

Supporting Information

“Synthesis of α -Trifluoromethylthio- α,β -unsaturated Carbonyl Compounds by DABCO-Mediated Electrophilic Trifluoromethylthiolation with N -SCF₃-dibenzenesulfonimide”

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*Department of Chemistry and Green-Nano Materials Research Center,
Kyungpook National University, Daegu 41566, Republic of Korea*

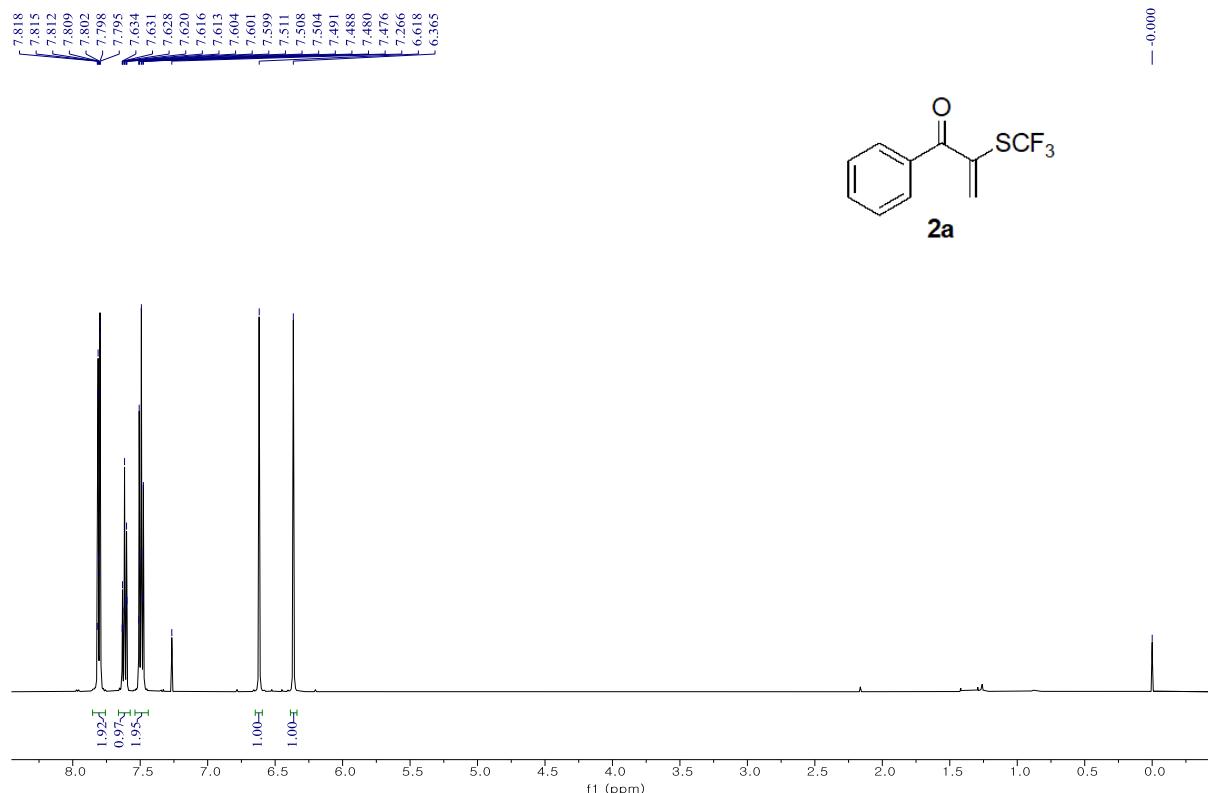
*E-mail: cwcho@knu.ac.kr

[†]J.Y. and H.-J.H. contributed equally to this work.

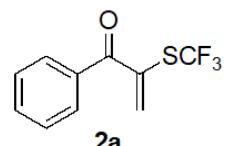
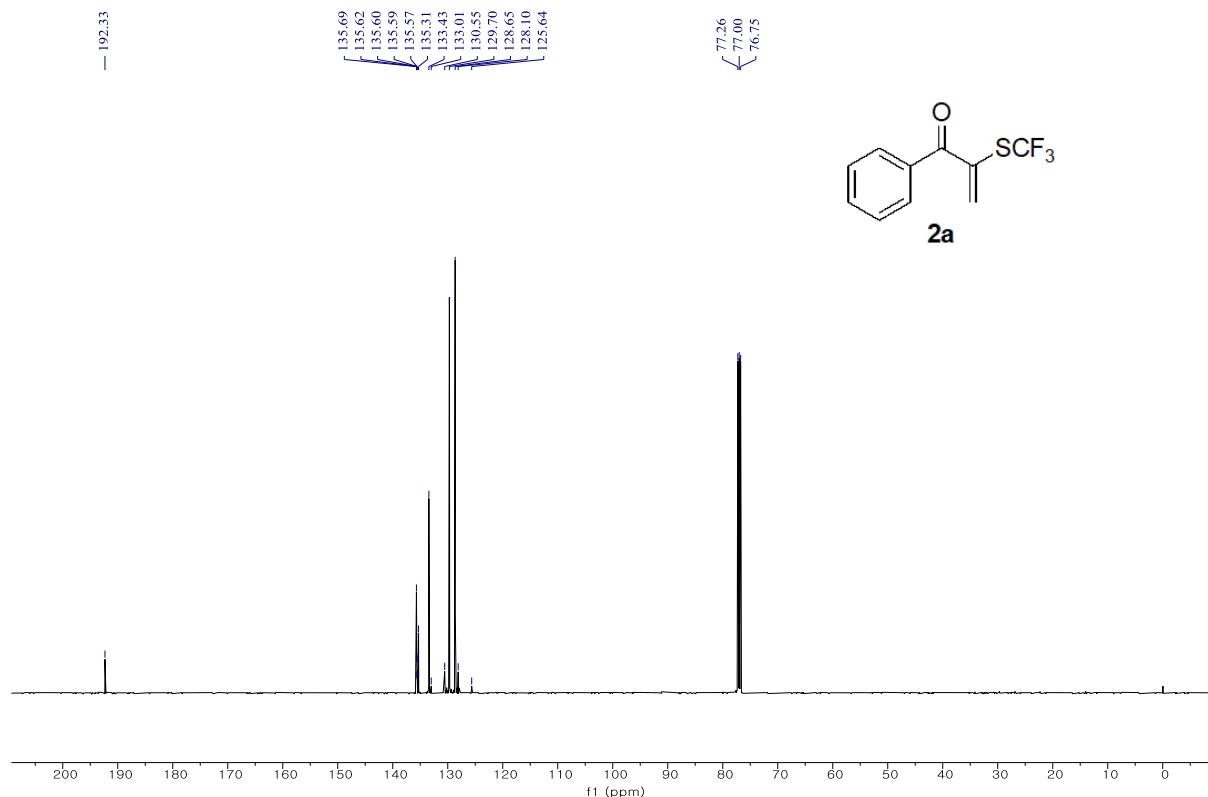
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¹H NMR of 2a (500 MHz, CDCl₃)



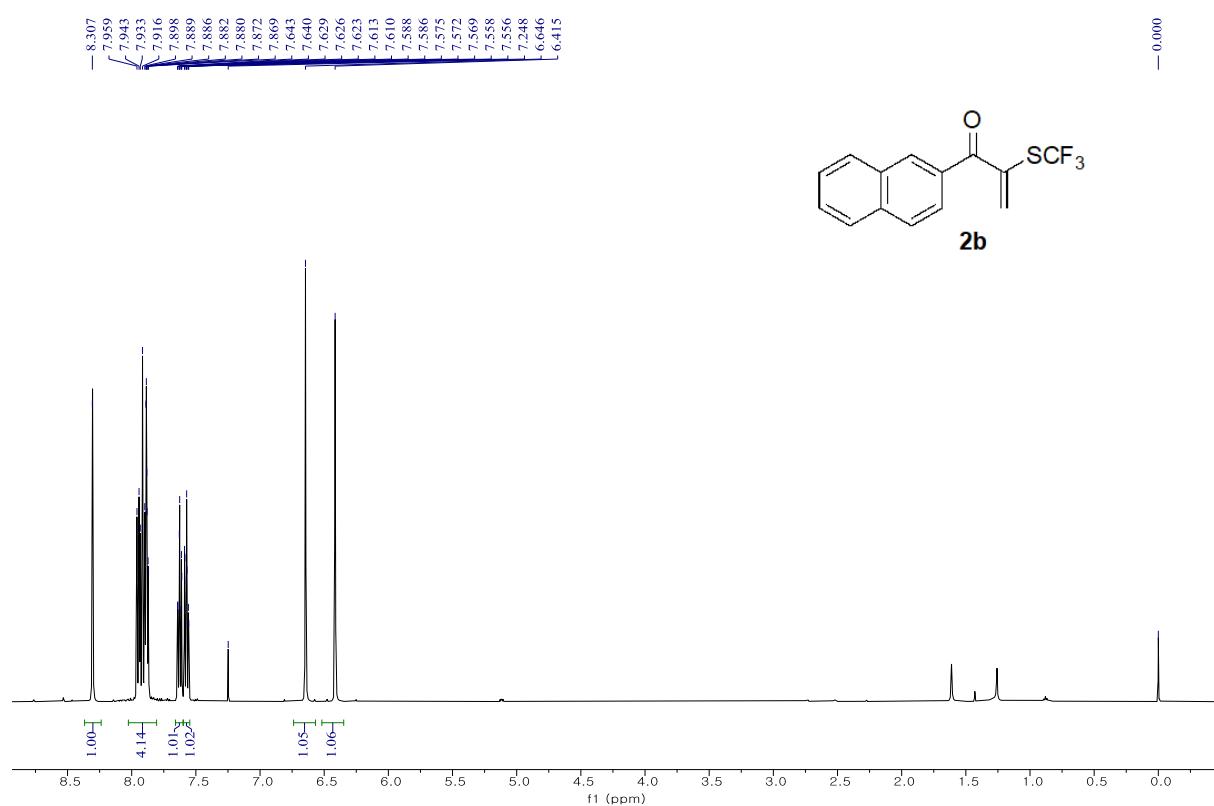
¹³C NMR of **2a** (125 MHz, CDCl₃)



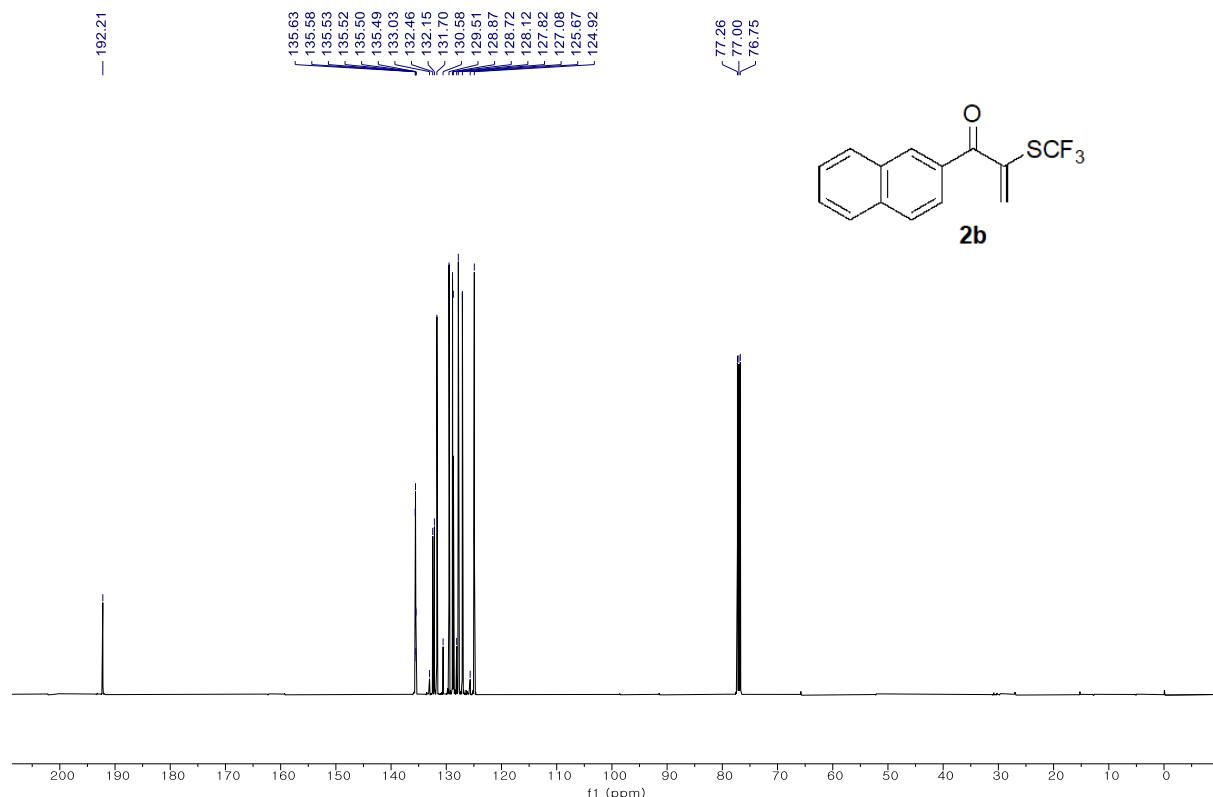
¹⁹F NMR of **2a** (564 MHz, CDCl₃)



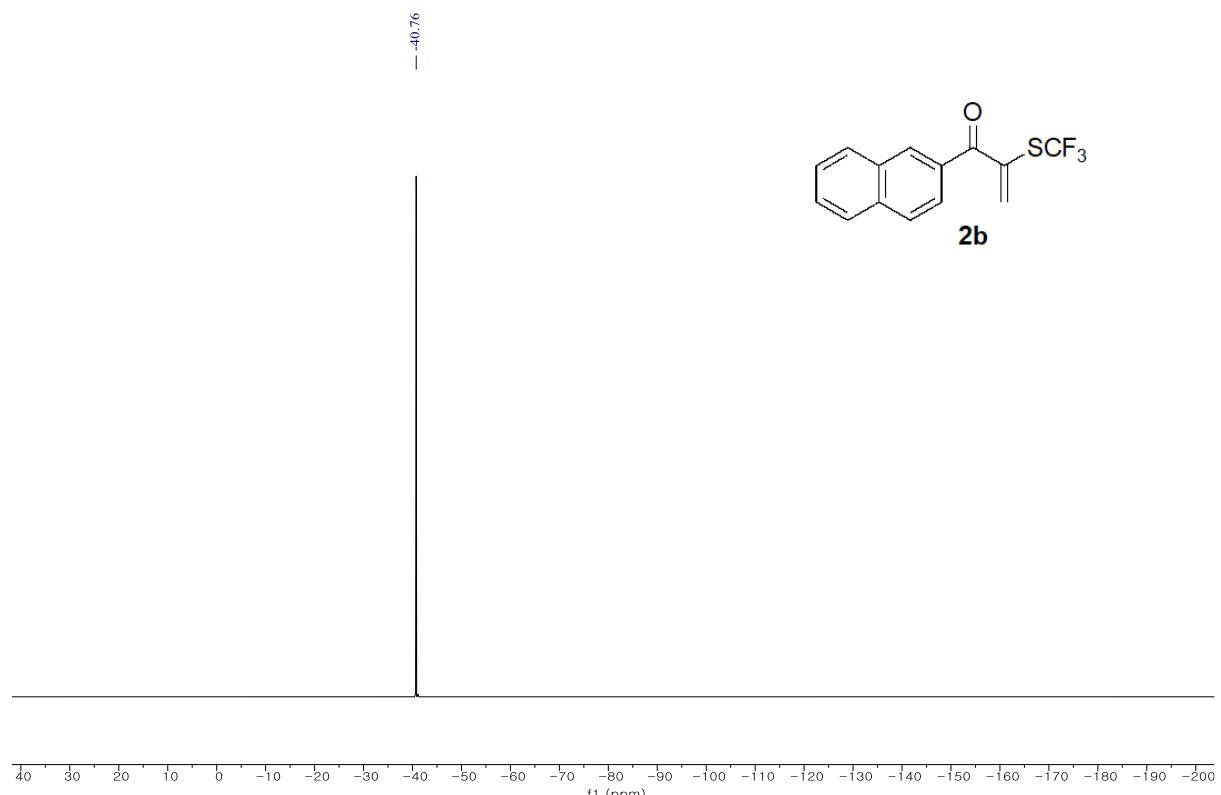
¹H NMR of **2b** (500 MHz, CDCl₃)



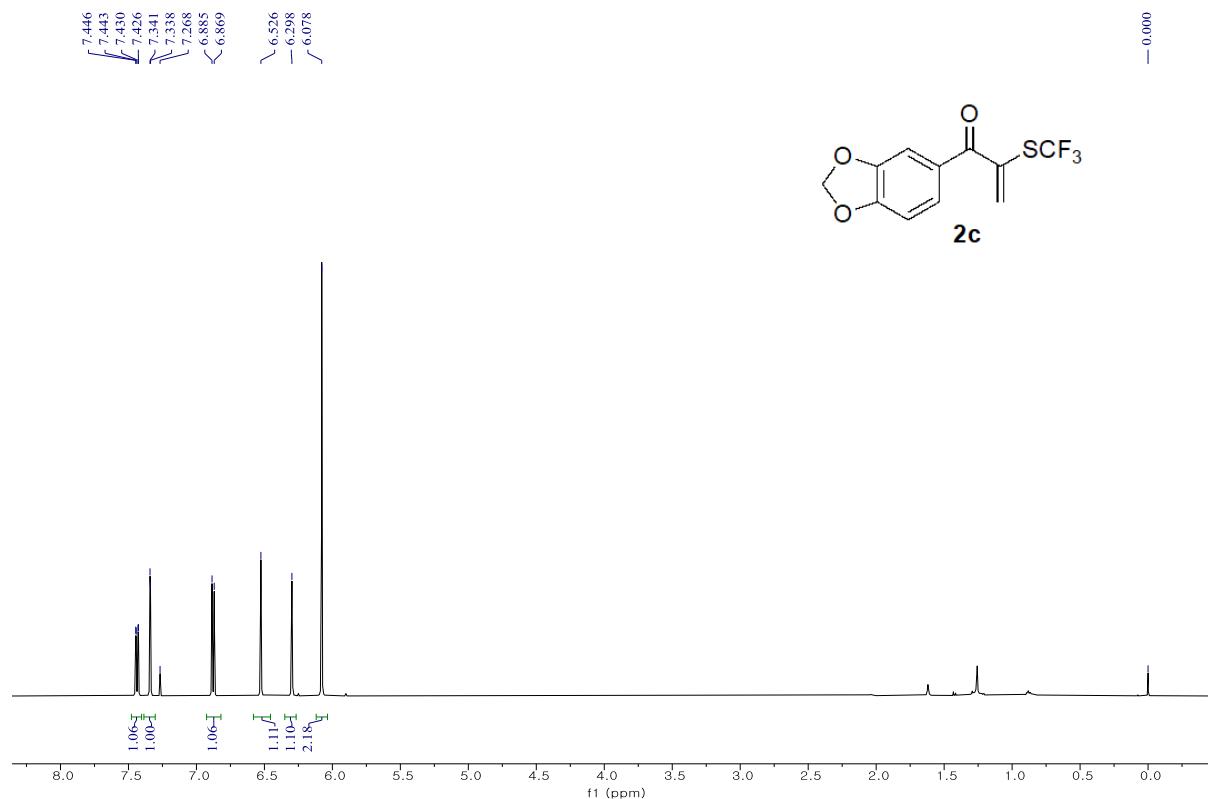
^{13}C NMR of **2b** (125 MHz, CDCl_3)



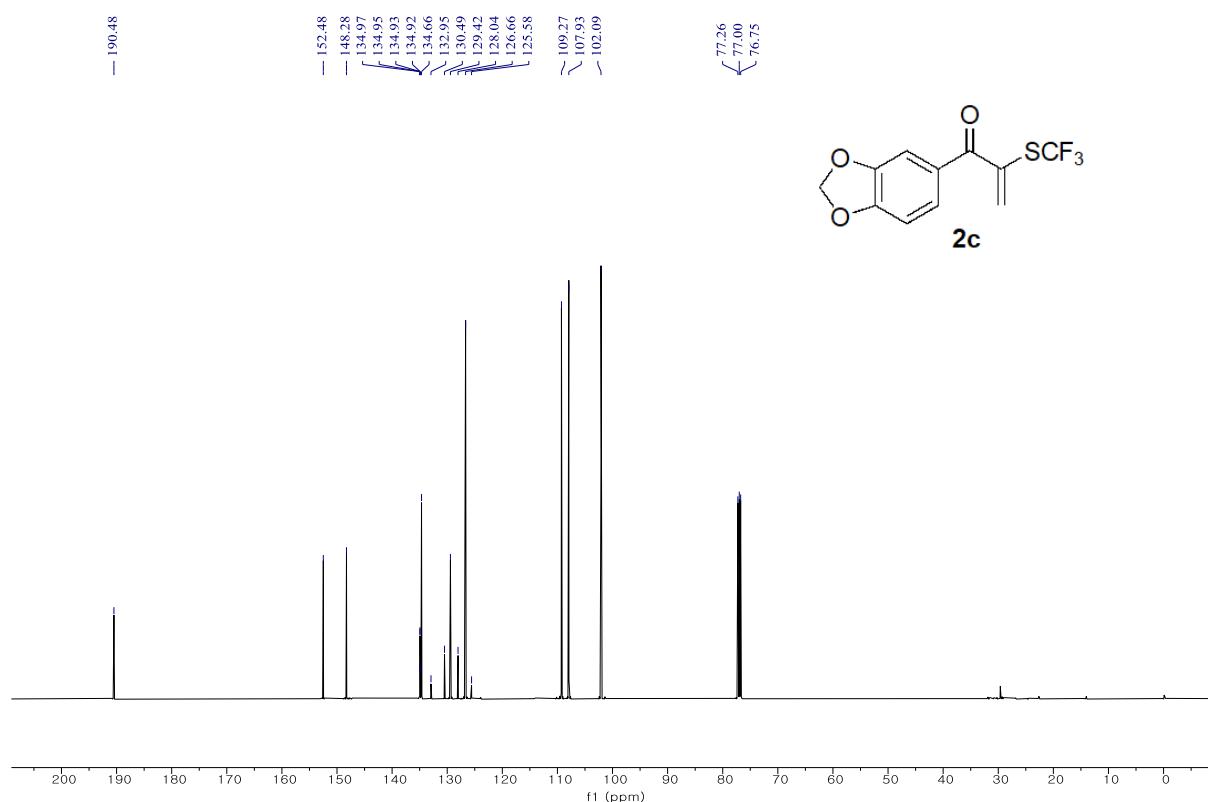
^{19}F NMR of **2b** (564 MHz, CDCl_3)



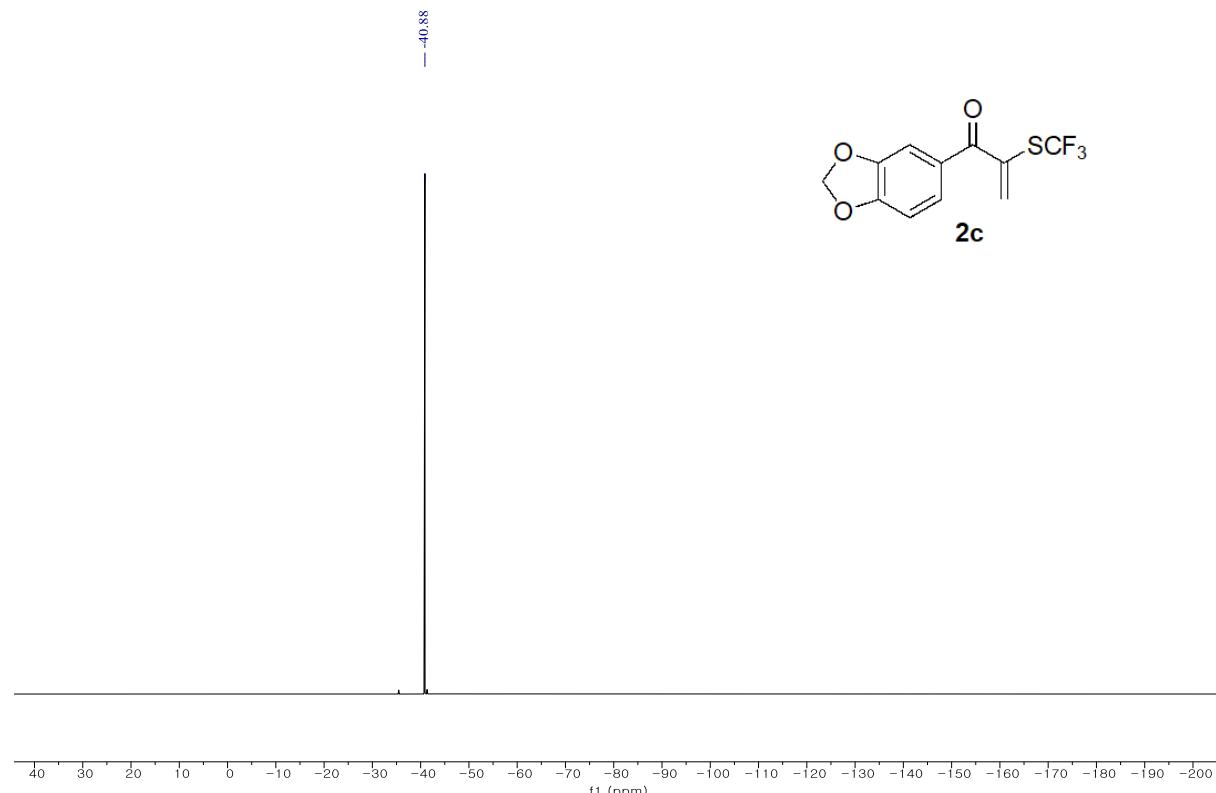
¹H NMR of 2c (500 MHz, CDCl₃)



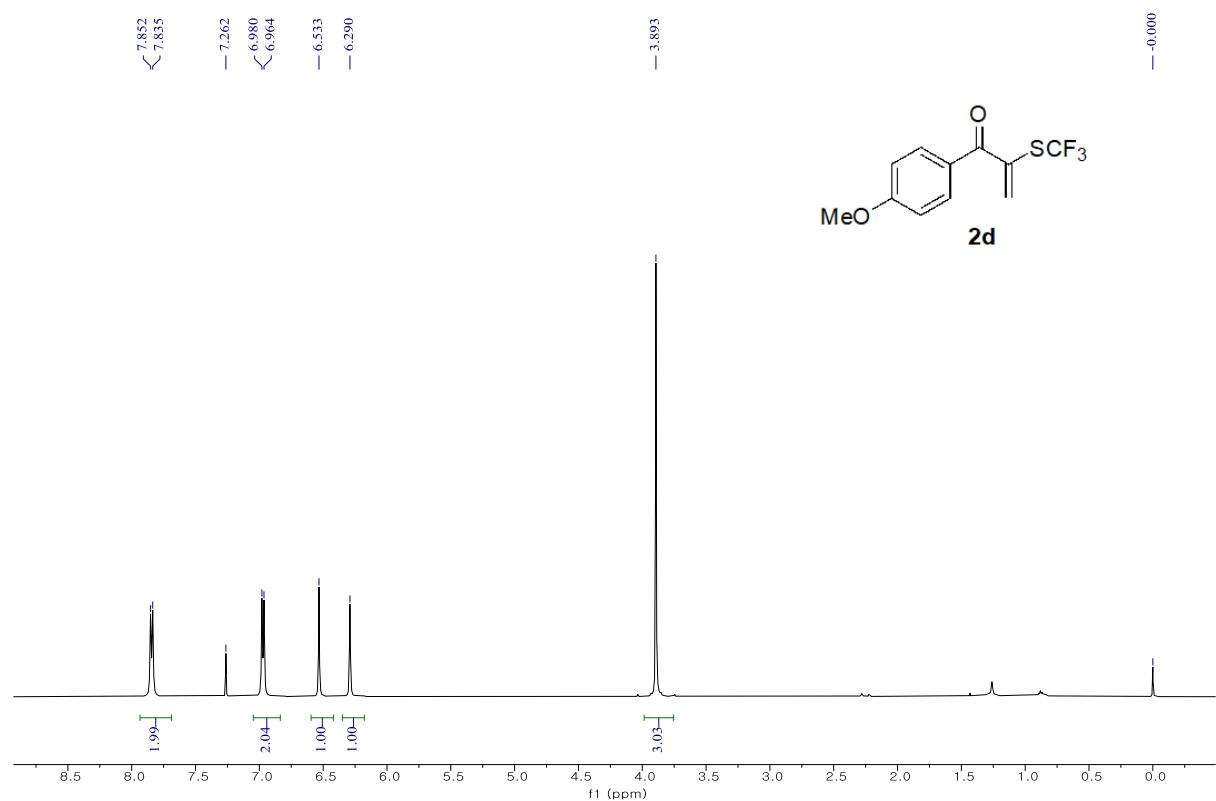
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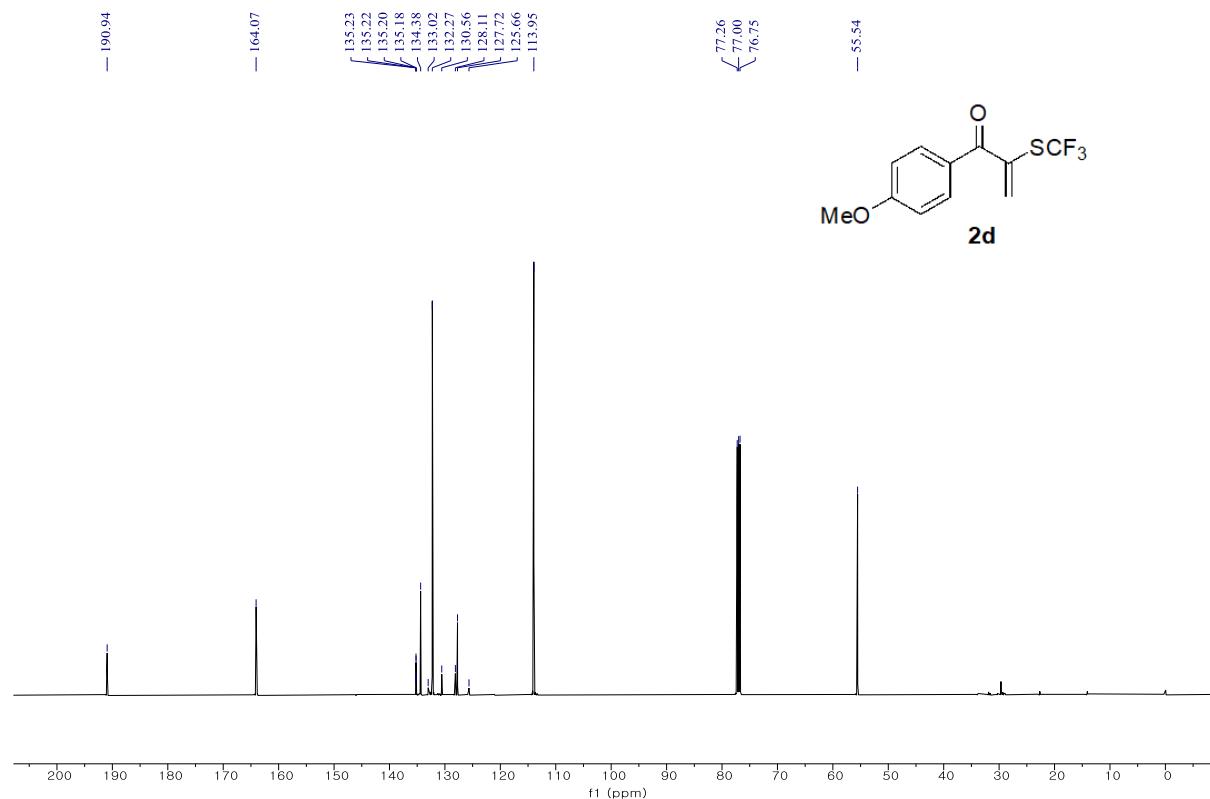
¹⁹F NMR of **2c** (564 MHz, CDCl₃)



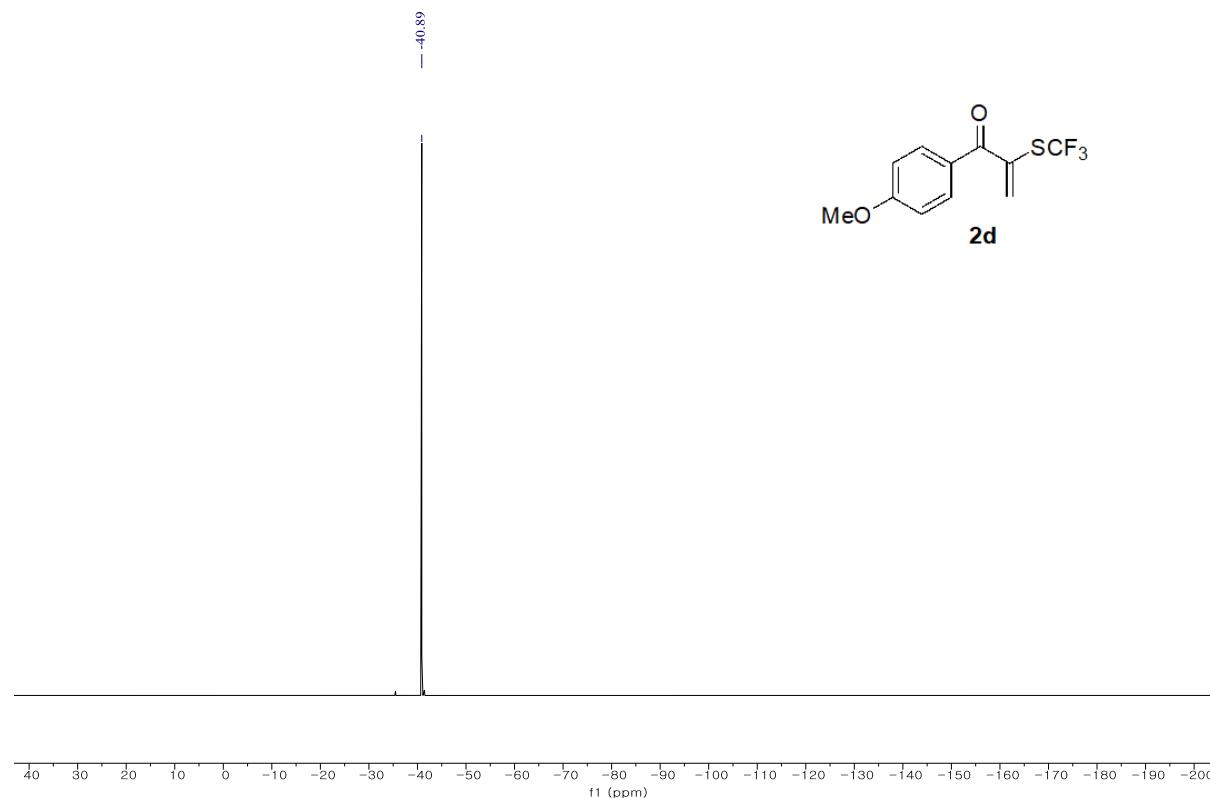
¹H NMR of **2d** (500 MHz, CDCl₃)



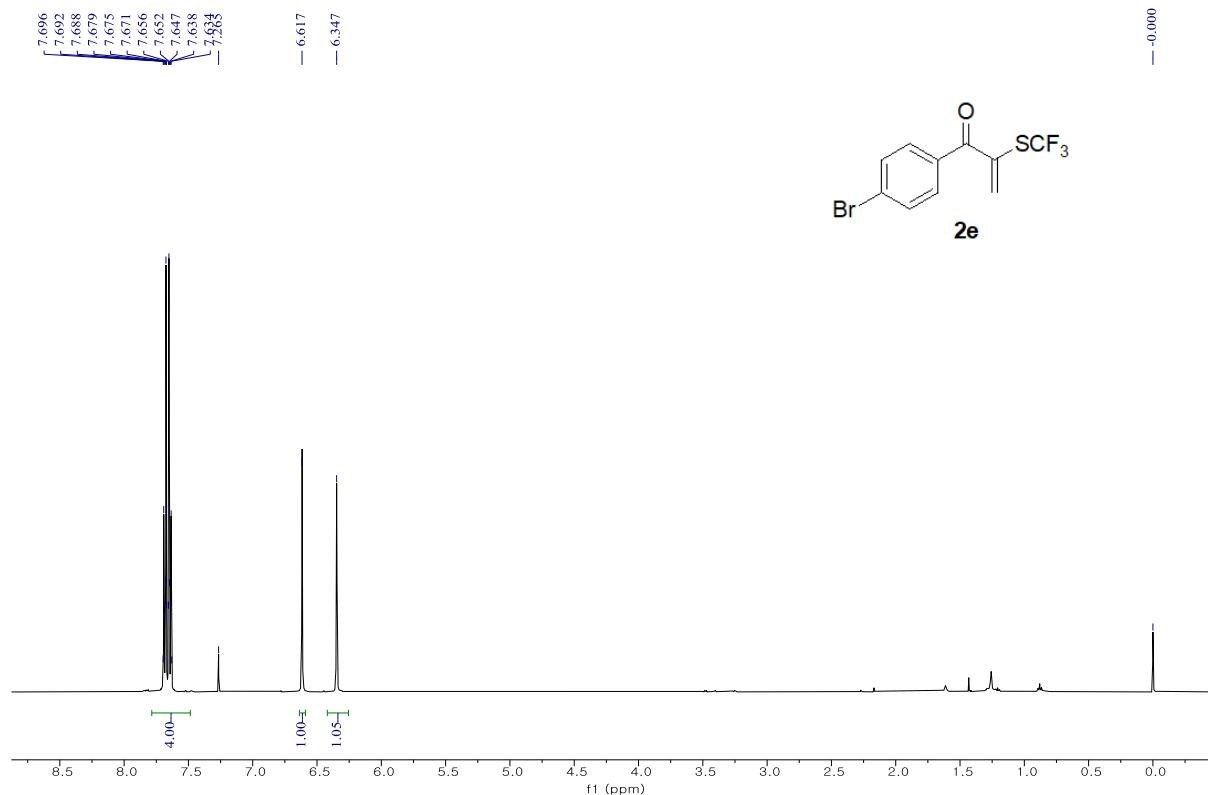
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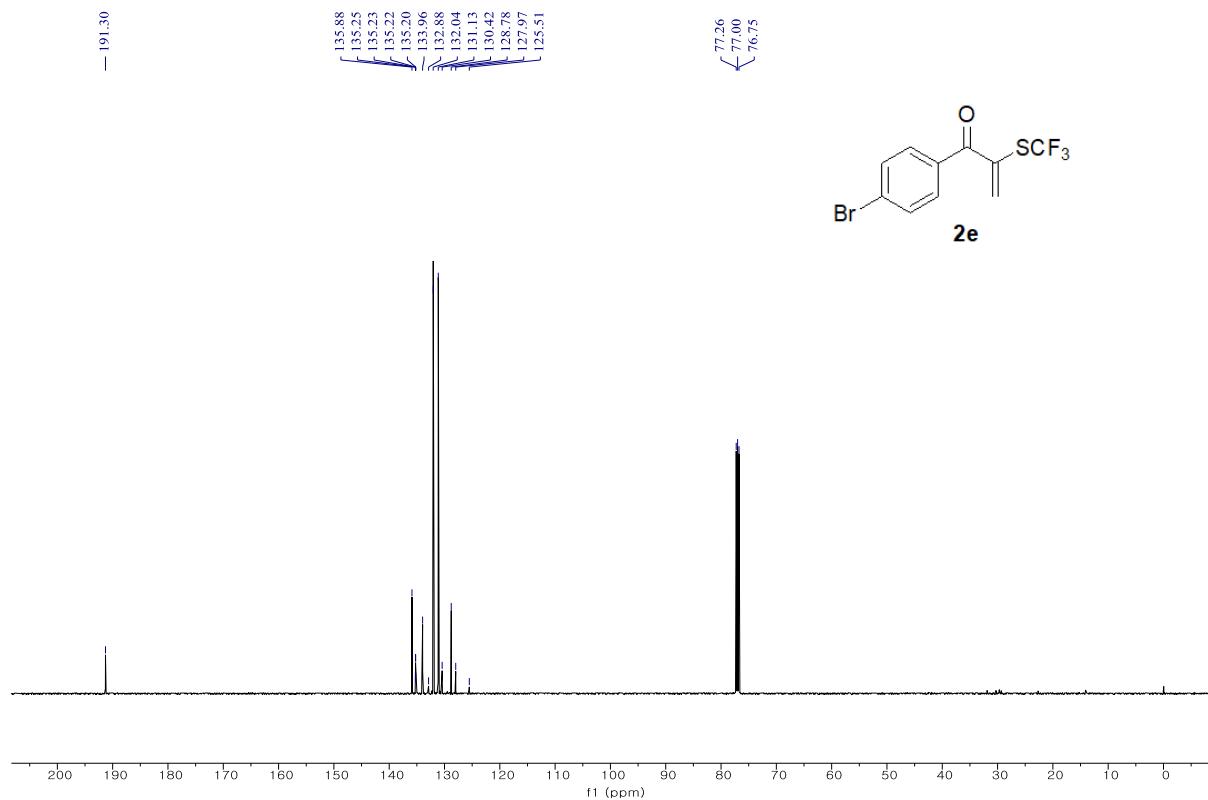
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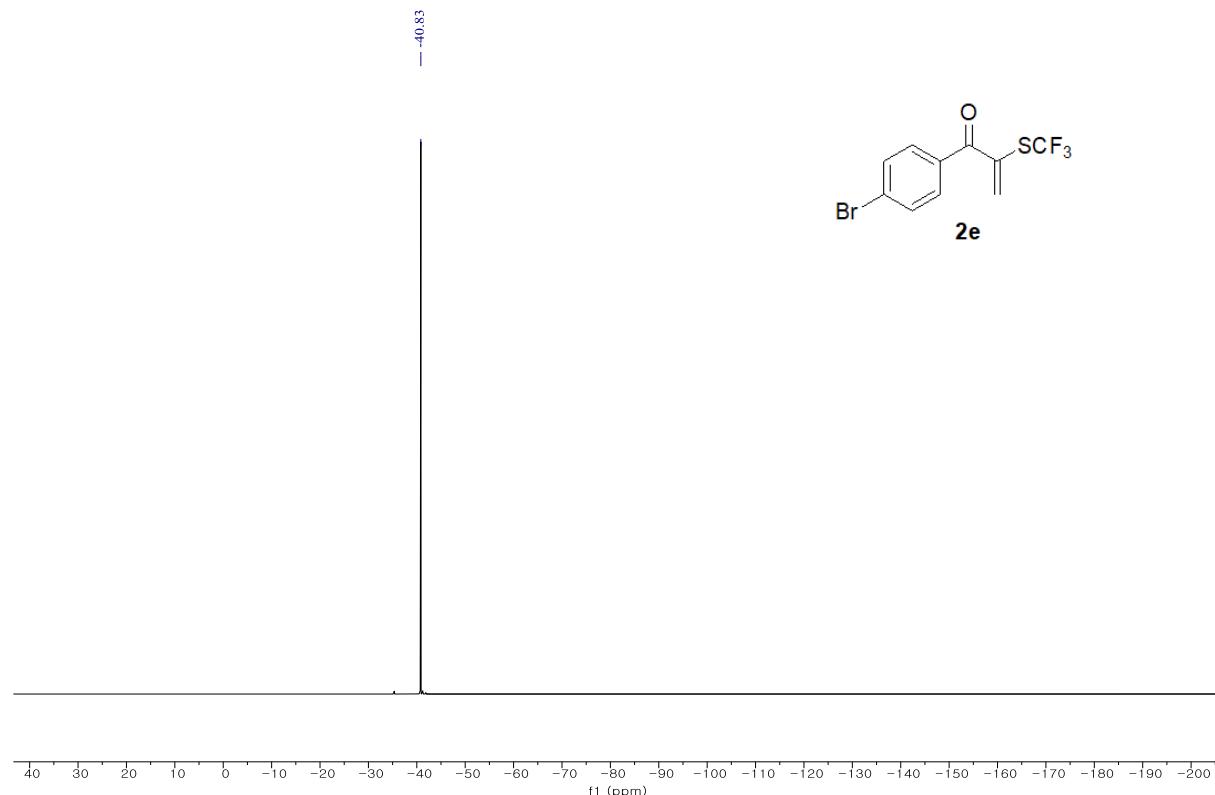
¹H NMR of **2e** (500 MHz, CDCl₃)



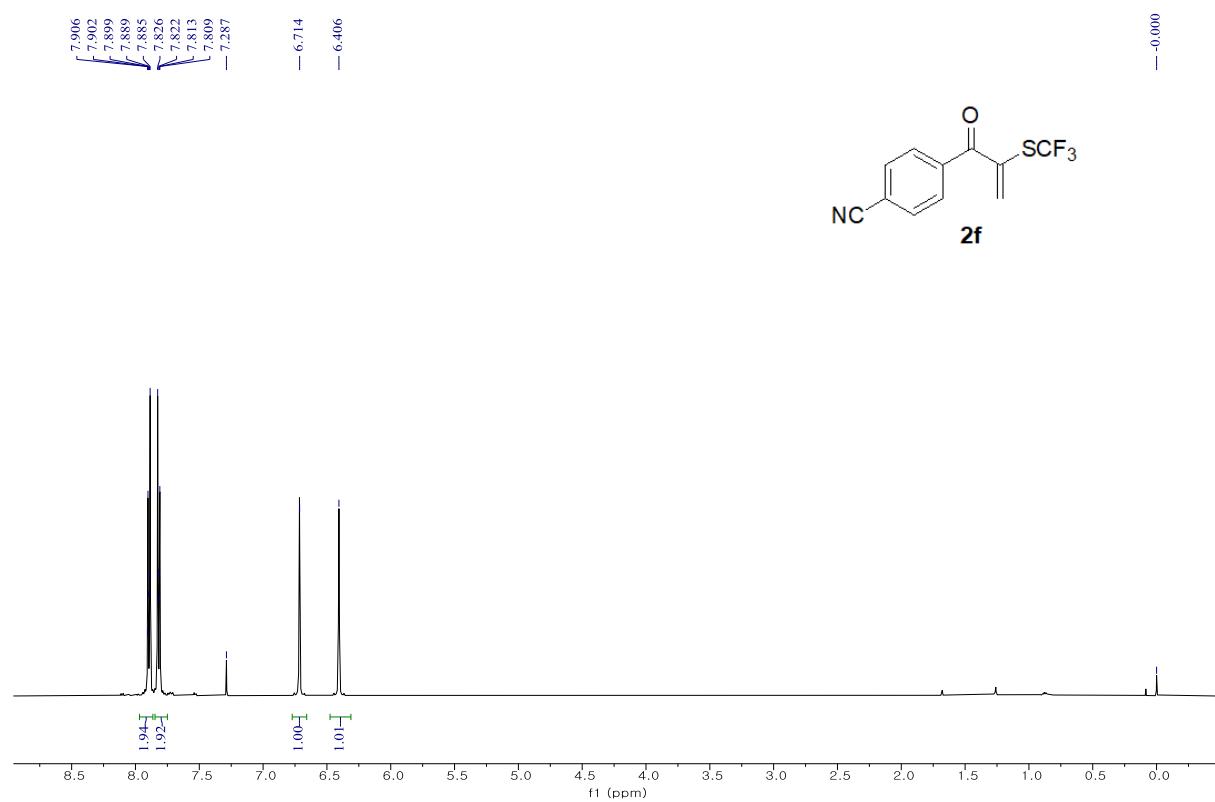
¹³C NMR of **2e** (125 MHz, CDCl₃)



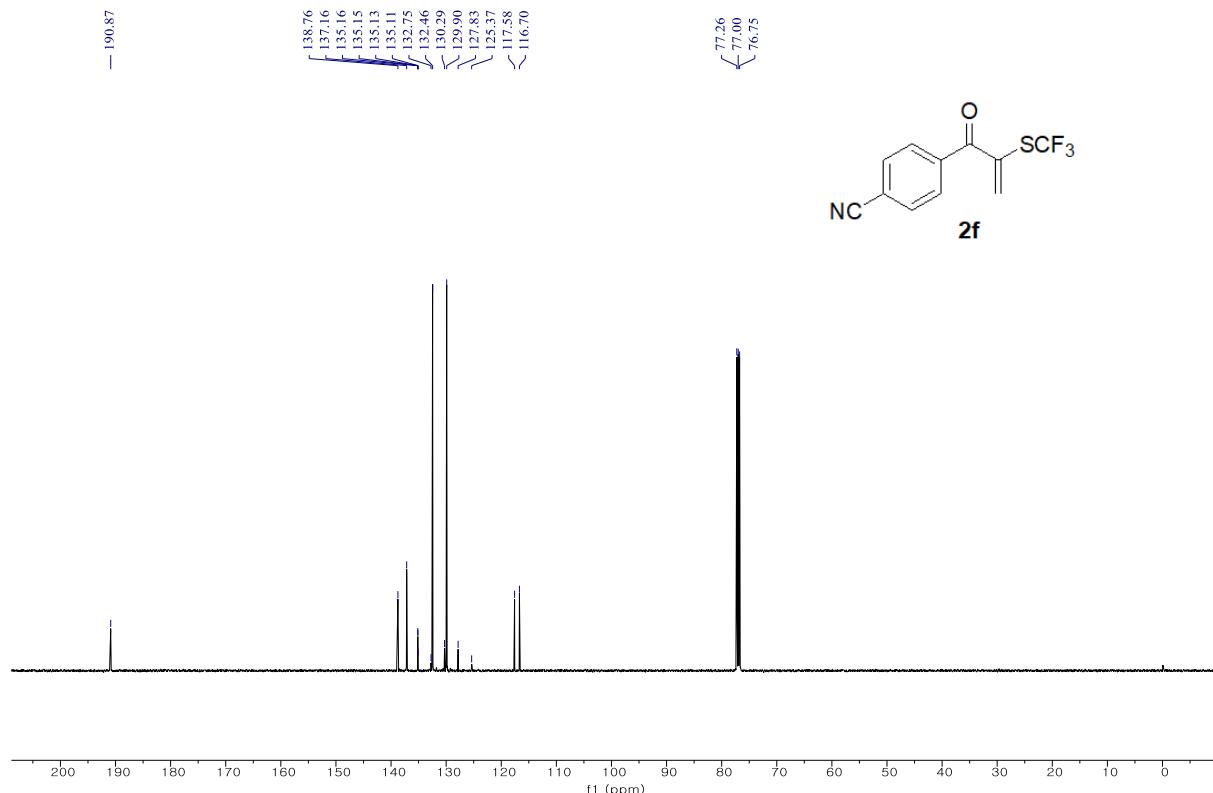
¹⁹F NMR of **2e** (564 MHz, CDCl₃)



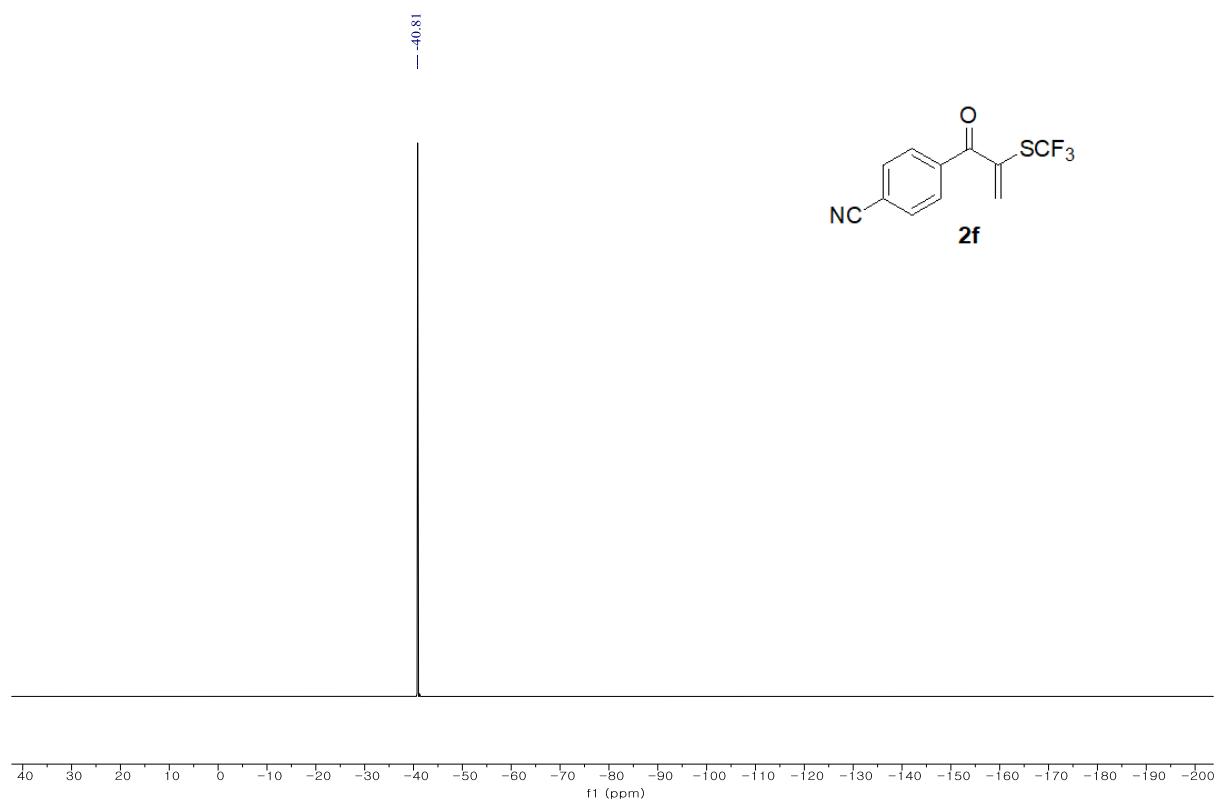
¹H NMR of **2f** (500 MHz, CDCl₃)



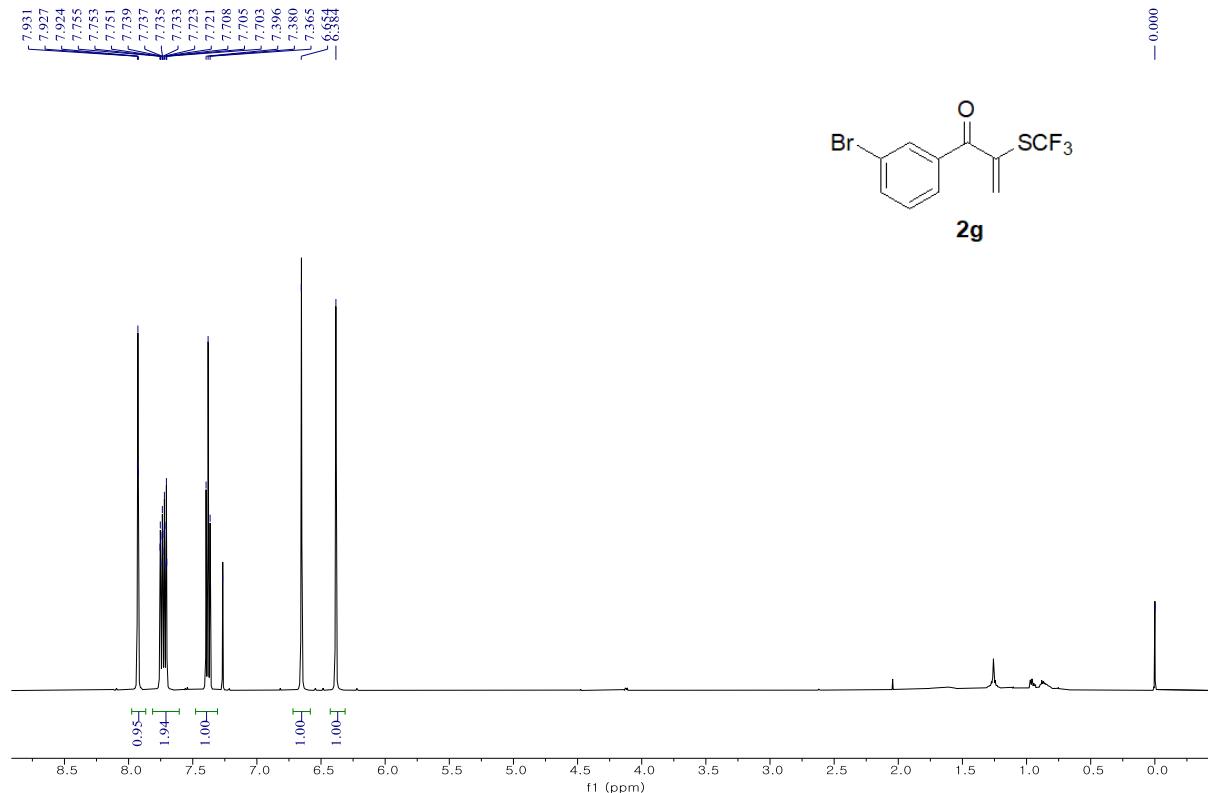
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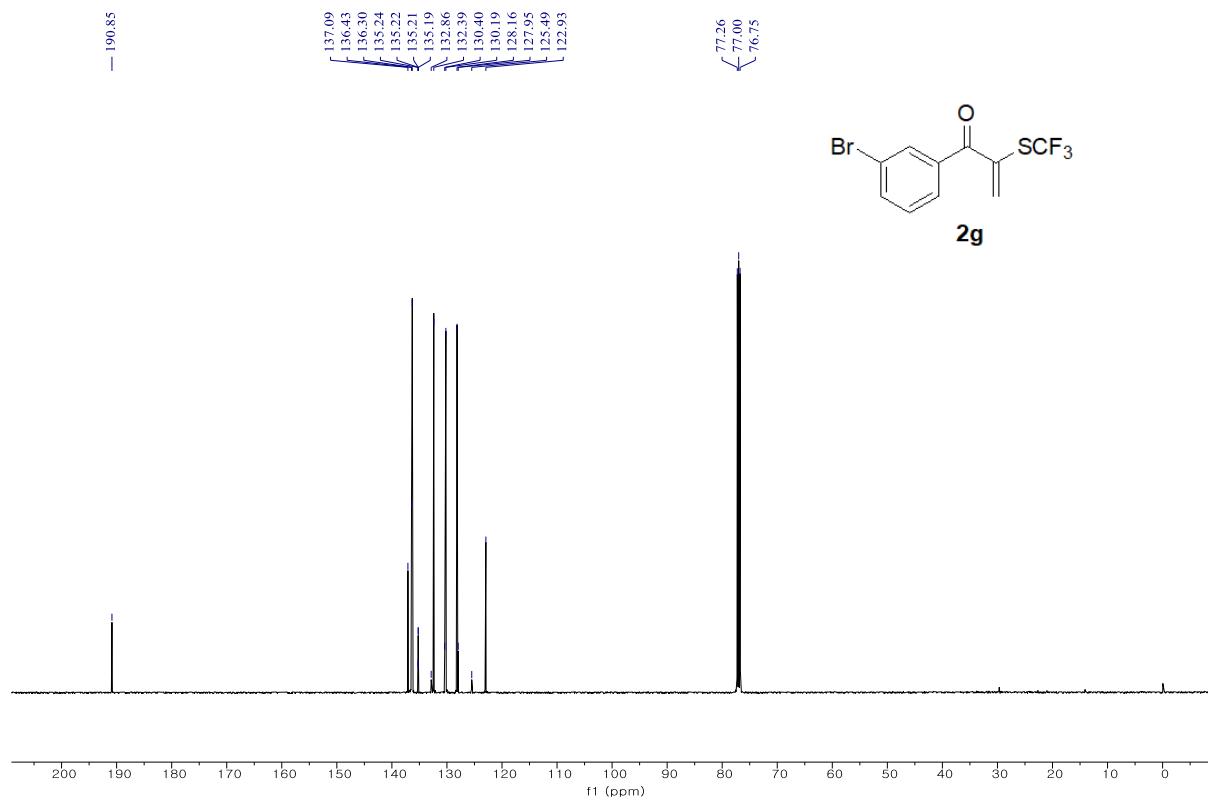
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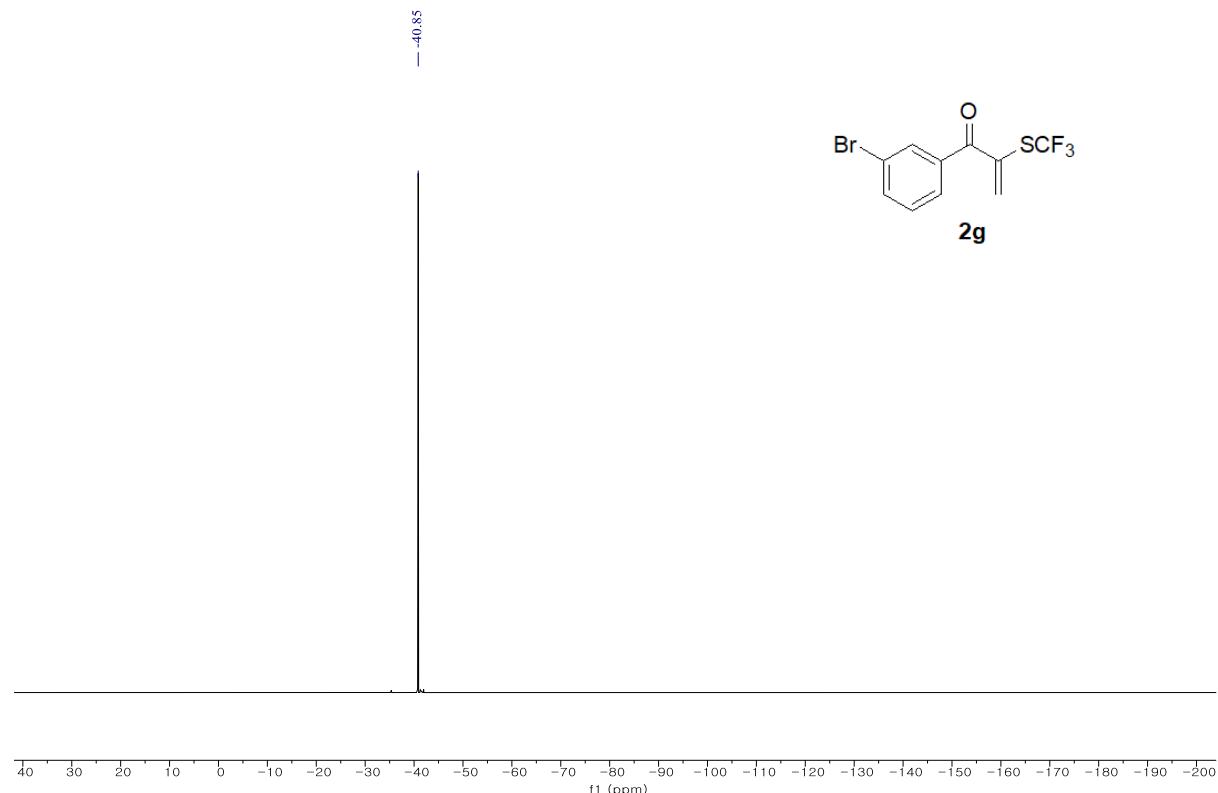
¹H NMR of 2g (500 MHz, CDCl₃)



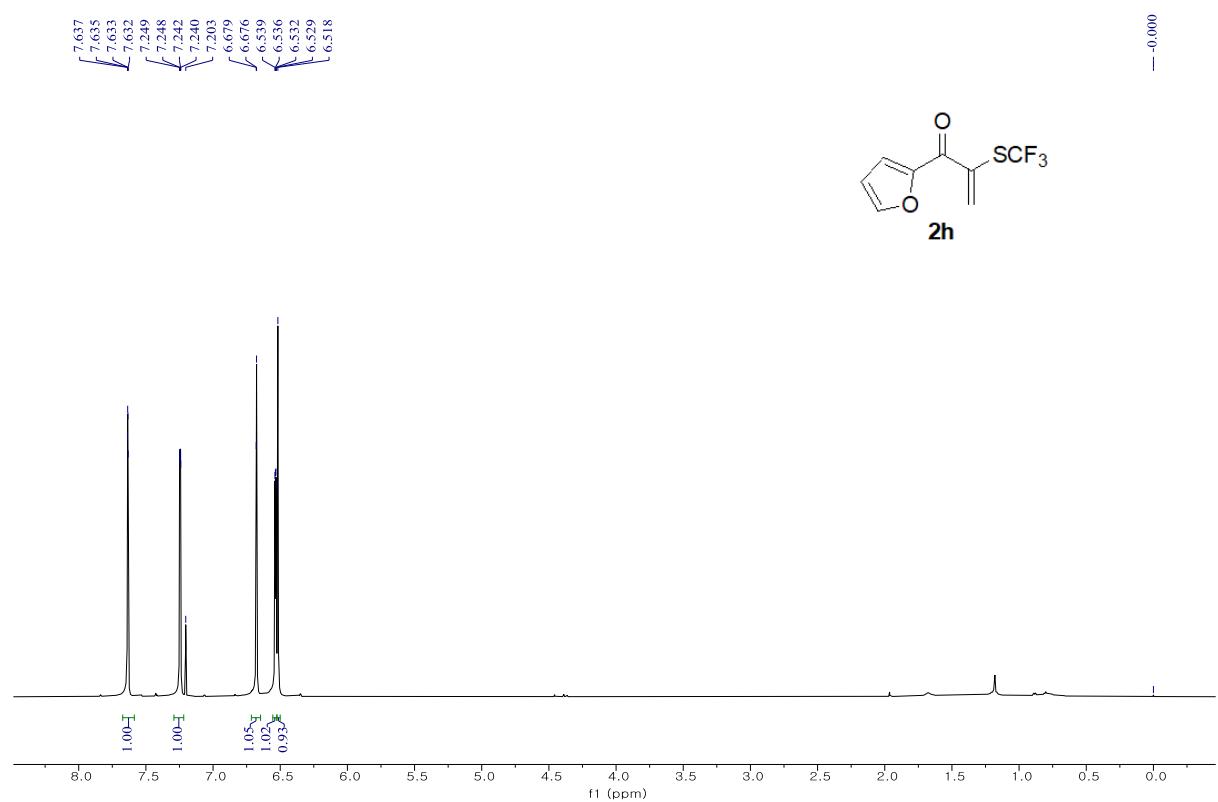
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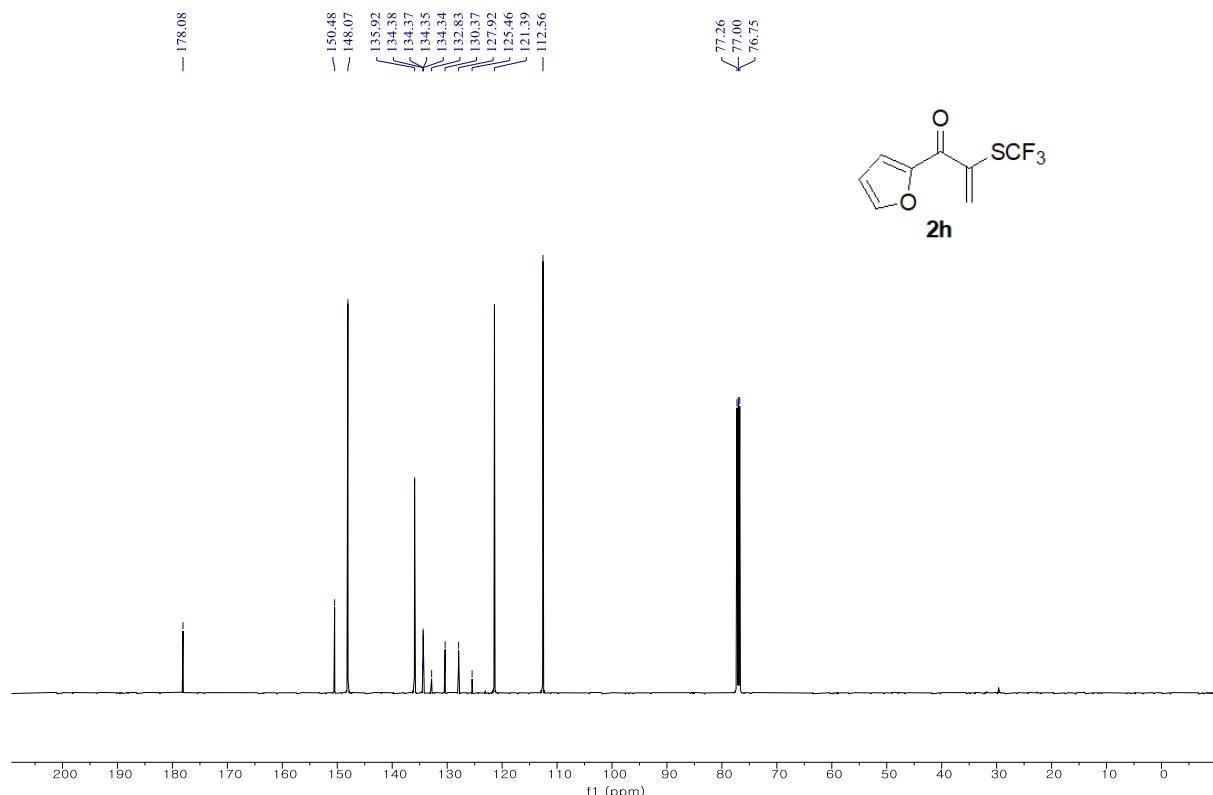
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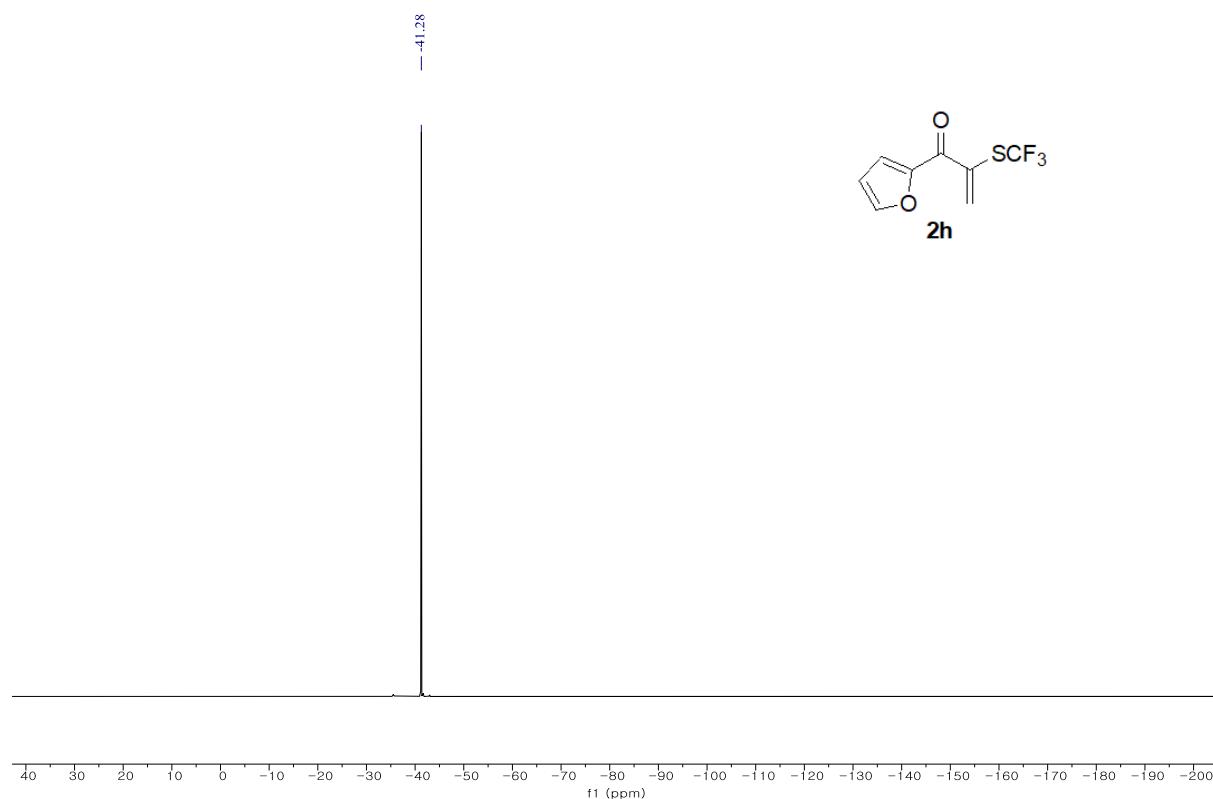
¹H NMR of **2h** (500 MHz, CDCl₃)



¹³C NMR of **2h** (125 MHz, CDCl₃)



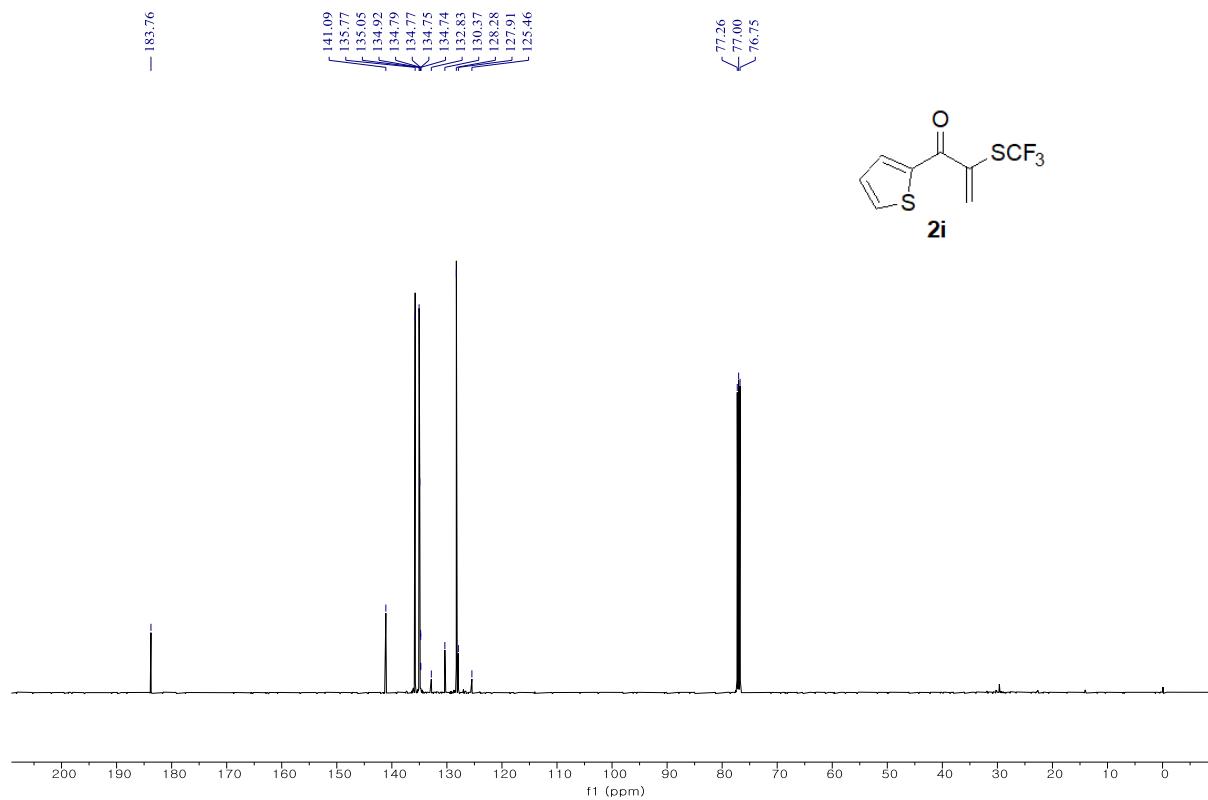
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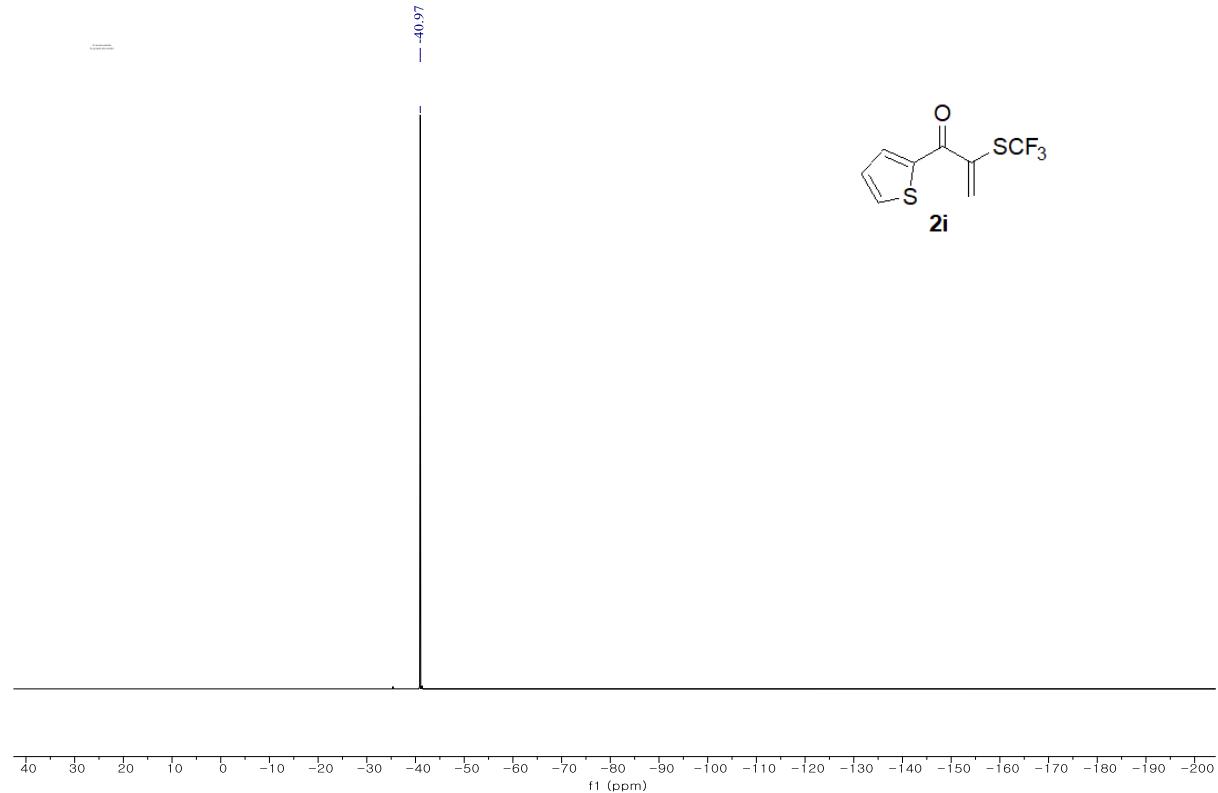
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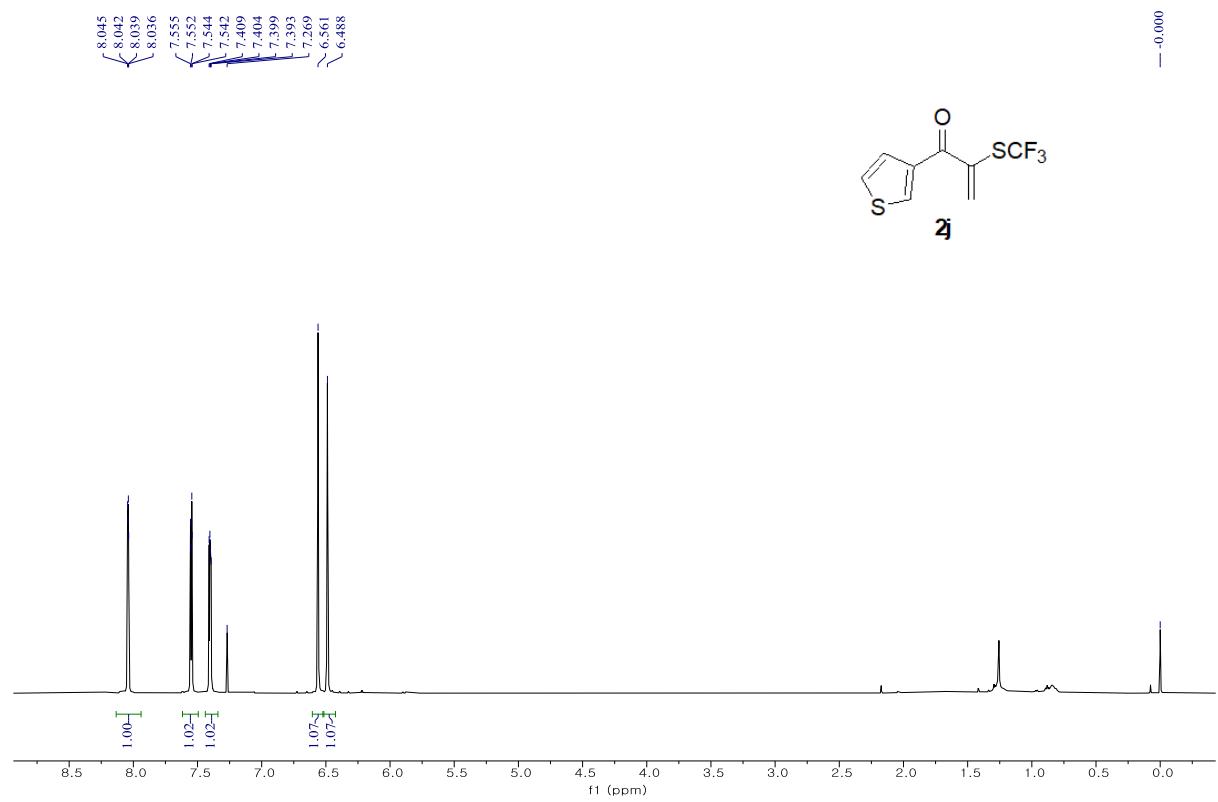
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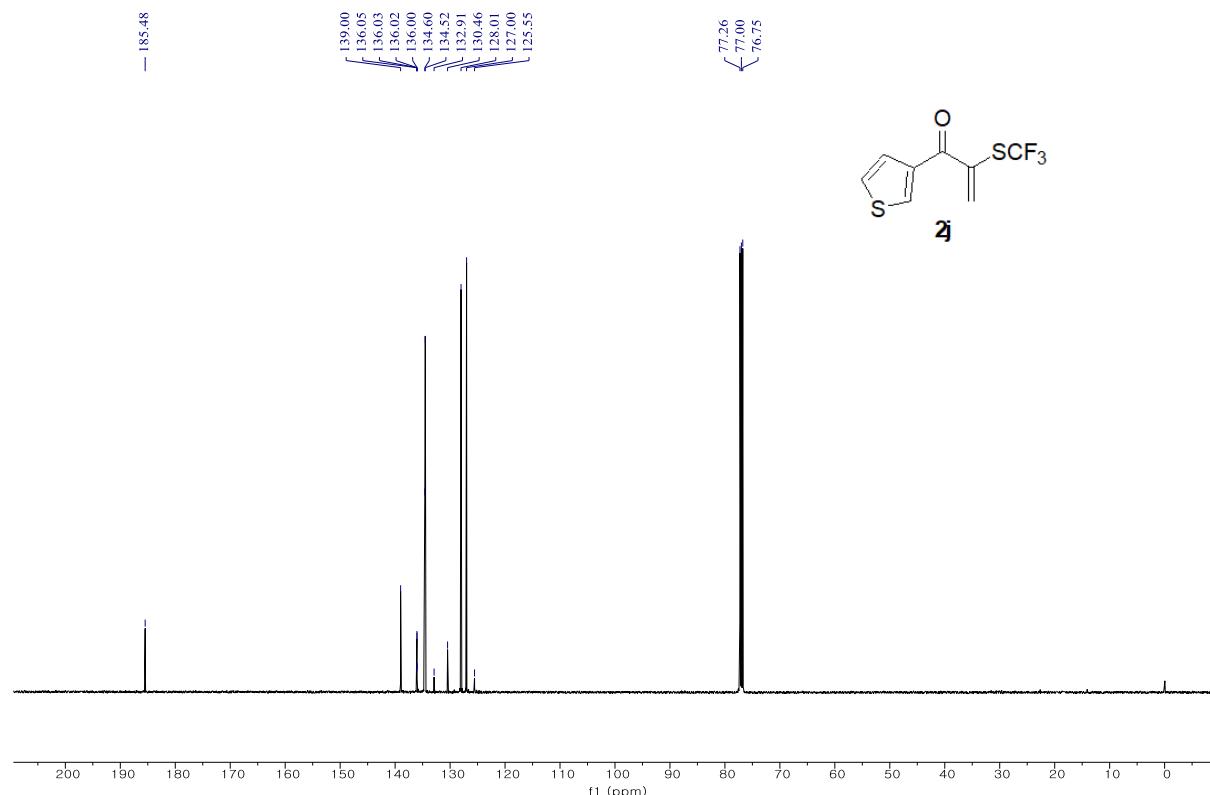
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¹H NMR of **2j** (500 MHz, CDCl₃)



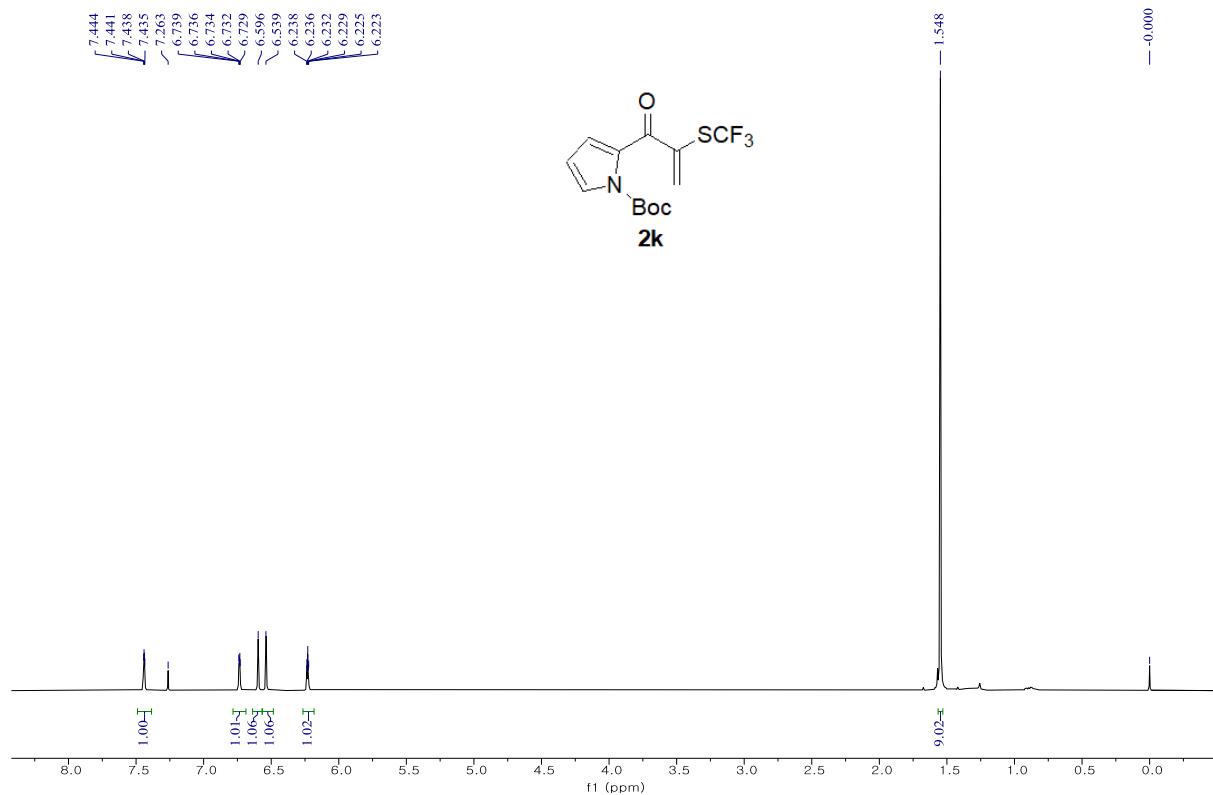
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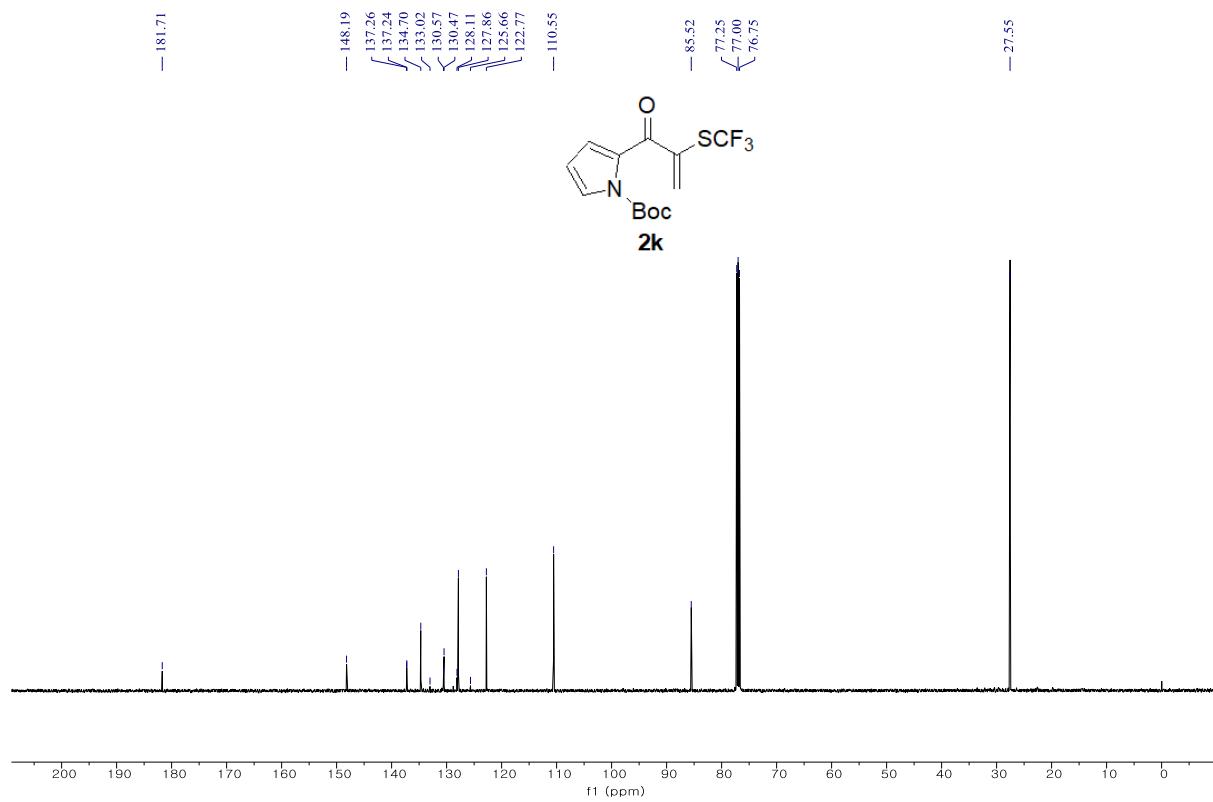
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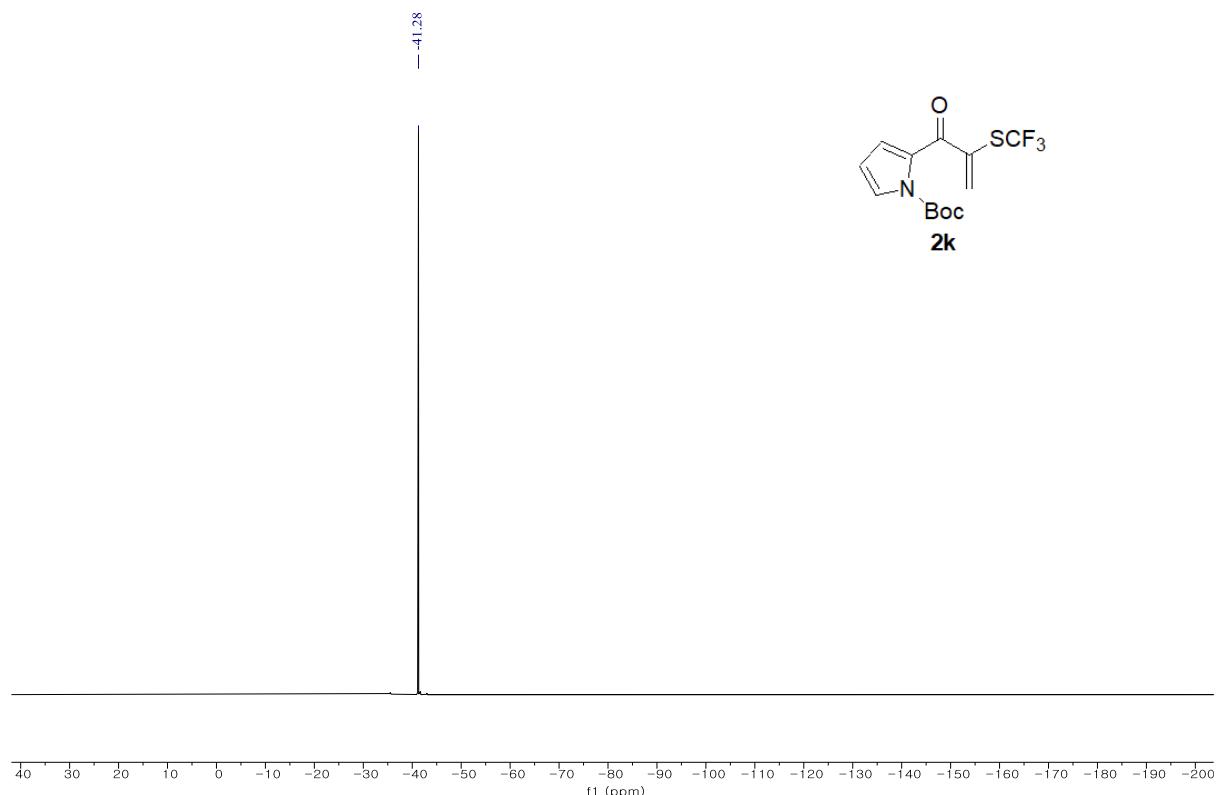
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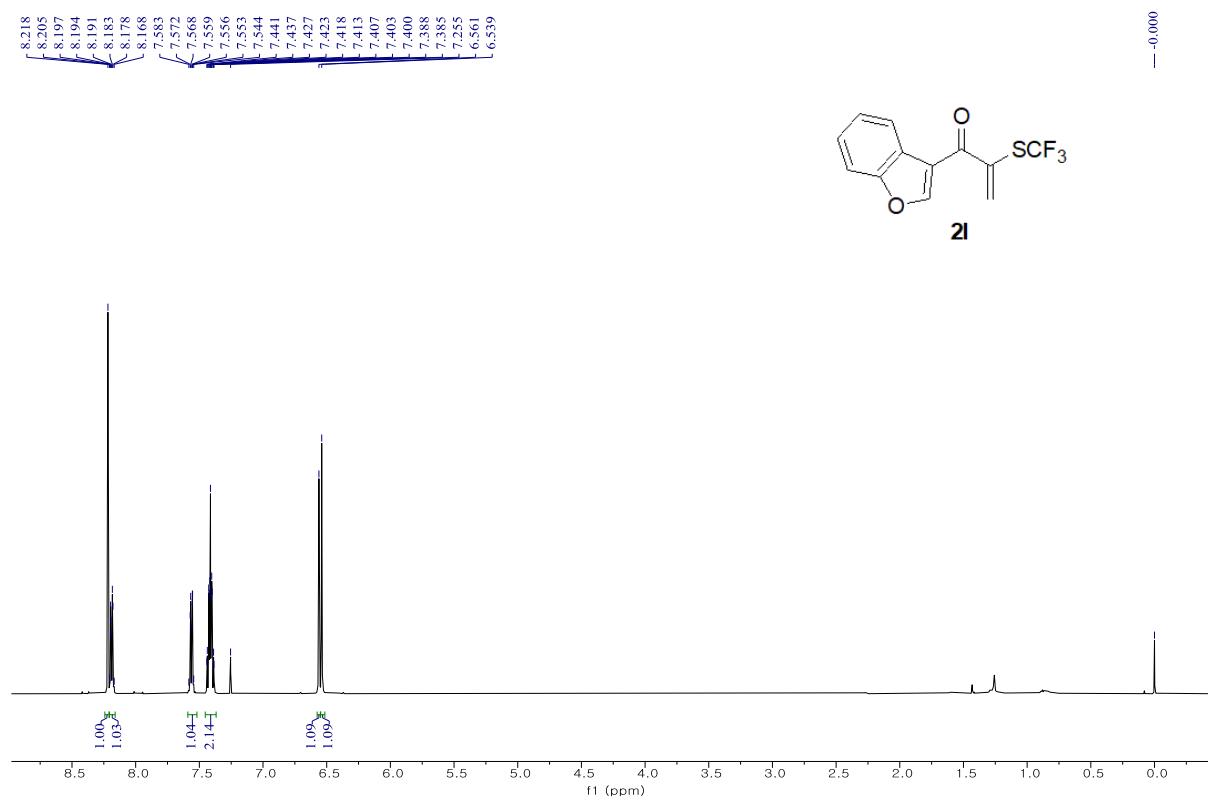
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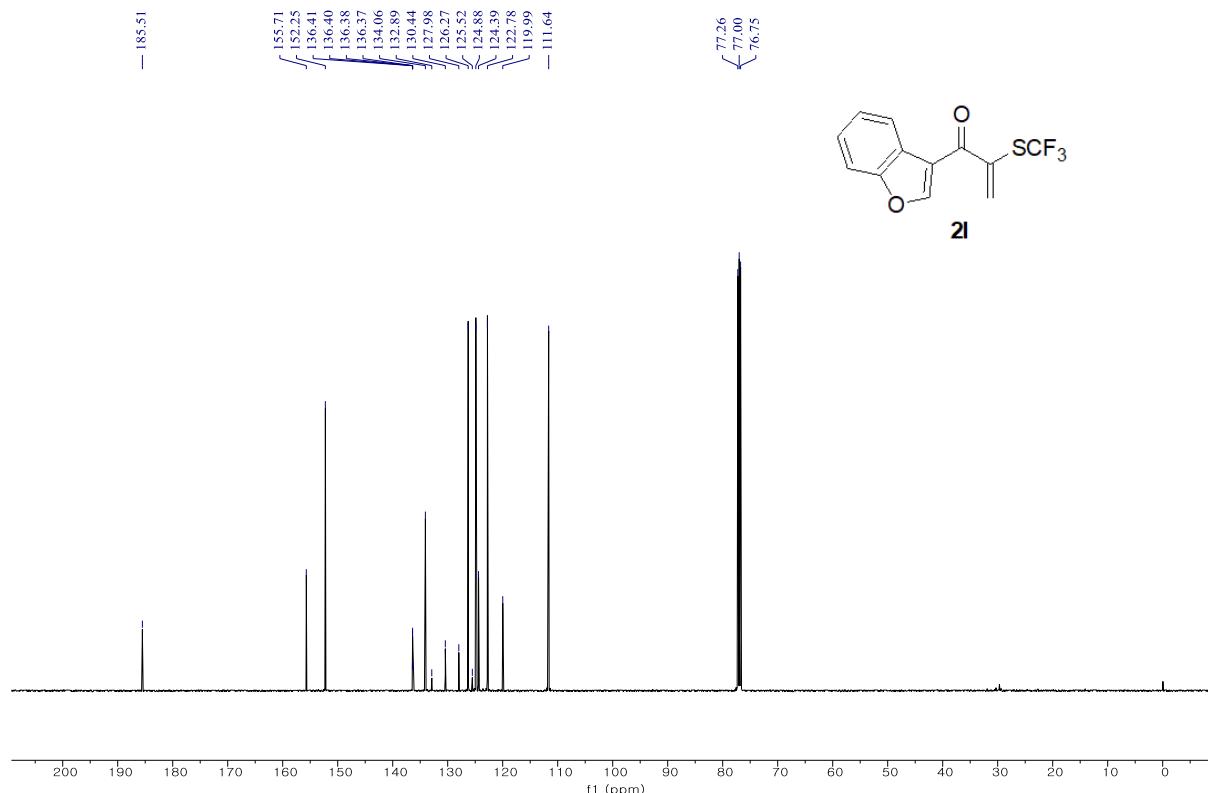
¹⁹F NMR of **2k** (564 MHz, CDCl₃)



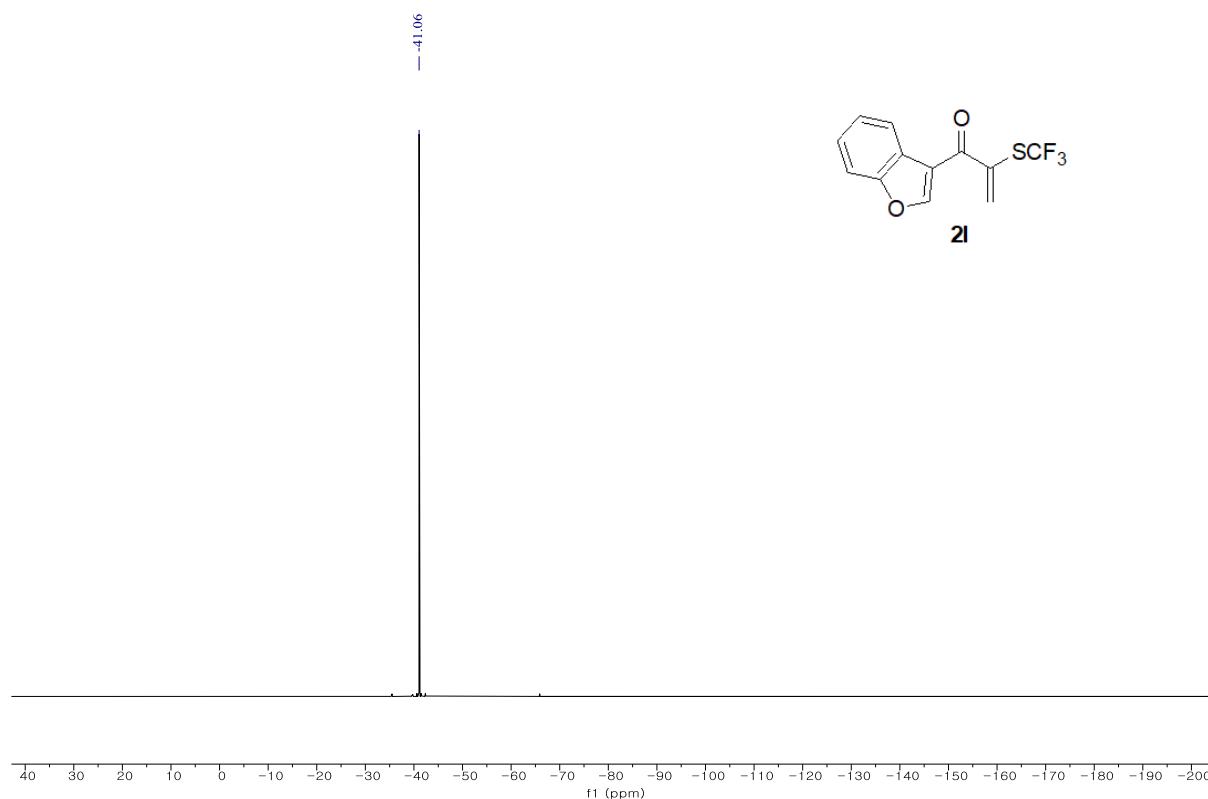
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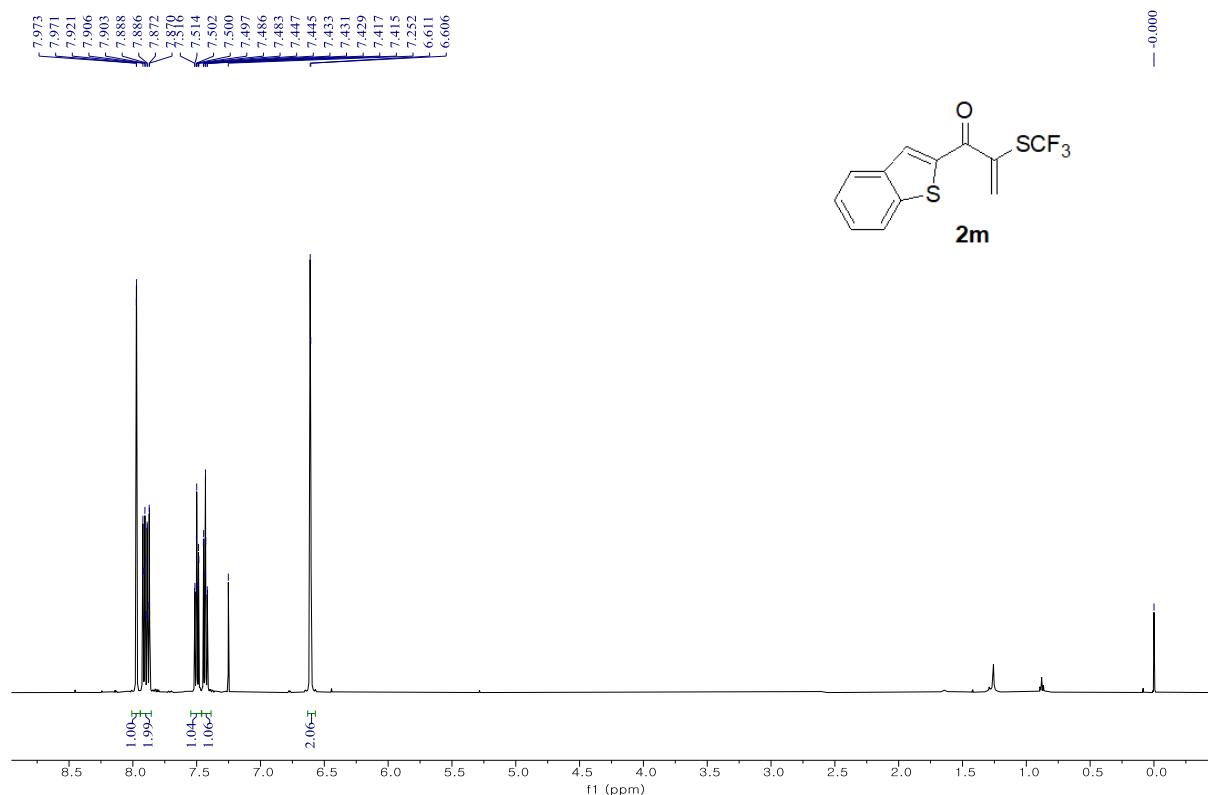
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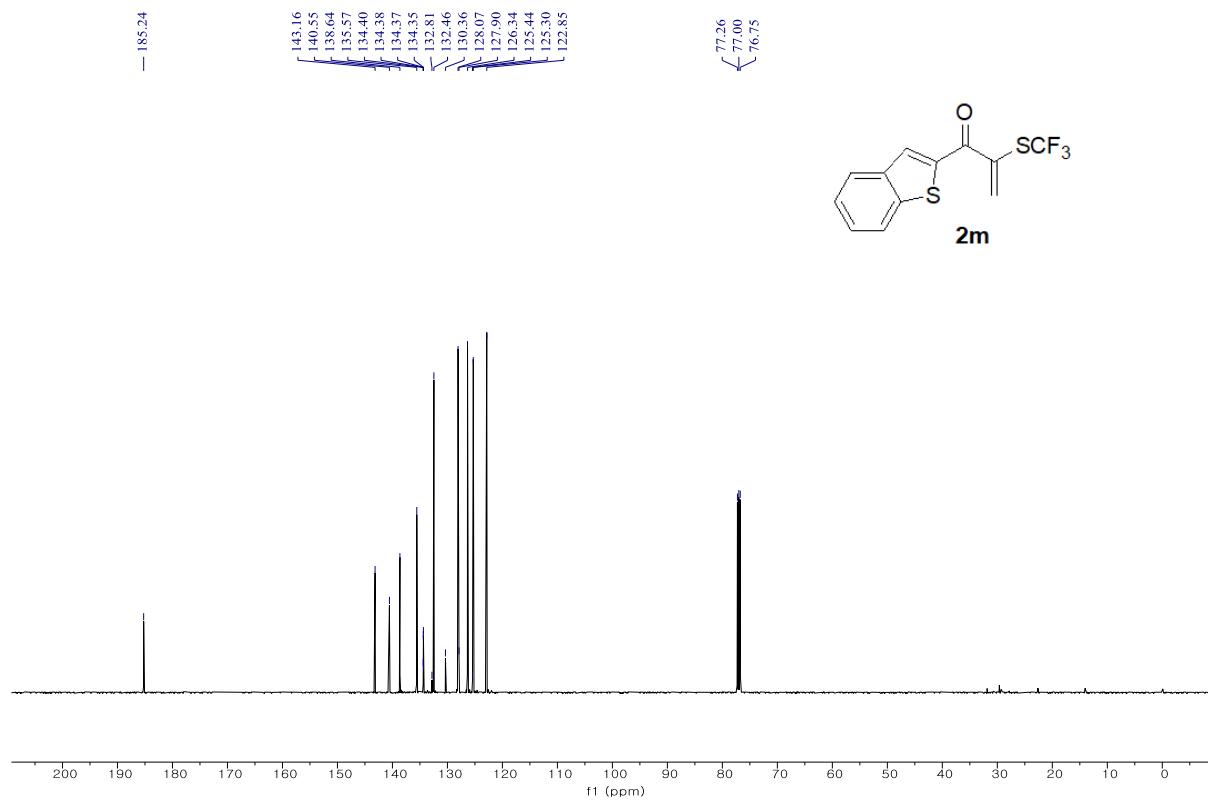
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¹H NMR of **2m** (500 MHz, CDCl₃)



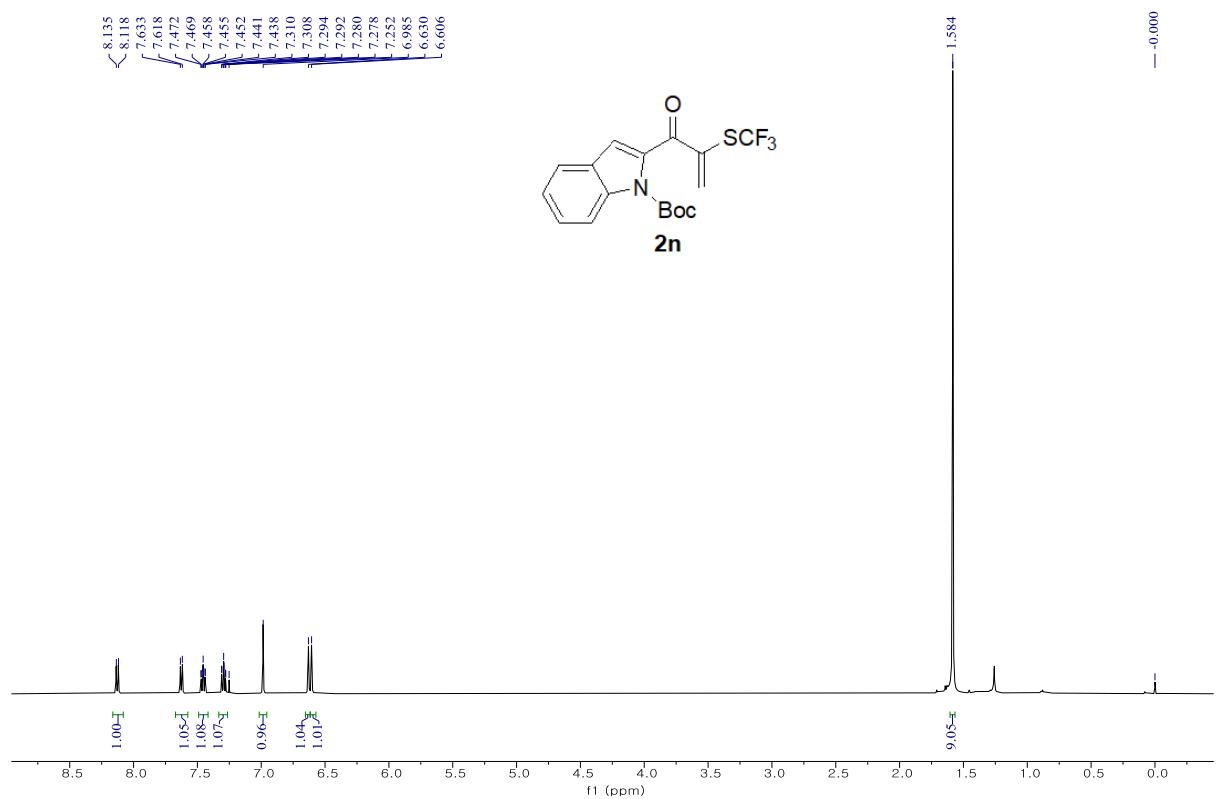
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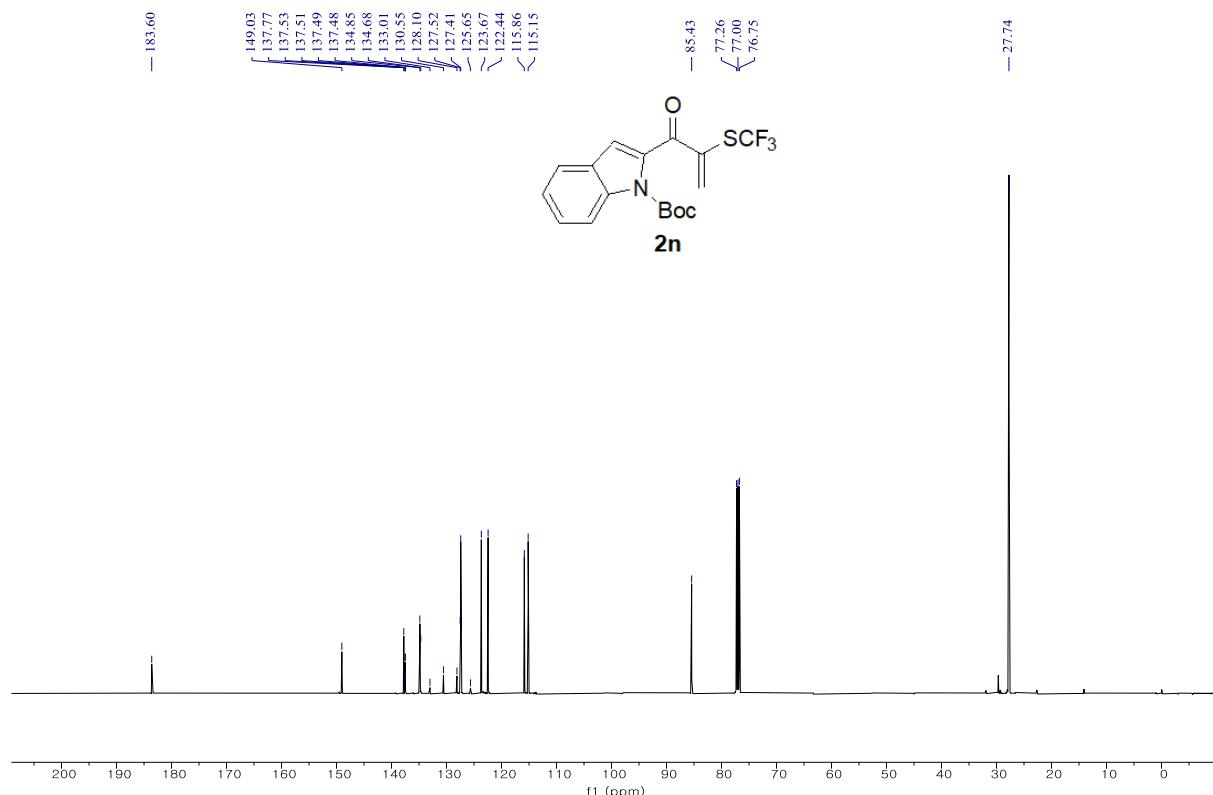
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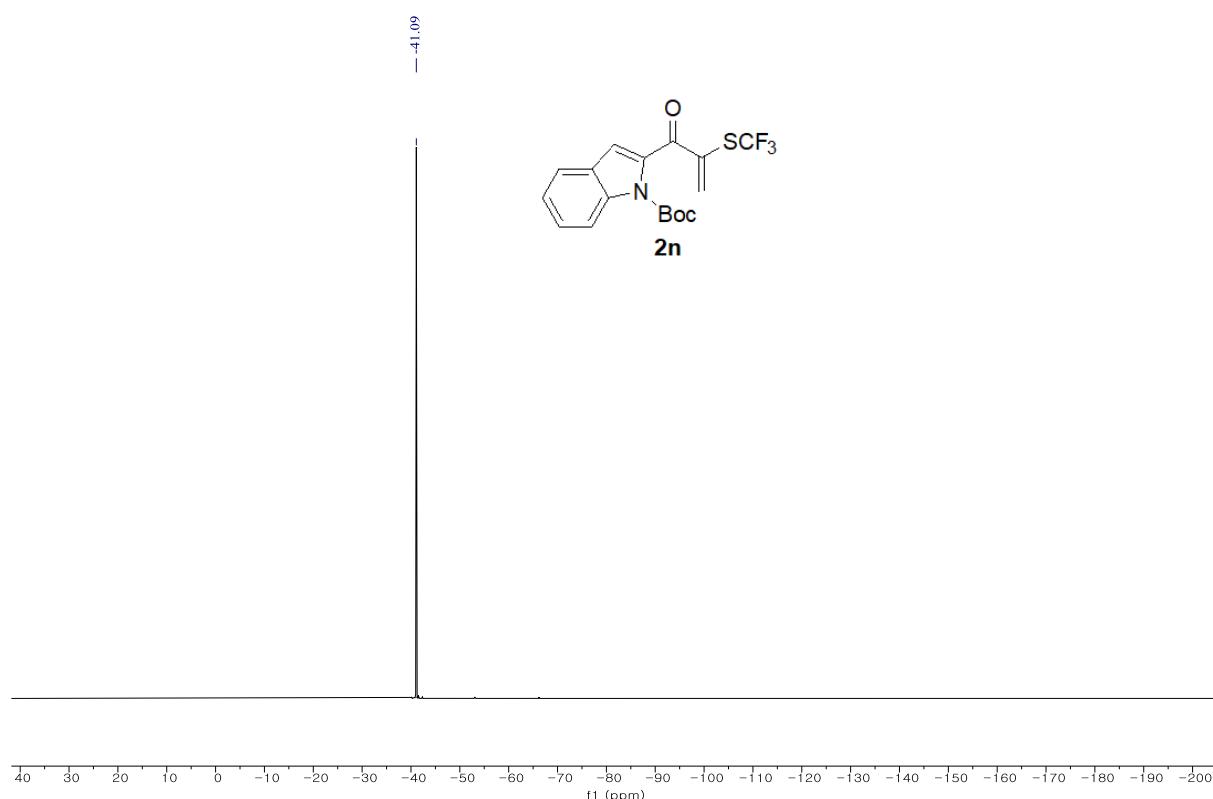
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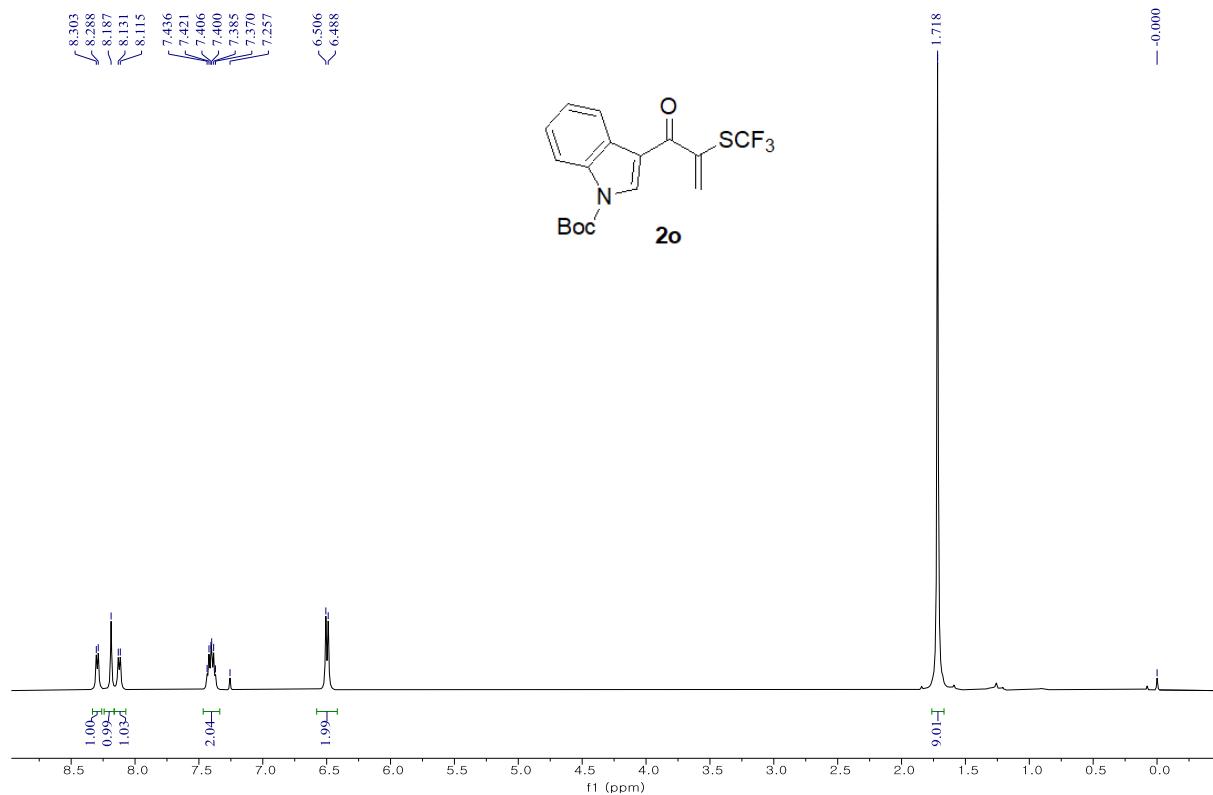
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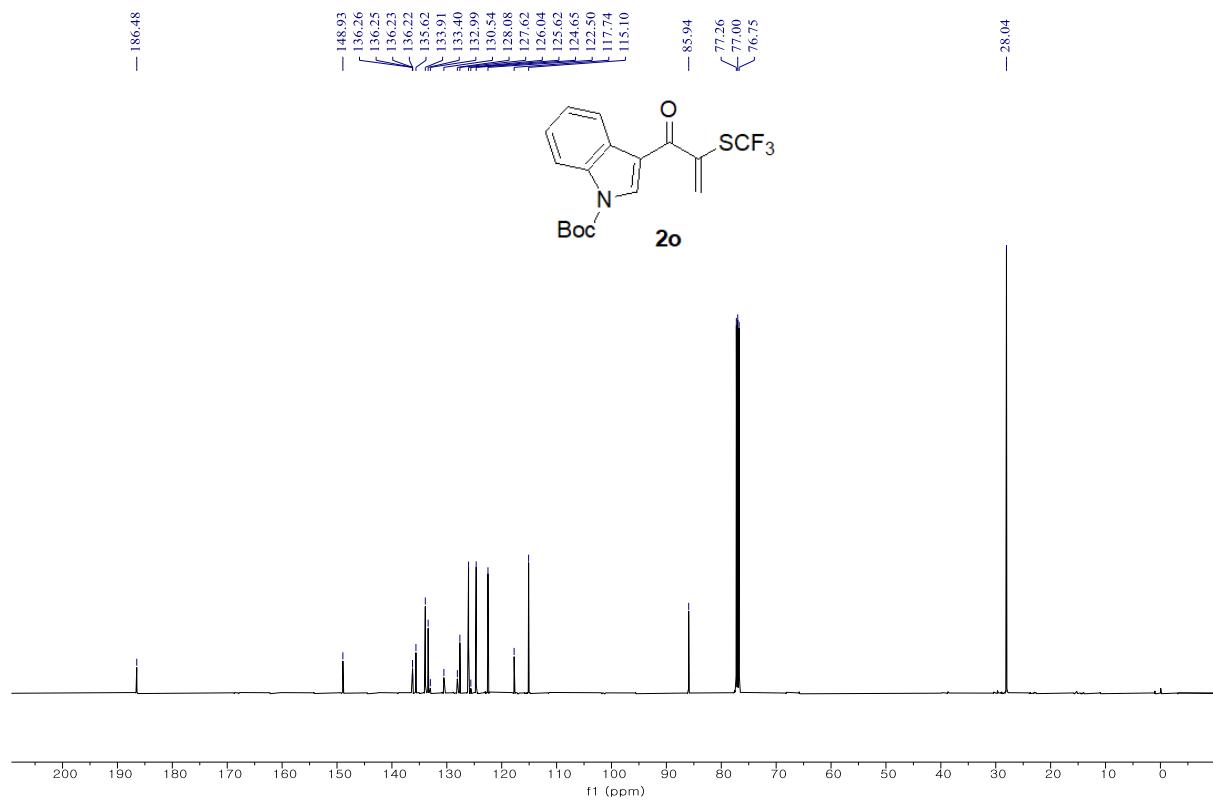
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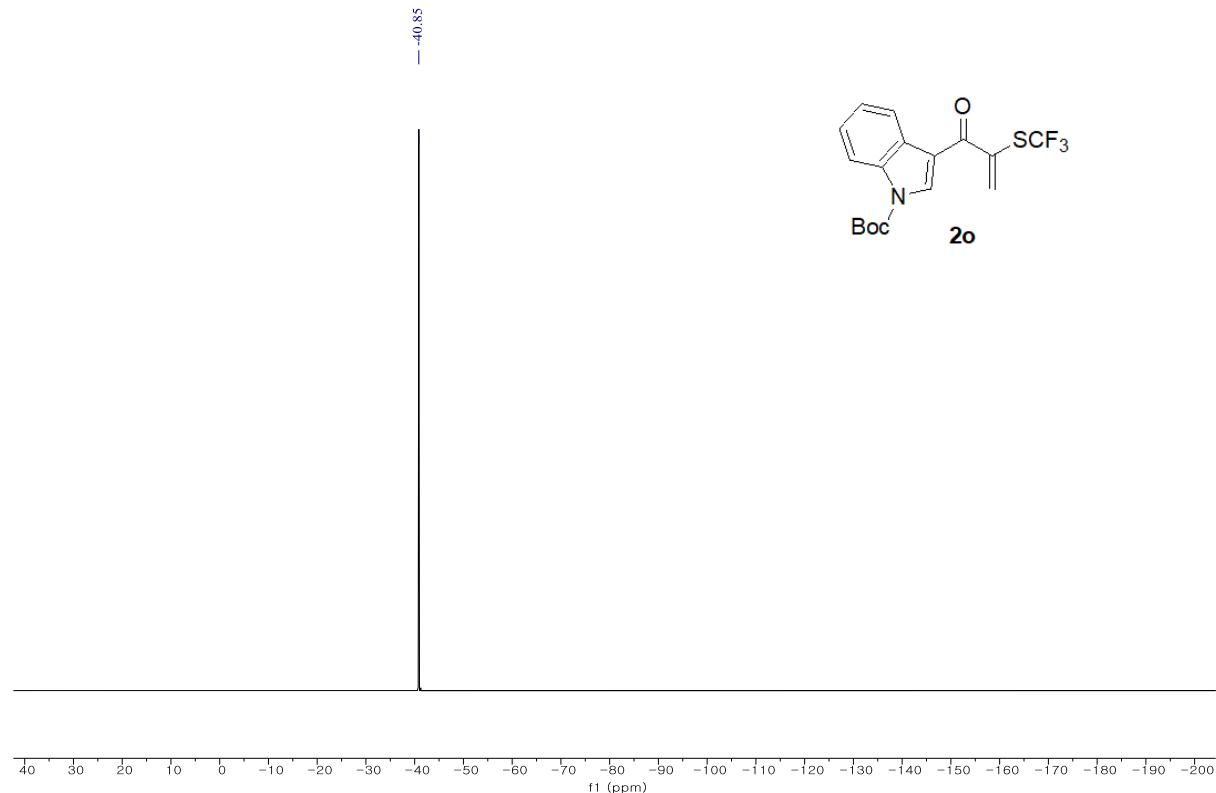
¹H NMR of **2o** (500 MHz, CDCl₃)



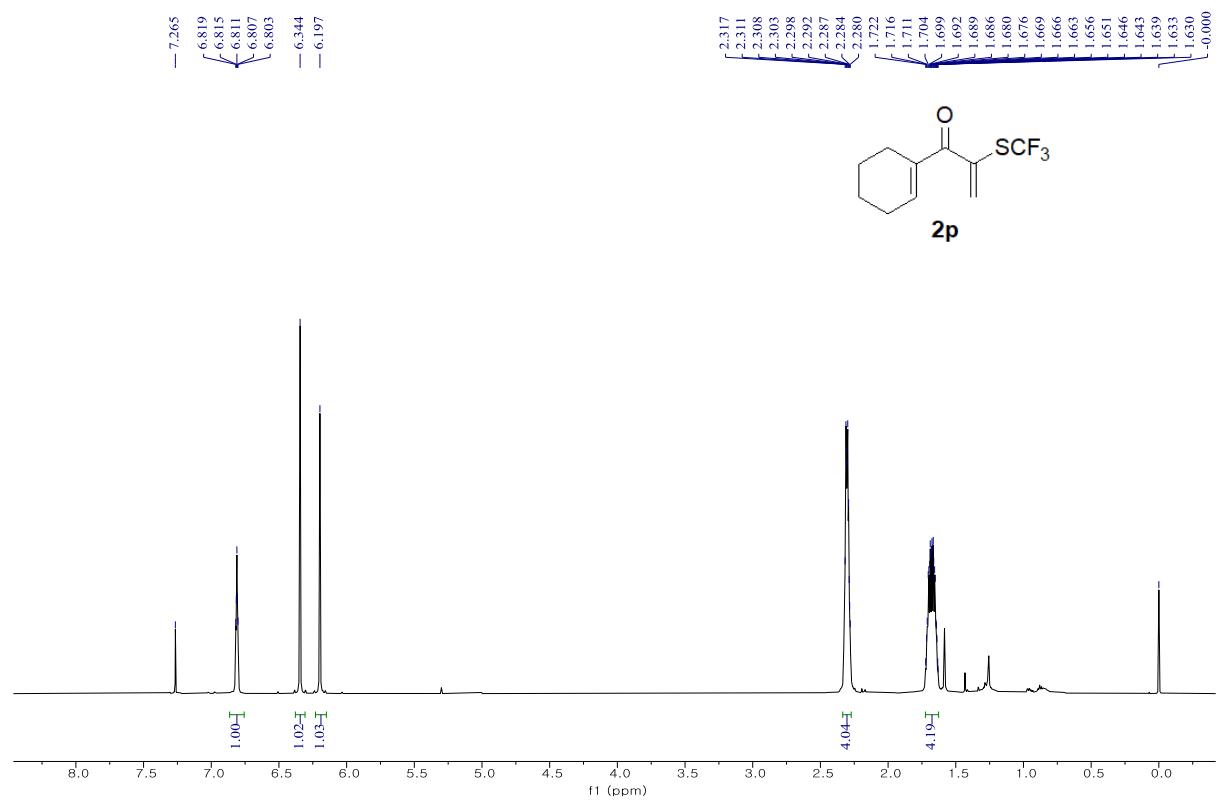
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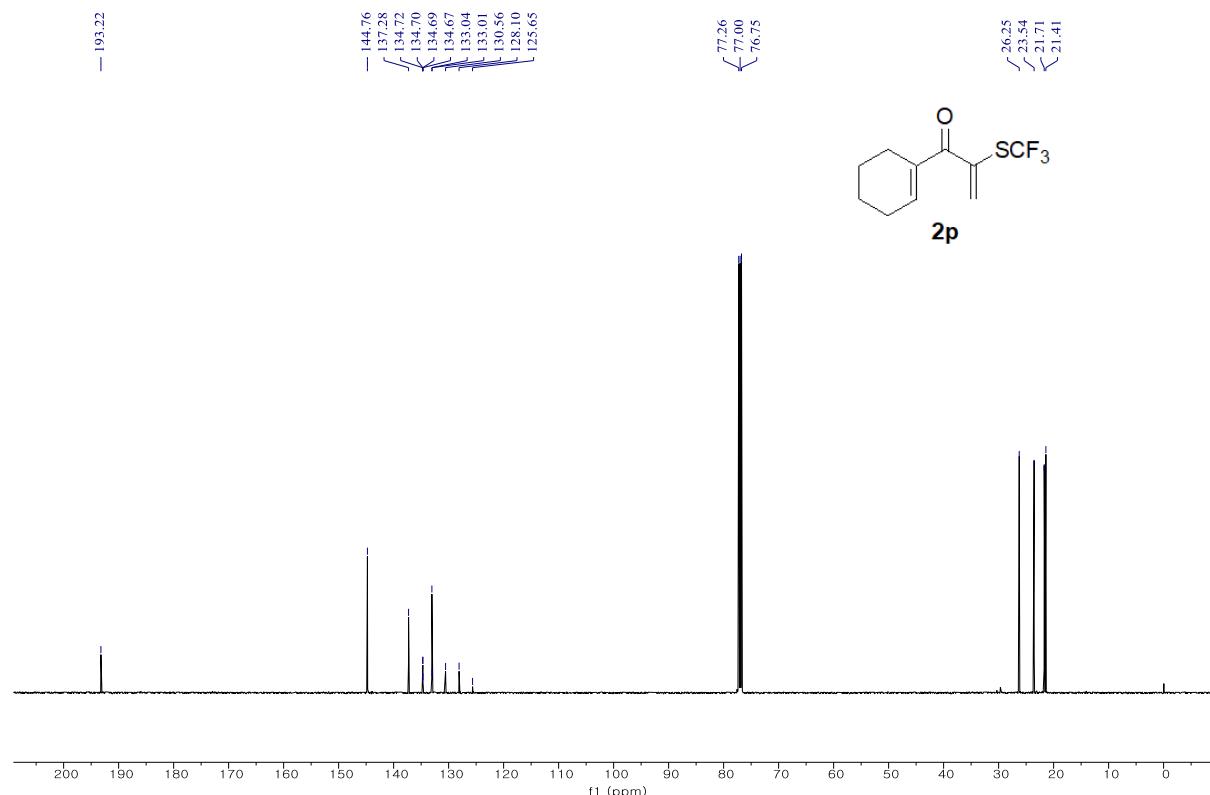
¹⁹F NMR of **2o** (564 MHz, CDCl₃)



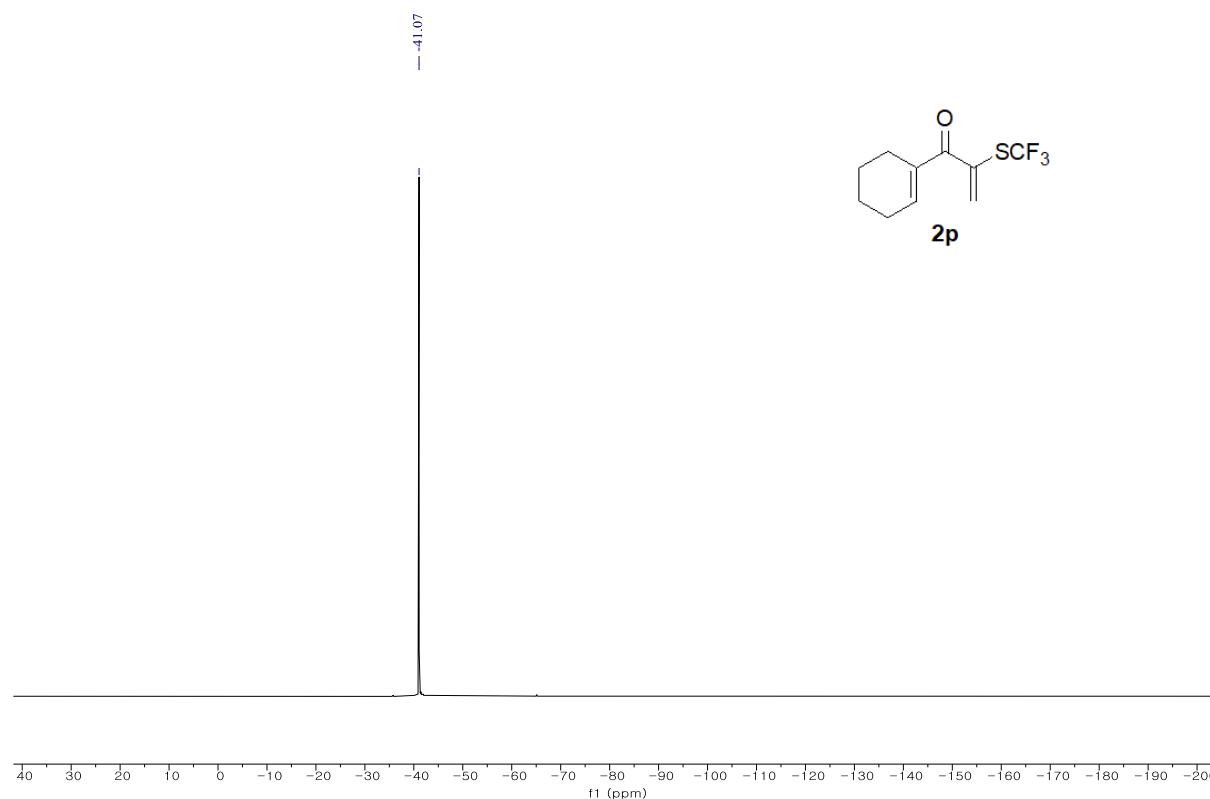
¹H NMR of **2p** (500 MHz, CDCl₃)



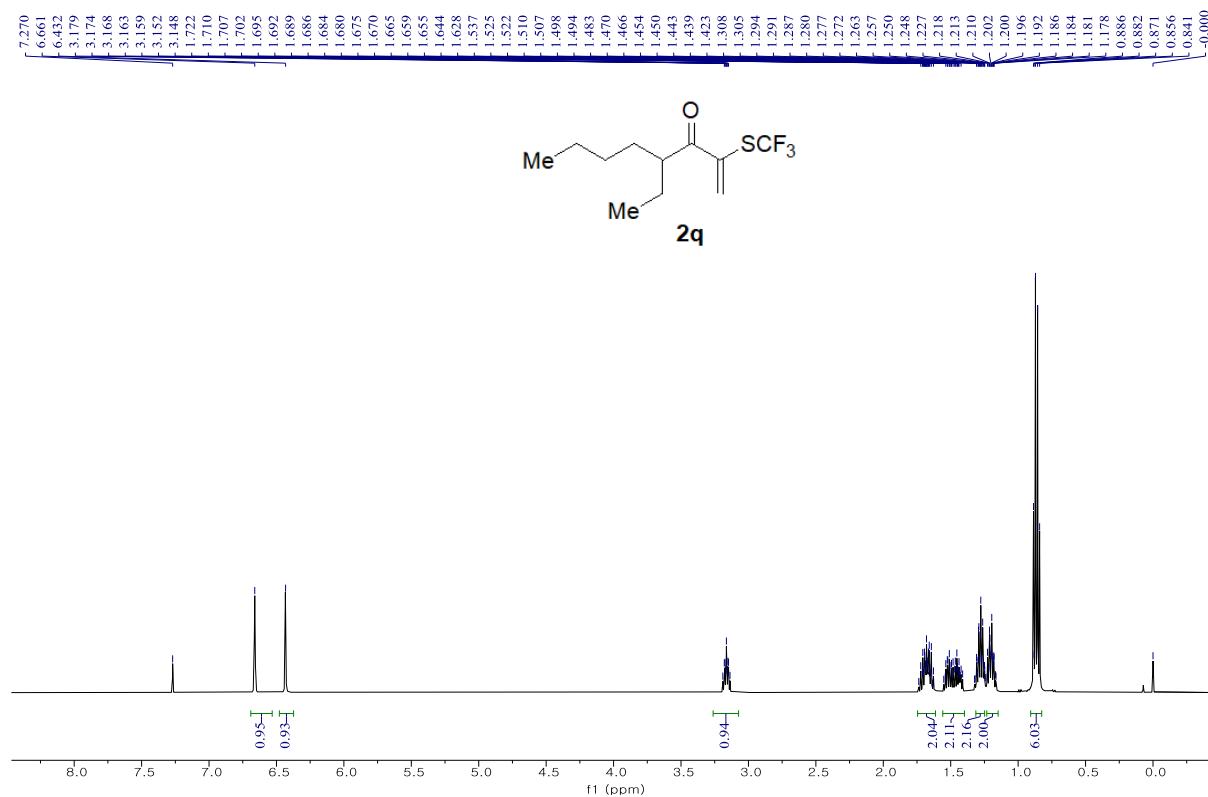
^{13}C NMR of **2p** (125 MHz, CDCl_3)



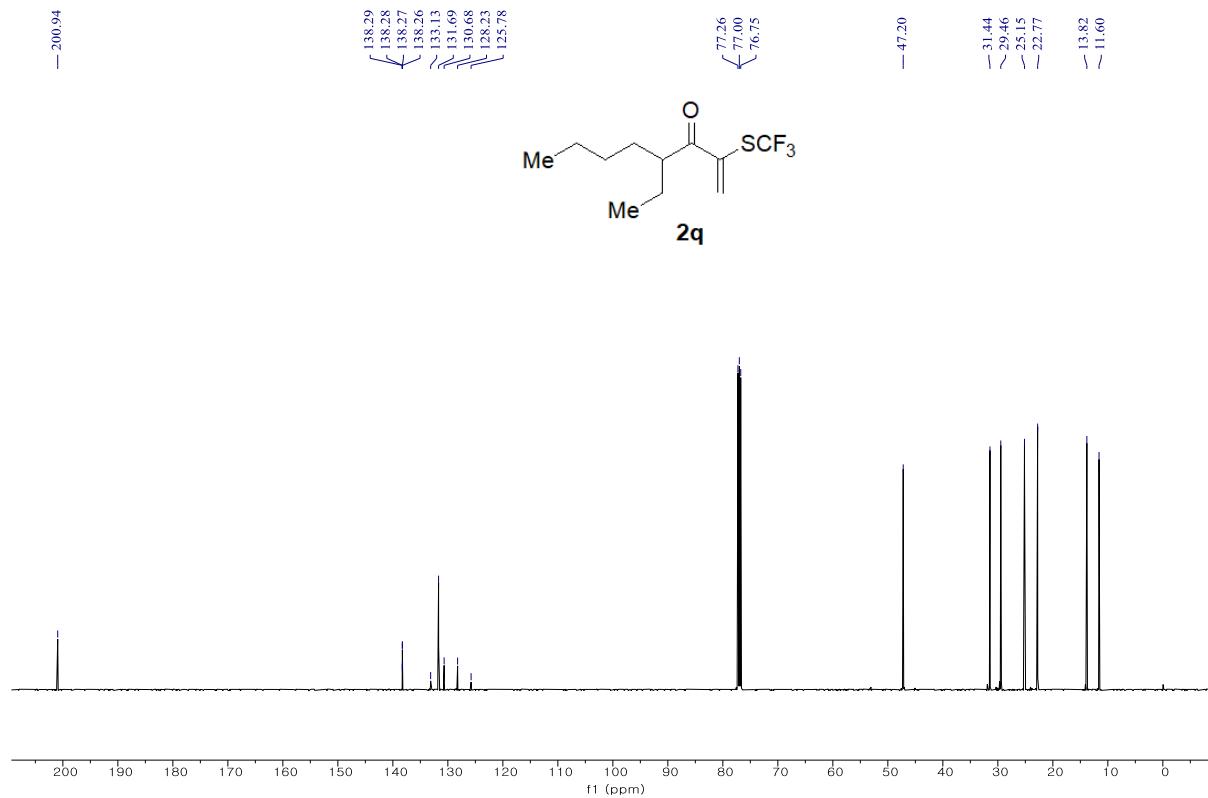
^{19}F NMR of **2p** (564 MHz, CDCl_3)



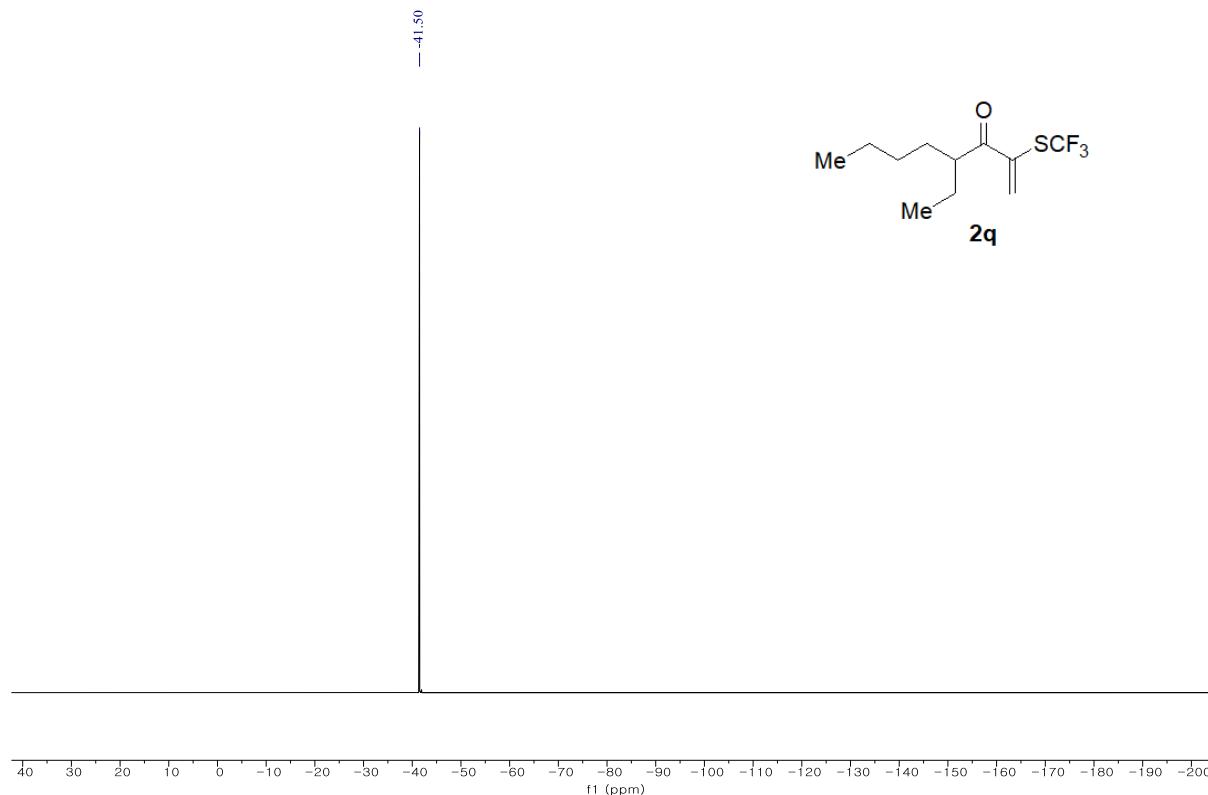
¹H NMR of **2q** (500 MHz, CDCl₃)



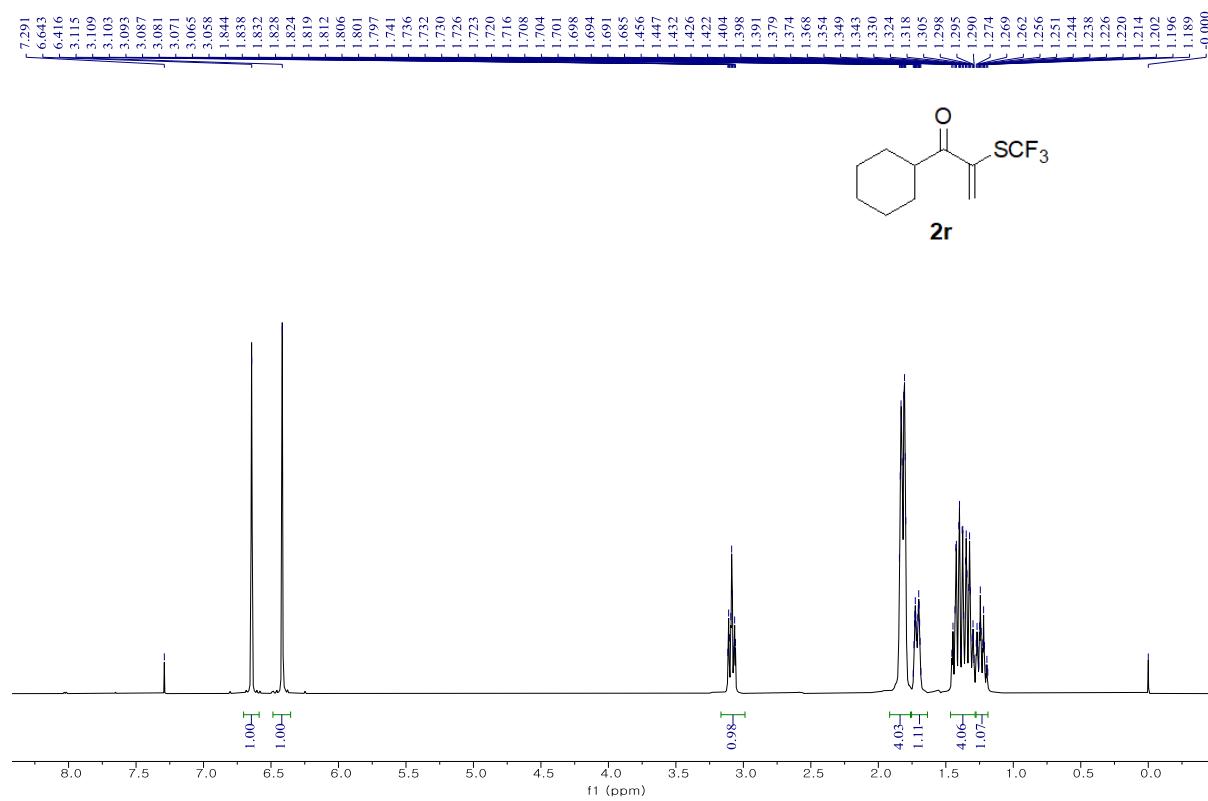
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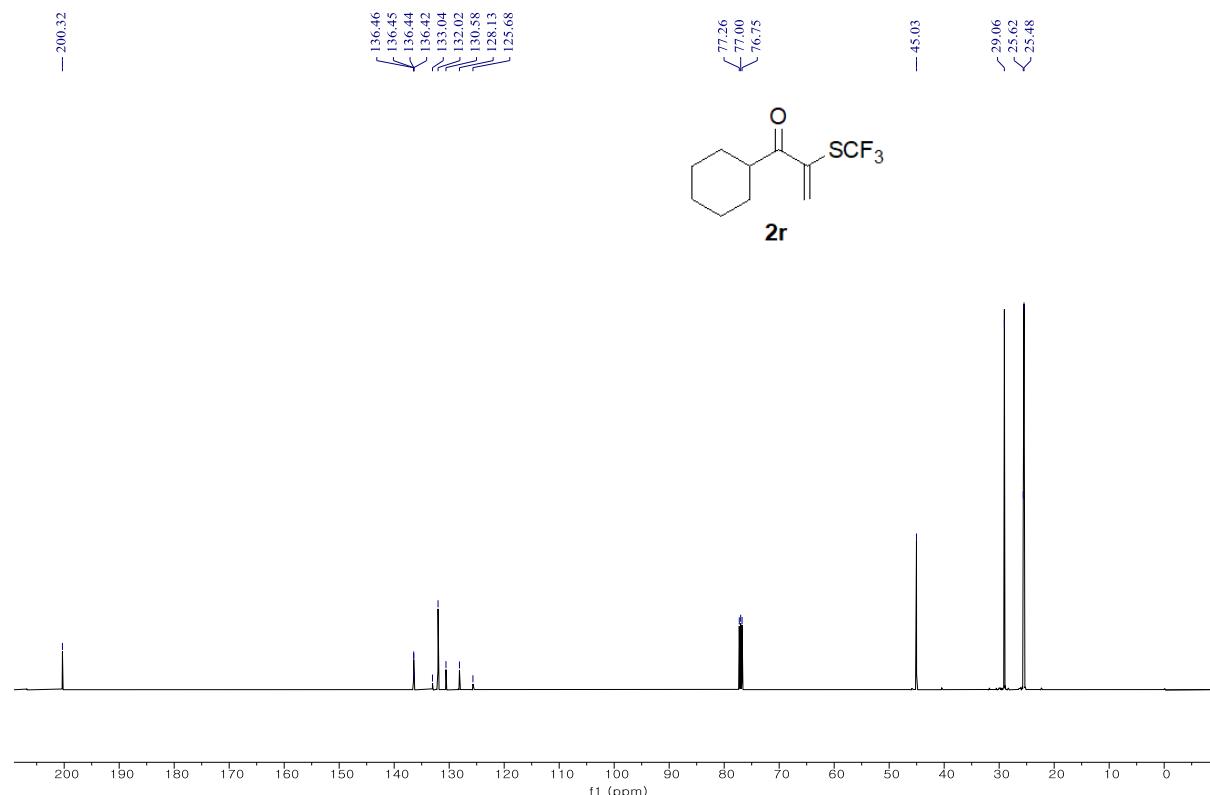
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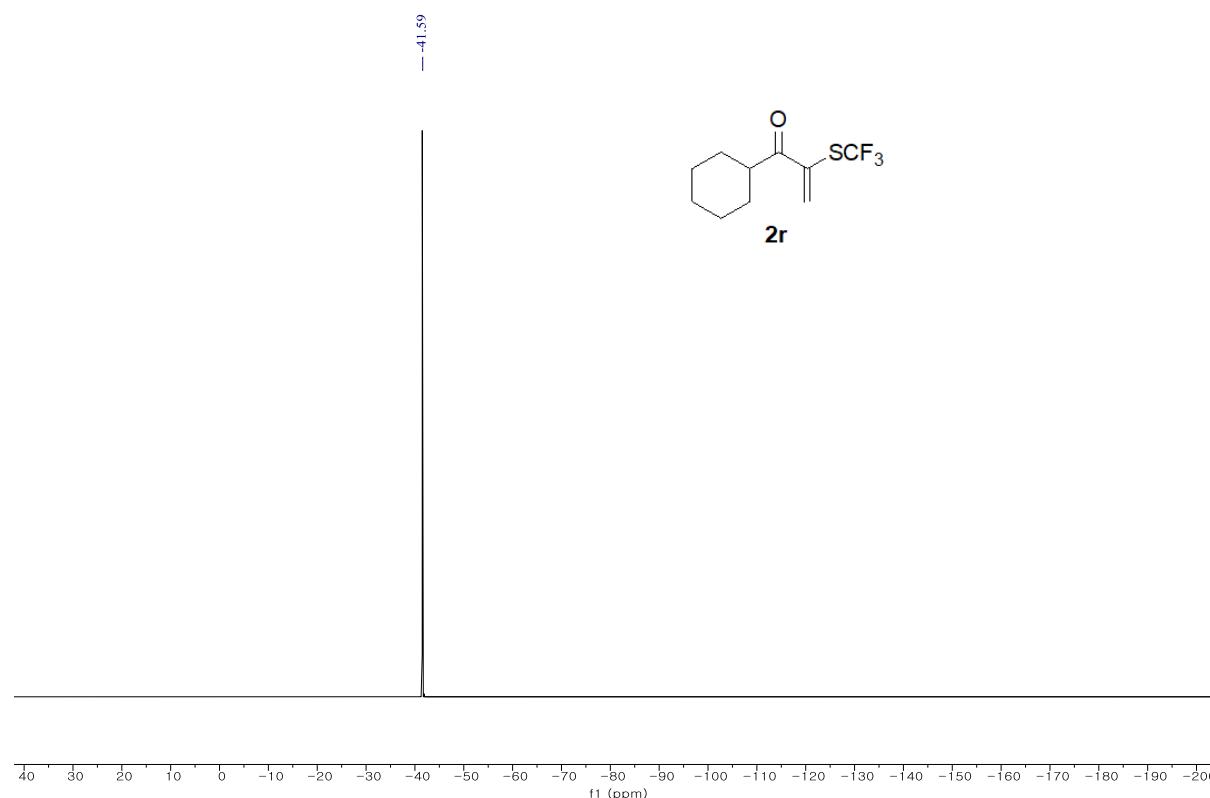
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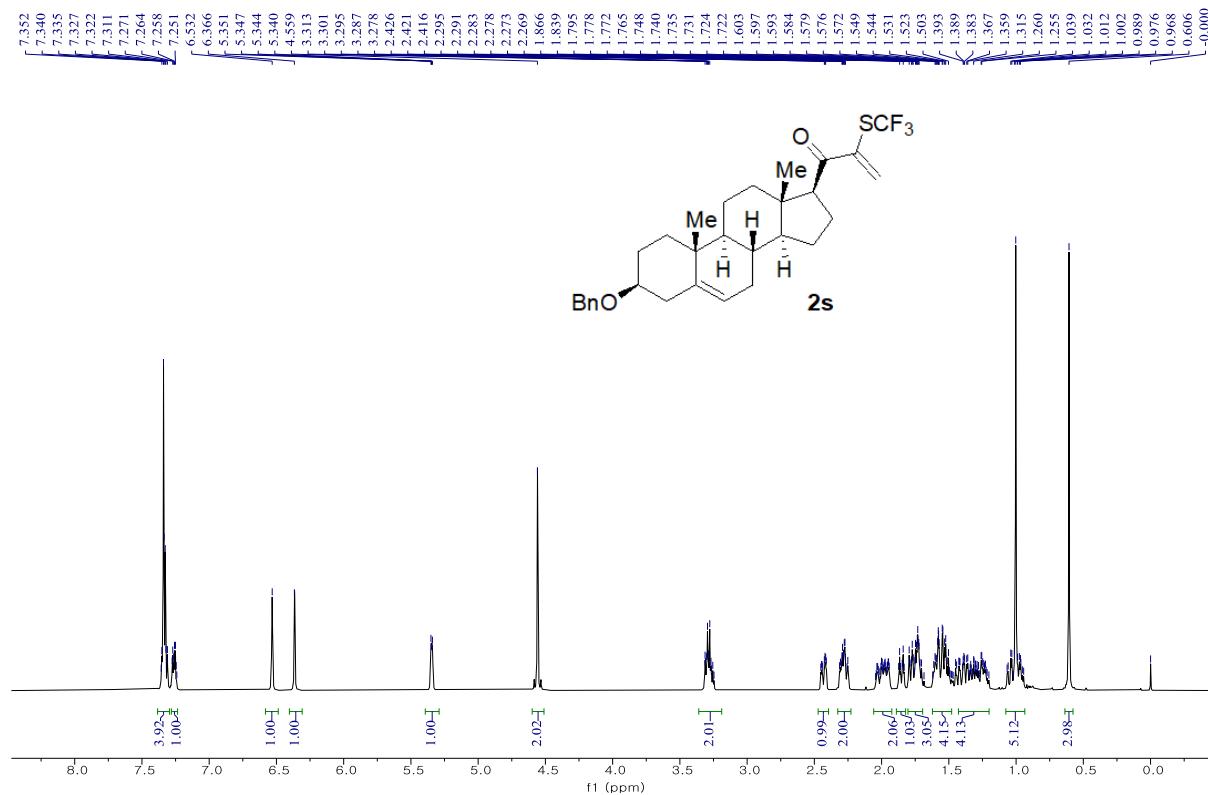
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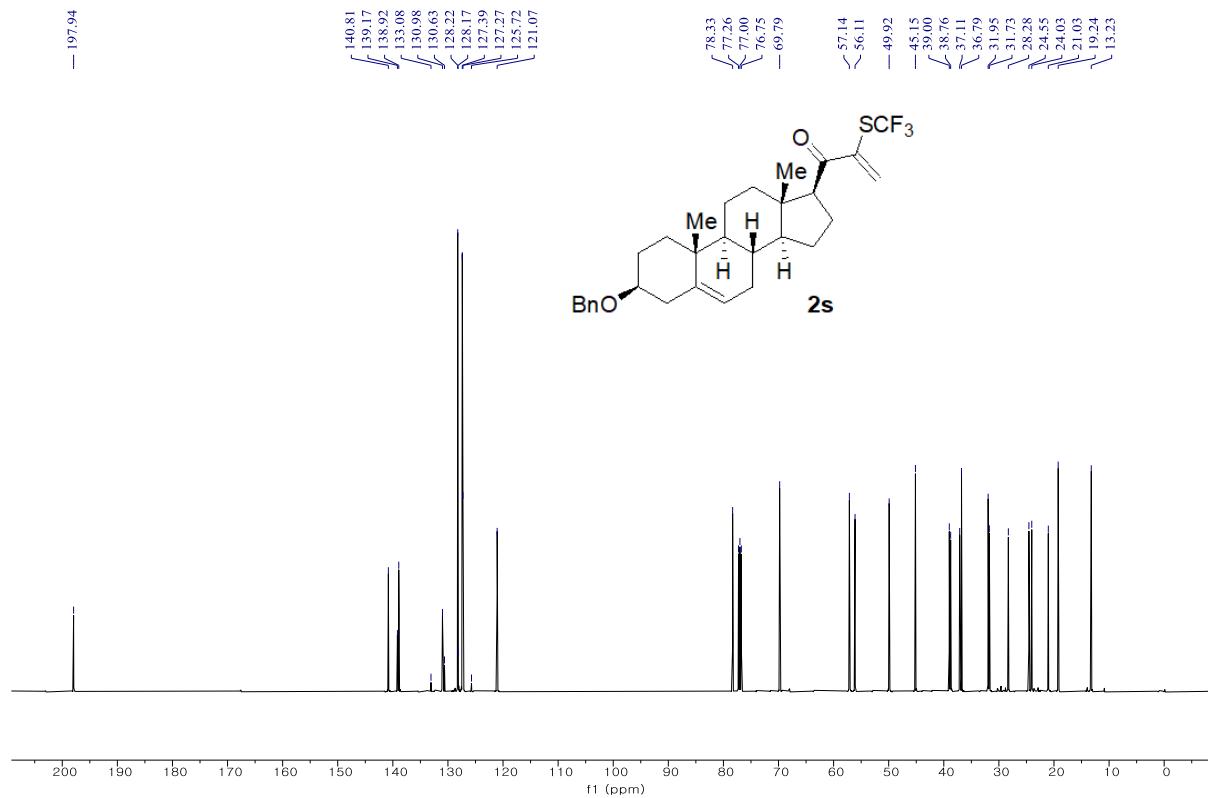
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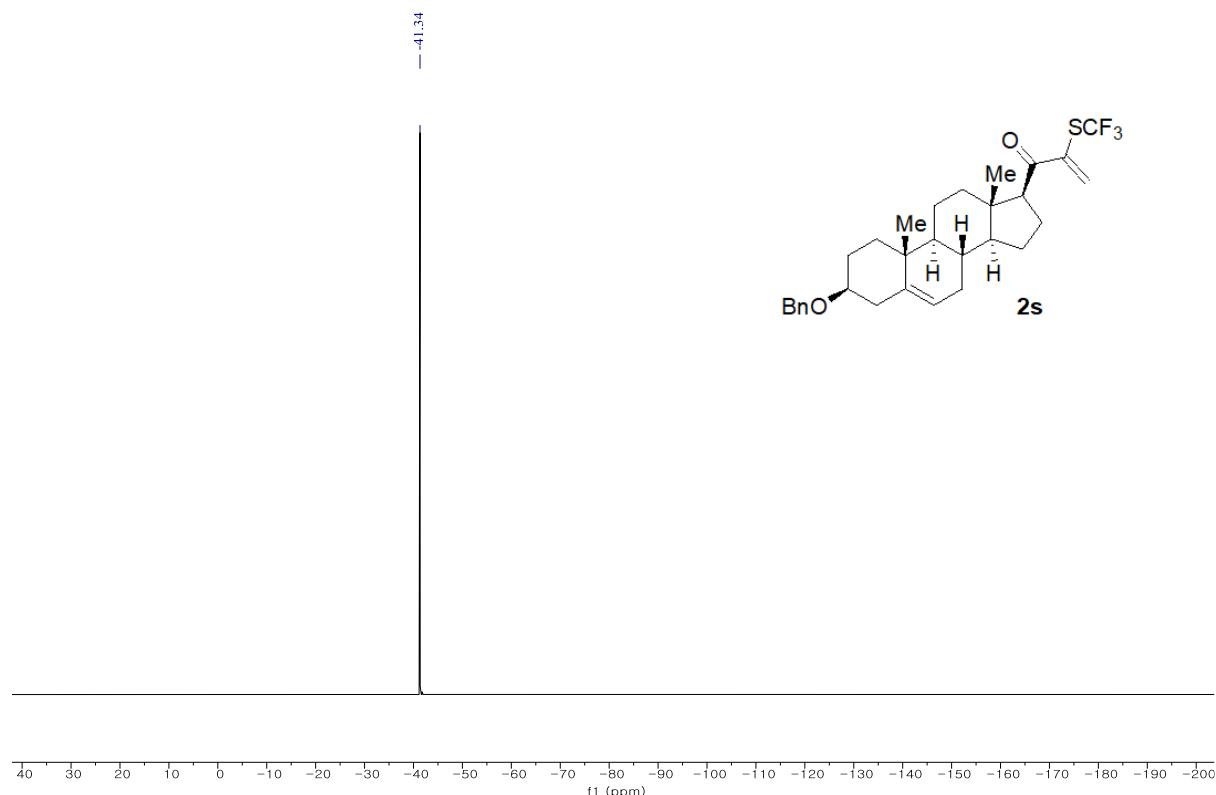
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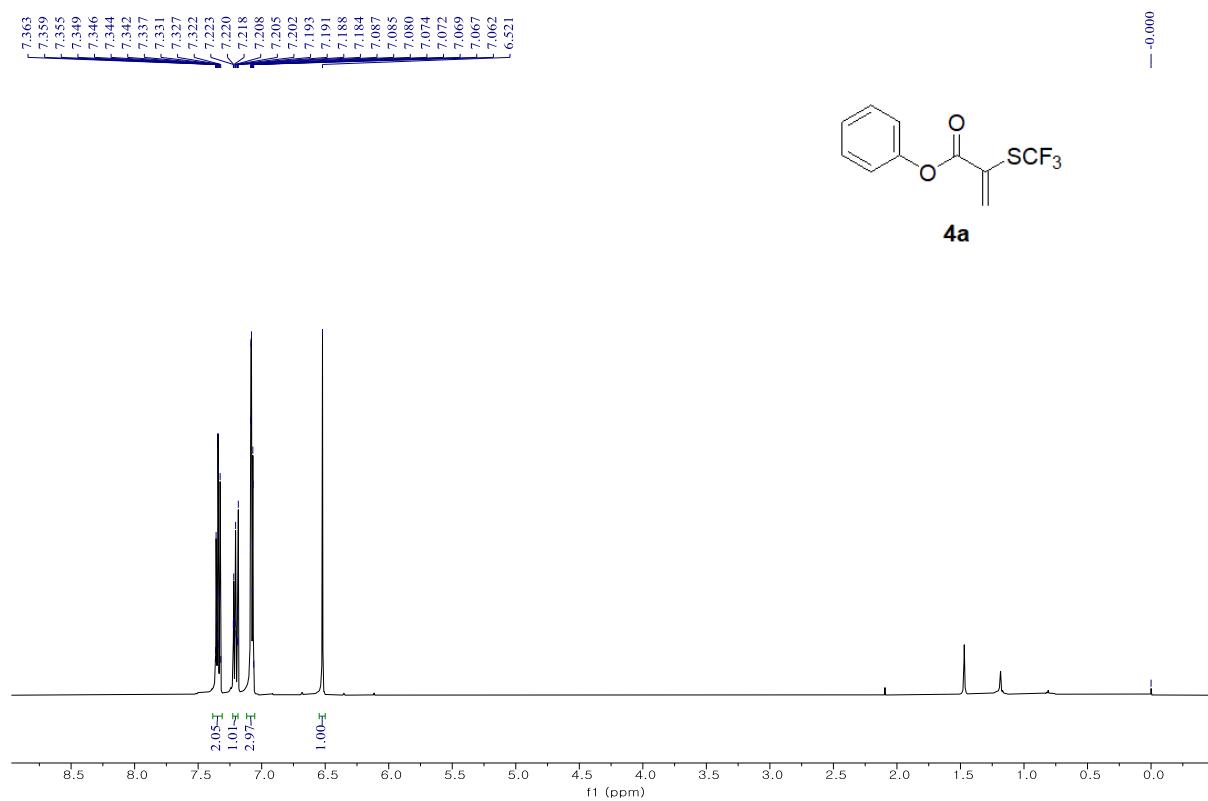
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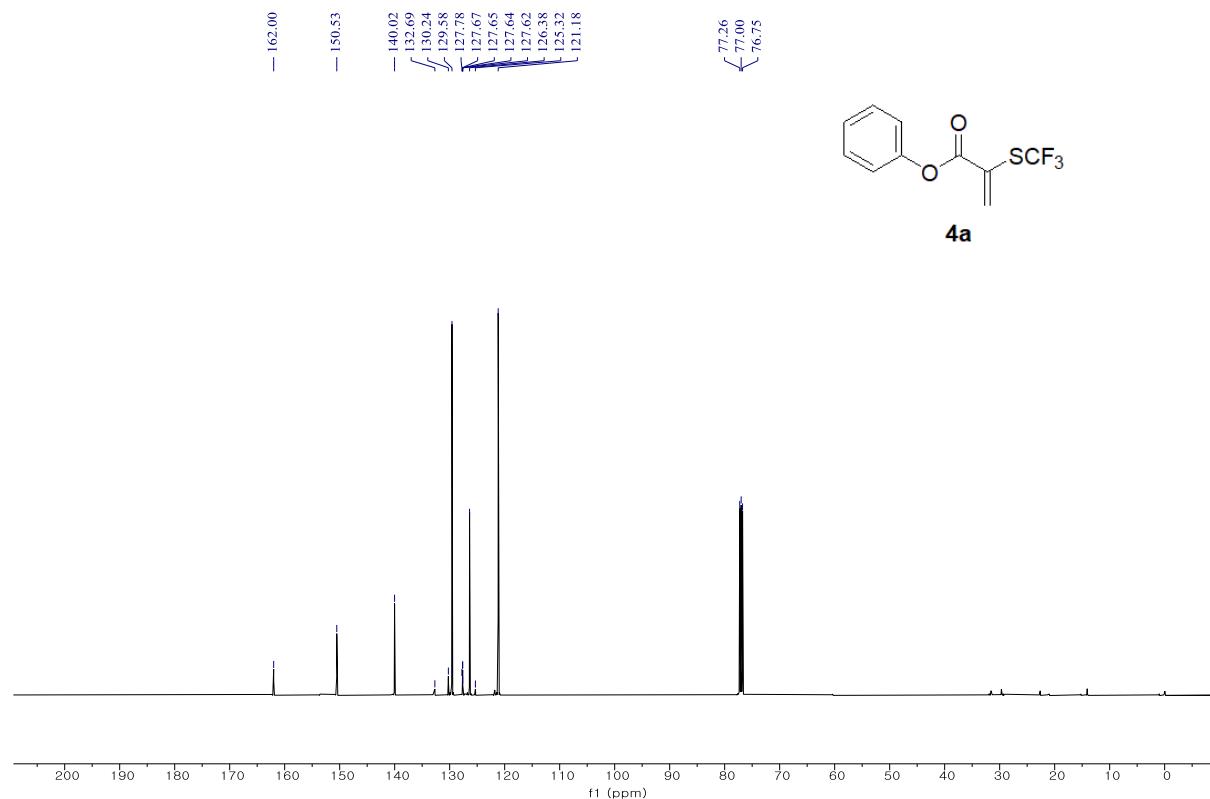
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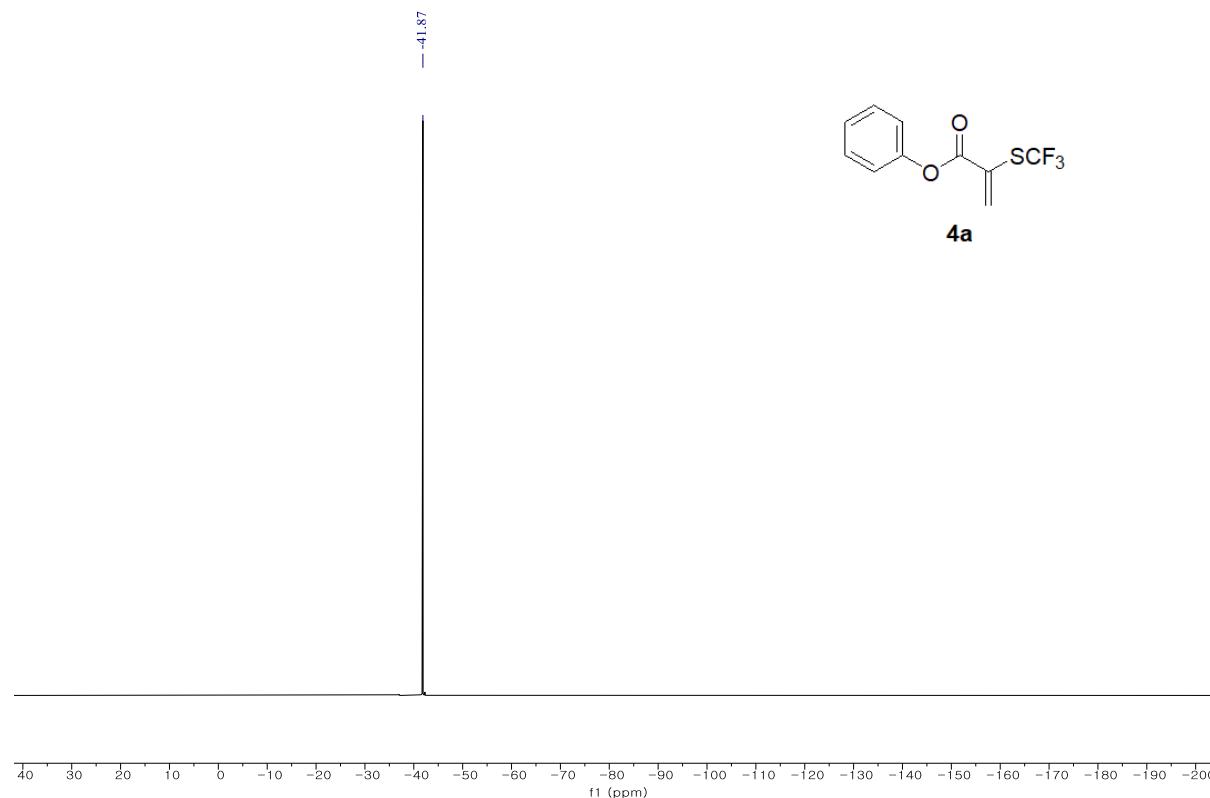
¹H NMR of **4a** (500 MHz, CDCl₃)



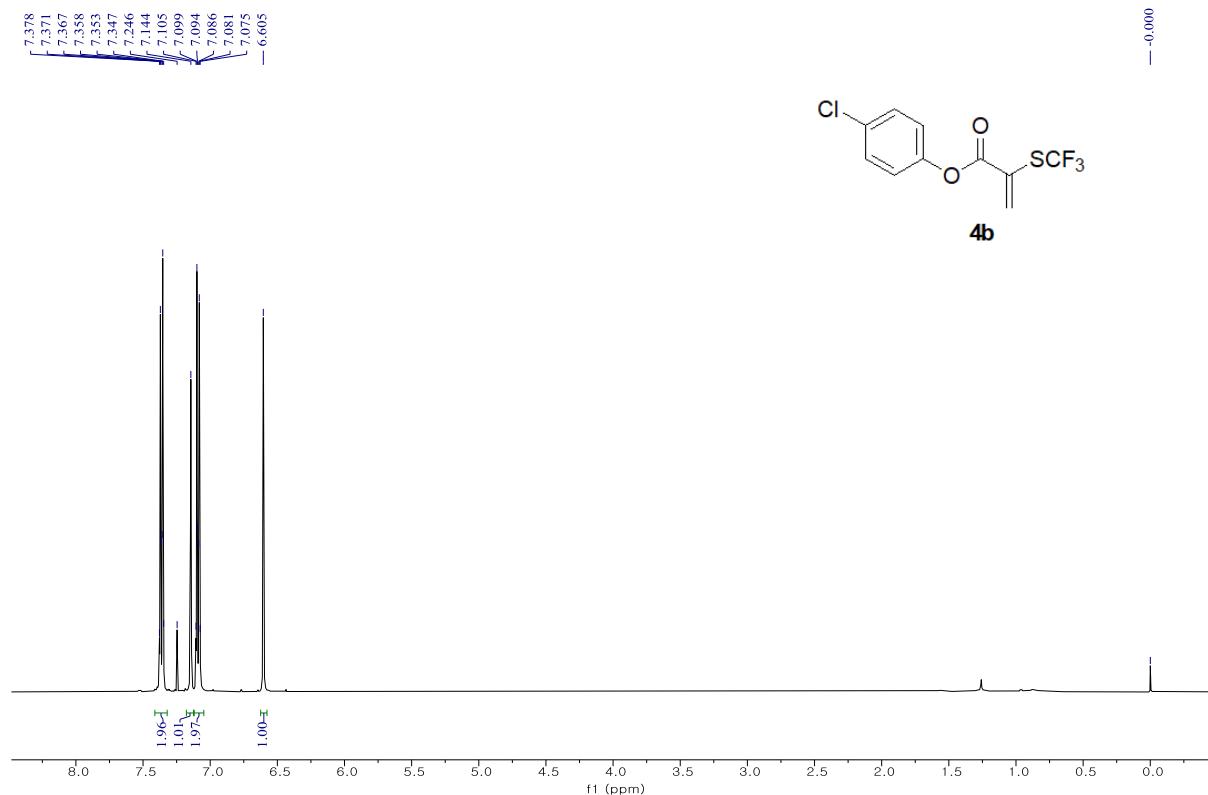
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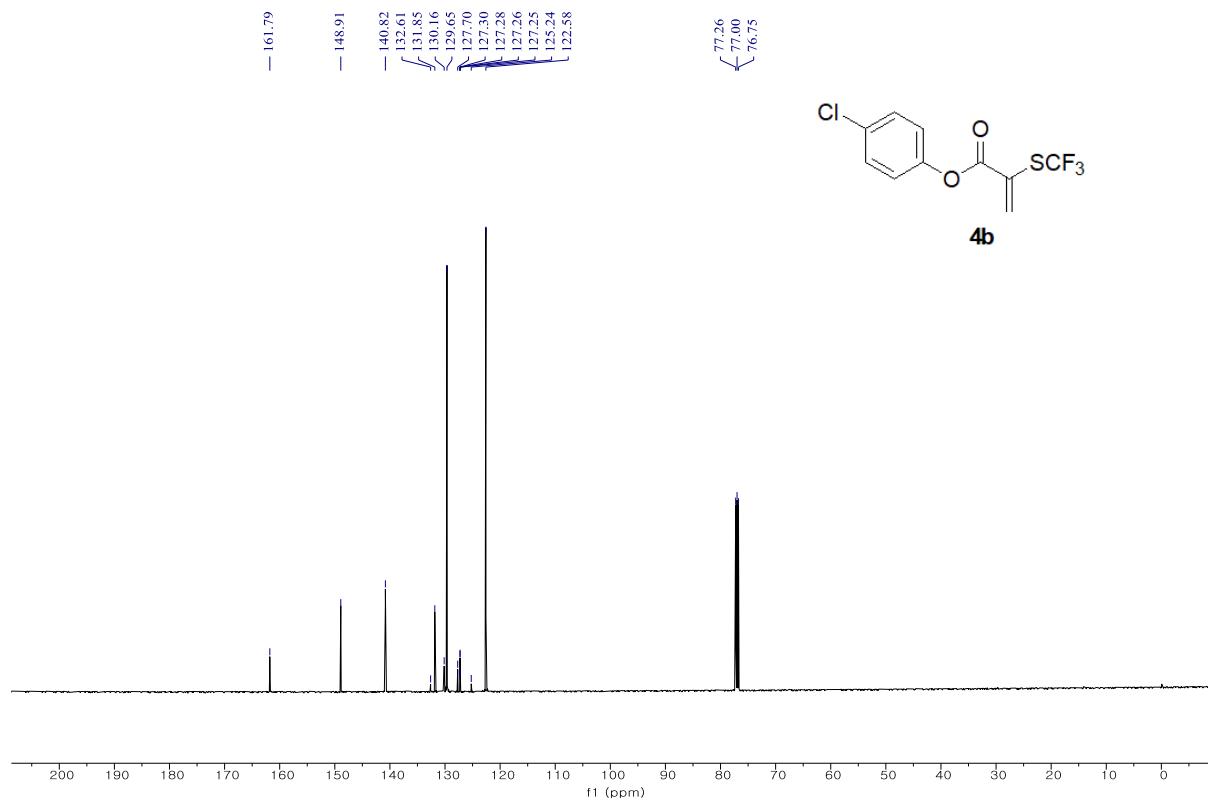
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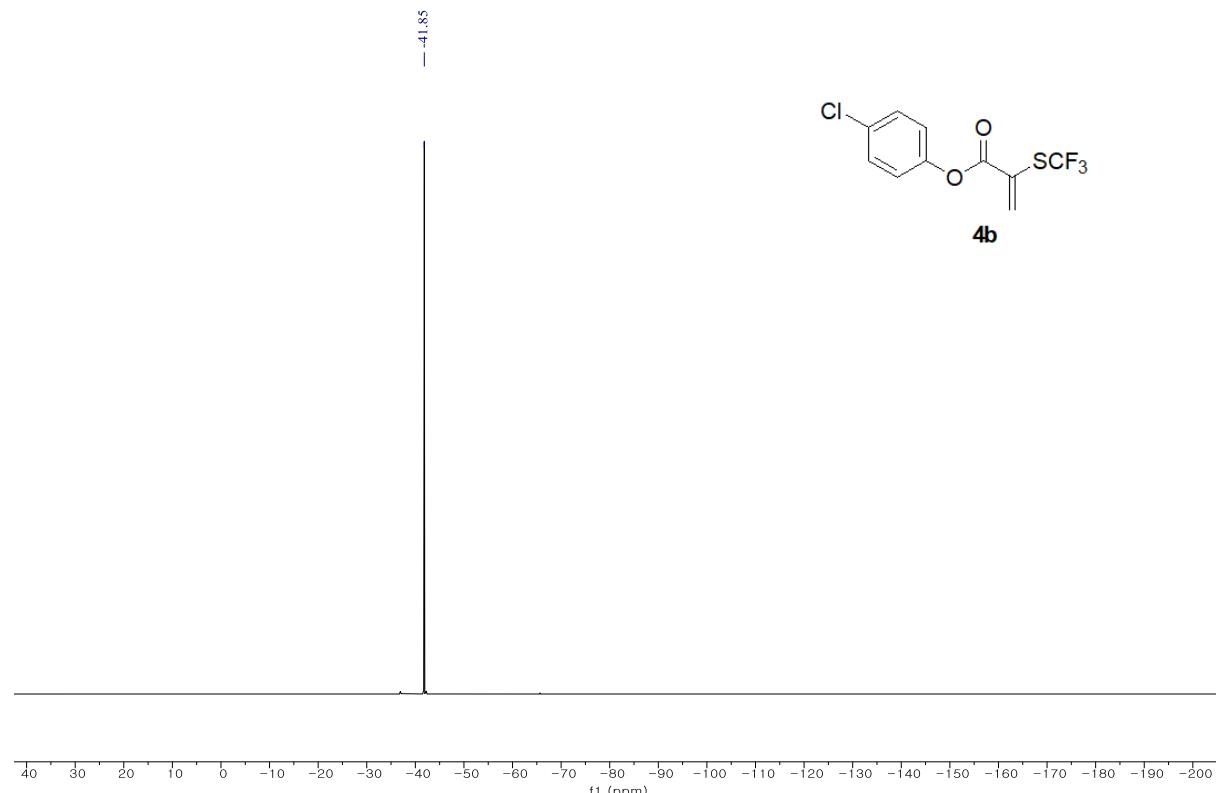
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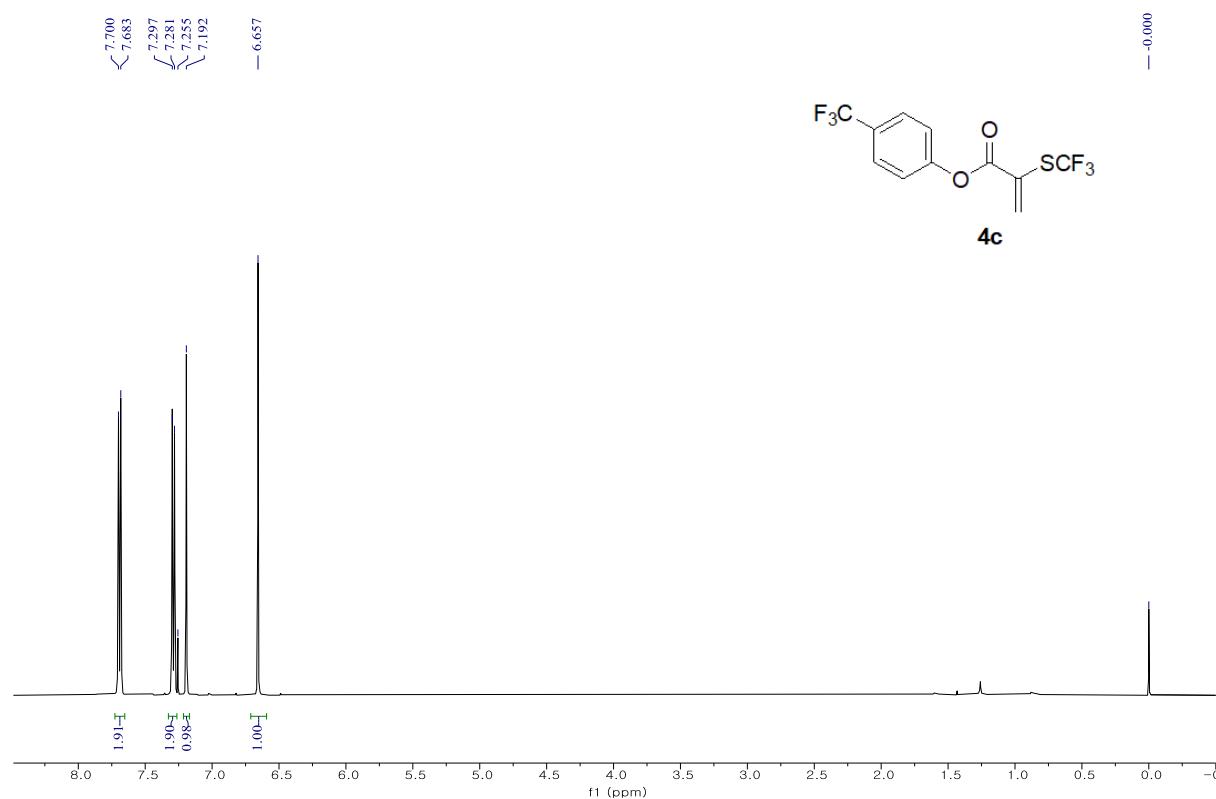
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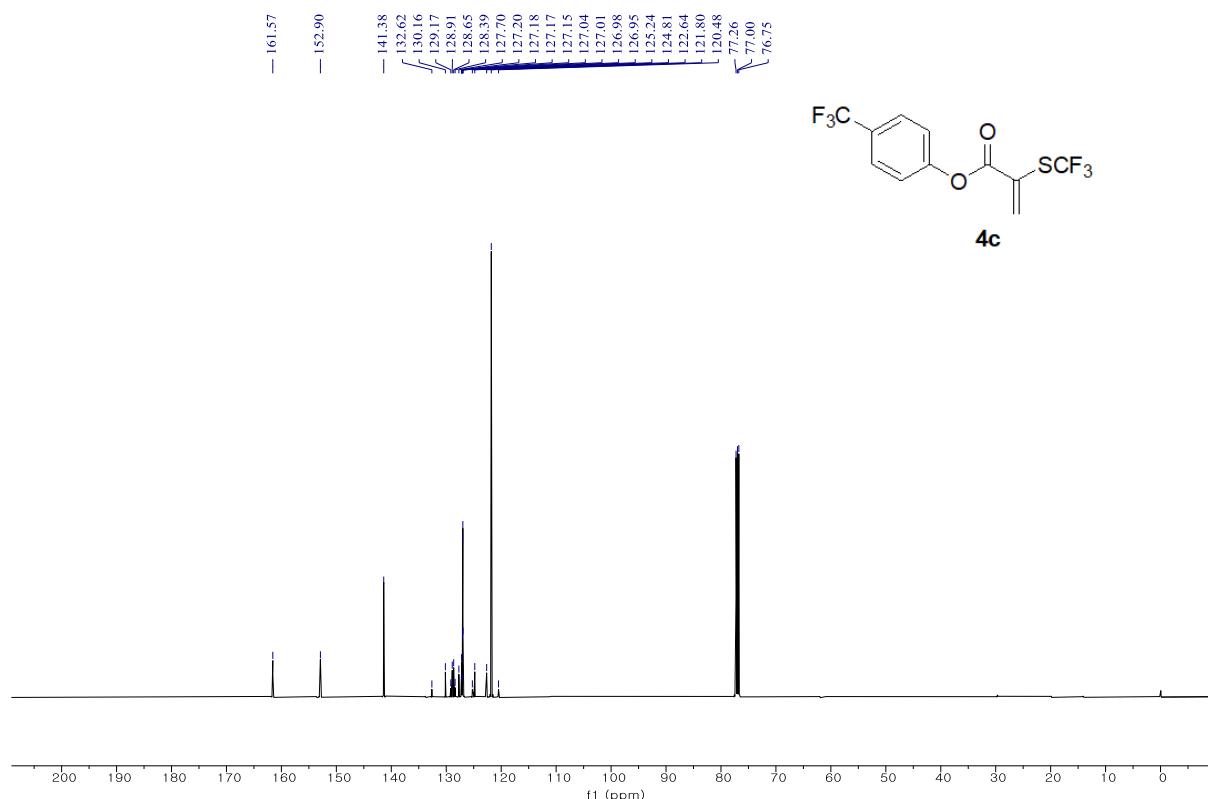
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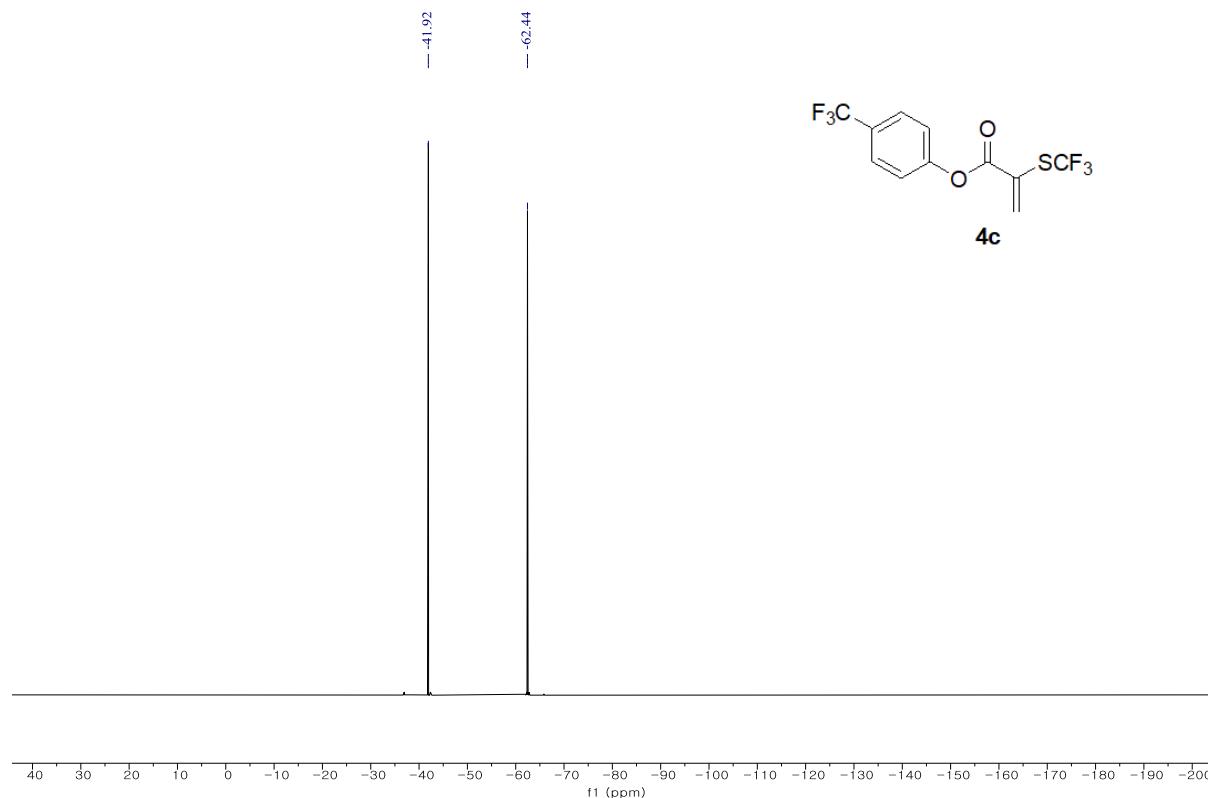
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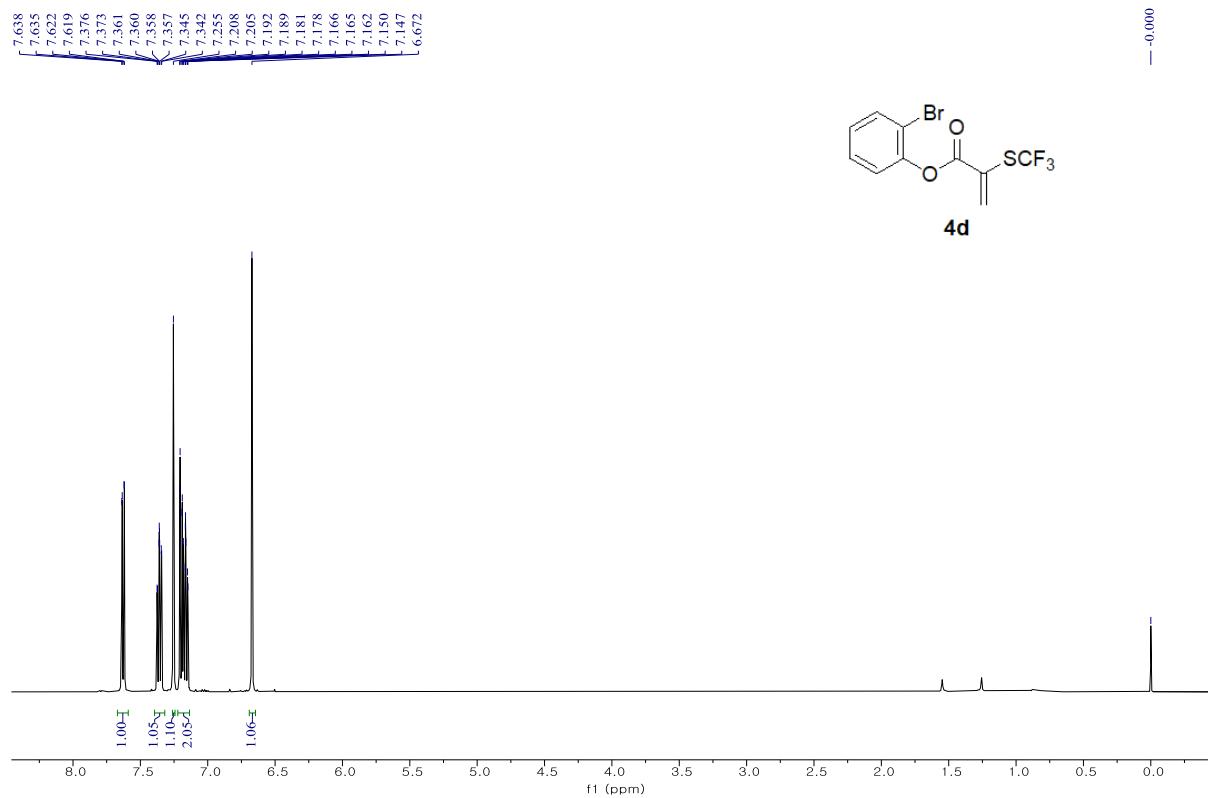
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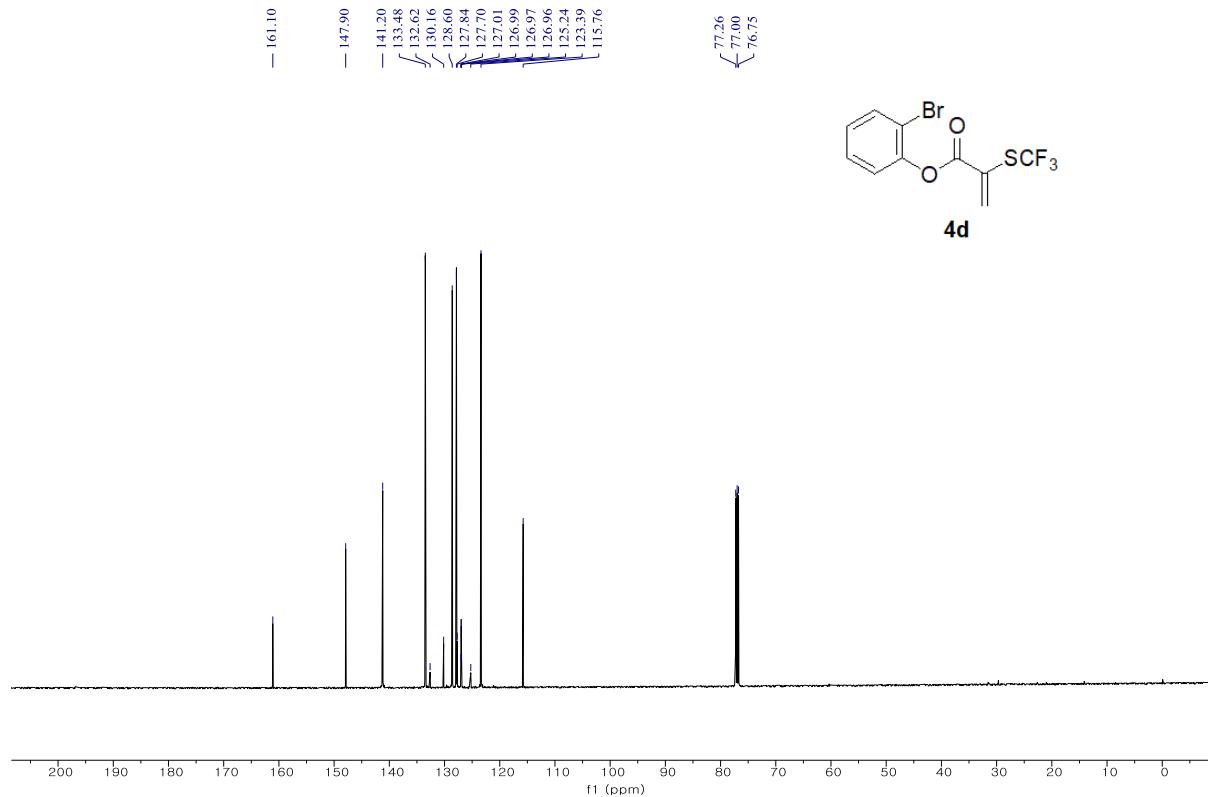
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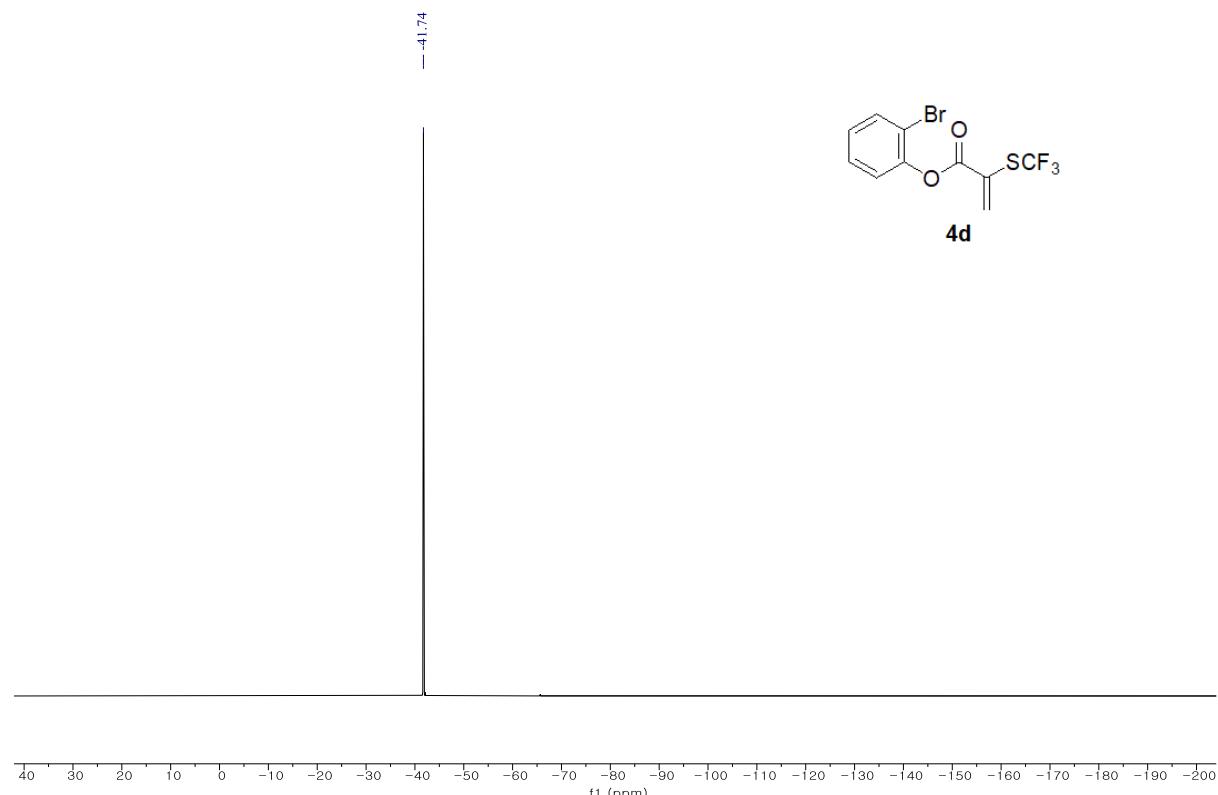
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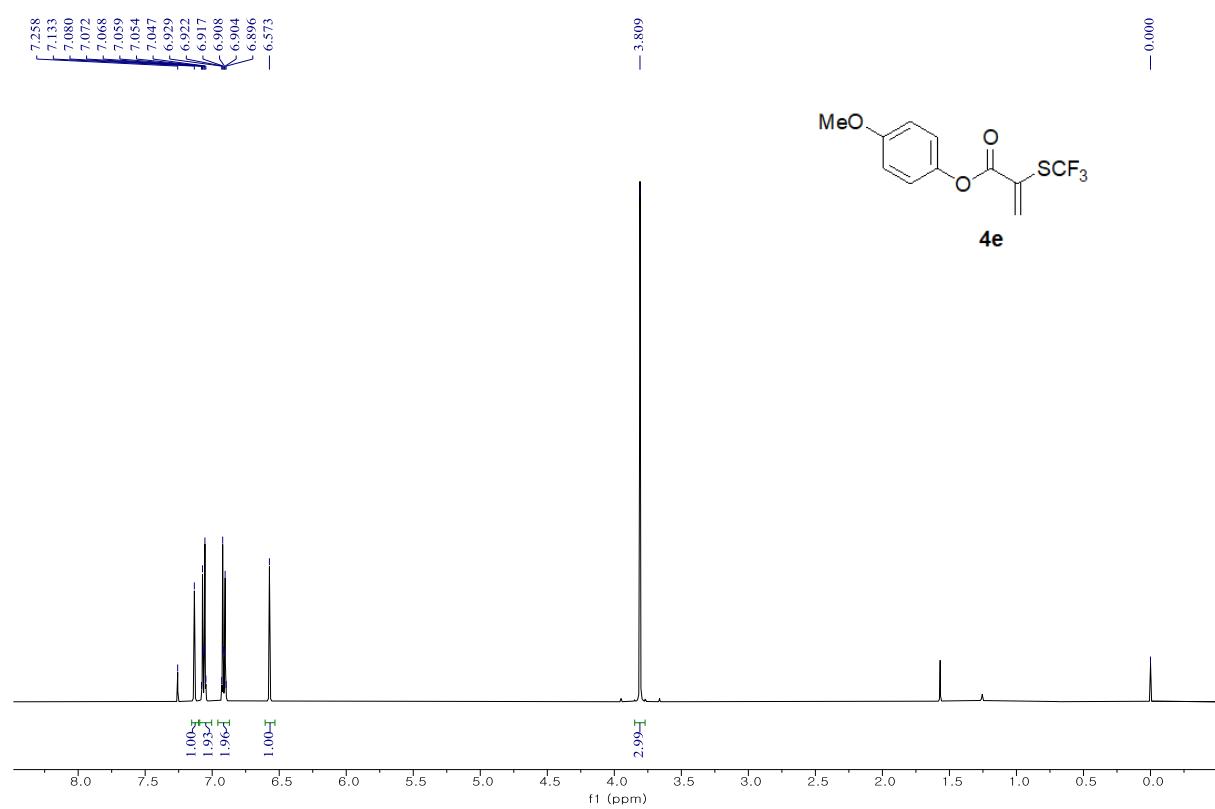
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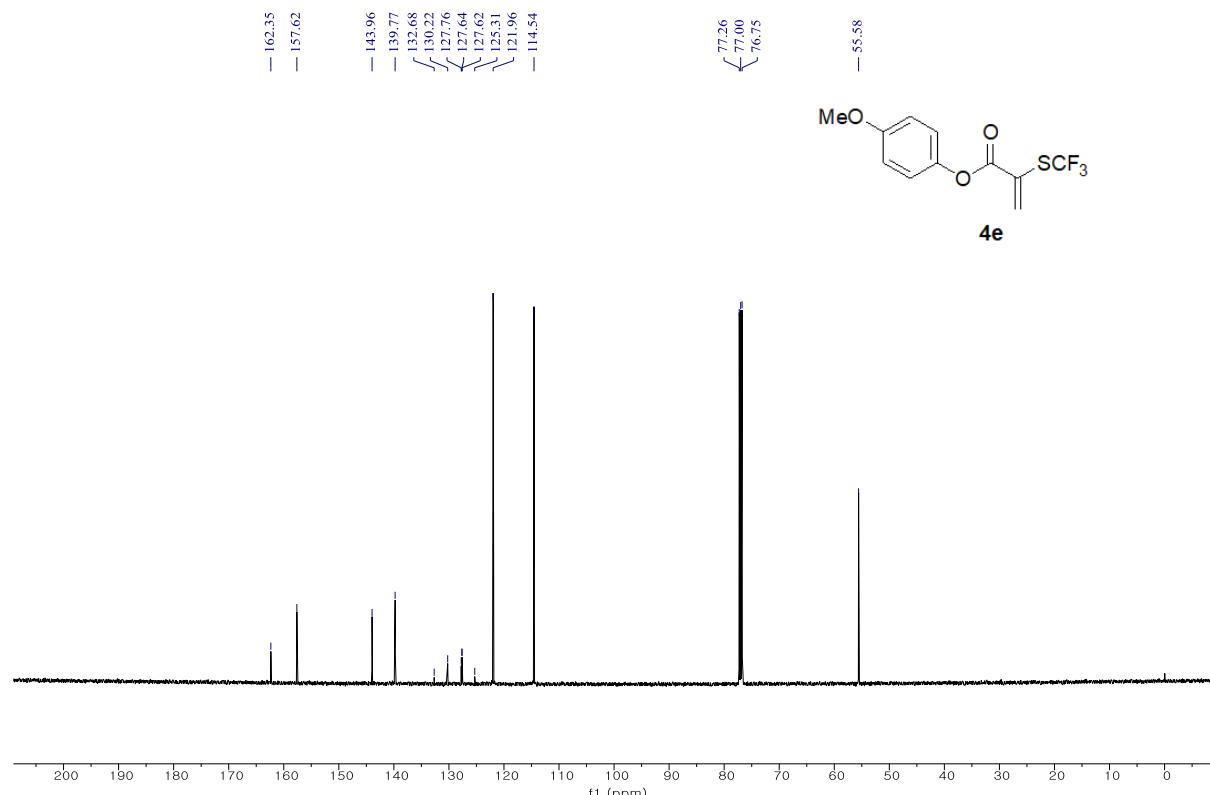
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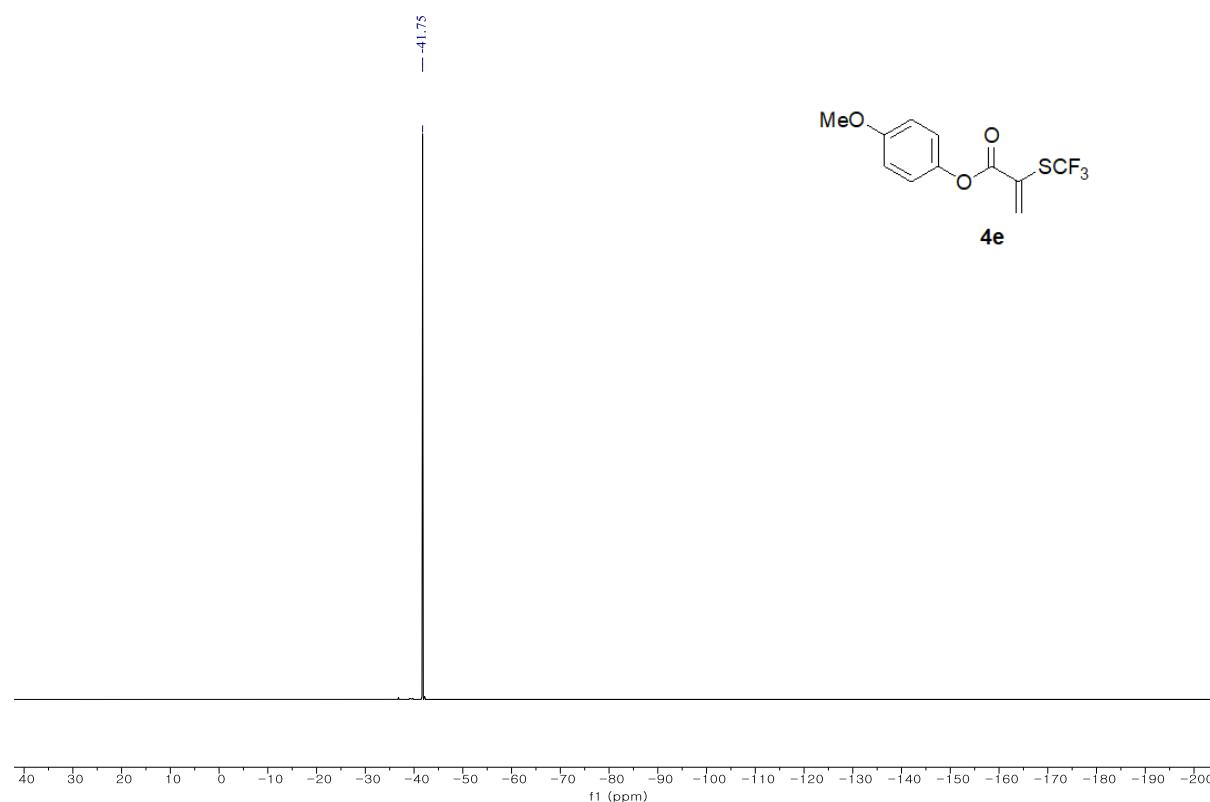
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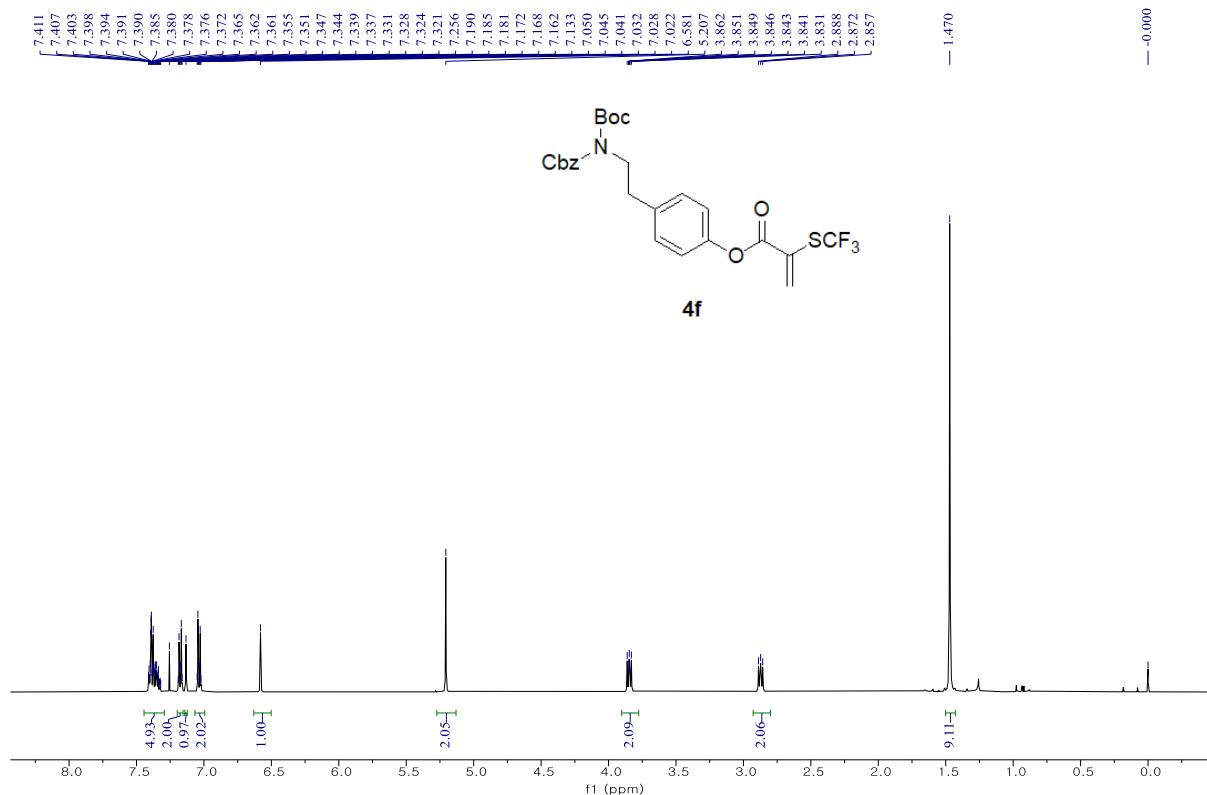
^{13}C NMR of **4e** (125 MHz, CDCl_3)



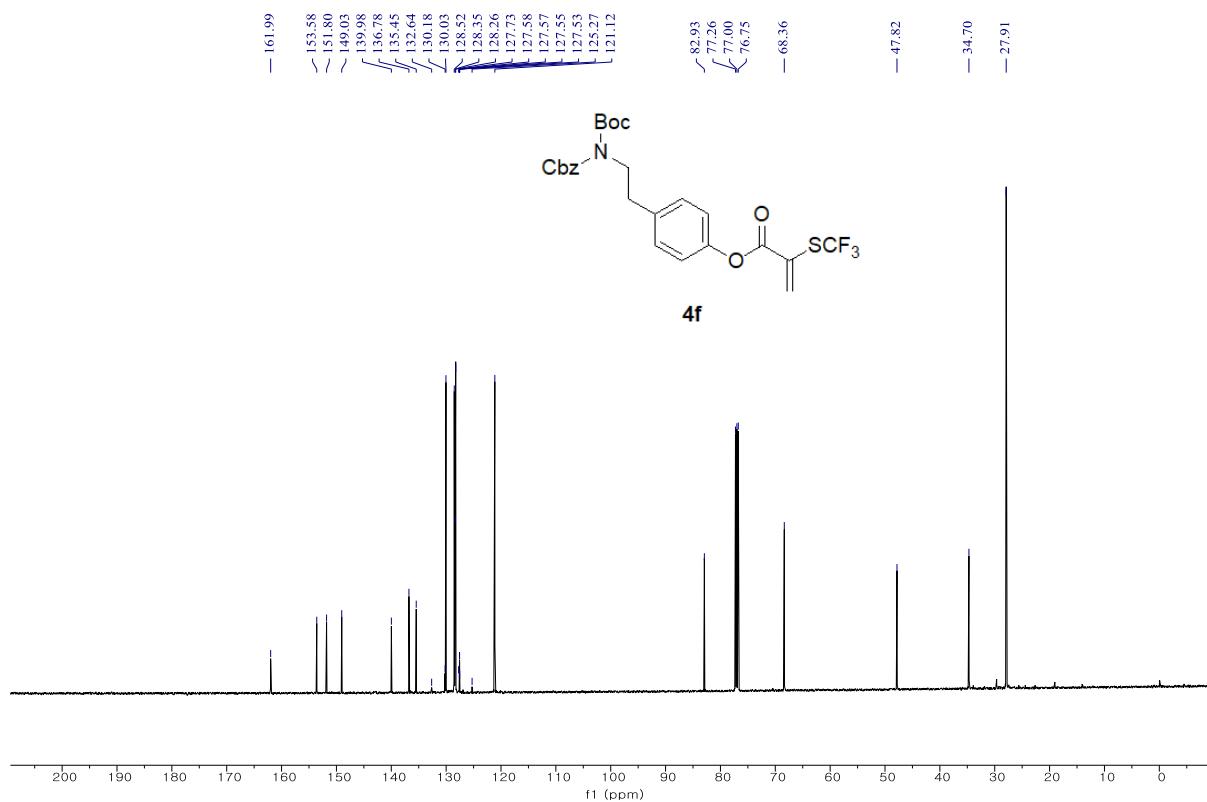
^{19}F NMR of **4e** (564 MHz, CDCl_3)



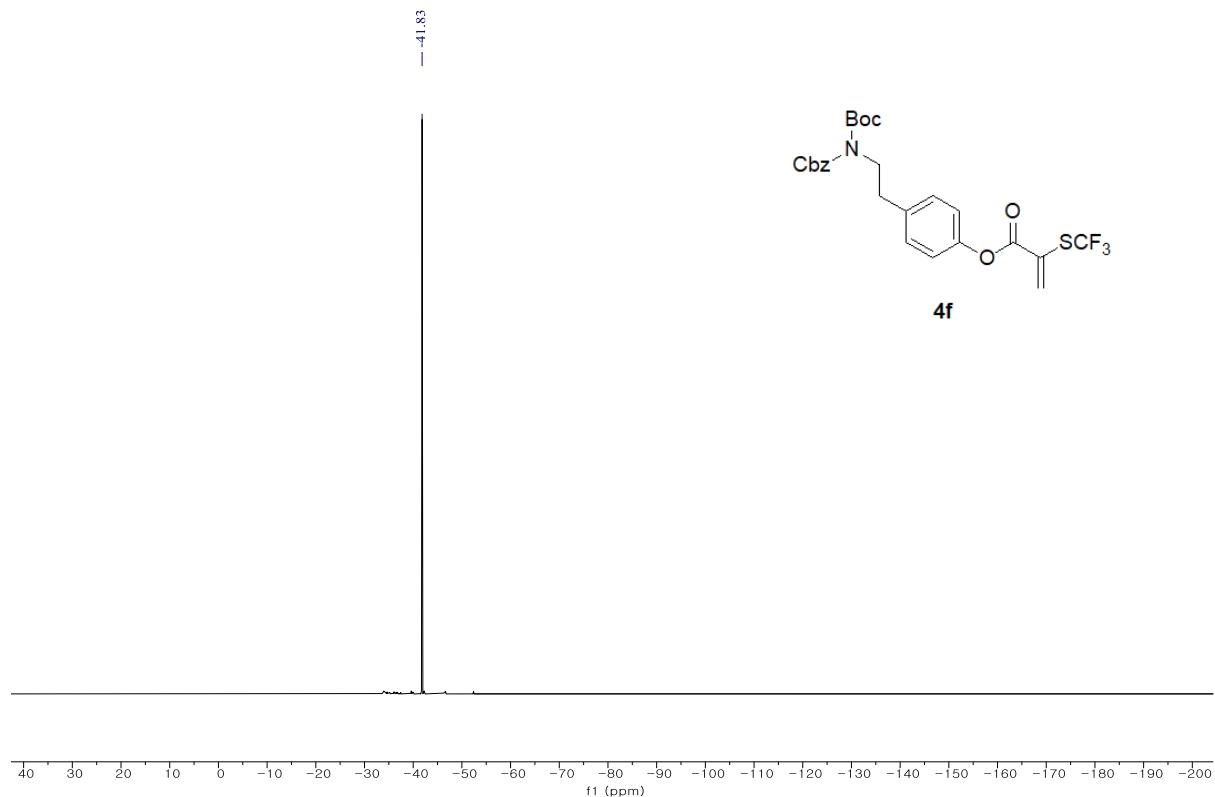
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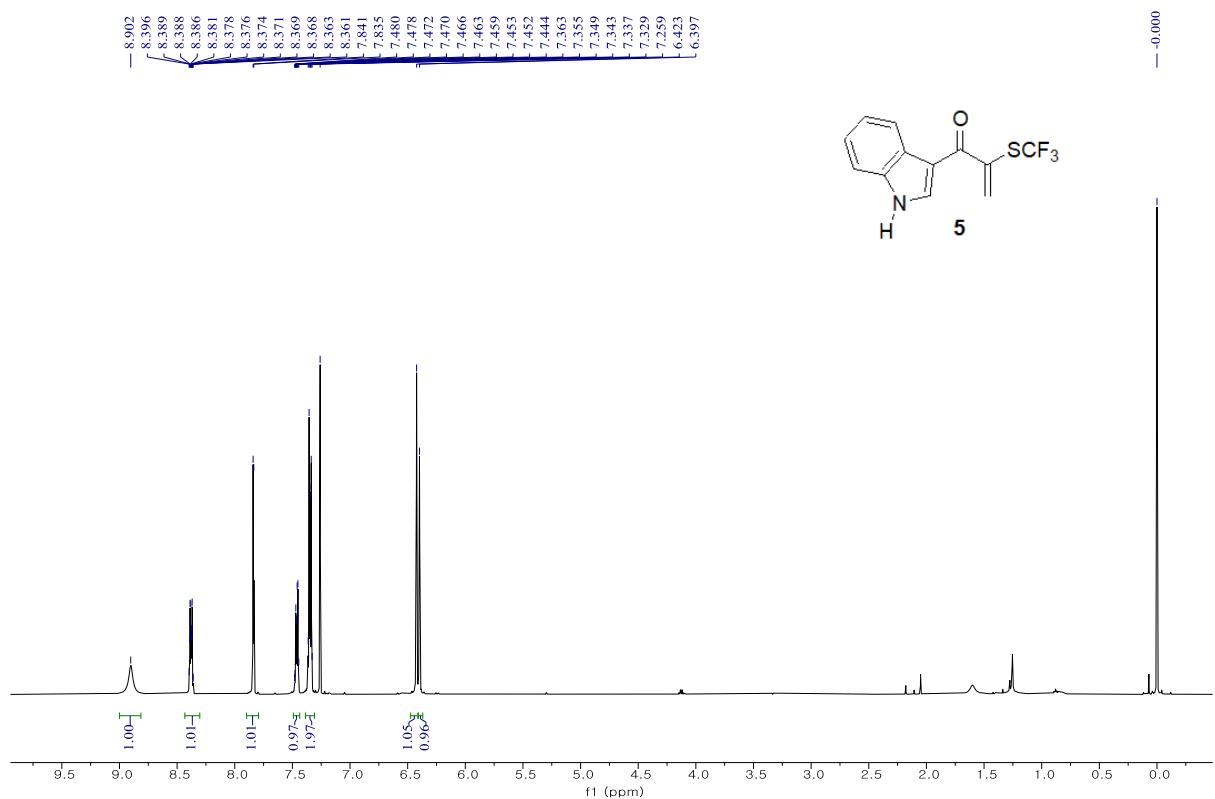
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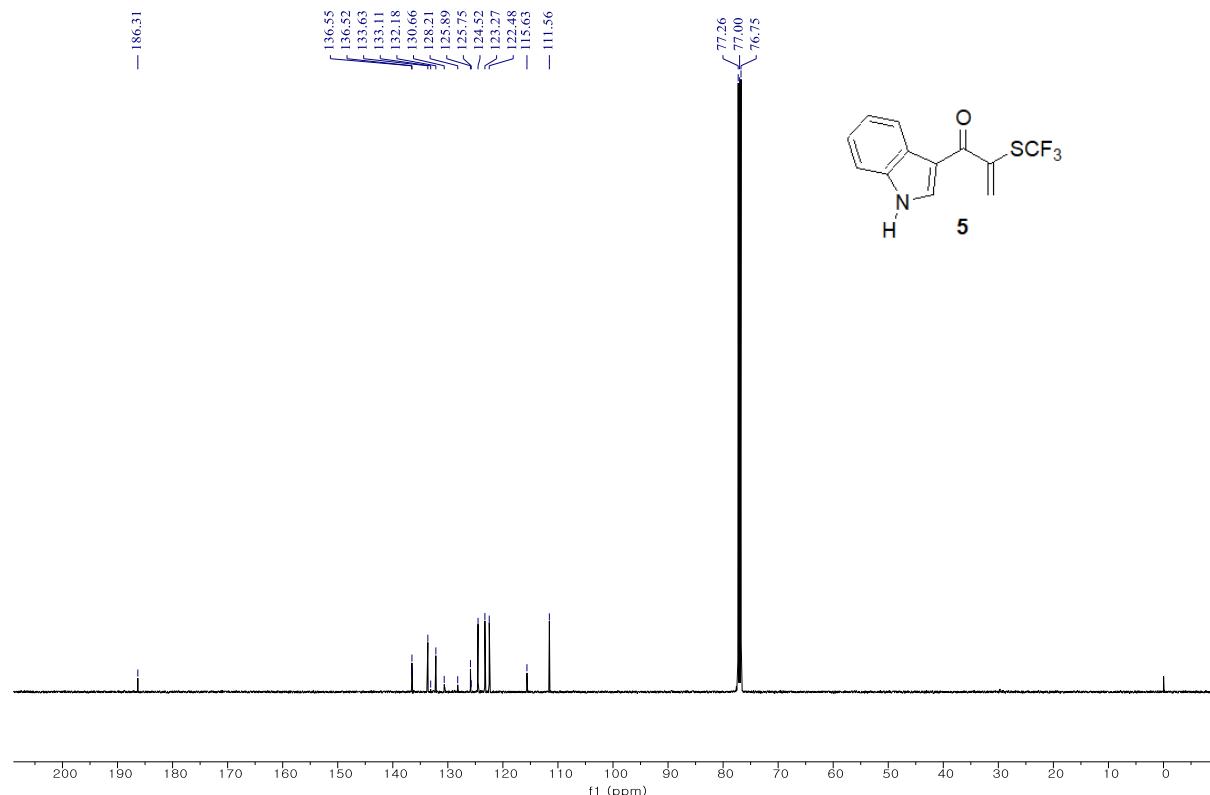
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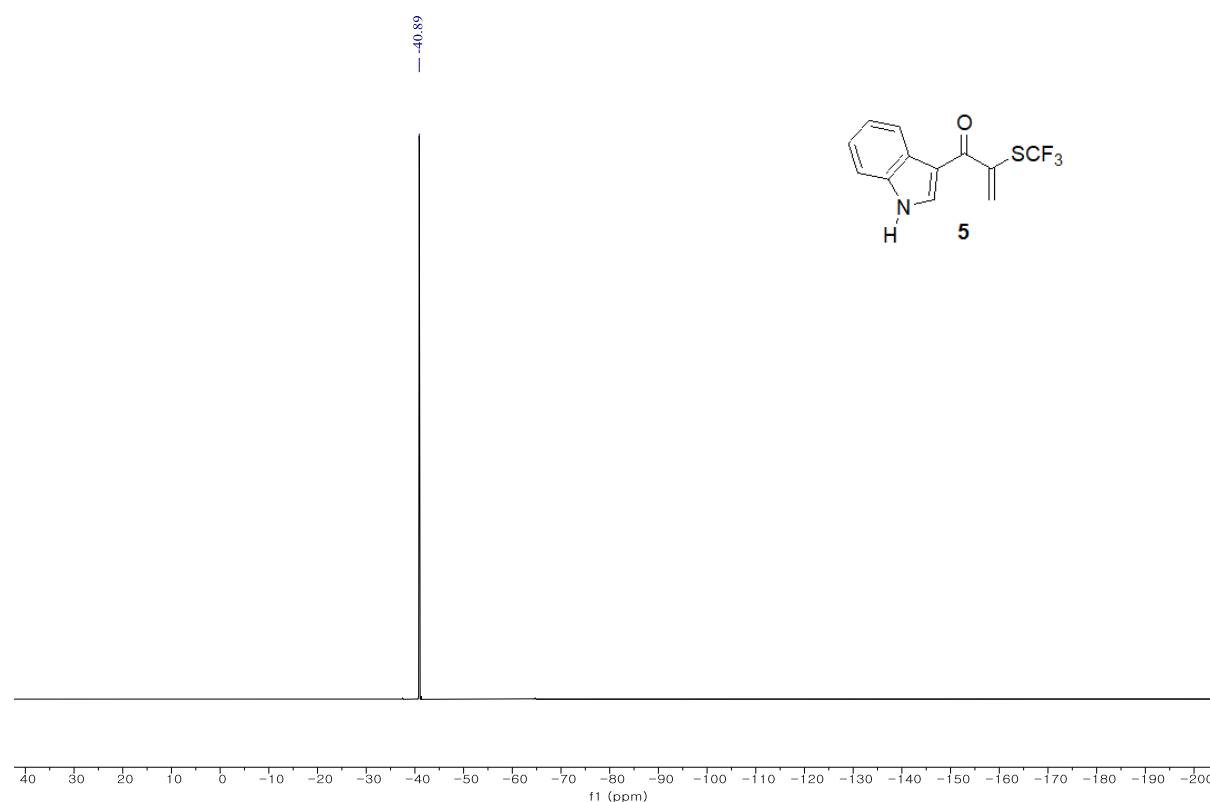
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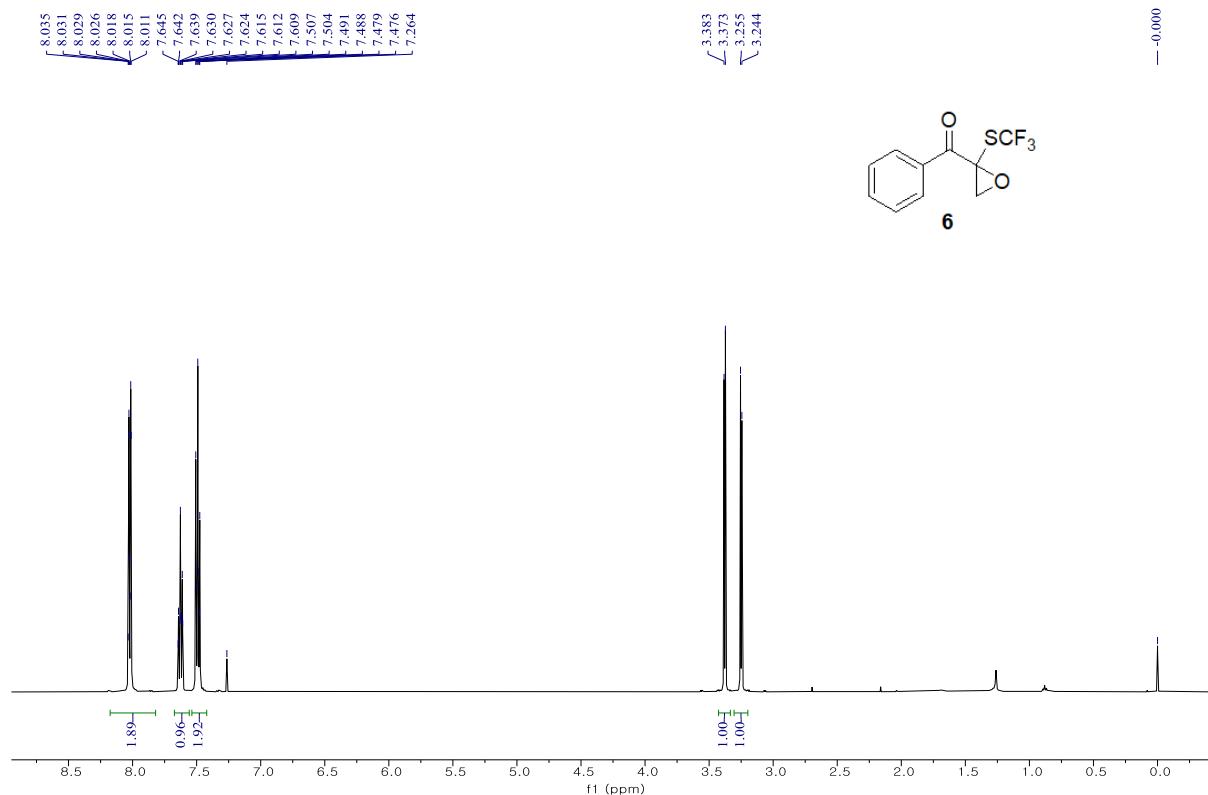
¹³C NMR of **5** (125 MHz, CDCl₃)



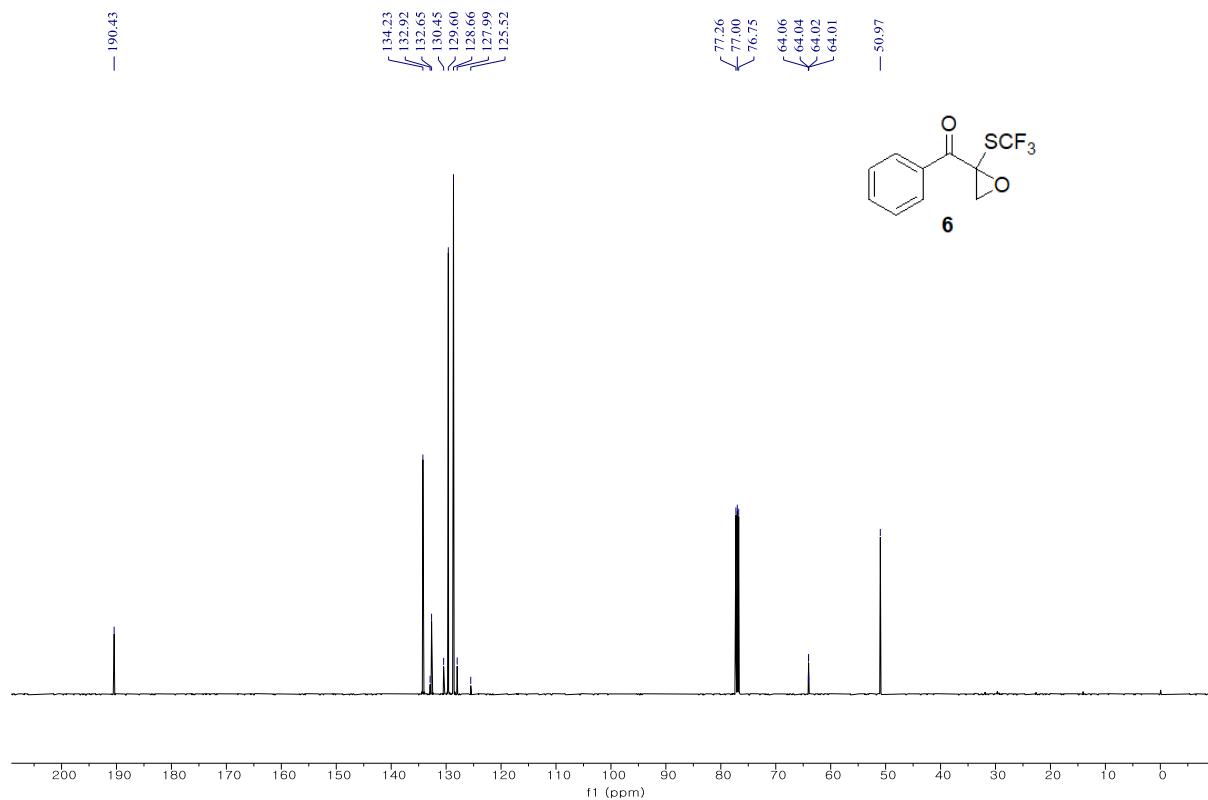
¹⁹F NMR of **5** (564 MHz, CDCl₃)



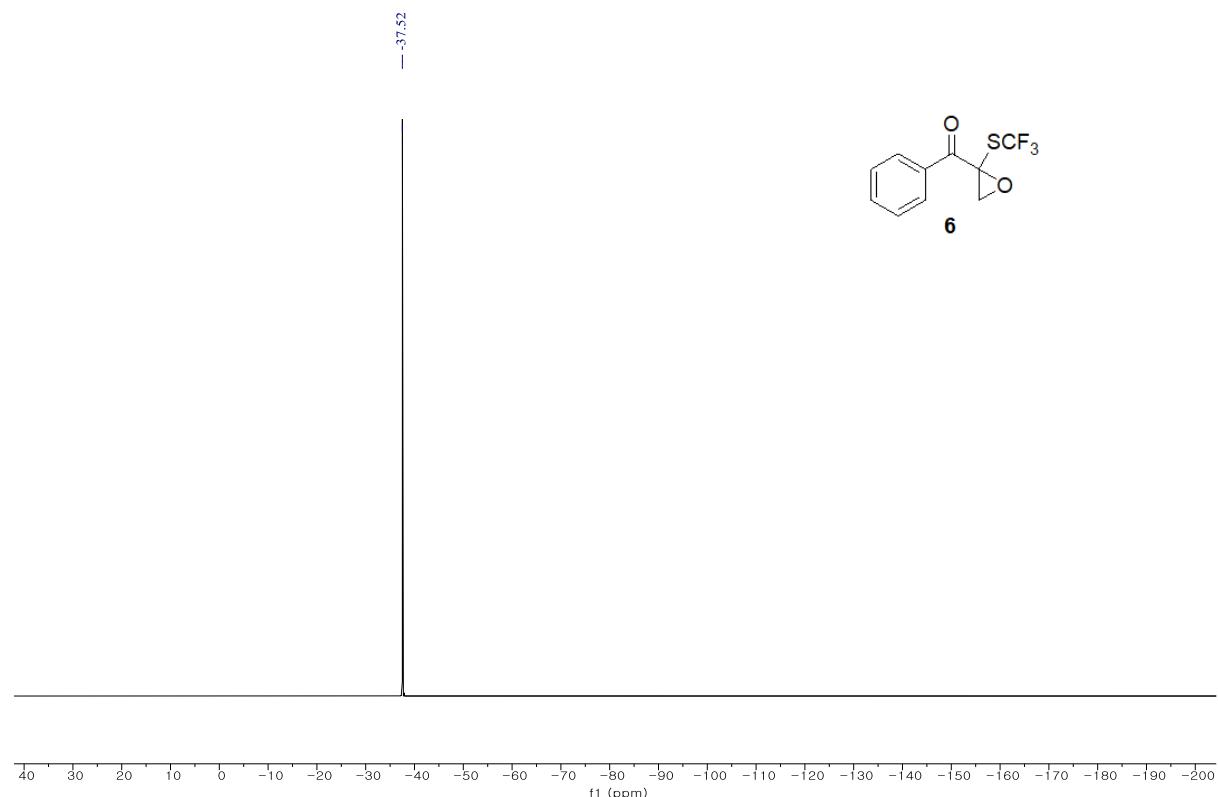
¹H NMR of **6** (500 MHz, CDCl₃)



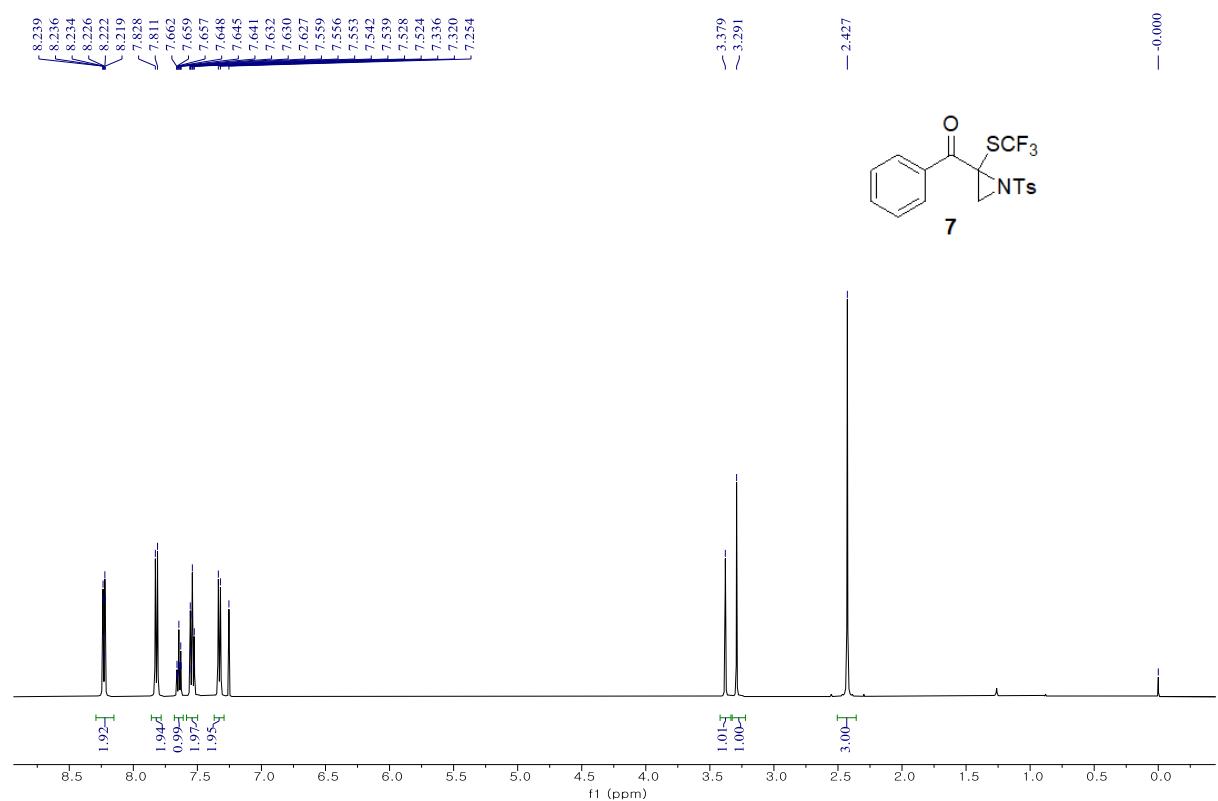
¹³C NMR of **6** (125 MHz, CDCl₃)



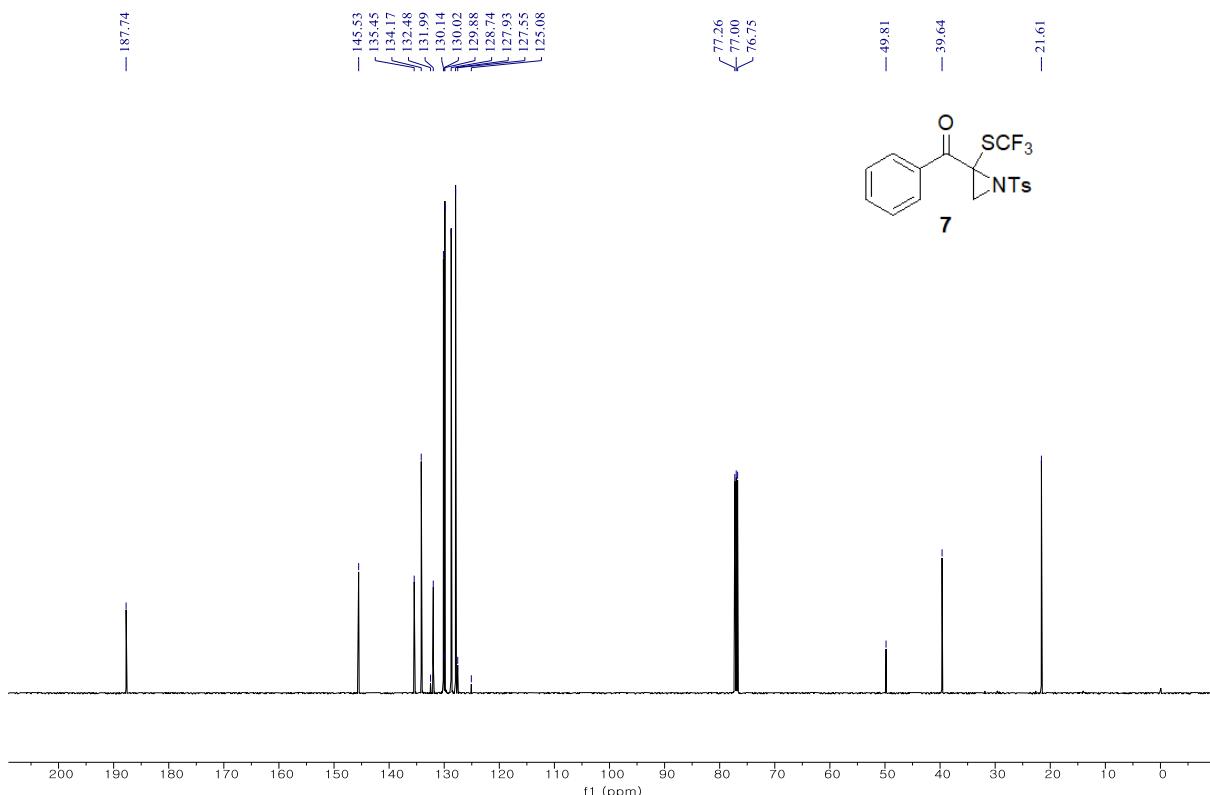
¹⁹F NMR of **6** (564 MHz, CDCl₃)



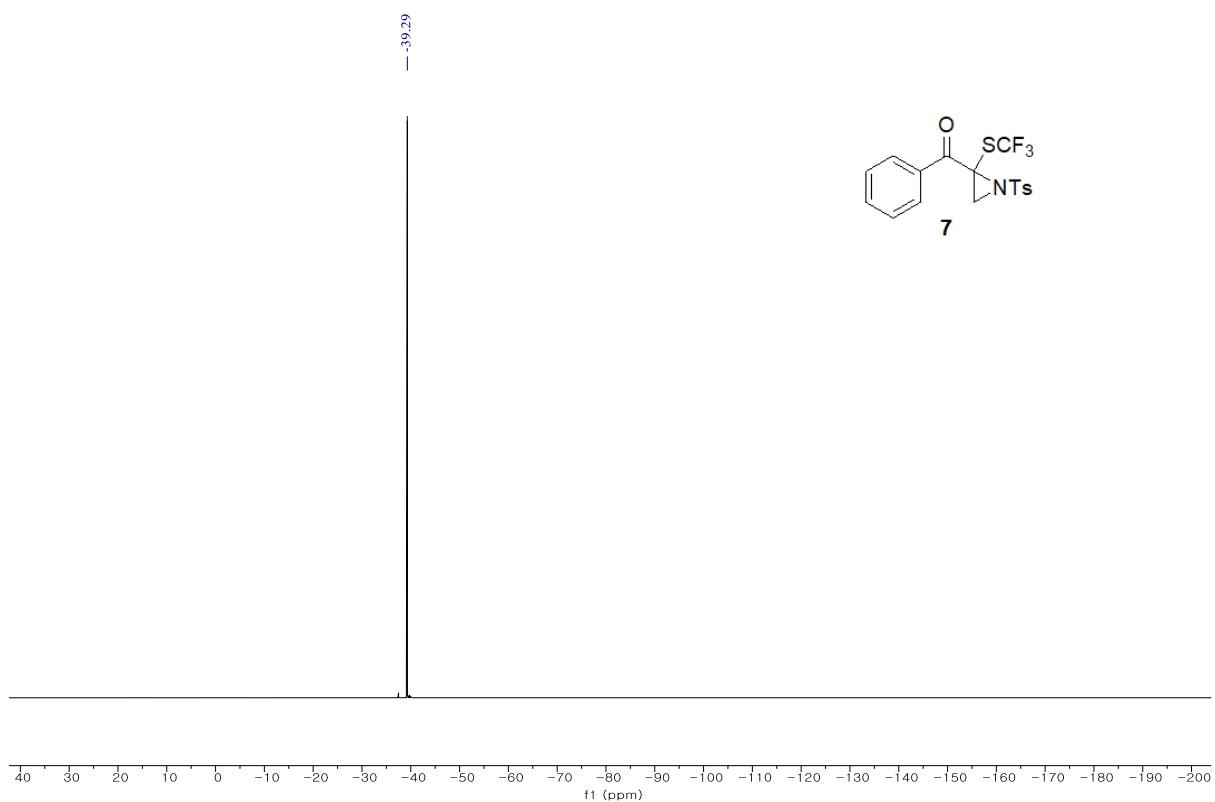
¹H NMR of **7** (500 MHz, CDCl₃)



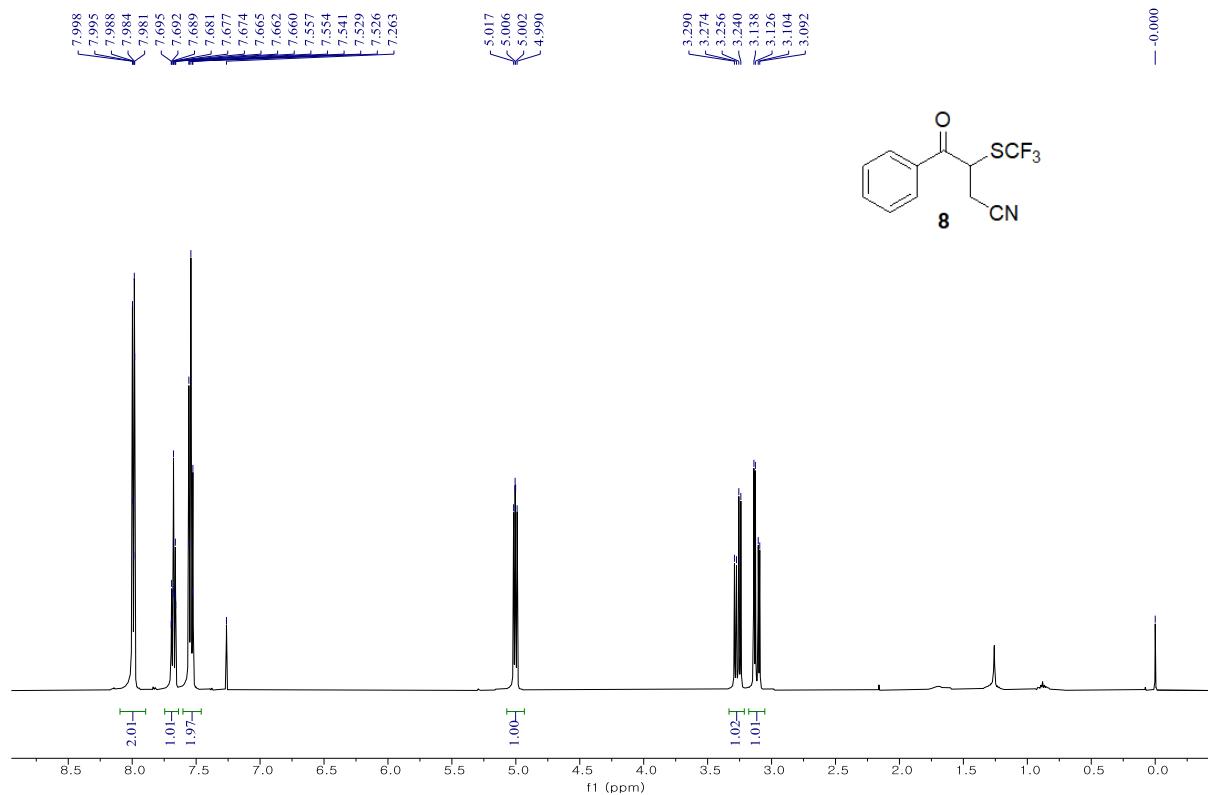
^{13}C NMR of 7 (125 MHz, CDCl_3)



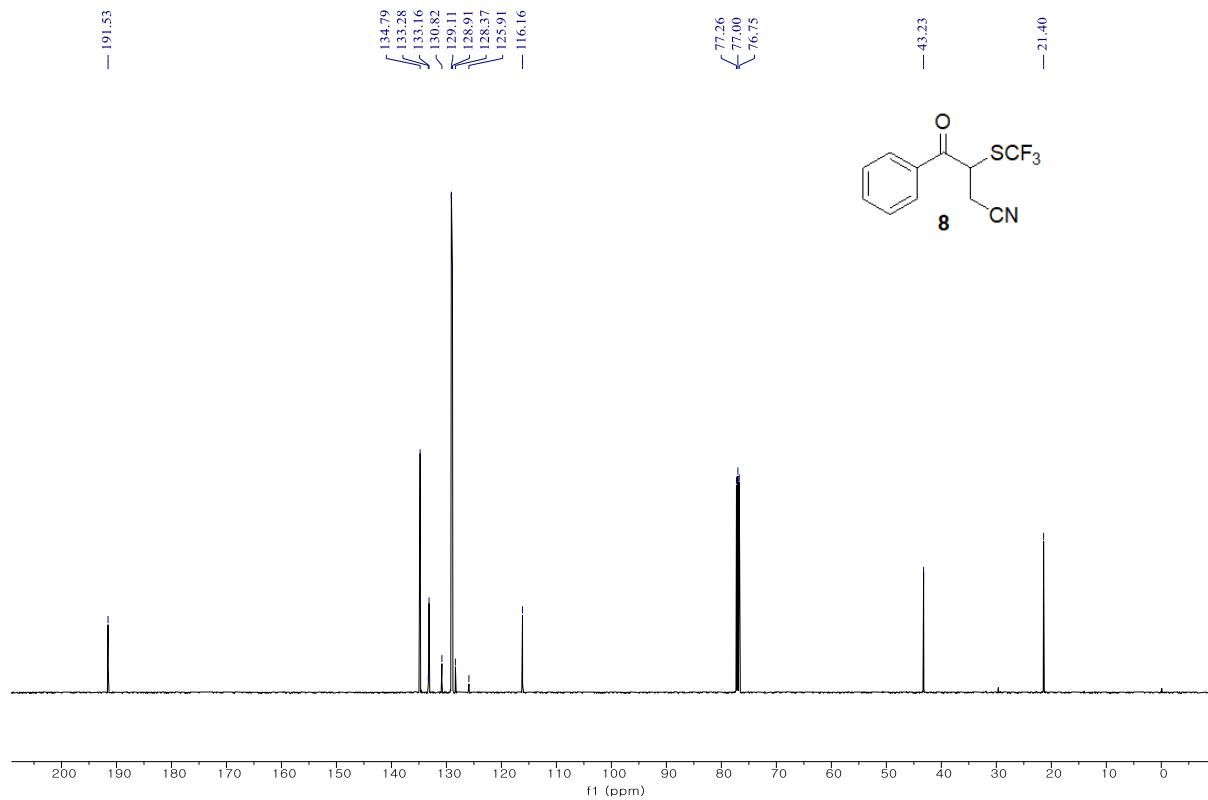
^{19}F NMR of 7 (564 MHz, CDCl_3)



¹H NMR of **8** (500 MHz, CDCl₃)



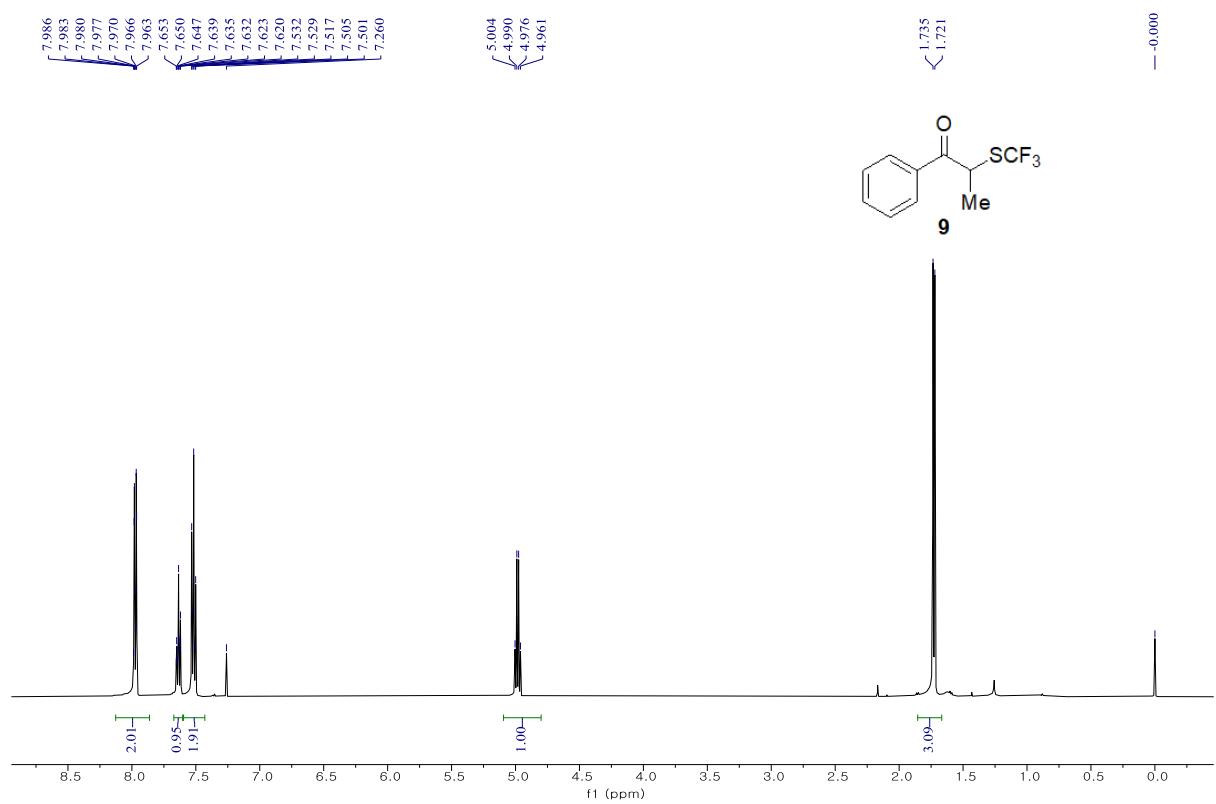
¹³C NMR of **8** (125 MHz, CDCl₃)



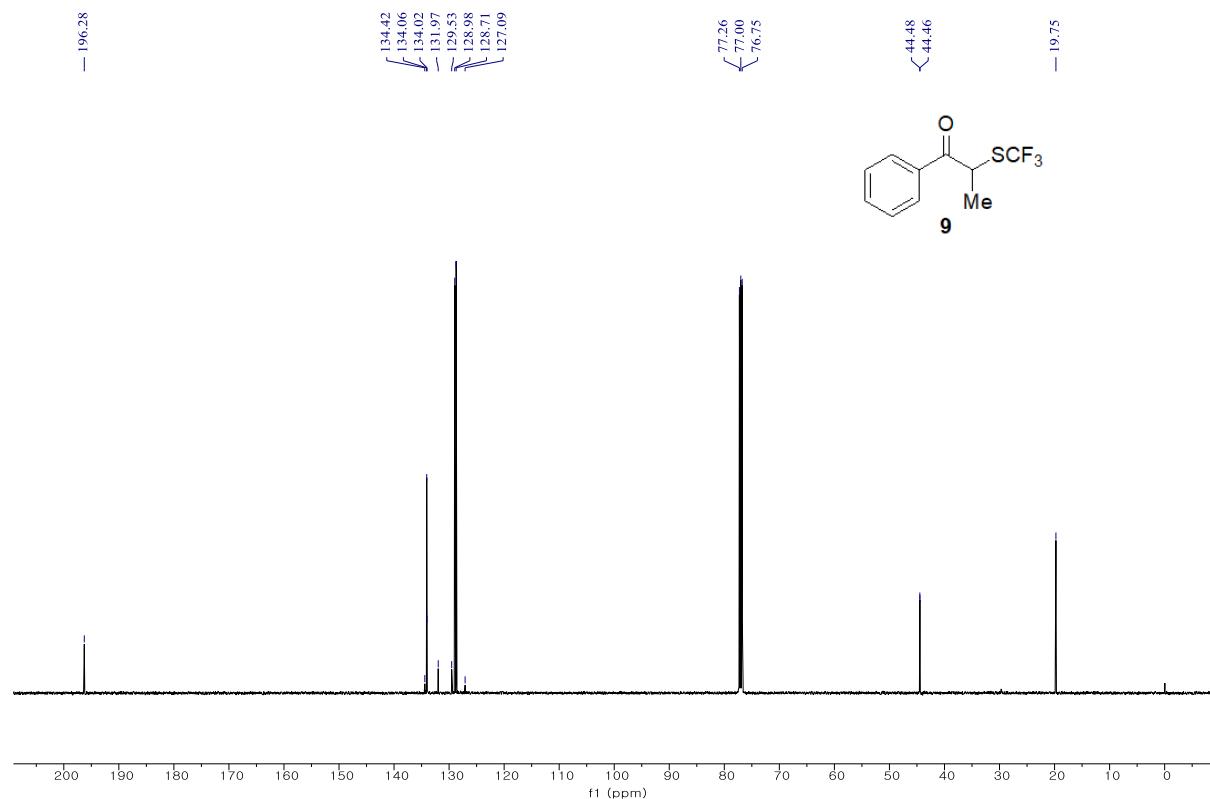
¹⁹F NMR of **8** (564 MHz, CDCl₃)



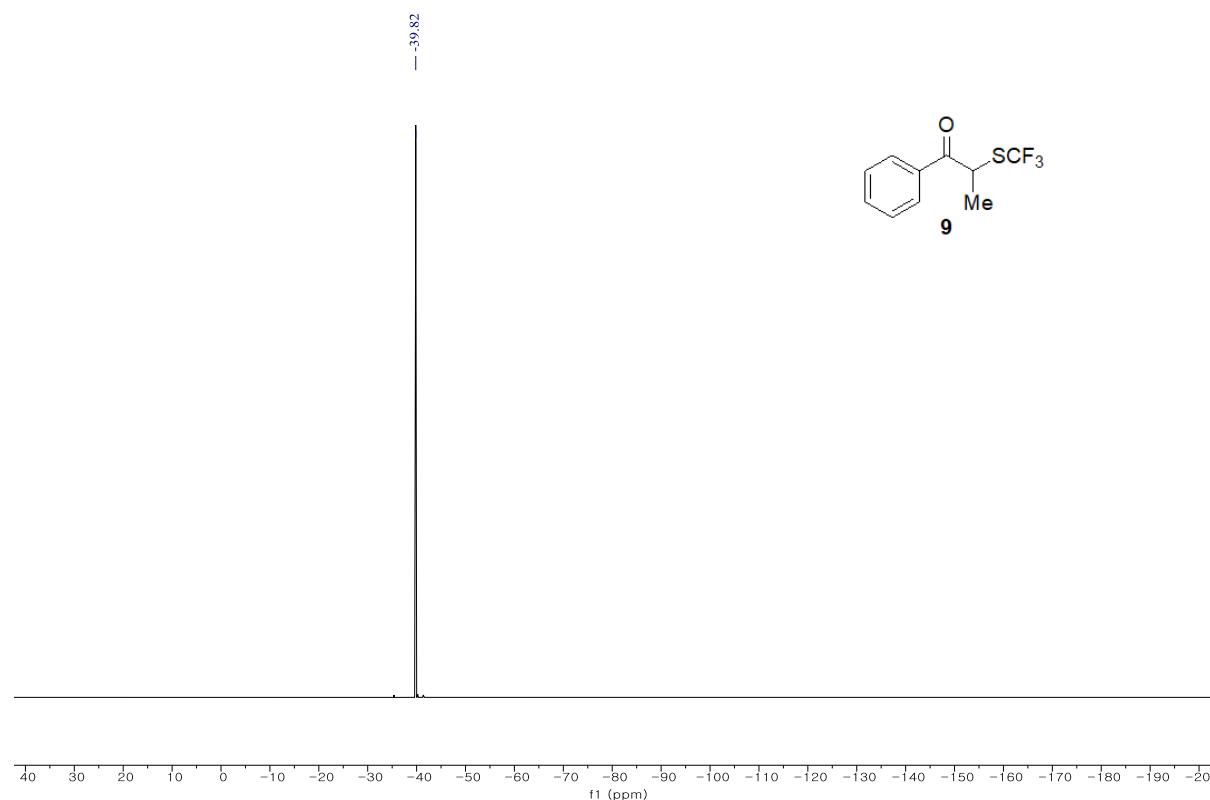
¹H NMR of **9** (500 MHz, CDCl₃)



^{13}C NMR of **9** (125 MHz, CDCl_3)



^{19}F NMR of **9** (564 MHz, CDCl_3)



X-Ray crystallographic data of compound 5

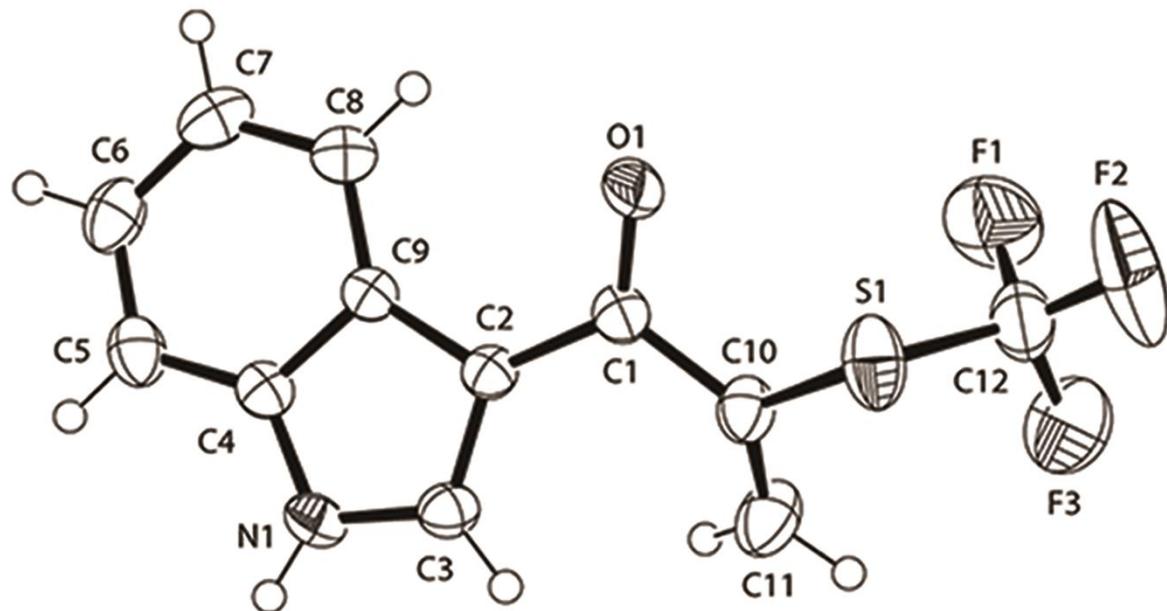


Figure S1. X-Ray crystal structure of **5**. Thermal ellipsoids are drawn at the 50% probability level.

Sample preparation

A single crystal of compound **5** was obtained by slow evaporation of a solution of **5** in dichloromethane–hexanes at room temperature.

Table S1. Crystal data and structure refinement for **5**

Identification code	5
Empirical formula	C12 H8 F3 N O S
Formula weight	271.25
Temperature	223(2) K
Wavelength	0.71073 Å
Crystal system	Orthorhombic
Space group	Pbca
Unit cell dimensions	a = 8.2239(10) Å α = 90°. b = 11.2158(11) Å β = 90°. c = 25.146(3) Å γ = 90°.
Volume	2319.4(5) Å ³
Z	8
Density (calculated)	1.554 Mg/m ³
Absorption coefficient	0.304 mm ⁻¹
F(000)	1104
Crystal size	0.230 x 0.150 x 0.050 mm ³
Theta range for data collection	2.960 to 28.376°.
Index ranges	-10<=h<=10, -14<=k<=14, -33<=l<=33
Reflections collected	72285
Independent reflections	2893 [R(int) = 0.0777]
Completeness to theta = 25.242°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7457 and 0.6950
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	2893 / 0 / 163
Goodness-of-fit on F ²	1.042
Final R indices [I>2sigma(I)]	R1 = 0.0588, wR2 = 0.1393
R indices (all data)	R1 = 0.0823, wR2 = 0.1554
Extinction coefficient	n/a
Largest diff. peak and hole	1.103 and -0.607 e.Å ⁻³