Supporting Information

Biodegradable Bacterial Cellulose-Supported Quasi-Solid Electrolyte for Lithium Batteries

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Figure S1. The N₂ adsorption and desorption curve and the Brunauer-Emmett-Teller test of the BC



Figure S2. DSC curves of electrolytes with different ratios



Figure S3. FT-IR spectra of the ILE $\space{-}$ BC and BC-ILE-2

	Ionic conductivity (mS cm ⁻¹)								
T(°C)	BC-IL-1.5	BC-IL-1.75	BC-IL-2	BC-IL-2.25	BC-IL-2.5				
0°C	0.00640198	0.01044672	0.004034747	0.0029917	0.007492				
10°C	0.01510682	0.06450494	0.054159601	0.0552765	0.091275				
20°C	0.03103435	0.1235111	0.112249363	0.1031988	0.17145				
25°C	0.04129147	0.16493383	0.137619033	0.1409692	0.238112				
30°C	0.05416995	0.21125441	0.176038312	0.1765631	0.299935				
40°C	0.07779372	0.30243165	0.269373309	0.2735112	0.483552				
50°C	0.12984538	0.47660534	0.399896108	0.3984073	0.694413				
60°C	0.1995618	0.72332731	0.597037583	0.590994	1.047607				
70°C	0.25423159	0.90992478	0.768151838	0.7301964	1.304314				
80°C	0.33913302	1.20431221	1.01412888	0.934425	1.787727				

Table S1. Conductivity values of electrolytes with different ratios at different temperatures



Figure S4. The relationship between conductivity and temperature of different ratios of BC-ILE.



Figure S5. Time evolution of the ac impedance plots for the Li/BC-ILE-2/Li battery and the schematic diagram of the partially amplified impedance changes.



Figure S6. Time evolution of the ac impedance plots for the Li/ILE/Li battery

Table S2. The calculation results of density fu	unctional theory (DFT))
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Structure	Li ⁺	TFSI [_]	LiTFSI	BC	BC-Li ⁺	BC-TFSI⁻	BC-LiTFSI
Total							
D	_	_	—	_	—	_	—
Energy	7 1220081	1822 1420026	1822 1420026	681 5800633	680 0350633	2502 7207802	2511 1810053
(Ha)	/.4329904	1822.1430020	1822.1430020	081.3899033	089.0330033	2303.1391893	2311.1019933



Figure S7. Schematic diagram of degradation experiment results