

Supporting Information

Tuning the encapsulation of simple fragrances with an amphiphilic graft copolymer

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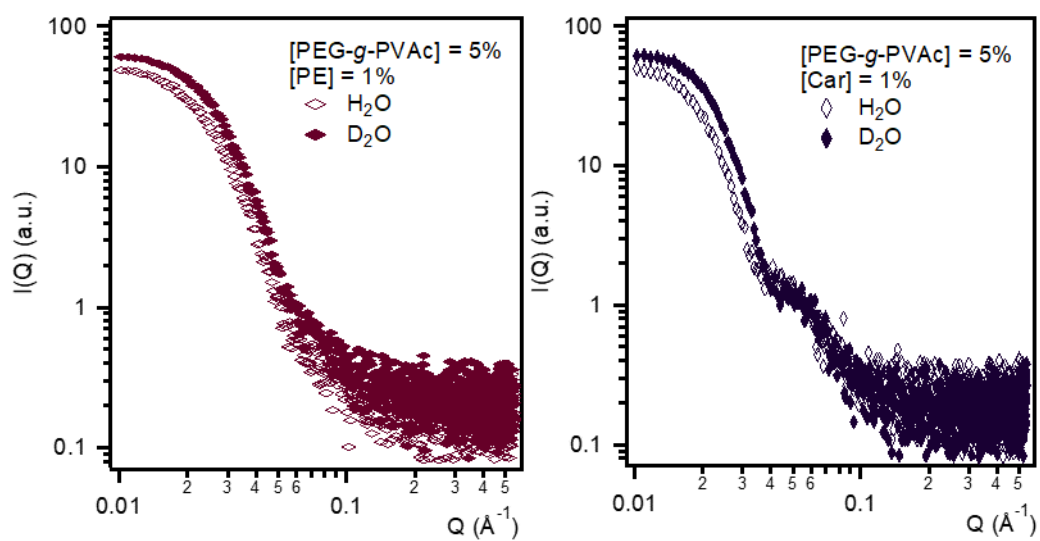


Figure S1. Comparison of SAXS curves for dilute PEG-*g*-PVAc/PE/water (left) and PEG-*g*-PVAc/Car/water (right) systems in normal water and heavy water.

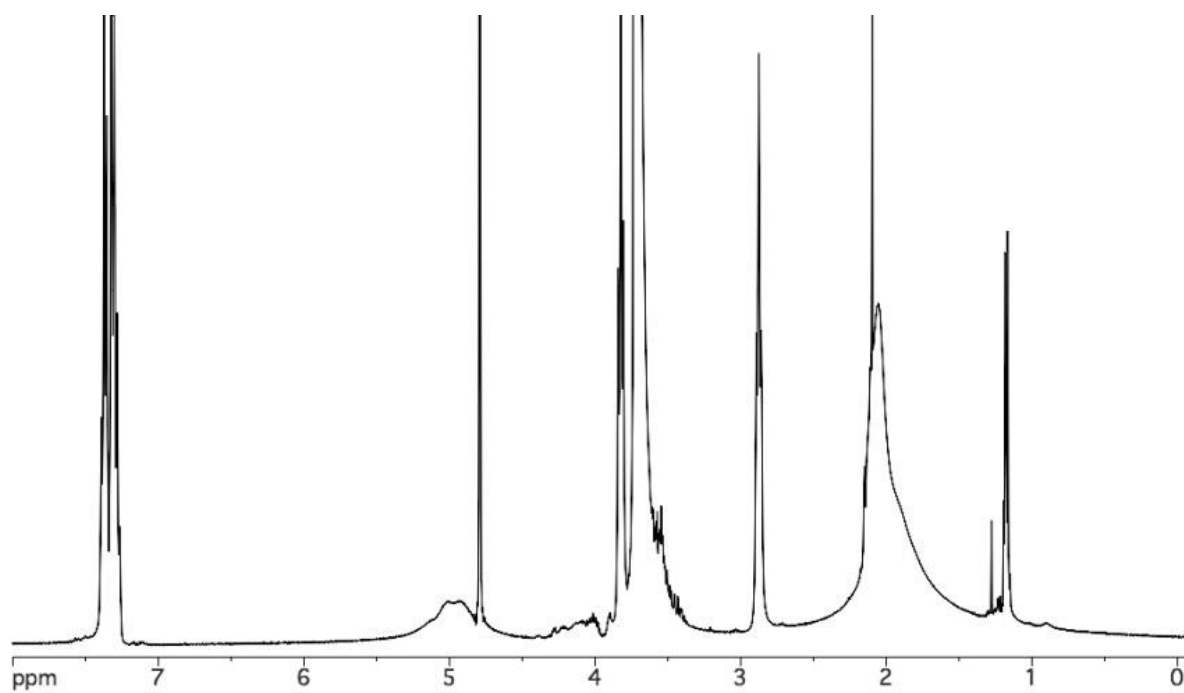


Figure S2: PEG-*g*-PVAc (5%)/2-phenylethanol (1%)/D₂O ¹H-NMR spectrum.

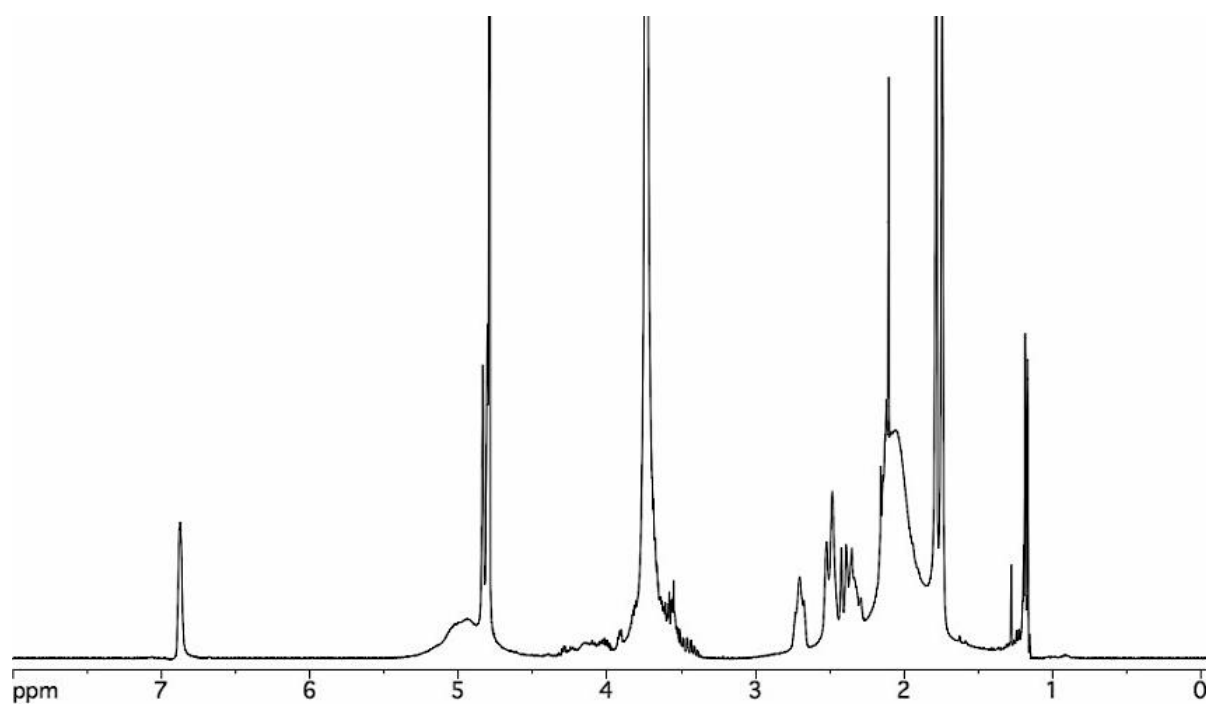


Figure S3: PEG-*g*-PVAc (5%)/Carvone (1%)/D₂O ¹H-NMR spectrum.

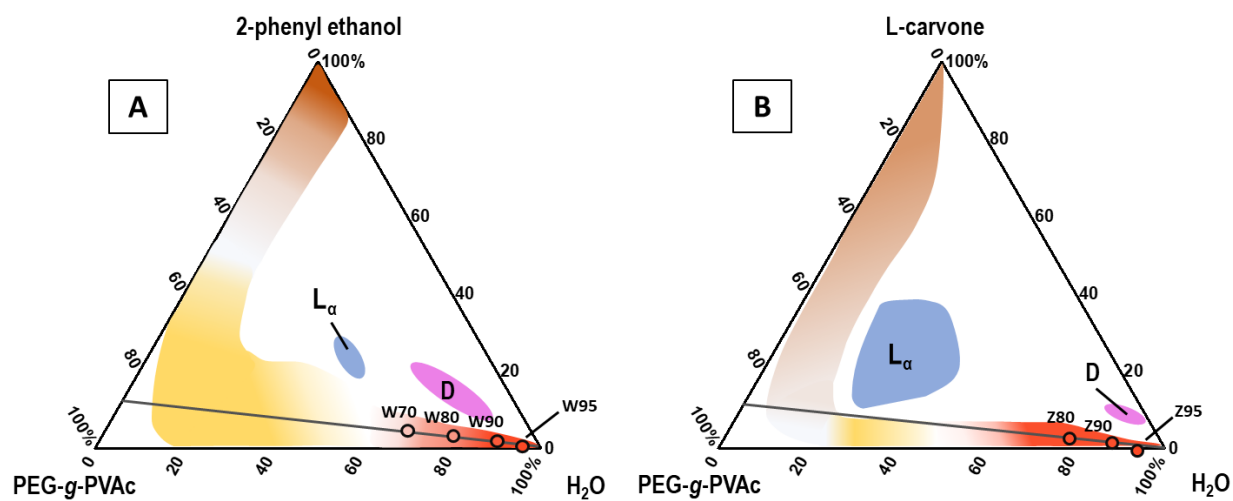


Figure S4. Positions of the samples investigated by means of NMR self-diffusion measurements in the PEG-*g*-PVAc/perfume/water phase diagrams with A) 2-phenyl ethanol and B) L-carvone.

Table S1. Compositions and extracted diffusion components for samples investigated by PGSTE NMR.

Sample name	Composition (%wt)			Self-Diffusion Coefficients, D_ϕ (m^2/s)			
	PE	D ₂ O	Polymer	PE	D ₂ O	Polymer	
						Fast	Slow
W90	1	90	9	$2.13 \cdot 10^{-10}$	$1.53 \cdot 10^{-09}$	$5.59 \cdot 10^{-11}$	$5.26 \cdot 10^{-12}$
W80	2	80	18	$1.16 \cdot 10^{-10}$	$1.27 \cdot 10^{-09}$	$2.11 \cdot 10^{-11}$	$5.52 \cdot 10^{-13}$
W70	3	70	27	$6.82 \cdot 10^{-11}$	$9.73 \cdot 10^{-10}$	$1.15 \cdot 10^{-11}$	$3.33 \cdot 10^{-13}$
Controls	PE	D ₂ O	Polymer	PE	D ₂ O	Polymer	
						Fast	Slow
PE/D ₂ O	1	99	-	$6.75 \cdot 10^{-10}$	-	-	-
W95	-	95	5	-	$1.76 \cdot 10^{-09}$	$7.35 \cdot 10^{-11}$	$1.29 \cdot 10^{-11}$

Table S2: Compositions and extracted diffusion components for samples investigated by PGSTE NMR.

Sample name	Composition (%wt)			Self-Diffusion Coefficients, D_ϕ (m^2/s)			
	Car	D ₂ O	Polymer	Car	D ₂ O	Polymer	
						Fast	Slow
Z90	1	90	9	$3.60 \cdot 10^{-10}$	$1.63 \cdot 10^{-09}$	$5.50 \cdot 10^{-11}$	$5.57 \cdot 10^{-12}$
Z80	2	80	18	$2.63 \cdot 10^{-10}$	$1.32 \cdot 10^{-09}$	$3.06 \cdot 10^{-11}$	$1.08 \cdot 10^{-12}$
Controls	Car	D ₂ O	Polymer	Car	D ₂ O	Polymer	
						Fast	Slow
Car/D ₂ O	1	99	-	$6.68 \cdot 10^{-10}$	-	-	-
Z95	-	95	5	-	$1.76 \cdot 10^{-09}$	$7.35 \cdot 10^{-11}$	$1.29 \cdot 10^{-11}$

Modelling of SAXS data

The following $P(Q)$ functions were used to fit the SAXS experimental data in this work:

-Form factor for spherical objects:

$$P(Q) = \frac{A}{V} \left[\frac{3V(\Delta\rho)(\sin(QR) - QR \cos(QR))}{(QR)^3} \right]^2 + bkg \quad (S1)$$

Here A is a scale factor, V and R are the volume and radius, respectively, of the scattering objects, $\Delta\rho$ is the so-called contrast i.e. the difference between the scattering length densities (SLDs) of the particles and the solvent, and bkg is the instrumental background signal.

-Form factor for core-shell spheres (with 2 shells):

$$P(Q) = \frac{A}{V_s} \left[\frac{3V_c(\rho_c - \rho_{s1})j_1(QR_c)}{QR_c} + \frac{3V_{s1}(\rho_{s1} - \rho_{s2})j_1(QR_{s1})}{QR_{s1}} + \frac{3V_{s2}(\rho_{s2} - \rho_{solv})j_1(QR_{s2})}{QR_{s2}} \right]^2 + bkg \quad (S2)$$

Here the subscripts have the following meanings: c = core, s_1 = first shell, s_2 = second shell, $solv$ = solvent; ρ is the SLD; $j_1(x) = (\sin x - x \cos x)/x^2$; $R_{s1} = R_c + t_1$ and $R_{s2} = R_c + t_1 + t_2$, with t the thickness of each shell; $V_i = (4\pi/3)R_i^3$.

Wherever relevant, a Schulz distribution of the radii was used:

$$f(R) = (z + 1)^{z+1} x^z \frac{\exp[-(z+1)x]}{R_{avg} \Gamma(z+1)} \quad (S3)$$

where $z = 1/(PDI^2 - 1)$ is a function of the polydispersity, $PDI = \sigma/R_{avg}$, with σ^2 the variance of the distribution, R_{avg} the mean radius, $x = R/R_{avg}$ and Γ the Gamma function.

The interaction between polymer aggregates was interpreted according to a hard-sphere structure factor and the Percus-Yevick closure to the Ornstein-Zernike equation:^{1,2}

$$S(Q) = \left[1 + \frac{24\phi_{HS}G(2QR_{HS})}{2QR_{HS}} \right]^{-1} \quad (S4)$$

where $G(2QR_{HS})$ is a trigonometric function depending on Q , R_{HS} (hard-sphere radius) and ϕ_{HS} (hard spheres volume fraction). The potential has the form:

$$U(r) = \begin{cases} \infty & r < 2R \\ 0 & r \geq 2R \end{cases} \quad (S5)$$

For systems in the high polymer concentration regime, the Teubner-Strey model was used:³

$$I(Q) = \frac{1}{a_2 + c_1 Q^2 + c_2 Q^4} + bkg \quad (S6)$$

The coefficients a , c_1 , and c_2 allow for calculation of the bicontinuous network's lattice size, or persistence length, ξ , and the repeat distance, d , through:^{4,5}

$$\frac{d}{2\pi} = \left[\frac{1}{2} \left(\frac{a_2}{c_2} \right)^{1/2} - \frac{c_1}{4c_2} \right]^{-1/2} \quad (S7)$$

$$\xi = \left[\frac{1}{2} \left(\frac{a_2}{c_2} \right)^{1/2} + \frac{c_1}{4c_2} \right]^{-1/2} \quad (S8)$$

From these, the amphiphilicity factor, f_a , is calculated:⁴

$$f_a = \frac{1 - \left(\frac{2\pi\xi}{d} \right)}{1 + \left(\frac{2\pi\xi}{d} \right)} \quad (S9)$$

In our data modelling, we used the following SLD values for the chemical species in the systems:

Table S3. X-ray (Cu $K\alpha$) SLD values calculated for the compounds used in the present work.

Compound	Molecular formula	SLD ($\text{\AA}^{-2}$)
PEG	$\text{C}_2\text{H}_4\text{O}$	$1.10 \cdot 10^{-5}$
PVAc	$\text{C}_4\text{H}_6\text{O}_2$	$1.08 \cdot 10^{-5}$
Water	H_2O	$9.36 \cdot 10^{-6}$
2-phenyl ethanol	$\text{C}_8\text{H}_{10}\text{O}$	$9.29 \cdot 10^{-6}$
L-carvone	$\text{C}_{10}\text{H}_{14}\text{O}$	$8.84 \cdot 10^{-6}$
α -pinene	$\text{C}_{10}\text{H}_{16}$	$8.07 \cdot 10^{-6}$

Table S4. Fitting results for the SAXS patterns of Figure 4A (main text). R_c = core radius; σ = Schulz polydispersity of R ; SLD_{core} , SLD_s = scattering length densities of the core and shell, respectively; t = shell thickness; R_{tot} = radius of the core-shell particle; Φ = hard-sphere volume fraction. Instrumental error associated to these results is ± 0.6 Å. ${}^{\ddagger}R_{tot}$ is not a fit model parameter.

Samples composition	PEG- <i>g</i> -PVAc (wt%)	19.0	28.5	38.0		47.5	57.0	66.5	76.0
	2-phenyl ethanol (wt%)	5.0	5.0	5.0		5.0	5.0	5.0	5.0
	Water (wt%)	76.0	66.5	57.0		47.5	38.0	28.5	19.0
Core-shell form factor + hard-sphere structure factor	R_c (Å)	48 ± 0.3	37 ± 0.2	35 ± 0.2	-	-	-	-	-
	σ	0.30	0.38	0.30	-	-	-	-	-
	SLD_{core} (10^{-6} Å ⁻²)	10.8	10.8	10.8	-	-	-	-	-
	t (Å)	64 ± 0.2	46 ± 0.3	40 ± 0.3	-	-	-	-	-
	SLD_s (10^{-6} Å ⁻²)	9.51	9.51	9.51	-	-	-	-	-
	${}^{\ddagger}R_{tot} = R_c + t$ (Å)	112	83	75	-	-	-	-	-
	Φ	0.14	0.22	0.23	-	-	-	-	-
Teubner-Strey model	Correlation length, ξ (Å)	-	-	-	73	70	63	50	42
	Repeat distance, d (Å)	-	-	-	172	152	140	143	148
	Amphiphilicity factor, f_a	-	-	-	-0.5	-0.5	-0.5	-0.4	-0.3

Table S5. Fitting results for the SAXS patterns shown in Figure 4B (main text). R_c : core radius; σ : Schulz polydispersity of R ; SLD_{core} , SLD_{s1} , SLD_{s2} , SLD_{solv} : scattering length densities of the core, first shell, second shell, and solvent, respectively; t_1 and t_2 : thickness of the first and second shell, respectively; R_{tot} = radius of the core-shell particle; Φ = hard sphere volume fraction. $^{\ddagger}R_{tot}$ is not a fit model parameter.

Samples composition	PEG- <i>g</i> -PVAc (wt%)	9.5	19.0	28.5	38.0	47.5	57.0		66.5
	L-carvone (wt%)	5.0	5.0	5.0	5.0	5.0	5.0		5.0
	Water (wt%)	85.5	76.0	66.5	57.0	47.5	38.0		29.5
Core-shell form factor + hard-sphere structure factor	R_c (Å)	87 ± 0.6	63 ± 0.8	64 ± 1.3	56 ± 0.8	48 ± 0.7	34 ± 1.0	-	-
	σ	0.31	0.31	0.32	0.30	0.35	0.4	-	-
	SLD_{core} (10^{-6} Å^{-2})	8.9	9.2	9.7	9.7	9.7	9.7	-	-
	t_1 (Å)	32 ± 0.7	25 ± 0.6	20 ± 1.2	22 ± 0.9	26 ± 0.9	19 ± 1.7	-	-
	SLD_{s1} (10^{-6} Å^{-2})	10.5	10.5	10.5	10.5	10.5	10.5	-	-
	t_2 (Å)	43 ± 1.2	40 ± 0.8	20 ± 1.4	14 ± 0.5	8 ± 0.5	18 ± 0.9	-	-
	SLD_{s2} (10^{-6} Å^{-2})	9.5	9.5	9.5	9.5	9.5	9.5	-	-
	$R_{tot} = R_c + t_1 + t_2$ (Å) [‡]	162	128	104	92	82	71	-	-
	Φ	0.15	0.24	0.33	0.35	0.37	0.32	-	-
Teubner-Strey model	Correlation length, ξ (Å)	-	-	-	-	-	-	89	70
	Repeat distance, d (Å)	-	-	-	-	-	-	160	148
	Amphiphilicity factor, f_a	-	-	-	-	-	-	-0.6	-0.5

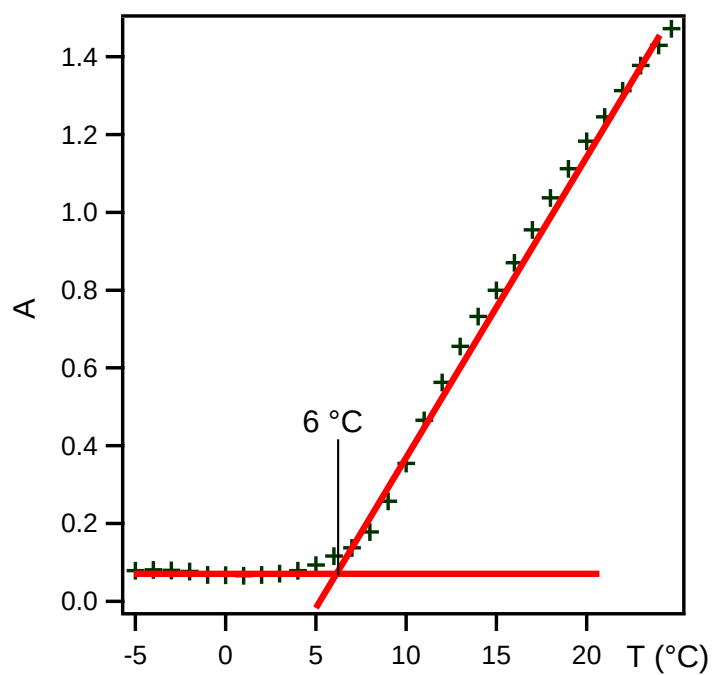


Fig. S5. Plot of the UV-vis absorbance at $\lambda = 500$ nm vs. temperature for PEG-*g*-PVAc/PE/water = 9.5/5.0/85.5 %wt. The two slopes were fitted to straight lines; the crossing point corresponds to the cloud point temperature of the polymer in this system.

References

- (1) Kinning, D. J.; Thomas, E. L. Hard-Sphere Interactions between Spherical Domains in Diblock Copolymers. *Macromolecules* **1984**, *17* (9), 1712–1718. <https://doi.org/10.1021/ma00139a013>.
- (2) Percus, J. K.; Yevick, G. J. Analysis of Classical Statistical Mechanics by Means of Collective Coordinates. *Phys. Rev.* **1958**, *110* (1), 1–13. <https://doi.org/10.1103/PhysRev.110.1>.
- (3) Teubner, M.; Strey, R. Origin of the Scattering Peak in Microemulsions. *The Journal of Chemical Physics* **1987**, *87* (5), 3195–3200. <https://doi.org/10.1063/1.453006>.
- (4) Koehler, R. D.; Schubert, K. -V.; Strey, R.; Kaler, E. W. The Lifshitz Line in Binary Systems: Structures in Water/C₄E₁ Mixtures. *The Journal of Chemical Physics* **1994**, *101* (12), 10843–10849. <https://doi.org/10.1063/1.467833>.
- (5) Leitão, H.; Telo da Gama, M. M.; Strey, R. Scaling of the Interfacial Tension of Microemulsions: A Landau Theory Approach. *The Journal of Chemical Physics* **1998**, *108* (10), 4189–4198. <https://doi.org/10.1063/1.475817>.