## **Supporting Information**

## Network Structure and Dissolution Properties of Phosphate-Doped Borosilicate Glasses

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Table S1. <sup>23</sup>Na MAS NMR fitting parameters for sample P4

$\delta_{iso}$ (ppm)	C <sub>Q</sub> (MHz)	$\eta_Q$	Int. (%) <sup>c</sup>
2.46(0.3)	2.54(0.05)	0.6(0.2)	13(5)
-11.3(0.2) <sup>a</sup>	28.3(1) <sup>b</sup>		87(5)

<sup>a</sup>Peak position (ppm) <sup>b</sup>Peak width (ppm) <sup>c</sup> Int. (%) = Integrated intensity

Table S2. Fitting parameters used in the deconvolution of <sup>31</sup>P MAS NMR spectra

		$\mathbf{P}^0$			$\mathbf{P}^1$			$P^2_{nB(Al)}$			$Na_4P_2O_7$	
	$\delta_{iso}$	FWHM <sup>a</sup>	(%) <sup>b</sup>	$\delta_{iso}$	FWHM <sup>a</sup>	(%) <sup>b</sup>	$\delta_{iso}(ppm)$	FWHM <sup>a</sup>	(%) <sup>b</sup>	$\delta_{iso}$	FWHM <sup>a</sup>	(%) <sup>b</sup>
	(ppm)			(ppm)						(ppm)		
P1	15.7(0.5)	3.6(1)	2(1)	3.4(0.3)	7.4(0.5)	69(5)	-6.5(0.5)	10.3(1)	29(5)	**	**	**
P2	14.5(0.2)	3.6(0.5)	11(2)	3.3(0.3)	7.2(0.3)	68(3)	-6.7(0.5)	10.4(1)	21(3)	**	**	**
Р3	14.5(0.2)	3.1(0.2)	24(2)	2.6(0.3)	3.1(0.3)	51(8)	-6.5(1)	10.4(2)	9(4)	5.6(0.6)	3.1(0.5)	16(6)
P4	15.3(0.2)	3.5(0.3)	6(2)	2.4(0.5)	4.9(1.5)	65(15)	-6.3(1.5)	10.4(2)	8(4)	5.6(1)	4.2(1.5)	21(10)
P4-Mo3	12.3(0.5)	4.4(1)	3(2)	-0.1(0.5)	8.1(2)	64(20)	-6.2(3)	11.3(3)	19(15)	5.8(1)	4.9(2)	13(8)

<sup>a</sup>FWHM = Peak width (ppm)

b(%) = Integrated intensity (%)

 Table S3. 27Al MAS NMR fit parameters

	<sup>[4]</sup> Al			<sup>[5]</sup> A1			<sup>[6]</sup> Al		
	$\delta_{iso}$	FWHM <sup>a</sup>	Int.	$\delta_{iso}$	FWHM <sup>a</sup>	Int.	$\delta_{iso}$	FWHM <sup>a</sup>	Int.
	(ppm)	(±1)	(%) <sup>b</sup>	(ppm)	(±0.5)	(%) <sup>b</sup>	(ppm)	(±2)	(%) <sup>b</sup>
	(±0.2)			(±0.2)			(±0.2)		
P0	56.9	13.7	81(3)	14.3	5.8	19(3)	**	**	**
P1	56.3	13.9	84(2)	14.4	5.3	16(2)	**	**	**
P2	56.2	14.0	84(2)	14.4	5.3	16(2)	**	**	**
Р3	56.4	14.1	90(1)	14.6	5.6	10(1)	**	**	**
P4	54.9	14.1	84(2)	14.1	5.4	16(2)	**	**	**
P4-Mo3	54.2	14.7	79(2)	14.4	5.9	15(1)	-6.1	14.9	6(2)

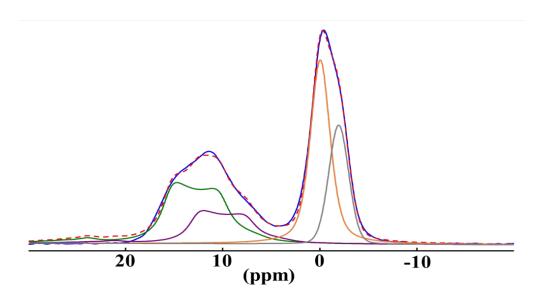
<sup>a</sup>FWHM = Peak width (ppm)

<sup>b</sup>Int. (%) = Integrated intensity

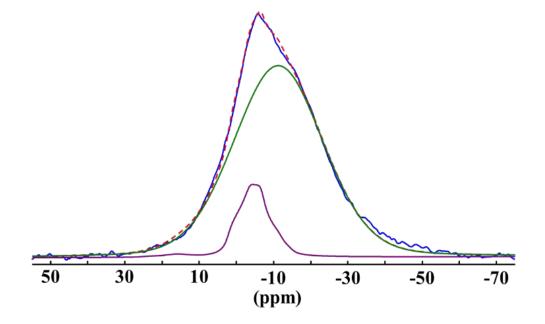
 Table S4.
 <sup>29</sup>Si MAS NMR fit parameters

	Q <sup>3</sup>		$Q^4$		
	Peak max (ppm)	Int. (%) <sup>a</sup>	Peak max (ppm)	Int. (%) <sup>a</sup>	
P0	-89(2)	63(17)	-101(3)	37(17)	
P2	-92(5)	41(14)	-102(3)	59(14)	
P4	-93(2)	10(10)	-102(1)	90(10)	
P4-Mo3	**	**	-105	100	

aInt. (%) = Integrated Intensity



**Figure S1.** Deconvolution of <sup>11</sup>B MAS NMR spectrum of sample P0. Green, magenta, orange, and grey components represent <sup>[3]</sup>B<sub>non-ring</sub>, <sup>[3]</sup>B<sub>ring</sub>, <sup>[4]</sup>B<sub>3Si</sub> and <sup>[4]</sup>B<sub>4Si</sub> units, respectively.



**Figure S2.** Deconvolution of the <sup>23</sup>Na MAS NMR spectrum of P4 representing glassy (green) and poorly-crystalline Na<sub>4</sub>P<sub>2</sub>O<sub>7</sub> (magenta) phases.

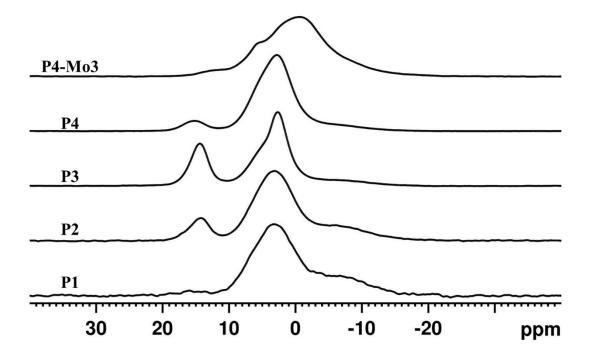
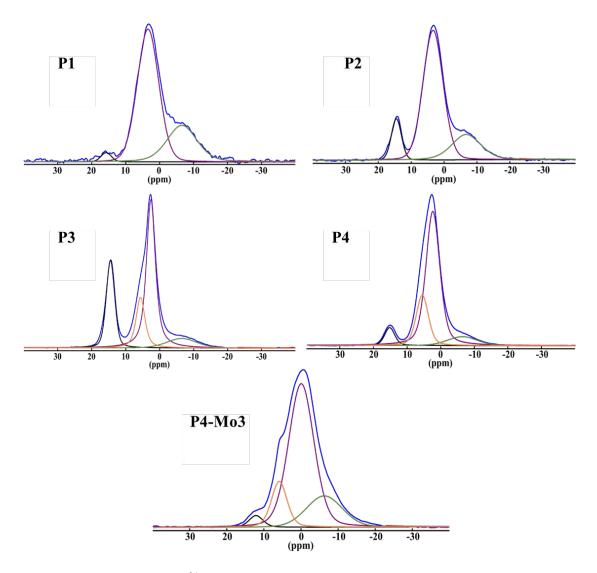
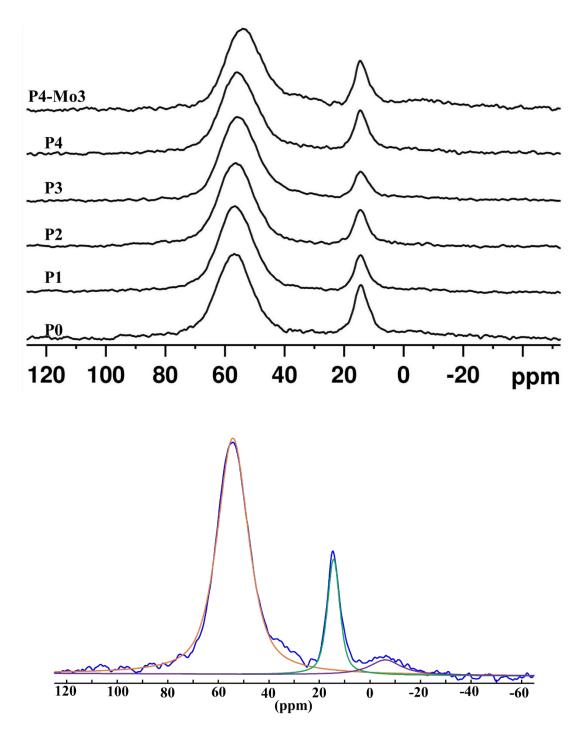


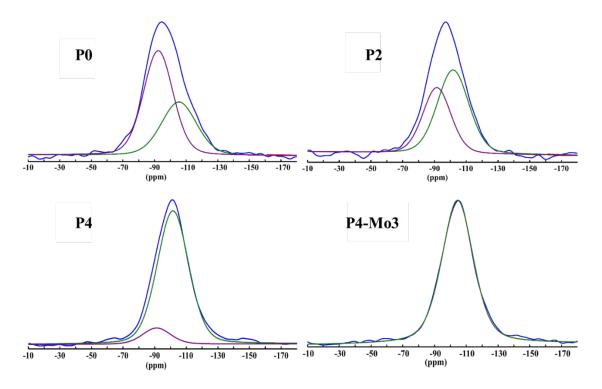
Figure S3. <sup>31</sup>P MAS NMR spectra of glasses



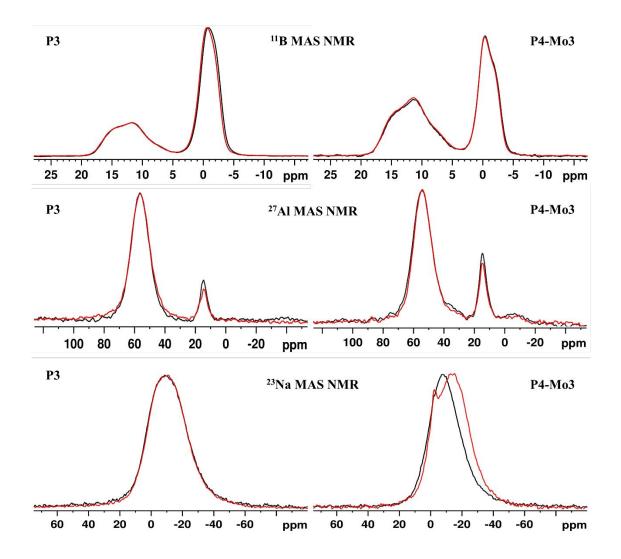
**Figure S4.** Deconvolution of <sup>31</sup>P MAS NMR spectra. Blue is the experimental spectrum and components in black, orange, purple and green represent P<sup>0</sup>, P<sup>1</sup>, P<sup>2</sup><sub>nB(Al)</sub> and poorly crystalline Na<sub>4</sub>P<sub>2</sub>O<sub>7</sub>, respectively.



**Figure S5.** <sup>27</sup>Al MAS NMR spectra of glasses (top) and deconvolution of <sup>27</sup>Al MAS NMR spectrum of P4-Mo3, where orange, green and purple components represent <sup>[4]</sup>Al, <sup>[5]</sup>Al and <sup>[6]</sup>Al units, respectively.



**Figure S6.** Deconvolution of <sup>29</sup>Si MAS NMR spectra. The subspectral components in purple and green represent Q<sup>3</sup> and Q<sup>4</sup> units, respectively.



**Figure S7.**<sup>11</sup>B, <sup>27</sup>Al and <sup>23</sup>Na MAS NMR spectra of P3 and P4-Mo3 before (black) and after (red) dissolution.

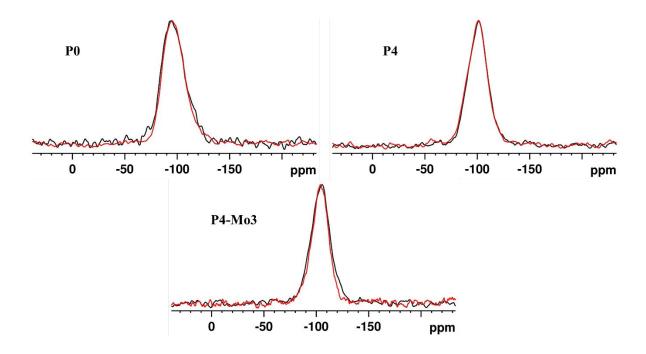


Figure S8. <sup>29</sup>Si MAS NMR spectra before (black) and after (red) dissolution

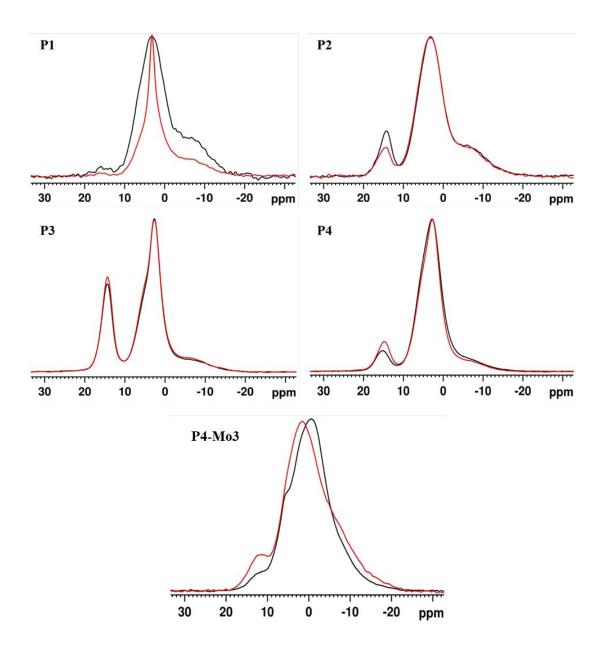
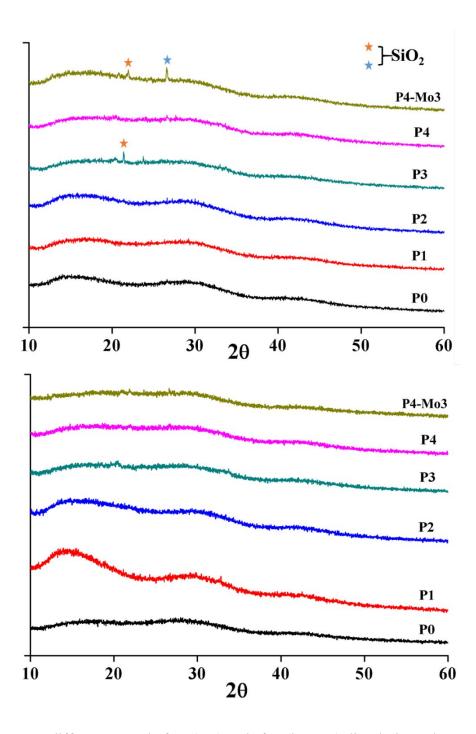
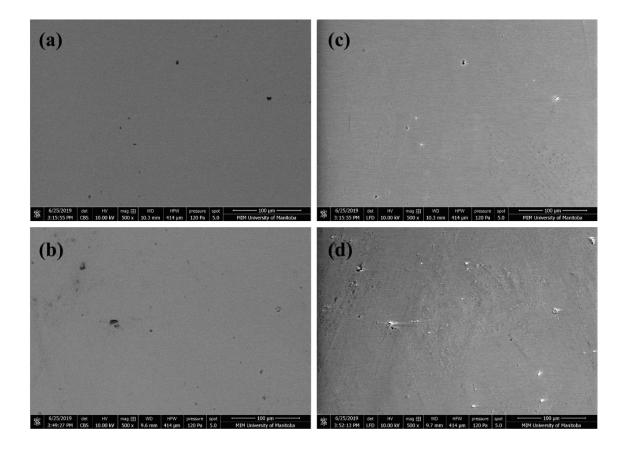


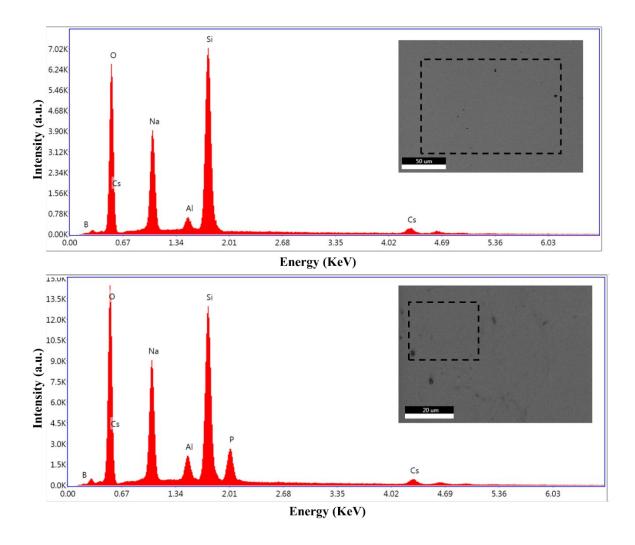
Figure S9. An overlay of <sup>31</sup>P MAS NMR spectra before (black) and after (red) the dissolution.



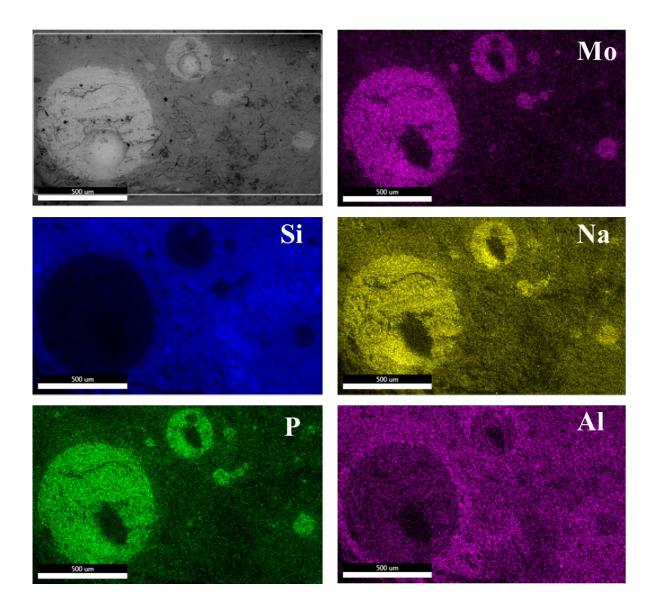
**Figure S10.** X-ray diffractograms before (top) and after (bottom) dissolution. The orange and blue stars represent reflections from the  $SiO_2$  phases, cristobalite and quartz, respectively.



**Figure S11.** SEM images (500x magnification) of glasses prior to dissolution: (a) BSE image of P0, (b) BSE image of P4, (c) SE image of P0, (d) SE image of P4.



**Figure S12.** EDS spectra of glasses P0 and P4 with their corresponding BSE images. The dotted rectangles show the analyzed areas.



**Figure S13.** BSE image of a cross-section of P4-Mo3 and the element maps showing the distribution of elements in the mapped area.

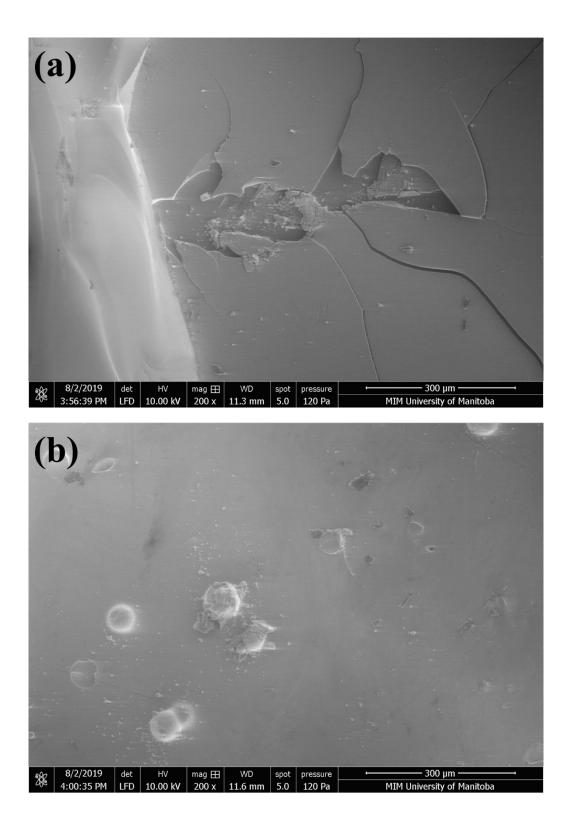
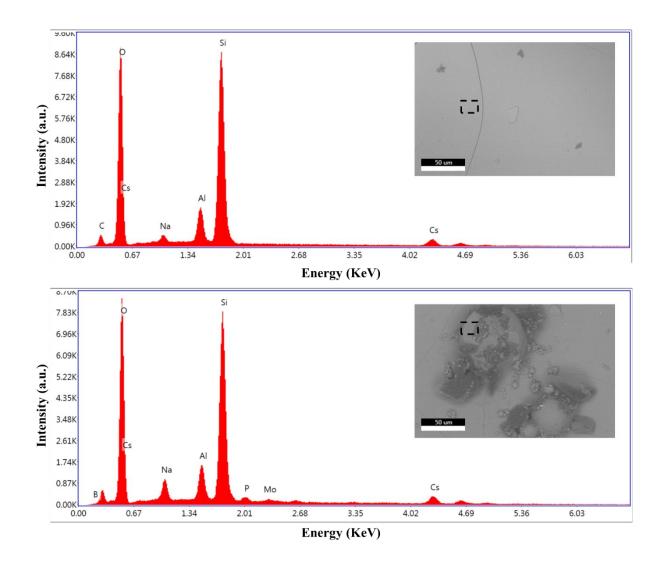
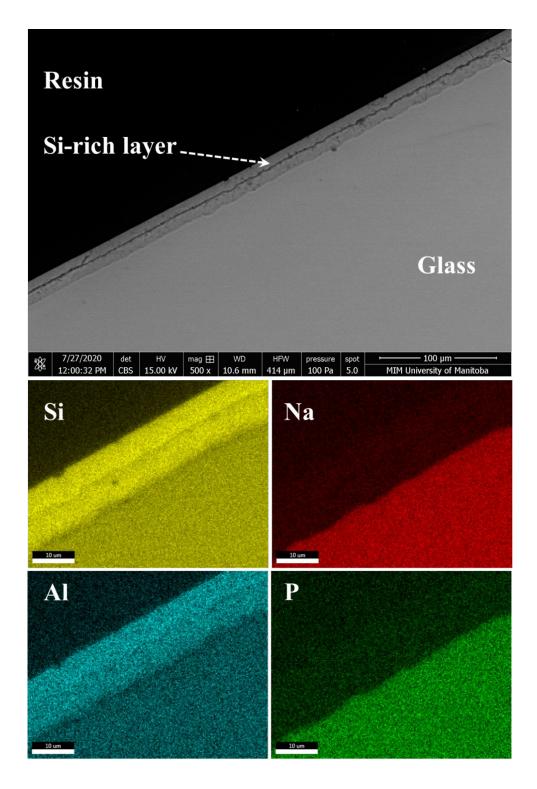


Figure S14. BSE images of the surfaces of (a) P4 and (b) P4-Mo3 after dissolution.



**Figure S15.** EDS spectra of the surfaces of glasses P4 (top) and P4-Mo3 (bottom) after dissolution. The analyzed areas are marked with a dotted square in the BSE images shown.



**Figure S16.** Back-scattered electron image of a cross-section of sample P4 after dissolution, and element maps of the Si-rich layer deposited on the glass surface