Cobalt nanoparticles embedded in N-doped carbon nanotubes on the reduced graphene oxide as efficient oxygen catalyst for Zn-air battery

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Fig. S1. (a) Survey XPS spectra and high-resolution. XPS spectra of N 1s (b) and Co 2p (c) in Co@N-CNT/rGO-0.1.



Fig. S2. CV curves of Co@N-CNT/rGO-0.1 in N_2 - and O_2 -saturated 0.1 M KOH solution.



Fig. S3. LSV curves of Co@N-CNT/rGO-0.05, Co@N-CNT/rGO-0.2 and Co@N-CNT/rGO-0.1



Fig. S4. (a) The ORR polarization curves of Co@N-CNT/rGO-0.1 at initial and after 5000 CV cycles. (b) The OER stability of Co@N-CNT/rGO-0.1 under a constant voltage of 1.68 V (vs RHE)



Fig. S5. The TEM image of Co@N-CNT/rGO-0.1 after long-term cycling test.

Catalysts	$\Delta \mathbf{E} (\mathbf{V}) = \mathbf{E}_{\mathbf{j}=10} - \mathbf{E}_$	Liquid Zn–air battery			References
	E _{1/2}	Charge/disc harge voltage gap (V)	Current density (mA cm ⁻²)	Cycling time (h)	- -
ZnCo-ZIF@ GO	0.90	1.05	10	25	1
ZnCo ₂ O ₄ /N-C NT	0.78	0.84	10	5.3	2
MnO@Co-N/ C	0.93	0.75	5	633	3
CoNi/BCF	0.80	0.95	10	30	4
NiCo2O4/CN Ts	0.88	0.84	10	200	5
Co ₉ S ₈ /P@CS- 1:2	/	0.9	5	350	6
Co-N-CNTs	0.81	0.80	5	10	7
Co@NCNT/r GO-0.1	0.87	0.80	5	125	This work

Table. S1. Comparison of the performances of Zn–air batteries with recently reported representative bifunctional electrocatalysts.

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