

Supporting Information

Complex pattern formation in solutions of protein and mixed salts using dehydrating sessile droplets

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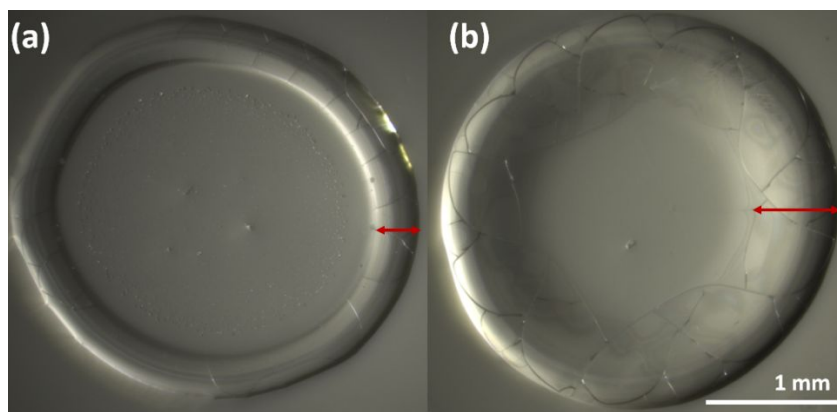


Figure S1. Final deposit of BSA droplets. We show the final deposit of droplets with an initial concentration of (a) 3% wt. BSA and (b) 7% wt. BSA in water.

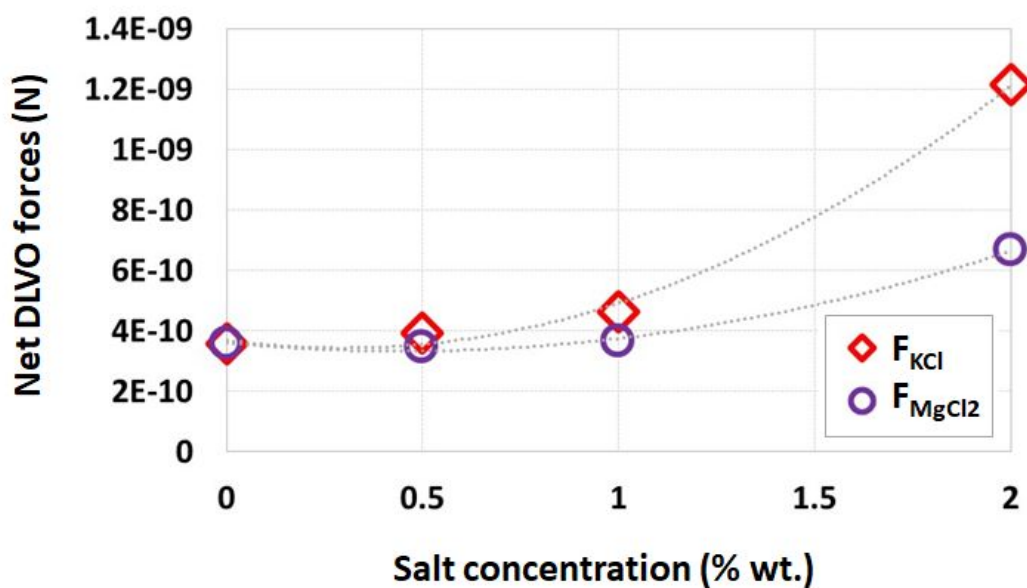
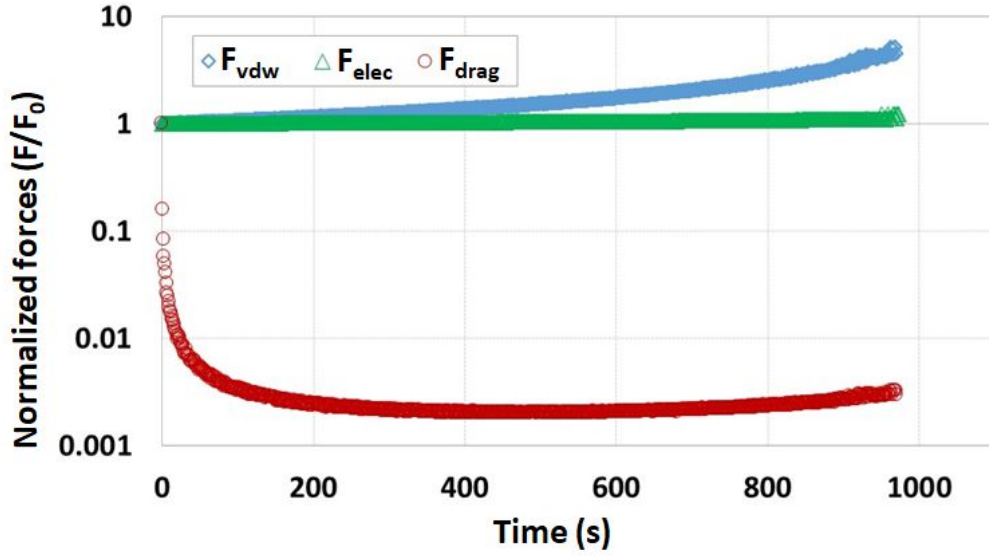


Figure S2. Comparison of the net DLVO forces. The net DLVO forces (magnitude) acting between particle-particle are plotted with the increase in the salt concentration (for BSA+KCl and BSA+MgCl₂ droplets; BSA: 7% wt.) [F_{KCl} and F_{MgCl2} are the net forces for KCl and MgCl₂ respectively]



F_{drag} : drag force

F_{vdw} : van der Waals forces between protein-protein

F_{elec} : electrostatic interaction forces between protein-protein

Figure S3. Temporal variation of different forces. The temporal variation of different forces are plotted for droplets of 1% wt. BSA + 2% wt. KCl salt [the forces are normalized with the forces at the initial state of the droplet (F_0 : forces at time= 0)].

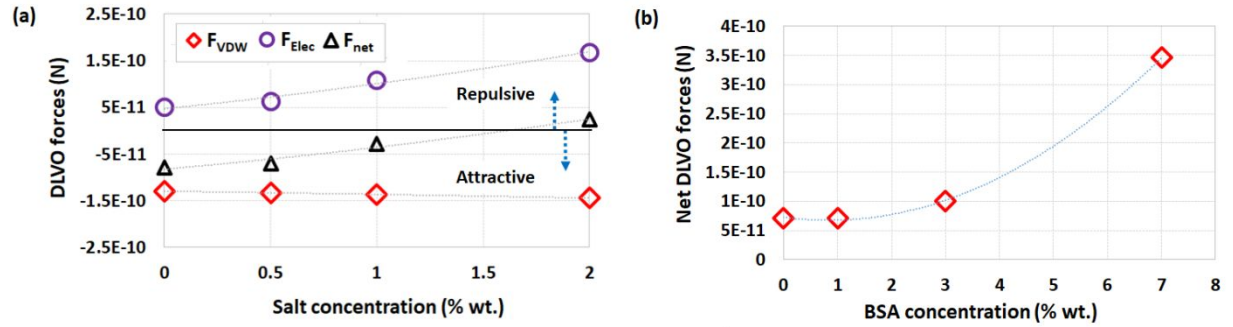


Figure S4. Variation of DLVO forces with salt concentration. The plot shows the variation of (a) DLVO forces for MgCl_2 +BSA droplets with salt concentration for (BSA: 1% wt.) and (b) net DLVO forces ($F_{\text{net}}=F_{\text{VDW}}+F_{\text{Elec}}$) (only magnitude) with variation in the BSA concentration (MgCl_2 : 0.5% wt.)

[van der Waals attraction: $F_{\text{VDW}} = \frac{-A}{6D^2} \frac{r_1 r_2}{(r_1 + r_2)}$ and electrostatic repulsion forces: $F_{\text{Elec}} = \kappa Z e^{-\kappa D}$ respectively]

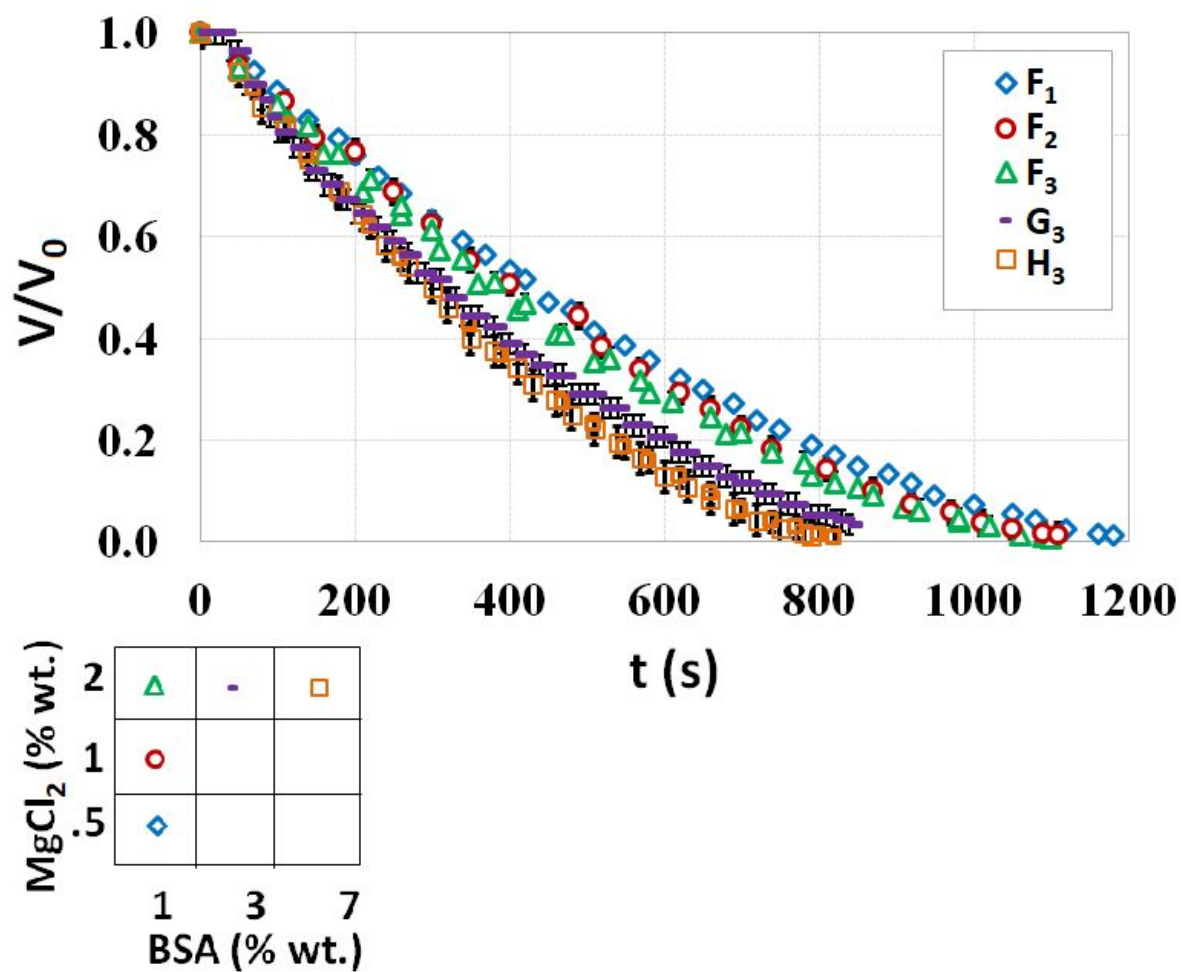


Figure S5. Temporal variation of droplet volume. The temporal variation of the volume of BSA+MgCl₂ droplets are plotted in the Figure. The volume is non-dimensionalized with the initial droplet volume

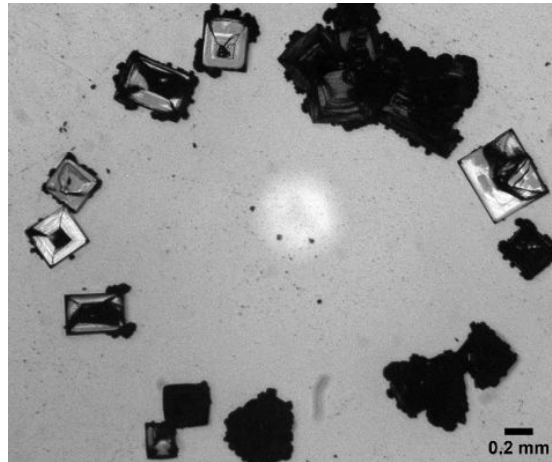


Figure S6. Final deposit of KCl crystals. The final precipitate shows the formation of ring-type peripheral deposit of KCl crystals (the initial concentration is 1 M solution in water) [unpublished data with permission from Efstratiou et. al].

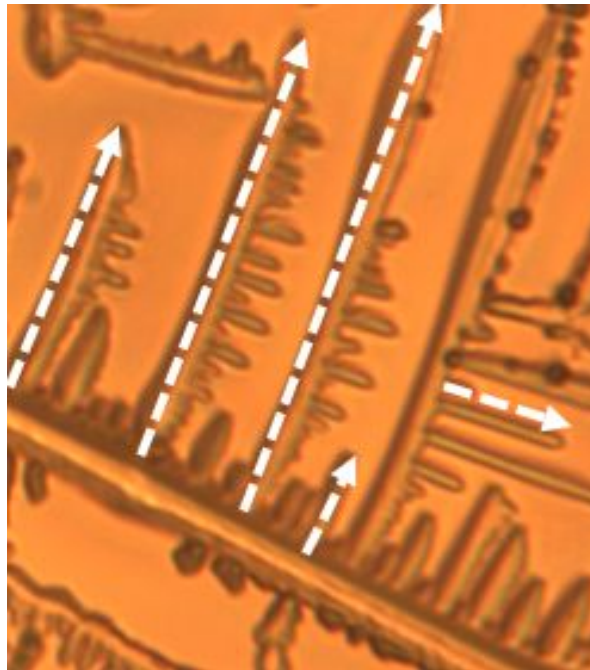


Figure S7: Estimation of l_{crys} . The l_{crys} is the axial length of the crystals measured from the point of initiation to the final growth (full grown crystal). The l_{crys} is calculated from the point of its initiation to the tip (maximum growth) for the main crystal as well as the branches.