Supplementary Information

Volcano-shaped SPM probe for combined force-electrogram recordings from excitable cells

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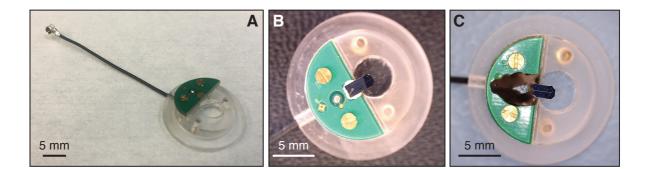
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13 **1** Mechanical and electrical interfacing

14 **1.1** Custom cantilever holder

15 A custom interface was necessary to connect the nanovolcano probe to both the AFM and the 16 electronic measurement system. A custom-made shielded PCB connected to a 800 µm-diameter 17 U. FL coaxial cable was screwed to a custom-made AFM cantilever holder as illustrated in 18 Figure S1-A. Figure S1-B shows the nanovolcano probe glued to the PCB and wire-bonded to 19 the gold coated pad of the PCB. The PCB directly connects the gold coated pad to the inner 20 wire of the coaxial output, therefore guaranteeing an standard electrical connection to the 21 nanovolcano. A glob top was used to electrically insulate the chip-PCB interface when the 22 system is immersed in liquid (cf. Figure S1-C). Overall, this interface prevents any electrical 23 shortcut when working in a liquid environment, and shields the recorded electrical signals from 24 external electromagnetic noises, therefore allowing for low-noise recordings.

The cantilever holder keeps the nanovolcano probe tilted at an angle of 11° with respect to the horizontal, as required for the optical deflection measurement with AFM.

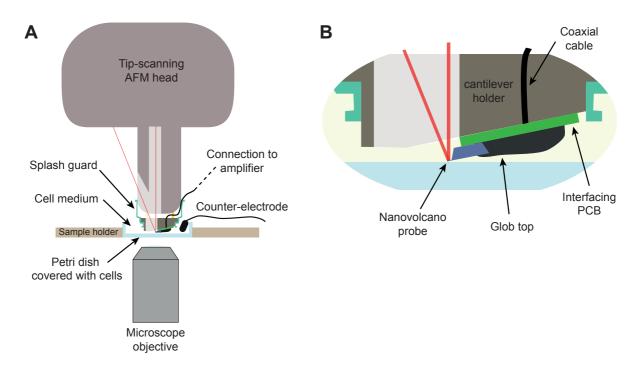


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Figure S1: Mechanical and electrical interfacing of the nanovolcano probe. A) Picture of the custom-made AFM holder without the nanovolcano probe assembled. B) Top-view image of the nanovolcano probe mechanically assembled and wire-bonded to the custom-made holder.
C) Top-view image of the interface with the glob top used to insulate the chip-PCB electrical contact.

33 1.2 Setup for simultaneous optical imaging, force, and electrical recordings

A custom setup was necessary to obtain simultaneously optical images as well as mechanical forces and electrical recordings (Figure S2-A). The cells were seeded to 6 cm in diameter petri dish, which was mounted in a custom sample holder. The cells were then covered with liquid medium (Figure S2-A). The sample holder rests on a large aluminum structure placed on a noise cancelling table (not represented in the figure) to ensure minimal mechanical noise. The 39 nanovolcano probe was placed on top of the sample using the custom cantilever holder 40 described above (Figure S1), connected to a commercial tip-scanning AFM head (Figure S2). 41 The cantilever deflection is measured using an optical laser readout. The force applied by the 42 cell to the cantilever is calculated based the cantilever deflection signal, knowing the cantilever 43 stiffness and deflection sensitivity. The nanovolcano cantilever electrode is connected through 44 a coaxial cable to a microelectrode amplifier and a digital acquisition system to record the 45 electrical signal. An inverted microscope was placed under the sample holder to allow for 46 simultaneous optical imaging.





- 48 Figure S2: Description of the custom AFM-optical system. A) Schematic drawing showing a
- 49 cross-section view of the custom-made AFM-optical system. B) Expanded cross-section view
- 50 of the custom-made cantilever holder mounted on the AFM head.

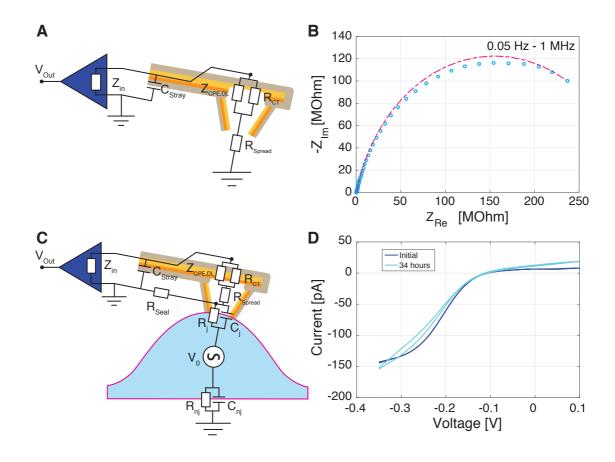
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52 2 Electrochemical characterization

53 2.1 Electrode-electrolyte interface

Figure S3-A shows the electrical equivalent model of the electrode-electrolyte interface. As previously described in the literature,¹ it is composed of a non-linear resistance, R_{CT} , that represents faradaic charge transfer secondary to redox reactions, in parallel with a constant phase element $Z_{CPE,DL}$ representing the double layer capacitance underlying the capacitive charge transfer. R_{Spread} represents the resistance induced the confinement of the electric field lines near the microelectrode. The stray capacitance, C_{stray} , denotes the capacitive current leaks along the insulated tracks.

The electrode-electrolyte interface properties have been experimentally measured using electrochemical impedance spectroscopy (EIS) based on a 100 mV sinusoidal signal applied to the nanovolcano probe in presence of phosphate buffer saline (PBS) at room temperature. The EIS data are shown in Figure S3-B and show a typical electrode-electrolyte behavior.² Values for each element composing the electrical equivalent circuit have been extracted and are summarized in Table S1.



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Figure S3: *Electrochemical characterization of the nanovolcano probe.* A) Equivalent electrical model of the electrode-electrolyte interface. B) Electrochemical impedance spectroscopy of a single nanovolcano probe in PBS. C) Equivalent electrical model of the cellelectrode interface. D) Cyclic voltammogram of the nanovolcano probe right after immersion in a solution of 5 mM hexaamine ruthenium chloride and 100 mM potassium nitrate in deionized water (dark blue) and after 34 hours of continuous chronoamperometry at a fixed potential of -0.35 V (light blue).

Table S1: Experimental values of every element composing the electrical equivalent circuit of
 the electrode-electrolyte interface

Element	Value		
R _{CT}	315 MΩ		
C _{DL} , n	4.15 nF, 0.82		
Rspread	74.4 kΩ		
CStray	30 pF		

77 2.2 Cell-electrode interface

Figure S3-C represents the electrical equivalent circuit once the nanovolcano probe is engaged onto a cell. In this situation, the junctional membrane resistance R_j and capacitance C_j are added in parallel with the seal resistance R_{seal} , representing the leaks at the cell-electrode interface, at the tip of the nanovolcano probe. V_0 is the cell transmembrane potential whereas R_{nj} and C_{nj} respectively represent the non-junctional cell membrane resistance and capacitance.

83 During engaging, no variation of the resistance is observed at low frequency; therefore 84 suggesting that the seal resistance R_{Seal} is much lower compared to the charge transfer resistance 85 R_{CT}. However, at higher frequency, the impedance of the electrode-electrolyte interface 86 (Z_{CPE,DL} || R_{CT}) becomes negligible as most of the current passes through the low double layer impedance. At this frequency, impedance measurements (as reported in Figure 2, main 87 88 manuscript) directly represent the spreading resistance in serial with the other components of 89 the cell-electrode interface ($R_{Seal} \parallel R_i \parallel C_i$). For this reason, the impedance and time constant 90 seen by the nanovolcano probe increases when approaching the cell surface.

91 2.3 Long-term characterization

92 The long-term stability of the nanovolcano probe was characterized by Scuba Probe 93 Technologies LLC. The nanovolcano probe was immersed into a solution of 5 mM hexaamine 94 ruthenium chloride and 100 mM potassium nitrate in deionized water. As shown in Figure S3-95 D, a first cyclic voltammogram was acquired between -0.35 V to 0.1 V vs. Ag/AgCl at a scan 96 rate of 50 mV/s using a 3 electrodes setup (dark blue curve). For the next 34 hours, the 97 nanovolcano potential was hold at -0.35 V while similar cyclic voltammograms were registered 98 every 20 minutes as control. The light blue cyclic voltammogram in Figure S3-D shows nearly 99 no significant differences compared to the initial one, thereby demonstrating that the 100 nanovolcano probe was functional for 34 hours.

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102 **3 Additional data**

103 In this section, additional data recorded with the nanovolcano probe are presented to 104 demonstrate the reliability of the method.

105 **3.1** Force-controlled impedance measurements

106 Overall, force-controlled impedance measurements were performed on seven human
107 embryonic kidney (HEK) cells and seven primary rat cardiomyocytes (CMCs) using two
108 different nanovolcano probes. Experimental results are summarized in Table S2.

109 Table S2: Summary of the force-controlled impedance measurements performed on human110 embryonic kidney cells and primary rat cardiomyocytes.

Cell	Cell	Probe	Impedance Z [kΩ]		Time constant τ [μs]			Peak	Ramp	
ID	type		Off cell	On cell	ΔZ [%]	Off cell	On cell	Δτ [%]	force [nN]	rate [Hz]
1	HEK	1	458	502	109	30	40	133	15	0.25
2	HEK	2	361	485	131	25	35	140	400	0.1
3	HEK	2	358	640	178	25	45	180	22.1	0.1
4	HEK	2	400	850	212	25	40	160	60	0.1
5	HEK	2	426	594	139	25	30	120	12	1
6	HEK	2	440	820	186	25	35	140	115	1
7	HEK	2	441	718	163	25	35	140	19.5	0.1
8	CMC	2	493	993	201	35	65	186	470	2
9	CMC	2	458	601	131	30	45	150	36	3
10	CMC	2	467	496	106	30	35	117	15.5	3
11	CMC	2	498	941	195	35	56	160	772	2
12	CMC	2	507	561	111	35	40	114	46	3
13	CMC	2	494	560	113	30	35	117	12.5	3
14	CMC	2	432	625	145	30	40	133	43	3

111 **3.2** Contraction displacement measurements

112	Contraction characteristics recorded from five different primary rat cardiomyocytes are
113	summarized in Table S3. Parameters include the contraction amplitude, duration at 50% of the
114	amplitude, contraction number, as well as the overall recording duration for each cell tested.

115 **Table S3:** Summary of the primary rat cardiomyocyte contraction characteristics recorded from

Cell ID	Contraction amplitude [nm]	Contraction duration [s]	Number of contractions	Recording duration [s]
1	382 ± 95	415 ± 117	11	50
2	262 ± 88	436 ± 26	12	25
3	207 ± 28	545 ± 78	6	7
4	26 ± 3	231 ± 124	3	3
5	121 ± 44	297 ± 18.1	19	50

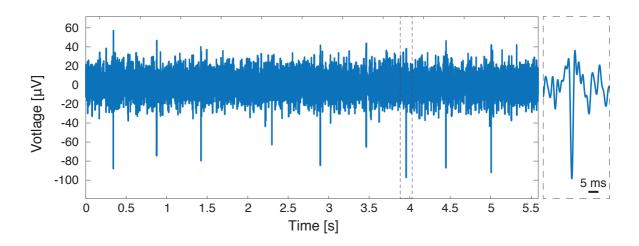
116 different cells with a single nanovolcano probe (mean \pm SD).

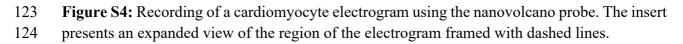
117 **3.3 Electrogram measurements**

Overall, two electrograms were successively recorded from two different cells using a single nanovolcano probe, therefore demonstrating the capability of successive recordings at different location with a single probe. The signals of the first recording are presented in main manuscript

121 Figure 4, those of the second recording are shown in Figure S4. Electrograms displayed

122 downstroke amplitudes of $64 \pm 17 \ \mu V \ (N = 42)$ for an overall recording duration of 25 s.





125 **4 References**

B. X. E. Desbiolles, E. De Coulon, A. Bertsch, S. Rohr and P. Renaud, Intracellular
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