

Supporting Information for

Chemical and structural evolution during synthesis of layered Li(Ni,Co,Mn)O₂ oxides

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S1 Additional results

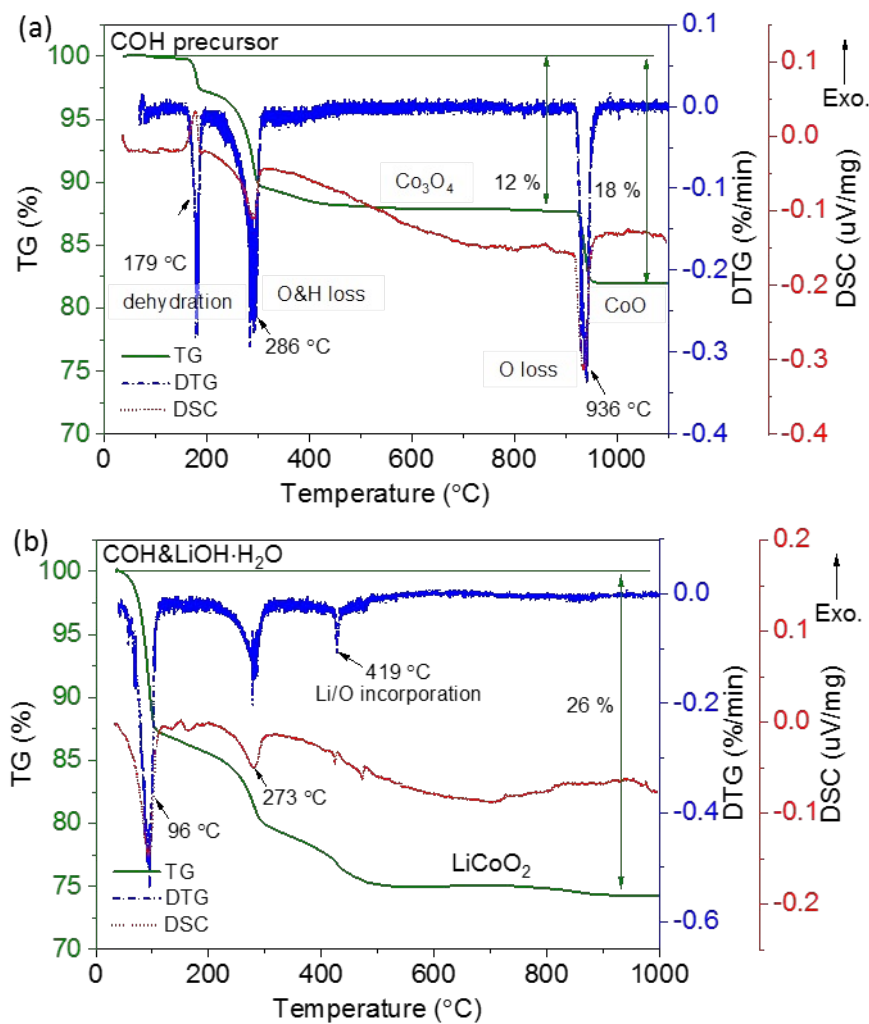
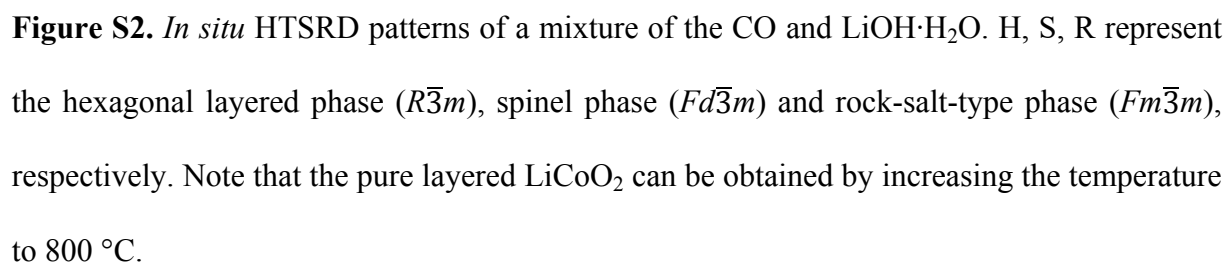


Figure S1. TG/DTG/DSC curves of (a) the COH precursor and (b) the mixture of COH and $\text{LiOH} \cdot \text{H}_2\text{O}$, respectively.



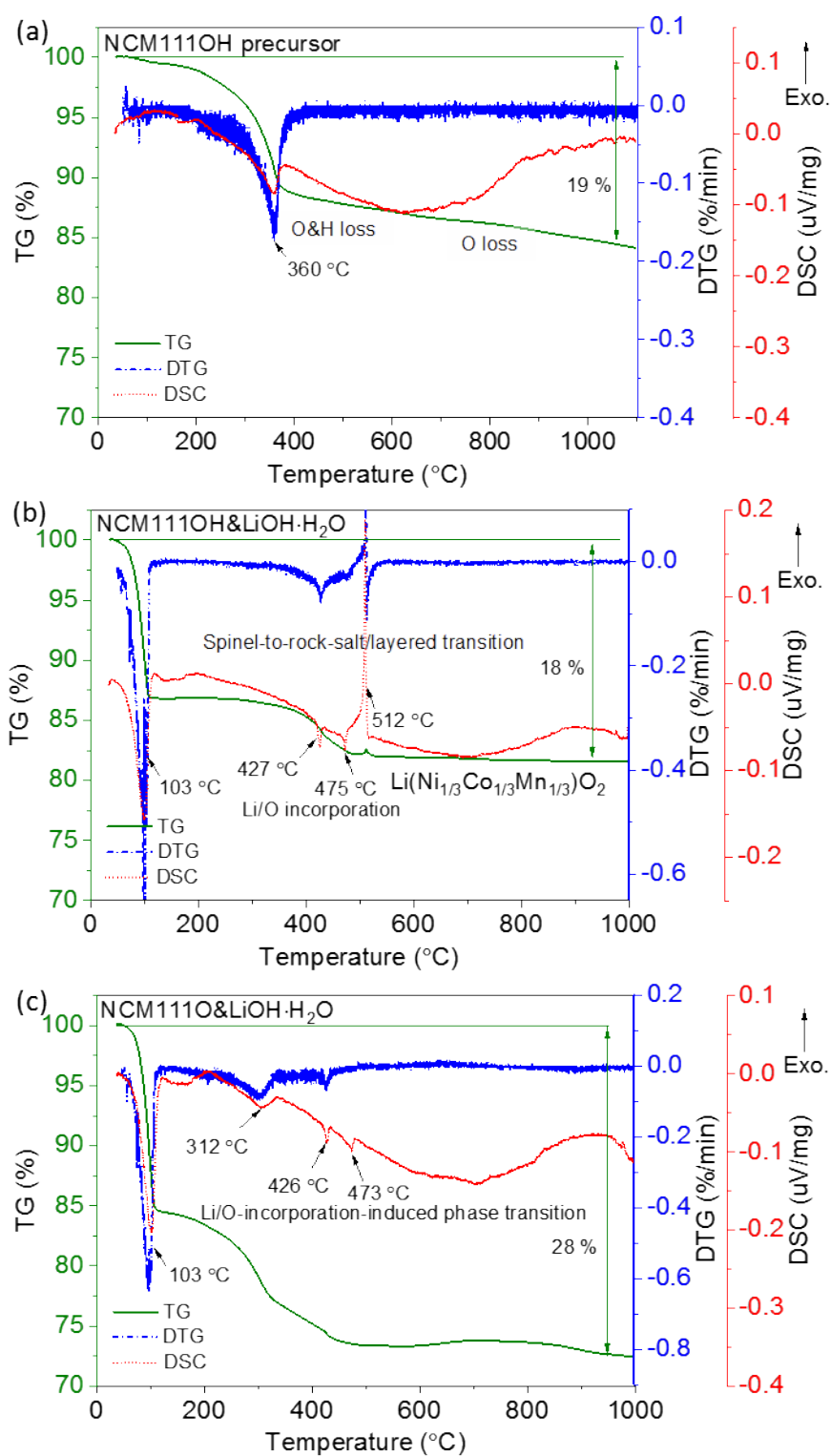


Figure S3. TG/DTG/DSC curves of (a) the NCM111OH precursor, (b) the mixture of NCM111OH and LiOH·H₂O, (c) a mixture of NCM111O together with LiOH·H₂O, respectively.

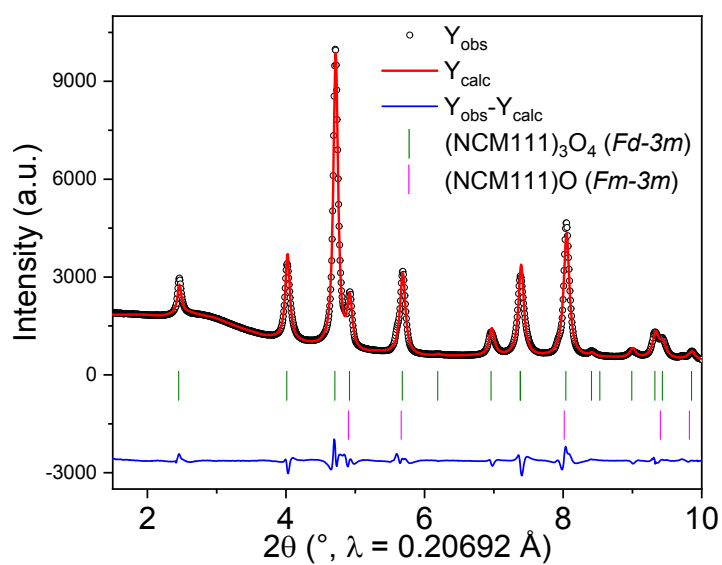


Figure S4. Rietveld refinement to the SRD pattern of NCM111OH precursor heated at 850 °C during in situ heating.

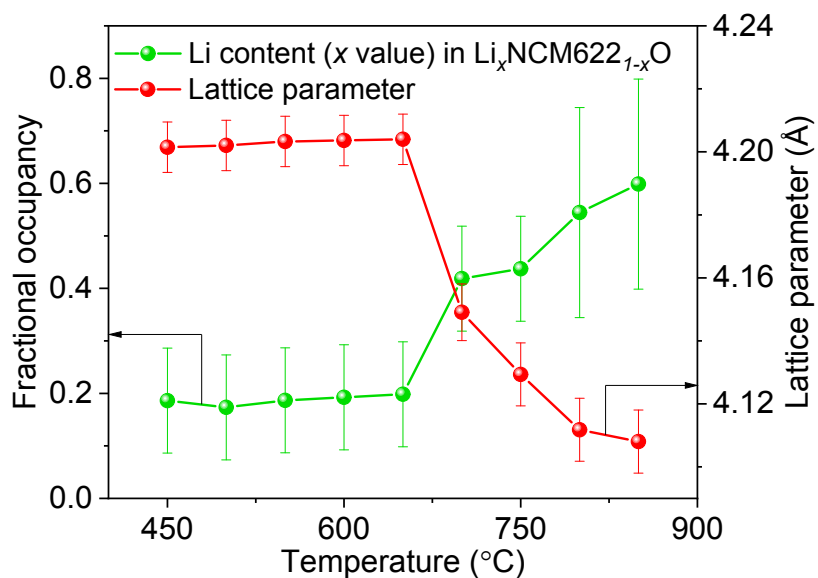


Figure S5. Evolution of lattice parameters in the fully disordered Li-containing rock-salt-type structure (*Fm-3m*) as a function of temperature during synthesis of LNCM622O.

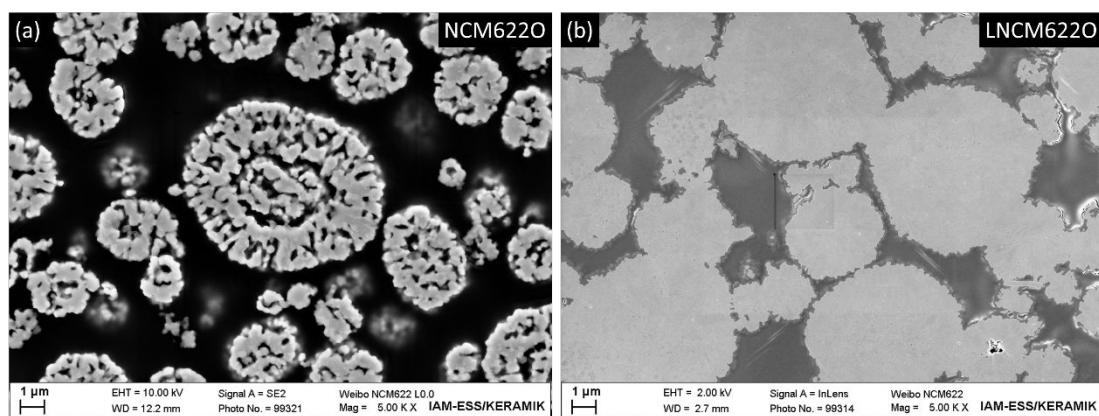


Figure S6. Cross section SEM images of (a) NCM622O and (b) LNCM622O.

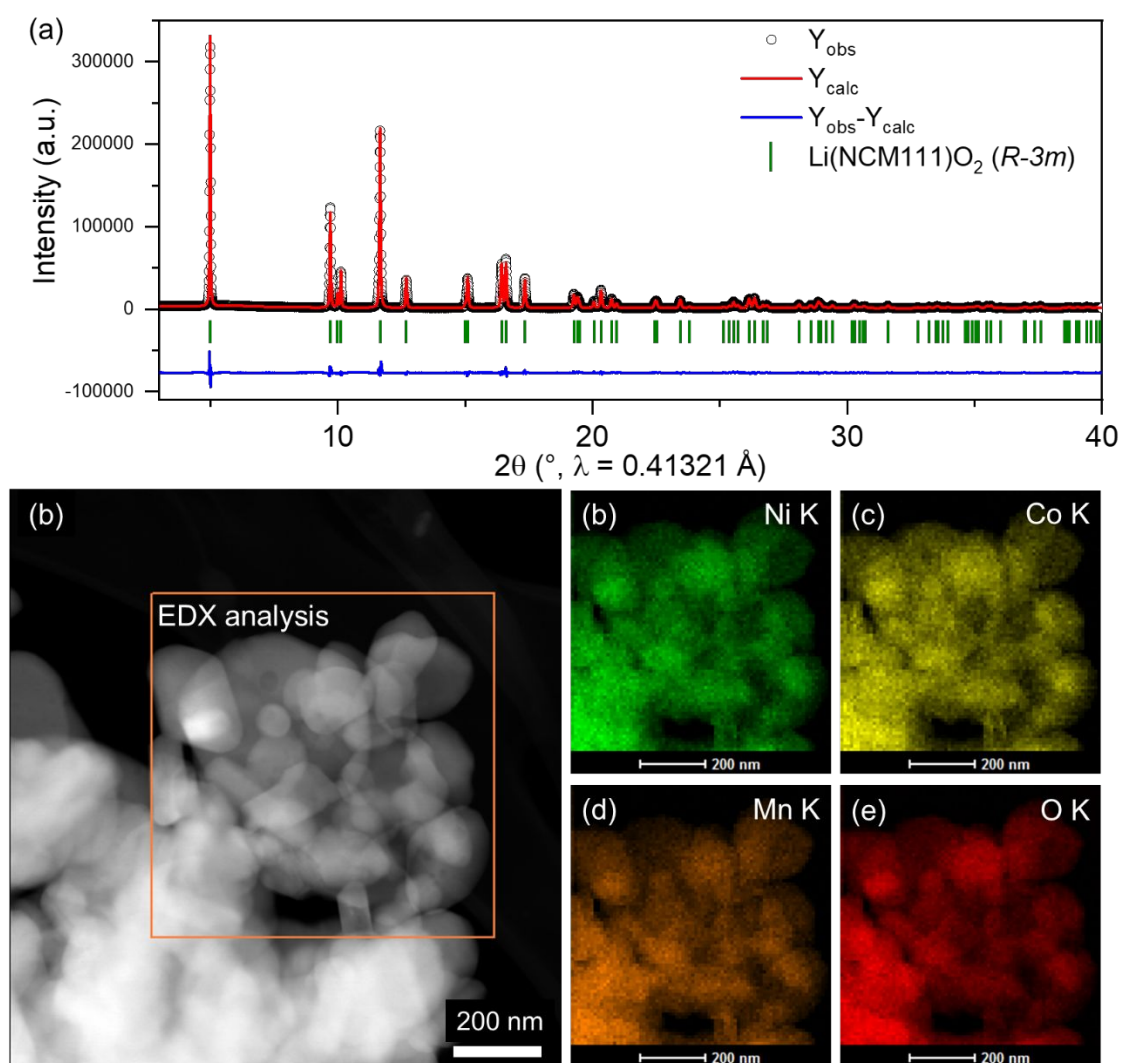


Figure S7. (a) Rietveld refinement against SRD patterns, (b) HAADF-STEM, (c-f) HAADF-STEM-EDX mapping images of LNCM111O.

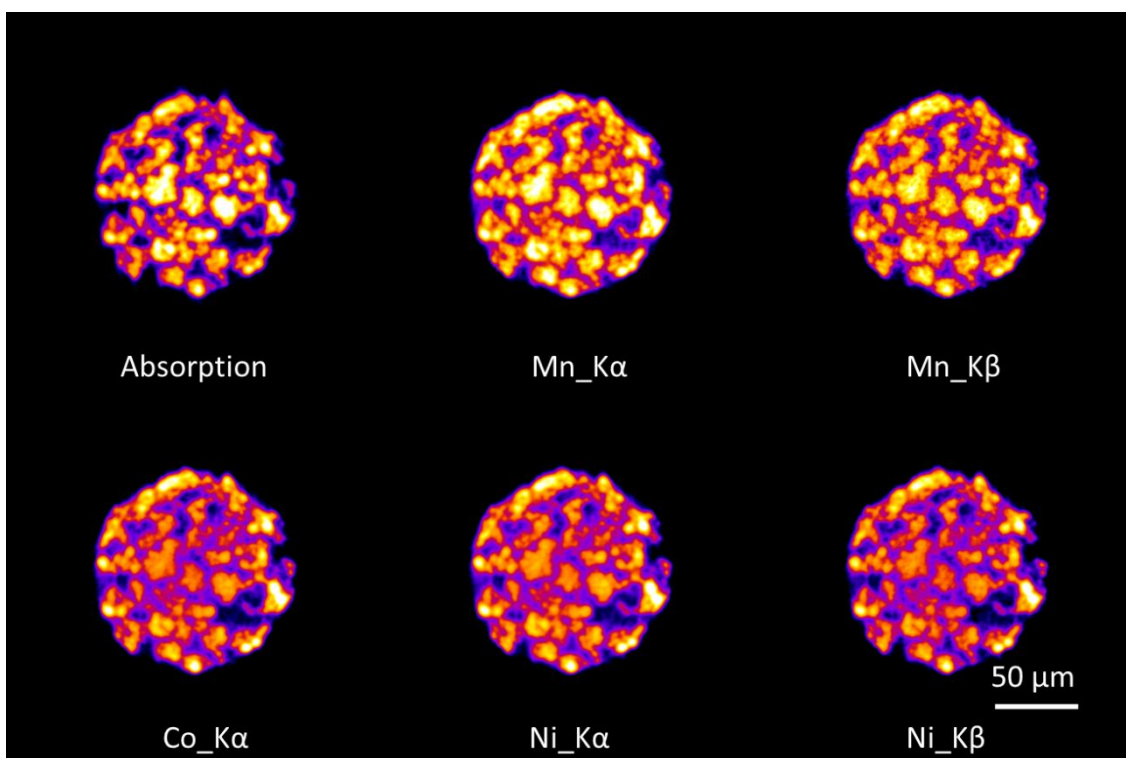


Figure S8. A virtual slice obtained from micro-XRF combined with absorption contrast computed scanning tomography projection slices of the sample LNCM622O, showing homogeneous spatial distributions of transition metals, i.e. Ni, Co, and Mn.

Table S1. Crystallographic parameters of NCM622OH.

Cell parameters					
Space group: $P-3m1$, $a = 3.1087 \text{ \AA}$, $b = 3.1087 \text{ \AA}$, $c = 4.6131 \text{ \AA}$, $V = 38.6081 \text{ \AA}^3$, gamma = 120° , $Z = 2$					
Atomic positions					
Name	site	x	y	z	Fract
Ni1	$1c$	0.000	0.000	0.000	0.600
Co1	$1c$	0.000	0.000	0.000	0.200
Mn1	$1c$	0.000	0.000	0.000	0.200
O1	$2d$	0.333	0.667	0.205	1.000
H1	$2d$	0.333	0.333	0.453	1.000

Refinement parameters					
$R_{wp} = 6.38 \%$, $R_p = 5.35 \%$, χ^2 : 2.68					

Note: Fract is the fractional occupancy of the atom on this site. Considering that synchrotron diffraction technique is not sensitive to distinguish Ni, Co and Mn, the ratio of Ni:Co:Mn is fixed according to their chemical composition.

Table S2. Crystallographic parameters of the NCM622O.

Cell parameters (Spinel phase, 56 wt%)					
Space group: $Fd-3m$, $a = b = c = 8.2876 \text{ \AA}$, $V = 569.2296 \text{ \AA}^3$, $Z = 8$, formula: AB_2O_4					
Atomic positions					
Name	site	x	y	z	Fract
Ni1	$8a$	0.125	0.125	0.125	0.600
Co1	$8a$	0.125	0.125	0.125	0.200
Mn1	$8a$	0.125	0.125	0.125	0.200
Ni2	$16d$	0.500	0.500	0.500	0.600
Co2	$16d$	0.500	0.500	0.500	0.200
Mn2	$16d$	0.500	0.500	0.500	0.200
O1	$32e$	0.261	0.261	0.261	1.000
Cell parameters (rock-salt-type phase, 44 wt%)					
Space group: $Fm-3m$, $a = b = c = 4.1834 \text{ \AA}$, $V = 73.2113 \text{ \AA}^3$, $Z = 4$, formula: AB					
Atomic positions					
Name	site	x	y	z	Fract
Ni1	$4a$	0.000	0.000	0.000	0.600
Co1	$4a$	0.000	0.000	0.000	0.200
Mn1	$4a$	0.000	0.000	0.000	0.200
O1	$4b$	0.500	0.500	0.500	1.000
Refinement parameters					
$R_{wp} = 9.08 \%$, $R_p = 7.40 \%$, χ^2 : 23.93					

Table S3. Crystallographic parameters of LNCM622O.

Cell parameters					
Space group: $R\bar{3}m$, $a = b = 2.8653 \text{ \AA}$, $c = 14.2117 \text{ \AA}$, $V = 101.0479 \text{ \AA}^3$, $\gamma = 120^\circ$, $Z = 3$, formula: ABO ₂					
Atomic positions					
Name	site	x	y	z	Fract
Li1	$3a$	0.000	0.000	0.000	0.981
Ni2	$3a$	0.000	0.000	0.000	0.019
Li2	$3b$	0.000	0.000	0.500	0.019
Ni1	$3b$	0.000	0.000	0.500	0.581
Co1	$3b$	0.000	0.000	0.500	0.200
Mn1	$3b$	0.000	0.000	0.500	0.200
O1	$6c$	0.241	0.241	0.241	1.000
Refinement parameters					
$R_{wp} = 10.80 \%$, $R_p = 8.73 \%$, χ^2 : 37.70					