Supporting Information

# Towards Sustained Product Formation in the Liquid Phase Hydrogenation of Mandelonitrile Over a Pd/C Catalyst 

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Figure S1.

Figure S2.


Figure S1. Reaction profile for the initial 3 additions of a 6-batch mandelonitrile hydrogenation reaction. Each reagent addition is comprised of mandelonitrile ( 1.46 g ) and 2 molar equivalents of $\mathrm{H}_{2} \mathrm{SO}_{4}$ reacted over the same $5 \mathrm{wt} \% \mathrm{Pd} / \mathrm{C}$ catalyst $(300 \mathrm{mg})$ in methanol. The reaction was conducted at $40^{\circ} \mathrm{C}$ under 6 barg hydrogen pressure and an agitation speed of 1050 rpm . Each addition (1-3) represents a fixed 2 hour time period. [MN = mandelonitrile; 2-APE = 2-amino-1-phenylethanol; 2AAP = 2-aminoacetophenone; PEA = phenethylamine].


Figure S2. Reaction profile for a 6-batch run for the hydrogenation reaction of mandelonitrile. Each reagent addition is comprised of mandelonitrile ( $1.46 \mathrm{~g}, 11 \mathrm{mmol}$ ) and 2 equivalents of sulphuric acid, reacted over the same $5 \% \mathrm{Pd} / \mathrm{C}$ catalyst $(300 \mathrm{mg})$ in methanol. The reaction was conducted under two different experimental regimes. The solid lines represent 'hydrogen-lean' conditions ( $40^{\circ} \mathrm{C}$ under 4 barg hydrogen pressure with an agitation speed of 600 rpm ), whilst the dashed lines represent 'hydrogen-rich' conditions ( $40{ }^{\circ} \mathrm{C}$ under 6 barg hydrogen pressure with an agitation speed of 1050 $\mathrm{rpm})$. Both reactions were conducted at an external line pressure of 8 barg. Each addition (1-6) represents a fixed 120 minute time period. [ $\mathrm{MN}=$ mandelonitrile; 2-APE = 2-amino-1-phenylethanol; PEA = phenethylamine].

