Supporting Information

Half-Sandwich Iridium Complexes for One-Pot Synthesis of Amides: Preparation, Structure and Diverse Catalytic Activity

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General information

All reagents obtained from Admas-beta were used without further purification. ¹H NMR and ¹³C spectra were recorded on Bruker AVANCE III 500 MHz (500 MHz for proton) spectrometer with tetramethylsilane as the internal reference using CDCl₃ as solvent in all cases, and chemical shifts were reported in parts per million (ppm, δ). FT-IR spectra were recorded on a Thermo fisher Nicolet 6700. GC analyses were performed on Shimadzu GC-2014 with a flame ionization detector equipped with an Rtx-1 capillary column (internal diameter = 0.25 mm, length = 30 m) or a Stabilwax capillary column (internal diameter = 0.25 mm, length = 30 m). GC mass spectra were recorded on Shimadzu GCMS-QP2010 with a capillary column (0.25 mm× 30 m). Column chromatography was performed using 200-300 mesh base-washed silica gel.

Characterizations of typical products

5a: ¹H NMR (500 MHz, CDCl₃): δ 7.78–7.74 (m, 2H), 7.50 (t, J = 7.5 Hz, 1H), 7.48 (t, J = 7.5 Hz, 2H), 5.83 (brs, 2H) ppm. Data are in accordance with that previously published.¹⁻⁵

5b: ¹H NMR (500 MHz, CDCl₃): δ 7.45 (d, J = 7.5 Hz, 1H), 7.35 (d, J = 7.5 Hz 1H), 7.25–7.20 (m, 2H), 6.45 (brs, 1H), 6,10 (brs, 1H), 2.45 (s, 3H) ppm. Data are in accordance with that previously published.¹⁻⁵

5c: ¹H NMR (500 MHz, CDCl₃): δ 8.13 (s, 1H), 7.61 (s, 1H), 7.43 (s, 1H), 7.38 (d, J = 8.0 Hz, 1H), 7.31 (d, J = 7.5 Hz, 1H), 7.20 (d, J = 7.5 Hz, 1H), 2.36 (s, 3H) ppm. Data are in accordance with that previously published.¹⁻⁵

5d: ¹H NMR (500 MHz, CDCl₃): δ 8.10 (s, 2H), 7.52 (d, J = 7.5 Hz, 2H), 6.93 (d, J = 7.5 Hz, 2H), 3.84 (s, 3H) ppm. Data are in accordance with that previously published.¹⁻⁵

5e: ¹H NMR (500 MHz, CDCl₃): δ 7.95 (s, 1H), 7.62 (s, 1H), 7.48–7.45 (m, 4H). ppm. Data are in accordance with that previously published.¹⁻⁵

5f: ¹H NMR (500 MHz, CDCl₃): δ 8.15 (s, 1H), 8.02 (d, J = 6.5 Hz, 1H), 7.85 (d, J = 7.0 Hz, 1H), 7.73–7.69 (m, 1H), 7.55 (s, 1H), 7.40 (t, J = 7.5 Hz, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5g: ¹H NMR (500 MHz, CDCl₃): δ 8.05 (s, 1H), 7.82 (d, J = 8.0 Hz, 2H), 7.62 (d, J = 8.0 Hz, 2H,), 7.47 (s, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5h: ¹H NMR (500 MHz, CDCl₃): δ 8.00 (s, 1H), 7.89 (d, J = 7.5 Hz, 2H), 7.60 (d, J = 7.5 Hz, 2H,), 7.41 (s, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5i: ¹H NMR (500 MHz, CDCl₃): δ 7.81 (d, J = 7.5 Hz, 2H), 7.53 (s, 1H), 7.03 (s, 1H), 6.99 (d, J = 7.5 Hz, 2H) 5.46 (s, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5j: ¹H NMR (500 MHz, CDCl₃): δ 7.71 (d, J = 8.0 Hz, 2H), 7.61 (s, 1H), 6.98 (s, 1H), 6.64 (d, J = 7.5 Hz, 2H), 2.90 (s, 6H) ppm. Data are in accordance with that previously published.¹⁻⁵

5k: ¹H NMR (500 MHz, CDCl₃): δ 8.25 (d, J = 7.5 Hz, 2H), 8.23 (s, 1H), 7.98 (s, 1H), 7.79 (d, J = 7.0 Hz, 2H) ppm. Data are in accordance with that previously published.¹⁻⁵

51: ¹H NMR (500 MHz, CDCl₃): δ 8.65–8.61 (m, 1H), 8.12 (brs, 1H), 8.02–7.97 (m, 2H), 7.63 (brs, 1H), 7.62–7.58 (m, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5m: ¹H NMR (500 MHz, CDCl₃): δ 7.82 (s, 1H), 7.78 (d, J = 5.0 Hz, 1H), 7.36 (s, 1H), 7.12 (d, J = 7.0 Hz, 1H), 6.56 (dd, J = 7.0, 3.0 Hz, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5n: ¹H NMR (500 MHz, CDCl₃): δ 7.95 (s, 1H), 7.79–7.73 (m, 2H), 7.39 (s, 1H), 7.12 (t, J = 7.0 Hz, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

50: ¹H NMR (500 MHz, CDCl₃): δ 6.16 (s, 1H), 5.90 (s, 1H), 2.09 (s, 3H) ppm. Data are in

accordance with that previously published.¹⁻⁵

5p: ¹H NMR (500 MHz, CDCl₃): δ 5.67 (brs, 1H), 5.48 (brs, 1H), 2.17 (t, J = 10.5 Hz, 1H), 1.87 (d, J = 12.0 Hz, 2H), 1.75 (d, J = 10.0 Hz, 2H), 1.65 (d, J = 10.5 Hz, 1H), 1.43 (q, J = 12.0 Hz, 2H), 1.23 (m, 3H) ppm. Data are in accordance with that previously published.¹⁻⁵

5q: ¹H NMR (500 MHz, CDCl₃): δ 7.95 (s, 1H), 7.76 (d, J = 7.5 Hz, 2H), 7.29 (s, 1H), 7.22 (d, J = 7.0 Hz, 2H), 2.33 (s, 3H) ppm. Data are in accordance with that previously published.¹⁻⁵

5r: ¹H NMR (500 MHz, CDCl₃): δ 9.12 (d, J = 7.0 Hz, 1H), 8.73 (s, 1H), 7.95 (d, J = 7.0 Hz, 1H), 7.83 (dd, J = 7.0 Hz, J = 4.0 Hz, 1H), 7.82 (d, J = 7.5 Hz, 1H), 7.59–7.55 (m, 1H) 7.51–7.45 (m, 2H,), 7.12 (s, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5s: ¹H NMR (500 MHz, CDCl₃): δ 10.05 (s, 1H), 8.17 (s, 1H), 8.03 (d, J = 7.0 Hz, 2H), 7.96 (d, J = 7.0 Hz, 2H), 7.62 (s, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

5u: ¹H NMR (500 MHz, CDCl₃): δ 8.11 (s, 1H), 7.95 (s, 2H), 7.51 (s, 1H) ppm. Data are in accordance with that previously published.¹⁻⁵

References

(1) Ambreen, N.; Wirth, T. High-Temperature Synthesis of Amides from Alcohols or Aldehydes by Using Flow Chemistry. *Eur. J. Org. Chem.* **2014**, 7590–7593.

(2) Zhang, S.; Xu, H.; Lou, C.; Senan, A. M.; Chen, Z.; Yin, G. Efficient Bimetallic Catalysis of Nitrile Hydration to Amides with a Simple Pd(OAc)₂/Lewis Acid Catalyst at Ambient Temperature. *Eur. J. Org. Chem.* **2017**, *2017*, 1870–1875.

- (3) Crestani, M. G.; García, J. J. Catalytic hydration of mono and dinitriles using nickel(0) and PTSA. *J. Mol. Catal. A. Chem.* **2009**, *299*, 26–36.
- (4) Guo, B.; deVaries, J. G.; Otten, E. Hydration of nitriles using a metal-ligand cooperative ruthenium pincer catalyst. *Chem. Sci.* **2019**, *10*, 10647–10652.
- (5) Lee, J.; Kim, M.; Chang, S.; Lee, H. -Y. Anhydrous Hydration of Nitriles to Amides using

Aldoximes as the Water Source. Org. Lett. 2009, 11, 5598–5601.

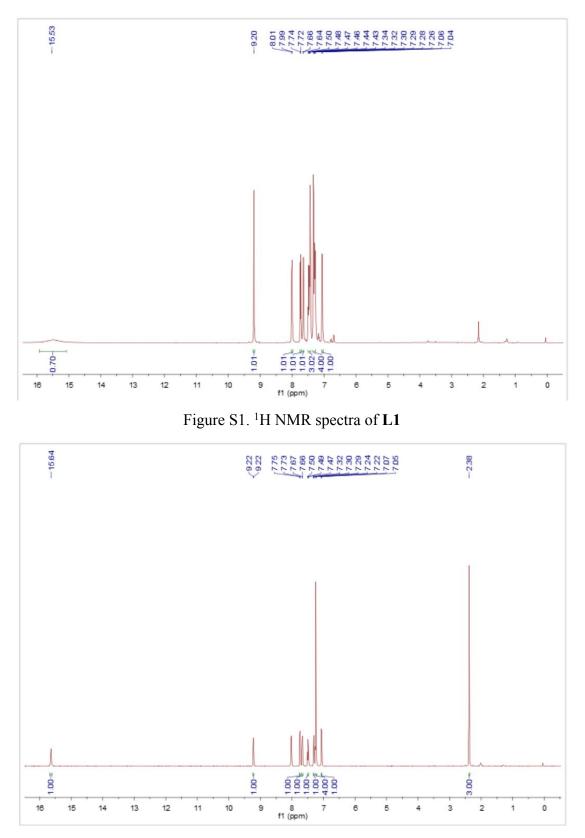


Figure S2. ¹H NMR spectra of L2

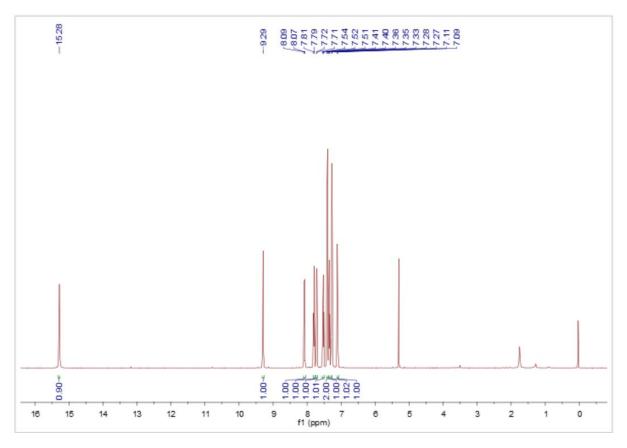
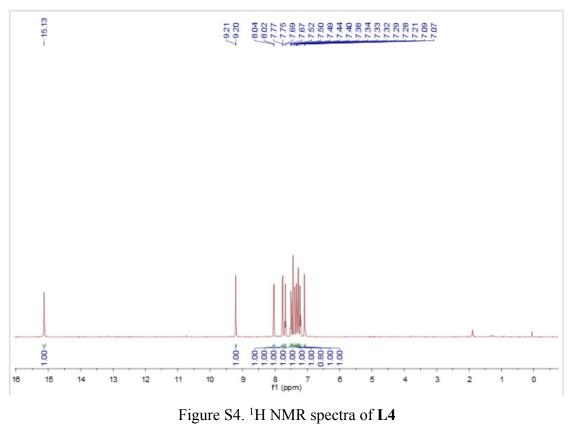


Figure S3. ¹H NMR spectra of L3



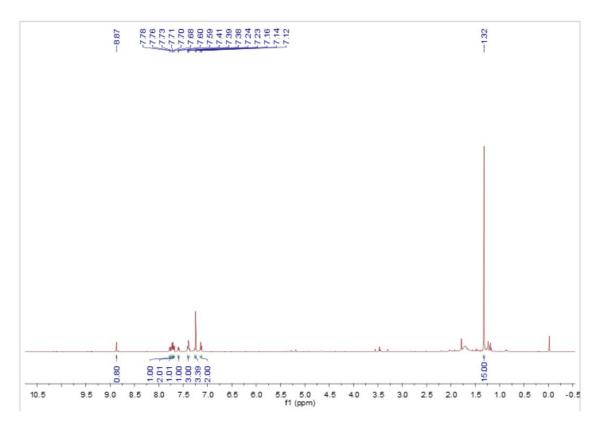


Figure S5. ¹H NMR spectra of **1**

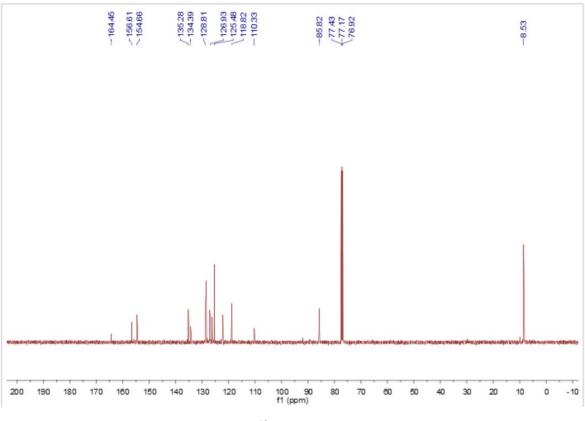


Figure S6. ¹³C NMR spectra of 1

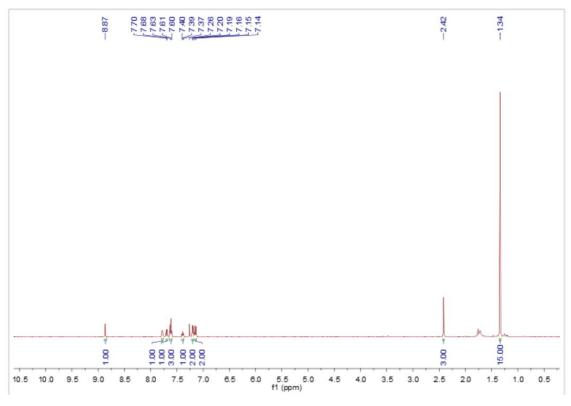


Figure S7. ¹H NMR spectra of 2

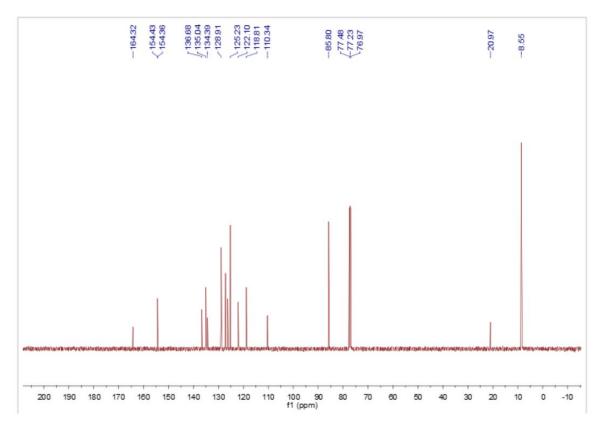


Figure S8. ¹³C NMR spectra of 2

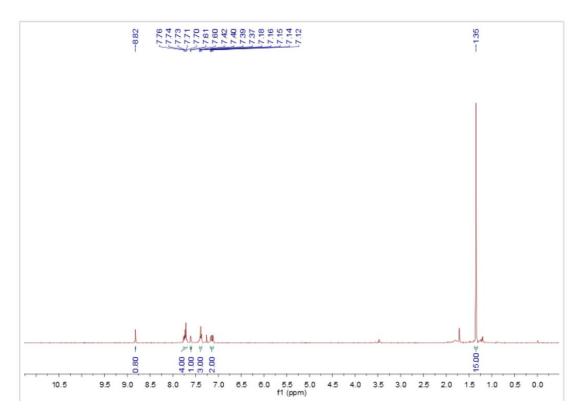


Figure S9. ¹H NMR spectra of **3**

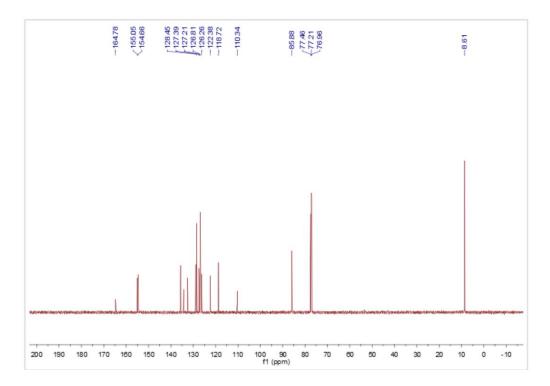


Figure S10. ¹³C NMR spectra of **3**

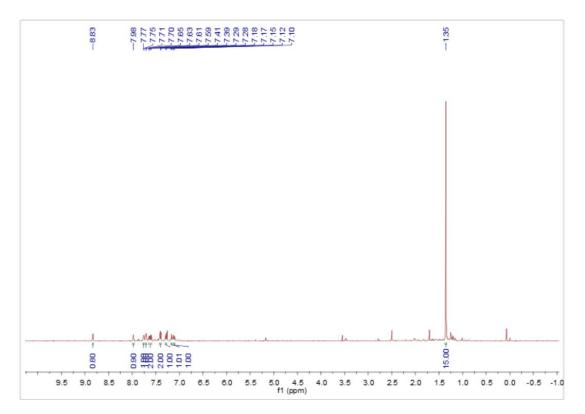


Figure S11. ¹H NMR spectra of 4

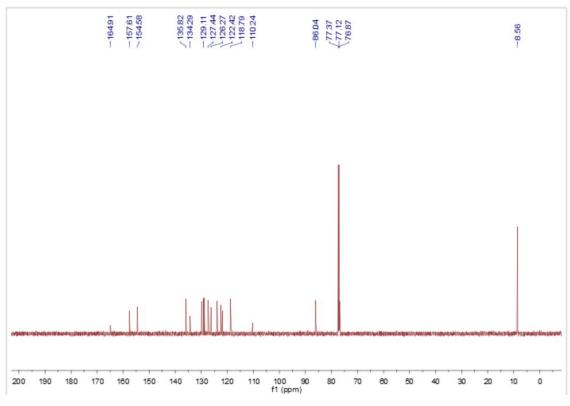


Figure S12. ¹³C NMR spectra of 4

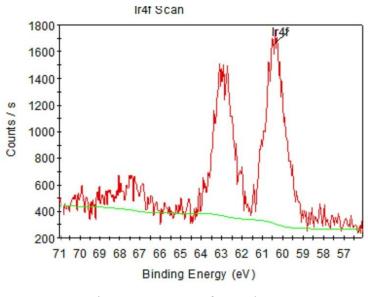


Figure S13. XPS of complex 1