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SUPPORTING INFORMATION

2 Arsenic Toxicity: Carbonate's Counteraction Revealed

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Figure S1. Control experiment for live dead staining assay - CaCO₃ treatment. (a) and (b) Fluorescence microscopy image of IEC-6 cells treated with 40 ppm of calcium carbonate at two different magnifications. 40 ppm of CaCO₃ used for the studies was found to be non-toxic to the cells. (c) and (d) Phase contrast microscopy images showing precipitates of CaCO₃ on the surface of the cells.



Figure S2. Control experiment for live dead staining assay - Na₂CO₃ treatment. Fluorescence microscopy images (a and b) and phase contrast microscopy image (c and d) of IEC-6 cells with 40 ppm of sodium carbonate. 40 ppm of sodium carbonate used for the studies was found to be non-toxic to the cells.



Figure S3. Characterization and chemical composition of the precipitates formed on cells during
CaCO₃ treatment. EDS spectrum (a) and the SEM image (b) of the precipitate along with (c-e)
elemental maps are shown. No complexes of arsenic were formed as *As* is absent. Scale bars in
c-e are the same as in b.



Figure S4. Phase-contrast images of control experiment for morphological study - 40 ppm of calcium carbonate and sodium carbonate. Treatment of IEC-6 cells with (a) 40 ppm of calcium carbonate and (b) 40 ppm of sodium carbonate (control experiments). Precipitates of CaCO₃ were found on the surface of the cells. 40 ppm of CaCO₃ and Na₂CO₃ used for the studies did not alter the cellular morphology.



Figure S5. Effect of Na₂CO₃ on the toxicity induced by varying concentrations (1 ppm, 3 ppm and 5 ppm) of arsenate (As^{5+}). The percentage of cell viability was measured using the MTT assay. The difference between the heights of dashed lines represent the percentage of increase in viability of the cells.



49 Figure S6. A comparative analysis of As^{5+} concentration in cells in presence of 40 ppm of

⁵⁰ CaCO₃ and Na₂CO₃.



Figure S7. Control experiment for ROS staining. IEC – 6 cells treated with 1 mM hydrogen
peroxide for 1 h induced ROS production and showed fluorescence on staining. (a) Phase
contrast microscopy image. (b) Fluorescence microscopy image.



Figure S8. pH profiling of treated cells using bromothymol blue (a) Bright-field optical microscopy images of cells stained with BTB; (i) Control (ii) 3 ppm As^{5+} (iii) 3 ppm $As^{5+} + 40$ ppm CaCO₃ (iv) 3 ppm $As^{5+} + 40$ ppm Na₂CO₃. (b) Histogram of the pH distribution for treated cells.



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Figure S9. CV curve of PBS (pH=7) used in the cyclic voltammetry experiment demonstrating

 H_2O_2 decomposition by CaCO₃.