Supporting Information

Ni₃N-Coated Ni Nanorod Arrays for Hydrogen and Oxygen Evolution in Electrochemical Water Splitting

Hongmei Wang,^a Jie Xiong,^a Xing Cheng,^b Mathias Fritz,^c Adriana Ispas,^d Andreas Bund,^c Ge Chen,^{b*} Dong Wang,^{a*} and Peter Schaaf^a

^{*a*} Fachgebiet Werkstoffe der Elektrotechnik, Institut für Werkstofftechnik und Institut für Mikro- und Nanotechnologien MacroNano®, TU Ilmenau, Gustav-Kirchhoff-Str. 5, 98693 Ilmenau, Germany

^b Beijing Key Laboratory for Green Catalysis Separation, Department of Chemistry and Chemical Engineering, College of Environmental and Energy Engineering, Beijing University of Technology, No. 100, Ping Le Yuan, Chaoyang District, Beijing 100124, P. R. China

^c Fachgebiet Elektrochemie und Galvanotechnik, Institut für Werkstofftechnik und Institut für Mikround Nanotechnologien MacroNano®, TU Ilmenau, Gustav-Kirchhoff-Str. 6, 98693 Ilmenau, Germany

^d Fachgebiet Chemie, Institut für Chemie und Biotechnik und Institut für Mikro- und Nanotechnologien MacroNano®, TU Ilmenau, Weimarer Str. 25, 98693 Ilmenau, Germany

*Corresponding authors

Dr. Dong Wang: dong.wang@tu-ilmenau.de

Prof. Ge Chen: chenge@bjut.edu.cn



Figure S1. Cross section SEM image of the nanostructured Si template.



Figure S2. Photographs of (a) Areas with deposited nickel on the Si wafer (4 inches) with nanostructured surface after about half an hour electroplating and (b) the finished Ni-electroplating layer.



Figure S3. Histogram plot for the diameter of nanorod arrays from the SEM image.



Figure S4. SEM images after electrochemical experiments: (a) B-Ni and (b) B-Ni_N2_30 min.



Figure S5. XRD patterns of P-Ni, B-Ni_N2_10 min and B-Ni_N2_30 min after electrochemical measurements.



Figure S6. High resolution spectra of Ni 2p for (a) B-Ni_N₂_30 min; (b)B-Ni; (c)P-Ni.



Figure S7. High resolution spectra of O 1s for (a) $B-Ni_N_2_30$ min; (b) B-Ni; (c) P-Ni. (d) High resolution spectrum of N 1s for $B-Ni_N_2_30$ min.



Figure S8. (a) Nyquist plots of as-prepared catalysts at -0.05 V versus RHE for HER, the inset is the equivalent circuit diagram and whole range plots. (b) Nyquist plots of as-prepared catalysts at 1.55 V versus RHE for OER, the inset is the equivalent circuit diagram.



Figure S9. Cyclic voltammetry measurements of (a) P-Ni, (b) B-Ni, (c) B-Ni_N₂_10 min and (d) B-Ni_N₂_30 min at different scan rates.

Electrode	HER		OER		
	$R_{S}(\Omega)$	$\operatorname{Ret}\left(\Omega\right)$	$R_{S}(\Omega)$	Rct (Ω)	
P-Ni	2.34	1152	2.06	63.75	
B-Ni	1.75	139.6	1.64	28.17	
B-Ni_N ₂ _10 min	1.56	119.3	1.45	22.13	
B-Ni_N ₂ _10 min	1.34	86.43	1.23	21.05	

Table S1 Fitted impedance parameters for the electrodes of HER and OER.

Table S2 Comparison of the electrocatalytic performance of Nickel-based electrocatalysts under alkaline conditions.

	Electrolysi	Electrolyte	Current	Overpotenti	Tafel	
Catalyst	s test		density (j)	al at j (mV)	slope	Ref.
			$(mA cm^{-2})$		(mV dec-	
					1)	
B-Ni_N ₂ _10min	HER	1 M KOH	10	161	189	This
	OER		10	350	67	work
NiNS	HER	1 M KOH	100	197	58.8	(7)
	OER		100	404	112	
Cu ₁ Ni ₂ -N	HER	1 M KOH	10	71.4	106.5	(15)
	OER		20	312	N.A.	
Fe ₂ Ni ₂ N	HER	1 M KOH	10	110	101	(46)
nanoarrays	OER		10	200	34	
FeNi ₃ N	HER	1 M KOH	10	75	98	(47)
	OER		10	202	40	
Co-Ni ₃ N	HER	1 M KOH	10	194	156	(48)
	OER		10	307	57	
NF@Ni/C-600	HER	1 M KOH	10	37	57	(49)
	OER		10	265	54	