# Supporting Information

# Discovery And Optimization of Novel SUCNR1 Inhibitors: Design of Zwitterionic Derivatives with Salt Bridge for Improvement of Oral Exposure

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#### **TABLE OF CONTENTS**

HPLC and NMR charts for compound 20	S2
Extended Data Table 1 for <b>20</b> bound to humanized rat SUCNR1	S4
Coordination data for crystal structures of 9, 10 and 20	S6





### EXTENDED DATA TABLE S1

Data collection and refinement statistics for the humanized ratSUCNR1-Nanobody6-20 complex.

	Humanized ratSUCNR1-
	Nanobody6-20 <sup>#</sup>
Data collection	
Space group	C2
Cell dimensions	
<i>a</i> , <i>b</i> , <i>c</i> (Å)	76.84, 151.0, 68.21
<i>α, β,</i> γ (°)	90, 112.29, 90
Wavelength (Å)	0.999
Wilson B-factor (Å <sup>2</sup> )	52.1
Reflections	139197 (7966)
Resolution (Å) <sup>a</sup>	63.12-2.27 (2.44-2.27)
R <sub>pim</sub>	0.089 (1.184)
Ι/σ(Ι)	5.6 (0.6)
<i>CC</i> <sub>1/2</sub>	0.995 (0.214)
Completeness (%)	

spherical	77.3 (19.5)
ellipsoidal	90.6 (45.2)
Redundancy	5.4 (6.2)

#### Refinement

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Resolution (Å)	63.12-2.27
No. reflections	25647
R <sub>work</sub> / R <sub>free</sub>	20.38 / 22.26
No. atoms	
Protein	3268
Ligands	255
Water	154
B factors	
Protein	59.20
Ligand/ion	79.13
Water	63.96
r.m.s deviations	

Bond lengths (Å)	0.008
Bond angles (°)	0.95
Ramachandran (%)	
favored	98.76
allowed	1.24
outliers	0

<sup>#</sup> Data obtained from a single crystal

<sup>a</sup> Values in parentheses are for highest-resolution shell.

#### **COORDINATION DATA FOR CRYSTAL STRUCTURES OF 9, 10 AND 20**

Compound 9 was crystallized as a hydrochloride from methanol by slow evaporation of the solvent, compound 10 was crystallized as a hydrochloride from acetonitrile / hydrochloric acid, compound 20 was crystallized as hydrochloride from a mixture of ethanol and isobutylmethylketone. Intensity data were collected at 100 K on a Bruker AXS three-circle diffractometer with monochromated Cu(K<sub>a</sub>)radiation (multilayer mirrors), microfocus rotating anode generator, and a Smart 6000 CCD detector using the SMART software. Several  $\omega$ -scans at different  $\Phi$ -positions were performed to ensure appropriate data redundancy. Data processing and global cell refinement were performed with Saint V7.36A. A semi-empirical absorption correction was applied, based on the intensities of symmetryrelated reflections measured at different angular settings (SADBS V2008/1). The structures were solved by dual space-recycling methods and subsequent DF syntheses and refined based on full-matrix leastsquares on F2 using the SHELXTL program suite. Anisotropic displacement parameters were used for all non-hydrogen atoms. Hydrogen atoms were located in DF maps and refined in idealized positions using a riding model. For **10**, an isolated residual electron density peak was refined as partially occupied water molecule without hydrogen atoms. Crystal data, data collection parameters, and convergence results are listed in Table **S2** Crystallographic data (excluding structure factors) have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 2009710 (**9**), CCDC 2009711 (**10**) and CCDC 2009712 (**20**). Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax (+44) 1223 336033, email: deposit@ccdc.cam.ac.uk]. This material is available free of charge via the internet at http://pubs.acs.org.

## Extended Data Table S2

Identification code	9	10	20
Empirical formula	C22 H22 CI N O3	C22 H20 Cl2 N2 O4	C32 H35 Cl2 F3 N2 O5
Formula weight	383.86	447.30	655.52
Temperature		100(2) K	
Wavelength		1.54178 Å	
Crystal system	monoclinic	orthorhombic	orthorhombic
Space group	P21/n	P212121	P212121
Unit cell dimensions	<i>a</i> = 19.865(2) Å	a = 4.989(3) Å	a = 7.396(3) Å
	b = 4.453(5) Å	b = 14.365(7) Å	b = 13.040(6) Å
	<i>c</i> = 21.536(2) Å	c = 28.185(14) Å	c = 32.182(15) Å
	$\beta = 99.147(5)$ °		
Volume	1881(2) Å <sup>3</sup>	2019.9(19) Å <sup>3</sup>	3104(2) Å <sup>3</sup>
Z	4	4	4
Density (calculated)	1.356 g/cm <sup>3</sup>	1.471 g/cm <sup>3</sup>	1.403 g/cm <sup>3</sup>
Absorption coefficient	1.982 mm <sup>-1</sup>	3.175 mm <sup>-1</sup>	2.415 mm <sup>-1</sup>
F(000)	808	928	1368
Crystal size [mm3]	0.28x0.03x0.02	0.30x0.02x0.01	0.19x0.11x0.02
Theta range	2.81 to 68.22°	3.14 to 68.31°	2.75 to 68.52°
Index ranges	-23,23; -4,5; -25,25	-5,5; -17,17; -33,33	-8,8; -15,15; -38,38
Reflections collected	37330	41796	61328
Independent reflections	3410, R(int) = 0.0721	3661, R(int) = 0.0576	5683, R(int) = 0.1207

Completeness	98.6 %	99.6 %	99.9 %
Absorption correction	Semi-empirical from equivalents		5
Max/min. transmission	0.9614 and 0.6069	0.9689 and 0.4493	0.9533 and 0.6569
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Data/restraints/param.	3410 / 0 / 246	3661 / 0 / 269	5683 / 0 / 401
Goodness-of-fit on F <sup>2</sup>	1.040	1.036	1.137
Final R [I>2s(I)]	R1 = 0.0363	R1 = 0.0312	R1 = 0.0669
	wR2 = 0.0871	wR2 = 0.0754	wR2 = 0.1616
R indices (all data)	R1 = 0.0471	R1 = 0.0347	R1 = 0.0725
	wR2 = 0.0934	wR2 = 0.0772	wR2 = 0.1643
Abs. structure parameter	n.a.	0.451(13)	0.08(3)
Largest diff. peak hole	0.31 / -0.24 e∙Å <sup>-3</sup>	0.20 / -0.15 e·Å⁻³	0.70 / -0.43 e∙Å <sup>-3</sup>



Figure S1: Structure of 9 in the crystal along with numbering scheme. Hydrogen bonds shown as dotted lines.



Figure S2: Structure of 10 in the crystal along with numbering scheme. Hydrogen bonds shown as dotted lines.



Figure S3: Structure of 20 in the crystal along with numbering scheme. Hydrogen bonds shown as dotted lines.