# HPHT annealing of Ni-containing nitrogen-rich synthetic diamonds and the formation of NE8 centre 

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The nitrogen concentrations in Table 2 are determined using the formula as follows:

The concentration of nitrogen only in the C-centre form $\left(\mathrm{N}_{\mathrm{C}}\right)$ in diamond can be calculated from the absorption coefficient $\alpha_{\mathrm{C}} / \mathrm{cm}^{-1}$ of the peak at $1130 \mathrm{~cm}^{-1[1]}$ to be

$$
\begin{equation*}
\mathrm{N}_{\mathrm{C}} / 10^{-6}=25 \alpha_{\mathrm{C}} \tag{1}
\end{equation*}
$$

The concentration of aggregated nitrogen only in the A-centre form $\left(\mathrm{N}_{\mathrm{A}}\right)$ is determined by measuring the absorption coefficient $\alpha_{\mathrm{A}} / \mathrm{cm}^{-1}$ of the peak at 1282 $\mathrm{cm}^{-1[2]}$ to be

$$
\begin{equation*}
\mathrm{N}_{\mathrm{A}} / 10^{-6}=16.5 \alpha_{\mathrm{A}} \tag{2}
\end{equation*}
$$

On the assumption that spectra are linearly overlapped, the concentration of A-centres and C-centres in mixed type of diamond $\mathrm{IaA}+\mathrm{Ib}$ is still determined by the absorption coefficients $\alpha_{C}$ and $\alpha_{A}$ which are expressed in the terms of $\alpha_{1130}$ and $\alpha_{1282}{ }^{[3]}$ to be

$$
\begin{align*}
& \alpha_{\mathrm{C}}=1.1 \alpha_{1130}-0.2 \alpha_{1282}  \tag{3}\\
& \alpha_{\mathrm{A}}=1.1 \alpha_{1282}-0.2 \alpha_{1130} \tag{4}
\end{align*}
$$

where $\alpha_{1130}$ and $\alpha_{1282}$ are absorption coefficients of the peaks at $1130 \mathrm{~cm}^{-1}$ and 1282 $\mathrm{cm}^{-1}$, respectively. The absorption coefficient of the peak at $2000 \mathrm{~cm}^{-1}$ is well known to be $12.3 \mathrm{~cm}^{-1}$, so $\alpha_{1130}$ and $\alpha_{1282}$ can be obtained by comparing their absorption intensity ( $\mu$ ) with that of the peak at $2000 \mathrm{~cm}^{-1}$ as follows:

$$
\begin{align*}
& \alpha_{1130}=\mu\left(1130 \mathrm{~cm}^{-1}\right) / \mu\left(2000 \mathrm{~cm}^{-1}\right) \times 12.3  \tag{5}\\
& \alpha_{1282}=\mu\left(1282 \mathrm{~cm}^{-1}\right) / \mu\left(2000 \mathrm{~cm}^{-1}\right) \times 12.3 \tag{6}
\end{align*}
$$

The values of absorption intensity are calculated according to the recorded value A in IR spectra to be

$$
\begin{align*}
& \mu\left(1130 \mathrm{~cm}^{-1}\right)=\mathrm{A}\left(1130 \mathrm{~cm}^{-1}\right)-\mathrm{A}\left(1370 \mathrm{~cm}^{-1}\right)  \tag{7}\\
& \mu\left(2000 \mathrm{~cm}^{-1}\right)=\mathrm{A}\left(2000 \mathrm{~cm}^{-1}\right)-\mathrm{A}\left(1370 \mathrm{~cm}^{-1}\right)  \tag{8}\\
& \mu\left(1282 \mathrm{~cm}^{-1}\right)=\mathrm{A}\left(1282 \mathrm{~cm}^{-1}\right)-\mathrm{A}\left(1370 \mathrm{~cm}^{-1}\right) \tag{9}
\end{align*}
$$

Using equations motioned above, NA and NC can be calculated. The calculated results with an uncertainty less than $5 \%$.

## References

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