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3	High-Throughput Screening for Engineered
4	Nanoparticles That Enhance Photosynthesis Using
5	Mesophyll Protoplasts
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32 **Chloroplast Isolation Assay.** The chloroplasts were isolated from hydroponic cultivated 3-week-old spinach leaves by a mechanical method according to Weise et al.¹ First, 4-5 33 pieces of spinach leaves were cut into 5 mm strips along the main vein, followed by 34 35 transfer to a pre-ice-cold mortar. The buffer used in the bowl consisted of 20 mL of icecold 30 mM HEPES buffer (pH 7.6) with 350 mM mannitol, 3 mM EDTA, and 0.05 g of 36 37 acetone-washed fraction-V BSA. When grinding, the grinding pestle should not be direct 38 contact with the wall of the grinding bowl. After grinding for two minutes to fully release chloroplasts, the solution should then be filtered through four layers of cheesecloth. The 39 suspension was then centrifuged using refrigerated centrifuge for 2 min at 100 g. After 40 removal of the supernatant, the tube (containing chloroplasts) was placed on ice. The 41 isolated chloroplasts were re-suspended in solution consisting of 30 mM Hepes buffer 42 (pH 7.6) with 350 mM mannitol, 2.5 mM MgCl₂, 5 mM KH₂PO₄ and 2.5 mg per ml BSA. 43 The isolated chloroplasts were then counted by Hemocytometer. 44

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Light-Absorption and Photoelectrochemical Measurements of NPs. NM lightadsorption was investigated by monitoring the absorbance spectra with a UV-vis spectrophotometer (UV-3010, Shimazu, Japan). Ten mg of the sample was dissolved in 100 mL of absolute ethanol and was subjected to ultrasonic oscillation for 30 minutes. Afterwards, 3 mL mixed suspension was transferred to a cuvette for measurement of the absorbance spectra by UV-visible spectrophotometry.

The photoelectrochemical performance of NPs was tested using a standard three-52 electrode system with Pt filament as a counter electrode, Ag/AgCl as a reference 53 electrode, and KCl (0.1 mol/L) as the electrolyte solution. The photoelectrochemical 54 properties of the NPs was tested by the bulk electrolysis with coulometry (BE) under 55 visible light irradiation. Ten mg of sample powder was mixed with 30 μ L of water and 56 57 30µL of PEDOT/PSS conductive ink so that a well dispersed solution was obtained. The solution was uniformly painted onto ITO conductive glass substrate ($\leq 7 \Omega/cm^2$) having 58 an area of 0.49 cm² with a glass rod. After the electrode was nearly dry, it was transferred 59 to an oven and dried at 80 °C for 30 min. A Pt filament was selected as the counter 60 electrode, an Ag/AgCl was used as the reference electrode, and KCl (0.1 mol/L) was used 61 as the electrolyte solution. A 150 W xenon lamp (Beijing Newbit Co., Ltd.) was used as 62 63 the light source, and a 420 nm filter was used to obtain visible light with a wavelength greater than 420 nm. The photoelectrode was irradiated from the back side (light intensity 64 was 300 W/m^2). The difference between the photocurrent and the dark current was 65 defined as the net current. 66

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ROS generated by MoS₂. The photodegradation of Rhodamine B was used to investigate if MoS₂ can generate hydroxyl radicals under visible light (λ > 420 nm, 150 W Xe lamp). Five milligrams of MoS₂ was dispersed in the Rhodamine B solution (80 mL, 1.625 mg/mL). Before the photocatalytic reaction, the suspensions were stirred under darkness for 120 min so that the adsorption of Rhodamine B reached saturation. For each test, 0.8 mL of the reaction solution was centrifuged and analyzed by UV-vis spectrophotometry (UV-3010, Shimazu, Japan). To determine the major active species in

75	the photocatalytic process, an active species trapping experiment was conducted under
76	the same conditions as the photodegradation experiment, except that an additional capture
77	agent was added. Ethylene diamine tetraacetic acid (EDTA) was used as the hydrogen ion
78	(h ⁺) trap and ethylene glycol (EG) was used as the hydroxyl radical trap.
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80	Reference
81 82 83	1. Weise, S. E.; Weber, A. P. M.; Sharkey, T. D., Maltose is the major form of carbon exported from the chloroplast at night. <i>Planta</i> 2004 , <i>218</i> , (3), 474-482.
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Figure S1. Light absorption: UV-vis light absorption spectra of (a) AgNPs, (b) Mn_3O_4

103 NPs, (c) FeNPs, (d) $MoS_2 NPs$ and (e) $SiO_2 NPs$.



Figure S2. Light-induced electron: photocurrent profiles of (**a**) AgNPs, (**b**) Mn₃O₄ NPs,

107 (c) FeNPs, (d) $MoS_2 NPs$ and (e) $SiO_2 NPs$.

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Figure S3. ROS generated by MoS₂: (a) the conversion of Rhodamine B in the
adsorption and catalytic processes, (b) the catalytic efficiency of Rhodamine B without
the capture agent, (c) the catalytic efficiency of Rhodamine B with the capture agent of
EDTA and (d) the catalytic efficiency of Rhodamine B with the capture agent of EG.





Figure S4 B. Box plots of GC-MS data showing the relative abundance of significantly changed metabolites in mesophyll protoplasts under dark (0 min) and light illumination (control). The *y*-axis indicates the absolute signal from GC-MS.



Figure S5. Box plots of GC-MS data showing the relative abundance of significantly
changed metabolites in mesophyll protoplasts exposed to MoS₂ (50 mg/L) for 2 hours
under illumination. The *y*-axis indicates the absolute signal from GC-MS.





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162 Figure S6. Phenotypic changes of spinach plants exposed to NPs; A) net photosynthetic
163 rate; B) respiration rate; C) stomatal conductance; D) chlorophyll content of spinach
164 leaves. Three-week-old spinach plants were foliar applied by NPs for 10 days. Data are
165 means ± SD (n=4).





178 Figure S7. Phenotypic images of spinach leaves and roots (A and C); fresh biomass of

- leaf and dry weight of root (**B** and **D**). Data are means \pm SD (n=4).



