

# Catalytic Transfer Hydrogenation of Levulinic Acid to $\gamma$ -Valerolactone over $\text{Ni}_3\text{P}$ - $\text{CePO}_4$ Catalysts

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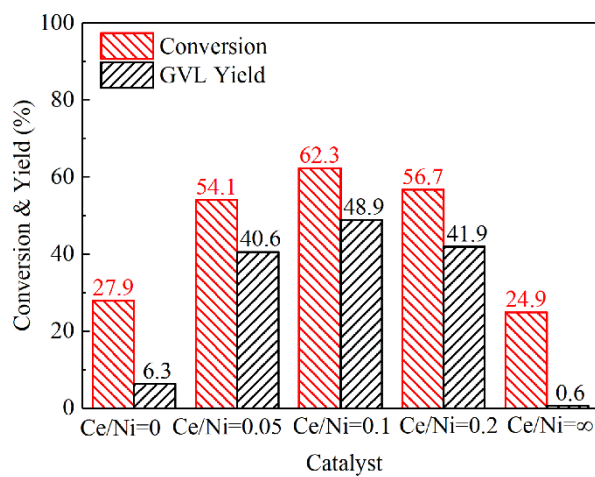
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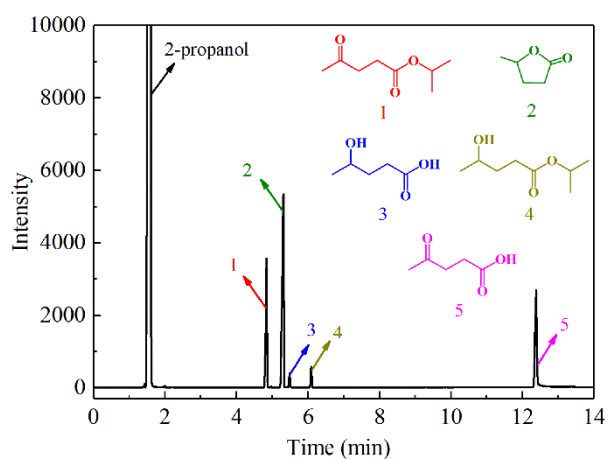
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## Results and Discussion



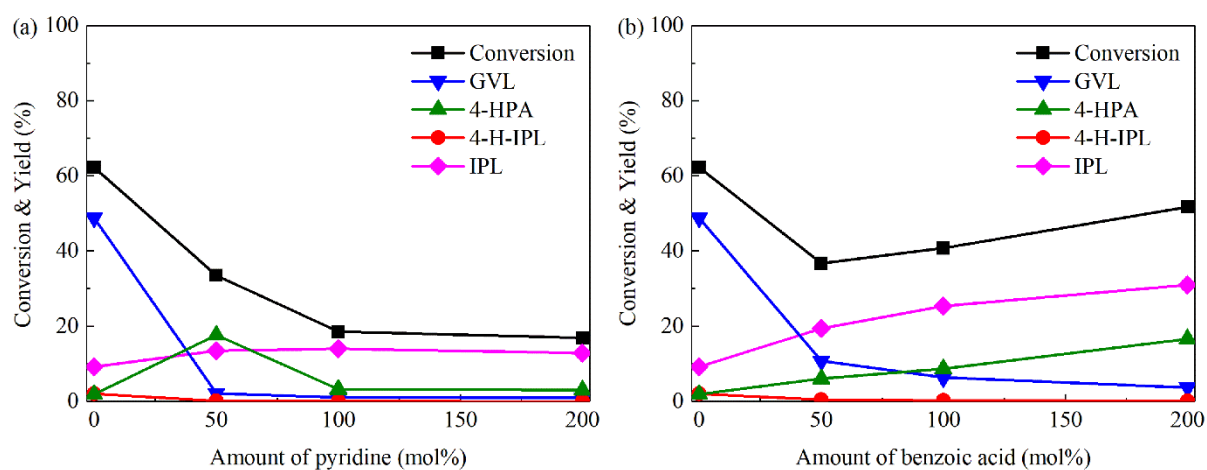
**Figure S1.** Catalytic performance of the  $\text{Ni}_3\text{P-CePO}_4(x)$  catalyst in the CTH of LA to GVL. Reaction conditions: LA 20 mmol,  $\text{Ni}_3\text{P-CePO}_4(x)$  0.2 g, 2-propanol 38 mL, 180 °C, 60 min, 750 rpm.



**Figure S2.** Representative gas chromatogram of the liquid product from CTH of LA.

**Table S1.** Kinetic parameters for the CTH of LA to GVL over Ni<sub>3</sub>P-CePO<sub>4</sub>(0.1) catalyst.

Temperature (°C)	Rate constant (mol·L <sup>-1</sup> ·min <sup>-1</sup> ) <sup>a</sup>	Apparent activation energy (kJ·mol <sup>-1</sup> ) <sup>b</sup>
160	$1.8 \times 10^{-3}$	73.9
170	$2.6 \times 10^{-3}$	
180	$4.6 \times 10^{-3}$	
190	$6.9 \times 10^{-3}$	
200	$9.8 \times 10^{-3}$	

<sup>a</sup> Assuming a pseudo zero order to the concentration of LA.<sup>b</sup> According to Arrhenius equation.**Figure S3.** Variation of LA conversion and product yield with amount of (a) pyridine or (b) benzoic acid in CTH of LA over Ni<sub>3</sub>P-CePO<sub>4</sub>(0.1) catalyst.