

*Supporting Information*

**Microwave Catalytic Degradation of Antibiotic Molecules by  
2D Sheets of Spinel Nickel Ferrite**

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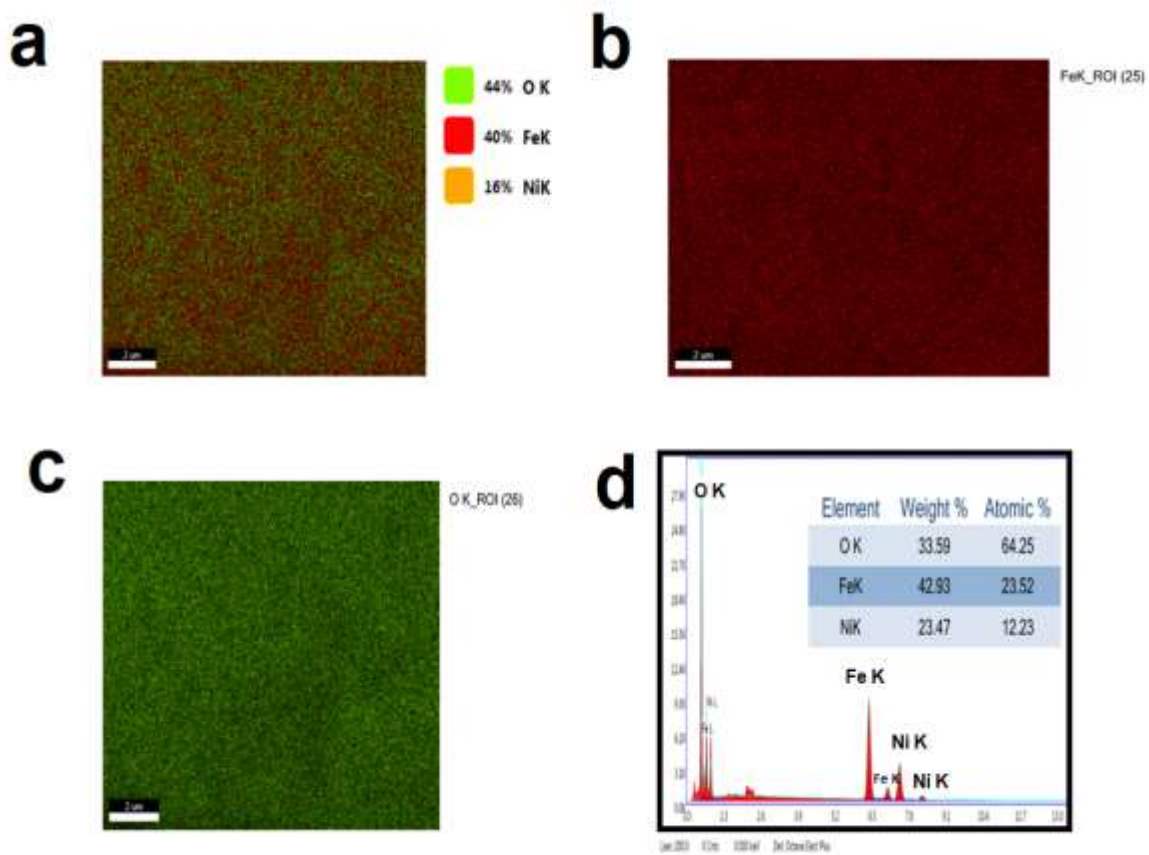
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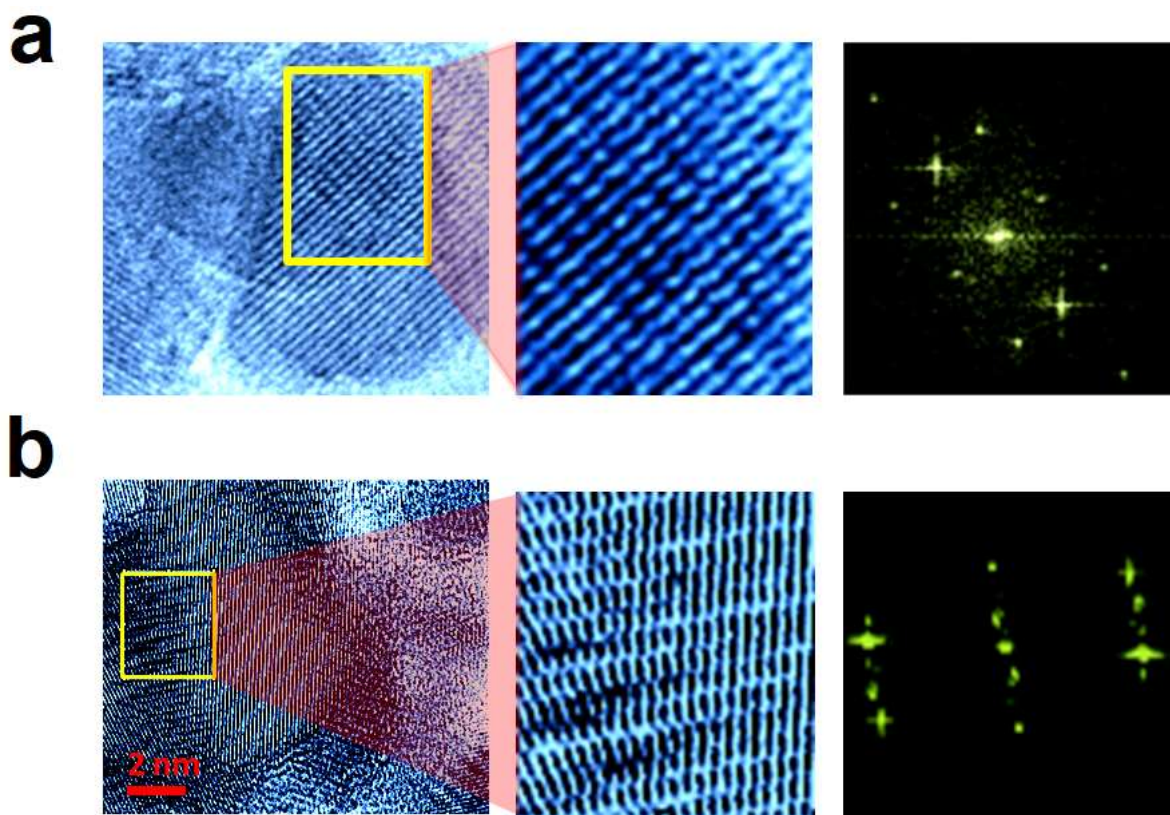
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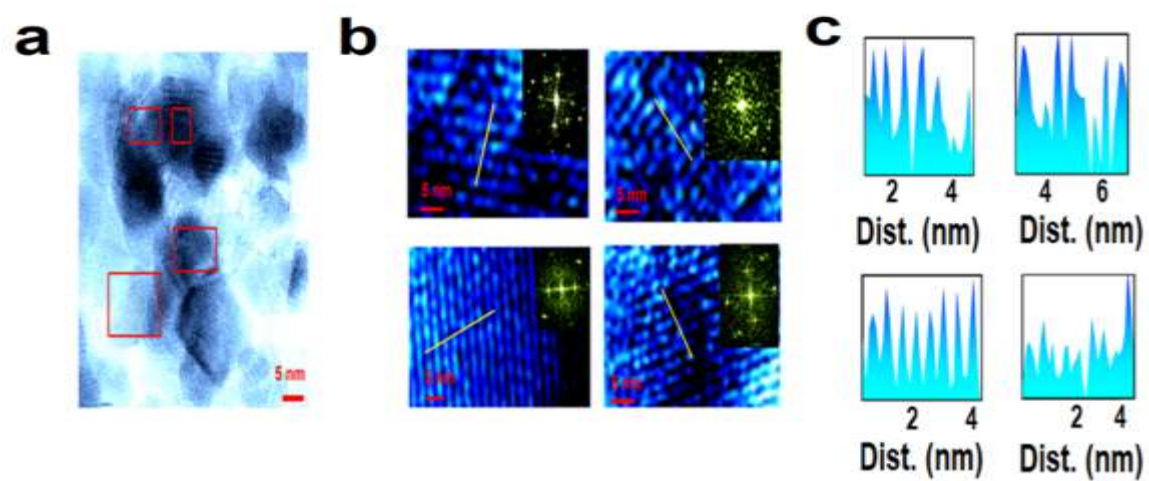
\*Dr. Sujoy Kumar Samanta: Phone: +91-612 302 8173, +91-9835355046, E-mail: [sksamanta@iitp.ac.in](mailto:sksamanta@iitp.ac.in), [krsanju07@gmail.com](mailto:krsanju07@gmail.com)



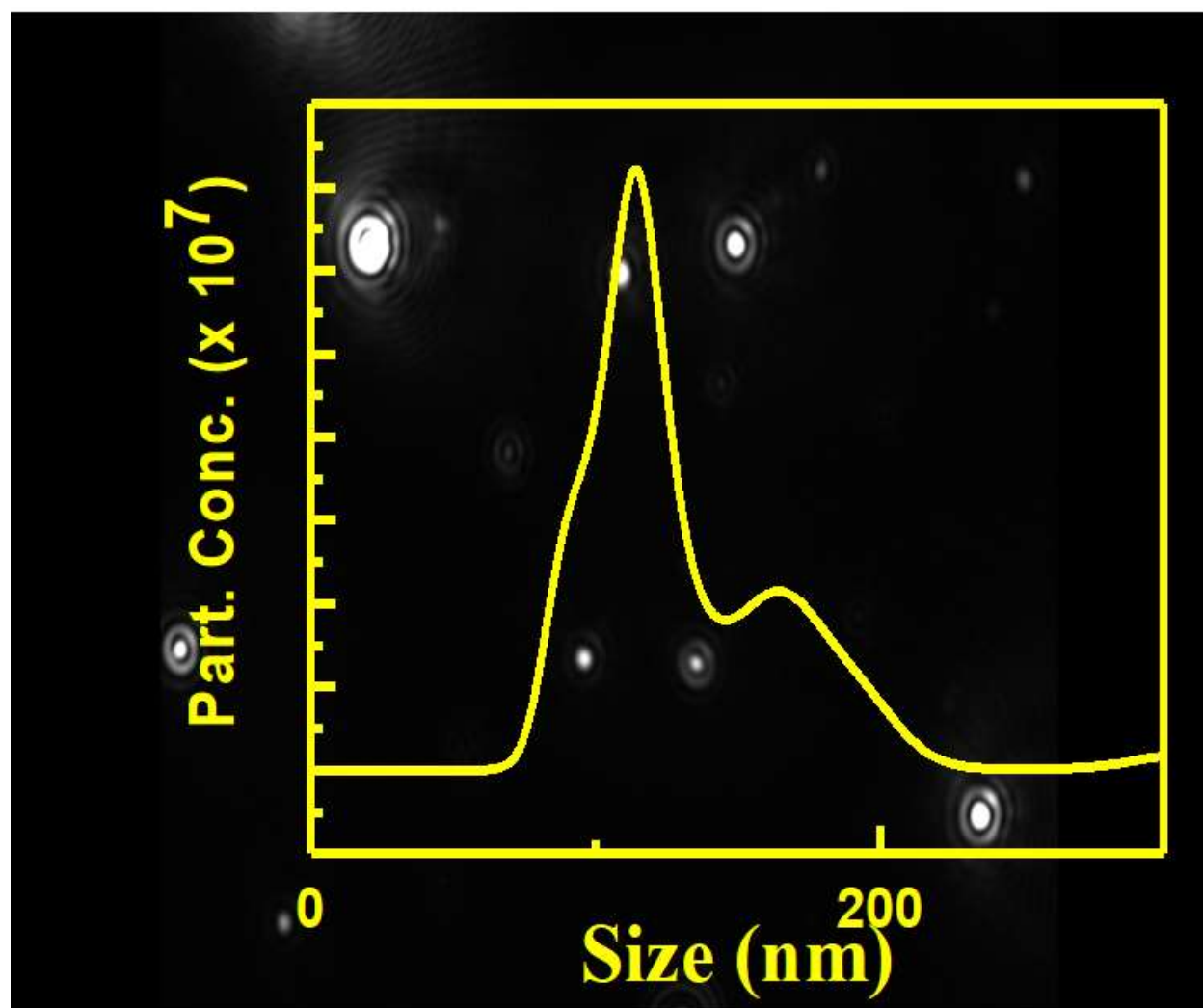
**Figure S1.** The elemental mapping and SEM-EDS of the SNFO catalyst.



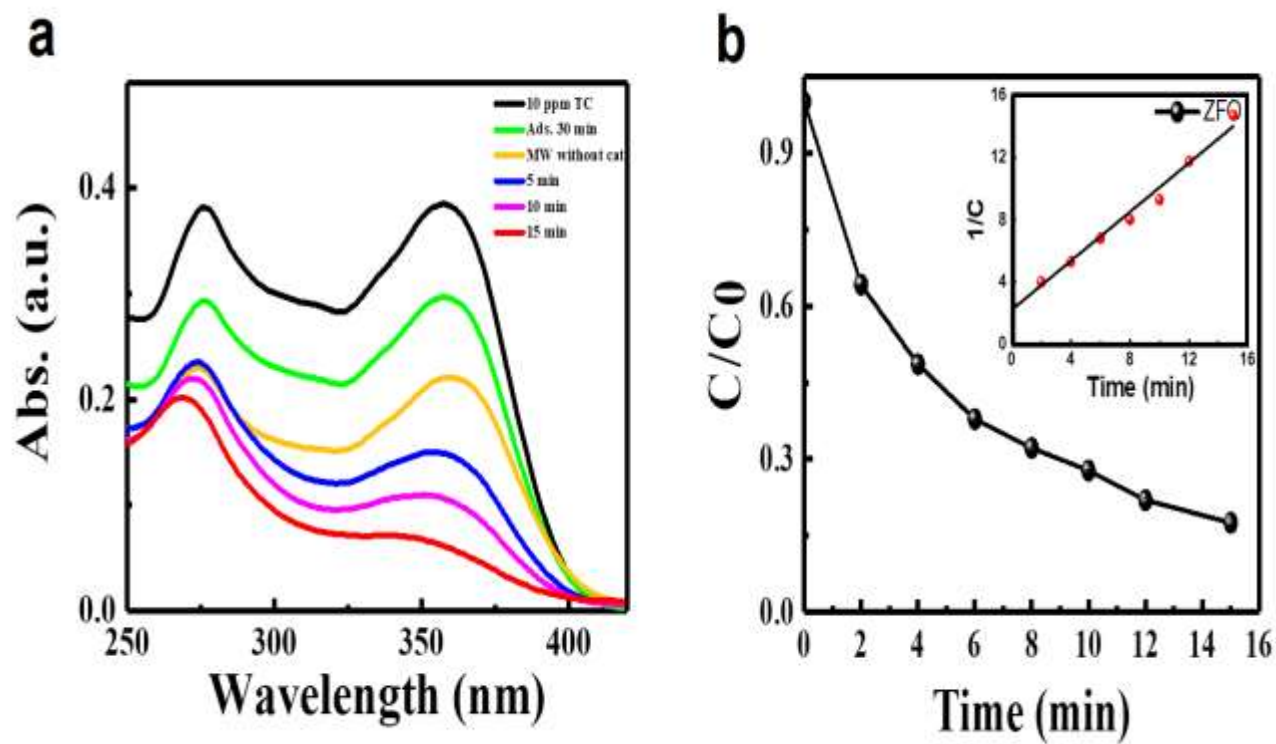
**Figure S2.** HRTEM images and FFT of SNFO catalyst.



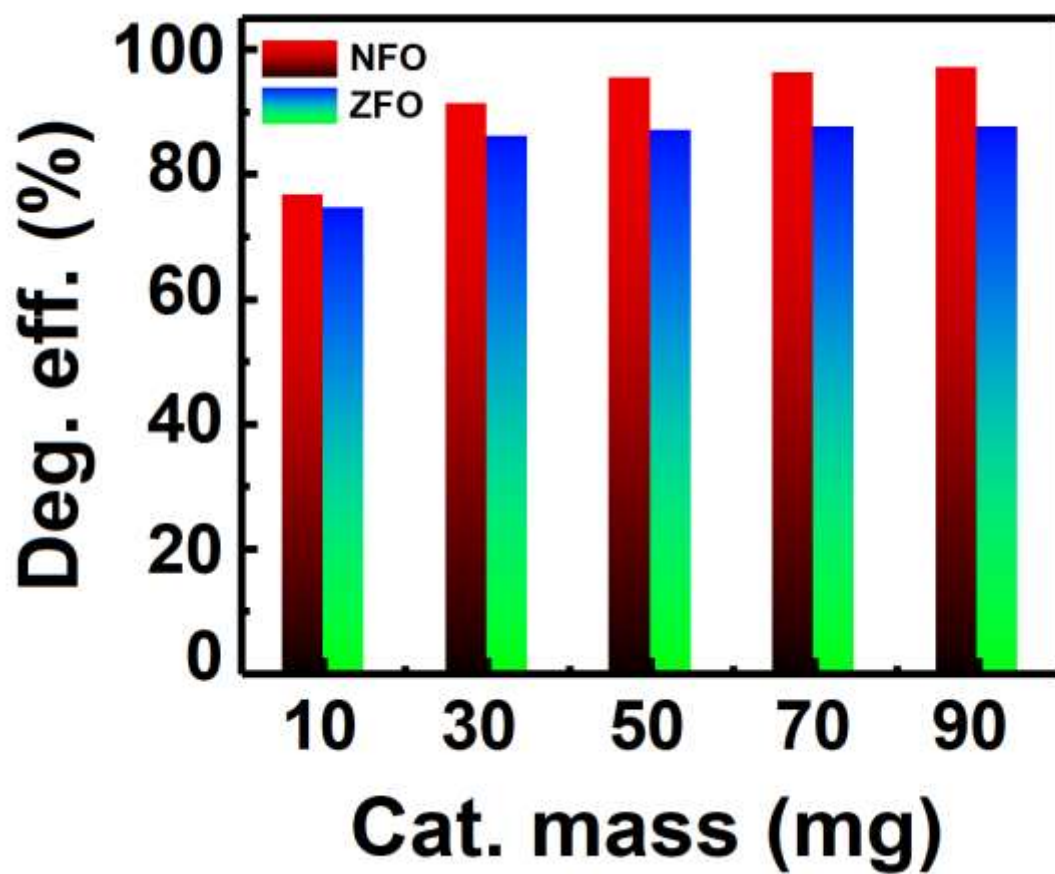
**Figure S3.** HRTEM images, FFT and the line profiles of the SNFO sheets.



**Figure S4.** The hydrodynamic particle size distribution of the magnetic SNFO sheets.

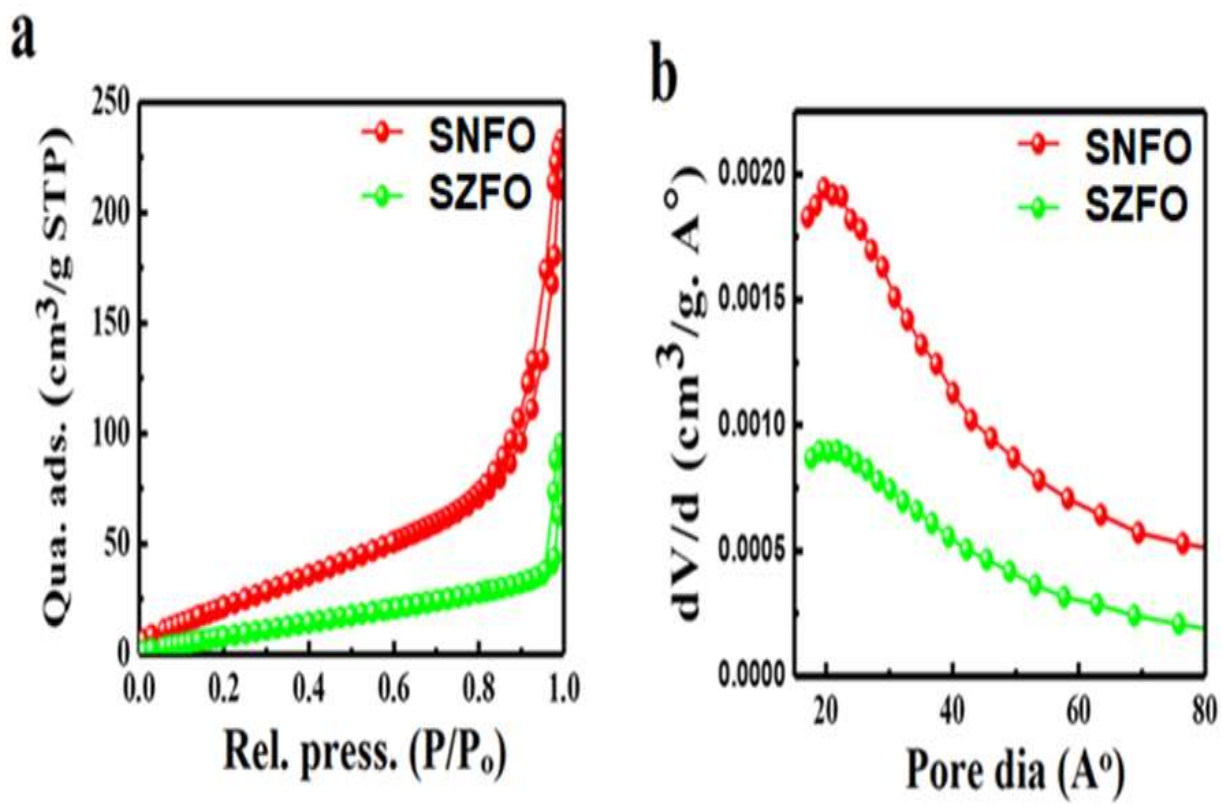


**Figure S5** (a) UV-Vis spectra recorded after the degradation with time; (b) the kinetics plot for the TCH antibiotic degradation with SZFO under the MW irradiation.



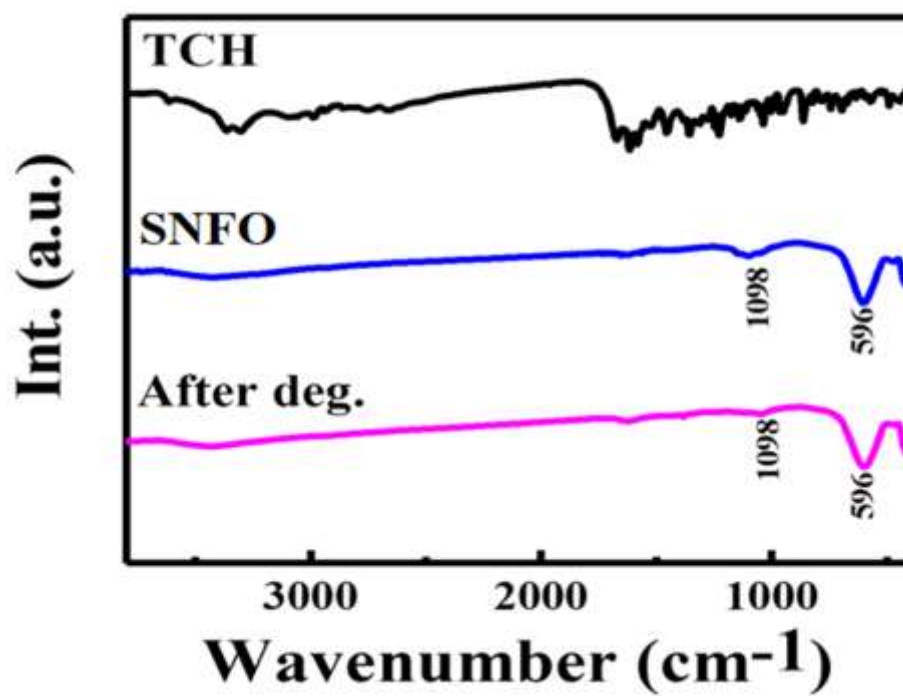
**Figure S6.** The effect of the catalyst mass on the TCH antibiotic degradation using SNFO and SZFO catalysts.



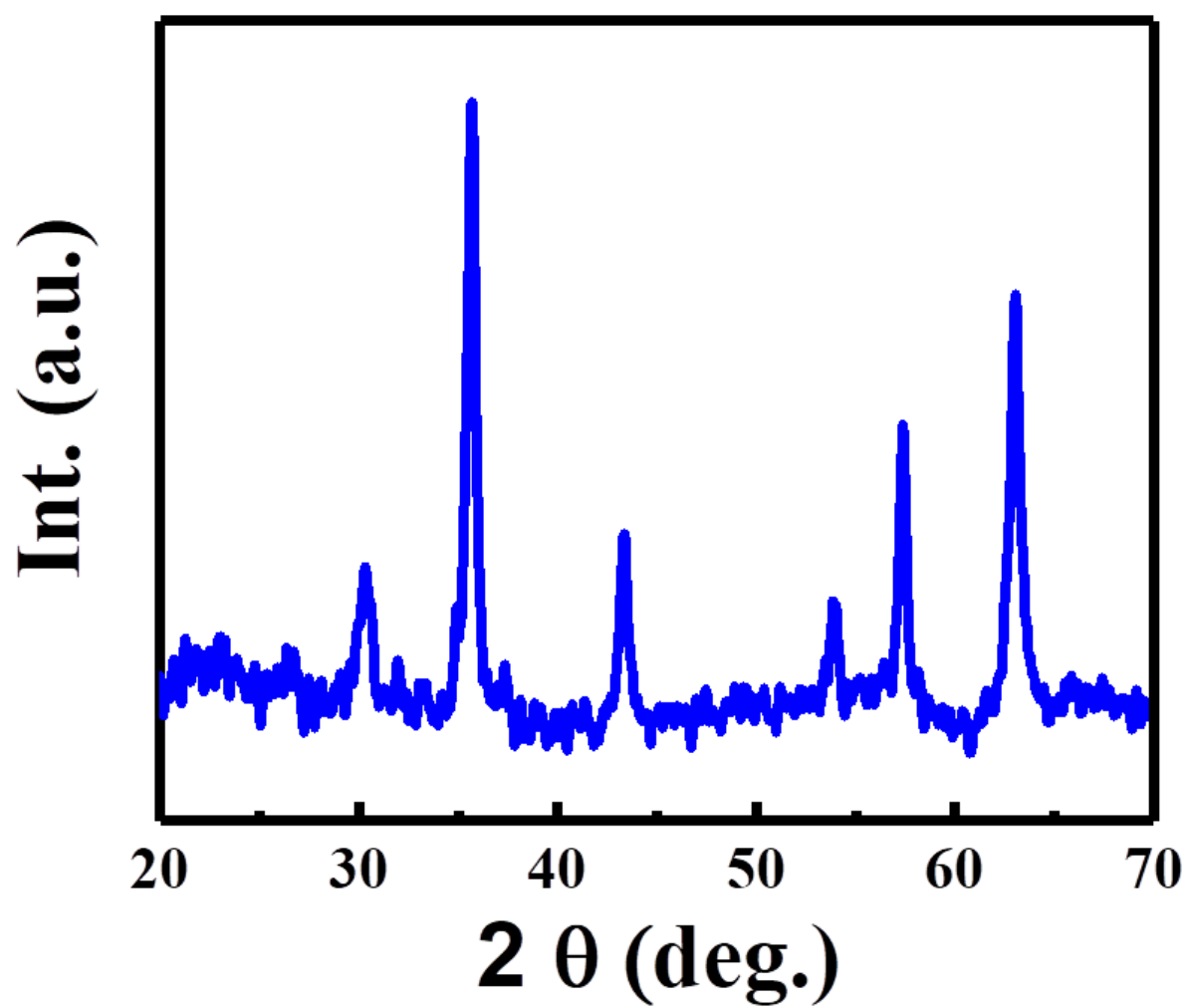


**Figure S7** (a)  $N_2$  adsorption-desorption isotherm; (b) pore volume vs pore diameter plots for SNFO and SZFO catalysts.

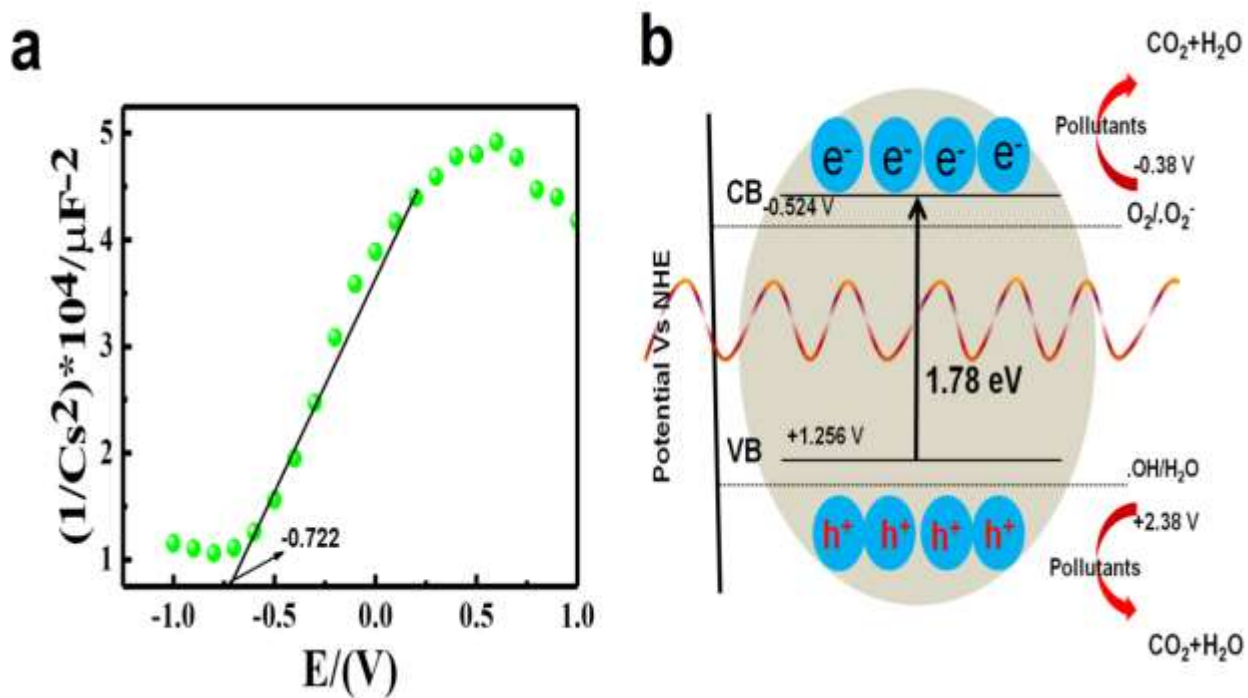




**Figure S8.** FTIR spectra of the TCH antibiotic molecule anchored on the SNFO catalyst before and after the degradation.



**Figure S9.** XRD pattern of SNFO after the MW treatment.



**Figure S10** (a) The Mott–Schottky plots for SNFO; (b) the mechanism of generations of the electrons and holes under the MW irradiation for SNFO.

**Table S1.** The comparisons of the degradation efficiency and TOC removal of other methods reported in the literature with those of the present method for the degradataion of the TCH antibiotic.

<b>Catalyst</b>	<b>Method</b>	<b>pH</b>	<b>Power (W)</b>	<b>Initial Concentrat ion</b>	<b>Time (min)</b>	<b>Degradation efficiency (%)</b>	<b>TOC (%)</b>	<b>References</b>
Ti <sub>4</sub> O <sub>7</sub>	Ozonation	7	NG*	5 mg/L	80	77.1	9.1	[64]
BiFeO <sub>3</sub>	Photo-Fenton	3	120	10 mg/L	120	40	21	[65]
Fe <sup>+2</sup> /H <sub>2</sub> O <sub>2</sub>	Ultrasound	6	100	100 mg/L	60	85.1	18.7	[66]
TiO <sub>2</sub>	Photocatalysis	6	120	40 mg/L	120	75	15	[67]
NiFe <sub>2</sub> O <sub>4</sub>	Photocatalysis	NG*	150	10 mg/L	90	47.43	NG*	[68]
Boron-doped diamond(BDD)/carbon-felt electrode, Fe <sup>3+</sup> , Fe <sup>2+</sup>	Electro-Fenton	7	18	100 mg/L	120	87.5	15	[69]
SNFO	MW assisted method	6	700	10 mg/L	15	90	40	Present work

\* NG=Not Given

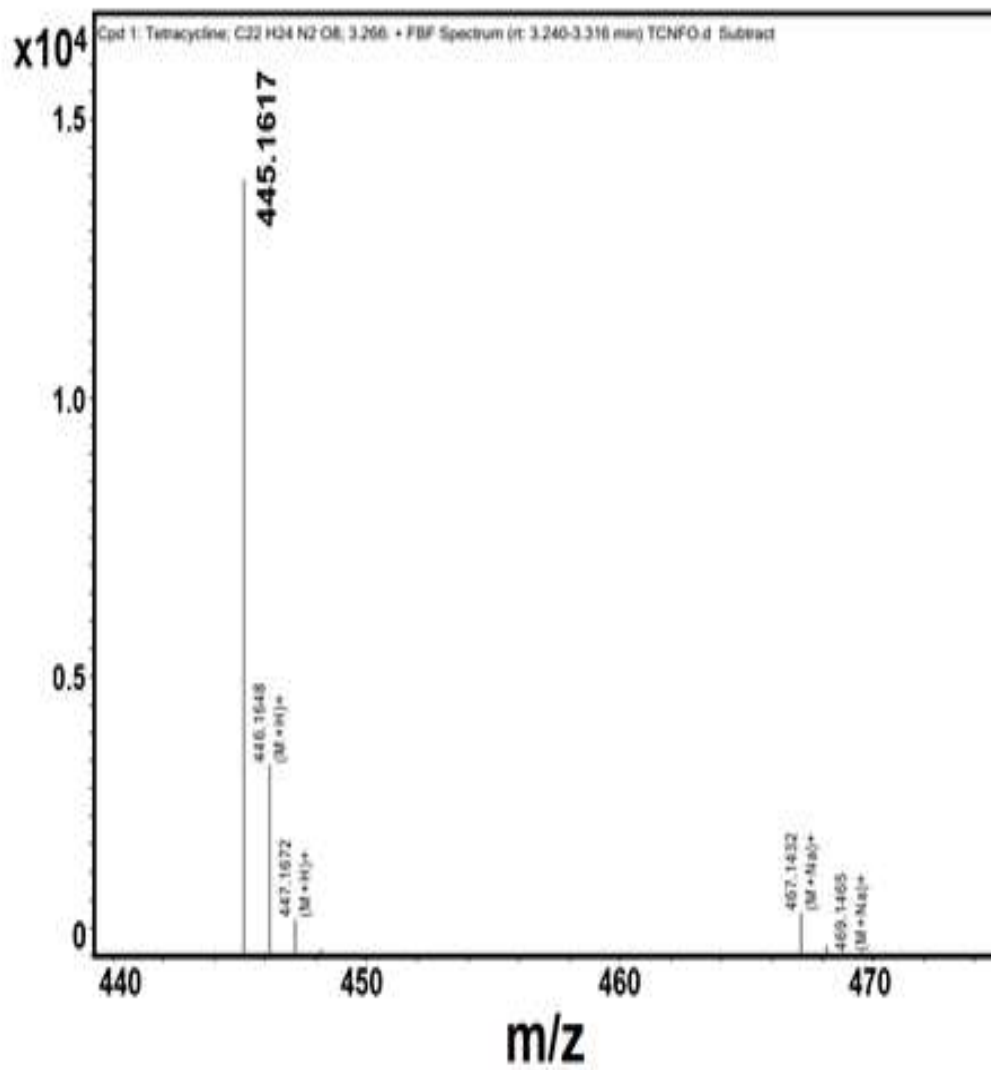


Figure S11 (a)

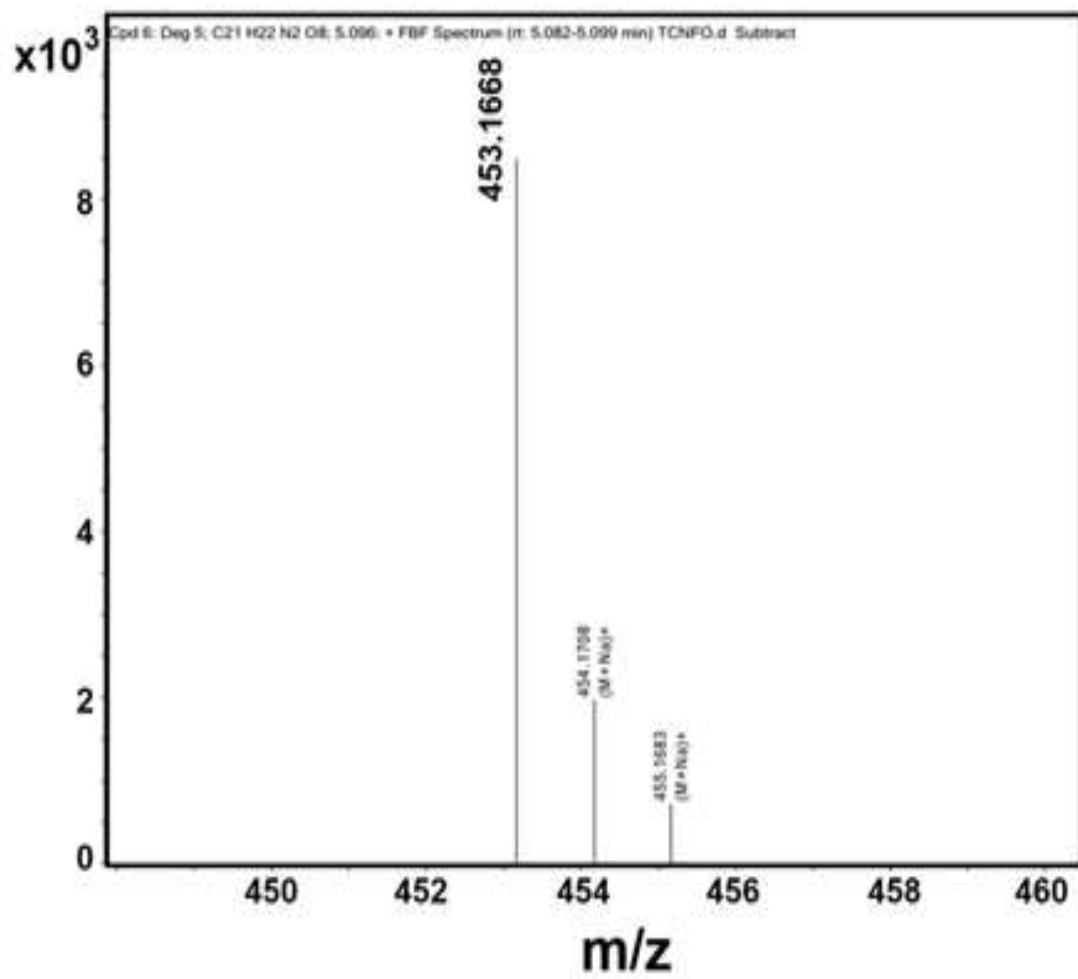


Figure S11(b)

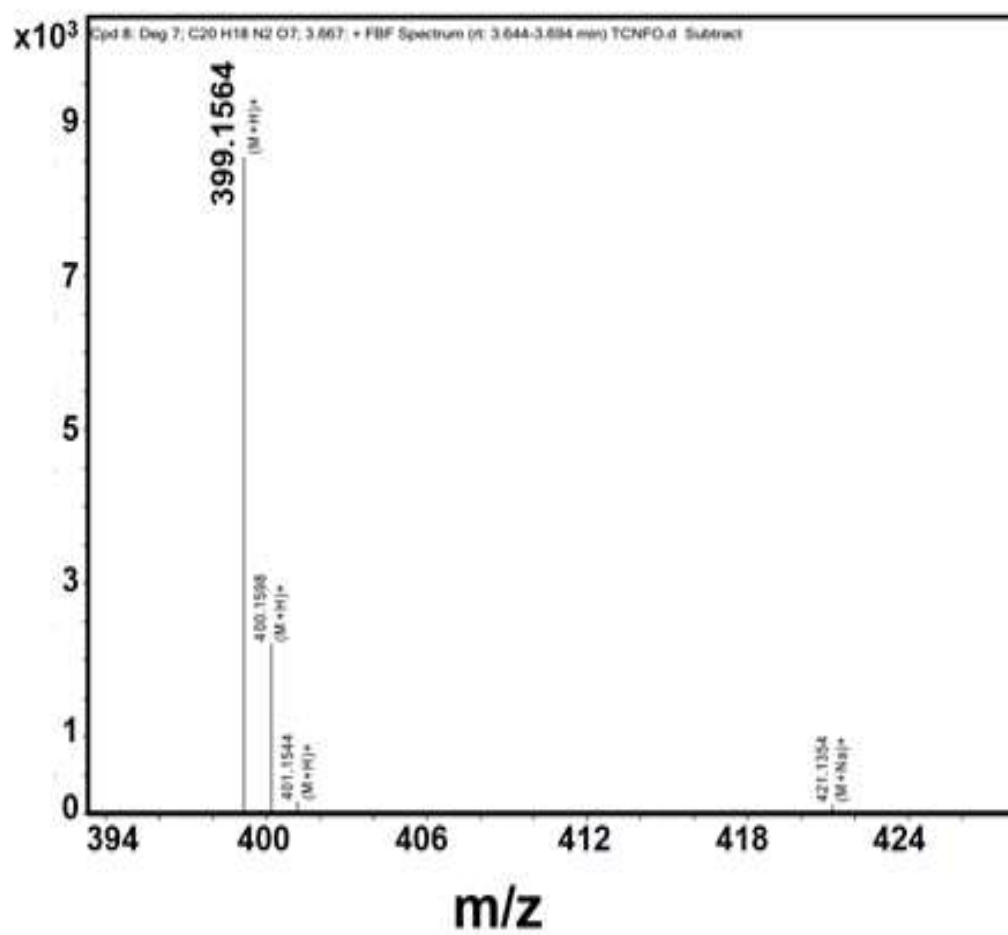


Figure S11 (c)



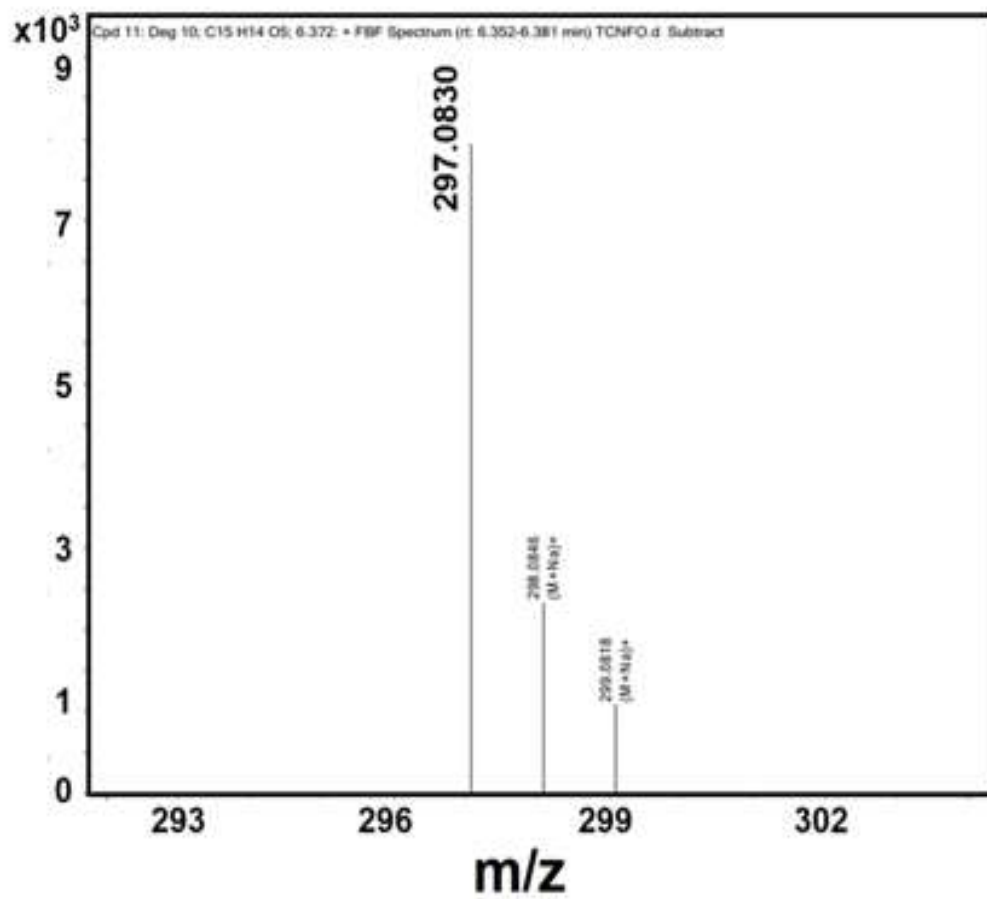


Figure S11 (d)

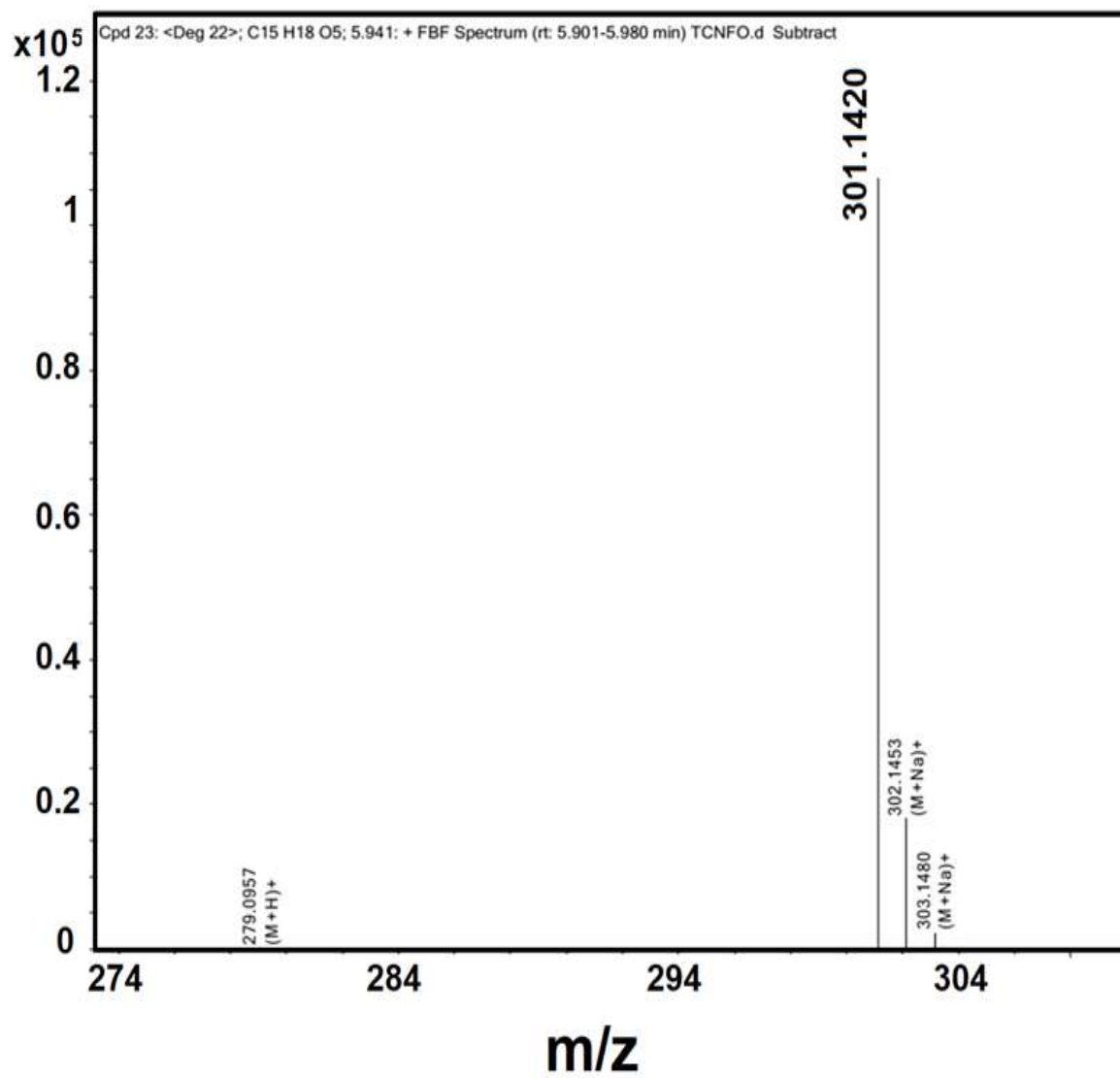


Figure S11(e)

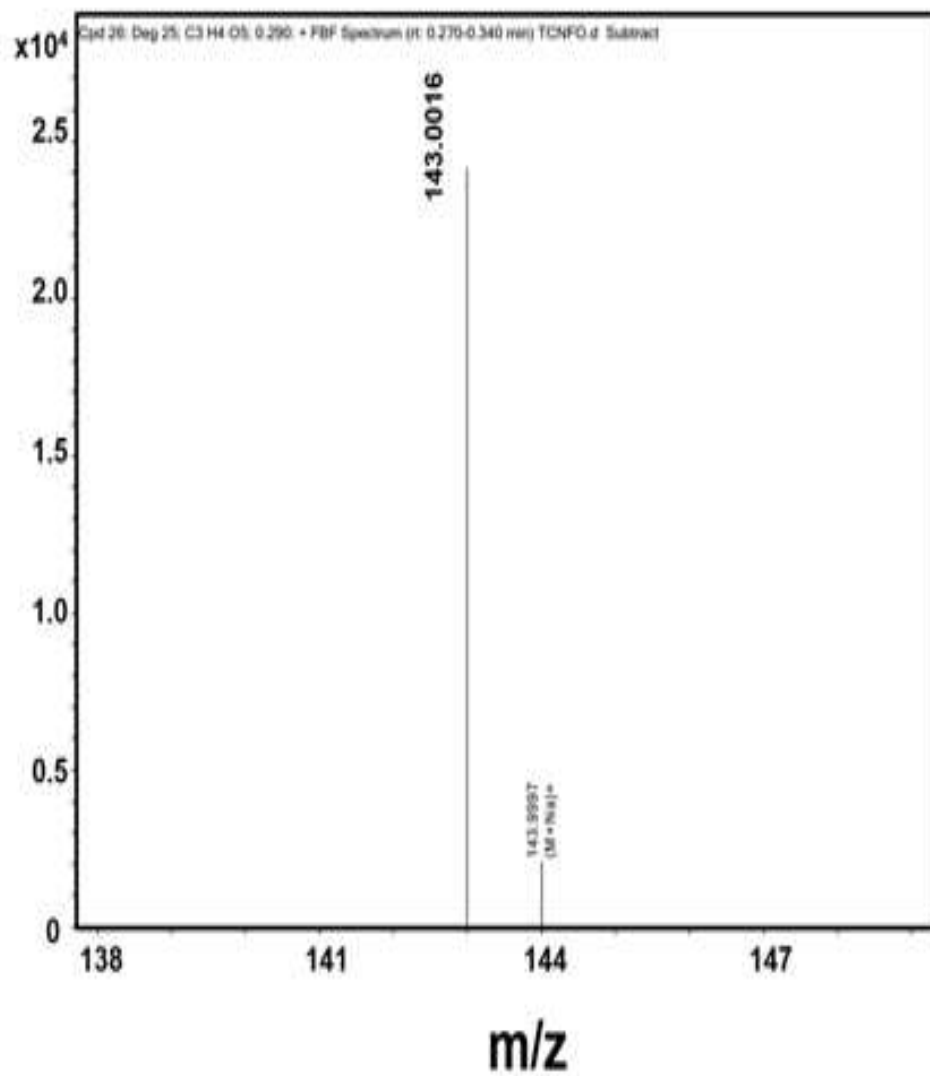


Figure S11(f)

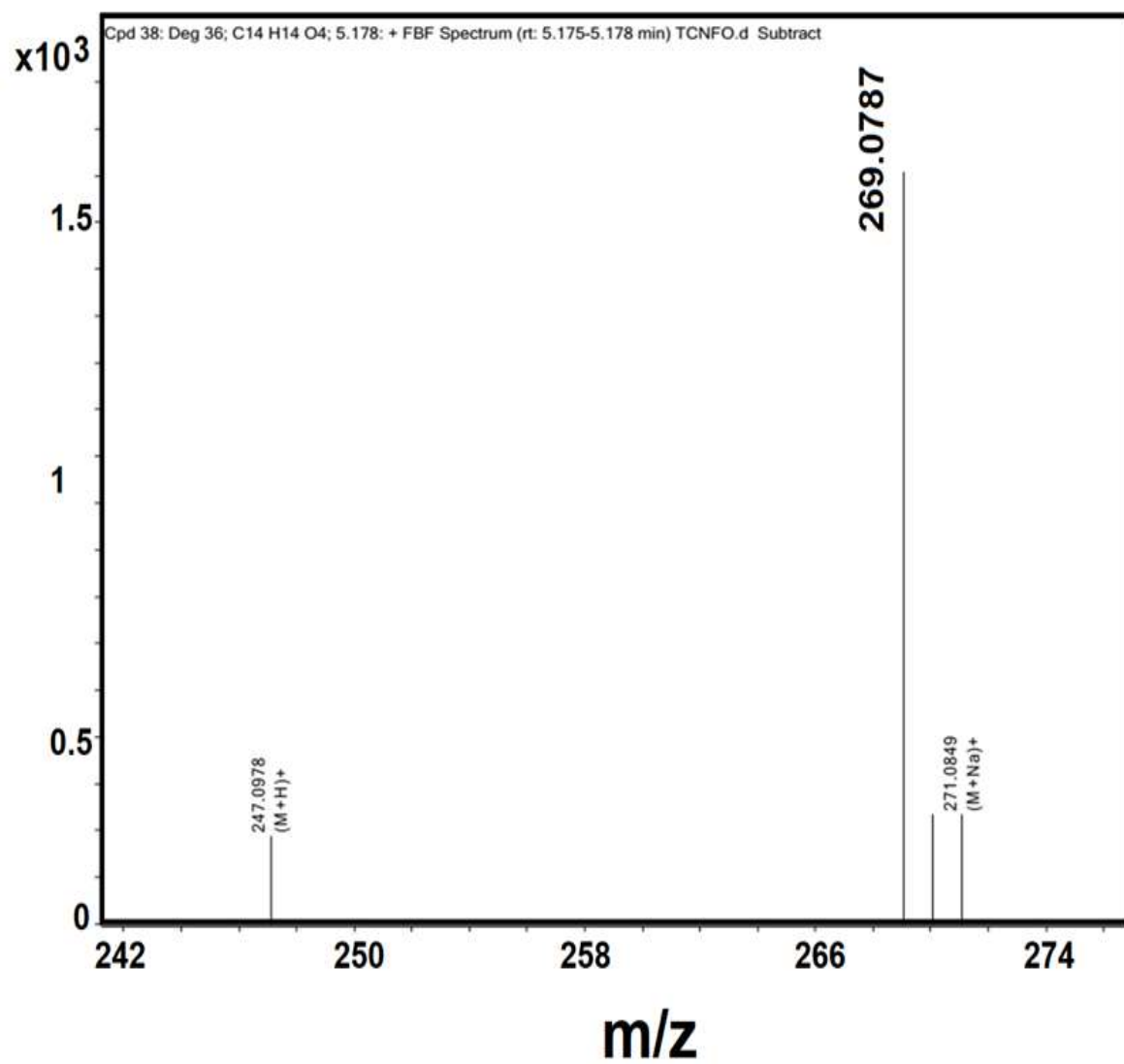
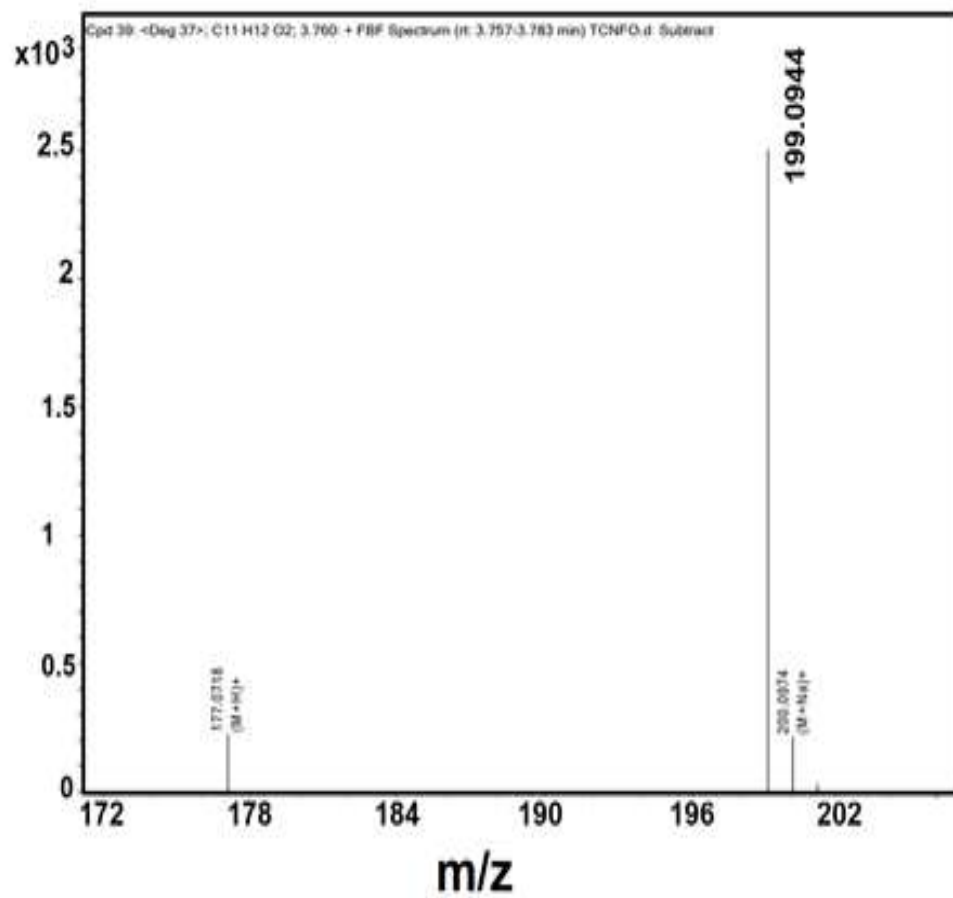
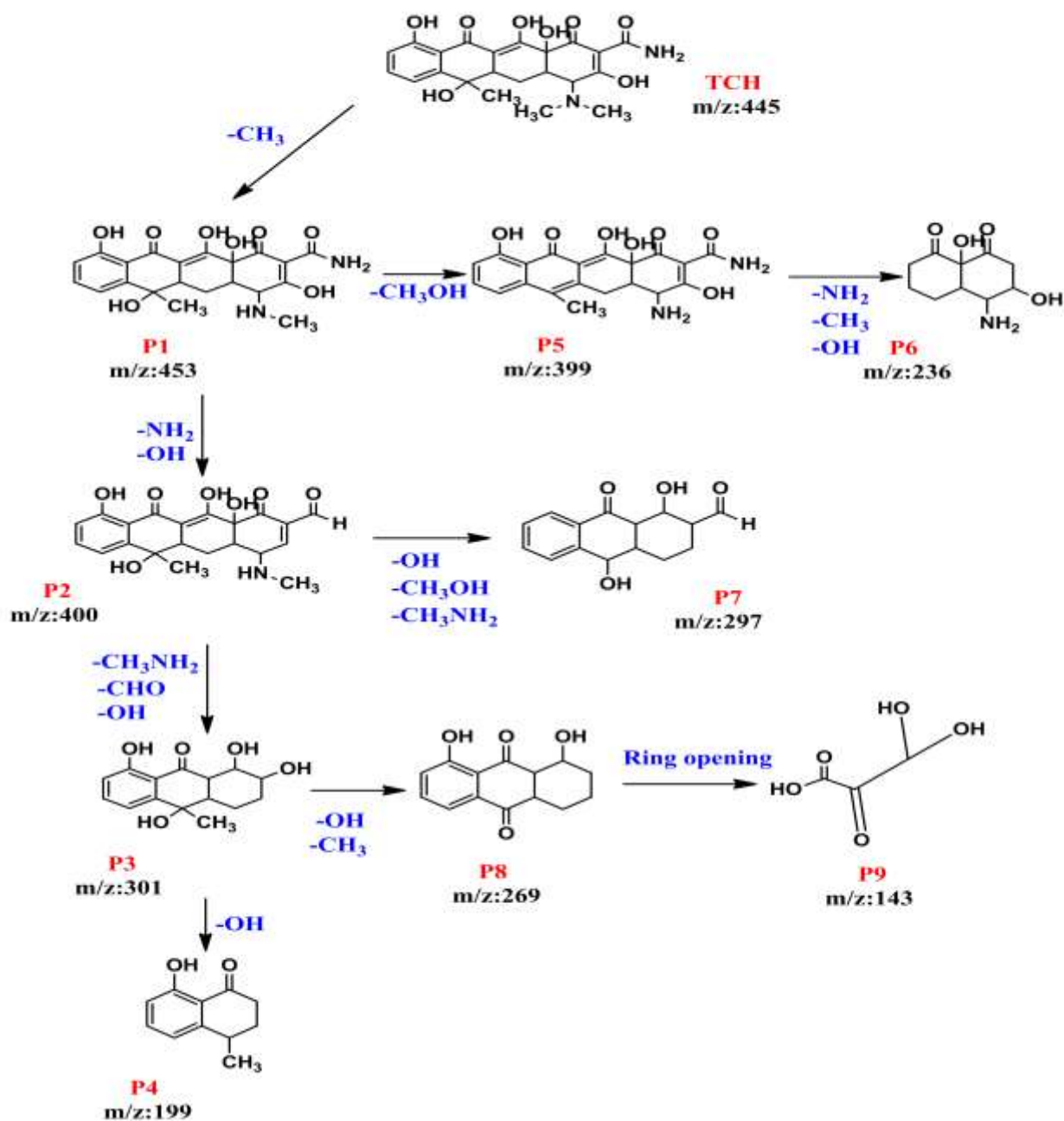


Figure S11(g)



**Figure S11(h)**

**Figure S11.** The mass spectra for the intermediates are shown for (a-h) at various time intervals during the degradation of the TCH antibiotic using the SNFO catalyst.



**Figure S12.** The proposed reaction pathways for the catalytic degradation of the TCH antibiotic molecule using the SNFO catalyst.

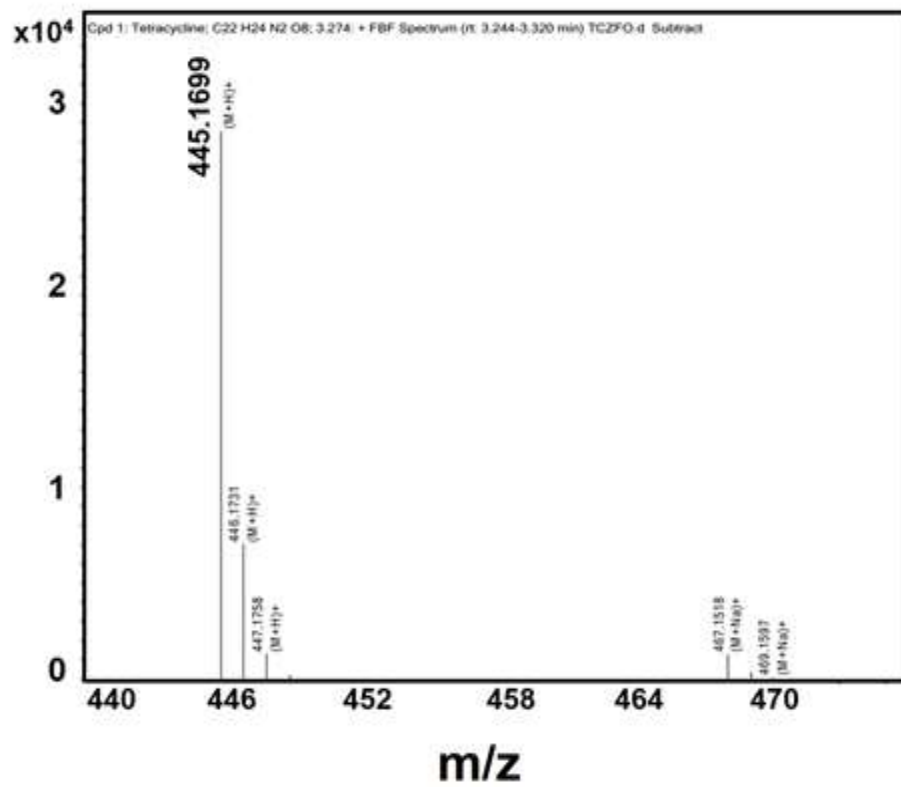


Figure S13(a)



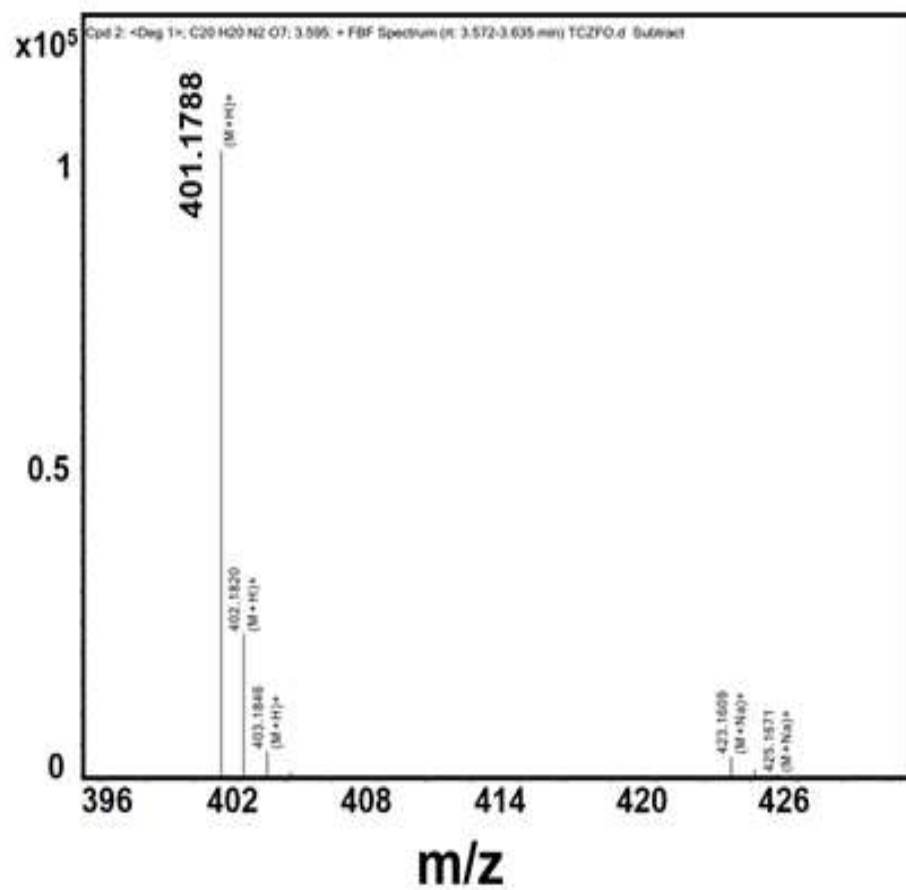


Figure S13(b)

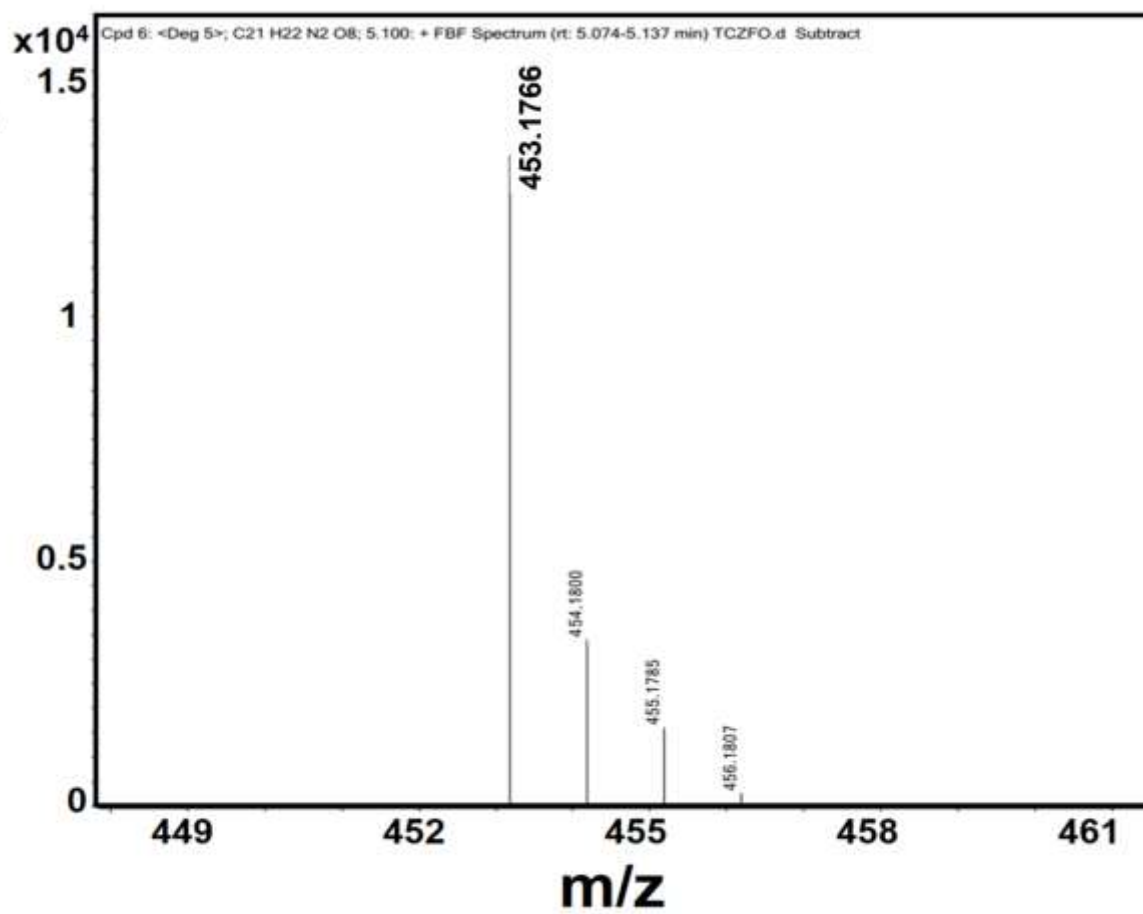


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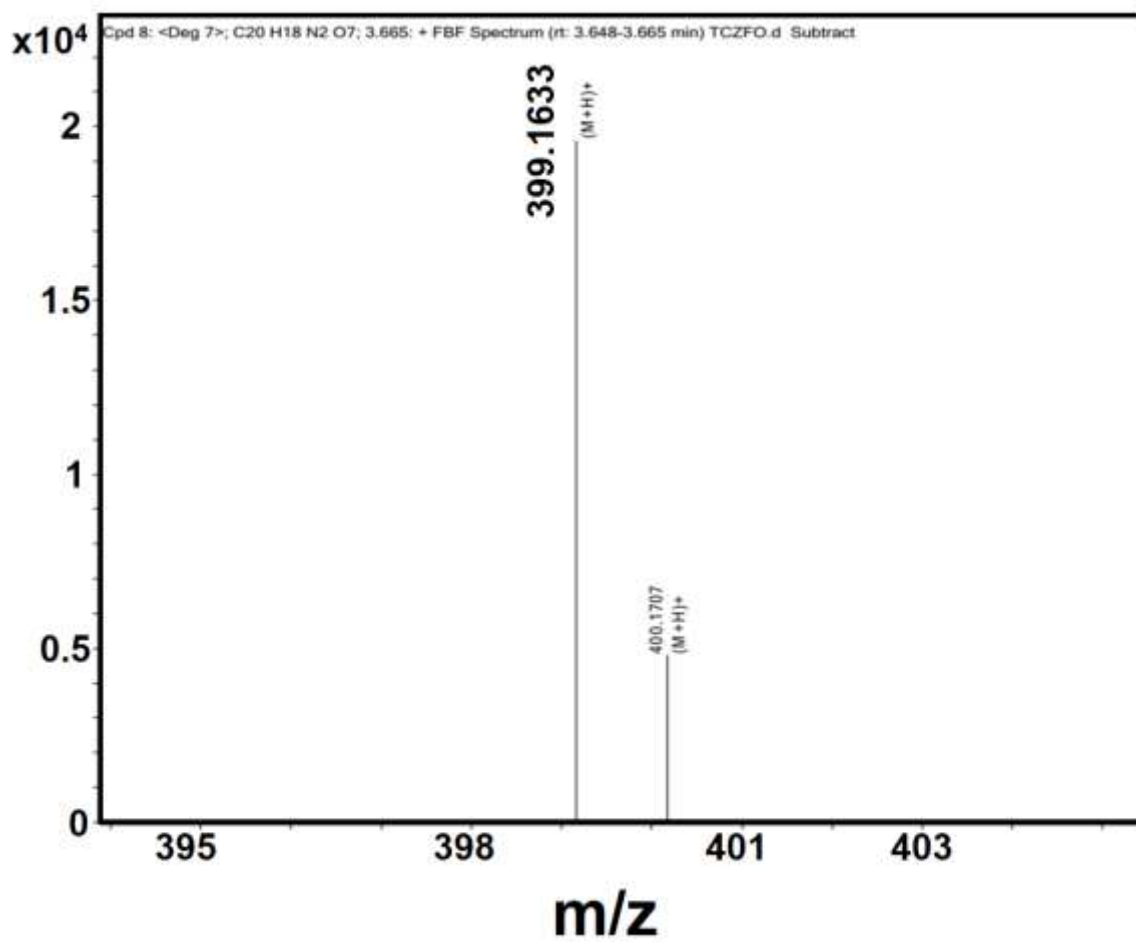


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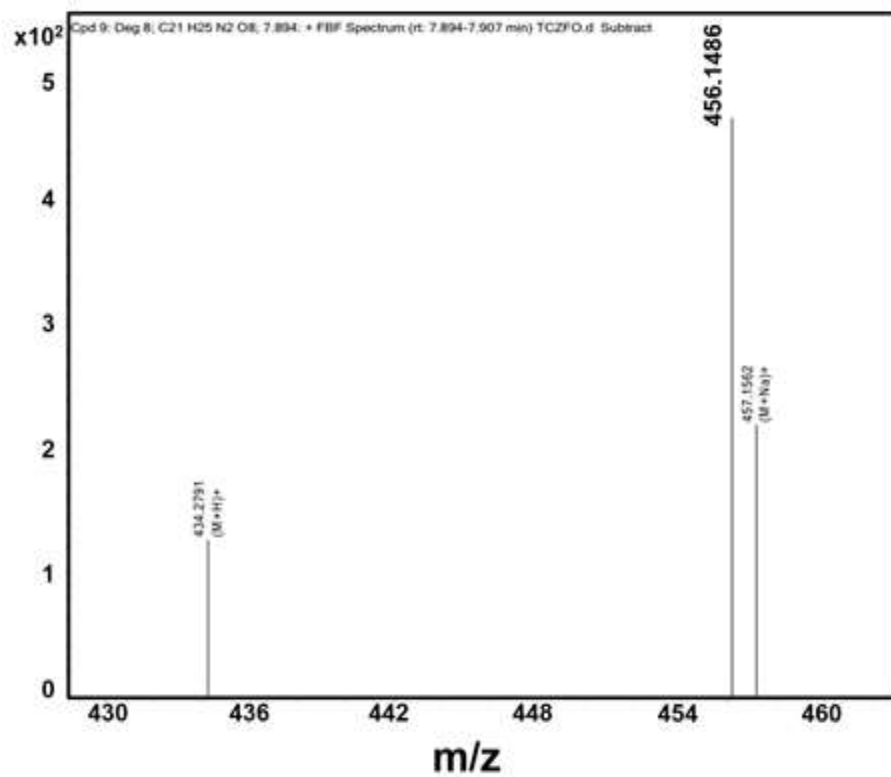


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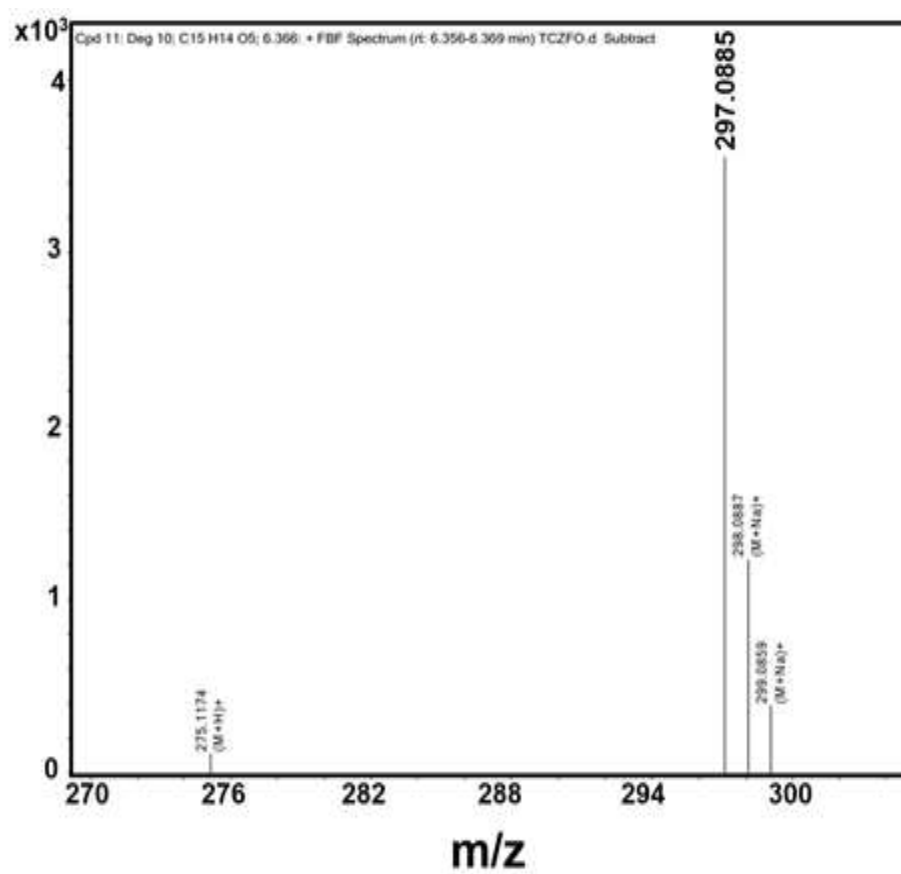


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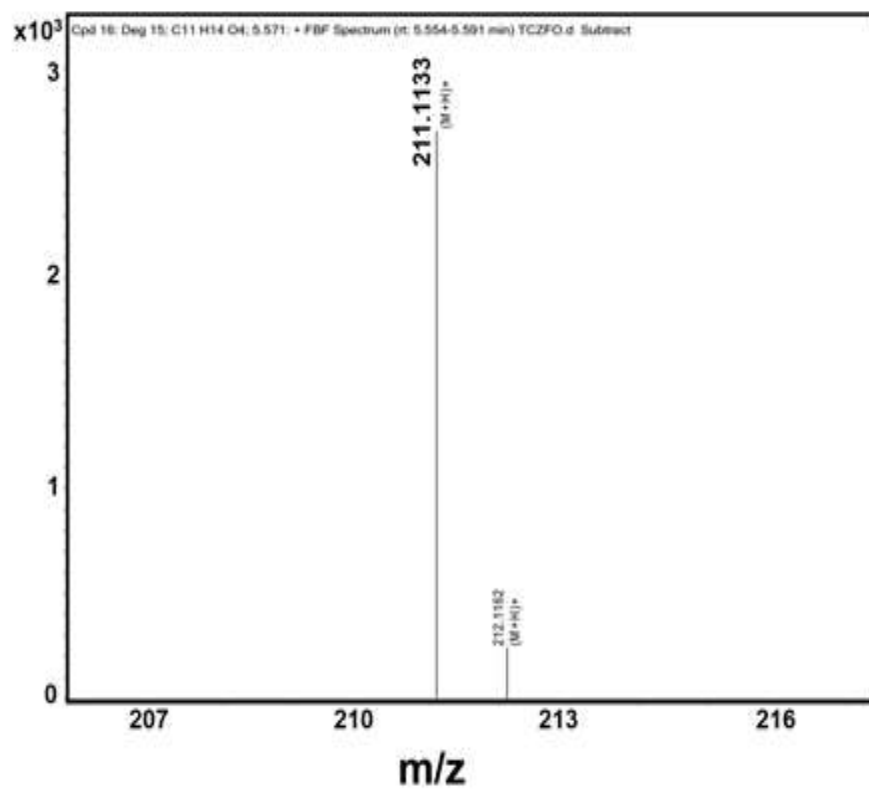


Figure S13(g)

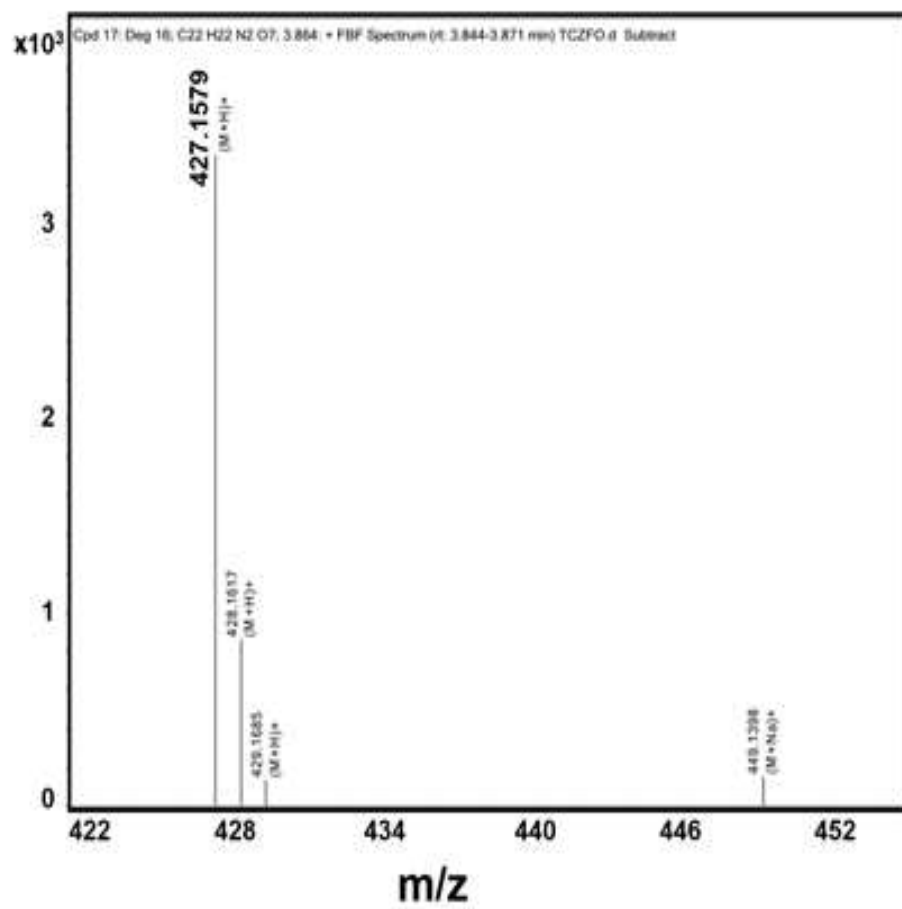


Figure S13 (h)



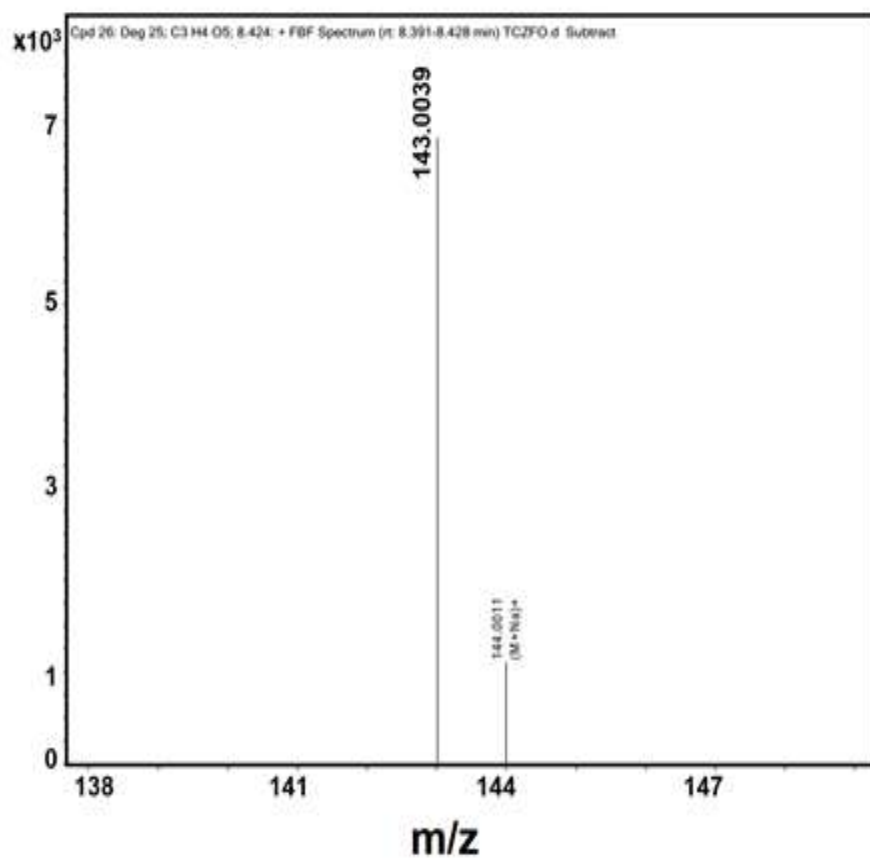


Figure S13(i)

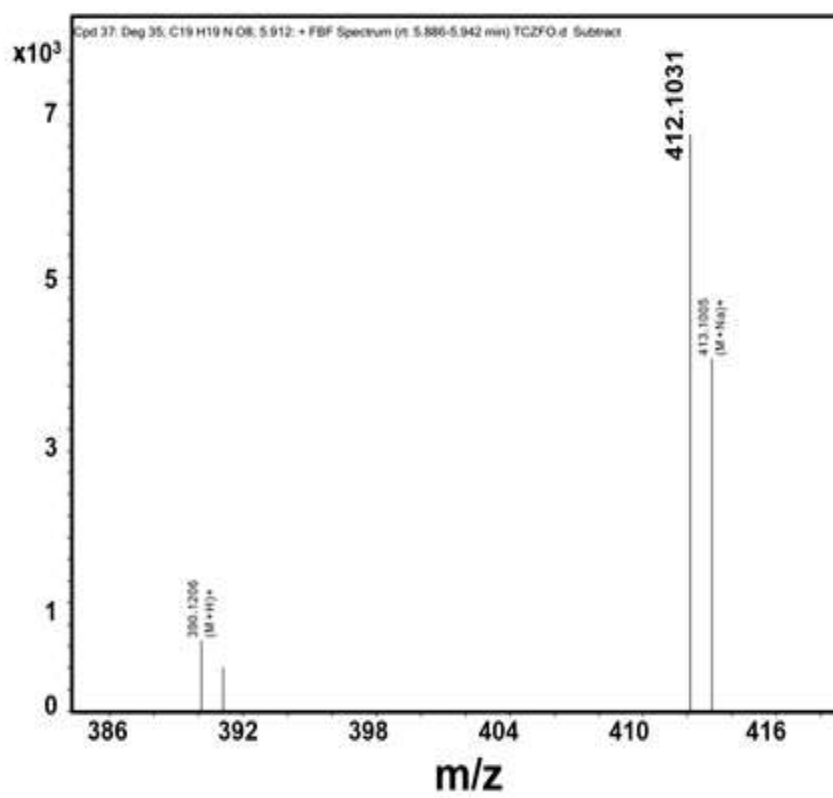


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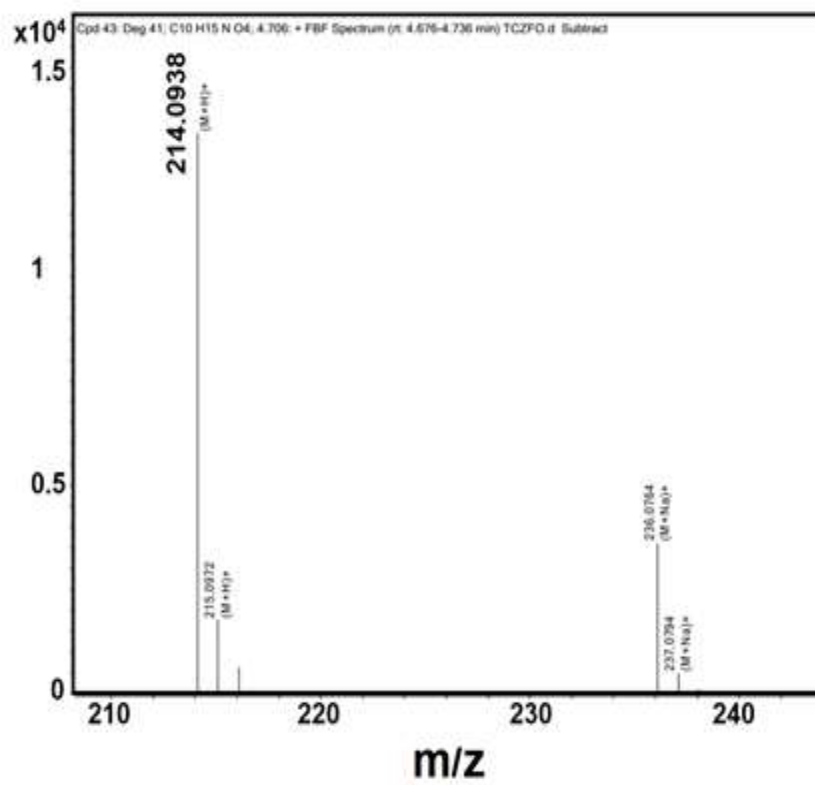
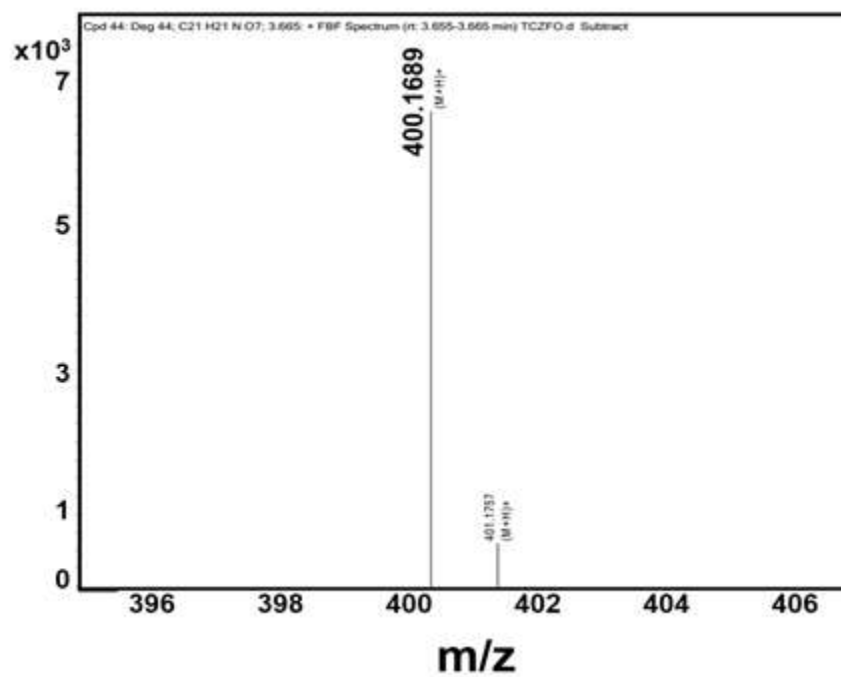
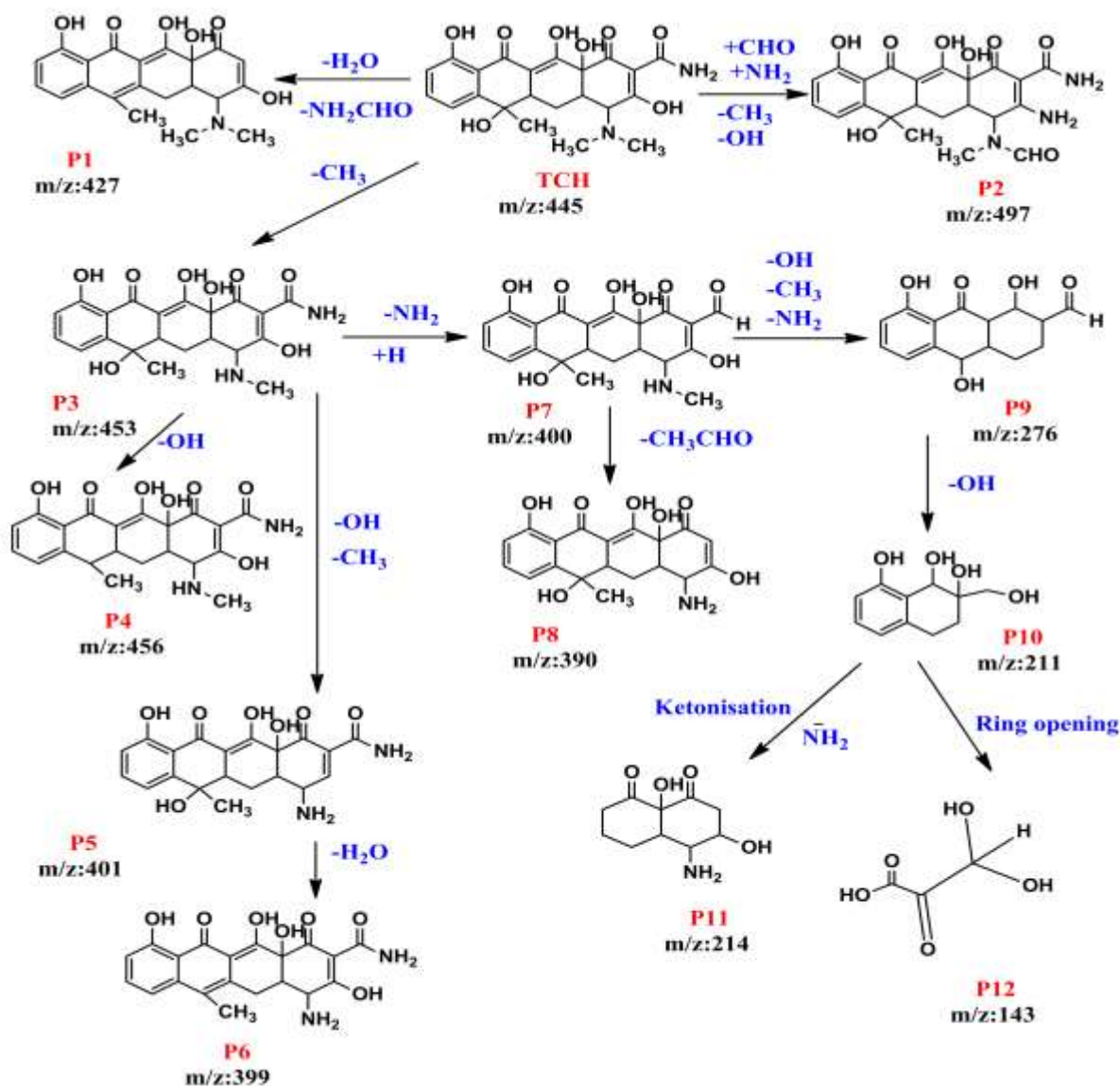


Figure S13 (k)



**Fig. S13 (l)**

**Figure S13.** The mass spectra for the intermediates are shown for (a-l) at various time intervals during the degradation of the TCH antibiotic using the SZFO catalyst.



**Figure S14.** The proposed reaction pathways for the catalytic degradation of the TCH antibiotic molecule using the SZFO catalyst.