

Supporting Information

An Endoplasmic Reticulum Targeted Ratiometric Fluorescent Probe for Sensing of Hydrogen Sulfide in Living Cells and Zebrafish

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1. Experimental reagents and instruments

All other chemicals were obtained from commercial suppliers and used without further purification. Silica gel (200-300 mesh, Qingdao Haiyang Chemical Co.) was used for column chromatography. ^1H NMR and ^{13}C NMR spectra were recorded on a Bruker Advance at 400MHz or at 100 MHz, δ values are in parts per million relatives to TMS in $\text{DMSO-}d_6$. Mass spectra (MS) were measured with Bruker Apex IV FTMS using electrospray ionization (ESI). Absorption spectra were recorded on a Purkinje TU-1901 spectrophotometer. Fluorescence measurements were taken on a Hitachi F-7000 fluorescence spectrometer with a 10mm quartz cuvette. pH measurements were carried out with a pH acidometer (Mettler Toledo FE-30). Fluorescence imaging was observed under an Olympus IX81 confocal fluorescence microscope.

HeLa cells were cultured in high-glucose DMEM supplemented with 10% fetal bovine serum, 1% penicillin, and 1% streptomycin with 5% CO_2 at 37°C .

4-day-old zebrafish were obtained from Shanghai Feixi Biotechnology Co., Ltd.

2. Normalized spectra of probe M- H_2S to H_2S

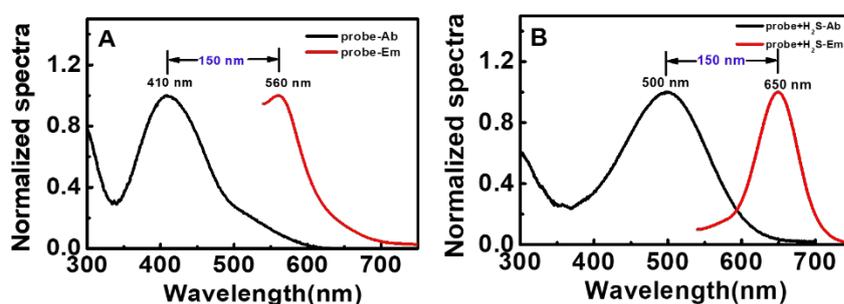


Fig. S1. (A) Normalized UV-*vis* absorption spectra and fluorescence spectra of probe **M- H_2S** .

(B) Normalized UV-*vis* absorption spectra and fluorescence spectra of **M- H_2S** recognized after H_2S .

$\lambda_{\text{ex}} = 480 \text{ nm}$, slit width: $d_{\text{ex}} = d_{\text{em}} = 10 \text{ nm}$, 10 mM PBS, pH 7.4, containing 10% DMSO, v/v at

room temperature.

3. The stability of M-H₂S at different pH values

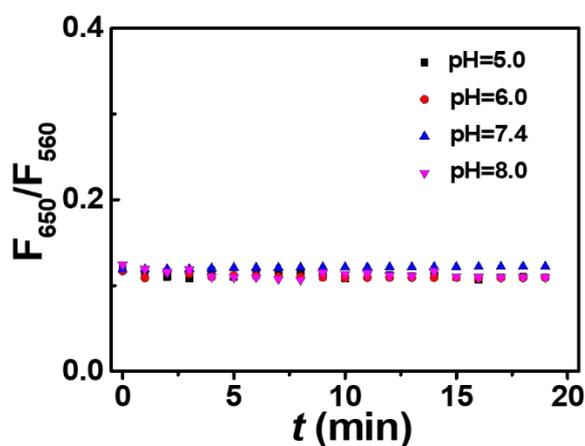


Fig. S2. Fluorescent intensity (F_{650}/F_{560}) responses of **M-H₂S** (10 μ M) at different pH values (pH=5.0, 6.0, 7.4, 8.0). Spacing interval is 1 min.

4. Time response spectra of probe M-H₂S to H₂S

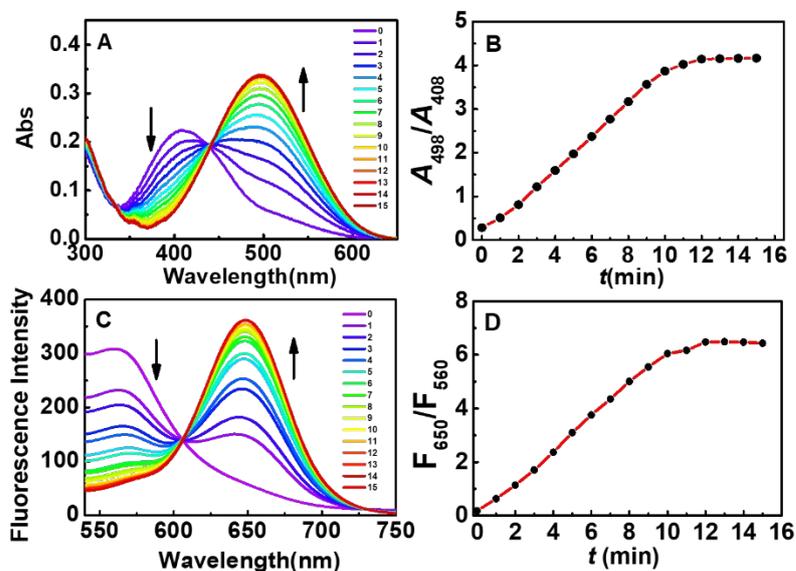


Fig. S3. UV-*vis* absorption A) and ratio intensity changes B) of **M-H₂S** (10 μ M) upon addition of H₂S (50 μ M). Fluorescence spectra C) and ratio intensity changes D) of **M-H₂S** (10 μ M) upon addition of H₂S (50 μ M).

5. Determination of the Detection Limit

According to IUPAC, the detection limits were determined based on the fluorescence titrations, carried out in PBS / DMSO (9:1, v/v), pH 7.4, using the following equation:

$$\text{Detection limit} = 3\sigma/k$$

where σ is the standard deviation of blank measurements and k is the slope of the plot of fluorescence intensity vs H_2S concentration. The standard deviations $\sigma = 0.0011$.

6. The capabilities of M-H₂S for detecting H₂S at different pH

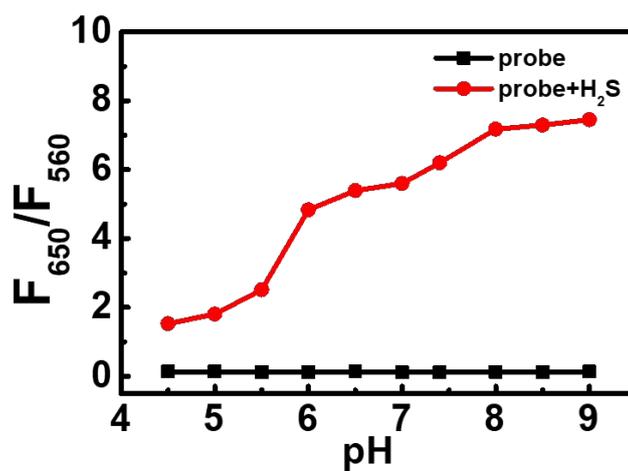


Fig. S4. Fluorescence responses of M-H₂S (10 μM) in the absence and presence of H₂S (50 μM) at different pH.

7. High-resolution mass spectra of reaction system

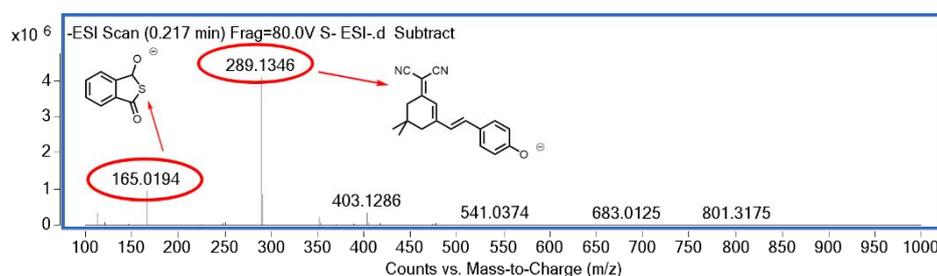


Fig. S5. The HRMS of the product after the reaction between M-H₂S and H₂S.

8. Cytotoxicity assay

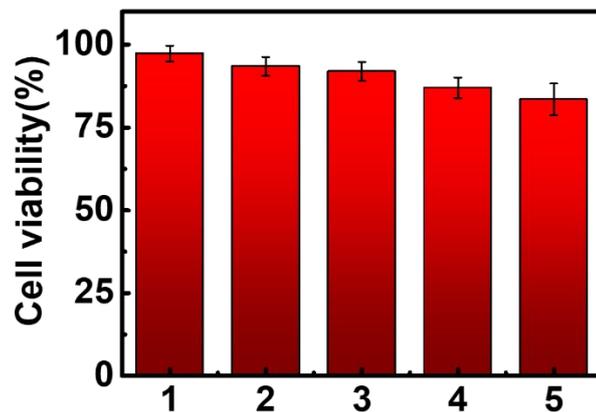


Fig. S6. MTT assay for estimating cell viability (%) of HeLa cells. The concentrations of probe

M-H₂S were used: 1: blank, 2: 5 μM, 3: 10 μM, 4: 15 μM, 5: 20 μM, respectively.

9. Characterization data

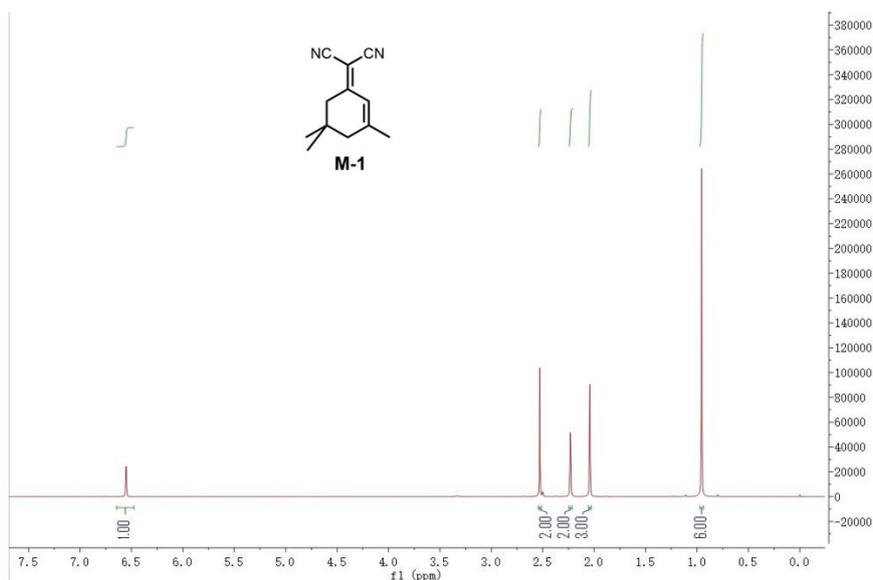


Fig. S7. ¹H NMR spectral of compound **M-1**.

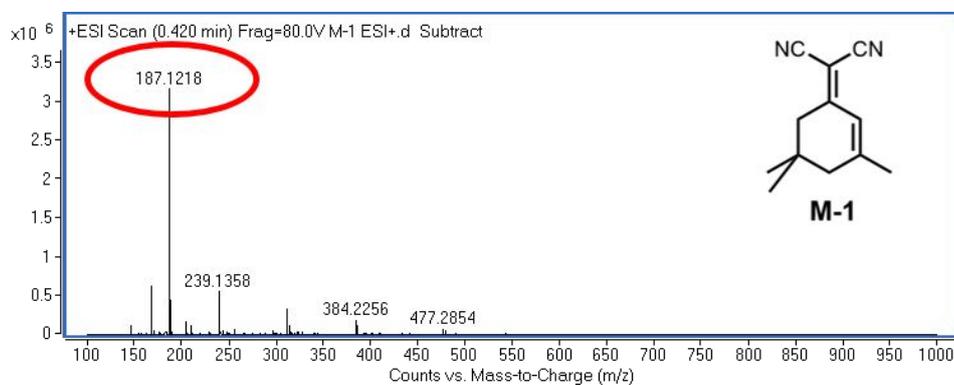


Fig. S8. HRMS spectral of compound **M-1**.

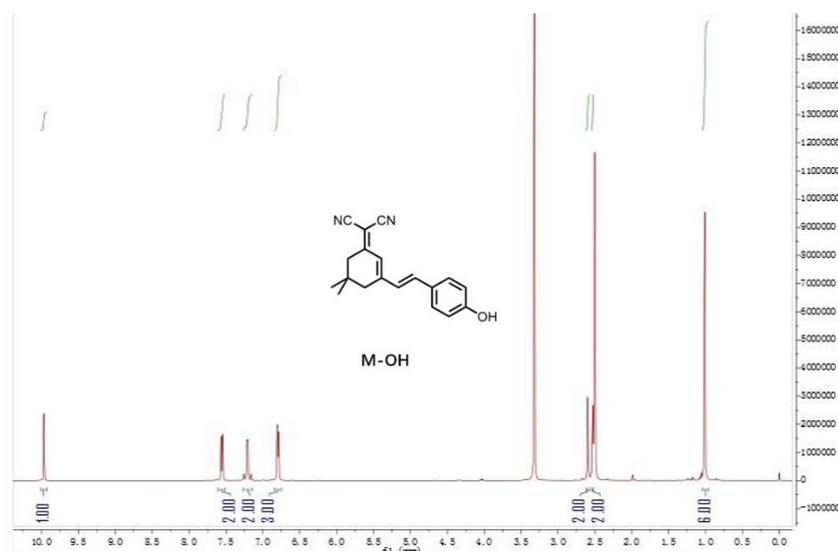


Fig. S9. ^1H NMR spectral of compound **M-OH**.

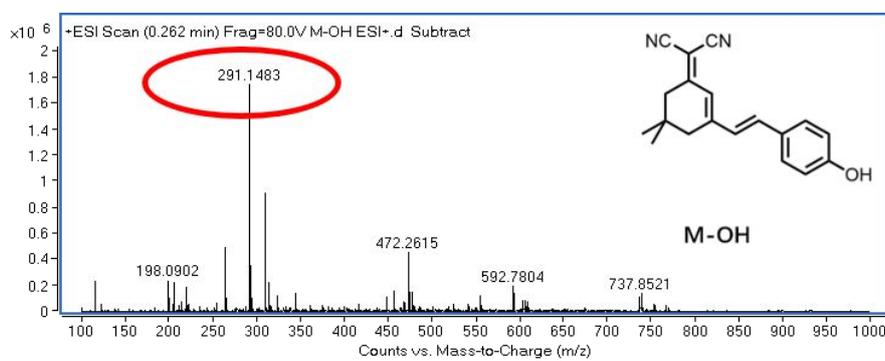


Fig. S10. HRMS spectral of compound **M-OH**.

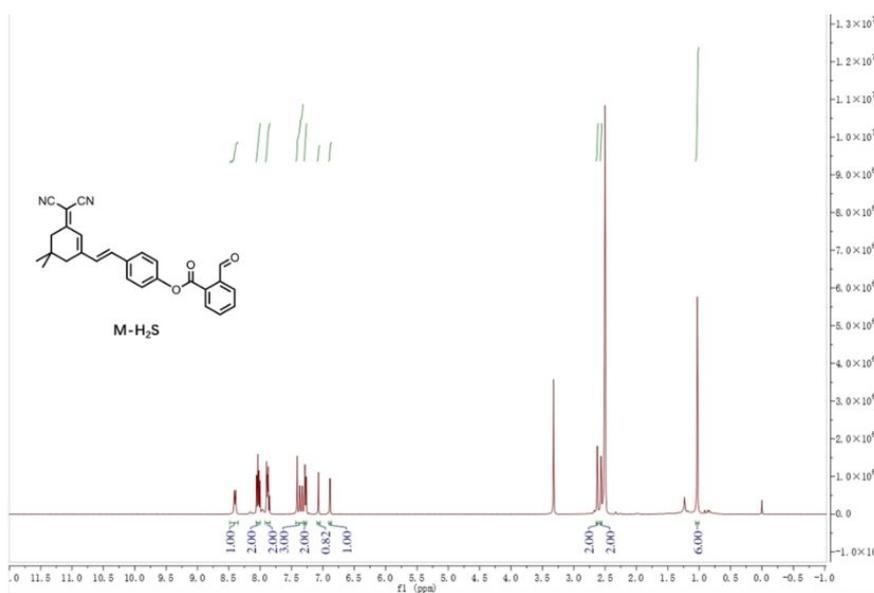


Fig. S11. ^1H NMR spectral of compound **M-H₂S**.

