

Supporting Information

A POROUS AROMATIC FRAMEWORK FUNCTIONALIZED WITH LUMINESCENT IRIDIUM(III) ORGANOMETALLIC COMPLEXES FOR TURN-ON SENSING OF $^{99}\text{TcO}_4^-$

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Figure S1. ^1H -NMR spectrum of $\text{Ir}(\text{ppy})_2(2,2'\text{-Br}_2\text{-5,5'-bpy})^+(\text{Cl})^-$ in CDCl_3 .

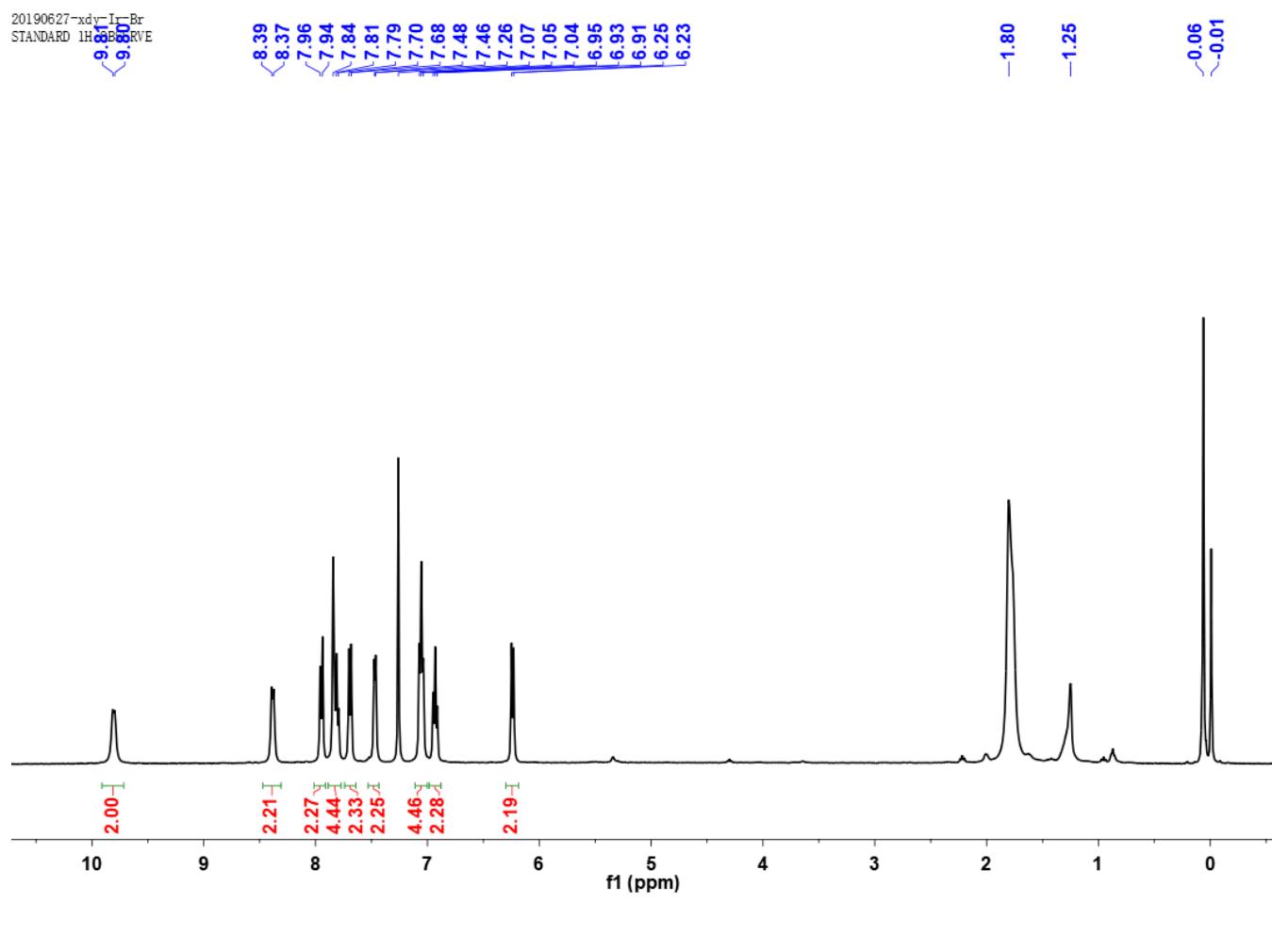


Figure S2. TGA curve of Ir-PAF.

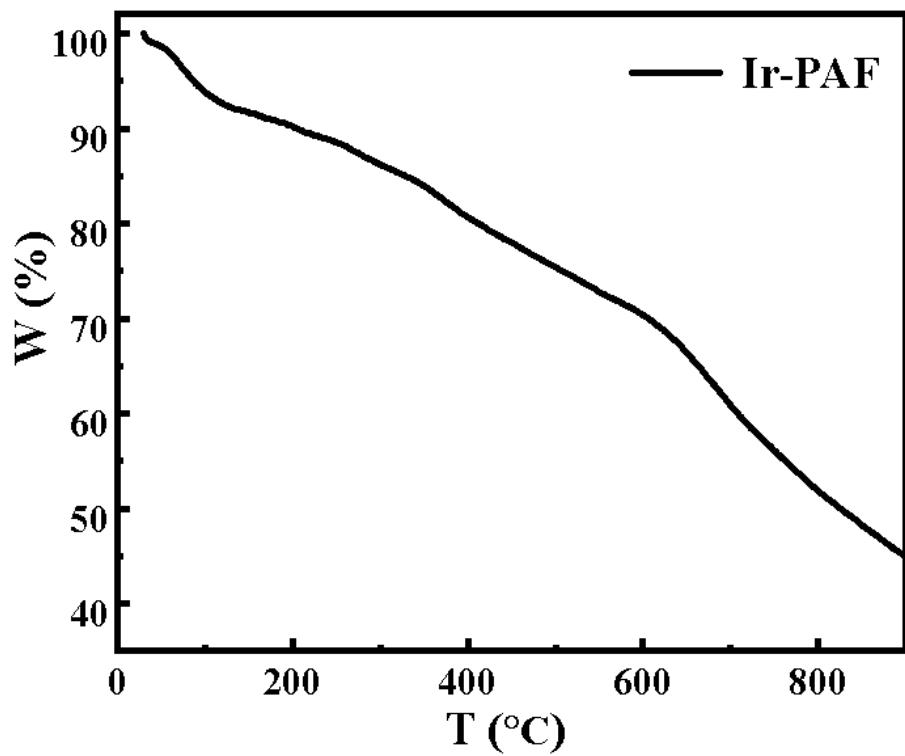


Figure S3. UV absorption spectrum of Ir-PAF.

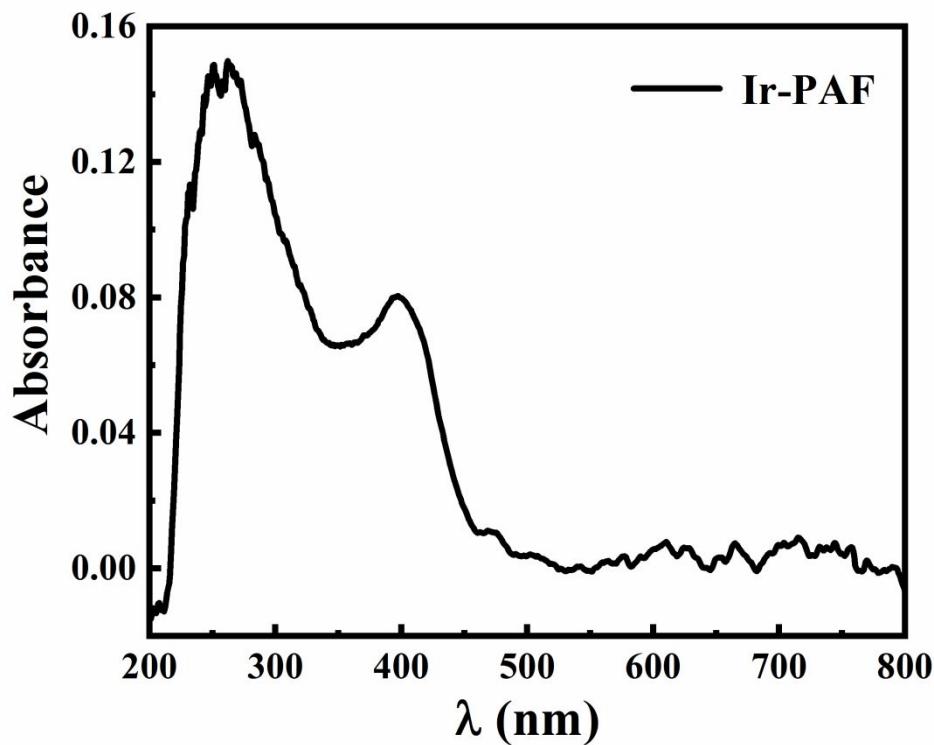


Figure S4. EDS of $\text{ReO}_4^-@\text{Ir-PAF}$.

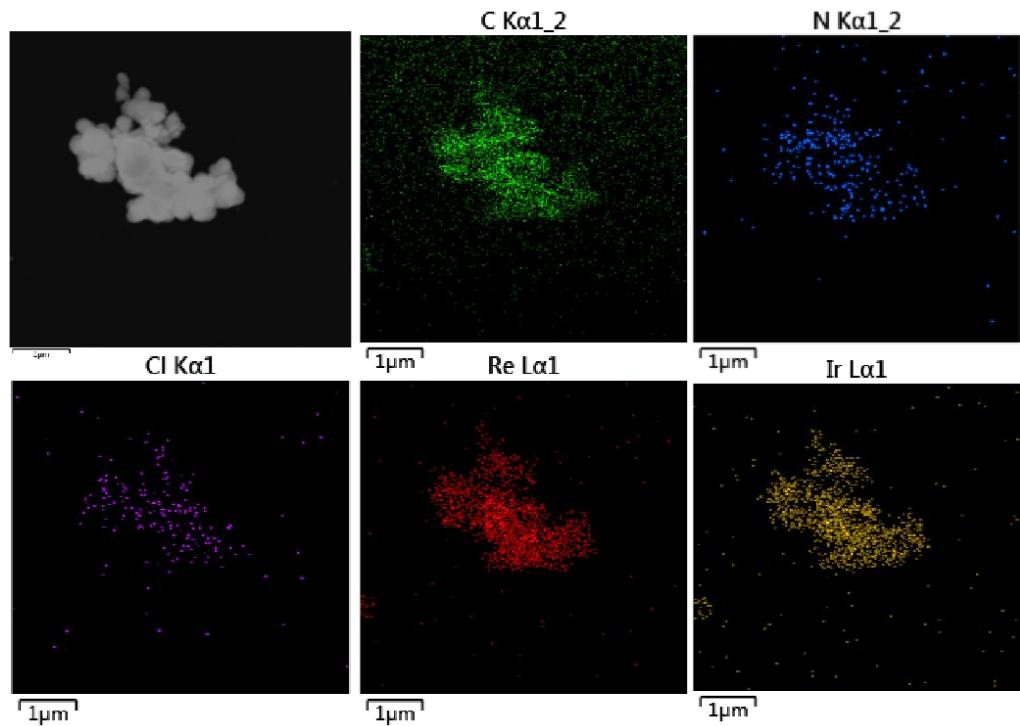


Figure S5. The emission intensity of Ir-PAF before and after adsorbing ReO_4^- in different solid-to-liquid ratios.

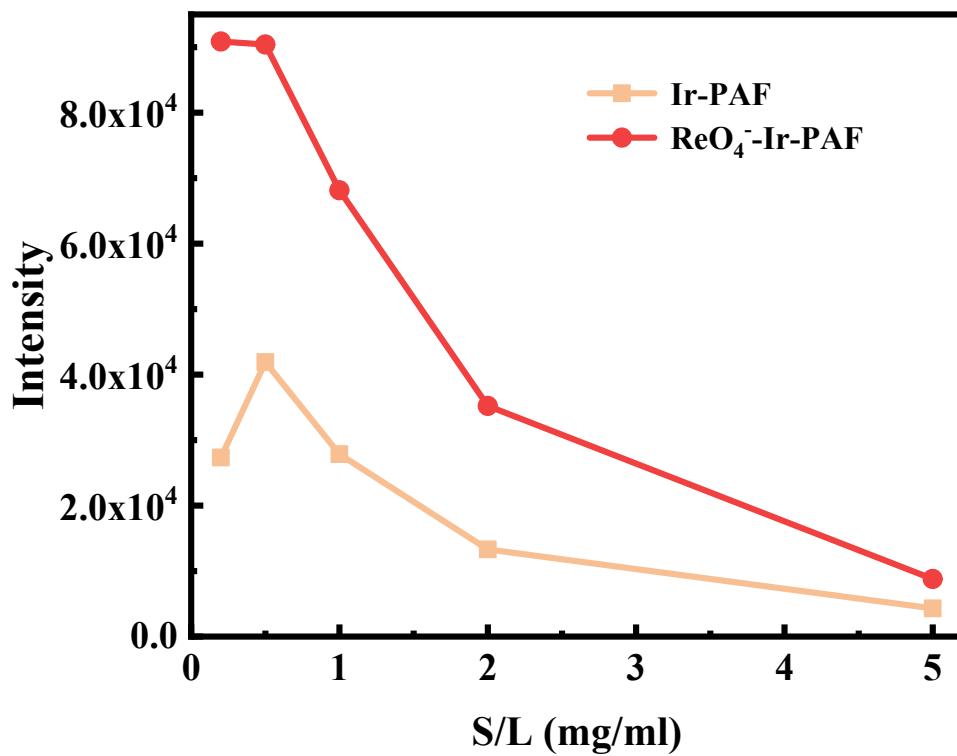


Figure S6. PXRD of Ir-PAF after irradiated by various doses of β ray.

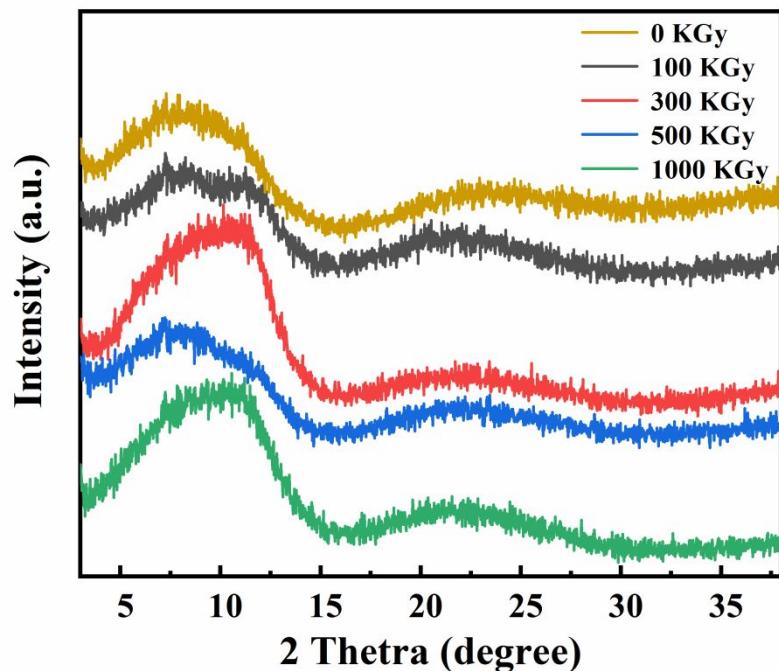


Figure S7. Fluorescence change ratio before and after perrhenate uptake with irradiated samples treated with various doses.

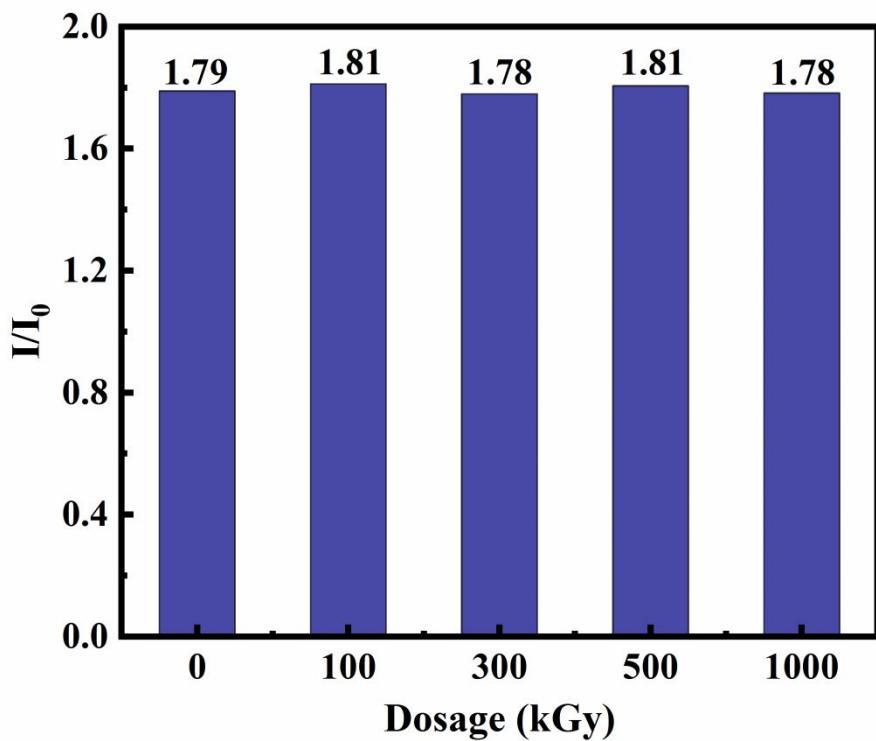


Figure S8. Solid fluorescence spectra of Ir-PAF with three adsorption/desorption cycles.

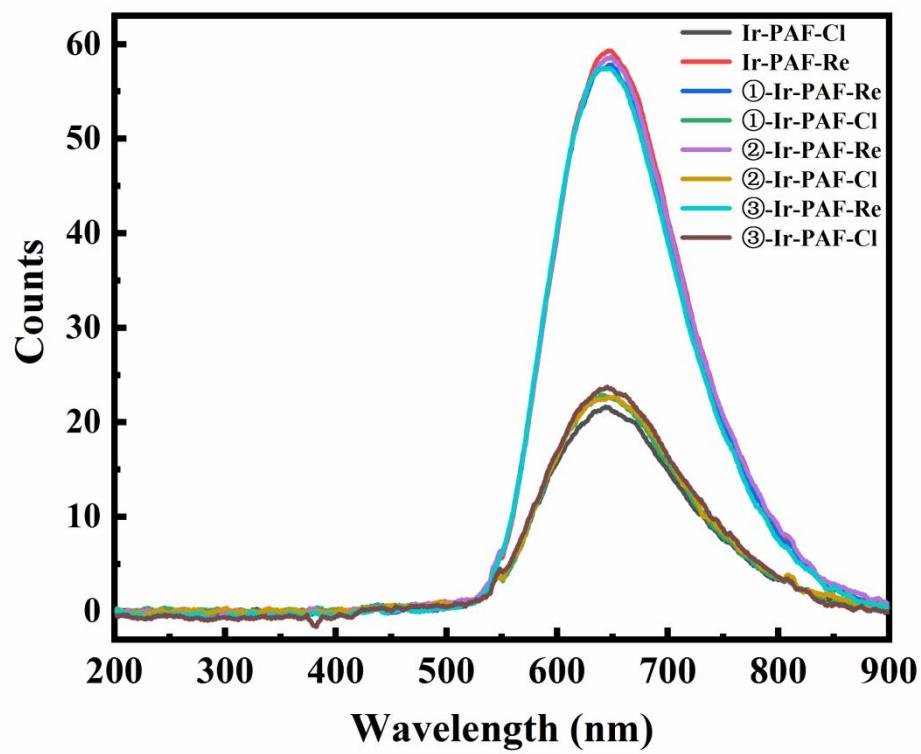
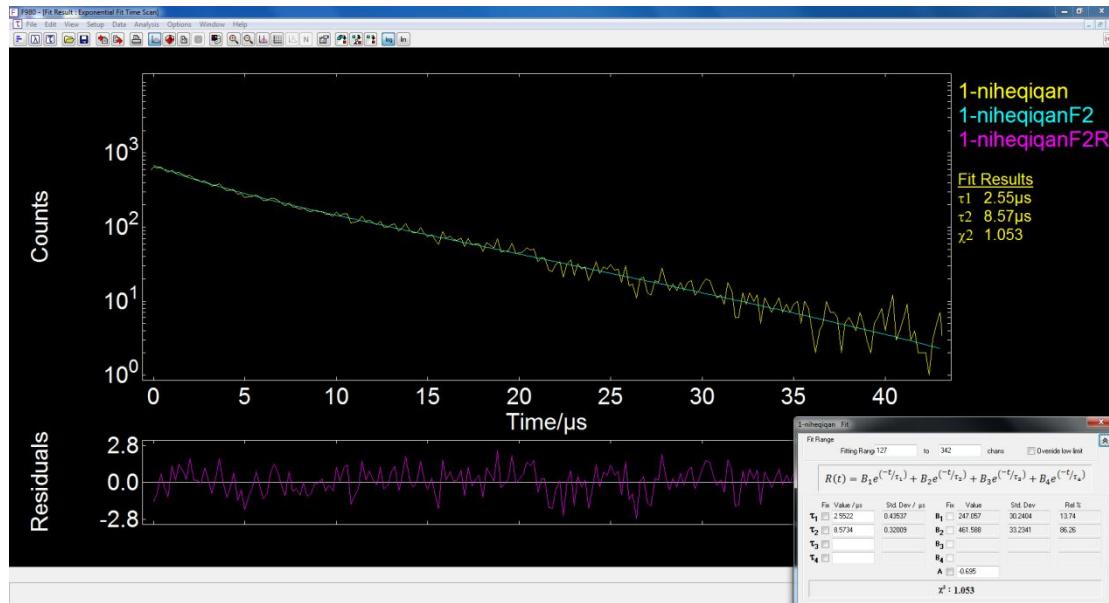
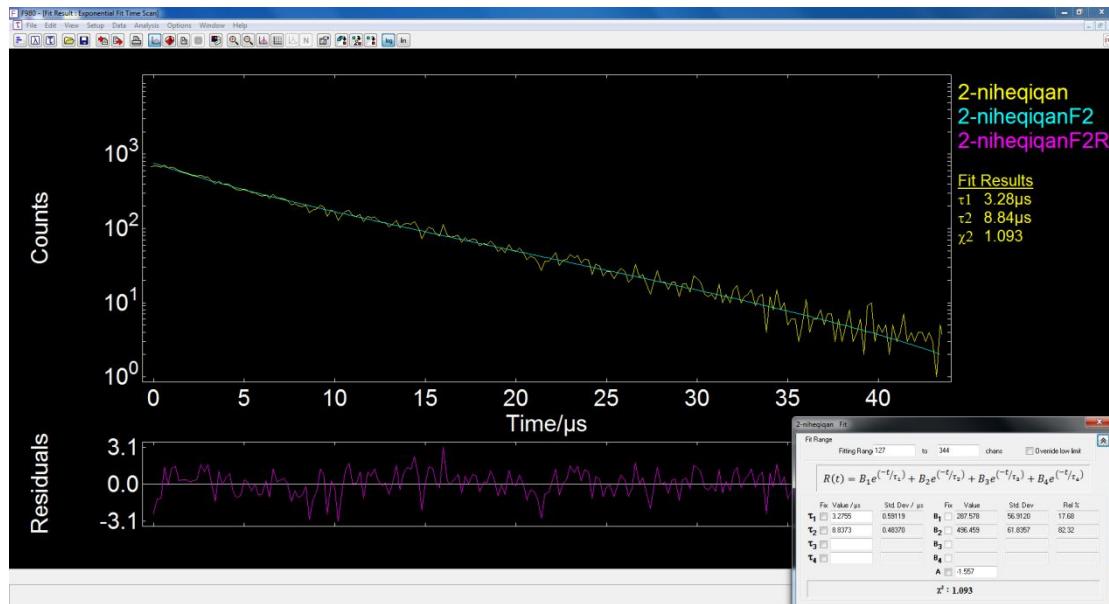


Figure S9. Fluorescence lifetime for a) Cl⁻@Ir-PAF, b) NO₃⁻@Ir-PAF and c) ReO₄⁻@Ir-PAF.

a) Cl⁻@Ir-PAF



b) NO₃⁻@Ir-PAF



c) ReO_4^- @Ir-PAF

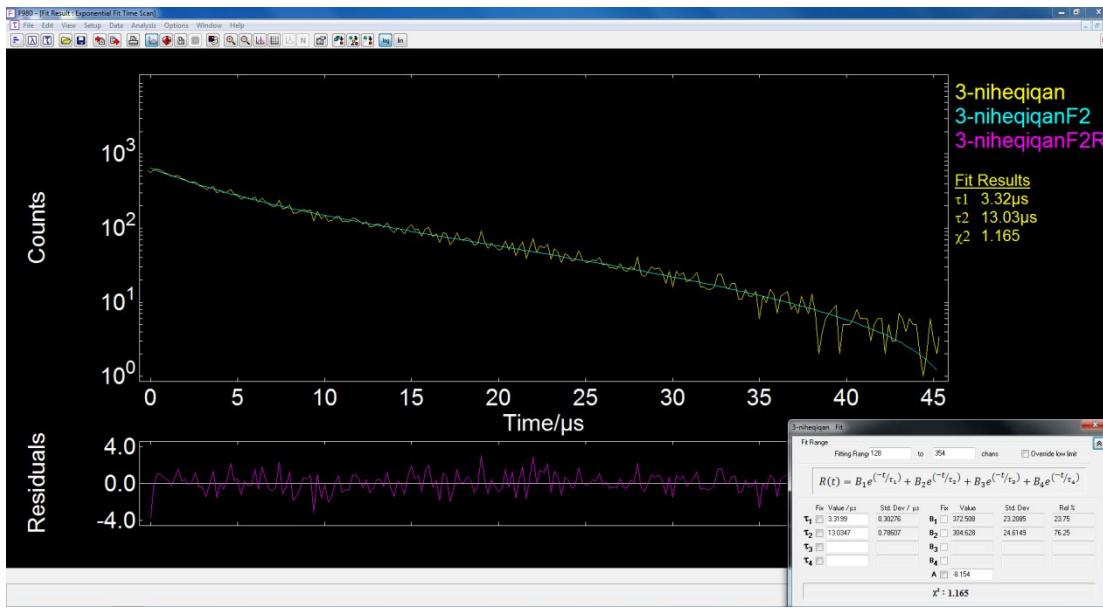
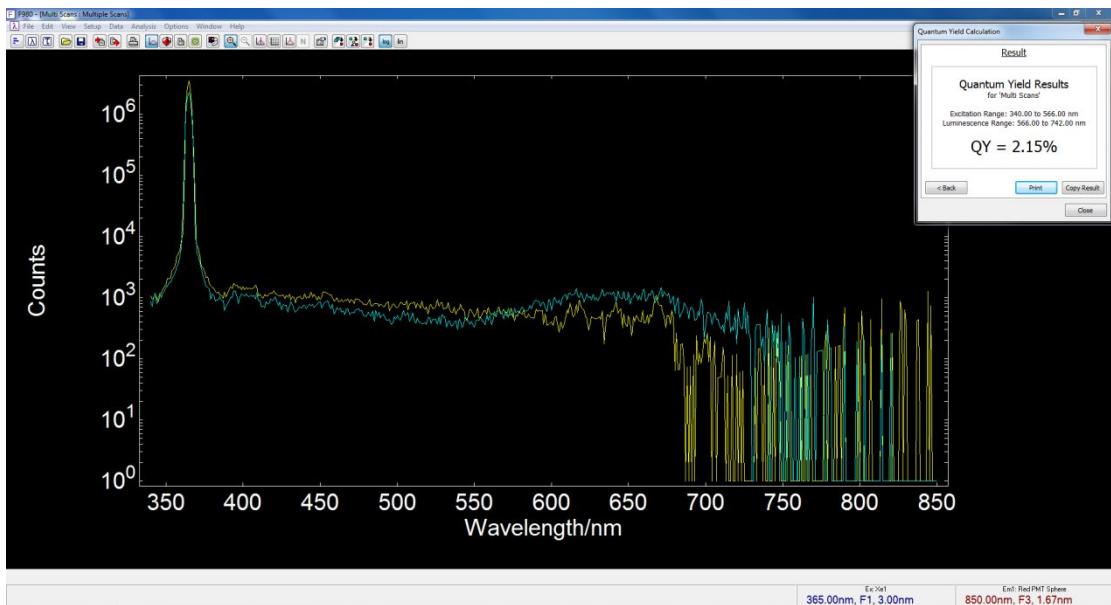
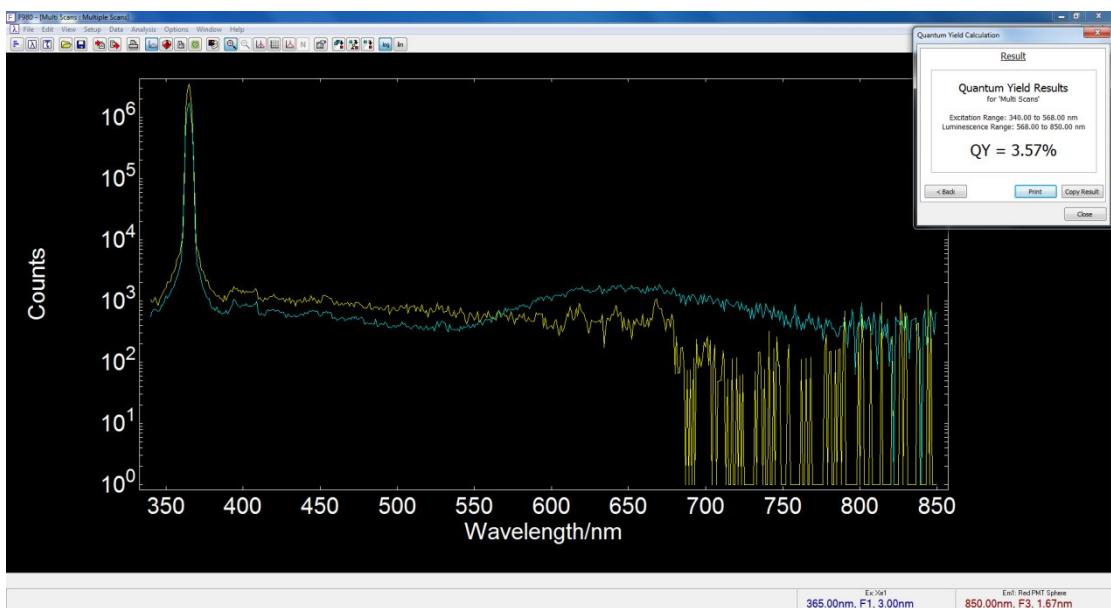


Figure S10. Fluorescence quantum yield for a) Cl⁻@Ir-PAF, b) NO₃⁻@Ir-PAF and c) ReO₄⁻@Ir-PAF.

a) Cl⁻@Ir-PAF



b) NO₃⁻@Ir-PAF



C) $\text{ReO}_4^-@\text{Ir-PAF}$

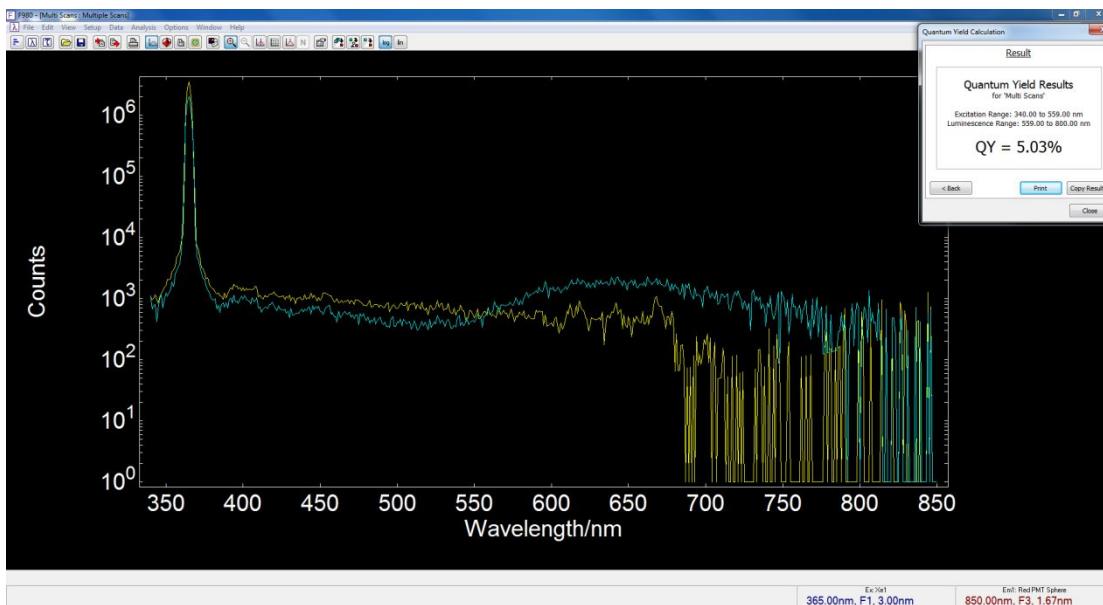


Table S1. Detection limits for different chemical sensors.

Sensors	Detection method	Target preparation	Detection limit
MOR-1	Fluorescent	Re solution	0.36 ppm ^[1]
MOR-2	Fluorescent	Re solution	0.15 ppm ^[1]
Cage 1H₆⁶⁺	Fluorescent	Tc/Re solution (2 < pH < 4)	--- ^[2]
Auramine O	Fluorescent	Re solution	270 μM ^[3]
Thioflavin-T	Fluorescent	Re solution	260 μM ^[4]
1·SbF₆	Fluorescent	Tc solution	2.6 × 10 ⁻¹⁰ M ^[5]
Ir-PAF	Fluorescent	Tc/Re solution	2.99 μM ^{This work}

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