

1                    **Supporting Information**

2                    **Using ESI FT-ICR MS to Characterize Dissolved Organic  
3                    Matter in Salt Lakes with Different Salinity**

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18     **1. Experimental Methods**

19     **Sample extraction for the FT-ICR MS measurement.** 500 mL filtered water  
20     samples were acidified with hydrochloric acid (p.a. grade, Merck, Germany) to pH 2  
21     and pumped at 5 mL/min through Bond Elut-PPL solid phase extraction cartridges  
22     (500 mg, 6 mL, Agilent, USA). The cartridges were previously rinsed with methanol  
23     and acidified (pH 2) Milli-Q water. To achieve complete salt removal, the cartridges  
24     were rinsed with 50 mL acidified Milli-Q water before elution. Upon the completion  
25     of extraction, DOM water samples were eluted with 10 mL LC-MS grade methanol  
26     immediately. The eluted samples were stored at -20 °C in the dark before MS  
27     analysis, which would be completed within 48 h.

28     **Microbial Community Composition. (1) DNA extraction and PCR amplification.**

29     Water samples from QHL and DHL were filtered onto a 0.2 µm Supor filter and  
30     frozen at -80 °C. DNA extraction was done using the Water DNA Kit (Omega  
31     Bio-tek, U.S.) according to manufacturer's protocols. PCR amplification was carried  
32     out by using the primers (338F 5'-ACTCCTACGGGAGGCAGCAG-3' and 806R  
33     5'-GGACTACHVGGGTWTCTAAT-3') in V3-V4 hypervariable regions of the  
34     bacteria 16S rRNA gene. The PCR reactions were performed under the following  
35     protocol: 3 min at 95 °C (denaturation), 27 cycles of 30 s at 95 °C, 30 s at 55 °C  
36     (annealing), and 45 s at 72 °C (elongation), and 10 min at 72 °C (final extension).  
37     Triplicate PCR reactions were pooled for each sample, and the amplification products  
38     were further purified. **(2) Sequencing analysis.** The purified amplicons were  
39     sequenced by Majorbio (Shanghai, China). The raw reads were uploaded to the NCBI

40 Sequence Read Archive database (Accession Number: SRP265086). Clustering of  
41 operational taxonomic units (OTUs) at 97% similarity was done using UPARSE  
42 (version7.1). Removal of chimeric sequences was done using UCHIME. Sequence  
43 taxonomy was analyzed by Ribosomal Database Project Classifier algorithm using the  
44 Silva (SSU123) 16S rRNA database.

45 **2. FT-ICR MS Data Processing**

46 **Van Krevelen diagrams boundary regions.** In this study, we divided the van  
47 Krevelen diagrams into regions corresponding to seven compound classes: lipids,  
48 aliphatic/proteins, lignins/carboxylic rich alicyclic molecules (CRAM)-like structures,  
49 carbohydrates, unsaturated hydrocarbons, aromatic structures, and tannins, according  
50 to previously published reports<sup>1, 2</sup>. The stoichiometric ranges used to establish the  
51 classification boundaries follow:

- 52 • lipids, H/C = 1.5–2.0, O/C = 0–0.3
- 53 • aliphatic/proteins, H/C = 1.5–2.2, O/C = 0.3–0.67
- 54 • lignins/CRAM-like structures, H/C = 0.7–1.5, O/C = 0.1–0.67
- 55 • carbohydrates, H/C = 1.5–2.4, O/C = 0.67–1.2
- 56 • unsaturated hydrocarbons, H/C = 0.7–1.5, O/C = 0–0.1
- 57 • aromatic structures, H/C = 0.2–0.7, O/C = 0–0.67
- 58 • tannins, H/C = 0.6–1.5, O/C = 0.67–1.0

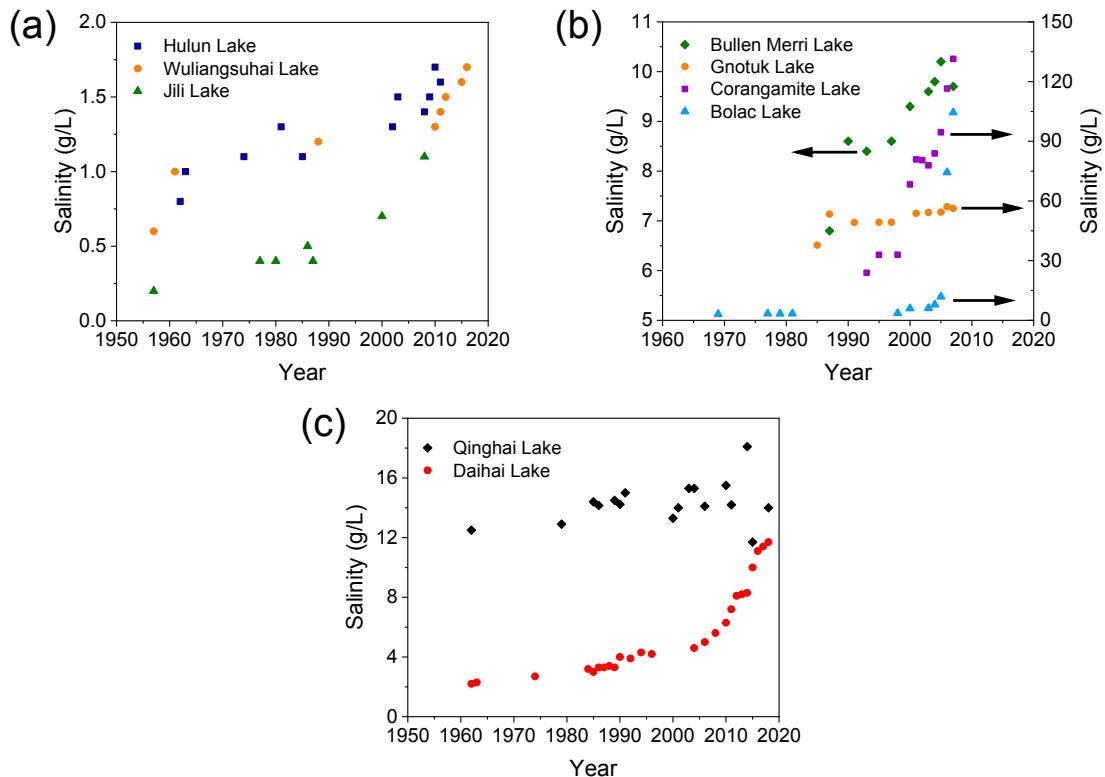
59 **Equations for average molecular mass, weighted averaged parameters, and**  
60 **aromaticity index (AI).** Molecular formulas were used to calculate the weighted

61 averaged values<sup>3, 4</sup> and aromaticity index (AI)<sup>5, 6</sup> for molecular composition with the  
62 equations below, where C, H, O, N, and S refer to the stoichiometric number of  
63 carbon, hydrogen, oxygen, nitrogen, and sulfur atoms in each formula, respectively.  
64 Relative magnitude ( $M_i$ ) is calculated by dividing the peak magnitude ( $I_i$ ) by the  
65 summed total peak magnitude for each sample.

- 66 •  $M_i = I_i / \sum I_i$
- 67 • Average molecular mass =  $\sum m/z_i \times M_i$
- 68 •  $X/C = \sum X/C_i \times M_i; X = H, O, N, S$
- 69 •  $DBE = \sum (C + 0.5N - 0.5H + 1)_i \times M_i$
- 70 •  $AI = \sum [(1 + C - O - S - 0.5H) / (C - O - N - S)]_i \times M_i$

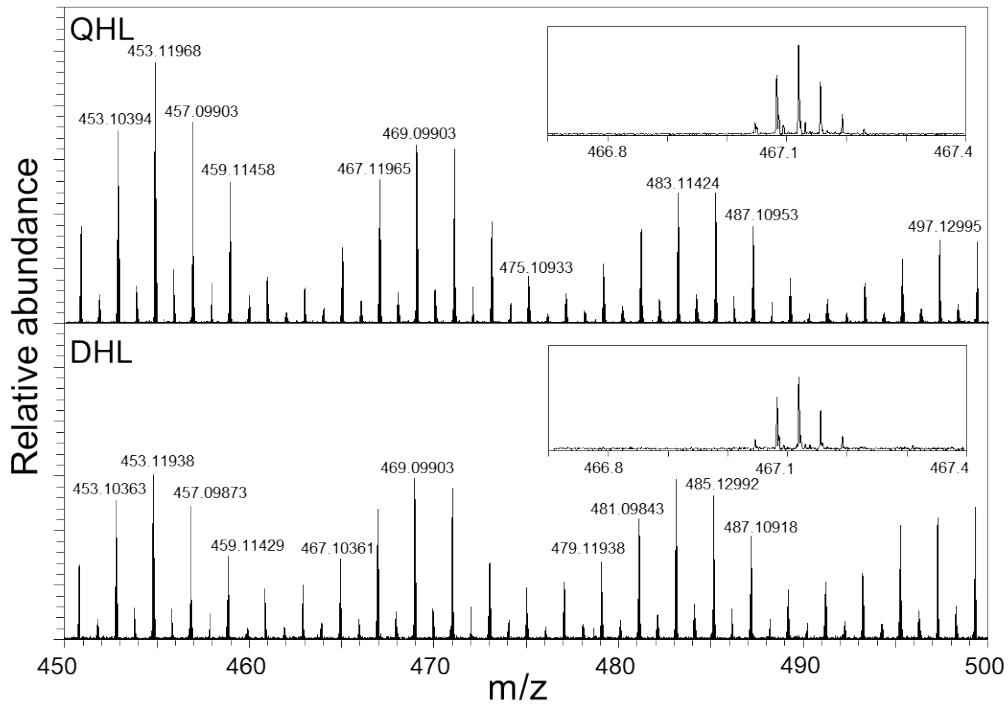
71 **Kendrick Mass Defect (KMD) analysis.** The KMD calculation equations for  
72 carboxyl group (COO) follow<sup>7-9</sup>:

- 73 • Kendrick Mass (COO) = exact m/z of peak × (nominal mass of COO/ exact  
74 mass of COO)
- 75 • KMD (COO) = Observed nominal mass - Kendrick Mass (COO)  
76 where nominal mass of nominal mass of COO = 44.00000, exact mass of COO =  
77 43.98983, and observed nominal mass is the integer value of the observed mass.
- 78



79  
80 **Figure S1.** Salinity variation of salt lakes in recent decades. (a) Hunlun Lake,  
81 Wuliangsuhai Lake, and Jili Lake from China in Asia<sup>10-12</sup>; (b) Bullen Merri Lake,  
82 Gnotuk Lake, Corangamite Lake, and Bolac Lake from Australia in Oceanica<sup>13</sup>; (c)  
83 Lakes in the present study: Qinghai Lake and Daihai Lake<sup>14-18</sup>.

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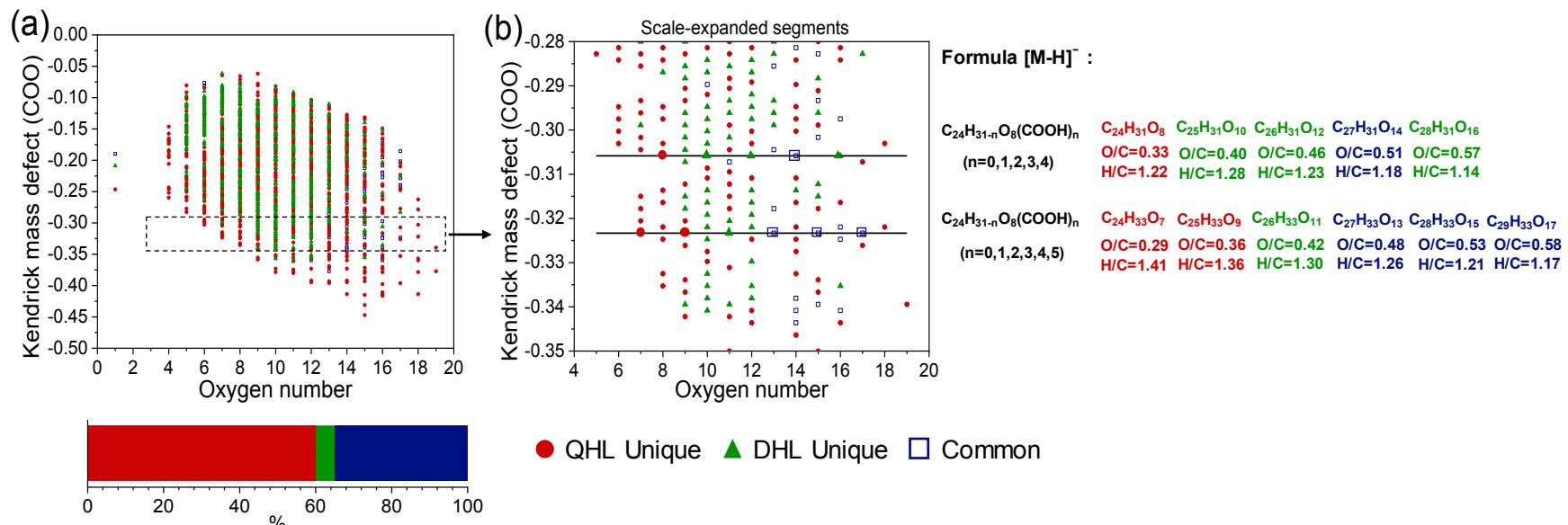


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86 **Figure S2.** An expanded region (450–500 m/z) of the mass spectra of QHL and DHL.

87 The insets expand the region of 466.7–467.4 m/z. The spectra have high magnitude  
 88 peaks with a low mass defect (0.0–0.5) of organic matters, and have no obvious peaks  
 89 with a high mass defect (0.7–0.8) of salts, confirming that PPL-SPE method used in  
 90 this work effectively removed salts in samples<sup>19</sup>.

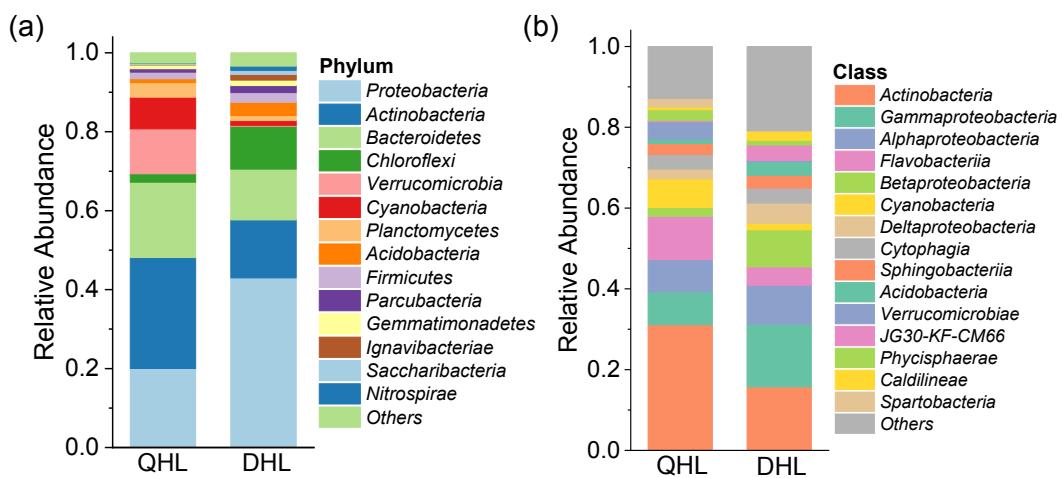
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93 **Figure S3.** KMD ( $\text{COO}$ )-number of O in the formulas plots (a) and its scale-expanded segments (b) for lignins/CRAM-like structures of CHO

94 formulas in the DOM of the two lakes. The assigned formulas (as  $[\text{M}-\text{H}]^-$  ions) of some randomly selected points in the segments are also  
 95 displayed. QHL Unique, the unique molecular formulas of QHL; DHL Unique, the unique molecular formulas of DHL; Common, the common  
 96 molecular formulas of both lakes.



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98 **Figure S4.** Microbial relative abundance of QHL and DHL at phylum level (a); class

99 level (b).

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**Table S1.** Geographical and climatic conditions of QHL and DHL.

Geographical and climatic conditions	QHL	DHL
Latitude and longitude	36°32'–37°15'N, 99°36'–100°47'E <sup>20</sup>	40°29'–40°37'N, 112°33'–112°46'E <sup>21</sup>
Altitude (m)	3193 <sup>20</sup>	1221 <sup>21</sup>
Surface area (km <sup>2</sup> )	4400 <sup>22</sup>	134 <sup>23</sup>
Average annual temperature (°C)	0.9–2.7 <sup>22</sup>	5.1 <sup>23</sup>
Average annual precipitation (mm)	357 <sup>20</sup>	423 <sup>23</sup>
Average annual evaporation (mm)	800–1000 <sup>20</sup>	1162 <sup>23</sup>
Annual average solar radiation (kcal/cm <sup>2</sup> )	~144.3 <sup>24</sup>	~141.1 <sup>25</sup>
Main soil type	Meadow soil <sup>26</sup>	Meadow soil <sup>27</sup>
Climate zone	East Asian Monsoon; semi-arid region <sup>20</sup>	East Asian Monsoon; semi-arid region <sup>28</sup>
Formation period	Late Tertiary <sup>29</sup>	Late Tertiary <sup>30</sup>

103 **Table S2.** Water properties of QHL and DHL.

Sample	pH	Temperature (°C)	DO (mg/L)	Electrical conductivity (mS/cm)	Salinity (g/L)	DOC (mg/L)
QHL	9.20	20.8	6.7	20.01	11.86	9.58
DHL	8.97	21.1	6.3	19.90	10.41	15.87

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105 **Table S3.** CHONS-Formula (as [M-H]<sup>-</sup> ions) from mass scale-expanded segments  
 106 (m/z = 319) in Figure 1 sorted by apparent molecular series.

Sample	m/z	Series	Formula	O/C	H/C	N/C	S/C
QHL	319.0461	CHO 1	C15H11O8	0.53	0.80	0.00	0.00
	319.0824	CHO 1	C16H15O7	0.44	1.00	0.00	0.00
	319.1188	CHO 1	C17H19O6	0.35	1.18	0.00	0.00
	319.1552	CHO 1	C18H23O5	0.28	1.33	0.00	0.00
	319.0672	CHO 2	C12H15O10	0.83	1.33	0.00	0.00
	319.1035	CHO 2	C13H19O9	0.69	1.54	0.00	0.00
	319.1399	CHO 2	C14H23O8	0.57	1.71	0.00	0.00
	319.0936	CHON 1	C15H15O6N2	0.40	1.07	0.13	0.00
	319.1299	CHON 1	C16H19O5N2	0.31	1.25	0.13	0.00
	319.0494	CHOS 1	C12H15O8S1	0.67	1.33	0.00	0.08
DHL	319.0858	CHOS 1	C13H19O7S1	0.54	1.54	0.00	0.08
	319.1222	CHOS 1	C14H23O6S1	0.43	1.71	0.00	0.07
	319.0459	CHO 1	C15H11O8	0.53	0.8	0	0
	319.0823	CHO 1	C16H15O7	0.44	1	0	0
	319.067	CHO 2	C12H15O10	0.83	1.33	0	0
	319.1033	CHO 2	C13H19O9	0.69	1.54	0	0
	319.1398	CHO 2	C14H23O8	0.57	1.71	0	0
CHS	319.0492	CHOS 1	C12H15O8S1	0.67	1.33	0	0.08
	319.0856	CHOS 1	C13H19O7S1	0.54	1.54	0	0.08
	319.1219	CHOS 1	C14H23O6S1	0.43	1.71	0	0.07

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108 **Table S4.** Total number of detected compounds, the contents of heteroatom, weighted averaged parameters, aromaticity index (AI) of two lakes.

Sample	Total number of compounds	Number of the O-compound	Number of the N-compound	Number of the S-compound	O/C	H/C	N/C	S/C	DBE	AI
QHL	3719	3639 (97.85%)	2024 (54.42%)	1497 (40.25%)	0.57	1.28	0.03	0.02	7.45	0.15
DHL	2058	1936 (94.07%)	1222 (59.38%)	919 (44.66%)	0.47	1.20	0.23	0.03	10.35	0.19

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110 **References**

- 111 1. Smith, C. R.; Sleighter, R. L.; Hatcher, P. G.; Lee, J. W., Molecular  
112 characterization of inhibiting biochar water-extractable substances using electrospray  
113 ionization Fourier transform ion cyclotron resonance mass spectrometry. *Environ. Sci.*  
114 *Technol.* **2013**, 47 (23), 13294-13302.
- 115 2. Xu, H.; Li, Y.; Ding, M. M.; Chen, W.; Wang, K.; Lu, C. H., Simultaneous  
116 removal of dissolved organic matter and nitrate from sewage treatment plant effluents  
117 using photocatalytic membranes. *Water Res.* **2018**, 143, 250-259.
- 118 3. Purcell, J. M.; Hendrickson, C. L.; Rodgers, R. P.; Marshall, A. G., Atmospheric  
119 pressure photoionization Fourier transform ion cyclotron resonance mass  
120 spectrometry for complex mixture analysis. *Anal. Chem.* **2006**, 78 (16), 5906-5912.
- 121 4. D'Andrilli, J.; Foreman, C. M.; Marshall, A. G.; McKnight, D. M.,  
122 Characterization of IHSS Pony Lake fulvic acid dissolved organic matter by  
123 electrospray ionization Fourier transform ion cyclotron resonance mass spectrometry  
124 and fluorescence spectroscopy. *Org. Geochem.* **2013**, 65, 19-28.
- 125 5. Koch, B. P.; Dittmar, T., From mass to structure: an aromaticity index for  
126 high-resolution mass data of natural organic matter. *Rapid Commun. Mass Spectrom.*  
127 **2006**, 20 (5), 926-932.
- 128 6. Kellerman, A. M.; Kothawala, D. N.; Dittmar, T.; Tranvik, L. J., Persistence of  
129 dissolved organic matter in lakes related to its molecular characteristics. *Nat. Geosci.*  
130 **2015**, 8 (6), 454-457.
- 131 7. Sleighter, R. L.; Hatcher, P. G., The application of electrospray ionization  
132 coupled to ultrahigh resolution mass spectrometry for the molecular characterization  
133 of natural organic matter. *J. Mass Spectrom.* **2007**, 42 (5), 559-574.
- 134 8. Kramer, R. W.; Kujawinski, E. B.; Hatcher, P. G., Identification of black carbon  
135 derived structures in a volcanic ash soil humic acid by Fourier transform ion cyclotron  
136 resonance mass spectrometry. *Environ. Sci. Technol.* **2004**, 38 (12), 3387-3395.
- 137 9. Li, L.; Zhu, B.; Yan, X.; Zhou, Q.; Wang, Y.; Jiang, G., Effect of silver sulfide  
138 nanoparticles on photochemical degradation of dissolved organic matter in surface  
139 water. *Chemosphere* **2018**, 193, 1113-1119.
- 140 10. Li, C.; Ma, W.; Shi, X. X.; Liao, W. G., Reconstruction of the hydrology series  
141 and simulation of salinity in ungauged Lake Hulun. *Sci. Limnol. Sin.* **2006**, 18 (1),  
142 13-20.
- 143 11. Lv, J.; Li, C. Y.; Jia, K. L.; Zhang, S.; Shi, X. H.; Zhao, S. N.; Sun, B., Impacts of  
144 salinity change on the effectiveness of heavy metals in the sediment of Wuliangsuhai.  
145 *Ecol. Environ. Sci.* **2017**, 26 (9), 1547-1553.
- 146 12. Zeng, H. A.; Wu, J. L., Lake status of water quality and the changes in Inner  
147 Mongolia-Xinjiang Plateau. *Sci. Limnol. Sin.* **2010**, 22 (6), 882-887.
- 148 13. Leahy, P.; Robinson, D.; Patten, R.; Kramer, A. *Lakes in the Western District of*  
149 *Victoria and climate change*; EPA Scientific Report: Victoria, **2010**.
- 150 14. Zhou, Y. K.; Jiang, J. H.; Huang, Q.; Sun, Z. D., Analysis on water salinization  
151 process of Daihai Lake in Inner Mongolia. *J. Arid Land Resour. Environ.* **2008**, 22  
152 (12), 51-55.
- 153 15. Wang, S. H.; Bai, M. X.; Chen, J. Y.; Zhao, L.; Zhang, B.; Guo, Y. Y.; Jiang, X.,  
154 Research on the ecological protection and restoration of  
155 mountain-river-forest-farmland-lake-grassland system in typical farming-pastoral  
156 ecotone: taking Daihai Lake Basin in Inner Mongolia as an example. *J. Lake Sci.*

- 157 2019, 9 (5), 515-519.
- 158 16. Yu, S. S., Chemical changes of Qinghaihu Lake and its rever water system in the  
159 last 30 years. *Oceanol. Limnol. Sin.* **1996**, 27 (02), 125-131.
- 160 17. Jin, Z. D.; You, C. F.; Wang, Y.; Shi, Y. W., Hydrological and solute budgets of  
161 Lake Qinghai, the largest lake on the Tibetan Plateau. *Quatern. Int.* **2010**, 218 (1-2),  
162 151-156.
- 163 18. Liu, W.; Jiang, H. C.; Yang, J.; Wu, G., Gammaproteobacterial diversity and  
164 carbon utilization in response to salinity in the lakes on the Qinghai-Tibetan plateau.  
165 *Geomicrobiol. J.* **2018**, 35 (5), 392-403.
- 166 19. Sleighter, R. L.; Hatcher, P. G., Fourier transform mass spectrometry for the  
167 molecular level characterization of natural organic matter: instrument capabilities,  
168 applications, and limitations. In *Fourier Transforms-Approach to Scientific  
169 Principles*, InTech: **2011**.
- 170 20. Cui, B. L.; Li, X. Y., Stable isotopes reveal sources of precipitation in the  
171 Qinghai Lake Basin of the northeastern Tibetan Plateau. *Sci. Total Environ.* **2015**,  
172 527, 26-37.
- 173 21. Wang, T.; Chen, J. S.; Xu, Y.; Zhan, L. C.; Huang, D. W., Isotopes and  
174 hydrochemistry of Daihai Lake recharging sources, Northern China. *J. Radioanal.  
175 Nucl. Ch.* **2017**, 312 (3), 615-629.
- 176 22. Liu, W.; Jiang, H. C.; Yang, J.; Wu, G., Gammaproteobacterial diversity and  
177 carbon utilization in response to salinity in the lakes on the Qinghai-Tibetan Plateau.  
178 *Geomicrobiol. J.* **2018**, 35 (5), 392-403.
- 179 23. Xiao, J. L.; Zhang, S. R.; Fan, J. W.; Wen, R. L.; Xu, Q. H.; Inouchi, Y.;  
180 Nakamura, T., The 4.2 ka event and its resulting cultural interruption in the Daihai  
181 Lake basin at the East Asian summer monsoon margin. *Quatern. Int.* **2018**, 527,  
182 87-93.
- 183 24. Wang, Z. L.; Yang, F. Y., Study on the change characteristics of total solar  
184 radiation in the main cities and towns of Qinghai Province in the past 55 years.  
185 *Qinghai Environ.* **2017**, 27 (3), 140-145+155.
- 186 25. Wang, S. M.; Yu, Y. S.; Wu, R. J.; Feng, M., Daihai: lake environment and  
187 climate change. USTC Press: **1990**.
- 188 26. Rui, J. P.; Li, J. B.; Wang, S. P.; An, J. X.; Liu, W.; Lin, Q. Y.; Yang, Y. F.; He,  
189 Z. L.; Li, X. Z., Responses of bacterial communities to simulated climate changes in  
190 alpine meadow soil of the Qinghai-Tibet Plateau. *Appl. Environ. Microb.* **2015**, 81  
191 (17), 6070-6077.
- 192 27. Evans, S. E.; Burke, I. C.; Lauenroth, W. K., Controls on soil organic carbon and  
193 nitrogen in Inner Mongolia, China: a cross-continental comparison of temperate  
194 grasslands. *Global Biogeochem. Cy.* **2011**, 25, 1-14.
- 195 28. Peng, Y. J.; Xiao, J.; Nakamura, T.; Liu, B. L.; Inouchi, Y., Holocene East Asian  
196 monsoonal precipitation pattern revealed by grain-size distribution of core sediments  
197 of Daihai Lake in Inner Mongolia of north-central China. *Earth Planet. Sci. Lett.*  
198 **2005**, 233 (3-4), 467-479.
- 199 29. Liang, S. H.; Guo, J.; Wu, P.; Feng, Y. Q.; Wang, X. S.; Wang, G. J.; Xu, W. L.;  
200 Luo, Y. F.; Wan, L., Hydrogeochemical and isotopic characteristics of surface water  
201 and groundwater in the Qinghai Lake catchment (China). *Arab. J. Geosci.* **2020**, 13  
202 (3), 1-16.
- 203 30. Yu, X. H.; Li, S. L.; Tan, C. P.; Jing, X.; Chen, B. T.; Fan, Y., The response of  
204 deltaic systems to climatic and hydrological changes in Daihai Lake rift basin, Inner  
205 Mongolia, northern China. *J. Palaeogeogr.-English* **2013**, 2 (1), 41-55.