Copolymerization of Nonpolar Olefins and Allyl Acetate Using a Nickel Catalyst Bearing a Methylene-bridged Bisphosphine Monoxide Ligand

Jin Jung, Hina Yasuda, and Kyoko Nozaki*

Department of Chemistry and Biotechnology, Graduate School of Engineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo-ku, Tokyo 113-8656, Japan

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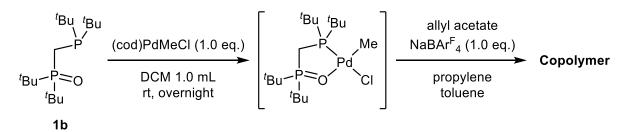
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- 5. X-ray Crystallographic Data of Complex 3
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1. Experimental Section

Copolymerization of propylene with allyl acetate by *in-situ* generated palladium complex (Table S1).

Designated amount of ligand **1b**, anhydrous dichloromethane (1.0 mL), and (cod)PdMeCl (1.0 equiv.) were added into a 20 mL Schlenk under inert atmosphere, and then the reaction mixture was stirred for overnight at room temperature. Subsequently, the reaction mixture was injected *via* syringe into the 50 mL autoclave charged with NaBAr^F₄ (1.0 equiv.), designated amount of propylene, allyl acetate, and anhydrous toluene. After the autoclave was sealed, the reaction mixture was stirred in an isothermal heating block under designated condition. After cooling to room temperature and venting residual propylene, the volatile matters of reaction mixture were removed under reduced pressure. The product was then dried under vacuum for 6 h at 100 °C to afford a copolymer. The molecular weight and polydispersity were determined by SEC. The contents of polar functional groups were determined by ¹H NMR spectra (Figures S32 and S33).

Table S1. Copolymerization with allyl acetate using *in-situ* generated Pd catalyst.



entry	propylene (g)	1b (mmol)	allyl acetate (mL)	temp. (°C)	time (h)	yield (g)	activity (g/mmol·h)	$M_{\rm n}$ (g/mol) ^a	$M_{ m w}/M_{ m n}{}^a$	FG content (mol%) ^b
1	6.4	0.020	1.0	50	12	0.053	0.22	870	1.5	0.25
2	9.3	0.010	0.1	100	6	0.026	0.43	740	1.4	0.16

cod = 1,5-cyclooctadiene, Ar^F = 3,5-(CF₃)₂C₆H₃. ^{*a*} Determined by size-exclusion chromatography using polystyrene standards and corrected by universal calibration. ^{*b*} Determined by ¹H NMR analysis.

Complexation trial of phenylnickel chloride complex 2b

Ligand **1b** (50.0 mg, 0.156 mmol), Ni(cod)₂ (42.9 mg, 0.156 mmol), and anhydrous chlorobenzene (1.0 mL) were added into a 20 mL Schlenk under inert atmosphere, and then the reaction mixture was stirred for 14 h at room temperature. The color of reaction mixture changed from yellow to orange. Subsequently, the reaction mixture was concentrated under reduced pressure, and then *n*-hexane was added to afford an orange precipitate, but it rapidly decomposed during separation to black-brown solid.

Complexation trial of cationic phenylnickel complex

Ligand **1b** (25.0 mg, 78.0 μ mol), Ni(cod)₂ (21.5 mg, 78.2 μ mol), and anhydrous chlorobenzene (1.0 mL) were added into a 10 mL J. Young sample flask under inert atmosphere, and then the reaction mixture was stirred for 14 h at room temperature. The color of reaction mixture changed from yellow to orange. Subsequently, to the reaction mixture was added NaBAr^F₄ (69.1 mg, 78.0 μ mol) and 2,6-lutidine (12.6 mg, 117 μ mol). After stirring for 3 h at room temperature, *n*-hexane was added to the reaction mixture to afford a sticky solid which exhibited a complicated ¹H NMR spectrum and no polymerization activity.

Homopolymerization trial of allyl acetate by *in-situ* generated nickel η^3 -allyl complex and MMAO.

Ligand **1b** (9.6 mg, 0.030 mmol), NaBAr^F₄ (28 mg, 0.030 mmol), $[Ni(\eta^3-allyl)Br]_2$ (5.4 mg, 0.015 mmol), and anhydrous dichloromethane (3.0 mL) were added into a 20 mL Schlenk under inert atmosphere, and then the reaction mixture was stirred for overnight at room temperature. Subsequently, the yellow suspension including sodium bromide salt was injected *via* syringe into the mixture of AAc (5.0 mL), MMAO (4.6 mL, Al/Ni = 500), and anhydrous toluene (3.4 mL). The reaction mixture was stirred for 47 h at 50 °C, then poured into MeOH (300 mL). Trace amount of precipitated was obtained by filter, but it was not determined as poly(AAc).

2. NMR Spectra

NMR spectra of complex 3

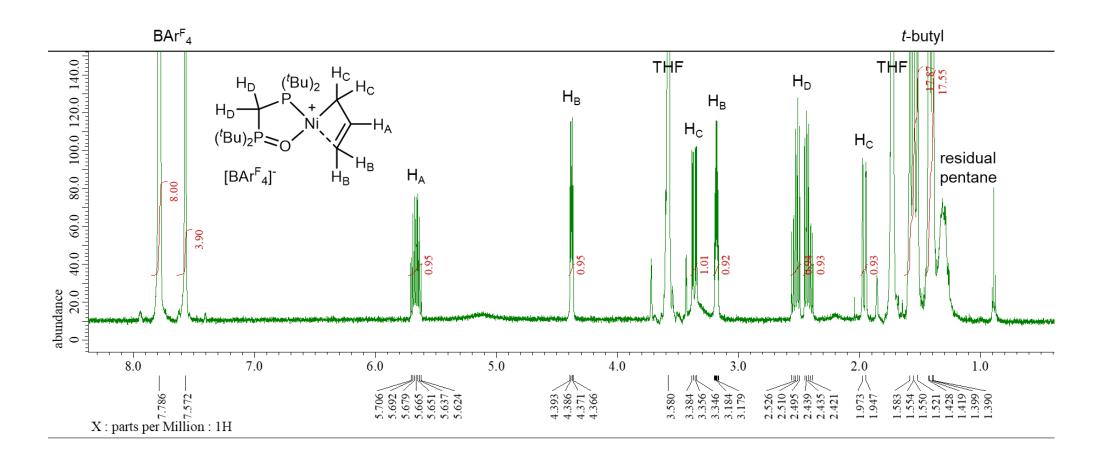


Figure S1. ¹H NMR spectrum of the complex **3** (THF-*d*₈, rt). Peak assignments were based on the previously reported palladium BPMO complex **VI**, ¹ palladium BPMO complex **VI**, ² and ¹H–¹H COSY NMR analysis.

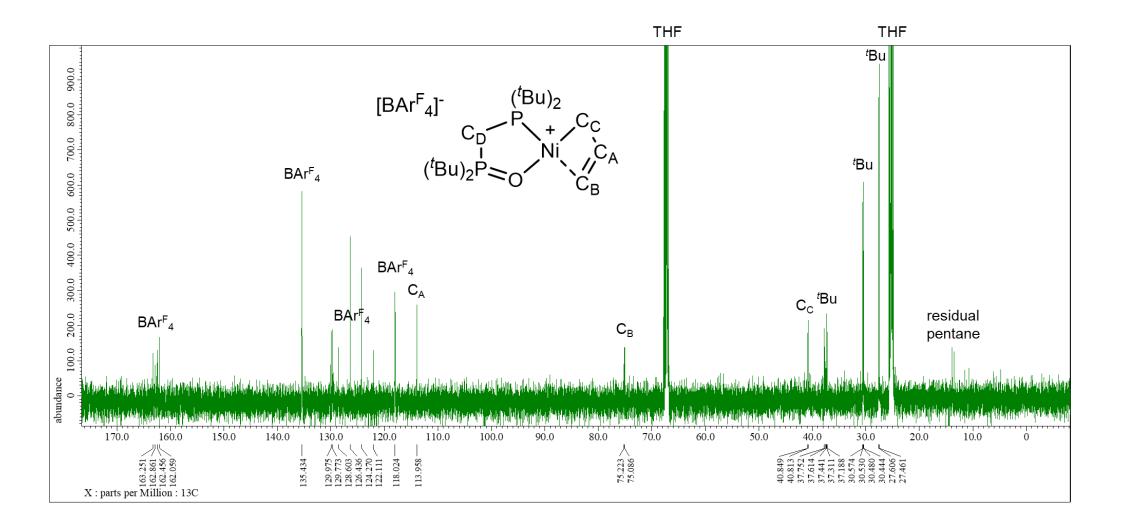


Figure S2. ¹³C{¹H} NMR spectrum of the complex 3 (THF- d_8 , rt). Peak assignments were based on the previously reported palladium BPMO complex VII¹ and palladium BPMO complex VI.²

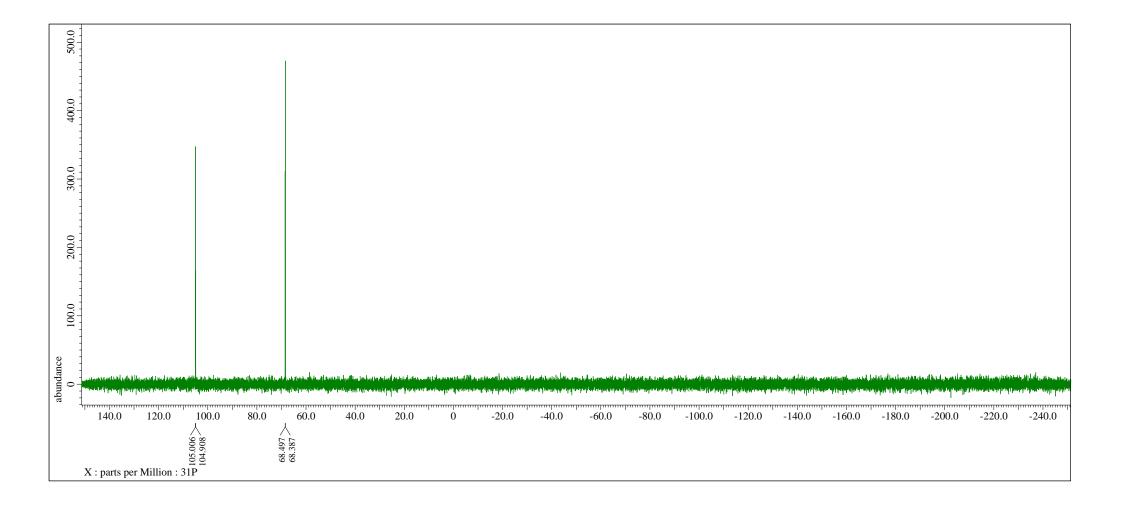


Figure S3. ${}^{31}P{}^{1}H$ NMR spectrum of the complex **3** (THF- d_8 , rt).

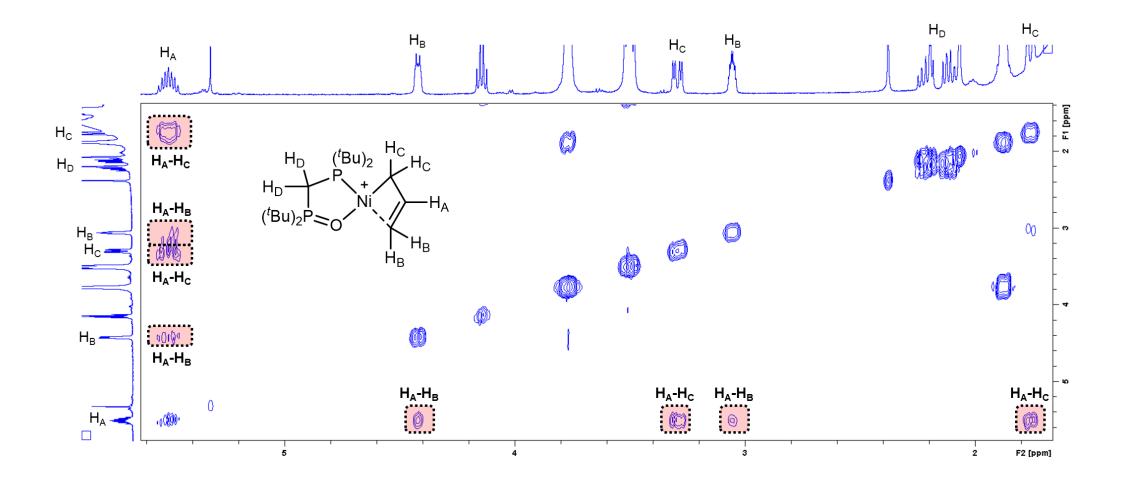


Figure S4. ¹H–¹H COSY NMR spectrum of the complex **3** (CDCl₃, rt).

NMR spectra of polymer from Table 2

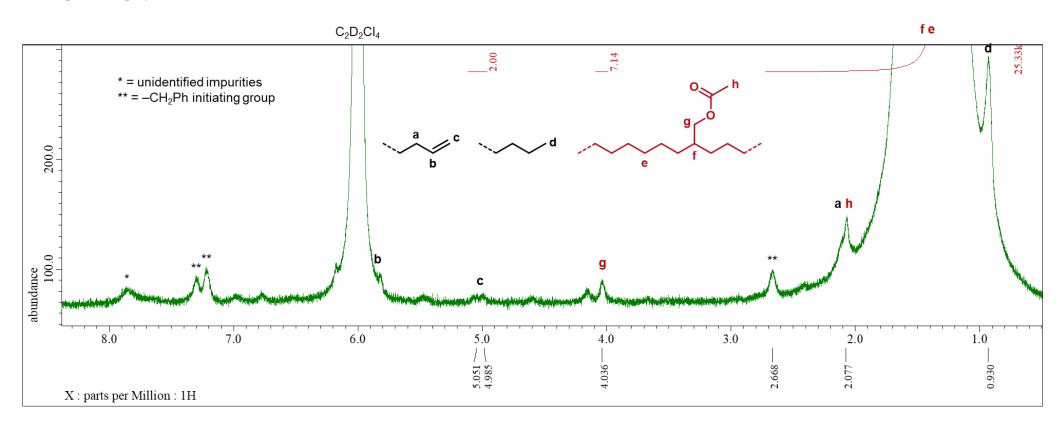


Figure S5. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 2, entry 2 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s). Peak assignments were based on the literature.³

Note: Calculation of acetate group content in the polyethylene³: Assume (1-x) mol% ethylene and x mol% allyl acetate in the copolymer,

$$x = \frac{(\mathbf{g} \times 2 \times 100)}{(\mathbf{b} + \mathbf{c} + \mathbf{j} + \mathbf{a} + \mathbf{h} + \mathbf{f} + \mathbf{e} + \mathbf{d} - \mathbf{g})}$$

Therefore, the acetate group content is calculated as follows: $(7.14 \times 2 \times 100)/(2.00+25330-7.14) = 0.06 \text{ mol}\%$. This calculation was applied to entire cases in this paper.

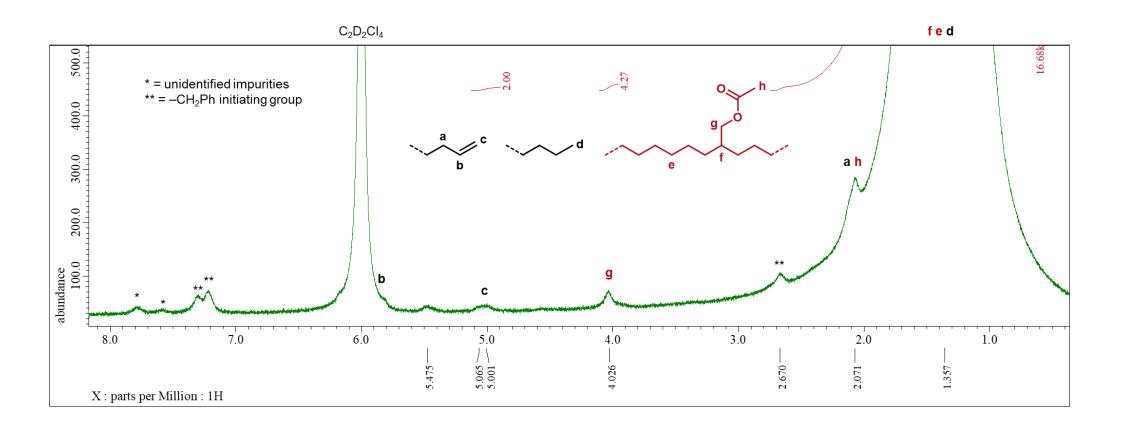


Figure S6. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 2, entry 3 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s)

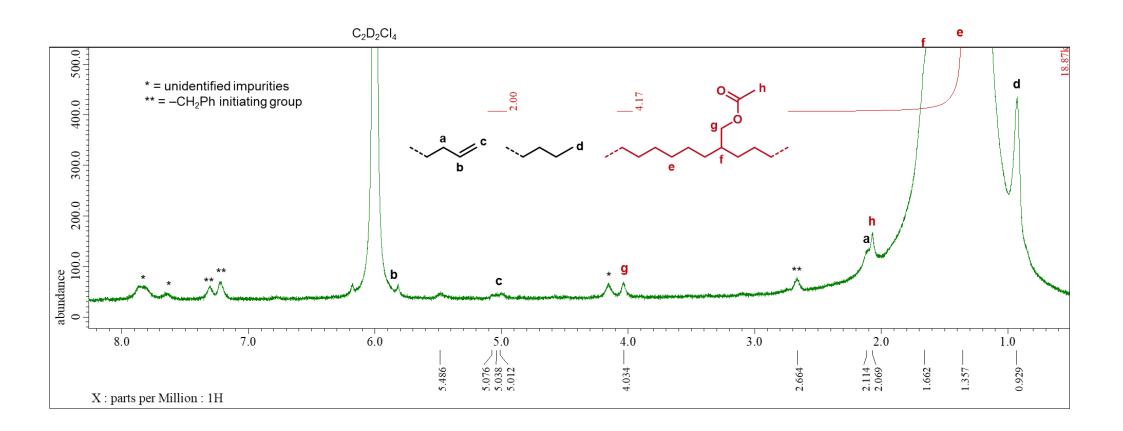


Figure S7. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 2, entry 4 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

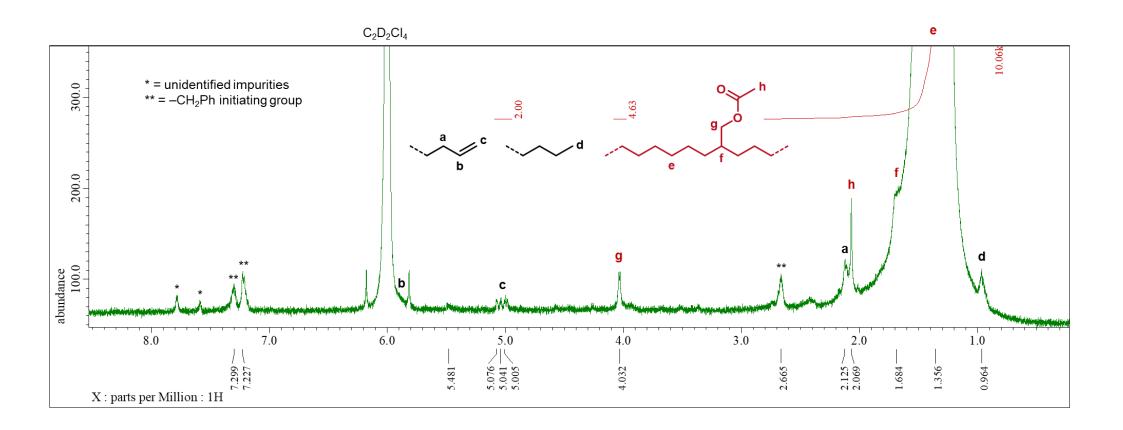


Figure S8. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 2, entry 5 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

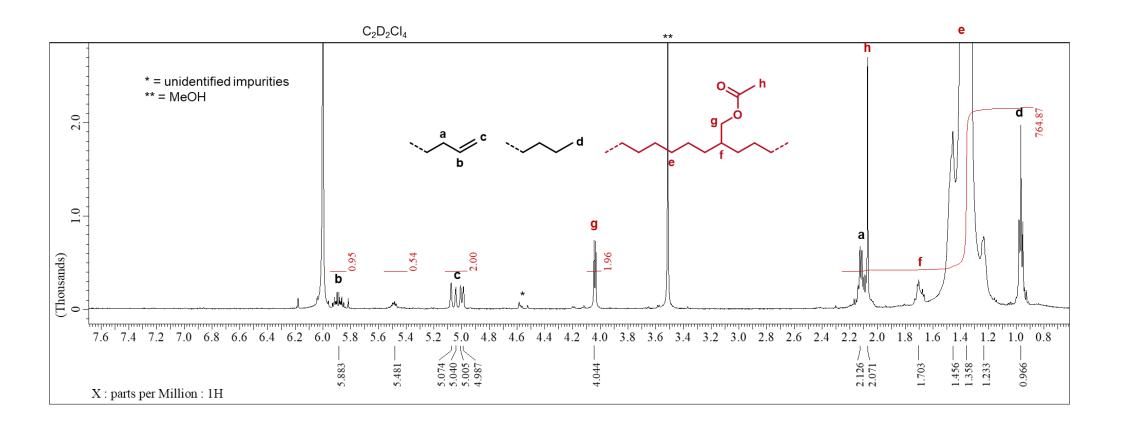


Figure S9. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 2, entry 7 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

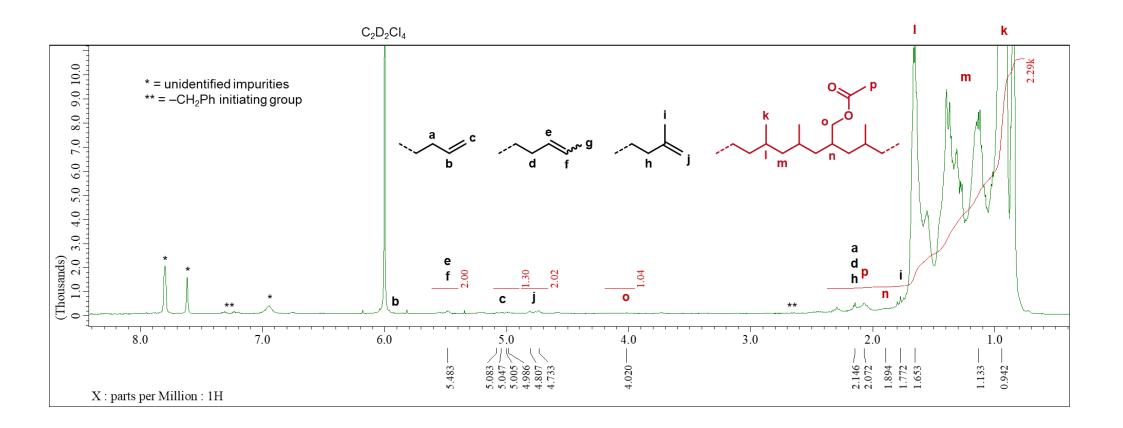


Figure S10. ¹H NMR spectrum of the propylene / allyl acetate copolymer obtained in Table 2, entry 6 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s). Peak assignments were based on the literatures.⁴⁻⁵

Note: Calculation of acetate group content in the polypropylene⁴⁻⁵:

Assume x mol% acetate group in the copolymer,

$$x = \frac{(\mathbf{o}/2) \times 100}{(\mathbf{b} + \mathbf{e} + \mathbf{f} + \mathbf{c} + \mathbf{j} + \mathbf{a} + \mathbf{d} + \mathbf{h} + \mathbf{p} + \mathbf{n} + \mathbf{i} + \mathbf{l} + \mathbf{m} + \mathbf{k})/6}$$

This calculation was applied to entire cases in this paper.

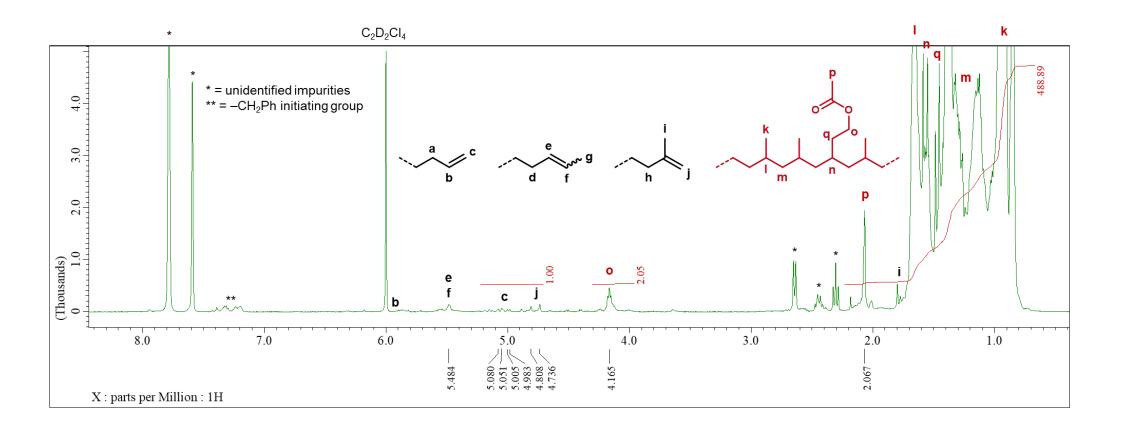


Figure S11. ¹H NMR spectrum of the propylene / 3-butenyl acetate copolymer obtained in Table 2, entry 7 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s). Peak assignments were based on the poly(propylene-*co*-allyl acetate)⁴⁻⁵ and model compound (3-methyl-1-butyl acetate).⁶

NMR spectra of polymer from Table 3

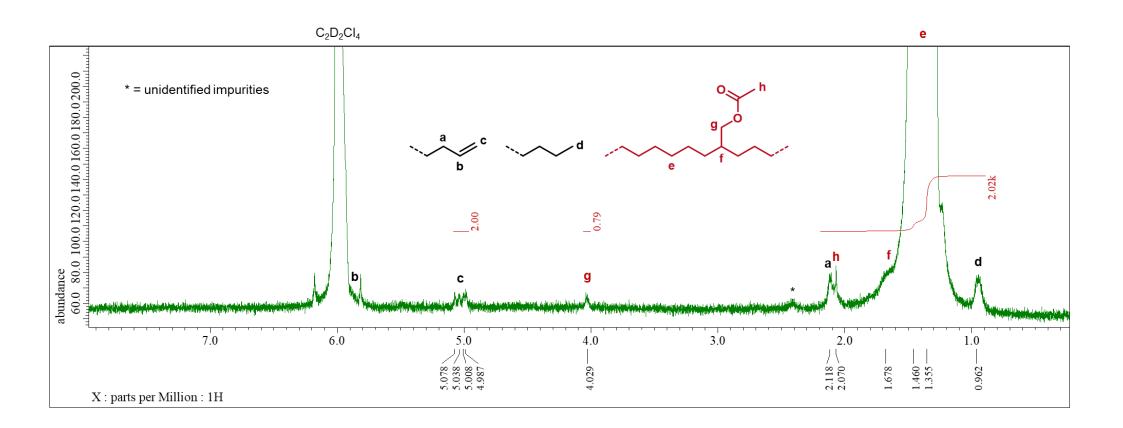


Figure S12. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 2 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

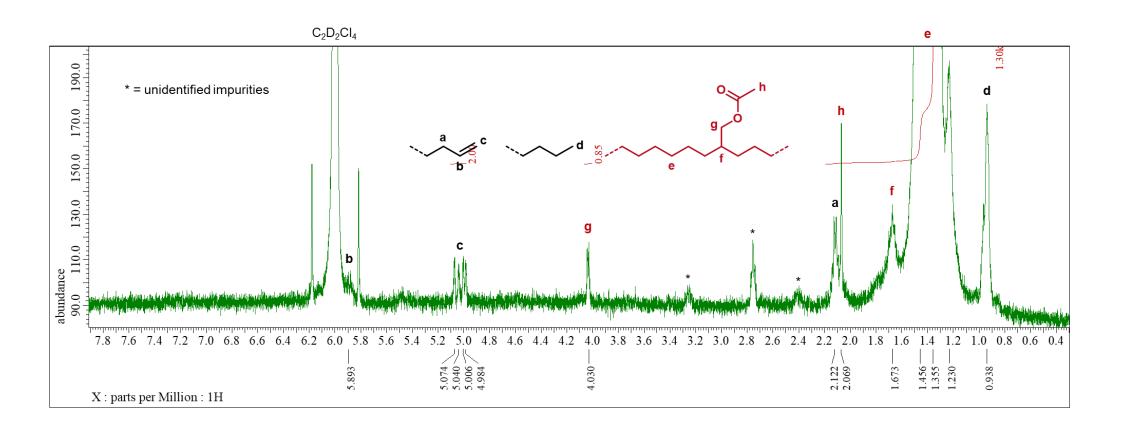


Figure S13. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 3 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

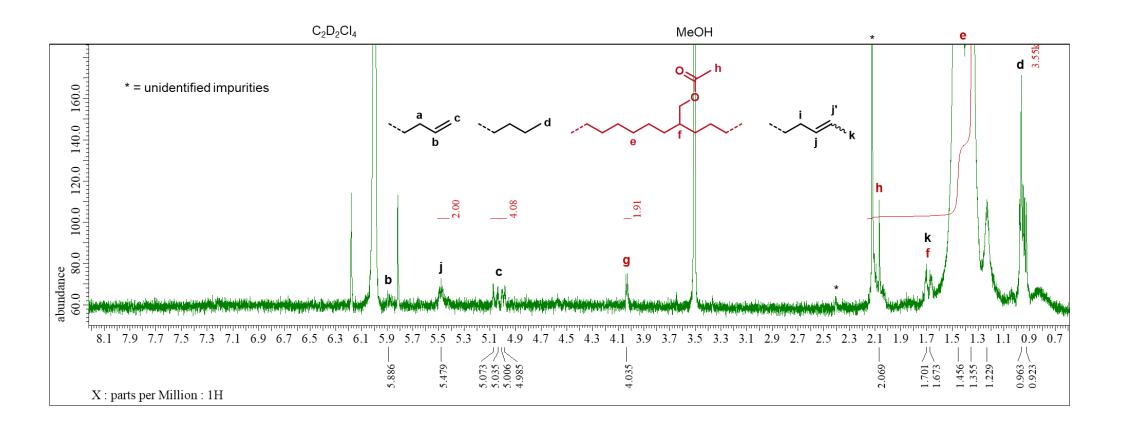


Figure S14. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 4 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

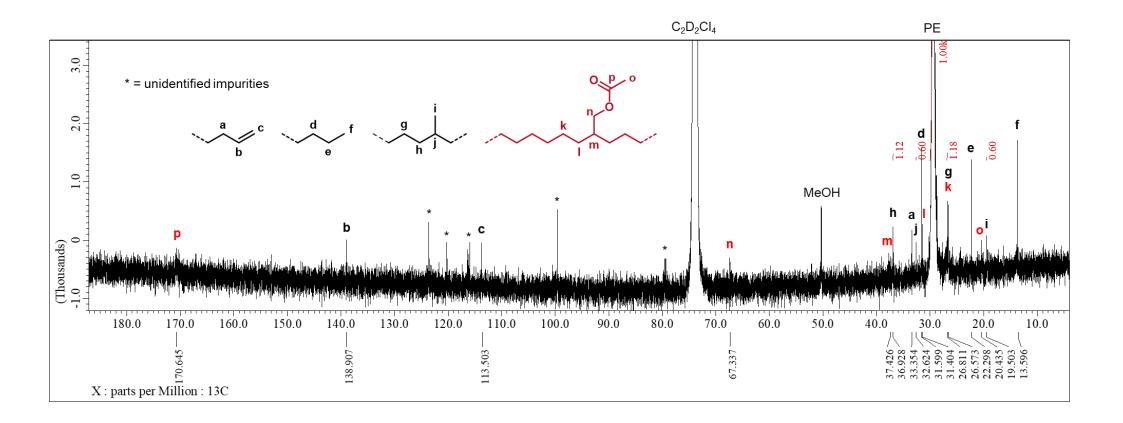


Figure S15. Quantitative ¹³C NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 4 (C₂D₂Cl₄, 120 °C). Peak assignments were based on the

literature.1

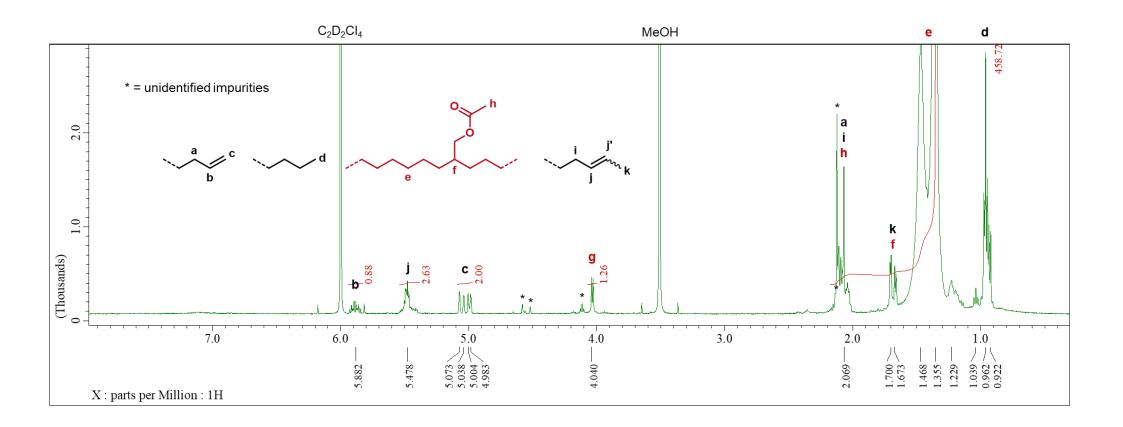


Figure S16. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 5 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

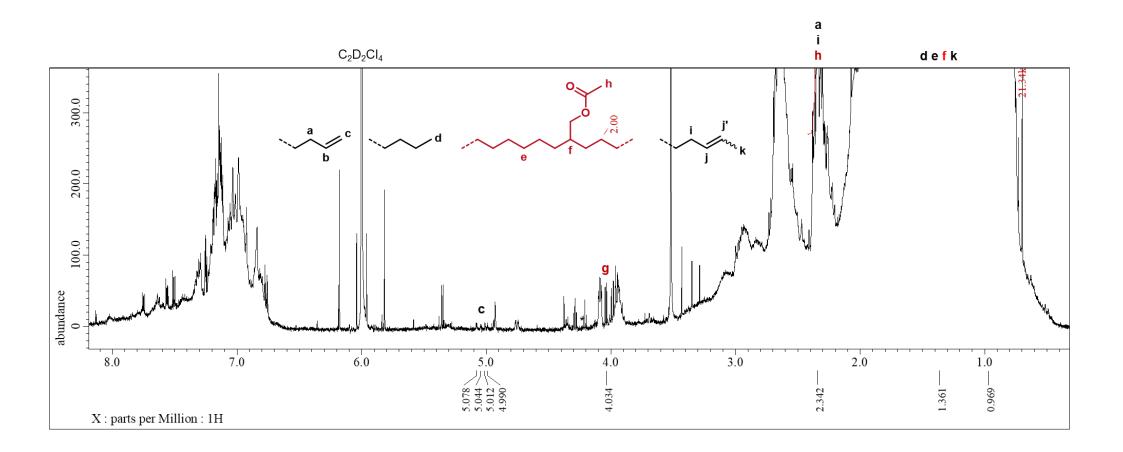


Figure S17. ¹H NMR spectrum of the polymer recovered from SEC eluent (retention time 18–20 min., C₂D₂Cl₄, 120 °C, relaxation delay 5 s). Polymer from Table 3, entry

5 was used to the SEC measurement. Unassigned peaks were incorporated during SEC analysis and recollecting sample.

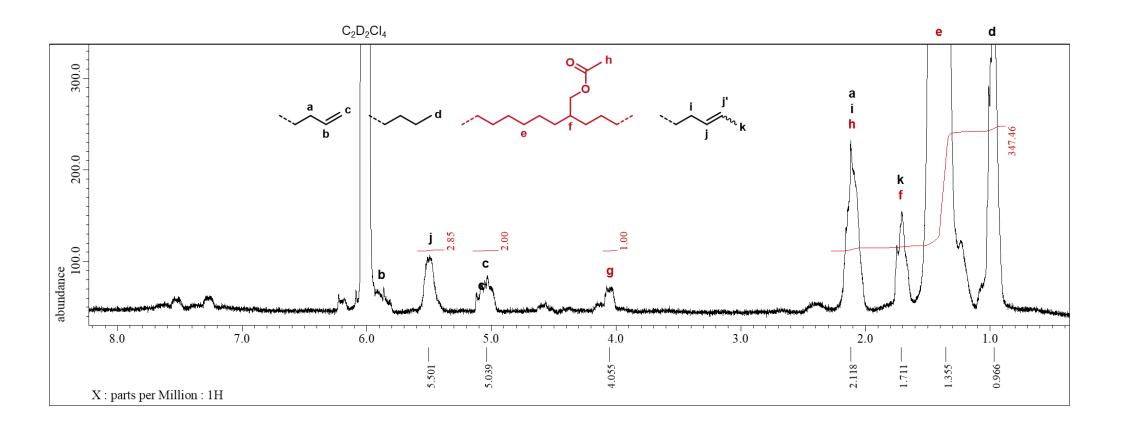


Figure S18. ¹H NMR spectrum of the polymer recovered from SEC eluent (retention time 21–23 min., C₂D₂Cl₄, 120 °C, relaxation delay 5 s). Polymer from Table 3, entry

5 was used to the SEC measurement. Unassigned peaks were incorporated during SEC analysis and recollecting sample.

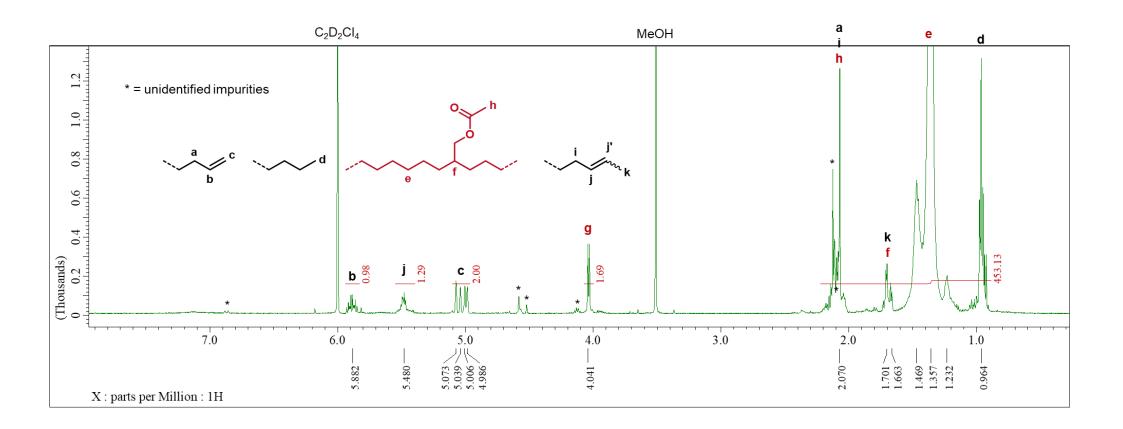


Figure S19. ¹H NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 6 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

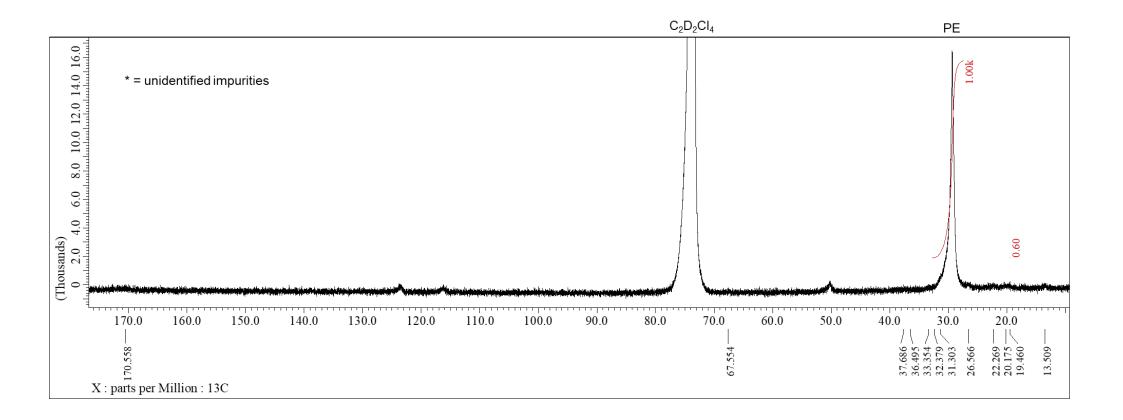


Figure S20. Quantitative ¹³C NMR spectrum of the ethylene / allyl acetate copolymer obtained in Table 3, entry 6 (C₂D₂Cl₄, 120 °C).

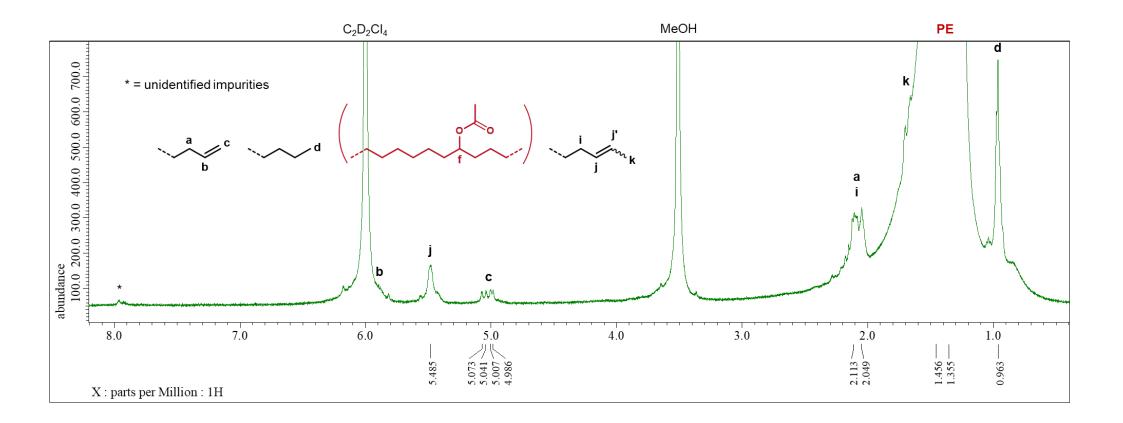


Figure S21. ¹H NMR spectrum of the polyethylene obtained in Table 3, entry 7 ($C_2D_2Cl_4$, 120 °C, relaxation delay 5 s). Characteristic peak of acetoxy group (f at 4.90–4.80 ppm)⁷ was not detected.

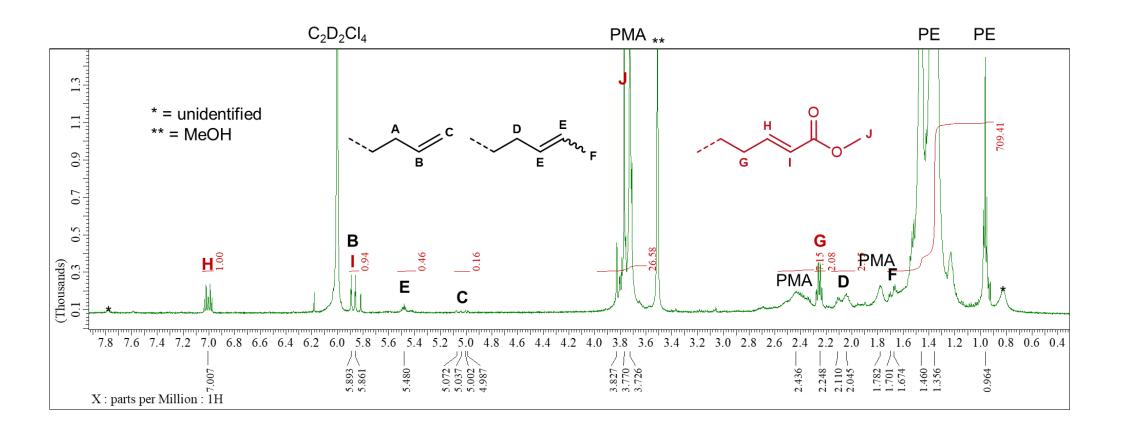


Figure S22. ¹H NMR spectrum of the ethylene / methyl acrylate copolymer obtained in Table 3, entry 8 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s). Peak assignments were based on the literatures.⁸⁻⁹

Note: Calculation of terminal α,β-unsaturated ester group content: Assume (1-x) mol% ethylene and x mol% terminal α,β-unsaturated ester group in the copolymer,

$$\frac{x}{x+4(1-x)} = \frac{\mathbf{H}}{\mathbf{H}+\mathbf{E}+\mathbf{C}+\mathbf{D}+\mathbf{F}+\mathbf{PE}} \qquad x = \frac{4\times\mathbf{H}\times100}{(4\times\mathbf{H})+\mathbf{E}+\mathbf{C}+\mathbf{D}+\mathbf{F}+\mathbf{PE}}$$

Therefore, the terminal α,β -unsaturated ester group content is calculated as follows: $(4 \times 1.00 \times 100)/{(4 \times 1.00)+0.46+0.16+2.15+709.41)} = 0.56 \text{ mol}\%$

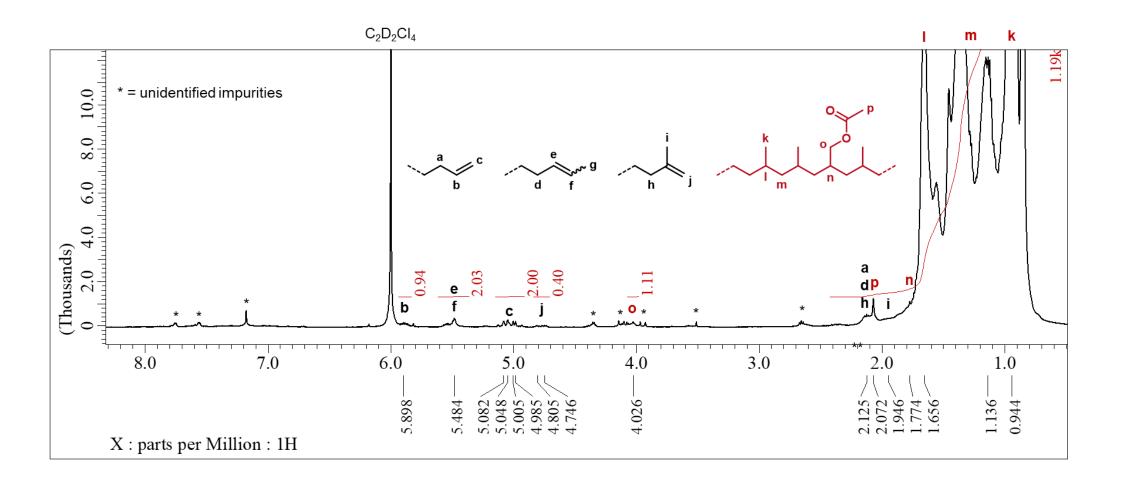


Figure S23. ¹H NMR spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

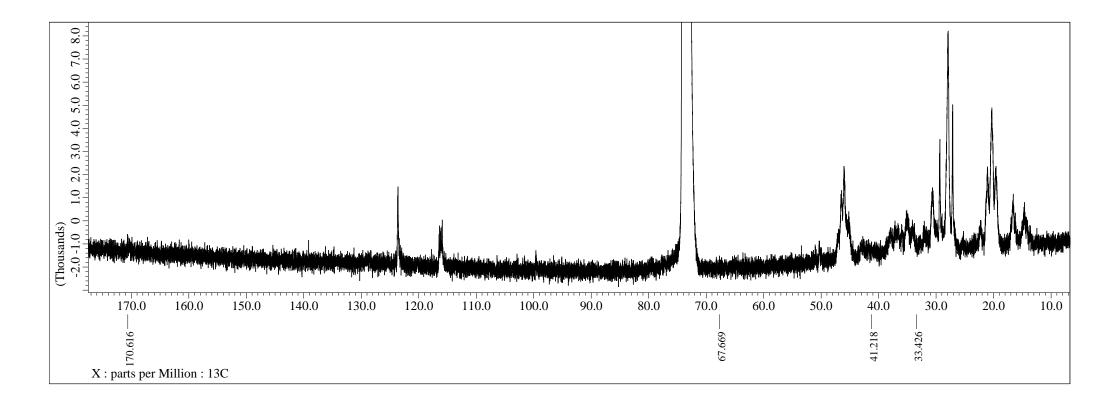


Figure S24. Quantitative ¹³C NMR spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 (C₂D₂Cl₄, 120 °C).

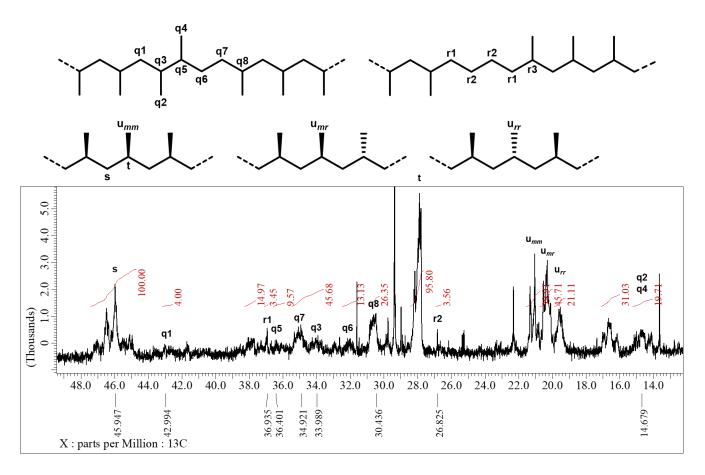


Figure S25. Amplified quantitative ¹³C NMR spectrum from Figure S24.

Note: Calculation of regiodefect (mol%)

$$x = \frac{[(q2+q4)/2] + (r2/2)}{\text{all carbons/3}} \times 100$$

Therefore, the regiodefect is calculated as follows: $[{(19.71/2)+(3.56/2)}/{(100+4+14.97+3.45+9.57+45.68+13.13+26.35+95.80+3.56+39.91+45.71+21.11+31.03+19.71)/3}]$

= 7.36 mol%. Among them, 1,2-/2,1-regiodefect (q1-q8) is estimated to be ca. 5.91 mol% while 1,3-enchainment (r1-r3) is estimated to be ca. 0.28 mol%

There are some other unidentified signals possibly from other defects structures. On the other hand, the triad abundances were determined by integral values of u_{mm}, u_{mr}, u_{rr}.

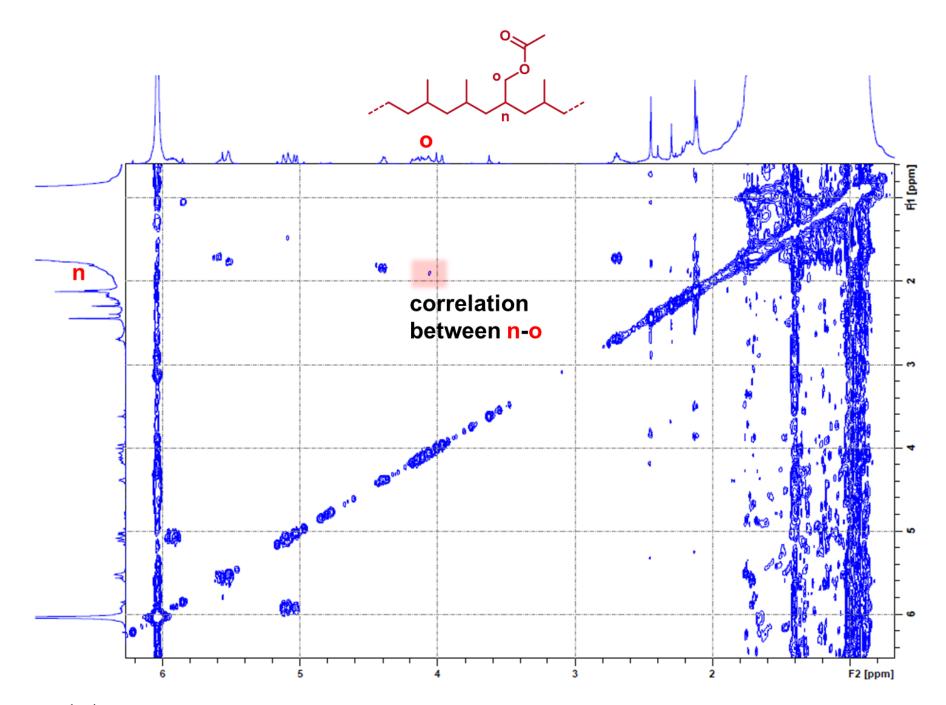


Figure S26. ¹H–¹H COSY spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 (C₂D₂Cl₄, 120 °C).

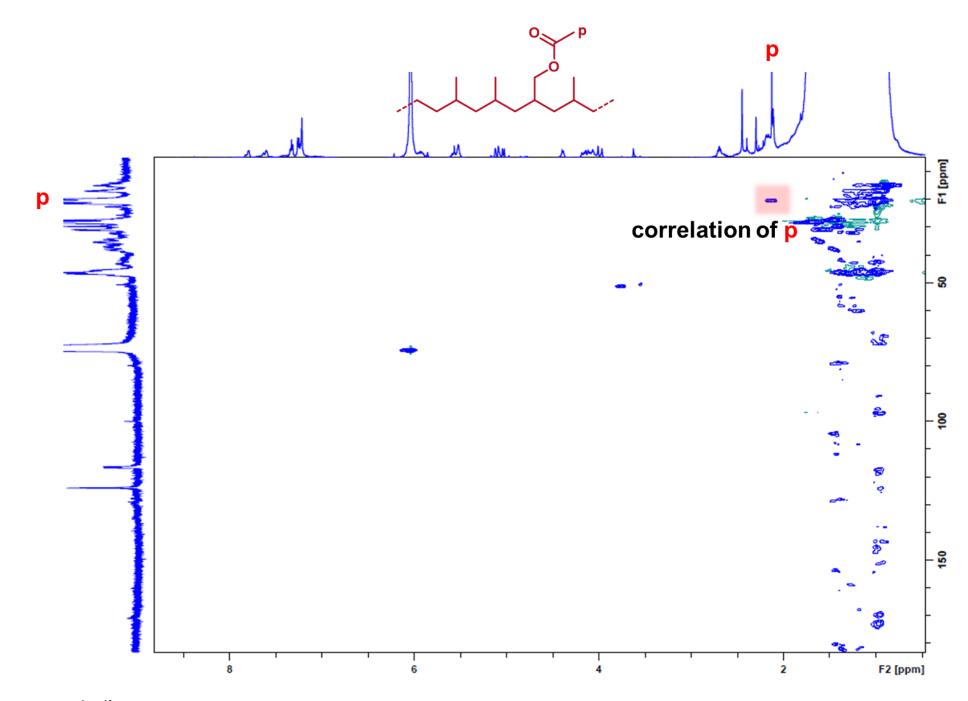


Figure S27. $^{1}\text{H}^{-13}\text{C}$ HSQC spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 (C₂D₂Cl₄, 120 °C).

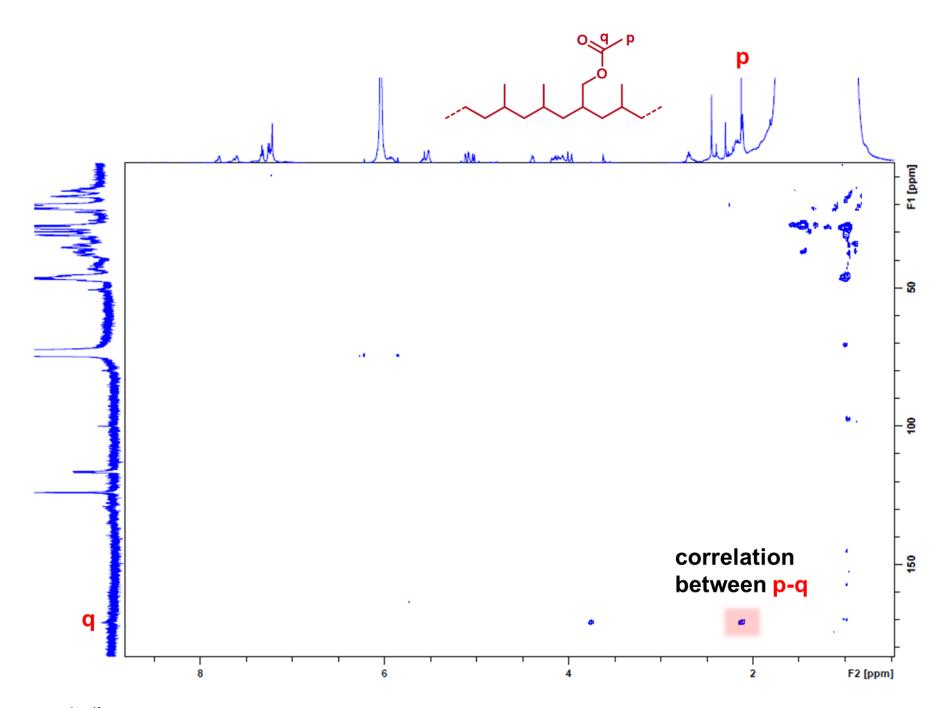


Figure S28. $^{1}\text{H}^{-13}\text{C}$ HMBC spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 ($C_{2}D_{2}Cl_{4}$, 120 °C).

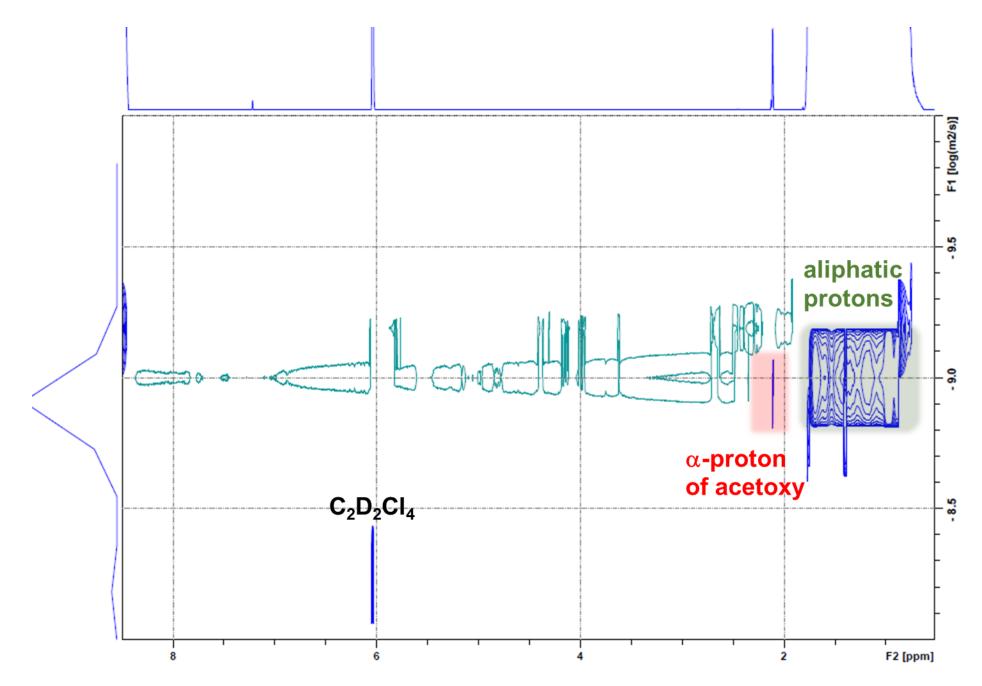


Figure S29. DOSY spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 (C₂D₂Cl₄, 120 °C).

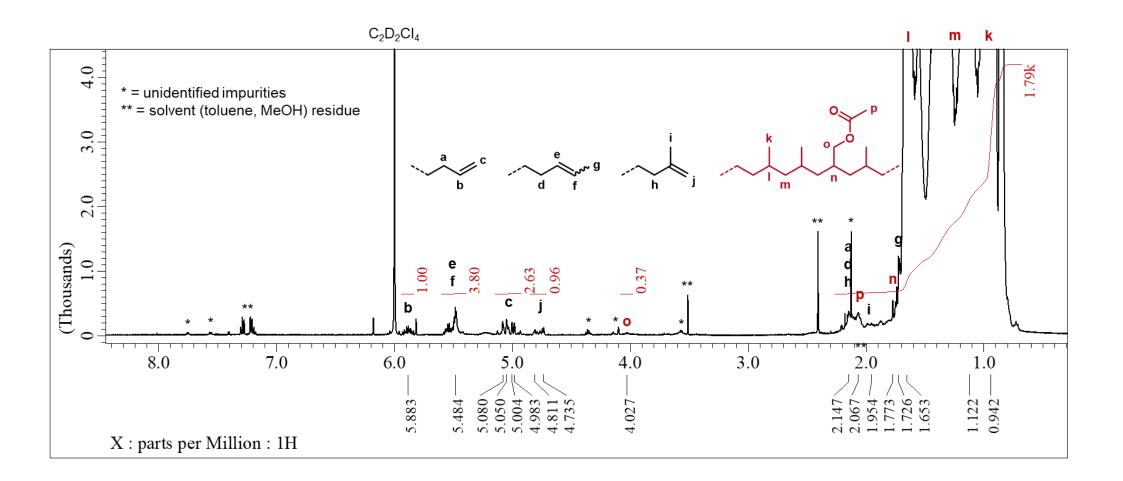


Figure S30. ¹H NMR spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 10 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

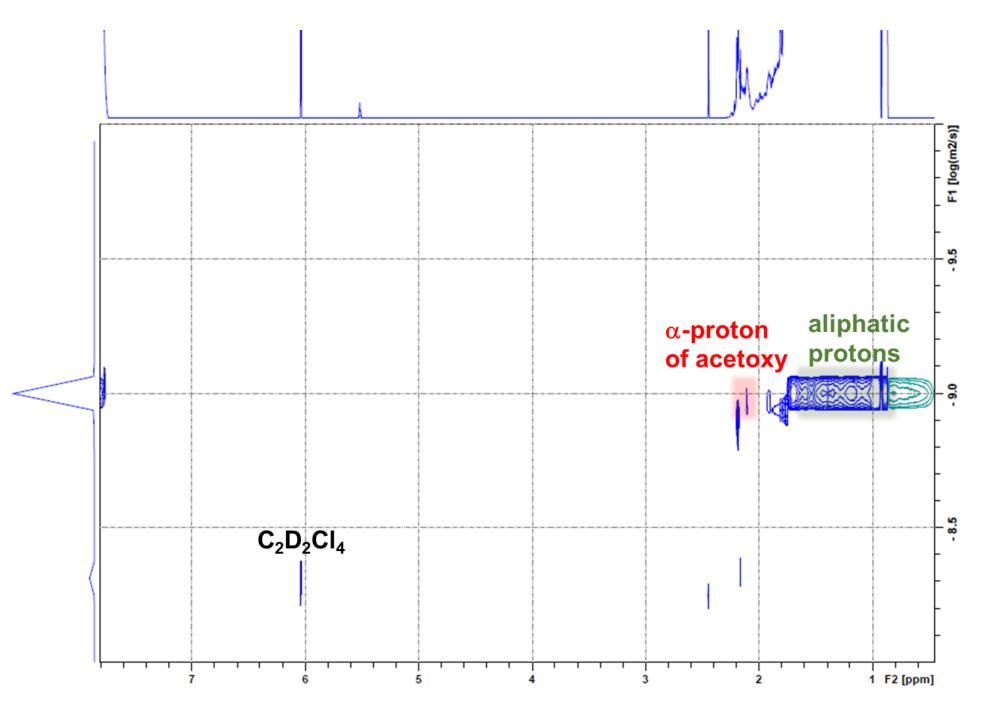


Figure S31. DOSY spectrum of the propylene / allyl acetate copolymer obtained in Table 3, entry 10 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

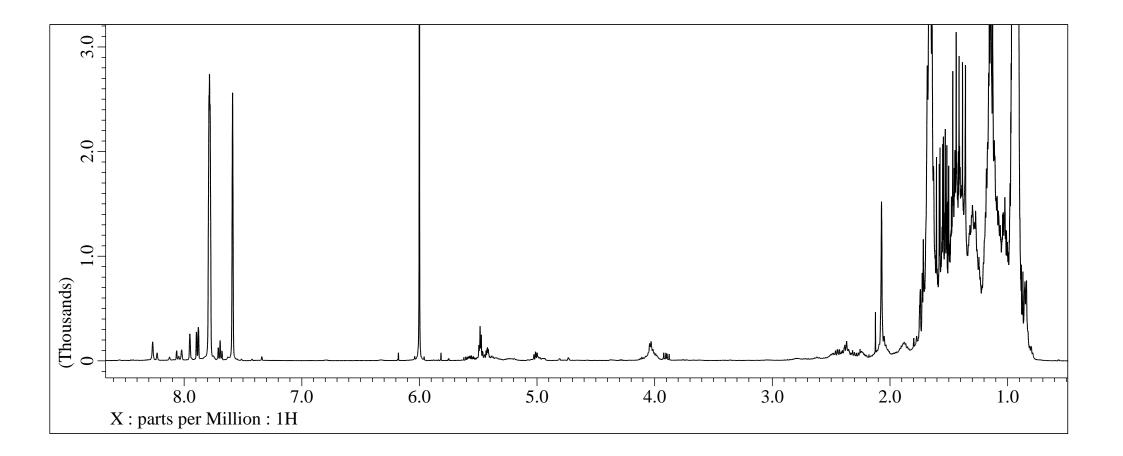


Figure S32. ¹H NMR spectrum of the propylene / allyl acetate copolymer obtained in Table S1, entry 1 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

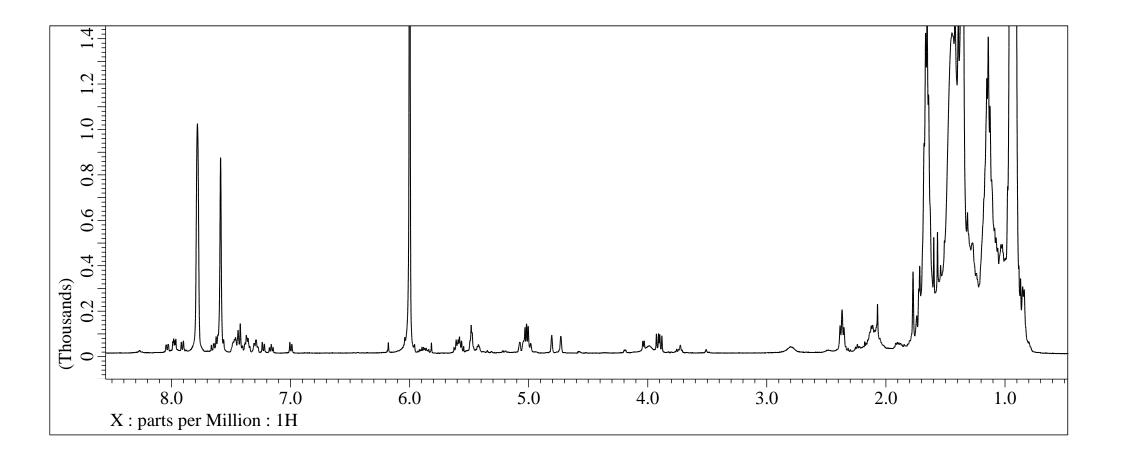


Figure S33. ¹H NMR spectrum of the propylene / allyl acetate copolymer obtained in Table S1, entry 2 (C₂D₂Cl₄, 120 °C, relaxation delay 5 s).

3. SEC Traces

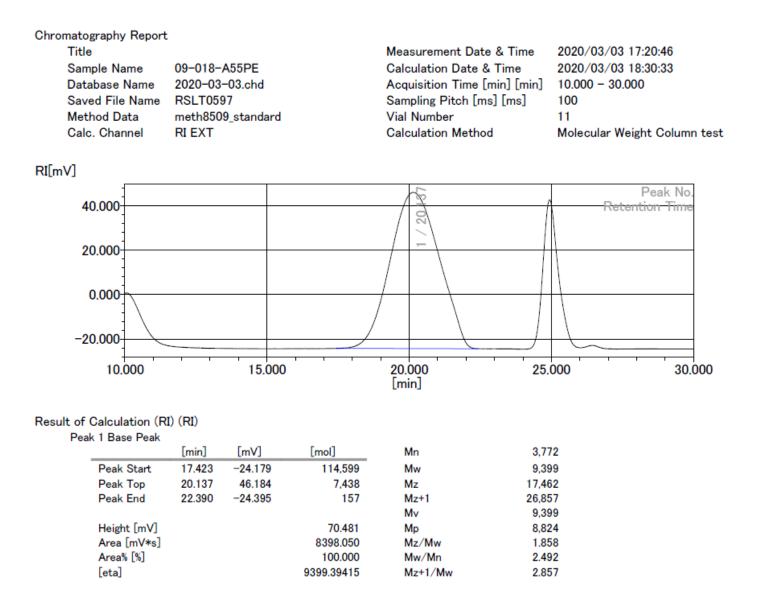
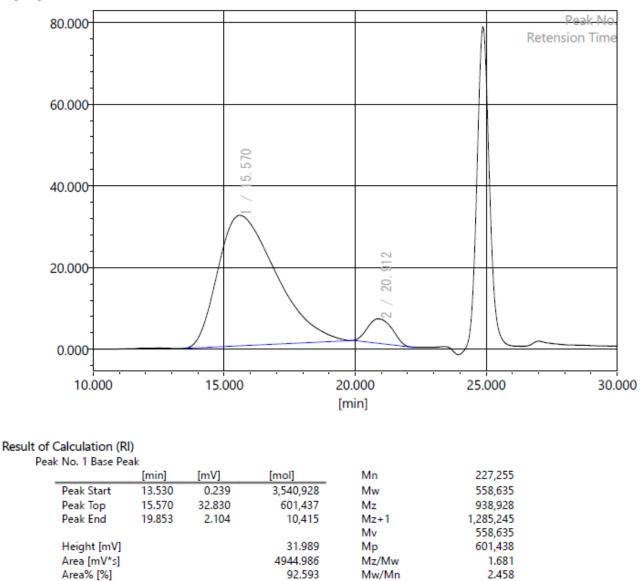


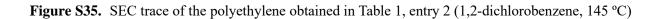
Figure S34. SEC trace of the polyethylene obtained in Table 1, entry 1 (1,2-dichlorobenzene, 145 °C)

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Database Name	2018-10-17.chd	Measurement Time [min]	10.000 - 30.000
Saved File Name	RSLT0332	Sampling Pitch [ms]	100
Method Data	meth8509_standard	Cup Number	4
Calc. Channel	RI EXT	Calculation Method	Molecular Weight Column test



[eta]





Mz+1/Mw

2.301

558635.26300

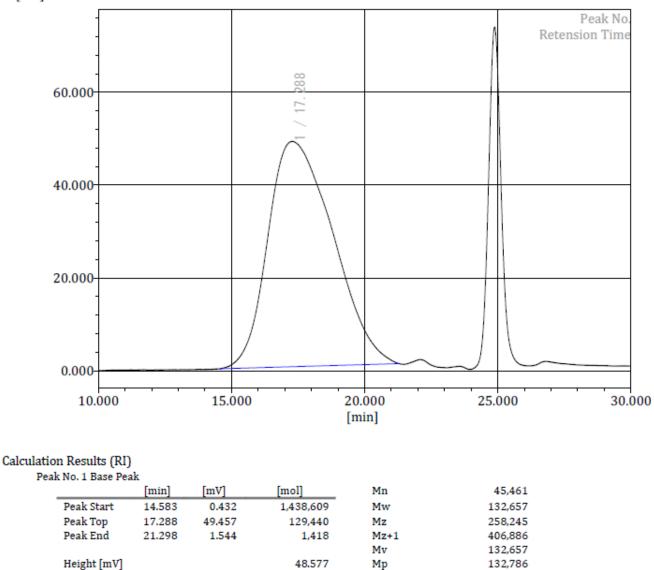
High-temperature GPC Report Title Measurement Date 2018/10/22 18:10:32 133-03-121 2018/10/22 18:26:13 Sample Name Calculation Date 10.000 - 30.000 Database Name 2018-10-22.chd Measurement Time [min] Saved File Name RSLT0334 Sampling Pitch [ms] 100 Cup Number Method Data meth8509_standard 2 Calc. Channel RI EXT Calculation Method Molecular Weight Column test

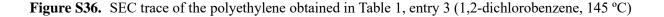


Area [mV*s]

Area% [%]

[eta]





8256.070

100.000

132656.70351

Mz/Mw

Mw/Mn

Mz+1/Mw

1.947

2.918

3.067

Chromatography Report Title Measurement Date & Time 2020/03/10 18:21:07 Sample Name 019-PP50 Calculation Date & Time 2020/03/10 18:56:03 Database Name 2020-03-10.chd Acquisition Time [min] [min] 10.000 - 30.000 Saved File Name RSLT0605 Sampling Pitch [ms] [ms] 100 Method Data meth8509_standard Vial Number 18 Calc. Channel **RI EXT** Calculation Method Molecular Weight Column test

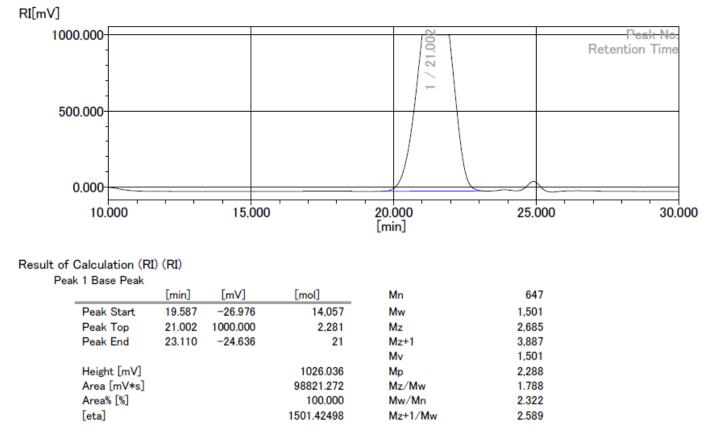
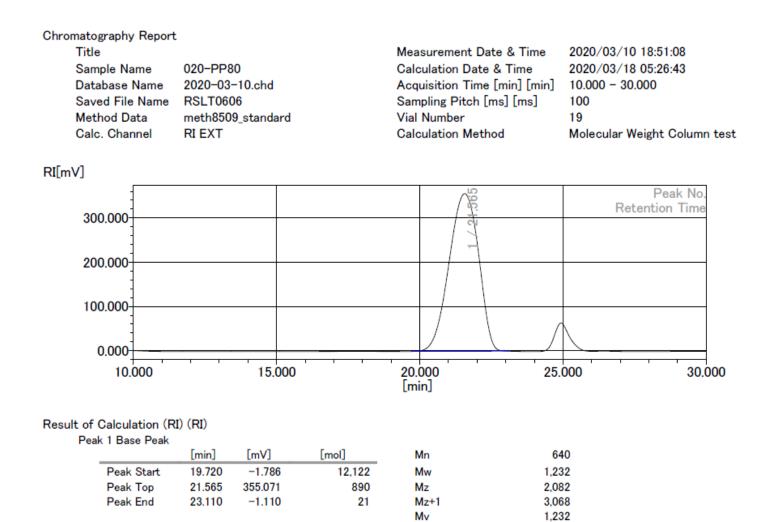


Figure S37. SEC trace of the polypropylene obtained in Table 1, entry 4 (1,2-dichlorobenzene, 145 °C)



356.489

100.000

Figure S38. SEC trace of the polypropylene obtained in Table 1, entry 5 (1,2-dichlorobenzene, 145 °C)

25594.820

1231.68501

Mp

Mz/Mw

Mw/Mn

Mz+1/Mw

1,038

1.690

1.925

2.491

Height [mV]

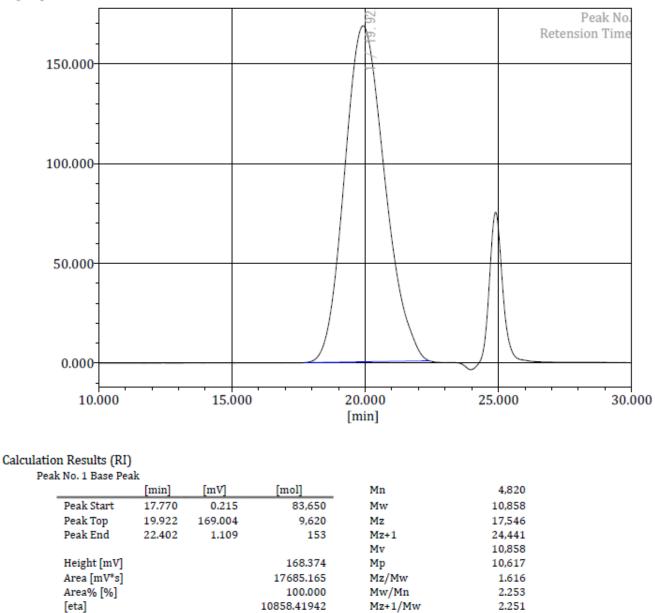
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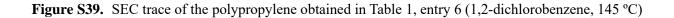
Area% [%]

[eta]

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meth8509_standard	Cup Number	1
RI EXT	Calculation Method	Molecular Weight Column test
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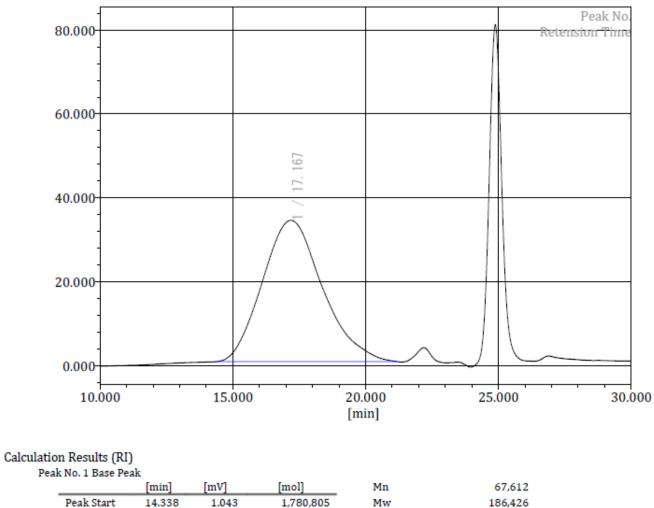




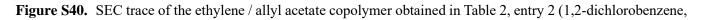


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Method Data	meth8509_standard	Cup Number	1
Calc. Channel	RI EXT	Calculation Method	Molecular Weight Column test





	[min]	[mv]	[mol]	Mn	67,612
Peak Start	14.338	1.043	1,780,805	Mw	186,426
Peak Top	17.167	34.721	144,416	Mz	356,813
Peak End	21.177	1.048	1,733	Mz+1	548,510
				Mv	186,426
Height [mV]			33.676	Мр	144,416
Area [mV*s]			5290.055	Mz/Mw	1.914
Area% [%]			100.000	Mw/Mn	2.757
[eta]			186425.75017	Mz+1/Mw	2.942



High-temperature GPC Report Title Measurement Date 2018/12/07 16:39:00 Sample Name 133-04-010 Calculation Date 2018/12/07 16:48:34 Database Name 2018-12-07.chd Measurement Time [min] 10.000 - 30.000 Saved File Name RSLT0384 Sampling Pitch [ms] 100 Method Data meth8509_standard Cup Number 1 Calc. Channel RI EXT Calculation Method Molecular Weight Column test

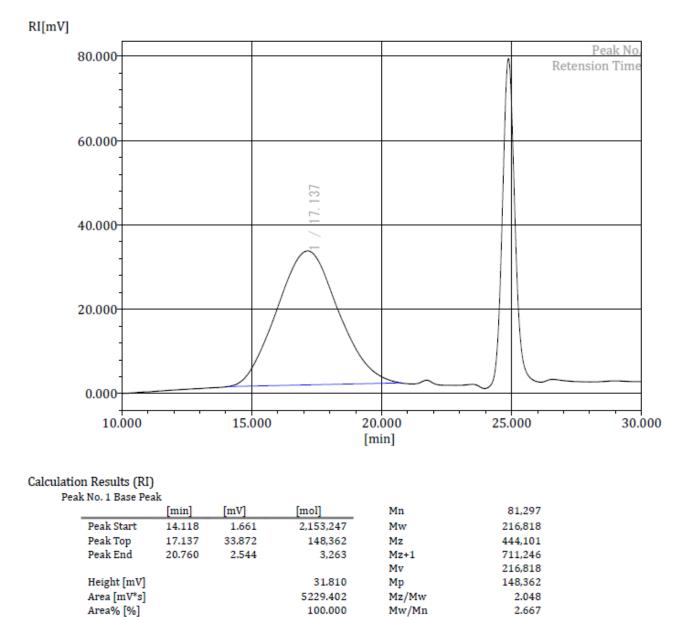


Figure S41. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 2, entry 3 (1,2-dichlorobenzene, 145 °C)

Mz+1/Mw

3.280

216818.11663

[eta]

High-temperature GPC Report

Area% [%]

[eta]

Title	
Sample Name	133-04-011
Database Name	2018-12-07.chd
Saved File Name	RSLT0385
Method Data	meth8509_standard
Calc. Channel	RI EXT

Measurement Date Calculation Date Measurement Time [min] Sampling Pitch [ms] Cup Number Calculation Method 2018/12/07 17:09:01 2018/12/07 17:14:21 10.000 - 30.000 100 2 Molecular Weight Column test



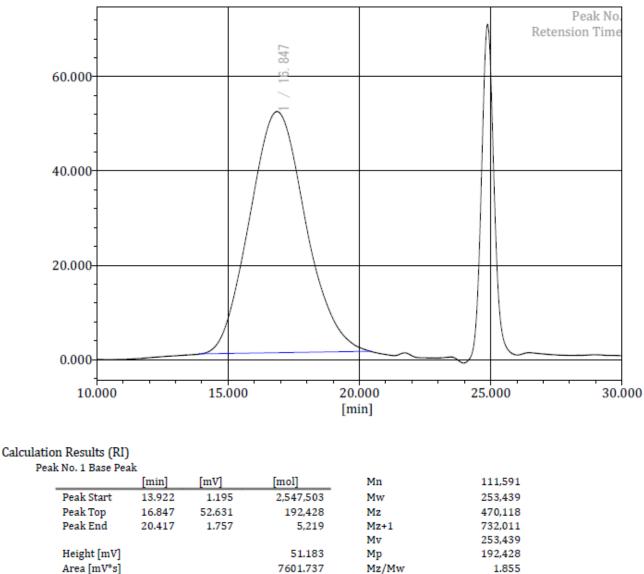


Figure S42. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 2, entry 4 (1,2-dichlorobenzene, 145 °C)

Mw/Mn

Mz+1/Mw

2.271

2.888

100.000

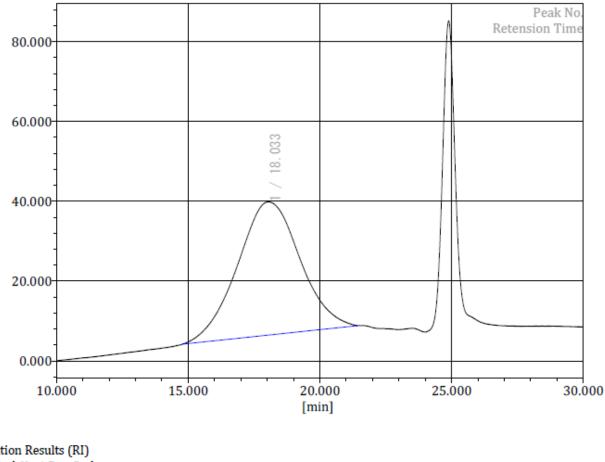
253439.01617

High-temperature GPC Report

4-028
01-11.chd
400
509_standard
[

Measurement Date Calculation Date Measurement Time [min] Sampling Pitch [ms] Cup Number Calculation Method 2019/01/11 16:27:12 2019/01/11 18:07:43 10.000 - 30.000 100 1 Molecular Weight Column test

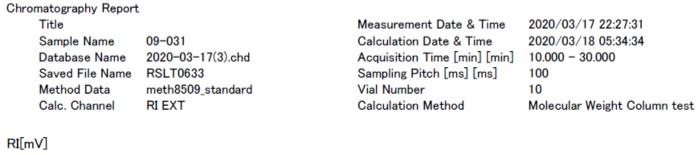




Calculation Results (RI) Peak No. 1 Base Peak

	[min]	[mV]	[mol]	Mn	33,187
Peak Start	14.803	4.213	1,186,113	Mw	101,151
Peak Top	18.033	39.893	65,675	Mz	231,507
Peak End	21.422	8.820	1,149	Mz+1	402,513
				Mv	101,151
Height [mV]			33.432	Мр	70,166
Area [mV*s]			5505.964	Mz/Mw	2.289
Area% [%]			100.000	Mw/Mn	3.048
[eta]			101150.56782	Mz+1/Mw	3.979

Figure S43. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 2, entry 5 (1,2-dichlorobenzene, 145 °C)



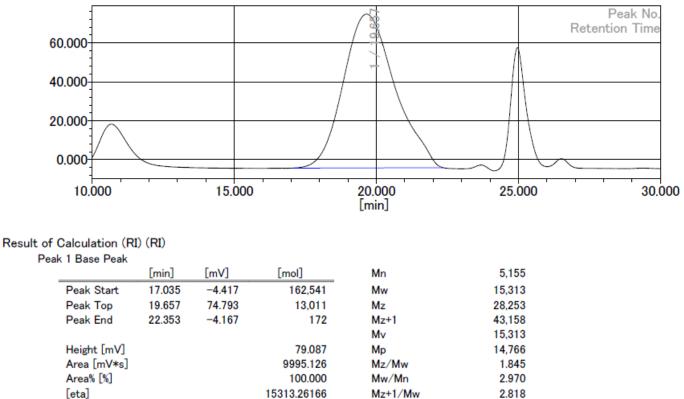


Figure S44. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 2, entry 7 (1,2-dichlorobenzene,

High-temperature GP	C Report		
Title		Measurement Date	2019/02/12 17:30:40
Sample Name	133-04-071	Calculation Date	2019/02/12 18:12:18
Database Name	2019-02-12.chd	Measurement Time [min]	10.000 - 30.000
Saved File Name	RSLT0423	Sampling Pitch [ms]	100
Method Data	meth8509_standard	Cup Number	1
Calc. Channel	RI EXT	Calculation Method	Molecular Weight Column test



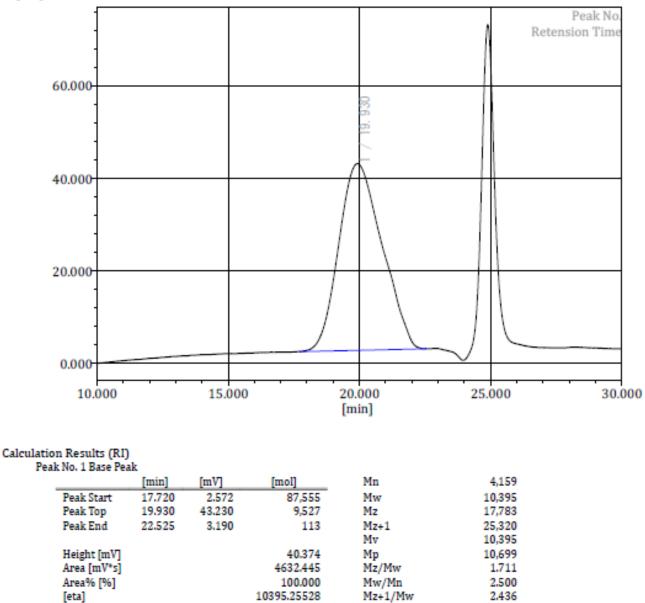


Figure S45. SEC trace of the propylene / allyl acetate copolymer obtained in Table 2, entry 9 (1,2-dichlorobenzene,

High-temperature GPC Report

Title	
Sample Name	133-03-128
Database Name	2018-11-02.chd
Saved File Name	RSLT0365
Method Data	meth8509_standard
Calc. Channel	RI EXT

Measurement Date Calculation Date Measurement Time [min] Sampling Pitch [ms] Cup Number Calculation Method

2018/11/02 17:33:55 2018/11/02 18:10:16 10.000 - 30.000 100 2 Molecular Weight Column test

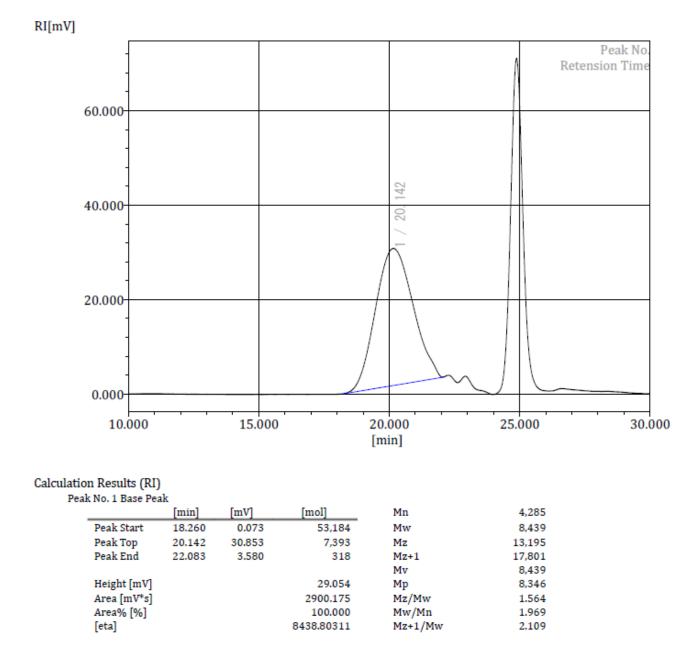


Figure S46. SEC trace of the propylene / 3-butetyl acetate copolymer obtained in Table 2, entry 10 (1,2dichlorobenzene, 145 °C)

Chromatography Report Title Measurement Date & Time 2019/09/06 19:24:39 Sample Name 068-Zn Calculation Date & Time 2019/12/24 15:59:56 Database Name 2019-09-06.chd Acquisition Time [min] [min] 10.000 - 30.000 100 Saved File Name RSLT0496 Sampling Pitch [ms] [ms] Method Data meth8509_standard Vial Number 24 Calc. Channel **RI EXT** Calculation Method Molecular Weight Column test

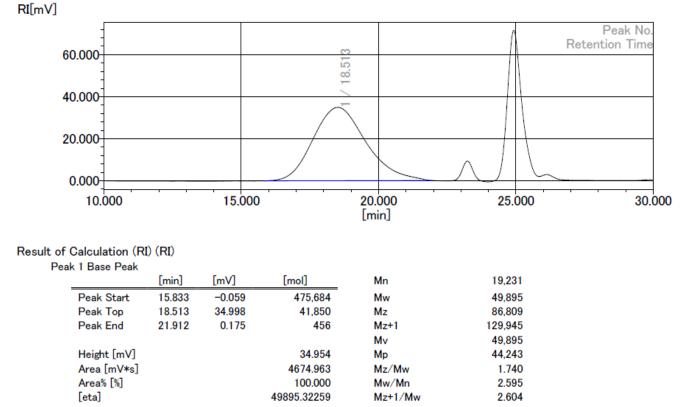


Figure S47. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 2 (1,2-dichlorobenzene,

Chromatography Report Measurement Date & Time 2019/09/06 18:24:37 Title Sample Name 062-BCF Calculation Date & Time 2019/12/24 16:02:08 Database Name 2019-09-06.chd Acquisition Time [min] [min] 10.000 - 30.000 Saved File Name RSLT0494 Sampling Pitch [ms] [ms] 100 Method Data Vial Number 22 meth8509_standard Calc. Channel **RI EXT** Calculation Method Molecular Weight Column test



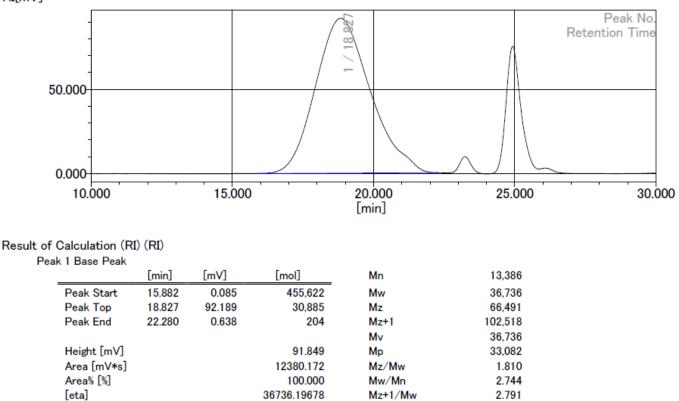
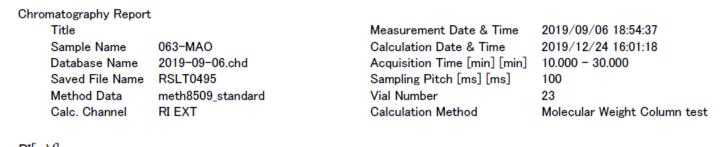


Figure S48. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 3 (1,2-dichlorobenzene,



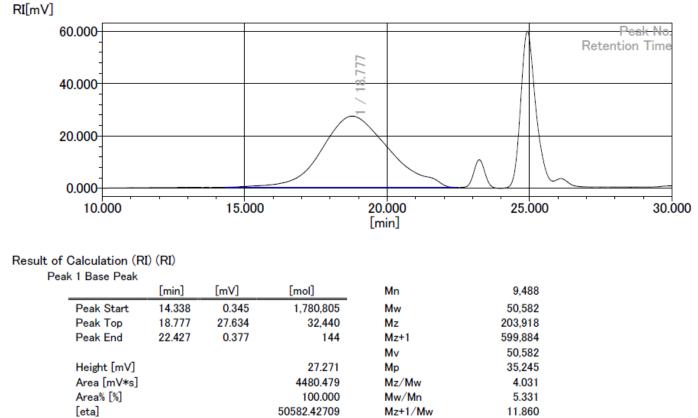


Figure S49. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 4 (1,2-dichlorobenzene,

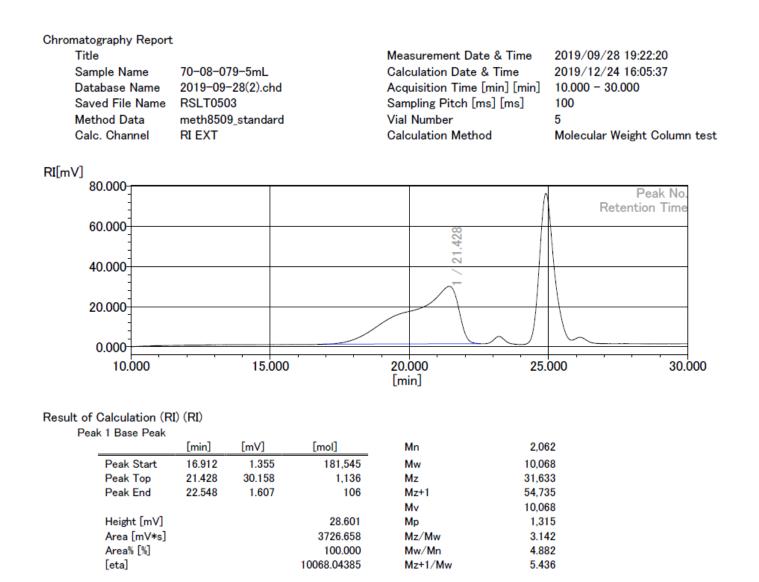


Figure S50. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 5 (1,2-dichlorobenzene,

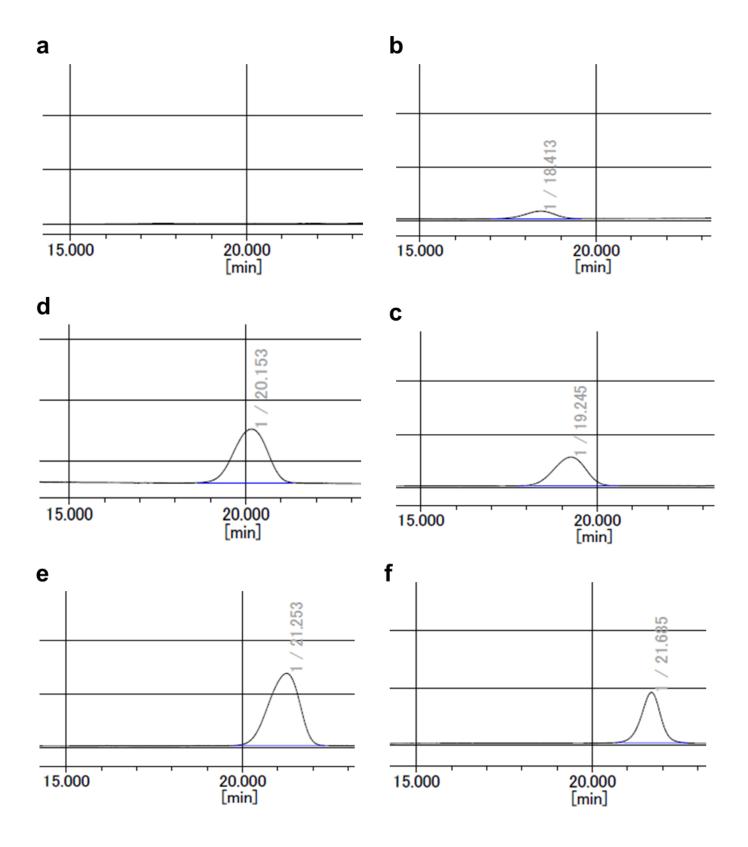
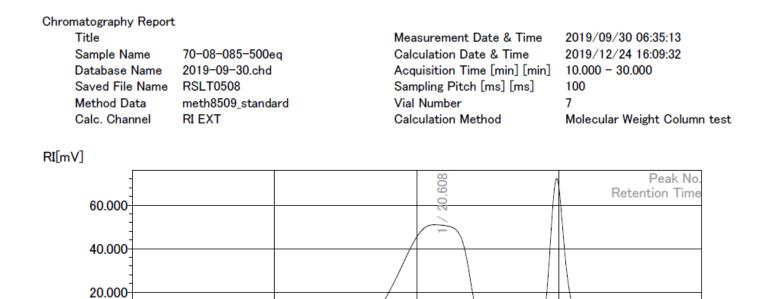


Figure S51. SEC traces of the collected eluents from SEC measurement of polymer from Table 3, entry 5 (1,2-dichlorobenzene, 145 °C). Eluent of a, retention time 17–18 min. b, retention time 18–19 min. c, retention time 19–20 min. d, retention time 20–21 min. e, retention time 21–22 min. and f, retention time 22–23 min.



20.000

[min]

25.000

30.000

Result of Calculation (RI) (RI)

10.000

0.000

Pea	k 1 Base Peak					
		[min]	[mV]	[mol]	Mn	2,062
	Peak Start	17.207	1.305	139,314	Mw	8,528
	Peak Top	20.608	51.047	4,036	Mz	22,870
	Peak End	22.770	2.352	59	Mz+1	39,886
					Mv	8,528
	Height [mV]			49.102	Mp	6,545
	Area [mV*s]			7794.272	Mz/Mw	2.682
	Area% [%]			100.000	Mw/Mn	4.136
	[eta]			8528.06827	Mz+1/Mw	4.677

15.000

Figure S52. SEC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 6 (1,2-dichlorobenzene,

Chromatography Report Title Measurement Date & Time 2019/11/08 23:20:14 Sample Name 70-08-111-EVA Calculation Date & Time 2019/11/09 11:10:46 Database Name 2019-11-08(2).chd Acquisition Time [min] [min] 10.000 - 30.000 Saved File Name RSLT0537 Sampling Pitch [ms] [ms] 100 Method Data meth8509_standard Vial Number 2 Calc. Channel **RI EXT** Calculation Method Molecular Weight Column test RI[mV]

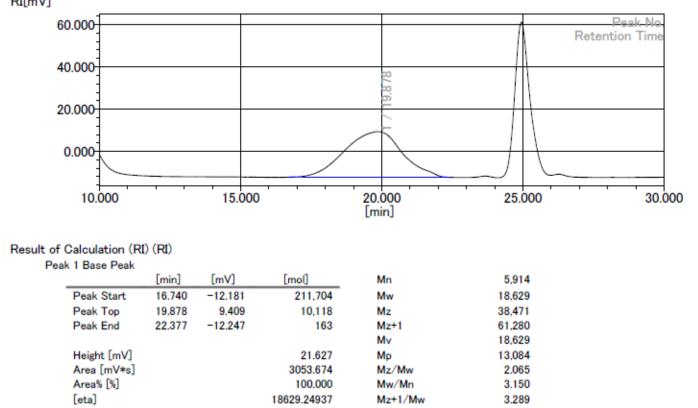
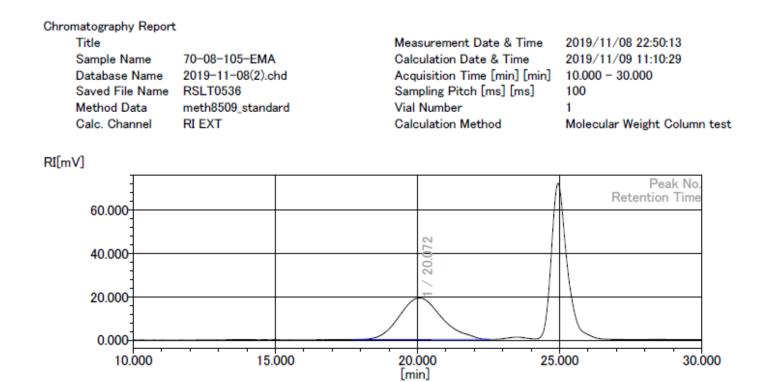


Figure S53. SEC trace of the polyethylene obtained in Table 3, entry 7 (1,2-dichlorobenzene, 145 °C)



Result of Calculation (RI) (RI)

Peak	1 Base Peak					
		[min]	[mV]	[mol]	Mn	4,010
	Peak Start	17.720	0.187	87,555	Mw	9,612
	Peak Top	20.072	19.497	8,050	Mz	16,205
	Peak End	22.500	0.373	120	Mz+1	23,813
					Mv	9,612
	Height [mV]			19.218	Mp	9,215
	Area [mV*s]			1999.713	Mz/Mw	1.686
	Area% [%]			100.000	Mw/Mn	2.397
	[eta]			9611.76027	Mz+1/Mw	2.478

Figure S54. SEC trace of the ethylene / methyl acrylate copolymer obtained in Table 3, entry 8 (1,2dichlorobenzene, 145 °C) Chromatography Report Title Measurement Date & Time 2020/01/14 22:25:47 Sample Name 70-08-133-2 Calculation Date & Time 2020/01/15 09:19:11 2020-01-14.chd 10.000 - 30.000 Database Name Acquisition Time [min] [min] Saved File Name RSLT0563 Sampling Pitch [ms] [ms] 100 Method Data meth8509_standard Vial Number 8 Calc. Channel **RI EXT** Calculation Method Molecular Weight Column test

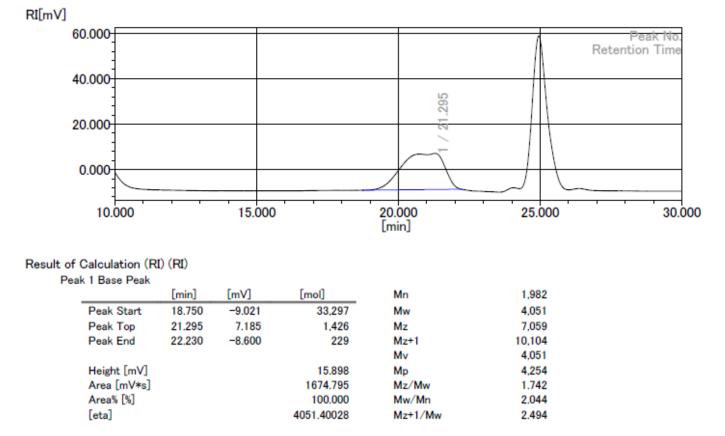
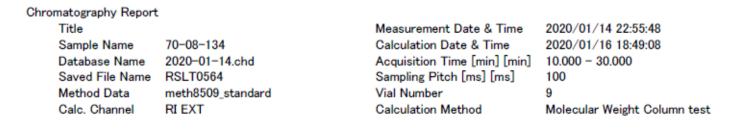


Figure S55. SEC trace of the propylene / allyl acetate copolymer obtained in Table 3, entry 9 (1,2-dichlorobenzene,

145 °C).



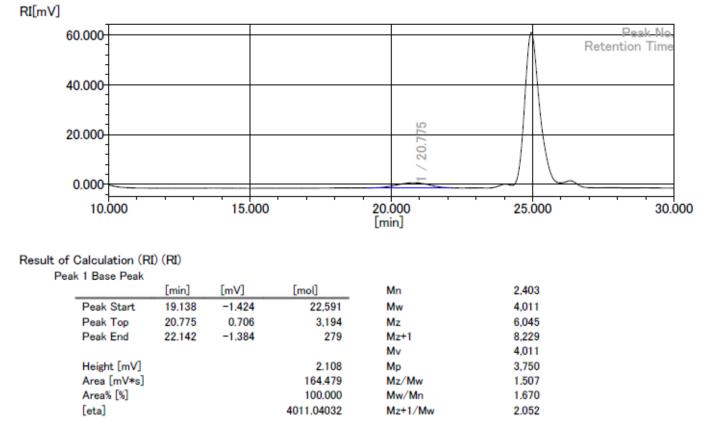


Figure S56. SEC trace of the propylene / allyl acetate copolymer obtained in Table 3, entry 10 (1,2-dichlorobenzene, 145 °C).

Chromatography Report	:		
Title		Measurement Date & Time	2020/03/17 19:52:10
Sample Name	09-032	Calculation Date & Time	2020/03/18 05:32:48
Database Name	2020-03-17(2).chd	Acquisition Time [min] [min]	10.000 - 30.000
Saved File Name	RSLT0631	Sampling Pitch [ms] [ms]	100
Method Data	meth8509_standard	Vial Number	7
Calc. Channel	RIEXT	Calculation Method	Molecular Weight Column test



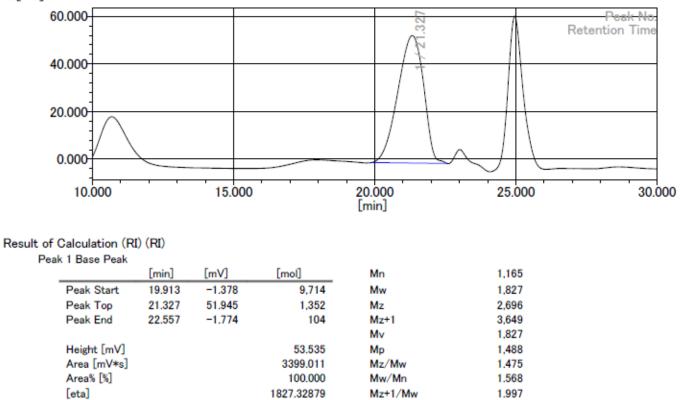
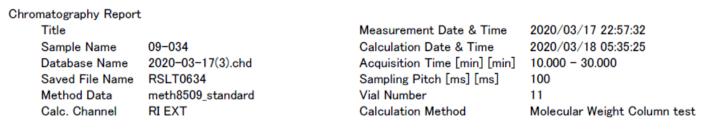


Figure S57. SEC trace of the propylene / allyl acetate copolymer obtained in Table S1, entry 1 (1,2-dichlorobenzene, 145 °C).





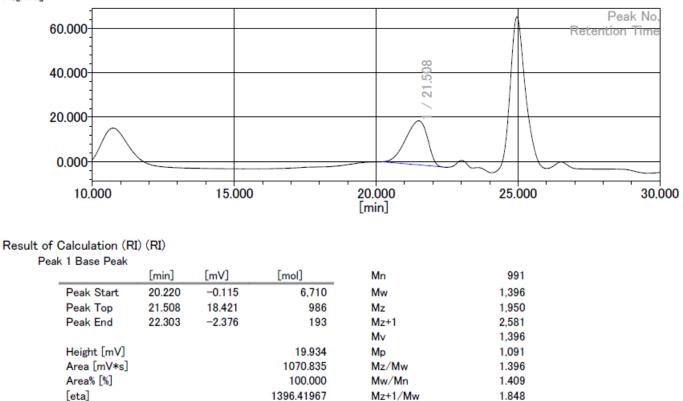


Figure S58. SEC trace of the propylene / allyl acetate copolymer obtained in Table S1, entry 2 (1,2-dichlorobenzene, 145 °C).

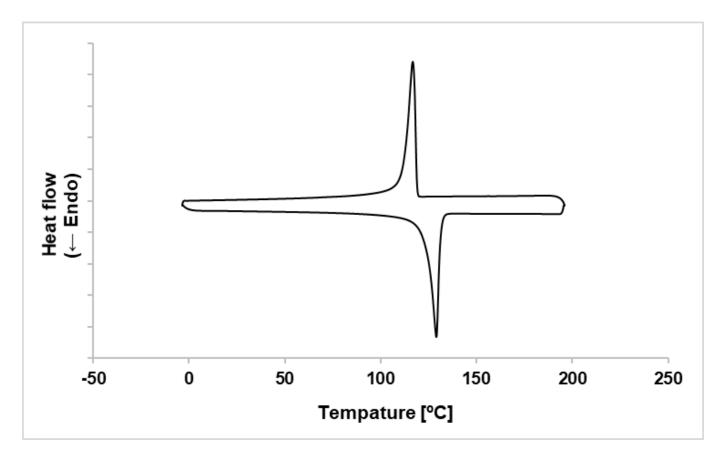


Figure S59. DSC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 2.

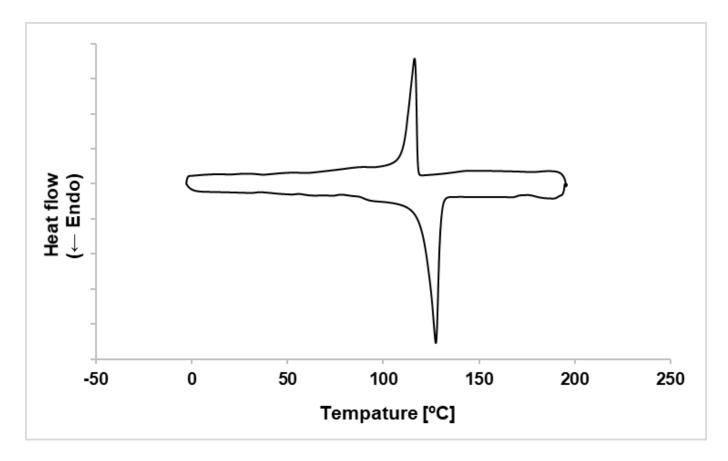


Figure S60. DSC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 3.

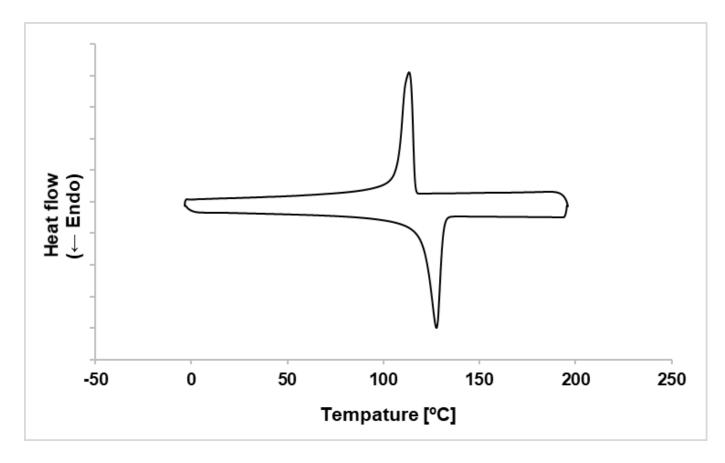


Figure S61. DSC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 4.

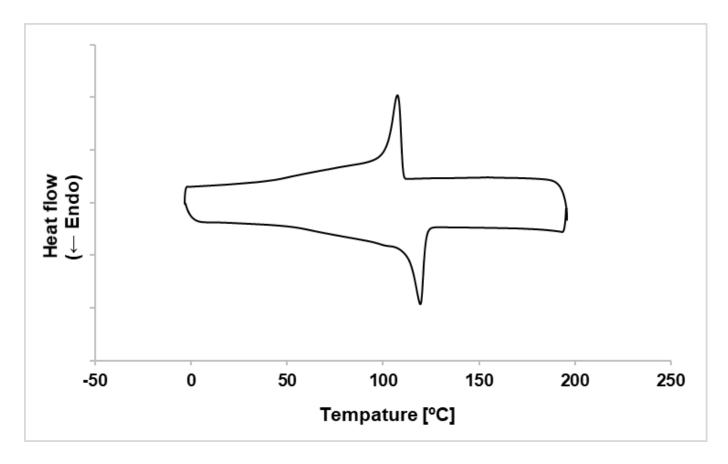


Figure S62. DSC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 5.

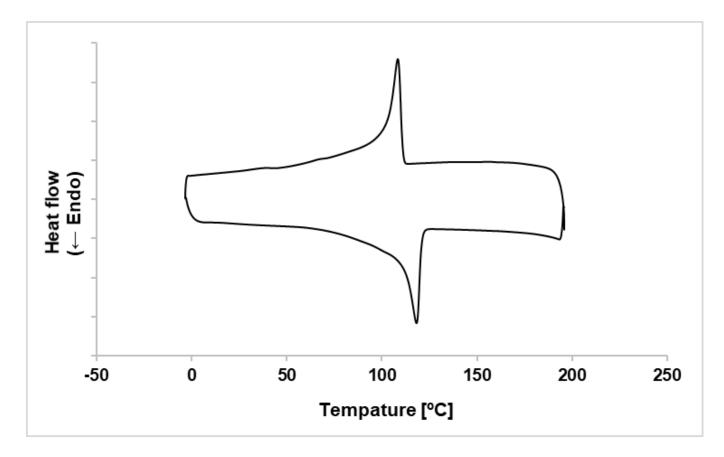


Figure S63. DSC trace of the ethylene / allyl acetate copolymer obtained in Table 3, entry 6.

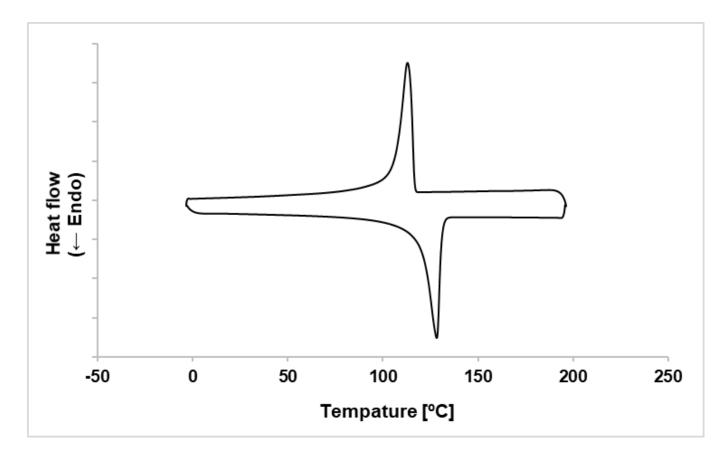


Figure S64. DSC trace of the polyethylene obtained in Table 3, entry 7.

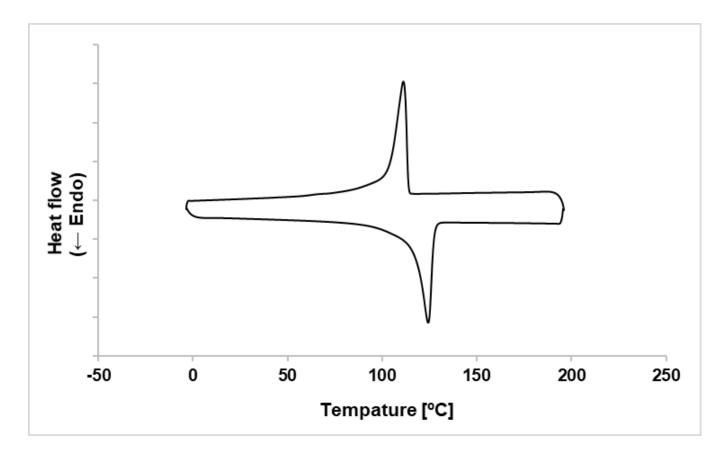


Figure S65. DSC trace of the polyethylene end-capped with ester obtained in Table 3, entry 8.

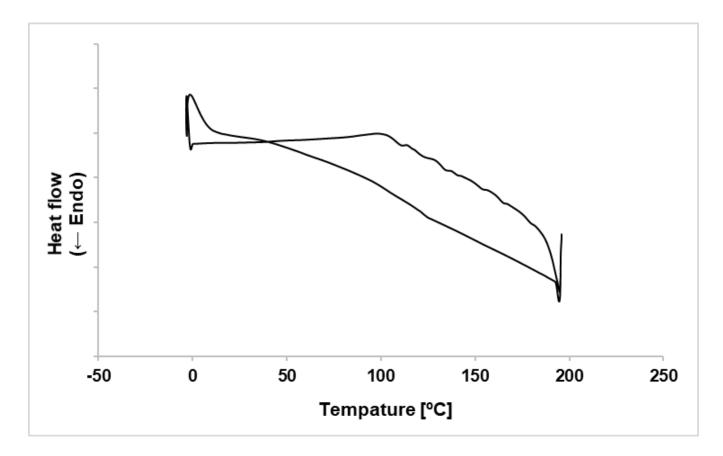


Figure S66. DSC trace of the propylene / allyl acetate copolymer obtained in Table 3, entry 9.

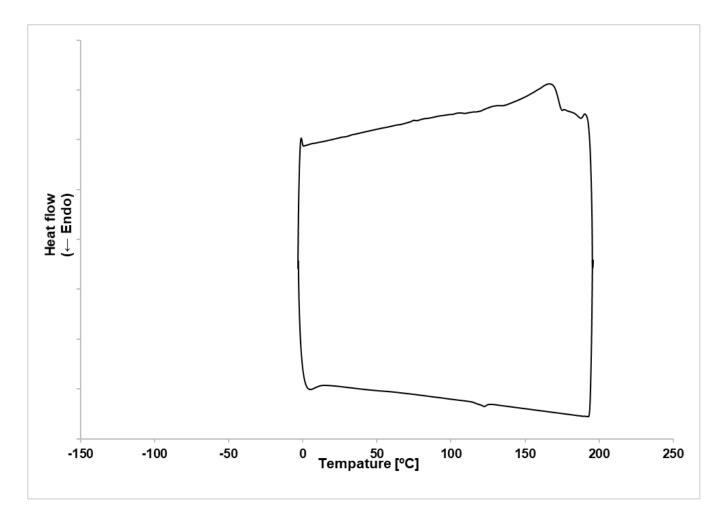


Figure S67. DSC trace of the propylene / allyl acetate copolymer obtained in Table 3, entry 10.

5. X-ray Crystallographic Data of Complex 3

Single crystal of **3** was mounted with mineral oil on a loop-type mount and transferred to the goniometer of a Rigaku Saturn CCD diffractometer. The radiation was performed with graphite-monochromated Mo K α ($\lambda = 0.71075$ Å). The structure was solved by direct methods with (SHELXT 2018)¹⁰ and refined by full-matrix least-squares techniques against F^2 (SHELXL 2018)¹⁰ on the Olex² program.¹¹ The intensities was corrected for Lorentz and polarization effects. The non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed using AFIX instructions. The data are summarized in Table S1.

Empirical formula		C ₅₂ H ₅₅ BF ₂₄ NiOP ₂
Formula weight		1283.42
Temperature (K)		293
Crystal system		triclinic
Space group		P-1
Unit cell dimensions	a (Å)	12.36430(10)
	b (Å)	21.2593(2)
	c (Å)	25.0961(3)
	α (°)	65.7360(10)
	β (°)	76.1750(10)
	γ (°)	76.5800(10)
Volume (Å ³)		5772.53(11)
Z		2
Density (calculated) (g/cm ³)		1.528
Absorption coefficient (mm ⁻¹)		0.50
F(000)		2714.0
Crystal size (mm ³)		0.50 imes 0.25 imes 0.40
Theta range		2.2960–28.9450
Index ranges		-16<=h<=16
		-28<=k<=28
		-34<=1<=34
Reflections collected		202560
Independent reflections		28130
R(int)		0.0299
Data / restraints / parameters		28130 / 0 / 1484
Goodness-of-fit on F^2		1.032
Final <i>R</i> indices $[I > 2\sigma(I)]$		$R_1 = 0.0953$
		$wR_2 = 0.2809$
R indices (all data) $[I > 2\sigma(I)]$		$R_1 = 0.1054$
		$wR_2 = 0.2919$

Table S1. Crystal Data and Structure Refinement for Complex 3.

5. References

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