

Supporting Information

A force field for poly(oxymethylene) dimethyl ethers (OME_n)

Aditya M. Kulkarni, Edder J. García, Angelo Damone, Michael Schappals, Simon Stephan, Maximilian Kohns* and Hans Hasse

Laboratory of Engineering Thermodynamics (LTD), Technische Universität Kaiserslautern, Erwin-Schrödinger-Str. 44, 67663 Kaiserslautern, Germany.

Email: maximilian.kohns@mv.uni-kl.de

1. Dihedral potential with Fourier equation

Some simulation codes use the Fourier equation instead of the Ryckaert-Bellemans (RB) equation discussed in the main text. We thus provide the equivalents of our dihedral parameters if the Fourier equation is used instead. The dihedral potential using the Fourier equation with six coefficients a_i is 1:

$$u_t = a_0 + \sum_{i=1}^5 a_i [1 + (-1)^{i+1} \cos(i\theta)] \quad (1)$$

The a_i coefficients for OME for the Fourier equation are given in Table S1. The dihedral angle in the RB equation φ is related to the angle θ in the Fourier equation by 2:

$$\varphi = \theta - 180^\circ \quad (2)$$

Table S1: Parameters for the dihedral potential using the Fourier equation.

a_i / K	CH ₃ -O-CH ₂ -O (edge)	CH ₂ -O-CH ₂ -O (center)
a_0	1037.95	1112.16
a_1	225.21	525.59
a_2	-866.08	-1060.64
a_3	917.08	725.12
a_4	-81.66	-146.37
a_5	53.16	51.24

2. Numerical values of liquid densities at 1 bar from *NPT* simulations

The liquid densities at 1 bar for OME_n ($n = 1 - 4$), calculated using *NPT* simulations with MedeA-Gibbs [1] are given in Table S2.

Table S2: Liquid densities at 1 bar for OME n ($n = 1 - 4$).

T / K	$\rho^{\text{liq}} / \text{kg m}^{-3}$
OME1	
293.05	850.69±1.15
333.16	797.92±2.39
363.15	755.73±1.38
OME2	
293.15	966.02±2.30
333.15	920.34±2.81
363.15	881.52±2.53
OME3	
293.15	1017.5±3.17
333.15	975.24±1.67
363.15	942.35±2.74
OME4	
293.15	1053.0±7.32
333.15	1018.9±4.00
363.15	993.46±3.11

3. Numerical values of VLE properties from GEMC simulations

VLE properties of OME1-OME4 modelled with the present force field, calculated using Gibbs Ensemble Monte Carlo simulations with MedeA-Gibbs [1] are given in Table S3.

Table S3: VLE properties for OME1-OME4.

T / K	$p^{\text{s}} / \text{bar}$	$\rho^{\text{liq}} / \text{kg m}^{-3}$	$\rho^{\text{vap}} / \text{kg m}^{-3}$
OME1			
336.455	2.179±0.073	794.76±0.968	6.276±0.219
360.490	4.071±0.110	759.59±1.402	11.300±0.376
384.520	7.042±0.107	723.34±0.641	19.129±0.312
408.553	11.563±0.104	683.97±0.532	31.605±0.466
432.585	17.979±0.238	638.52±1.523	50.690±0.877
442.200	21.253±0.241	618.69±1.482	61.127±0.864
OME2			
386.540	1.237±0.032	850.37±1.267	3.895±0.046
414.150	2.549±0.051	814.25±1.295	8.411±0.209
441.760	5.057±0.152	773.59±1.072	16.556±0.680
469.370	8.630±0.193	727.34±2.425	28.112±0.891
496.980	14.414±0.226	678.29±1.694	45.799±1.110
508.024	17.120±0.278	652.33±2.041	55.977±1.718
OME3			
452.550	2.029±0.061	831.53±0.853	7.908±0.294
482.720	3.819±0.148	789.59±2.034	14.498±0.862
512.890	6.630±0.128	740.11±1.315	24.730±0.413
543.060	11.620±0.214	690.99±2.920	45.799±1.110
OME4			
485.175	1.005±0.026	858.07±1.677	4.244±0.116
517.520	2.252±0.063	821.65±2.256	9.470±0.326
549.865	4.480±0.138	769.15±1.756	18.816±0.802
582.210	7.809±0.206	715.17±0.970	33.648±1.442

4. Comparison between two independent MC codes

In addition to the GEMC results computed with MedeA-Gibbs [1], which are discussed in the main text, for several state-points we also calculated the VLE properties of OME1-OME4 with Cassandra [2]. Excellent agreement was observed between the two simulation codes, as illustrated in Figure S1.

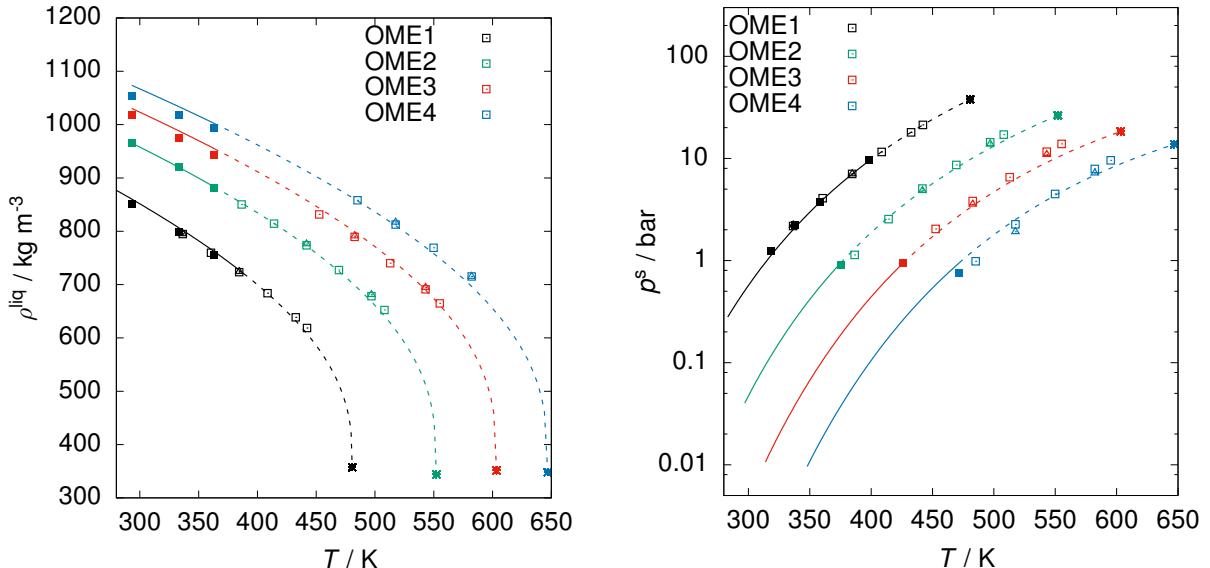


Figure S1: Liquid densities (left) and saturated vapour pressures (right) for OME1-OME4 as a function of temperature. Full lines are correlations of experimental data and dashed lines are extrapolations of these correlations. Filled squares indicate the molecular simulation results obtained with MedeA-Gibbs, at temperatures that were considered for parametrisation. Open squares indicate the molecular simulation results obtained with MedeA-Gibbs, at temperatures that were not considered for parametrisation. Triangles indicate the molecular simulation results obtained with Cassandra. Error bars are within symbol size.

5. Numerical results for solubility from BPE-MC simulations

The numerical results for the pressures in vapour phase for CO₂-OMEn mixtures at 353.15 K, obtained with BPE-MC simulations with MedeA-Gibbs [1] are given in Table S4

Table S4: Solubilities of CO₂ in OME_n for $n = 2 - 4$

x_{CO_2}	p/bar
OME2	
0.064	5.6709±0.4166
0.150	13.083±0.2655
0.280	26.921±0.2492
OME3	
0.030	2.5757±0.0321
0.078	7.2016±0.0530
0.122	10.152±0.3517
0.264	25.559±0.8138
OME4	
0.034	2.7402±0.1413
0.088	7.9732±0.3603
0.192	15.700±0.6535
0.294	27.961±0.8630

6. Shear viscosity

The numerical values of the shear viscosity and density for OME1-OME4 modelled with the present force field are provided in Table S5.

Table S5: Shear viscosity for OME1-OME4.

T / K	ρ^{liq} / kg m ⁻³	η / mPa s
OME1		
293.15	851.181±0.025	0.2954±0.0151
303.15	838.642±0.029	0.2564±0.0136
313.15	825.579±0.024	0.2265±0.0115
323.15	812.746±0.028	0.2026±0.0103
OME2		
298.15	960.930±0.033	0.5914±0.0333
303.15	955.111±0.024	0.5585±0.0273
308.15	949.402±0.038	0.5288±0.0253
313.15	943.536±0.042	0.4959±0.0255
318.15	937.635±0.060	0.4646±0.0215
323.15	931.690±0.038	0.4350±0.0193
OME3		
298.15	1014.860±0.040	0.9125±0.0336
303.15	1009.340±0.031	0.8547±0.0306
308.15	1003.880±0.028	0.8011±0.0270
313.15	998.372±0.027	0.7515±0.0260
318.15	992.858±0.021	0.7103±0.0245
323.15	987.366±0.038	0.6671±0.0197
OME4		
298.15	1067.770±0.036	1.7227±0.0497
303.15	1062.650±0.041	1.5934±0.0460
308.15	1057.520±0.052	1.4124±0.0435
313.15	1052.280±0.045	1.3385±0.0415
318.15	1047.110±0.045	1.2445±0.0394
323.15	1041.990±0.038	1.1229±0.0376

7. Quantum-Mechanical Calculations

7.1. Bond Lengths and Partial Charges

This section explains the procedure to calculate the bond lengths and the partial charges. The procedure is divided into three subsequent steps: As the first step, a DFT calculation with the 6-311G** Basis set was carried out for obtaining the polarisability α and the permanent dipolar moment μ_0 for each molecule. As

the second step, the Clausius-Mossotti-Debye equation 3

$$\frac{4}{3}\pi \frac{N_A}{V_m} \left(\alpha + \frac{\mu_0^2}{3k_B T} \right) = \frac{\epsilon_r - 1}{\epsilon_r + 2} \quad (3)$$

was applied in order to calculate the dielectric constant ϵ_r using the reference data [5] for temperature T and molar volume V_m given in Table S6.

Table S6: DFT Results of the first calculation.

Molecule	T / K	V_m / cm^3	$\alpha / \text{\AA}^3$	μ_0 / D	ϵ_r
OME1	293.15	88.56	7.29	0.0632	2.650
OME2	298.15	110.55	9.79	0.5078	4.471
OME3	298.15	132.98	12.30	0.0053	1.913
OME4	298.15	155.75	14.85	0.4914	3.407
OME5	298.15	178.32	17.42	0.1214	2.039

As the third step, the dielectric constants were used in a DFT calculation combined with the COSMO [6] solvation model to calculate the partial charges, the bond lengths and the angles. The RESP Method [7] uses 4 layers of spherical isosurfaces with a scaling factor of 1.4 VDW radii for the first shell. The step for each layer is 0.2 VDW radii with a distance of 0.8 Å for each point in the single layer.

In the next subsection of the supporting information the details about the Quantum-Chemical calculations for OMEs, i.e. bond lengths, angles and point charges for the united-atom model are shown. For each OME n (with $n = 1$ to 5), the atom coordinates corresponding to the lowest energy configuration are provided including the center of mass and the Z-Matrix.

7.2. Angle Conformational Analysis

In this section, the results about the angle conformational analysis are provided for each angle type. The Quantum-chemical calculations were carried out using Hartree-Fock method with 6-311G** basis set. The angle bending is modelled using a harmonic potential:

$$u_b = k_b(\theta - \theta_0)^2/2 \quad (4)$$

where u_b is the bending energy, k_b is the bending constant, and θ_0 and θ are the equilibrium and the bending angle, respectively.

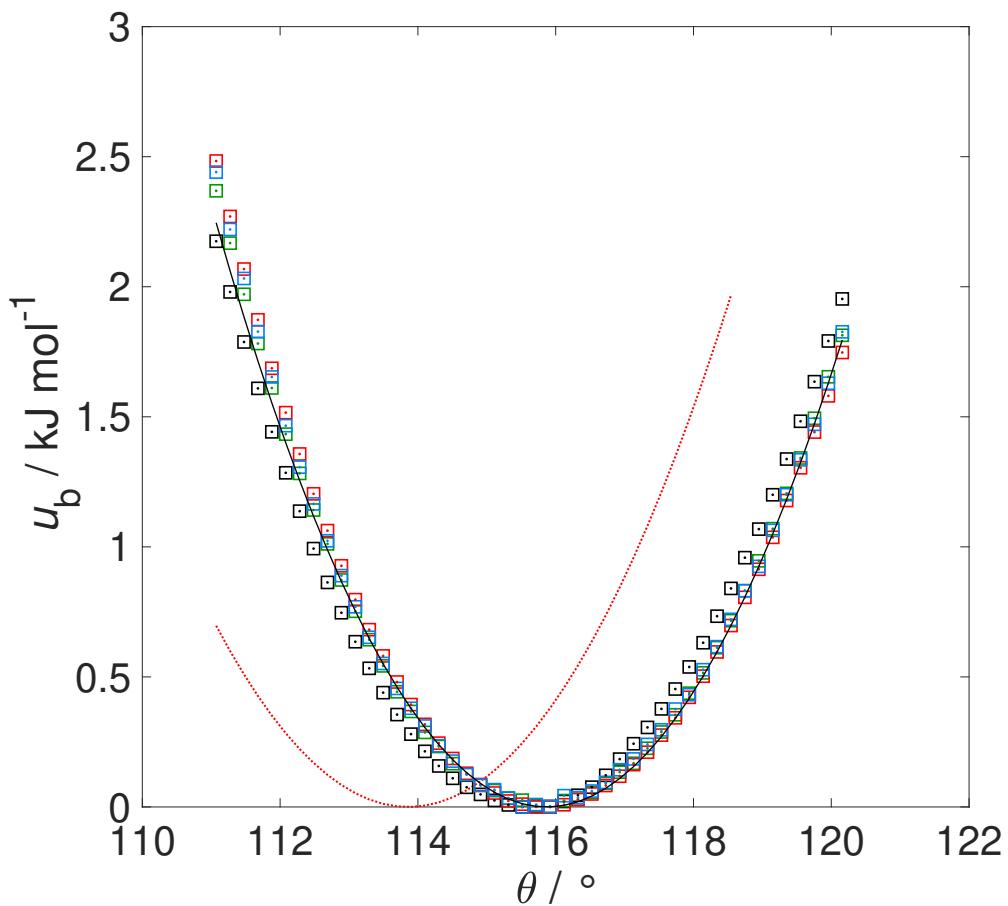


Figure S2: Angle conformational analysis obtained from QM calculations for the $\text{CH}_3\text{-O-CH}_2$ angle type. OME1 (black), OME2 (green), OME3 (red) and OME4 (blue). The solid black line (-) represents the fitted harmonic function (eq. 4) for all the OMEs QM calculation and the dotted red line (---) is the harmonic potential function from Dasgupta et al. [8].

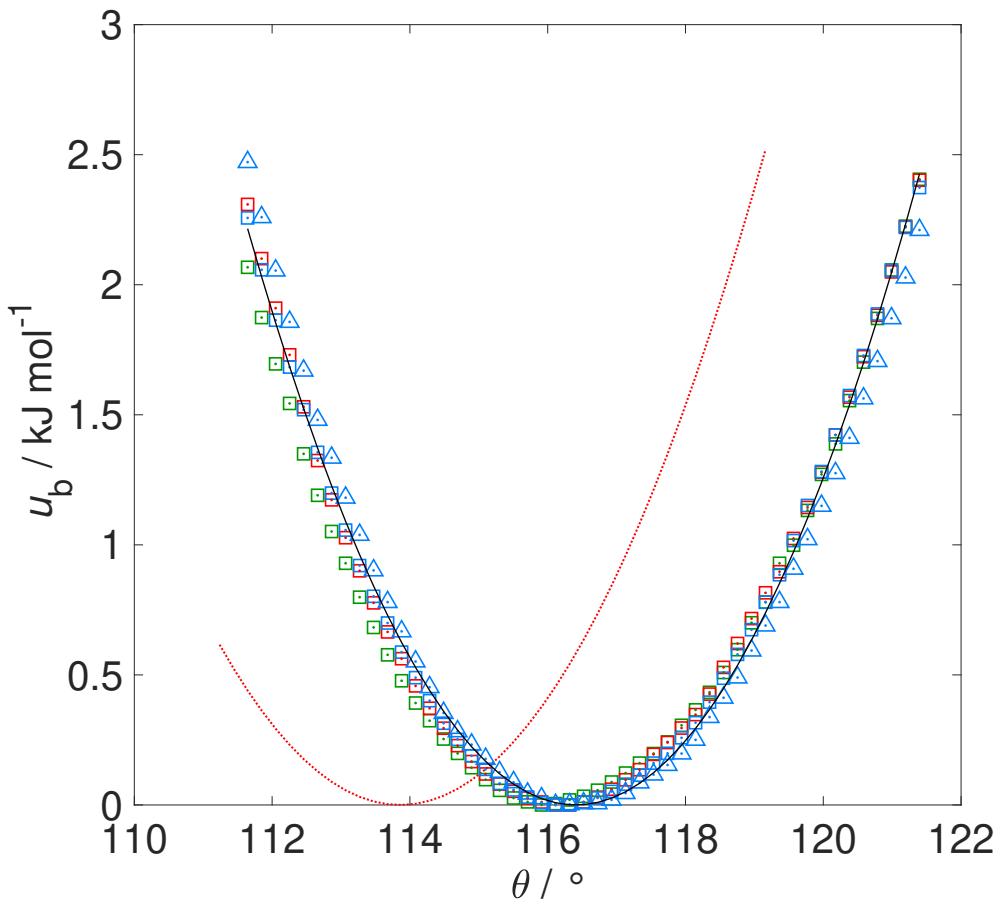


Figure S3: Angle conformational analysis obtained from QM calculations for the $\text{CH}_2\text{-O-CH}_2$ angle type. Squares indicate edge angles and triangles indicate central angles. OME1 (black), OME2 (green), OME3 (red) and OME4 (blue). The solid black line (-) represents the fitted harmonic function (eq. 4) for all the OMEs QM calculation and the dotted red line (---) is the harmonic potential function from Dasgupta et al. [8].

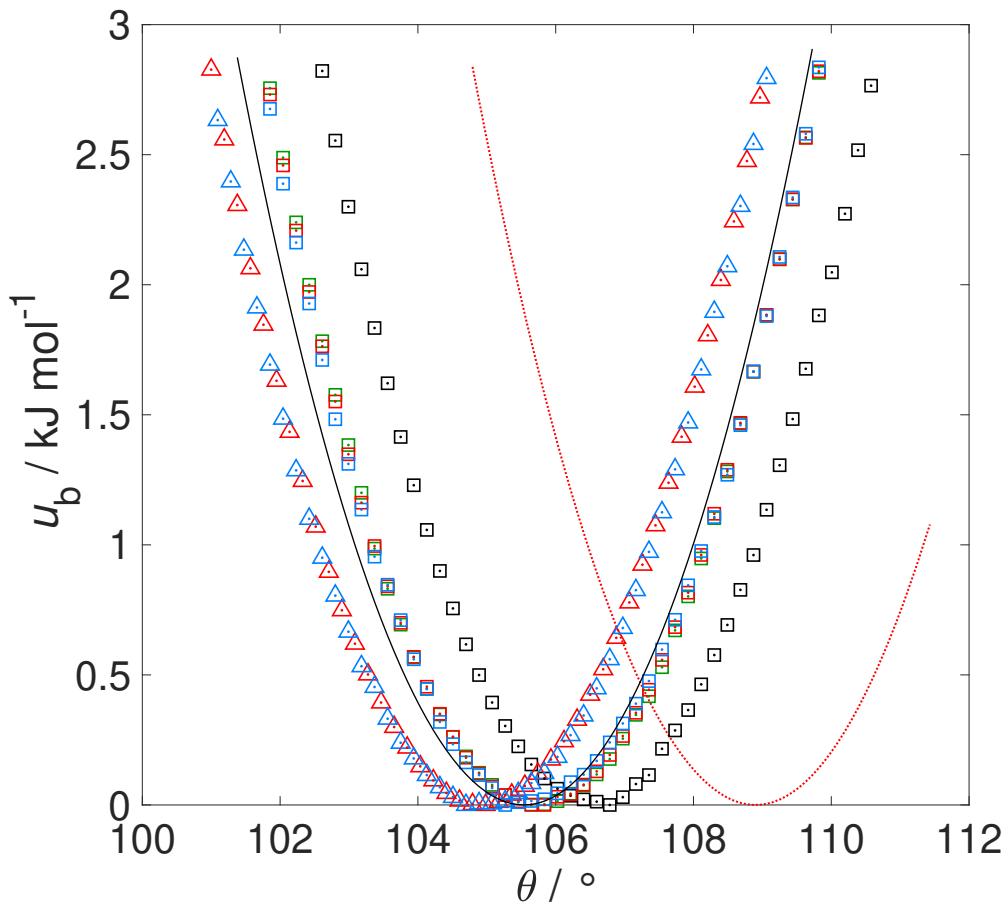


Figure S4: Angle conformational analysis obtained from QM calculations for the O-CH₂-O angle type. Squares indicate edge angles and triangles indicate central angles. OME1 (black), OME2 (green), OME3 (red) and OME4 (blue). The solid black line (-) represents the fitted harmonic function (eq. 4) for all the OMEs QM calculation and the dotted red line (---) is the harmonic potential function from Dasgupta et al. [8].

7.3. OME1

In this section the full atom and center of mass coordinates for OME1 are shown in table S7. The table S8 represents the Z-Matrix for equilibrium geometry of united-atom model, and related partial charges are shown in table S9.

Table S7: Full atom coordinates and center of mass (CM) for equilibrium geometry of OME1.

Atom	$x/\text{\AA}$	$y/\text{\AA}$	$z/\text{\AA}$	Group	$x_{\text{CM}}/\text{\AA}$	$y_{\text{CM}}/\text{\AA}$	$z_{\text{CM}}/\text{\AA}$
C	1.856	0.553	0.308	CH ₃	1.903	0.588	0.354
H	2.312	-0.161	1.004				
H	1.311	1.309	0.878				
H	2.638	1.034	-0.277				
O	1.004	-0.122	-0.616	O	1.004	-0.122	-0.616
C	0.000	-0.890	-0.003	CH ₂	0.000	-0.980	-0.004
H	0.423	-1.515	0.790				
H	-0.422	-1.510	-0.801				
O	-1.006	-0.128	0.615	O	-1.006	-0.128	0.615
C	-1.854	0.557	-0.304	CH ₃	-1.901	0.592	-0.350
H	-1.305	1.313	-0.870				
H	-2.315	-0.150	-1.004				
H	-2.633	1.039	0.285				

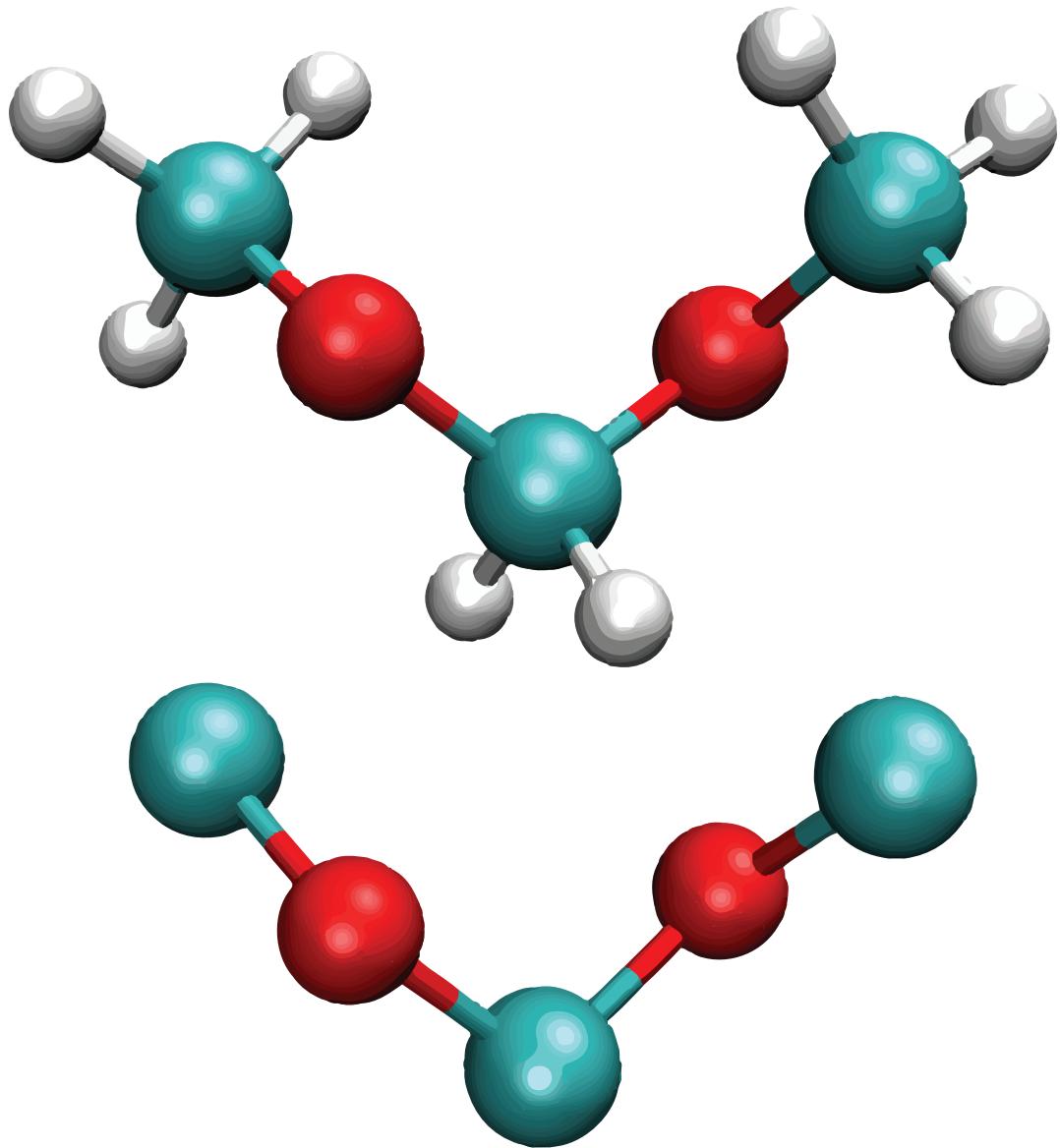


Figure S5: Equilibrium geometry of OME1. Top: full atom representation, bottom: united-atom model.

Table S8: Z-Matrix for equilibrium geometry of OME1 using united-atom model.

Group	ID	Bonds / Å	ID	Angles / °	ID	Dihedrals / °
CH ₃						
O	1	1.500				
CH ₂	2	1.456	1	114.811		
O	3	1.456	2	108.100	1	71.900
CH ₃	4	1.500	3	114.804	2	71.600

Table S9: Point Charges for OME1.

Atom	Charge	Group	Total Charge
C	0.0265	CH ₃	0.2195
H	0.0491		
H	0.0772		
H	0.0667		
O	-0.3455	O	-0.3455
C	-0.0342	CH ₂	0.2539
H	0.1436		
H	0.1445		
O	-0.3469	O	-0.3469
C	0.0312	CH ₃	0.2189
H	0.0742		
H	0.0489		
H	0.0646		

7.4. OME2

In this section the full atom and center of mass coordinates for OME2 are shown in table S10. The table S11 represents the Z-Matrix for equilibrium geometry of united-atom model, and related partial charges are shown in table S12.

Table S10: Full atom coordinates and center of mass (CM) for equilibrium geometry of OME2.

Atom	$x/\text{\AA}$	$y/\text{\AA}$	$z/\text{\AA}$	Group	$x_{\text{CM}}/\text{\AA}$	$y_{\text{CM}}/\text{\AA}$	$z_{\text{CM}}/\text{\AA}$
C	-2.870	-0.359	0.267	CH ₃	-2.922	-0.386	0.312
H	-3.273	0.431	0.912				
H	-2.437	-1.146	0.889				
H	-3.672	-0.776	-0.340				
O	-1.903	0.177	-0.638	O	-1.903	0.177	-0.638
C	-0.848	0.851	-0.007	CH ₂	-0.835	0.939	-0.012
H	-1.227	1.580	0.715				
H	-0.289	1.347	-0.802				
O	-0.003	-0.008	0.742	O	-0.003	-0.008	-0.742
C	0.838	-0.835	-0.047	CH ₂	0.824	-0.922	-0.056
H	0.275	-1.288	-0.864				
H	1.213	-1.601	0.638				
O	1.897	-0.137	-0.644	O	1.897	-0.137	-0.644
C	2.888	0.314	0.281	CH ₃	2.941	0.337	0.327
H	2.483	1.061	0.968				
H	3.688	0.759	-0.307				
H	3.286	-0.528	0.859				

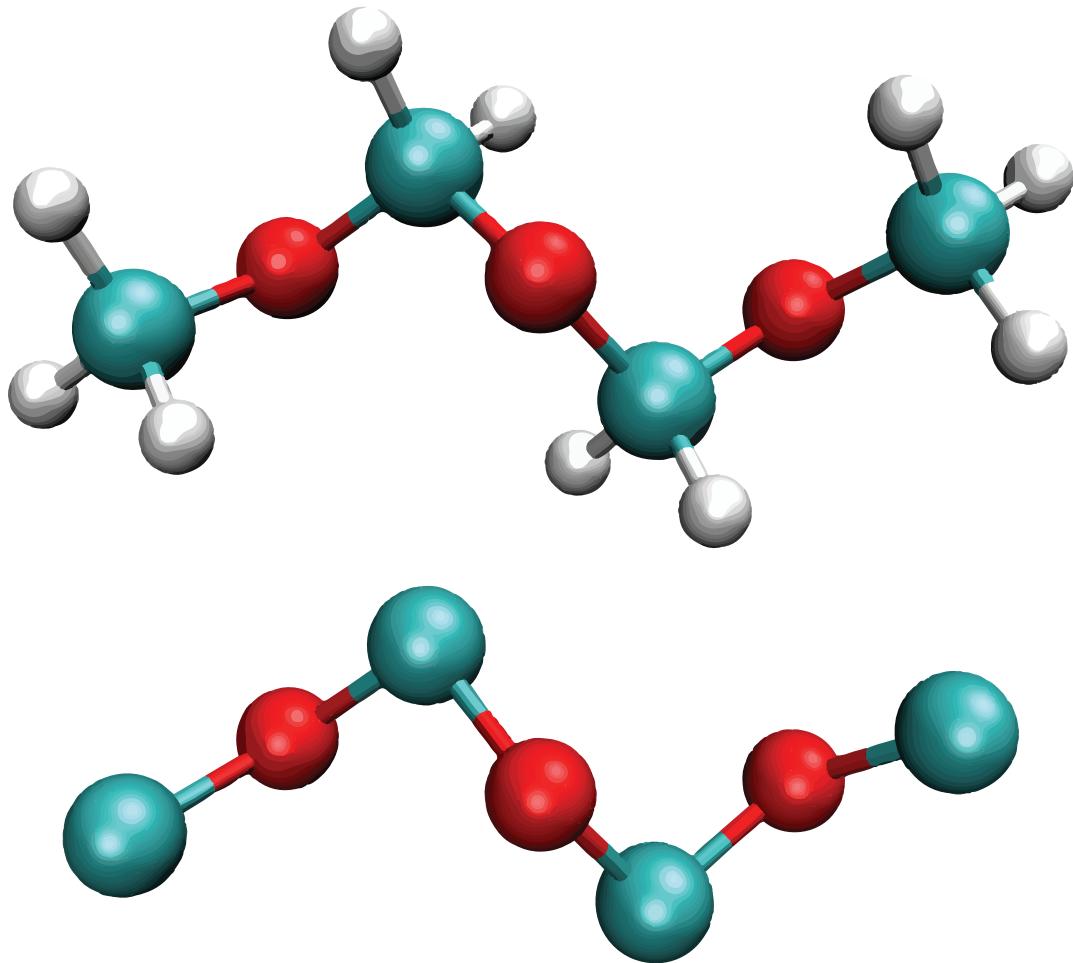


Figure S6: Equilibrium geometry of OME2. Top: full atom representation, bottom: united-atom model.

Table S11: Z-Matrix for equilibrium geometry of OME2 using united-atom model.

Group	ID	Bonds / Å	ID	Angles / °	ID	Dihedrals / °
CH ₃						
O	1	1.502				
CH ₂	2	1.454	1	115.008		
O	3	1.469	2	107.424	1	69.100
CH ₂	4	1.469	3	116.161	2	71.000
O	5	1.453	4	107.375	3	71.700
CH ₃	6	1.502	5	114.999	4	70.900

Table S12: Point Charges for OME2.

Atom	Charge	Group	Total Charge
C	-0.0110		
H	0.0632	CH ₃	0.2097
H	0.0669		
H	0.0906		
O	-0.4134	O	-0.4134
C	0.2150		
H	0.0907	CH ₂	0.4040
H	0.0983		
O	-0.4083	O	-0.4083
C	0.2338		
H	0.0956	CH ₂	0.4140
H	0.0846		
O	-0.4090	O	-0.4090
C	-0.0395		
H	0.0762	CH ₃	0.2029
H	0.0997		
H	0.0665		

7.5. OME3

In this section the full atom and center of mass coordinates for OME3 are shown in table S13. The table S14 represents the Z-Matrix for equilibrium geometry of united-atom model, and related partial charges are shown in table S15.

Table S13: Full atom coordinates and center of mass (CM) for equilibrium geometry of OME3.

Atom	$x/\text{\AA}$	$y/\text{\AA}$	$z/\text{\AA}$	Group	$x_{\text{CM}}/\text{\AA}$	$y_{\text{CM}}/\text{\AA}$	$z_{\text{CM}}/\text{\AA}$
C	-3.731	-0.612	0.333	CH ₃	-3.779	-0.642	0.381
H	-3.189	-1.237	1.048				
H	-4.450	-1.223	-0.211				
H	-4.263	0.178	0.876				
O	-2.850	-0.049	-0.637	O	-2.850	-0.049	-0.637
C	-1.894	0.821	-0.098	CH ₂	-1.893	0.909	-0.112
H	-1.413	1.306	-0.948				
H	-2.361	1.560	0.560				
O	-0.933	0.164	0.714	O	-0.933	0.164	-0.714
C	0.000	-0.618	-0.005	CH ₂	0.000	-0.707	-0.005
H	-0.501	-1.232	-0.753				
H	0.500	-1.241	0.738				
O	0.934	0.169	-0.715	O	0.934	0.169	-0.715
C	1.894	0.821	0.102	CH ₂	1.893	0.909	0.117
H	2.363	1.563	-0.551				
H	1.412	1.301	0.954				
O	2.848	-0.052	0.639	O	2.848	-0.052	0.639
C	3.731	-0.611	-0.333	CH ₃	3.779	-0.641	-0.381
H	3.190	-1.233	-1.050				
H	4.449	-1.223	0.210				
H	4.263	0.181	-0.872				

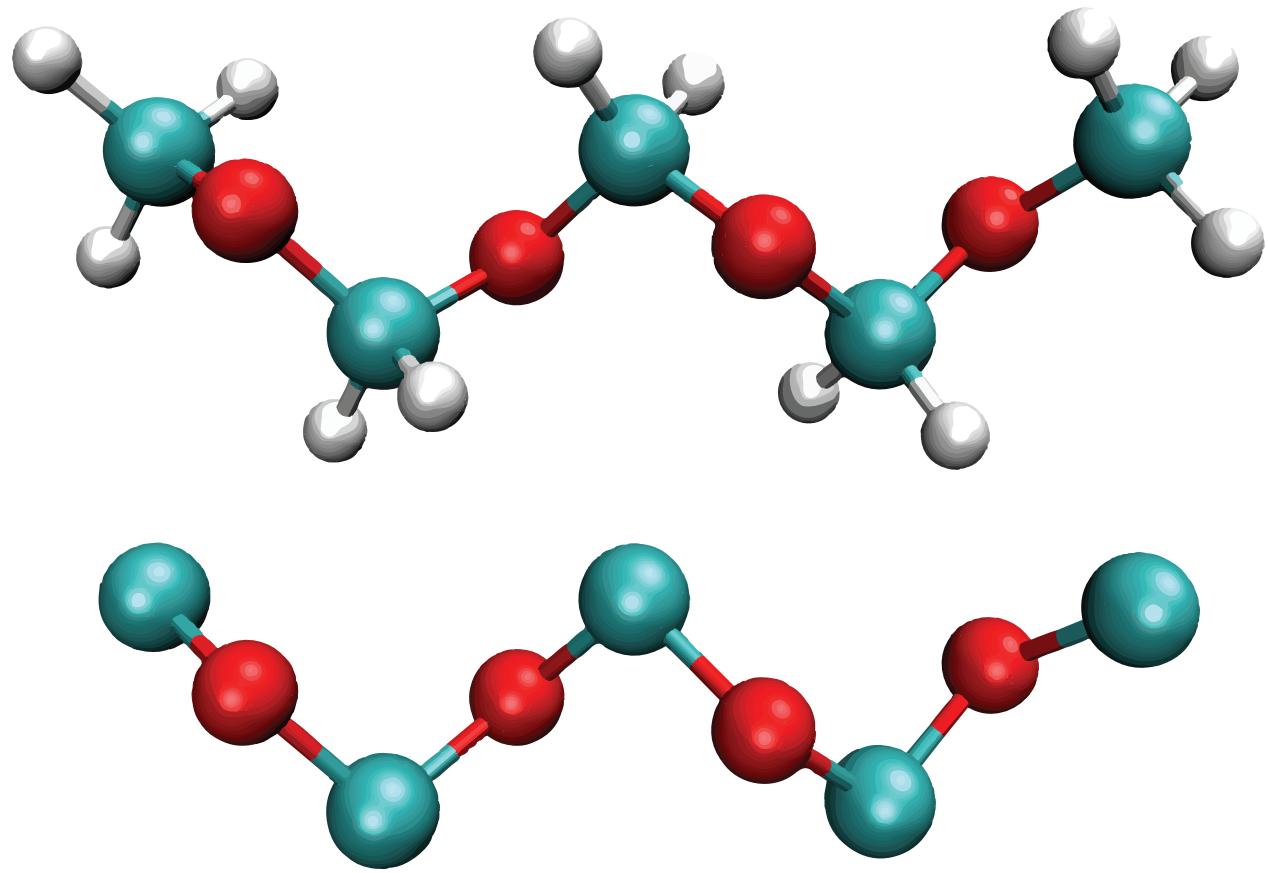


Figure S7: Equilibrium geometry of OME3. Top: full atom representation, bottom: united-atom model.

Table S14: Z-Matrix for equilibrium geometry of OME3 using united-atom model.

Group	ID	Bonds / Å	ID	Angles / °	ID	Dihedrals / °
CH ₃						
O	1	1.500				
CH ₂	2	1.452	1	115.014		
O	3	1.469	2	107.436	1	69.500
CH ₂	4	1.465	3	116.231	2	70.700
O	5	1.465	4	106.765	3	70.500
CH ₂	6	1.469	5	116.278	4	70.300
O	7	1.452	6	107.456	5	70.600
CH ₃	8	1.500	7	115.025	6	69.400

Table S15: Point Charges for OME3.

Atom	Charge	Group	Total Charge
C	0.0054		
H	0.0537	CH ₃	0.1941
H	0.0802		
H	0.0548		
O	-0.3720	O	-0.3720
C	0.1694		
H	0.0763	CH ₂	0.3476
H	0.1019		
O	-0.4398	O	-0.4398
C	0.4858		
H	0.0267	CH ₂	0.5368
H	0.0243		
O	-0.4472	O	-0.4472
C	0.2219		
H	0.0866	CH ₂	0.3713
H	0.0628		
O	-0.3781	O	-0.3781
C	-0.0252		
H	0.0586	CH ₃	0.1873
H	0.0908		
H	0.0631		

7.6. OME4

In this section the full atom and center of mass coordinates for OME4 are shown in table S16. The table S17 represents the Z-Matrix for equilibrium geometry of united-atom model, and related partial charges are shown in table S18.

Table S16: Full atom coordinates and center of mass (CM) for equilibrium geometry of OME4.

Atom	$x/\text{\AA}$	$y/\text{\AA}$	$z/\text{\AA}$	Group	$x_{\text{CM}}/\text{\AA}$	$y_{\text{CM}}/\text{\AA}$	$z_{\text{CM}}/\text{\AA}$
C	4.736	-0.430	-0.285	CH ₃	4.778	-0.452	-0.333
H	5.464	-1.044	0.242				
H	5.241	0.440	-0.722				
H	4.276	-1.017	-1.084				
O	3.768	-0.017	0.680	O	3.768	-0.017	0.680
C	2.796	0.854	0.174	CH ₂	2.790	0.941	0.195
H	2.256	1.244	1.037				
H	3.253	1.667	-0.398				
O	1.900	0.227	-0.734	O	1.900	0.227	-0.734
C	0.987	-0.666	-0.127	CH ₂	0.992	-0.754	-0.136
H	0.538	-1.231	-0.944				
H	1.496	-1.331	0.571				
O	-0.001	0.003	0.633	O	-0.001	-0.003	0.633
C	-0.989	0.665	-0.134	CH ₂	-0.993	0.753	-0.144
H	-1.498	1.336	0.558				
H	-0.538	1.223	-0.956				
O	-1.901	-0.233	-0.735	O	-1.901	-0.233	-0.735
C	-2.796	-0.855	0.177	CH ₂	-2.790	-0.941	0.198
H	-3.256	-1.670	-0.390				
H	-2.255	-1.241	1.041				
O	-3.766	0.020	0.681	O	-3.766	0.020	0.681
C	-4.734	0.431	-0.286	CH ₃	-4.786	0.453	-0.333
H	-4.273	1.009	-1.090				
H	-5.458	1.053	0.239				
H	-5.245	-0.439	-0.714				

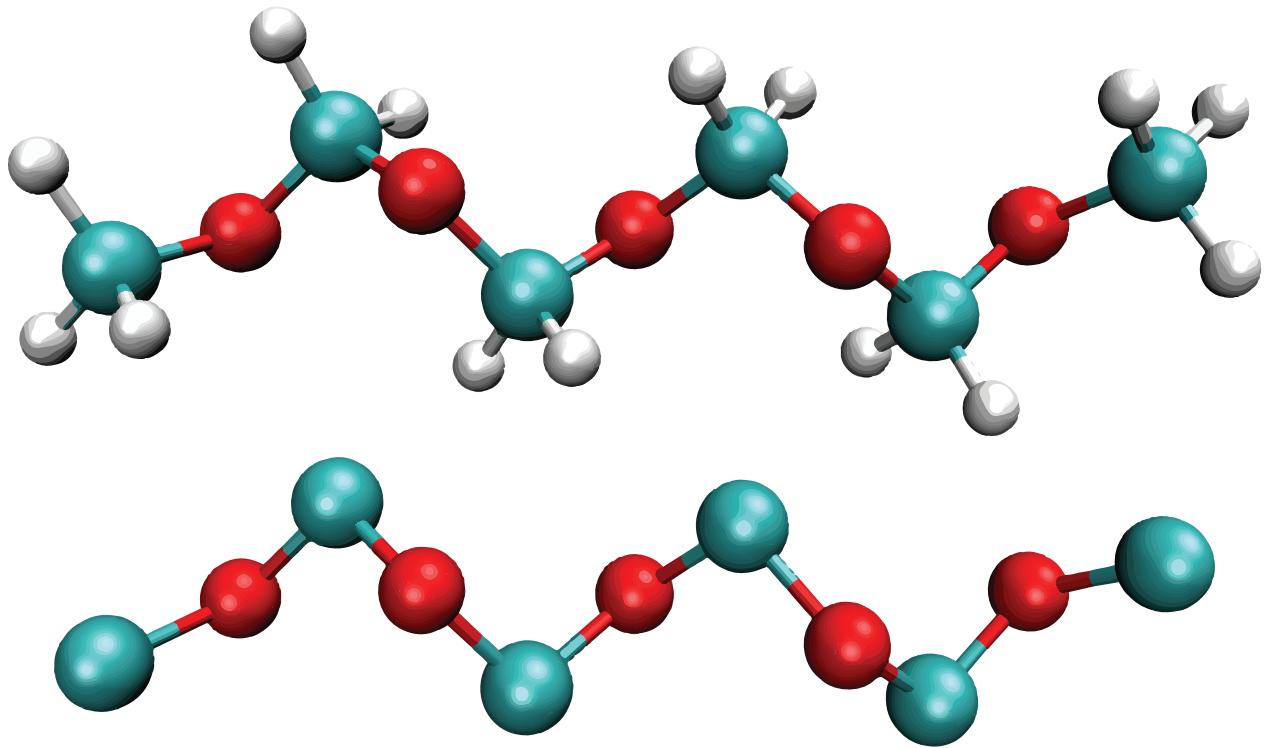


Figure S8: Equilibrium geometry of OME4. Top: full atom representation, bottom: united-atom model.

Table S17: Z-Matrix for equilibrium geometry of OME4 using united-atom model.

Group	ID	Bonds / Å	ID	Angles / °	ID	Dihedrals / °
CH ₃						
O	1	1.502				
CH ₂	2	1.452	1	115.003		
O	3	1.471	2	107.395	1	71.000
CH ₂	4	1.465	3	116.279	2	70.000
O	5	1.466	4	106.722	3	69.600
CH ₂	6	1.466	5	116.415	4	71.800
O	7	1.465	6	106.752	5	71.800
CH ₂	8	1.471	7	116.309	6	69.500
O	9	1.452	8	107.390	7	69.800
CH ₃	10	1.502	9	114.998	8	70.900

Table S18: Point Charges for OME4.

Atom	Charge	Group	Total Charge
C	-0.0867		
H	0.1037	CH ₃	0.1839
H	0.0852		
H	0.0817		
O	-0.3428	O	-0.3428
C	0.1024		
H	0.0922	CH ₂	0.3186
H	0.1240		
O	-0.4268	O	-0.4268
C	0.4672		
H	0.0112	CH ₂	0.5197
H	0.0413		
O	-0.4840	O	-0.4840
C	0.3818		
H	0.0675	CH ₂	0.4808
H	0.0315		
O	-0.4171	O	-0.4171
C	0.1243		
H	0.1181	CH ₂	0.3296
H	0.0872		
O	-0.3532	O	-0.3532
C	-0.0513		
H	0.0759	CH ₃	0.1912
H	0.0924		
H	0.0742		

7.7. OME5

In this section the full atom and center of mass coordinates for OME5 are shown in table S19. The table S20 represents the Z-Matrix for equilibrium geometry of united-atom model, and related partial charges are shown in table S21.

Table S19: Full atom coordinates and center of mass (CM) for equilibrium geometry of OME5.

Atom	$x/\text{\AA}$	$y/\text{\AA}$	$z/\text{\AA}$	Group	$x_{\text{CM}}/\text{\AA}$	$y_{\text{CM}}/\text{\AA}$	$z_{\text{CM}}/\text{\AA}$
C	5.650	0.558	0.476	CH ₃	5.697	0.575	0.531
H	5.100	0.974	1.324				
H	6.368	1.292	0.114				
H	6.183	-0.343	0.801				
O	4.777	0.261	-0.613	O	4.777	0.261	-0.613
C	3.822	-0.720	-0.320	CH ₂	3.821	-0.802	-0.356
H	3.351	-0.979	-1.270				
H	4.288	-1.600	0.134				
O	2.850	-0.288	0.621	O	2.850	-0.288	0.621
C	1.897	0.614	0.098	CH ₂	1.898	0.701	0.114
H	1.425	1.085	0.961				
H	2.371	1.359	-0.541				
O	0.934	-0.032	-0.713	O	0.934	-0.032	0.713
C	-0.001	-0.814	0.004	CH ₂	-0.001	-0.903	0.005
H	0.499	-1.428	0.754				
H	-0.500	-1.437	-0.737				
O	-0.935	-0.023	0.712	O	-0.935	-0.023	0.712
C	-1.898	0.613	-0.107	CH ₂	-1.898	0.701	-0.124
H	-2.372	1.366	0.524				
H	-1.427	1.075	-0.975				
O	-2.850	-0.295	-0.618	O	-2.850	-0.295	-0.618
C	-3.821	-0.717	0.327	CH ₂	-3.821	-0.799	0.363
H	-4.289	-1.600	-0.119				
H	-3.350	-0.968	1.279				
O	-4.776	0.267	0.612	O	-4.776	0.267	0.612
C	-5.649	0.554	-0.480	CH ₃	-5.696	-0.570	-0.534
H	-5.100	0.962	-1.332				
H	-6.367	1.290	-0.123				
H	-6.183	-0.350	-0.796				

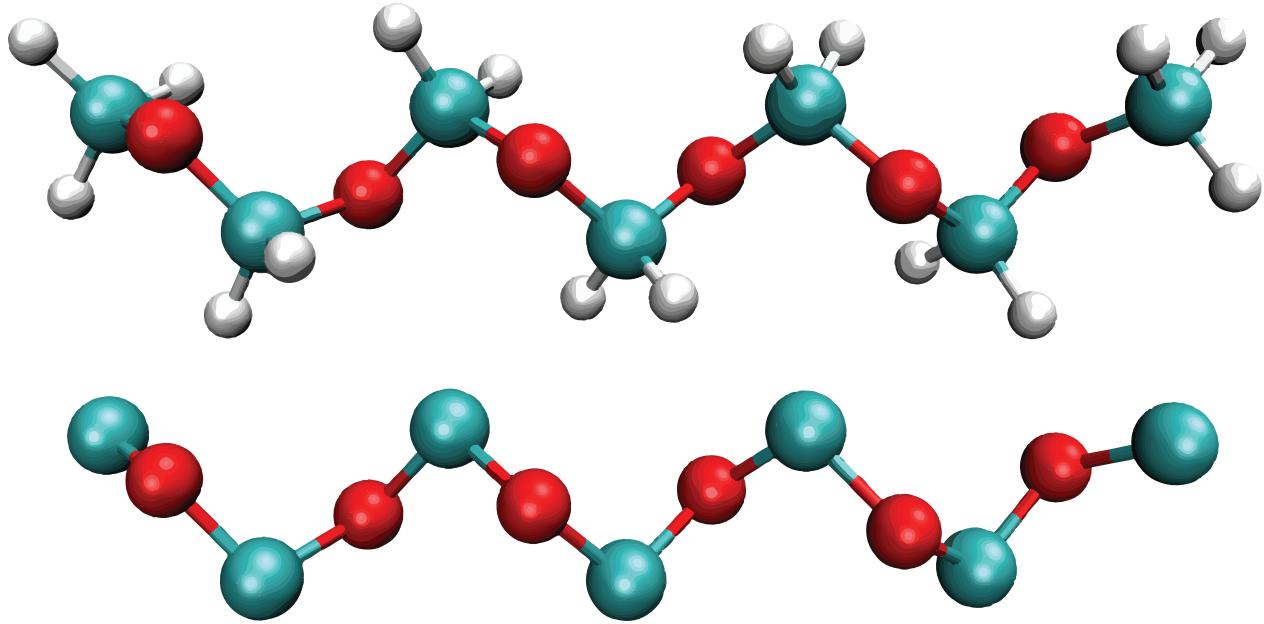


Figure S9: Equilibrium geometry of OME5. Top: full atom representation, bottom: united-atom model.

Table S20: Z-Matrix for equilibrium geometry of OME5 using united-atom model.

Group	ID	Bonds / Å	ID	Angles / °	ID	Dihedrals / °
CH ₃						
O	1	1.501				
CH ₂	2	1.452	1	114.907		
O	3	1.470	2	107.250	1	69.800
CH ₂	4	1.464	3	115.872	2	72.600
O	5	1.467	4	106.545	3	73.900
CH ₂	6	1.466	5	116.069	4	70.900
O	7	1.465	6	106.640	5	71.000
CH ₂	8	1.467	7	116.094	6	71.100
O	9	1.464	8	106.539	7	70.900
CH ₂	10	1.470	9	115.911	8	73.800
O	11	1.452	10	107.261	9	72.400
CH ₃	12	1.501	11	114.884	10	69.800

Table S21: Point Charges for OME5.

Atom	Charge	Group	Total Charge
C	-0.0455		
H	0.0698	CH ₃	0.1855
H	0.0912		
H	0.0700		
O	-0.3301	O	-0.3301
C	0.0376		
H	0.1115	CH ₂	0.2808
H	0.1317		
O	-0.3701	O	-0.3701
C	0.3910		
H	0.0223	CH ₂	0.4677
H	0.0544		
O	-0.4730	O	-0.4730
C	0.4516		
H	0.0169	CH ₂	0.4823
H	0.0138		
O	-0.4780	O	-0.4780
C	0.3761		
H	0.0646	CH ₂	0.4638
H	0.0231		
O	-0.3686	O	-0.3686
C	0.0549		
H	0.1253	CH ₂	0.2908
H	0.1106		
O	-0.3404	O	-0.3404
C	-0.0343		
H	0.0685	CH ₃	0.1893
H	0.0890		
H	0.0661		

8. References

- [1] M. Yiannourakou, P. Ungerer, B. Leblanc, N. Ferrando, J.-M. Teuler, Overview of medea®-gibbs capabilities for thermodynamic property calculation and vle behaviour description of pure compounds and mixtures: application to polar compounds generated from ligno-cellulosic biomass, *Molecular Simulation* 39 (14-15) (2013) 1165–1211.
- [2] J. K. Shah, E. Marin-Rimoldi, R. G. Mullen, B. P. Keene, S. Khan, A. S. Paluch, N. Rai, L. L. Romaniello, T. W. Rosch, B. Yoo, E. J. Maginn, Cassandra: An open source monte carlo package for molecular simulation, *Journal of computational chemistry* 38 (19) (2017) 1727–1739.
- [3] T. Darden, D. York, L. Pedersen, Particle mesh ewald: An nlog(n) method for ewald sums in large systems, *The Journal of Chemical Physics* 98 (12) (1993) 10089–10092. doi:10.1063/1.464397.
- [4] B. Hess, H. Bekker, H. J. C. Berendsen, J. G. E. M. Fraaije, Lincs: A linear constraint solver for molecular simulations, *Journal of Computational Chemistry* 18 (12) (1997) 1463–1472. doi:10.1002/(SICI)1096-987X(199709)18:12<1463::AID-JCC4>3.0.CO;2-H.
- [5] Boyd, R. H., Some physical properties of polyoxymethylene dimethyl ethers, *J. Pol. Chem.* 50 (153) (1961) 133–141. doi:10.1002/pol.1961.1205015316.
- [6] S. A., Klamt; G., Cosmo: a new approach to dielectric screening in solvents with explicit expressions for the screening energy and its gradient, *J. Chem. Soc., Perkin Trans. 2* (1993) 799–805doi:10.1039/P29930000799.
- [7] Bayly, C. I.; Cieplak, P.; Cornell, W.; Kollman, P. A., A well-behaved electrostatic potential based method using charge restraints for deriving atomic charges: the RESP model, *J. Phys. Chem.* 97 (40) (1993) 10269–10280. doi:<https://doi.org/10.1021/j100142a004>.
- [8] S. Dasgupta, K.A. Smith, W.A. Goddard, Polyoxymethylene: the Hessian biased force field for molecular dynamics simulations., *J. Phys. Chem.* 97 (1993) 10891–10902. doi:doi:10.1021/j100144a001.