Supporting Information

Amorphous Ni-Based Nanoparticles for Alkaline Oxygen Evolution

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| Table S1: List of material | s used for synthesis and | as standards for analysis |
|----------------------------|--------------------------|---------------------------|
|----------------------------|--------------------------|---------------------------|

| Material | Source | Purity | Particle Size |
|--------------------------------|-------------------|------------|---------------|
| Ni | Alfa Aesar | 99.9 wt % | < 149 µm |
| Со | Alfa Aesar | 99.8 wt % | < 149 µm |
| Nb | Alfa Aesar | 99.99 wt % | < 44 µm |
| Y | Goodfellow | 99.9 wt % | <500 µm* |
| NiO | Fisher Scientific | 99.8 wt % | |
| β-Ni(OH) ₂ | Sigma Aldrich | | |
| β-NiOOH | This work** | | |
| Co ₃ O ₄ | Sigma Aldrich | 99.5 wt % | < 50 nm |
| Co(OH) ₂ | Sigma Aldrich | 95 wt % | |
| Y ₂ O ₃ | This work*** | | |
| Y(OH) ₃ | Muse Chem | 99.99 wt% | |

*Milled at room temperature to reduce the particle size to < 149 μ m prior to alloying **Synthesized by adding a solution of 47.5 g KOH and 11 mL Br₂ in 500 mL of H₂O dropwise to 50 g of Ni(NO₃)₂·H₂O in 750 mL of H₂O at 100 °C. The sample was then rinsed with H₂O, centrifuged, filtered, and dried under vacuum for ca. 48 hours. The presence of β-NiOOH was verified using XRD.

***Produced by oxidizing Y powder and verified using XRD

| Table S2: Nominal and measured (ICP-OES) compositions for as-produced n | nanoparticles |
|---|---------------|

| Material | | Ni (at %) | Co (at %) | Nb (at %) | Y (at %) |
|--|----------------|--------------|-------------|-----------------|---------------|
| Crystalline | Nominal | 95 | 5 | | |
| Ni95C05 | ICP-AES | 95.2 ± 2.1 | 4.8 ± 0.3 | | |
| Amorphous | Nominal | 79.2 | | 12.5 | 8.3 |
| Ni _{79.2} Nb _{12.5} Y _{8.3} | ICP-AES | 82.0 ± 2.0 | | 10.8 ± 2.6 | 7.3 ± 0.8 |
| Amorphous | Nominal | 74.2 | 5 | 12.5 | 8.3 |
| Ni74.2C05Nb12.5Y8.3 | ICP-AES | 73.4 ± 2.1 | 4.9 ± 0.3 | $12.1\ \pm 2.8$ | 9.6 ± 1.4 |

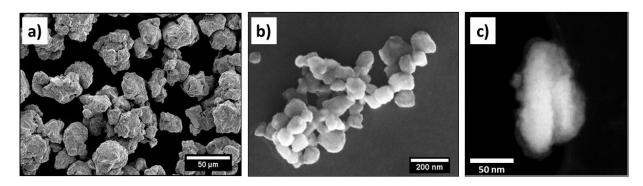


Figure S1: SEM images of amorphous $Ni_{74.2}Co_5Nb_{12.5}Y_{8.3}$ particles after (a) cryomilling and (b) SA-HEBM, along with (c) TEM image of a single particle.

Table S3: Tafel slope, current density at $\eta = 400 \text{ mV}$, and η at $j = 10 \ \mu\text{A cm}^{-2}_{\text{ECSA}}$ values for the OER of crystalline Ni, Ni₉₅Co₅ and amorphous Ni_{79.2-x}Co_xNb_{12.5}Y_{8.3} (x = 0 and 5 at %) after anodically cycling 50 and 10000 times with a scan rate of 50 mV s⁻¹ between 0.1 and 0.7 V_{Hg/Hg0}. Experiments were conducted in deaerated 1 M KOH at 30 °C. Values and error bars are based on a minimum of three tests.

| Material | Cycles | Tafel Slope (mV) | jη = 400 mV (μΑ cm ⁻² εcsA) | η _{j = 10 μA cm⁻²_{ECSA} (mV)} |
|---------------------------------|--------|---------------------|---|--|
| Crystalline | 50 | 59 ± 8.2 | 252 ± 89.7 | 299 ± 1.80 |
| Ni | 10000 | 71 ± 0.2 | 169 ± 44.4 | 276 ± 13.4 |
| Crystalline | 50 | 49 ± 1.0 | 852 ± 180 | 235 ± 3.50 |
| Ni95Co5 | 10000 | 81 ± 5.9 | 93.3 ± 43.6 | 297 ± 22.1 |
| Amorphous | 50 | 50 ± 3.8 | 299 ± 18.8 | 281 ± 17.0 |
| $Ni_{79.2}Nb_{12.5}Y_{8.3}$ | 10000 | 73 ± 2.7 | 76.1 ± 18.1 | 288 ± 12.5 |
| Amorphous | 50 | 71 ± 3.1 | 287 ± 48.1 | 267 ± 17.0 |
| $Ni_{74.2}Co_5Nb_{12.5}Y_{8.3}$ | 10000 | 58 ± 1.3 | 289 ± 73.3 | 265 ± 18.1 |

X-ray Photoelectron Spectroscopy (XPS) Analysis

Tables S4 (Ni 2p) and S5 (O 1 s) were used to aid with the identification of surface species produced on crystalline Ni₉₅Co₅, amorphous Ni_{79.2}Nb_{12.5}Y_{8.3}, and amorphous Ni_{74.2}Co₅Nb_{12.5}Y_{8.3} after anodic cycling 50 times. The standards produced and tested within this report were confirmed to be in good agreement with literature (NIST X-ray Photoelectron Spectroscopy Database, NIST Standard Reference Database Number 20, National Institute of Standards and Technology, Gaithersburg MD, 20899 (2000), doi:10.18434/T4T88K).

| Material | Binding Energy (eV) | Peak Width FWHM (eV) | Δsplit (eV) | ∆sat (eV) | Reference |
|---|------------------------|-------------------------|------------------|--------------|-----------|
| Crystalline Ni95C05 | 852.1, 854.6, 855.1 | 1.57, 1.76, 2.83 | 18.4, 17.6, 17.8 | 6.10 | This work |
| Amorphous Ni _{79.2} Nb _{12.5} Y _{8.3} | 851.9, 855.0 | 1.24, 2.70 | 17.3, 17.5 | 5.88 | This work |
| Amorphous Ni _{74.2} Co ₅ Nb _{12.5} Y _{8.3} | 852.0, 854.9 | 1.13, 3.03 | 17.2, 17.5 | 5.95 | This work |
| Ni | 852.5 | 1.08 | 17.3 | 5.65 | This work |
| NiO | 854.1, 855.7 | 0.98, 3.30 | 17.3, 17.3 | 6.30, 7.10 | This work |
| β-Ni(OH) ₂ | 855.4, 856.2 | 2.06, 3.13 | 17.4, 18.0 | 5.38, 6.10 | This work |
| β-ΝίΟΟΗ | 854.8, 856.0 | 1.33, 2.05 | 17.4 | 5.11 | This work |
| γ-ΝίΟΟΗ | 855.3 | 2.42 | 17.5, 18.3 | 5.80 | 1 |

Table S4: Ni 2p_{3/2} XPS fitting parameters for alloys after anodic cycling and relevant standards

 Δ split is the difference between Ni 2p_{3/2} and Ni 2p_{1/2} peaks

 Δsat is the difference between Ni $2p_{3/2}$ and Ni $2p_{3/2}$ satellite peak

| Material | Binding Energy(eV) | Peak Width FWHM (eV) | Reference |
|---|----------------------------|-------------------------|-----------|
| Crystalline Ni95C05 | 528.9, 530.8, 533.1 | 1.37, 1.85, 1.98 | This work |
| Amorphous Ni _{79.2} Nb _{12.5} Y _{8.3} | 529.2, 530.6, 532.7 | 1.31, 1.68, 1.49 | This work |
| Amorphous Ni _{74.2} Co ₅ Nb _{12.5} Y _{8.3} | 529.3, 530.7, 531.4 | 1.25, 1.61, 2.30 | This work |
| NiO | 529.7, 531.2, 531.3 | 0.88, 3.19, 2.14 | This work |
| β-Ni(OH) ₂ | 530.7, 530.9 | 1.39, 2.14 | This work |
| β-ΝίΟΟΗ | 529.3, 530.7, 532.1 | 0.87, 1.74, 2.68 | This work |
| γ-ΝίΟΟΗ | 528.9, 530.8, 532.6 | 1.27, 1.92, 2.18 | 1 |
| C03O4 | 529.4, 529.8, 531.2, 532.5 | 1.49, 0.92, 1.86, 2.36 | This work |
| Co(OH) ₂ | 531.2, 532.7 | 1.55, 1.85 | This work |
| CoO ₂ | 528.7, 531.2, 533.0 | | 2 |
| Y ₂ O ₃ | 529.2, 531.5, 533.3 | 1.45, 2.01, 1.99 | This work |
| Y(OH) ₃ | 531.4, 532.4, 533.2 | 1.72, 1.76, 2.28 | 3 |

Table S5: O 1 s XPS fitting parameters for alloys after anodic cycling and relevant standards

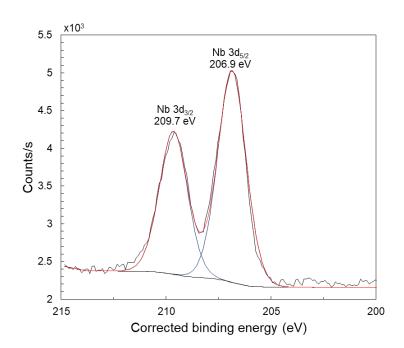


Figure S2: Nb 3d XPS spectra corresponding to the presence of Nb₂O₅ on the surface of amorphous Ni_{74.2}Co₅Nb_{12.5}Y_{8.3} prior to anodic cycling

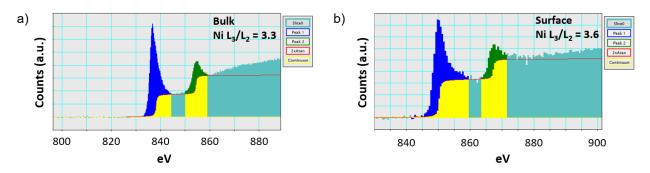


Figure S3: Ni L_3/L_2 white-line intensity ratios from electron energy-loss near-edge spectroscopy for pristine amorphous Ni_{74.2}Co₅Nb_{12.5}Y_{8.3} a) bulk and b) surface

Table S6: Comparison of L_3/L_2 white-line intensity ratios for pristine Ni_{74.2}Co₅Nb_{12.5}Y_{8.3} bulk and surface with Ni, NiO and Ni(OH)₂ references

| Sample | Estimated L ₃ /L ₂ ratios | Reference |
|--|---|-----------|
| Ni _{74.2} Co ₅ Nb _{12.5} Y _{8.3} bulk | 3.3 | This work |
| Ni _{74.2} Co ₅ Nb _{12.5} Y _{8.3} surface | 3.6 | This work |
| Ni reference | 3.3 | 4 |
| NiO reference | 3.8 | 5 |
| NiO reference | 4.0 | 6 |
| Ni(OH) ₂ reference | 3.6 | 5 |

Table S7: Quantification of Nb and Y elemental edges in EELS for surface and core regions in pristine, 1000 cycled, 5000 cycled, and submerged Ni_{74.2}Co₅Nb_{12.5}Y_{8.3} nanoparticles.

| Area | Sample | Nb Concentration (at%) | Y Concentration (at%) |
|---------|-------------|---------------------------|--------------------------|
| | Pristine | 45 | 55 |
| Core | Submerged | 39 | 61 |
| COLE | 1000 Cycles | 41 | 59 |
| | 5000 Cycles | 48 | 52 |
| | Pristine | 69 | 31 |
| Surface | Submerged | 68 | 32 |
| Surrace | 1000 Cycles | 25 | 75 |
| | 5000 Cycles | 0 | 100 |

References

- 1. A. N. Mansour and C. A. Melendres, Surf. Sci. Spectra 3, 271 (1994).
- 2. L. Dahéron, R. Dedryvère, H. Martinez, M. Ménétrier, C. Denage, C. Delmas, and D.
- Gonbeau, Chem. Mater. 20, 583 (2008).
- 3. K. M. Cole, D. W. Kirk, and S. J. Thorpe, Surf. Sci. Spectra 27, (2020).
- 4. P. A. Van Aken and B. Liebscher, Phys. Chem. Miner. 29, 188 (2002).
- 5. M. Hao, S. Garbarino, S. Prabhudev, T. Borsboom-Hanson, G. A. Botton, D. A. Harrington, and D. Guay, J. Phys. Chem. C **123**, 1082 (2019).
- 6. R. D. Leapman, L. A. Grunes, and P. L. Fejes, Phys. Rev. B 26, 614 (1982).