Supporting Information

Pt Nanoparticles Supported on Mesoporous Graphitic Carbon Nitride as Catalysts for Hydrolytic Dehydrogenation of Ammonia Borane

Merve Aksoy^a, Önder Metin^{a,b,c}*

^aDepartment of Chemistry, College of Sciences, Koç University 34450, Istanbul, Turkey. ^bKoç University TÜPRAŞ Energy Center (KUTEM), 34450 Sarıyer, Istanbul, Turkey. ^cKoç University Surface Science and Technology Center (KUYTAM), 34450 Sarıyer, Istanbul, Turkey

e-mail: <u>ometin@ku.edu.tr</u>



Figure S1. TEM images of mpg-CN/Pt nanocomposites with different experimental Pt loadings (a) 5.94 wt%, (b) 3.82 wt%, (c) 2.08 wt% and (d) 1.16 wt% Pt.



Figure S2. XPS survey spectrum of mpg-CN/Pt nanocatalysts (3.82 wt% Pt loading).



Figure S3. Pt 4f_{7/2} XPS spectra of mpg-CN/Pt nanocatalysts reduced by Sodium Borohydride instead of Ammonia Borane.



Figure S4. Pt 4f_{7/2} XPS spectra of mpg-CN/Pt (IV) composite (3.82 wt% Pt loading) before AB dehydrogenation, after AB dehydrogenation, together with the same catalyst after 10th run reusability test.



Figure S5. Pt 4f XPS spectra of rGO/Pt (0) nanocatalysts reduced by AB.



Figure S6. High-resolution XPS spectra of N1s region of mpg-CN, mpg-CN/Pt (IV) composite and mpg-CN/Pt nanocatalysts (3.82 wt% Pt loading)



Figure S7: Plot of volume of H₂ generated from AB hydrolysis catalyzed by mpg-CN/Pt nanocatalysts (3.82 wt% Pt loading) versus time at reaction conditions: [AB] = 50-200 mM, T = 25 °C, [Pt] = 0.2 mM (a) and logarithmic plots of H₂ generation rate versus AB concentrations (b).

Pt loading experimentally	Pt loading by ICP-MS		
(w/w %)	(w/w %)		
10	5.94		
5	3.82		
2.5	2.08		
1.25	1.16		

Table S1. Pt loading weight percent (w/w %) of mpg-CN/Pt composites experimentally and
by ICP-MS.

Table S2. The corresponding binding energy values for each nanocomposite.

N groups	mpg-CN	mpg-CN/Pt (IV)	mpg-CN/Pt (0)
C=N-C	398.47	398.58	398.68
N(-C-) ₃	399.87	399.98	399.98
(-C-NH ₂)	400.97	401.08	401.08

Table S3. Catalytic activity of reported platinum based catalysts used for the hydrolysis of AB together with the percentage of initial catalytic activity maintained after the reusability test.

Entry	Catalyst	E _a (kJ/mol)	TOF (min ⁻¹)	Cycle number	Maintaining initial Catalytic Activity	Ref.
1	SiO ₂ @Pt _{0.1} Co _{0.9}	37.05	25.59	5	70	1
2	Pt ₁₂ Ni ₄₈ DENS		32			2
3	Pt(8%)/CCF-500	39.2	35.0			3
4	SNP-Pt ₆₅ Ti ₃₅	39.4	51.4	5	63	4
5	$NP-Pt_{70}Ru_{30}$	38.9	59.6	5	70	5
6	Pt-CeO ₂ /rGO NCs	64.7	93.8	10	92	6
7	Cu ₅₀ Pt ₅₀ NPs	36	102.5	3	75	7
8	hnp–Pt ₃₅ Cu ₆₅	40.5	108	5	68	8
9	Pt/C		111			9
10	PC3/SAG	30.2	123.1	2	100	10
11	PdPt @PVP NPs	51.7	125.0	5	61	11
12	NP–Pt ₄₀ Co ₆₀ /Co ₃ O ₄ composite	38.8	131	5	65	12
13	Pt@PC-POP	56.4	133.2 (at 35 °C)	8	96	13
14	Pt/CNT-G	35.34	135			14
15	Pt@CF-12	30.7	139.3	4	100	15
16	Pt@SiO ₂		158.6	5	100	16
17	Pt/CeO ₂		182			17
18	mpg-CN/Pt (3.82 %)	40.7	202.89	10	78	This study
19	$Pt/\gamma -Al_2O_3$	21	222			18
20	mpg-CN/Pt (5.94 %)		274.18			This study
21	Pt–Ru@PVP NPs	56.3	308.0	5	72	19
22	Pt20/CNT (by ALD)	48.3	416.5	4	40	20
23	Pt-Co@PG	32.79	461.17	5	81.2	21
24	3 wt%Pt-3 wt%Ru/CNT	36	547			22
25	Pt/CNTs-O-HT		567 (at 30 °C)			23
26	PtCo20/CNTs	42.5	675.1			24
	(by ALD)					
27	Cu ₆₈ Pt ₃₂ /KB	39	859	5	75	25
28	PtNi/NiO	43	1240.3			26
29	Pt@NiO/Ni-CNT	24.9	2665			27
DENS=dendrimer-encapsulated nanoparticles CCE=cotton derived carbon fiber SNP=Stratified						

DENS=dendrimer-encapsulated nanoparticles, CCF=cotton derived carbon fiber, SNP=Stratified nanoporous, NC=nanocomposite, hnp= hierarchical nanoporous, PC/SAG=Platinum-Cobalt/ Mesoporous Silica Aerogel, PVP= Poly(N-vinyl-2-pyrrolidone), POP=Porous Organic Polymer, CNT-G= carbon nanotube-graphene, CF-12=Carboxylic acid Functionalized Cage-Type Mesoporous Silica FDU-12, CNTs-O-HT= carbon nanotubes acid oxidation and high temperature treated, ALD= Atomic Layer Deposition, PG= Nanoporous graphene, KB= Ketjen Black

Entry	Catalyst	Ea	TOF	Cycle	Maintained	
		(kJ/mol)	(min ⁻¹)	number	Activity	Ref.
1	Ni/g-C ₃ N ₄	36	18.7	4	75	28
2	Pd(0)/g-C ₃ N ₄ -CS	35.3	27.7 (at 30 °C)	8	72.9	29
3	Co/ g-C ₃ N ₄		55.6			30
4	mpg-C ₃ N ₄ /Pd	53.6	66.3	5	75	31
5	FeCo/g-C ₃ N ₄		68.2			30
6	CuCo/g-C ₃ N ₄		75.1			30
7	$Cu_{0.4}Co_{0.6}MoO_4/g-C_3N_4$	14.46	75.7			32
8	Ni _{0.5} Co _{0.5} O-NCN	43.18	76.1	6	83.2	33
9	$\begin{array}{c} mpg-C_3N_4 @Ag_{42}Pd_{58}-\\ AAt \end{array}$	28.2	94.1	5	70	34
10	mpg-CN/Pt (3.82 %)	40.7	202.89	10	77.5	This study
11	Pd ₇₄ Ni ₂₆ /MCN	54.1	246.8			35
12	Ag _{0.1} Co _{0.9} / g-C ₃ N ₄	40.91	249.02	5	52	36
13	mpg-CN/Pt (5.94 %)		274.18			This study
14	Rh/g-C ₃ N ₄	24.2	969	4	44.2	37
15	Au-Co@CN		2897			38
CS=Chitosan, NCN= Nitric-acid treated Carbon Nitride, AAt= Acetis acid treated, MCN= Mesoporous Carbon Nitride,						

Table S4: Catalytic activity of reported mpg-CN based catalysts used for the hydrolysis of AB together with the percentage of initial catalytic activity maintained after the reusability test.

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