

Supporting Information for  
**Facile Fabrication of Robust and Stretchable Cellulose  
Nanofibers/Polyurethane Hybrid Aerogels**

Yating Wang<sup>†,‡</sup>, Xiang Li<sup>†,‡</sup>, Huan Cheng<sup>†,‡</sup>, Bijia Wang<sup>†,‡</sup>, Xueling Feng<sup>†,‡</sup>, Zhiping  
Mao<sup>†,‡</sup>, Xiaofeng Sui<sup>†,\*\*</sup>

† Key Lab of Science and Technology of Eco-Textile, Ministry of Education, College  
of Chemistry, Chemical Engineering and Biotechnology, Donghua University, No.  
2999 North Renmin Road, Shanghai, 201620, People's Republic of China

‡ Innovation Center for Textile Science and Technology of DHU, Donghua University,  
No. 2999 North Renmin Road, Shanghai, 201620, People's Republic of China

\*Corresponding authors: Tel.: +86 21 67792605. Fax: +86 21 67792707.

E-mail: suixf@dhu.edu.cn (Xiaofeng Sui)

Supporting information consists of 5 pages, 4 Tables, 5 Figures and 2 Videos.

**Table S1.** The compositions of the CNF/PU hybrid aerogels

Samples	CNF aerogel (wt%)	PU (wt%)	Density (g/cm <sup>3</sup> )
CNF aerogel	100	0	0.03
CNF/PU-48	52	48	0.06
CNF/PU-66	34	66	0.09
CNF/PU-75	25	75	0.12

**Table S2.** Thermal parameters of the CNF/PU/octadecane composites

Samples	PCM content (wt%)	$\Delta H_m$ (J/g)	$\Delta H_c$ (J/g)
CNF/PU-48/octadecane	92.5	223.3	222.8
CNF/PU-66/octadecane	87.1	207.6	207.8

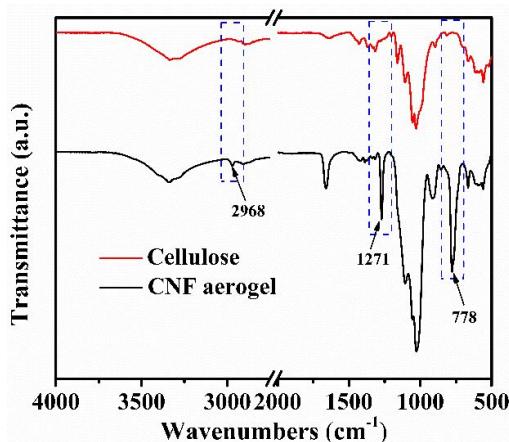
**Table S3.** Thermal conductivity of octadecane and CNF/PU/octadecane composites

Samples	Thermal conductivity (Wm <sup>-1</sup> K <sup>-1</sup> )
octadecane	0.32±0.01
CNF/PU-48/octadecane	0.23±0.02
CNF/PU-66/octadecane	0.25±0.06
CNF/PU-75/octadecane	0.23±0.03

**Table S4.** Thermal parameters of CNF/PU-75/octadecane after different thermal cycling numbers

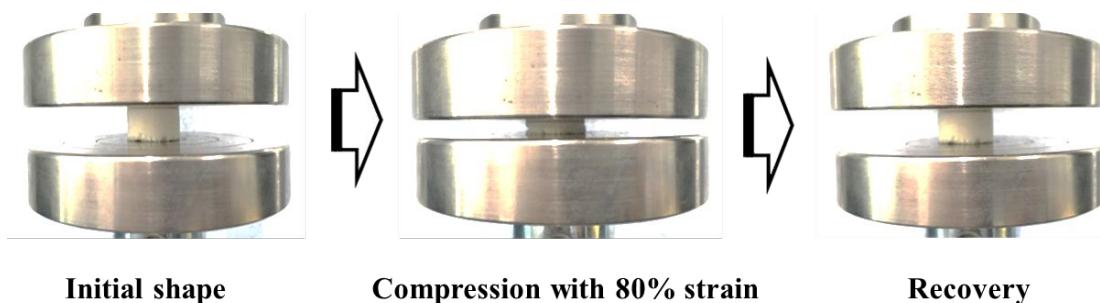
Cycling numbers	$T_m$ (°C)	$T_c$ (°C)	$\Delta H_m$ (J/g)	$\Delta H_c$ (J/g)
1 <sup>th</sup>	37.6	16.6	190.0	189.7
25 <sup>th</sup>	36.6	18.1	190.1	188.9
50 <sup>th</sup>	36.6	17.9	189.0	188.7
75 <sup>th</sup>	36.6	18.4	188.3	187.4
100 <sup>th</sup>	36.6	18.4	187.1	186.2

$T_m$ : melting temperature,  $T_c$ : crystallization temperature,  $\Delta H_m$ : melting enthalpy,  $\Delta H_c$ : crystallization enthalpy.

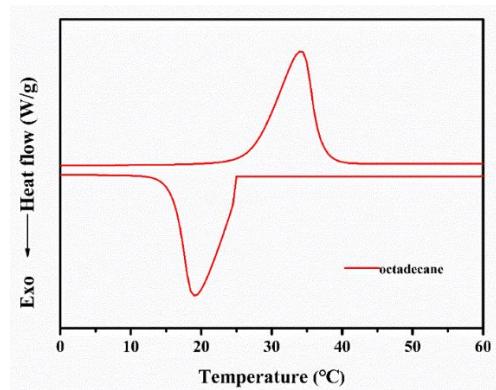


**Fig.S1.** FTIR spectra of the cellulose and the CNF aerogel

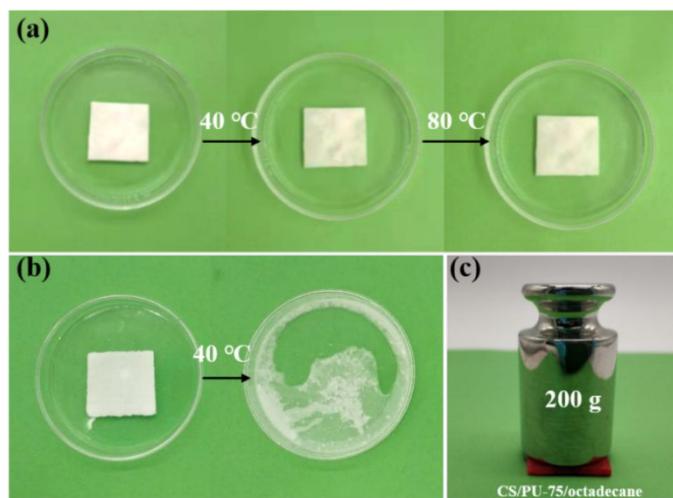
The FTIR spectra of the cellulose and CNF aerogel are compared in Fig.S1. The new peaks at 778 cm<sup>-1</sup> in the CNF aerogel spectrum is assigned to C-Si asymmetric stretching vibration. The peak at 1100 cm<sup>-1</sup>-1130 cm<sup>-1</sup> was belonged to the stretching absorption peaks of the Si-O-Si bonds, overlapping with the peaks of the C–O bonds of cellulose. The new peak at 1271 cm<sup>-1</sup> and 2968 cm<sup>-1</sup> correspond to the stretching and bending vibration of -CH<sub>3</sub> in the siloxane compounds. The above results confirmed the successful occurrence of polysiloxane reaction between cellulose and methyl-trimethoxy silane.



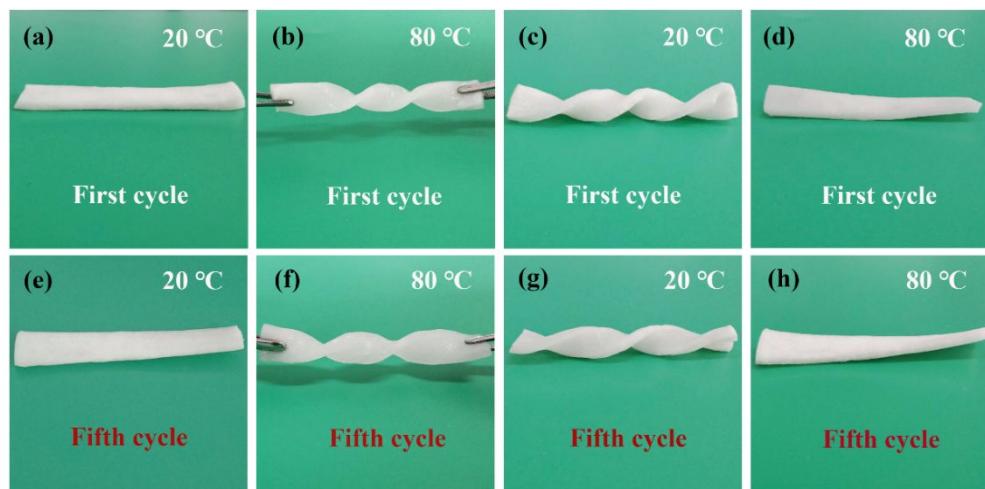
**Fig.S2.** Shape recovery process of CNF/PU-75



**Fig.S3.** DSC curves of octadecane



**Fig.S4.** Shape stability of (a) CNF/PU-75/octadecane and (b) octadecane with increasing temperature. (c) The mechanical stability of CNF/PU-75/octadecane at 40 °C.



**Fig.S5.** Shape memory behavior of the CNF/PU-75/octadecane at 80 °C. a-d Images showed the shape memory process of the CNF/PU-75/octadecane. e-h Images showed the CNF/PU-75/octadecane recovered its permanent shape after five cycles.