

Supporting Information

Quenching of Singlet Oxygen by Carotenoids via Ultrafast Super-exchange Dynamics

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Rotation of π_g^* orbitals of O₂

There is arbitrariness in the rotation of the degenerate π_g^* orbitals (π_y^* and π_z^*) along the bond axis of isolated O₂. The ¹O₂ (¹Δ_g) CASSCF wavefunction with 45-degree rotated π_g^* orbitals is equivalent to the ¹O₂ (¹Δ_g') wavefunction, and vice versa, as shown below.

The 45 and −45 degree rotated π_g^* orbitals are expressed with the original π_y^* and π_z^* orbitals as follows.

$$\pi_{45}^* = \frac{1}{\sqrt{2}}\pi_x^* + \frac{1}{\sqrt{2}}\pi_y^*, \quad \pi_{-45}^* = \frac{1}{\sqrt{2}}\pi_x^* - \frac{1}{\sqrt{2}}\pi_y^*$$

In the ¹Δ_g configuration, either π_{45}^* or π_{-45}^* orbital is occupied by two electrons (electrons 1 and 2).

$$\begin{aligned} {}^1\Delta_g &= \frac{1}{\sqrt{2}}\pi_{45}^*(1)\pi_{45}^*(2) - \frac{1}{\sqrt{2}}\pi_{-45}^*(1)\pi_{-45}^*(2) \\ &= \frac{1}{\sqrt{2}}\left[\frac{1}{\sqrt{2}}\pi_x^*(1) + \frac{1}{\sqrt{2}}\pi_y^*(1)\right]\left[\frac{1}{\sqrt{2}}\pi_x^*(2) + \frac{1}{\sqrt{2}}\pi_y^*(2)\right] - \frac{1}{\sqrt{2}}\left[\frac{1}{\sqrt{2}}\pi_x^*(1) - \frac{1}{\sqrt{2}}\pi_y^*(1)\right]\left[\frac{1}{\sqrt{2}}\pi_x^*(2) - \frac{1}{\sqrt{2}}\pi_y^*(2)\right] \\ &= \frac{1}{\sqrt{2}}\left[\frac{1}{2}\pi_x^*(1)\pi_x^*(2) + \frac{1}{2}\pi_x^*(1)\pi_y^*(2) + \frac{1}{2}\pi_y^*(1)\pi_x^*(2) + \frac{1}{2}\pi_y^*(1)\pi_y^*(2)\right] \\ &\quad - \frac{1}{\sqrt{2}}\left[\frac{1}{2}\pi_x^*(1)\pi_x^*(2) - \frac{1}{2}\pi_x^*(1)\pi_y^*(2) - \frac{1}{2}\pi_y^*(1)\pi_x^*(2) + \frac{1}{2}\pi_y^*(1)\pi_y^*(2)\right] \\ &= \frac{1}{\sqrt{2}}\left[\pi_x^*(1)\pi_y^*(2) + \pi_y^*(1)\pi_x^*(2)\right] \end{aligned}$$

$\frac{1}{\sqrt{2}} \quad -\frac{1}{\sqrt{2}}$
 $\uparrow\uparrow \text{---} \uparrow\uparrow$
¹O₂ (¹Δ_g)

$\frac{1}{\sqrt{2}} \quad \frac{1}{\sqrt{2}}$
 $\uparrow \text{---} \uparrow \text{---} \uparrow \text{---} \uparrow$
¹O₂ (¹Δ_g')

The final expression corresponds to the ¹Δ_g' configuration consisting of the original π_y^* and π_z^* orbitals, where the respective orbitals are occupied by one electron.