

# Supporting Information

## Highly efficient lead-free (Bi,Ce)-codoped $\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{InCl}_6$ double perovskites for white LEDs

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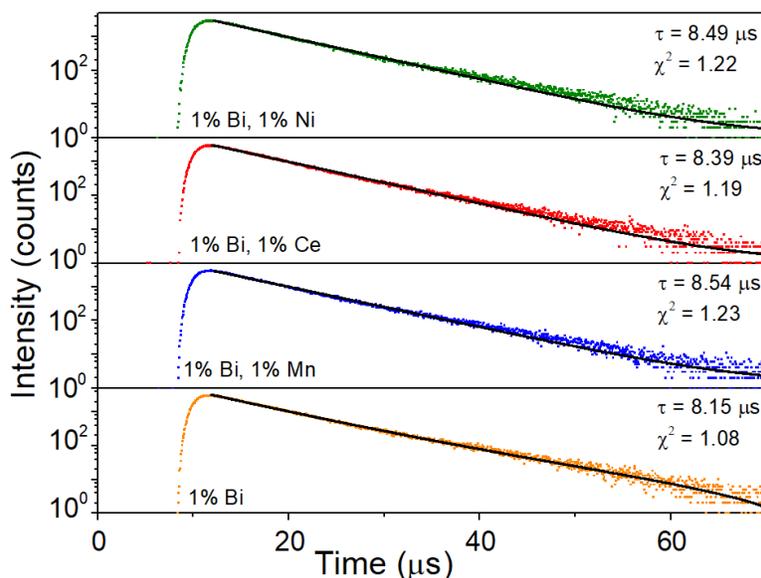
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**Figure S1.** Time resolved photoluminescence spectra of  $\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{InCl}_6:\text{X}$  ( $\text{X} = \text{Bi}; \text{Bi}, \text{Mn}; \text{Bi}, \text{Ce}; \text{Bi}, \text{Ni}$ ) perovskite phosphors ( $\lambda_{\text{ex}} = 350 \text{ nm}$ ).

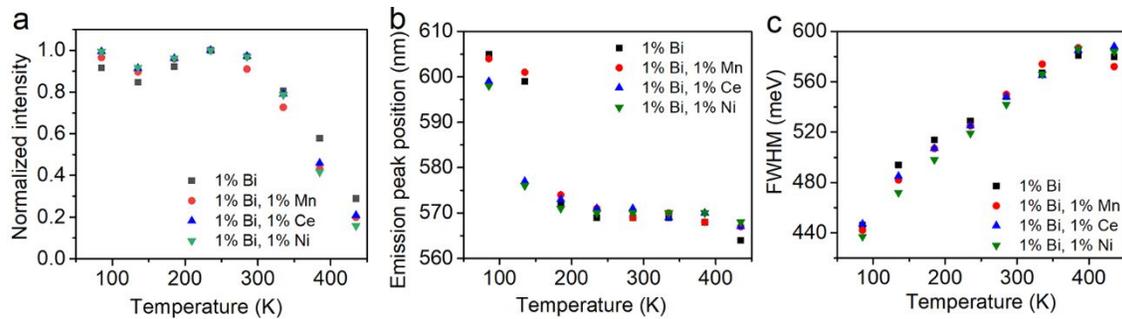
**Table S1.** Comparison of designed composition and measured composition by XPS for each sample.

Sample	Designed composition	Measured composition by XPS	Ag/In molar ratio measured by ICP-MS
Bi doped	$\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{In}_{0.99}\text{Cl}_6\text{Bi}_{0.01}$	$\text{Cs}_{1.92}\text{Ag}_{0.22}\text{Na}_{0.53}\text{In}_{0.99}\text{Cl}_{5.68}\text{Bi}_{0.016}$	0.32
(Bi,Mn)-codoped	$\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.61}\text{In}_{0.98}\text{Cl}_6\text{Bi}_{0.01}\text{Mn}_{0.01}$	$\text{Cs}_{1.79}\text{Ag}_{0.21}\text{Na}_{0.63}\text{In}_{0.98}\text{Cl}_{5.62}\text{Bi}_{0.015}\text{Mn}_{<0.01}$	0.47
(Bi,Ce)-codoped	$\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{In}_{0.98}\text{Cl}_6\text{Bi}_{0.01}\text{Ce}_{0.01}$	$\text{Cs}_{1.81}\text{Ag}_{0.22}\text{Na}_{0.71}\text{In}_{0.98}\text{Cl}_{5.80}\text{Bi}_{0.018}\text{Ce}_{0.018}$	0.40
(Bi,Ni)-codoped	$\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.61}\text{In}_{0.98}\text{Cl}_6\text{Bi}_{0.01}\text{Ni}_{0.01}$	$\text{Cs}_{1.92}\text{Ag}_{0.20}\text{Na}_{0.65}\text{In}_{0.98}\text{Cl}_{5.82}\text{Bi}_{0.015}\text{Ni}_{0.031}$	0.38

**Table S2.** Summary of calculation details for the total formation energies.

Chemical Formula	Atom Number	Total Energy ( $E_{\text{total}}$ , eV)	Formation Energy Equations <sup>a)</sup>
$\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{InCl}_6$	$\text{Cs}_{40}\text{Ag}_8\text{Na}_{12}\text{In}_{20}\text{Cl}_{120}$	144897.513	Referenced energy
$\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{Bi}_{0.05}\text{In}_{0.95}\text{Cl}_6$	$\text{Cs}_{40}\text{Ag}_8\text{Na}_{12}\text{BiIn}_{19}\text{Cl}_{120}$	145339.653	$E_{\text{total}} - E_{\text{ref}} - \mu_{\text{Bi}} + \mu_{\text{In}}$
$\text{Cs}_2\text{Mn}_{0.05}\text{Ag}_{0.4}\text{Na}_{0.55}\text{Bi}_{0.05}\text{In}_{0.95}\text{Cl}_6$	$\text{Cs}_{40}\text{MnAg}_8\text{Na}_{11}\text{BiIn}_{19}\text{Cl}_{120}$	146883.112	$E_{\text{total}} - E_{\text{ref}} - \mu_{\text{Mn}} + \mu_{\text{Na}} - \mu_{\text{Bi}} + \mu_{\text{In}}$
$\text{Cs}_2\text{Ce}_{0.05}\text{Ag}_{0.4}\text{Na}_{0.55}\text{Bi}_{0.05}\text{In}_{0.95}\text{Cl}_6$	$\text{Cs}_{40}\text{CeAg}_8\text{Na}_{11}\text{BiIn}_{19}\text{Cl}_{120}$	145466.068	$E_{\text{total}} - E_{\text{ref}} - \mu_{\text{Ce}} + \mu_{\text{Na}} - \mu_{\text{Bi}} + \mu_{\text{In}}$
$\text{Cs}_2\text{Ni}_{0.05}\text{Ag}_{0.4}\text{Na}_{0.55}\text{Bi}_{0.05}\text{In}_{0.95}\text{Cl}_6$	$\text{Cs}_{40}\text{NiAg}_8\text{Na}_{11}\text{BiIn}_{19}\text{Cl}_{120}$	148358.726	$E_{\text{total}} - E_{\text{ref}} - \mu_{\text{Ni}} + \mu_{\text{Na}} - \mu_{\text{Bi}} + \mu_{\text{In}}$

<sup>a)</sup>  $E_{\text{ref}} = -144897.513$  eV;  $\mu$  is chemical potential of an element,  $\mu_{\text{Bi}} = -1865.037$  eV;  $\mu_{\text{In}} = -1423.122$  eV;  $\mu_{\text{Na}} = -1159.212$  eV;  $\mu_{\text{Ce}} = -1283.719$  eV;  $\mu_{\text{Mn}} = -2707.688$  eV;  $\mu_{\text{Ni}} = -4182.087$  eV.



**Figure S2.** Temperature dependency of normalized PL integral intensity (a), emission peak position (b) and FWHM (c) in the temperature range of 85 - 435 K for  $\text{Cs}_2\text{Ag}_{0.4}\text{Na}_{0.6}\text{InCl}_6$  phosphors doped with 1% Bi; 1% Bi, 1% Mn; 1% Bi, 1% Ce and 1% Bi, 1% Ni.