = 8.0, 1.5 Hz, 2H, 7.66–7.71 (m, 2H), 7.77 (dt, J = 3.0, 1.5 Hz, 1H), 7.83 (d, J = 8.0 Hz, 1H), 7.91–7.95 (m, 2H), 8.08 (d, J = 1.0 Hz,1H), 8.14 (t, J = 7.5, 1.5 Hz, 2H),8.18 (d, J = 8.0 Hz, 1H), 8.41 (dd, J = 7.5, 1.5Hz, 1H), 8.49 (d, J = 5.0, 2.0 Hz, 1H), 8.71 (m, 1H). ¹³C NMR (125 MHz, CDCl₃): § 109.99, 110.55, 111.15, 116.11, 116.41, 119.24, 120.32, 120.53, 120.71, 120.78, 120.89, 120.96, 121.10, 121.40, 123.85, 124.01, 126.08, 126.37, 126.51, 126.96, 127.00, 128.32, 129.98, 136.68, 138.59, 138.88, 140.14, 140.17, 140.18, 143.74, 146.54, 149.77, 151.73, 152.03. HRMS (ESI): calcd for $C_{34}H_{23}N_4[M+H]^+$ 487.1917, found 487.1914.

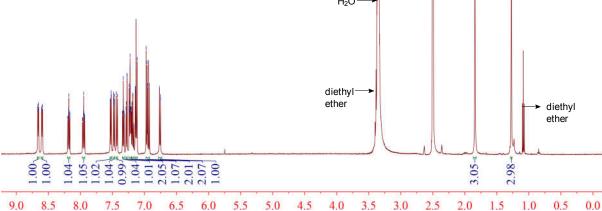
Synthesis of Pt(bp-8): A mixture of L(bp-8) (200 mg, 0.41 mmol, 1.00 equiv), K₂PtCl₄ (188 mg, 0.45 mmol, 1.10 equiv) and *n*-Bu₄NBr (13 mg, 0.04 mmol, 0.1 equiv) were added sequentially to a dry three-necked flask equipped with a magnetic stir bar. The flask was evacuated and backfilled with nitrogen, this evacuation and backfill procedure was repeated twice. Then AcOH (30 mL) were added into the flask under nitrogen atmosphere at room temperature. The reaction mixture was bubbled with nitrogen for 30 minutes and then stirred at room temperature for 12 h. Then the flask was placed in an oil bath and heated at 120 °C with stirring. After two days, the resulting mixture was cooled down to room temperature and concentrated in vacuum and the residue was diluted with dichloromethane. The mixture was washed with water, the organic layer was separated and dried over anhydrous Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The residue obtained in last step, THF (15 mL) and t-BuOK (460 mg, 4.1 mmol, 10.00 equiv) were added to the mixture under nitrogen atmosphere and stirred in an oil bath at 76 °C for 5 h. The resulting mixture was cooled down to room temperature and concentrated in vacuum. The residue was diluted with dichloromethane, and washed with water. The organic layer was separated, and the aqueous phase was extracted with dichloromethane twice. The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and the filtrate was concentrated under reduced pressure. The crude solid was purified through column chromatography on silica gel using petroleum ether/dichloromethane = 3:1-1:1 as eluent to afford the desired product as a yellow solid 25 mg in 9% yield. m.p.: 221.1–222.0 °C. ¹H NMR (500 MHz, DMSO-*d*₆): δ 7.07–7.11 (m, 1H),7.23–7.27 (m, 1H), 7.32 (d, J = 7.0 Hz, 1H), 7.37 (t, J = 8.0 Hz, 1H), 7.42 (d, J = 8.0 Hz, 1H), 7.46–7.50 (m, 1H),7.51–7.54 (t, J = 7.5 Hz, 1H), 7.61–7.66 (m, 3H), 7.69–7.73 (m, 1H), 8.04 (d, J = 8.5 Hz, 1H), 8.11-8.13 (m, 1H), 8.15-8.19 (m, 1H), 8.25-8.31 (t, J = 6.5 Hz, 2H), 8.45-8.48 (m, 1H), 8.53-8.56 (dd, J = 6.0, 2.0 Hz, 1H), 8.88 (dd, J = 5.5, 1.5 Hz, 1H), 9.05 (dd, J = 7.5, 1.5 Hz, 1H).¹³C NMR (150 MHz, DMSO-d₆): δ 113.97, 114.89, 115.60, 116.42, 116.66, 116.73, 118.38, 119.64, 119.84, 120.65, 120.88, 121.97, 122.16, 122.36, 122.71, 124.43, 125.88, 127.85, 128.11, 129.50, 130.50, 135.63, 136.10, 138.26, 138.30, 139.36, 139.68, 146.39, 146.44, 149.00, 152.44, 153.61, 157.00. HRMS (DART positive ion mode): calcd for $C_{34}H_{21}N_4Pt [M+H]^+ 680.1408$, found 680.1411. Anal. for C₃₄H₂₀N₄Pt, Calcd.: C, 60.09, H, 2.97, N, 8.24; Found: C, 59.60, H, 3.30, N, 7.88.



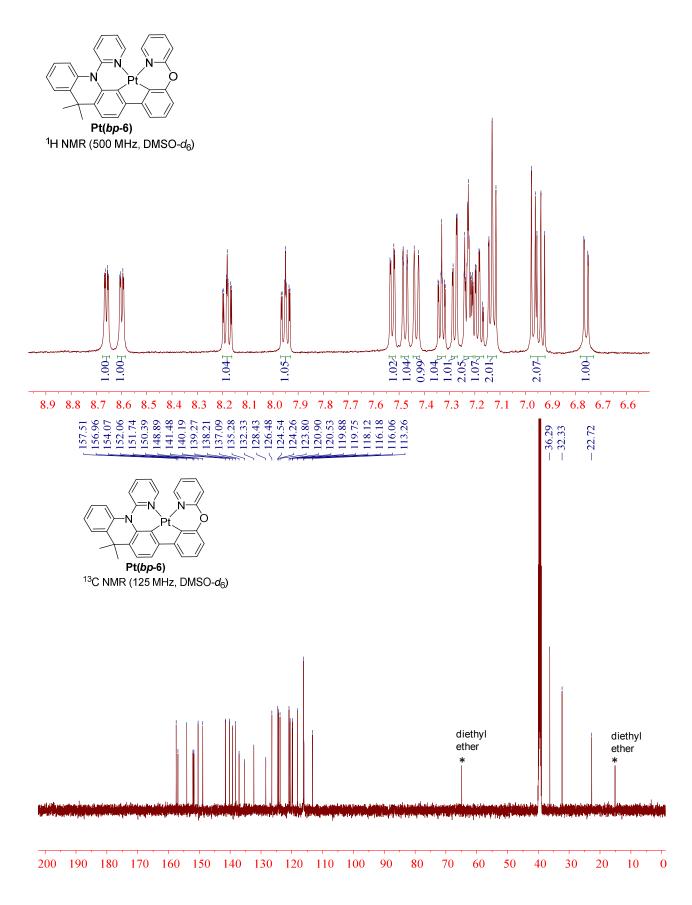
S-**26** / S-**52**







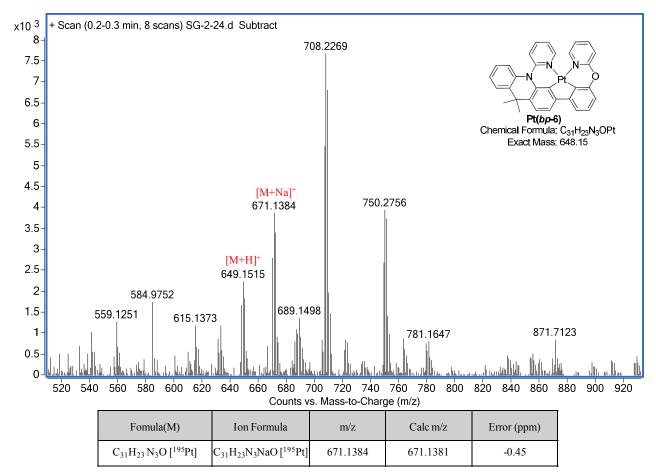


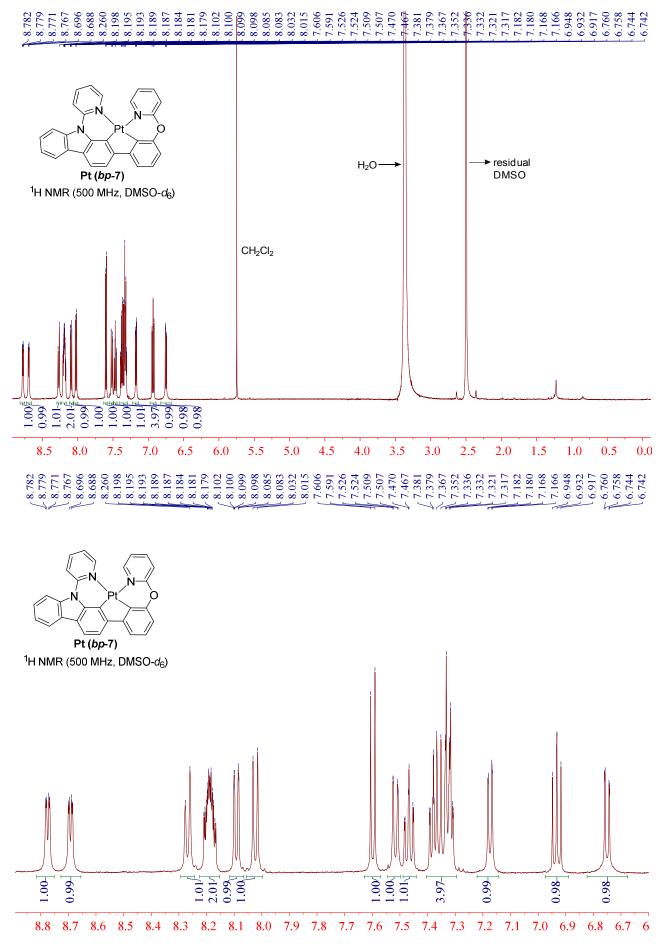




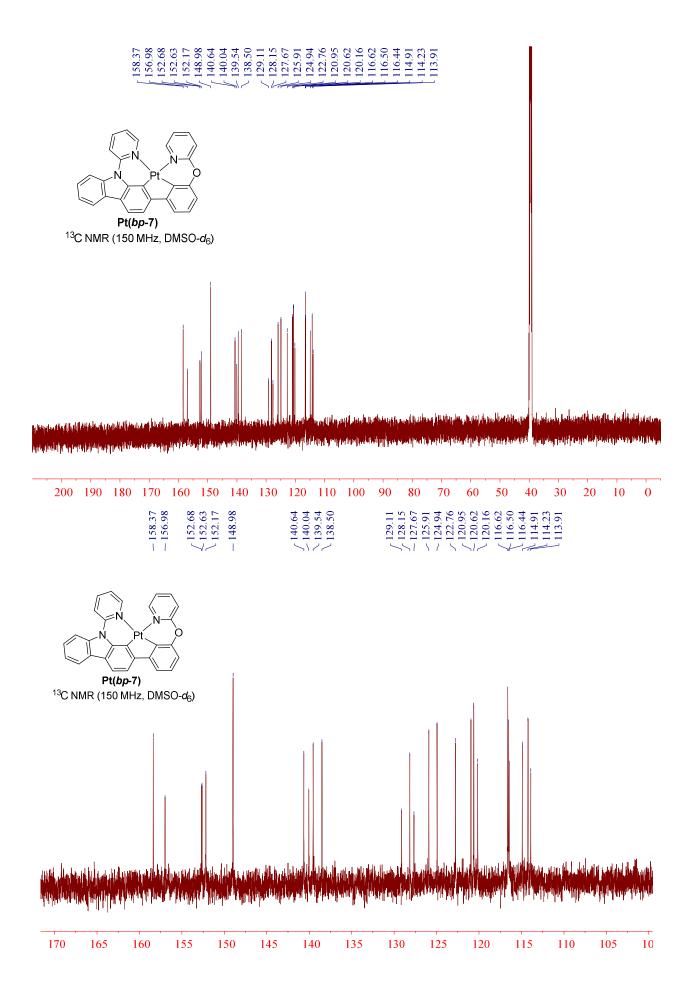


160 158 156 154 152 150 148 146 144 142 140 138 136 134 132 130 128 126 124 122 120 118 116 114 112 110 10





S-**31** / S-**52**



Card Serial Number : D172118+

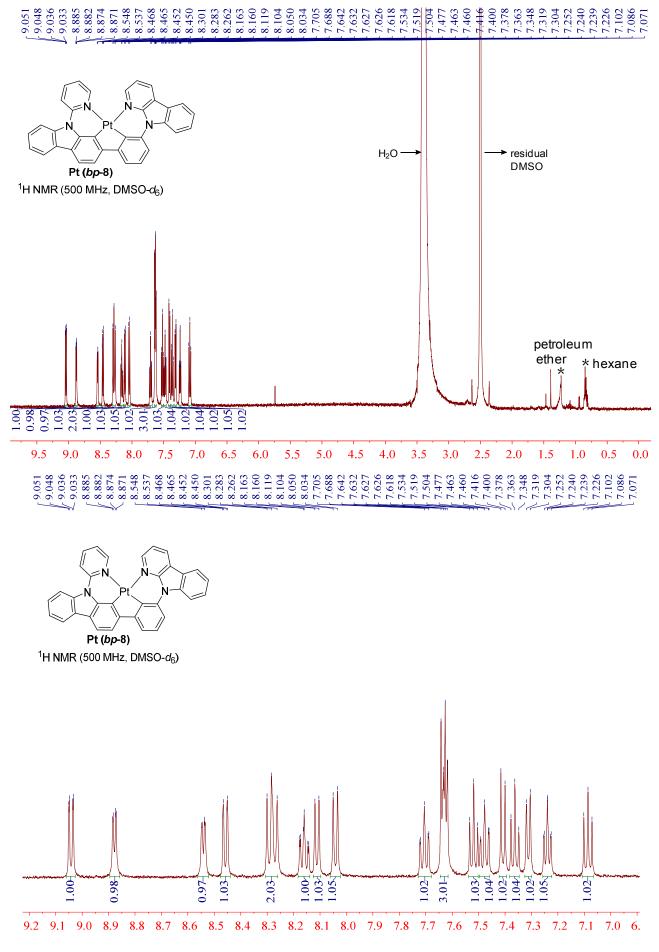
Sample Serial Number: Z-9-2.

Operator : DONG Date: 2017/12/20.

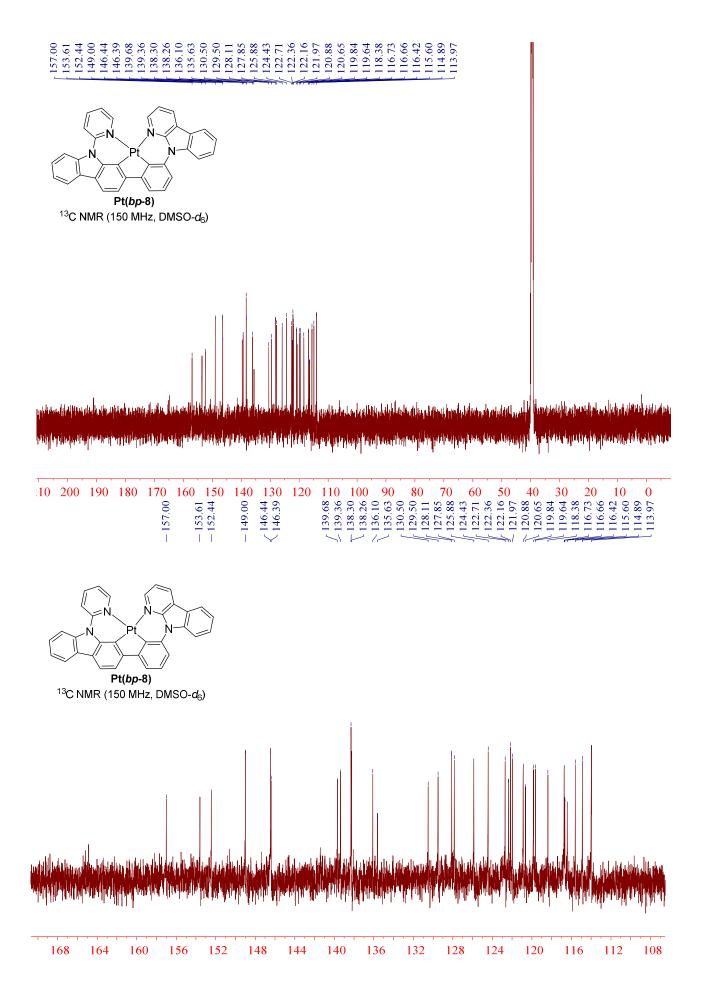
Operation Mode: DART POSITIVE Ion Mode-

Elemental composition search on mass 607.11

m/z	Theo. Mass	Delta (ppm)	RDB equiv.	Composition
07.1092	607.1092	-0.00	21.5	C ₂₈ H ₁₈ O N ₃ Pt
	607.1097	-0.88	33.0	C ₃₈ H ₁₇ O ₂ N ₅ S
	607.1086	1.07	12.5	C ₂₀ H ₂₂ O ₃ N ₅ Pt S
	607.1099	-1.14	12.0	C22 H24 O4 N2 Pt S
	607.1106	-2.22	21.0	C30 H20 O2 Pt
	607.1077	2.46	37.5	C43H15O3N2
	607.1111	-3.10	32.5	C40 H19 O3 N2 S
	607.1071	3.53	28.5	C 35 H 19 O 5 N 4 S Pt(bp
	607.1065	4.41	17.0	C ₂₅ H ₂₀ O ₄ N ₂ Pt
	607.1064	4.67	38.0	C ₄₁ H ₁₃ O ₂ N ₅



S-**34** / S-**52**



S-**35** / S-**52**

Card Serial Number : D172122+

Sample Serial Number: Z-10-72+

Operator : DONG Date: 2017/12/20-

Operation Mode: DART POSITIVE Ion Mode, Elemental composition search on mass 680.14

m/z	Theo. Mass	Delta (ppm)	RDB equiv.	Composition	
680.1411	680.1408	0.43	26.5	C 34 H 21 N 4 Pt	
	680.1415	-0.59	17.0	C ₂₈ H ₂₇ O ₃ N ₃ Pt S	DO
	680.1422	-1.54	26.0	C ₃₆ H ₂₃ ONPt	NNN N
	680.1400	1.61	33.0	C43 H24 O5 N2 S	
	680.1395	2.40	21.5	C ₃₃ H ₂₅ O ₄ Pt	
	680.1429	-2.56	16.5	C 30 H 29 O 4 Pt S	Pt(<i>bp</i> -8)
	680.1387	3.59	33.5	C41 H22 O4 N5 S	ι (<i>μ</i> ρ-ο)
	680.1382	4.37	22.0	C ₃₁ H ₂₃ O ₃ N ₃ Pt	
	680.1442	-4.53	21.5	C 31 H 25 N 4 Pt S	

Cartesian Coordinates of the Structures

Pt(*ppy*-1)_S₀

110 /= *			
С	-3.53863900	1.78471700	-0.71644000
С	-4.01847400	0.74328600	0.30010000
С	-4.78225700	-1.22522200	2.16696900
С	-5.23098100	0.83379800	0.99444000
С	-3.18901600	-0.36432600	0.55728400
С	-3.56171200	-1.32713700	1.50590600
С	-5.62358700	-0.14280100	1.90925800
Н	-5.88463600	1.68090500	0.82004600
Н	-2.88881600	-2.14699600	1.73366200
Н	-6.57408400	-0.04936200	2.42657500
Н	-5.06023700	-1.97875900	2.89836900
С	-2.03968900	1.97843500	-0.46019400
С	0.75872900	2.20223200	-0.13528800
С	-1.24721100	0.83073800	-0.24559100
С	-1.40222000	3.22703900	-0.46763200
С	-0.02236100	3.34925000	-0.29160100
С	0.13335900	0.92355300	-0.13344100
Н	-1.98725700	4.12748000	-0.61332300
Н	0.42813300	4.33857100	-0.28676600
С	-4.32642000	3.10167700	-0.61849900
Н	-5.38682100	2.93548300	-0.83275100
Н	-3.96634700	3.81955600	-1.36155400
Н	-4.24173900	3.55901300	0.37300300
С	-3.72979900	1.22040100	-2.15516500
Н	-4.79296300	1.03765000	-2.34969100
Н	-3.18865900	0.28093100	-2.29771700
Н	-3.35928200	1.93922900	-2.89451600
Ν	-1.91320300	-0.43226100	-0.09210500

S-**37** / S-**52**

С	2.22139800	2.16962800	0.05931900
С	4.94144000	1.59160000	0.40321600
С	3.14791100	3.21372700	0.12335500
С	4.50381400	2.91528600	0.30004200
С	3.98994500	0.57744500	0.33597900
Н	2.81778400	4.24307100	0.03201800
С	-1.38762200	-1.68833500	-0.42074700
С	-0.43889400	-4.24192700	-1.01961300
С	-2.27029300	-2.69321300	-0.87786400
Ν	-0.06009800	-1.96888900	-0.31729400
С	0.39082800	-3.21832400	-0.61062800
С	-1.80292800	-3.96043900	-1.17060100
Н	-3.31731800	-2.44793300	-1.00250500
Н	1.46195100	-3.34894300	-0.49578200
Н	-2.49085500	-4.71979100	-1.53093000
Н	-0.02629500	-5.22172100	-1.23113600
Pt	1.38622900	-0.56336600	0.03448100
Н	5.22357800	3.72773500	0.34930100
Н	5.98385400	1.32170100	0.52599200
Ν	2.69927900	0.90131600	0.18636000
С	4.28187700	-0.93389500	0.38648400
0	3.23915800	-1.70239400	0.25697000
0	5.44822700	-1.28005800	0.52992300
Pt(<i>ppy</i> -1)_ T ₁			
С	3.53422300	-1.77954600	-0.73707700
С	4.02734700	-0.74329500	0.27820000
С	4.79885100	1.21325100	2.15836200
С	5.24616700	-0.83424300	0.96189100
С	3.19928400	0.36127100	0.55214900
С	3.57426800	1.31833500	1.50608800

S-**38** / S-**52**

С	5.64113600	0.13535600	1.88283400
Н	5.90231900	-1.67634300	0.77319700
Н	2.89911300	2.13240700	1.74682300
Н	6.59649600	0.04123000	2.39107000
Н	5.07964700	1.95962500	2.89573200
С	2.04298700	-1.96074400	-0.44886700
С	-0.81237100	-2.23419600	-0.10132900
С	1.25291700	-0.82088300	-0.23966800
С	1.37550300	-3.22001000	-0.43460500
С	0.00852300	-3.37083200	-0.25746600
С	-0.14346300	-0.90969600	-0.11960300
Н	1.96743700	-4.11884800	-0.57194800
Н	-0.41909100	-4.36929900	-0.24499700
С	4.31478200	-3.10167000	-0.64194300
Н	5.37432400	-2.94475600	-0.86748200
Н	3.94260400	-3.81981100	-1.37910800
Н	4.23709100	-3.55525500	0.35207700
С	3.72475900	-1.21686000	-2.17634700
Н	4.78762000	-1.03873500	-2.37987400
Н	3.18775100	-0.27424500	-2.31676200
Н	3.34499300	-1.93296800	-2.91386200
N	1.90880500	0.43175700	-0.07283000
С	-2.22174500	-2.18869000	0.08026200
С	-4.95859000	-1.54347500	0.38989900
С	-3.19501100	-3.21912800	0.14235000
С	-4.53046000	-2.90385800	0.29974900
С	-4.00351900	-0.54955000	0.32835800
Н	-2.88148100	-4.25548500	0.05938800
С	1.38856500	1.69543200	-0.40183200
С	0.44068400	4.24183700	-0.99892800

S-**39** / S-**52**

С	2.27388600	2.69685500	-0.85287400
Ν	0.05763100	1.96512600	-0.30933100
С	-0.39195700	3.21946700	-0.59831900
С	1.80779500	3.96537800	-1.14859000
Н	3.32156800	2.45182100	-0.97123900
Н	-1.46361600	3.34982100	-0.48953700
Н	2.49503500	4.72586700	-1.50635600
Н	0.02757400	5.22204900	-1.20848600
Pt	-1.38850800	0.54571500	0.03015600
Н	-5.26782900	-3.69940400	0.34646500
Н	-6.00089500	-1.26799200	0.49870900
Ν	-2.70337600	-0.88151200	0.20375700
С	-4.25665500	0.95636300	0.36818300
0	-3.16786600	1.69881900	0.25385000
0	-5.39412100	1.38315000	0.48834400
Pt(<i>bp</i> -6) _ S ₀			
Рt(<i>bp</i>-6)_S 0 С	-4.23357600	1.29923200	-0.33079600
	-4.23357600 -4.37509700	1.29923200 0.12043000	-0.33079600 0.63702300
C	-4.37509700		
C C	-4.37509700 -4.52245600	0.12043000	0.63702300
c c c	-4.37509700 -4.52245600 -5.49056000	0.12043000 -2.13246800	0.63702300 2.32851300
C C C C	-4.37509700 -4.52245600 -5.49056000 -3.32128100	0.12043000 -2.13246800 -0.10147600	0.63702300 2.32851300 1.45334100
	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100	0.12043000 -2.13246800 -0.10147600 -0.80980100	0.63702300 2.32851300 1.45334100 0.69276100
	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300
	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900 -6.31372400	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100 -1.21861200	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300 2.28525700
С С С С С С Н	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900 -6.31372400 -2.54938600	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100 -1.21861200 0.60419200	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300 2.28525700 1.43943300
С С С С С С Н Н	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900 -6.31372400 -2.54938600 -6.45803900	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100 -1.21861200 0.60419200 -2.60344300	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300 2.28525700 1.43943300 1.60307000
С С С С С С Н Н Н	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900 -6.31372400 -2.54938600 -6.45803900 -4.56749400	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100 -1.21861200 0.60419200 -2.60344300 -1.36741200	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300 2.28525700 1.43943300 1.60307000 2.90177200 2.98695300
С С С С С С Н Н Н Н	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900 -6.31372400 -2.54938600 -6.45803900 -4.56749400 -2.77841400	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100 -1.21861200 0.60419200 -2.60344300 -1.36741200 -2.99559800	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300 2.28525700 1.43943300 1.60307000 2.90177200 2.98695300 -0.24211500
С С С С С С Н Н Н Н Н С	-4.37509700 -4.52245600 -5.49056000 -3.32128100 -3.39081100 -5.57596900 -6.31372400 -2.54938600 -6.45803900 -4.56749400 -2.77841400 -0.06460700	0.12043000 -2.13246800 -0.10147600 -0.80980100 -1.91977100 -1.21861200 0.60419200 -2.60344300 -1.36741200 -2.99559800 1.77909400	0.63702300 2.32851300 1.45334100 0.69276100 1.54626300 2.28525700 1.43943300 1.60307000 2.90177200 2.98695300 -0.24211500 -0.17638900

S-**40** / S-**52**

С	-2.41435200	3.12821500	-0.23649400
С	-1.07431400	3.52011000	-0.18903300
С	-0.40016700	1.18059100	-0.21749600
Н	-3.18055400	3.89510300	-0.25803200
Н	-0.83356400	4.57951100	-0.14682600
С	-5.24107400	2.42009800	-0.02402200
Н	-6.26870600	2.05436200	-0.11820000
Н	-5.13360300	3.23929500	-0.74099200
Н	-5.10827600	2.82557700	0.98459700
С	-4.51134400	0.79733900	-1.77872600
Н	-5.53622900	0.41556700	-1.86125700
Н	-3.82323800	-0.00125500	-2.06996300
Н	-4.38807300	1.62165400	-2.49012600
Ν	-2.14655900	-0.56589200	-0.07552900
С	1.38401700	2.81308600	-0.04773600
С	4.17269100	2.99665200	0.15234600
С	2.15339300	1.62202700	0.07025500
С	2.00773600	4.06223400	-0.04644100
С	3.40133000	4.15080200	0.05765000
С	3.52926500	1.75218100	0.15308200
Н	1.42031300	4.97204300	-0.14007200
С	-1.52035100	-1.63123900	-0.71787500
С	-0.33290000	-3.68823200	-2.15532700
С	-2.31521700	-2.65484100	-1.28810100
Ν	-0.16890300	-1.67441000	-0.82770400
С	0.38631000	-2.66391600	-1.57042600
С	-1.72516800	-3.68434500	-1.99469700
Н	-3.39156900	-2.59780700	-1.18513200
Н	1.46300500	-2.60689300	-1.68080200
Н	-2.34102200	-4.45650800	-2.44707200

S-**41** / S-**52**

Н	0.17759400	-4.45568300	-2.72668800
С	4.17267300	-0.62176400	0.45440700
С	3.92938700	-3.29932900	1.05436100
Ν	2.96228300	-1.20102900	0.33222000
С	5.31672200	-1.35994800	0.83151000
С	5.19724000	-2.70190200	1.12717600
С	2.86244700	-2.51614000	0.66209000
Н	6.06876500	-3.27686800	1.42644000
Н	1.86242900	-2.92831700	0.60263400
Н	3.77385300	-4.34372200	1.30139300
Pt	1.14002200	-0.06729800	-0.12867000
Н	3.88651300	5.12304700	0.05768300
Н	5.25534700	3.03941800	0.22307000
0	4.42305900	0.67002900	0.21813500
Н	6.26127400	-0.83220700	0.89183100
н Pt(<i>bp-</i> 6)_T ₁	6.26127400	-0.83220700	0.89183100
		-0.83220700 1.25765600	
Pt(<i>bp</i> -6)_T ₁	-4.22338300	1.25765600	
Рt(<i>bp-</i>6)_Т 1 С	-4.22338300	1.25765600	-0.40716300
Рt(<i>bp</i>-6)_T 1 С С	-4.22338300 -4.37411700 -4.52484100	1.25765600 0.11277500	-0.40716300 0.60121000
Pt(<i>bp</i>-6)_T 1 С С С	-4.22338300 -4.37411700 -4.52484100 -5.49170100	1.25765600 0.11277500 -2.07412200	-0.40716300 0.60121000 2.37592800
Pt(<i>bp</i>-6)_T 1 С С С	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900	1.25765600 0.11277500 -2.07412200 -0.07617300	-0.40716300 0.60121000 2.37592800 1.42399800
Pt(<i>bp</i> -6)_T ₁ C C C C	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300
Pt(<i>bp</i> -6)_T ₁ C C C C C	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100 -5.57776000	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900 -1.89225900	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300 1.58865600
Pt(<i>bp</i> -6)_T ₁ C C C C C C	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100 -5.57776000 -6.31536300	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900 -1.89225900 -1.16088400	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300 1.58865600 2.29691500
Pt(<i>bp</i>-6)_T 1 С С С С С С Н	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100 -5.57776000 -6.31536300 -2.55042200	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900 -1.89225900 -1.16088400 0.62781800	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300 1.58865600 2.29691500 1.38036000
Pt(<i>bp</i>-6)_T 1 С С С С С С Н Н	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100 -5.57776000 -6.31536300 -2.55042200 -6.46154300	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900 -1.89225900 -1.16088400 0.62781800 -2.57369700	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300 1.58865600 2.29691500 1.38036000 1.67010300 2.91603700
Pt(<i>bp</i>-6)_T 1 С С С С С С Н Н	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100 -5.57776000 -6.31536300 -2.55042200 -6.46154300 -4.57232300	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900 -1.89225900 -1.16088400 0.62781800 -2.57369700 -1.28745900	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300 1.58865600 2.29691500 1.38036000 1.67010300 2.91603700 3.06607300
<pre>Pt(bp-6)_T1 C C C C C C H H H<</pre>	-4.22338300 -4.37411700 -4.52484100 -5.49170100 -3.31975900 -3.39138100 -5.57776000 -6.31536300 -2.55042200 -6.46154300 -4.57232300 -2.77671400	1.25765600 0.11277500 -2.07412200 -0.07617300 -0.81386900 -1.89225900 -1.16088400 0.62781800 -2.57369700 -1.28745900 -2.91190300	-0.40716300 0.60121000 2.37592800 1.42399800 0.69613300 1.58865600 2.29691500 1.38036000 1.67010300 2.91603700 3.06607300 -0.28908200

S-**42** / S-**52**

С	-1.74361800	0.78425700	-0.21701500
С	-2.41242100	3.12974900	-0.28583100
С	-1.11055200	3.54747900	-0.23293800
С	-0.39940500	1.15237900	-0.21592900
Н	-3.19880400	3.87564700	-0.32274200
Н	-0.88323200	4.60936100	-0.21176800
С	-5.24489200	2.38116700	-0.15884100
Н	-6.26689700	2.00235200	-0.26100100
Н	-5.12991300	3.17646500	-0.90147600
Н	-5.13836200	2.82185300	0.83821800
С	-4.46943800	0.70731100	-1.84448700
Н	-5.49247200	0.32170300	-1.93561900
Н	-3.77496500	-0.09989000	-2.09239900
Н	-4.33234100	1.50719900	-2.58118900
N	-2.13916000	-0.58962100	-0.06928900
С	1.31530800	2.83327800	-0.00686100
С	4.12957700	3.01215200	0.18212300
С	2.13266000	1.61281500	0.06861800
С	1.95840900	4.11701300	0.06454700
С	3.32468100	4.18880800	0.15135300
С	3.51020100	1.75983600	0.13633300
Н	1.36431600	5.02544000	0.03186000
С	-1.50061400	-1.66941900	-0.68049300
С	-0.28309900	-3.75388600	-2.04451600
С	-2.27904700	-2.72111600	-1.20837800
Ν	-0.14387400	-1.68440700	-0.79543800
С	0.42224100	-2.69658600	-1.51105500
С	-1.67749200	-3.77313700	-1.87759700
Н	-3.35583300	-2.67486300	-1.10297100
Н	1.49543600	-2.62166000	-1.64065500

S-**43** / S-**52**

Н	-2.28179400	-4.57140000	-2.29771500
Н	0.23901200	-4.53179300	-2.59123400
С	4.17612100	-0.60582700	0.39977300
С	3.95149000	-3.27076100	1.03811100
Ν	2.95508500	-1.17816500	0.33264400
С	5.33138400	-1.33776400	0.73032100
С	5.22599900	-2.68081600	1.04483500
С	2.86816200	-2.49254800	0.68875200
Н	6.10833200	-3.25511300	1.31009600
Н	1.86563200	-2.90223700	0.68400600
Н	3.80307600	-4.31134400	1.30576900
Pt	1.13631400	-0.05652500	-0.12629300
Н	3.81793000	5.15625800	0.19557100
Н	5.21070100	3.07509600	0.24335300
0	4.40962300	0.68981800	0.14004000
Н	6.27762100	-0.80962800	0.74341500
Pt(<i>bp</i> -7)_S ₀			
С	-4.25611300	1.06317900	0.36138000
С	-6.19355600	-0.83478800	1.00410700
С	-5.55353900	1.48013400	0.67336100
С	-3.94640400	-0.32165900	0.33657700
С	-4.90502400	-1.27455300	0.69117600
С	-6.52270000	0.52705400	0.98104200
Н	-5.79560100	2.53925600	0.69176800
Н	-4.66430600	-2.32934100	0.75825100
Н	-7.53324300	0.84175800	1.22573900
Н	-6.94700900	-1.56726000	1.28009800
С	-3.02816000	1.78954900	0.10227000
С	-0.34706300	2.53706300	-0.21326700
0			
С	-2.00812700	0.82786300	-0.05798800

S-**44** / S-**52**

С	-2.69326900	3.14819700	0.05583400
С	-1.35530000	3.51194000	-0.10893600
С	-0.67527100	1.15870300	-0.20199500
Н	-3.45869000	3.91099600	0.17008500
Н	-1.09305300	4.56699300	-0.11356000
Ν	-2.57038700	-0.47894000	0.02545200
С	1.11035100	2.77075700	-0.25975500
С	3.90924500	2.88808800	-0.40081800
С	1.86907200	1.56881300	-0.15797300
С	1.75595300	3.99919800	-0.41181200
С	3.15406900	4.05431400	-0.48062300
С	3.24644800	1.66431100	-0.23942800
Н	1.17987900	4.91732300	-0.49524800
С	-1.97363300	-1.63421700	-0.47740400
С	-0.81098000	-3.92886200	-1.52200400
С	-2.77382900	-2.62517100	-1.08843800
Ν	-0.61999000	-1.77502900	-0.42782900
С	-0.08100500	-2.89684300	-0.96020000
С	-2.19794900	-3.77534100	-1.59591100
Н	-3.83460200	-2.45133400	-1.19965700
Н	1.00083600	-2.94265100	-0.93880200
Н	-2.81994800	-4.52734900	-2.07303800
Н	-0.30442400	-4.80231400	-1.91731900
С	3.88353800	-0.65543700	0.31480900
С	3.67279300	-3.14085000	1.48349500
Ν	2.65919400	-1.20055300	0.45392900
С	5.05292600	-1.34853100	0.69368200
С	4.94889300	-2.59733500	1.27291700
С	2.57716200	-2.41039000	1.06673300
Н	5.84149200	-3.13562600	1.57814300

S-**45** / S-**52**

Н	1.56962900	-2.77906400	1.21996700
Н	3.53301300	-4.10305400	1.96366300
Pt	0.82492100	-0.11187500	-0.07004300
Н	3.65481800	5.01079100	-0.60302000
Н	4.99318300	2.90644600	-0.46105100
0	4.11186800	0.55505600	-0.21018000
Н	6.00640200	-0.85925500	0.53292600

Pt(*bp***-**7**)**_**T**₁

С	4.23435200	1.06387400	-0.33636200
С	6.17583400	-0.83330700	-1.02496000
С	5.54731400	1.48443100	-0.64751600
С	3.93223400	-0.33932400	-0.33085900
С	4.88941400	-1.28088700	-0.70927800
С	6.50322400	0.53647700	-0.97578400
Н	5.79376800	2.54249300	-0.64514300
Н	4.64934900	-2.33442700	-0.79427900
Н	7.51384200	0.85223600	-1.22046500
Н	6.92821900	-1.55668200	-1.32504600
С	3.03249300	1.76437000	-0.06742200
С	0.32877200	2.56524100	0.22611900
С	2.00422400	0.80079900	0.08842900
С	2.68561500	3.16338200	-0.00039100
С	1.37925100	3.54267100	0.15822100
С	0.67700000	1.12843200	0.21391200
Н	3.46824400	3.91215100	-0.08639700
Н	1.12917500	4.59954700	0.19033700
Ν	2.56683400	-0.50081800	-0.00109800
С	-1.06477700	2.78873100	0.23276700
С	-3.88370000	2.89746900	0.33619500

С	-1.85015400	1.56150900	0.15681400
С	-1.73710700	4.04464200	0.32341900
С	-3.11384800	4.08476400	0.37768000
С	-3.23151800	1.66790500	0.22887300
Н	-1.16410600	4.96605100	0.37394300
С	1.95560800	-1.66531000	0.46374200
С	0.76531400	-3.97222400	1.44145800
С	2.74512000	-2.68746600	1.03088200
Ν	0.58997700	-1.77264200	0.42970400
С	0.04472500	-2.91261500	0.93403100
С	2.16256000	-3.84665300	1.50685200
Н	3.80917100	-2.52850100	1.13589700
Н	-1.03786500	-2.93945700	0.93693600
Н	2.77621900	-4.62042600	1.95801000
Н	0.24831200	-4.84957100	1.81442400
С	-3.87599100	-0.65100800	-0.29223500
С	-3.67270600	-3.12480600	-1.47631100
Ν	-2.64918100	-1.19055100	-0.44353700
С	-5.04643100	-1.33934500	-0.66014100
С	-4.94867200	-2.58820900	-1.24750800
С	-2.57210200	-2.39706900	-1.07010100
Н	-5.84330800	-3.12624300	-1.54605300
Н	-1.56626900	-2.76178200	-1.24049300
Н	-3.53581900	-4.08307400	-1.96538100
Pt	-0.81781500	-0.10471700	0.08873900
Н	-3.62725200	5.03913600	0.45780300
Н	-4.96724700	2.92866400	0.38347200
0	-4.09206100	0.56097800	0.24486000
Н	-5.99805700	-0.85078000	-0.48670900

S-**47** / S-**52**

· • / •			
С	-5.07770300	1.10717800	0.08530000
С	-7.09462400	-0.75960300	0.54530500
С	-6.39121100	1.54479500	0.27949400
С	-4.78824700	-0.28207000	0.08460700
С	-5.79034300	-1.21978800	0.34938000
С	-7.39928900	0.60728800	0.49539000
Н	-6.61679800	2.60772500	0.27846100
Н	-5.57472500	-2.27840300	0.43583200
Н	-8.42288600	0.93776000	0.64756400
Н	-7.88135300	-1.48040300	0.75002400
С	-3.82087300	1.81414900	-0.05959900
С	-1.11450100	2.52320700	-0.12505000
С	-2.80448900	0.83816500	-0.13765000
С	-3.46572900	3.16822400	-0.06404600
С	-2.11414000	3.51277200	-0.09943800
С	-1.45812400	1.15100400	-0.16958900
Н	-4.22825800	3.94074100	-0.01352700
Н	-1.83923800	4.56395200	-0.06705300
Ν	-3.39271700	-0.45812200	-0.10332300
С	0.34138400	2.74441900	-0.02591400
С	3.12100200	2.89425500	0.18200300
С	1.10230400	1.53988900	0.04408200
С	0.97130400	3.98874100	0.01518400
С	2.35899100	4.05712000	0.15324800
С	2.49798100	1.63566000	0.06170600
Н	0.38913600	4.90498100	-0.03666100
N	3.32233500	0.45380900	0.01337200
С	5.48297600	1.33278600	-1.01835200
С	6.61192300	-1.09281500	-0.07387000

С	4.69871600	0.41737400	-0.30497900
С	6.82720200	1.02838900	-1.22322500
С	7.39835200	-0.16195500	-0.74269000
С	5.25683900	-0.81071700	0.13313900
Н	7.44277400	1.72948100	-1.77995100
Н	8.45123600	-0.36394100	-0.91583400
С	-2.76408600	-1.63309800	-0.51031400
С	-1.54162200	-3.98102800	-1.34117700
С	-3.51116800	-2.62798700	-1.17895000
Ν	-1.42701500	-1.79130300	-0.30869500
С	-0.85699400	-2.94209700	-0.73615800
С	-2.90681400	-3.80501300	-1.58071800
Н	-4.54818200	-2.43776100	-1.41613100
Н	0.21330400	-3.00768800	-0.58878200
Н	-3.48638000	-4.56131900	-2.10242700
Н	-1.01254600	-4.87748500	-1.64494200
С	2.99492100	-0.79275600	0.52189100
С	2.78485000	-3.25879600	1.66639300
Ν	1.75966100	-1.25143900	0.77935600
С	4.17145800	-1.58013700	0.68774200
С	4.05771300	-2.83376800	1.27554000
С	1.68121800	-2.45583500	1.39031200
Н	4.93001100	-3.46213100	1.43223300
Н	0.67860900	-2.78047700	1.64515800
Н	2.63744000	-4.21619000	2.15444100
Pt	0.01458000	-0.12936800	0.09834400
Н	2.85345200	5.02057700	0.24321700
Н	5.06013300	2.23964100	-1.43041200
Н	4.19045700	2.96695700	0.33166200

Pt(*bp***-8)**_**T**₁

С	5.04608700	1.11961400	-0.08208400
С	7.07085300	-0.73886800	-0.61190500
С	6.37097100	1.56538100	-0.28366900
С	4.76932600	-0.28717400	-0.10419300
С	5.77186000	-1.21072400	-0.40433000
С	7.36836200	0.63615900	-0.53359100
Н	6.59652700	2.62783100	-0.25951500
Н	5.55962800	-2.26813000	-0.51108400
Н	8.38969100	0.97104700	-0.69353400
Н	7.85848500	-1.44763000	-0.85009300
С	3.81243000	1.79810500	0.09042200
С	1.08441900	2.54932500	0.16224100
С	2.79344000	0.81749200	0.16444300
С	3.43984800	3.18881000	0.12822000
С	2.11780500	3.54526800	0.17609900
С	1.45163300	1.12233200	0.19218300
Н	4.21401500	3.95092600	0.10665000
Н	1.84905200	4.59779500	0.18613000
Ν	3.38482800	-0.47214900	0.11267500
С	-0.30607100	2.75501900	0.02541700
С	-3.09941200	2.87763600	-0.24685600
С	-1.08955000	1.52318800	-0.04209100
С	-0.96056300	4.01668000	-0.06672100
С	-2.32511200	4.06179900	-0.22768600
С	-2.49024300	1.62593300	-0.08056800
Н	-0.38358800	4.93577800	-0.02511600
Ν	-3.31225000	0.45117400	-0.01328200
С	-5.48132800	1.35212500	0.97922200
С	-6.60425500	-1.08479800	0.05566800
С	-4.69376600	0.42422700	0.28957800

S-**50** / S-**52**

С	-6.83007300	1.05423500	1.17300600
С	-7.39605400	-0.14171000	0.70236700
С	-5.24584800	-0.80928400	-0.13986200
Н	-7.45097100	1.76438500	1.71192500
Н	-8.45149900	-0.33940200	0.86515200
С	2.75412000	-1.65437700	0.49922600
С	1.52378000	-4.00308800	1.30683900
С	3.50710800	-2.67419100	1.11804600
Ν	1.40135700	-1.78273300	0.33504400
С	0.83339800	-2.94447300	0.75622300
С	2.90483600	-3.85414300	1.51055700
Н	4.55270000	-2.49806400	1.32807500
Н	-0.24250900	-2.99245000	0.64917100
Н	3.48778400	-4.62772500	2.00113600
Н	0.99110600	-4.89890100	1.60658400
С	-2.98320200	-0.80185000	-0.50828100
С	-2.76042200	-3.27392900	-1.62657400
Ν	-1.74097800	-1.25537900	-0.75130300
С	-4.15564500	-1.58714100	-0.67440800
С	-4.03896500	-2.85140500	-1.24765400
С	-1.65908900	-2.46834400	-1.35940400
Н	-4.90890600	-3.48190400	-1.40554900
Н	-0.65637600	-2.79029000	-1.61491800
Н	-2.60791100	-4.23459600	-2.10706400
Pt	-0.01088600	-0.12751600	-0.07236300
Н	-2.83193800	5.01638000	-0.34148000
Н	-5.05989000	2.26522700	1.37987700
Н	-4.16392400	2.95134500	-0.42569300
Н	-7.02712800	-2.02874000	-0.27760400

S-**51** / S-**52**

References:

- Li, G.; Wolfe, A.; Brooks, J.; Zhu, Z.-Q.; Li, J. Modifying Emission Spectral Bandwidth of Phosphorescent Platinum(II) Complexes Through Synthetic Control. *Inorg. Chem.* 2017, 56, 8244–8256.
- Connelly, N. G.; Geiger, W. E. Chemical Redox Agents for Organometallic Chemistry. *Chem. Rev.* 1996, *96*, 877–910.
- (3) Harris, D. C. In Quantitative Chemical Analysis, 6th ed.; W. H. Freeman: New York, 2002; pp 394–396.
- (4) Li, G.; Chen, Q.; Zheng, J.; Wang, Q.; Zhan, F.; Lou, W.; Yang, Y.-F.; She, Y. Metal-assisted delayed fluorescent Pd(II) complexes and phosphorescent Pt(II) complex based on [1,2,4]triazolo[4,3-a]pyridine-containing ligands: synthesis, characterization, electrochemistry, photophysical studies, and application. *Inorg. Chem.* 2019, 58, 21, 14349–14360.
- (5) Li, G.; Zhan, F.; Zheng, J.; Yang, Y.-F.; Wang, Q.; Chen, Q.; Shen, G.; She, Y. Highly Efficient Phosphorescent Tetradentate Platinum(II) Complexes Containing Fused 6/5/6 Metallocycles. *Inorg. Chem.* 2020, 59, 3718–3729.
- (6) Zhao, X.; She, Y.; Fang, K.; Li, G. CuCl-catalyzed Ullmann-Type C–N Cross-Coupling Reaction of Carbazoles and 2-Bromopyridine Derivatives. *J. Org. Chem.* **2017**, *82*, 1024–1033.