

## **Supporting Information**

# Ultrasonic Assisted Synthesis of Size-Controlled Cu-Metal–Organic Framework Decorated Graphene Oxide Composite: Sustainable Electrocatalyst for the Trace-Level Determination of Nitrite in Environmental Water Samples

*P. Arul<sup>a</sup>, N.S.K. Gowthaman<sup>b</sup>, S. Abraham John<sup>a\*</sup> and Hong Ngee Lim<sup>b,c\*</sup>*

<sup>a</sup> Centre for Nanoscience and Nanotechnology, Department of Chemistry  
The Gandhigram Rural Institute  
Gandhigram-624 302, Dindigul, Tamilnadu, India

<sup>b</sup> Materials Synthesis and Characterization Laboratory, Institute of Advanced Technology,  
Universiti Putra Malaysia, 43400 UPM Serdang, Selangor, Malaysia

<sup>c</sup> Department of Chemistry, Faculty of Science, Universiti Putra Malaysia,  
43400 UPM Serdang, Selangor, Malaysia

\*Corresponding Authors

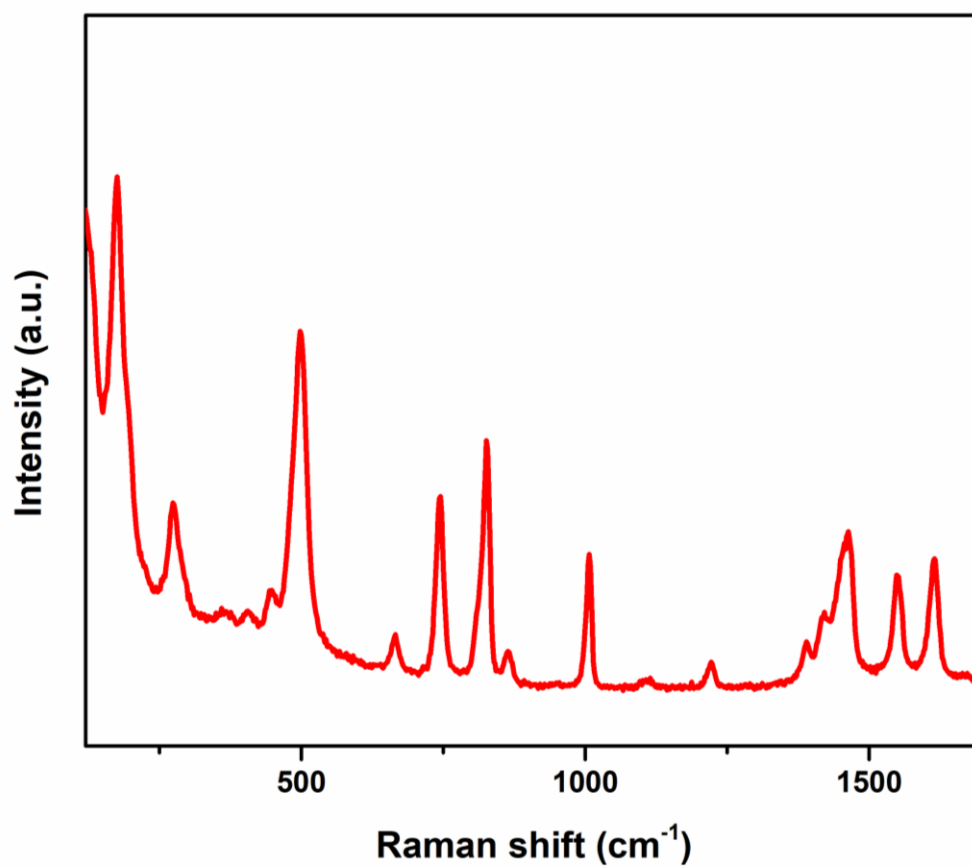
Email: [hongngee@upm.edu.my](mailto:hongngee@upm.edu.my) (H.N. Lim), [abrajohn@yahoo.co.in](mailto:abrajohn@yahoo.co.in) (S.A. John)

**Number of Pages: 11**

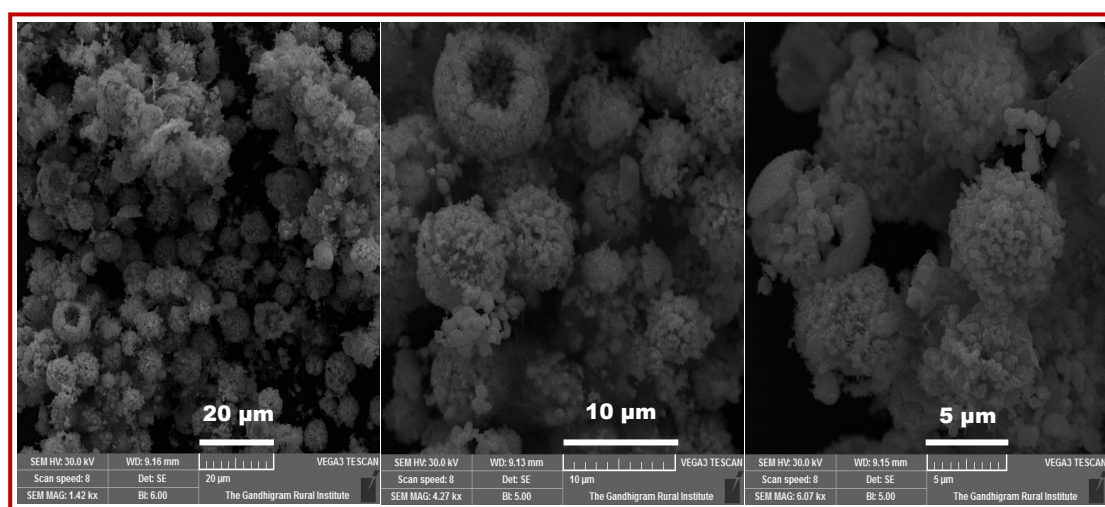
**Number of Figures: 8**

**Number of Tables: 2**

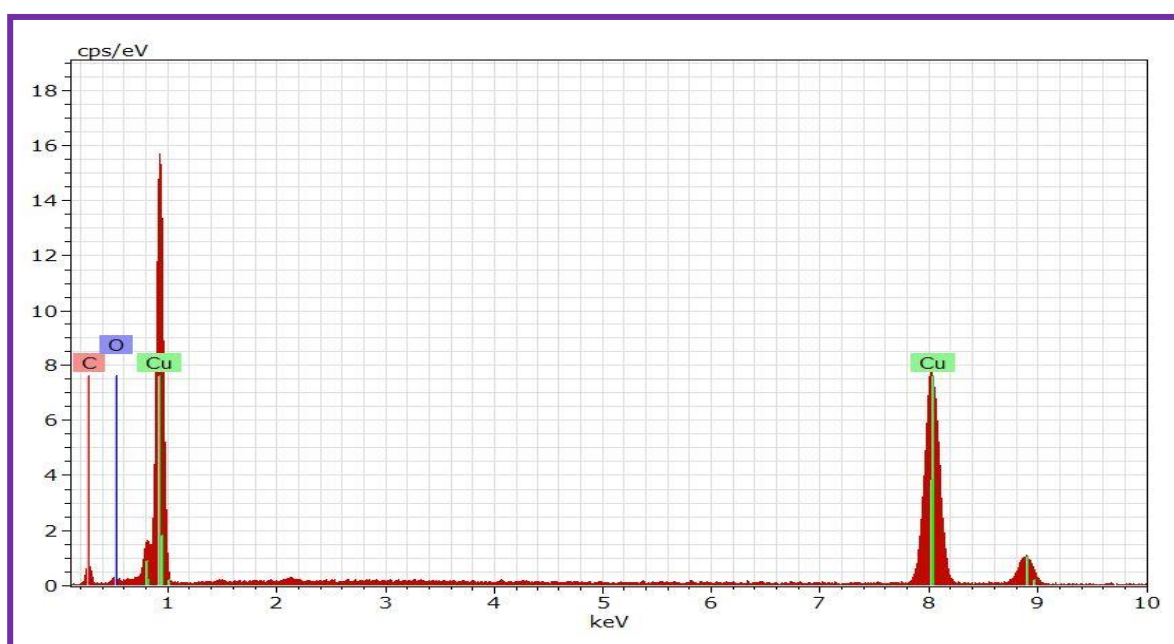
S.No.	Content	Page No.
I.	Additional characterizations and results	S2 to S11



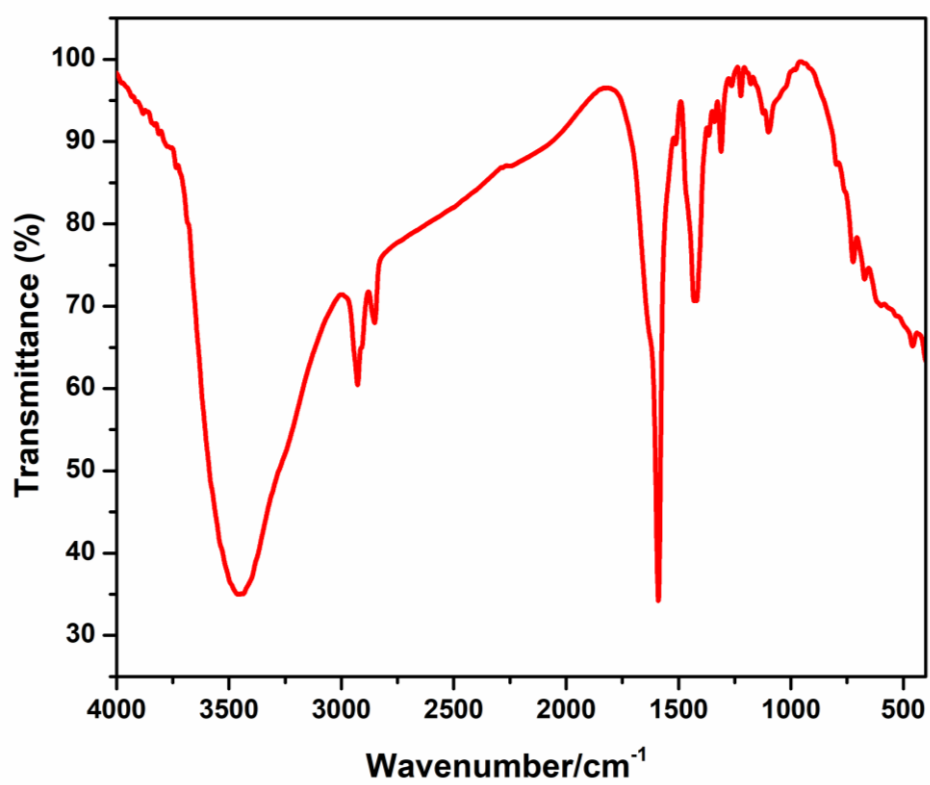
**Figure S1.** Raman spectrum of Cu-MOF-GO.



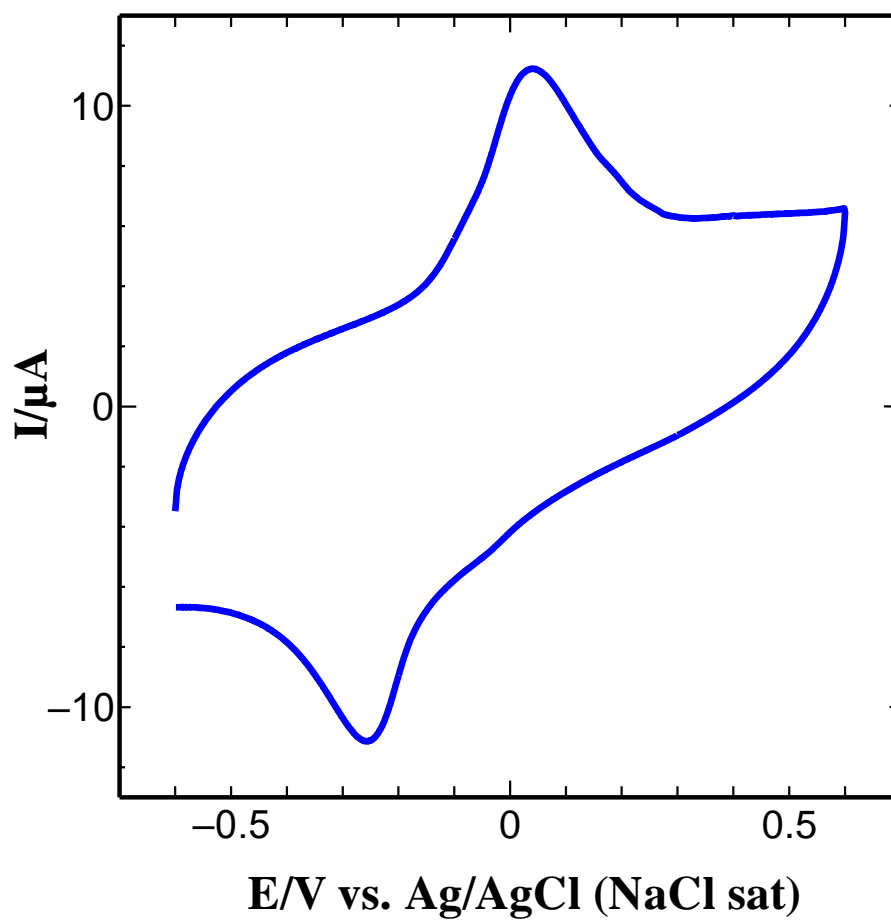
**Figure S2.** SEM images of Cu-MOF-GO in absence of ultrasonication.



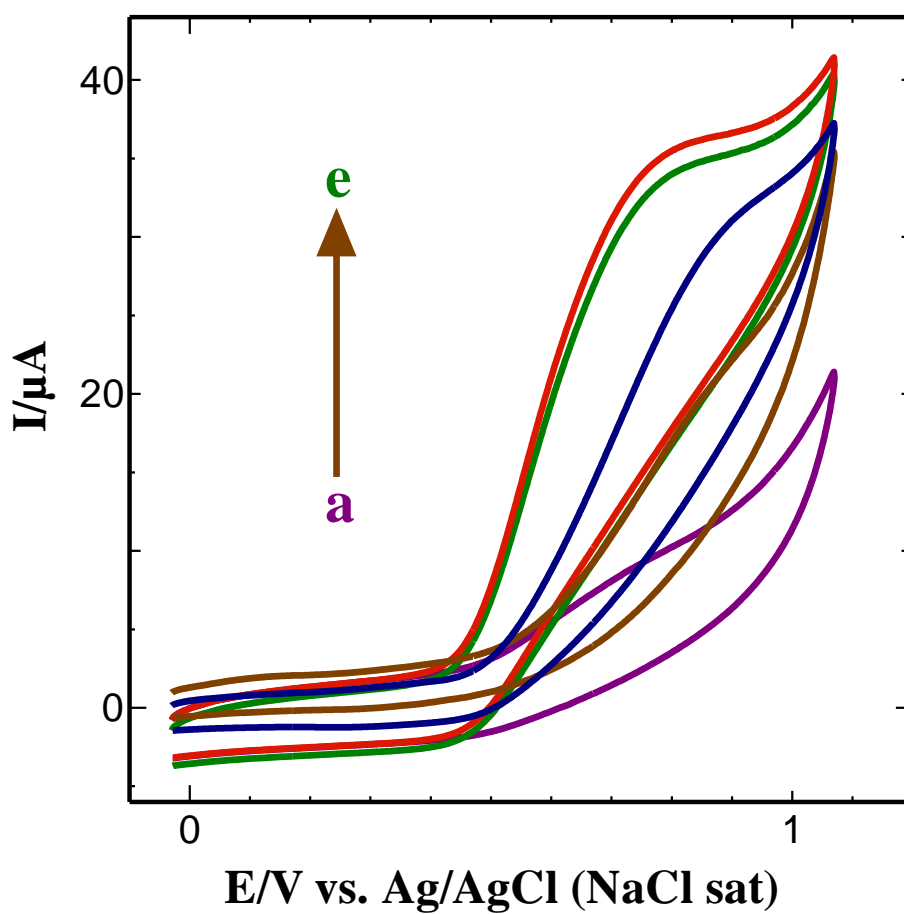
**Figure S3.** EDX spectrum of solid Cu-MOF-GO.



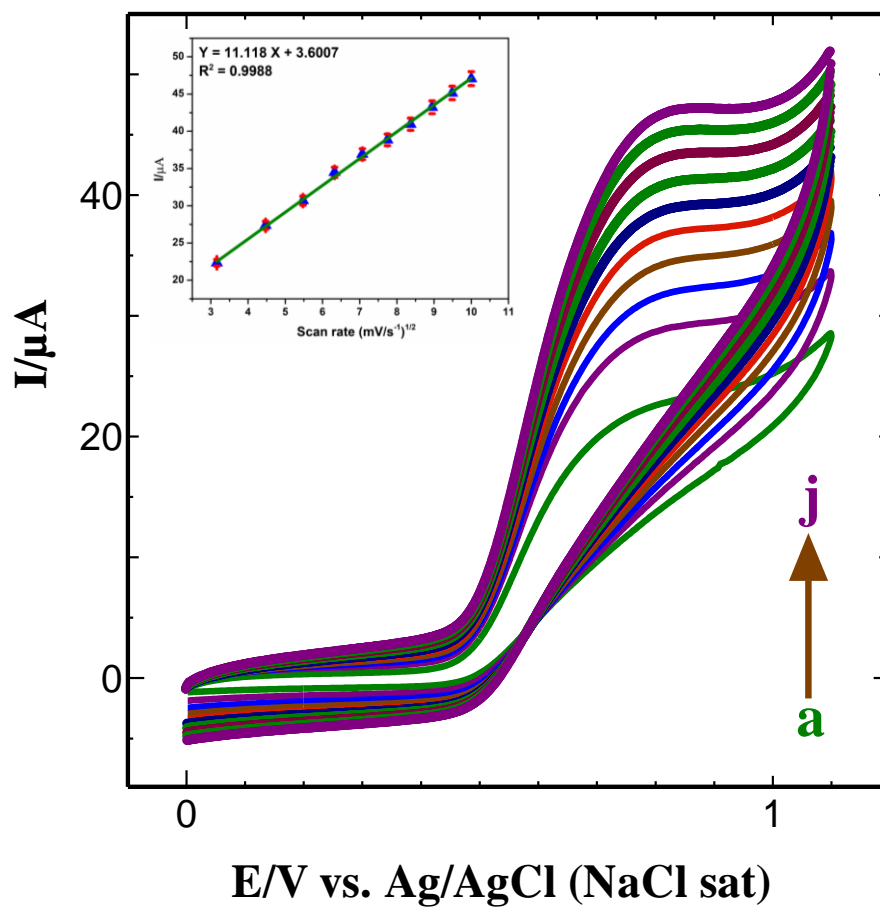
**Figure S4.** ATR-FT-IR spectrum of Cu-MOF-GO on GC film.



**Figure S5.** CV obtained at Cu-MOF-GO electrode in 0.2 M PB solution (pH 7.2) at a scan rate of 50 mV/s.

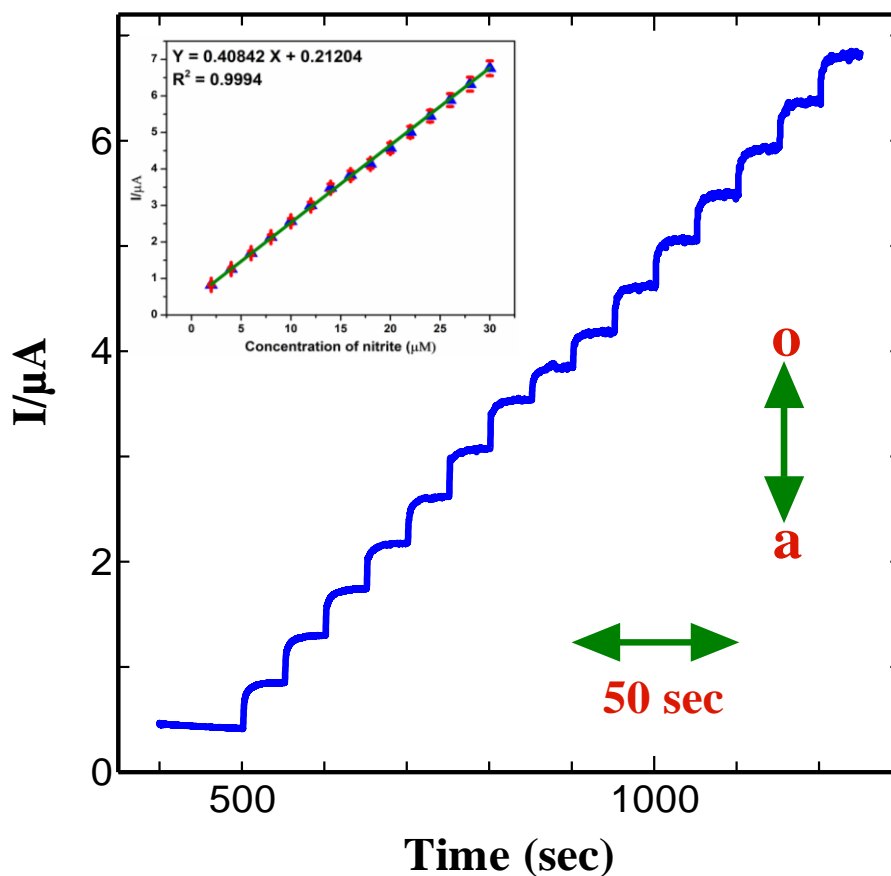


**Figure S6.** CVs obtained for 0.5 mM nitrite at Cu-MOF-GO/GC electrode prepared at different loading level of (a) 1, (b) 2, (c) 3, (d) 4 and (e) 5 mg/mL in 0.2 M PB solution (pH 7.2) at a scan rate of 50 mV/s.



**Figure S7.** CVs obtained for 0.5 mM of nitrite at Cu-MOF-GO/GC electrode in 0.2 M PB solution (pH 7.2) at different scan rates: 10 to 100 mV/s (a-j). **Inset:** plot of current vs. square root of scan rate.





**Figure S8.** Amperometric i-t curve for the determination of nitrite at Cu-MOF-GO modified GC electrode in 0.2 M PB solution (pH 7.2) at an applied potential of +1.0 V. Each increment of 2  $\mu\text{M}$  nitrite (a-o) at a regular interval of 50 s. ***Inset:*** plot of current vs. concentration of nitrite.

Wavenumber/cm <sup>-1</sup>				Peak Assignments
AZA	Cu-MOF	GO	Cu-MOF-GO	
-	614	-	626	Metal stretching vibration of Cu-O
677	-	-	-	Bending vibration of -CH
724	725	-	716	-C-H out plane bending vibration
1086	1101	1090	1107	Stretching vibration of -C-O-
1420	1436	1387	1422	-OH bending vibration
-	1492	-	1495	Stretching vibration of -CH <sub>2</sub>
1703	-	1634	-	Conjugated C=O stretching
2936	2847 and 2929	-	2848 and 2922	-C-H stretching vibration
3440	3448	3459	3432	-OH stretching vibration

**Table S1.** FT-IR spectral data and their assignments for powder AZA, Cu-MOF, GO and Cu-MOF-GO.

Sample	Added (nitrite, $\mu\text{M}$ )	Present method	
		Found ( $\mu\text{M}$ )	Recovery (%)
Lake water	0	-	-
	5	4.98	99.6
	10	14.96	99.8
	20	34.92	99.8
	-	2.30	-
Industrial effluent	5	7.27	99.6
	10	17.23	99.7
	20	37.18	99.8
	-	-	-

**Table S2.** Determination of nitrite in lake and industrial effluent samples using Cu-MOF-GO modified electrode.