

Efficient Solar Light Driven Degradation of Tetracycline by Fe-EDTA Modified g-C₃N₄ Nanosheets

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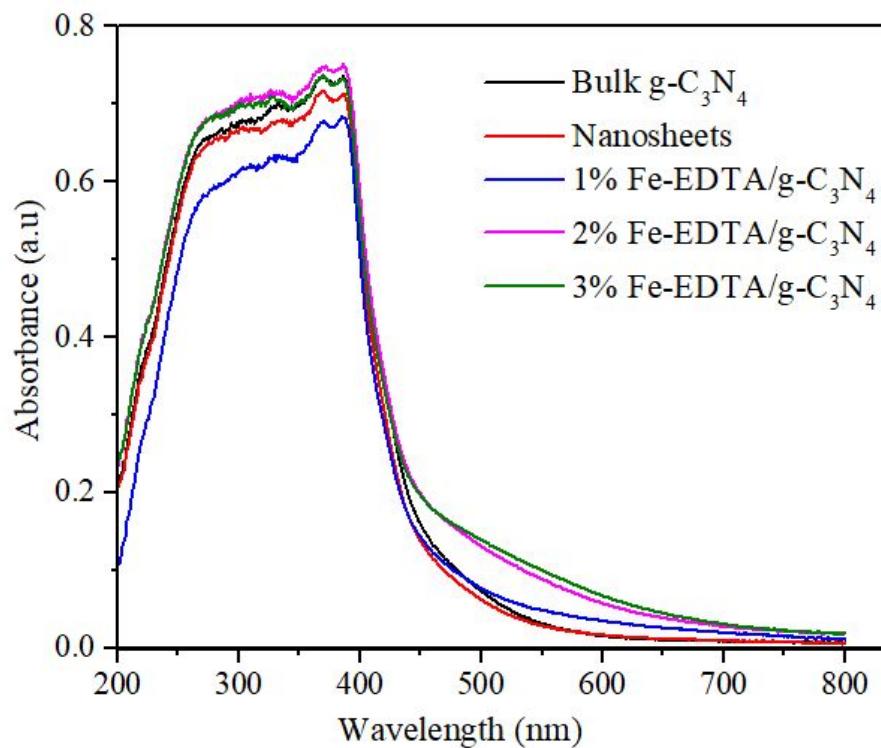


Figure S1. UV-vis-NIR diffuse reflectance spectra of the as-prepared samples.

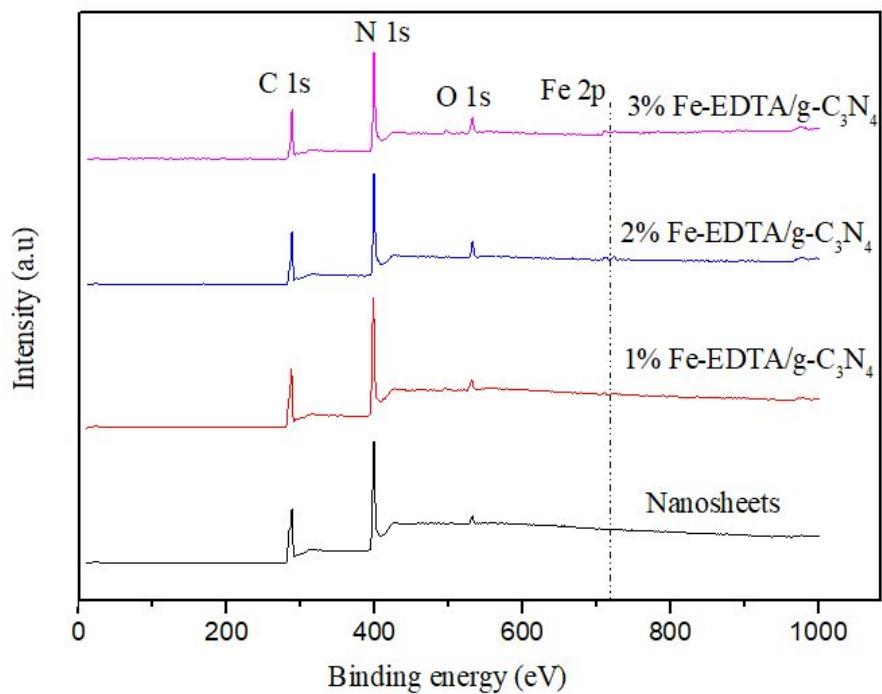


Figure S2. XPS survey spectra of as-prepared samples.

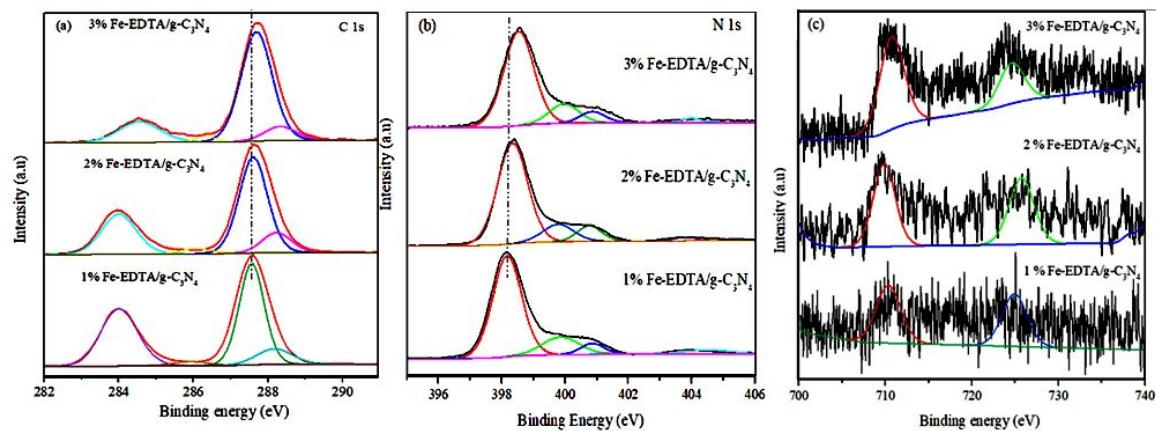


Figure S3. (a) C1s, (b) N1s and (c) Fe 2p spectra for 1 % Fe-EDTA/g-C₃N₄, 2% Fe-EDTA/g-C₃N₄ and 3% Fe-EDTA/g-C₃N₄.

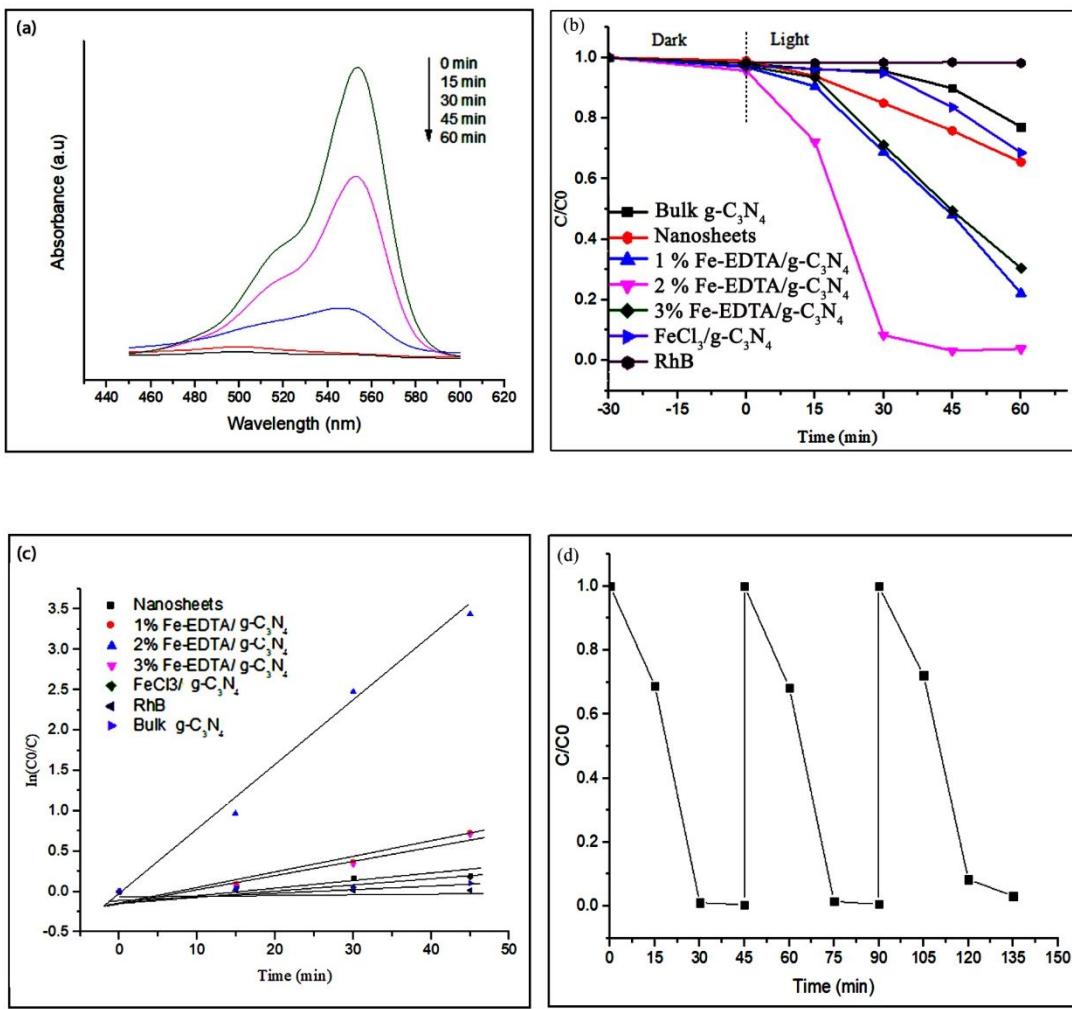


Figure S4. (a) Time-dependent absorption spectra for photodegradation of RhB,
(b) Photodegradation of RhB over different as-prepared samples under simulated solar irradiation,
(c) The corresponding first-order kinetics plots of the prepared samples for RhB degradation,
(d) Recycling experiment for RhB degradation over 2% Fe-EDTA/g-C₃N₄ (Catalyst = 10 mg, RhB = 80 mg/L).

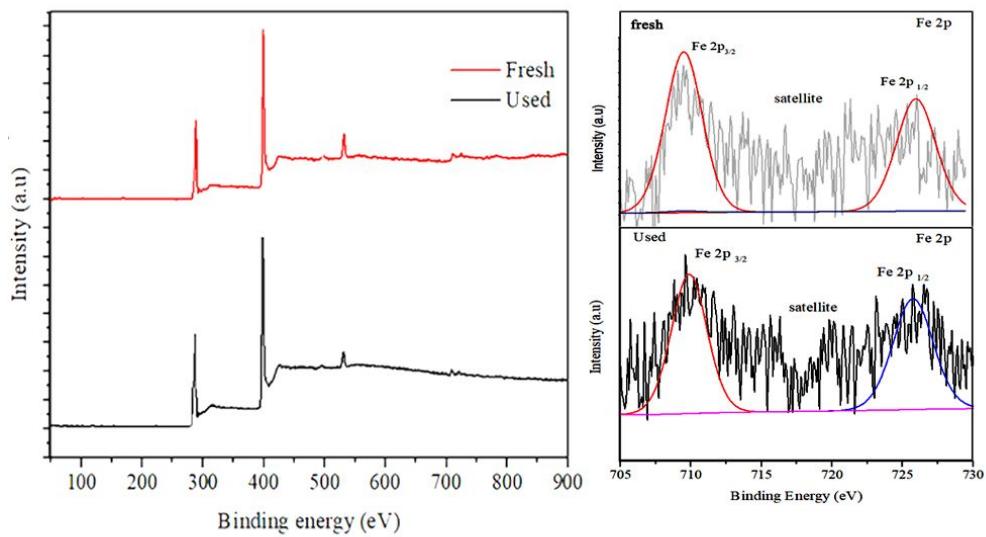


Figure S5. XPS spectra of fresh and used sample of 2% Fe-EDTA/g-C₃N₄.

Table S1. Comparison of degradation efficiencies for organic pollutants with different photocatalysts

Photocatalysts	Organic pollutant	Initial concentration	Photocatalyst dosage	Activation system	Degradation performance	Ref.
g-C ₃ N ₄ /Fe ⁰ (1%) /TiO ₂	TC	10 mgL ⁻¹	1 gL ⁻¹	-	90 min, 98%	1
Co ₃ O ₄ @CoO/g-C ₃ N ₄	TC	10 mgL ⁻¹	0.6 gL ⁻¹	-	120 min, 97%	2
g-C ₃ N ₄ /PDI@MOF	TC	50 mgL ⁻¹	0.4 gL ⁻¹	H ₂ O ₂	1 h, 90%	3
MIL (53) Fe	TC-H	20 mgL ⁻¹	0.16 gL ⁻¹	-	100 min 95%	4
g-C ₃ N ₄ /K/OH/Fe	TC	20 mgL ⁻¹	0.5gL ⁻¹	-	80 min, 63.7%	5
FeCNS	MB	10 mgL ⁻¹	1 gL ⁻¹	-	3 h	6
Fe-gC ₃ N ₄	RhB	10 μM	0.5 gL ⁻¹	H ₂ O ₂	9 h	7
CNNSS/MIL88B	MB	10 mgL ⁻¹	0.4 gL ⁻¹	-	120 min,	8

(Fe)					98%	
Fe-SN	RhB	12 mgL ⁻¹	0.5 gL ⁻¹	-	50 min, 95%	⁹
Fe-EDTA/g-C ₃ N ₄	TC	40 mgL ⁻¹	0.5 gL ⁻¹	-	45 min, 90%	Present work
Fe-EDTA/g-C ₃ N ₄	RhB	80 mgL ⁻¹	0.5 gL ⁻¹	-	45 min, 99.7%	Present work

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