

Elemental Sulfur-Incorporated Cyclizations of Pyrrolidines Leading to Thienopyrroles

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Table S1. Optimization of the reaction conditions.^a

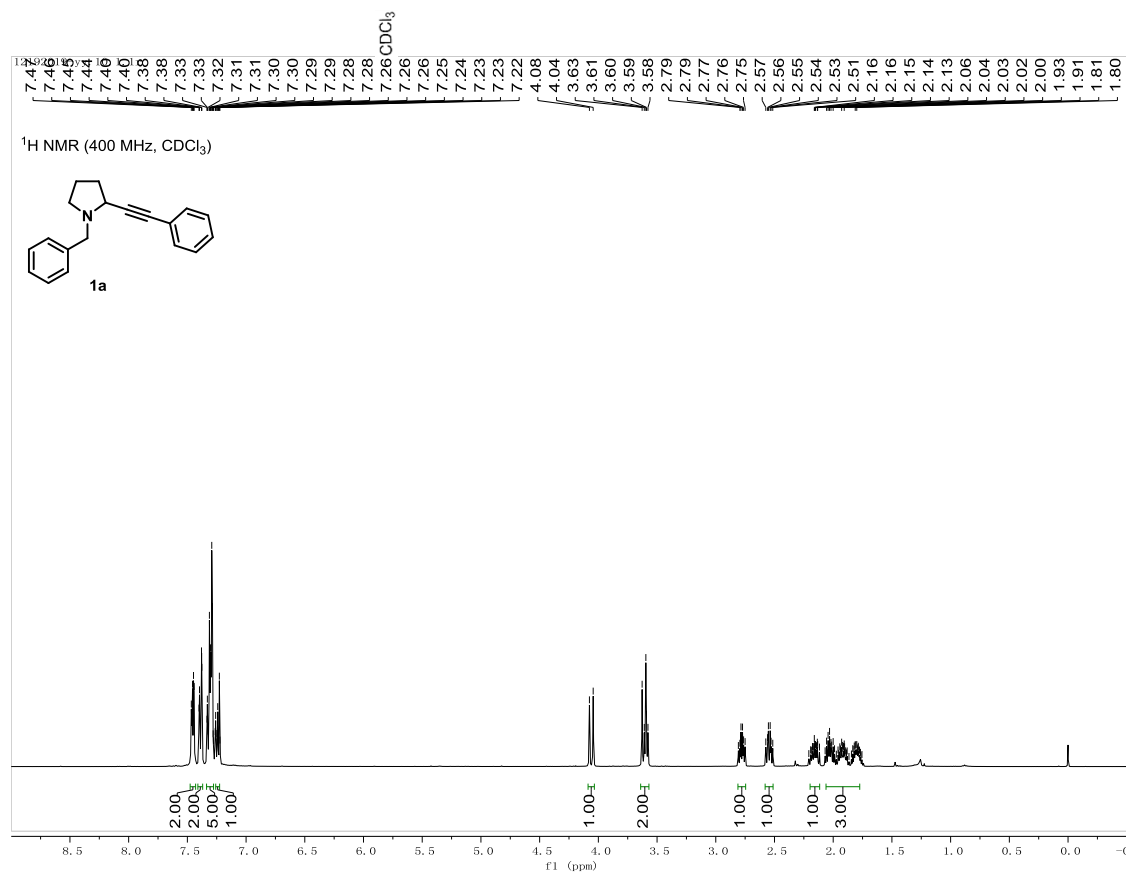
Entry	S ₈ (equiv)	Base (equiv)	Solvent	t (h)	T (°C)	Yield (%)	
						2a'	2a
1	S ₈ (1.0)	K ₂ CO ₃ (2.0)	1,4-dioxane	24	100	27	0
2	S ₈ (1.5)	K ₂ CO ₃ (2.0)	1,4-dioxane	24	100	41	8
3	S ₈ (1.0)	NaO ^t Bu (2.0)	1,4-dioxane	24	100	15	7
4	S ₈ (1.0)	K ₂ CO ₃ (2.0)	1,4-dioxane	36	100	22	4
5	S ₈ (1.0)	K ₂ CO ₃ (2.0)	1,4-dioxane	48	100	32	13
6	S ₈ (1.5)	K ₂ CO ₃ (2.0)	1,4-dioxane	36	100	30	15
7	S ₈ (1.5)	K ₂ CO ₃ (1.0)	1,4-dioxane	24	100	29	18
8	S ₈ (1.5)		1,4-dioxane	24	100	39	17
9	S ₈ (1.5)		1,4-dioxane	24	110	37	30
10	S₈ (1.5)		1,4-dioxane	24	120	0	74
11	S ₈ (1.5)		1,4-dioxane	36	100	0	64
13	S ₈ (1.5)		1,4-dioxane	36	110	0	64
14	S ₈ (1.5)		1,4-dioxane	36	120	0	63
15	S ₈ (1.0)		1,4-dioxane	24	120	0	43
16	S ₈ (0.50)		1,4-dioxane	24	120	0	41
17	S ₈ (1.5)		THF	24	120	19	55
18	S ₈ (1.5)		Toluene	24	120	0	37
19	S ₈ (1.5)		DMF	24	120	0	53
20	S ₈ (1.5)		DME	24	120	0	71

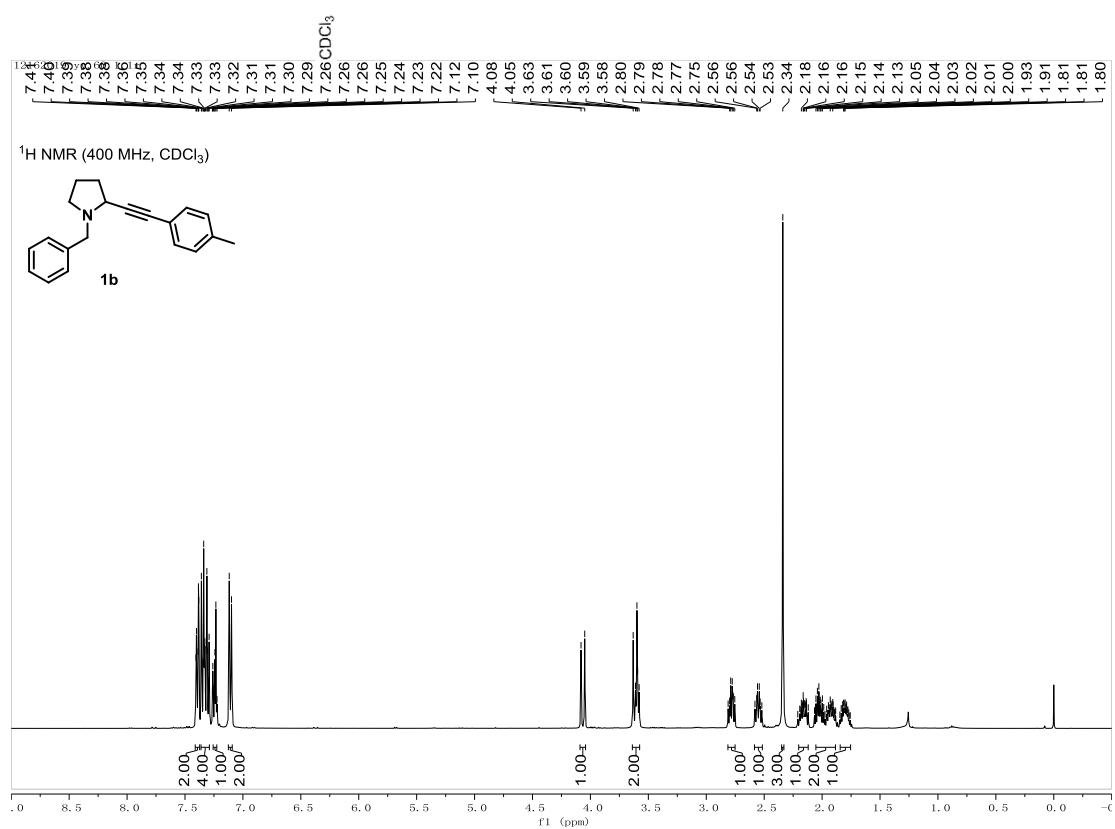
^a Standard condition: **1a** (0.30 mmol), S₈ (1.5 equiv), solvent (2.0 mL), N₂, 120 °C, 24 h.

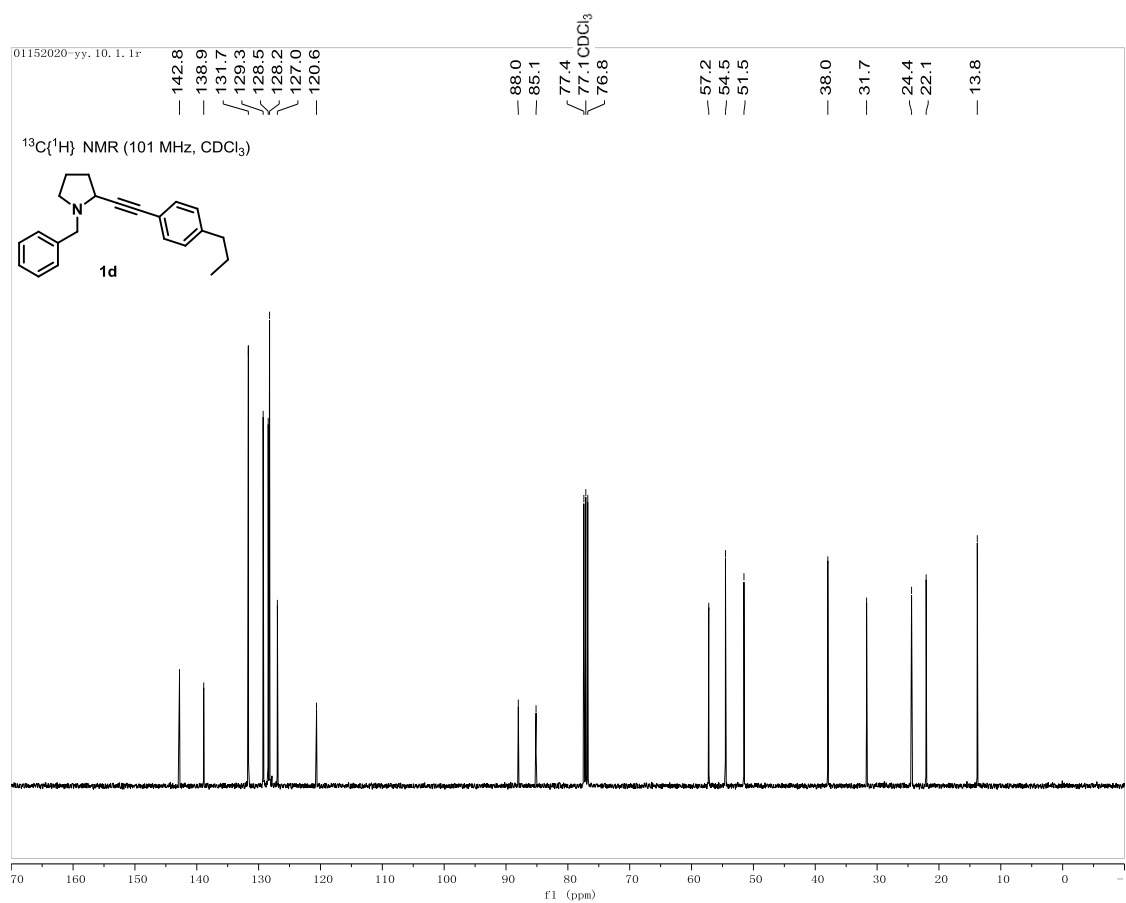
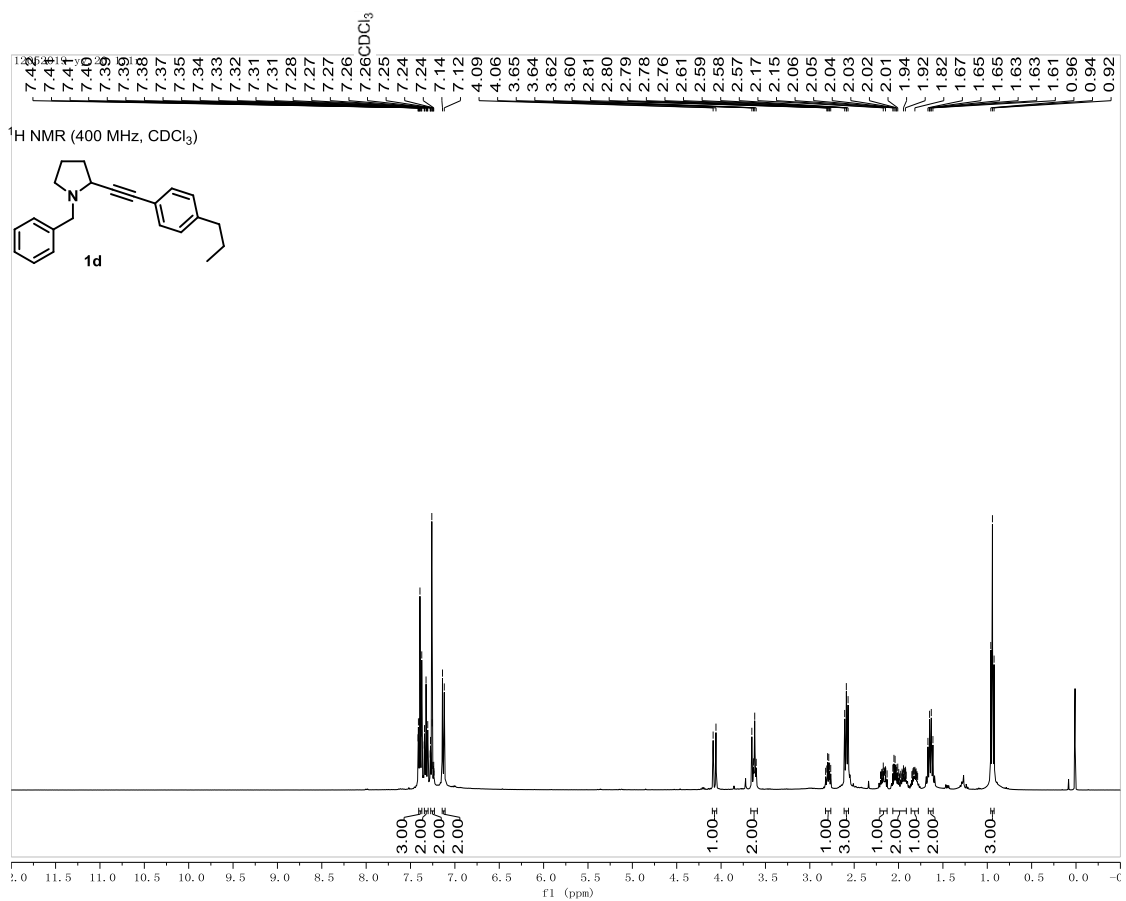
2. EPR experiment procedure for interaction of 4-benzyl-2-phenyl-5,6-dihydro-4*H*-thieno[3,2-*b*]pyrrole (1a) with elemental sulfur.

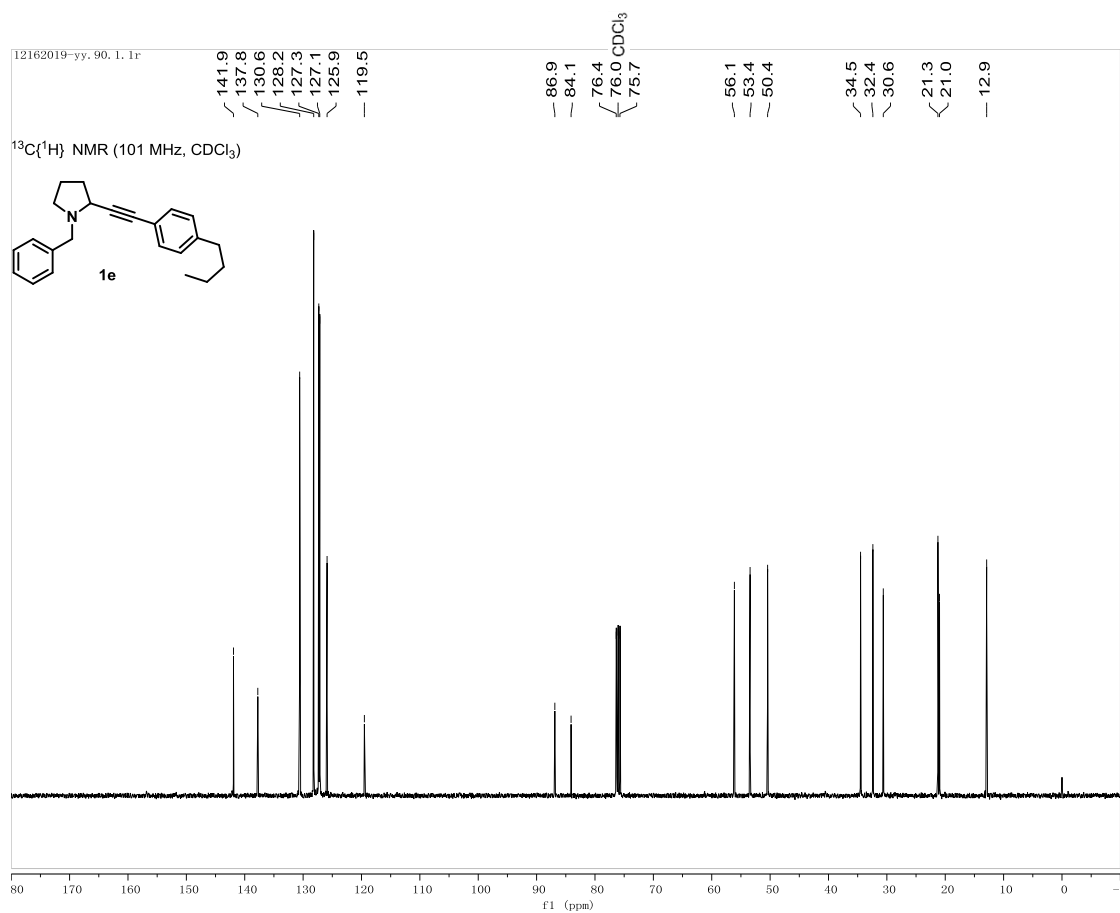
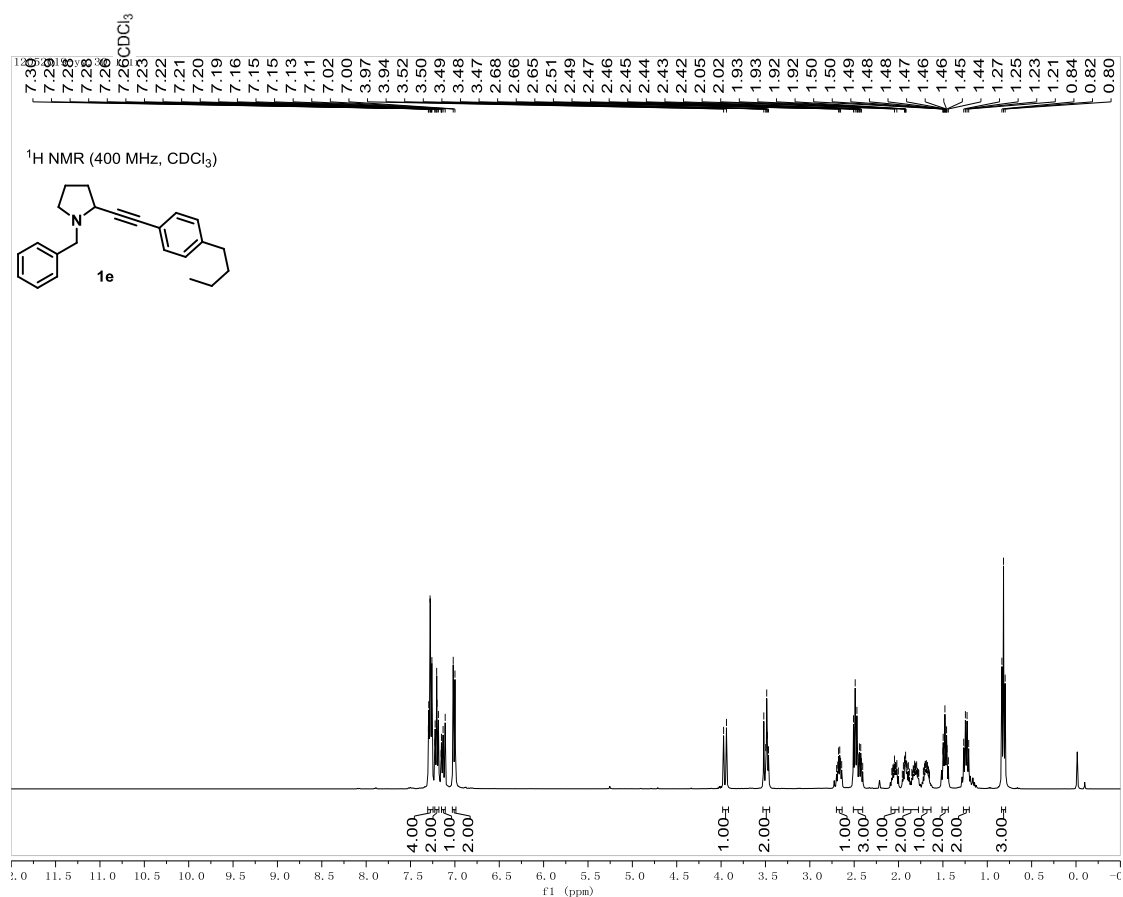
1-benzyl-2-(phenylethynyl)pyrrolidine (78.4 mg, 0.30 mmol) and elemental sulfur S₈ (115.2 mg, 0.45 mmol) were combined in a 50 mL flame-dried Young-type tube equipped with a stir bar, and then the tube was sealed. Next, the Schlenk tube was purged three times with N₂. Then, 1,4-dioxane (2.0 mL) was injected into the Schlenk tube with a syringe under N₂ atmosphere. The contents of the Schlenk tube were then allowed to stir at 120 °C by using a heating mantle for 1.0 h. Then, DMPO was added to the mixture and preserved in liquid nitrogen for EPR examination. No organic radical was observed.

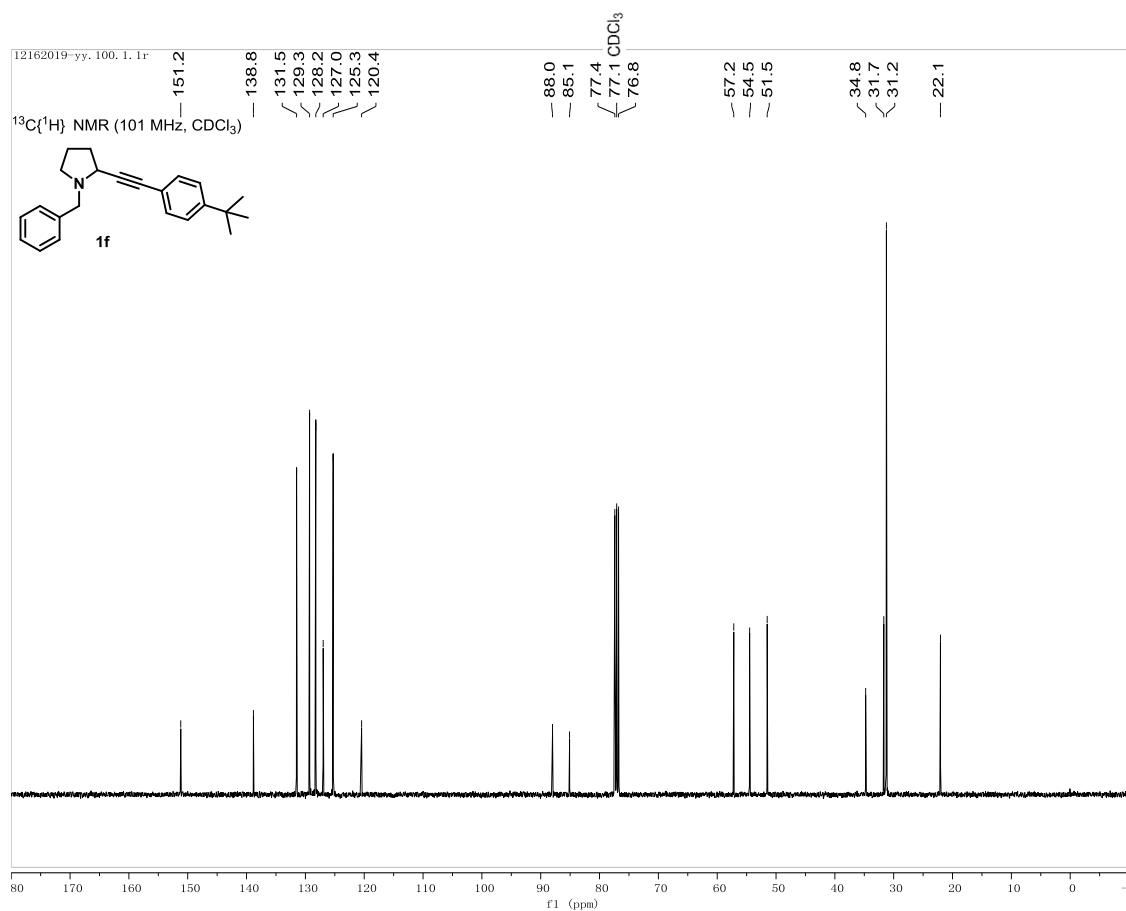
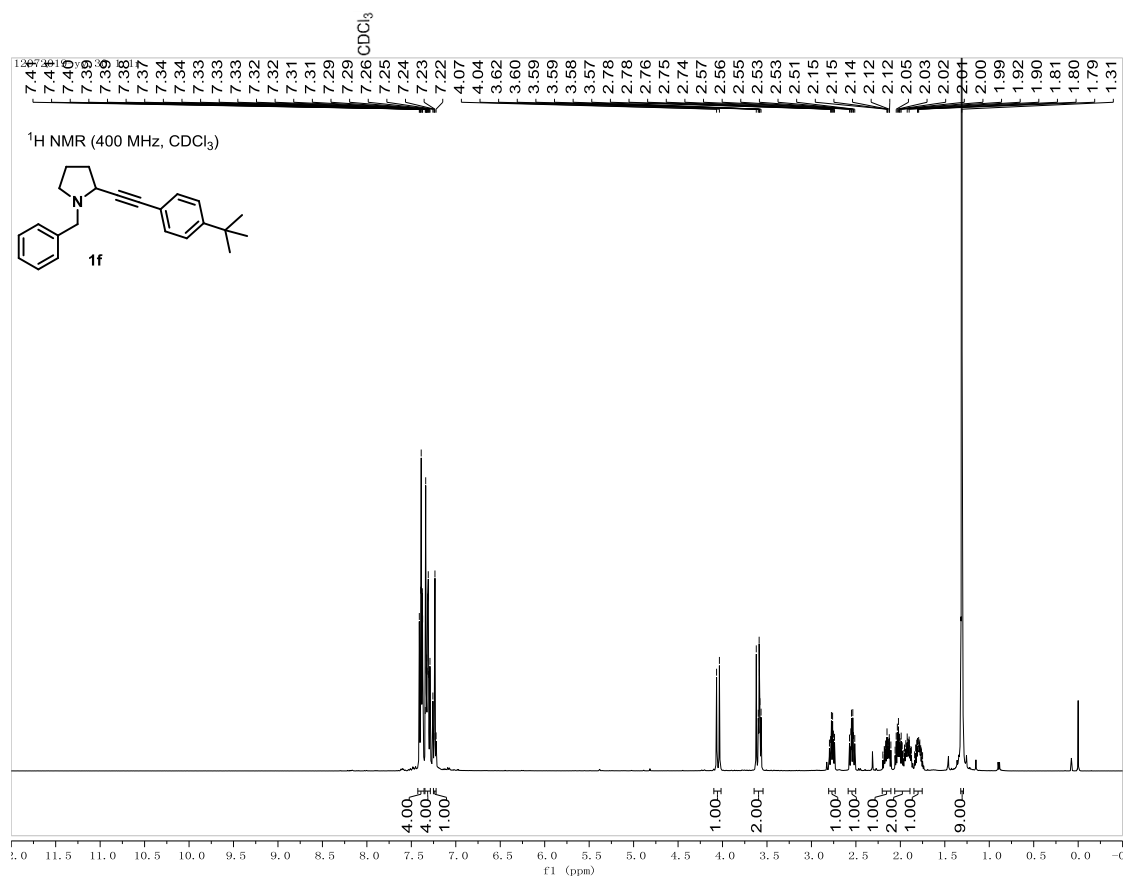
3. ^1H NMR and ^{13}C $\{^1\text{H}\}$ NMR copies of substrates

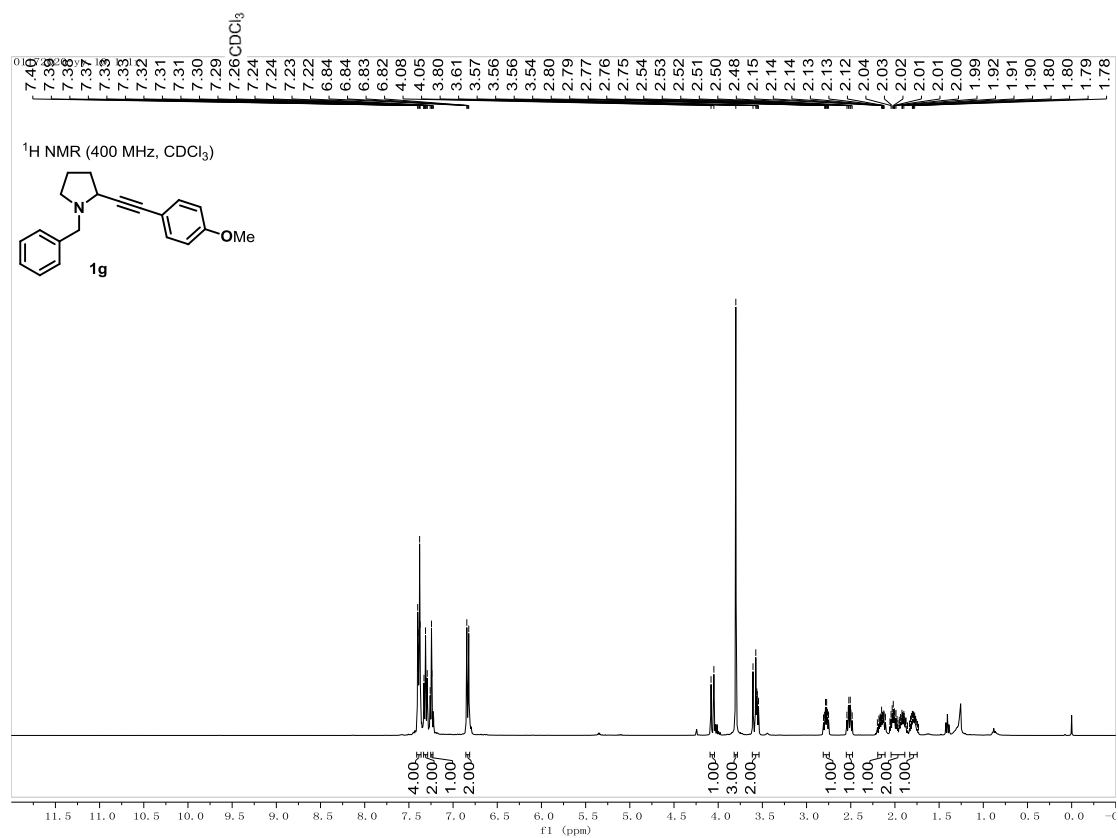


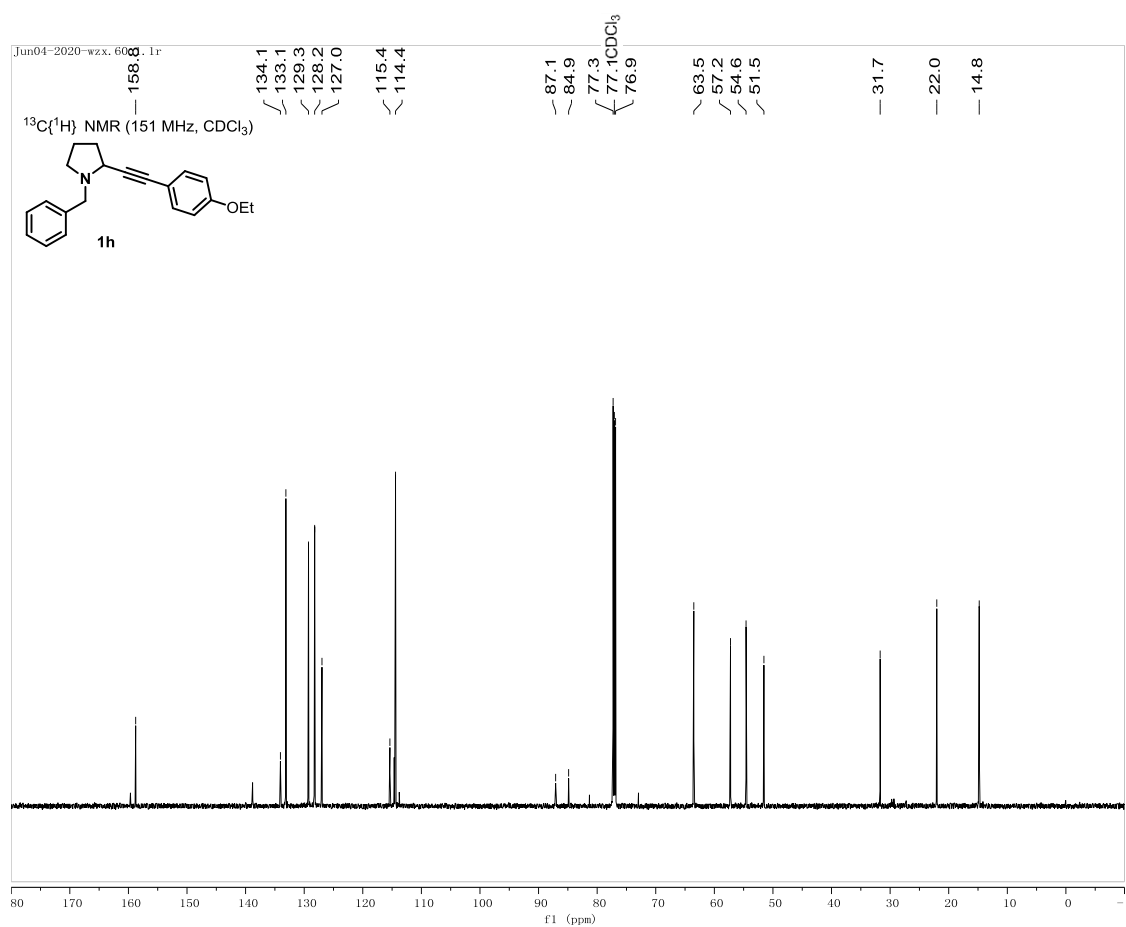
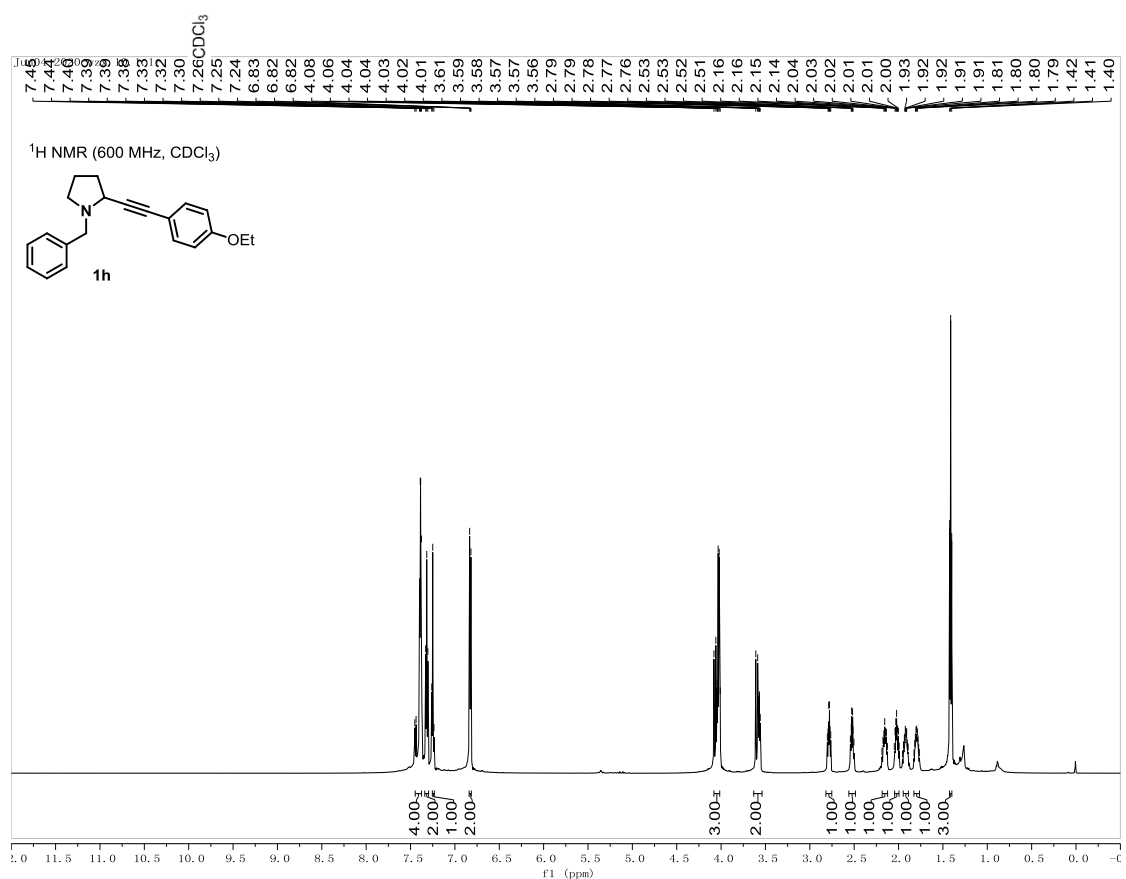


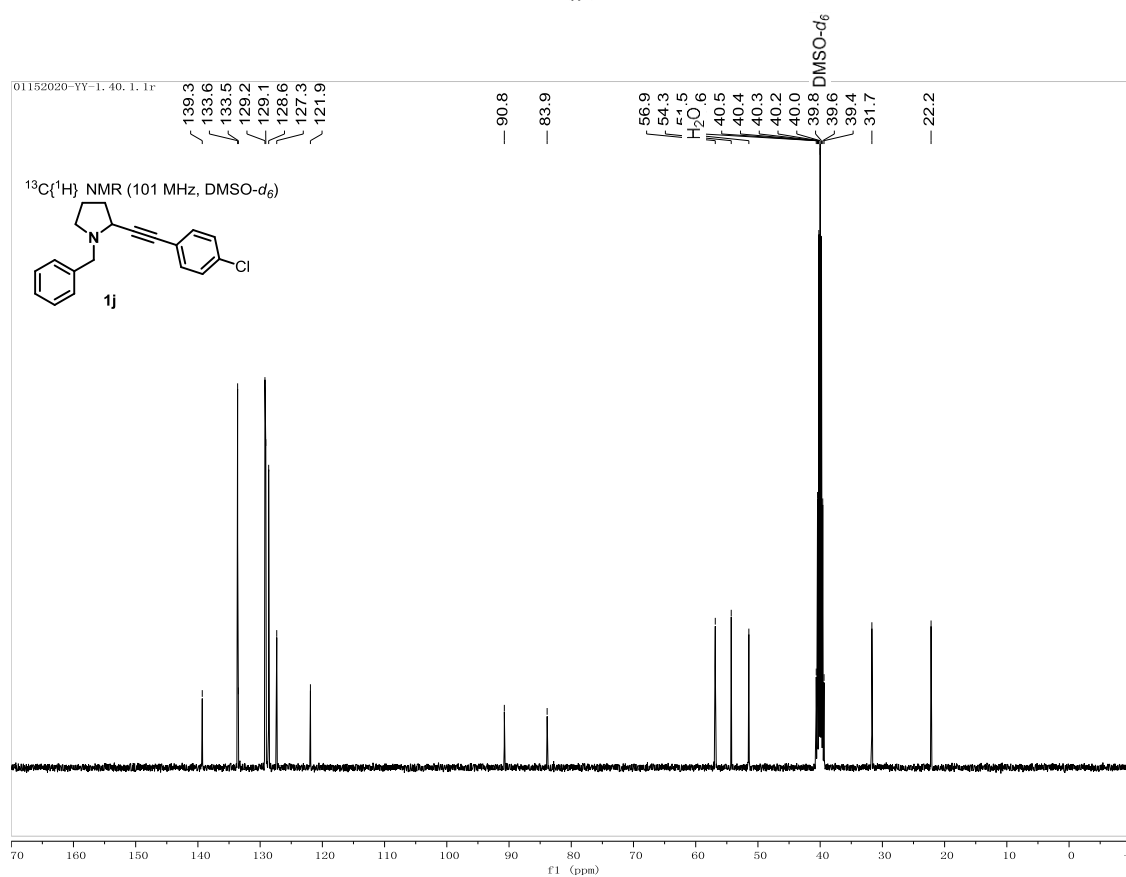
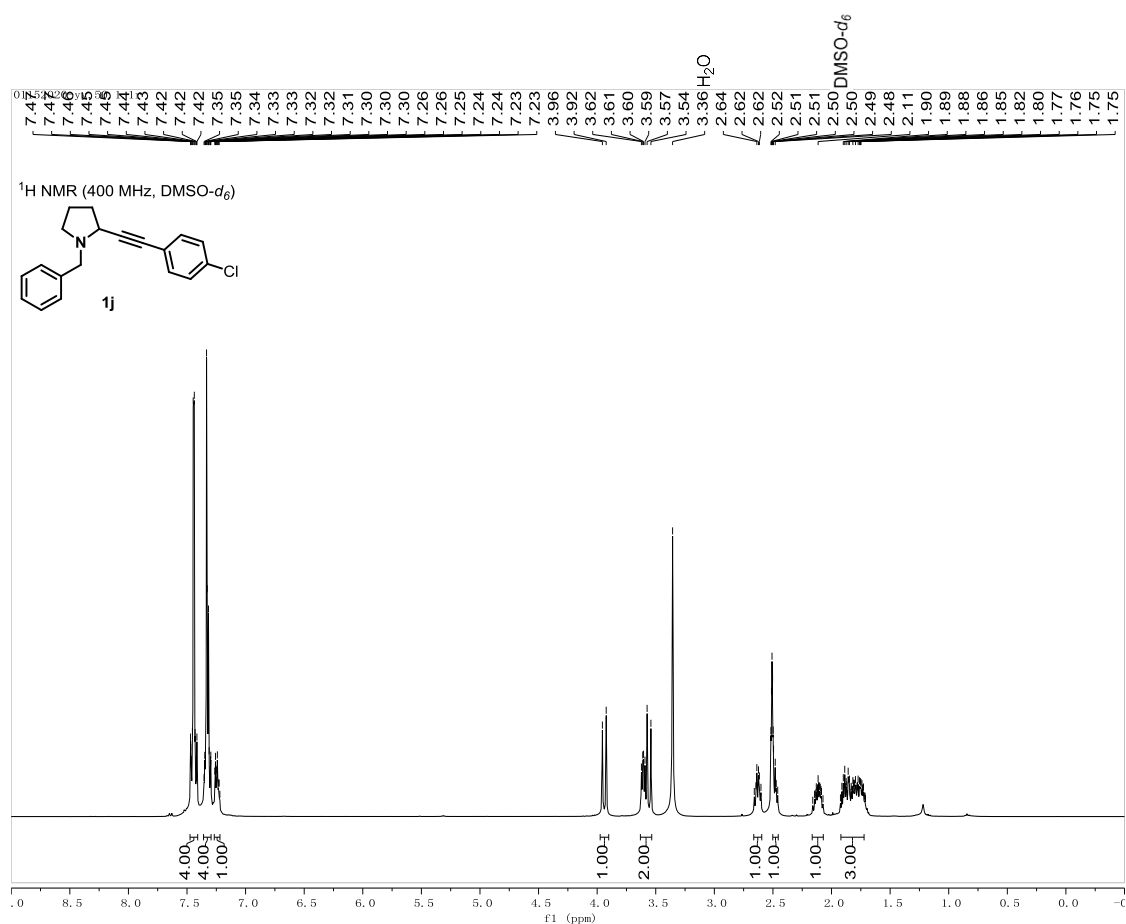


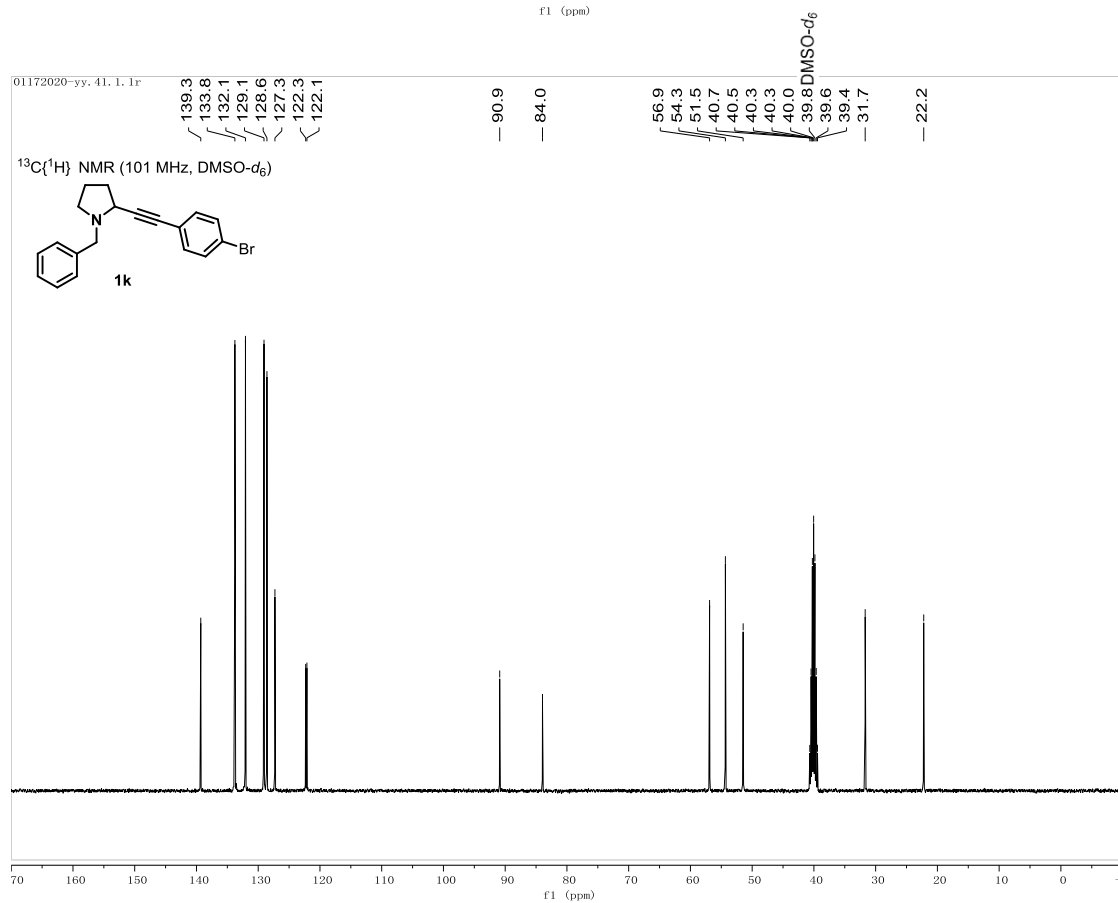
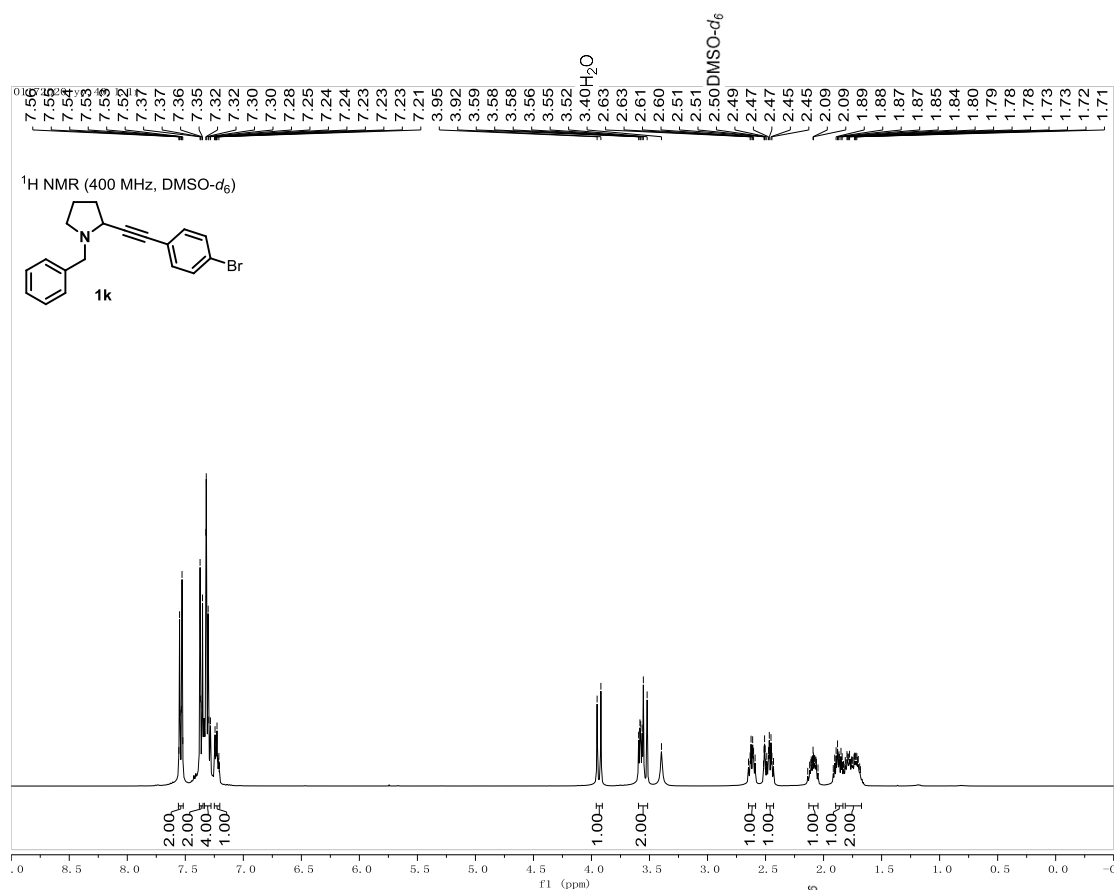


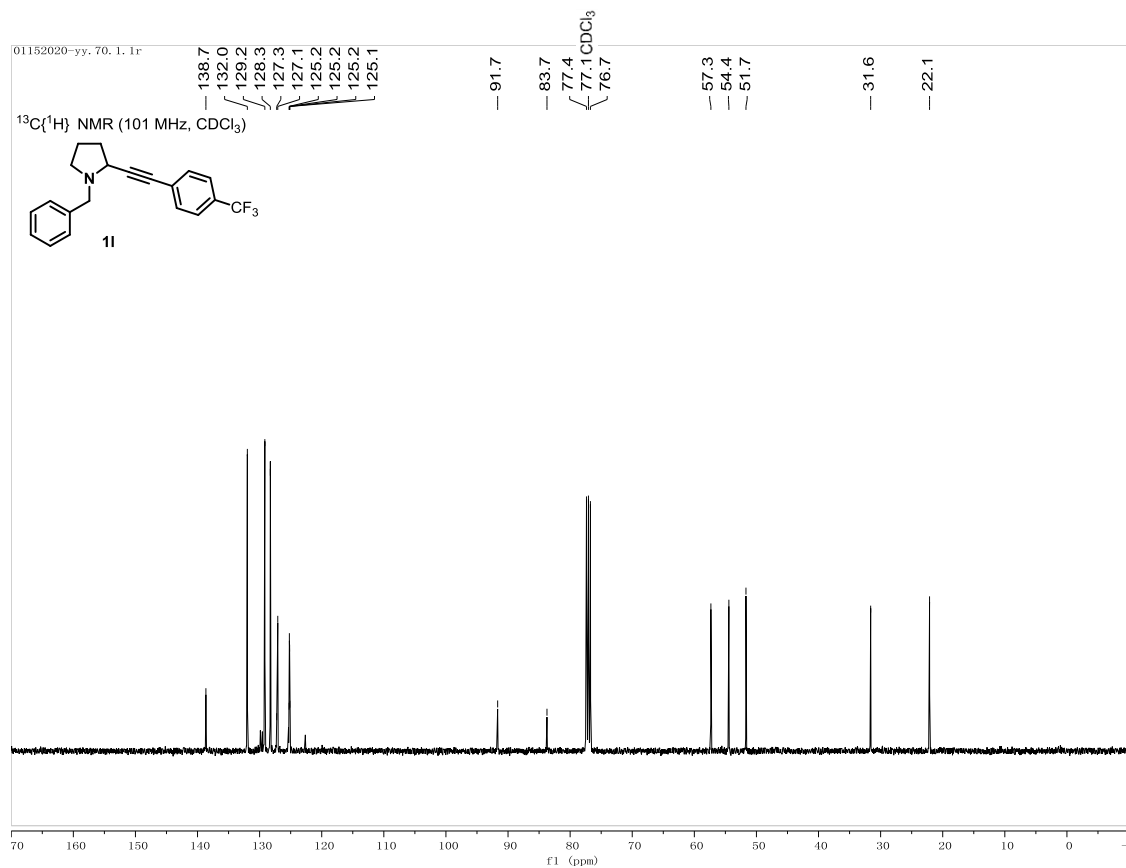
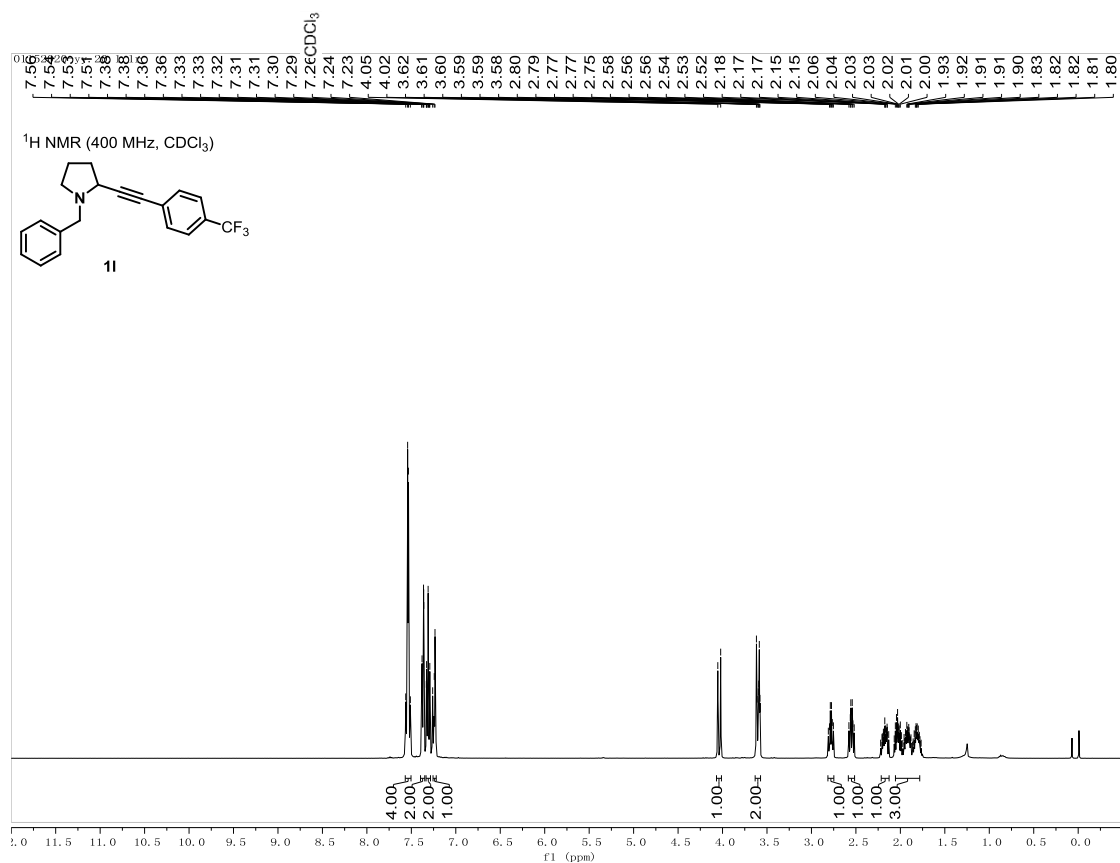


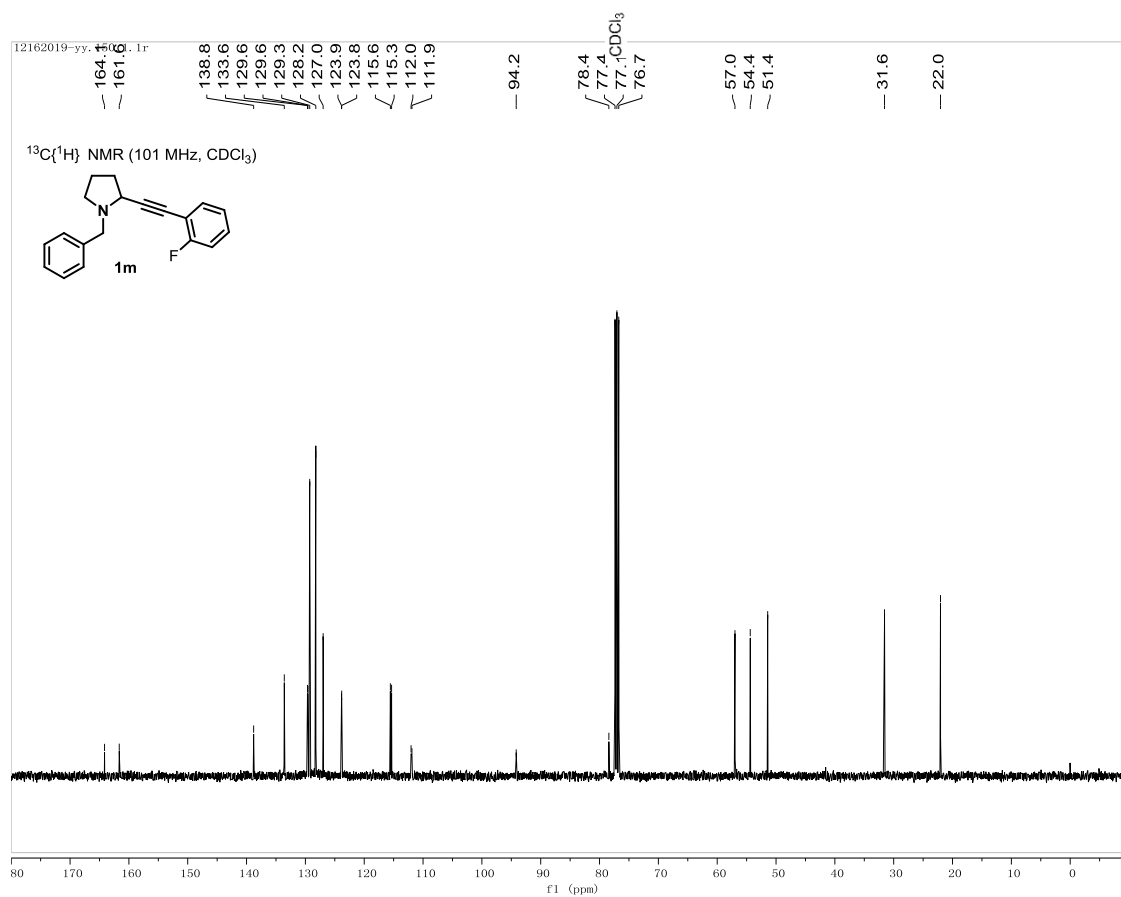
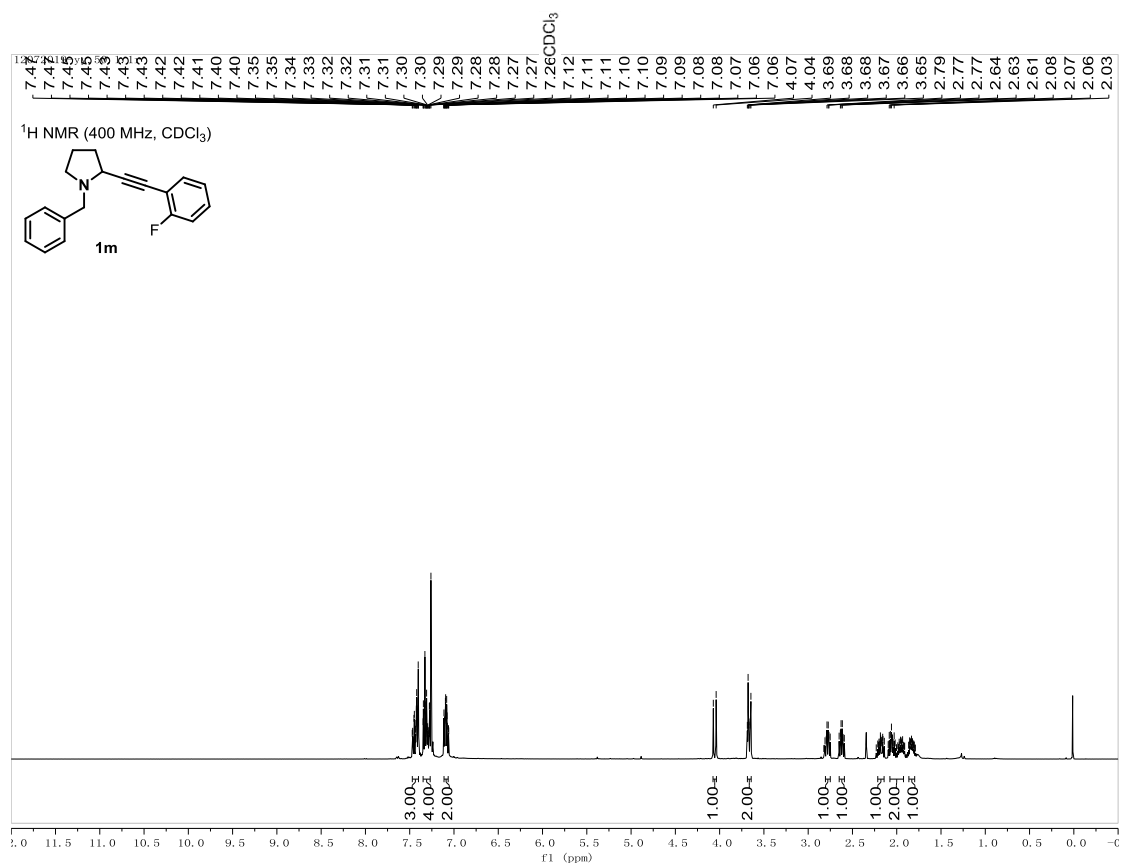


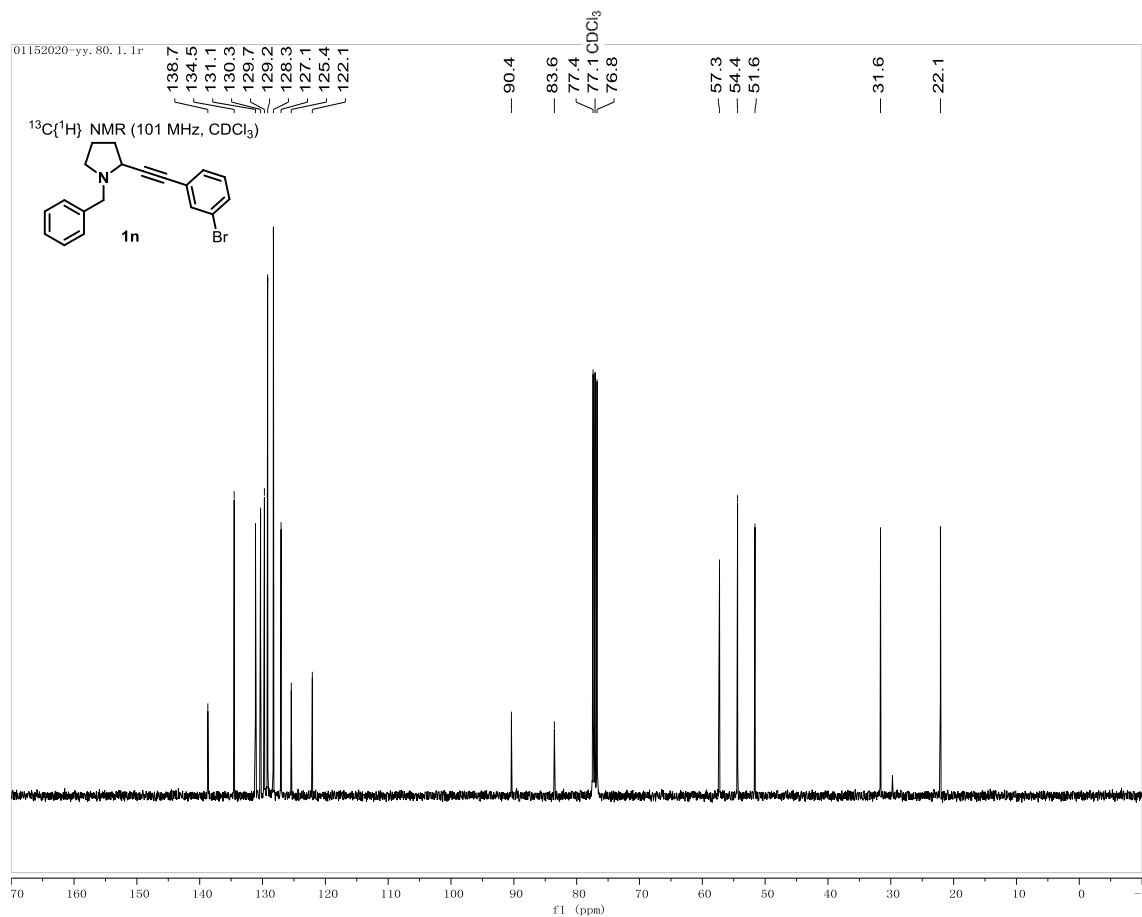
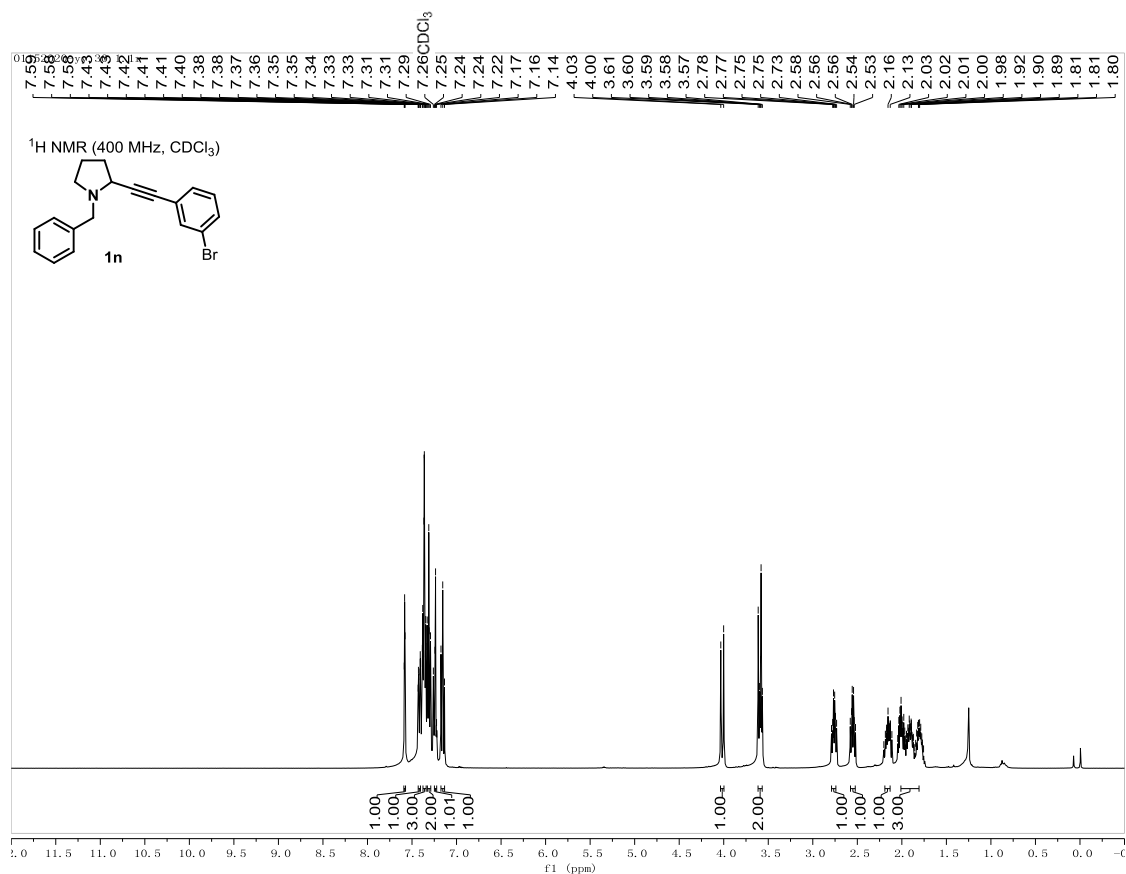


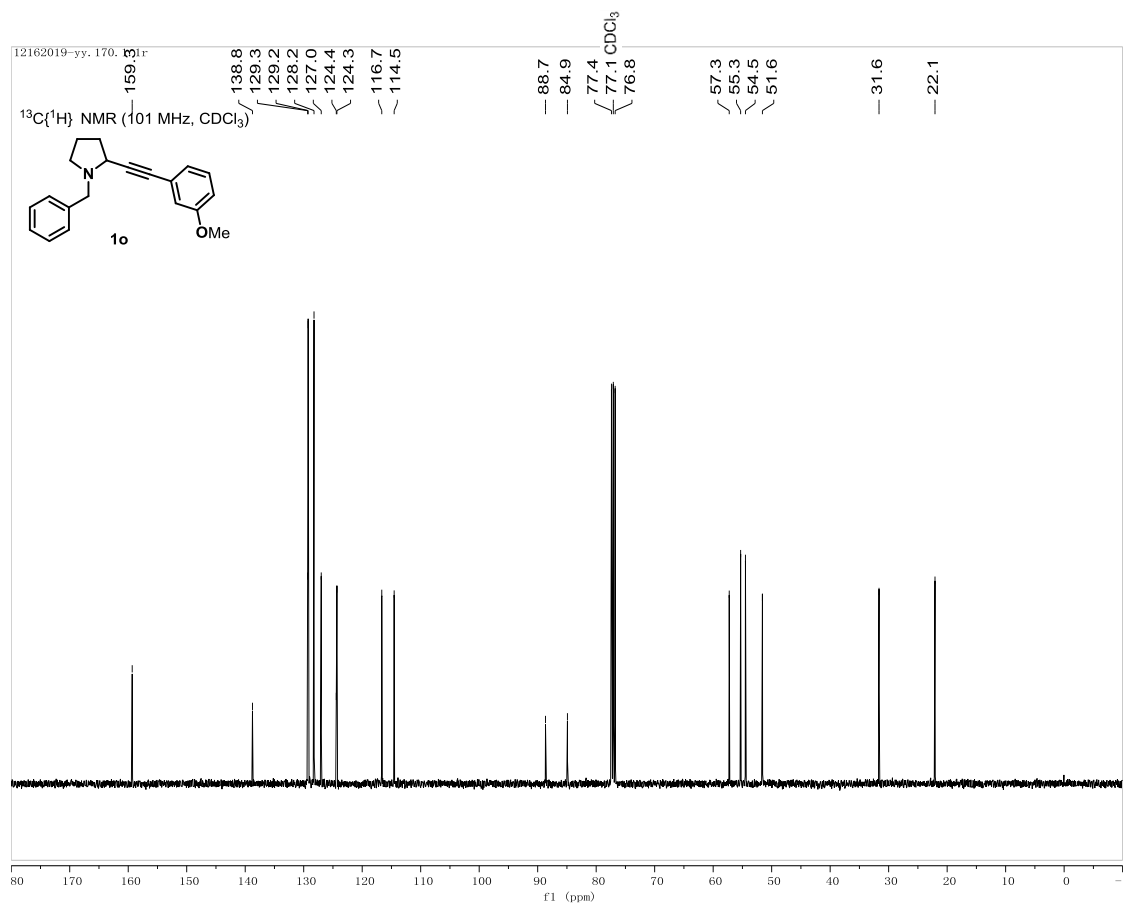
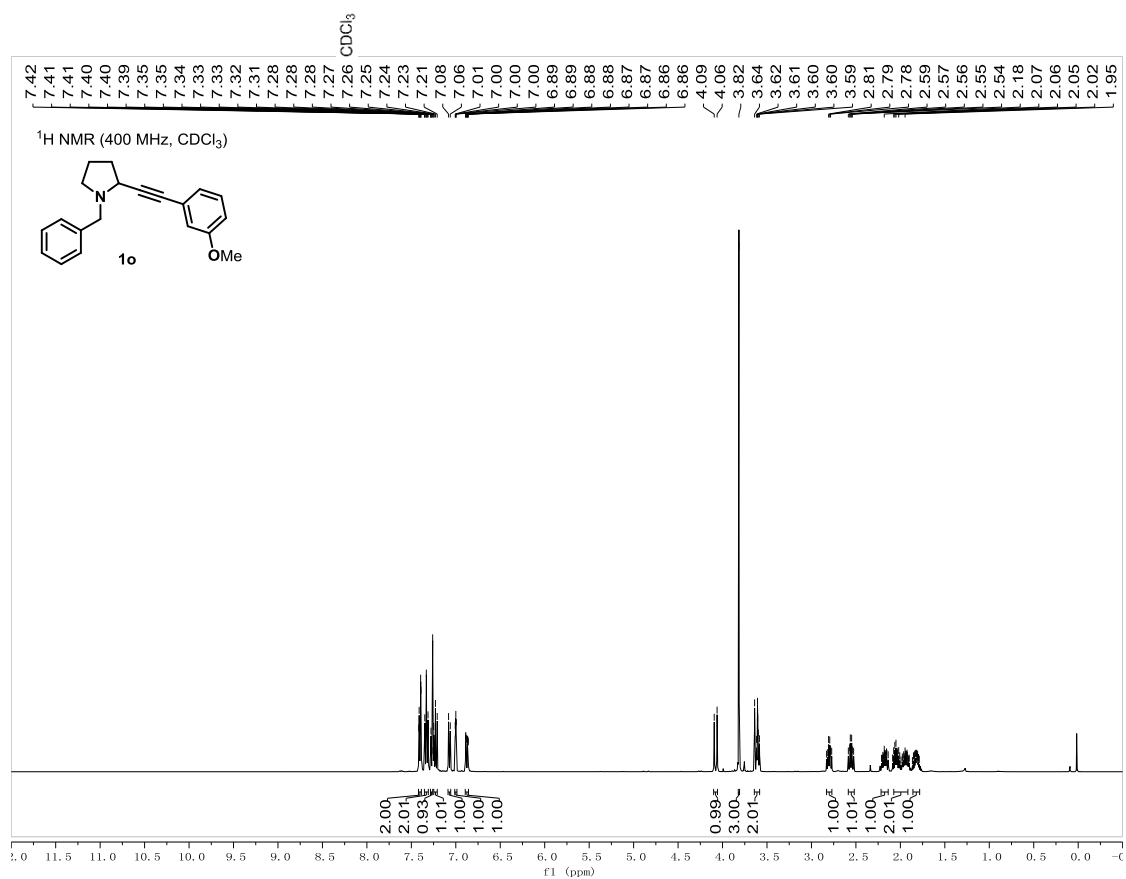


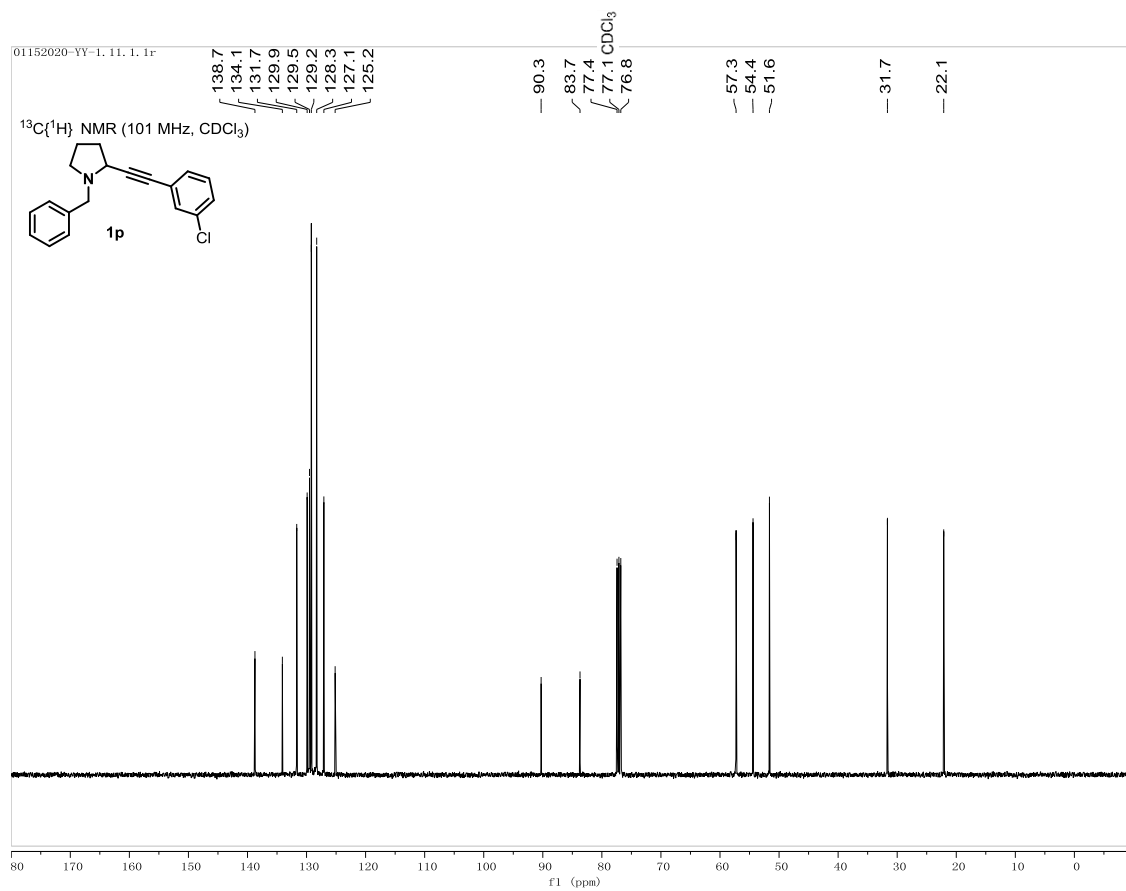
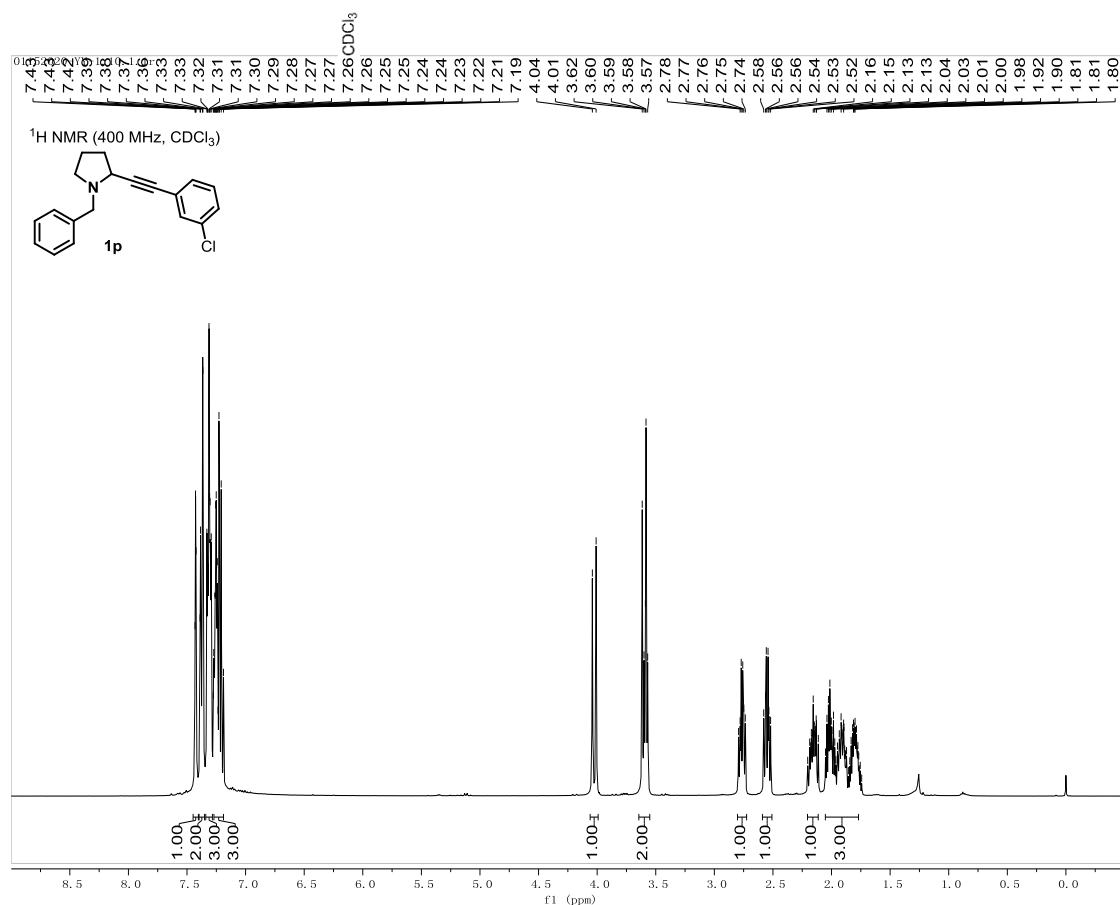


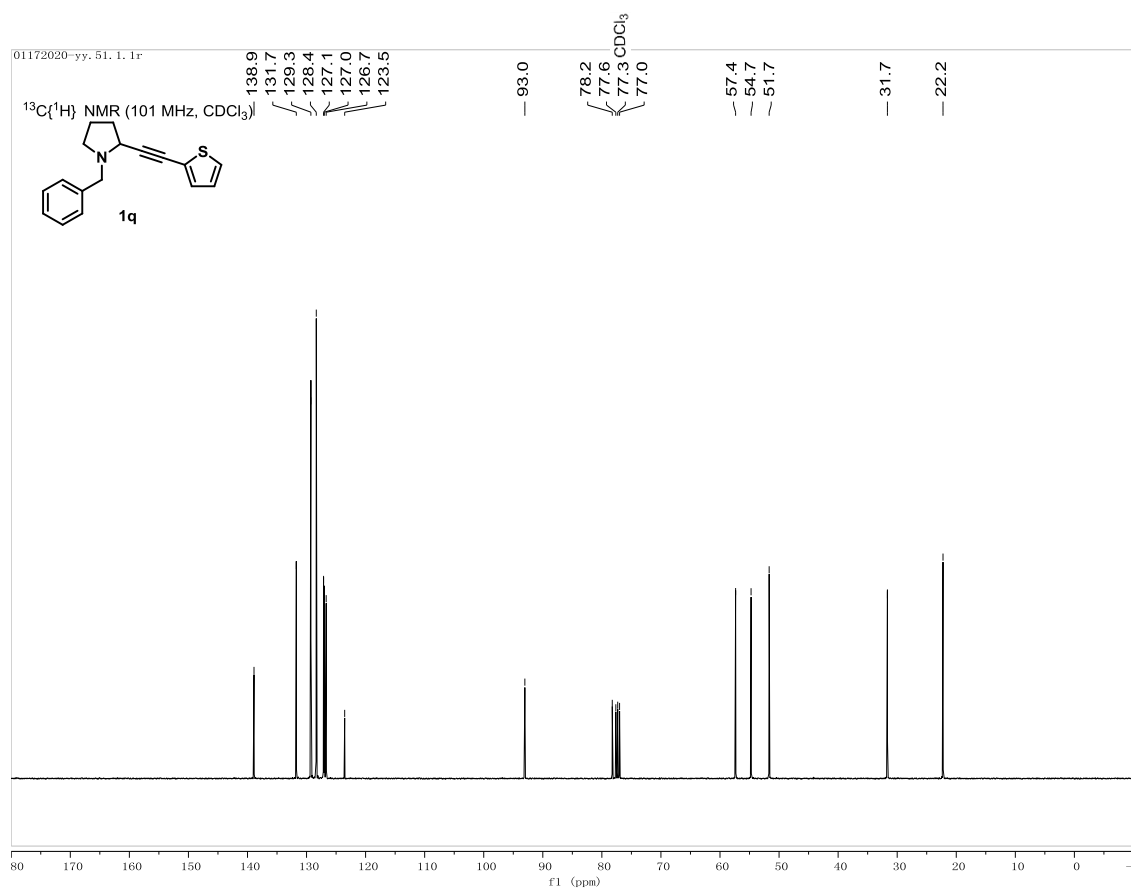
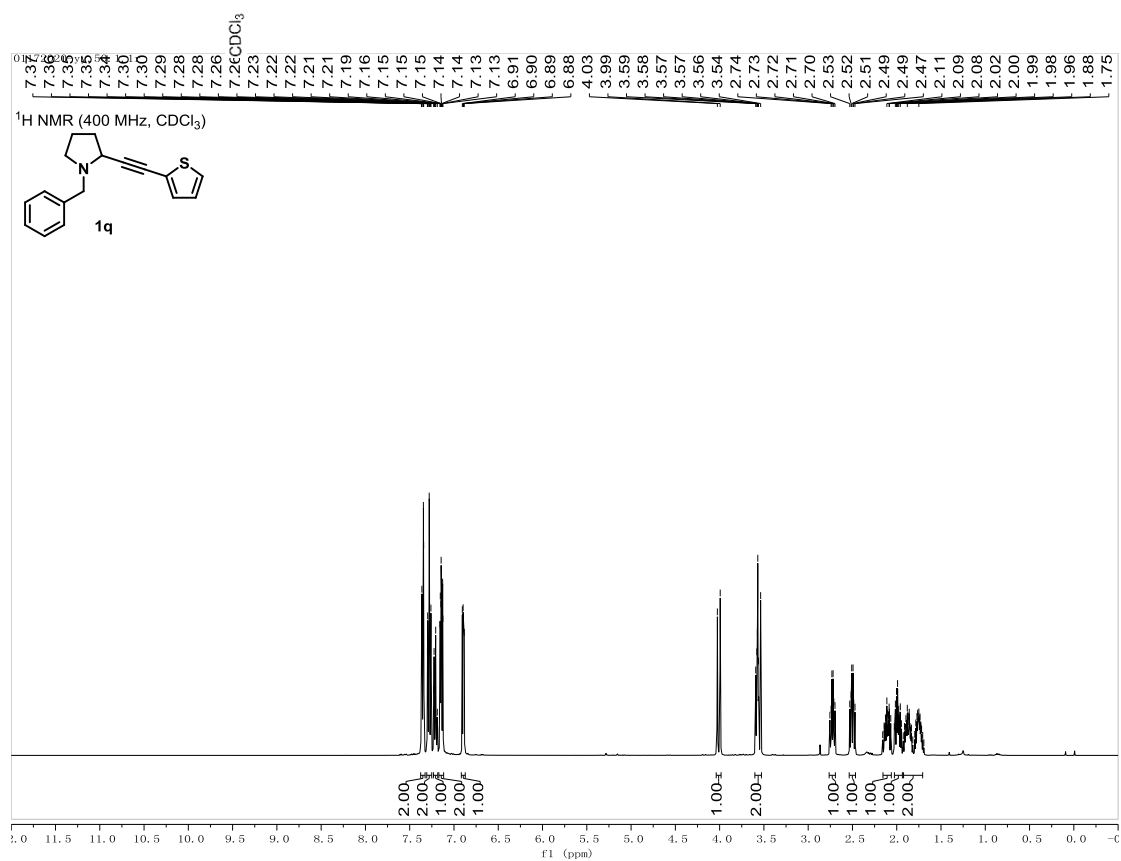


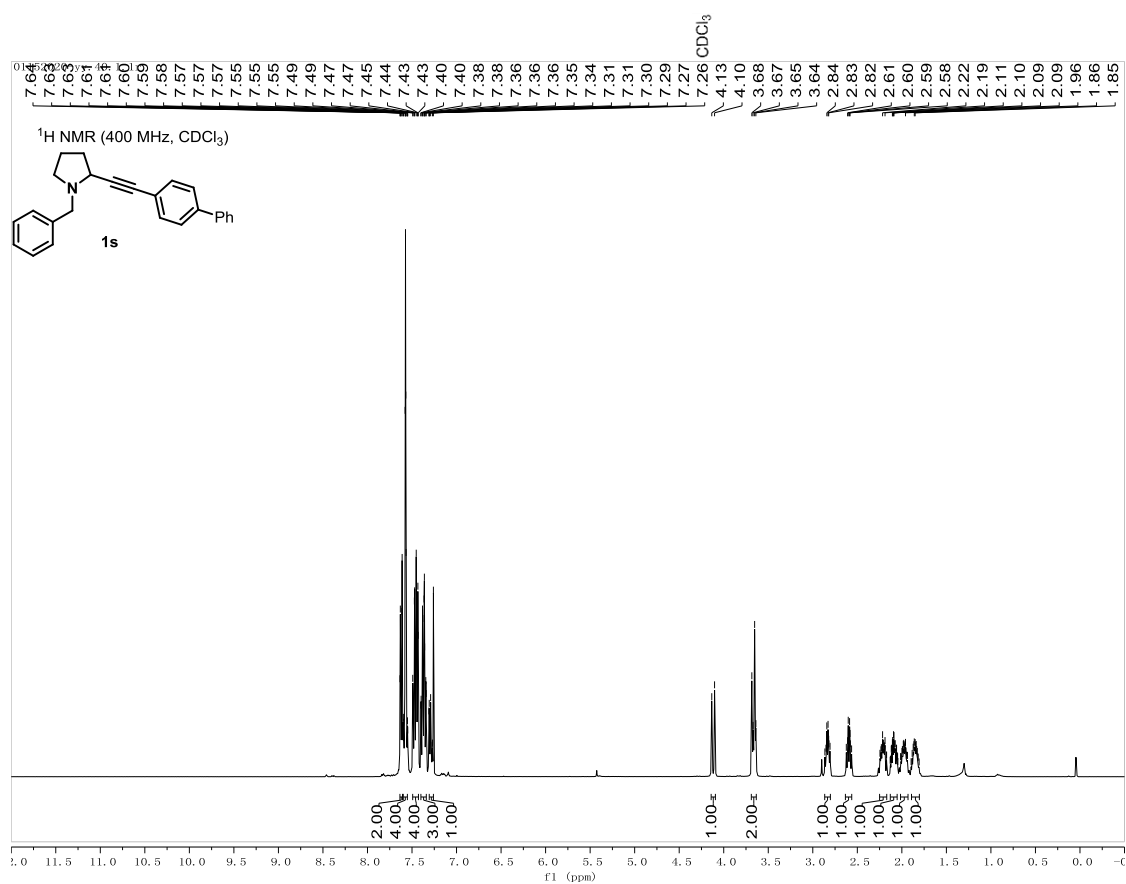


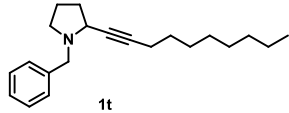
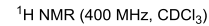


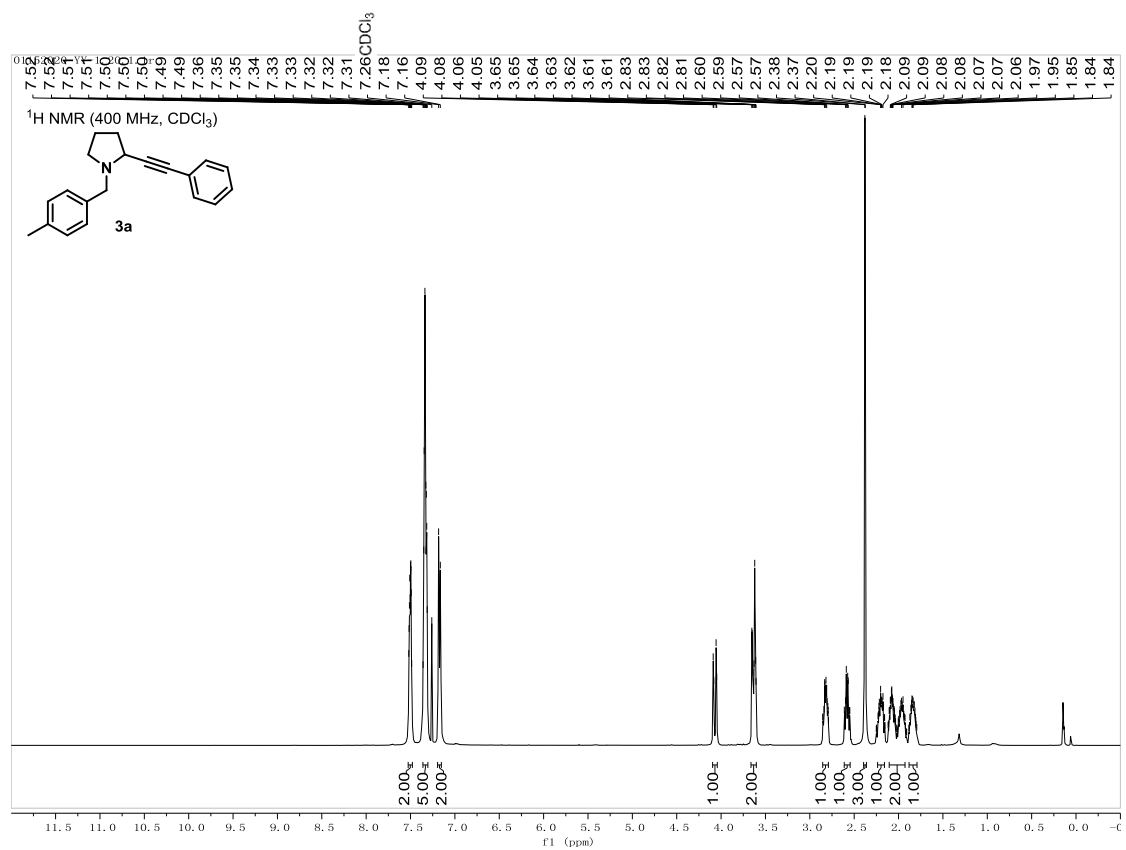


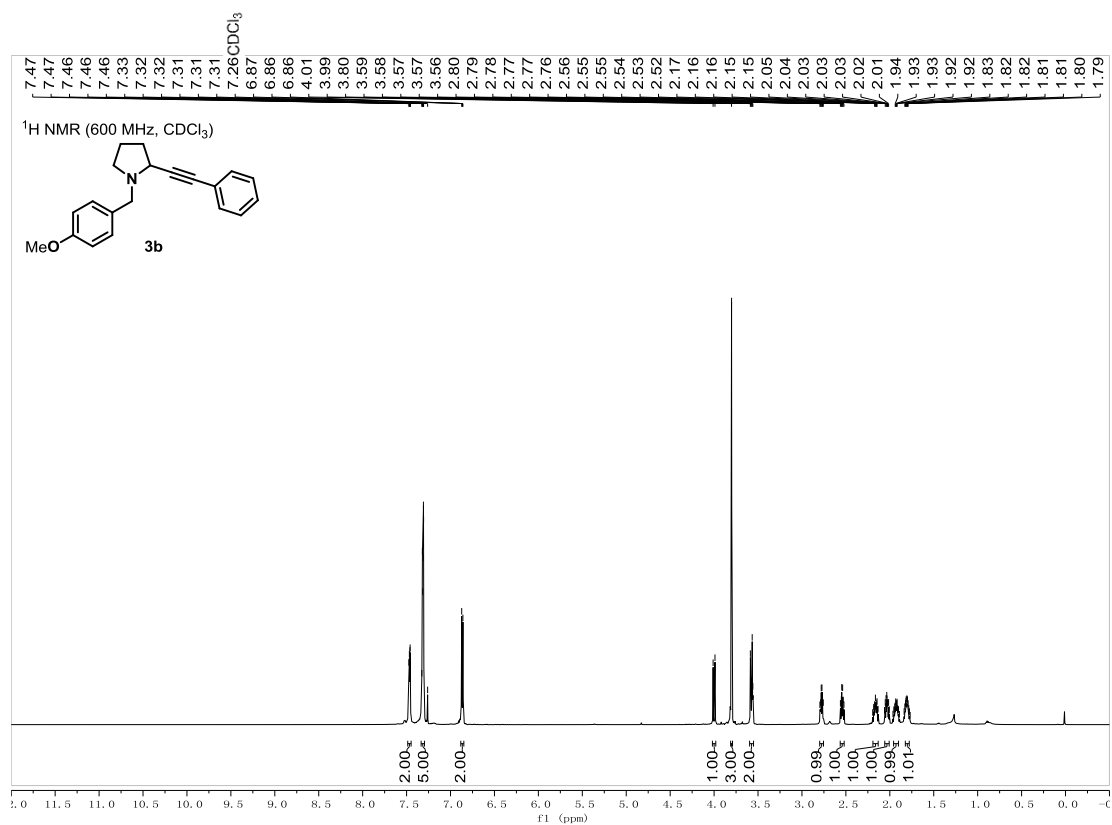


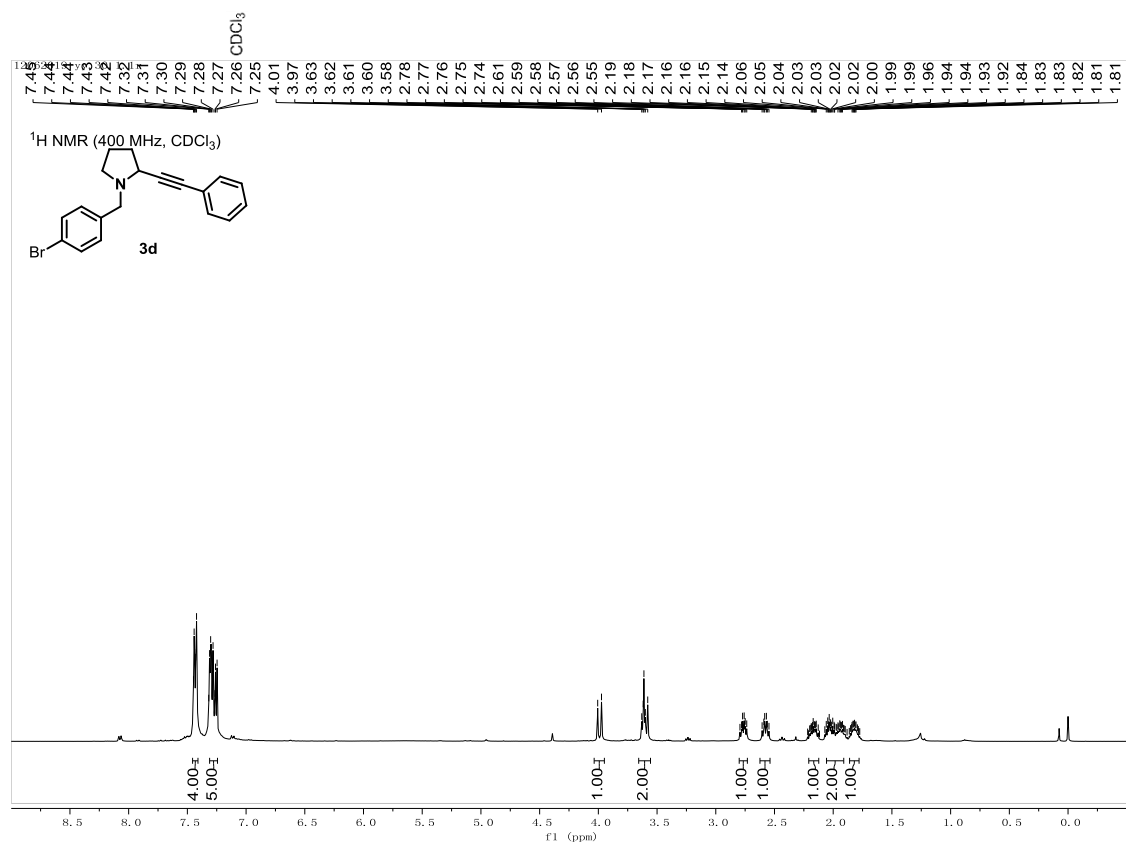


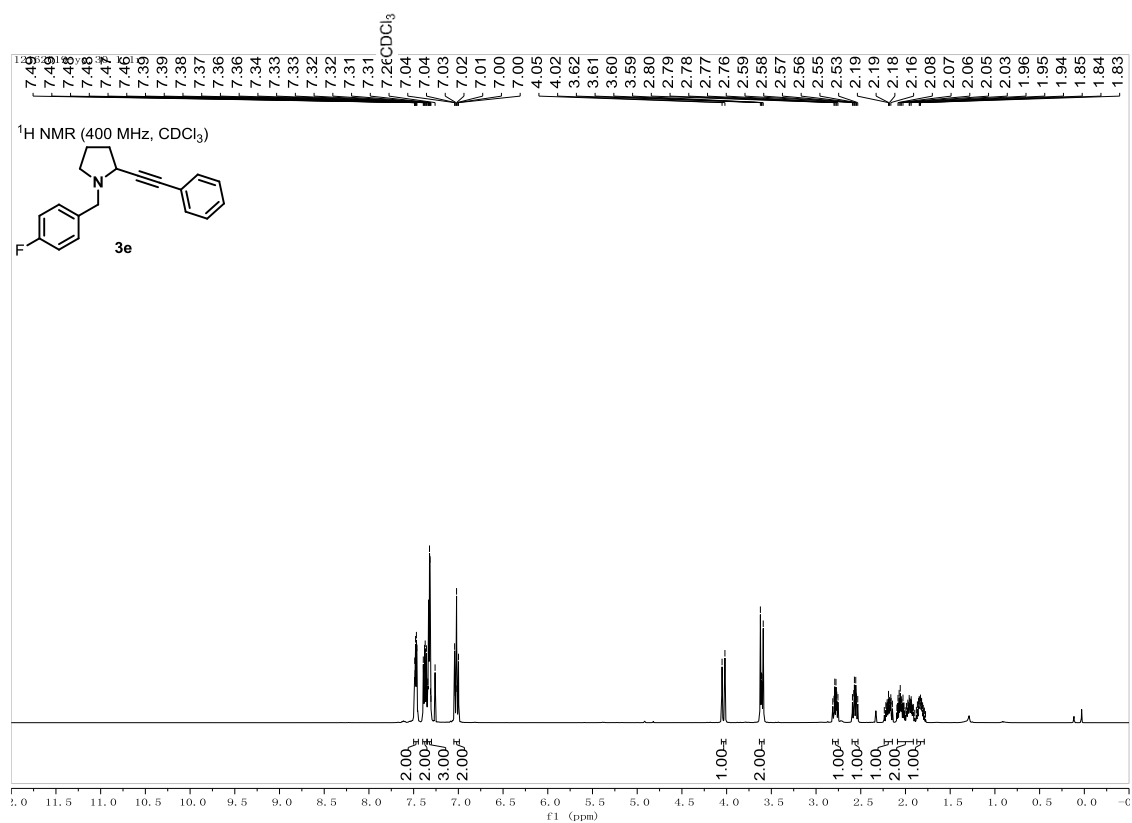


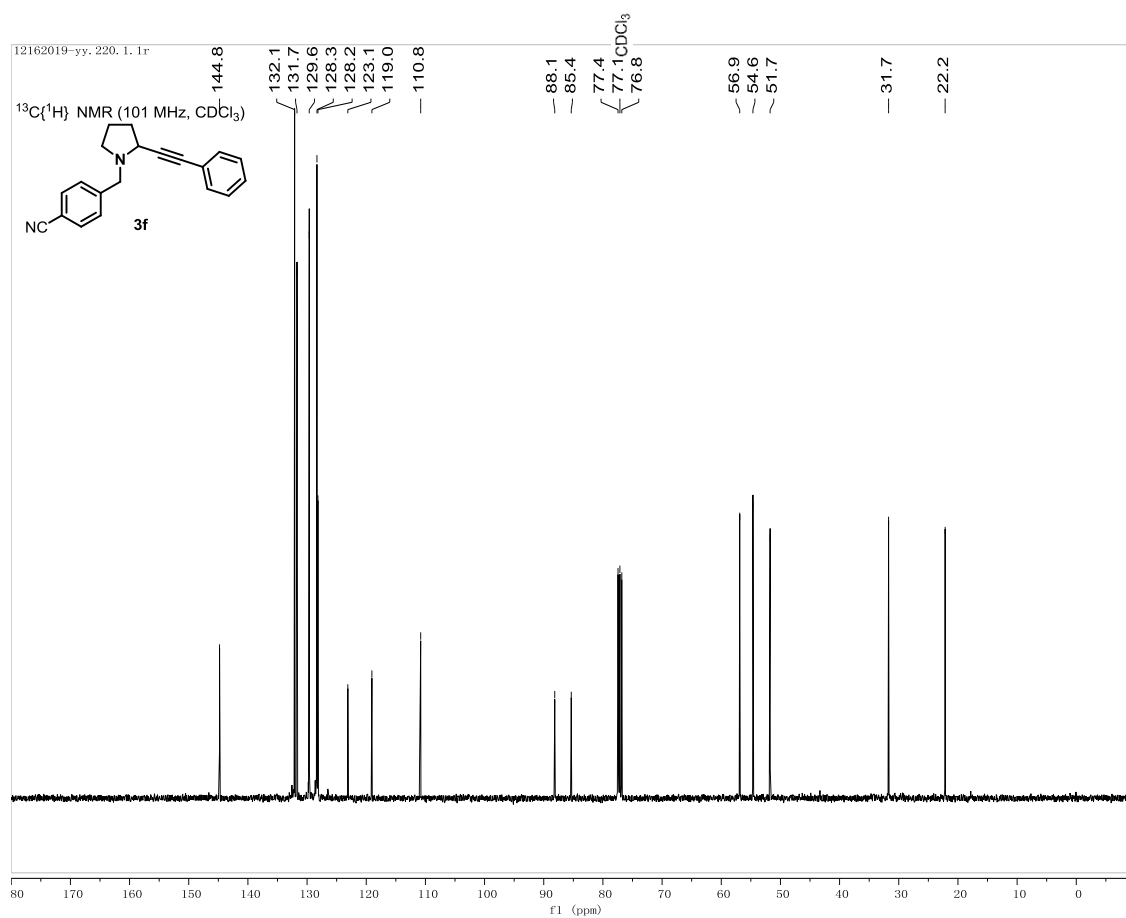
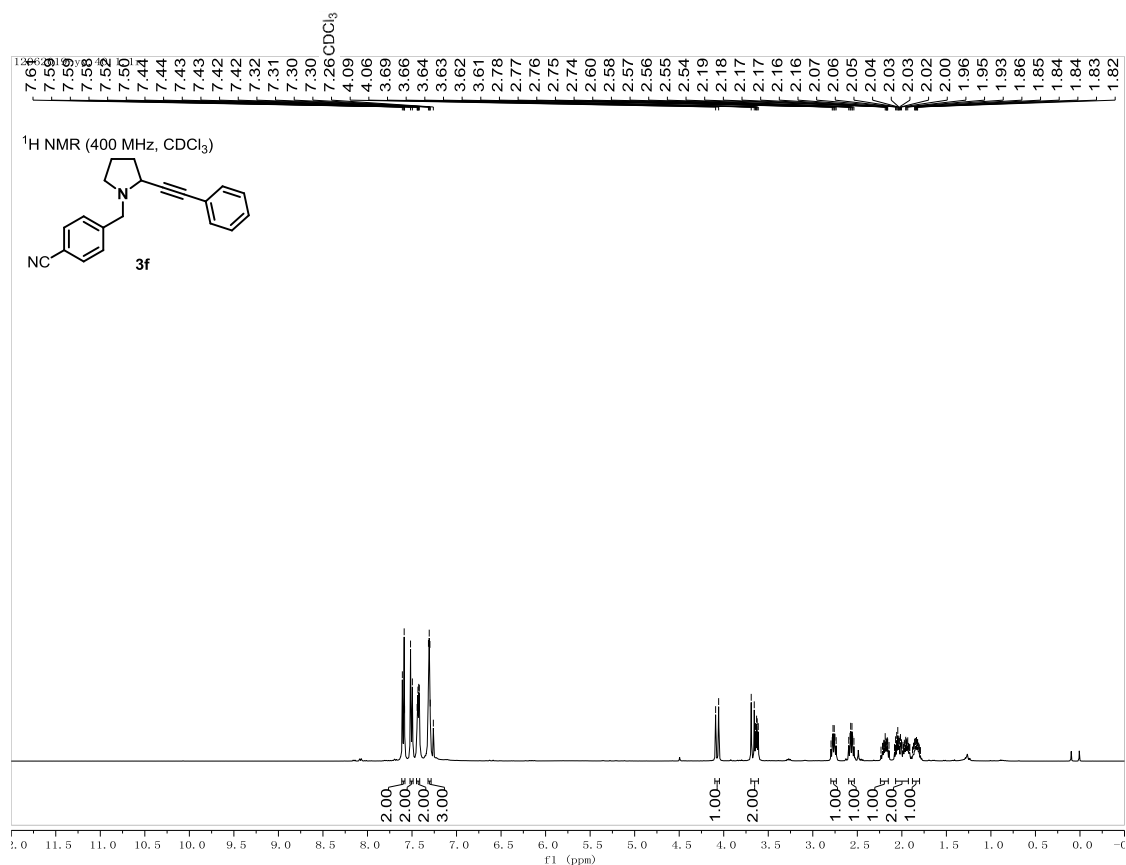


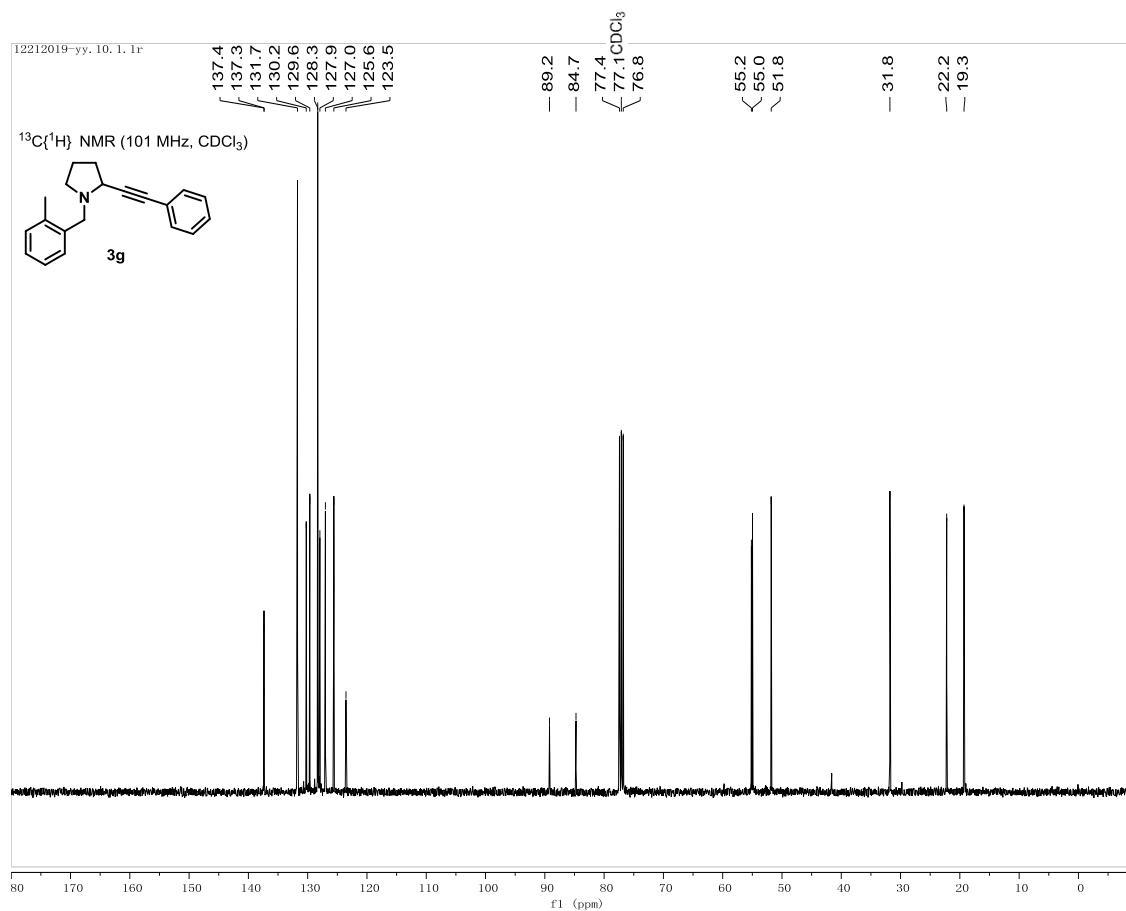
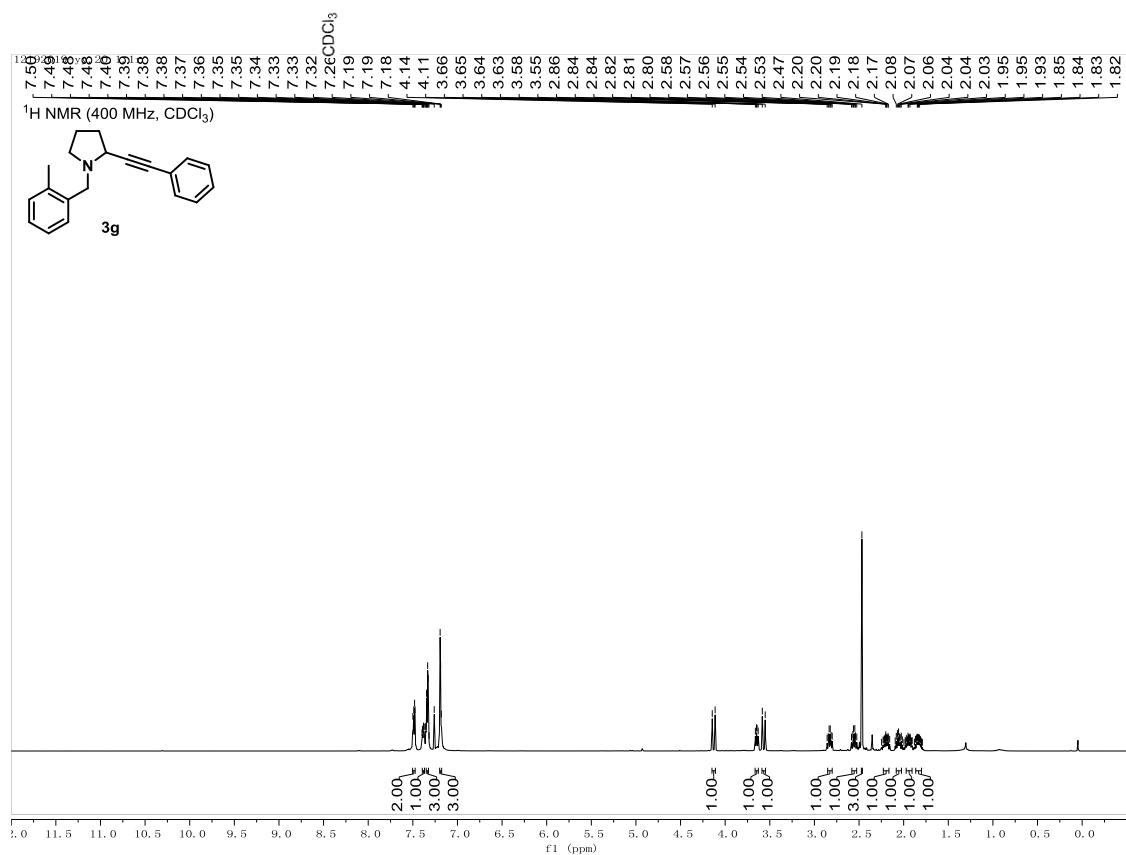


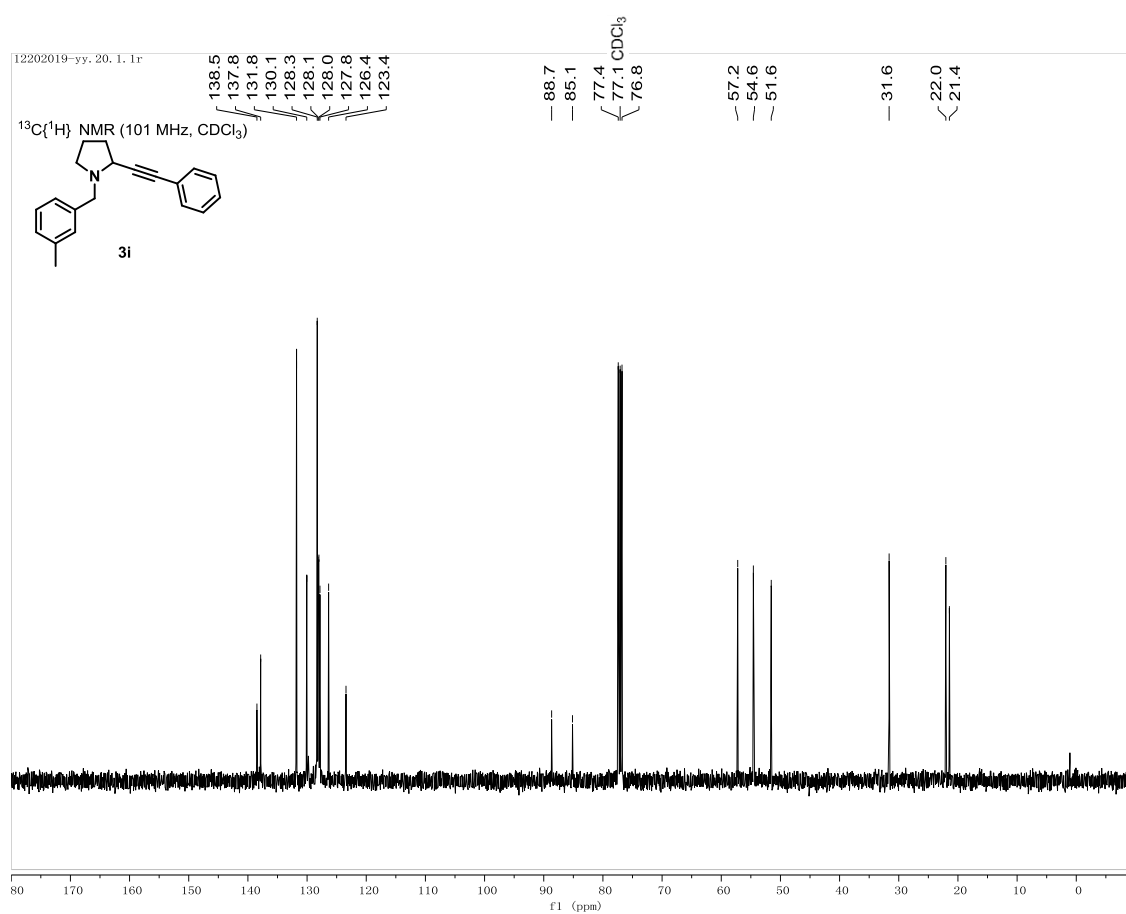
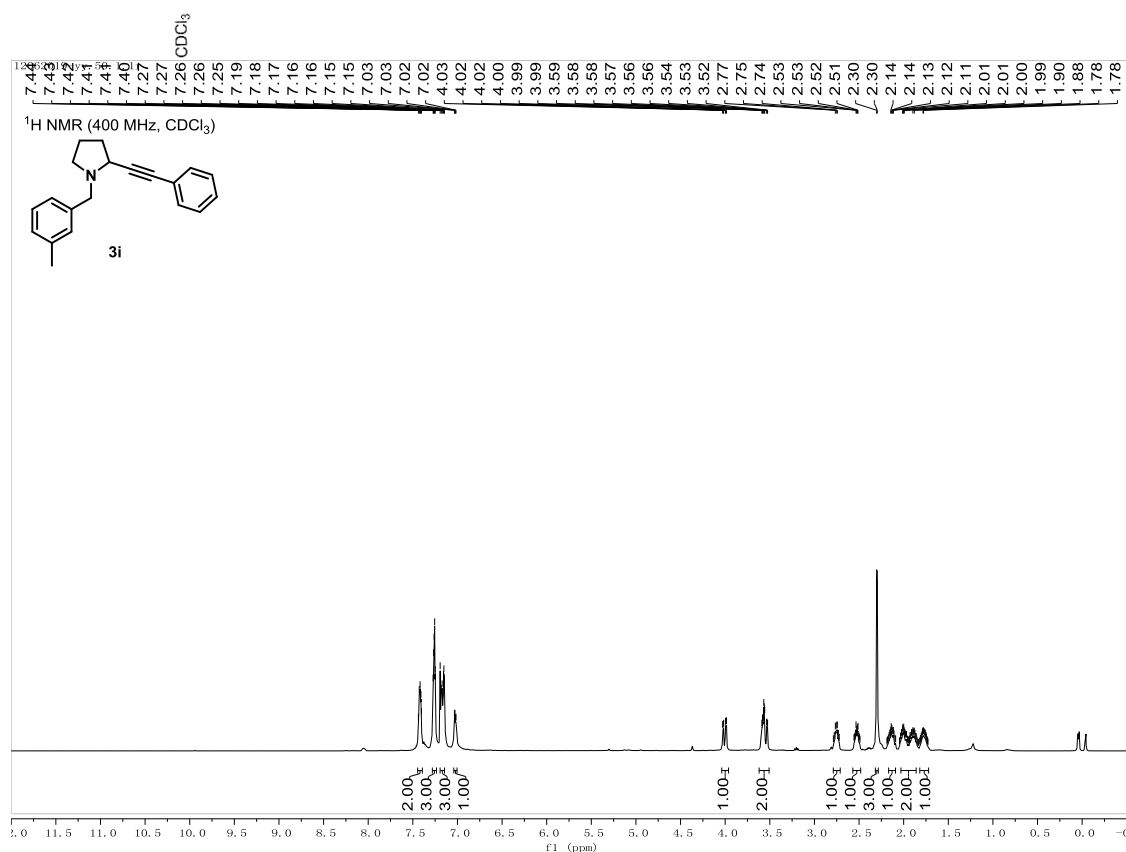


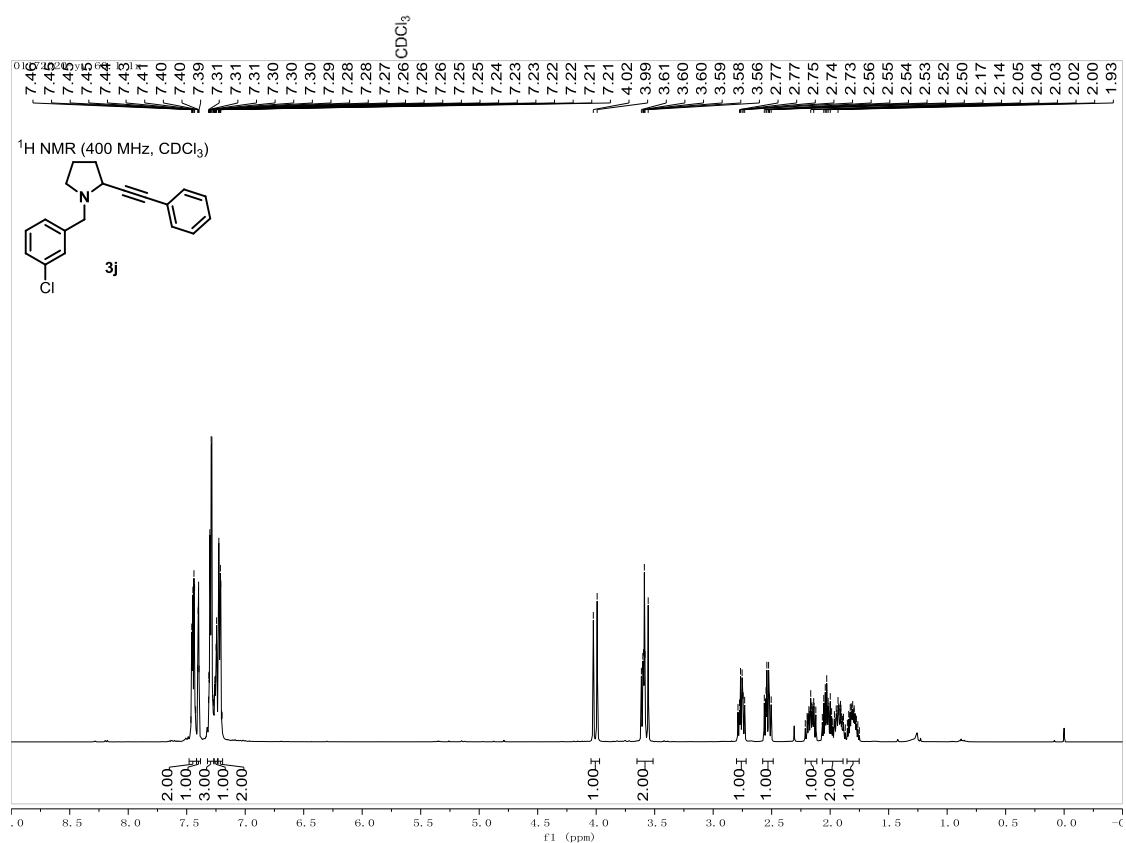


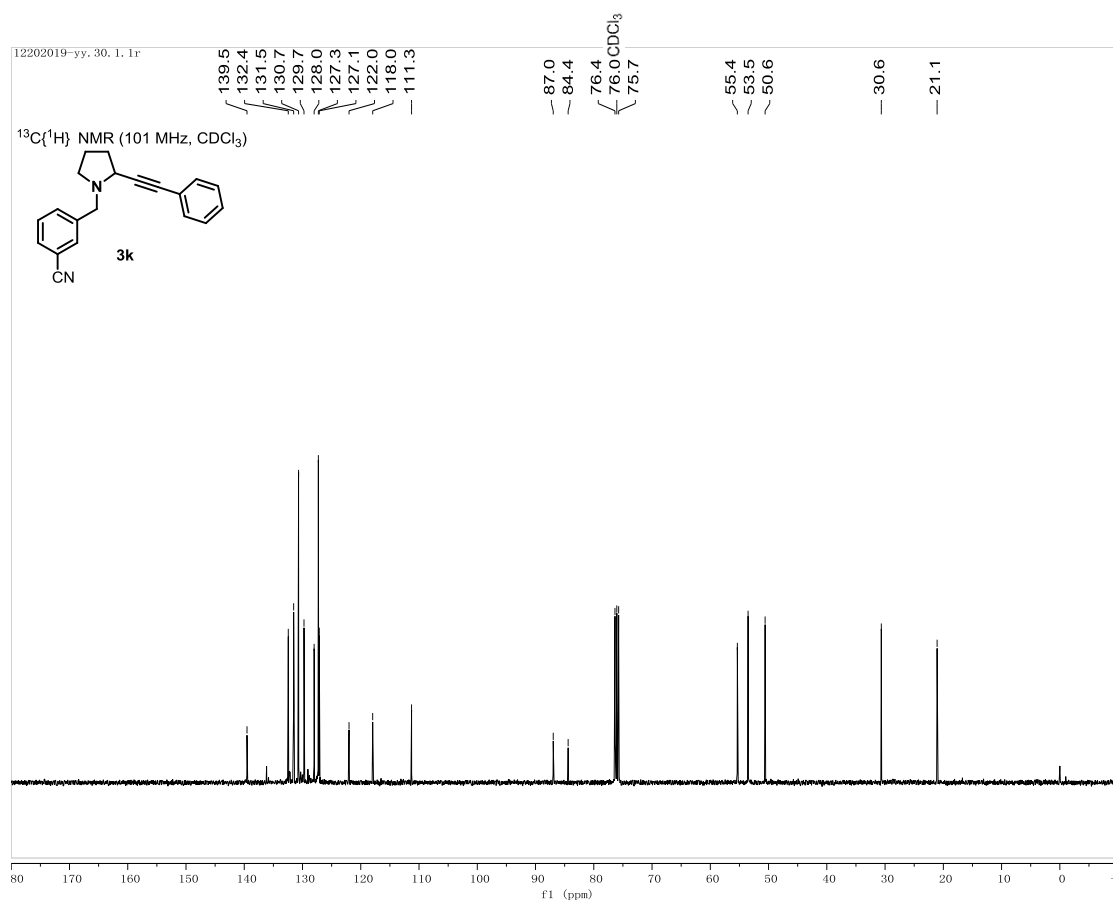
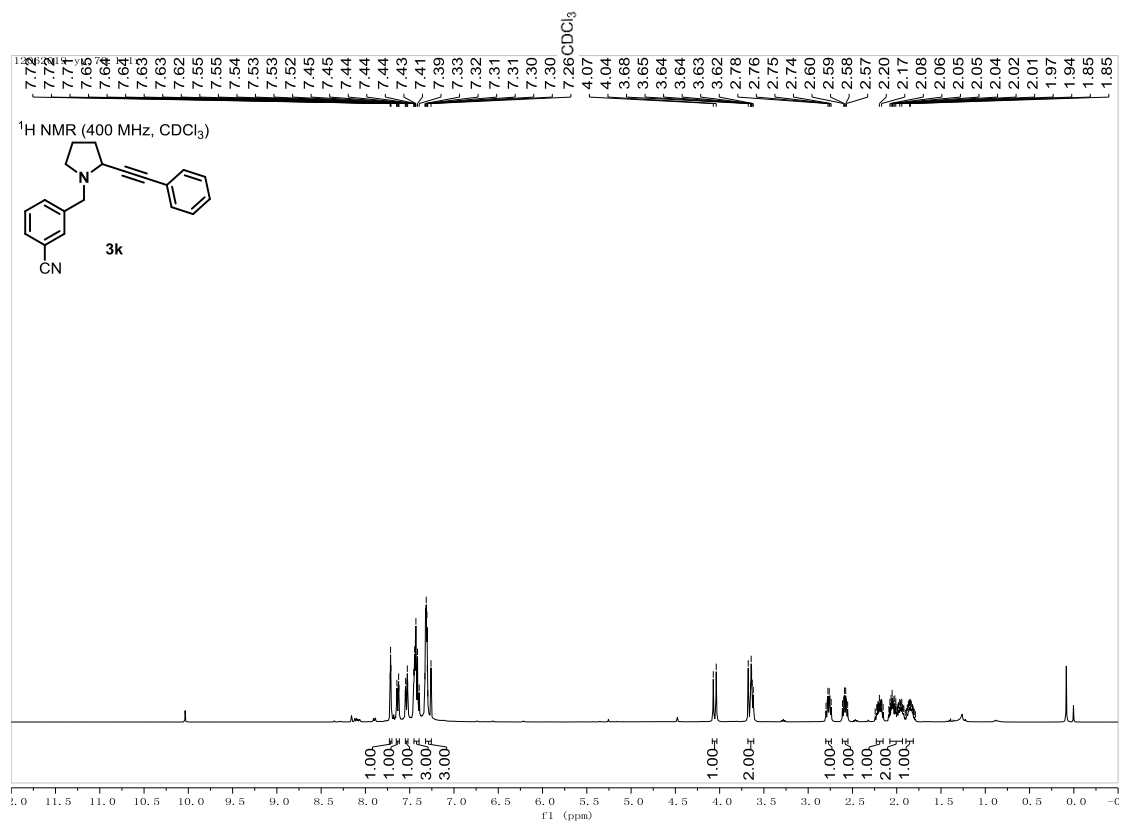


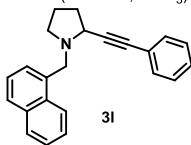


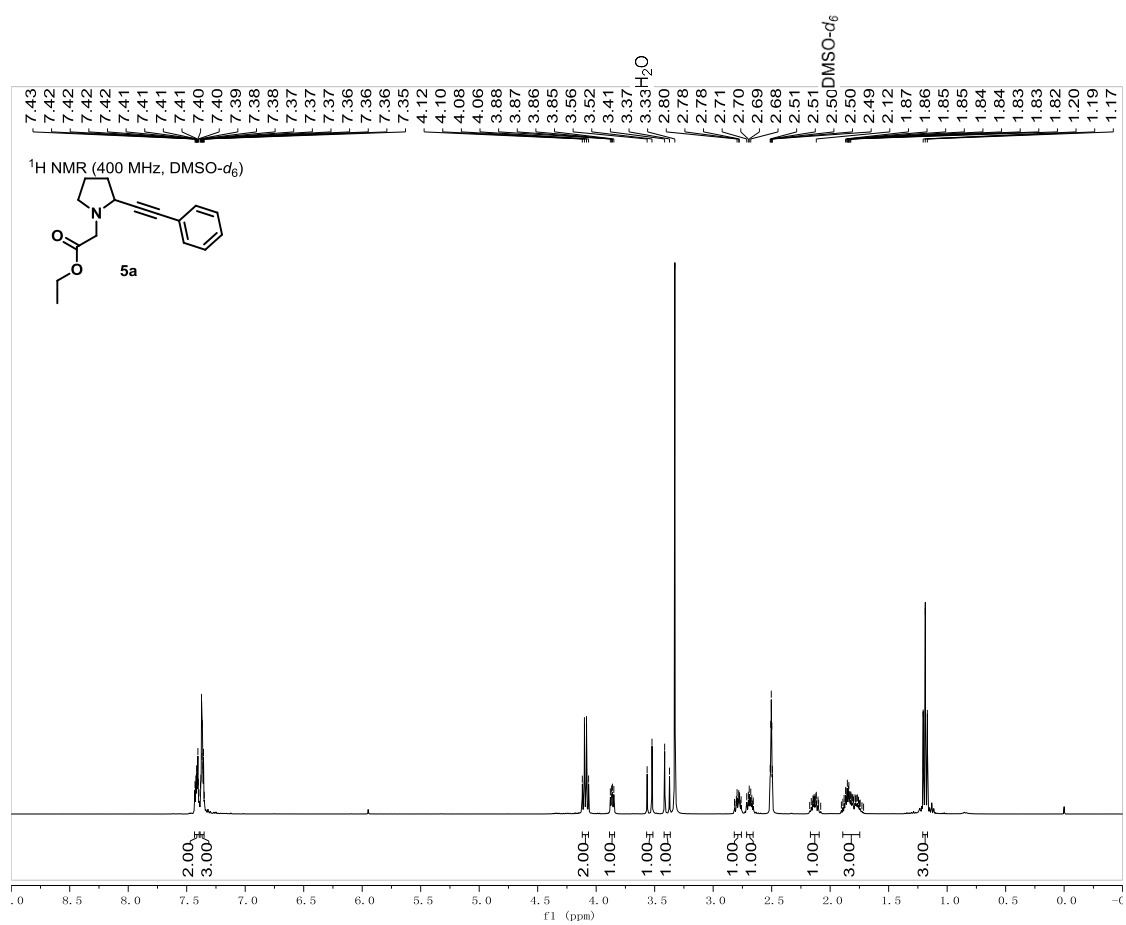


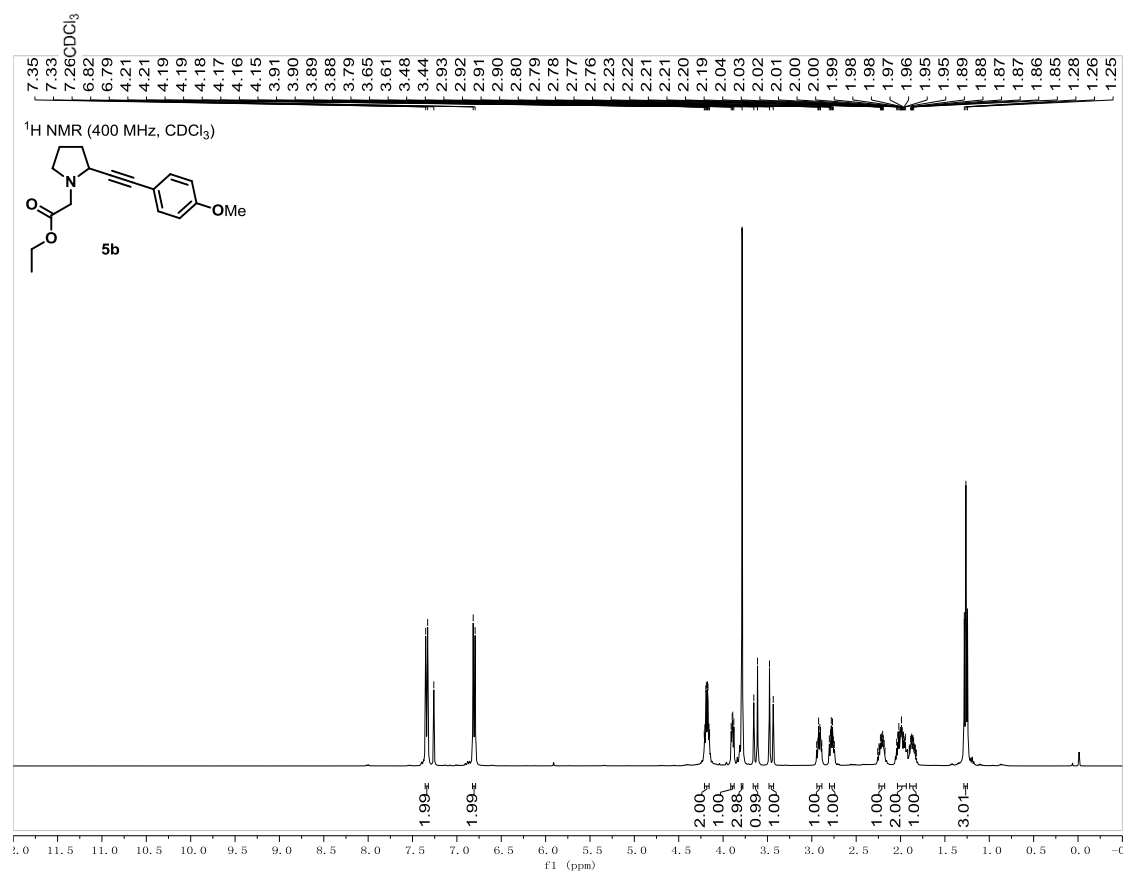


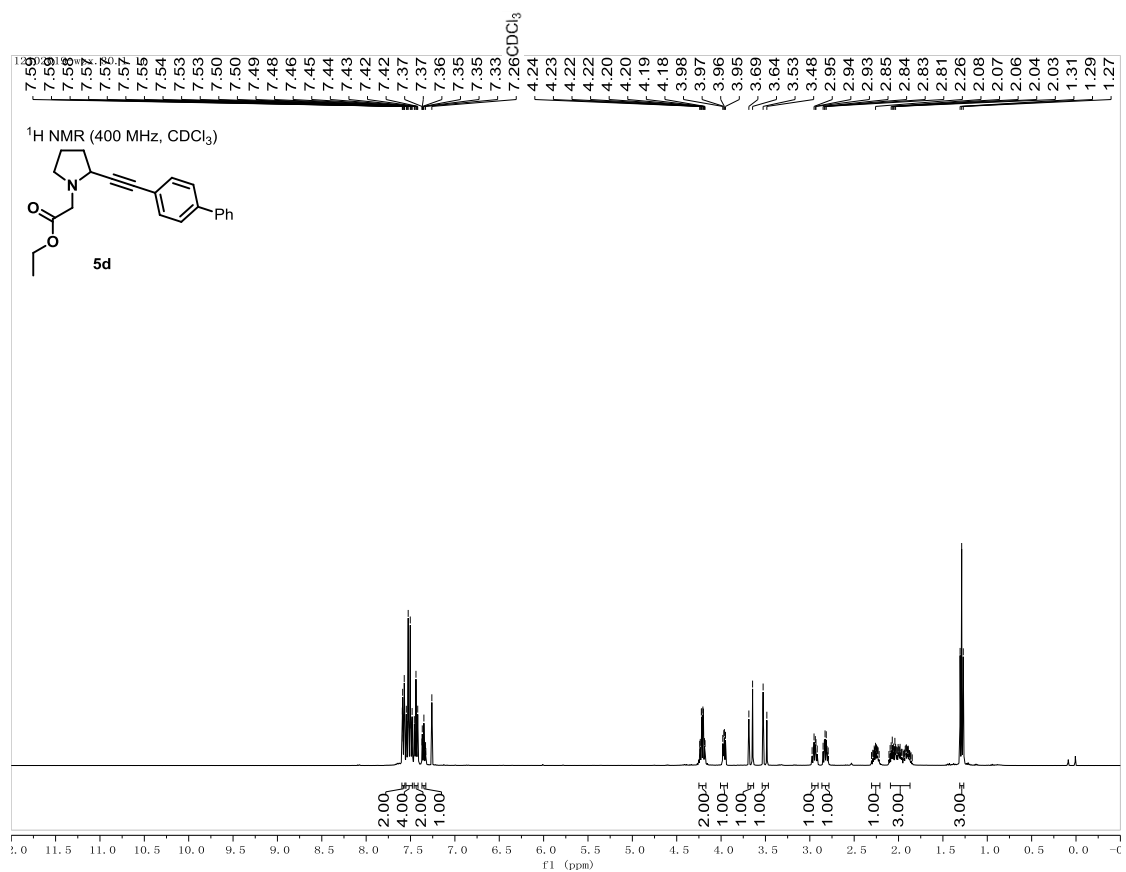


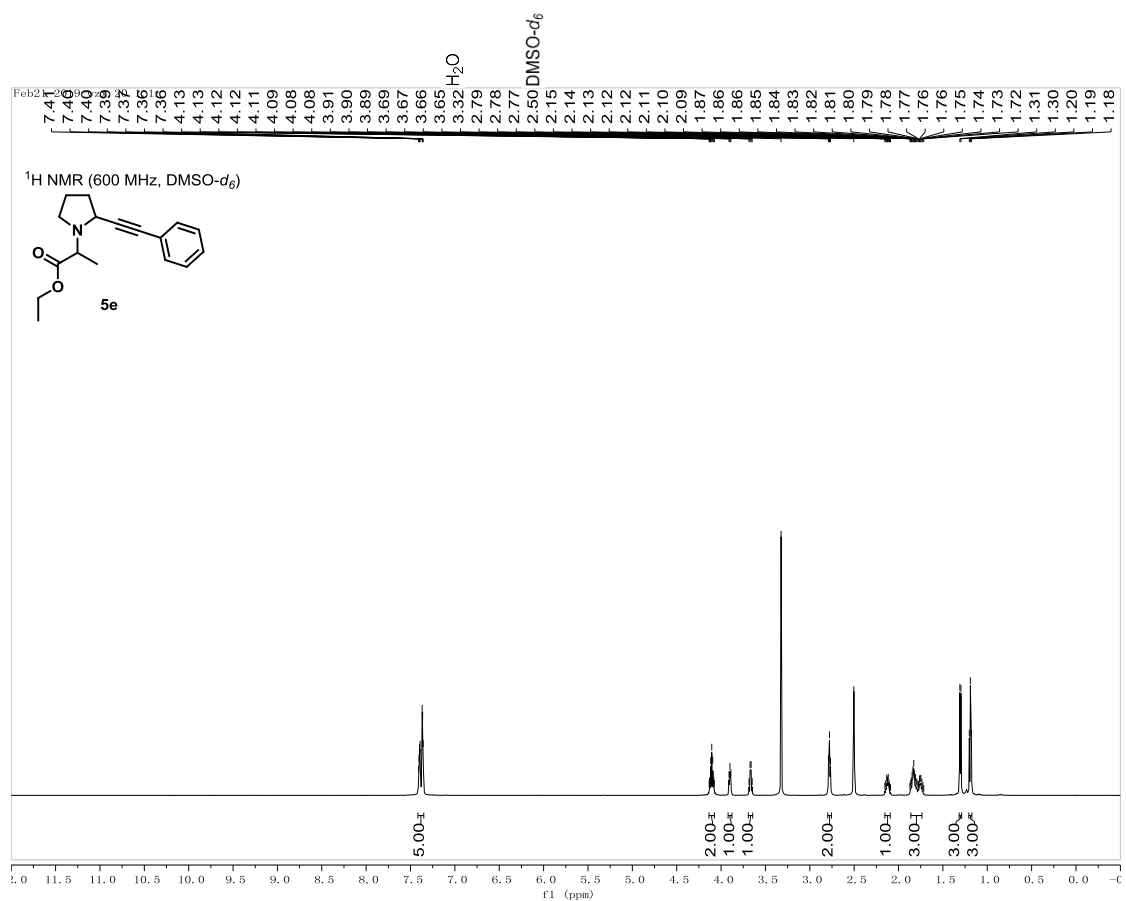


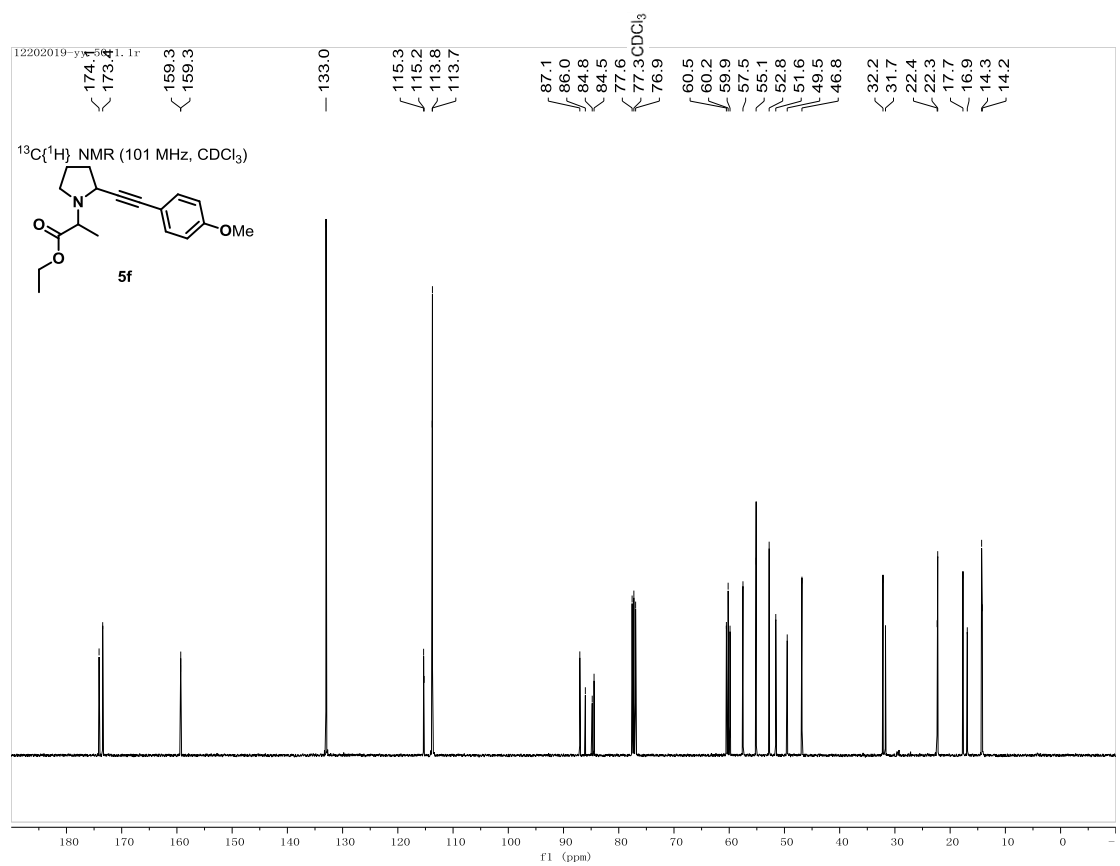
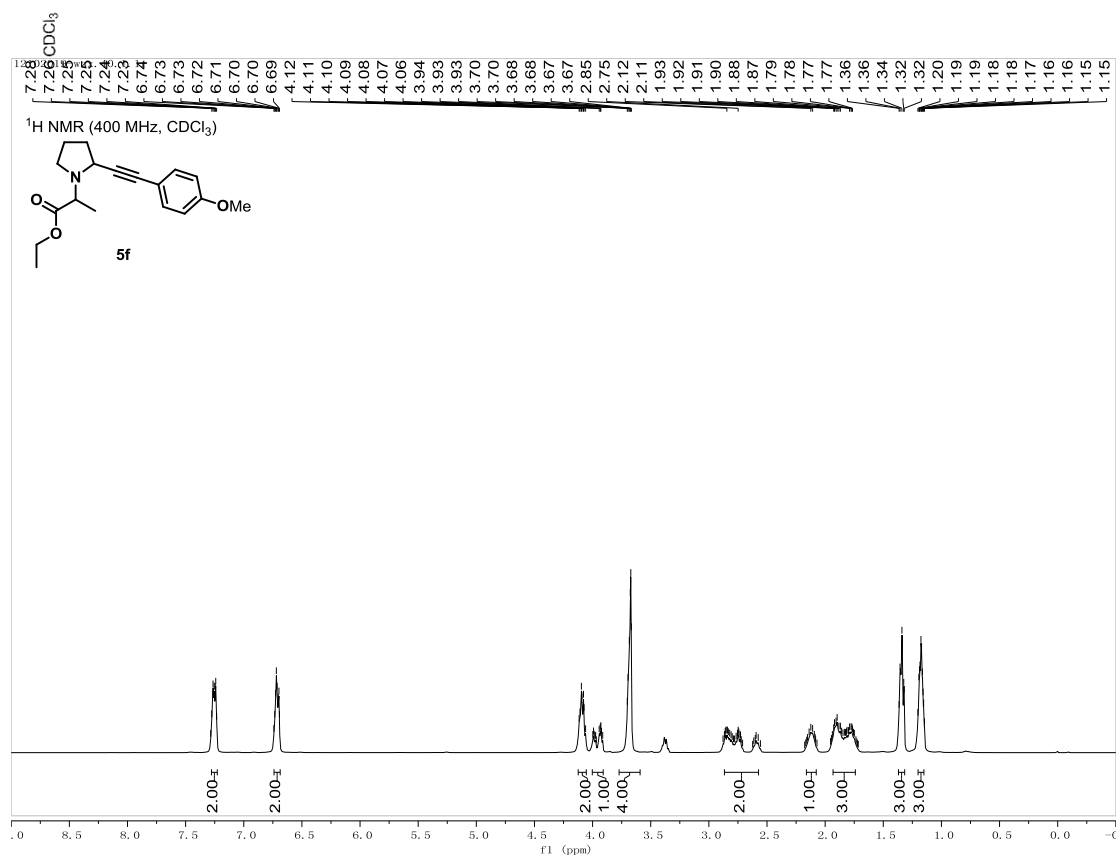


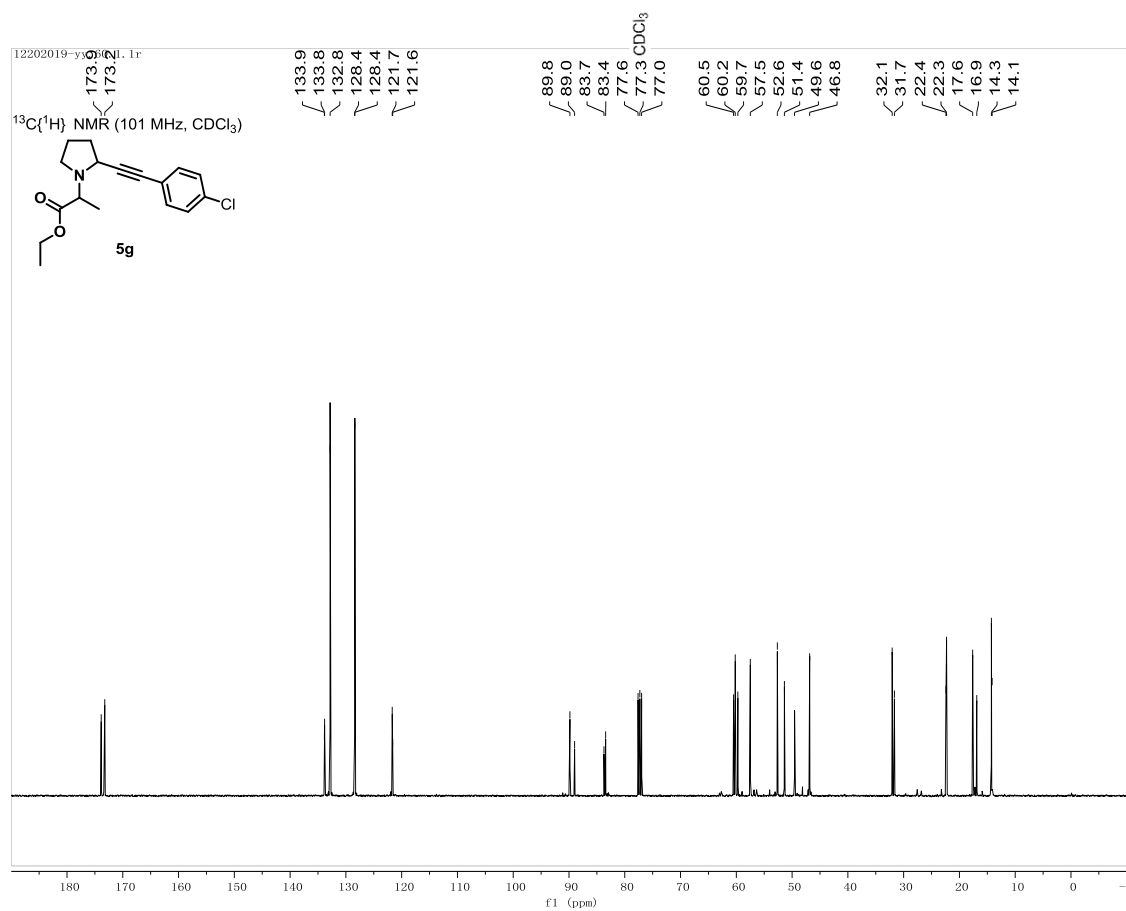
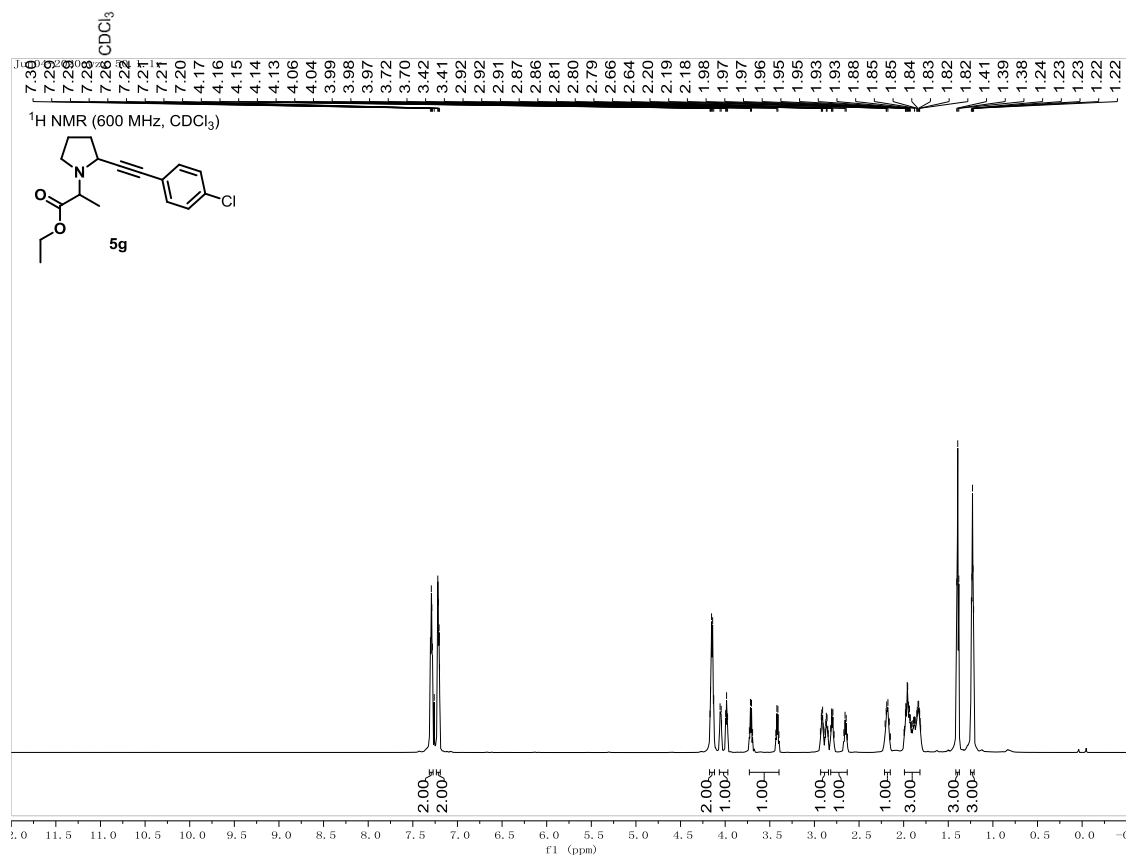


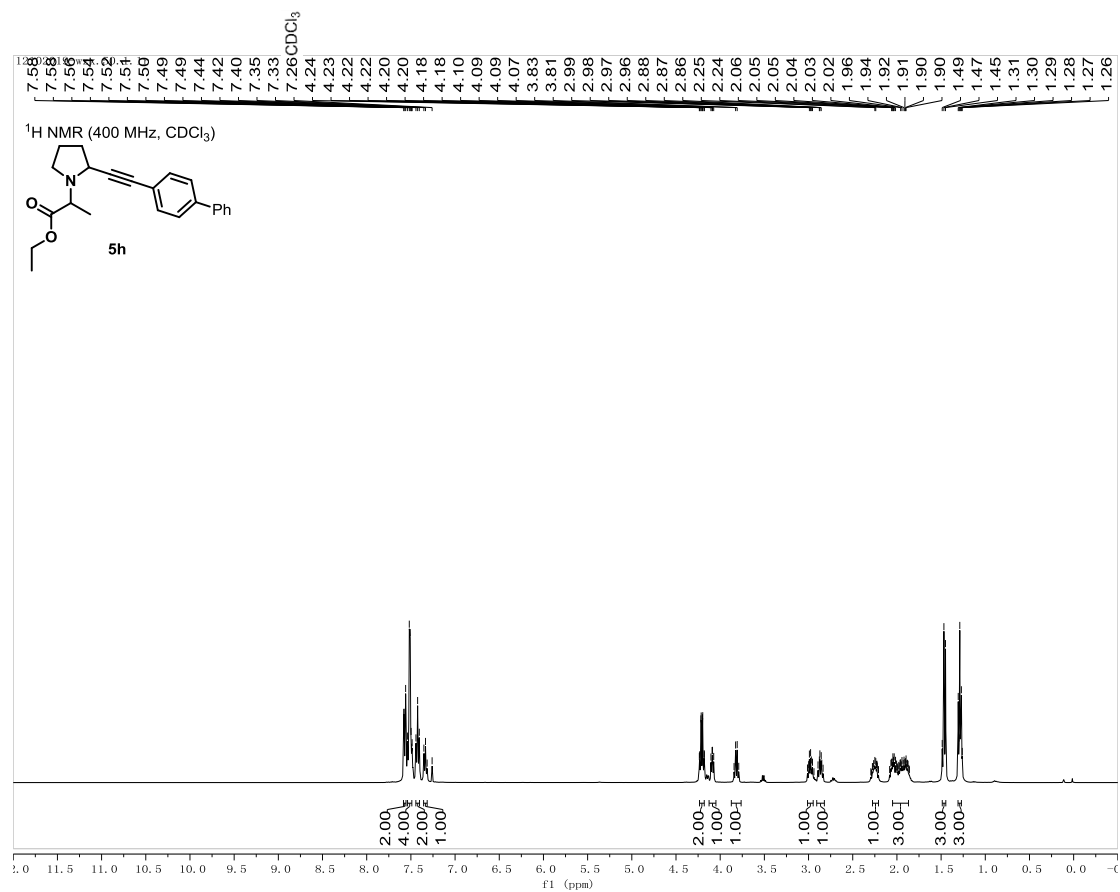


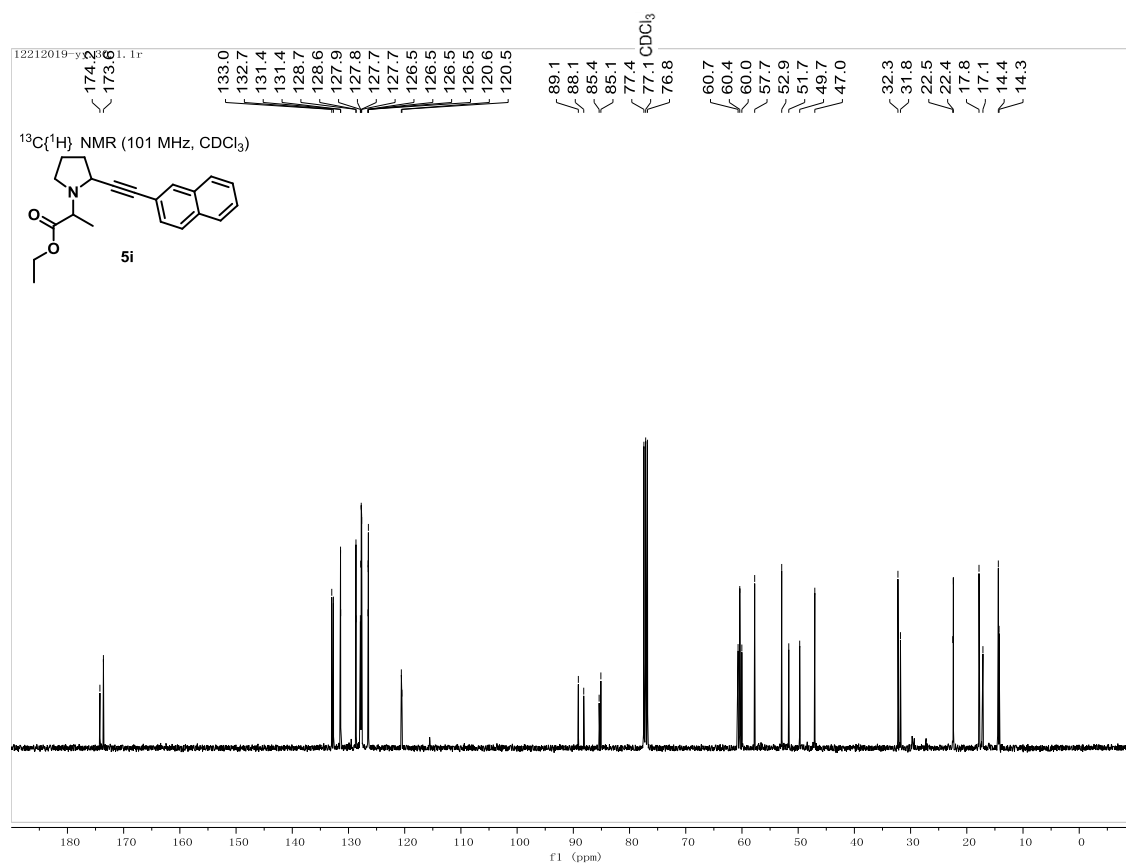
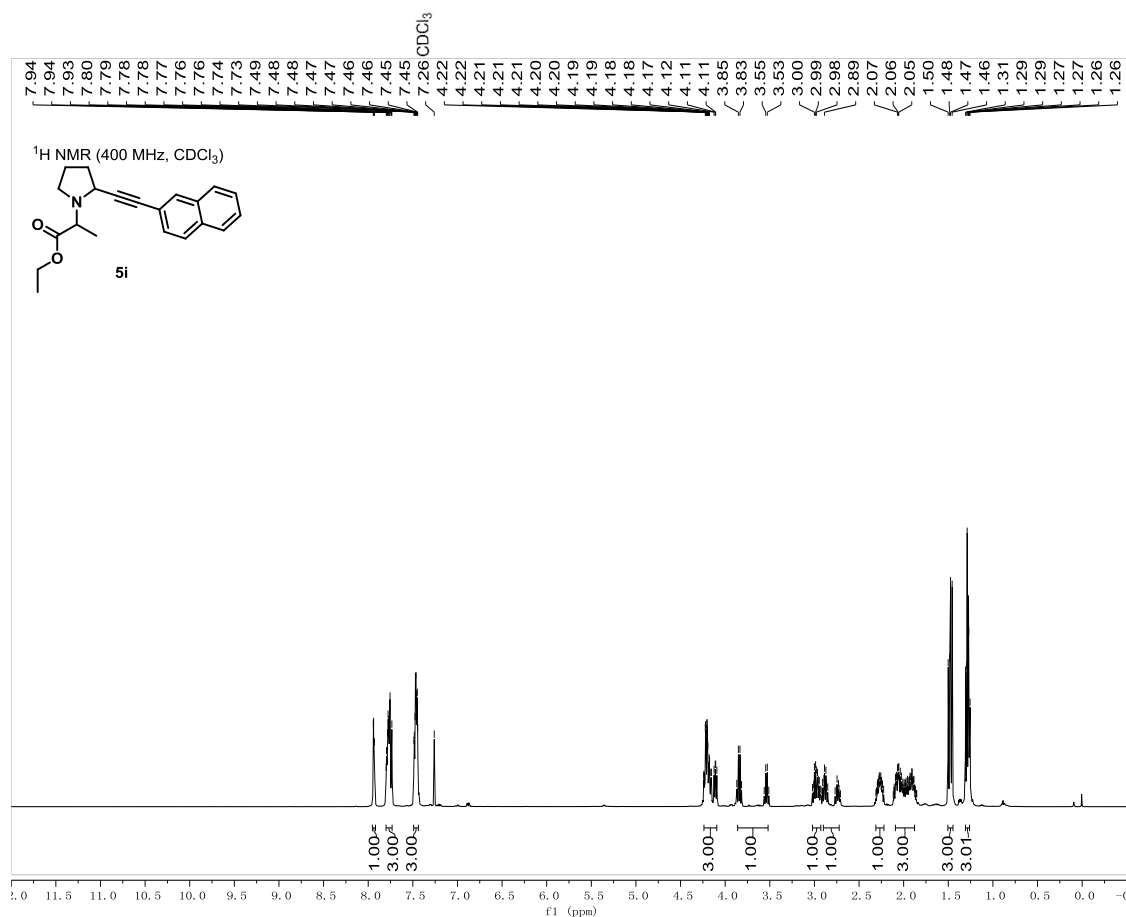




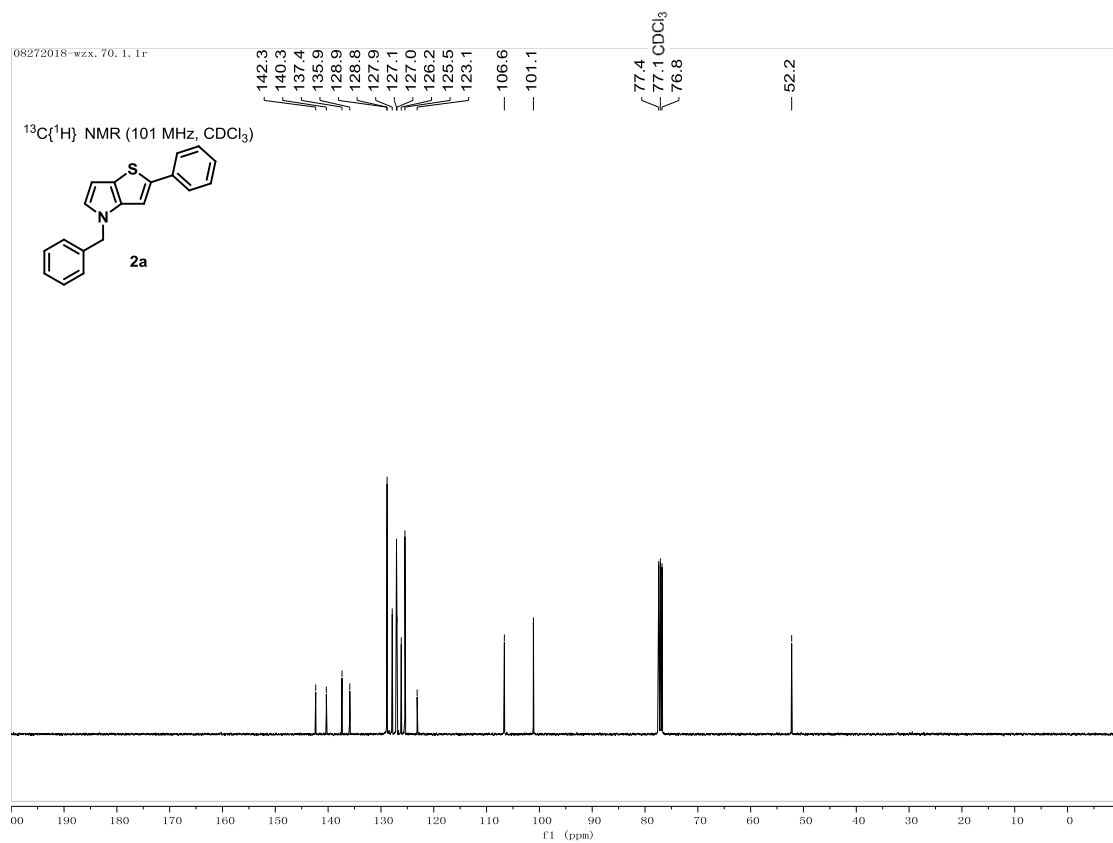
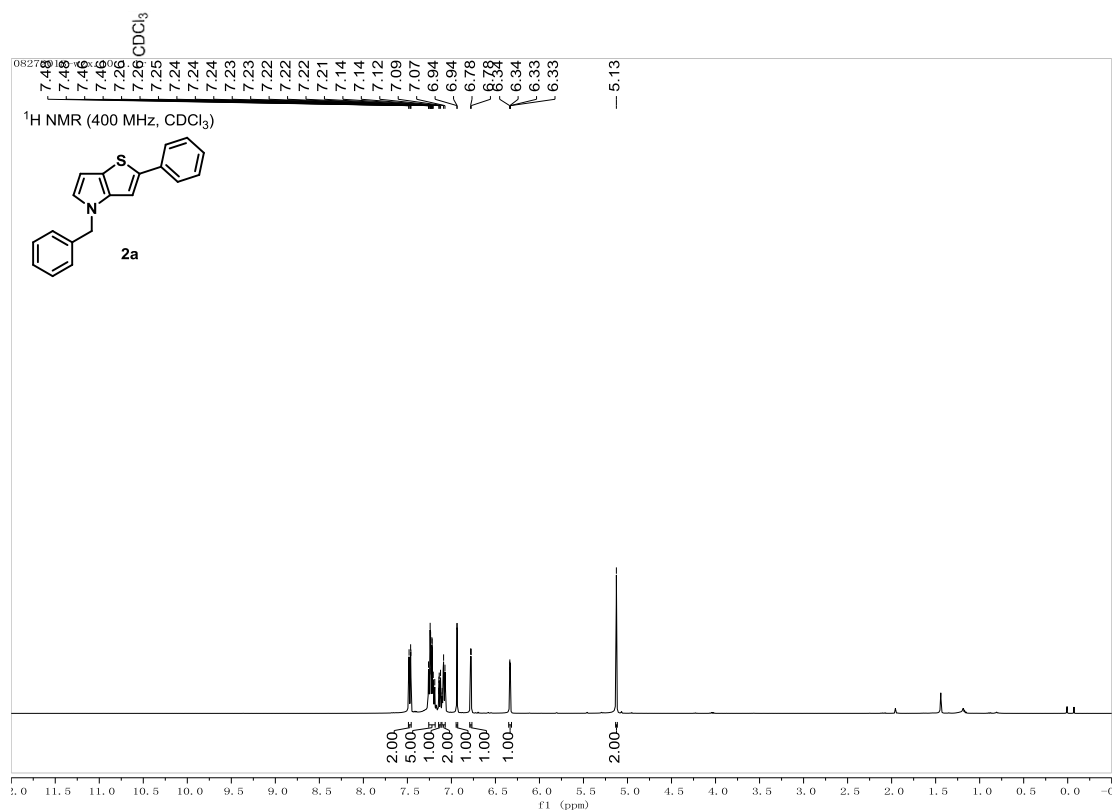


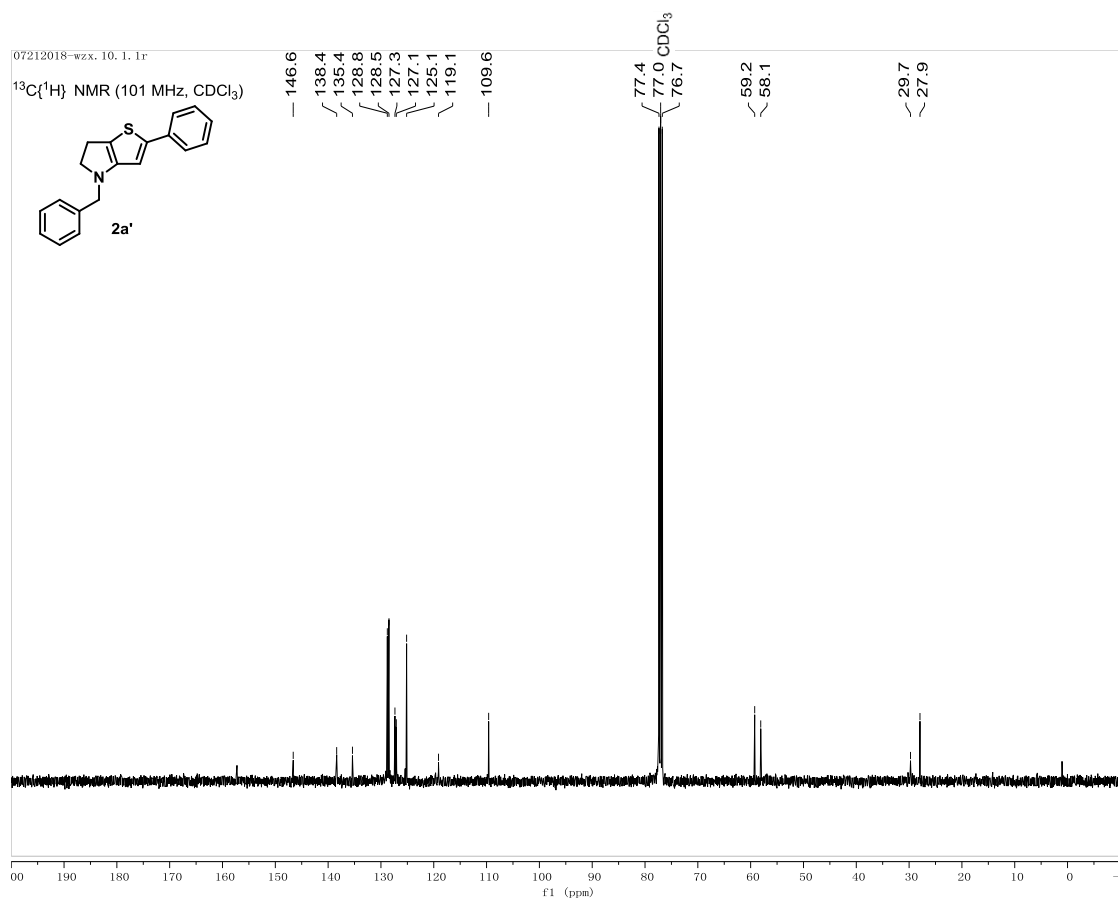
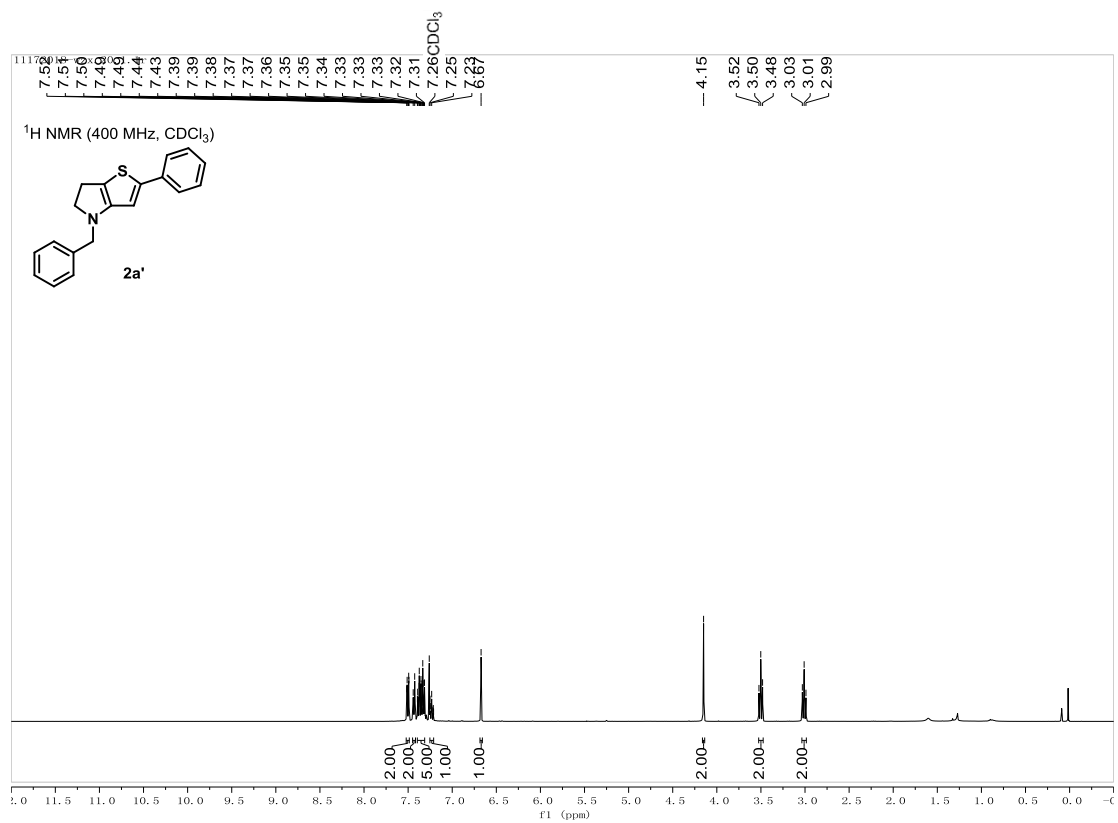


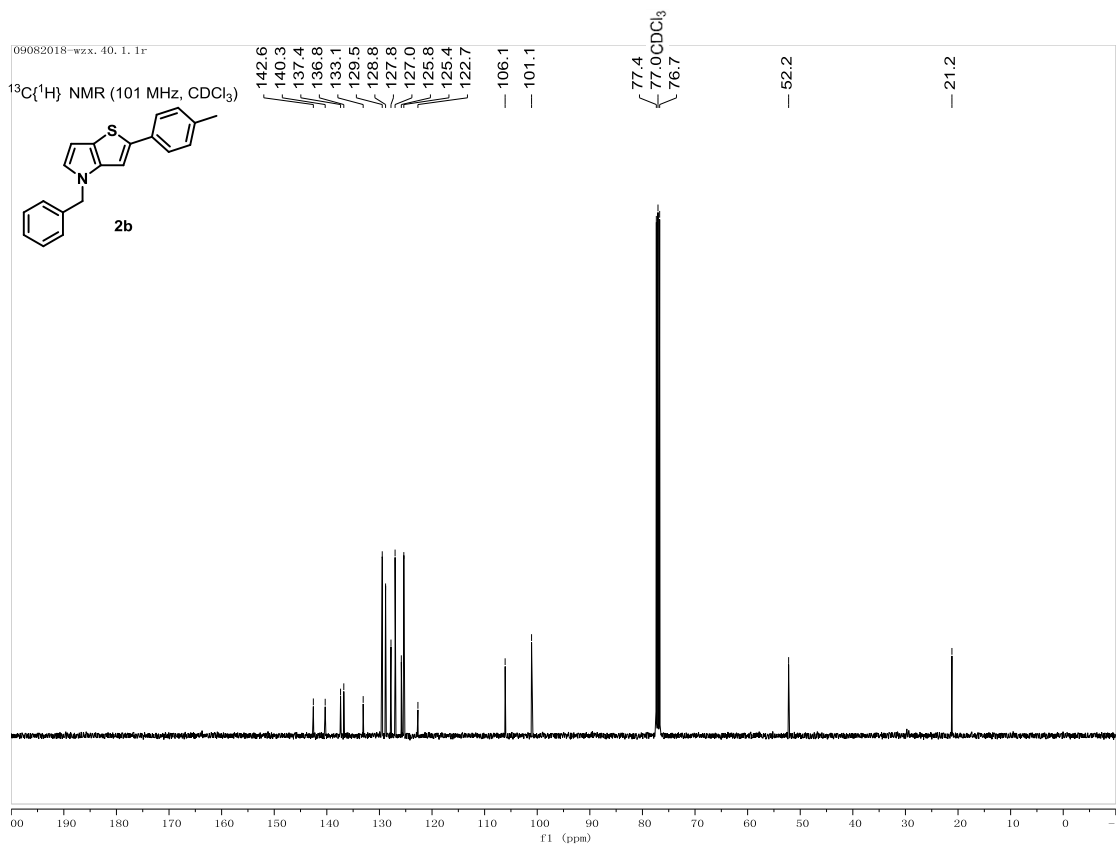
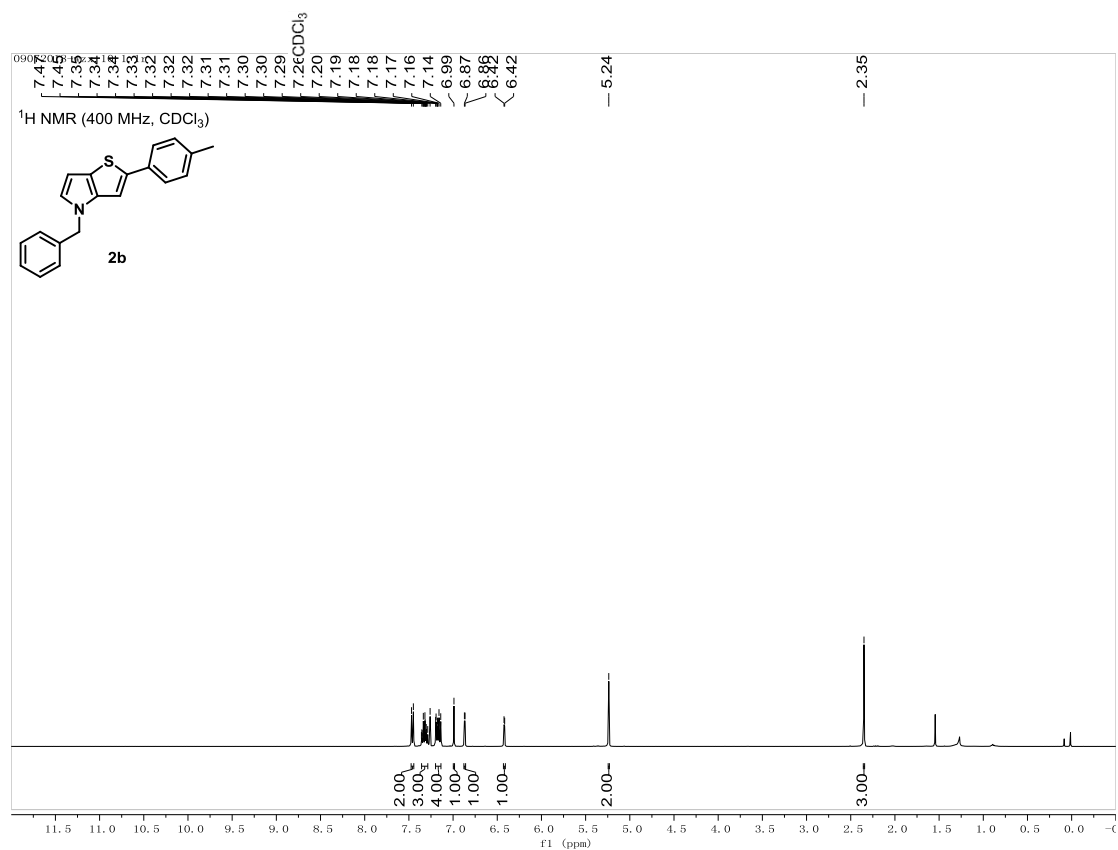


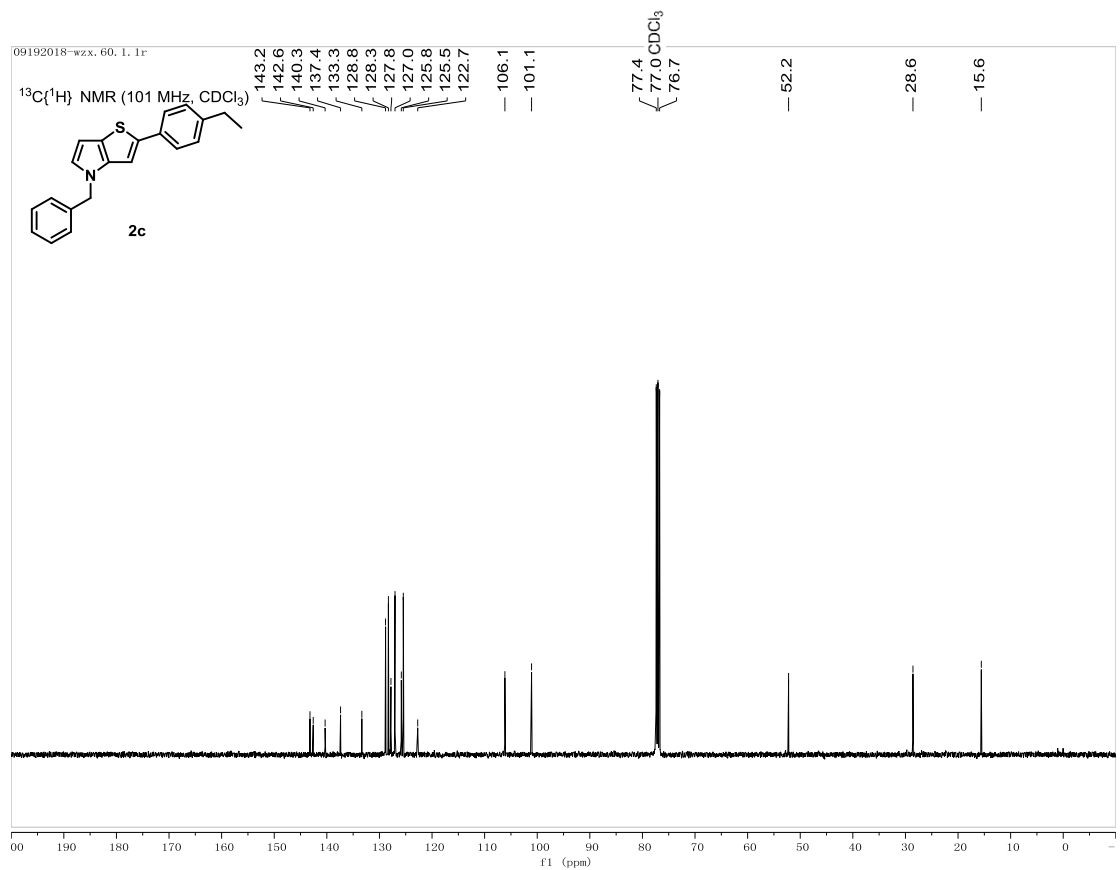
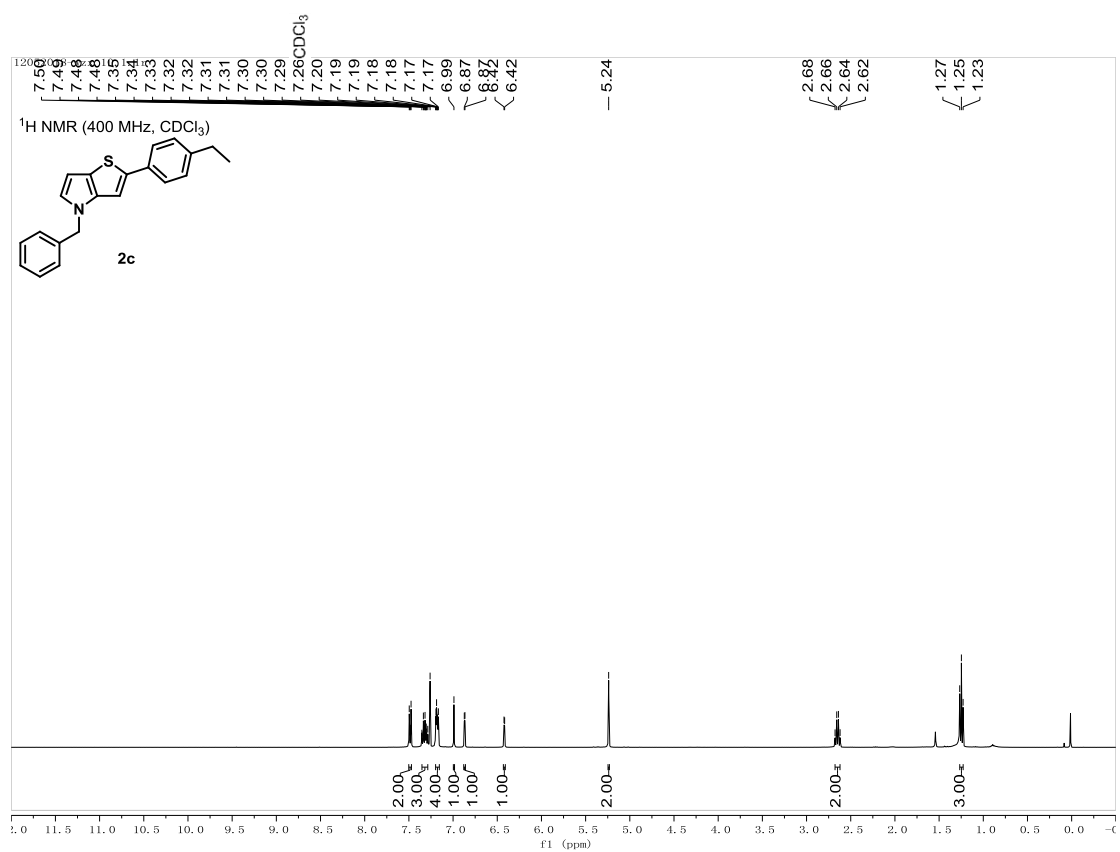


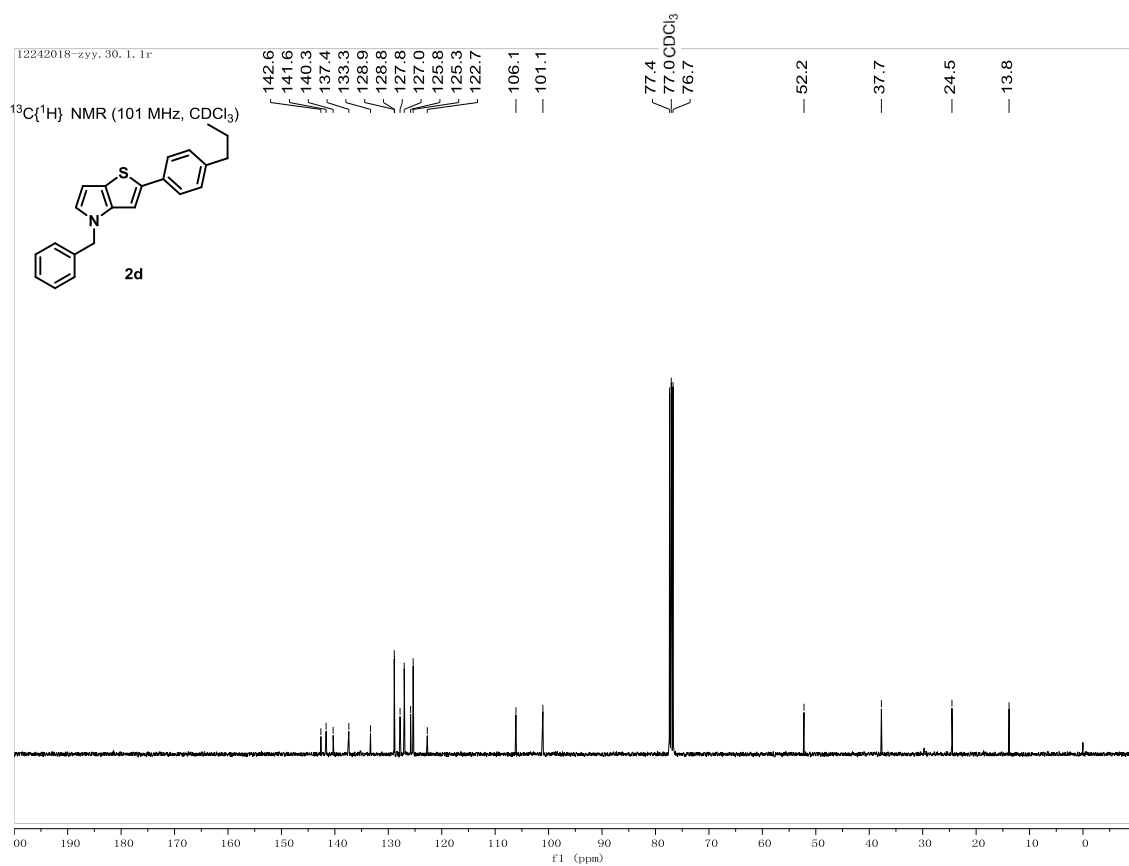
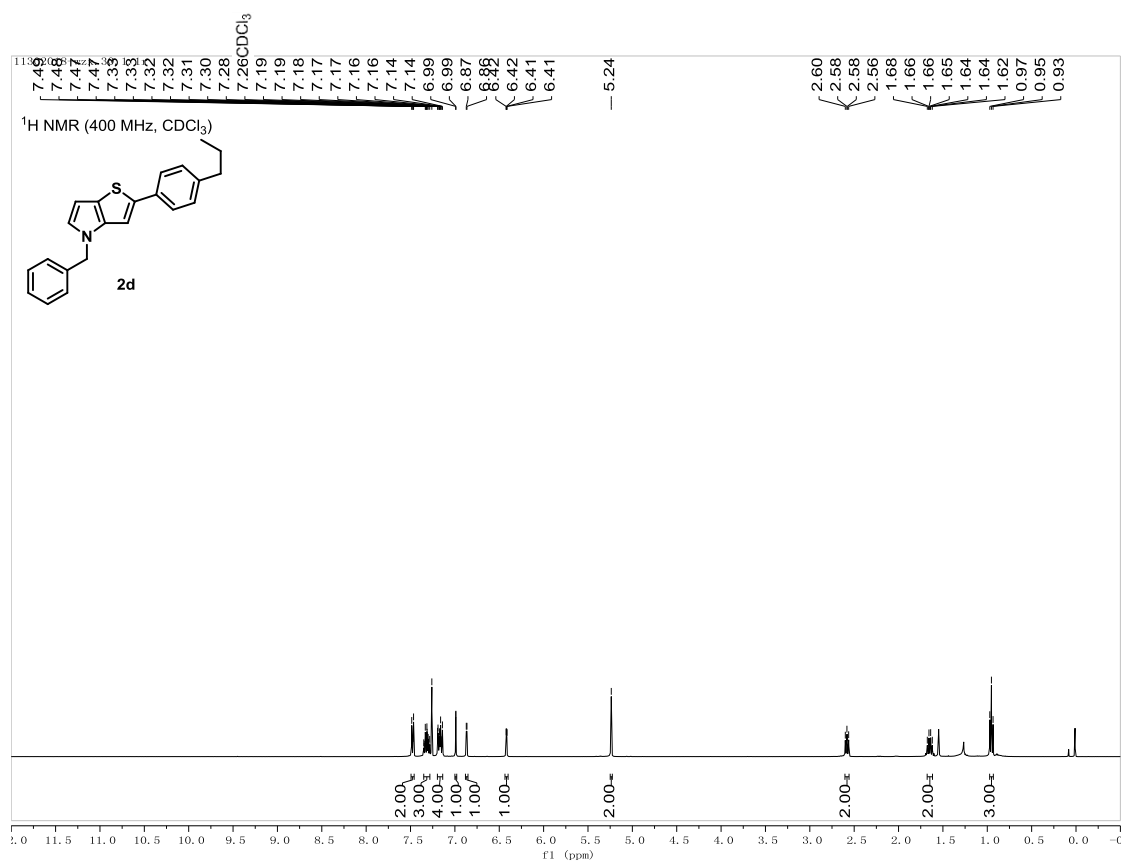
4. ^1H NMR and ^{13}C $\{^1\text{H}\}$ NMR copies of products

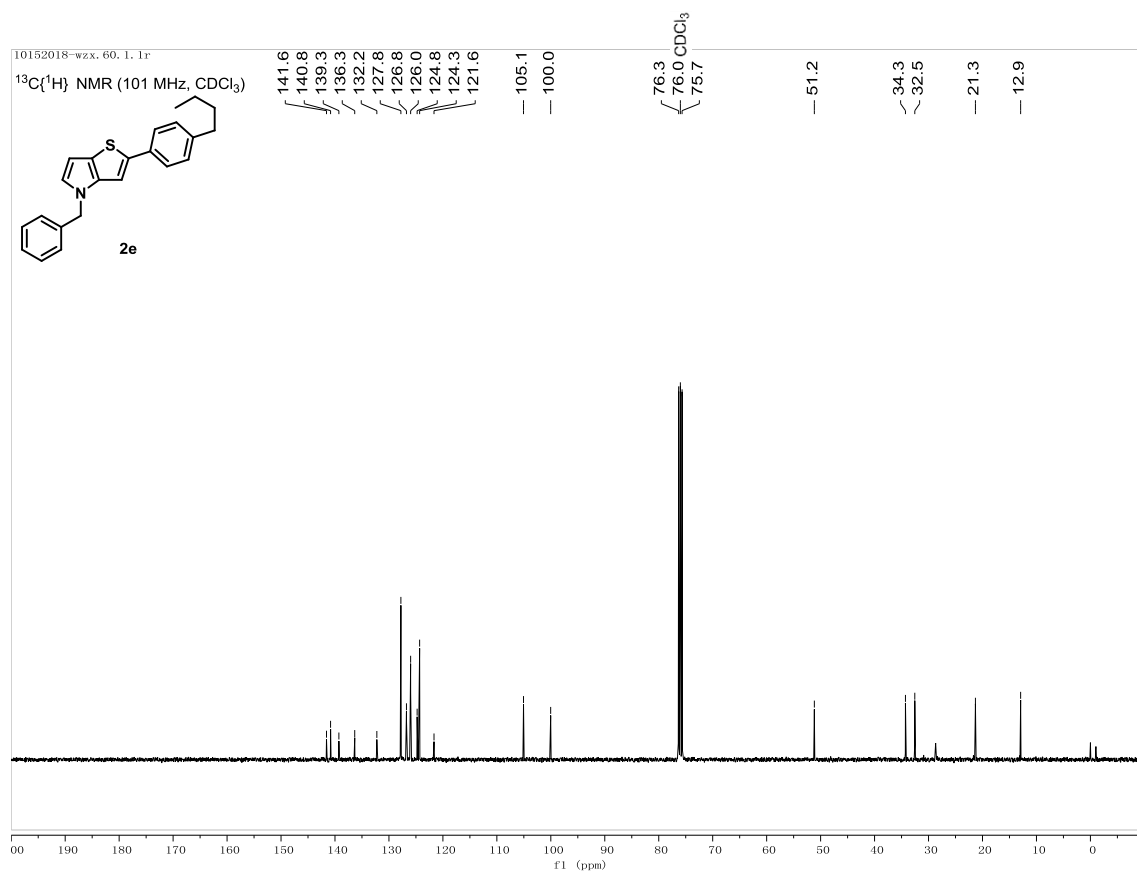
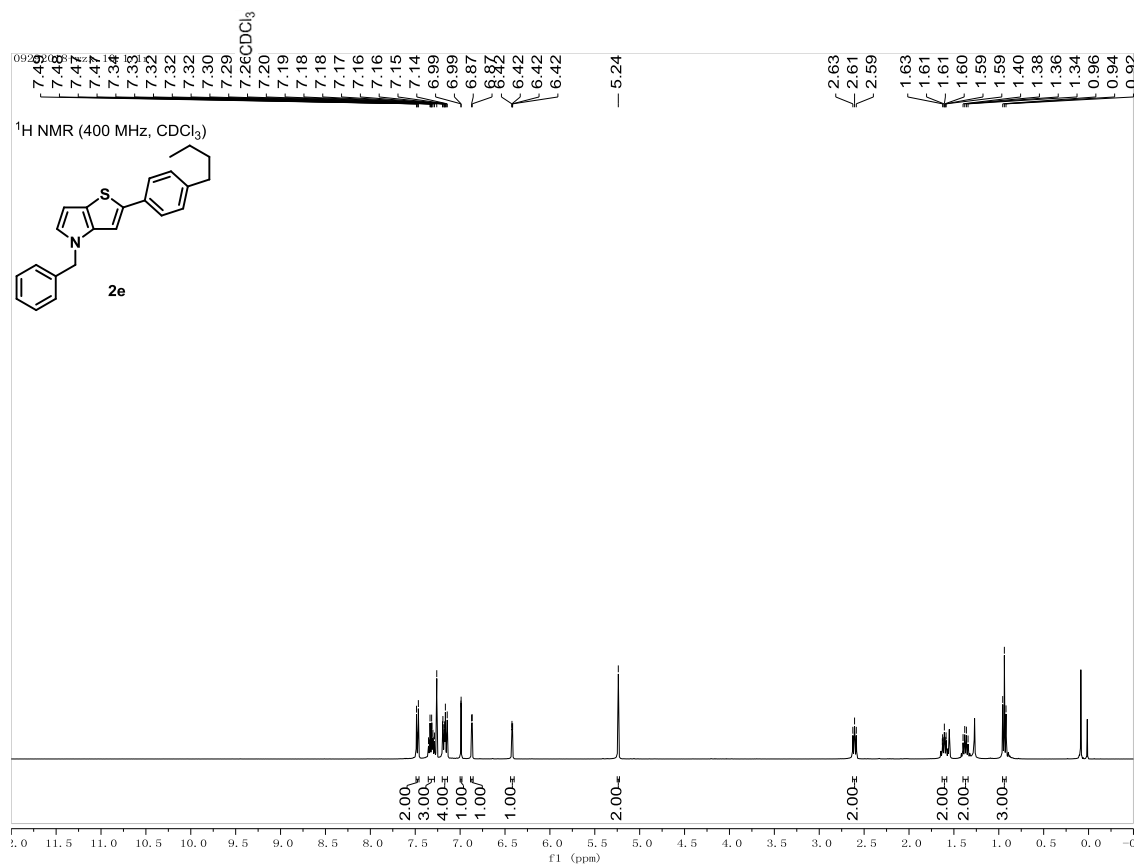


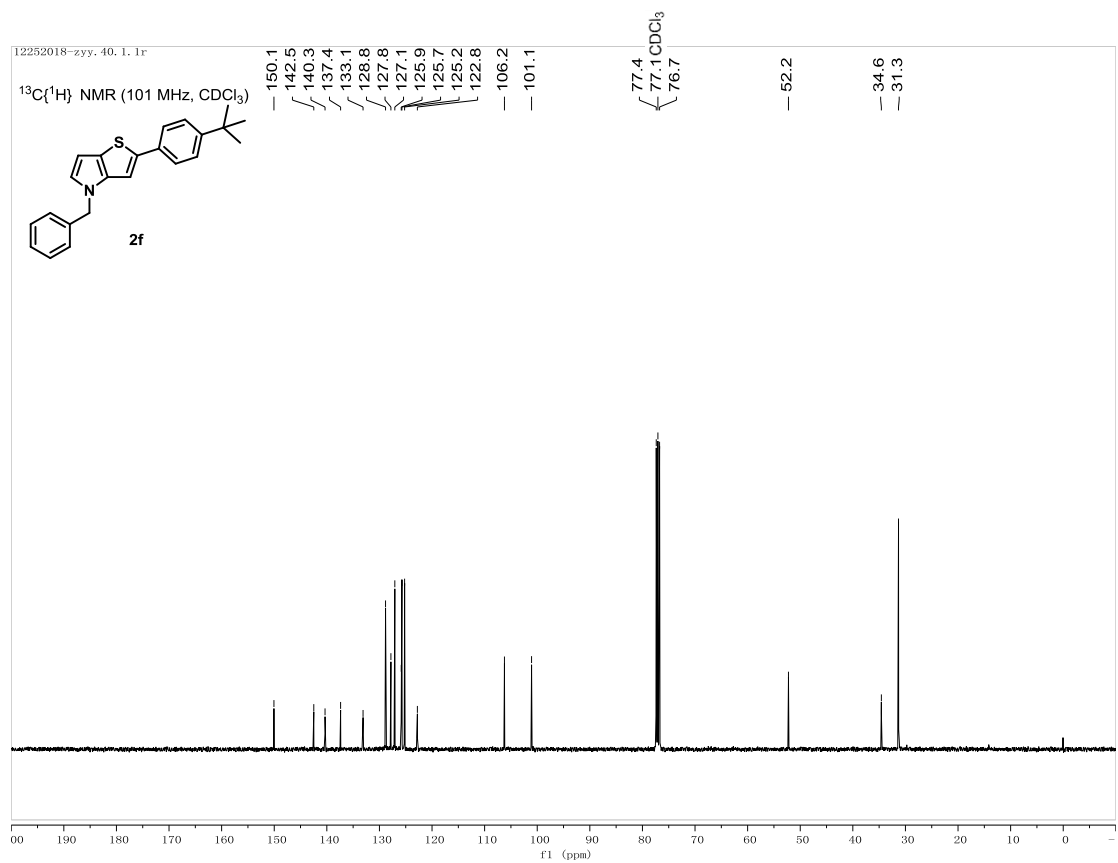
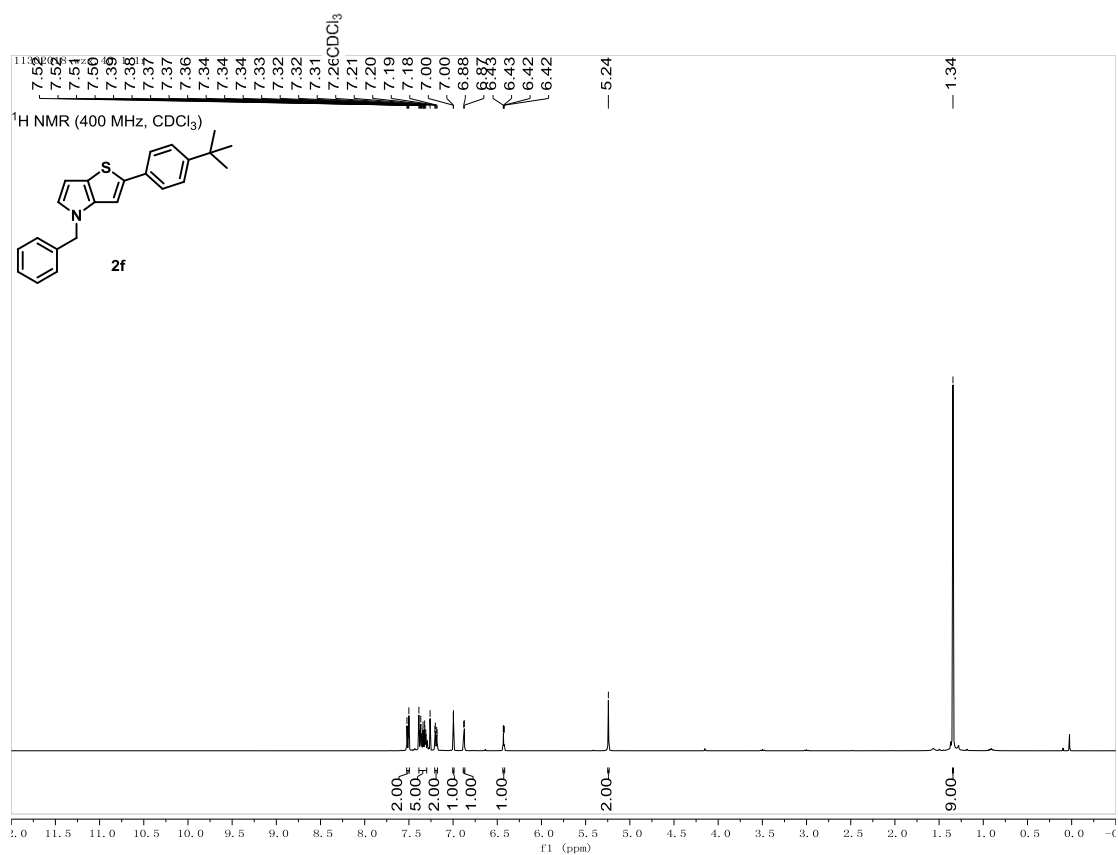


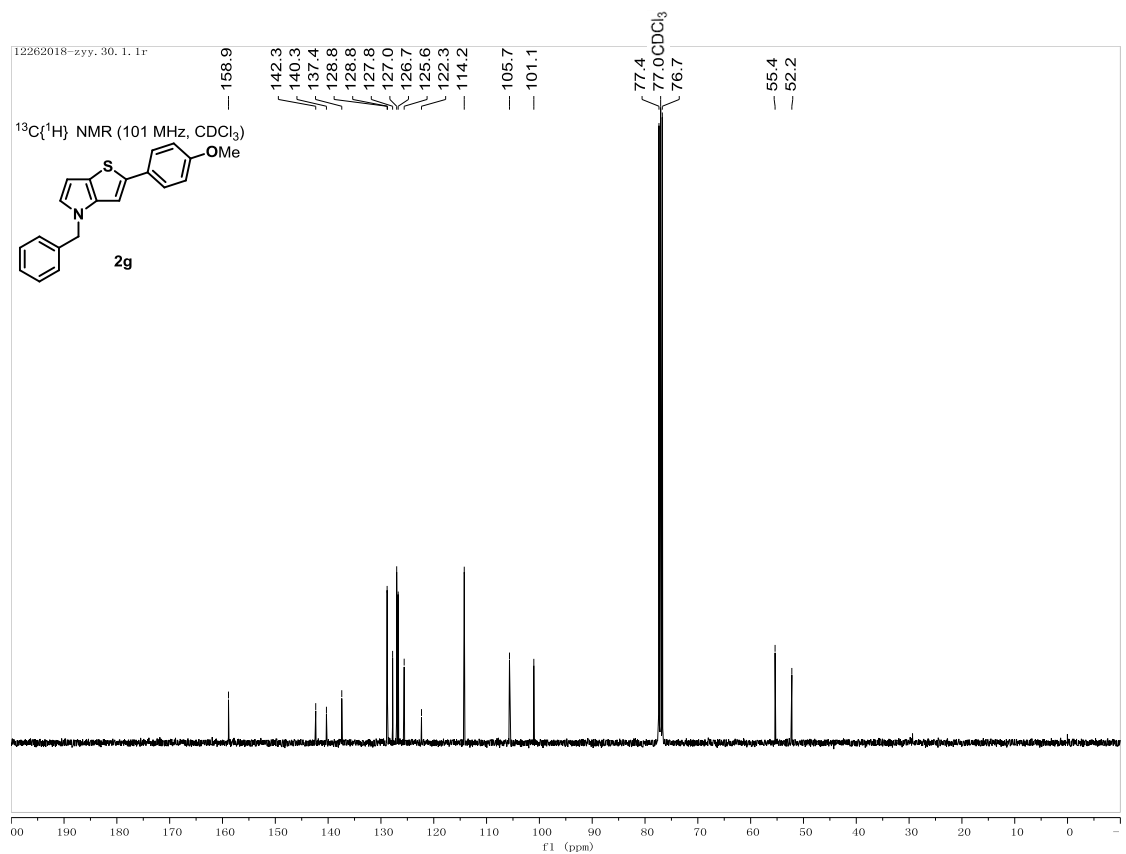
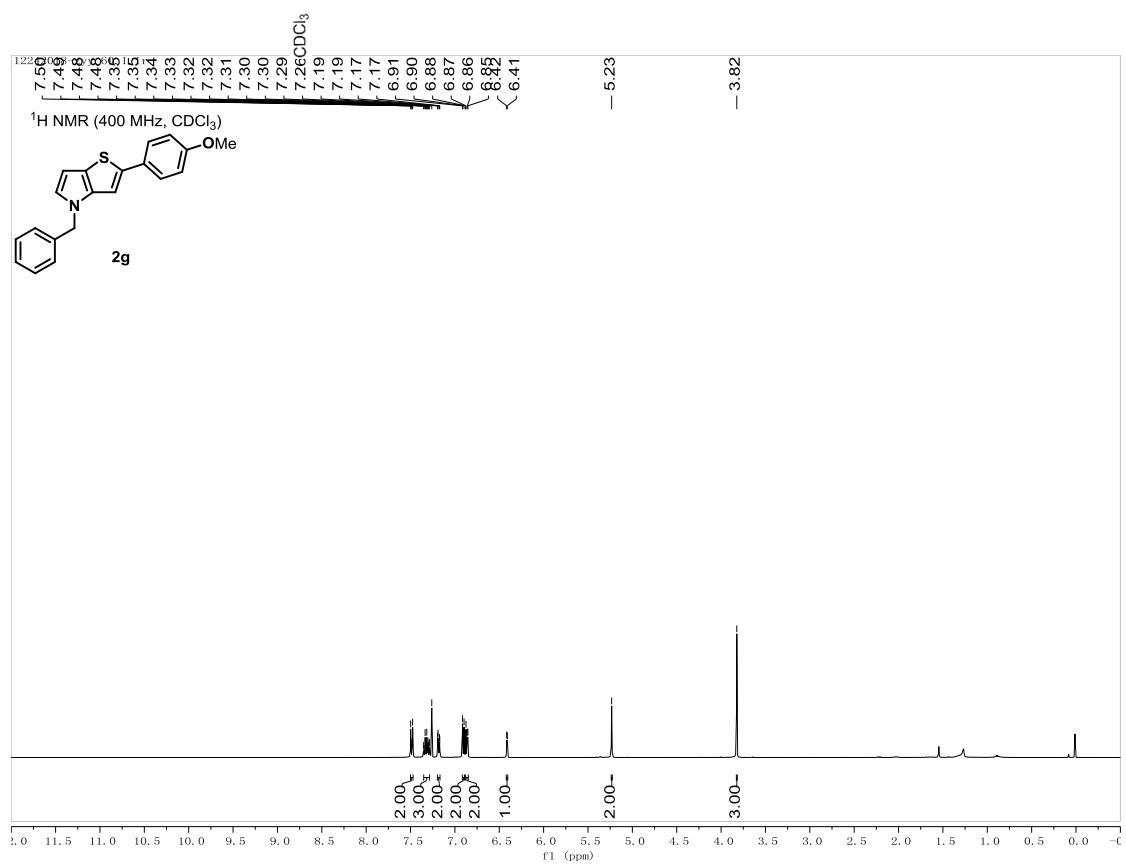


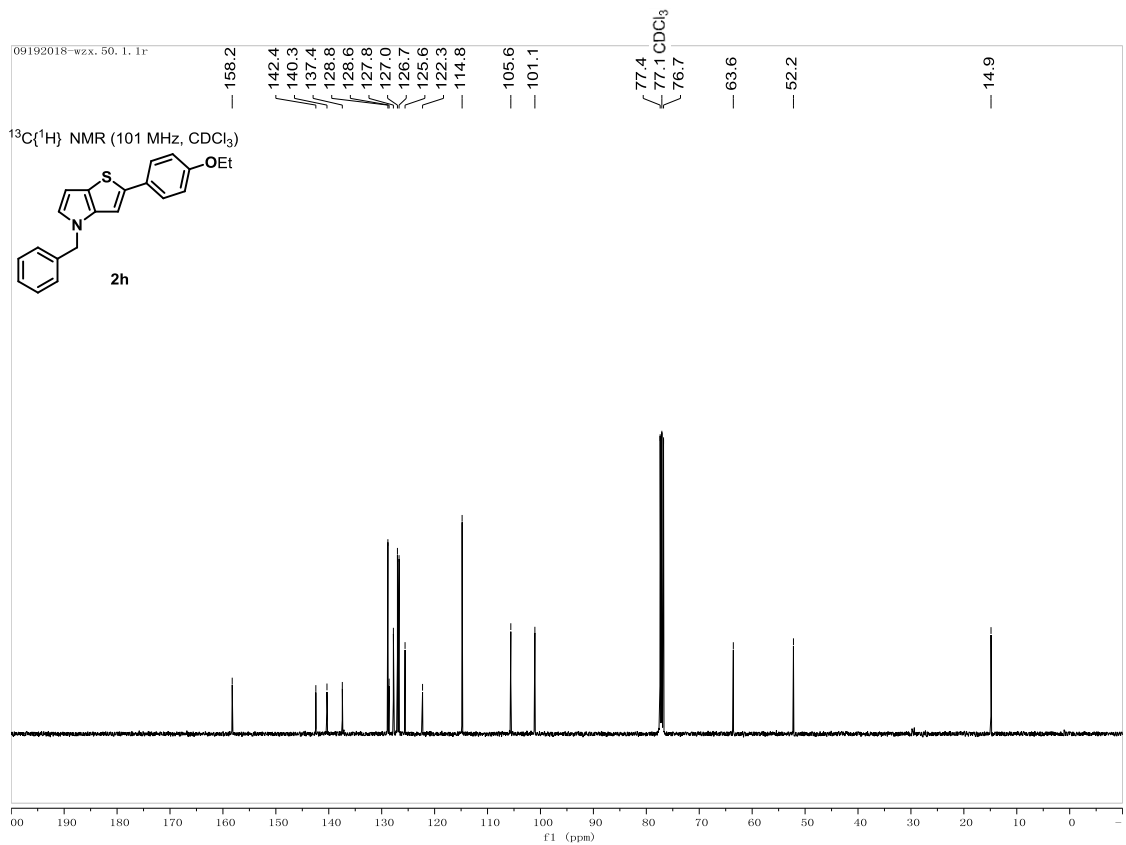
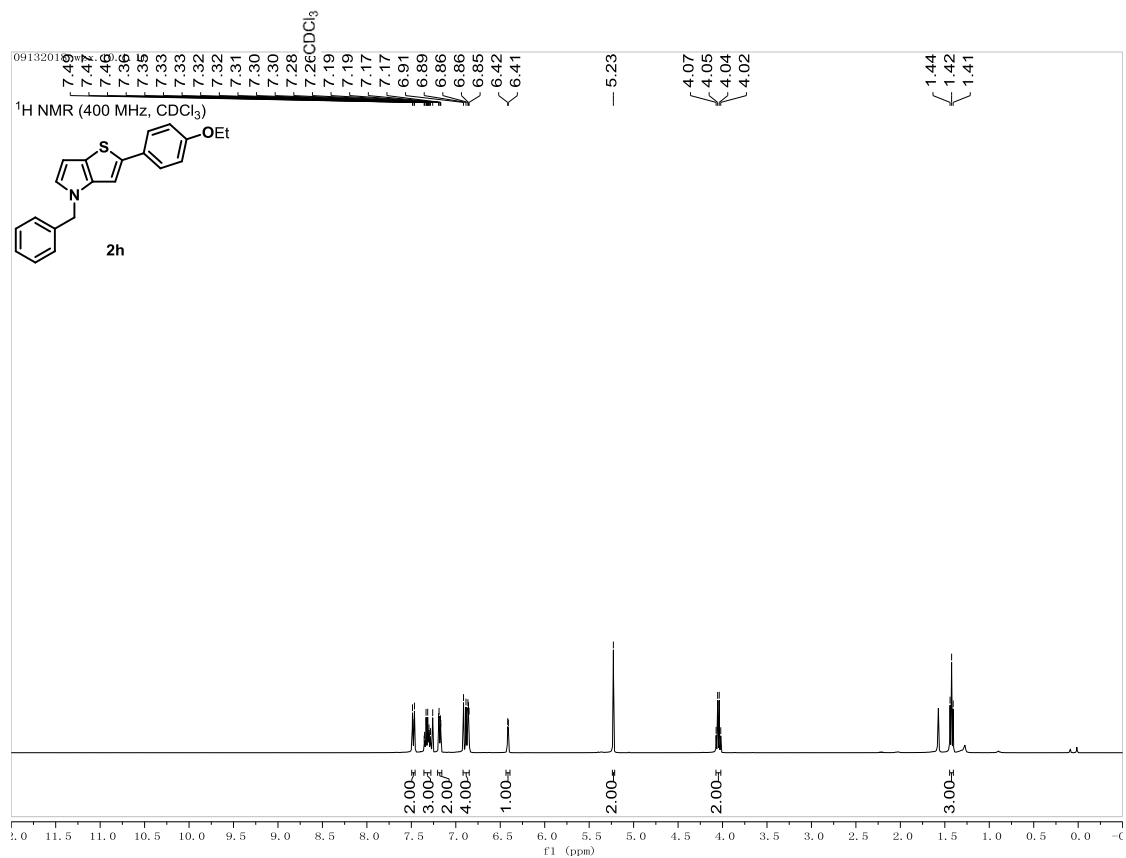


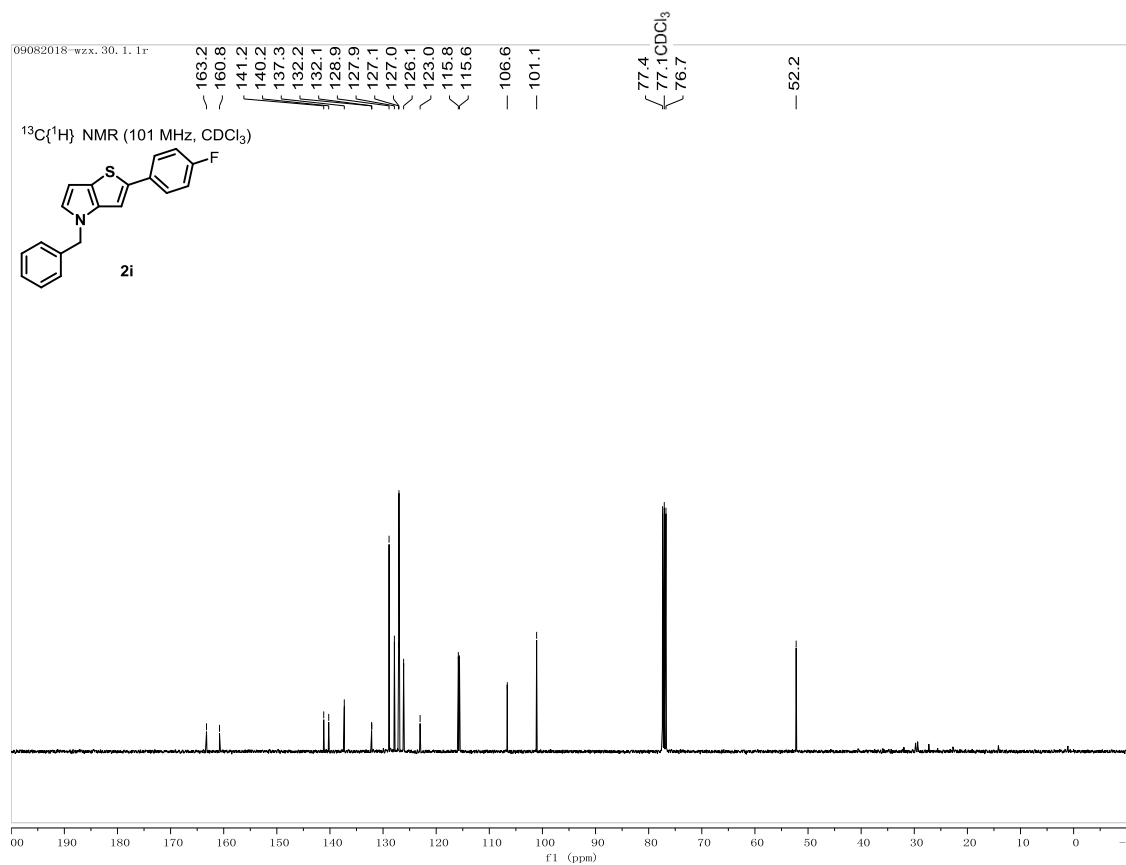
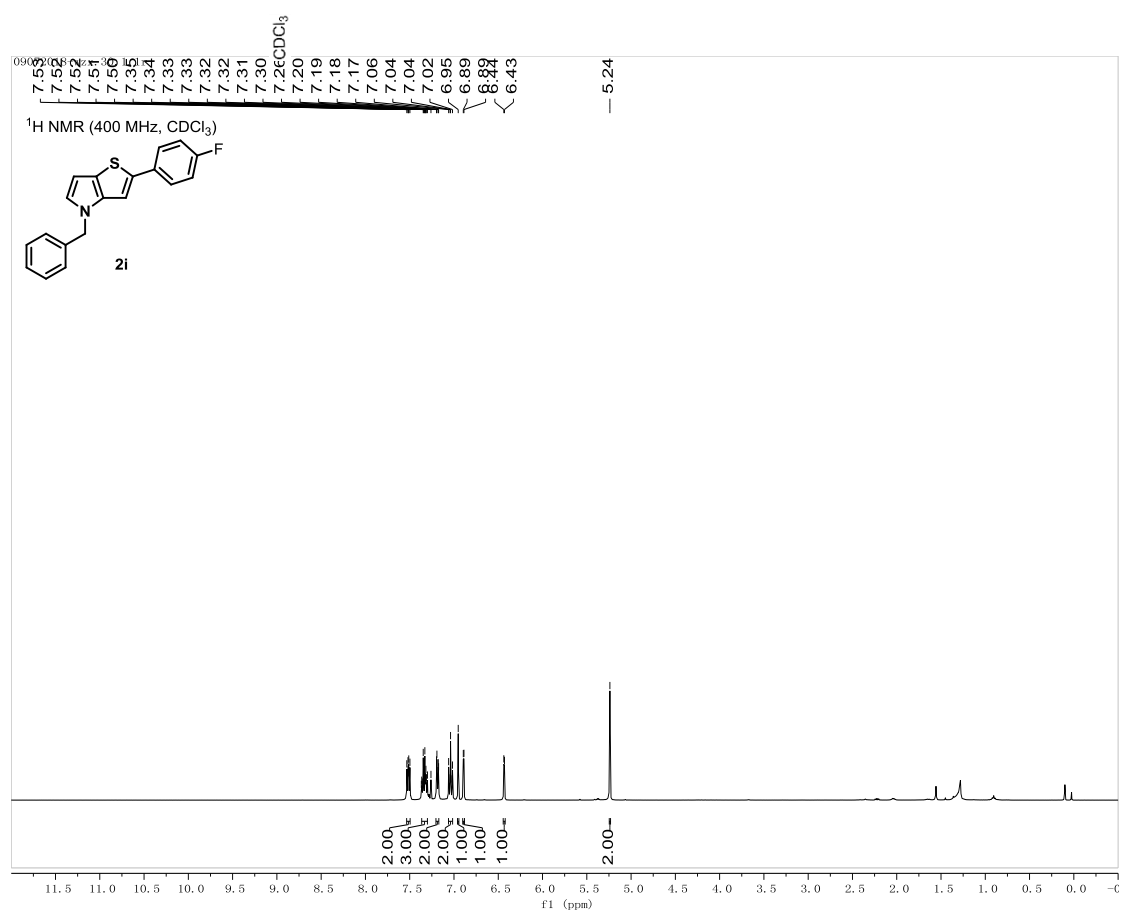






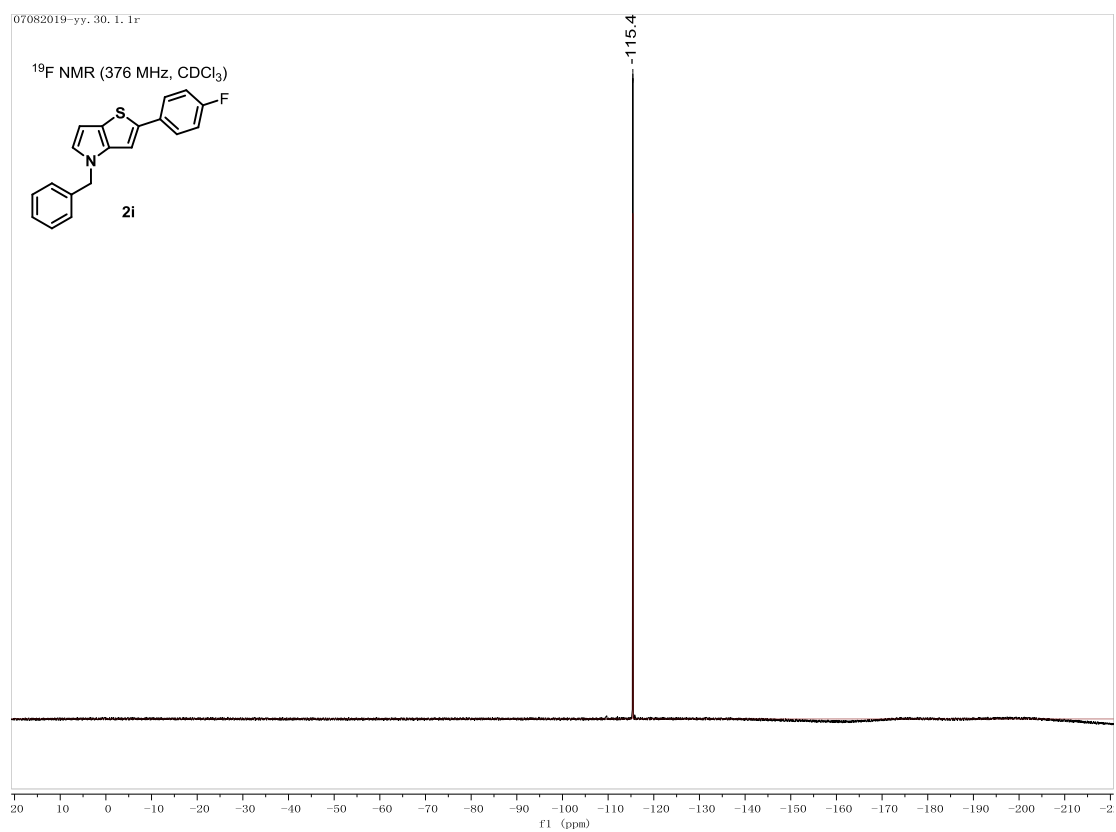
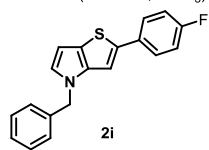


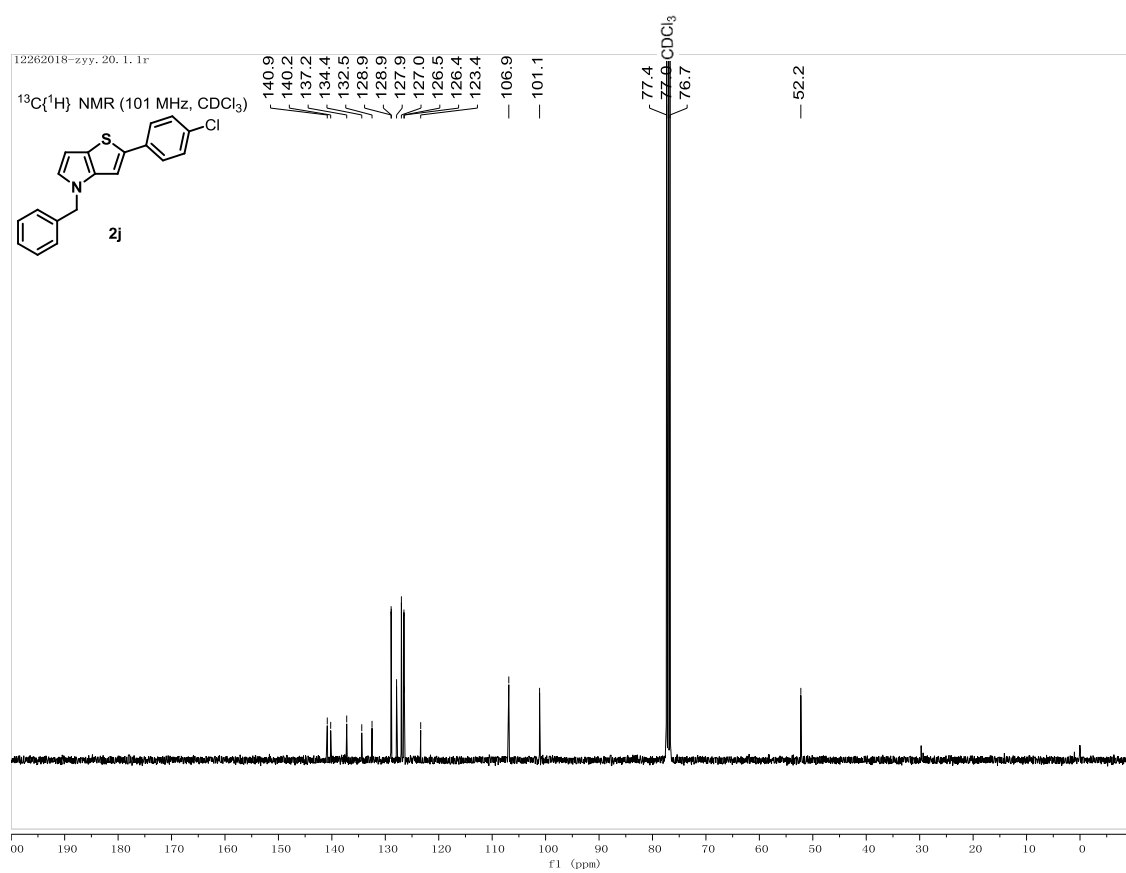
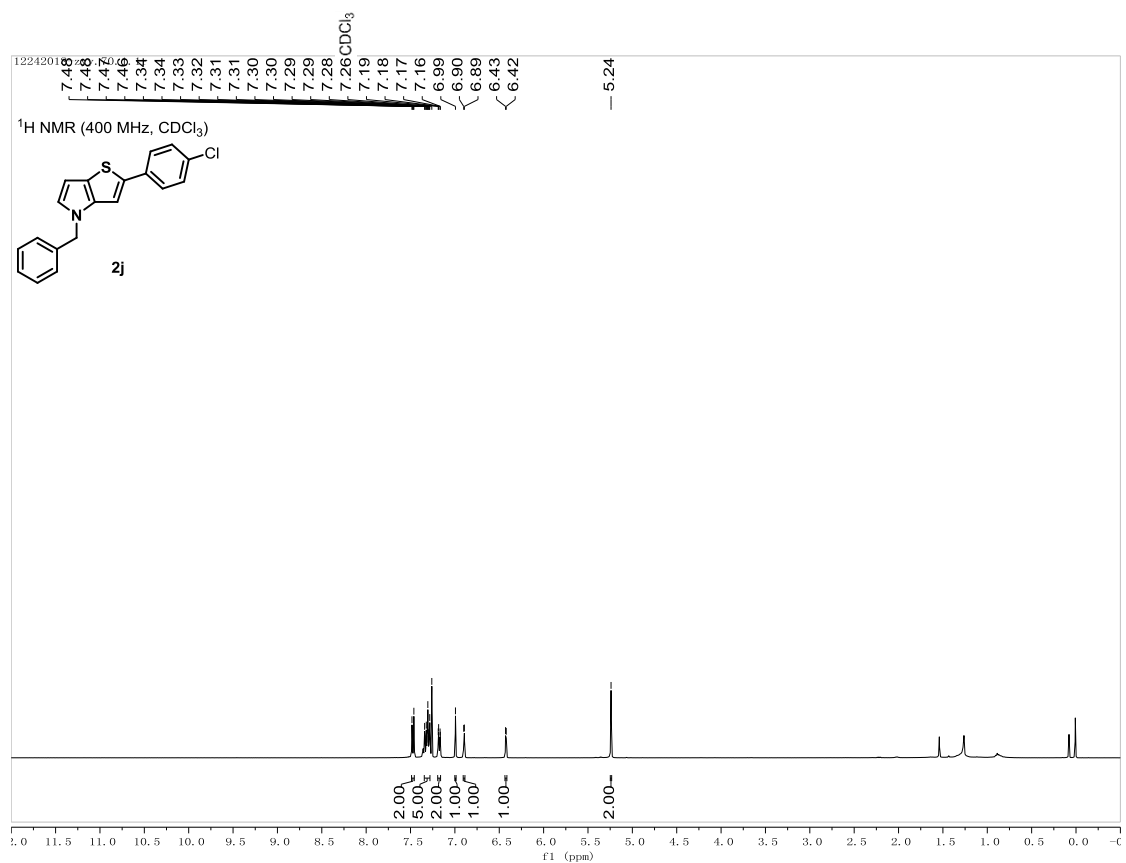


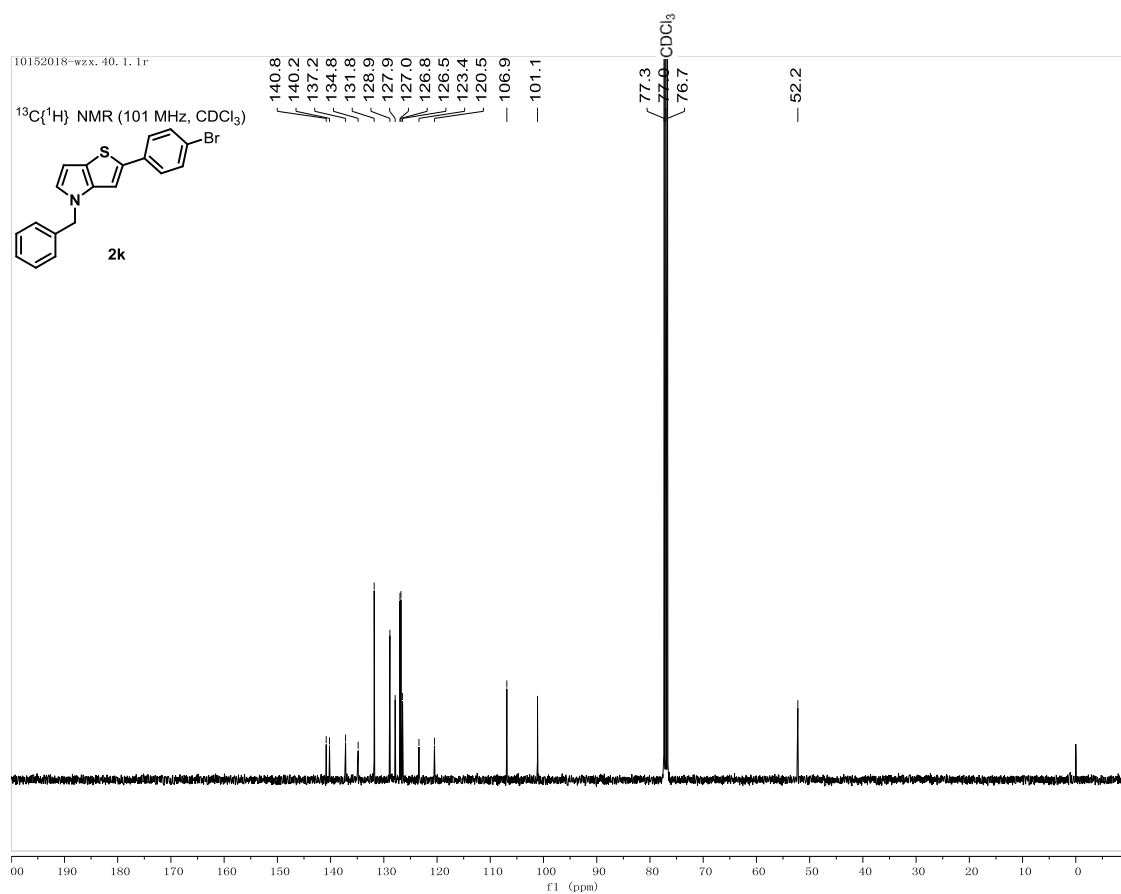
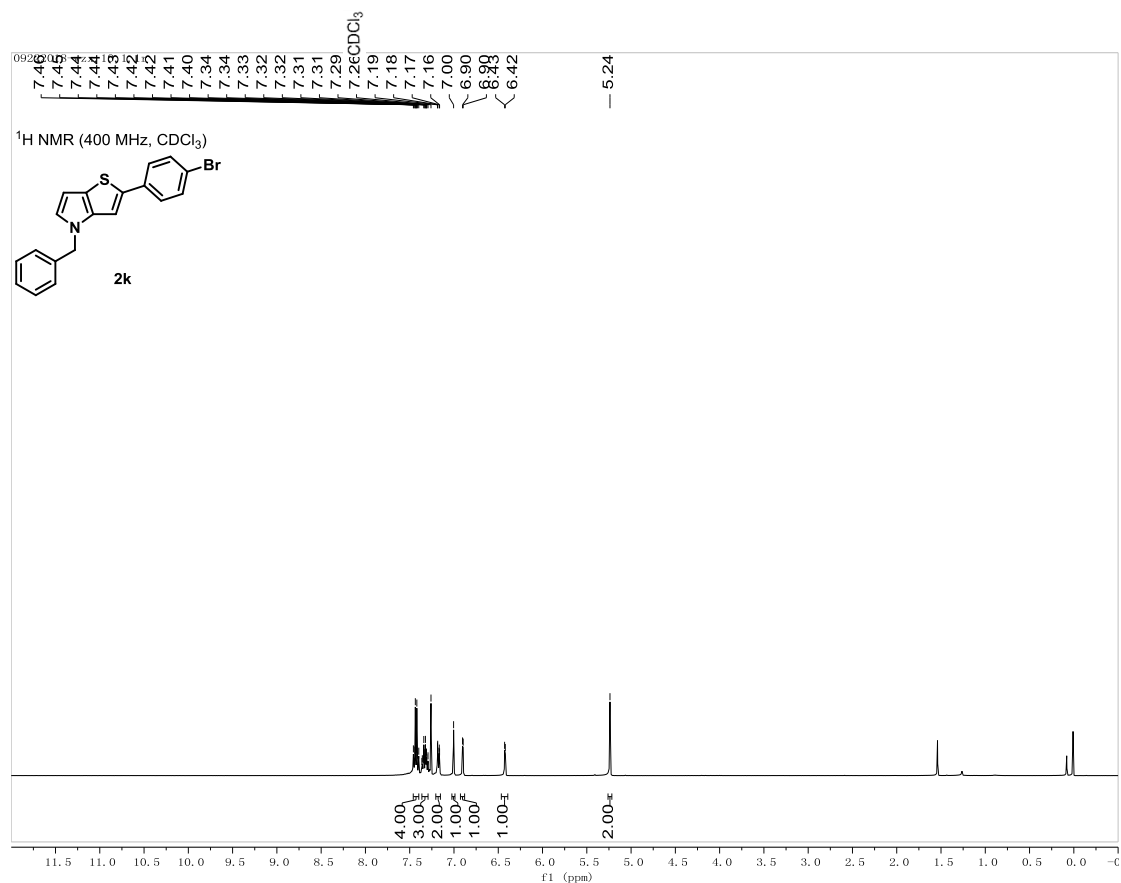


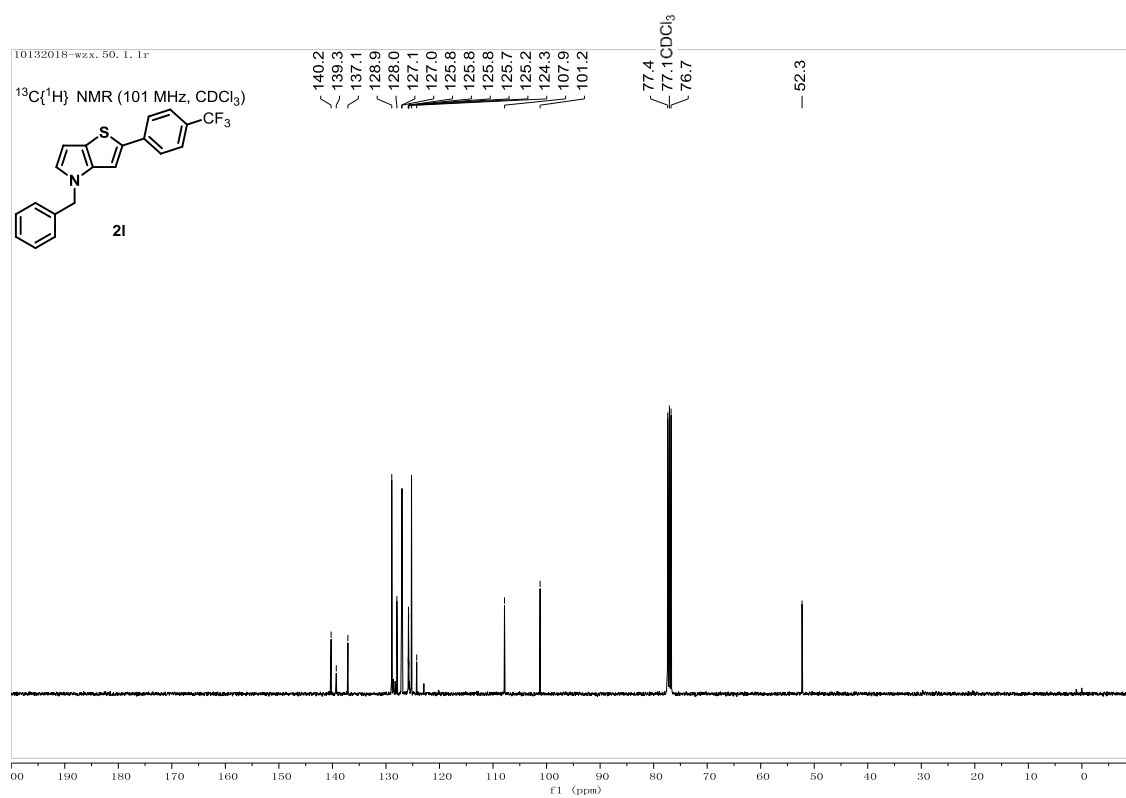
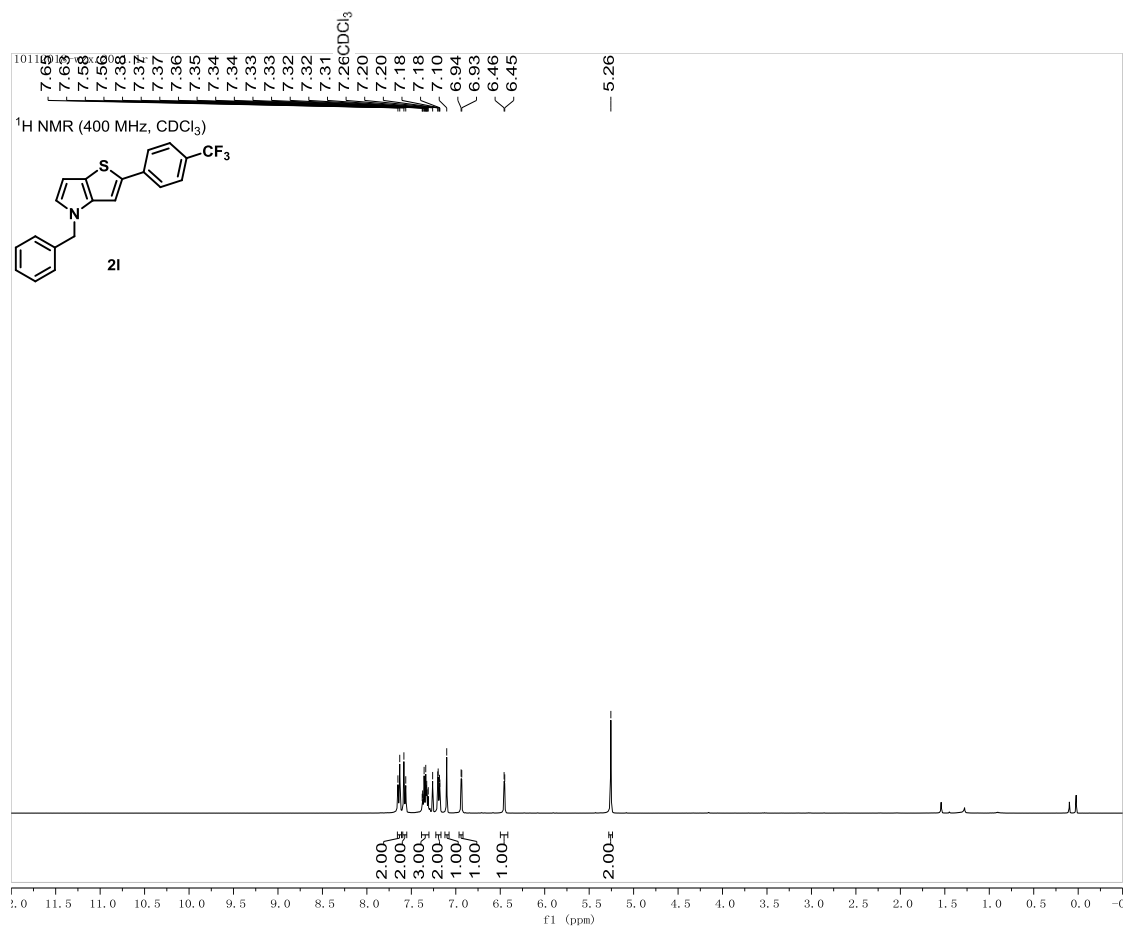
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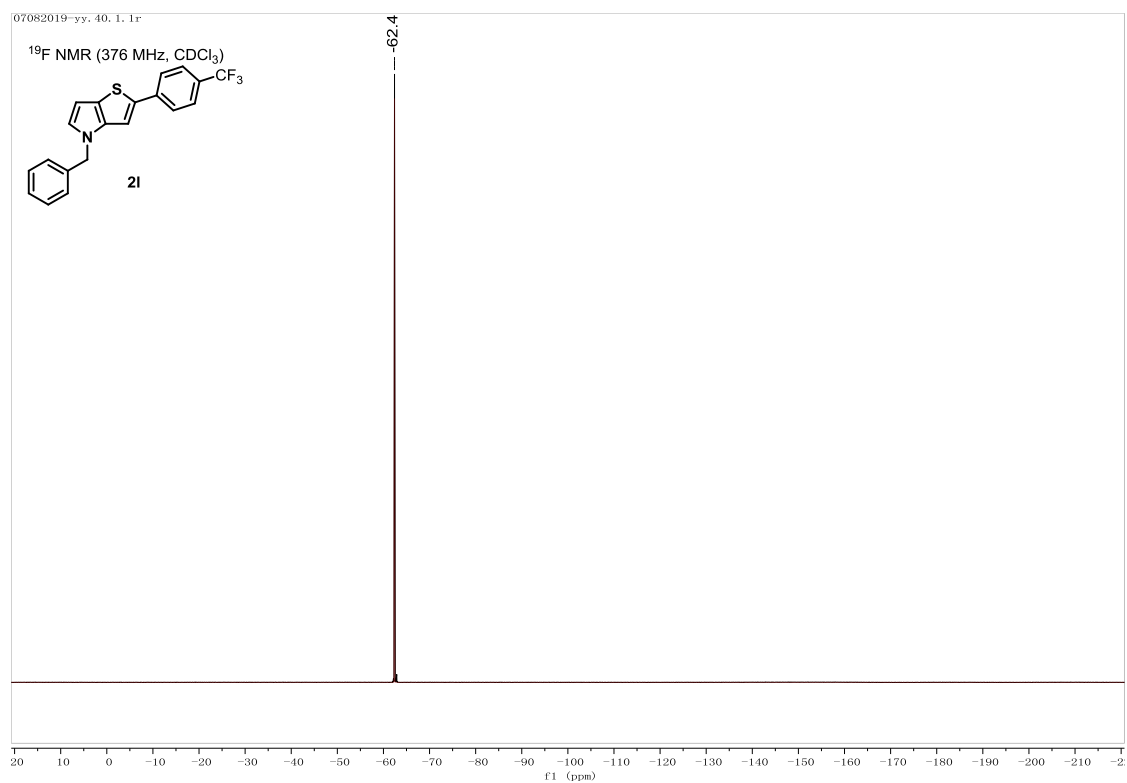
^{19}F NMR (376 MHz, CDCl_3)

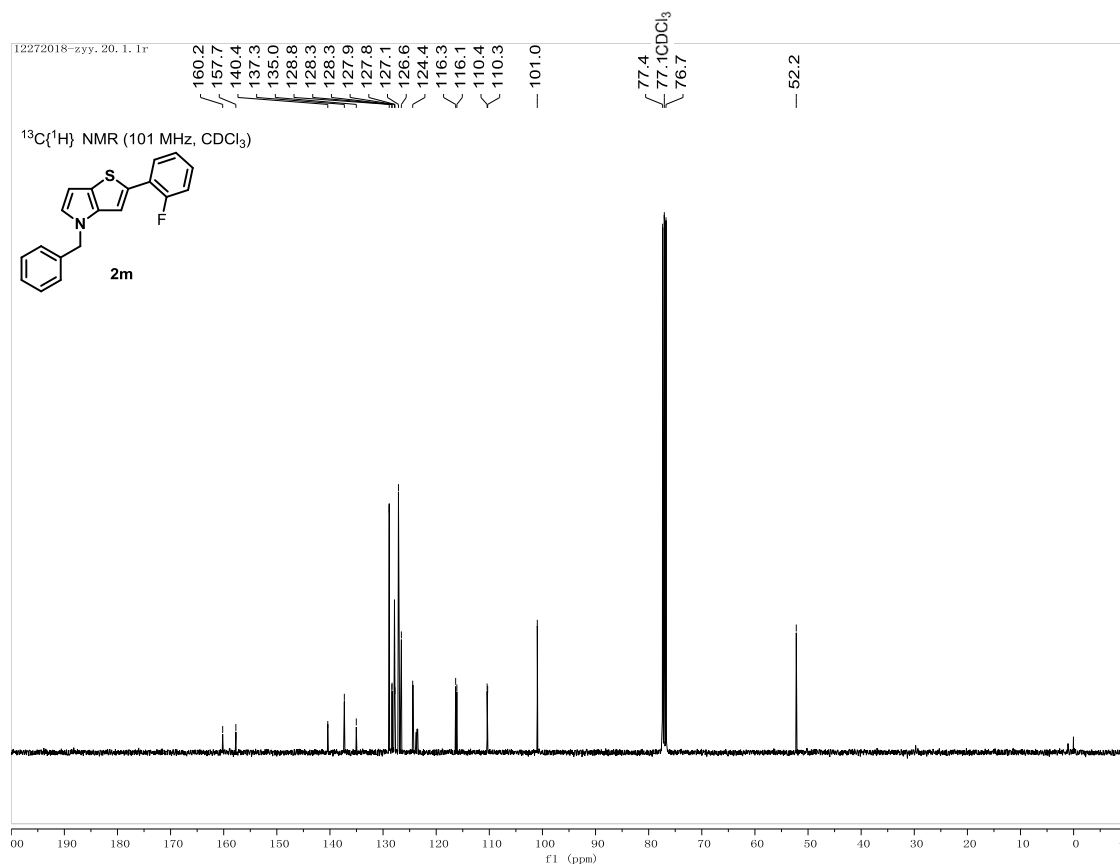
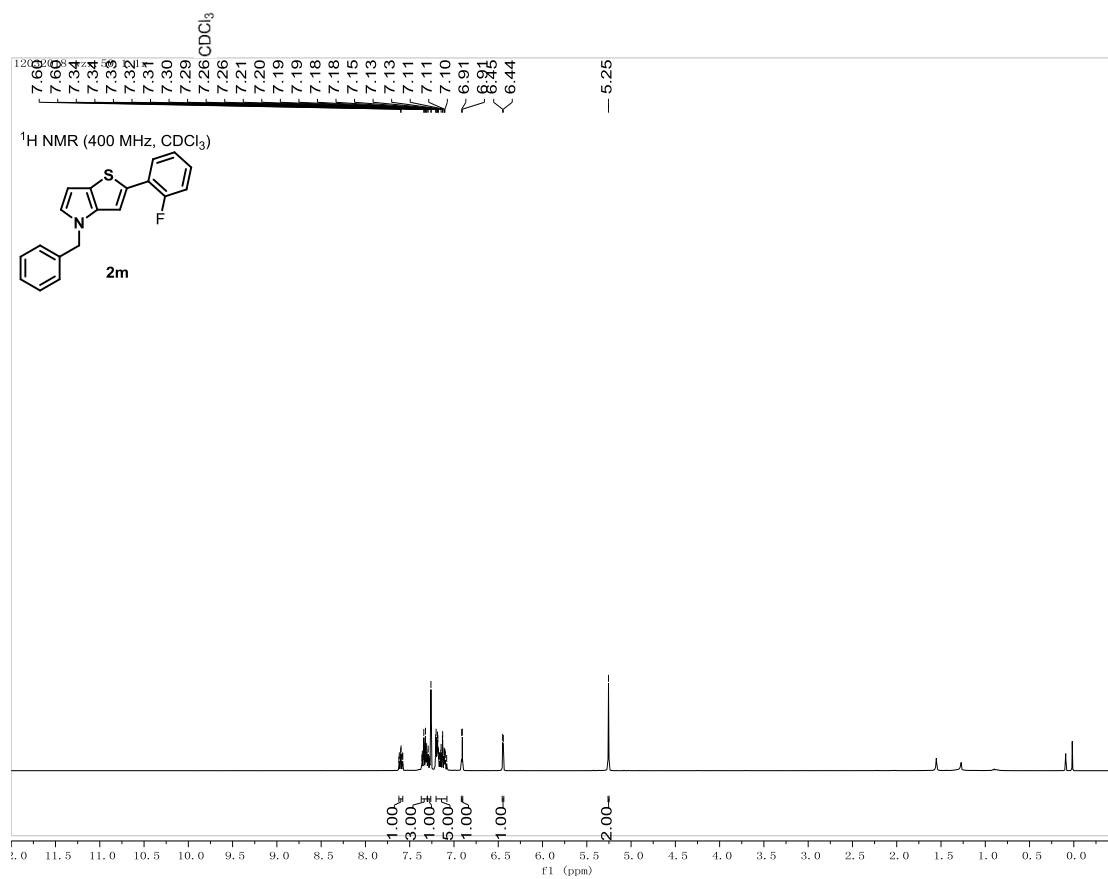






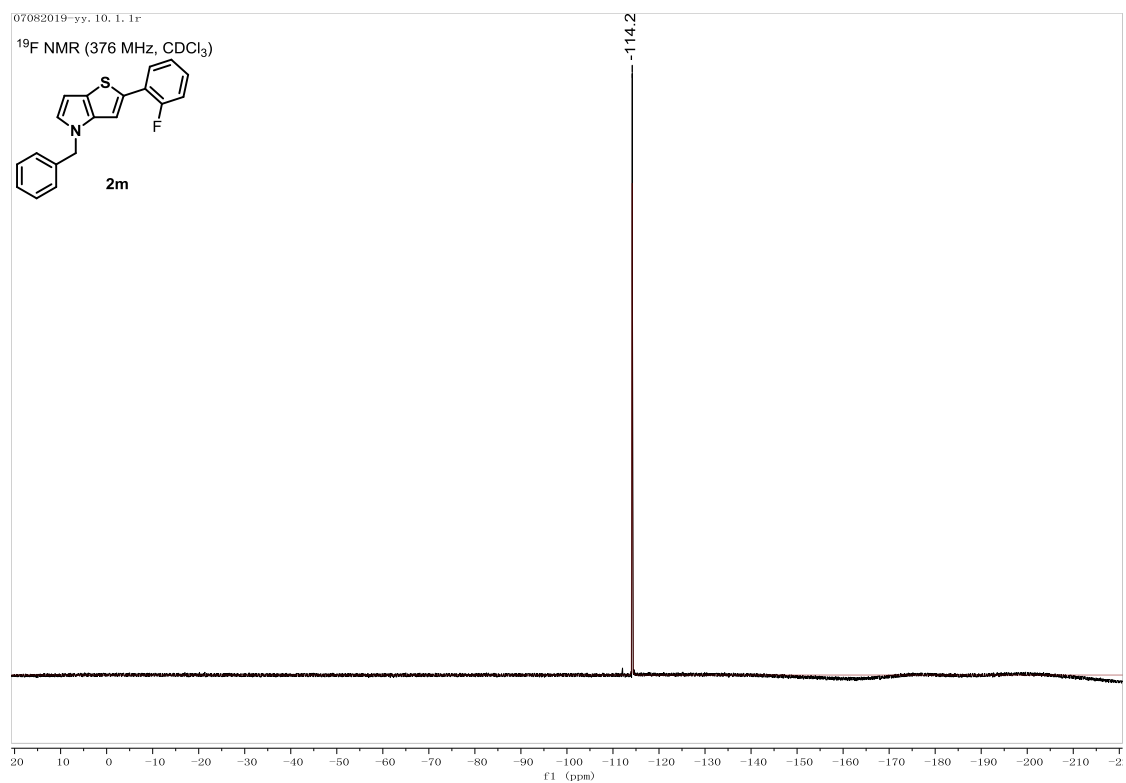
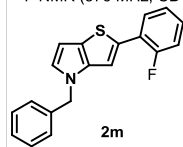


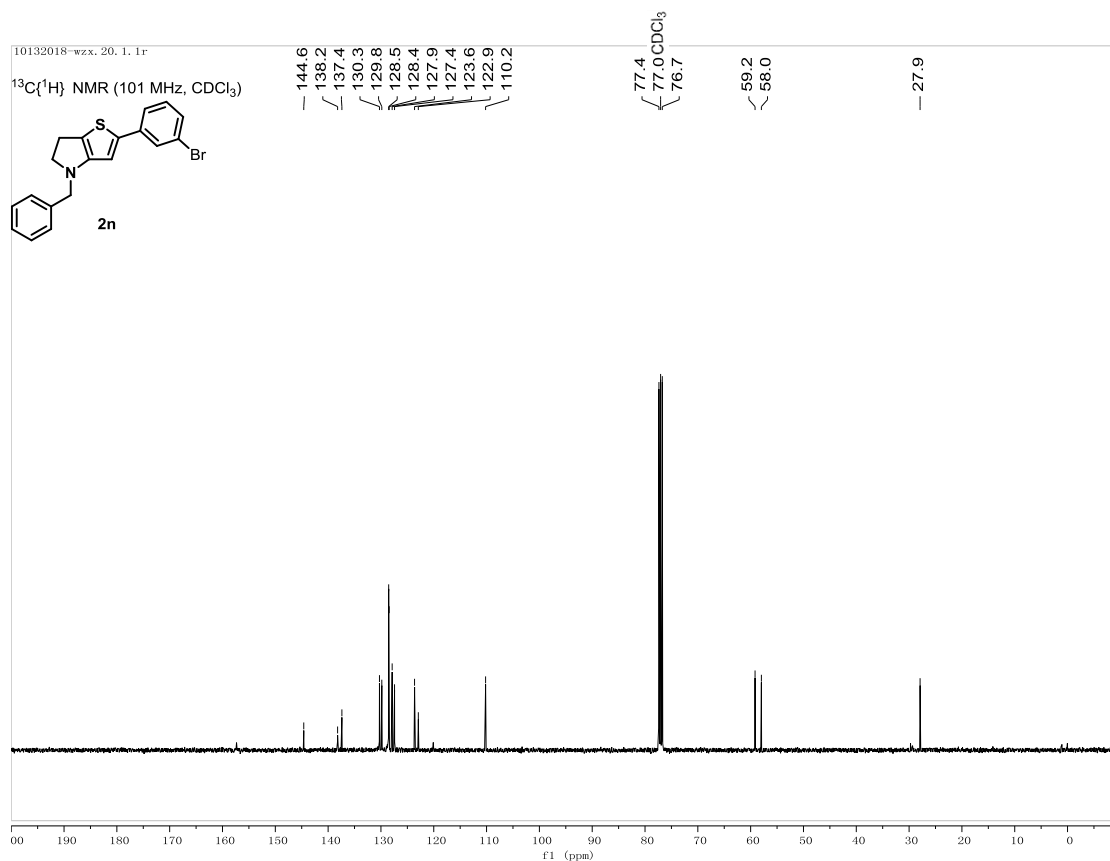
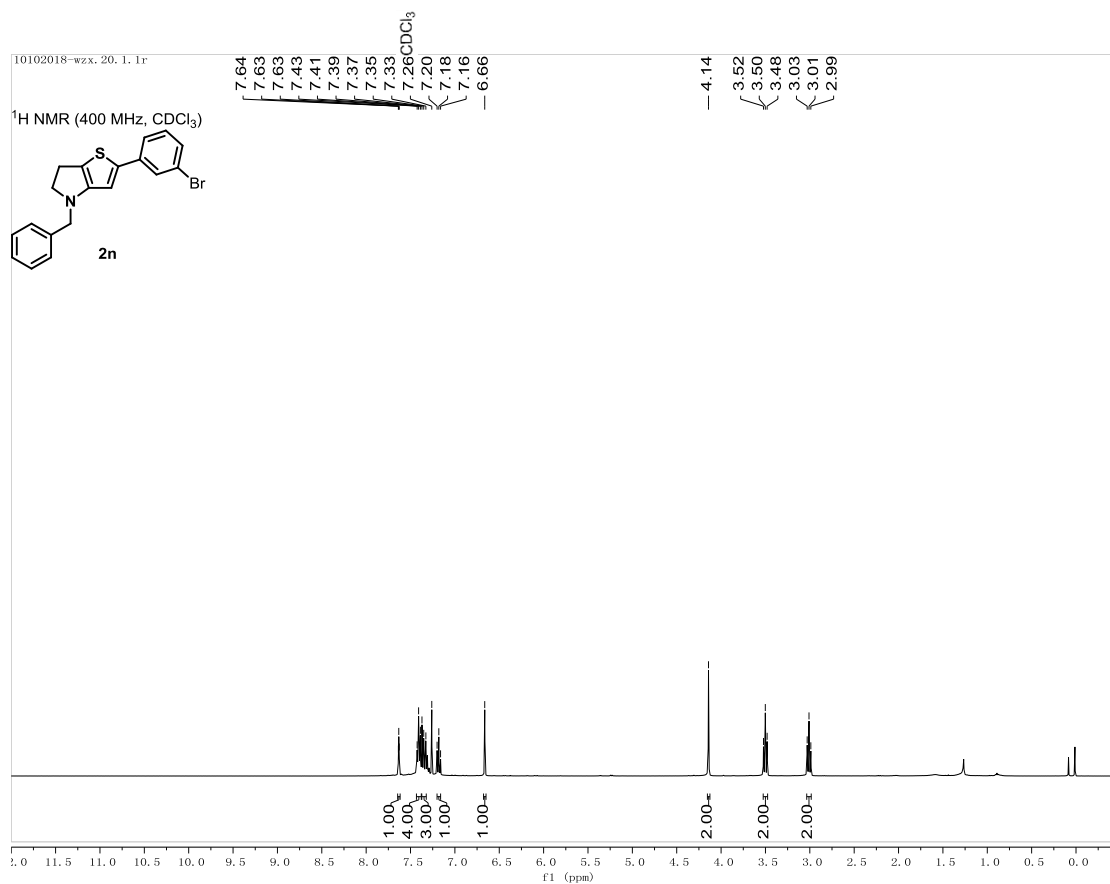


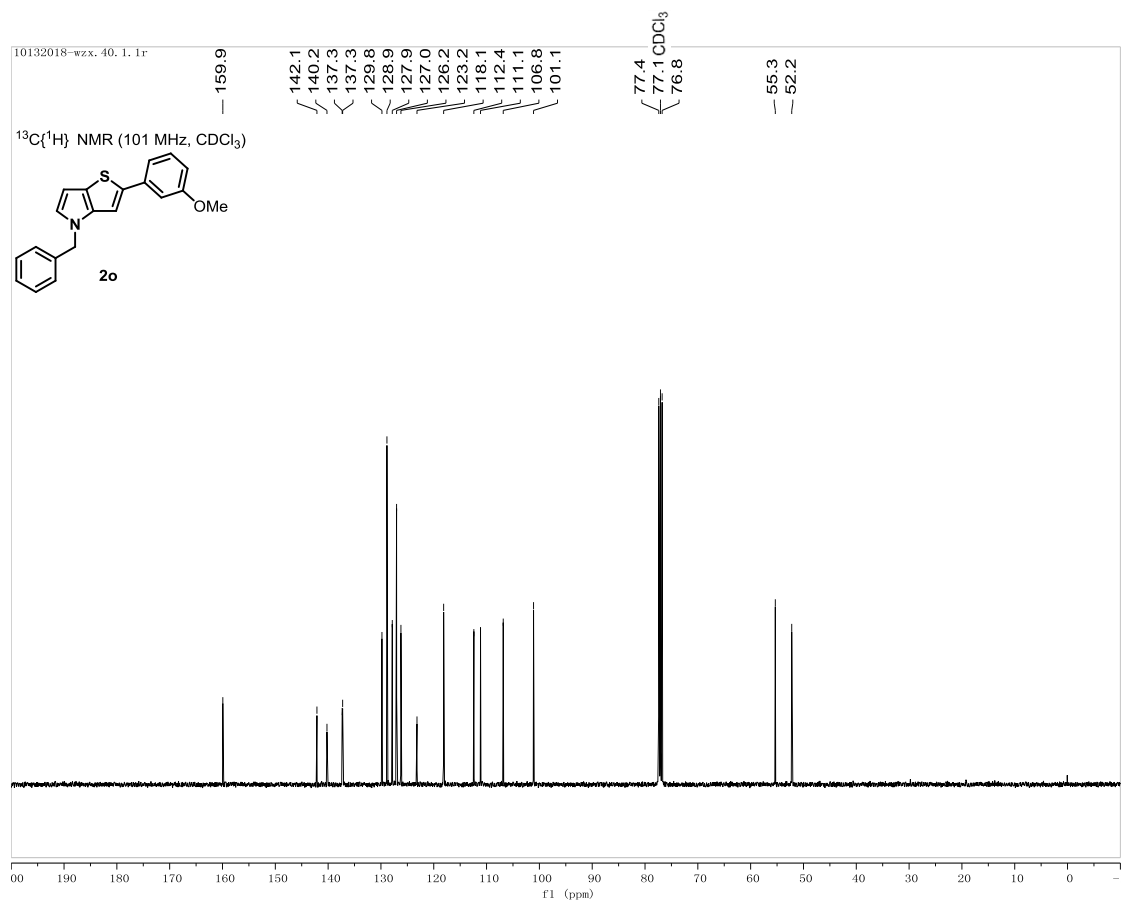
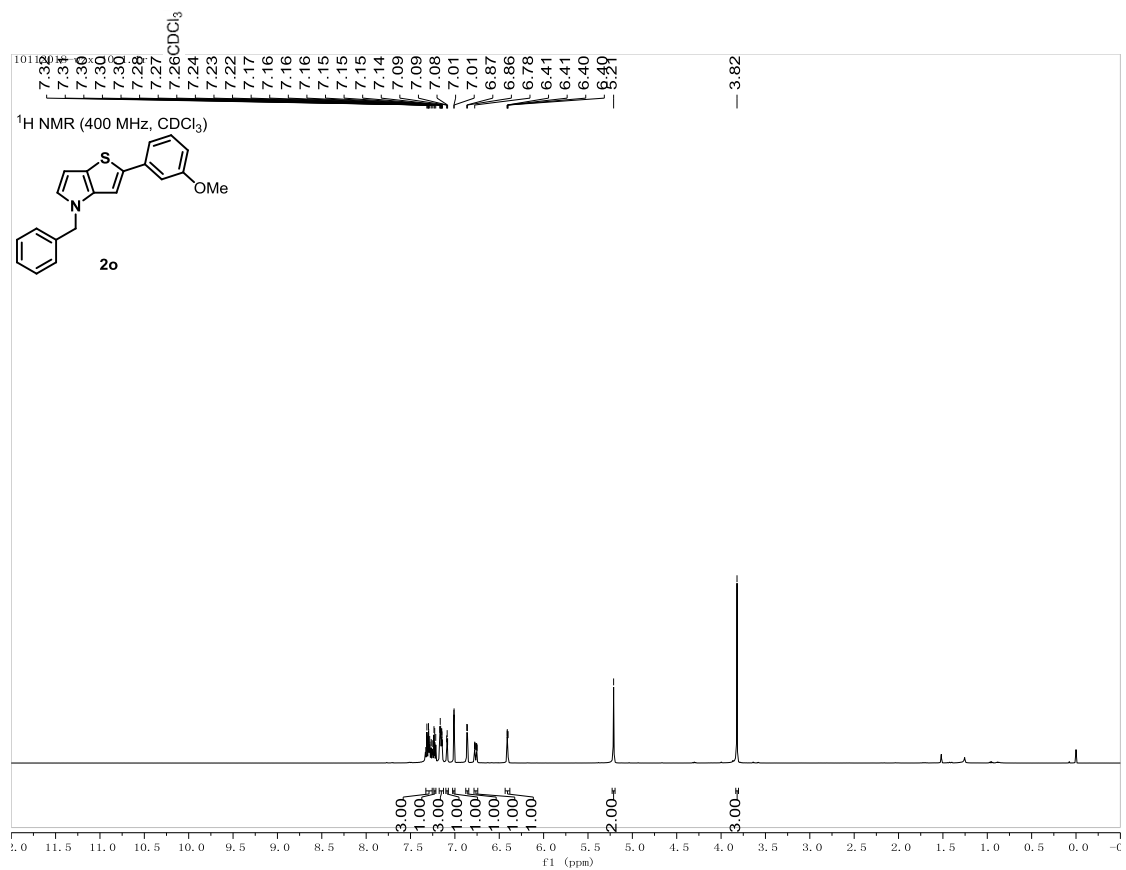


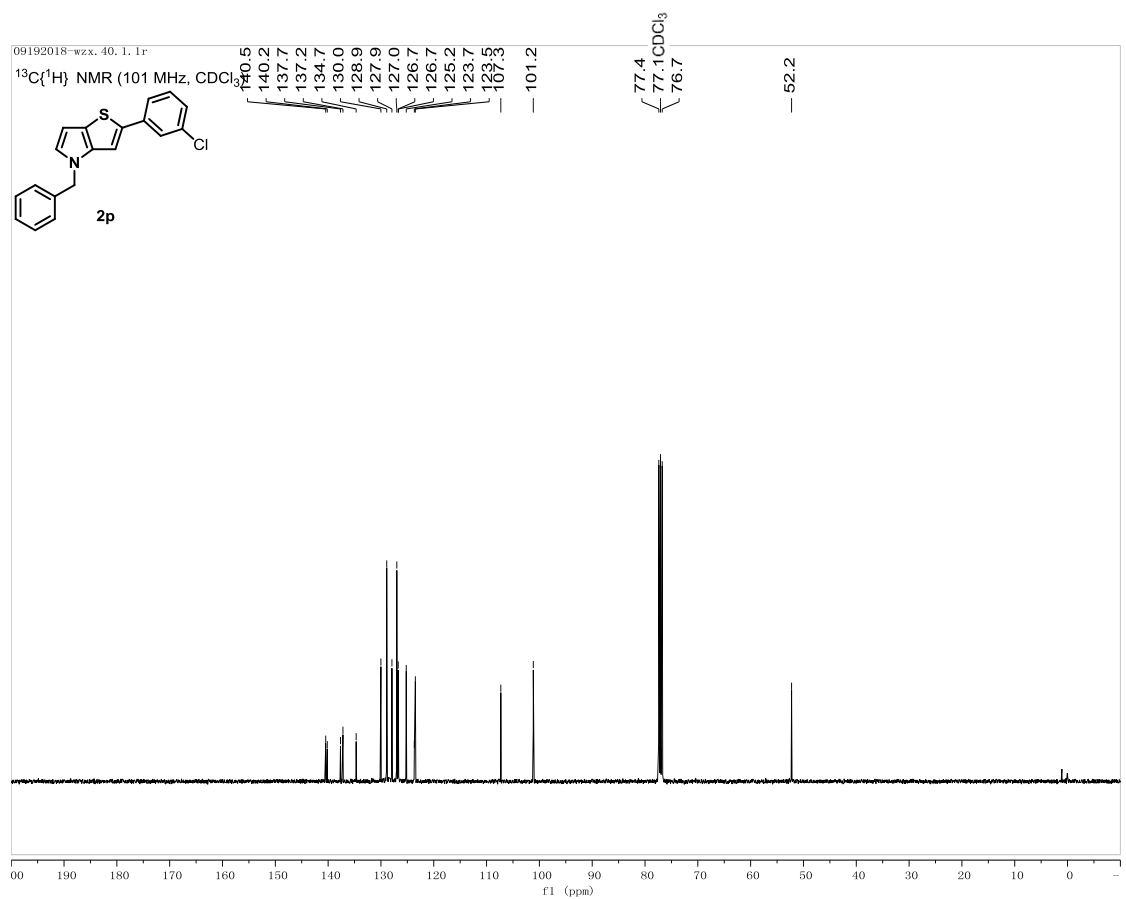
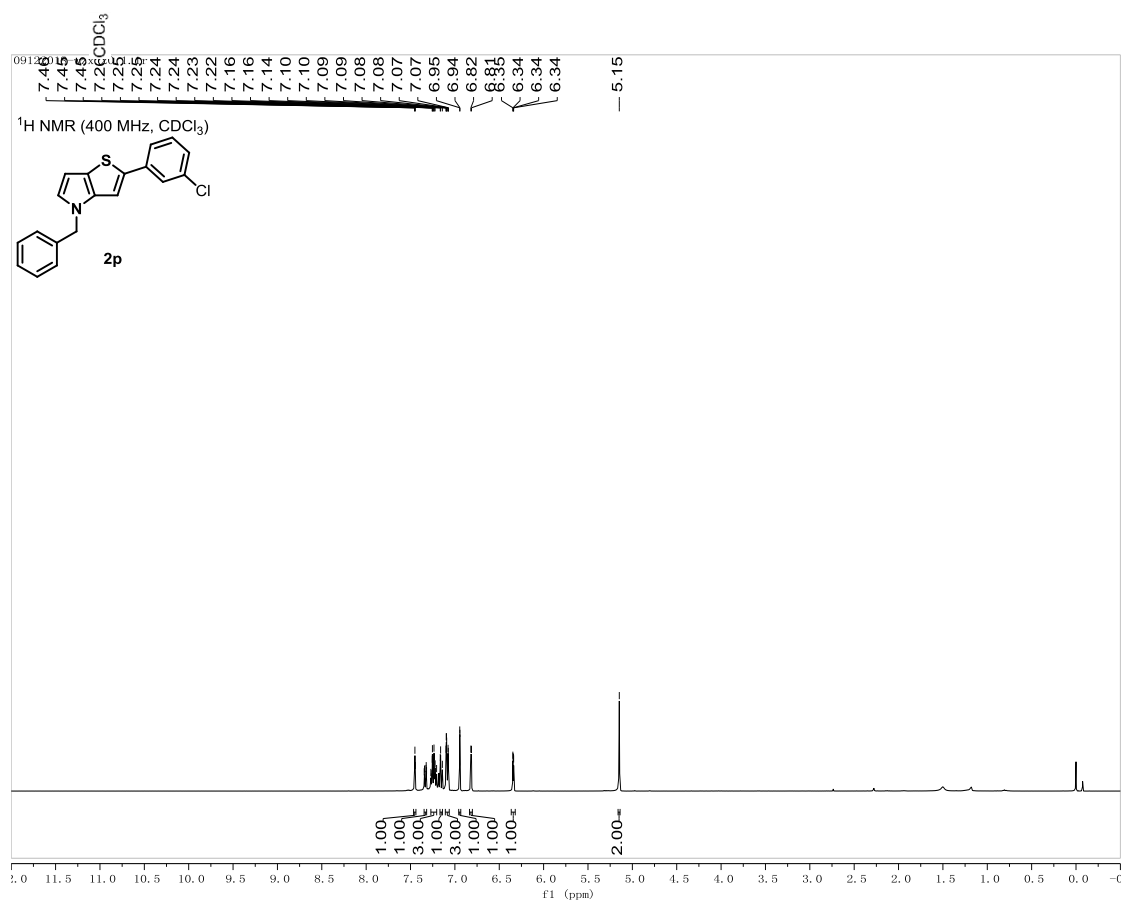
07082019-yy, 10, 1, 1r

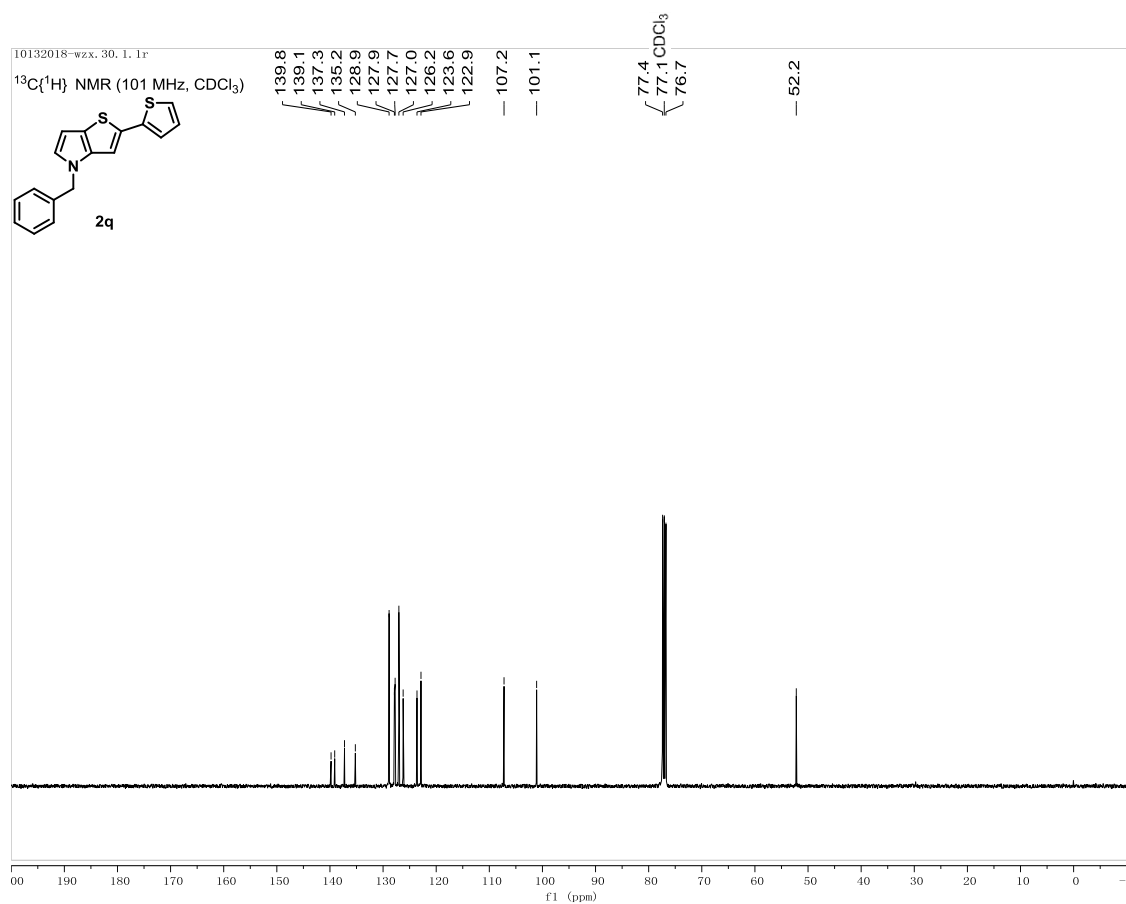
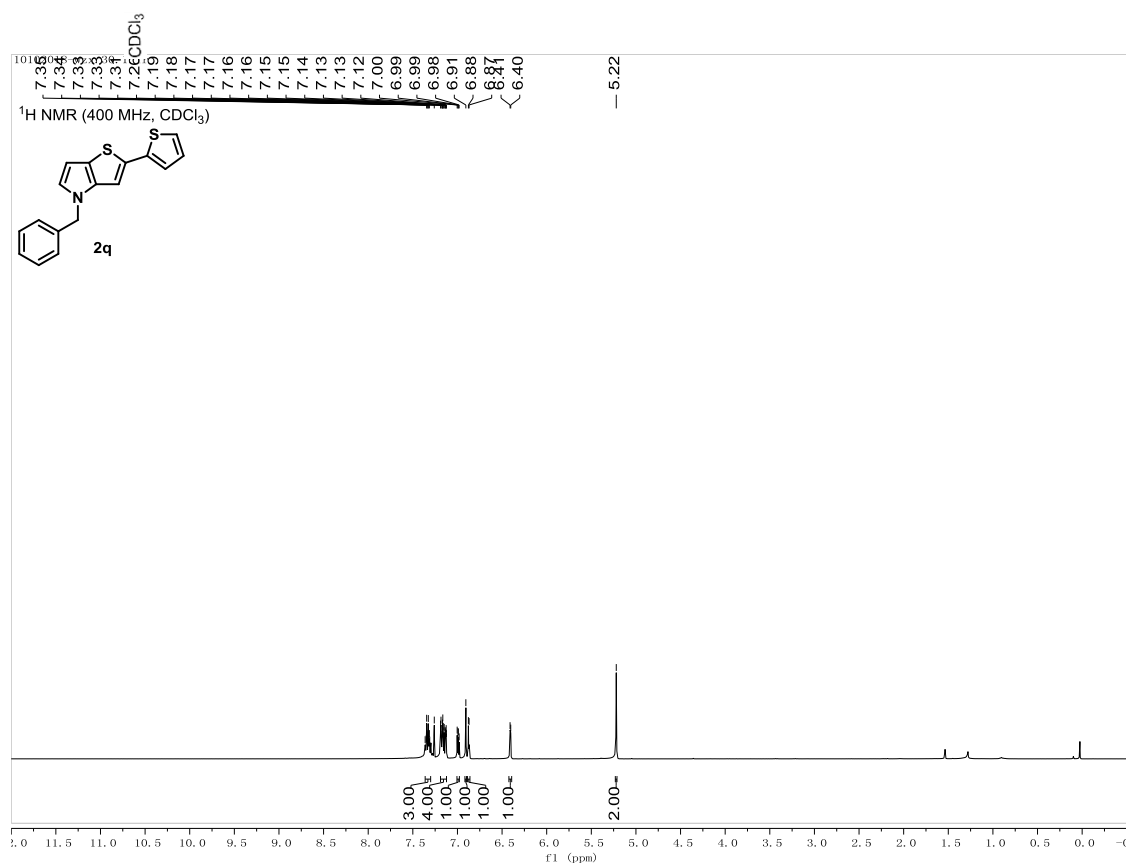
^{19}F NMR (376 MHz, CDCl_3)

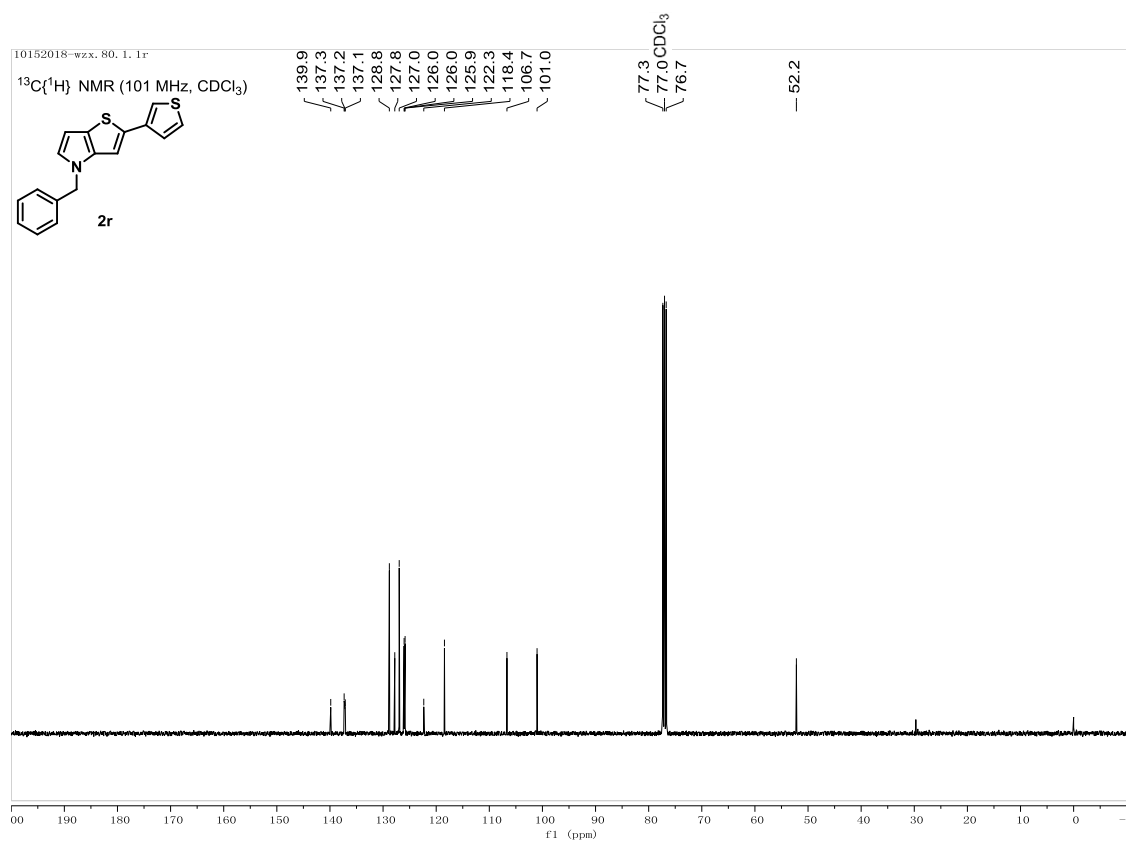
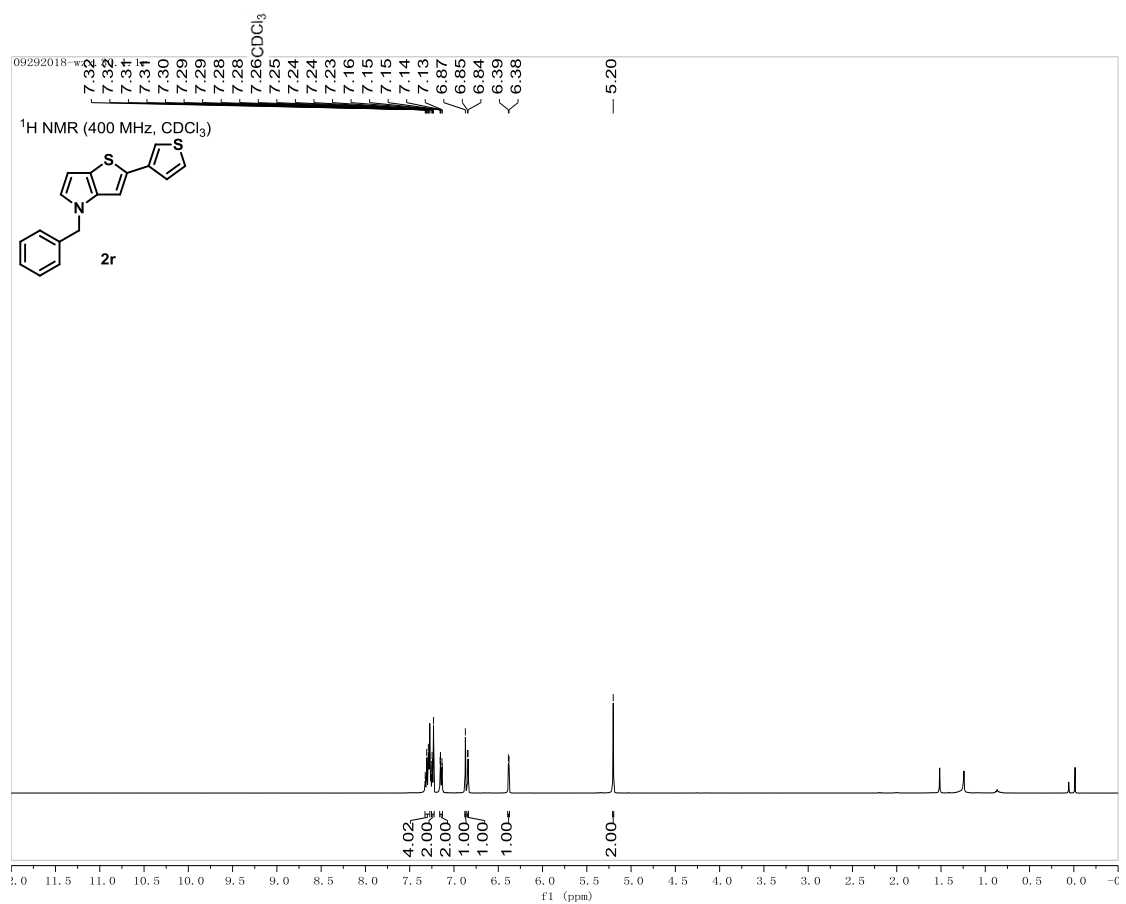


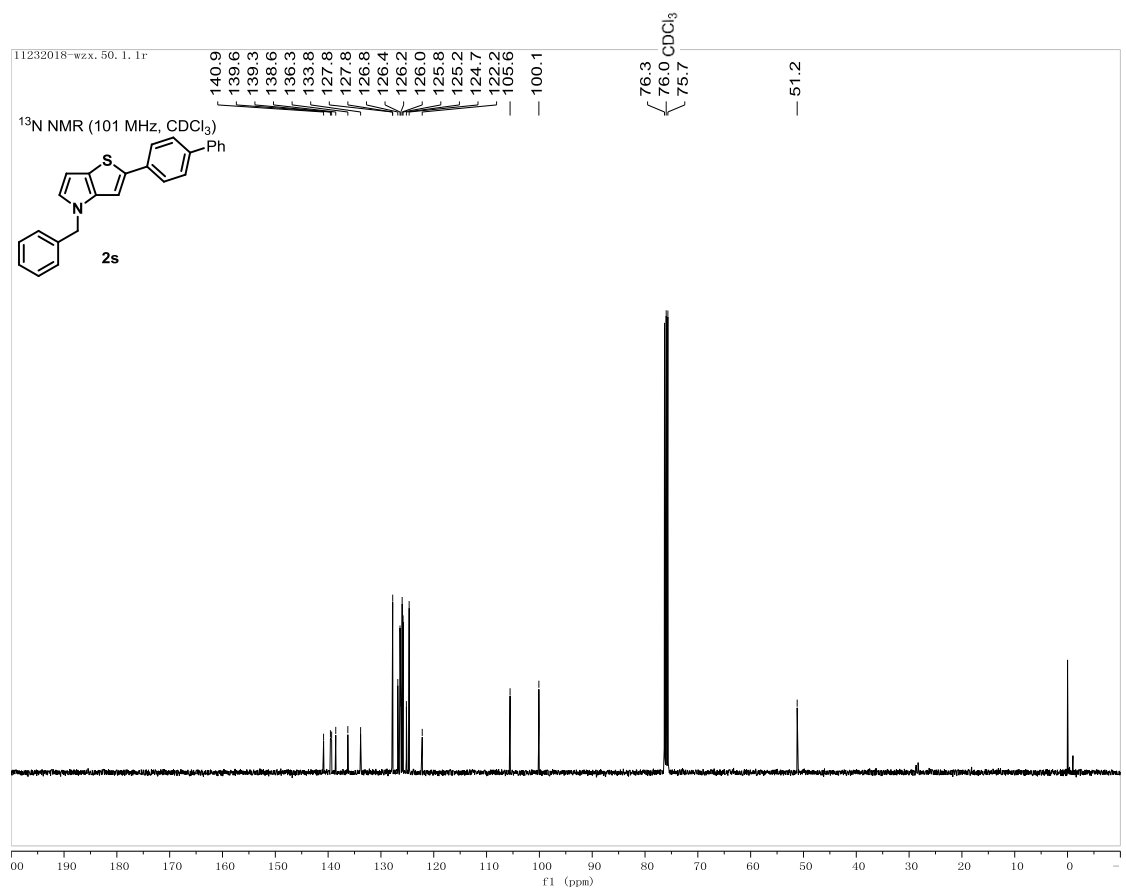
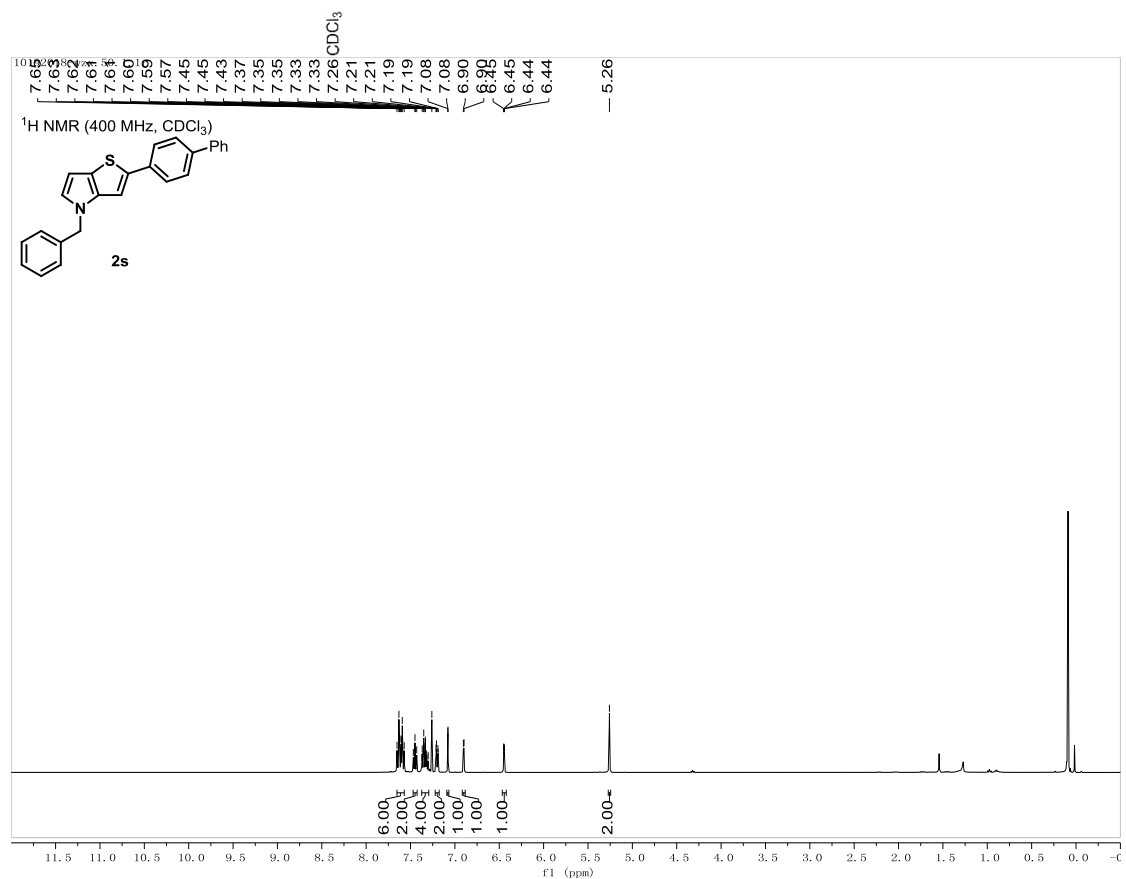






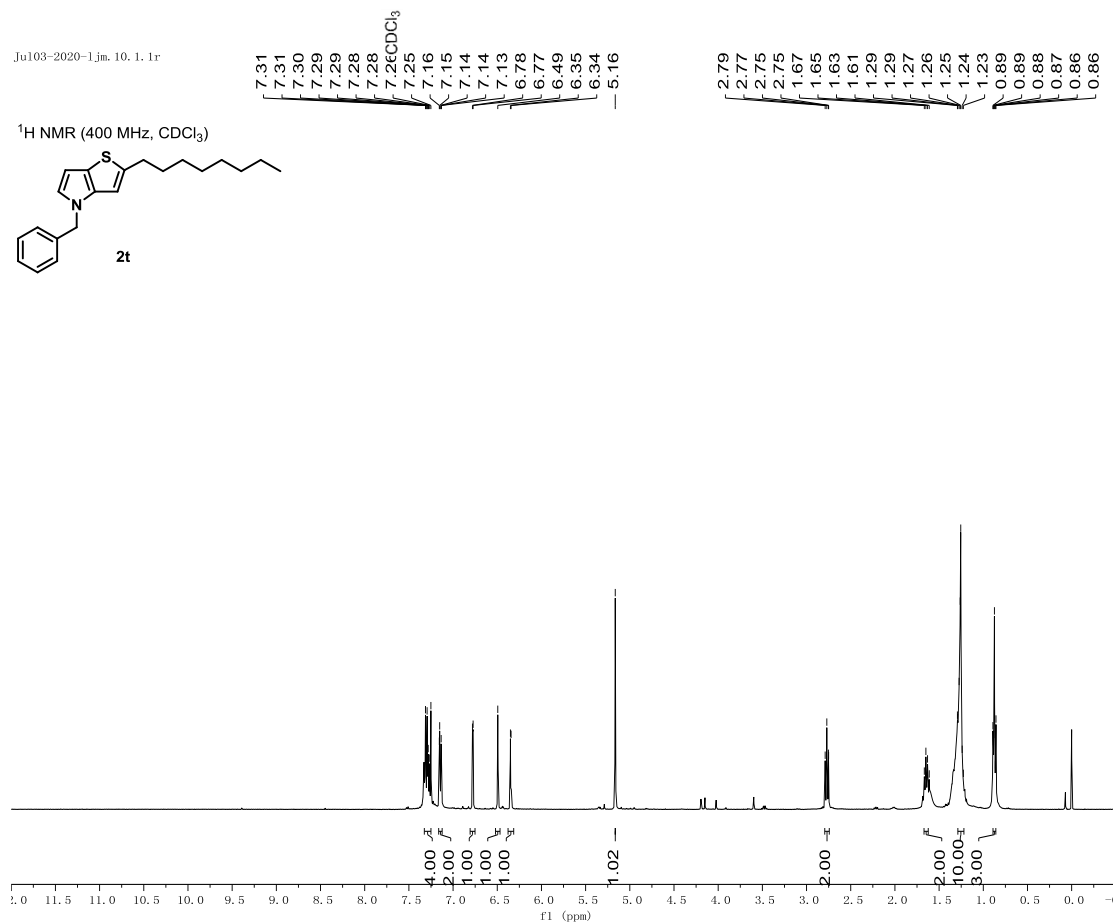
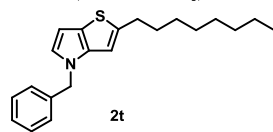






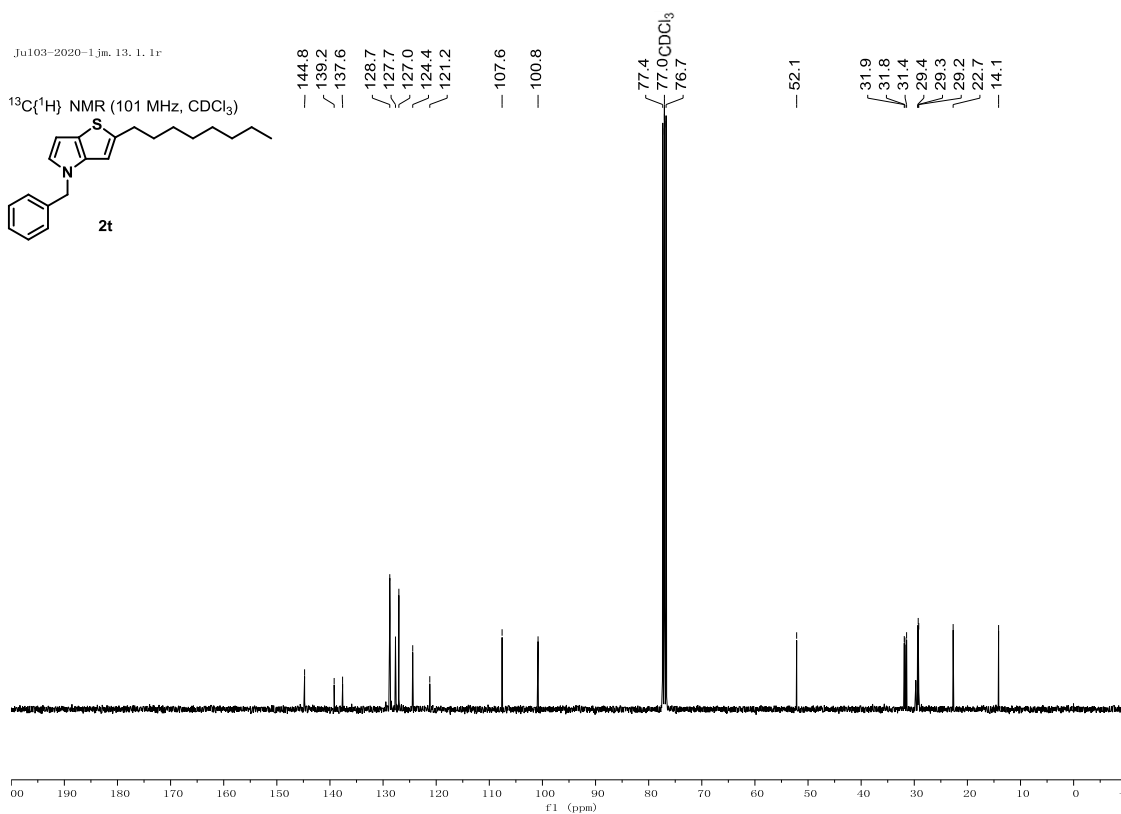
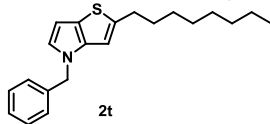
Ju103-2020-1.jm. 10. 1. 1r

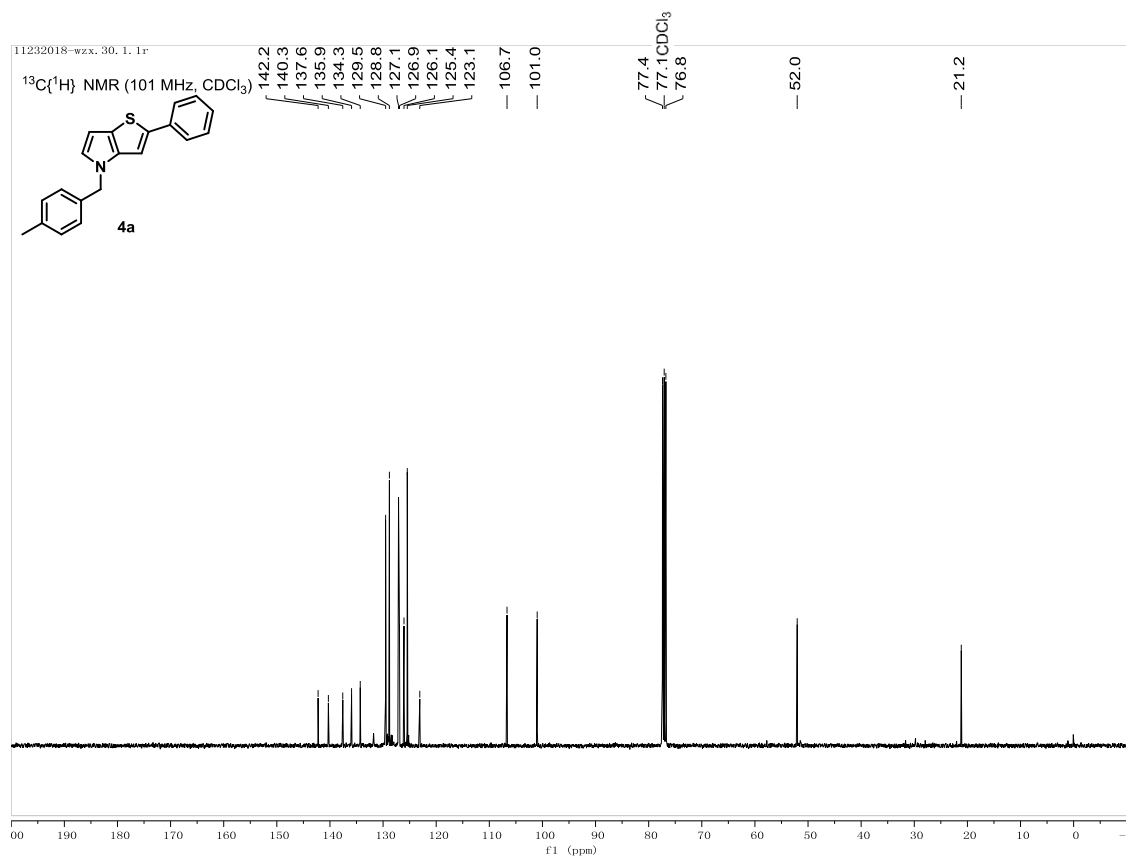
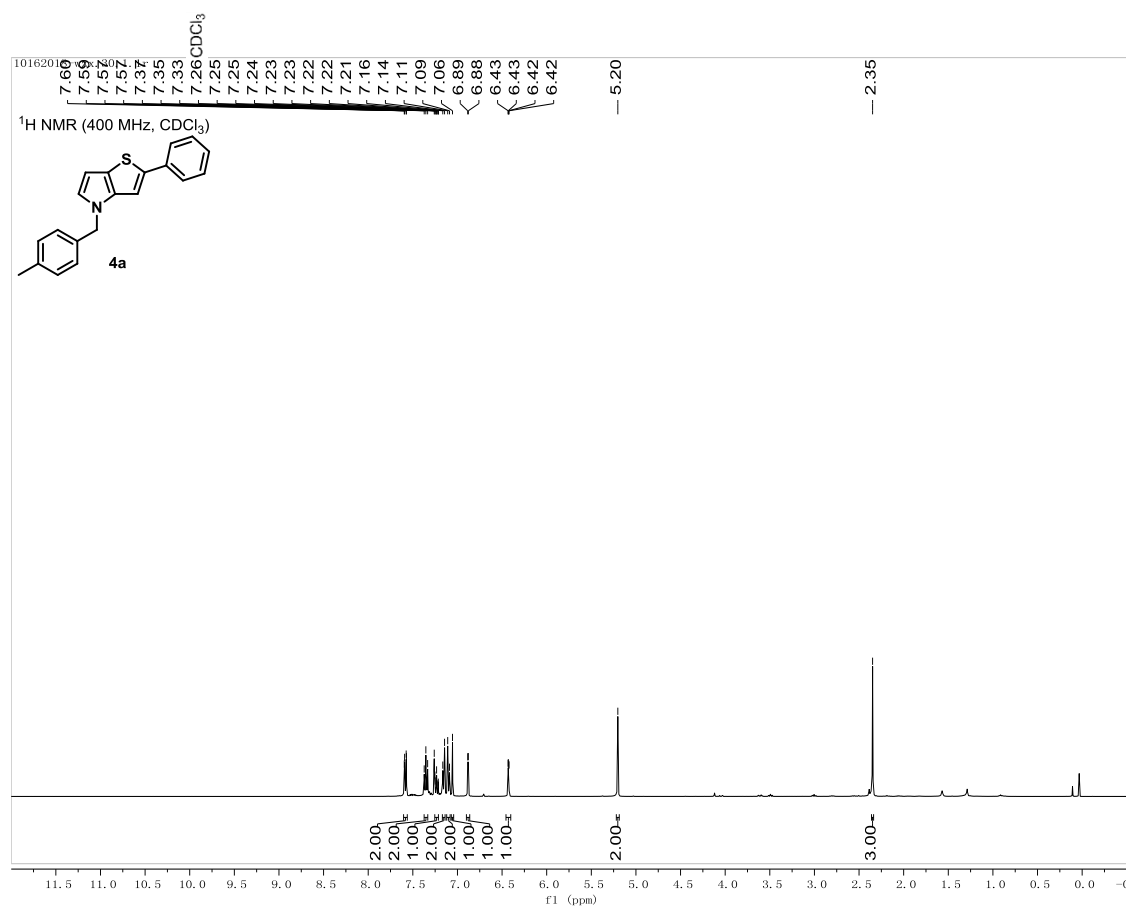
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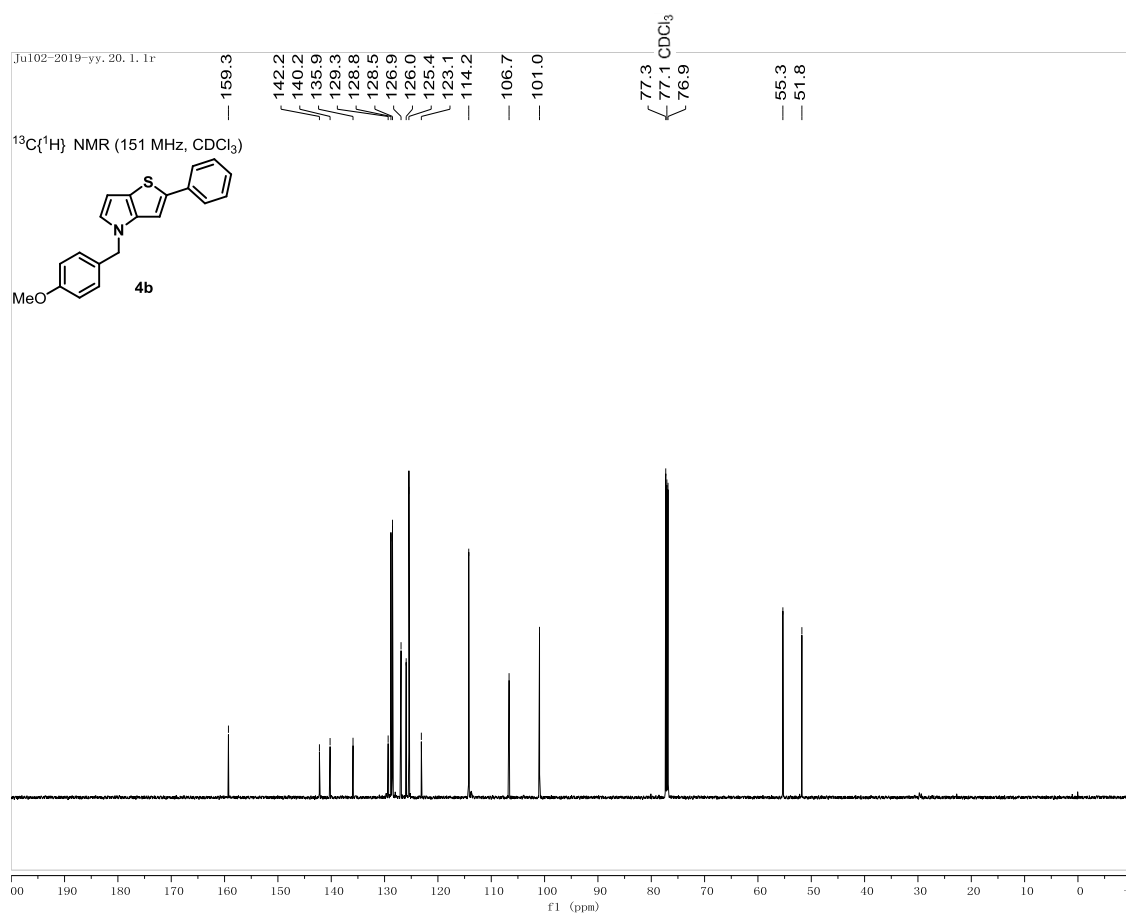
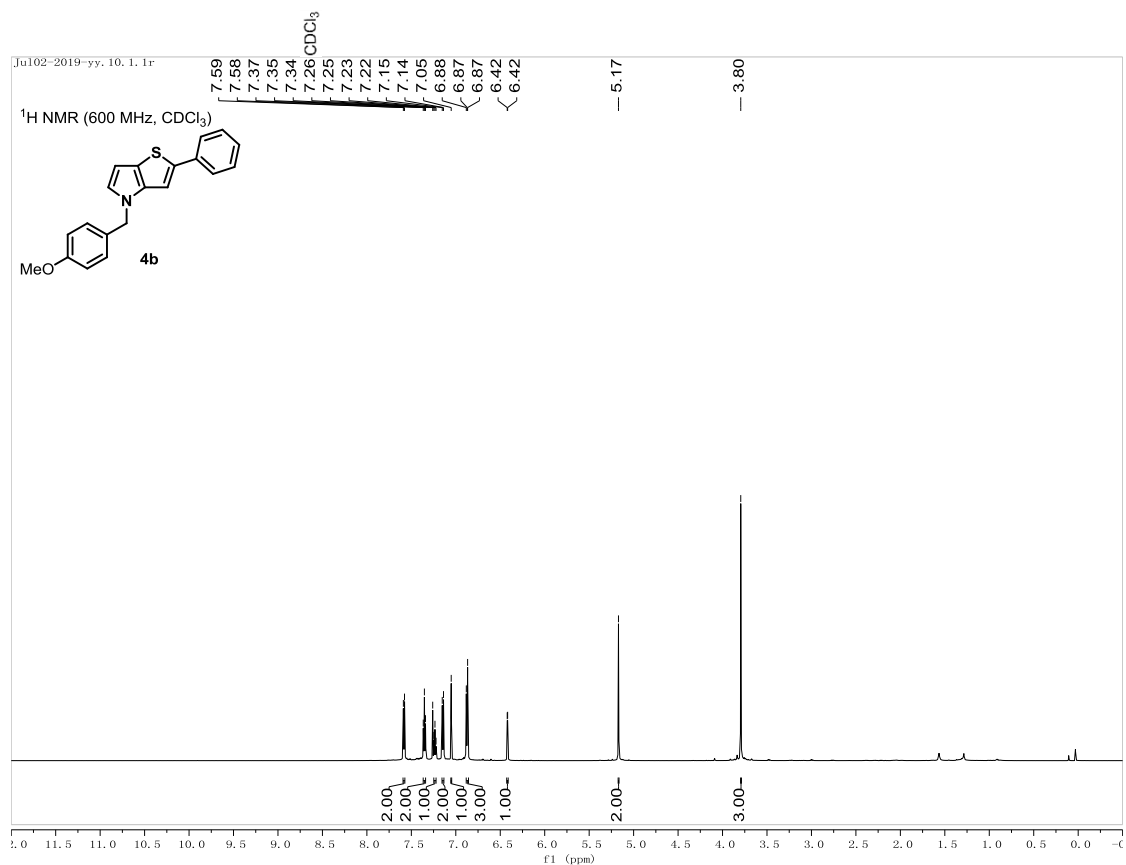


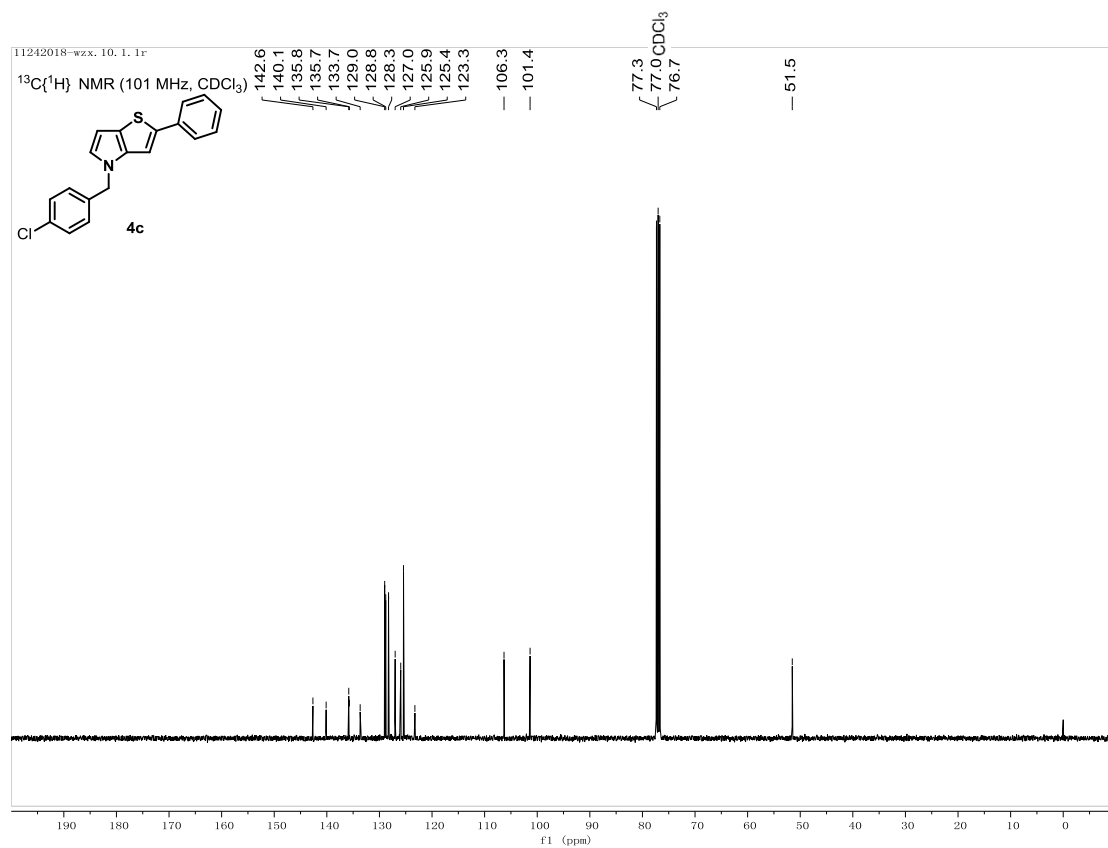
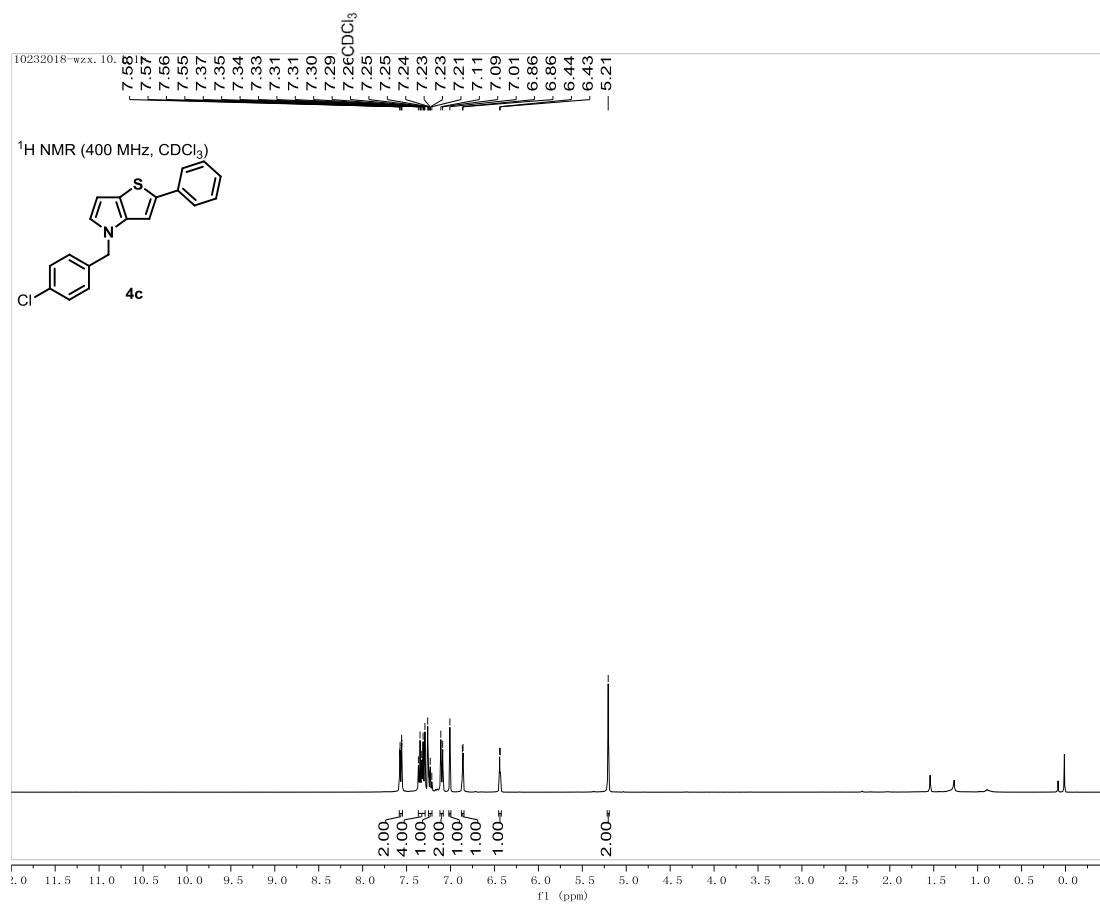
Ju103-2020-1.jm. 13. 1. 1r

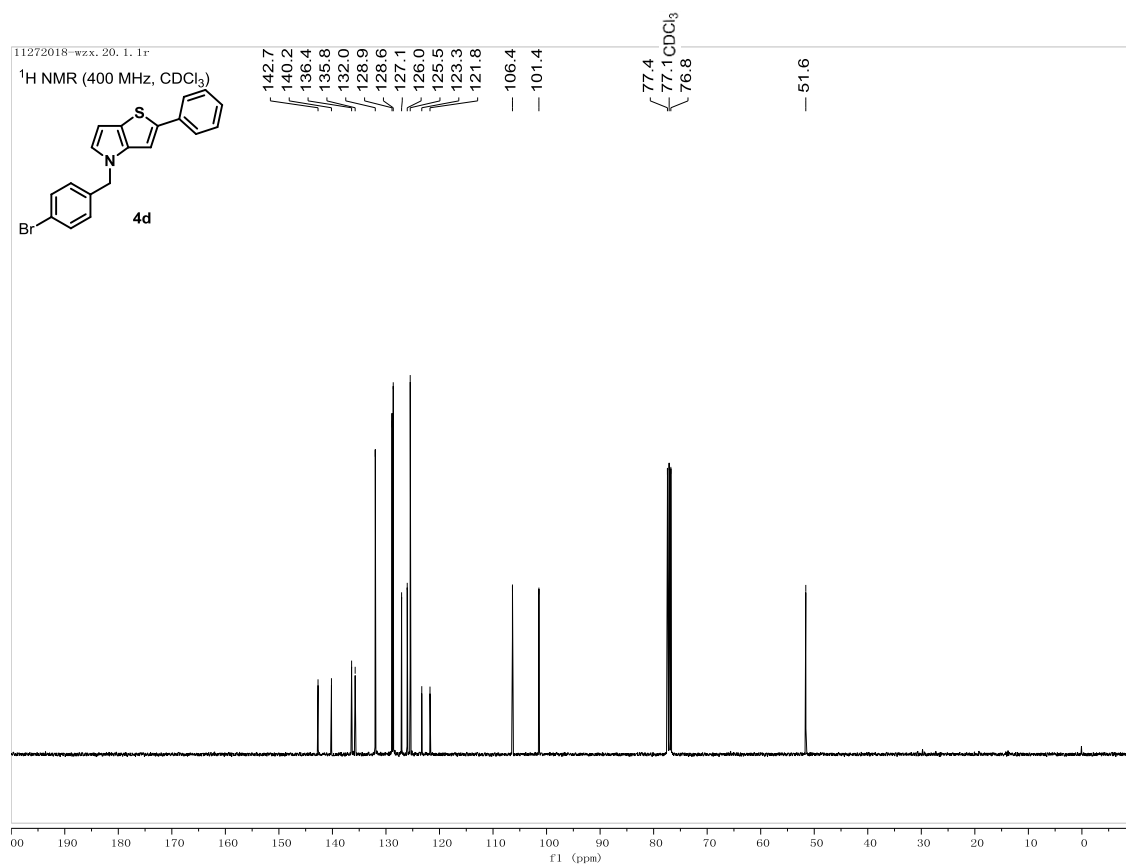
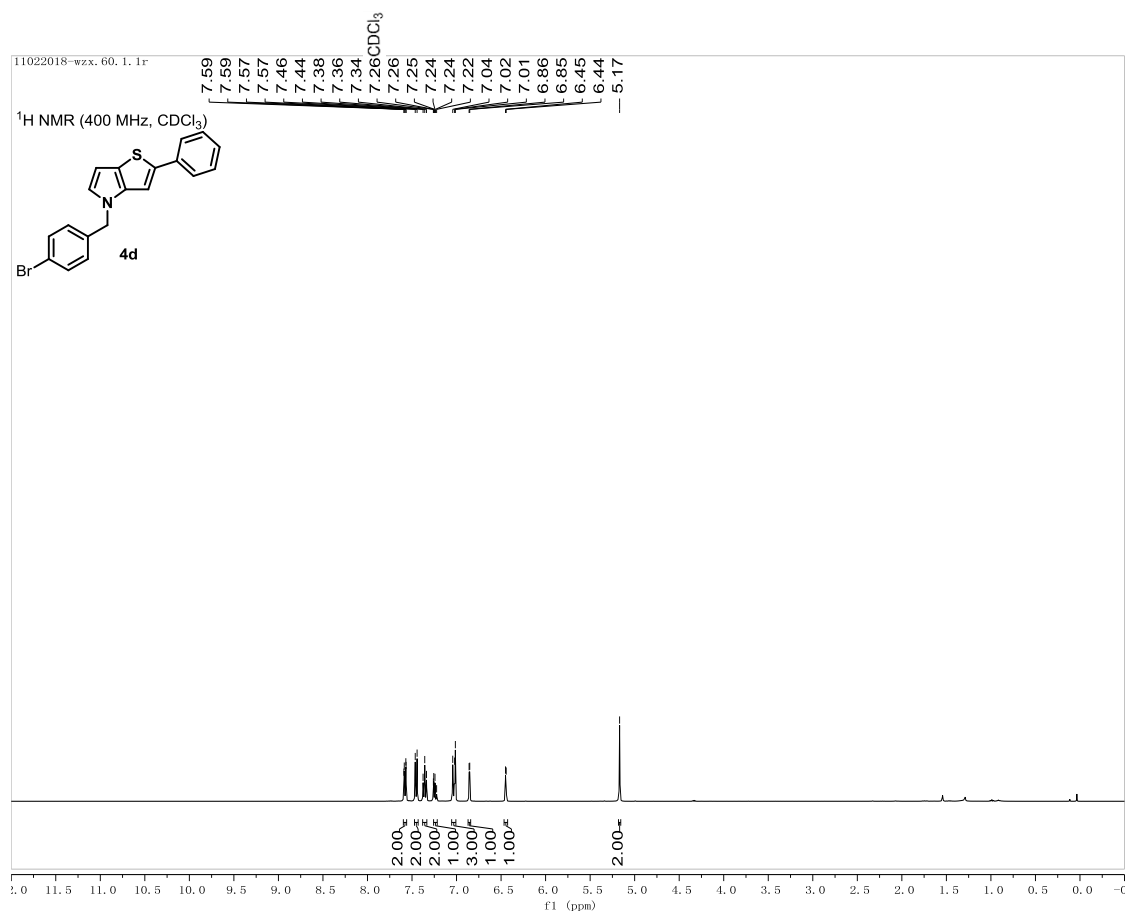
¹³C{¹H} NMR (101 MHz, CDCl₃)

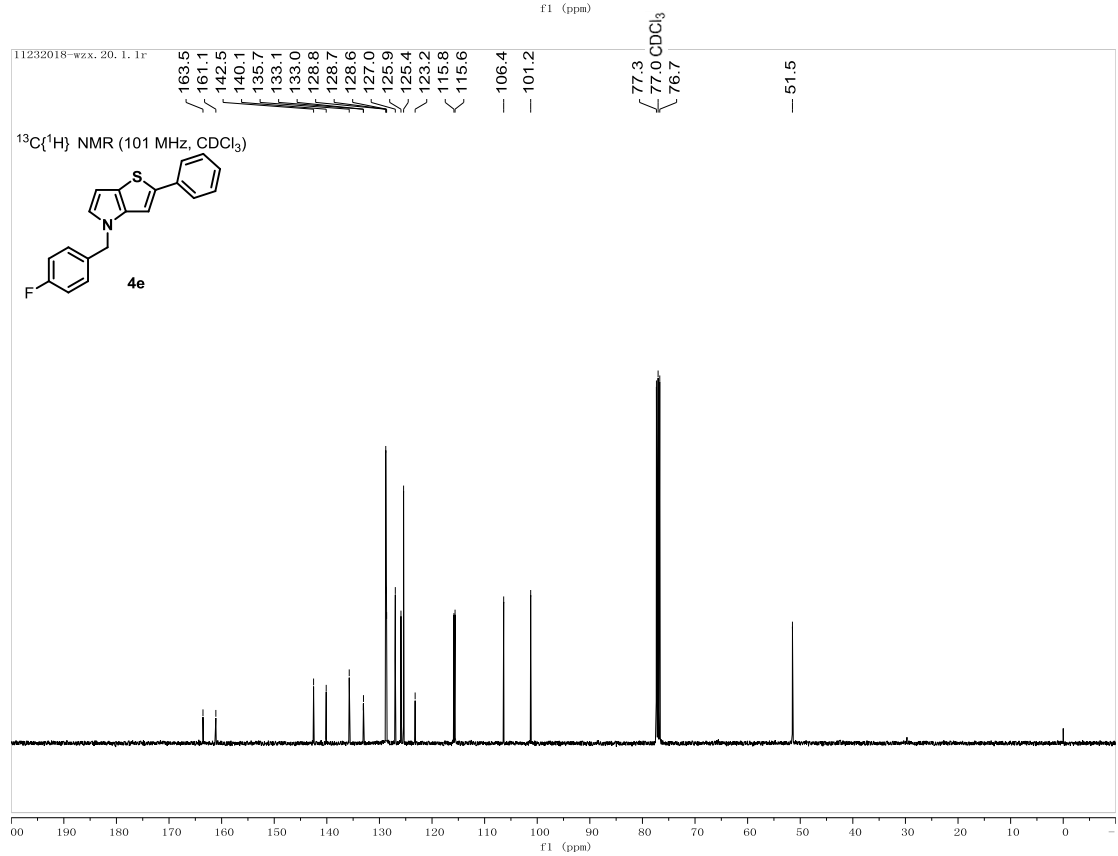
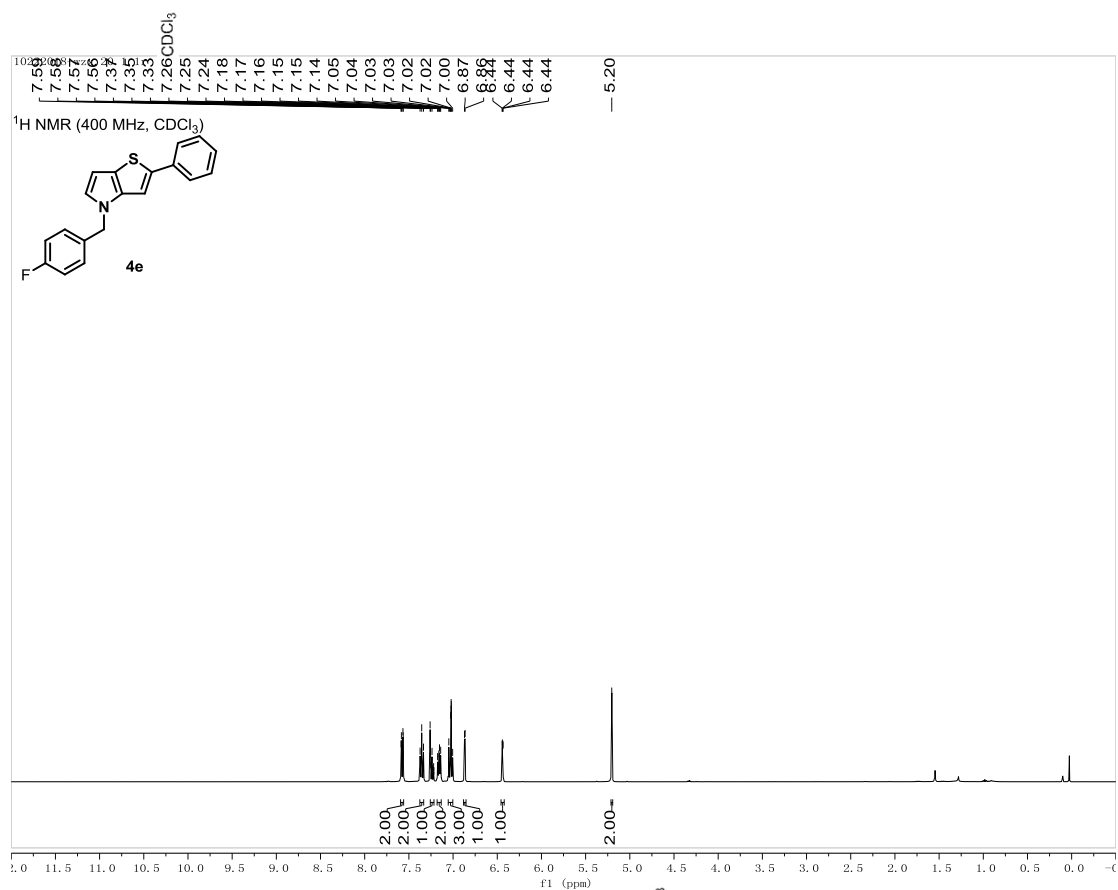


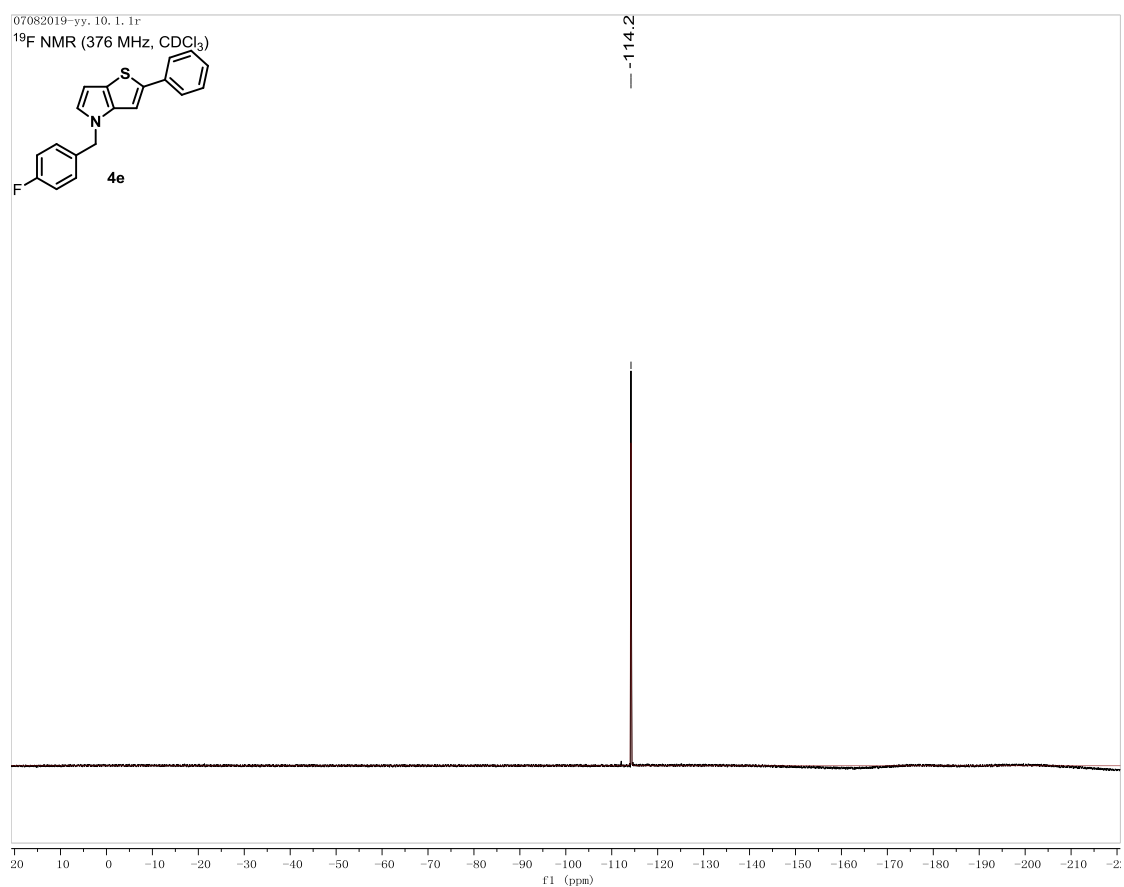


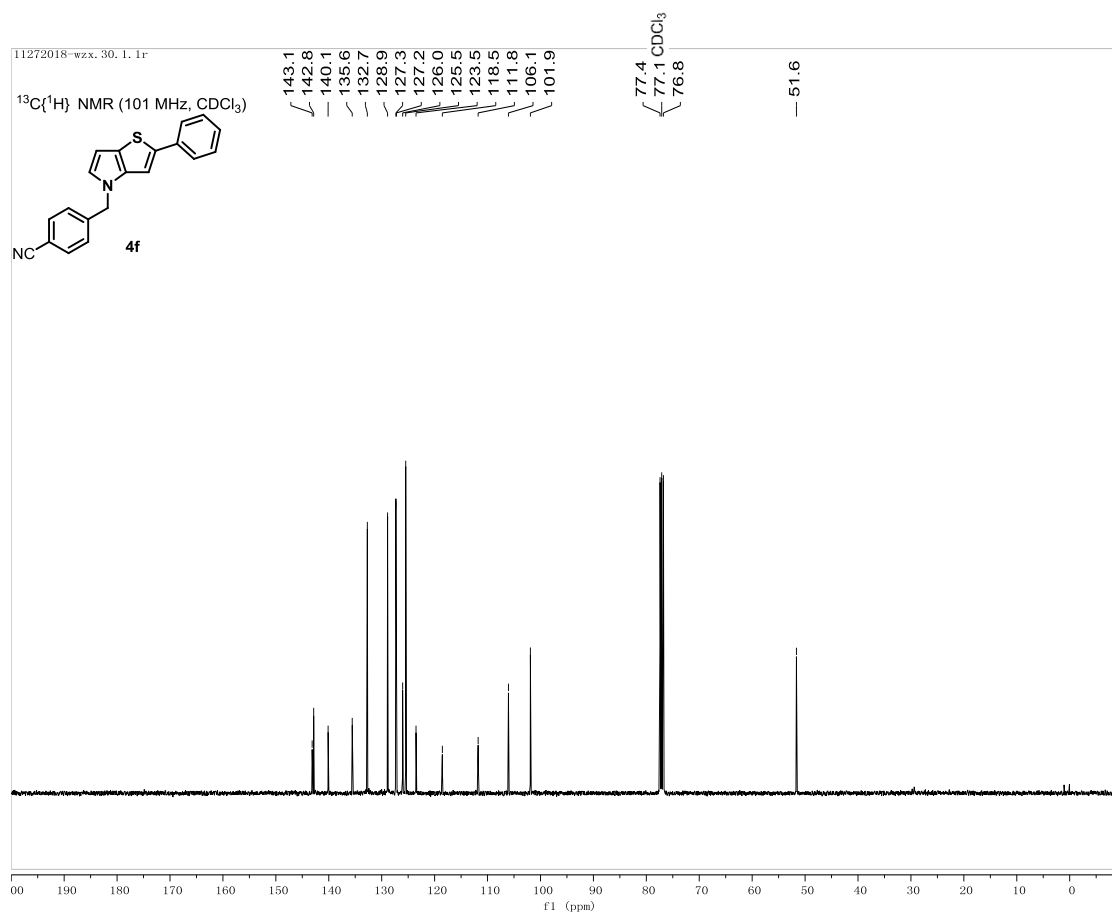
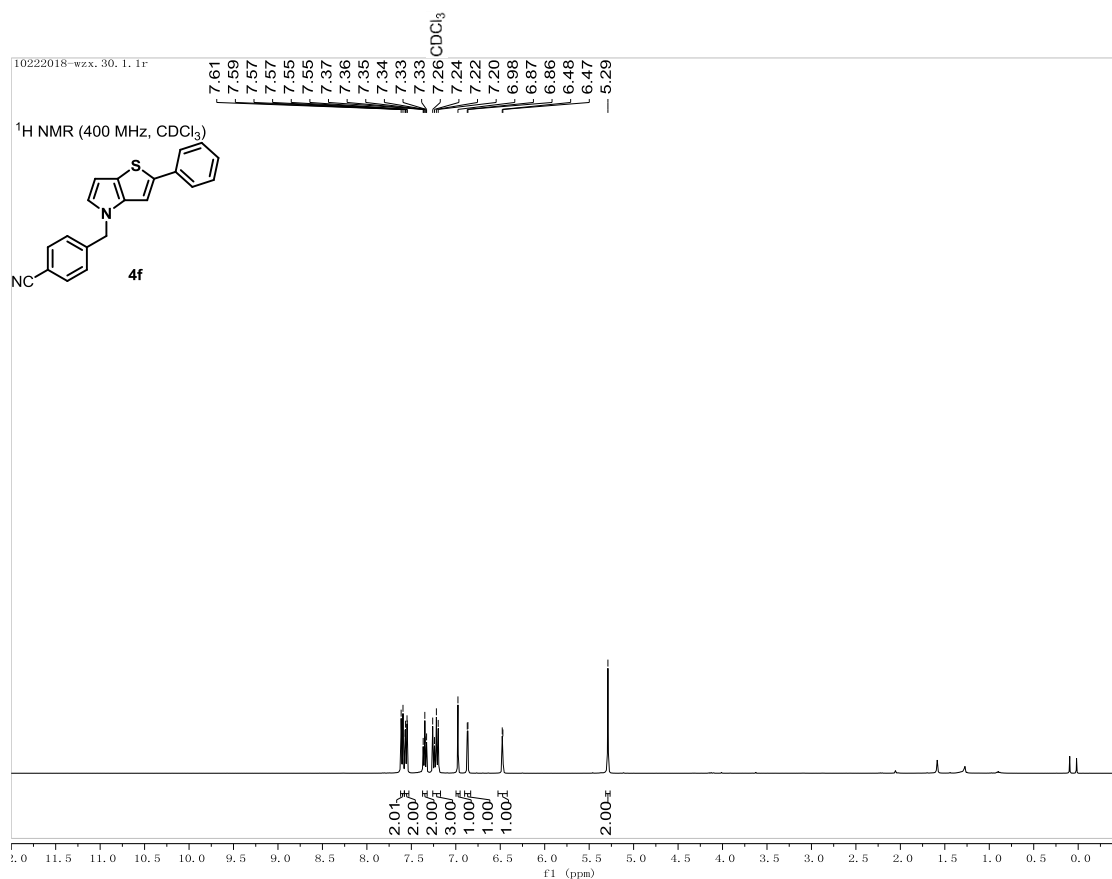


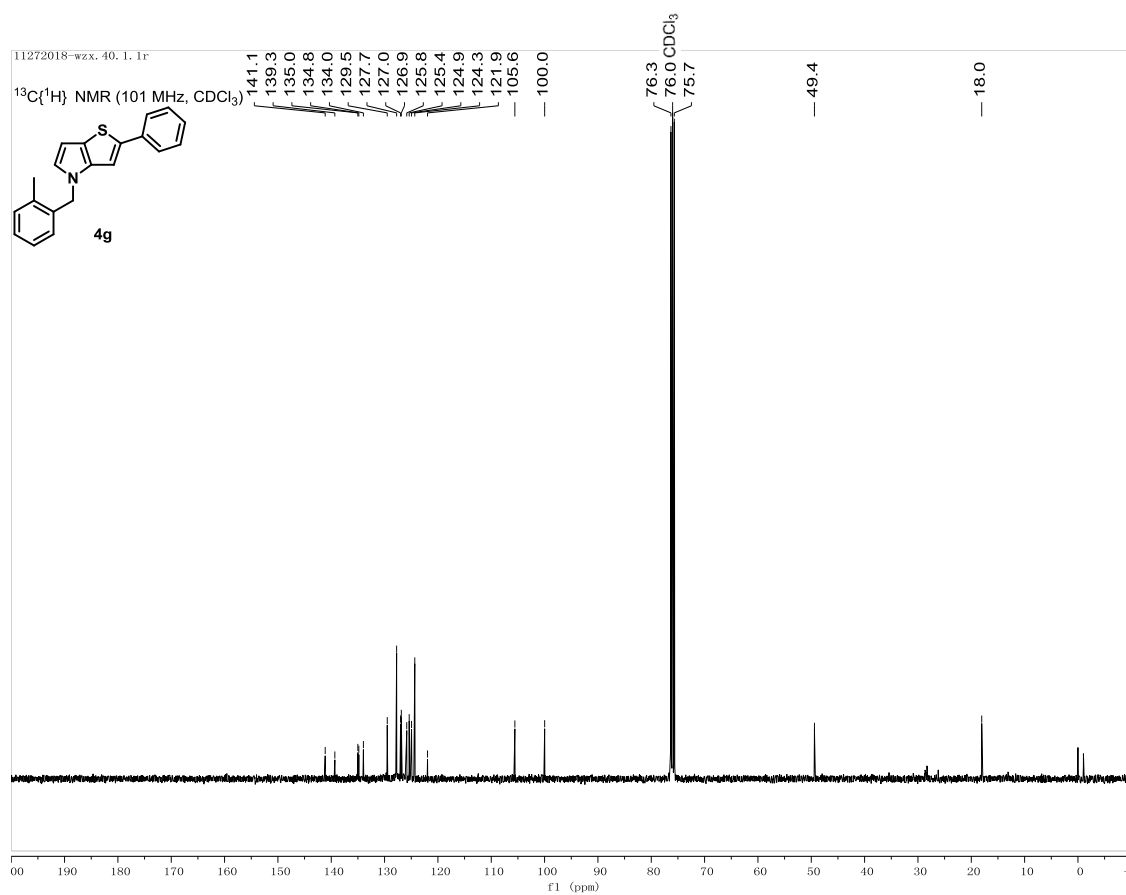
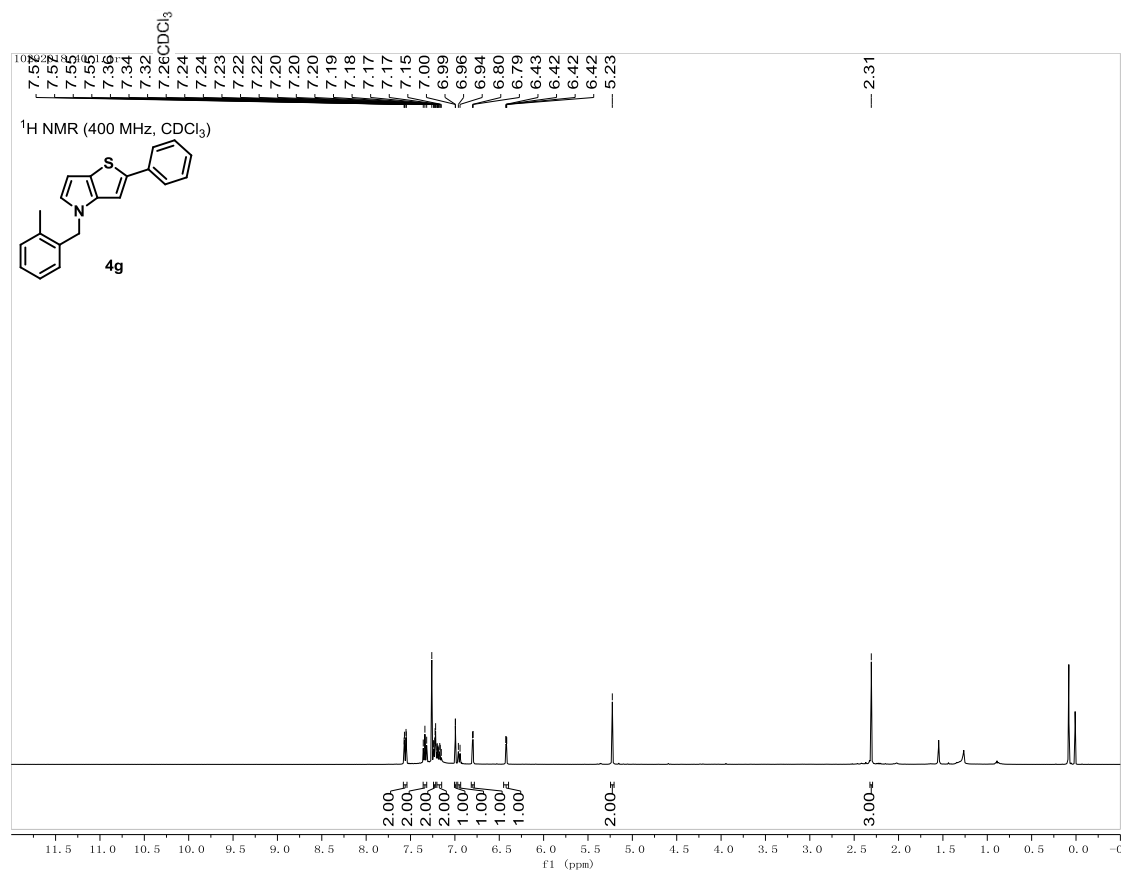


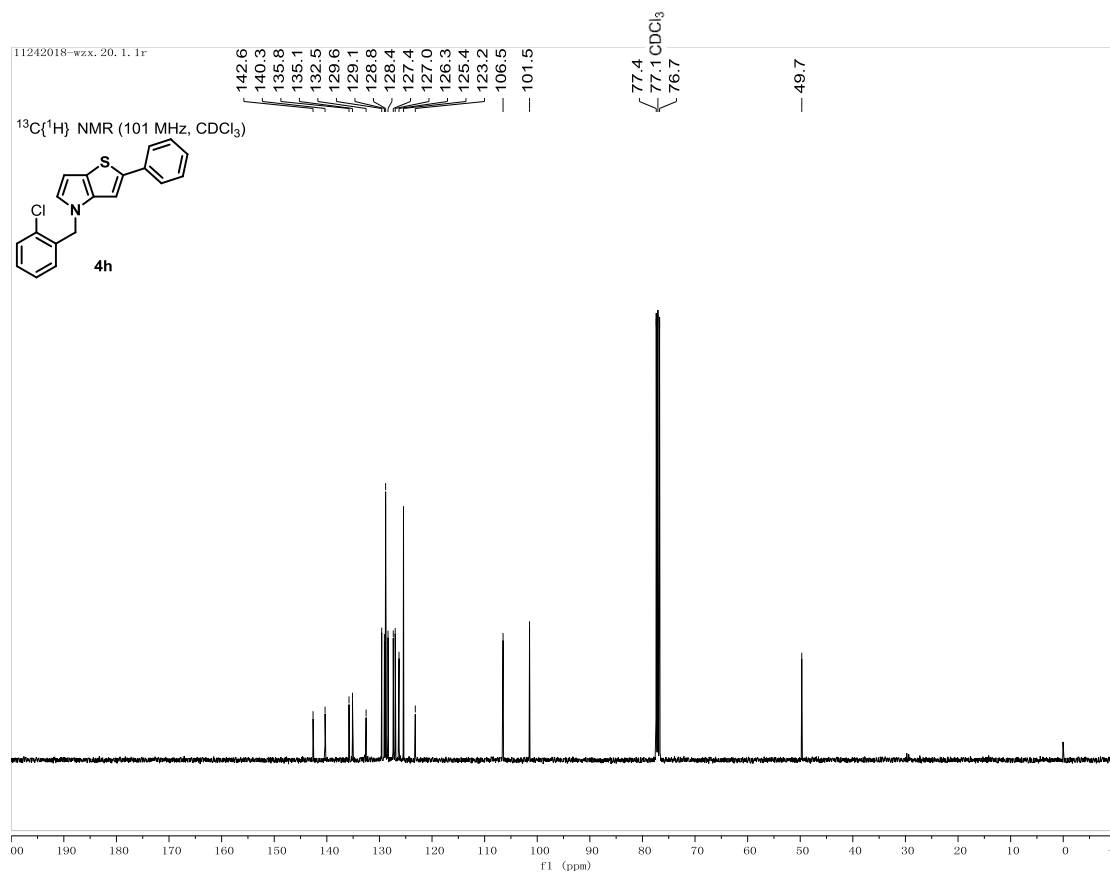
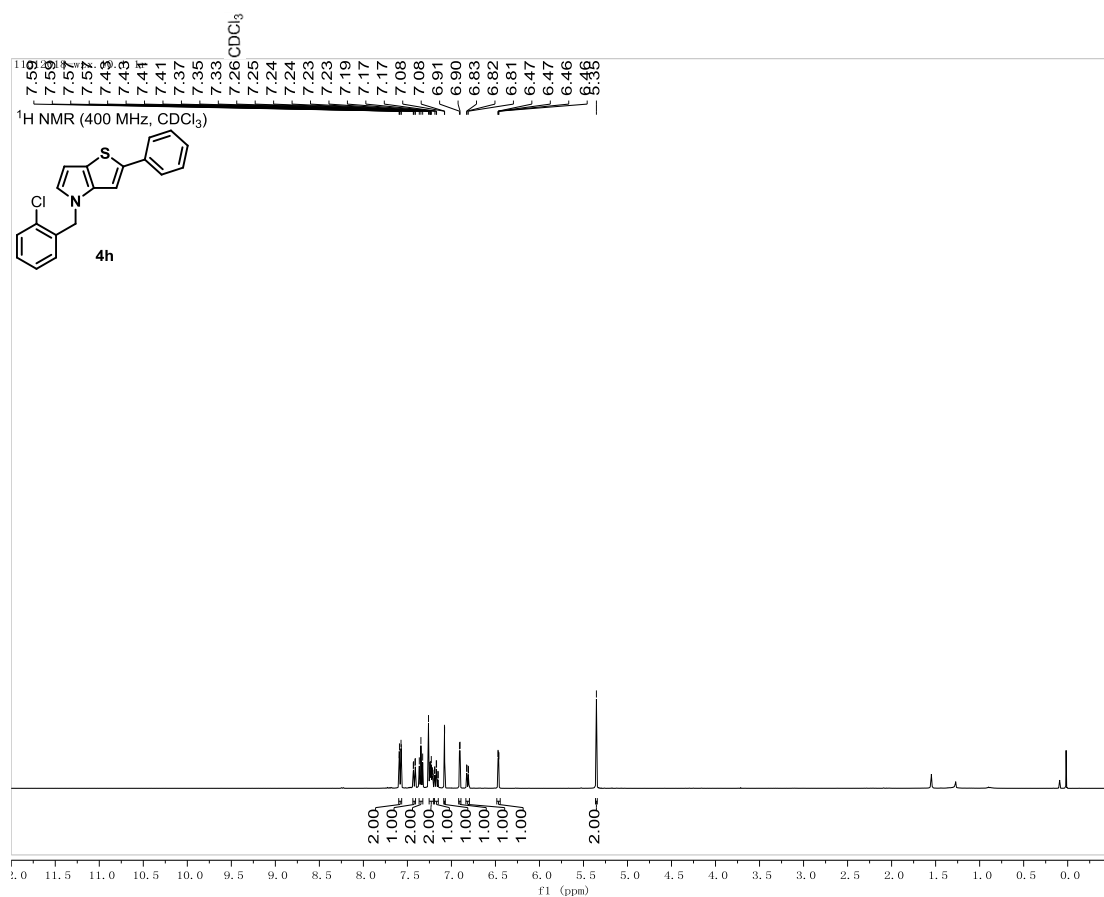


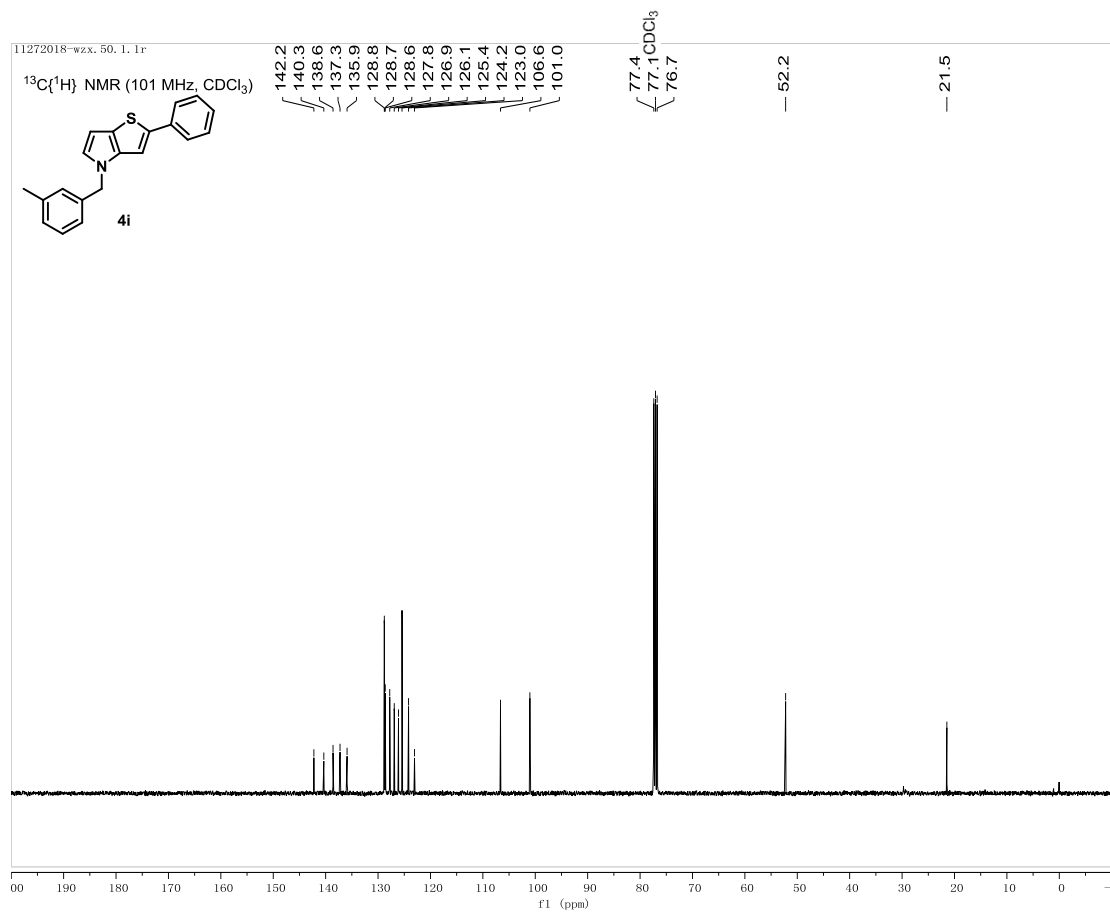
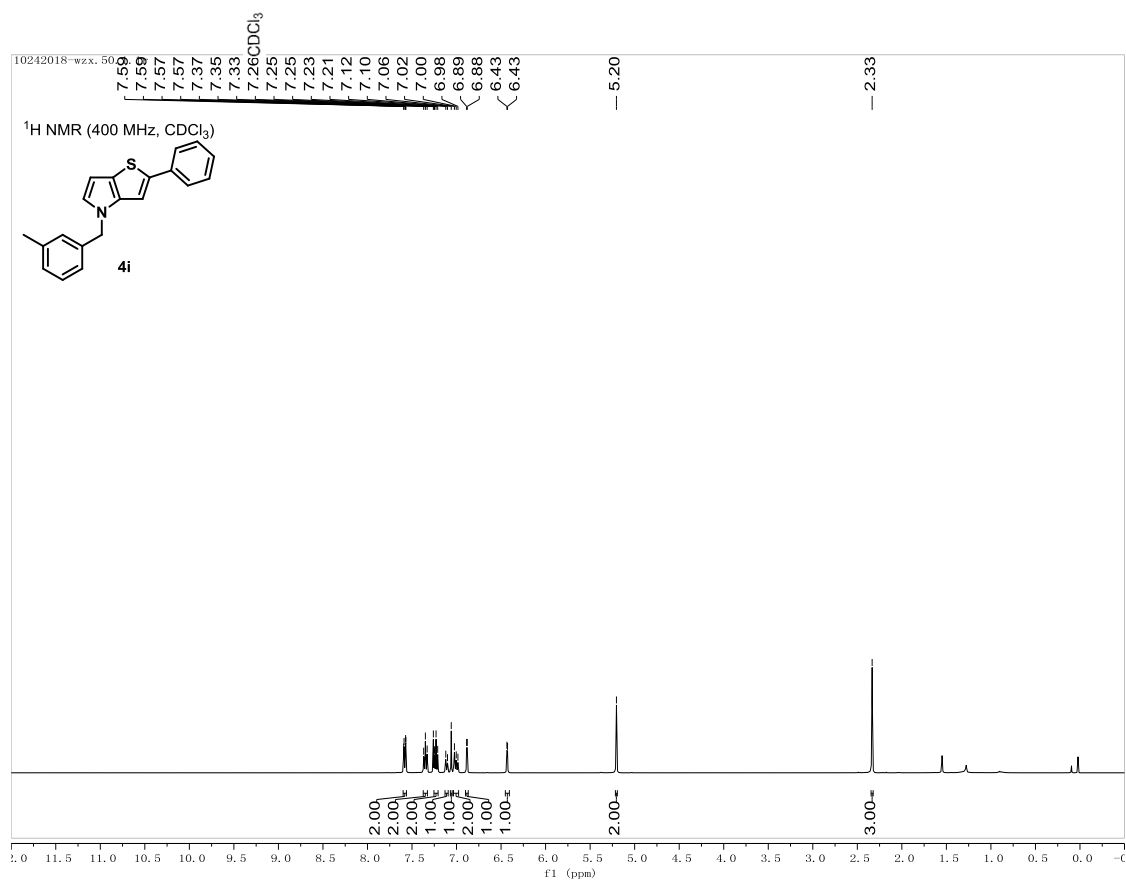


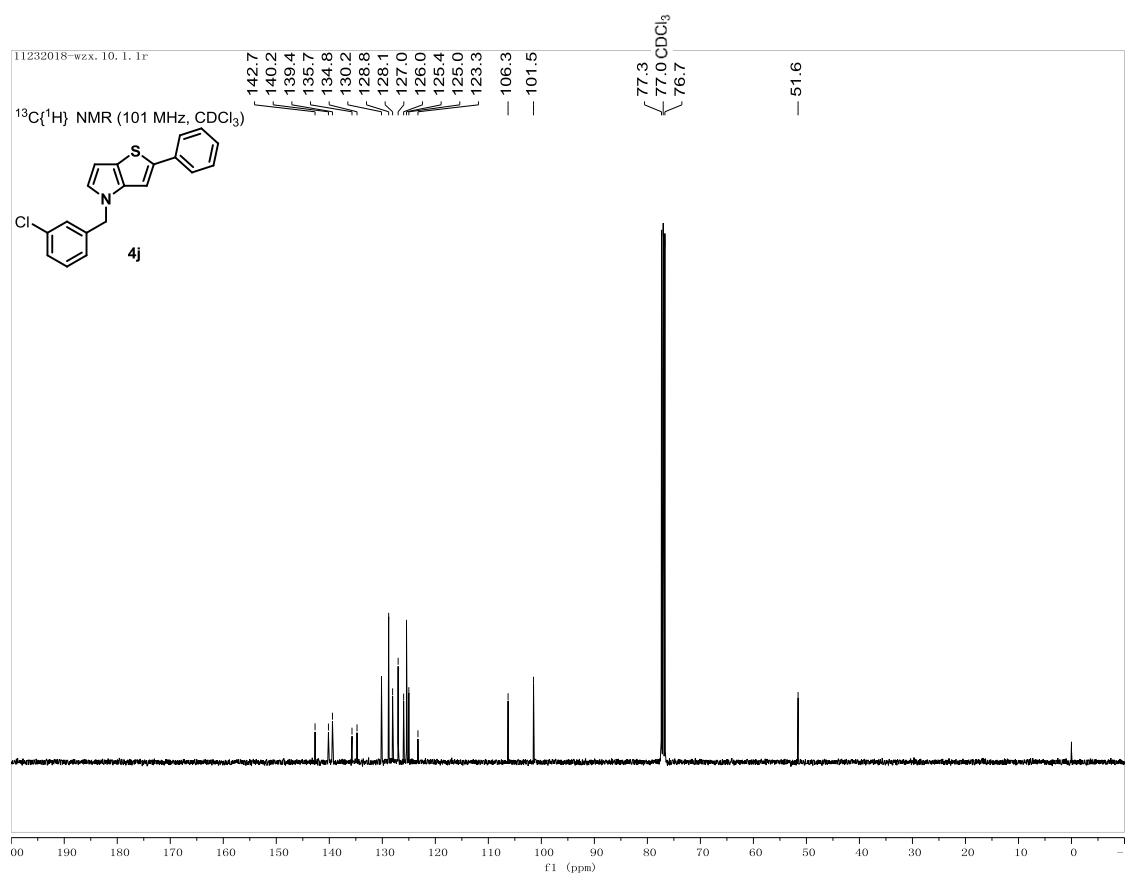
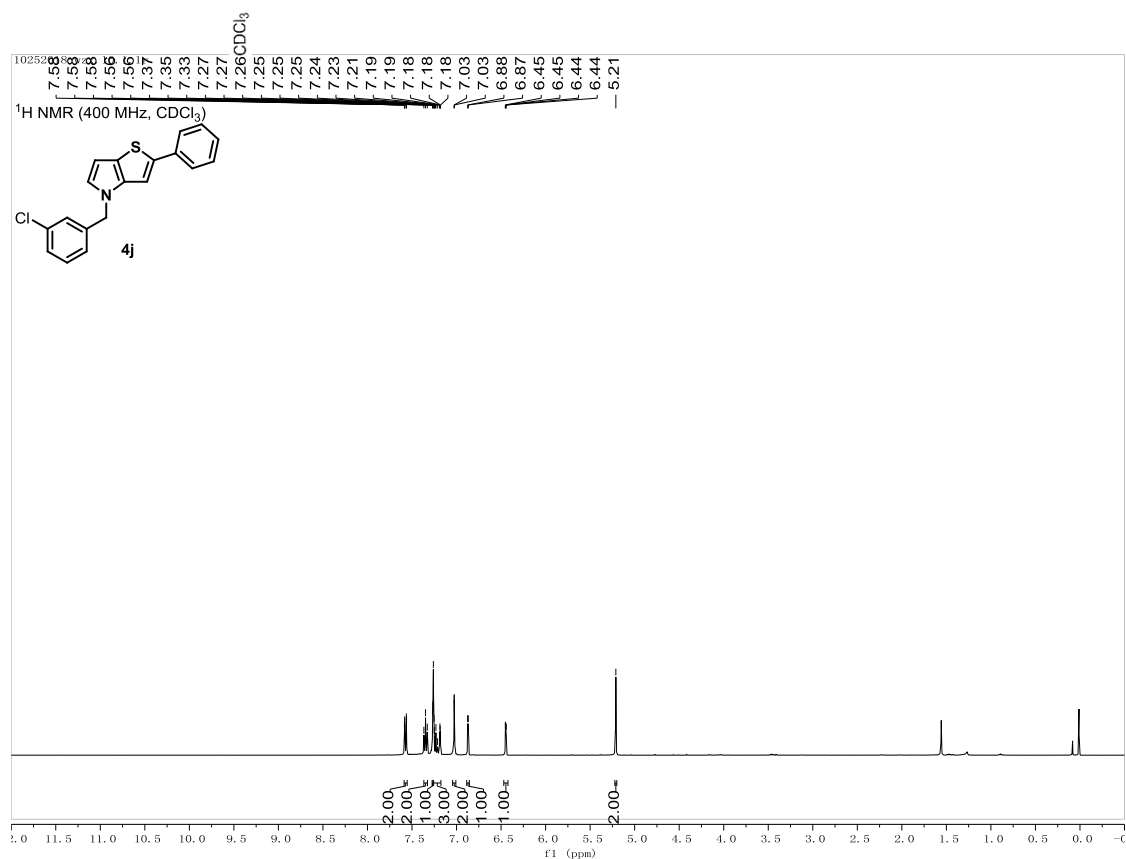


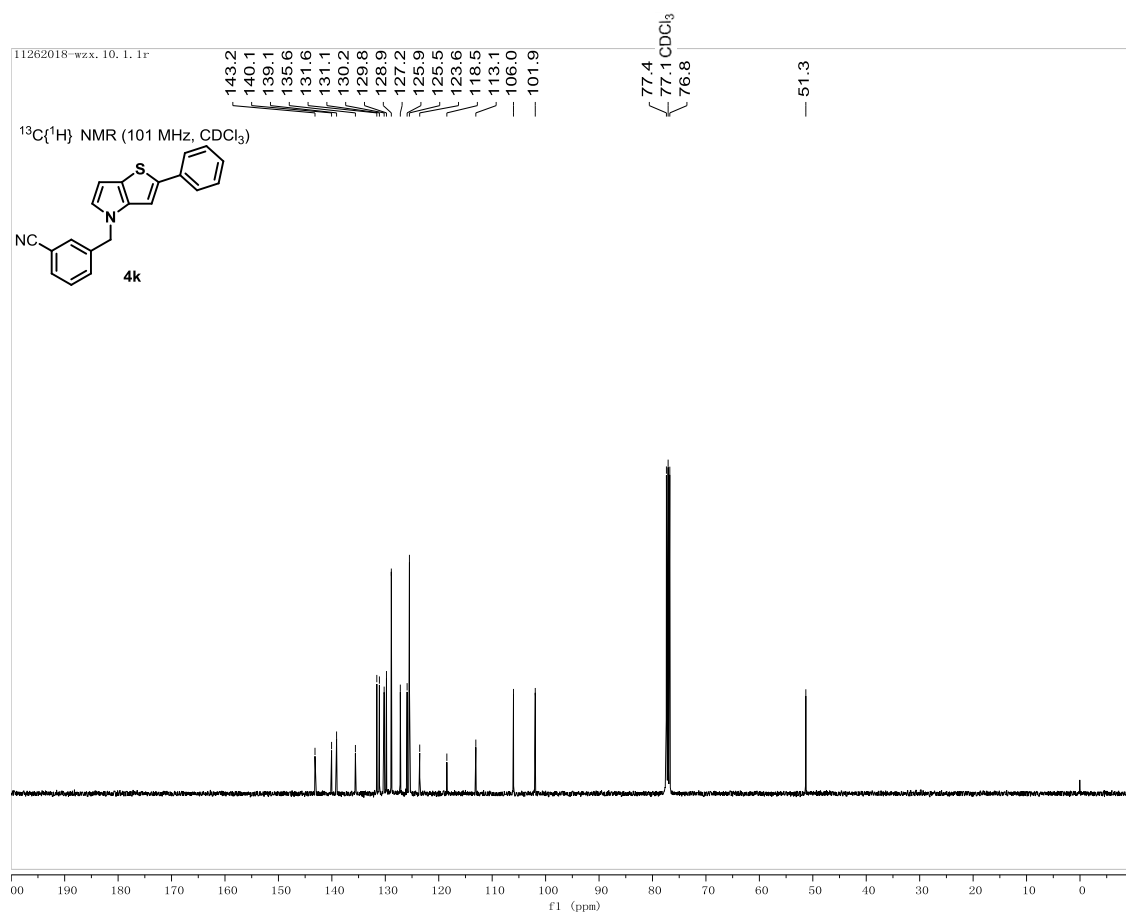
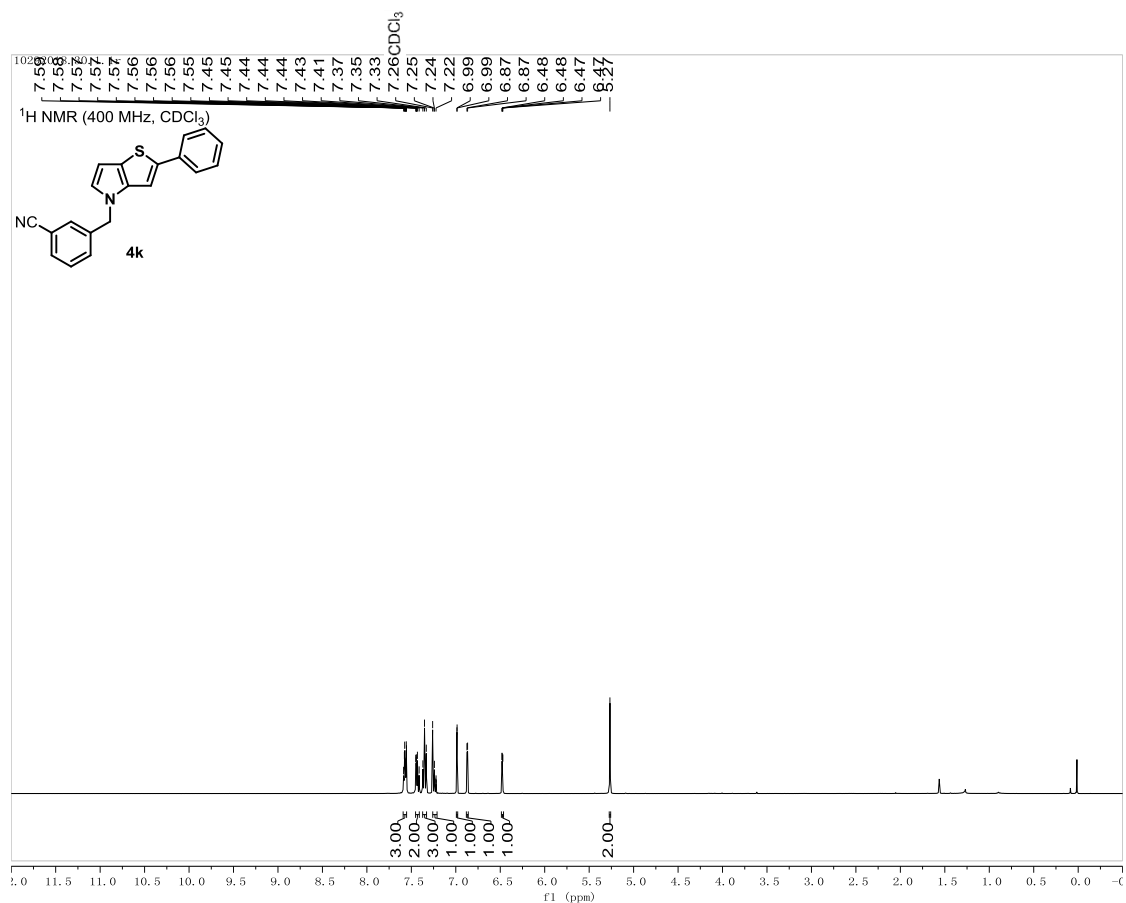


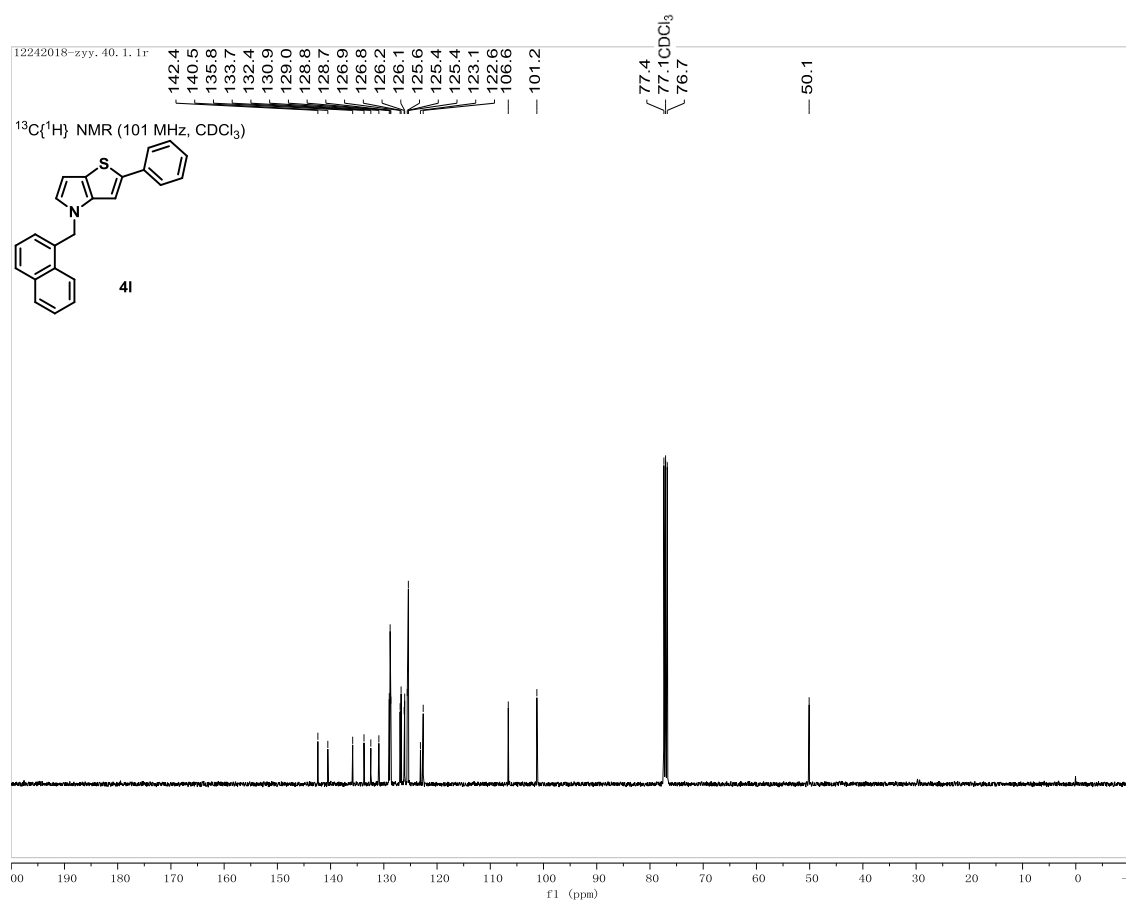
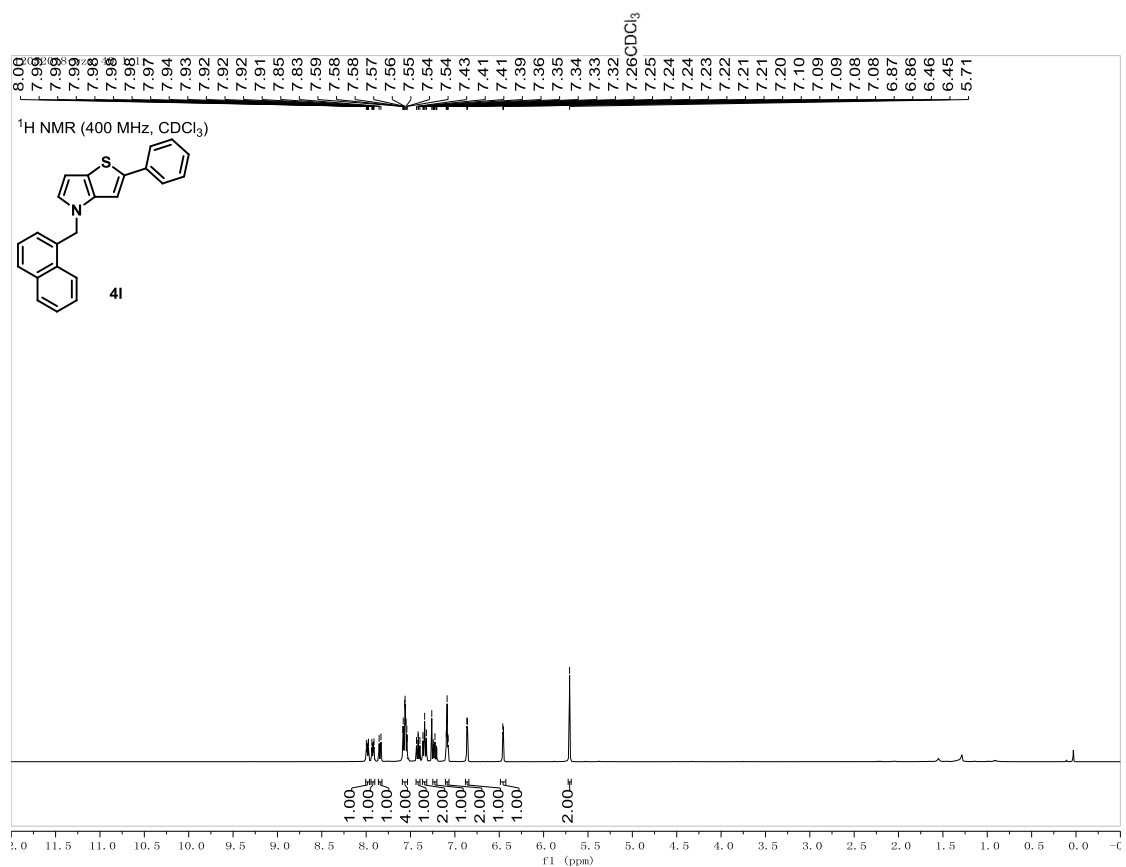


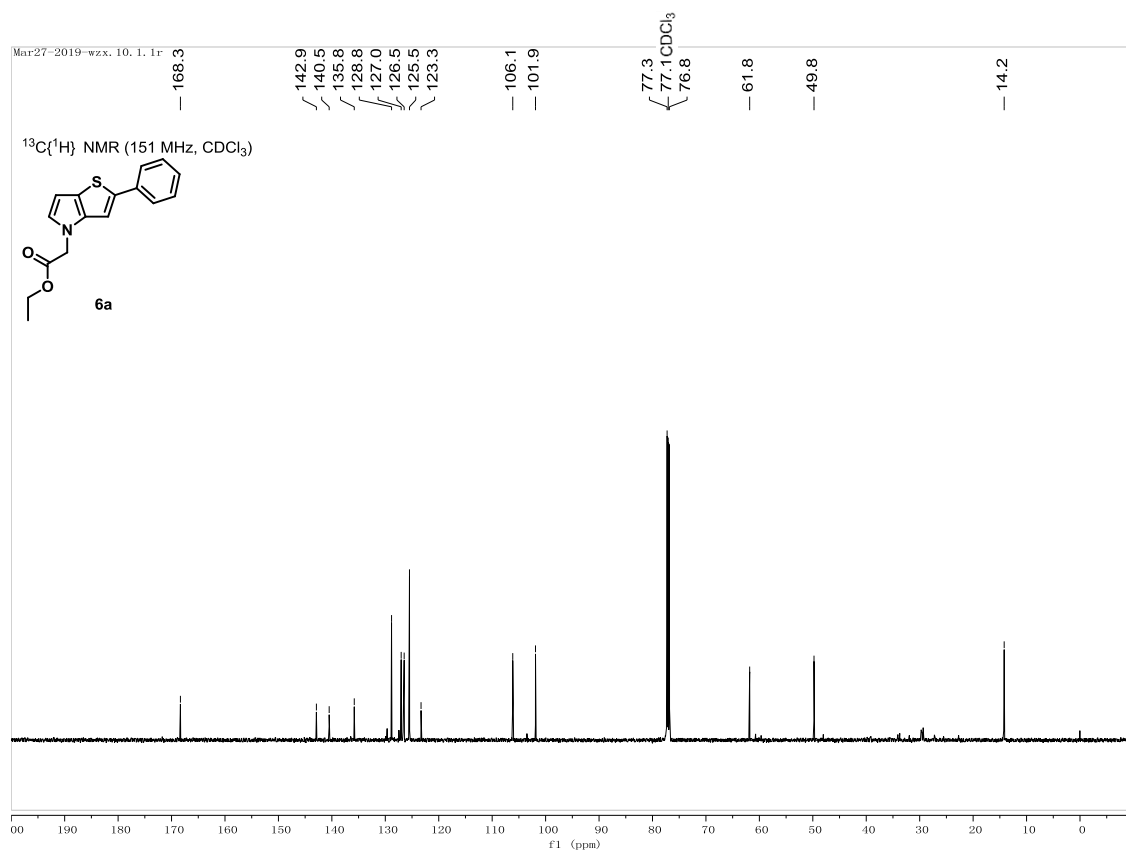
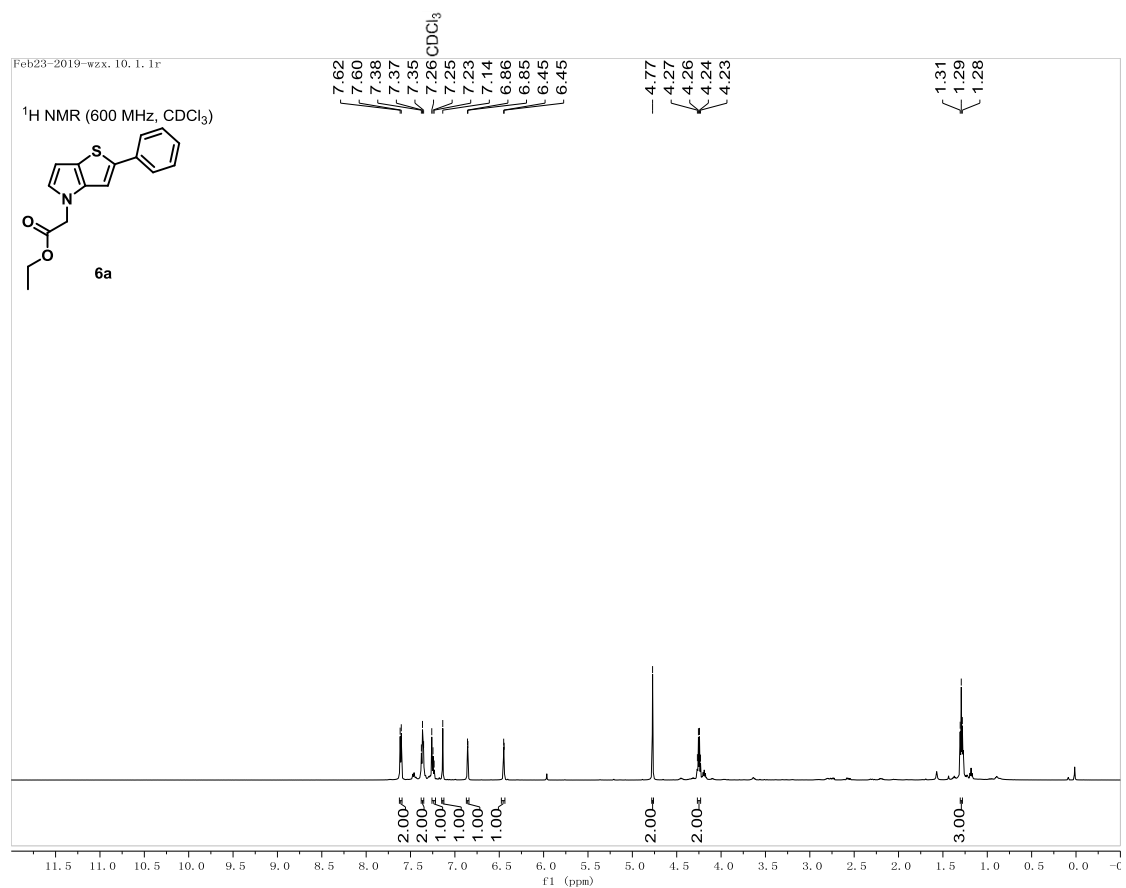


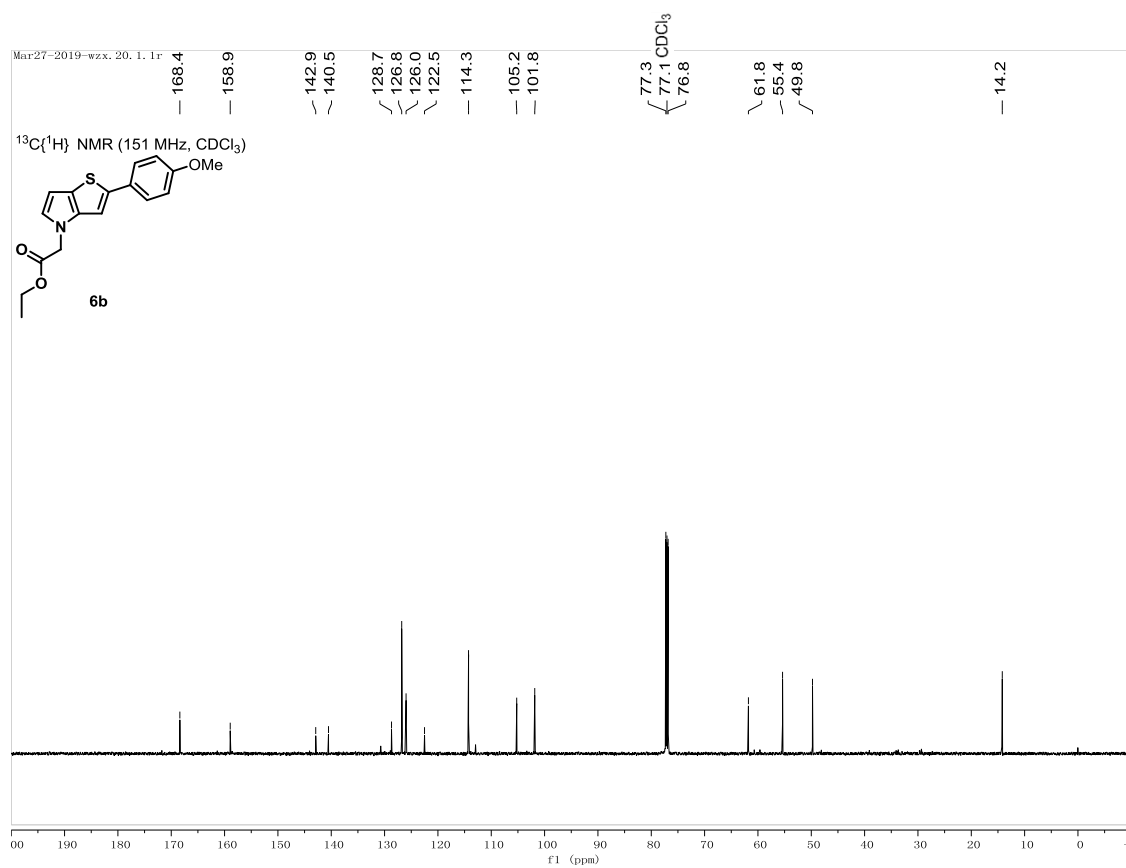
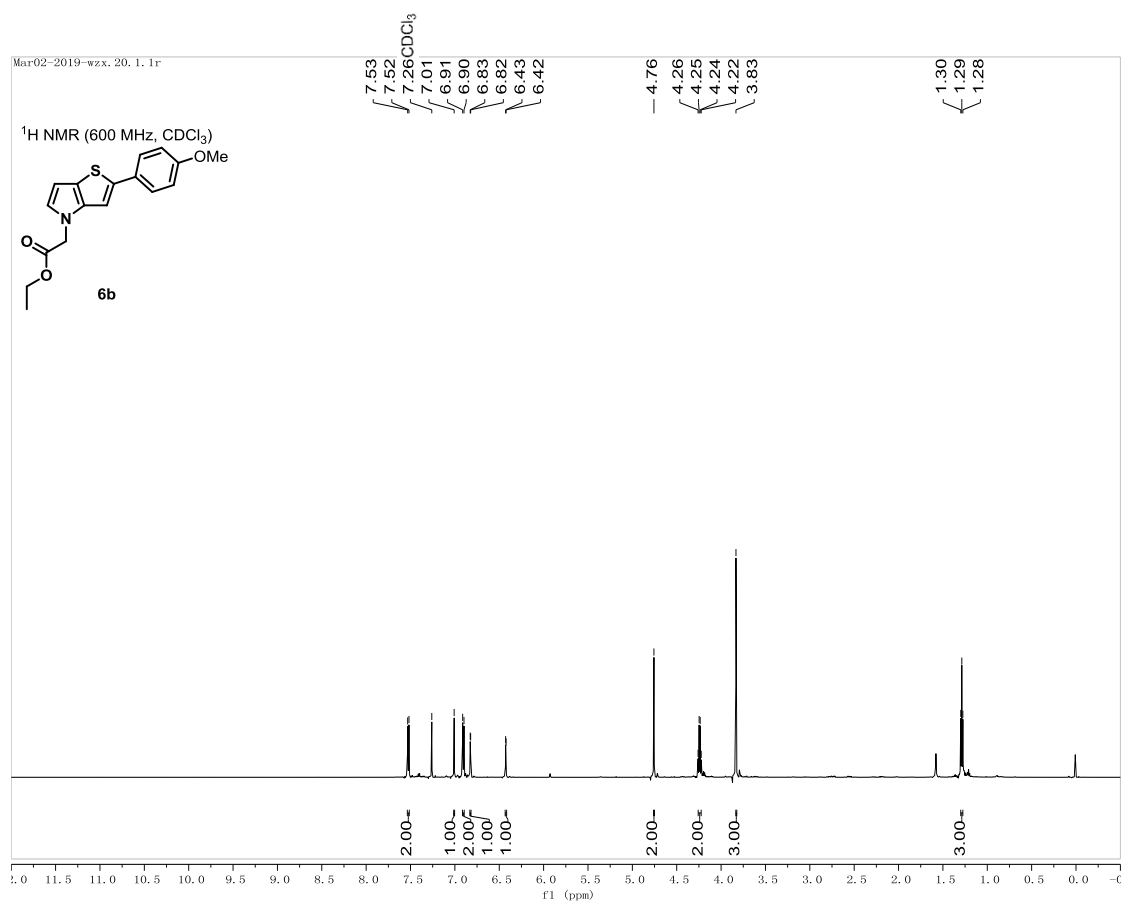


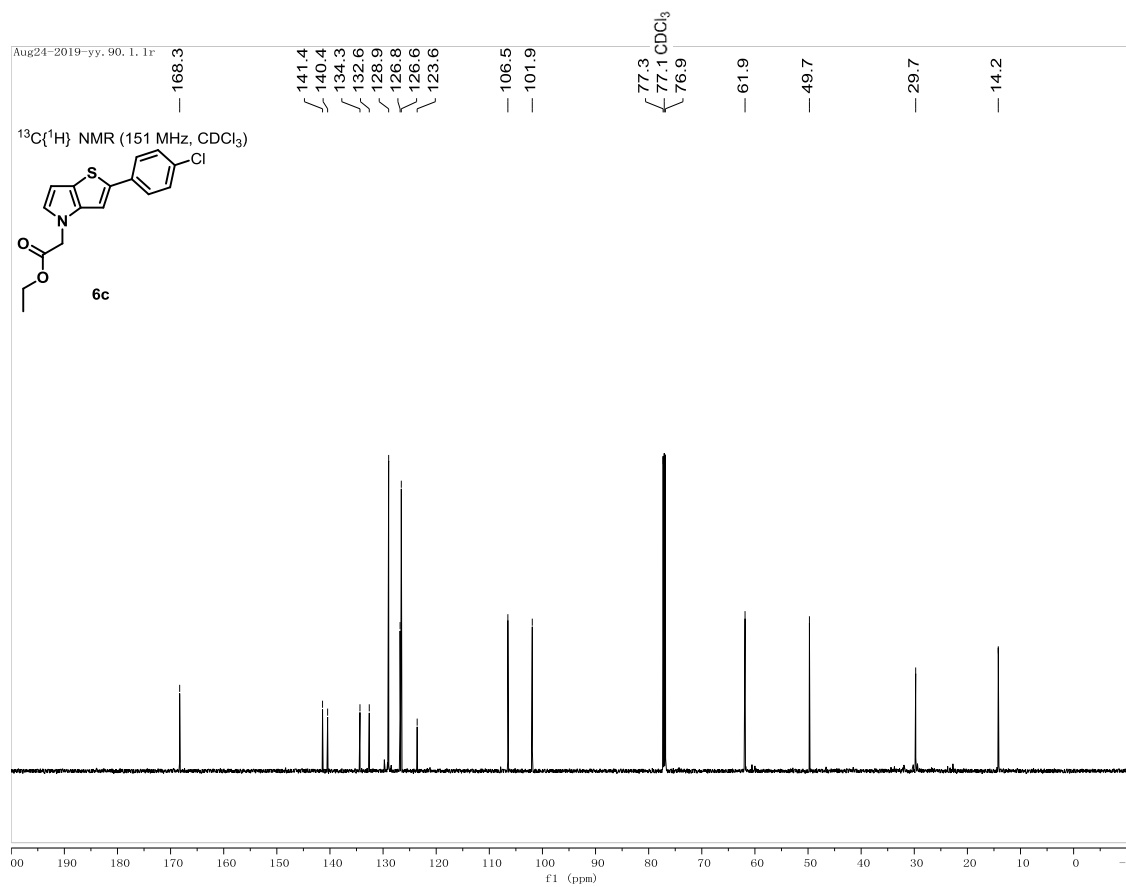
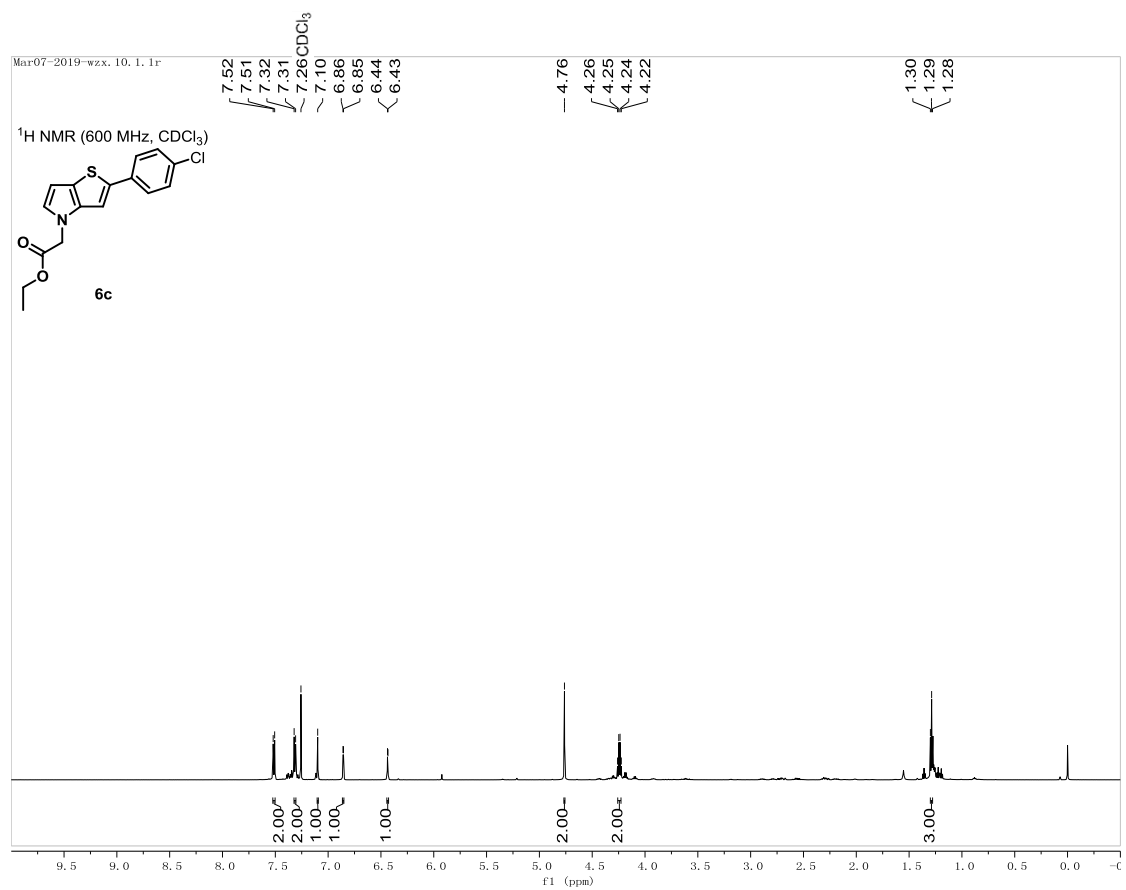


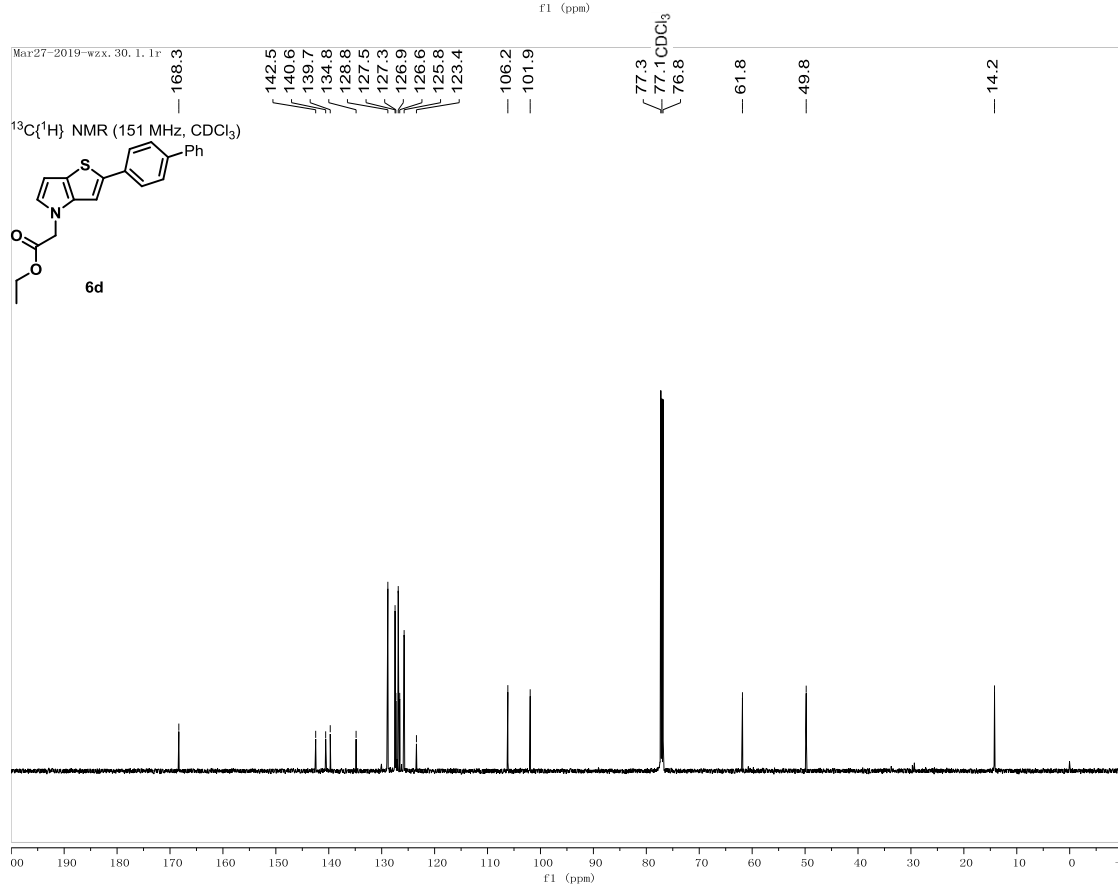
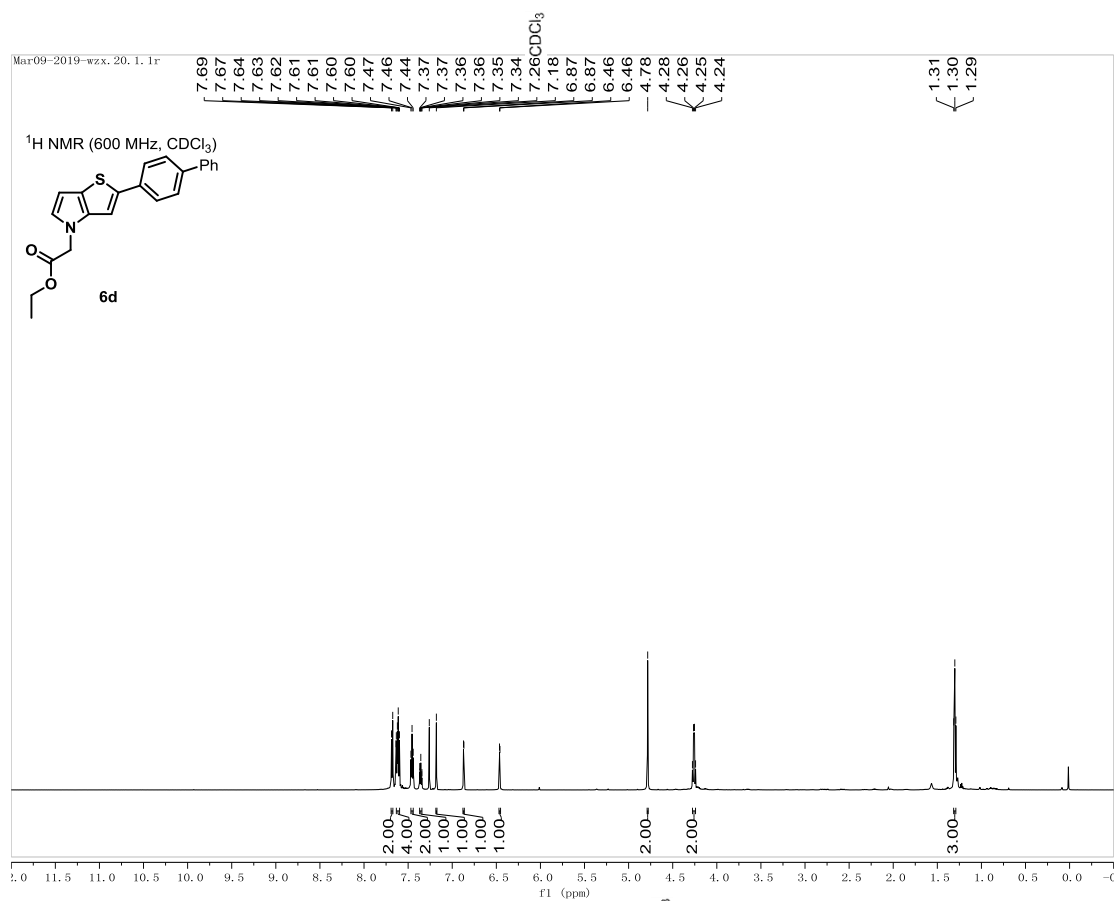


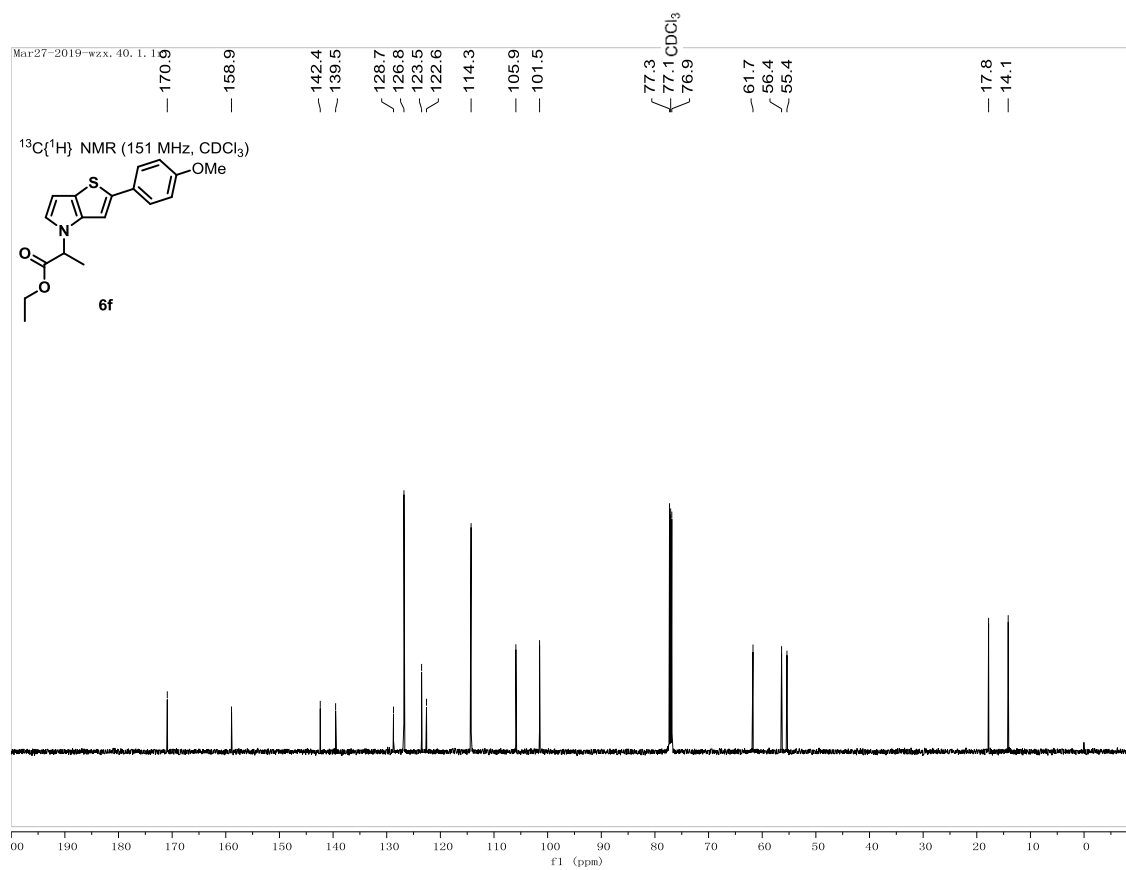
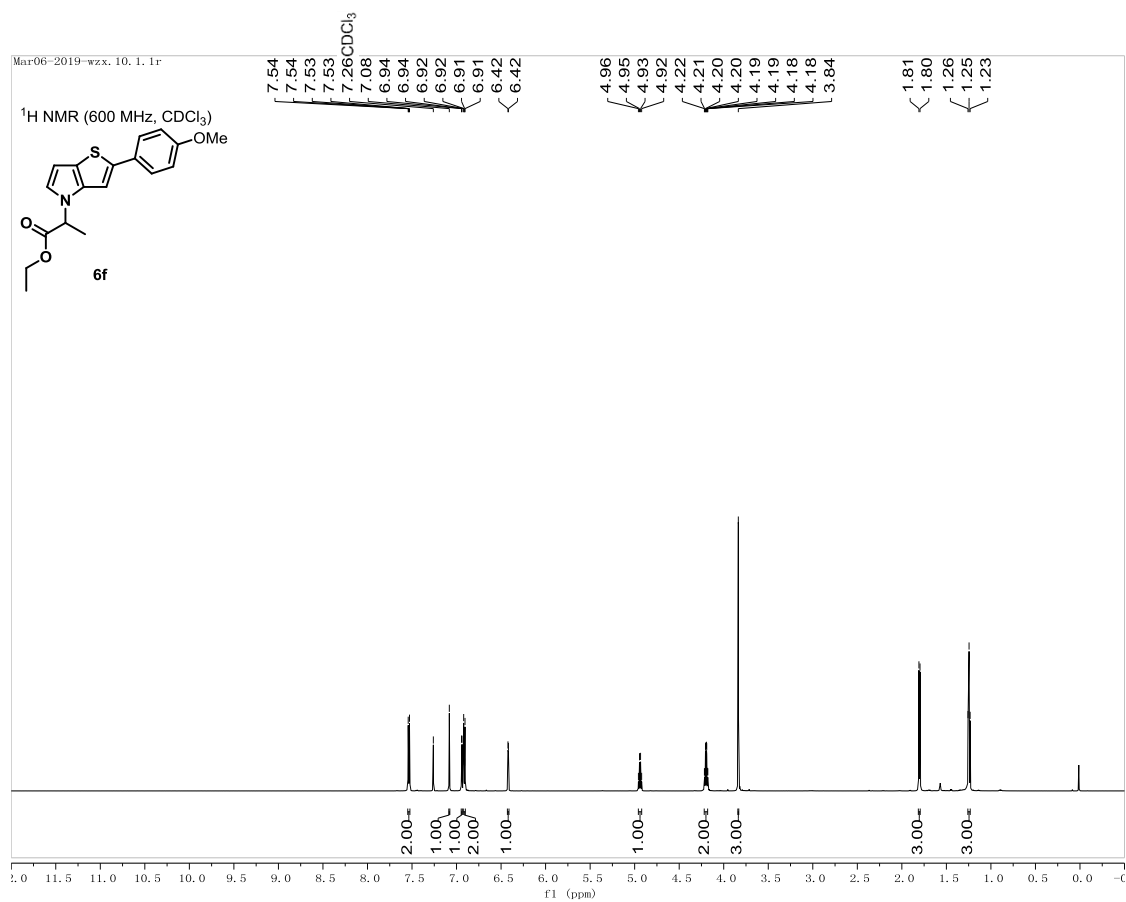


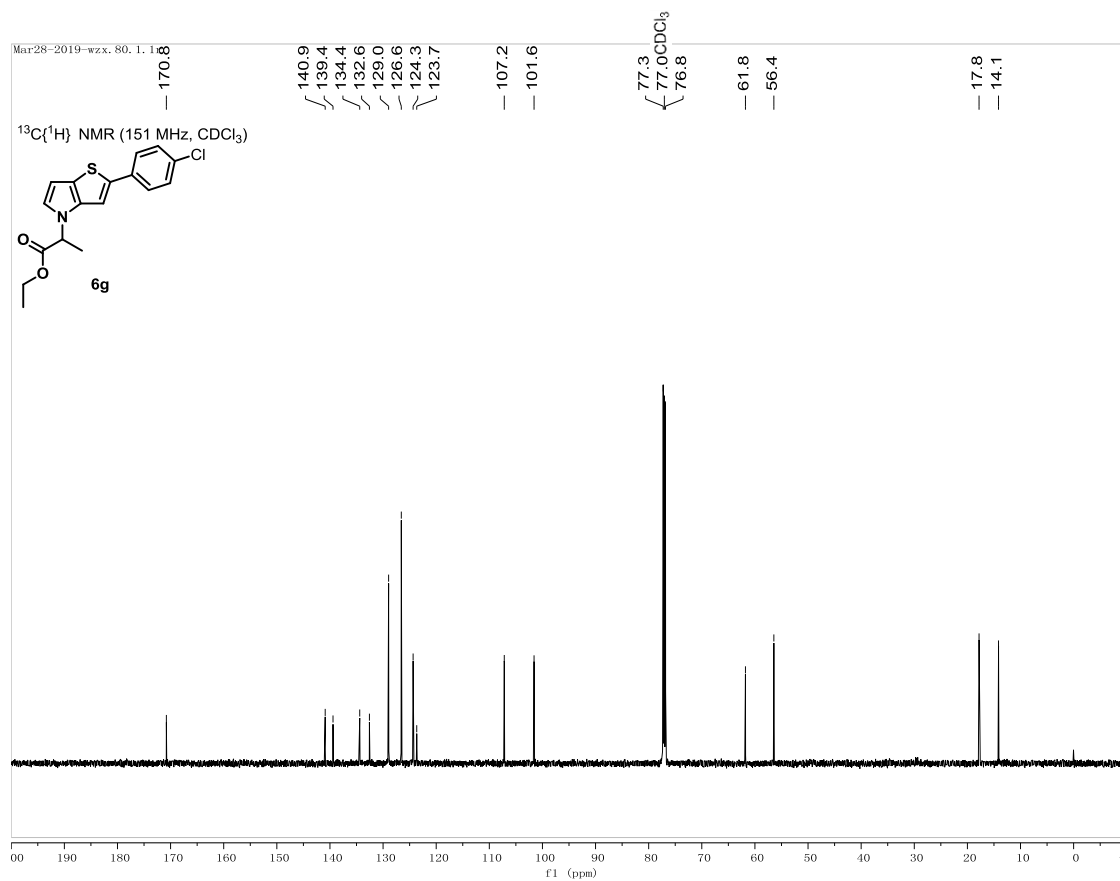
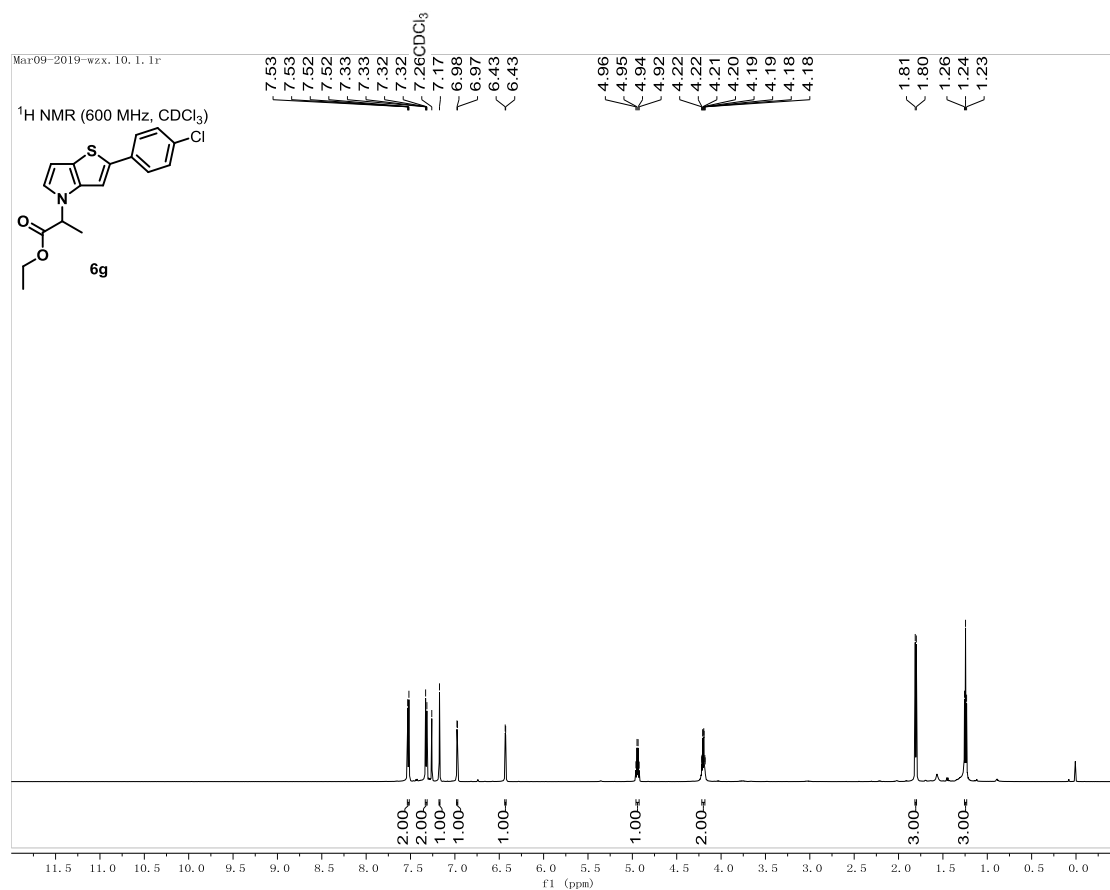












5. Single-Crystal X-ray diffraction.

Single-crystal XRD studies on compounds **2b** and **2h** were performed on a Supernova CCD diffractometer at 293(2) K. Determination of unit cell parameters and data collection were performed with Mo-K α radiation at a wavelength of 0.71073 Å using the ω -scan technique. The structures were solved by direct methods and refined by full matrix least-squares on F^2 using SHELXS-97 and SHELXL-97 programs.¹ The metal atoms in each compound were located from the E-maps, and other non-hydrogen atoms were located in successive difference Fourier syntheses and refined with anisotropic thermal parameters on F^2 . The hydrogen atoms were added theoretically, riding on the concerned atoms and refined with fixed thermal factors. The SQUEEZE function in PLATON was utilized during the refinement of **2b** and **2h** owing to the disordered solvents.²

The structure was then refined again using the data generated. Crystal data and details of the data collection are given in **Table S1-S2**. CCDC 1946767 (**2b**) and 1946766 (**2h**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.

(1) M. Sheldrick, G. SHELXS-97. Program for X-ray crystal structure determination, Gottingen University, Germany, **1997**.

(2) L. Spek, A. Single-crystal structure validation with the program PLATON. *J. Appl. Crystallogr.* **2003**, 36, 7-13.

Single crystals (**2b** and **2h**) suitable for X-ray analysis were obtained by slow evaporation of CH₂Cl₂ solvent.

Table S1. Crystal data and structure refinement for 2b

Identification code	shs-20180927
Empirical formula	C ₂₀ H ₁₇ NS
Formula weight	303.41
Temperature/K	293(2) K
Crystal system	monoclinic
Space group	P 1 21/c 1
a/Å	13.4137(6) Å

b/Å	5.9329(2) Å
c/Å	20.1416(8) Å
$\alpha/^\circ$	90
$\beta/^\circ$	90.210(4)
$\gamma/^\circ$	90
Volume/Å ³	1602.90(11)
Z	4
$\rho_{\text{calc}}/\text{g cm}^{-3}$	1.257
μ/mm^{-1}	0.198
F(000)	640
Crystal size/mm ³	0.45 × 0.28 × 0.11
Radiation	MoK α (λ = 0.71073)
2 θ range for data collection/ $^\circ$	3.58 to 29.21
Index ranges	-17 ≤ h ≤ 16, -8 ≤ k ≤ 6, -26 ≤ l ≤ 25
Reflections collected	8879 / 3702 [R(int) = 0.0224]
Independent reflections	unique 3702
Data/restraints/parameters	3702 / 0 / 200
Goodness-of-fit on F ²	1.038
Final R indexes [I ≥ 2 σ (I)]	R1 = 0.0439, wR2 = 0.0992
Final R indexes [all data]	R1 = 0.0604, wR2 = 0.1081
Largest diff. peak/hole / e Å ⁻³	0.218 and -0.236

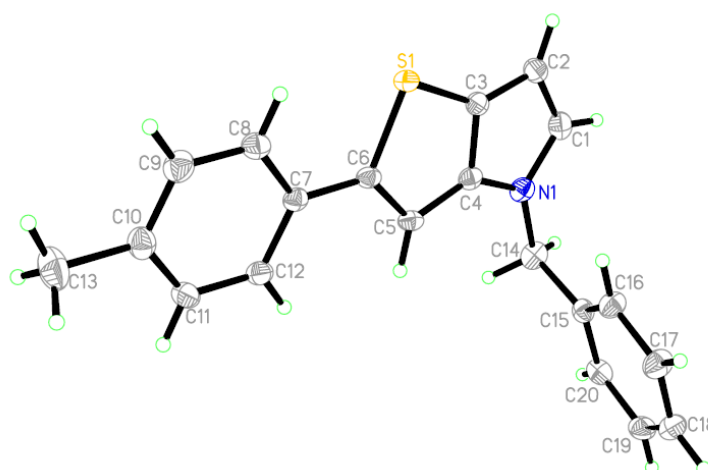


Figure S1. Additional X-ray crystallographic structures **2b** with 30% probability ellipsoid.

Table S2. Crystal data and structure refinement for **2h**

Identification code	shs-2-20180919
Empirical formula	C ₂₁ H ₁₉ NOS
Formula weight	333.43
Temperature/K	293(2) K

Crystal system	monoclinic
Space group	C1c1
a/Å	28.9068(5) Å
b/Å	5.75310(10) Å
c/Å	10.5484(2) Å
$\alpha/^\circ$	90
$\beta/^\circ$	98.531(2)
$\gamma/^\circ$	90
Volume/Å ³	1734.83(5)
Z	4
$\rho_{\text{calc}}/\text{g cm}^{-3}$	1.277
μ/mm^{-1}	1.694
F(000)	704
Crystal size/mm ³	0.20 × 0.20 × 0.20
Radiation	MoK α (λ = 0.71073)
2 Θ range for data collection/ $^\circ$	6.19 to 71.18
Index ranges	-34 ≤ h ≤ 35, -5 ≤ k ≤ 6, -12 ≤ l ≤ 9
Reflections collected	5226 / 2646 [R(int) = 0.0166]
Independent reflections	unique 2646
Data/restraints/parameters	2646 / 2 / 218
Goodness-of-fit on F ²	1.079
Final R indexes [I ≥ 2 σ (I)]	R1 = 0.0363, wR2 = 0.1052
Final R indexes [all data]	R1 = 0.0369, wR2 = 0.1060
Largest diff. peak/hole / e Å ⁻³	0.108 and -0.191

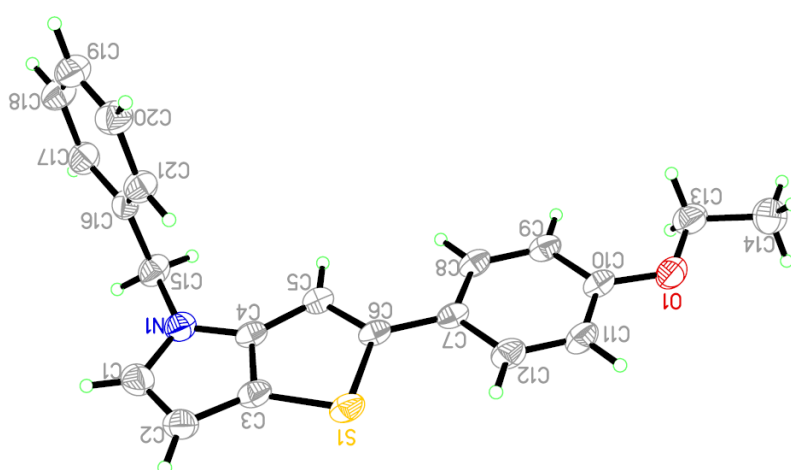


Figure S2. Additional X-ray crystallographic structures **2h** with 30% probability ellipsoid.