Comprehensive Multiphase NMR - A Powerful Tool to Understand and Monitor Molecular Processes During Biofuel Production

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15 Supporting Information

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17 S1. NMR Experimental Parameters

Spectral Editing has been previously introduced and discussed in details by Courtier-Murias et al..1 As such 18 19 only key parameters are provided here. ¹H experiments were performed with presaturation for water 20 suppression. 90° pulses were calibrated for each sample and garp-4 decoupling was used during acquisition to remove ¹³C J-coupling. A standard inversion recovery experiment was also performed for each sample 21 22 to measure T_1 and the recycle delay was set to 3.5 s, which is ~5 times the measured T_1 value (note the 23 relaxation was fast due to the ¹³C enrichment). All experiments were performed with 256 scans and 16K 24 time domain points. Spectral width was set to 16 ppm. Spectral editing was performed with a bipolar pulse pair longitudinal encode-decode (BPPLED) sequence with encoding/decoding gradients of 1.8 ms at 50 25 gauss/cm and a diffusion time of 180 ms. ¹H spectral editing is shown in supporting information Figure S3 26 27 and more general editing details are provided in the main paper (section "Spectral Editing"). A line broadening of 0.3 Hz was used for the reference and IDE spectra while 5 Hz was used for other ¹H 28 29 experiments.

30 All ${}^{13}C$ experiments were recorded with a spectral width of 400 ppm. 90° pulses and T₁ were measured as 31 described for the ¹H experiments. 1944 (243*8) scans were recorded for all ¹³C observe experiments. This 32 value was chosen because the sideband suppression technique used in CP experiment (a modified version of TOSS termed TOSS.243)^{2,3} requires a multiple of 243 scans. For stepped decoupling experiments, 33 34 decoupling was performed by using high-power spinal64 and low-power waltz16 in succession as 35 previously described.⁴ Sideband suppression for stepped decoupling experiments was done through various speed magic angle spinning (VSMAS).^{4,5} For swollen fractions, spectral editing employed the same 36 37 BPPLED sequence as mentioned above with the same gradient strength and diffusion time. In solid-state experiments, a Carr-Purcell-Meiboom-Gill (CPMG) relaxation filter on ¹H (2 \times 15 µs echoes) was 38 39 employed prior to a fixed cross-polarization contact time of 1ms to attenuate ¹H magnetisation in crystalline 40 components. These values were found optimal for the differentiation of dynamic and crystalline solid 41 standards as discussed by Courtier-Murias *et al.*.¹ A line broadening of 15 Hz was used for all ¹³C observe 42 experiments except for CP and CP_T₂ where 25 Hz was used.

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44 S2. Quantitative Estimates of Different Phases

45 The distribution of biomass in each phase can provide vital insights into how much of the organic material 46 is accessible for solvent extraction (i.e. swellable) and how much is protected/inaccessible (i.e. nonswellable). The estimation of such distribution in each phase has been introduced previously,^{1,6} however, it 47 is rather complex and thus worth revisiting here. Readers should be reminded that this is not absolute 48 49 quantification; for example, it is challenging to define exactly where the solid domain stops and the gel 50 domain starts. This stated, if applied rigorously, the approaches here provide an estimation of the amount 51 of organic materials of each phase in a sample, and certainly allow relative comparison between samples in 52 a series. As examples, different feedstock species and changes in extraction treatment are discussed in the paper. The following section outlines the steps needed to differentiate solution, gels, rigid gels, dynamic 53 54 solids and true rigid solids by CMP-NMR. In this example, a whole, non-extracted intact C. reinhardtii 55 algae sample is used.

56 Step 1 - Estimating the Solvent Accessible/Swellable Fraction. First, the multiphase sample 57 needs to be divided into two fractions – swellable and non-swellable. To do so, the freeze-dried sample is 58 first placed into a 4 mm MAS rotor. The rotor contains inserts (in this case, a bottom insert made of zirconia 59 and an extended-length Kel-F top plug), both designed to maintain the sample within the lower section of the coil. Only a relatively small amount of each sample (~10 mg) is used which only fills the 20% of the 60 coil region (see Figure S1a). Next, a cross-polarization (CP) experiment is performed on the dry sample 61 62 (Figure S1c). Cross polarization is highly efficient when strong dipoles form between ¹H and ¹³C nuclei in 63 the sample, and in the dry state, all components would undergo efficient cross polarization.¹ Next, an excess 64 of $D_2O(30 \,\mu\text{L})$ is added to the same rotor to swell the sample and it is thoroughly mixed and allowed to swell for 12 hrs. Great care is taken to ensure all the sample stays within the coil region (see Figure S1b). 65

66 The extended screw plug is now replaced with a normal length screw plug to account for the volume increase during swelling. Next, the CP experiment is performed again (see Figure S1d). For the swollen 67 components, bonds become more dynamic, and in turn, this modulates the ${}^{1}H{}^{-13}C$ dipoles, decreasing the 68 69 efficiency of cross polarization. For example, it was shown in a multiphase sample, CP does not occur in solutions, is inefficient in swollen gels, and retains normal efficiency for true solids.^{1,7} As such, the 70 71 reduction of signal between the dry CP experiment (everything) and the swollen CP experiment (dominated 72 by solid components) gives an estimation as to the percentage of organic matter that interacts with the 73 solvent. In this particular C. reinhardtii sample, the signal is decreased by 36.7% when the sample becomes 74 swollen (Figure S1e). This indicates ~37% of the sample is swellable while 63% has no interaction with the solvent (i.e. non-swellable).⁶ Again, it is important to understand these values are best estimates and 75 76 should not be taken as absolute values; however, these values are valid for comparison to estimate relative 77 change in phases across a series of samples.

78 Step 2 - Estimating the Subfractions within the Non-swellable Fraction. The second step is to 79 provide an estimation of the subfractions (dynamic solids and true solids) within the non-swellable fraction 80 (Figure S2). Components with varying degrees of local mobility and dynamics have different spin-spin 81 relaxation behavior (T₂). A very short T₂ filter $(2 \times 15 \,\mu s \text{ echoes})^1$ is employed on the ¹H channel prior to 82 cross polarization such that the signal with extremely broad ¹H profiles (i.e. true rigid solids) relaxes, leaving signals only from components that are solid but have local mobility (Figure S2b).⁶ While not 83 84 directly relevant to this study, to give the readers some context, rubber is an example of a dynamic solid, where the material is a solid but at the molecular level the bonds are flexible and dynamic.⁷ As such, the 85 86 signal loss observed between standard CP and CP with a T_2 filter (CP $_T_2$) is proportional to the amount of 87 true rigid solids in the sample. The spectrum for the dynamic solids is represented by the CP_{T_2} experiment (Figure S2b) while the rigid solids spectrum (Figure S2c) is created by difference (normal CP minus CP_T₂, 88 "CP Inv T_2 ") which is essentially the rigid solids.^{1,6} 89

90 Step 3 - Estimating the Subfractions within the Swellable Fraction. The last step is to 91 differentiate the contribution of components in the swellable fraction. Because of strong ¹H-¹H dipolar 92 interactions in solids, ¹H spectra give very broad lineshape for true solids which can be many KHz wide. As such, solids cannot be detected using ¹H NMR without the use of multiple approaches such as combined 93 rotation and multiple-pulse NMR spectroscopy (CRAMPS).8 Thus, when conventional low power "solution 94 95 state" ¹H NMR experiments are performed, only the dynamic fractions with at least some interaction with the solvent (i.e. gels and solution) are observed.^{1,9} This swellable fraction is further broken down into 3 96 97 subfractions or domains: swollen solids/rigid gels (Recovery Arising from Diffusion Editing, RADE), 98 restricted diffusion such as dynamic gels and macromolecules (Diffusion Editing, DE), and soluble and 99 mobile components (Inverse Diffusion Editing, IDE). Estimations as to the contribution of each subfraction 100 can be performed though it is not obvious.

101 The RADE spectrum represents the signals lost when the delays required for diffusion editing are 102 set. These components are the fastest relaxing components in the swellable fraction and could be attenuated 103 if diffusion editing is used alone. These components relax quite rapidly (thus have some "solid-like" 104 character) but have enough dynamics to be narrow enough for easy ¹H NMR detection. This fraction is 105 recovered by subtracting the delays-on spectrum from the reference spectrum (delays off, gradients off) as 106 shown in Figure S3A. The contribution from this domain is easy to estimate as it simply is represented by 107 the intensity lost when the delays are set (Figure S3A-3).

Obtaining the contribution from soluble (IDE) and restricted diffusion (DE, dynamic gel/macromolecules) domains is slightly more challenging. As seen from above, the "delays-on" spectrum equals Reference minus RADE. In turn this means the "delays-on" spectrum is the sum of the remaining domains i.e. restricted diffusion (DE, dynamic gel/macromolecules Figure S3B-2) and the soluble components (IDE) (Figure S3B-1). The components with restricted diffusion (dynamic gel/macromolecules) are easily obtained as these are the components detected during the diffusion editing experiment (Figure S3B-2). However, in addition to suppressing the soluble molecules, the diffusion gradient also supresses
the signal from large components to some extent. As such, the weighting of DE cannot be estimated directly
from its signal intensity and must be given additional considerations.

117 The soluble components are selected via a weighted subtraction of the diffusion editing result from 118 the "delays-on" spectrum (Figure S3A-2) but again, as the DE spectrum is used in the weighting, it is not 119 possible to take the signal intensity directly from IDE. However, the relative contributions of IDE and DE 120 to the mixture can be obtained by weighting their relative contribution until their sum best matches the 121 "delays-on" spectrum. Figure S3 shows the concept and that when the weighting of DE and IDE is correct, 122 the "weighted sum" spectrum (Figure S3B-3) equals the "delays-on" spectrum (Figure S3B-4). The correct 123 weighting of the DE and IDE spectrum can then be used estimate is contribution to the swellable fraction. The result is that contributions of 3 swellable domains can be estimated. When combined with the non-124 swellable fraction, all 5 domains can be combined to show their approximate contributions to the mixture 125 126 as whole (see Figure S4).

Interestingly, Ning et al.⁴ introduced a ¹³C-detect experiment based on stepped decoupling 127 128 technique that allows all phases to be detected equally and quantitatively. As such, while stepped 129 decoupling cannot be used for editing or phase differentiation, it does provide a quantitative overview of 130 the whole sample. Combining editing steps described above, the percentage of biomass distributed in each 131 of the five domains (true solid (Figure 1f), dynamic solid (Figure 1e), swollen solid/rigid gel (Figure 1d), 132 gel (Figure 1c), and solution (Figure 1b)) can be estimated. To validate the estimates are representative, the 133 13 C spectra of the above domains were summed together with their appropriate weightings (Figure S5) and 134 compared to the quantitative spectrum. The summed spectrum (Figure S5f) matches the stepped decoupling 135 spectrum quite well (Figure S5g), confirming that the weightings estimated based on the above methods 136 are relative accurate and there is no significant exaggeration or discrimination in any domains (solids, 137 solution etc.).



Figure S1 Differentiation of swellable and non-swellable fractions by CP experiments on intact C. reinhardtii. Spectrum (c) is obtained from freeze-dried algae sample (a) and the spectrum represents all carbon in the sample. Spectrum (d) is obtained when sample (a) is swollen with excess amount of water (b). Because CP experiments detect based on strong dipolar interaction (rigid, solid structures), spectrum (d) represents the remaining solid fraction after swelling (non-swellable fraction). The difference between the percentage represented by spectra (c) and (d) is the amount of carbon whose signal is lost after swelling (i.e., swellable fraction (e)).



Figure S2 Differentiation of domains within the non-swellable fraction. By deploying a T_2 relaxation filter prior to CP (CP_ T_2), signals from fast-relaxing components (rigid solids) are filtered out, leaving only slower-relaxing, dynamic solids (middle). Conversely, subtracting this spectrum from the CP spectrum of the non-swellable domain (left) yields the signals from the rigid solid domain (right spectrum, CP inverse T_2 filter).



Figure S3 Differentiation of domains within the swellable fraction using ¹H diffusion editing. Delays associated with this technique would filter out fast-relaxing components (rigid gels). Subtracting the delays-on spectrum (A-2) from the reference spectrum (A-1) gives the signals from the fast-relaxing components (spectrum A-3, recovery arising from diffusion editing (RADE)). With appropriate weightings of the mobile components (B-1) and gel components (B-2), the sum of these two spectra shows an identical spectrum as the delays-on spectrum that contains signals from gel and mobile components.



Figure S4 Differentiating components in all phases. Combining the swellable and non-swellable fractions, information on the distribution of different chemical components in the multiphase sample can be extracted with the help of CMP-NMR.

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Figure S5 Comparison between the sum of signals from all domains (a - e) and a reference quantitative stepped decoupling spectrum. With appropriate weighting for spectra a - e, the sum resembles the quantitative spectrum, indicating an accurate weighting of different phases.

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Figure S6 CP spectra of model biopolymers to represent major groups within the biomass – lipids (aliphatic rich), carbohydrates, and protein. The model compounds, cutin, xylan, and bovine serum albumin, showcase the key spectral features, such as the alpha carbon, and can help identifying corresponding groups within the algae extract. Note these spectra were collected as dry solids and thus the lineshape is broader than most algal materials in this paper, the majority of which were swollen to some extent during the experiment.

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