Supporting Information for

# Zwitterionic Ru(III) Complexes: Stability of Metal-Ligand Bond and Host-Guest Binding with Cucurbit[7]uril

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**Figure S1**. <sup>1</sup>H NMR spectrum of compound **1a** dissolved in  $D_2O$  at 298 K. The polar water environment stabilizes the ion pair formed by the dissociation of zwitterionic compound **1a**. The DABCO-R ligand in the ion pair is not exposed directly to the paramagnetic environment of the RuCl<sub>4</sub>DMSO core in contrast to that in **1a**. Color code: green - zwitterionic system **1a** (not detected in the aqueous solution, possibly due to its poor solubility), magenta - suggested ion pair resulting from the decomposition of **1a**.



**Figure S2.** <sup>1</sup>H NMR spectrum of compound **1a** in (**a**) dry DMF- $d_7$  (**b**) DMF- $d_7$  with a few drops of D<sub>2</sub>O (**c**) D<sub>2</sub>O, temperature 298 K.



Figure S3. Portions of the <sup>1</sup>H NMR spectra of compounds 1a, 2c, and 3d in DMSO- $d_6$  at 298 K.



Figure S4. The NOCV channels for the formation of the Ru–N bond in compounds 1a, 2c, and 3d with their corresponding contributions to the EDA orbital term ( $\Delta E$ ) and the charge transfer ( $\Delta \rho$ ), calculated at the ZORA level of theory (1c/PBE0/TZ2P/vacuum) and separated into  $\alpha$ - and  $\beta$ -spin parts. The iso-surface of the electron deformation density is plotted for the value 0.001 a.u. The accumulation and depletion of electron density are shown in blue and red, respectively.



**Figure S5**. Portion of the <sup>1</sup>H NMR spectrum of compound **2c** measured in DMSO- $d_6$  solvent at elevated temperatures (293-338 K). Note the <sup>1</sup>H NMR signals of products of decomposition (labeled \*), which appear clearly at temperatures of 323 K and above.



**Figure S6**. Portion of the <sup>1</sup>H-<sup>13</sup>C HSQC spectrum of compound **2c** measured in DMF- $d_7$  (left) or DMSO- $d_6$  (right) at 298 K.



**Figure S7**. Portion of <sup>1</sup>H NMR spectrum of compound **3d** measured in DMSO- $d_6$  at several elevated temperatures (298-318 K).



Figure S8. <sup>1</sup>H NMR spectrum of compound 2c in free and bound (with CB7) forms in DMSO- $d_6$  at 298 K. The signals of the complex 2c@CB7 that appear upon addition of the CB7 macrocycle to a solution of 2c are shown in bold.



Figure S9. <sup>1</sup>H NMR spectrum of compound 3d in free and bound (with CB7) forms in DMSO- $d_6$  at 298 K. The signals of the complex 3d@CB7 that appear upon addition of the CB7 macrocycle to a solution of 3d are shown in bold.

The NMR spectra for systems in the slow-exchange regime (Figures S8 and S9), with the signals of the free and bound forms make it possible to calculate the binding constant directly from the relative integrals of the NMR signals. The modified Job's plots<sup>1</sup> constructed from the NMR spectra of compounds 2c and 3d are shown in Figure S10 and S11, respectively.



Figure S10. Modified Job's plot for the mixing of compound 2c with CB7, see Figure S8.



Figure S11. Modified Job's plot for the mixing of compound 3d with CB7, see Figure S9.

The binding constants<sup>1</sup> estimated for guest compounds **2c**, **3d**, and some published data for several structurally related organic ligands with a CB7 host in DMSO are summarized in Table S1.

Compounds	Binding Constant (M <sup>-1</sup> ) in DMSO
2c	$(2.18\pm0.16) \times 10^3$
3d	$(1.59\pm0.83) \times 10^3$
Me-BiPy <sup>2</sup>	$(0.59\pm0.13) \times 10^3$
Hex-BiPy <sup>2</sup>	$(1.3\pm0.5) \times 10^3$
Bz-BiPy <sup>2</sup>	$(1.9\pm0.04) \times 10^3$

**Table S1**. The binding constants for a CB7 host with guest compounds 2c, 3d, and several structurally related organic ligands<sup>2</sup> in DMSO at room temperature.



**Figure S12**. Portion of the 2D <sup>1</sup>H-<sup>1</sup>H NOESY spectrum with cross-peaks between the cyclohexylmethyl-based (Cyhm) ligand of **2c** and CB7 in DMSO- $d_6$  at 298 K. The NOE contacts can result from the superposition of differently encapsulated states of the Cyhm moiety. Note that cross-peaks with H<sub>a</sub> are missing from the spectrum because of efficient paramagnetic relaxation and significant broadening of the NMR signal. The minor form of the **2c**@CB7 complex (labeled with \*, the corresponding cross-peaks are blurred) is assumed to have been produced by the solvolysis of **2c**.



**Figure S13**. The effects of increased and decreased rigid tiltin of the ruthenium guest (0 corresponds to the optimized geometry) relative to the CB7 portal without geometry reoptimization on the  $\delta^{PD}$  values (the g-tensor was not recalculated) of 2c@CB7 (left) and 3d@CB7 (right). Ha is more shielded in the more horizontal orientation of the ruthenium core (CB7 buried deeper in the [red] shielding region) and the effect is more pronounced for 3d@CB7 because the distance is smaller. Hb shows a tendency to transition, and the remaining protons of CB7 show a tiny increase (the small effect of the close [blue] deshielding region). The numerical SO-ZORA values for 2c@CB7 in the range  $\pm 5^{\circ}$  are shown in Table S2 below.

2c@CB[7]		5	0(6	eq)	+	5
	$\delta^{\scriptscriptstyle HF}_{\scriptscriptstyle DFT}$	$\delta_{DFT}^{HFa}$	$\delta^{\scriptscriptstyle HF}_{\scriptscriptstyle DFT}$	$\delta_{DFT}^{HFa}$	$\delta^{HF}_{DFT}$	$\delta_{DFT}^{HFa}$
На	-0.46	-0.28	-0.51	-0.31	-0.53	-0.33
Hb	-0.21	-0.14	-0.28	-0.18	-0.24	-0.16
Hc	-0.18	-0.11	-0.20	-0.13	-0.20	-0.13
Hb′	-0.16	-0.09	-0.16	-0.10	-0.16	-0.10
Ha'	-0.16	-0.09	-0.17	-0.10	-0.17	-0.10

**Table S2**. Hyperfine <sup>1</sup>H NMR shifts calculated for 2c@CB7 from the equilibrium position obtained by a rigid tilting of  $\pm 5^{\circ}$ . The geometry has not been re-optimized and the g-tensor has not been recalculated.

# X-ray analysis of [Cyhm-DABCO-H]<sup>+</sup> *trans*-[RuCl4(DMSO-S)(ACN)]<sup>-</sup> obtained from the crystallization of 1c

The significantly reduced stability of the Ru–N bond in compounds of type **1** can be supported indirectly by observing the molecular structure shown in Figure S14, which formed during our attempts to prepare high-quality monocrystals by slowly evaporating the organic solvent from a reaction solution of **1c** in acetonitrile, see Table S3.



**Figure S14**. Molecular structure of Ru-complex [Cyhm-DABCO-H]<sup>+</sup> *trans*-[RuCl<sub>4</sub>(DMSO-S)(ACN)]<sup>-</sup> (Br<sub>0.36</sub> Cl<sub>0.64</sub>)<sup>-</sup> obtained from the crystallization of **1c**.

CCDC No	1998055
chemical formula	$C_{17}H_{35}$
	$Br_{0.4}Cl_{4.6}N_3ORuS$
formula weight	623.94
crystal system	Monoclinic
space group	C2/c
<i>a</i> (Å)	20.8698 (4)
<i>b</i> (Å)	12.1608 (2)
<i>c</i> (Å)	22.7784 (5)
$\alpha$ (deg)	90
$\beta$ (deg)	116.647 (2)
γ (deg)	90
$V(\text{\AA}^3)$	5166.99 (19)
Ζ	8
$D_{\text{calcd.}}$ (g cm <sup>-3</sup> )	1.604
$\mu \text{ (mm}^{-1}\text{)}$	1.741
measured/unique	15549/4890
reflections	
data/parameters/restrai	4890/278/19
nts	
$R_1/wR_2 [I > 2\sigma(I)]$	0.0424/0.1212
$R_1/wR_2$ [all data]	0.0463/0.1246
GoF	1.049
$\Delta  ho_{ m max}/\Delta  ho_{ m min}$ (e Å <sup>-3</sup> )	2.013/-0.689

**Table S3**. Selected crystallographic data for the Ru(III) complex  $[Cyhm-DABCO-H]^+$  *trans*- $[RuCl_4(DMSO-S)(ACN)]^-$  (Br<sub>0.36</sub> Cl<sub>0.64</sub>)<sup>-</sup> formed during the crystallization of **1c**.

#### References

- K. Hirose: A Practical Guide for the Determination of Binding Constants. J. Incl. Phenom. Macrocycl. Chem., 2001, 39, 193–209. http://dx.doi.org/10.1023/A:1011117412693.
- A. Thangavel, A. M. M. Rawashdeh, C. Sotiriou-Leventis, N. Leventis: Simultaneous Electron Transfer from Free and Intercalated 4-Benzoylpyridinium Cations in Cucurbit[7]uril. *Org. Lett.*, 2009, 11, 1595–1598. https://doi.org/10.1021/ol9002459.

# Cartesian coordinates for DFT optimized HG complexes

## 2c@CB7

181

Ru	11.37203	1.46436	-1.02033
Cl	11.19974	3.82817	-1.23518
Cl	11.61014	1.61005	1.31015
Cl	11.49657	-0.89844	-0.78388
Cl	11.08464	1.13996	-3.40379
S	13.63065	1.65969	-1.35095
0	14.42001	0.41634	-1.49369
Ν	9.28118	1.36108	-0.69595
Ν	2.32152	1.95683	-0.12533
С	14.40105	2.65066	-0.08374
Н	15.46244	2.75069	-0.34593
Н	14.26256	2.11567	0.86303
Н	13.88599	3.61956	-0.04409
С	14.02810	2.66716	-2.77130
Н	15.12231	2.74220	-2.81521
Н	13.53801	3.64318	-2.66397
Н	13.61821	2.13489	-3.63839
С	8.49019	0.46989	-1.33139
Н	9.00616	-0.24239	-1.97625
С	7.10973	0.47077	-1.18234
Н	6.50858	-0.27941	-1.70430
С	6.51016	1.40404	-0.35747
С	7.33635	2.29242	0.33656
Н	6.93066	3.06054	0.99600
С	8.70101	2.23737	0.12564
Н	9.37285	2.93571	0.62540
С	5.03895	1.53996	-0.25420
С	4.22802	1.33317	-1.36797
Н	4.62310	0.99822	-2.32687
С	2.88056	1.55863	-1.28431
Н	2.22577	1.45965	-2.15009
С	4.43142	1.94559	0.92664
Н	4.99579	2.07945	1.84877
С	3.06873	2.14647	0.96829
Н	2.55574	2.45135	1.88353
С	0.84075	2.08855	-0.04533
Н	0.53716	2.69473	-0.90850
Н	0.62153	2.65383	0.86968
С	0.18223	0.71840	-0.04455

Н	0.49017	0.19	193	-0.96881
С	0.58837	-0.13	3248	1.15041
Η	0.30873	0.40	155	2.07837
Η	1.68201	-0.27	206	1.18412
С	-0.11105	-1.49	0162	1.12499
Н	0.23899	-2.06	6769	0.25038
Н	0.18312	-2.07	/159	2.01201
С	-1.61237	-1.33	3397	1.06520
Н	-1.98146	-0.87	921	1.99988
Н	-2.07905	-2.32	2979	1.00246
С	-2.03497	-0.47	/185	-0.11479
Н	-1.80009	-0.99	0316	-1.06068
Н	-3.12380	-0.30	)844	-0.11097
C	-1.34233	0.87	546	-0.09191
H	-1.66613	1.44	829	0.79631
Н	-1.63231	1.47	/813	-0.96934
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0	1.01316	3.44158	-3.123	82
0	-4.28917	0.59626	-3.518	347
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0	-3.83548	0.51610	3.216	542
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0	-1.68160	-2.58883	4.400	)88
Ν	3.89651	-3.54393	1.111	90
Ν	1.87269	-4.86632	1.457	21
Ν	1.59235	-4.13194	3.516	64
Ν	3.34628	-2.49584	3.000	193
Ν	3.10553	-3.00582	-3.278	333
Ν	0.96817	-4.09259	-2.919	943
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Ν	0.94446	0.36606	-5.247	'65
Ν	-1.19627	-0.78077	-5.202	236
Ν	-0.08083	-2.56281	-4.475	500
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Ν	-1.23343	3.93722	-3.369	931
Ν	-3.32439	2.69541	-3.417	735
Ν	-2.52086	1.23325	-4.870	)73
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Ν	-1.75643	5.01331	1.00179
Ν	-3.72534	3.55571	1.01189
Ν	-3.79300	3.60061	-1.20995
Ν	-1.73870	4.90670	-1.21425
Ν	-0.16953	2.83327	4.70167
Ν	-2.06831	1.29765	4.49312
Ν	-3.20678	2.73396	3.22940
Ν	-1.15622	4.06452	3.13272
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Η	3.84388	-5.47990	1.93098
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Η	3.61413	-4.29340	4.07830
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Н	5.39507	-3.03411	-0.21137
Н	5.06855	-4.80082	-0.05129
С	1.57623	-5.62614	0.25772
Н	2.25996	-6.48811	0.18355
Н	0.53894	-5.97189	0.36115
С	3.97534	-2.75079	-2.25509
С	2.39028	-4.25583	-3.10719
Н	2.60369	-4.94364	-3.94250
С	0.55737	-4.49211	-1.67200
С	2.88679	-4.75953	-1.72599
Н	3.40271	-5.73336	-1.75393
С	3.20125	-2.34152	-4.55655
Н	4.09247	-1.70066	-4.51861
Н	3.31125	-3.09211	-5.35599
С	0.05112	-3.93343	-4.02779
Н	-0.94045	-4.25428	-3.68143
Н	0.38605	-4.56500	-4.86781
С	2.11040	-0.12960	-4.73069
С	0.09752	-0.65410	-5.82384
Н	-0.00624	-0.50765	-6.91199
С	-1.29127	-1.92032	-4.44954
С	0.83436	-1.96542	-5.42902
Н	1.02300	-2.64928	-6.27331
С	0.72367	1.77763	-5.44539
Н	0.48397	1.97480	-6.50331
Н	1.65388	2.28996	-5.16707
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Η	-3.24614	-0.61611	-5.39174
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Η	-2.83392	4.30759	-4.68031
С	-3.45191	1.41757	-3.88565
C	-1.71125	2.41218	-5.10057
Н	-1.72393	2.69585	-6.16554
С	-1.25325	5.12534	-2.55466
H	-1.87203	5.88732	-3.05625
Н	-0.22018	5.48646	-2.46237
С	-4.29713	3.28794	-2.52731
Н	-4.69834	4.20831	-2.98359
Н	-5.09989	2.54908	-2.40226
С	-0.95233	5.09348	-0.10543
С	-3.15309	4.85206	0.67081
Н	-3.75316	5.66426	1.11353
С	-4.18959	2.89990	-0.10008
С	-3.15275	4.84212	-0.88394
Н	-3.69159	5.69227	-1.33559
С	-1.27898	5.27176	2.33761
Η	-1.95768	5.97884	2.84169
Н	-0.27506	5.70794	2.24749
С	-4.22371	3.24288	2.34598
Η	-4.67579	4.14530	2.78787
Η	-4.98123	2.45652	2.22889
С	0.04900	3.68719	3.66105
С	-1.57934	2.60162	4.93281
Η	-1.83508	2.77279	5.99129
С	-3.10749	1.41754	3.60042
С	-2.26440	3.57547	3.94955
Η	-2.78567	4.41569	4.43636
С	0.90137	2.26411	5.47867
Η	0.64562	2.32408	6.54781
Η	1.79893	2.86152	5.27275
С	-1.95085	0.10493	5.31034
Η	-2.05673	0.38232	6.37421
Η	-2.76321	-0.57391	5.01794
С	2.31225	0.51811	4.44694
С	1.45958	-1.42134	5.43980
Η	1.98381	-1.82575	6.32205
С	-0.70861	-1.94533	4.75475
С	0.53033	-0.22950	5.74793
Н	0.37874	-0.04685	6.82558
С	3.59660	-1.54649	4.06416
Η	4.28825	-0.78797	3.67262

4.05176	-2.07044	4.92253
0.91955	-3.80730	4.75489
-0.01636	-4.38104	4.76986
1.55830	-4.09964	5.60233
	4.05176 0.91955 -0.01636 1.55830	4.05176-2.070440.91955-3.80730-0.01636-4.381041.55830-4.09964

## 3d@CB7

182

Ru	-2.80289	9.73819	1.12516
S	-3.75261	11.78346	1.47027
С	-2.51710	13.00769	1.83493
Cl	-3.90338	9.45602	-0.91817
Cl	-0.98352	10.72690	0.00003
Cl	-1.65316	9.95107	3.17947
Cl	-4.57389	8.64621	2.20982
Ν	-1.89633	7.85858	0.83872
С	-1.38369	7.51609	-0.34491
С	-0.75701	6.30234	-0.55165
С	-0.64199	5.40766	0.50006
С	-1.18365	5.76741	1.72688
С	-1.80512	6.99202	1.85229
С	-4.64455	11.82008	3.00642
0	-4.63544	12.35614	0.43658
Н	-3.03823	13.92709	2.09878
Н	-1.91819	13.13228	0.93663
Н	-1.89635	12.64210	2.65060
Н	-5.49791	11.15629	2.89716
Η	-4.95718	12.84910	3.17913
Н	-3.98793	11.45672	3.79507
Н	-1.14117	5.10194	2.58032
Н	-2.24526	7.30701	2.78595
Η	-0.36527	6.06646	-1.53280
С	0.04102	4.08744	0.31506
Н	-1.48856	8.24069	-1.13779
Ν	-0.93840	2.98204	0.47312
С	-0.42955	1.58002	0.22900
С	-0.14917	1.38511	-1.25586
С	0.83501	1.30878	1.03257
С	-1.54135	0.63514	0.67853
Н	-1.04538	1.61256	-1.83599
С	0.28338	-0.06272	-1.49092
Η	0.63884	2.06097	-1.59669
С	1.26527	-0.14086	0.79762

Η	0.65142	2 1.48	3903	2.09371
Η	1.6430	1 1.97	/510	0.72281
Н	-1.3385	0 3.03	3826	1.41782
С	-1.1083	4 -0.81	1001	0.43927
Η	-2.4624	3 0.84	954	0.12754
Η	-1.7503	9 0.79	0147	1.74262
С	-0.8313	4 -1.00	)673	-1.04940
Η	-1.73792	2 -0.81	1354	-1.62985
Η	-0.5432	1 -2.04	4166	-1.24879
С	1.5487	5 -0.35	5151	-0.68769
Η	1.8724′	7 -1.38	8110	-0.86220
Н	2.36032	2 0.30	)489	-1.01428
Н	0.48382	2 -0.19	9736	-2.55616
С	0.1569	1 -1.08	3771	1.24733
Н	-1.9121	9 -1.47	7596	0.76092
Н	0.4753	1 -2.12	2414	1.11184
Н	-0.0399	5 -0.95	5384	2.31478
Н	2.1734	1 -0.33	3009	1.37381
Н	-1.7318	8 3.15	5642	-0.16929
Н	0.47494	4 4.01	141	-0.67930
Н	0.82408	8 3.94	740	1.05878
Н	6.29579	0.62138	-2.8997	76
Н	6.77635	-1.05870	-1.1360	51
Н	5.22108	-0.88097	-4.5200	54
Н	5.30561	2.00797	-2.3724	12
С	5.34859	0.92187	-2.4405	58
Н	5.81596	-2.57805	-2.8178	87
С	5.77556	-0.89158	-0.7280	00
Н	6.70965	-1.01225	1.2898	36
С	4.26231	-0.73411	-4.0149	93
Ν	5.29507	0.40981	-1.1040	)0
Ν	4.26809	0.50479	-3.2847	74
Н	3.33829	-0.66723	-6.0390	52
С	4.81981	-2.56904	-2.3652	24
С	5.04556	1.21619	-0.0245	54
С	5.73459	-0.85224	0.8204	7
С	3.06572	-0.57104	-4.9847	71
С	3.32074	1.38862	-3.7351	.3
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Ν	3.91619	-1.90160	-3.254	18
Ν	4.91327	-1.99324	-1.0550	55
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Ν	5.25423	0.47064	1.1068	34
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Ν	2.59336	0.75591	-4.70956
Η	5.23502	2.13651	2.28535
С	5.26170	1.05550	2.41400
Н	1.77184	2.51248	-5.35183
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N	2.17916	-1.63624	-4.59905
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C	4.67279	-2.43204	2.48504
Č	1.03535	-2.02758	-5.36800
Õ	2.22508	-3.46647	-3.19871
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N	4.14126	0.69725	3.23308
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N	0.26803	1 13795	-5 22391
н	-0 26934	-0.08254	-6 82459
C	4 07040	-0 50809	4 01141
N	3 73326	-1 70398	3 28725
$\hat{\mathbf{C}}$	-0.43033	-0.00282	-5 74560
$\tilde{0}$	3 04557	2 72323	3 12440
N	-0 14801	-1 26056	-5 10539
C	3 15856	1.20050	3 55534
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Č	-0.56216	2.02862	-4 59744
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C	-1.26239	-1.82218	-4.53594
C	-1.90411	0.26815	-5.35549
N	-1 82132	1 48049	-4 58480
Н	-2.56653	0.40727	-6.21509
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C	1.33735	1.73433	5.18395
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Η	-3.65648	2.31951	-5.05915
С	0.88888	-1.80775	5.45111
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Η	-3.11981	3.36168	4.23234
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Ν	-5.22760	-0.46249	-1.22794
С	-4.09526	2.57834	2.07944
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Η	-4.91874	3.60320	-0.01390
С	-3.38506	2.31866	4.39777
Ν	-4.11949	1.88388	3.25212
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Η	-2.73049	0.49240	6.32632
С	-4.57896	-1.22890	2.03740
Ν	-4.99359	1.99901	1.23265
Η	-6.51153	1.03590	-1.92388
Ν	-4.34075	-0.53223	3.19587
С	-3.82170	-1.17199	4.37302
Η	-3.83036	-2.24621	4.19432
Ν	-5.28586	-0.41154	1.19151
Η	-4.02219	2.24196	5.28229

С	-5.87524	-0.89899	-0.02454
Η	-5.81617	-1.98613	-0.00625
Η	-6.41042	2.63212	-0.13223
С	-4.99832	0.74843	3.21309
С	-5.63557	0.84362	1.80035
Η	-4.47787	-0.93849	5.21714
Η	-6.92495	-0.59173	-0.05093
Η	-5.72352	0.79657	4.02905
Η	-6.72046	0.97454	1.81317